

22 **Text S1. Chemical analyses and calculation**

- 23 COD concentration was measured using the standard test kits (range 25–1500 mg/L, Merck). 24 NH₄⁺-N, NO₂ -N, NO₃⁻⁻N, and PO₄^{3–}-P concentrations were analysed using a Flow Injection
- 25 Analyser (Lachat Instrument, Milwaukee, Wisconsin). Analyses of total and dissolved iron,
- 26 TP, and total kjeldahl nitrogen (TKN) concentrations were conducted by using Inductive
- 27 Column Plasma Optical Emission Spectroscopy (ICP-OES, Perkin Elmer Optima 7300DV,
- 28 Waltham, USA). The pH was measured using a portable pH monitor and probe (pH 5+,
- 29 Oakton). Organic micropollutants were measured by liquid chromatography tandem mass
- 30 spectrometry (LC-MS/MS). The dissolved oxygen (DO) concentration was measured using a
- 31 portable DO monitor and probe (Optical DO sensor inPro 6960i, Mettler Toledo). Organic
- 32 nitrogen was calculated as the difference between TKN and NH_4^+ -N. Alkalinity was 33 determined by titration with an ending pH value of 4.3 according to the standard method.¹
- 34 FNA concentration was calculated according to the equation FNA (mg $HNO₂ N/L$) =

- 36 of linear regression of Arrhenius plot: $\ln r = -\frac{E_a}{R} \cdot \frac{1}{T} + \ln A$, where r is the biomass activity, $\frac{E_a}{R} \cdot \frac{1}{T} + \ln A$
- 37 *R* is the gas constant (8.32 J/mol/K), *T* is the temperature in kelvin, and *A* is the reaction 38 frequency factor.
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40 **Text S2. Measurement of maximal activity for AOB, NOB and anammox bacteria**

- 41 The maximal activities of AOB and NOB were measured in the aerobic MBBR, and the 42 maximal anammox activity was assessed in the anoxic MBBR. Each test lasted for 3 h, 43 during which all the controllers including two feeding pumps and two pH control systems 44 were temporarily turned off. A NH₄HCO₃ stock solution (10 g N/L) of 10 mL and a Na₂NO₂ 45 stock solution (10 g N/L) of 8 mL were added into the aerobic MBBR to increase NH₄⁺-N 46 and $NO₂$ -N concentrations to about 60–80 mg N/L. The DO concentration was maintained 47 above 7.0 mg/L by constantly supplying compressed air to the reactor at a flow rate of 1.0 48 L/min via an air pump (whisper 100, China). The pH of aerobic MBBR was controlled 49 between 7.0 and 7.5 by adding 0.1 M HCL and 0.1 M NaOH manually. Liquid samples were 50 collected every 0.5 h to the end of test, and then filtered by 0.22 μm disposable sterile 51 Millipore filters (Merck) for the analysis of NH_4^+ -N, NO_2^- -N and NO_3^- -N concentrations. The 52 maximal AOB and NOB activities were represented by the volumetric NH_4^+ -N oxidation and 53 NO₃-N production rates, which were determined through linear regression of corresponding 54 profiles obtained from the batch test. In the anoxic MBBR, 6 mL $NH₄HCO₃$ stock solution 55 (10 g N/L) and 6 mL Na₂NO₂ stock solution (10 g N/L) were added to increase NH₄⁺-N and 56 NO₂ -N concentrations to 60–70 mg N/L. To remove oxygen in anoxic MBBR, compressed 57 pure dinitrogen (N_2) gas was continually flushed in the reactor at a flow rate of 1.0 L/min 58 during the test. The pH control and sampling strategy of anoxic MBBR were similar to the 59 test performed in aerobic MBBR. The volumetric NH_4 ⁺-N oxidation rate, which was 60 determined through linear regression of the ammonium profile obtained from the batch test, 61 represented the maximal anammox activity.
	- S2

 $\frac{NO_2^-(mg N/L)}{2.2300/273 + Temp(C)}$ \times 100H. The apparent activation energy (*Ea*, kJ/mol) was estimated by the slope $e^{-2300/273 + Temp(^{\circ}C)} \times 10^{pH}$

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63 **Text S3. DNA extraction, 16S rRNA gene amplicon sequencing, and data analyses**

64 Two aerobic and two anoxic biofilm samples were collected at the end of Phase Ⅰ (i.e., 65 operating temperature of 23°C) and Ⅳ (i.e., operating temperature of 12°C), respectively. 66 After collection, all samples were stored at -80°C before being assessed by the Australian 67 Centre for Ecogenomics at The University of Queensland (https://ecogenomic.org/). DNA 68 was extracted from 50-200 mg of raw sample using Qiagen DNeasy Powersoil Pro Kit (cat 69 #7016) following the manufacturer's protocol and checked with gel electrophoresis. The 16S 70 rRNA gene encompassing the V6 to V8 regions was targeted using the 926F (5'- AAA CTY 71 AAA KGA ATT GRC GG -3') and 1392wR (5'- ACG GGC GGT GWG TRC -3') primers 72 modified to contain Illumina specific adapter sequence (926F: 5'- TCG TCG GCA GCG 73 TCA GAT GTG TAT AAG AGA CAG AAA CTY AAA KGA ATT GRC GG -3' and 74 1392wR: 5'- GTC TCG TGG GCT CGG AGA TGT GTA TAA GAG ACA GAC GGG 75 CGG TGW GTR C -3'). The universal primer pair 926F-1392wR amplifies the small submit 76 (SSU) ribosomal RNA of eukaryotes (18S) and prokaryotes (16S) specifically the V6, V7 and 77 V8 regions. Raw sequencing data was processed by Quantitative Insights Microbial Ecology 78 Ⅱ (QIIME Ⅱ) in multiple steps, including poor-sequences removal. After that, the sequences 79 were clustered into operational taxonomic units (OTUs) at 97% identify threshold.

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81 **Text S4. Mass and energy balance assessments**

82 The mass and energy balance assessments were carried out to further evaluate the feasibility 83 of applying this novel system for domestic wastewater treatment. The evaluation of mass

84 balance was performed based on the measured data of the laboratory-scale treatment system.

85 The energy balance assessment was performed in a hypothetical WWTP with a treatment

86 capacity of 10,000 m³/d, according to Wu et al.² The main parameters used for the calculation

- 87 are summarized in Table S5.
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90 **Figure S1.** The schematic diagram of the laboratory-scale wastewater treatment system.

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93 **Figure S2.** Changes of wastewater composition with the addition of FeCl₃ at different 94 concentrations. (a) The change of wastewater CaCO₃-alkalinity with the increased Fe dosage. 95 (b) The linear relationship between decreased $CaCO₃$ -alkalinity and dosed Fe. (c) The 96 changes of CaCO₃-alkalinity/NH₄⁺-N molar ratio and NH₄⁺-N concentration with the 97 increased Fe dosage. (d) The profile of wastewater pH with the increased Fe dosage. (e) the 98 shifts of TCOD and SCOD concentrations with the increased Fe dosage. (f) The changes of 99 TP, PO_4^3 -P, and organic nitrogen concentrations with the increased Fe dosage.

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111 units of the wastewater treatment system.

- 114 **Figure S6.** The microbial composition of the two-stage PN/A system. (a) The top ten phyla
- 115 of aerobic MBBR at 23°C. (b) The top ten phyla of aerobic MBBR at 12°C. (c) The top ten
- 116 phyla of anoxic MBBR at 23°C. (d) The top ten phyla of anoxic MBBR at 12°C. (e) The top
- 117 five genera in both aerobic and anoxic MBBRs at 23°C and 12°C.
- 118

Parameter	Unit Average value \pm standard error		Number of samples
TCOD	mg/L	461.7 ± 46.1	27
SCOD	mg/L	218.3 ± 13.2	27
NH_4^+ -N	mg/L	46.1 ± 5.7	76
$NO2 - N$	mg/L	ND^a	76
$NO3-N$	mg/L	ND^a	76
$PO43-P$	mg/L	6.8 ± 1.3	76
Alkalinity	$mg \text{CaCO}_3/\text{L}$	382.4 ± 35.6	35
pH		7.1 ± 0.1	38

119 **Table S1**. Main characteristics of raw domestic wastewater used in this study.

120 a ND: Not detected as the nitrite and nitrate was always below 0.5 mg/L and could be

121 neglected during the experiments.

123 **Table S2.** Removals of organic carbon and nutrients by the integrated wastewater treatment system (CEPT, acidic PN and anammox) at different 124 temperatures.

Parameter	Unit	Temperature	Raw wastewater	CEPT effluent	Acidic PN effluent	Anammox Effluent
	COD concentration mg COD/L	23° C	468.3 ± 50.7	176.3 ± 19.2	73.9 ± 13.9	38.8 ± 10.8
		20° C	432.5 ± 4.9	168.3 ± 3.8	92.3 ± 7.6	50.0 ± 5.6
		15° C	439.5 ± 33.2	167.7 ± 21.0	66.0 ± 23.6	52.3 ± 12.1
		12° C	449.7 ± 20.6	178.0 ± 8.5	93.5 ± 7.7	49.8 ± 5.3
	mg P/L	$23^{\circ}C$	6.6 ± 1.5	0.6 ± 0.4	0.7 ± 0.3	0.3 ± 0.2
$PO43-P$		20° C	7.1 ± 1.2	0.7 ± 0.3	0.5 ± 0.2	0.4 ± 0.1
concentration		15° C	6.5 ± 0.8	0.5 ± 0.3	0.5 ± 0.1	0.5 ± 0.2
		12° C	6.8 ± 0.7	0.5 ± 0.3	0.5 ± 0.2	0.4 ± 0.1
		$23^{\circ}C$	49.2 ± 4.9	48.4 ± 5.3	16.9 ± 4.2	2.7 ± 1.4
NH_4^+ -N		20° C	42.9 ± 4.7	43.4 ± 4.5	17.4 ± 3.4	2.4 ± 0.7
concentration	mg N/L	15° C	44.1 ± 0.9	42.9 ± 0.6	17.3 ± 1.6	1.8 ± 0.7
		12° C	46.8 ± 1.2	45.4 ± 1.8	20.1 ± 1.5	2.1 ± 0.6
	mg N/L	$23^{\circ}C$	ND^a		28.8 ± 4.2	1.5 ± 1.1
$NO2-N$		20° C		ND^a	24.3 ± 0.6	1.8 ± 0.3
concentration		15° C			25.9 ± 2.1	2.4 ± 0.4
		12° C			25.5 ± 1.9	1.9 ± 0.6
	mg N/L	$23^{\circ}C$			0.6 ± 0.8	1.5 ± 1.1
$NO3-N$		20° C	ND^a	ND^a	0.6 ± 0.6	1.8 ± 0.3
concentration		15° C			0.2 ± 0.2	2.4 ± 0.4
		12° C			0.2 ± 0.1	1.9 ± 0.6
	mg N/L	$23^{\circ}C$	49.5 ± 4.9	48.6 ± 5.3	47.2 ± 6.2	5.1 ± 1.8
TN concentration		20° C	43.0 ± 4.7	43.7 ± 4.5	42.1 ± 3.3	5.5 ± 1.1
		15° C	44.3 ± 0.9	43.0 ± 0.6	43.3 ± 1.4	5.1 ± 1.1
		12° C	47.1 ± 1.2	45.7 ± 1.8	45.6 ± 2.2	5.1 ± 1.2

125 *a* ND: Not detected.

Parameter	Total iron	Dissolved iron
Unit		(mg Fe/L)
Iron dosage	50	
CEPT effluent	2.8 ± 1.2	0.3 ± 0.2
Acidic MBBR effluent	2.4 ± 1.2	0.2 ± 0.1
Anoxic MBBR effluent	2.3 ± 0.8	0.2 ± 0.1

126 **Table S3.** The iron concentration in each part of system.

Influent ammonium concentration (mg N/L)	Type of wastewater	HRT (h)	NLR $(kg N/(m^3 \cdot d))$	Type of reactor	Temperature $({}^{\circ}C)$	TN removal efficiency $(\%)$	Biomass growth type	Reference
53 ± 5	diluted sludge digester	12	0.11 ± 0.01	SBR	15	70.6 ± 19.5	granules	$\overline{3}$
	aerobically pre-treated	$9 - 14$	$0.04 - 0.06$			73.1	biofilm	
21.2 ± 5.2	municipal wastewater	$12 - 14$	0.04	SBR	15	62.3	biofilm + suspended	$\overline{4}$
		1.57	$0.18 - 0.35$	RBC	15	36 ± 9	biofilm	$\overline{5}$
23-46	synthetic wastewater	1.09	$0.25 - 0.51$		14	42 ± 4		
61	synthetic wastewater	5.4	0.27 ^a	SBAR	10	39	granules	6
50	synthetic wastewater	$64.8 -$ 139.2^a	0.0 $1 - 0.02$	MBBR	10	71.8 ^a	biofilm	$\overline{7}$
	synthetic wastewater	>240	< 0.05	SBR	10	~ 30	suspended	8
		24-48	$0.03 - 0.05$	SBR		~140	granules	
50		24-48	$0.03 - 0.05$	MBBR		~140	biofilm (2 mm)	
		24-48	$0.03 - 0.05$	MBBR		~1	biofilm (10 mm)	
~1	synthetic wastewater	60	~10.03	SBR	12	>90	suspended	9
60-80	synthetic wastewater	${\sim}8$	~10.2	SBR	15	$<$ 50	granules	$10\,$
45.4 ± 1.8	CEPT pre-treated municipal wastewater	12.2	-0.09	MBBR	12	88.9 ± 2.5	biofilm	This study

128 **Table S4.** The performance of mainstream PN/A process in treating low-strength wastewater at low temperature.

129 RBC: rotating biological contactor; SBR: sequencing batch reactor; SBAR: sequencing batch air-lift reactor

Parameters	Unit	Value
Wastewater flow rate	m^3/d	10000
Fe dosage	mg/L	50
Influent COD concentration	mg/L	468
Effluent COD concentration	mg/L	40
Influent NH_4 ⁺ concentration	mg N/L	46
Effluent NH_4 ⁺ concentration	mg N/L	$\overline{3}$
COD removal efficiency of CEPT	$\frac{0}{0}$	62.4
Ratio of NH_4^+ oxidized by AOB	$\frac{0}{0}$	57
Autotrophic bacteria yield ¹¹	g cell formed / $g NH_3-N$ oxidized	0.24
Autotrophic bacteria decay ¹¹	/day	0.1
Heterotrophic bacteria yield ¹¹	g cell COD formed / g COD removed	0.64
Heterotrophic bacteria decay ¹¹	/day	0.2
Anammox bacteria yield ¹¹	g cell formed / $g NH_3-N$ oxidized	0.13
Anammox bacteria decay ¹¹	/day	0.1
Energy recovery efficiency of AD	$\frac{0}{0}$	30
Energy density of $CH4$	MJ/m^3	36

131 **Table S5.** Main parameters used in the energy balance assessment.

Type of wastewater	Country	Ammonia nitrogen concentration (mg N/L)	Alkalinity concentration (mg CaCO ₃ /L)	Molar ratio of alkalinity to ammonia nitrogen	Reference
HRAS (high rate activated sludge) effluent	Australia	44.7 ± 4.5	380 ± 10	1.17 ± 0.03	12
Raw domestic wastewater	China	79.1	510.6	0.90	13
Raw domestic wastewater	Korea	35.2	195	0.78	14
Raw domestic wastewater	Greece	150	350	0.33	15
Raw domestic wastewater	Venezuela	21.3	115.2	0.76	16
Raw domestic wastewater	Spain	29 ± 5	250 ± 28	~1.21	17
		8.6 ± 0.5	73.5 ± 2.4	~1.20	
Raw domestic wastewater	India	8.0 ± 0.2	58.8 ± 3.3	~1.03	18
		7.2 ± 0.1	62.8 ± 2.8	~1.22	
Raw domestic wastewater	France	44.2	342	1.08	19
		12	50	0.58	
Raw domestic wastewater	United States	25	100	0.56	
		50	200	0.56	
Raw domestic wastewater	Brazil	17 ± 3	155 ± 17	~1.28	20
Raw domestic wastewater	India	$30 - 45$	$230 - 300$	$0.72 - 1.40$	21
Raw domestic wastewater	Australia	46.1 ± 5.7	382.4 ± 35.6	1.1 ± 0.1	This study

133 **Table S6**. Ammonium nitrogen and alkalinity concentrations in domestic wastewater in the literature.

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