

## Supporting Information

**Title:** Aggregation in aqueous solutions of 2-(tetrafluoro(trifluoromethyl)- $\lambda^6$ -sulfanyl-ethan-1-ol (CF<sub>3</sub>SF<sub>4</sub>-ethanol): A comparison with aqueous trifluoroethanol and hexafluoroisopropanol using molecular dynamics simulations and dynamic light scattering experiments.

**Authors:** Samadrita Biswas<sup>†1</sup>, Simi Kaur<sup>†1,2</sup>, Christopher A. Myers<sup>2,3</sup>, Alan A. Chen<sup>\*1,2</sup> and John T. Welch<sup>\*1</sup>

<sup>1</sup>Department of Chemistry, University at Albany, State University of New York, 1400 Washington Ave, Albany, NY 12222, USA

<sup>2</sup>RNA Institute, University at Albany, State University of New York, 1400 Washington Ave, Albany, NY 12222, USA

<sup>3</sup>Department of Physics, University at Albany, State University of New York, 1400 Washington Ave, Albany, NY 12222, USA

† equal contributions

\* corresponding author

## Table of Contents

### General Computational information

Table S1. Parameters for bonds, angles and dihedrals for CF <sub>3</sub> SF <sub>4</sub> -ethanol	S3-4
Table S2: Parameters for bonds, angles and dihedrals for TFE	S5
Table S3: Parameters for bonds, angles and dihedrals for HFIP	S6
Table S4 : Summary of charges for HFIP conformers	S7
Table S5 : Summary of charges for TFE conformers	S7
Table S6: Bonded parameter used for TFE	S7-8
Table S7: Bonded parameter used for HFIP	S8

### Figures representing Radial Distribution Function for TFE, HFIP and CF<sub>3</sub>SF<sub>4</sub>-ethanol

Figure S1: Radial Distribution Function g (r) for :TFE-TFE	S9
Figure S2: Radial Distribution Function g (r) for : HFIP-HFIP	S10
Figure S3: Radial Distribution Function g (r) for :CF <sub>3</sub> SF <sub>4</sub> -ethanol-CF <sub>3</sub> SF <sub>4</sub> -ethanol	S11
Figure S4. RDFs for aqueous mixtures of fluorinated alcohols. RDFs for the O <sub>TFE</sub> -H <sub>Water</sub> and O <sub>TFE</sub> -O <sub>Water</sub>	S12

### Presentation of NMR and DART-MS Spectra for trifluoromethyl tetrafluoro-λ<sup>6</sup>-sulfanyl-ethan-1-ol

Figure S5 : NMR spectra of 2-(tetrafluoro(trifluoromethyl)-λ <sup>6</sup> -sulfanyl)ethan-1-ol	
(a) <sup>1</sup> H Nmr spectra of CF <sub>3</sub> SF <sub>4</sub> -ethanol.	13
(b) <sup>19</sup> F Nmr spectra of CF <sub>3</sub> SF <sub>4</sub> -ethanol.	14
(c) HRMS (DART-ESI, m/z,positive) spectrum	15

**Table S1: Parameters for bonds, angles and dihedrals for CF<sub>3</sub>SF<sub>4</sub>-ethanol**

<b>Parameter<sup>a</sup></b>	<b>Gauche</b>	<b>Antiperiplanar</b>
<i>r</i> (S-C)	1.85	1.85
<i>r</i> (S-F)	1.68	1.68
<i>r</i> (S-F1)	1.68	1.68
<i>r</i> (S-F2)	1.69	1.65
<i>r</i> (S-F3)	1.72	1.65
<i>r</i> (S-C2)	1.98	1.92
<i>r</i> (C-C1)	1.53	1.53
<i>r</i> (C-H3)	1.09	1.09
<i>r</i> (C-H4)	1.09	1.09
<i>r</i> (C1-O)	1.41	1.42
<i>r</i> (C1-H1)	1.10	1.10
<i>r</i> (C1-H2)	1.10	1.10
<i>r</i> (O-H)	0.96	0.97
<i>r</i> (C2-F4)	1.32	1.32
<i>r</i> (C2-F5)	1.31	1.32
<i>r</i> (C2-F6)	1.32	1.32
$\theta$ (S-C-C1)	115.1	115.5
$\theta$ (S-C-H1)	101.97	104.05
$\theta$ (S-C-H4)	104.16	104.8
$\theta$ (S-C2-F4)	108.88	108.4
$\theta$ (S-C2-F5)	109.51	109.9
$\theta$ (S-C2-F6)	108.19	109.9
$\theta$ (C-S-F)	92.64	92.30
$\theta$ (C-S-F1)	90.52	89.93
$\theta$ (C-S-F2)	90.48	90.23
$\theta$ (C-S-F3)	93.51	93.12
$\theta$ (C-S-C2)	177.58	177.19
$\theta$ (C-C1-O)	114.26	108.47
$\theta$ (C-C1-H1)	104.71	110.07
$\theta$ (C-C1-H2)	105.83	111.28
$\theta$ (C1-C-H3)	111.57	109.89
$\theta$ (C1-C-H4)	111.49	112.21
$\theta$ (C1-O-H)	108.17	108.12
$\theta$ (O-C1-H1)	106.09	112.06
$\theta$ (O-C1-H2)	112.78	106.65
$\theta$ (H1-C1-H2)	107.36	108.18

$\theta(\text{H3-C-H4})$	111.56	109.38
$\theta(\text{F-S-F1})$	176.86	176.65
$\theta(\text{F-S-F2})$	89.24	89.31
$\theta(\text{F-S-F3})$	89.50	90.59
$\theta(\text{F-S-C2})$	89.33	90.68
$\theta(\text{F1-S-F2})$	90.38	90.23
$\theta(\text{F2-S-F3})$	175.83	177.48
$\theta(\text{F1-S-C2})$	90.91	89.36
$\theta(\text{F1-S-F3})$	90.65	93.12
$\theta(\text{F2-S-C2})$	87.55	89.93
$\theta(\text{F3-S-C2})$	85.10	92.27
$\theta(\text{F4-C2-F5})$	110.39	109.3
$\theta(\text{F5-C2-F6})$	110.17	109.34
$\theta(\text{F4-C2-F6})$	109.66	109.7
$\phi(\text{C-S-C2-F5})$	-124.18	-165.71
$\phi(\text{C-S-C2-F4})$	-4.97	-46.11
$\phi(\text{C-S-C2-F6})$	115.01	73.40
$\phi(\text{F3-S-C-C1})$	108.51	-103.63
$\phi(\text{F2-S-C-C1})$	162.05	76.25
$\phi(\text{F1-S-C-C1})$	71.84	167.03
$\phi(\text{F-S-C-C1})$	-18.85	-14.35
$\phi(\text{S-C-C1-O})$	79.03	-172.43
$\phi(\text{C-C1-O-H})$	-72.58	-76.22
$\phi(\text{C2-S-C-C1})$	38.16	70.14
$\phi(\text{F4-C2-S-F})$	52.18	44.82
$\phi(\text{F4-C2-S-F1})$	-38.68	-45.95
$\phi(\text{F4-C2-S-F3})$	141.74	133.93
$\phi(\text{F4-C2-S-F2})$	-128.86	-136.68
$\phi(\text{F5-C2-S-F})$	-67.01	-74.68
$\phi(\text{F5-C2-S-F1})$	-157.9	-165.47
$\phi(\text{F5-C2-S-F2})$	111.93	14.43
$\phi(\text{F5-C2-S-F3})$	22.59	103.79
$\phi(\text{F6-C2-S-F})$	172.18	164.35
$\phi(\text{F6-C2-S-F1})$	81.03	73.57
$\phi(\text{F6-C2-S-F2})$	-8.87	-106.53
$\phi(\text{F6-C2-S-F3})$	-98.27	-17.16

---

a. All distances are reported in Angstroms ( $\text{\AA}$ ), and all angles and dihedrals are in degrees ( $^\circ$ ).

**Table S2: Parameters for bonds, angles and dihedrals for TFE**

<b>Parameter<sup>a</sup></b>	<b>Gauche</b>	<b>Antiperiplanar</b>
$r(\text{C1-F})$	1.35	1.35
$r(\text{C1-F2})$	1.34	1.34
$r(\text{C-C1})$	1.51	1.51
$r(\text{C-H1})$	1.10	1.10
$r(\text{C-H2})$	1.09	1.09
$r(\text{C-O})$	1.40	1.40
$r(\text{O-H})$	0.97	0.97
$\theta(\text{F-C1-F1})$	107.9	107.5
$\theta(\text{F1-C1-F2})$	108.1	107.2
$\theta(\text{F-C1-F2})$	107.4	107.2
$\theta(\text{F-C1-C})$	109.95	110.9
$\theta(\text{F1-C1-C})$	110.75	110.4
$\theta(\text{F2-C1-C})$	112.53	112.7
$\theta(\text{C1-C-H1})$	107.38	107.3
$\theta(\text{C1-C-H2})$	108.42	107.85
$\theta(\text{H1-C-O})$	113.23	112.99
$\theta(\text{H2-C-O})$	107.6	107.5
$\theta(\text{C-O-H})$	107.5	109.4
$\theta(\text{C1-C-O})$	112.41	112.4
$\phi(\text{F-C1-C-O})$	58.67	180
$\phi(\text{F1-C1-C-O})$	-61.33	60.76
$\phi(\text{F2-C1-C-O})$	177.79	-60.76
$\phi(\text{C2-C-O-H})$	62.74	180

a. All distances are reported in Angstroms (Å), and all angles and dihedrals are in degrees (°).

**Table S3: Parameters for bonds, angles and dihedrals for HFIP**

Parameter <sup>a</sup>	Gauche	Antiperiplanar
$r(\text{O-C})$	1.39	1.39
$r(\text{C-H1})$	1.10	1.09
$r(\text{C-C1})$	1.53	1.53
$r(\text{C-C2})$	1.53	1.53
$r(\text{F-C1})$	1.34	1.34
$r(\text{F1-C1})$	1.33	1.33
$r(\text{F2-C1})$	1.33	1.33
$r(\text{F3-C2})$	1.33	1.33
$r(\text{F4-C2})$	1.33	1.33
$r(\text{F5-C2})$	1.33	1.33
$\theta(\text{C-O-H})$	109.14	107.5
$\theta(\text{F-C1-C})$	107.83	110.2
$\theta(\text{H1-C-O})$	113.11	107.6
$\theta(\text{F1-C1-C})$	113.4	109.7
$\theta(\text{F2-C1-C})$	111.6	112.3
$\theta(\text{F-C1-F2})$	107.5	108.3
$\theta(\text{F-C1-F1})$	107.6	107.8
$\theta(\text{F1-C1-F2})$	108.6	108.3
$\theta(\text{C1-C-C2})$	113.7	113.3
$\theta(\text{C-C2-F3})$	110.7	110.2
$\theta(\text{C-C2-F4})$	112.4	109.7
$\theta(\text{C-C2-F5})$	109.3	112.4
$\theta(\text{F3-C2-F4})$	108.4	107.8
$\theta(\text{F3-C2-F5})$	107.8	108.2
$\theta(\text{F4-C2-F5})$	108	108.2
$\phi(\text{F-C1-C-O})$	-53.42	-179.9
$\phi(\text{F1-C1-C-O})$	65.52	59.5
$\phi(\text{F2-C1-C-O})$	-171.67	-59.03
$\phi(\text{F3-C2-C-O})$	-177.76	58.53
$\phi(\text{F4-C2-C-O})$	-57.02	-60.02
$\phi(\text{F5-C2-C-O})$	63.32	179.4
$\phi(\text{F-C1-C-C2})$	-173.31	-55.9
$\phi(\text{F1-C1-C-C2})$	-54.37	-176.5
$\phi(\text{F2-C1-C-C2})$	68.43	64.9
$\phi(\text{F3-C2-C-C1})$	-55.68	-65.5
$\phi(\text{F4-C2-C-C1})$	65.06	175.9
$\phi(\text{F5-C2-C-C1})$	-174.59	55.4
$\phi(\text{H-O-C-C1})$	58.65	62.9
$\phi(\text{H-O-C-C2})$	-177.83	-62.9
$\phi(\text{H1-C-O-H})$	-57.09	179.99

a. All distances are reported in Angstroms (Å), and all angles and dihedrals are in degrees (°).

**Table S4 : Summary of Charges for HFIP conformers**

Atom ID	Charges <sup>a</sup> (Gauche)	Charges <sup>a</sup> (antiperiplanar)
C	-0.179624	-0.220327
C	0.485153	0.401674
F	-0.151961	-0.133323
F	-0.151961	-0.133323
F	-0.151961	-0.133323
C	0.485153	0.401674
F	-0.151961	-0.133323
F	-0.151961	-0.133323
F	-0.151961	-0.133323
O	-0.389807	-0.443433
H	0.354832	0.4016661
H	0.156059	0.258687

a. Values for partial Charge are presented in atomic units

**Table S5 : Summary of charges for TFE conformers**

Atom ID	Charges <sup>a</sup> (Gauche)	Charges <sup>a</sup> (Antiperiplanar)
C	0.543783	0.565457
C	-0.093684	-0.077413
O	-0.455792	-0.502283
H	0.350009	0.395821
H	0.119999	0.094675
H	0.119999	0.094675
F	-0.194771	-0.190311
F	-0.194771	-0.190311
F	-0.194771	-0.190311

a. Values for partial Charge are presented in atomic units

**Table S6: Bonded parameters used for TFE**

Bond stretching parameters for CF <sub>3</sub> SF <sub>4</sub> -ethanol used in simulation		
Bond	$r_{eq}$ (Å)	$k_r$ (KJ/mol.nm <sup>2</sup> )
C-C	1.53	253632.1
C1-H1	1.10	281081.0
C1-O	1.40	262846.0
C-F	1.34	304431.0
O-H	0.96	309287.4

Bond angle parameters used in simulation

Angle	$\theta_{eq}$ / degree	$k_{\theta}$ (KJ/mol.rad <sup>2</sup> )
F-C-F	109.06	644.336
F-C-C1	109.24	418.400
C-C1-O	110.19	418.440
C-C1-H1	109.56	313.800
C1-O-H	109.5	460.250
O-C1-H1	110.26	292.880

Dihedral parameters used in simulations (Ryckaert-Bellemans type)				
Dihedral	$C_0$	$C_1$	$C_2$	$C_3$
H3-C-C1-F	0.65689	1.97066	0.0	-2.62755
O-C1-C-H3	1.40600	-1.04600	0.0	0.0
H-O-C1-H2	0.995779	2.98736	0.0	-3.98316
C-C1-O-H	0.26778	-9.41032	9.10021	0.0

Table S7. Bonded parameters used for HFIP

Bond stretching parameters for CF3SF4-ethanol used in simulation		
Bond	$r_{eq}$ (Å)	$k_r$ (KJ/mol.nm <sup>2</sup> )
C-C1	1.53	253632.1
C-H1	1.09	281081.0
C-O	1.40	262846.0
C1-F	1.34	304431.0
O-H	0.96	309287.4

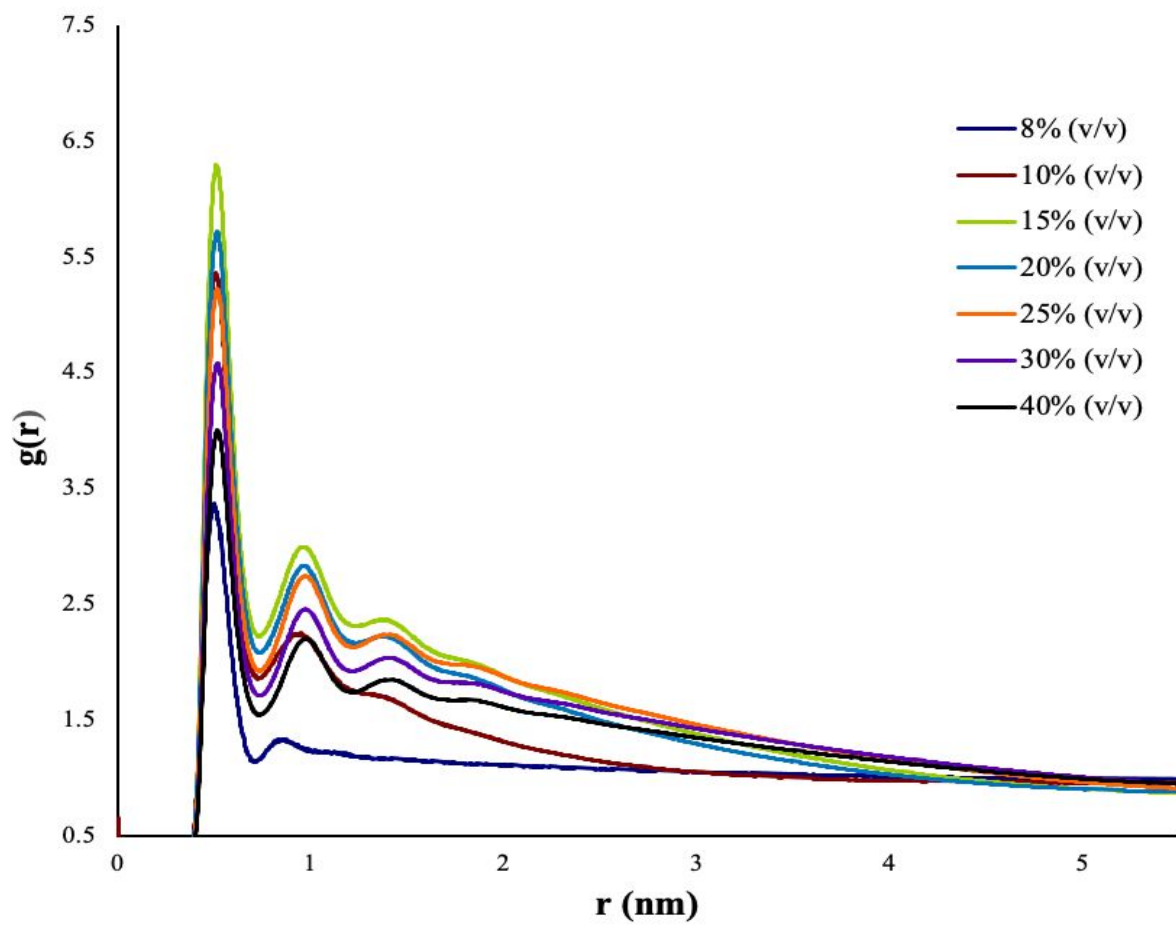
  

Bond angle parameters used in simulation		
Angle	$\theta_{eq}$ / degree	$k_{\theta}$ (KJ/mol.rad <sup>2</sup> )
F-C1-F1	109.06	644.336
F-C1-C	109.24	418.400
C1-C-O	109.50	418.440
C1-C-H1	109.56	313.800
C-O-H	109.50	460.250
O-C-H1	110.26	292.880

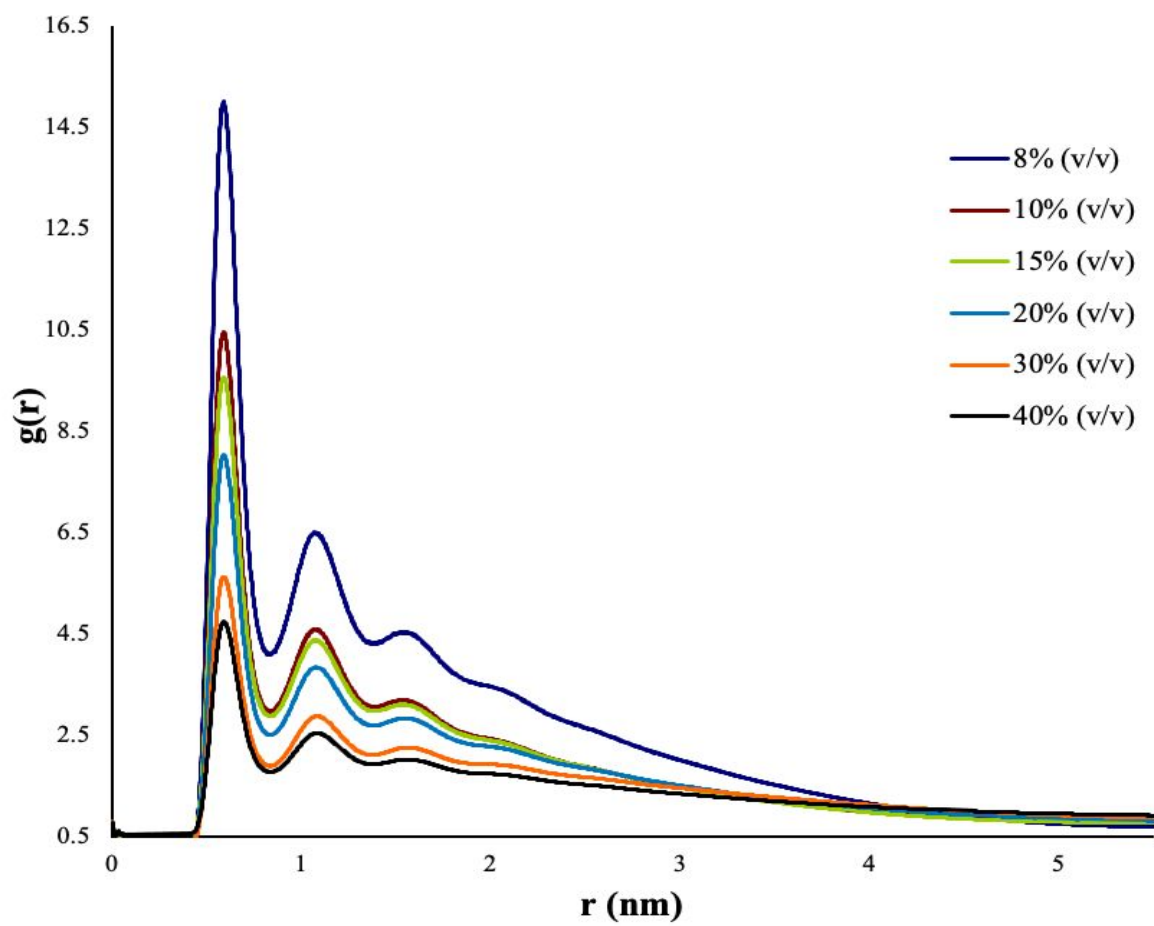
  

Dihedral parameters used in simulations (Ryckaert-Bellemans type)				
Dihedral	$C_0$	$C_1$	$C_2$	$C_3$
H1-C-C1-F	0.79496	-0.79496	0.0	0.0
C1-C-C2-F3	0.65084	1.95253	0.0	-2.60338
H-O-C-H1	0.69733	2.09200	0.0	-2.78933
C2-C-O-H	0.26778	-9.41032	9.10021	0.0

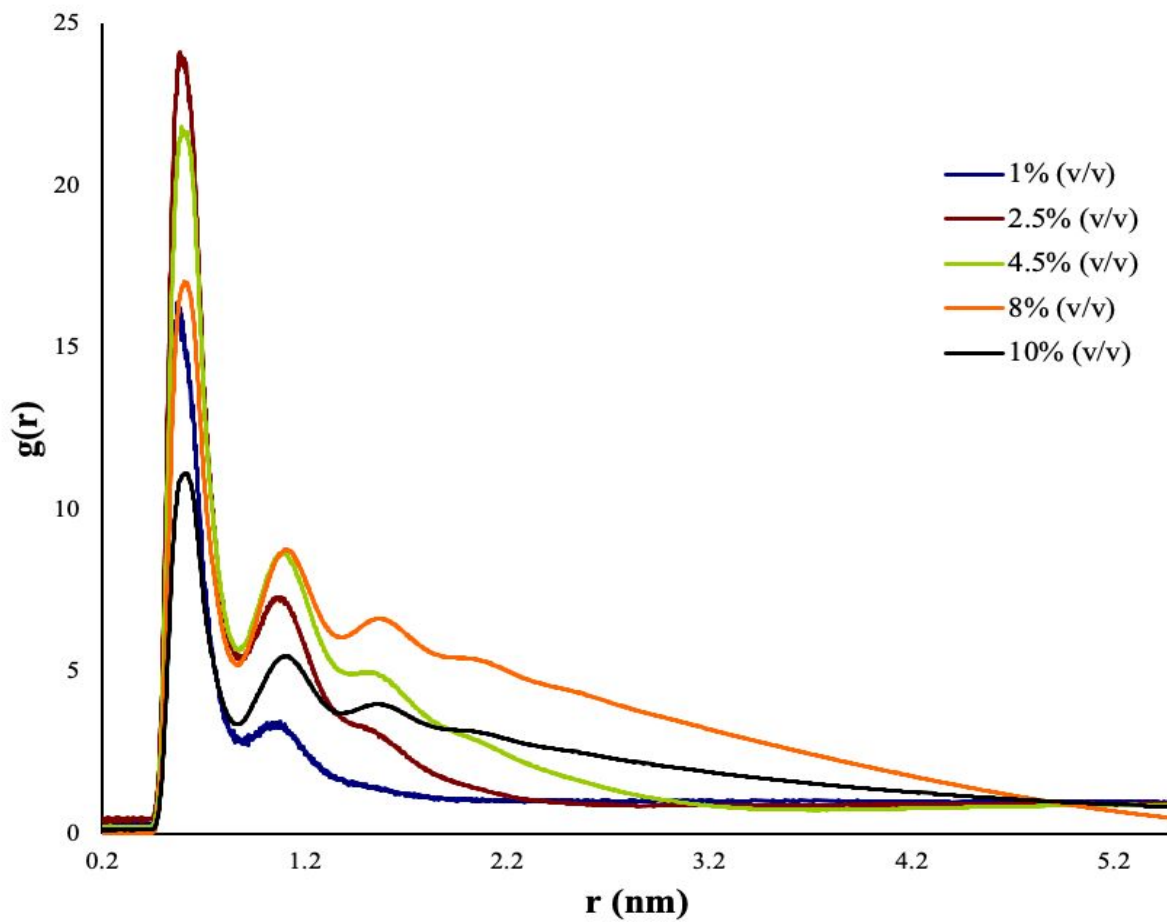




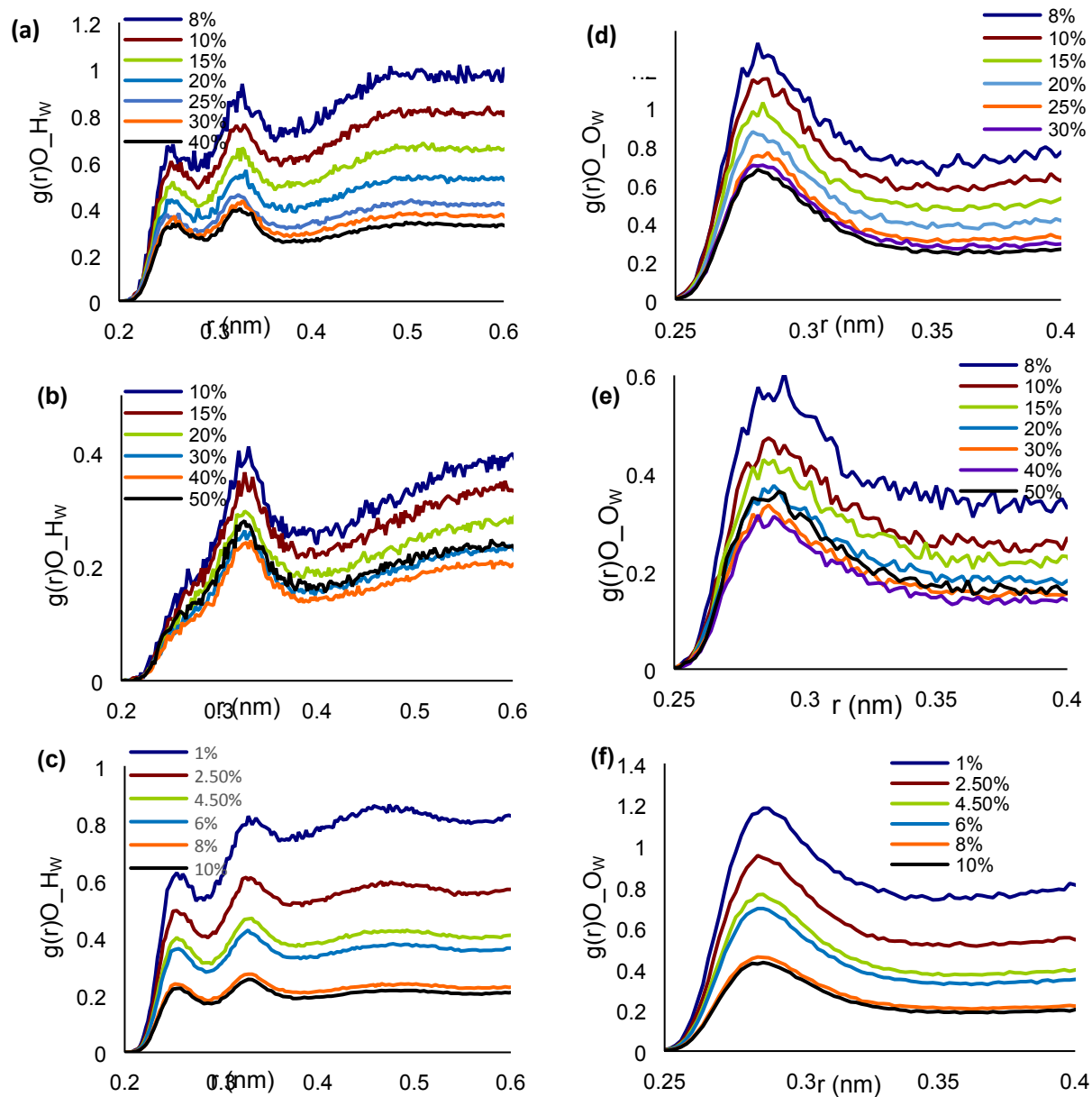
**Figure S1:** Radial Distribution Function  $g(r)$  for TFE-TFE



**Figure S2:** Radial Distribution Function  $g(r)$  for HFIP-HFIP



**Figure S3:** Radial Distribution Function  $g(r)$  for  $CF_3SF_4$ -ethanol- $CF_3SF_4$ -ethanol



**Figure S4.** RDFs for aqueous mixtures of fluorinated alcohols. Presented above are the RDFs for the  $O_{\text{TFE}}\text{-}H_{\text{Water}}$  distribution in TFE (a), HFIP (b) and  $\text{CF}_3\text{SF}_4\text{-ethanol}$  (c) and for the  $O_{\text{TFE}}\text{-}O_{\text{Water}}$  distribution in TFE (d) HFIP, (e) and  $\text{CF}_3\text{SF}_4\text{-ethanol}$  (f).

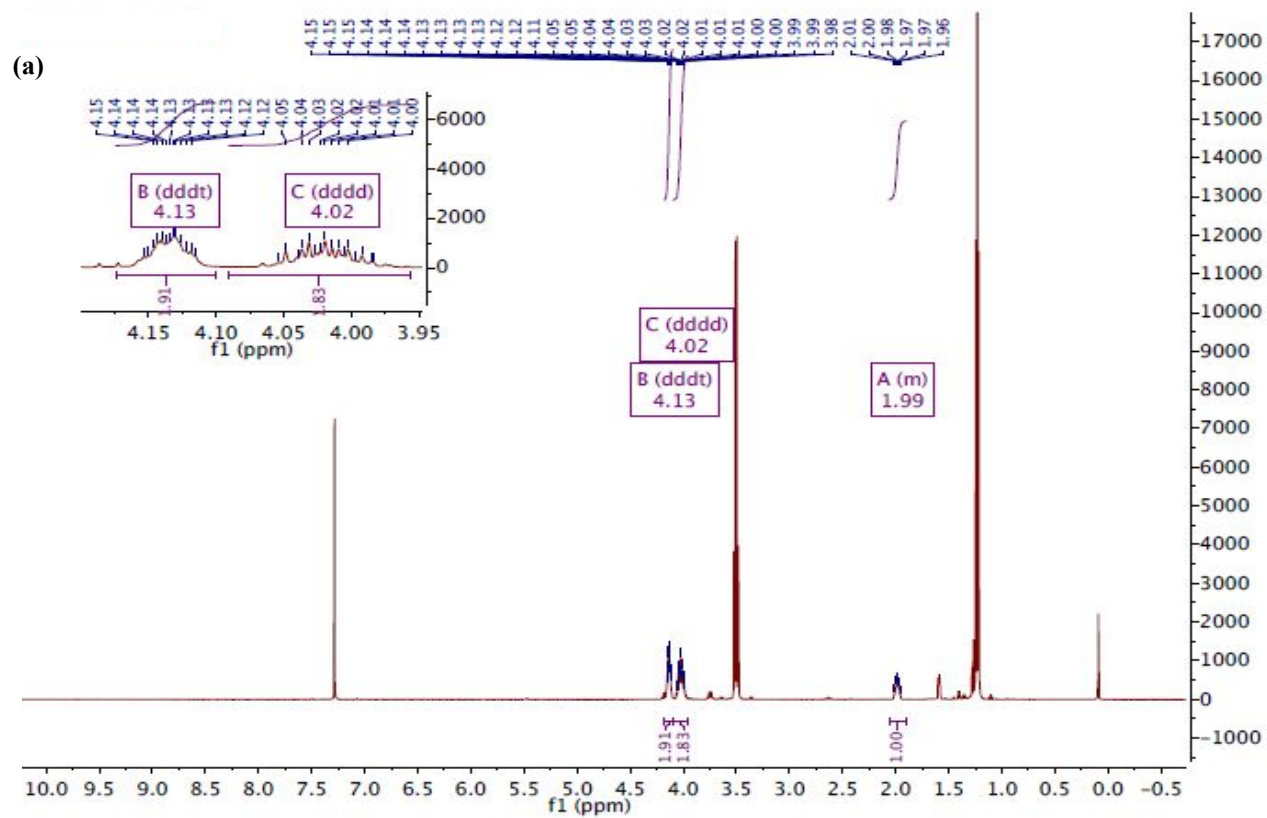
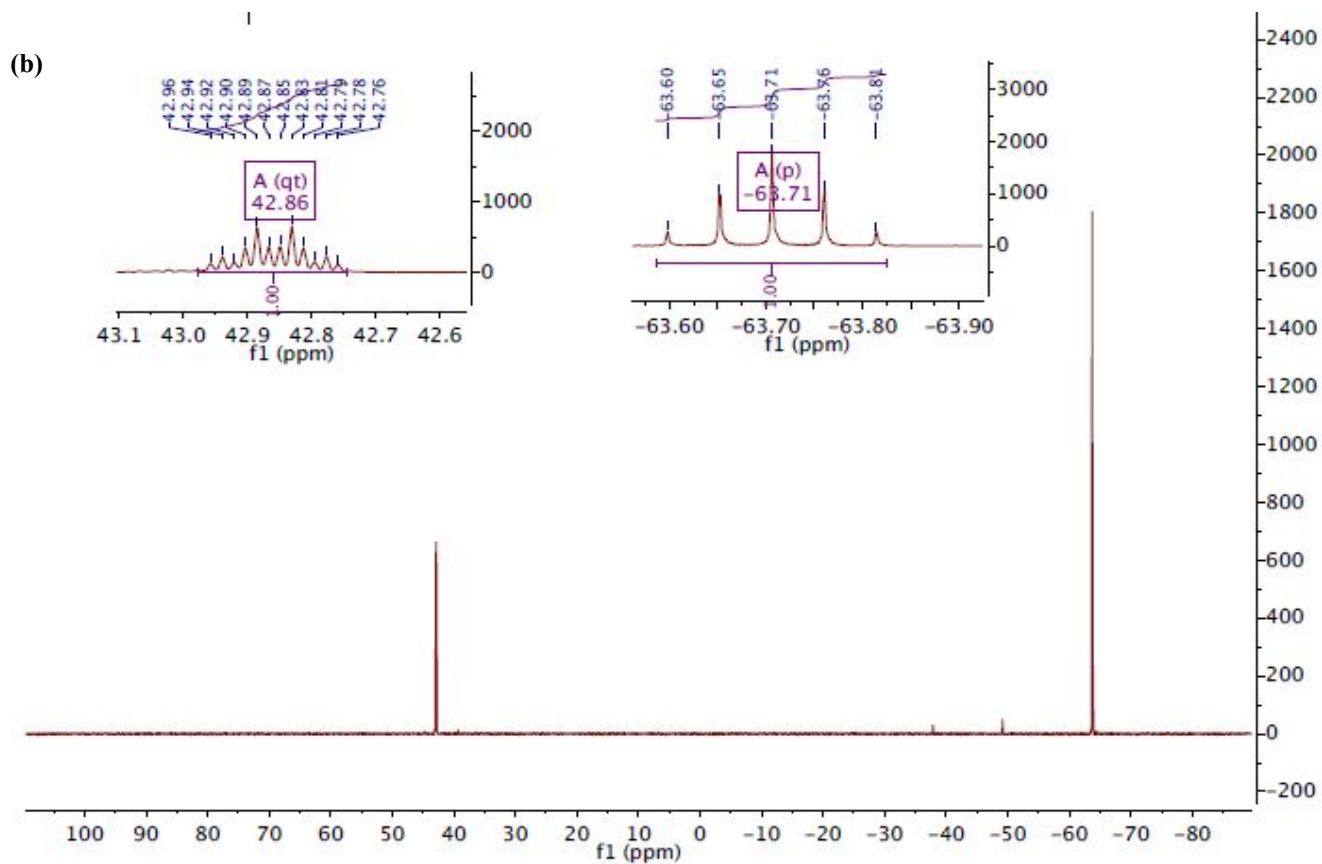
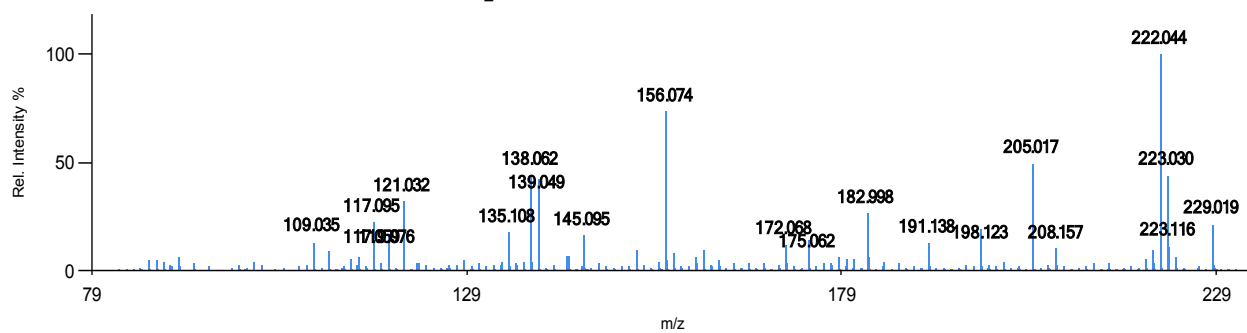


Figure S5 : (a)  $^1\text{H}$  Nmr spectrum of 2-(tetrafluoro(trifluoromethyl)- $\lambda^6$ -sulfanyl)ethan-1-ol



**Figure S5 :** (b)  $^{19}\text{F}$  Nmr spectra of 2-(tetrafluoro(trifluoromethyl)- $\lambda^6$ -sulfanyl)ethan-1-ol



**Figure S5** : (c) HRMS (DART-ESI, m/z, positive) spectrum of 2-(tetrafluoro(trifluoromethyl)-λ<sup>6</sup>-sulfanyl)ethan-1-ol