Supplementary Information

Asymmetric transformation of achiral gold nanoclusters with negative nonlinear dependence between chiroptical activity and enantiomeric excess

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Supplementary Methods

Reactions of Au₂₃ with phosphines

(1) Synthesis of [Au₂₃(SC₆H₁₁)₁₆]⁻TOA⁺

HAuCl₄·3H₂O (0.3 mmol, 118 mg) and tetraoctylammonium bromide (TOAB, 0.35 mmol, 190 mg) were dissolved in methanol (15 mL) in a 100 mL round-bottom flask. After vigorously stirring for 15 min, the solution color changed from yellow to dark reddish orange. Then, cyclohexanethiol (1.6 mmol, 196 μL) was added to the mixture at room temperature. The reddish orange solution turned yellowish immediately, indicating the conversion of Au(III) to Au(I) complexes. After 15 min, NaBH₄ (3 mmol, 114 mg dissolved in 6 mL of cold Nanopure water) was rapidly added to the solution under vigorous stirring. The solution turned black immediately, indicating the formation of Au clusters.^[11] The reaction mixture was allowed to stir overnight and finally gave rise to Au₂₃ (20% yield, Au atom basis), which was further purified by recrystalization.

(2) Synthesis of phosphoramidites



Triethylamine (5.0 eq., 25 mmol, 3.5 mL) was added dropwise to a stirred ice-cooled solution of PCl₃ (5 mmol, 436 µL) in dichloromethane (DCM, 35 mL). The ice bath was removed and the solution was warm to room temperature before diethylamine (5 mmol, 517 µL) was added. After 5 h, (*S*)-binaphthol (5 mmol, 1.43 g) was added to the suspension and the resulted mixture was left to stir for an additional 18 h. The purification of the reaction was achieved via column chromatography on silica gel (eluent: Pentane/EtOAc = 60/1) to give (*S*)-L₁ (1.30 g, 3.4 mmol, 68%), colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 7.98-7.90 (m, 4H), 7.52 (d, *J* = 8.8 Hz, 1H), 7.44-7.35 (m, 5H), 7.30-7.23 (m, 2H), 3.12-3.01 (m, 2H), 2.92-2.80 (m, 2H), 1.06 (t, *J* = 6.8 Hz, 6H). The ¹H NMR data is consistent with that in the reported literature.^[2]

The general method from above was used for the preparation of $(Rac)-L_1$, $(R)-L_1$ and L₂₋₅ with racemic, R and S configurations.



L₂ (1.32 g, 3.18 mmol, 64%), colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 7.92-7.89 (m, 4H), 7.52 (d, *J* = 8.8 Hz, 1H), 7.46-7.38 (m, 4H), 7.33-7.21 (m, 3H), 3.44-3.34 (m, 2H), 1.23 (d, *J* = 6.8 Hz, 6H), 1.19 (d, *J* = 6.8 Hz, 6H). The ¹H NMR data is consistent with that in the reported literature.^[3]



L₃ (1.30 g, 2.90 mmol, 58%), colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 7.98 (d, *J* = 8.8 Hz, 1H), 7.92 (d, *J* = 8.0 Hz, 1H), 7.88-7.85 (m, 2H), 7.57 (d, *J* = 8.8 Hz, 1H), 7.43-7.31 (m, 9H), 7.26-7.22 (m, 3H), 4.28-4.22 (m, 1H), 3.82-3.76 (m, 1H), 3.04-2.93 (m, 1H), 2.67-2.65 (m, 1H), 1.06 (t, *J* = 6.8 Hz, 3H). The ¹H NMR data is consistent with that in the reported literature.^[4]



L₄ (1.18 g, 2.44 mmol, 49%), colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 7.96 (d, *J* = 8.8 Hz, 1H), 7.90 (d, *J* = 8.4 Hz, 1H), 7.75 (d, *J* = 8.4 Hz, 1H), 7.53-7.47 (m, 2H), 7.42-7.33 (m, 2H), 7.26-7.17 (m, 4H), 7.08-7.06 (m, 4H), 7.04-7.00 (m, 4H), 6.97-6.90 (m, 3H). The ¹H NMR data is consistent with that in the reported literature.^[5]



L₅ (1.35 g, 3.25 mmol, 65%), colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 7.96 (d, J = 8.8 Hz, 1H), 7.91 (t, J = 6.8 Hz, 3H), 7.51 (d, J = 8.8 Hz, 1H), 7.43-7.38 (m, 4H),

7.35 (d, J = 8.8 Hz, 1H), 7.29-7.22 (m, 2H), 2.95-2.85 (m, 2H), 2.82-2.72 (m, 2H), 1.54-1.44 (m, 4H), 0.78 (t, J = 7.2 Hz, 6H). The ¹H NMR data is consistent with that in the reported literature.^[6]

(3) The reactions of different phosphine ligands with Au₂₃

*Reaction of Au*₂₃ *with butane-2,3-diybis(diphenylphosphine)*

3.0 equivalents of butane-2,3-diybis(diphenylphosphine) (2.80 mg) were added into the DCM (1.5 mL) solution of Au₂₃ (15.0 mg). The reaction was monitored by UV-vis. After 4 h, the mixture was purified by prepared thin layer chromatography (PTLC) with DCM/MeOH = 20/1. As shown in Supplementary Figure 1, at least three components were observed based on the PTLC. Their UV-vis spectra were also shown in Supplementary Figure 1a. A lower equivalent of the phosphine ligand or reaction temperature did not improve the selectivity either. These results indicated that butane-2,3-diybis(diphenylphosphine) with two chiral sp³ carbons was reactive to Au₂₃, but showed poor selectivity and resulted in wide product distribution.

Reaction of Au₂₃ with BINAP

5.0 equivalents of BINAP (6.83 mg) were added into the DCM (1.5 mL) solution of Au_{23} (15.0 mg). The reaction was monitored by UV-vis, and it was found that the UV-vis spectra kept unchanged even after four days. A higher equivalent of BINAP did not improve the reactivity either. The mixture was purified by PTLC with Pentane/DCM = 2/1 and Au₂₃ was found to be recovered based on the UV-vis spectrum (Supplementary Figure 1b). This result indicated that axially chiral BINAP with sterically hindered phosphine sites displayed low activity to Au₂₃, and led to the recovery of the gold nanocluster.

Reaction of Au₂₃ with phosphoric acid

3.0 equivalents of phosphoric acid (2.29 mg) were added into the DCM (1.5 mL) solution of Au_{23} (15.0 mg). The reaction was monitored by UV-vis, and it was found that the mixture turned to colorless quickly, demonstrating the decomposition of Au_{23} nanocluster. A lower equivalent of the phosphoric acid did not give better results. This

result indicated that phosphoric acid triggered the decomposition of Au₂₃, probably due to its strong acidity.

Reaction of Au_{23} with phosphoramidites (L_1 - L_5)

5.0 equivalents of L_1 (4.24 mg) were added into 15.0 mg of Au₂₃, which were dissolved in 1.5 mL of DCM. After the completion of the reaction in 4 h, the mixture was concentrated, and washed with MeOH for 2-3 times to give the crude LC₁ (precipitate). Recrystallization of the crude product in the system of DCM/n-hexane gave red hexagon crystals (Supplementary Figure 2b), which were suitable for UV-vis (Supplementary Figure 2a), ESI-MS, SCXRD and the other characterizations. Notably, purification of the reaction was also achieved via PTLC or column chromatography on silica gel (eluent: DCM/MeOH = 20/1), giving the pure LC₁ (10.0 mg, yield: 65%, based on Au atom). PTLC showed that three components were included in the products (Supplementary Figure 2c): component A (soluble complexes), component B (LC₁) and component C (insoluble nanoparticles).

The procedure from above was also applied for the preparation of LC₂-LC₅ by using L₂-L₅. The reactions were isolated by column chromatography on silica gel (eluent: DCM/MeOH = 20/1) to give the yields of three components of products (Supplementary Table 2).

Supplementary Tables

Supplementary Table 1. Reaction results of different chiral phosphine ligands with

 Au_{23}



Supplementary Table 2. Reaction results of L1-L5 with Au23

$\begin{array}{c} & L_{1-5} (5.0 \text{ equiv.}) \\ Au_{23}(SC_{6}H_{11})_{16} & \longrightarrow & Au_{24}(L)_{2}(SC_{6}H_{11})_{16} \\ 15 \text{ mg} & DCM, \text{ rt, 4 h} & LC_{1-5} \end{array}$						
1	Υ	ields of products	oducts			
∟ 1-5	Component A (complexes)	Component B (LC₁₋₅)	Component C (nanoparticles)			
L ₁	3.0 mg	10.0 mg, 65%	1.3 mg			
L ₂	3.8 mg	9.20 mg, 59%	1.5 mg			
L ₃	3.3 mg	10.2 mg, 65%	1.3 mg			
L ₄	3.0 mg	9.00 mg, 57%	1.5 mg			
L_5	3.5 mg	10.0 mg, 64%	1.4 mg			

Supplementary Table 3. Amount of (R)-L₁ and (S)-L₁ used for the preparation of L₁ with different ee

ee/%	-100	-80	-60	-40	-20	0	20	40	60	80	100
(R)-L ₁	0	1	2	3	4	5	6	7	8	9	10
(S)-L ₁	10	9	8	7	6	5	4	3	2	1	0

	(R)-LC ₁	(S)-LC ₁
CCDC Number	2216153	2216156
Empirical formula	$C_{144}H_{220}Au_{24}N_2O_4P_2S_{16}$	$C_{144}H_{220}Au_{24}N_2O_4P_2S_{16}$
Formula weight	7345.30	7345.30
Temperature (K)	296.15	296.15
Crystal system	monoclinic	monoclinic
Space group	P21	P21
A (Å)	26.7353(18)	26.555(2)
B (Å)	33.348(2)	33.331(3)
C (Å)	43.034(3)	43.127(3)
A (°)	90	90
B (°)	108.051(2)	107.875(4)
Γ (°)	90	90
Volume (Å ³)	36479(4)	36329(5)
Z	6	6
Pcalcg (cm ³)	2.006	2.014
M (mm ⁻¹)	14.594	19.355
F(000)	19872.0	19872.0
Crystal size (mm ³)	0.1 imes 0.1 imes 0.08	$0.09 \times 0.09 \times 0.1$
Radiation	MoKa ($\lambda = 0.71073$)	GaK α (λ = 1.34139)
20 range for data collection (°)	3.43 to 55.136	2.97 to 102.498
Index ranges	$-19 \le h \le 34, -41 \le k \le 43, -$	$-30 \le h \le 24, -38 \le k \le 33, -$
	$54 \leq l \leq 55$	$45 \le 1 \le 50$
Reflections collected	275282	194226
In doman dans to flastions	157258 [$R_{int} = 0.0843$, R_{sigma}	$105347 [R_{int} = 0.1644,$
independent reflections	= 0.1358]	$R_{sigma} = 0.1691$]

Supplementary Table 4. Crystal data and structure refinement for LC_1

Data/restraints/parameters	157258/17193/4910	105347/28014/5028
Goodness-of-fit on F ²	1.019	1.140
Final R indexes [I>=2σ (I)]	$R_1 = 0.0841, wR_2 = 0.2103$	$R_1 = 0.0891, wR_2 = 0.2161$
Final R indexes [all data]	$R_1 = 0.1817, wR_2 = 0.2659$	$R_1 = 0.1794, wR_2 = 0.2494$
Largest diff. peak/hole (e Å-3)	2.86/-2.00	3.12/-2.53
Flack parameter	0.224(9)	0.319(17)

Explanation for Alerts

 (\mathbf{R}) -LC₁

Alert level B

PLAT342_ALERT_3_B Low Bond Precision on C-C Bonds 0.03214 Ang.

Author Response: Large Uiso presented on the alkyl side chains, solvent molecules, and counteranions showed the potentiality of disorder, resulting in this alert.

(S)-LC₁

Alert level B

PLAT241_ALERT_2_B High 'MainMol' Ueq as Compared to Neighbors of S1_16 Check PLAT241_ALERT_2_B High 'MainMol' Ueq as Compared to Neighbors of S1_19 Check PLAT241_ALERT_2_B High 'MainMol' Ueq as Compared to Neighbors of S1_54 Check PLAT242_ALERT_2_B Low 'MainMol' Ueq as Compared to Neighbors of Au18_2 Check PLAT242_ALERT_2_B Low 'MainMol' Ueq as Compared to Neighbors of Au22_2 Check PLAT242_ALERT_2_B Low 'MainMol' Ueq as Compared to Neighbors of Au23_2 Check PLAT242_ALERT_2_B Low 'MainMol' Ueq as Compared to Neighbors of Au23_2 Check PLAT242_ALERT_2_B Low 'MainMol' Ueq as Compared to Neighbors of Au21_1 Check PLAT242_ALERT_2_B Low 'MainMol' Ueq as Compared to Neighbors of Au21_1 Check Author Response: These alerts are due to the disorder of the corresponding atom, and the temperature factor of an atom is higher or lower than that of nearby atoms.

Supplementary Figures



Supplementary Figure 1. (a) UV-vis spectra of the products in the reaction of Au_{23} with butane-2,3-diybis(diphenylphosphine). Inset: PTLC photographs of the reaction of Au_{23} with butane-2,3diybis(diphenylphosphine). (b) UV-vis spectra of the the product in the reaction of Au_{23} with BINAP. Inset: PTLC photographs of the reaction of Au_{23} with BINAP.



Supplementary Figure 2. (a) UV-vis spectra of (Rac)-LC₁, (R)-LC₁ and (S)-LC₁. (b) Crystal photographs of (R)-LC₁ and (S)-LC₁. (c) PTLC photographs of the reaction between Au₂₃ and L₁.



Supplementary Figure 3. Unit cells of Au₂₃ (a), (\mathbf{R})-LC₁ (b) and (S)-LC₂ (c) crystal structures. Color label: Au = pink, blue; S = yellow; P = green; N = orange; O = light blue; C = gray. H atoms are all omited for clearity.



Supplementary Figure 4. Packing mode of (\mathbf{R})-LC₁ molecules along the [100], [010] and [001] directions. Color label: Au = pink, blue; S = yellow; P = green; N = orange; O = light blue; C = gray. H atoms are all omited for clearity.



Supplementary Figure 5. Packing mode of (*S*)-LC₁ molecules along the [100], [010] and [001] directions. Color label: Au = pink, blue; S = yellow; P = green; N = orange; O = light blue; C = gray. H atoms are all omited for clearity.



Supplementary Figure 6. (a) Au_{13} kernel of (*R*)-LC₁ and Au-Au distances from the central gold atom to the surrounded gold atoms. (b) Top view of the Au_{13} kernel. Color label: Au = pink, green.



Supplementary Figure 7. Au–Au distances in the Au₁₃ kernels of Au₂₃ (a and b) and (R)-LC₁ (c and d). Color label: Au = pink, orange, green.



Supplementary Figure 8. The proposed transformation mechanism from Au_{23} to LC_1 . (a) Structure of Au_{23} (the two red cycle represent the open sites being attacked by L_1). (b) L_1 attack and S-Au-S motifs dissociation. (c) Rearrangement of a -S-Au-S-Au-S-Au-S- motif. (d) Association of a -S-Au-S-Au-S-Au-S- motif.



Supplementary Figure 9. The representation of the Au₁₃ kernels of (*R*)-LC₁ and (*S*)-LC₁.



Supplementary Figure 10. The representation of the left- and right-handed helix of (*R*)-LC₁ and (*S*)-LC₁.



(*R*)-LC₁ molecule

Supplementary Figure 11. The representative $H \cdots H$ interactions between two (*R*)-LC₁ molecules. Color label: Au, S, P, N, O = yellow; C = gray; H = white.



Supplementary Figure 12. (a-e) Combined CD-absorption spectra of LC_1-LC_5 . (f) UV-vis spectra of L_1-L_5 . Source data are provided as a Source Data file.



Supplementary Figure 13. (a) Anisotropy factors of **LC**₁ prepared by different ee values of **L**₁. (b and c) Representation of the negative nonlinear CD-ee dependence at 266, 322, 500 and 610 nm. The solid and border highlighted colored lines in (b) and (c) are not fitted. They are mere guides to the eye.



Supplementary Figure 14. CD spectra of the isolated (R)- and (S)- LC_1 , and the crude product obtained from the reaction of Au₂₃ with (R)- and (S)- L_1 (Inset: the enlarged view from 400 to 700 nm, and the magnification factors is 7). Source data are provided as a Source Data file.



Supplementary Figure 15. Time-dependent UV-vis spectra of Au_{23} and LC_1 at 80°C (in toluene).

Source data are provided as a Source Data file.



Supplementary Figure 16. DPV spectrum of Au_{23} (a) and LC_1 (b). Source data are provided as a Source Data file.



Supplementary Figure 17. (a) Excitation spectra of Au_{23} and LC_1 . (b) Emission spectra of Au_{23} and LC_1 . The inset shows the photographs of Au_{23} and LC_1 solution under UV lamp (365nm). Source data are provided as a Source Data file.



Supplementary Figure 18. ¹H NMR spectrum of L₁.



Supplementary Figure 19. ¹H NMR spectrum of L₂.



Supplementary Figure 20. ¹H NMR spectrum of L₃.



Supplementary Figure 21. ¹H NMR spectrum of L₄.



Supplementary Figure 22. ¹H NMR spectrum of L₅.

Supplementary references

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