SUPPORTING INFORMATION: Coherent X-ray Scattering Reveals Nanoscale Fluctuations in Hydrated Proteins

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1 Detection of crystallization



Figure S1: Wide-angle X-ray Scattering (WAXS) 2D intensity for the hydrated proteins at T = 290 K (left) and T = 175 K (right). The latter shows sharp rings, feature of crystallization.

2 Azimuthal dependence analysis



Figure S2: a) Layout of the performed azimuthal dependence analysis. The image depicts the different Q-bins as well as the azimuthal angles. b) The autocorrelation functions g_2 calculated for the horizontal component (black) and the vertical component (orange). The corresponding areas in the detector are marked with the same colors in panel a. The analysis shows no major difference between the horizontal and the vertical component.

3 WAXS during exposure



Figure S3: The normalised WAXS intensity I(Q) as a function of momentum transfer Q (left panel) during a measurement with 1000 s exposure time measured with $F = 1.5 \cdot 10^6 \text{ ph/s}/\mu\text{m}^2$ at temperature T = 290 K. The I(Q) does not show any significant changes and the solid lines depict Gaussian fits used to extract the peak position shown in the right-hand panel.

4 Temperature increase and dose estimation

Temperature increase

The transmission of lysozyme ($T_{\text{lys}} = 0.495$) for the current experimental conditions (photon energy E = 12.4 keV, sample thickness $d_s = 1.5$ mm) is calculated by using the molecular formula $C_{125}H_{196}N_{40}O_{36}S_2$ based on the known atomic data tables [1]. The total sample transmission is estimated by averaging the transmission of lysozyme with that of water ($T_w = 0.673$) weighted by the corresponding mass fraction (h = 0.3).

The absorbed energy dQ for a given exposure time t_e is calculated by including the incident beam flux I (in μ J s⁻¹) and the sample transmission, which gives $dQ = (1 - T_r) \cdot I \cdot t_e$. This estimation corresponds to a maximum temperature increase of $T_{max}(t) = dQ/m/c_p$ where m is the mass and c_p the heat capacity. Here we used the following values for the lysozyme: isobaric heat capacity $c_p = 1260 \text{ Jkg}^{-1} \text{ K}^{-1}$ [2], density $\rho = 2200 \text{ kg} \text{ m}^{-1}$ [3] and heat conductivity $k_w = 0.42 \text{ W} \text{ K}^{-1} \text{ m}^{-1}$ [4]. The heat dissipation time [5] is calculated by

$$t_0 = c_p \cdot \rho \cdot a^2 / (2 \cdot k_w) \approx 3 \text{ ms}$$
⁽¹⁾

where a is the beamsize (30 µm). The corresponding temperature rise [5] is

$$\delta T = \int T_{max}(t) \cdot [1 - \exp(-\frac{1}{[2 \cdot (1 + t/t_0)]})] dt.$$
⁽²⁾

This estimation gives values below $\Delta T = 10$ K for the highest flux used here, $I = 4 \cdot 10^9$ ph s⁻¹.

Dose

In order to quantify the amount of energy absorbed by the sample, we calculate the dose \mathcal{D} absorbed by the system by

$$\mathcal{D} = \frac{F \cdot E \cdot A \cdot t_e}{a^2 \cdot d_s \cdot \rho_w},\tag{3}$$

where *F* denotes the flux, *E* the photon energy, *A* is the absorption, t_e the exposure time, *a* the beam size, d_s the sample thickness, and ρ_w the weighted averaged density. The values utilized are summarized in Table S1. The calculated dose is $\mathcal{D} = 1.58$ kGy.

The absorption is computed from the transmission as $A = 2 - \log(\% T_{lys})$, with $T_{lys} = 0.495$ (see previous paragraph), which is calculated from atomic data tables and is in agreement with the transmission measurements performed during the experiment. Furthermore, the weighted average density ρ_w is computed based on the mass fraction h = 0.28. In particular,

$$\rho_w = \frac{\sum_{i=1}^n w_i \rho_i}{\sum_{i=1}^n \rho_i} = \frac{w_{lys} \rho_{lys} + w_{wat} \rho_{wat}}{\rho_{lys} + \rho_{wat}}$$
(4)

and the weights w_i are calculated from

$$h = \frac{m_{wat}}{m_{lys}} = \frac{\rho_{wat} V_{wat}}{\rho_{lys} V_{lys}}.$$
(5)

By rearranging Equation (5), one obtains that $w_{lys} = 0.73$ and $w_{wat} = 1 - w_{lys}$.

Quantity	Symbol	Value	Unit
Flux	F	$6 imes 10^9$	ph/s
Photon energy	Ε	9	keV
Absorption	Α	0.31	
Exposure time	t_e	1	S
Beam size	а	$30 imes 10^{-6}$	m
Sample thickness	d_s	$1.5 imes 10^{-3}$	m
Sample density	$ ho_w$	1.25	kg/m ³

Table S1: Parameters used for the dose estimation

References

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