Ketone Incorporation Extends the Emission Properties of the Xanthene Scaffold Beyond 1000 nm

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Section S	1: Sup	plement	tary Fig	ures and	Tables

R Group	Structure	Year	Solvent	$\lambda_{abs}(nm)$	ϵ (M ⁻¹ cm ⁻¹)	$\lambda_{em}(nm)$	$\Phi_{ m F}$	Lit. ref
R1	Me ₂ N	2015	Aq. buffer	493	50,000	528	0.21	(1)
R2	Me ₂ N	2011	HEPES, 7.4	547	-	562	0.4	(2)
R3	Me Me Me NMe2	2011	EtOH	605	-	627	0.71	(3)
R4	Me_Ne_Ne_NMe_	2014	PBS	634	48,600	648	0.42	(4) (SI) (2)
R5	Me ₂ N Me ₂ N Me	2016	PBS 10mM pH 7.4 1% DMSO	700	71,000	722	0.11	(5)
R6	Me ₂ N S NMe ₂	2016	PBS pH 7.4 (0.25% CH ₃ CN)	703	39,000	734	0.067	(6)

R7	Me ₂ N Me ₂ N Me	2021	CH ₂ Cl ₂	855	28,400	911	0.013	This work
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Table S1: Structures, photophysical properties and literature references of previously reported compounds in Figure 1A.

R Group	Structure	Year	Solvent	$\lambda_{abs}(nm)$	ϵ $(M^{-1}cm^{-1})$	$\lambda_{em}(nm)$	$\Phi_{ m F}$	Lit ref
R1	Me ₂ N	2015	Aq. buffer	493	50,000	528	0.21	(1)
R2	Me ₂ N O Me	2011	PBS pH 7.4	549	72,000	570	0.31	(6)
R3	Me_Ne_NMe_ CO ₂ H	2011	Aq. buffer	606	121,000	626	0.52	(1)
R4	Me, Me Me ₂ N Si Me Me	2014	20mM PBS pH 7.4	647	110,000	661	0.31	(6)



Table S2: Structures, photophysical properties and literature references of dyes used in absorbance oscillator strengths (S0 \rightarrow S1^{*}) and (S1^{*} \rightarrow S0) vs. extinction coefficient (Figure 1E/ F).



Figure S1: Exo- to endo-switching of dye 11 (200 μ M) in acidic media.

Section S2: General Experimental and Materials

Unless stated otherwise, reactions were conducted in oven- dried glassware under an atmosphere of nitrogen or argon using anhydrous solvents (passed through activated alumina columns). All commercially obtained reagents were used as received. Flash column chromatography was performed using reversed phase (100 Å, 20-40-micron particle size, RediSep® Rf Gold® Reversed-phase C18 or C18Aq) and silica on a CombiFlash® Rf 200i (Teledyne Isco, Inc.) Highresolution LC/MS analyses were conducted on a ThermoFisher LTQ-Orbitrap-XL hybrid mass spectrometer system with an Ion MAX API electrospray ion source in negative ion mode. Analytical LC/MS was performed using a Shimadzu LCMS-2020 Single Quadrupole utilizing a Kinetex C18 column (100 Å, 2.6 µm, 2.1 x 50 mm) obtained from Phenomenex, Inc (Torrance, CA). Runs employed a gradient of 0-90% MeCN/0.1% aqueous formic acid over 4.5 min at a flow rate of 0.2 mL/min. ¹H NMR and ¹³C NMR spectra were recorded on Bruker spectrometers (at 400 or 500 MHz or at 100 or 125 MHz) and are reported relative to deuterated solvent signals. Data for ¹ H NMR spectra are reported as follows: chemical shift (δ ppm), multiplicity, coupling constant (Hz), and integration. Data for ¹³C NMR spectra are reported in terms of chemical shift. Absorbance curves were obtained on a PerkinElmer UV/Vis Photodiode Array Spectrophotometer Lambda 465 (Scan No. 30. Integration No. 1). Molar absorption coefficients (ɛ) were determined in PBS (50 mM, pH 7.4), CH₂Cl₂, CH₂Cl₂ supplemented with 5 % Trifluoroacetic Acid or Acetate Buffer (0.05M) at pH 3.6 using Beer's law, from plots of absorbance vs. concentration. Measurements were performed in 10 mm path length quartz cuvettes (Hellma 111-QS) that were maintained at room temperature. Fluorescence traces were recorded on aPTI QuantaMaster steady state spectrofluorometer operated by FelixGX 4.2.2 software, with 15 nm excitation and 20nm emission slit widths with a 1 nm step size and 1 s integration rate. Microplates were read on Synergy Mx microplate reader. pH values were determined using a Fisherbrand[™] accumet[™] AB150 pH Benchtop Meter. All statistical analyses were carried out by Graphpad Prism version 9.0 (Graphpad Software). See section S7 and S8 for computational chemistry methods and results.

Section S3 Synthetic Procedures

bis(2-bromo-5-(dimethylamino)phenyl)methanone (7)



To a solution of bis(2-bromo-5-(dimethylamino)phenyl)methanone (5.0 g, 0.012 mmol, 1.0 eq.) in DMF (19 mL, 0.12 M) at 0 °C under a N₂ atmosphere Cs₂CO₃ (46 g, 0.14 mol, 6.0 eq.) was added. The reaction mixture was sonicated and set to stir at room temperature before addition of iodomethane (7.3 mL, 0.12 mol, 5 eq.) *via* syringe. The suspension was then heated to 100 °C for 48 hours. The reaction was quenched with an excess of NaHCO₃ (sat.) and transferred to a separatory funnel with additional H₂O used to dissolve any salts present. The crude mixture was extracted with EtOAc × 2 with the organic phase then washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo* with water heating bath set at 60 °C. Bis(3-(dimethylamino)phenyl)methanone **1** was isolated at a brown oil (5.98 g) was used without further purification.

¹H NMR (400 MHz, CDCl₃) δ : 7.30 (t, *J* = 7.9 Hz, 2H), 7.23 – 7.18 (m, 2H), 7.10 (d, *J* = 7.6 Hz, 2H), 6.93 (dd, *J* = 8.3, 2.8, Hz, 2H), 2.99 (12H) ppm.¹³C NMR (100 MHz, CDCl₃) δ : 198.3, 150.5, 138.9, 128.7, 118.9, 116.3, 113.6, 40.7 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₁₇H₂₀N₂O 268.1576; Found 268.1571.

To the resultant crude material 1 (5.98 g, 0.0222 mol, 1.0 eq.) CH₂Cl₂ (225 mL, 0.1 M) was added. The reaction mixture was sonicated to ensure dissolution and set to stir at room temperature under a N₂ atmosphere for 10 minutes before cooling to 0 °C in an ice bath. *N*-bromosuccinimide (7.94 g, 0.0446 mol, 2.0 eq.) was then added portion wise and the reaction was followed by TLC (4:1, Hexanes: EtOAc, R_f: 0.45) with typical reaction times of 30 minutes. Once the reaction was judged to have gone to completion it was quenched with NaHCO₃ (sat.), extracted with EtOAc × 2. The organic phase washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The residue was purified by normal-phase column chromatography (120 g C18 column, 0 \rightarrow 10% hexanes /EtOAc) to afford bis(2-bromo-5 (dimethylamino)phenyl)methanone 7 (7.68 g, 77 %) as a bright yellow solid.

¹H NMR (400 MHz, CDCl₃) δ : 7.40 (d, *J* = 8.8 Hz, 2H), 6.81 (d, *J* = 3.2 Hz, 2H), 6.68 (dd, *J* = 8.9, 3.2 Hz, 2H), 2.92 (s, 12H) ppm.¹³C NMR (100 MHz, CDCl₃) δ : 196.9, 149.5, 139.8, 134.3, 116.4, 115.1, 107.1, 40.6 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₁₇H₁₈Br₂N₂O 425.9765; Found 425.9766.

bis(2-bromo-5-(dimethylamino)phenyl)methanol (12)



To bis(2-bromo-5-(dimethylamino)phenyl)methanone 7 (2.55 g, 5.89 mmol, 1.0 eq.) in a 500 mL round bottom flask under a N₂ atmosphere (N₂ balloon) THF (100 mL) was added and set to stir at room temperature. Once fully dissolved the solution was cooled to 0 °C in an ice bath and MeOH (17 mL, 0.05 M) was added. After 15 minutes NaBH₄ (2.3 g, 0.059 mol, 10 eq.) was added portion wise. The reaction mixture was kept at 0 °C for 1 hour and allowed to gradually warm to room temperature. The reaction was followed by TLC (4:1, Hexanes: EtOAc, R_f: 0.23) and once judged to have gone to completion the reaction mixture was cooled in an ice bath and quenched carefully using NH₄Cl (sat.). Additional H₂O was used to dissolve any salts present and the reaction mixture was extracted with EtOAc × 2. The organic phase then washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. Bis(2-bromo-5-(dimethylamino)phenyl)methanol **12** (2.45g, 96 %) was isolated as a light yellow solid and used without further purification.

¹H NMR (400 MHz, CDCl₃) δ 7.36 (d, J = 8.8 Hz, 2H), 6.73 (d, J = 3.2 Hz, 2H), 6.53 (dd, J = 8.8, 3.2 Hz, 2H), 6.28 (s, 1H), 2.87 (s, 12H) ppm.¹³C NMR (100 MHz, CDCl₃) δ 150.1, 141.20, 133.0, 113.5, 113.0, 110.1, 74.7, 40.7 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₁₇H₂₀Br₂N₂O 427.9922; Found 427.9920.

3,3'-((methoxymethoxy)methylene)bis(4-bromo-N,N-dimethylaniline) (2a)



To a stirred solution of bis(2-bromo-5-(dimethylamino)phenyl)methanol **12** (500 mg, 1.17 mmol, 1.0 eq.) in DMF (4.0 mL, 0.29 M) under an Ar atmosphere at 0 °C, sodium hydride (84 mg, 3.5 mol, 3.0 eq.) was added and the reaction was stirred at room temperature for 30 minutes. To the crude reaction mixture at 0 °C chloromethyl methyl ether (130 μ L, 1.8 mmol, 1.5 eq.) was added *via* syringe and the reaction set to stir at room temperature for 3.5 hours. Once the reaction was judged to have gone to completion (TLC, 4:1, Hexanes: EtOAc, R_f: 0.4) the reaction mixture was cooled in an ice bath and quenched using NH₄Cl (sat.). The reaction mixture was extracted with EtOAc × 2 with the organic phase then washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The residue was purified by normal-phase column chromatography (12 g C18 gold column, 0 \rightarrow 10% hexanes /EtOAc) to afford 3,3'-(methoxymethylene)bis(4-bromo-*N*,*N*-dimethylaniline) **2a** (460 mg, 83 %) as an off-white solid.

¹H NMR (400 MHz, CDCl₃) δ 7.35 (d, *J* = 8.8 Hz, 2H), 6.80 (d, *J* = 3.1 Hz, 2H), 6.53 (dd, *J* = 8.8, 3.2 Hz, 2H), 6.26 (s, 1H), 4.80 (s, 2H), 3.46 (s, 3H), 2.88 (s, 12H); ¹³C NMR (100 MHz, CDCl₃) δ 149.9, 139.7, 133.2, 113.7, 113.6, 110.8, 95.8, 78.5, 56.8, 40.7 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₁₉H₂₄Br₂N₂O₂ 472.0184; Found 472.0187.

3,3'-(methoxymethylene)bis(4-bromo-*N*,*N*-dimethylaniline) (2b)



To a stirred solution of bis(2-bromo-5-(dimethylamino)phenyl)methanol **12b** (500 mg, 1.17 mmol, 1.0 eq.) in DMF (4.0 mL, 0.29 M) under an Ar atmosphere at 0 °C, sodium hydride (84 mg, 3.5 mol, 3.0 eq.) was added and the reaction was stirred at room temperature for 30 minutes. To the crude reaction mixture at 0 °C iodomethane (110 μ L, 1.8 mmol, 1.5 eq.) was added *via* syringe and the reaction set to stir at room temperature for 3.5 hours. Once the reaction was judged to have gone to completion (TLC, 4:1, Hexanes: EtOAc, R_f: 0.5) the reaction mixture was cooled in an ice bath and quenched using NH₄Cl (sat.). The reaction mixture was extracted with EtOAc × 2 with the organic phase then washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The residue was purified by normal-phase column chromatography (12 g C18 gold column, 0 \rightarrow 10% hexanes /EtOAc) to afford 3,3'-(methoxymethylene)bis(4-bromo-*N*,*N*-dimethylaniline) **2b** (400 mg, 77 %) as light brown solid.

¹H NMR (400 MHz, CDCl₃) δ : 7.37 (d, *J* = 8.8 Hz, 2H), 6.69 (d, *J* = 3.1 Hz, 2H), 6.53 (dd, *J* = 8.8, 3.1 Hz, 2H), 5.79 (s, 1H), 3.52 (s, 3H), 2.87 (s, 12H); ¹³C NMR (100 MHz, CDCl₃) δ : 150.0, 139.35, 133.1, 113.6, 113.3, 111.3, 83.5, 58.3, 40.7 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₁₈H₂₂Br₂N₂O₂ 442.0078; Found 442.0080.

3,3'-(((tert-butyldimethylsilyl)oxy)methylene)bis(4-bromo-N,N-dimethylaniline) (2c)



To bis(2-bromo-5-(dimethylamino)phenyl)methanol **12** (2.5 g, 0.0058 mol, 1.0 eq.), imidazole (1.19 g, 0.0175 mol, 3.0 eq.) and *tert*-butyldimethylsilyl chloride (1.76 g, 0.00117 mol, 2.0 eq.) in an Ar flushed round-bottom flash DMF (25 mL, 0.23 M) was added. The reaction was set to stir at room temperature before heating to 70 °C overnight. The cooled reaction mixture was suspended in 2:1 EtOAc/ NaHCO₃ (sat.) and extracted with EtOAc. The aqueous phase was re-extracted with additional EtOAc and combined organic phase were washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The residue was purified by normal-phase column chromatography (40 g silica gold column, $0 \rightarrow 10\%$ hexanes /EtOAc) and dried under hi-vaccum to afford 3,3'-(((*tert*-butyldimethylsilyl)oxy)methylene)bis(4-bromo-*N*,*N*-dimethylaniline) **2c** (2.7 g, 84 %) as an off-white solid.

¹H NMR (400 MHz, DMSO-d₆) δ : 7.30 (d, *J* = 8.8 Hz, 2H), 6.83 (d, *J* = 3.2 Hz, 2H), 6.59 (dd, *J* = 8.9, 3.2 Hz, 2H), 6.15 (s, 1H), 2.83 (s, 12H), 0.87 (s, 9H), 0.02 (s, 6H); ¹³C NMR (100 MHz, DMSO-d₆) δ : 149.5, 141.6, 132.6, 113.6, 112.9, 108.3, 74.5, 39.9, 25.7, 17.8, – 4.40 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₂₃H₃₄Br₂N₂O₂Si 542.0787; Found 542.0790.

9-(methoxymethoxy)-N²,N²,N⁷,N⁷-tetramethyl-10-(o-tolyl)anthracene-2,7-diamine (3a)



To 9-(methoxymethoxy)- N^2 , N^2 , N^7 , N^7 -tetramethyl-10-(o-tolyl)anthracene-2,7-diamine 2a (200 mg, 0.423 mmol, 1.0 eq.) in a flame-dried crimped 25 mL microwave vial under an Ar atmosphere, THF (10 mL, 0.042 M) was added. The solution was stirred at room temperature before cooling to -78 °C in a dry ice/ acetone bath. The solution was then treated with *tert*-Butyllithium solution (1.1 mL, 1.9 mmol, 4.4 eq.) and allowed stir at - 78 °C for 0.5 h. A separately prepared solution of methyl 2-methylbenzoate (130 µL, 0.93 mmol, 2.2 eq.) in THF (5.0 mL, 0.17 M) under an Ar atmosphere was then added dropwise to the -78 °C reaction mixture. The reaction mixture was allowed stir for 1 hour at -78 °C before removal from the cold bath, covering with aluminum foil and allowing to gradually equilibrate to room temperature overnight for a total reaction time of 16 hours. The reaction mixture was then quenched at -78 °C with an excess of NH₄Cl (sat.) and transferred to a separatory funnel with additional H₂O used to dissolve any salts present. The crude mixture was extracted with EtOAc \times 2 with the organic phase then washed with brine \times 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The yellow residue was purified by normal-phase column chromatography (12 g silica gold column, $0 \rightarrow 10\%$ hexanes /EtOAc) to afford 9-(methoxymethoxy)- N^2 , N^2 , N^7 , N^7 -tetramethyl-10-(o-tolyl)anthracene-2,7-diamine **3a** (101 mg, 58 %) as a bright yellow solid.

¹H NMR (400 MHz, DMSO-d₆) δ : 7.47 – 7.41 (m, 2H), 7.40 – 7.34 (m, 1H), 7.19 – 7.06 (m, 7H), 5.31 (s, 2H), 3.74 (s, 3H), 3.03 (s, 12H), 1.78 (s, 3H); ¹³C NMR (100 MHz, DMSO-d₆) δ 147.5, 143.3, 138.0, 137.0, 131.6, 130.9, 129.9, 127.7, 127.0, 126.3, 125.9, 122.44, 116.7, 99.2, 97.7, 57.2, 40.2, 19.3 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₂₇H₃₀N₂O₂ 414.2307; Found 414.2302.

9-methoxy-N²,N²,N⁷,N⁷-tetramethyl-10-(o-tolyl)anthracene-2,7-diamine (3b)



To 3.3'-(methoxymethylene)bis(4-bromo-N,N-dimethylaniline) **2b** (200 mg, 0.45 mmol, 1.0 eq.) in a flame-dried crimped 25mL microwave vial under an Ar atmosphere, THF (10 mL, 0.045 M) was added. The solution was stirred at room temperature before cooling to -78 °C in a dry ice/ acetone bath. The solution was then treated with tert-Butyllithium solution (1.2 mL, 2.0 mmol, 4.4 eq.) and allowed stir at - 78 °C for 0.5 h. A separately prepared solution of methyl 2methylbenzoate (140 µL, 1.0 mmol, 2.2 eq.) in THF (5 mL, 0.2 M) under an Ar atmosphere was then added dropwise to the -78 °C reaction mixture. The reaction mixture was allowed stir for 1 hour at -78 °C before removal from the cold bath, covering with aluminum foil and allowing to gradually equilibrate to room temperature overnight for a total reaction time of 16 hours. The reaction mixture was then quenched at -78 °C with an excess of NH₄Cl(sat.), and transferred to a separatory funnel with additional H₂O used to dissolve any salts present. The crude mixture was extracted with EtOAc \times 2 with the organic phase then washed with brine \times 2, dried over Na₂SO₄, filtered and evaporated to dryness in vacuo. The yellow residue was purified by normal-phase column chromatography (12 g silica gold column, $0 \rightarrow 10\%$ hexanes /EtOAc) to afford 9-methoxy- N^2 , N^2 , N^7 , N^7 -tetramethyl-10-(o-tolyl)anthracene-2,7-diamine (85 mg, 54 %) **3b** as a bright yellow solid.

¹H NMR (400 MHz, CDCl₃) δ 7.43 – 7.39 (m, 2H), 7.37 – 7.31 (m, 3H), 7.27 – 7.22 (m, 3H), 7.04 (dd, *J* = 9.5, 2.6 Hz, 2H), 4.17 (s, 3H), 3.11 (s, 12H), 1.91 (s, 3H); ¹³C NMR (100 MHz, DMSO-d₆) δ : δ 147.8, 146.8, 138.9, 138.2, 132.4, 131.6, 129.9, 128.2, 127.6, 126.2, 125.7, 123.7, 116.6, 98.3, 60.8, 41.0, 19.9 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₂₆H₂₈N₂O₂ 384.2202; Found 384.2196.

N-(7-(dimethylamino)-9-oxo-10-(*o*-tolyl)anthracen-2(9H)-ylidene)-*N*-methylmethanaminium (KR-1)



A solution of **3a** (10 mg, 0.024 mmol, 1.0 eq.) in CH₂Cl₂ (2.4 mL, 0.01 M) or **3b** (10 mg, 0.026 mmol, 1.0 eq.) in CH₂Cl₂ (2.6 mL, 0.01 M) in an oven-dried crimp vial under an Ar atmosphere were cooled to 0 °C in an ice bath. The solution was then treated with either trifluoroacetic acid (260 μ L) or BBr₃ (0.52 mL, 0.52 mmol, 20 eq.). The resulting suspension was sonicated and set to stir at room temperature in the absence of light overnight. The reaction mixture was reduced to dryness under a stream on N₂ before resuspending in CH₂Cl₂ (3 mL) with this process repeated a further two times. The concentrated product was suspended in CH₂Cl₂ before adding NaHCO₃ (sat.). The crude mixture was extracted with CH₂Cl₂ × 2 with the organic phase then washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The dark brown residue was then purified by normal phase chromatography (4 g silica column, 0 \rightarrow 15% MeOH/ CH₂Cl₂) to afford **KR-1** (6.2 mg, 70 %) as a brown black solid. Yield from **3b** = 5.9 mg, 62 %.

¹H NMR (400 MHz, CDCl₃) δ : 7.75 (d, J = 2.5 Hz, 2H), 7.45 (t, J = 7.5 Hz, 1H), 7.40 – 7.31 (m, 2H), 7.11 (d, J = 7.4 Hz, 1H), 7.03 – 6.88 (m, 4H), 3.52 (s, 12H), 2.11 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ : 183.4, 164.8, 156.7, 138.5, 136.0, 135.0, 133.7, 130.9, 130.0, 129.0, 126.2, 124.0, 116.9, 116.6, 42.2, 19.9 ppm. HRMS (ESI) m/z: [M]⁺ Calcd. for C₂₅H₂₅N₂O⁺ 369.1961; Found 369.1963.

2-(3,6-bis(dimethylamino)-10-(methoxymethoxy)anthracen-9-yl)benzoic acid (4)



To 9-(methoxymethoxy)- N^2 , N^7 , N^7 -tetramethyl-10-(o-tolyl)anthracene-2,7-diamine **2c** (300) mg, 0.64 mmol, 1.0 eq.) in a flame-dried crimped 25 mL microwave vial under an Ar atmosphere, THF (15 mL, 0.042 M) was added. The solution was stirred at room temperature before cooling to -78 °C in a dry ice/ acetone bath. The solution was then treated with *tert*-Butyllithium solution (1.64 mL, 2.79 mmol, 4.4 eq.) and allowed stir at $-78 \text{ }^{\circ}\text{C}$ for 0.5 h. A separately prepared solution of phthalic anhydride (207 mg, 1.40 mmol, 2.2 eq.) in THF (5 mL, 0.28 M) under an Ar atmosphere was then added dropwise to the - 78 °C reaction mixture. The reaction mixture was allowed stir for 1 hour at -78 °C before removal from the cold bath, covering with aluminum foil and allowing to gradually equilibrate to room temperature overnight for a total reaction time of 16 hours. The reaction mixture was then quenched at -78 °C with an excess of NH₄Cl(sat.) and transferred to a separatory funnel with additional H₂O used to dissolve any salts present. The crude mixture was extracted with EtOAc \times 2 with the organic phase then washed with brine \times 2, dried over Na₂SO₄, filtered and evaporated to dryness in vacuo. The yellow residue was purified by normal-phase column chromatography (24 g silica gold column, $0 \rightarrow 10\%$ CH₂Cl₂ /MeOH) to afford 2-(3,6bis(dimethylamino)-10-(methoxymethoxy)anthracen-9-yl)benzoic acid 4 (282 mg, 53 %) as a bright yellow solid.

¹H NMR (400 MHz, DMSO-d₆) δ 8.05 (dd, J = 7.7, 1.6 Hz, 1H), 7.70 (td, J = 7.5, 1.6 Hz, 1H), 7.63 (td, J = 7.6, 1.4 Hz, 1H), 7.27 (dd, J = 7.5, 1.4 Hz, 1H), 7.18 – 7.00 (m, 6H), 5.31 (s, 2H), 3.75 (s, 3H), 3.02 (s, 12H); ¹³C NMR (100 MHz, DMSO-d₆) δ 168.0, 147.4, 143.3, 138.9, 133.7, 132.6, 132.4, 131.3, 129.8, 127.8, 127.4, 126.2, 122.8, 116.4, 99.2, 97.6, 57.3, 40.3 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₂₇H₂₈N₂O₄ 444.2049; Found 444.2044.

3,6-bis(dimethylamino)-3'H,10H-spiro[anthracene-9,1'-isobenzofuran]-3',10-dione (KR-2)



A solution of 4 (50 mg, 0.11 mmol, 1.0 eq.) in CH₂Cl₂ (11 mL, 0.01 M) under an Ar atmosphere was cooled to 0 °C in an ice bath. The solution was then treated with trifluoroacetic acid (1.2 mL). The resulting suspension was sonicated and set to stir at room temperature in the absence of light overnight. The reaction mixture was reduced to dryness under a stream on N₂ before resuspending in CH₂Cl₂ (10 mL) with this process repeated a further two times. The crude material was then suspended in CH₂Cl₂ before adding NaHCO₃ (sat.). The crude mixture was extracted with CH₂Cl₂ × 2 with the organic phase then washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The dark brown residue was then purified by normal phase chromatography (12 g silica column, $0 \rightarrow 5\%$ MeOH/ CH₂Cl₂) to afford **KR-2** (36 mg, 80 %) as a yellow-orange solid.

¹H NMR (500 MHz, CDCl₃) δ : 8.03 – 7.95 (m, 1H), 7.60 (d, J = 2.9 Hz, 2H), 7.55 – 7.44 (m, 2H), 6.96 – 6.87 (m, 3H), 6.83 (dd, J = 8.8, 2.9 Hz, 2H), 3.02 (s, 12H); ¹³C NMR (125 MHz, CDCl₃) δ ; 184.1, 171.2, 154.7, 150.7, 135.1, 131.9, 129.2, 128.0, 126.7, 125.6, 125.2, 122.8, 117.8, 109.4, 84.5, 40.4 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₂₅H₂₂N₂O₃ 398.1630; Found 398.1633 ppm.

3-(2-(3,6-bis(dimethylamino)-10-(methoxymethoxy)anthracen-9-yl)-*N*-(3-(*tert*-butoxy)-3-oxopropyl)benzamido)propane-1-sulfonate (9)



A solution of 4 (50 mg, 0.11 mmol, 1.0 eq.), HATU (130 mg, 0.34 mmol, 3.0 eq.) and *N*,*N*-diisopropylethylamine (59 µL, 0.33 mmol, 3.0 eq.) in DMF (3 mL) under an Ar atmosphere was set to stir at room temperature. To the solution was then added 3-((3-(*tert*-butoxy)-3-oxopropyl)amino)propane-1-sulfonic acid (45 mg, 0.17 mmol, 1.5 eq.) and the reaction mixture was sonicated for 5 minutes before stirring at room temperature for 1 hour. The reaction mixture was then added to a separatory funnel containing of 4:1, EtOAc: Hexanes and NH₄Cl (sat.). Once extracted the separatory funnel was allowed to rest for 15-20 minutes to aid separation. This process was repeated until minimal crude material was detected in the organic layer by TLC. The organic layers were then combined and washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The residue was purified by normal-phase column chromatography (12 g silica gold column, $0 \rightarrow 10\%$ CH₂Cl₂ /MeOH) to afford 3-(2-(3,6-bis(dimethylamino)-10-(methoxymethoxy)anthracen-9-yl)-*N*-(3-(*tert*-butoxy)-3-oxopropyl)benzamido)propane-1-sulfonate **9** (68 mg, 87 %) as a bright yellow solid.

¹H NMR (500 MHz, DMSO- d_6) δ 7.68 – 7.56 (m, 2H), 7.55 – 7.41 (m, 1H), 7.41 – 7.35 (m, 1H), 7.33 – 7.19 (m, 2H), 7.16 – 6.99 (m, 4H), 5.28 (s, 2H), 3.74 (s, 3H), 3.03 (s, 12H), 2.92 (t, J = 7.6 Hz, 1H), 2.16 (br. s, 2H), 1.53 (br. s, 2H), 1.32 (d, J = 42.6 Hz, 9H), 0.97 (t, J = 8.1 Hz, 2H); ¹³C NMR (125 MHz, DMSO- d_6) δ 170.1, 168.8, 147.4, 143.5, 138.2, 135.7, 132.2, 132.0, 129.7, 128.6, 127.8, 126.5, 126.1, 116.0, 99.2, 97.3, 79.7, 57.3, 48.4, 44.4, 40.3, 34.0, 30.8, 27.6, 24.3 ppm.HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₃₇H₄₇N₃O₈S 693.3084; Found 693.3081.

3-(*N*-(2-carboxyethyl)-2-(6-(dimethylamino)-3-(dimethyliminio)-10-oxo-3,10dihydroanthracen-9-yl)benzamido)propane-1-sulfonate (KR-3)



To solution of **9** (66 mg, 0.096 mmol, 1.0 eq.) in CH₂Cl₂ (9.5 mL, 0.01 M) under an Ar atmosphere was added DDQ (24 mg, 0.1057 mmol, 1.1 eq.). The reaction mixture was stirred at room temperature for 1 hour before addition of NaHCO₃ (sat.). The crude mixture was extracted with CH₂Cl₂ × 2 with the organic phase then washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The residue was then purified by normal phase chromatography (12 g silica column, $0 \rightarrow 10\%$ CH₂Cl₂ /MeOH) to afford 3-(*N*-(3-(*tert*-butoxy)-3-oxopropyl)-2-(6-(dimethylamino)-3-(dimethyliminio)-10-oxo-3,10-dihydroanthracen-9-yl)benzamido)propane-1-sulfonate **12** (40.8 mg, 91 %) as a dark brown solid.

To a solution of **12** (110 mg, 0.17 mmol, 1.0 eq.) in CH₂Cl₂ (15.3 mL, 0.01 M) under an Ar atmosphere was cooled to 0 °C in an ice bath. The solution was then treated with trifluoroacetic acid (1.7 mL). The resulting solution was sonicated and set to stir at room temperature in the absence of light for 16 hours. The reaction mixture was reduced to dryness under a stream on N₂ before resuspending in CH₂Cl₂ (15 mL) with this process repeated a further two times before drying under high vacuum to afford **KR-3** (138 mg) as a crude TFA salt. The residue could be further purified by reverse-phase chromatography (C18 20 ×150mm, dimensions: 21 mm x 250 mm 10 µm, 0 \rightarrow 50 % H₂O /CH₃CN) to afford **KR-3** (68 mg, 63 %) as a brown-black solid.

¹H NMR (400 MHz, MeOD/ TFA-*d*) δ 7.81 (dd, J = 9.5, 2.8 Hz, 1H), 7.76 – 7.53 (m, 3H), 7.46 (dd, J = 14.8, 6.0, 2.8 Hz, 1H), 7.08 (dd, J = 19.4, 9.4 Hz, 2H), 6.89 (dd, J = 20.8, 9.3 Hz, 2H), 3.53 – 3.47 (m, 1H), 3.41 (s, 12H), 2.65 (t, J = 7.2 Hz, 1H), 2.54 (t, J = 7.4 Hz, 1H), 2.45 – 2.31 (m, 1H), 2.14 (t, J = 6.5 Hz, 1H), 2.06 – 1.84 (m, 1H), 1.70 – 1.46 (m, 1H) ppm. HRMS (ESI) m/z: [M]⁺ Calcd. for C₃₁H₃₃N₃O₇S⁺ 592.2112; Found 592.2256.

Methyl 2-(10-((tert-butyldimethylsilyl)oxy)-3,6-bis(dimethylamino)anthracen-9-yl)acetate (10)



To 3,3'-(((*tert*-butyldimethylsilyl)oxy)methylene)bis(4-bromo-*N*,*N*-dimethylaniline) (50 mg, 0.092 mol, 1.0 eq.), Pd(PPh₃)₄ (5.3 mg, 0.0046 mol, 0.05 eq.) and triethylamine (65 µL, 0.46 mmol, 5.0 eq.) in a flame dried, Ar flushed 1 dram vial was added DMF (190 µL, 0.5 M). The reaction was set to stir at room temperature for 5 minutes before the addition of methyl acrylate (12.5 µL, 0.0138 mmol, 1.5 eq.). The vial was flushed with Ar, sealed carefully with parafilm and black electrical tape and heated to 150 °C overnight in the absence of light. Once at room temperature the crude reaction mixture was diluted with EtOAc and passed through a celite plug eluting with additional EtOAc. The eluant was transferred to a separatory funnel and washed with NaHCO₃ (sat.). The aqueous phase was extracted with EtOAc × 2 with the organic phase then washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The yellow residue was purified by normal-phase column chromatography (4 g silica, 0 → 10% hexanes /EtOAc) to afford methyl 2-(10-((tert-butyldimethylsilyl)oxy)-3,6-bis(dimethylamino)anthracen-9-yl)acetate (35 mg, 40 %) **10** as a bright yellow solid.

¹H NMR (400 MHz, DMSO-d₆) δ : 8.01 (d, *J* = 9.6 Hz, 2H), 7.26 (dd, *J* = 9.6, 2.7 Hz, 2H), 6.98 (d, *J* = 2.6 Hz, 2H), 4.48 (s, 2H), 3.58 (s, 3H), 3.03 (s, 12H), 1.13 (s, 9H), 0.21 (s, 6H); ¹³C NMR (100 MHz, DMSO-d₆) δ 171.9, 146.5, 141.1, 125.7, 124.8, 123.4, 119.5, 117.1, 99.2, 51.7, 40.4, 32.9, 25.9, 18.4, - 2.6 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₂₇H₃₈N₂O₃Si 466.2652; Found 466.2645.

Methyl 2-(3,6-bis(dimethylamino)-10-oxoanthracen-9(10H)-ylidene)acetate (11)



To 10 (10 mg, 0.021 mmol, 1.0 eq.) in a solution in 2:1 ACN/ H₂O (3 mL) formic acid (150 μ L) was added. The reaction was set to stir at room temperature overnight. The solution was dried under vacuum to give methyl 2-(3,6-bis(dimethylamino)-10-oxoanthracen-9(10*H*)-ylidene)acetate 11 (7.1 mg, 95%) as a yellow-orange solid.

¹H NMR (400 MHz, CD₃CN) δ 7.81 (d, *J* = 8.9 Hz, 1H), 7.77 (d, *J* = 8.9 Hz, 1H), 7.38 (d, *J* = 2.9 Hz, 2H), 7.01 (dd, *J* = 8.9, 3.0 Hz, 1H), 6.93 (dd, *J* = 8.9, 3.0 Hz, 1H), 6.46 (s, 1H), 3.73 (s, 3H), 3.05 (d, *J* = 8.0 Hz, 12H); ¹³C NMR (125 MHz, CDCl₃) δ 185.8, 168.4, 151.2, 150.9, 143.5, 133.1, 132.1, 131.0, 127.5, 125.1, 122.6, 116.7, 114.7, 112.1, 109.0, 108.7, 51.7, 40.4, 40.3 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₂₁H₂₂N₂O₃ 350.1630; Found 350.1637.

Tert-butyl 2-(10-((*tert*-butyldimethylsilyl)oxy)-3,6-bis(dimethylamino)anthracen-9-yl)acetate (5)



To 3,3'-(((*tert*-butyldimethylsilyl)oxy)methylene)bis(4-bromo-*N*,*N*-dimethylaniline) **2c** (200 mg, 0.37 mol, 1.0 eq.), Pd(PPh₃)₄ (21 mg, 0.018 mol, 0.05 eq.) and triethylamine (260 μ L, 1.8 mmol, 5.0 eq.) in a flame dried, Ar flushed 1 dram vial was added DMF (740 μ L, 0.5 M). The reaction was set to stir at room temperature for 5 minutes before the addition of *tert*-butyl acrylate (82 μ L, 0.55 mmol, 1.5 eq.). The vial was flushed with Ar, sealed carefully with parafilm and black electrical tape and heated to 150 °C overnight in the absence of light. Once at room temperature the crude reaction mixture was diluted with EtOAc and passed through a celite plug eluting with additional EtOAc. The eluant was transferred to a separatory funnel and washed with NaHCO₃ (sat.). The aqueous phase was extracted with EtOAc × 2 with the organic phase then washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The yellow residue was purified by normal-phase column chromatography (12 g silica, 0 \rightarrow 5% hexanes /EtOAc) to afford *tert*-butyl 2-(10-((*tert*-butyldimethylsilyl)oxy)-3,6-bis(dimethylamino)anthracen-9-yl)acetate **5** (130 mg, 71 %) as a bright yellow solid.

¹H NMR (500 MHz, CD₃CN) δ : 8.03 (d, *J* = 9.7 Hz, 2H), 7.26 (d, *J* = 2.6 Hz, 1H), 7.24 (d, *J* = 2.6 Hz, 1H), 7.10 (d, *J* = 2.6 Hz, 2H), 4.34 (s, 2H), 3.06 (s, 12H), 1.38 (s, 9H), 1.16 (s, 9H), 0.22 (s, 6H); ¹³C NMR (126 MHz, CD₃CN) δ : 172.1, 148.0, 142.7, 126.7, 126.3, 124.8, 121.0, 118.0, 100.9, 81.3, 41.2, 35.4, 28.2, 26.5, 19.4, – 2.2 ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₃₀H₄₄N₂O₃Si 508.3121; Found 508.3111.

1-(3,6-bis(dimethylamino)-10-oxoanthracen-9(10H)-ylidene)-2-oxo-6,9,12,15,18-pentaoxa-3azaicosan-20-oic acid (KR-4)



To *tert*-butyl 2-(10-((*tert*-butyldimethylsilyl)oxy)-3,6-bis(dimethylamino)anthracen-9-yl)acetate **5** (50 mg, 0.098 mmol, 1.0 eq.) in an Ar flushed round bottom flask CH₂Cl₂ (9mL, 0.01M) was added and the solution was cooled to 0 °C in an ice-bath. To this TFA (1 mL) was added dropwise and the reaction mixture was stirred overnight wrapped in aluminum foil to exclude light. The reaction was reduced to dryness under a stream of N₂ before being redissolved in CH₂Cl₂ (9mL) with this process being repeated for a total of 3 times. The residue was then suspended in EtOAc, transferred to a separatory funnel and washed with NaHCO₃ (sat.). The aqueous phase was extracted with EtOAc × 2 with the organic phase then washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The yellow residue was used without purification.

To 2-(10-((*tert*-butyldimethylsilyl)oxy)-3,6-bis(dimethylamino)anthracen-9-yl)acetic acid (45 mg, 0.010 mmol, 1.0 eq.) and TSTU (60 mg, 0.20 mmol, 2.0 eq.) in an Ar flushed 1 dram vial was added DMF (0.5 mL). The reaction mixture was sonicated and set to stir at room temperature for 5 min before adding DIPEA (35 μ L, 0.20 mmol, 2.0 eq.). After 1 hour the reaction mixture was added to a partition of EtOAc and 1M HCl in a separatory funnel, the organic layer was recovered and washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The yellow residue was used without purification.

To 2,5-dioxopyrrolidin-1-yl 2-(10-((*tert*-butyldimethylsilyl)oxy)-3,6bis(dimethylamino)anthracen-9-yl)acetate (54 mg, 0.098 mmol, 1 eq.) and Amino-PEG₅-CH₂CO₂H (CAS: 141282-35-1) (35 mg, 0.17 mmol, 1.2 eq.) in an Ar flushed 1 dram vial was added DMF (1 mL, 0.1M). The reaction mixture was sonicated and set to stir at room temperature for 5 min before adding DIPEA (34 μ L, 0.20 mmol, 2.0 eq.). After 1 hour the reaction mixture was dissolved in CH₂Cl₂, transferred to a separatory funnel and washed with NaHCO₃ (sat.). The aqueous phase was extracted with CH₂Cl₂ × 2 and the organic phase then washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The yellow residue was then dissolved in a 5% formic acid solution in 2:1 ACN/ H₂O and stirred for 6 hours at room temperature. Upon drying *in vacuo* the residue was purified by normal-phase column chromatography (12 g silica gold column, 0 \rightarrow 30 % CH₂Cl₂ /MeOH) to afford 1-(3,6bis(dimethylamino)-10-oxoanthracen-9(10*H*)-ylidene)-2-oxo-6,9,12,15,18-pentaoxa-3azaicosan-20-oic acid **KR-4** (38 mg, 63 % over 4 steps) as a bright yellow solid.

¹H NMR (400 MHz, CD₃CN/ D₂O; 9:1) δ : 7.80 (d, *J* = 9.0 Hz, 1H), 7.74 (d, *J* = 8.9 Hz, 1H), 7.36 (d, *J* = 3.0 Hz, 2H), 7.04 (dd, *J* = 8.9, 3.0 Hz, 1H), 6.95 (dd, *J* = 8.9, 3.0 Hz, 1H), 6.60 (s, 1H), 3.83 (s, 2H), 3.61 – 3.43 (m, 19H), 3.39 (t, *J* = 5.4 Hz, 2H), 3.00 (d, *J* = 2.9 Hz, 12H); ¹³C NMR (100 MHz, CD₃CN/ D₂O; 9:1) δ : 186.4, 175.8, 170.7, 151.7, 151.7, 135.8, 133.3, 132.3,

130.5, 127.9, 126.0, 124.3, 119.5, 118.6, 117.0, 108.8, 108.7, 70.7, 70.5, 70.5, 70.4, 70.4, 70.3, 40.5, 40.4, 40.0 ppm. HRMS (ESI) m/z: $[M+H]^+$ Calcd. for $C_{32}H_{43}N_3O_9$ 613.2999; Found 613.2988.

Section S4: Retro-Brook Rearrangement Experiment and NMR Analysis



Scheme S1: Synthetic scheme of H_2O/D_2O quench experiment with predicted products 2c-1 and 2c-2.

To 3,3'-(((*tert*-butyldimethylsilyl)oxy)methylene)bis(4-bromo-*N*,*N*-dimethylaniline) **2c** (20 mg, 0.03687 mmol, 1.0 eq.) in a flame-dried crimped microwave vial under an Ar atmosphere, THF (1 mL) was added. The solution was stirred at room temperature before cooling to -78 °C in a dry ice/ acetone bath. The solution was then treated with *tert*-Butyllithium solution (95 µL, 0.01622 mmol, 4.4 eq.) and allowed stir at -78 °C for 0.5 h. The reaction mixture was stirred for an additional 0.5 h at -10 °C to ensure halogen-metal exchange. The reaction mixture was then quenched at -10 °C with an H₂O (3 mL) or D₂O (3 mL) and extracted with EtOAc × 2. The organic phase was washed with brine × 2, dried over Na₂SO₄, filtered and evaporated to dryness *in vacuo*. The resulting colorless oils **2c-1** and **2c-2** were used without further purification and submitted for ¹H NMR analysis in CD₃CN. Additionally, to an NMR sample of H₂O quench experiment **2c**-1, a few drops of MeOD were added and it was also analyzed ¹H NMR.

<u> A - Starting Material</u>



3,3'-(((tert-butyldimethylsilyl)oxy)methylene)bis(4-bromo-N,N-dimethylaniline) 2c

¹H NMR (400 MHz, CD₃CN) δ : 7.30 (d, *J* = 8.8 Hz, 2H), 6.92 (d, *J* = 3.2 Hz, 2H), 6.57 (dd, *J* = 8.8, 3.2 Hz, 2H), 6.24 (s, 1H), 2.86 (s, 12H), 0.91 (s, 9H), 0.07 (s, 6H) ppm. HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₂₃H₃₄Br₂N₂O₂Si 542.0787; Found 542.0790.

B - H₂O quench



(2-(*tert*-butyldimethylsilyl)-5-(dimethylamino)phenyl)(3-(dimethylamino)phenyl-6-*H*)methanol **2c-1**

¹H NMR (400 MHz, CD₃CN) δ : 7.37 (dt, *J* = 8.4, 1.2 Hz, 1H), 7.11 (t, *J* = 7.9 Hz, 1H), 6.77 (s, 1H), 6.69 – 6.64 (m, 2H), 6.61 (dd, *J* = 8.2, 2.7 Hz, 1H), 6.56 (d, *J* = 7.5 Hz, 1H), 5.86 (d, *J* = 5.1 Hz, 1H), 3.42 (d, *J* = 5.2 Hz, 1H), 2.85 (d, *J* = 13.8 Hz, 12H), 0.91 (s, 9H), 0.31 (d, *J* = 36.9 Hz, 6H). HRMS (ESI) m/z: [M+H]⁺ Calcd. for C₂₃H₃₅¹HN₂OSi 384.2597; Found 384.2272 ppm.

C - D₂O quench



(2-(*tert*-butyldimethylsilyl)-5-(dimethylamino)phenyl)(3-(dimethylamino)phenyl-6-*d*)methanol **2c-2**

¹H NMR (400 MHz, CD₃CN) δ 7.37 (dt, *J* = 8.4, 1.2 Hz, 1H), 7.11 (d, *J* = 8.2 Hz, 1H), 6.77 (d, *J* = 2.6 Hz, 1H), 6.69 - 6.64 (m, 2H), 6.61 (dd, *J* = 8.3, 2.7 Hz, 1H), 5.86 (d, *J* = 5.2 Hz, 1H), 3.41 (d, *J* = 5.2 Hz, 1H), 2.85 (d, *J* = 13.5 Hz, 12H), 0.91 (s, 8H), 0.31 (d, *J* = 37.0 Hz, 6H) ppm. C₂₃H₃₅DN₂OSi 385.2660; Found 385.2284.

D - ²H exchange experiment in MeOD



(2-(*tert*-butyldimethylsilyl)-5-(dimethylamino)phenyl)(3-(dimethylamino)phenyl-6-*H*)methanol*d* **2c-3**

¹H NMR (400 MHz, CD₃CN) δ : 7.45 – 7.29 (m, 1H), 7.10 (t, *J* = 7.9 Hz, 1H), 6.77 (s, 1H), 6.68 – 6.64 (m, 2H), 6.61 (dd, *J* = 8.2, 2.7 Hz, 1H), 6.56 (d, *J* = 7.3 Hz, 1H), 5.85 (s, 1H), 2.84 (d, *J* = 12.4 Hz, 12H), 0.90 (s, 9H), 0.30 (d, *J* = 38.6 Hz, 6H) ppm.



Figure S2: Stacked ¹H NMR spectra of compounds 2c, 2c-1, 2c-2 and 2c-3 in ACN-d₃.



Figure S3: Stacked ¹H NMR spectra of compounds **2c**, **2c-1**, **2c-2** and **2c-3** in ACN-d₃ (zoom 7.5 - 3.3 ppm).

Section S5: Photophysical Measurements

Determination of molar absorption coefficients and fluorescence quantum yields.

All spectroscopic measurements were carried out in duplicate at room temperature using 10 mm × 10 mm quartz cuvettes (Hellma GmbH) and high-purity spectroscopic grade solvents. High purity water ($\sigma < 55 \ \mu$ S) was obtained using a MilliPore filter system. Dichloromethane (CH₂Cl₂; anhydrous, \geq 99.8 %), Dimethylsulfoxide (DMSO; for molecular biology) and Trifluoroacetic acid (TFA; HPLC grade) were purchased from Millipore Sigma. Phosphate buffered saline (PBS 1x; gibco) was purchased from Fisher Scientific. Acetate buffer (pH 3.6) was prepared by adding 0.099 g of Sodium Acetate (Millipore Sigma - 791741-500g) and 1.429 g of Acetic Acid (VWR - BDH3094-2.5LG) to 400 mL H₂O and adjusting the solution to pH 3.6 using HCl or NaOH before adding H₂O to a final volume of 500 mL in a volumetric flask,

The reference dyes IR140 was purchased from Lambda Physik and used to determine the fluorescence quantum yield of **KR-1** (see main body text).

Molar absorption coefficients (ϵ) at the respective absorbance maximum were determined using a PerkinElmer UV/Vis Photodiode Array Spectrophotometer Lambda 465 (Scan No. 30, Integration No. 1) in CH₂Cl₂, PBS (pH 7.4), CH₂Cl₂ supplemented with 5 % TFA or acetate buffer at pH 3.6 using Beer's law, from plots of absorbance vs. concentration. Measurements were performed in 10 mm path length quartz cuvettes (Hellma 111-QS), maintained at 25 °C, with absorbance at the highest concentration ≤ 0.20 .

Fluorescence traces were recorded on a PTI QuantaMaster steady state spectrofluorometer operated by FelixGX 4.2.2 software, with 827 nm excitation, 15 nm excitation and 20 nm emission slit widths with a 1 nm step size, 1 s integration rate and a collection range of 855 nm - 1300 nm (organic solvent) or 855 nm - 1500 nm (aqueous solvent).

Previously measured KR-1 quantum yield value of 1.3 % in CH₂Cl₂ was used as the internal standard (ST below) for all subsequent measurements which were determined using a relative method and calculated according to the following equation:

$$\phi_x = \phi_{ST} \left(\frac{(Grad_x)}{(Grad_{ST})} \right) \left(\frac{(n_x^2)}{(n_{ST}^2)} \right)$$

where Φ is the fluorescence quantum yield, subscript x and ST denote the test and standard respectively, Grad is the gradient from the plot of the integrated intensity vs. absorbance at excitation wavelength (827 nm) and n is the refractive index of the solvent used.

All fluorescent measurements were manually background subtracted using fluorescent traces of the appropriate blank taken under identical conditions to that of the corresponding samples (827 nm excitation, 15 nm excitation and 20 nm emission slit widths with a 1 nm step size and 1 s integration rate).

Figure 2, Panel D: A 2mM stock of **KR-2** was prepared in DMSO. 10 μ M - 50 μ M samples (5 mL) were prepared in CH₂Cl₂ supplemented with 5 % TFA and separately a 50 μ M sample of **KR-2** was prepared in CH₂Cl₂ with all samples containing a final DMSO concentration of 2.5 %. All samples were vortexed and sonicated before allowing to equilibrate at ambient temperature in the absence of light before absorbance/ fluorescence measurements.

Figure 2, Panel E: A 10 mM stock of **KR-4** in DMSO was diluted to 250 μ M in PBS pH 7.4 (10 mL). This solution was adjusted from pH 7.4 to pH 3.5 using 0.1 M, 0.5 M and 1M NaOH/ HCl in decrements of pH 0.5. Once the desired pH was obtained a 200 μ L sample was removed and absorbance curve was generated using a Synergy Mx microplate reader (300 nm - 999nm, step size 5 nm).

Figure 2, Panel F: pK_a was determined by plotting the λ_{max} Abs (870 nm) vs. each pH value. Sigmoidal plot fit resulted in apparent pKa = 4.4.

Section S6: ¹ H, ¹³C and Associated 2D NMR Spectra

¹H (400 MHz) and ¹³C (100 MHz) NMR spectra of **bis(3-(dimethylamino)phenyl)methanone** (1)



¹H (400 MHz) and ¹³C (100 MHz) NMR spectra of **bis(2-bromo-5-(dimethylamino)phenyl)methanone (7)**



¹H (400 MHz) and ¹³C (100 MHz) NMR spectra of **bis(2-bromo-5-(dimethylamino)phenyl)methanol (12)**





¹H (400 MHz) and ¹³C (100 MHz) NMR spectra of **3,3'-((methoxymethoxy)methylene)bis(4-bromo-***N*,*N*-**dimethylaniline) (2a)**



¹H (400 MHz) and ¹³C (100 MHz) NMR spectra of **3,3'-(methoxymethylene)bis(4-bromo-***N*,*N***-dimethylaniline) (2b)**





¹H (400 MHz) and ¹³C (100 MHz) NMR spectra of **3,3'-(((***tert***-butyldimethylsilyl)oxy)methylene)bis(4-bromo-***N***,***N***-dimethylaniline) (2c)**






¹H (400 MHz) and ¹³C (100 MHz) NMR spectra of **9-methoxy**-*N*²,*N*²,*N*⁷,*N*⁷-tetramethyl-10-(*o*-tolyl)anthracene-2,7-diamine (3b)



¹H (400 MHz) and ¹³C (100 MHz) NMR spectra of *N*-(7-(dimethylamino)-9-oxo-10-(*o*-tolyl)anthracen-2(9*H*)-ylidene)-*N*-methylmethanaminium (KR-1)



¹H (400 MHz) and ¹³C (100 MHz) NMR spectra of **2-(3,6-bis(dimethylamino)-10-(methoxymethoxy)anthracen-9-yl)benzoic acid (4)**



¹H (500 MHz) and ¹³C (125 MHz) NMR spectra of **3,6-bis(dimethylamino)-3'H,10H-spiro[anthracene-9,1'-isobenzofuran]-3',10-dione (KR-2)**



¹H (500 MHz) and ¹³C (125 MHz) NMR spectra of **3-(2-(3,6-bis(dimethylamino)-10-(methoxymethoxy)anthracen-9-yl)**-*N*-(**3-(***tert***-butoxy)-3-oxopropyl)**benzamido)propane-1-sulfonate (9)



¹H (400 MHz) and ¹³C (100 MHz) NMR spectra of **3-(***N***-(2-carboxyethyl)-2-(6-(dimethylamino)-3-(dimethyliminio)-10-oxo-3,10-dihydroanthracen-9-yl)benzamido)propane-1-sulfonate (KR-3)**



¹H (400 MHz) and ¹³C (100 MHz) NMR spectra of methyl 2-(10-((*tert*-butyldimethylsilyl)oxy)-3,6-bis(dimethylamino)anthracen-9-yl)acetate (10)



¹H (400 MHz) and ¹³C (100 MHz) NMR spectra of methyl 2-(3,6-bis(dimethylamino)-10-oxoanthracen-9(10*H*)-ylidene)acetate (11)



¹H (500 MHz) and ¹³C (125 MHz) NMR spectra of *tert*-butyl 2-(10-((*tert*-butyldimethylsilyl)oxy)-3,6-bis(dimethylamino)anthracen-9-yl)acetate (5)



¹H (400 MHz) and ¹³C (100 MHz) spectra of **1-(3,6-bis(dimethylamino)-10-oxoanthracen-9(10H)-ylidene)-2-oxo-6,9,12,15,18-pentaoxa-3-azaicosan-20-oic acid (KR-4)**



HSQC spectra of 1-(3,6-bis(dimethylamino)-10-oxoanthracen-9(10*H*)-ylidene)-2-oxo-6,9,12,15,18-pentaoxa-3-azaicosan-20-oic acid (KR-4)



¹H (400 MHz) NMR spectra of **3,3'-(((***tert***-butyldimethylsilyl)oxy)methylene)bis(4-bromo-***N*,*N*-dimethylaniline) (2c)

¹H (400 MHz) NMR spectra of (2-(*tert*-butyldimethylsilyl)-5-(dimethylamino)phenyl)(3-(dimethylamino)phenyl-6-H)methanol (2c-1)





¹H (400 MHz) NMR spectra of (2-(*tert*-butyldimethylsilyl)-5-(dimethylamino)phenyl)(3-(dimethylamino)phenyl-6-*d*)methanol (2c-2)

¹H (400 MHz) NMR spectra of (2-(*tert*-butyldimethylsilyl)-5-(dimethylamino)phenyl)(3-(dimethylamino)phenyl-6-*H*)methanol-*d* (2c-3)



Section S7: Computational Chemistry

Quantum chemistry calculations were performed using a local version of the GAMESS package(7, 8) where cc-pVTZ basis sets (spherical harmonics) were used throughout(9, 10). Molecular orbitals were illustrated with MacMolPlt(11).

Density functional theory (DFT)(12) utilizing the B3LYP functional(13-15) was used to optimize ground state (S0) geometries. The time-dependent density functional theory (TDDFT) method(16-18) was used to compute vertical excitation, or absorbance, energies (S0 \rightarrow S1*) at ground state (S0) optimized geometries, as well as deexcitation, or emission, energies (S1 \rightarrow S0*) at singlet excited state optimized geometries (S1)(19) Ground state (S0) DFT and excited state (S1) TDDFT computations used the Euler-MacLaurin radial(20) and Lebedev angular(21) grids with respective dimensions (96, 302) and (48, 110).

The occupation restricted multiple active space (ORMAS)(22, 23) method with second-order perturbation theory correction (ORMAS-PT2)(24) was also used to compute vertical excitation (absorbance) energies and deexcitation (emission) energies. ORMAS-PT2 calculations were performed at B3LYP ground state (S0; vertical excitation energy) and TDDFT-B3LYP excited state (S1; deexcitation energy) optimized geometries. ORMAS active spaces consisted of all valence π electrons and orbitals partitioned into three groups as follows:

1. All valence π bonding/lone-pair orbitals not including HOMO-like orbital (numbering N) with allowed total minimum & maximum occupation of (2N-2) & 2N, i.e., single & double excitations are allowed from this first group of N orbitals.

2. HOMO-like orbital and LUMO-like orbital with total minimum & maximum occupation of 0 and 4, i.e., single & double electron excitations are allowed into and out of this second group of two orbitals.

3. All remaining valence π antibonding orbitals with total minimum & maximum occupation of 0 and 2, i.e., single & double electron excitations are allowed into this third group.

For example, for R = NH and O the \$DET/\$ORMAS GAMESS input specifications are:

\$det ncore=62 nels=18 nact=16 nstate=3 wstate(1)=1,1 \$end \$ormas nspace=3 mstart(1)=63,71,73 mine(1)=14,0,0 maxe(1)=16,4,2 \$end

The numbers of orbitals in each group and associated occupancy minima/maxima are given in **Table S3** below:

R-group	No. orbitals in group 1 (min/max e ⁻)	No. of orbitals in group 2 (min/max e ⁻)	No. orbitals in group 3 (min/max e ⁻)
NH,O	8 (14/16)	2 (0/4)	6 (0/2)
CMe ₂ , SiMe ₂ , SO ₂ , POOEt	7 (12/14)	2 (0/4)	6 (0/2)
C=O	8 (14/16)	2 (0/4)	7 (0/2)

The ORMAS ground state (S0) and singlet excited state (S1) wavefunctions were simultaneously optimized (for orbitals and configuration interaction coefficients) by minimizing the average of

the two energies. Final improved ORMAS-PT2 energies for each state (S0 and S1) were then computed by modeling all single and double valence electron excitations out of the individual S0 and S1 ORMAS wavefunctions using second-order perturbation theory (1s orbitals were kept core).

The polarizable continuum model (PCM)(25-27) approach was used to model water solvent effects for the DFT(28), TDDFT(29), ORMAS(26), and ORMAS-PT2 methods. All calculations used a high density of tesserae (NTSALL = 960 in TESCAV).

Mulliken electron population analysis(30) was used to analyze the B3LYP-PCM highest occupied molecular orbitals (HOMOs) and lowest unoccupied molecular orbitals (LUMOs). Specifically, for each orbital we determined 1) $C-C_{10}$ bond orders (see Figure 1A and Table SZ), and 2) the total electron populations on the R-groups that were then multiplied by the sign of the $C-C_{10}$ bond order, i.e., the R-group electron populations were penalized to be negative if they introduced a node between ring-bound R-group atom and neighboring carbon atoms.

Table S4: Computed absorbance and emission energies (in eV with oscillator strengths in parentheses) vs. experimentally measured. Calculations included water solvent effects via the Polarizable Continuum Model (PCM). Mean unsigned errors (MUEs) for each method shown at bottom. See Methods for full details.

R-Group	Computed abso	rbance Experimental		Computed emission		Experimental
	TDDFT-B3LYP	ORPT2	absorbance	TDDFT-B3LYP	ORPT2	emission
N(H)	2.791 (0.90)	2.588	2.515 ^x	2.517 (1.22)	2.465	2.348
0	2.666 (1.06)	2.386	2.267 ^x	2.366 (1.32)	2.288	2.206
$C(Me)_2$	2.498 (1.14)	2.060	2.050 ^x	2.204 (1.39)	1.996	1.978
Si(Me) ₂	2.401 (1.14)	1.972	1.956 ^x	2.100 (1.39)	1.903	1.914
$S(O)_2$	2.247 (0.95)	1.846	1.771 ^x	1.924 (1.14)	1.746	1.717
P(OOEt)	2.223 (0.92)	1.824	1.764 ^x	1.863 (1.03)	1.684	1.689
C(O)	1.727 (0.41)	1.547	1.450ª	1.405 (0.48)	1.373	1.361
MUE	0.397	0.065		0.167	0.039	



Figure S4. Experimental vs. predicted absorbance (A) and emission (B) energies (eV) using TDDFT-B3LYP-PCM(H₂O) and ORMAS-PT2-PCM(H₂O) methods. Mean unsigned errors (MUEs) in absorbance/emission predictions are 0.397/0.167 and 0.065/0.039 for TDDFT-B3LYP and ORMAS-PT2, respectively.

Table S5: Experimental and computed (TDDFT-B3LYP-PCM) absorbance energies (eV), C-C10' bond orders, and signed R-group electron populations in the LUMOs. Properties determined at ground state optimized geometries. Calculations included water solvent effects via the Polarizable Continuum Model (PCM). See Methods for full details. Correlation coefficients (R^2) between experimental (left) and computed (right) absorbances vs. LUMO C-C10' bond orders and signed R-group electron populations are shown at bottom.

R-group	Experimental absorbance	Computed absorbance	LUMO C-C10' bond orders	LUMO R-group e^{-} population
N(H)	2.515	2.791	-0.049	-0.065^{a}
0	2.267	2.666	-0.039	-0.042^{a}
$C(Me)_2$	2.050	2.498	-0.001	-0.026^{a}
Si(Me) ₂	1.956	2.401	0.002	0.035
$S(O)_2$	1.771	2.247	0.024	0.045
P(OOEt)	1.764	2.223	0.021	0.047
C(O)	1.450	1.727	0.054	0.181
R^2			0.94 / 0.87	0.88 / 0.97

^a These Mulliken populations are penalized to be negative since each R-group introduces a node to the LUMO that passes between the ring-bound R-group atom (not in parentheses) and neighboring carbon atoms, i.e., in these systems there is a negative $C-C_{10}$, bond order in the LUMO. For the other molecules the $C-C_{10}$ bond order is positive. See **Table S6** for further details.

Table S6: Examinations of HOMOs and LUMOs in the C10' substituted series via Mulliken electronic population analysis. For each orbital, the C–C10' bond order is provided along with the total electron population on the R group. Analyses were performed at ground state optimized geometries.



	HOM	ЛО	LUMO		
R-group C-	C–C10' bond order	R <i>e</i> [−] population	C–C10' bond order	R e^- population	
N(H)	0.000	0.000	-0.049	0.065	
0	0.000	0.000	-0.039	0.042	
C(Me) ₂	0.000	0.001	-0.001	0.026	
Si(Me) ₂	0.000	0.001	0.022	0.035	
$S(O)_2$	0.000	0.001	0.024	0.045	
P(OOEt)	0.000	0.000	0.021	0.047	
C(O)	0.000	0.000	0.054	0.181	

Section S8: Raw Computational Chemistry Data

I. Energies (S0 & S1* in hartree), oscillator strengths (S0 \rightarrow S1*), HOMO and LUMO energies (hartree), excitation/de-excitation amplitudes, and S0 optimized structures (Å)

a. R1 (NH)



TDDFT-B3LYP-PCM(H₂O)

E(S0) = -823.753609 $E(S1^*) = -823.651047$ Oscillator strength = 0.8956E(HOMO) = -0.2084; E(LUMO) = -0.0954

EXCITATION DE-EXCITATION

OCC VIR AMPLITUDE AMPLITUDE

Ι	А	X(I->A)	Y(A->I)
68	72	-0.042187	-0.006089
71	72	-0.987340	0.086689
69	73	0.058654	0.025205
70	73	-0.149761	-0.038580
69	74	-0.043578	-0.007822

С	6.0 -1.2226311806	0.6677218834	-0.0449552379
С	6.0 -2.2602567210	-0.3138660898	-0.0065562156
С	6.0 -1.9160583604	-1.6623065593	-0.0310673043
С	6.0 -0.5895440976	-2.0777994523	-0.0884399778
С	6.0 0.4451749848	-1.0932229219	-0.1241529030
Ν	7.0 0.0708497181	0.2236533337	-0.1041878502
С	6.0 -2.8431158499	2.4716235052	0.0534042466
С	6.0 -3.8879748719	1.4842292919	0.0885132213
С	6.0 -3.5984866918	0.1557159806	0.0588000280
С	6.0 -1.5108919399	2.0262393918	-0.0196271326
Η	1.0 -4.9183632801	1.7962451348	0.1372421984
Η	1.0 -4.3991506569	-0.5717731078	0.0852000967
Η	1.0 -0.6939709348	2.7299110812	-0.0487530416
Ν	7.0 -3.1461413002	3.7896397426	0.0907972073
С	6.0 -2.0865431514	4.7892171070	0.0221113938

С	6.0 -4.5320767499 4.2	425551450 0.1815757554
Η	1.0 -2.5302397588 5.7	784039093 0.0572403549
Η	1.0 -1.5172808350 4.7	010862109 -0.9063260075
Η	1.0 -4.5411875231 5.3	247165599 0.2568657121
Η	1.0 -5.1077180282 3.9	572341900 -0.7016281706
Η	1.0 -1.3944086994 4.6	934882674 0.8621437480
Η	1.0 -5.0246961059 3.8	372617767 1.0665588697
Η	1.0 -2.7011442403 -2.4	083928427 -0.0025864590
С	6.0 -0.1938077300 -3.4	413052146 -0.1097146466
С	6.0 1.7849360127 -1.4	548510171 -0.1774965618
С	6.0 1.1158649676 -3.8	025630614 -0.1631911177
С	6.0 2.1568978413 -2.8	114980010 -0.2017775402
Ν	7.0 3.4550608180 -3.1	876737532 -0.2585483562
С	6.0 3.8320064862 -4.5	982206395 -0.3147623786
С	6.0 4.5110148565 -2.1	821756937 -0.2592907958
Η	1.0 2.5310759978 -0.6	0767154968 -0.2038791914
Η	1.0 -0.9627426483 -4.2	2020470302 -0.0792650060
Η	1.0 1.3718119483 -4.8	494197921 -0.1722586369
Η	1.0 4.9072856813 -4.6	651270539 -0.4403333001
Н	1.0 3.3604990868 -5.1	010197246 -1.1602336895
Η	1.0 5.4742746497 -2.6	804928738 -0.2710962992
Η	1.0 4.4608252562 -1.5	548855485 0.6336144981
Η	1.0 4.4469772502 -1.5	368629744 -1.1390060544
Н	1.0 3.5612070847 -5.1	262034023 0.6023532392
Н	1.0 0.8037349552 0.9	194231997 -0.1284159547

E(S0) = -822.284283: ORMAS-PCM natural orbitals 71 / 72 occupations = 1.940 / 0.072 $E(S1^*) = -822.189168$: ORMAS-PCM natural orbitals 71 / 72 occupations = 1.183 / 0.830

b. R2 (O)



TDDFT-B3LYP-PCM(H₂O)

E(S0) = -843.601869 $E(S1^*) = -843.503894$ Oscillator strength = 1.0637E(HOMO) = -0.2123; E(LUMO) = -0.1072

EXCITATION DE-EXCITATION OCC VIR AMPLITUDE AMPLITUDE I A $X(I_{->A})$ $Y(A_{->I})$

I	Α	X(1->A)	Y(A->1)
68	72	0.037796	0.004803
71	72	0.996944	-0.111461
69	73	0.065997	0.028503
70	73	-0.086145	-0.029931
71	74	-0.033024	-0.004508
69	75	-0.032036	-0.005688

С	6.0	-1.2543120786	0.6019886662	-0.0407150587
С	6.0	-2.2996166459	-0.3604024924	-0.0015587917
С	6.0	-1.9677077307	-1.7111897978	-0.0295958563
С	6.0	-0.6376981422	-2.1138261050	-0.0894503835
С	6.0	0.3785562827	-1.1207484173	-0.1221968550
0	8.0	0.0443767480	0.1986608014	-0.1018446185
С	6.0	-2.8205310698	2.4350584267	0.0579192632
С	6.0	-3.8857721896	1.4735837228	0.0982021627
С	6.0	-3.6266960571	0.1375419941	0.0677812265
С	6.0	-1.4908650575	1.9547420499	-0.0168869061
Н	1.0	-4.9084645653	1.8088340750	0.1508437686
Н	1.0	-4.4429735724	-0.5720258797	0.0968743584
Н	1.0	-0.6478566172	2.6251325968	-0.0500200502
Ν	7.0	-3.0839076551	3.7560248397	0.0905678010
С	6.0	-1.9981005865	4.7295093908	0.0126120473
С	6.0	-4.4587586880	4.2475838233	0.1727649378
Н	1.0	-2.4171677927	5.7288521952	0.0522197851
Н	1.0	-1.4419808079	4.6256204490	-0.9213360050
Н	1.0	-4.4397453274	5.3302136449	0.2332311007
Н	1.0	-5.0367431958	3.9632448922	-0.7086717364
Н	1.0	-1.3033764978	4.6147461893	0.8469909511
Н	1.0	-4.9618072205	3.8652494350	1.0617491144
Н	1.0	-2.7520870330	-2.4575543639	-0.0021650819
С	6.0	-0.2135802718	-3.4676000873	-0.1158475113
С	6.0	1.7156490169	-1.4309408313	-0.1747839187
С	6.0	1.1052402344	-3.7996510618	-0.1700243631
С	6.0	2.1219061857	-2.7866803957	-0.2031614103
Ν	7.0	3.4256526459	-3.1229939878	-0.2586497345
С	6.0	3.8400555974	-4.5241565133	-0.3153094490
С	6.0	4.4584007538	-2.0904537320	-0.2519936081
Η	1.0	2.4309264791	-0.6255045667	-0.1977520688
Н	1.0	-0.9661360652	-4.2444825592	-0.0886430178
Н	1.0	1.3841599022	-4.8403382007	-0.1832668828

Н	1.0	4.9168761730	-4.5639553655	-0.4372824952
Η	1.0	3.3819302371	-5.0352507148	-1.1625529342
Н	1.0	5.4323897242	-2.5669530111	-0.2650807017
Η	1.0	4.3909393489	-1.4713409913	0.6446989716
Η	1.0	4.3790110355	-1.4441278809	-1.1288107189
Н	1.0	3.5778518533	-5.0564326569	0.6012735192

E(S0) = -842.115491: ORMAS-PCM natural orbitals 71 / 72 occupations = 1.945 / 0.068 $E(S1^*) = -842.027793$: ORMAS-PCM natural orbitals 71 / 72 occupations = 1.023 / 0.988

c. R3 (CMe₂)



TDDFT-B3LYP-PCM(H₂O)

E(S0) = -886.268540 $E(S1^*) = -886.176729$ Oscillator strength = 1.1385E(HOMO) = -0.2070; E(LUMO) = -0.1104

EXCITATION DE-EXCITATION OCC VIR AMPLITUDE AMPLITUDE $X(I \rightarrow A)$ $Y(A \rightarrow I)$ Ι Α -------_____ 79 80 1.002508 -0.134081 76 81 0.035302 0.019047 79 81 -0.055473 -0.008565 82 -0.049856 77 -0.024708ATOM CHARGE Ζ Х Y С 6.0 -1.1846279517 0.8008604074 -0.1045345506 С 6.0 -2.1738499430 -0.2142125284 0.0486986666 С 6.0 -1.8065520253 -1.5592849299 0.0743685659 С 6.0 -0.4889366835 -1.9986045685 -0.0432123114 С 6.0 0.5738408138 -1.0611756156 -0.1981089499 С 6.0 0.2985324569 0.4456010379 -0.2491399530 С 6.0 -2.9441095056 2.5048835651 0.0062272515 С 6.0 -3.9204822941 1.4750465993 0.1532425941

6.0	-3.5347478180	0.1677837592	0.1729581040
6.0	-1.5809721817	2.1165183842	-0.1240414266
1.0	-4.9649639667	1.7232828512	0.2486453804
1.0	-4.2806156373	-0.6086378280	0.2858980411
1.0	-0.8380708498	2.8879977900	-0.2390160583
7.0	-3.3012650497	3.8050474800	-0.0066116748
6.0	-2.2983755020	4.8572021671	-0.1600998549
6.0	-4.7003279119	4.1969727197	0.1552771003
1.0	-2.7972501444	5.8199267933	-0.1604053389
1.0	-1.7563690540	4.7523338559	-1.1014070042
1.0	-4.7620424975	5.2795083674	0.1552957885
1.0	-5.3166612772	3.8172193935	-0.6617388975
1.0	-1.5808611235	4.8419145920	0.6624911984
1.0	-5.1045515368	3.8294280194	1.0997009607
1.0	-2.5866248116	-2.3030178772	0.1918493860
6.0	-0.1822117957	-3.3841679000	-0.0107611515
6.0	1.8607417769	-1.5299840908	-0.3078132007
6.0	1.0962927656	-3.8419627749	-0.1243908035
6.0	2.1732108114	-2.9190502726	-0.2769205760
7.0	3.4458087578	-3.3488793265	-0.3888976909
6.0	3.7549995951	-4.7779258412	-0.3999163708
6.0	4.5511912325	-2.4000774187	-0.5028813856
1.0	2.6676679776	-0.8264377074	-0.4280826639
1.0	-0.9944325796	-4.0898847598	0.1091028065
1.0	1.2868284340	-4.9023675181	-0.0919474259
1.0	4.8188342306	-4.9036872780	-0.5676033325
1.0	3.2196878188	-5.2897779924	-1.2008902452
1.0	5.4858208656	-2.9495706429	-0.5142293983
1.0	4.5683669329	-1.7142059267	0.3452118427
1.0	4.4834198147	-1.8180352059	-1.4243461836
1.0	3.4976074562	-5.2496505498	0.5506803341
6.0	0.8005712542	0.9887804141	-1.6140591336
6.0	1.0841368063	1.1280133926	0.9025574706
1.0	0.2655339819	0.5185862705	-2.4393259322
1.0	0.6492439852	2.0649946936	-1.6770730984
1.0	1.8637113378	0.7903199009	-1.7379491462
1.0	2.1503943749	0.9295622702	0.8103852679
1.0	0.7491704223	0.7595668082	1.8722250189
1.0	0.9410384973	2.2069033606	0.8788168302
	$\begin{array}{c} 6.0\\ 6.0\\ 1.0\\ 1.0\\ 1.0\\ 1.0\\ 7.0\\ 6.0\\ 1.0\\ 1.0\\ 1.0\\ 1.0\\ 1.0\\ 1.0\\ 1.0\\ 1$	6.0 -3.5347478180 6.0 -1.5809721817 1.0 -4.9649639667 1.0 -4.2806156373 1.0 -4.2806156373 1.0 -0.8380708498 7.0 -3.3012650497 6.0 -2.2983755020 6.0 -4.7003279119 1.0 -2.7972501444 1.0 -1.7563690540 1.0 -4.7620424975 1.0 -5.3166612772 1.0 -5.3166612772 1.0 -5.3166612772 1.0 -5.1045515368 1.0 -2.5866248116 6.0 -0.1822117957 6.0 1.8607417769 6.0 2.1732108114 7.0 3.4458087578 6.0 3.7549995951 6.0 4.5511912325 1.0 2.6676679776 1.0 1.2868284340 1.0 4.8188342306 1.0 3.2196878188 1.0 5.4858208656 1.0 4.834198147 1.0 3.4976074562 6.0 0.8005712542 6.0 1.0841368063 1.0 0.2655339819 1.0 0.7491704223 1.0 0.7491704223 1.0 0.9410384973	6.0 -3.5347478180 0.1677837592 6.0 -1.5809721817 2.1165183842 1.0 -4.9649639667 1.7232828512 1.0 -4.2806156373 -0.6086378280 1.0 -0.8380708498 2.8879977900 7.0 -3.3012650497 3.8050474800 6.0 -2.2983755020 4.8572021671 6.0 -4.7003279119 4.1969727197 1.0 -2.7972501444 5.8199267933 1.0 -1.7563690540 4.7523338559 1.0 -4.7620424975 5.2795083674 1.0 -5.3166612772 3.8172193935 1.0 -1.5808611235 4.8419145920 1.0 -5.1045515368 3.8294280194 1.0 -5.1045515368 3.8294280194 1.0 -2.5866248116 -2.3030178772 6.0 0.1822117957 -3.3841679000 6.0 1.8607417769 -1.5299840908 6.0 1.0962927656 -3.8419627749 6.0 2.1732108114 -2.9190502726 7.0 3.4458087578 -3.3488793265 6.0 3.7549995951 -4.7779258412 6.0 4.5511912325 2.4000774187 1.0 2.6676679776 -0.8264377074 1.0 -0.9944325796 -4.0898847598 1.0 1.2868284340 -4.9023675181 1.0 4.58360856 -2.9495706429 1.0 3.4976074562 -5.2496505498 6.0 0.8005712542 0.9887804141

E(S0) = -884.672184: ORMAS-PCM natural orbitals 79 / 80 occupations = 1.945 / 0.068 $E(S1^*) = -884.596467$: ORMAS-PCM natural orbitals 79 / 80 occupations = 1.020 / 0.991

d. R4 (SiMe₂)



TDDFT-B3LYP-PCM(H₂O)

E(S0) = -1137.678861 $E(S1^*) = -1137.590633$ Oscillator strength = 1.1388E(HOMO) = -0.2078; E(LUMO) = -0.1151

EXCITATION DE-EXCITATION OCC VIR AMPLITUDE AMPLITUDE I A X(I->A) Y(A->I)

80	84	0.044641	0.005492
83	84	1.001235	-0.138984
80	85	-0.048737	-0.025170
83	85	-0.065898	-0.011229
81	86	0.039006	0.019489

ATOM CHARGE X Y

Ζ

С	6.0 -1.2082030246 1.0527659644 -0.0210967970
С	6.0 -2.0911782649 -0.0747052931 0.0143045388
С	6.0 -1.6527920117 -1.4082039944 -0.0137065050
С	6.0 -0.3481095455 -1.9234454159 -0.0815575095
С	6.0 0.8266262931 -1.1057832558 -0.1338037208
SI	14.0 0.6609955540 0.7763323390 -0.1042139194
С	6.0 -3.1454826175 2.5662371675 0.0753362072
С	6.0 -4.0082144181 1.4336418227 0.1115844827
С	6.0 -3.4892053961 0.1709954872 0.0812847196
С	6.0 -1.7378715518 2.3198347786 0.0099458316
Н	1.0 -5.0769061918 1.5642935958 0.1644139350
Н	1.0 -4.1637580664 -0.6757631339 0.1099575203
Н	1.0 -1.0742276996 3.1713484042 -0.0179997515
Ν	7.0 -3.6353604214 3.8197096016 0.1017703524
С	6.0 -2.7439677435 4.9784192066 0.0802869147
С	6.0 -5.0778081849 4.0523730688 0.1614121241
Н	1.0 -3.3417252657 5.8815858925 0.1314359157
Н	1.0 -2.1546574495 5.0063934062 -0.8378024100

Η	1.0 -5.2651221277 5.1203263198 0.1429754081
Η	1.0 -5.5834290936 3.5995590162 -0.6927785740
Η	1.0 -2.0642693744 4.9680103262 0.9337641304
Η	1.0 -5.5048687480 3.6420391755 1.0783294160
Н	1.0 -2.4416374169 -2.1527847009 0.0204980554
С	6.0 -0.1828785154 -3.3343037813 -0.1023522498
С	6.0 2.0597809297 -1.7068108866 -0.2049023466
С	6.0 1.0471334936 -3.9243850373 -0.1712409909
С	6.0 2.2255957076 -3.1261996430 -0.2268619379
Ν	7.0 3.4481487259 -3.6859201445 -0.2967405603
С	6.0 3.6024435267 -5.1395997951 -0.2887739128
С	6.0 4.6523498047 -2.8597270685 -0.3728449103
Η	1.0 2.9443433705 -1.0887474378 -0.2419259284
Η	1.0 -1.0656310960 -3.9603928435 -0.0635583856
Η	1.0 1.1163831417 -5.0000407192 -0.1866928931
Η	1.0 4.6596150674 -5.3800138423 -0.3052732169
Н	1.0 3.1311486935 -5.5918226312 -1.1634552845
Η	1.0 5.5150213070 -3.5075479785 -0.4821322330
Η	1.0 4.7819018380 -2.2623885673 0.5318212420
Η	1.0 4.6161321668 -2.1903676480 -1.2333443046
Η	1.0 3.1641764903 -5.5757401757 0.6101183130
С	6.0 1.3638600178 1.5209718678 -1.6778276615
С	6.0 1.4928244901 1.4865273858 1.4217405085
Η	1.0 1.1999814780 2.6001735404 -1.7027622833
Н	1.0 2.4393652285 1.3451003253 -1.7445491854
Η	1.0 2.5659530666 1.2862969546 1.4071556782
Η	1.0 1.3556748605 2.5689695092 1.4666654492
Η	1.0 1.0768842211 1.0531229216 2.3322862694
Н	1.0 0.8923819315 1.0851867863 -2.5597952704

E(S0) = -1135.721772: ORMAS-PCM natural orbitals 83 / 84 occupations = 1.945 / 0.072 $E(S1^*) = -1135.649301$: ORMAS-PCM natural orbitals 83 / 84 occupations = 1.020 / 0.993

e. R5 (POOEt)



<u>TDDFT-B3LYP-PCM(H₂O)</u>

E(S0) = -1339.411043

E(*S*1*) = -1339.328473 Oscillator strength = 0.9473 *E*(*HOMO*) = -0.2159; *E*(*LUMO*) = -0.1287

EXCITATION DE-EXCITATION

OCC VIR AMPLITUDE AMPLITUDE

1
6
7
8

С	6.0 -1.0187910832 1.001117562	6 0.1968931977
С	6.0 -1.9414543636 -0.079582057	1 0.3489633220
С	6.0 -1.5446340044 -1.423026283	0.3684354462
С	6.0 -0.2508791853 -1.934449976	0.1916797199
С	6.0 0.9010504585 -1.105710950	0.0422867827
Р	15.0 0.7600578120 0.676236622	0.2916746947
С	6.0 -2.8529272474 2.607007479	9 0.0672066504
С	6.0 -3.7682047315 1.534688907	1 0.2728261056
С	6.0 -3.3211359376 0.250258490	6 0.3998100471
С	6.0 -1.4563668749 2.288070269	8 0.0472836124
Н	1.0 -4.8280322080 1.727602289	0.3084713433
Н	1.0 -4.0394528762 -0.549706442	0.5252690898
Н	1.0 -0.7333100914 3.075753551	8 -0.0952521925
Ν	7.0 -3.2752069732 3.868792068	3 -0.1059254563
С	6.0 -2.3275318433 4.953241833	1 -0.3682239064
С	6.0 -4.7030287115 4.188020355	2 -0.0698003999
Н	1.0 -2.8835293193 5.869047360	9 -0.5328735231
Н	1.0 -1.7343722516 4.745606711	9 -1.2593004054
Н	1.0 -4.8256929720 5.258267720	5 -0.1890249728
Н	1.0 -5.2373308647 3.683751391	7 -0.8769604669
Η	1.0 -1.6561921930 5.104981615	0.4782976172
Η	1.0 -5.1443466724 3.893192038	0.8828152571
Н	1.0 -2.3368690158 -2.154897429	0.4787078477
С	6.0 -0.0506657214 -3.336303683	0 0.1011087440
С	6.0 2.1327760939 -1.638863101	8 -0.2163403159
С	6.0 1.1771609703 -3.883229552	0 -0.1457195979
С	6.0 2.3207784937 -3.053495894	4 -0.3300709771
Ν	7.0 3.5319235840 -3.567946448	9 -0.5948647625
С	6.0 3.7178437246 -5.013834205	2 -0.7181896255
С	6.0 4.7058497720 -2.704078848	1 -0.7329505216
Η	1.0 2.9765832931 -0.977842584	6 -0.3363801321

Н	1.0	-0.9053484466	-3.9888472230	0.2241630872
Н	1.0	1.2720890009	-4.9553092233	-0.2026060070
Н	1.0	4.7343901025	-5.2086313133	-1.0405916227
Н	1.0	3.0358543197	-5.4308714598	-1.4590089409
Н	1.0	5.5759322718	-3.3251757759	-0.9124534916
Н	1.0	4.8776008554	-2.1267765473	0.1765313749
Н	1.0	4.5925641689	-2.0169264181	-1.5726585194
Н	1.0	3.5520561055	-5.5151101760	0.2374062477
0	8.0	1.5794396157	1.5410963810	-0.5856839307
0	8.0	1.1228597274	0.7462706044	1.8444796534
С	6.0	1.1984065019	2.0081745657	2.5748150631
С	6.0	1.8368795710	1.7356160420	3.9166054140
Н	1.0	1.7866168732	2.7132236715	1.9885446282
Н	1.0	0.1861793038	2.3975983769	2.6892198186
Н	1.0	2.8464068255	1.3439021035	3.7932138539
Н	1.0	1.2480209489	1.0191486176	4.4893013053
Н	1.0	1.8937134640	2.6658655008	4.4833297140

E(S0) = -1337.259355: ORMAS-PCM natural orbitals 91 / 92 occupations = 1.942 / 0.079 $E(S1^*) = -1337.191502$: ORMAS-PCM natural orbitals 91 / 92 occupations = 1.024 / 0.992

f. R6 (SO₂)



TDDFT-B3LYP-PCM(H₂O)

E(S0) = -1316.949052 $E(S1^*) = -1316.867343$ Oscillator strength = 0.9217 E(HOMO) = -0.2217; E(LUMO) = -0.1356

EXCITATION DE-EXCITATION AMPLITUDE OCC VIR AMPLITUDE Ι А X(I->A) Y(A->I)---_____ --- ------78 84 0.040368 -0.004842 0.012996 81 84 0.086272 83 84 -0.999111 0.150971 81 85 -0.054551 -0.028631

83	85	0.078604	0.013445
79	93	-0.032413	-0.019220

С	6.0	-1.1715837556	0.9561906046	0.1479396286
С	6.0	-2.0868354438	-0.1256352919	0.2617883714
С	6.0	-1.6826254294	-1.4628743511	0.3048939614
С	6.0	-0.3722527928	-1.9357579812	0.1853070106
С	6.0	0.7558830334	-1.0795077950	0.0615338272
S	16.0	0.5800212648	0.6742290049	0.3285971512
С	6.0	-2.9632747052	2.5822249286	0.0049712537
С	6.0	-3.8936056558	1.5103196129	0.1461946208
С	6.0	-3.4646956734	0.2202259859	0.2660191565
С	6.0	-1.5698871329	2.2528242515	0.0167077799
Η	1.0	-4.9518484741	1.7137803561	0.1564335150
Η	1.0	-4.1922405912	-0.5753884475	0.3579323031
Η	1.0	-0.8289471759	3.0302772233	-0.0779348205
Ν	7.0	-3.3655568312	3.8524416720	-0.1330702623
С	6.0	-2.3956365997	4.9453636590	-0.2331472189
С	6.0	-4.7906185783	4.1832756037	-0.1834924902
Η	1.0	-2.9321410396	5.8867640531	-0.2522961606
Η	1.0	-1.8052306421	4.8642320242	-1.1473050607
Η	1.0	-4.8947808531	5.2418419732	-0.3913501125
Η	1.0	-5.2906060498	3.6252884242	-0.9754137855
Η	1.0	-1.7254316858	4.9493353483	0.6262996925
Η	1.0	-5.2791446552	3.9641897594	0.7676115682
Η	1.0	-2.4640944383	-2.2080216867	0.3916401210
С	6.0	-0.1078898835	-3.3290492529	0.1097464005
С	6.0	2.0184813348	-1.5437032076	-0.1582086236
С	6.0	1.1464309432	-3.8247318088	-0.1000908994
С	6.0	2.2623592840	-2.9510552894	-0.2615566373
Ν	7.0	3.4959382464	-3.4161575766	-0.4995881861
С	6.0	3.7415041727	-4.8544710806	-0.6190491164
С	6.0	4.6387245829	-2.5069101781	-0.6114006276
Η	1.0	2.8339463585	-0.8451310428	-0.2567142319
Η	1.0	-0.9392046386	-4.0141628141	0.2113478339
Η	1.0	1.2858516936	-4.8921734667	-0.1475859393
Η	1.0	4.7666508551	-5.0073938010	-0.9361767136
Η	1.0	3.0805773073	-5.2985879961	-1.3626216033
Η	1.0	5.5349802404	-3.0941404854	-0.7743712560
Η	1.0	4.7682931458	-1.9274982506	0.3034592559
Η	1.0	4.5153993086	-1.8232167743	-1.4523415376
Η	1.0	3.5911293185	-5.3585510798	0.3373446565
0	8.0	0.9148975928	0.9402383782	1.7121753226
0	8.0	1.3001207225	1.3995081749	-0.6926710178

E(S0) = -1314.939256: ORMAS-PCM natural orbitals 83 / 84 occupations = 1.941 / 0.082 $E(SI^*) = -1314.872218$: ORMAS-PCM natural orbitals 83 / 84 occupations = 1.021 / 0.997

g. R7 (CO)



<u>TDDFT-B3LYP-PCM(H_2O)</u>

E(S0) = -881.696896 $E(S1^*) = -881.633414$ *Oscillator strength* = 0.4134E(HOMO) = -0.2145; E(LUMO) = -0.1409

EXCITATION DE-EXCITATION

OCC VIR AMPLITUDE AMPLITUDE

Ι	А	X(I->A)	Y(A->I)
69	75	-0.061780	-0.000915
71	75	-0.100900	-0.020839
74	75	0.991644	-0.145842
71	76	0.053784	0.034598
74	76	-0.139133	-0.028770

ATOM CHARGE X Y

Ζ

С	6.0 -1.2034425071	0.7691982394	-0.0449898645
С	6.0 -2.2053617971	-0.2437510631	-0.0081977412
С	6.0 -1.8447886866	-1.5945321176	-0.0286292581
С	6.0 -0.5174591035	-2.0275096269	-0.0833140722
С	6.0 0.5464688429	-1.0802773752	-0.1233196937
С	6.0 0.2402299540	0.3843377050	-0.1088102282
С	6.0 -2.9078844556	2.5103030043	0.0495966452
С	6.0 -3.9058611984	1.4930654817	0.0770387468
С	6.0 -3.5568122123	0.1702007706	0.0497504223
С	6.0 -1.5424017398	2.0950304407	-0.0200080341
Н	1.0 -4.9493530441	1.7595754845	0.1179770638

Η	1.0	-4.3322845391	-0.5844359892	0.0720884996
Η	1.0	-0.7485276554	2.8236531784	-0.0482766655
Ν	7.0	-3.2282632483	3.8149854620	0.0902347498
С	6.0	-2.1946889100	4.8497905108	0.0316122047
С	6.0	-4.6259786846	4.2329631518	0.1854749735
Н	1.0	-2.6730558161	5.8219501618	0.0663557091
Н	1.0	-1.6214366684	4.7810459892	-0.8938959631
Η	1.0	-4.6639532583	5.3121408369	0.2806956627
Η	1.0	-5.1859975104	3.9424833377	-0.7055927929
Η	1.0	-1.5107291744	4.7716580335	0.8779849091
Н	1.0	-5.1036707277	3.7914636651	1.0606727089
Η	1.0	-2.6304308232	-2.3403286275	0.0006052367
С	6.0	-0.1766386152	-3.4011946346	-0.0999410104
С	6.0	1.8510075985	-1.4899090557	-0.1767804179
С	6.0	1.1241363445	-3.8205039629	-0.1544174782
С	6.0	2.1922720311	-2.8781300090	-0.1968948318
Ν	7.0	3.4757827598	-3.2708627268	-0.2559706272
С	6.0	3.8175682012	-4.6913148057	-0.3206917558
С	6.0	4.5651932272	-2.2937580483	-0.2663143500
Н	1.0	2.6196352214	-0.7348876802	-0.2091887771
Н	1.0	-0.9706424758	-4.1359233887	-0.0666366604
Н	1.0	1.3350951580	-4.8775081640	-0.1605544964
Н	1.0	4.8894249776	-4.7873640332	-0.4500487483
Н	1.0	3.3232701028	-5.1718903867	-1.1657171644
Н	1.0	5.5096660324	-2.8247509101	-0.2354114803
Н	1.0	4.5116202125	-1.6389562274	0.6036698310
Н	1.0	4.5398473588	-1.6838451303	-1.1710615061
Н	1.0	3.5296237724	-5.2089765296	0.5964929630
0	8.0	1.1254564564	1.2226474892	-0.1462959683

E(S0) = -880.145356: ORMAS-PCM natural orbitals 74 / 75 occupations = 1.944 / 0.086 $E(S1^*) = -880.088493$: ORMAS-PCM natural orbitals 74 / 75 occupations = 1.019 / 1.003 II. Energies (S0* & S1 in hartree), oscillator strengths (S1 \rightarrow S0*), HOMO and LUMO energies (hartree), excitation/de-excitation amplitudes, and S1 optimized structures (Å)

a. R1 (NH)

Η



TDDFT-B3LYP-PCM(H2O)

 $E(S0^*) = -823.751804$ E(S1) = -823.659290 Oscillator strength = 1.2158E(HOMO) = -0.2093; E(LUMO) = -0.1003

EXCITATION DE-EXCITATION

OCC	V	'IR AMPLI	TUE	DE AMPLITU	JDE
Ι	А	X(I->A)	Y	(A->I)	
 71	72	0.995506	-0	.059008	
69	73	0.031813	0	.019036	
70	73	-0.072873	-(0.025075	
69	74	0.041775	0	.008802	
ATOM	1 C	HARGE	Х	Y	Z
С	6.0	-1.2434941	764	0.6555102058	-0.0470436165
С	6.0	-2.2849583	8635	-0.3221392607	-0.0094916940
С	6.0	-1.9488236	6656	-1.6931271856	-0.0361514567
С	6.0	-0.5992220	0001	-2.1014072238	-0.0912159301
С	6.0	0.4319643	031	-1.1127785979	-0.1266506782
Ν	7.0	0.0612407	653	0.2151048405	-0.1089456657
С	6.0	-2.8626633	3144	2.4704940255	0.0541532164
С	6.0	-3.9072635	5687	1.5017494106	0.0891659377
С	6.0	-3.6155763	3237	0.1602782365	0.0566597082
С	6.0	-1.5270806	5977	2.0112463067	-0.0188639316
Н	1.0	-4.9376733	3835	1.8141158068	0.1378537912
Н	1.0	-4.4220314	1293	-0.5607226128	0.0827094972
Н	1.0	-0.7070481	1430	2.7119583358	-0.0454343876
Ν	7.0	-3.1408826	5402	3.8056123098	0.0919623724
С	6.0	-2.0736144	639	4.7911982048	0.0077771919
С	6.0	-4.5152879	9088	4.2730323055	0.1977735191

1.0 -2.5065743503 5.7849462620 0.0346481781

Н	1.0 -1.50	094798942	4.6831146773	-0.9222461730
Н	1.0 -4.5	167547630	5.3542806678	0.2797862059
Н	1.0 -5.09	993311742	3.9902136367	-0.6826077957
Н	1.0 -1.3	770287898	4.6939891017	0.8448927601
Н	1.0 -5.00	028866764	3.8564039613	1.0815632200
Н	1.0 -2.73	319711434	-2.4375366216	-0.0093471997
С	6.0 -0.19	903185536	-3.4580038252	-0.1111116148
С	6.0 1.76	594077354	-1.4696894384	-0.1782058544
С	6.0 1.13	322751929	-3.8219990191	-0.1637057352
С	6.0 2.15	548963291	-2.8302110626	-0.2013157805
Ν	7.0 3.47	715577710	-3.1831039576	-0.2585109589
С	6.0 3.86	525937964	-4.5839294501	-0.3185188371
С	6.0 4.51	47770921	-2.1689187618	-0.2499128484
Н	1.0 2.51	30533547	-0.6887163586	-0.2052291369
Н	1.0 -0.95	532009590	-4.2247835495	-0.0813648343
Н	1.0 1.38	889439993	-4.8688166817	-0.1719644143
Н	1.0 4.93	384235399	-4.6445704411	-0.4386137931
Н	1.0 3.38	395534222	-5.0863384229	-1.1648074022
Н	1.0 5.48	827955990	-2.6571643221	-0.2617732805
Н	1.0 4.45	519521422	-1.5453116641	0.6454987212
Н	1.0 4.44	39331705	-1.5185374636	-1.1263042165
Н	1.0 3.58	841974448	-5.1151032511	0.5962441918
Н	1.0 0.79	926669646	0.9096043368	-0.1324805349

 $E(S0^*) = -822.282254$: ORMAS-PCM natural orbitals 71 / 72 occupations = 1.938 / 0.074 E(S1) = -822.191649 : ORMAS-PCM natural orbitals 71 / 72 occupations = 1.197 / 0.817

b. R2 (O)



TDDFT-B3LYP-PCM(H₂O)

 $E(S0^*) = -843.600340$ E(S1) = -843.513375 Oscillator strength = 1.3147E(HOMO) = -0.2133; E(LUMO) = -0.1115

EXCITATION DE-EXCITATION

OCC VIR AMPLITUDE AMPLITUDE

Ι	А	X(I->A)	Y(A->I)
71	72	-0.998267	0.072716
69	73	0.035667	0.020514
70	73	0.036749	0.017419
69	75	0.030021	0.006552

6.0 -1.2755569776 0.5844106185 -0.0428430322	
6.0 -2.3232886781 -0.3732412156 -0.0043803752	
6.0 -2.0059623276 -1.7471524338 -0.0344944940	
$6.0 \ \textbf{-}0.6517921717 \ \textbf{-}2.1360525889 \ \textbf{-}0.0921887659$	
6.0 0.3599687787 -1.1403048662 -0.1245159556	
8.0 0.0356781724 0.1910678717 -0.1067265126	
6.0 -2.8329741871 2.4342599670 0.0582204072	
6.0 -3.8994189434 1.4875761093 0.0988488350	
6.0 -3.6434030322 0.1412727236 0.0663831647	
6.0 -1.5059614275 1.9394599721 -0.0163423069	
1.0 -4.9216919852 1.8251348164 0.1517610877	
1.0 -4.4668601101 -0.5597919165 0.0956283762	
1.0 -0.6568503901 2.6029402032 -0.0476213187	
7.0 -3.0772241971 3.7711631859 0.0901765200	
6.0 -1.9841754063 4.7286824523 -0.0009688478	
6.0 -4.4406013999 4.2750682210 0.1878619840	
1.0 -2.3908090385 5.7333640807 0.0231281964	
1.0 -1.4271702163 4.6004182393 -0.9325254445	
1.0 -4.4140910291 5.3568015348 0.2557149037	
1.0 -5.0287097102 3.9953079211 -0.6905143736	
1.0 -1.2884281229 4.6155194183 0.8345351648	
1.0 -4.9399533576 3.8825983089 1.0760888352	
1.0 -2.7880831193 -2.4916019249 -0.0089440824	
$6.0 \ \textbf{-}0.2107822065 \ \textbf{-}3.4837409977 \ \textbf{-}0.1178779634$	
6.0 1.6998736888 -1.4444715605 -0.1756209977	
6.0 1.1185432816 -3.8132652163 -0.1711174221	
6.0 2.1206711783 -2.7982423682 -0.2032262072	
7.0 3.4413870101 -3.1166454530 -0.2586938435	
6.0 3.8677254140 -4.5082110147 -0.3163503377	
6.0 4.4596872714 -2.0760137148 -0.2453134002	
1.0 2.4087224760 -0.6327535355 -0.1985448871	
$1.0 \ -0.9553993827 \ -4.2681846719 \ -0.0918658488$	
1.0 1.3998544405 -4.8536336462 -0.1844849361	
1.0 4.9455710624 -4.5424251391 -0.4279331605	
1.0 3.4133144793 -5.0207821618 -1.1665011213	
	6.0 -1.2755569776 0.5844106185 -0.0428430322 6.0 -2.3232886781 -0.3732412156 -0.0043803752 6.0 -2.0059623276 -1.7471524338 -0.0344944940 6.0 -0.6517921717 -2.1360525889 -0.0921887659 6.0 0.3599687787 -1.1403048662 -0.1245159556 8.0 0.0356781724 0.1910678717 -0.1067265126 6.0 -2.8329741871 2.4342599670 0.0582204072 6.0 -3.8994189434 1.4875761093 0.0988488350 6.0 -3.6434030322 0.1412727236 0.0663831647 6.0 -1.5059614275 1.9394599721 -0.0163423069 1.0 -4.9216919852 1.8251348164 0.1517610877 1.0 -4.4668601101 -0.5597919165 0.0956283762 1.0 -0.6568503901 2.6029402032 -0.0476213187 7.0 -3.0772241971 3.7711631859 0.0901765200 6.0 -1.9841754063 4.7286824523 -0.0009688478 6.0 -4.4406013999 4.2750682210 0.1878619840 1.0 -2.3908090385 5.7333640807 0.0231281964 1.0 -1.4271702163 4.6004182393 -0.9325254445 1.0 -4.4140910291 5.3568015348 0.2557149037 1.0 -5.0287097102 3.9953079211 -0.6905143736 1.0 -2.7880831193 -2.4916019249 -0.0089440824 6.0 -0.2107822065 -3.4837409977 <t< td=""></t<>

Н	1.0	5.4391812314	-2.5403571844	-0.2644347969
Н	1.0	4.3834788719	-1.4618659345	0.6555264922
Н	1.0	4.3684516733	-1.4214636080	-1.1163772174

H 1.0 3.5951157371 -5.0448669118 0.5966625321

<u>ORMAS-PT2-PCM(H₂O) at TDDFT-B3LYP-PCM(H₂O) optimized geometry</u>

 $E(S0^*) = -842.114045$: ORMAS-PCM natural orbitals 71 / 72 occupations = 1.945 / 0.069 E(S1) = -842.029965: ORMAS-PCM natural orbitals 71 / 72 occupations = 1.023 / 0.988

c. R3 (CMe₂)



TDDFT-B3LYP-PCM(H₂O)

 $E(S0^*) = -886.267502$ E(S1) = -886.186503 Oscillator strength = 1.3897E(HOMO) = -0.2075; E(LUMO) = -0.1132

С	6.0 -1.	1968613423	0.7819047415	-0.1091392652
С	6.0 -2.1	1925478885	-0.2295478929	0.0423808544
С	6.0 -1.8	8449829415	-1.5952281862	0.0687908457
С	6.0 -0.5	5059503406	-2.0162359651	-0.0482293453
С	6.0 0.5	535810235	-1.0722760695	-0.2020719300
С	6.0 0.2	2885975717	0.4367611209	-0.2615194248
С	6.0 -2.9	9430358280	2.5066536808	0.0100146921
С	6.0 -3.9	9274306817	1.4842818351	0.1535305592
С	6.0 -3.5	5521725327	0.1713878550	0.1682358073
С	6.0 -1.5	5877056856	2.1061341086	-0.1233321802
Н	1.0 -4.9	9704905296	1.7393112618	0.2513585369
Н	1.0 -4.	3064711279	-0.5968151203	0.2793921531
Н	1.0 -0.3	8371560453	2.8705823539	-0.2380452970

Ν	7.0	-3.2938329355	3.8192100418	0.0037873075
С	6.0	-2.2857209571	4.8604519825	-0.1462817150
С	6.0	-4.6861118761	4.2193677039	0.1653402665
Н	1.0	-2.7737948294	5.8284674850	-0.1391285182
Н	1.0	-1.7442194466	4.7522364009	-1.0890619866
Н	1.0	-4.7424450475	5.3020169094	0.1686994802
Н	1.0	-5.3047808175	3.8421167943	-0.6529311087
Н	1.0	-1.5633095682	4.8293706864	0.6733784633
Н	1.0	-5.0944866831	3.8469166439	1.1074952016
Н	1.0	-2.6233966771	-2.3368055147	0.1845881525
С	6.0	-0.1799736627	-3.4017522415	-0.0134430907
С	6.0	1.8494700346	-1.5364028315	-0.3082025782
С	6.0	1.1050483465	-3.8494046912	-0.1223401175
С	6.0	2.1748030366	-2.9183955768	-0.2734988718
Ν	7.0	3.4603390350	-3.3430640497	-0.3807805711
С	6.0	3.7783736415	-4.7655679820	-0.4014595430
С	6.0	4.5560407850	-2.3880992142	-0.4758298945
Н	1.0	2.6495052790	-0.8247714921	-0.4296802488
Н	1.0	-0.9843038942	-4.1162818978	0.1055542644
Н	1.0	1.3030470353	-4.9087814639	-0.0857706057
Н	1.0	4.8415775103	-4.8848223996	-0.5768239387
Н	1.0	3.2363799093	-5.2774137572	-1.1993172981
Н	1.0	5.4964409626	-2.9274906947	-0.4743129234
Н	1.0	4.5495652289	-1.7003372319	0.3723303782
Н	1.0	4.4940218793	-1.8020988837	-1.3971045594
Н	1.0	3.5272073681	-5.2446394948	0.5490311118
С	6.0	0.7866335375	0.9762498802	-1.6308403155
С	6.0	1.0857648594	1.1305025134	0.8769516866
Н	1.0	0.2473291229	0.5040612228	-2.4523330658
Н	1.0	0.6382561462	2.0530369396	-1.6984986693
Н	1.0	1.8495551767	0.7781294760	-1.7608117239
Н	1.0	2.1522451715	0.9345209636	0.7784976499
Н	1.0	0.7596594182	0.7703317109	1.8528394045
Η	1.0	0.9454895189	2.2098846795	0.8468208198

 $E(S0^*) = -884.671127$: ORMAS-PCM natural orbitals 79 / 80 occupations = 1.945 / 0.069 E(S1) = -884.597771: ORMAS-PCM natural orbitals 79 / 80 occupations = 1.020 / 0.991

d. R4 (SiMe₂)



TDDFT-B3LYP-PCM(H₂O)

 $E(S0^*) = -1137.677685$ E(S1) = -1137.600506*Oscillator strength* = 1.3907E(HOMO) = -0.2084; E(LUMO) = -0.1180

EXCITATION DE-EXCITATION OCC VIR AMPLITUDE AMPLITUDE A $X(I \rightarrow A)$ $Y(A \rightarrow I)$ Ι ------ -----84 -0.034581 0.005232 78 83 84 0.999707 -0.090040

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С	6.0 -1.2147714315	5 1.0275961553	0.0069642133	
С	6.0 -2.1102480726	6 -0.0962401736	0.0626204198	
С	6.0 -1.6913247871	1 -1.4482053224	0.0546472410	
С	6.0 -0.3675570788	3 -1.9434693186	-0.0305589655	
С	6.0 0.8034334456	5 -1.1123787090	-0.1033821997	
SI	14.0 0.646747759	8 0.7601549945	-0.0491282835	
С	6.0 -3.1388426899	9 2.5657320163	0.0567151636	
С	6.0 -4.0141005705	5 1.4382793175	0.1197553704	
С	6.0 -3.5103774762	0.1700592234	0.1225627309	
С	6.0 -1.7414964495	5 2.3067335817	0.0045472919	
Н	1.0 -5.082087146	5 1.5806824158	0.1651279218	
Н	1.0 -4.1947400414	4 -0.6676194802	0.1686944133	
Н	1.0 -1.067907225	1 3.1502467830	-0.0380503092	
Ν	7.0 -3.6310705944	4 3.8286349879	0.0497976656	
С	6.0 -2.7356381932	4.9766932509	-0.0062922732	
С	6.0 -5.0682473383	3 4.0707891071	0.1175399938	
Н	1.0 -3.3255271564	4 5.8861312974	-0.0179568072	
Н	1.0 -2.1191755803	5 4.9502248372	-0.9078855770	
Н	1.0 -5.2494651258	8 5.1393662841	0.0868648882	
Н	1.0 -5.5850170568	8 3.6073497006	-0.7256892599	
Н	1.0 -2.073776025	1 5.0012544084	0.8630847815	
Η	1.0	-5.4908094532	3.6730004812	1.0432478300
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Η	1.0	-2.4770109385	-2.1913964263	0.1056391448
С	6.0	-0.1823943148	-3.3572634838	-0.0555337069
С	6.0	2.0467532609	-1.7097880187	-0.2070297134
С	6.0	1.0521131376	-3.9314286970	-0.1552070457
С	6.0	2.2246898405	-3.1191420187	-0.2386422101
Ν	7.0	3.4547538122	-3.6796032831	-0.3432386605
С	6.0	3.6195118246	-5.1288053395	-0.3404259729
С	6.0	4.6451375612	-2.8470393601	-0.4600307500
Η	1.0	2.9237039563	-1.0817324312	-0.2611939134
Η	1.0	-1.0559073033	-3.9942611190	0.0012767885
Η	1.0	1.1326772456	-5.0066038371	-0.1760443540
Η	1.0	4.6778463180	-5.3623512176	-0.3695505066
Η	1.0	3.1391922751	-5.5834871648	-1.2106392683
Η	1.0	5.5122702002	-3.4855970548	-0.5853997247
Η	1.0	4.7893138214	-2.2356390190	0.4348882683
Η	1.0	4.5753026941	-2.1820785195	-1.3237712105
Η	1.0	3.1917990612	-5.5703435802	0.5617749129
С	6.0	1.3998102355	1.5502879595	-1.5824691920
С	6.0	1.4704531729	1.4624522059	1.4914171181
Η	1.0	1.2398497502	2.6306467912	-1.5806259881
Η	1.0	2.4776388473	1.3783438027	-1.6206954458
Η	1.0	2.5446623921	1.2660970869	1.4823012146
Η	1.0	1.3319542805	2.5446347036	1.5448853572
Η	1.0	1.0512236991	1.0217811835	2.3970954711
Η	1.0	0.9580706372	1.1428218684	-2.4929125928

<u>ORMAS-PT2-PCM(H₂O) at TDDFT-B3LYP-PCM(H₂O) optimized geometry</u>

 $E(S0^*) = -1135.720747$: ORMAS-PCM natural orbitals 83 / 84 occupations = 1.944 / 0.073E(S1) = -1135.650797: ORMAS-PCM natural orbitals 83 / 84 occupations = 1.020 / 0.993

e. R5 (POOEt)



<u>TDDFT-B3LYP-PCM(H₂O)</u>

 $E(S0^*) = -1339.409126$ E(S1) = -1339.338413 Oscillator strength = 1.1349E(HOMO) = -0.2161; E(LUMO) = -0.1327

EXCITATION DE-EXCITATION

ATOM CHARGE X Y Z

С	6.0	-1.0005154787	0.9876924978	0.2657291401
С	6.0	-1.9374171612	-0.0884804299	0.4419522455
С	6.0	-1.5570610279	-1.4445888201	0.5153546202
С	6.0	-0.2518440397	-1.9353687919	0.2995080562
С	6.0	0.8949037286	-1.0886550546	0.1250413241
Р	15.0	0.7582160937	0.6699144454	0.4243489462
С	6.0	-2.8221663232	2.5955628811	0.0197523553
С	6.0	-3.7510312512	1.5346685195	0.2701056897
С	6.0	-3.3180870863	0.2557589029	0.4681452873
С	6.0	-1.4379793881	2.2758005585	0.0400875774
Н	1.0	-4.8100234856	1.7370555317	0.2834269330
Н	1.0	-4.0441640370	-0.5313389065	0.6233919414
Н	1.0	-0.7052668035	3.0520214295	-0.1157967908
Ν	7.0	-3.2525117977	3.8535716687	-0.2167547189
С	6.0	-2.3022318560	4.9161110177	-0.5259552093
С	6.0	-4.6747530000	4.1837537872	-0.1826840020
Н	1.0	-2.8494387552	5.8213536562	-0.7623324840
Н	1.0	-1.6848669559	4.6456271938	-1.3846114488
Н	1.0	-4.7892870354	5.2501860939	-0.3380417253
Н	1.0	-5.2192005931	3.6562544457	-0.9695232823
Н	1.0	-1.6462640427	5.1167040968	0.3251863743
Н	1.0	-5.1147671634	3.9234640336	0.7814521979
Н	1.0	-2.3431811117	-2.1739713539	0.6570022452
С	6.0	-0.0378191118	-3.3378182046	0.1869252956
С	6.0	2.1235009420	-1.6152284287	-0.2117429607
С	6.0	1.1815985576	-3.8630241163	-0.1289420610
С	6.0	2.3115821654	-3.0145272806	-0.3634515962
Ν	7.0	3.5140489274	-3.5329606823	-0.6955836901
С	6.0	3.7079799611	-4.9760681758	-0.8083356907
С	6.0	4.6647138851	-2.6618566625	-0.9081217972
Η	1.0	2.9558790158	-0.9426411591	-0.3472494943
Η	1.0	-0.8782173253	-4.0022019552	0.3382080142
Н	1.0	1.2852254111	-4.9331714832	-0.2090384259
Η	1.0	4.7184415653	-5.1662649982	-1.1513343241
Η	1.0	3.0109187532	-5.4091415613	-1.5271045693

Н	1.0	5.5206641162	-3.2678245482	-1.1814126113
Н	1.0	4.9065665452	-2.1041698340	-0.0000201001
Н	1.0	4.4705534859	-1.9481380029	-1.7116179069
Н	1.0	3.5678710997	-5.4685019756	0.1569685638
0	8.0	1.6213265589	1.5624795285	-0.3871783692
0	8.0	1.0830128272	0.7187515041	1.9954290858
С	6.0	1.1963942253	1.9802273796	2.7158418215
С	6.0	1.7972687906	1.6980344917	4.0737955084
Н	1.0	1.8226950928	2.6568479373	2.1353508608
Н	1.0	0.1993077060	2.4138745234	2.8081325251
Н	1.0	2.7921768941	1.2638667008	3.9745121309
Н	1.0	1.1695686605	1.0115940463	4.6423336123
Η	1.0	1.8814100916	2.6306820938	4.6333107764

<u>ORMAS-PT2-PCM(H₂O) at TDDFT-B3LYP-PCM(H₂O) optimized geometry</u>

 $E(S0^*) = -1337.257894$: ORMAS-PCM natural orbitals 91 / 92 occupations = 1.941 / 0.081 E(S1) = -1337.193723: ORMAS-PCM natural orbitals 91 / 92 occupations = 1.023 / 0.995

f. R6 (SO₂)



TDDFT-B3LYP-PCM(H₂O)

 $E(S0^*) = -1316.946196$ E(S1) = -1316.877747 Oscillator strength = 1.0264E(HOMO) = -0.2216; E(LUMO) = -0.1404

EXCITATION DE-EXCITATION						
OCO	C V	IR AMPLI	TUDE	AMPL	ITUDE	
Ι	А	X(I->A)	Y(A	>I)		
79	84	0.046159	-0.0	01455		
81	84	0.061224	0.0	08771		
83	84	0.999317	-0.1	06741		
81	85	-0.031286	-0.0	21279		
83	85	-0.037629	-0.0	05749		
ATO	M C	HARGE	Х	Y	Ζ	

С	6.0	-1.1568498409	0.9115469366	0.4079459283
С	6.0	-2.1033103295	-0.1616204350	0.4900852224
С	6.0	-1.7243881673	-1.5143487408	0.5621022522
С	6.0	-0.3996575764	-1.9608514473	0.3913166976
С	6.0	0.7238475108	-1.0751797233	0.3151333521
S	16.0	0.5227806774	0.5964284073	0.8337325779
С	6.0	-2.8999680271	2.5421149763	-0.0109245949
С	6.0	-3.8633177432	1.4928334439	0.1544780773
С	6.0	-3.4732965803	0.2066125719	0.3909854728
С	6.0	-1.5280394413	2.2077012680	0.1443488216
Η	1.0	-4.9154122978	1.7117373383	0.0692759266
Η	1.0	-4.2219605639	-0.5696163424	0.4744913586
Η	1.0	-0.7663089236	2.9681787879	0.0837709541
Ν	7.0	-3.2872012012	3.8034935587	-0.2866255299
С	6.0	-2.2959600565	4.8515142486	-0.5091109960
С	6.0	-4.7020938525	4.1589405721	-0.3666611971
Η	1.0	-2.8033474208	5.7571598515	-0.8202955659
Η	1.0	-1.5943885224	4.5564333101	-1.2907985177
Η	1.0	-4.7833437149	5.2266666192	-0.5332969421
Η	1.0	-5.1908168925	3.6390080237	-1.1936819925
Η	1.0	-1.7344354510	5.0607657324	0.4051318757
Η	1.0	-5.2205561967	3.9106716922	0.5607041540
Η	1.0	-2.5049772180	-2.2586100542	0.6333079236
С	6.0	-0.1221355467	-3.3392195994	0.1817577255
С	6.0	1.9842312212	-1.5035923155	-0.0264355643
С	6.0	1.1255907113	-3.7866076477	-0.1431580912
С	6.0	2.2265485707	-2.8792242926	-0.2847036382
Ν	7.0	3.4511106597	-3.3231722407	-0.6333071168
С	6.0	3.6999267034	-4.7452928737	-0.8610133252
С	6.0	4.5679969734	-2.3940131780	-0.7670023014
Η	1.0	2.7863736714	-0.7850835164	-0.0779570192
Η	1.0	-0.9385143097	-4.0451516519	0.2534665339
Η	1.0	1.2744351705	-4.8413010364	-0.3085161551
Η	1.0	4.7418565857	-4.8791083970	-1.1269123459
Η	1.0	3.0801851984	-5.1224210034	-1.6765373238
Η	1.0	5.4410211885	-2.9377294226	-1.1082018988
Η	1.0	4.8000936302	-1.9230041953	0.1914846877
Η	1.0	4.3388048126	-1.6111601176	-1.4927479625
Η	1.0	3.4903052145	-5.3281763109	0.0378439468
0	8.0	0.6319401299	0.6331732089	2.2814577316
0	8.0	1.4062879245	1.4679353741	0.0866239876

<u>ORMAS-PT2-PCM(H₂O) at TDDFT-B3LYP-PCM(H₂O) optimized geometry</u>

 $E(S0^*) = -1314.937967$: ORMAS-PCM natural orbitals 83 / 84 occupations = 1.938 / 0.086E(S1) = -1314.876096: ORMAS-PCM natural orbitals 83 / 84 occupations = 1.029 / 0.992 g. R7 (CO)



TDDFT-B3LYP-PCM(H₂O)

E(S0) = -881.693547 $E(S1^*) = -881.641925$ Oscillator strength = 0.4808E(HOMO) = -0.2137; E(LUMO) = -0.1461

EXCITATION DE-EXCITATION

OCC VIR AMPLITUDE AMPLITUDE

Ι	А	X(I->A)	Y(A->I)
69	75	-0.057007	-0.001654
71	75	0.068986	0.014047
74	75	0.998298	-0.112927
71	76	-0.031047	-0.026474
74	76	-0.071029	-0.015802

ATOM CHARGE X Y Z

6.0 -1.1741695982	0.7639262670	-0.0474247369
6.0 -2.1981986903	-0.2511697300	-0.0089774924
6.0 -1.8589870663	-1.6081777584	-0.0281540467
6.0 -0.5242123799	-2.0193735948	-0.0834717457
6.0 0.5428377495	-1.0497694651	-0.1249938904
6.0 0.2398903392	0.3846032459	-0.1107223583
6.0 -2.8812765802	2.5041179819	0.0499233022
6.0 -3.8963327945	1.4822486903	0.0782845280
6.0 -3.5567551698	0.1653341627	0.0497011499
6.0 -1.5265380379	2.0981691811	-0.0203959539
1.0 -4.9375803402	1.7576035718	0.1192201207
1.0 -4.3313967513	-0.5893872824	0.0712995844
1.0 -0.7315430490	2.8246028326	-0.0476469458
7.0 -3.2244820113	3.8041225151	0.0901573563
6.0 -2.1987540054	4.8429415944	0.0283845531
6.0 -4.6234716097	4.2225807909	0.1876968876
	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

Н	1.0	-2.6781236070	5.8142934488	0.0576645156
Н	1.0	-1.6219850549	4.7631092384	-0.8948131290
Н	1.0	-4.6571265565	5.2999189482	0.3007018971
Н	1.0	-5.1785009431	3.9495896637	-0.7118385682
Н	1.0	-1.5141823634	4.7628845037	0.8751241266
Н	1.0	-5.1049274473	3.7686625512	1.0535265600
Н	1.0	-2.6432121871	-2.3528792721	0.0015134948
С	6.0	-0.1811407504	-3.4003363610	-0.0993637953
С	6.0	1.8550522564	-1.4735483694	-0.1804987030
С	6.0	1.1141453346	-3.8098361883	-0.1548631885
С	6.0	2.1875940916	-2.8505538510	-0.1999985330
Ν	7.0	3.4650604450	-3.2661337134	-0.2572910319
С	6.0	3.8068629654	-4.6881479506	-0.3230967586
С	6.0	4.5589215820	-2.2980063532	-0.2593516070
Н	1.0	2.6215481863	-0.7173903862	-0.2154502280
Н	1.0	-0.9751919465	-4.1340883534	-0.0637301494
Н	1.0	1.3340577987	-4.8650479876	-0.1589559806
Н	1.0	4.8760078997	-4.7802573874	-0.4737221022
Н	1.0	3.2978217880	-5.1729527396	-1.1558675039
Н	1.0	5.5016803013	-2.8308902463	-0.2236246364
Н	1.0	4.4941907314	-1.6408575136	0.6088310245
Н	1.0	4.5340903531	-1.6857423199	-1.1635777437
Н	1.0	3.5402004196	-5.2003192384	0.6035112653
0	8.0	1.1447620984	1.2420393241	-0.1484187969

<u>ORMAS-PT2-PCM(H₂O) at B3LYP-PCM(H₂O) optimized geometry</u>

E(S0) = -880.142357: ORMAS-PCM natural orbitals 74 / 75 occupations = 1.940 / 0.094 $E(S1^*) = -880.091888$: ORMAS-PCM natural orbitals 74 / 75 occupations = 1.019 / 1.006

Section S9: References

- Grimm, J. B., B. P. English, J. Chen, J. P. Slaughter, Z. Zhang, A. Revyakin, R. Patel, J. J. Macklin, D. Normanno, R. H. Singer, T. Lionnet and L. D. Lavis (2015) A general method to improve fluorophores for live-cell and single-molecule microscopy. *Nature Methods* 12, 244-250.
- Koide, Y., Y. Urano, K. Hanaoka, T. Terai and T. Nagano (2011) Evolution of Group 14 Rhodamines as Platforms for Near-Infrared Fluorescence Probes Utilizing Photoinduced Electron Transfer. ACS Chemical Biology 6, 600-608.
- 3. Arden-Jacob, J., J. Frantzeskos, N. U. Kemnitzer, A. Zilles and K. H. Drexhage (2001) New fluorescent markers for the red region. *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy* **57**, 2271-2283.
- Pastierik, T., P. Šebej, J. Medalová, P. Štacko and P. Klán (2014) Near-Infrared Fluorescent 9-Phenylethynylpyronin Analogues for Bioimaging. *The Journal of Organic Chemistry* 79, 3374-3382.
- 5. Zhou, X., R. Lai, J. R. Beck, H. Li and C. I. Stains (2016) Nebraska Red: a phosphinate-based near-infrared fluorophore scaffold for chemical biology applications. *Chemical Communications* **52**, 12290-12293.
- Deng, F., L. Liu, W. Huang, C. Huang, Q. Qiao and Z. Xu (2020) Systematic study of synthesizing various heteroatom-substituted rhodamines from diaryl ether analogues. Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy 240, 118466.
- 7. Schmidt, M. W., K. K. Baldridge, J. A. Boatz, S. T. Elbert, M. S. Gordon, J. H. Jensen, S. Koseki, N. Matsunaga, K. A. Nguyen, S. J. Su, T. L. Windus, M. Dupuis and J. A. Montgomery (1993) General Atomic and Molecular Electronic-Structure System. J Comput Chem 14, 1347-1363.
- Barca, G. M. J., C. Bertoni, L. Carrington, D. Datta, N. De Silva, J. E. Deustua, D. G. Fedorov, J. R. Gour, A. O. Gunina, E. Guidez, T. Harville, S. Irle, J. Ivanic, K. Kowalski, S. S. Leang, H. Li, W. Li, J. J. Lutz, I. Magoulas, J. Mato, V. Mironov, H. Nakata, B. Q. Pham, P. Piecuch, D. Poole, S. R. Pruitt, A. P. Rendell, L. B. Roskop, K. Ruedenberg, T. Sattasathuchana, M. W. Schmidt, J. Shen, L. Slipchenko, M. Sosonkina, V. Sundriyal, A. Tiwari, J. L. G. Vallejo, B. Westheimer, M. Wloch, P. Xu, F. Zahariev and M. S. Gordon (2020) Recent developments in the general atomic and molecular electronic structure system. J Chem Phys 152.
- 9. Dunning, T. H. (1989) Gaussian-Basis Sets for Use in Correlated Molecular Calculations .1. The Atoms Boron through Neon and Hydrogen. *J Chem Phys* **90**, 1007-1023.
- Dunning, T. H., K. A. Peterson and A. K. Wilson (2001) Gaussian basis sets for use in correlated molecular calculations. X. The atoms aluminum through argon revisited. J Chem Phys 114, 9244-9253.
- 11. Bode, B. M. and M. S. Gordon (1998) MacMolPlt: A graphical user interface for GAMESS. *J Mol Graph Model* 16, 133-+.
- 12. Kohn, W. and L. J. Sham (1965) Self-Consistent Equations Including Exchange and Correlation Effects. *Phys Rev* 140, 1133-&.
- 13. Becke, A. D. (1993) Density-Functional Thermochemistry .3. The Role of Exact Exchange. J Chem Phys 98, 5648-5652.

- Stephens, P. J., F. J. Devlin, C. F. Chabalowski and M. J. Frisch (1994) Ab-Initio Calculation of Vibrational Absorption and Circular-Dichroism Spectra Using Density-Functional Force-Fields. J Phys Chem-Us 98, 11623-11627.
- 15. Hertwig, R. H. and W. Koch (1997) On the parameterization of the local correlation functional. What is Becke-3-LYP? *Chem Phys Lett* **268**, 345-351.
- 16. Runge, E. and E. K. U. Gross (1984) Density-Functional Theory for Time-Dependent Systems. *Phys Rev Lett* **52**, 997-1000.
- 17. Gross, E. K. U. and W. Kohn (1990) Time-Dependent Density-Functional Theory. In *Advances in Quantum Chemistry*, Vol. 21. pp. 225-291. Acad.
- 18. LEEUWEN, R. V. (2001) KEY CONCEPTS IN TIME-DEPENDENT DENSITY-FUNCTIONAL THEORY. *International Journal of Modern Physics B* **15**, 1969-2023.
- 19. Chiba, M., T. Tsuneda and K. Hirao (2006) Excited state geometry optimizations by analytical energy gradient of long-range corrected time-dependent density functional theory. *J Chem Phys* **124**.
- 20. Murray, C. W., N. C. Handy and G. J. Laming (1993) Quadrature Schemes for Integrals of Density Functional Theory. *Mol Phys* **78**, 997-1014.
- 21. Lebedev, V. I. and D. N. Laikov (1999) Quadrature formula for the sphere of 131-th algebraic order of accuracy. *Dokl Akad Nauk*+ **366**, 741-745.
- 22. Ivanic, J. (2003) Direct configuration interaction and multiconfigurational self-consistentfield method for multiple active spaces with variable occupations. I. Method. *J Chem Phys* **119**, 9364-9376.
- Ivanic, J. (2003) Direct configuration interaction and multiconfigurational self-consistentfield method for multiple active spaces with variable occupations. II. Application to oxoMn(salen) and N2O4. J Chem Phys 119, 9377-9385.
- Roskop, L. and M. S. Gordon (2011) Quasi-degenerate second-order perturbation theory for occupation restricted multiple active space self-consistent field reference functions. J Chem Phys 135.
- 25. Miertus, S., E. Scrocco and J. Tomasi (1981) Electrostatic Interaction of a Solute with a Continuum a Direct Utilization of Abinitio Molecular Potentials for the Prevision of Solvent Effects. *Chem Phys* 55, 117-129.
- 26. Li, H. and J. H. Jensen (2004) Improving the efficiency and convergence of geometry optimization with the polarizable continuum model: New energy gradients and molecular surface tessellation. *J Comput Chem* **25**, 1449-1462.
- 27. Tomasi, J., B. Mennucci and R. Cammi (2005) Quantum mechanical continuum solvation models. *Chem Rev* **105**, 2999-3093.
- 28. Li, H. (2009) Quantum mechanical/molecular mechanical/continuum style solvation model: Linear response theory, variational treatment, and nuclear gradients. *J Chem Phys* 131.
- 29. Wang, Y. L. and H. Li (2010) Excited state geometry of photoactive yellow protein chromophore: A combined conductorlike polarizable continuum model and time-dependent density functional study. *J Chem Phys* **133**.
- 30. Mulliken, R. S. (1955) Electronic Population Analysis on Leao-Mo Molecular Wave Functions .1. *J Chem Phys* 23, 1833-1840.