

Title: Redox double switch theranostics against cancer: Pt(IV) prodrug-functionalised MnO₂ nanostructures

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Supporting Information

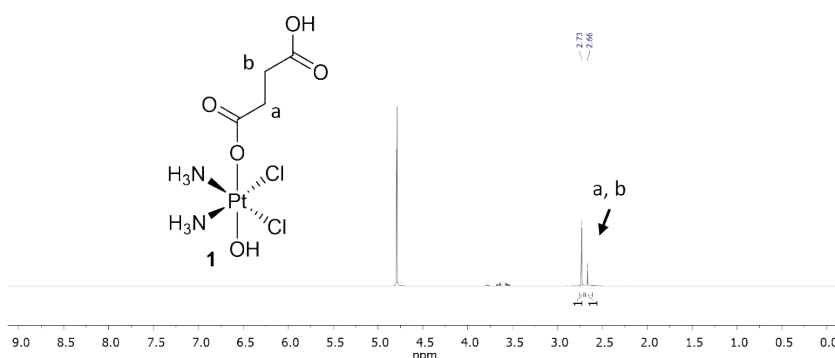


Figure S1. ¹H NMR of Pt(IV) complex **1**, in D₂O, at 400 MHz.

Table S1. Optimisation of the preparation of MnO₂-Pt NPs. Reaction conditions, hydrodynamic size (*D_H*), surface charge (ζ -Pot), and ICP measurements. Relaxivity measurements performed in the absence (native) and presence of ascorbic acid (10 mM, 1.5 T).

| Reaction | [KMnO ₄] (mM) | Mn/Pt reaction | <i>D_H</i> (nm) | ζ -Pot (mV) | Mn/Pt ICP | Native <i>r</i> ₁ (mM ⁻¹ .s ⁻¹) | Redox <i>r</i> ₁ (mM ⁻¹ .s ⁻¹) | Fold increase | |
|----------|---------------------------|----------------|---------------------------|-------------------|-------------|---|--|---------------|-----|
| A | 9 | 0.50 | 187 ± 54 | -43.0 ± 1.5 | 0.5 | 0.046 ± 0.002 | 0.343 ± 0.026 | 7 | |
| B | 1.6 | 2 | 198 ± 44 | -46.3 ± 1.4 | 9.1 | 0.066 ± 0.007 | 7.201 ± 0.351 | 108 | |
| C | | 1 | 123 ± 38 | -47.8 ± 1.3 | 8.3 | 0.078 ± 0.006 | 6.147 ± 0.141 | 79 | |
| D | | 0.70 | 160 ± 61 | -52.4 ± 0.9 | 4.0 | 0.057 ± 0.001 | 5.666 ± 0.134 | 100 | |
| E | | 0.5 | 156 ± 37 | -45.9 ± 0.9 | 4.0 | 0.104 ± 0.005 | 6.236 ± 0.227 | 60 | |
| F | | 0.25 | | 75 ± 21 | -53.2 ± 2.1 | 5.3 | 0.035 ± 0.007 | 4.698 ± 0.112 | 136 |
| G | | | | 145 ± 35 | -54.6 ± 1.9 | 4.5 | 0.070 ± 0.002 | 5.747 ± 0.077 | 82 |
| H | | | | 135 ± 35 | -26.2 ± 0.3 | 4.5 | 0.067 ± 0.011 | 7.258 ± 0.158 | 108 |
| I | | | | 140 ± 20 | -53.0 ± 0.4 | 2.8 | 0.065 ± 0.003 | 4.700 ± 0.008 | 72 |

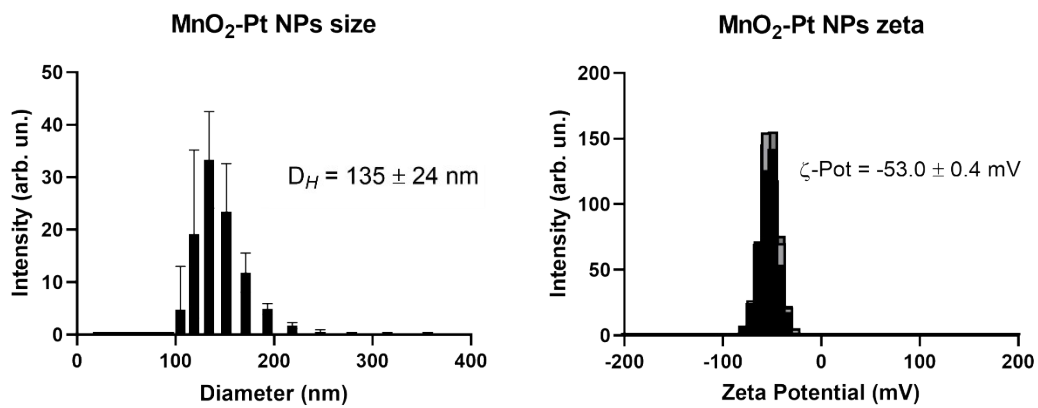


Figure S2. DLS and zeta measurements of MnO₂-Pt(IV) NPs.

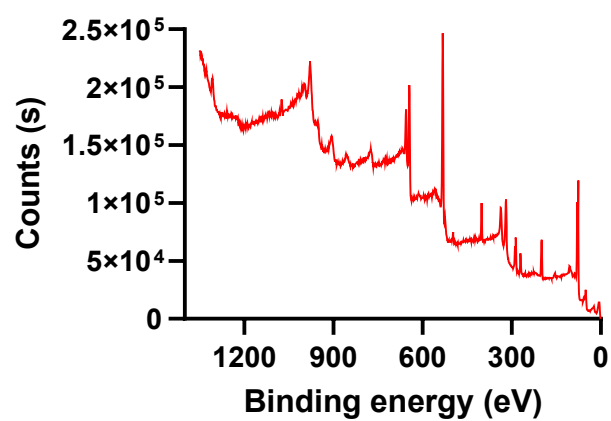


Figure S3. XPS spectra of MnO₂-Pt(IV) NPs.

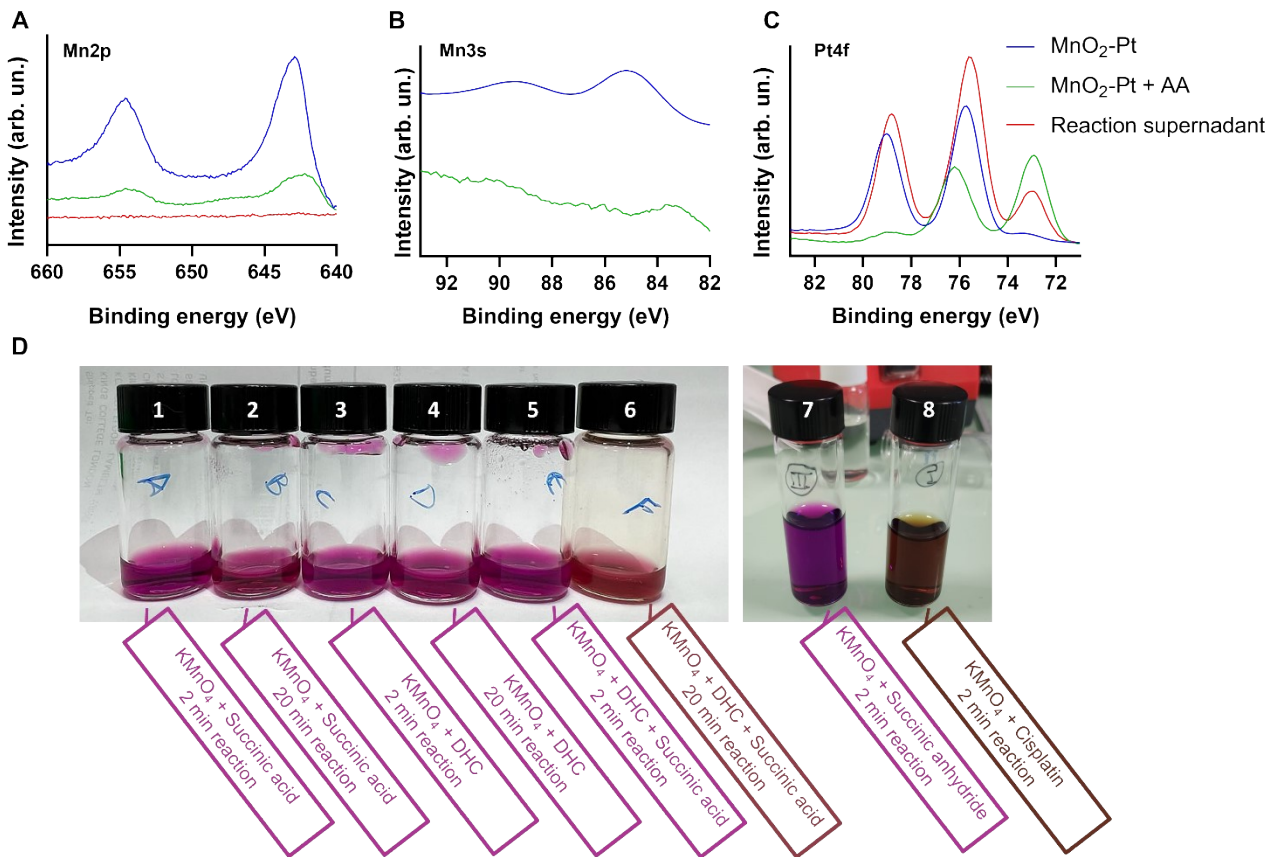
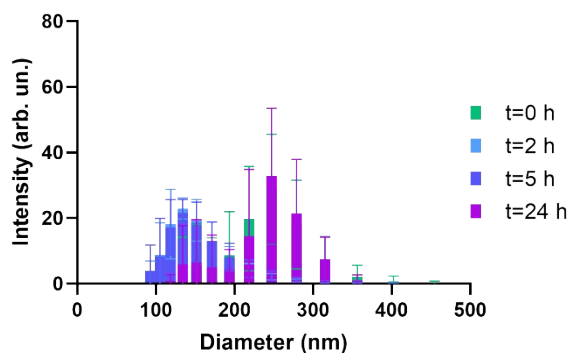


Figure S4. Expanded A) Mn2p, B) Mn3s and C) Pt4f regions of XPS spectra of MnO_2 -Pt(IV) nanoparticles. Blue line: MnO_2 -Pt(IV) NPs (pH = 7.4); green line: MnO_2 -Pt(IV) NPs (pH = 7.4, 100 μM AA); red line: reaction supernatant obtained during purification of MnO_2 -Pt(IV) NPs (pH = 7.4). D) Control KMnO_4 ultrasonication reactions with different ligands. Reaction mixtures that remained purple (1-5 and 7) show that Mn(VII) was not reduced, and nanoparticles were not formed. Reaction mixtures that turned brown after ultrasonication (6 and 8) indicate that the Mn(VII) in solution was reduced into nanoparticles.

Stability of MnO_2 -Pt(IV) NPs in PBS with 10% FBS



Stability of MnO_2 -Pt(IV) NPs in RPMI

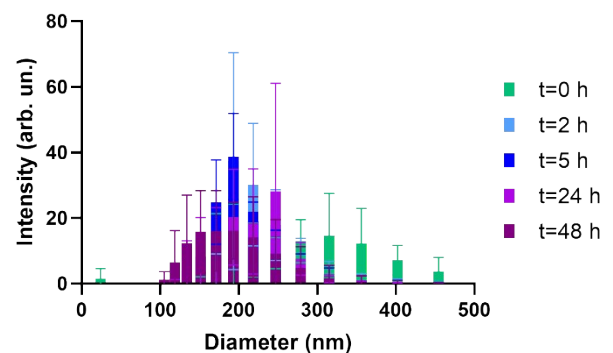


Figure S5. DLS measurements of MnO_2 -Pt nanoparticles in PBS with 10% FBS (left) or RPMI with 10% FBS (right) at different time points.

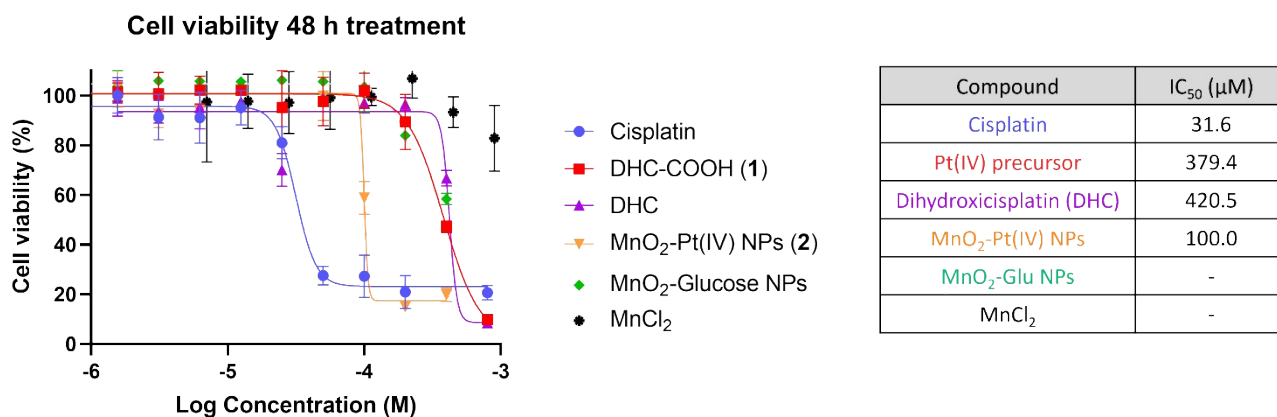


Figure S6. Cell viability study employing 2D model of A549 cells after 48 h of treatments.

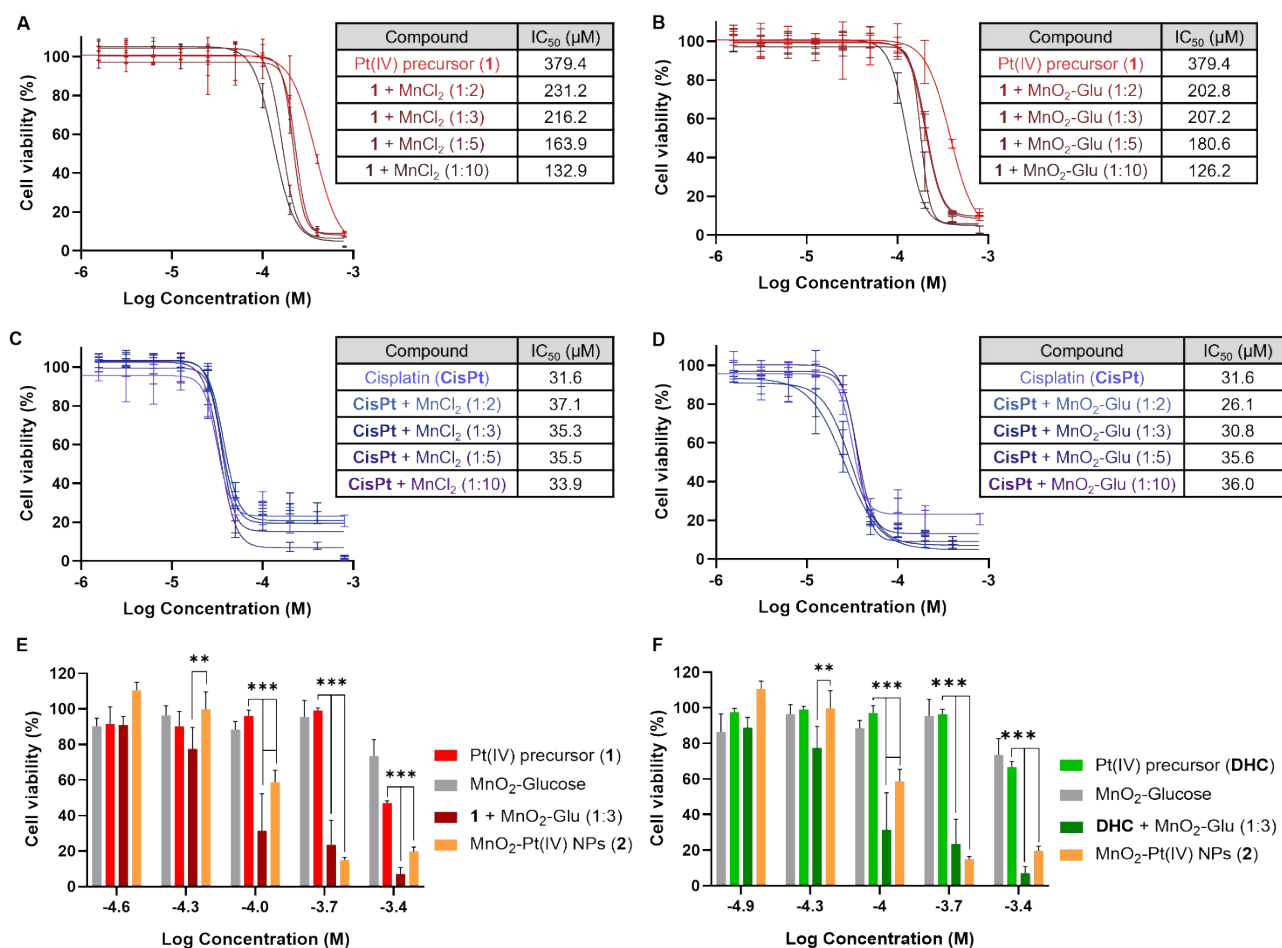


Figure S7. Cell viability studies employing 2D models of A549 cells after 48 h of treatment. Concentration refers to Pt concentration in all conditions, except for MnO₂-Glu NPs and MnCl₂, where Mn concentration is shown. Comparison of cell viability of Pt(IV) prodrug **1** after addition of **A**) MnCl₂ salts and **B**) MnO₂-Glucose nanoparticles (Mn concentration equivalent to 2, 3, 5 and 10 times the Pt concentration) and corresponding calculated IC₅₀ values. Comparison of cell viability of cisplatin after addition of **A**) MnCl₂ salts and **B**) MnO₂-Glucose nanoparticles (Mn concentration equivalent to 2, 3, 5 and 10 times the Pt concentration) and corresponding calculated IC₅₀ values. **E** and **F**) Comparison of cell viability of Pt(IV) prodrugs **1** and oxoplatin (DHC) after addition of MnO₂-Glucose nanoparticles (Mn/Pt = 3). **p < 0.0034, ***p < 0.001 (two-way ANOVA).

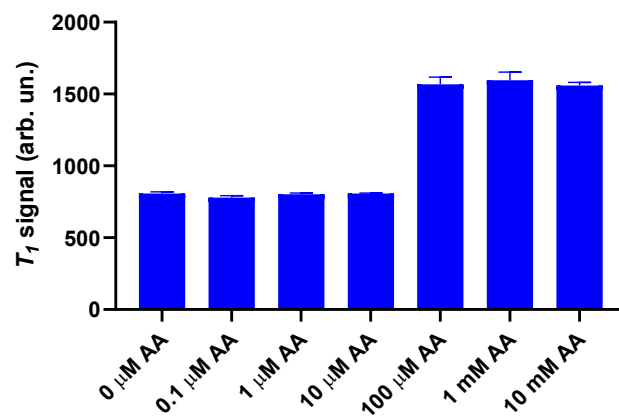


Figure S8. MR signals of MnO₂-Pt(IV) NPs in the presence of different concentrations of AA (0-10 mM) at pH=7.4, at 3 T.

Table S2. ICP-MS results from the analysis of Pt and Mn content of DNA samples extracted from cell incubated in media or in the presence of the NPs.

| | DNA measure / ng | Pt / ppm | Mn / ppm |
|----------------------|------------------|------------|-----------|
| Media control | 168.0 ± 53.7 | <LOD | 0.6 ± 0.2 |
| NPs | 103.3 ± 38.5 | 12.4 ± 2.6 | 1.3 ± 0.7 |

T_1 MR signal of MnO₂-Pt with \neq reducing agents

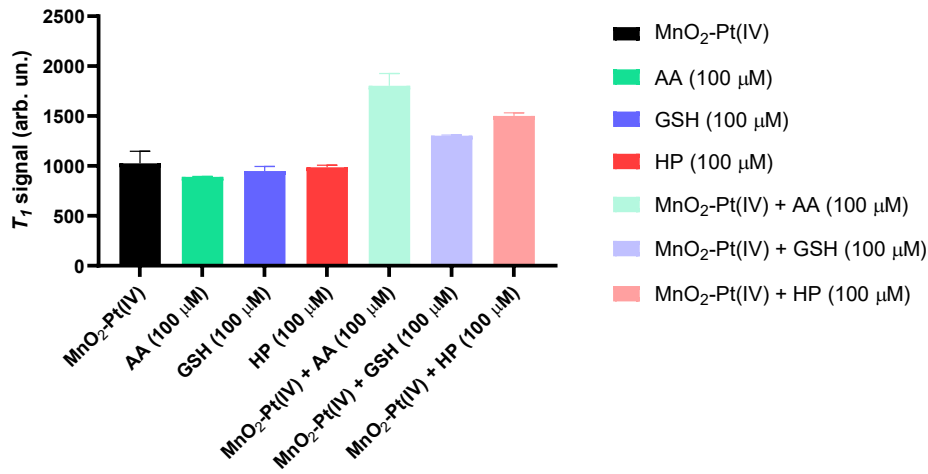


Figure S9. MR signals of MnO₂-Pt(IV) NPs in the presence of 100 μ M of different reducing agents: ascorbic acid (AA), glutathione (GSH) and hydrogen peroxide (HP) at pH=7.4, at 3 T.

Stability of MnO₂-Pt(IV) NPs in murine plasma

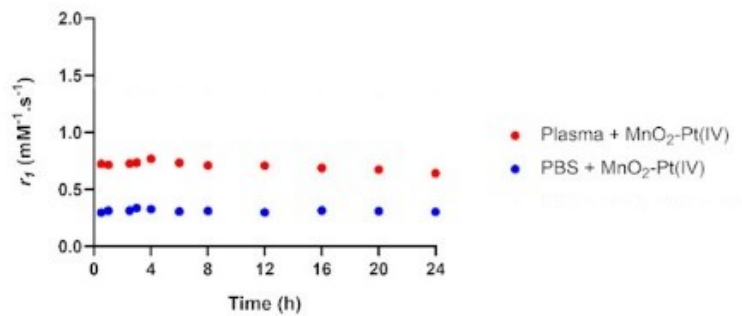
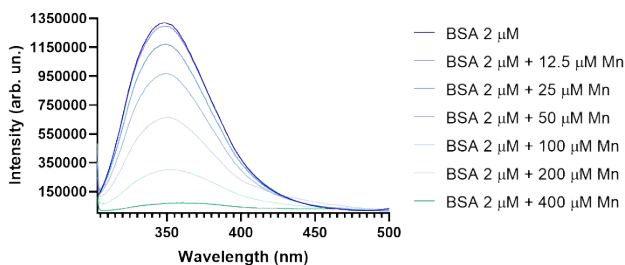


Figure S10. Relaxometry measurements of MnO₂-Pt nanoparticles in PBS or murine plasma at different time points, at 9.4 T.

Emission of 2 μ M BSA in PBS (pH 7.4) with MnO₂-Pt NPs (λ_{ex} =295 nm)



Stern–Volmer plot of BSA fluorescence quenching by MnO₂-Pt NPs

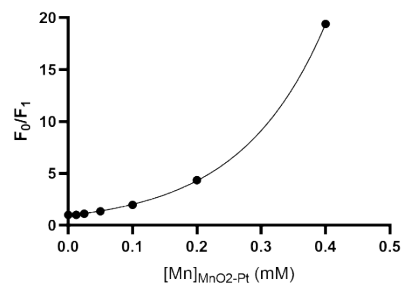


Figure S11. Left: Emission spectra of 2 μ M BSA in PBS (pH=7.4) with varying concentration of MnO₂-Pt nanoparticles at λ_{ex} =295 nm. Right: Stern–Volmer plot relating [MnO₂-Pt nanoparticles] added *versus* F_0/F_1 , where F_0 is the fluorescence intensity of BSA (2 μ M) in PBS (pH=7.4) without quenching agent and F_1 is the fluorescence intensity of samples with MnO₂-Pt NPs added, showing dynamic quenching, indicative of dynamic binding, of MnO₂-Pt(IV) NPs to BSA.

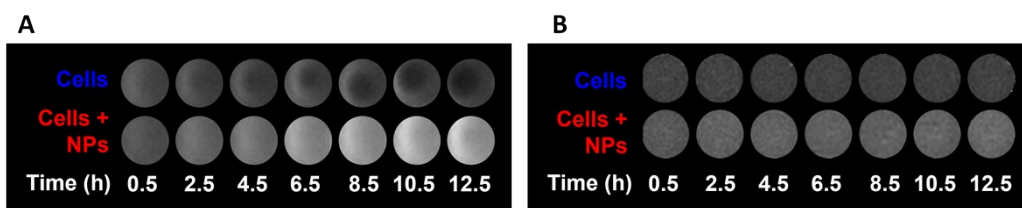


Figure S12. Representative MR phantom images of A) 2D and B) 3D A549 cells treated with MnO₂-Pt NPs (637 μM and 500 μM of Mn, respectively) over time, at 3.0 T.

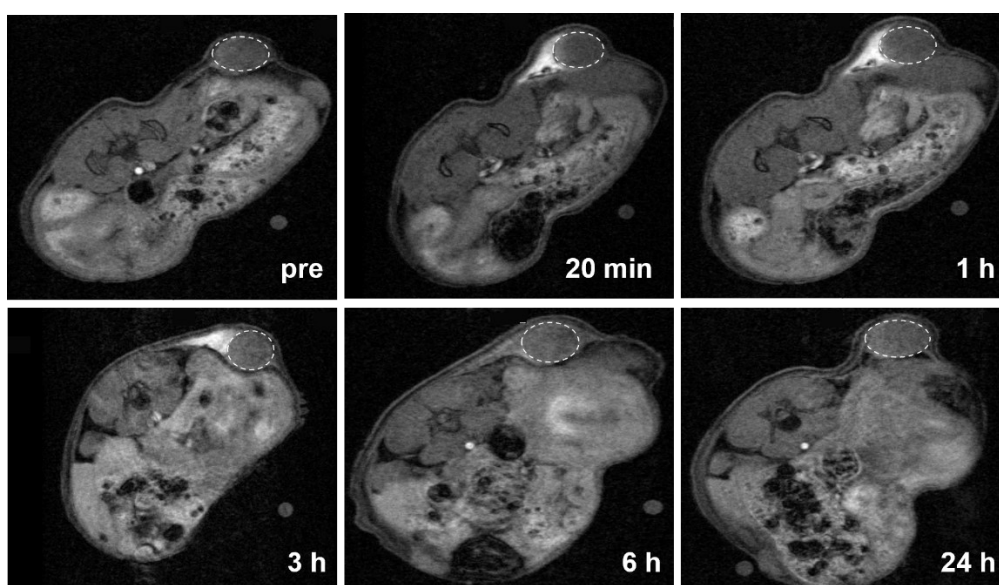


Figure S13. *T*₁-weighted axial MR images of tumour-bearing Balb/c nude mice before (pre) and 20 min, 1 h, 3 h, 6 h and 24 h after intratumoral injection of MnCl₂, acquired using a 9.4 T MR scanner. White circles highlight tumour sites.

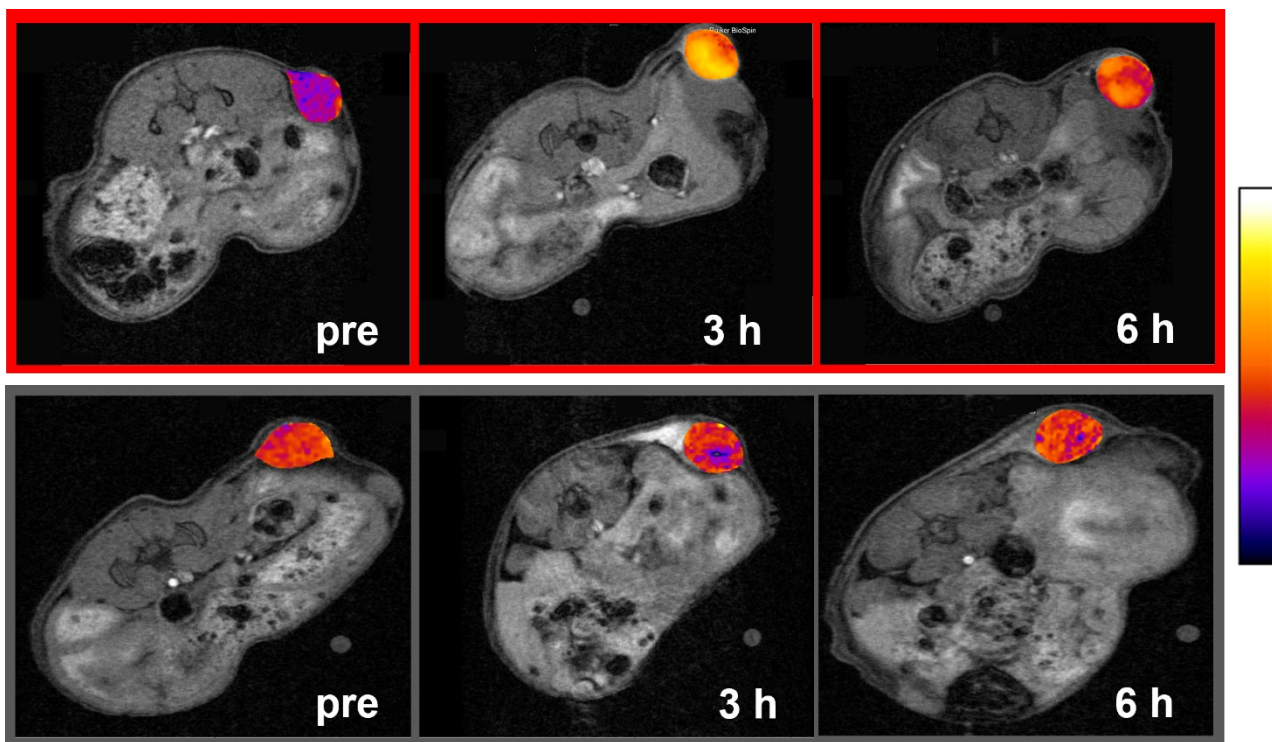


Figure S14. T_1 -weighted axial MR images of tumour-bearing Balb/c nude mice before (pre), 3 h and 6 h after intratumoral injection of MnO_2 -Pt(IV) NPs (above) MnCl_2 (below), overlapped with T_1 maps of tumours, acquired using a 9.4 T MR scanner.

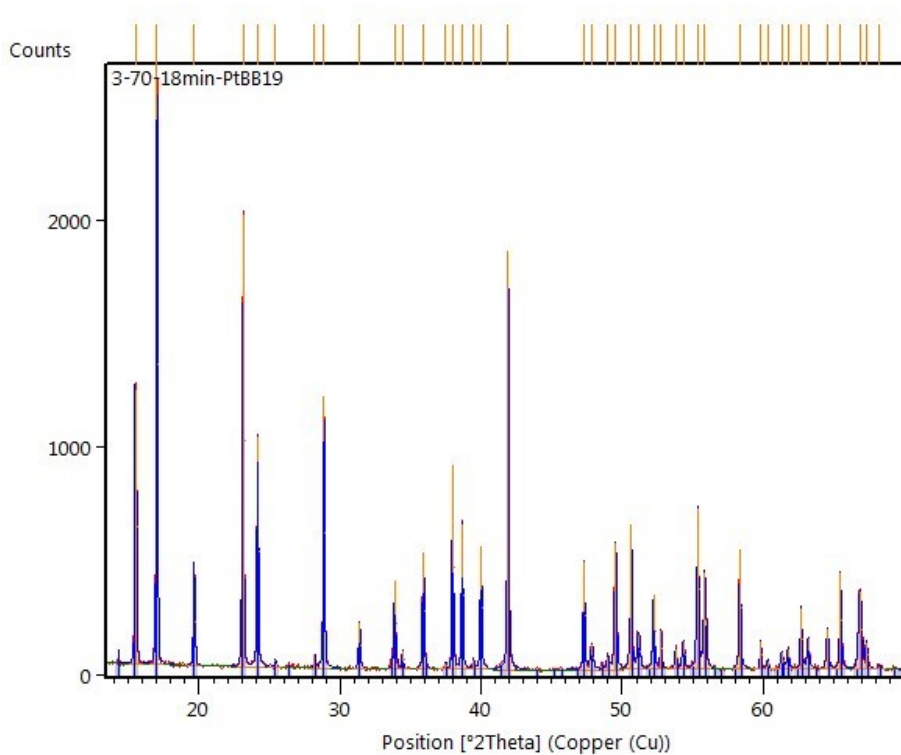


Figure S15. XRD spectrum of Pt(IV) prodrug oxoplatin (*cis, cis, trans*-Diamminedichlorodihydroxyplatinum(IV)).