Supporting Information

Electrochemical Synthesis of Glycosyl Fluorides Using Sulfur(VI) Hexafluoride as the Fluorinating Agent

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I. General Information

Methods and Reagents:

Unless otherwise stated, all reagents were purchased from commercial suppliers and used without further purification. All reactions were carried out in flame- or oven-dried glassware with magnetic stirring. Reactions were cooled using external cooling baths: ice water $(0^{\circ}C)$. Deionized water was used in the preparation of all aqueous solutions and for all aqueous extractions. Solvents used for extraction and chromatography were ACS or HPLC grade. Purification of reaction mixtures was performed by flash column chromatography (FCC) using SiliCycle SilicaFlash P60 (230-400 mesh). Diastereomeric ratios were determined by ¹H NMR or ¹⁹F NMR analysis.

Instrumentation:

¹H NMR spectra were recorded on Varian vnmrs 700 (700 MHz), Varian vnmrs 600 (600 MHz), or Varian vnmrs 500 (500 MHz) spectrometers and chemical shifts (δ) are reported in parts per million (ppm) with solvent resonance as the internal standard (CDCl₃ at δ 7.26, CD₃CN at δ 1.94). Data are reported as (br = broad, $s =$ singlet, $d =$ doublet, $t =$ triplet, $q =$ quartet, $m =$ multiplet; coupling constant(s) in Hz; integration). Protondecoupled 13C NMR spectra were recorded on Varian vnmrs 700 (700 MHz), Varian vnmrs 600 (600 MHz), or Varian vnmrs 500 (500 MHz) spectrometers and chemical shifts (δ) are reported in ppm with solvent resonance as the internal standard (CDCl₃ at δ 77.0, CD₃CN at δ 118.7). ¹⁹F NMR spectra were recorded on Varian vnmrs 600 (600 MHz), Varian vnmrs 500 (500 MHz), or Varian vnmrs 400 (400 MHz) spectrometers and chemical shifts (δ) are reported in parts per million (ppm) and are referenced to CFCl₃ (δ 0.0). High resolution mass spectra (HRMS) were recorded on Micromass AutoSpec Ultima or VG (Micromass) 70-250-S Magnetic sector mass spectrometers in the University of Michigan mass spectrometry laboratory. Infrared (IR) spectra were recorded as thin films on NaCl plates on a Perkin Elmer Spectrum BX FT-IR spectrometer. Absorption peaks were reported in wavenumbers $(cm⁻¹)$.

Information of the DC power supply equipment:

- a. Tekpower TP3005P
	- Output Voltage: 0-30 V
	- Output Current: 0-5 A
	- Setup Resolution: 10 mV / 1 mA
	- Setup Accuracy: $\leq 0.5 \% + 20$ mV $\frac{1}{\leq} 0.5 \% + 10$ mA

Assignment of Stereochemistry:

Stereochemistry of glycosyl fluoride products (except mannose and rhamnose) are characterized by: 1. One-bond C-F coupling of anomeric carbon in H-coupled ¹³C NMR; α -glycosyl fluoride, ¹J_{C-F} = 224 - 228 Hz and β -glycosyl fluoride, 1 J_{C-F} = 213 - 219 Hz.

2. Three-bond H-H coupling between anomeric hydrogen and C2-hydrogen in ¹H NMR; α -glycosyl fluoride, 3 _H. $H = 5.6 - 7.8$ Hz, and β-glycosyl fluoride, ${}^{3}J_{H-H} = 5.6 - 7.8$ Hz.

For D-mannosyl fluorides and L-rhamnosyl fluorides:

1. One-bond C-F coupling of anomeric carbon in H-coupled ¹³C NMR; α-anomers, ¹J_{C-F} = 221 - 224 Hz and βanomers, ${}^{1}J_{C-F} = 213-218$ Hz

When previously reported NMR spectra were available, the stereochemistry of the product was confirmed again by comparing with them.

II. Optimization

a. Reaction Condition Optimization

Table S1. Solvent Screening

Table S2. Electrode Screening

^aYield based on the recovered starting material.

Table S3. DIPEA Loading and Reaction Time Screening

Figure S1. Amine base prevents the electrode passivation.

Table S4. Electrolyte Screening

a Voltage applied to the electrolytic cell by DC power supply.

Table S5. Amine Base Screening

Table S6. TEA vs DIPEA

III. Experimental Procedure

a. Synthesis of Substrates

 1_m

 $1n$

 1_o

.o

Ph⁻

BnO

OH

Preparation of 4-Methoxybenzyl bromide (S1)

Bı

4-Methoxybenzyl bromide (S1)

In a dry round-bottom flask, 4-methoxybenzyl alcohol (112 mmol, 2.8 equiv.) was dissolved with Et₂O (40 mL). After cooling the mixture by placing the flask in an ice bath, PBr₃ (40 mmol, 1 equiv.) was added dropwise into the reaction mixture. The reaction was

stirred for 5 hours while it was gradually warmed up to room temperature and quenched by adding aqueous NaCl solution (50 mL). The crude mixture was washed with aqueous NaHCO₃ (saturated, 50 mL) and then aqueous NaCl solution (saturated, 50 mL). The organic layer is concentrated *in vacuo* to yield 4-methoxybenzyl bromide S1 as colorless liquid. The product was confirmed by ¹H NMR and was subjected to the next synthesis without further purification. **¹ H NMR** (700 MHz, Chloroform-*d*) δ 7.33 (d, *J* = 8.6 Hz, 2H), 6.87 (d, *J* = 8.6 Hz, 2H), 4.51 (s, 2H), 3.81 (s, 3H).

Preparation of 2,3,4,6-tetrakis-*O***-(phenylmethyl)-D-galactopyranoside (1a)**

Purchased from CarboSynth. Lot#: MT040861801.

 ORn **BnC BnO** BnÒ $1a$

Preparation of 2,3,4,6-Tetrakis-*O***-(phenylmethyl)-D-glucopyranoside (1b)**

Purchased from CarboSynth. Batch#: MT066911502.

BnC BnO BnO **BnO** 1_b

Preparation of 2,3,4,6-tetrakis-*O***-(phenylmethyl)-D-Mannopyranoside (1c)**

1,2,3,4,6-Penta-*O***-acetyl-D-mannopyranoside (S2)**

In an oven dried round-bottom flask, D-mannose (27.8 mmol, 1.0 equiv.) and pyridine (40 mL) were added. The reaction mixture was stirred until the solution became homogeneous, and then the round-bottom flask was placed in an ice-bath. After the

reaction was cooled down, acetic anhydride (166.8 mmol, 6.0 equiv.) and DMAP (1.35 mmol, 0.05 equiv.) were added in that order. The reaction was stirred overnight while it was gradually warmed up to the room temperature. The reaction was quenched by diluting with DCM (200 mL) followed by washing with 1N HCl (200 mL x 2), water (200 mL x 2), and then brine (200 mL). Extracted organic layer was dried with Na₂SO₄ and was concentrated *in vacuo* to yield 1,2,3,4,6-penta-*O*-acetyl-D-mannopyranoside **S2**. The product was subjected to the next synthesis without further purification. ¹H NMR (500 MHz, CDCl₃) δ 6.09 (d, $J = 1.9$ Hz, 1H (α -C₁-H)), 5.35 (d, *J* = 6.2 Hz, 2H), 5.30 (s, 1H), 5.27 – 5.26 (m, 1H), 4.29 (dd, *J* = 12.4, 4.9 Hz, 1H), 4.11 (dd, *J* = 12.4, 2.5 Hz, 1H), 4.06 (q, *J* = 6.9, 6.2 Hz, 1H), 2.18 (s, 3H), 2.17 (s, 3H), 2.10 (s, 3H), 2.06 (s, 3H), 2.01 (s, 3H).

Ac_O OAc AcO AcO **SPh** $S₃$

Phenyl 2,3,4,6-tetra-*O***-acetyl-1-thio-α-D-mannopyranoside (S3)**

In an oven dried round-bottom flask, **S2** (27.8 mmol, 1.0 equiv.) and DCM (30 mL) were added. After the addition of thiophenol (41.7 mmol, 1.5 equiv.), the round-bottom flask was sealed with a rubber-septum cap and purged with nitrogen gas. The reaction mixture was

stirred until it became homogeneous, and then $BF_3 \cdot OEt_2$ (83.4 mmol, 3.0 equiv.) was slowly injected into the flask. The reaction was stirred overnight at room temperature and quenched with aqueous NaHCO3 (saturated, 50 mL). After extracting the mixture with DCM (100mL x 3), the combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane \rightarrow 40% EtOAc in hexane) to afford phenyl 2,3,4,6-tetra-*O*-acetyl-1-thio-D-mannopyranoside **S3**.

Phenyl 1-thio-α-D-mannopyranoside (S4)

A dry round-bottom flask was filled with **S3** (26.5 mmol, 1.0 equiv.) and anhydrous methanol (50 mL). After the reaction mixture became homogeneous, a chip of sodium metal (2.7 mmol, 0.1 equiv.) was added. The reaction was stirred overnight at room temperature,

and then quenched by adding acetic acid until the pH of the reaction became neutral. The crude mixture was concentrated *in vacuo* to yield **S4** as yellowish white solid. The product was subjected to the next synthesis without further purification.

Phenyl 2,3,4,6-tetrakis-*O***-(phenylmethyl)-1-thio-D-mannopyranoside (S5)**

An oven dried and nitrogen filled round-bottom flask was charged with **S4** (11 mmol, 1.0 equiv.) and DMF (30 mL). After cooling down the reaction in an ice bath, NaH (88 mmol, 8.0 equiv.) was added portion wise. Once H₂ gas stops generating, PMBBr (88)

mmol, 8.0 equiv.) was added followed by TBAI (0.55 mmol, 0.05 equiv.). The reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (100 mL x 3), and the organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane \rightarrow 30% EtOAc in hexane) to afford phenyl 2,3,4,6-tetrakis-*O*-(phenylmethyl)-1-thio-D-mannopyranoside **S5** as product

2,3,4,6-Tetrakis-*O***-(phenylmethyl)-D-mannopyranoside (1c)**

In a clean round-bottom flask, **S5** (10.9 mmol, 1.0 equiv.) was dissolved with acetone/water mixture (9:1, 55.5 mL). After cooling down the mixture in an ice bath,

NBS (22.9 mmol, 2.1 equiv.) was added. The reaction was stirred overnight while it was gradually warmed up to the room temperature. Water (50 mL) was added to quench the reaction followed by the extraction with EtOAc (120 mL x 3). The combined organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 40% EtOAc in hexane \rightarrow 60% EtOAc in hexane) to afford 2,3,4,6tetrakis-*O*-(phenylmethyl)-D-mannopyranoside **1c** as yellowish syrup.

Preparation of 2,3,4-tris-*O***-(phenylmethyl)-D-xylopyranoside (1d)**

AcO

1,2,3,4-Tetra-*O***-acetyl-D-xylopyranoside (S6)**

In an oven dried round bottom flask, D-xylopyranose (13 mmol, 1.0 equiv.) and anhydrous pyridine (40 mL) were added. After stirring the mixture until it became

homogeneous, the round bottom flask was placed in an ice bath and was cooled down to 0°C. Into the cooled down reaction, acetic anhydride (65 mmol, 5.0 equiv.) and DMAP (0.65 mmol, 0.05 equiv.) were added in that order. The reaction was slowly warmed up to room temperature with stirring until starting materials are gone. The mixture was then diluted with DCM (150 mL) and washed with 1N HCl solution (2 x 120 mL). The organic layer was washed with water (2 x 120 mL) and then saturated brine (2 x 120 mL). The extracted organic layer was dried with Na2SO4 and concentrated *in vacuo*. The product 1,2,3,4-Tetra-*O*-acetyl-D-xylopyranose **S6** is subjected to the next synthesis without further purification.

Phenyl 2,3,4-tri-*O***-acetyl-1-thio-D-xylopyranoside (S7)**

 ACO In an oven dried round-bottom flask, **S6** (13 mmol, 1.0 equiv.) and DCM (15 mL) were Ac SPh added. After the addition of thiophenol (19.5 mmol, 1.5 equiv.), the round-bottom flask **S7** was sealed with a rubber-septum cap and purged with nitrogen gas. The reaction mixture was stirred until it became homogeneous, and then BF₃·OEt₂ (33 mmol, 3.0 equiv.) was slowly injected into the flask. The reaction was stirred overnight at room temperature and quenched with saturated NaHCO₃ solution (50 mL). After extracting the mixture with DCM (3 x 70mL), the combined organic layer was dried with Na_2SO_4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane) to afford phenyl 2,3,4-tri-*O*-acetyl-1-thio-D-xylopyranose **S7** as yellowish oil.

Phenyl 1-thio-D-xylopyranoside (S8)

A dry round-bottom flask was filled with **S7** (13 mmol, 1.0 equiv.) and anhydrous methanol (40 mL). After the reaction mixture became homogeneous, a chip of sodium S₈ metal (1.3 mmol, 0.1 equiv.) was added. The reaction was stirred overnight at room temperature, and then quenched by adding acetic acid until the pH of the reaction became neutral. The crude mixture was concentrated *in vacuo* to yield yellowish white solid. The product phenyl 1-thio-D-xylopyranose **S8** was subjected to the next synthesis without further purification.

Phenyl 2,3,4-tris-*O***-(phenylmethyl)-1-thio-D-xylopyranoside (S9)**

 Rn An oven dried and nitrogen filled round-bottom flask was charged with **S8** (13 mmol, **BnO** 1.0 equiv.) and DMF (60 mL). After cooling down the reaction in an ice bath, NaH (78 $S₀$ mmol, 6.0 equiv.) was added portion wise. Once H2 gas stops generating, BnBr (78 mmol, 6.0 equiv.) was added followed by TBAI (0.65 mmol, 0.05 equiv.). The reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 100 mL), and the organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane) to afford phenyl 2,3,4-tris-*O*-(phenylmethyl)-1-thio-D-xylopyranoside **S9** as yellowish oil.

 BnO

2,3,4-tris-*O***-(phenylmethyl)-D-xylopyranoside (1d)**

In a clean round-bottom flask, **S9** (13 mmol, 1 equiv.) was dissolved with acetone/water mixture (9:1, 50 mL). After cooling down the mixture in an ice bath, NBS (26 mmol, 2

equiv.) was added. The reaction was stirred overnight while it was gradually warmed up to the room temperature. Water (50 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 120 mL). The combined organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, 10% EtOAc in DCM) to afford 2,3,4-tris-*O*-(phenylmethyl)-D-xylopyranose **1d** as white solid, which is confirmed by comparing with literature NMR spectra.¹

¹Zhang, J.; Fu, J.; Si, W.; Wang, X.; Wang, Z.; Tang, J. *Carbohydrate Research.* **2011**, 346, 2290-2293

Preparation of 2,3,4-tris-*O***-(phenylmethyl)-L-fucopyranoside (1e)**

1,2,3,4-Tetra-*O***-acetyl-L-fucopyranoside (S10)**

In an oven dried round bottom flask, L-fucose (3.05 mmol, 1.0 equiv.) and anhydrous pyridine (10 mL) were added. After stirring the mixture until it became homogeneous, the round bottom flask was placed in an ice bath and was cooled down to 0°C. Into the cooled

down reaction, acetic anhydride (18.3 mmol, 6.0 equiv.) and DMAP (0.15 mmol, 0.05 equiv.) were added in that order. The reaction was slowly warmed up to room temperature with stirring until starting materials are gone. The mixture was then diluted with DCM (100 mL) and washed with 1N HCl solution (2 x 80 mL). The organic layer was washed with water (2 x 80 mL) and then saturated brine (2 x 80 mL). The extracted organic layer was dried with Na2SO4 and concentrated *in vacuo*. The product 1,2,3,4-tetra-*O*-acetyl-L-fucopyranoside **S10** is subjected to the next synthesis without further purification.

Phenyl 2,3,4-tri-*O***-acetyl-1-thio-L-fucopyranoside (S11)**

In an oven dried round bottom flask, **S8** (3.0 mmol, 1.0 equiv.) and anhydrous DCM (5 mL) were added. After stirring until the mixture became homogeneous, PhSH (4.5 mmol, 1.5 equiv.) and BF_3 OEt₂ (9.0 mmol, 3.0 equiv.) were added. Then, the reaction mixture was

stirred overnight and quenched with saturated NaHCO₃ solution (20 mL) . The aqueous layer was extracted with DCM (3 x 35 mL) and dried with Na₂SO₄. The extracted organic layer was concentrated in vacuo and subjected to FCC (SiO2, 30% EtOAc in hexane → 50% EtOAc in hexane) to afford phenyl 2,3,4-tri-*O*-acetyl-1-thio-Lfucopyranoside **S11**.

Phenyl 1-thio-L-fucopyranoside (S12)

In an oven dried round bottom flask, **S9** (2.9 mmol, 1 equiv.) and MeOH (20 mL) were added. After stirring until the mixture became homogeneous, Na metal (0.29 mmol, 0.1 equiv.) was added and stirred at room temperature until all starting materials were

consumed. Once all starting materials are gone, the reaction was neutralized with acetic acid and was concentrated *in vacuo* to afford phenyl 1-thio-L-fucopyranoside **S10** as yellowish syrup. The product was subjected to next reaction without further purification.

Phenyl 2,3,4-tri-*O***-(phenylmethyl)-1-thio-L-fucopyranoside (S13)**

An oven dried and nitrogen filled round-bottom flask was charged with **S10** (3.0 mmol, 1.0 equiv.) and DMF (23 mL). After cooling down the reaction in an ice bath, NaH (18 mmol, 6.0 equiv.) was added portion wise. Once H₂ gas stops generating, BnBr (18 mmol,

6.0 equiv.) was added followed by TBAI (0.15 mmol, 0.05 equiv.). The reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 100 mL), and the organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, 20% EtOAc in hexane) to afford phenyl 2,3,4-tris-*O*-(phenylmethyl)-1-thio-D-fucopyranoside **S11** as yellowish oil.

2,3,4-Tris-*O***-(phenylmethyl)-L-fucopyranoside (1e)**

In a clean round-bottom flask, **S11** (2.6 mmol, 1 equiv.) was dissolved with acetone/water mixture (9:1, 25 mL). After cooling down the mixture in an ice bath, NBS (5.2 mmol, 2 equiv.) was added. The reaction was stirred overnight while it was gradually warmed up to

the room temperature. Water (20 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 100 mL). The combined organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, 30% EtOAc in hexane) to afford **1e** as white solid, which is confirmed by comparing with literature NMR spectra.²

²Nishi, Y.; Tanimoto, T. *Biosci. Biotechnol. Biochem.* **2009**, 73 (3), 562–569

SPh Me $M \approx$ о́Ас ÓАс ÓAc **S14 S15** NaH. BnBr SPh SPh NaOMe **TBAI** MeOH, r.t DMF, $0^\circ \rightarrow r.t$ ்⊦ ÓBn S₁₆ **S17** ΩH Acetone/H₂ ÒRn

Preparation of 2,3,4-tri-*O***-(phenylmethyl)-α-L-rhamnopyranoside (1f)**

 $1f$

An oven dried round bottom flask was charged with L-rhamnose (3.0 mmol, 1.0 equiv.) and anhydrous pyridine (10 mL). After stirring the mixture until it became homogeneous, the round bottom flask was placed in an ice bath and was cooled down to 0°C. Into the

cooled down mixture, acetic anhydride (18 mmol, 6.0 equiv.) and DMAP (0.15 mmol, 0.05 equiv.) were added in that order. The reaction was slowly warmed up to room temperature with stirring until all starting materials were gone. The mixture was then diluted with DCM (100 mL) and washed with 1N HCl solution (2 x 80 mL). The organic layer was washed with water $(2 \times 80 \text{ mL})$ and then saturated brine $(2 \times 80 \text{ mL})$. The extracted organic layer was dried with Na2SO4 and concentrated *in vacuo* to afford 1,2,3,4-tetra-*O*-acetyl-L-rhamnopyranoside **S14** as yellowish syrup. The product was then subjected to the next synthesis without further purification.

Phenyl 2,3,4-tri-*O***-acetyl-1-thio-L-rhamnopyranoside (S15)**

In an oven dried round bottom flask, **S14** (3.0 mmol, 1.0 equiv.) and anhydrous DCM (5 mL) were added. After stirring until the mixture became homogeneous, PhSH (4.5 mmol, 1.5 equiv.) and BF_3 OEt₂ (9.0 mmol, 3.0 equiv.) were added. The reaction mixture was

stirred overnight and quenched with saturated NaHCO₃ solution (20 mL). The aqueous layer was extracted with DCM (3 x 35 mL) and dried with Na2SO4. The extracted organic layer was concentrated *in vacuo* and subjected to FCC (SiO2, 30% EtOAc in hexane) to afford phenyl 2,3,4-tri-*O*-acetyl-1-thio-L-rhamnopyranoside **S15** as white solid.

Phenyl 1-thio-L-rhamnopyranoside (S16)

In an oven dried round bottom flask, **S15** (3.0 mmol, 1 equiv.) and MeOH (20 mL) were added. After stirring until the mixture became homogeneous, a chip of sodium metal (0.3 mmol, 0.1 equiv.) was added and stirred at room temperature until all starting materials

were consumed. Once all starting materials are gone, the reaction was neutralized with acetic acid and was concentrated *in vacuo* to afford phenyl 1-thio-L-rhamnopyranoside **S16** as white syrup. The product was then subjected to next reaction without further purification.

Phenyl 2,3,4-tri-*O***-(phenylmethyl)-1-thio-L-rhamnopyranoside (S17)**

An oven dried and nitrogen filled round-bottom flask was charged with **S16** (3.0 mmol, 1.0 equiv.) and DMF (20 mL). After cooling down the reaction in an ice bath, NaH (18 mmol, 6.0 equiv.) was added portion wise. Once H₂ gas stops generating, BnBr (18 mmol,

6.0 equiv.) was added followed by TBAI (0.15 mmol, 0.05 equiv.). The reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 60 mL), and the organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, 5% EtOAc in hexane) to afford phenyl 2,3,4-tri-*O*-(phenylmethyl)-1-thio-L-rhamnopyranoside **S17** as yellowish oil.

2,3,4-tri-*O***-(phenylmethyl)-α-L-rhamnopyranoside (1f)**

In a clean round-bottom flask, **S7** (2.5 mmol, 1 equiv.) was dissolved with acetone/water mixture (9:1, 22.2 mL). After cooling down the mixture in an ice bath, NBS (5.0 mmol, 2.0 equiv.) was added. The reaction was stirred overnight while it was gradually warmed up to

the room temperature. Water (20 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 70 mL). The combined organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, 30% EtOAc in hexane) to afford 2,3,4-tri-*O*-(phenylmethyl)-α-Lrhamnopyranoside **1f** as white solid, which is confirmed by comparing with literature NMR spectra.3

Preparation of 2,3,5-Tris-*O***-(phenylmethyl)-β-D-ribofuranose (1g)**

Purchased from Combi-Blocks. Batch#: A98634.

³Zhang, J.; Fu, J.; Si, W.; Wang, X.; Wang, Z.; Tang, J. *Carbohydrate Research.* **2011**, 346, 2290-2293

Preparation of 2,3,4,6-tetrakis-*O***-[(4-methoxyphenyl)methyl]-D-galactopyranose (1h)**

1,2,3,4,6-Penta-*O***-acetyl-α-D-galactopyranoside (S18)**

In an oven dried round-bottom flask, D-galactose (5.55 mmol, 1.0 equiv.) and pyridine (20 mL) were added. The reaction mixture was stirred until the solution became homogeneous, and then the round-bottom flask was placed in an ice-bath. After the reaction was cooled

down, acetic anhydride (33.3 mmol, 6.0 equiv.) and DMAP (0.278 mmol, 0.05 equiv.) were added in that order. The reaction was stirred overnight while it was gradually warmed up to the room temperature. The reaction was quenched by diluting with DCM (150 mL) followed by washing with 1N HCl (2 x 150 mL), water (2 x 150 mL), and then brine (150 mL). Extracted organic layer was dried with Na₂SO₄ and was concentrated *in vacuo* to yield 1,2,3,4,6-penta-*O*-acetyl-α-D-galactopyranoside **S18** as yellowish syrup. The product was subjected to the next synthesis without further purification. **¹ H NMR** (500 MHz, Chloroform-*d*) δ 6.38 (d, *J* = 1.8 Hz, 1H), 5.50 (d, *J* = 1.5 Hz, 1H), 5.36 – 5.32 (m, 2H), 4.34 (td, *J* = 6.6, 1.4 Hz, 1H), 4.14 – 4.05 (m, 2H), 2.16 (s, 3H), 2.16 (s, 3H), 2.04 (s, 3H), 2.02 (s, 3H), 2.01 (s, 3H).

Phenyl 2,3,4,6-tetra-*O***-acetyl-1-thio-D-galactopyranoside (S19)**

In an oven dried round-bottom flask, **S18** (11 mmol, 1.0 equiv.) and DCM (15 mL) were added. After the addition of thiophenol (16.5 mmol, 1.5 equiv.), the round-bottom flask was sealed with a rubber-septum cap and purged with nitrogen gas. The reaction mixture

was stirred until it became homogeneous, and then BF₃·OEt₂ (33 mmol, 3.0 equiv.) was slowly injected into the flask. The reaction was stirred overnight at room temperature and quenched with saturated NaHCO₃ solution (50 mL). After extracting the mixture with DCM (3 x 70mL), the combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane \rightarrow 40% EtOAc in hexane) to afford phenyl 2,3,4,6-tetra-*O*-acetyl-1-thio-D-galactopyranoside **S19** as product. **¹ H NMR** (500 MHz, Chloroform-*d*) δ 7.54 – 7.49 (m, 2H), 7.34 – 7.30 (m, 3H), 5.42 (dd, *J* = 3.4, 1.1 Hz, 1H), 5.24 (t, *J* = 10.0 Hz, 1H), 5.05 (dd, *J* = 9.9, 3.3 Hz, 1H), 4.72 (d, *J* = 10.0 Hz, 1H), 4.19 (dd, *J* = 11.3, 7.0 Hz, 1H), 4.12

(dd, *J* = 11.4, 6.3 Hz, 2H), 3.96 – 3.92 (m, 1H), 2.12 (s, 3H), 2.10 (s, 3H), 2.04 (s, 3H), 1.98 (s, 3H).

Phenyl 1-thio-D-galactopyranoside (S20)

A dry round-bottom flask was filled with **S19** (11 mmol, 1.0 equiv.) and dry methanol (45 mL). After the reaction mixture became homogeneous, a chip of sodium metal (1.1 mmol, 0.1 equiv.) was added. The reaction was stirred overnight at room temperature, and then

quenched by adding acetic acid until the pH of the reaction became neutral. The crude mixture was concentrated *in vacuo* to afford phenyl 1-thio-D-galactopyranoside **S20** as yellowish white solid. The product was subjected to the subsequent reaction without further purification.

Phenyl 2,3,4,6-tetrakis-*O***-[(4-methoxyphenyl)methyl]-1-thio-D-galactopyranoside (S21)**

An oven dried and nitrogen filled round-bottom flask was charged with **S20** (11 mmol, 1.0 equiv.) and DMF (60 mL). After cooling down the reaction in an ice bath, NaH (88 mmol, 8.0 equiv.) was added portion wise. Once H₂ gas stops generating, PMBBr (88 mmol, 8.0 equiv.) was added followed by TBAI (1.1 mmol, 0.1 equiv.). The reaction

mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 100 mL), and the organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane \rightarrow 30% EtOAc in hexane) to afford phenyl 2,3,4,6-tetrakis-*O*-[(4-methoxyphenyl)methyl]-1-thio-D-galactopyranoside **S21** as yellowish syrup.

2,3,4,6-tetrakis-*O***-[(4-methoxyphenyl)methyl]-D-galactopyranoside (1h)**

In a clean round-bottom flask, **S21** (10 mmol, 1.0 equiv.) was dissolved with acetone/water mixture (9:1, 55.5 mL). After cooling down the mixture in an ice bath, NBS (21 mmol, 2.1 equiv.) was added. The reaction was stirred overnight while it was

gradually warmed up to the room temperature. Water (50 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 120 mL). The combined organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, 30% EtOAc in hexane → 40% EtOAc in hexane) to afford 2,3,4,6-tetrakis-*O*-[(4-methoxyphenyl)methyl]-D-galactopyranoside **1h** as white solid product, which is confirmed by comparing with literature NMR spectra.⁴

⁴Whalen, L. J.; Halcomb, R. L. *Org. Lett.* **2004**, 6, 19, 3221-3224

Preparation of 2,3,4,6-tetrakis-*O***-[(4-methoxyphenyl)methyl]-α-D-glucopyranoside (1i)**

Phenyl 2,3,4,6-tetra-*O***-acetyl-1-thio-D-glucopyranoside (S22)**

In an oven dried round-bottom flask, 1,2,3,4,6-penta-*O*-acetyl-D-glucopyranoside (5.0 mmol, 1.0 equiv.) and DCM (5 mL) were added. After the addition of thiophenol (7.5 mmol, 1.5 equiv.), the round-bottom flask was sealed with a rubber-septum cap and

purged with nitrogen gas. The reaction mixture was stirred until it became homogeneous, and then BF₃·OEt₂ (15 mmol, 3.0 equiv.) was slowly injected into the flask. The reaction was stirred overnight at room temperature and quenched with saturated NaHCO₃ solution (10 mL). After extracting the mixture with DCM (3 x 50mL), the combined organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, pure hexane → 30% EtOAc in hexane) to afford phenyl 2,3,4,6-tetra-*O*-acetyl-1-thio-Dglucopyranoside **S22** as white solid.

Phenyl 1-thio-D-glucopyranoside (S23)

A dry round-bottom flask was filled with **S22** (5.0 mmol, 1.0 equiv.) and dry methanol (30 mL). After the reaction mixture became homogeneous, a chip of sodium metal (0.5 mmol, 0.1 equiv.) was added. The reaction was stirred overnight at room temperature,

and then quenched by adding acetic acid until the pH of the reaction became neutral. The crude mixture was concentrated *in vacuo* to yield **S23** as yellowish white solid. The product was subjected to the subsequent reaction without further purification.

Phenyl 2,3,4,6-tetrakis-*O***-[(4-methoxyphenyl)methyl]-1-thio-D-glucopyranoside (S24)**

An oven dried and nitrogen filled round-bottom flask was charged with **S23** (5.0 mmol, 1.0 equiv.) and DMF (30 mL). After cooling down the reaction in an ice bath, NaH $(25 \text{ mmol}, 5.0 \text{ equiv.})$ was added portion wise. Once H₂ gas stops generating, PMBBr (25 mmol, 5.0 equiv.) was added followed by TBAI (0.5 mmol, 0.1 equiv.). The

reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 100 mL), and the organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 40% EtOAc in hexane) to afford phenyl 2,3,4,6-tetrakis-*O*-[(4-methoxyphenyl)methyl]-1-thio-D-glucopyranoside **S24** as yellowish white syrup.

2,3,4,6-Tetrakis-*O***-[(4-methoxyphenyl)methyl]-α-D-glucopyranoside (1i)**

In a clean round-bottom flask, **S24** (5.0 mmol, 1.0 equiv.) was dissolved with acetone/water mixture (9:1, 33.3 mL). After cooling down the mixture in an ice bath, NBS (10 mmol, 2.0 equiv.) was added. The reaction was stirred overnight while it was

gradually warmed up to the room temperature. Water (35 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 100 mL). The combined organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, 40% EtOAc in hexane) to afford 2,3,4,6tetrakis-*O*-[(4-methoxyphenyl)methyl]-α-D-glucopyranoside **1i** as white solid, which is confirmed by comparing with literature NMR spectra.⁵

⁵Lucchetti, N.; Gilmour, R. *Chem. Eur. J.* **2018**, 24,16266 –16270

Preparation of 2,3,4,6-tetrakis-*O***-[(4-methoxyphenyl)methyl]-α-D-mannopyranoside (1j)**

Phenyl 2,3,4,6-tetrakis-*O***-[(4-methoxyphenyl)methyl]-1-thio-α-D-mannopyranoside (S25)**

An oven dried and nitrogen filled round-bottom flask was charged with **S4** (9.5 mmol, 1.0 equiv.) and DMF (50 mL). After cooling down the reaction in an ice bath, NaH (52 mmol, 5.5 equiv.) was added portion wise. Once H₂ gas stops generating, PMBBr (47.5) mmol, 5.0 equiv.) was added followed by TBAI (0.95 mmol, 0.1 equiv.). The reaction

mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 100 mL), and the organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 40% EtOAc in hexane) to afford phenyl 2,3,4,6tetrakis-*O*-[(4-methoxyphenyl)methyl]-1-thio-α-D-mannopyranoside **S25**.

2,3,4,6-Tetrakis-*O***-[(4-methoxyphenyl)methyl]-α-D-mannopyranoside (1j)**

In a clean round-bottom flask, **S25** (9.5 mmol, 1.0 equiv.) was dissolved with acetone/water mixture (9:1, 55.5 mL). After cooling down the mixture in an ice bath, NBS (20 mmol, 2.1 equiv.) was added. The reaction was stirred overnight while it was gradually

warmed up to the room temperature. Water (50 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 120 mL). The combined organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 50% EtOAc in hexane \rightarrow 70% EtOAc in hexane) to afford 2,3,4,6-Tetrakis-*O*-[(4-methoxyphenyl)methyl]-α-D-mannopyranoside **1m** as white solid. **¹ H NMR** (700 MHz, CDCl3) δ 7.27 (d, *J* = 10.2 Hz, 6H), 7.06 (dd, *J* = 8.7, 2.6 Hz, 2H), 6.90 – 6.78 (m, 8H), 5.21 (d, *J* = 1.9 Hz, 1H), 4.77 (d, *J* = 10.4 Hz, 1H), 4.66 (d, *J* = 20.2 Hz, 2H), 4.55 (d, *J* = 3.8 Hz, 2H), 4.52 (d, *J* = 11.9 Hz, 1H), 4.48 (d, *J* = 11.9 Hz, 1H), 4.39 (d, *J* = 10.5 Hz, 1H), 3.96 (ddd, *J* = 8.9, 6.3, 2.0 Hz, 1H), 3.90 (dd, *J* = 9.5, 3.0 Hz, 1H), 3.81 (s, 3H), 3.80 (s, 3H), 3.79 (s, 3H), 3.79 (s, 3H), 3.76 (t, *J* = 2.5 Hz, 1H), 3.68 – 3.66 (m, 1H), 3.65 (d, *J* = 1.9 Hz, 1H), 3.62 (dd, *J* = 10.5, 6.4 Hz, 1H); **13C NMR** (176 MHz, CDCl3) δ 159.1, 159.1, 159.1, 159.1, 130.7, 130.6, 130.4, 130.1, 129.8, 129.7, 129.6, 129.6, 129.6, 129.5, 129.2, 129.2, 113.7, 113.7, 113.7, 113.6, 92.7, 79.4, 74.9, 74.6, 74.3, 72.8, 72.2, 71.8, 71.4, 69.1, 55.2, 55.2.

Preparation of 2,3,4,6-tetra-*O***-acetyl-α-D-galactopyranose (1k)**

In an oven dried round-bottom flask, **S18** (5.55 mmol, 1.0 equiv.) and THF (20 mL) were added. The mixture was stirred until the solution became homogeneous, and then benzylamine (8.33 mmol, 1.5 equiv.) was added. The reaction was stirred at room

temperate for 20 hours and quenched by evaporating the solvent under reduced pressure. The crude mixture was diluted with DCM (150 mL) and washed with 1N HCl (100 mL x 2), saturated NaHCO₃ solution (100 mL), and then water (100 mL). The organic layer was dried with Na₂SO₄ and was concentrated *in vacuo* to give ambercolored oil. This oil was subjected to FCC (SiO₂, 20% EtOAc in hexane \rightarrow 60% EtOAc in hexane) to afford 2,3,4,6-tetra-*O*-acetyl-D-galactopyranoside **1k** as yellowish solid, which was confirmed by comparing with reported NMR spectra.6

⁶Cai, T. B.; Lu, D.; Tang, X.; Zhang, Y.; Landerholm, M.; Wang, P. G. *J. Org. Chem.* **2005**, 70, 9, 3518-3524

Preparation of 2,3,4,6-tetra-*O***-acetyl-D-glucopyranoside (1l)**

2,3,4,6-Tetra-*O***-acetyl-D-glucopyranoside (1l)**

In an oven dried round-bottom flask, 1,2,3,4,6-penta-*O*-acetyl-α-D-glucopyranoside (5.12 mmol, 1.0 equiv.) and THF (20 mL) were added. The mixture was stirred until the solution became homogeneous, and then benzylamine (7.69 mmol, 1.5 equiv.) was added.

The reaction was stirred at room temperate for 24 hours and quenched by evaporating the solvent under reduced pressure. The crude mixture was diluted with DCM (150 mL) and washed with 1N HCl (2 x 100 mL), saturated NaHCO₃ solution (100 mL), and then water (100 mL). The organic layer was dried with Na₂SO₄ and was concentrated *in vacuo* to give amber-colored oil. This oil was subjected to FCC (SiO2, 60% EtOAc in hexane) to afford 2,3,4,6-tetra-*O*-acetyl-D-glucopyranoside **1l** as yellowish solid, which is confirmed by comparing with literature NMR spectra.⁷

⁷Ikeda, K.; Morimoto, T.; Kakiuchi, K. *J. Org. Chem.* **2010**, 75, 18, 6279–6282

Preparation of 2,3,4,6-tetra-*O***-acetyl-D-mannopyranose (1m)**

2,3,4,6-Tetra-*O***-acetyl-α-D-mannopyranoside (1m)**

In an oven dried round-bottom flask, **S2** (2.17 g, 5.55 mmol) and THF (20 mL) were added. The mixture was stirred until the solution became homogeneous, and then benzylamine (0.91 mL, 8.33 mmol) was added. The reaction was stirred at room temperate for 20 hours

and quenched by evaporating the solvent under reduced pressure. The crude mixture was diluted with DCM (150 mL) and washed with HCl $(1.0 \text{ M}, 100 \text{ mL} \times 2)$, NaHCO₃ (saturated, 100 mL), and then water (100 mL) . The organic layer was dried with Na2SO4 and was concentrated *in vacuo* to give amber-colored oil. This oil was subjected to FCC (SiO₂, 20% EtOAc in hexane \rightarrow 60% EtOAc in hexane) to afford 2,3,4,6-tetra-O-acetyl-Dmannopyranoside **1m** as yellowish solid, which was confirmed by comparing with reported NMR spectra. 8 **1 H NMR** (700 MHz, Chloroform-*d*) δ 5.42 (dd, *J* = 10.1, 3.4 Hz, 1H), 5.30 (t, *J* = 9.8 Hz, 1H), 5.28 – 5.24 (m, 2H), 4.27 – 4.22 (m, 2H), 4.12 (q, *J* = 7.4 Hz, 2H), 3.19 (br, 1H), 2.16 (s, 3H), 2.11 (s, 3H), 2.04 (s, 3H), 2.00 (s, 3H).

⁸Carcabal, P.; Hunig, I.; Gamblin, D. P.; Liu, B.; Jockusch, R. A.; Kroemer, R. T.; Snoek, L. C.; Fairbanks, A. J.; Davis, B. G.; Simons, J. P. *J. Am. Chem. Soc.* **2006**, 128, 6, 1976-1981

Preparation of 2,3,4-tri-*O***-acetyl-α-L-rhamnopyranoside (1n)**

2,3,4-Tri-*O***-acetyl-α-L-rhamnopyranoside (1n)**

In an oven dried round-bottom flask, **S14** (3.0 mmol, 1.0 equiv.) and THF (15 mL) were added. The mixture was stirred until the solution became homogeneous, and then benzylamine (4.5 mmol, 1.5 equiv.) was added. The reaction was stirred at room temperate

for 20 hours and quenched by evaporating the solvent under reduced pressure. The crude mixture was diluted with DCM (70 mL) and washed with 1N HCl (2 x 50 mL), saturated NaHCO₃ solution (50 mL), and then water (50 mL). The organic layer was dried with Na2SO4 and was concentrated *in vacuo* to give amber-colored oil. This oil was subjected to FCC (SiO2, 20% EtOAc in hexane → 50% EtOAc in hexane) to afford 2,3,4-tri-*O*-acetyl-α-Lrhamnopyranoside **1n** as yellowish solid, which is confirmed by comparing with literature NMR spectra.9

⁹Donahue, M. G.; Johnston, J. N. *Bioorg. Med. Chem. Lett.* **2006**, 16, 5602–5604

Preparation of 2,3,4-tri-*O***-benzoyl-L-rhamnopyranoside (1o)**

1,2,3,4-Tetra-*O***-benzoyl-L-rhamnopyranoside (S26)**

BzO S26

An oven dried round bottom flask was charged with L-rhamnose (5.5 mmol, 1.0 equiv.) and anhydrous pyridine (12 mL). After stirring the mixture until it became homogeneous, the round bottom flask was placed in an ice bath and was cooled down

to 0°C. Into the cooled down mixture, benzoyl chloride (33 mmol, 6.0 equiv.) was added, and then the reaction was slowly warmed up to room temperature with stirring until all starting materials were gone. The mixture was then diluted with DCM (100 mL) and washed with 1N HCl solution (2 x 80 mL). The organic layer was washed with water (2 x 80 mL) and then saturated brine (2 x 80 mL). The extracted organic layer was dried with Na2SO₄ and concentrated *in vacuo*. The worked-up mixture was then subjected to FCC (SiO2, 20% EtOAc in hexane) to afford 1,2,3,4-tetra-*O*-benzoyl-L-rhamnopyranoside **S26** as white solid.

2,3,4-tri-*O***-benzoyl-L-rhamnopyranoside (1o)**

An oven dried round bottom flask was charged with 1,2,3,4-tetra-*O*-benzoyl-Lrhamnopyranoside **S26** (3.7 mmol, 1.0 equiv.) and THF:MeOH solution (7:3). After stirring the mixture until it became homogeneous, the round bottom flask was placed in

an ice bath and was cooled down to 0° C. Into the cooled down mixture, ammonia gas was bubbled for 1 minute, and then the reaction was stirred for 10 hours while slowly warmed up to room temperature. The mixture was then concentrated *in vacuo,* and then was subjected to FCC (SiO2, 30% EtOAc in hexane) to afford 2,3,4-tri-*O*-benzoyl-L-rhamnopyranoside **1o** as white solid.

Preparation of 2,3-bis-*O***-(phenylmethyl)-4,6-***O***-[(R)-phenylmethylene]-α-D-mannopyranoside (1p)**

Phenyl 4,6-*O***-[(R)-phenylmethylene]-1-thio-α-D-manopyranoside (S27)**

An oven dried and nitrogen filled round-bottom flask was charged with **S4** (4.5 mmol, 1.0 equiv.) and anhydrous acetonitrile (26 mL). PhCH(OMe)₂ (6.75 mmol, 1.5 equiv.) and p-TsOH (0.27 mmol, 0.06 equiv.) were added and the mixture was refluxed for 3

hours. Once all starting materials are consumed, the reaction was quenched with Et₃N (3 mL), diluted with EtOAc (50 mL), and then washed with brine. Collected organic layer was dried with NaSO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, 5% EtOAc in hexane) to afford phenyl 4,6-*O*-[(R) phenylmethylene]-1-thio-α-D-manopyranoside **S18** as white solid.

Phenyl 2,3-bis-*O***-(phenylmethyl)-4,6-***O***-[(R)-phenylmethylene]-1-thio-α-Dmannopyranoside (S28)**

An oven dried and nitrogen filled round-bottom flask was charged with **S27** (3.0 mmol, 1.0 equiv.) and DMF (20 mL). After cooling down the reaction in an ice bath, NaH (18

mmol, 6.0 equiv.) was added portion wise. Once H₂ gas stops generating, BnBr (12 mmol, 4 equiv.) was added followed by TBAI (0.15 mmol, 0.05 equiv.). The reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 60 mL), and the organic layer was dried with Na2SO4 and concentrated *in vacuo*.

2,3-Bis-*O***-(phenylmethyl)-4,6-***O***-[(R)-phenylmethylene]-α-D-mannopyranoside (1p)** In a clean round-bottom flask, **S19** (3.0 mmol, 1.0 equiv.) was dissolved with

acetone/water mixture (9:1, 22.2 mL). After cooling down the mixture in an ice bath, NBS (6.3 mmol, 2.1 equiv.) was added. The reaction was stirred overnight while it was

gradually warmed up to the room temperature. Water (20 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 60 mL). The combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 40% EtOAc in hexane) to afford 2,3-bis-*O*-(phenylmethyl)-4,6-*O*-[(R)-phenylmethylene]-α-D-mannopyranoside **1p** as white solid, which is confirmed by comparing with literature NMR spectra.¹⁰

¹⁰Codée, J. D. C.; Hossain, L. H.; Seeberger, P. H. *Org. Lett.* **2005**, 7, 15, 3251–3254

Preparation of 2,3,6-tris-*O***-(phenylmethyl)-4-***O***-[2,3,4,6-tetrakis-***O***-(phenylmethyl)-β-D-glucopyranosyl]- D-glucopyranoside (1q)**

Phenyl 4-*O***-(2,3,4,6-tetra-***O***-acetyl-β-D-glucopyranosyl)-1-thio-D-glucopyranoside 2,3,6-triacetate (S29)**

In an oven dried round-bottom flask, α-D-cellobiose octaacetate (3.0 mmol, 1.0 equiv.) and DCM (10 mL) were added. After the addition of thiophenol (4.5 mmol, 1.5 equiv.), the round-bottom flask was sealed with a rubber-septum cap and purged with nitrogen gas. The reaction mixture was stirred until it became homogeneous, and then BF_3 ·OEt₂ (9.0 mmol, 3.0 equiv.) was slowly injected into the flask. The reaction was stirred overnight at room temperature and quenched

with saturated NaHCO₃ solution (10 mL). After extracting the mixture with DCM (3 x 70mL), the combined organic layer was dried with Na2SO4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, 50% EtOAc in hexane) to afford phenyl 4-*O*-(2,3,4,6-tetra-*O*-acetyl-β-D-glucopyranosyl)-1-thio-D-glucopyranoside 2,3,6-triacetate **S29** as fluffy white solid.

Phenyl 4-*O***-β-D-glucopyranosyl-1-thio-D-glucopyranoside (S30)**

A dry round-bottom flask was filled with **S29** (3.0 mmol, 1.0 equiv.) and dry methanol (20 mL). After the reaction mixture became homogeneous, a chip of sodium metal (0.3 mmol, 0.1 equiv.) was added. The reaction was stirred overnight at room temperature, and then quenched by adding acetic acid until the pH of the reaction became neutral. The crude mixture was concentrated *in vacuo* to yield **S30**

as yellowish white solid. The product was subjected to the subsequent reaction without further purification.

Phenyl 2,3,6-tris-*O***-(phenylmethyl)-4-***O***-[2,3,4,6-tetrakis-***O***-(phenylmethyl)-β-D-glucopyranosyl]-1-thio-Dglucopyranoside (S31)**

An oven dried and nitrogen filled round-bottom flask was charged with **S30** (3.0 mmol, 1.0 equiv.) and DMF (30 mL). After cooling down the reaction in an ice bath, NaH (48 mmol, 16 equiv.) was added portion wise. Once H2 gas stops generating, BnBr (42 mmol, 14 equiv.) was added followed by TBAI (0.15 mmol, 0.05 equiv.). The reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 80 mL), and the organic layer was dried with

Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, hexane → 20% EtOAc in hexane) to afford phenyl 2,3,6-tris-*O*-(phenylmethyl)-4-*O*-[2,3,4,6-tetrakis-*O*-(phenylmethyl)-β-Dglucopyranosyl]-1-thio-D-glucopyranoside **S31** as yellowish white solid.

2,3,6-Tris-*O***-(phenylmethyl)-4-***O***-[2,3,4,6-tetrakis-***O***-(phenylmethyl)-β-D-glucopyranosyl]-D-**

glucopyranoside (1q)

In a clean round-bottom flask, **S31** (2.0 mmol, 1.0 equiv.) was dissolved with acetone/water mixture (9:1, 22 mL). After cooling down the mixture in an ice bath, NBS (4.2 mmol, 2.1 equiv.) was added. The reaction was stirred overnight while it was gradually warmed up to the room temperature. Water (20 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 70 mL). The combined organic layer was dried with Na2SO4 and concentrated *in vacuo*. The

concentrated crude mixture was subjected to FCC (SiO2, 30% EtOAc in hexane) to afford 2,3,6-tris-*O*- (phenylmethyl)-4-*O*-[2,3,4,6-tetrakis-*O*-(phenylmethyl)-β-D-glucopyranosyl]-D-glucopyranoside **1q** as white solid, which is confirmed by comparing with literature NMR spectra.¹¹

¹¹Liu, B.; Mechelen, J.; Berg, R.; Nieuwendijk, A.; Aerts, J.; Marel, G.; Codée, J. Overkleeft, H. S. *Eur. J. Org. Chem.* **2019**, 118–129

b. List of components for the experimental setup

Figure S2. Components of experimental setup.

- A) Mudder soldering tin wire (1.5 mm diameter) Amazon
	- Content parameter: Lead-free, Sn 99% / Ag 0.3% / Cu 0.7%

Flux 2% (washed away by sonicating in acetone)

B) Zinc wire (2.0 mm diameter) – Beantown Chemical

C) 5 cm zinc and tin wire

- D) 1 dram vial Sigma and PTFE/Silicon seal with a cap Fisher Scientific
- E) Hamilton MTB valve Thomas Scientific

F) Balloon for SF₆

c. Synthesis of glycosyl fluorides

General Procedure

An oven-dried 1 dram vial was charged with a glycosyl substrate (0.1 mmol, 1 equiv.) and tetrabutylammonium perchlorate (0.45 mmol, 0.15 M). The vial containing the mixture was sealed with PTFE/Silicone septum cap and was further dried under high vacuum for 5 minutes. Into the sealed vial, THF (3 mL) and triethylamine (0.3 mmol, 3 equiv.) were added, and the vial was stirred until the mixture became homogeneous. After plugging in the electrodes (2.3 cm submerged in the solution), the mixture was sparged with a balloon containing SF_6 gas for 30 seconds. Once the sparging was done, the valve of the balloon was adjusted to sparge small bubbles and left open until the reaction was stopped.

The reaction was stirred at high rpm while the electric current was applied to cathode and anode for 1 hour. The reaction mixture was then diluted with 30 mL of ethyl acetate, and then washed with 1N HCl (25 mL x 3), saturated NaHCO3 (25 mL), water (25 mL), and brine (25 mL). The extracted organic layer was dried with Na2SO4 and concentrated *in vacuo*. The worked-up reaction mixture was then subjected to FCC (SiO2) for purification.

Figure S3. Experimental setup for substrate screening

Cleaning Procedure for the Electrodes

Before each use, the surface of electrodes is polished with 1000 grit sandpaper and sonicated in acetone. After wiping the surface with a paper towel, the electrodes are dried and kept under high vacuum.

 2_c

 $2d$

 $2a$

 BnO $2e$

 $2j$

 $2m$

 $2n$

 2_o

 $2p$

Synthesis of 2,3,4,6-tetrakis-*O***-(phenylmethyl)-D-galactopyranosyl fluoride (2a)**

The product was synthesized employing the general procedure. After a work-up, the reaction mixture was subjected to FCC (SiO2, pure hexane → 20% EtOAc in hexane) to afford **2a** in average of 94% yield as yellowish cloudy syrup ($R_{f-a} = 0.36$ (20% EtOAc/Hex), $R_{f-a} = 0.30$ (20% EtOAc/Hex)).

2,3,4,6-Tetrakis-*O***-(phenylmethyl)-α-D-galactopyranosyl fluoride** (minor product).

 ORr **BnO BnC**

 RnC

 BnO

1 H NMR (700 MHz, CDCl3) δ 7.39 – 7.26 (m, 20H), 5.59 (dd, *J* = 53.7, 2.7 Hz, 1H), 4.94 (d, *J* = 11.3 Hz, 1H), 4.83 (dd, *J* = 23.1, 11.7 Hz, 2H), 4.74 (dd, *J* = 20.8, 11.8 Hz, 2H), 4.57 (d, *J* $= 11.3$ Hz, 1H), 4.48 (d, $J = 11.8$ Hz, 1H), 4.42 (d, $J = 11.8$ Hz, 1H), 4.13 – 4.09 (m, 1H), 4.08

– 4.01 (m, 2H), 3.95 (dd, *J* = 10.1, 2.7 Hz, 1H), 3.55 (d, *J* = 5.9 Hz, 2H); **13C NMR** (176 MHz, CDCl3) δ 138.4, 138.3, 138.0, 137.7, 128.4, 128.4, 128.3, 128.1, 128.0, 127.9, 127.8, 127.8, 127.7, 127.6, 127.5, 106.2 (d, *J* = 226.1 Hz, α-C1), 78.4, 75.7 (d, *J* = 23.8 Hz), 74.8, 74.3, 73.7, 73.5, 73.1, 71.7 (d, *J* = 2.8 Hz), 68.2; **19F NMR** (564 MHz, CDCl3) δ -150.13 (dd, *J* = 53.7, 25.3 Hz); **IR** (thin film, cm-1): 3030, 2870, 1496, 1453, 1363, 1302, 1208, 1099, 1053, 1027, 910, 820, 732, 695; **HRMS (ESI+)** (m/z): [M+Na] ⁺ calcd for C34H35FNaO5 565.2392, found 565.2353; $\left[\alpha\right]_0^{23} = +26.2$ (c = 0.4, CH₂Cl₂)

2,3,4,6-Tetrakis-*O***-(phenylmethyl)-β-D-galactopyranosyl fluoride** (major product).

1 H NMR (700 MHz, CDCl3) δ 7.39 – 7.27 (m, 20H), 5.19 (dd, *J* = 53.1, 7.0 Hz, 1H), 4.95 (d, *J* = 11.5 Hz, 1H), 4.86 (d, *J* = 11.0 Hz, 1H), 4.80 – 4.71 (m, 3H), 4.61 (d, *J* = 11.5 Hz,

1H), 4.50 (d, *J* = 11.8 Hz, 1H), 4.43 (d, *J* = 11.8 Hz, 1H), 3.99 – 3.91 (m, 2H), 3.69 – 3.61 $2a$ (m, 3H), 3.56 (ddd, *J* = 9.6, 2.9, 0.9 Hz, 1H); **13C NMR** (176 MHz, CDCl3) δ 138.3, 138.1, 138.0, 137.6, 128.4, 128.4, 128.3, 128.2, 128.1, 128.1, 127.9, 127.9, 127.7, 127.7, 127.6, 127.5, 110.2 (d, *J* = 215.4 Hz, β-C1), 81.0 (d, *J* = 11.3 Hz), 79.1 (d, *J* = 20.8 Hz), 75.0, 75.0, 74.6, 73.6, 73.6, 73.1, 73.0, 68.3; **19F NMR** (564 MHz, CDCl3) δ -138.91 (dd, *J* = 54.0, 12.8 Hz); **IR** (thin film, cm⁻¹): 3030, 2870, 1496, 1453, 1363, 1302, 1208, 1099, 1053, 1027, 910, 820, 732, 695; **HRMS (ESI**+) (m/z): [M+Na]⁺ calcd for C₃₄H₃₅FNaO₅ 565.2392, found 565.2353; [α]_D²³ = $+29.8$ (c = 1.1, CH₂Cl₂)

Synthesis of 2,3,4,6-tetrakis-*O***-(phenylmethyl)-D-glucopyranosyl fluoride (2b)**

The product was synthesized employing the general procedure. After a work-up, the reaction mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 20% EtOAc in hexane) to afford 2b as yellowish cloudy syrup in average of 98% yield ($R_f = 0.38$ (20% EtOAc/Hex)). The NMR spectrum of the product is in a good agreement with our previously published NMR data.12

2,3,4,6-Tetrakis-*O***-(phenylmethyl)-D-glucopyranosyl fluoride**

(Isolated as α : β = 1 : 1.4) ¹H NMR¹H NMR (600 MHz, Chloroform-*d*) δ 7.42 – 7.28 (m, 41.4H), 7.21 – 7.14 (m, 4.8H), 5.59 (dd, *J* = 53.2, 2.6 Hz, 1H, C1-Hα), 5.29 (dd, *J* = 52.8, 6.8 Hz, 1.4H, C₁-H_β), 4.99 (d, $J = 10.9$ Hz, 1H), 4.93 (d, $J = 11.0$ Hz, 1.4H), 4.91 – 4.80

(m, 7.3H), 4.74 (d, *J* = 11.3 Hz, 2.5H), 4.64 (dd, *J* = 16.1, 12.1 Hz, 2.5H), 4.60 – 4.54 (m, 3.4H), 4.54 – 4.48 (m, 1.6H), 4.02 (t, *J* = 9.4 Hz, 1H), 3.98 (dt, *J* = 10.2, 2.6 Hz, 1H), 3.81 – 3.73 (m, 6.3H), 3.73 – 3.67 (m, 2.5H), 3.66 – 3.57 (m, 3.9H); **19F NMR** (564 MHz, CDCl3) δ -138.06 (dd, *J* = 52.9, 11.9 Hz, C1-Fβ), -149.49 (dd, *J* = 53.1, 25.6 Hz, C₁-F_a); **IR** (thin film, cm⁻¹): 3030, 2867, 1496, 1453, 1360, 1308, 1208, 1155, 1088, 1062, 909, 821, 733, 695; **HRMS (ESI+)** (m/z): [M+Na] ⁺ calcd for C34H35FNaO5 565.2392, found 565.2353

¹²Kim, S.; Khomutnyk, Y.; Bannykh, A.; Nagorny, P. *Org. Lett*. **2021**, 23, 1, 190-194

Synthesis of 2,3,4,6-tetraki-O-(phenylmethyl)-D-mannopyranosyl fluoride (2c)

The product was synthesized employing the general procedure. After a work-up, the reaction mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 30% EtOAc in hexane) to afford 2c as cloudy syrup in average of 95% yield ($R_{f\alpha}$ = 0.57 (30% EtOAc/Hex), $R_{f\beta}$ = 0.5 (30% EtOAc/Hex)). Stereochemistry of the product was confirmed by comparing with previously reported NMR spectra.¹³

2,3,4,6-Tetrakis-*O***-(phenylmethyl)-α-D-Mannopyranosyl fluoride** (major product)

1 H NMR (700 MHz, CDCl3) δ 7.38 – 7.27 (m, 18H), 7.20 – 7.17 (m, 2H), 5.61 (d, *J* = 50.8 Hz, 1H), 4.89 (d, *J* = 10.8 Hz, 1H), 4.81 (d, *J* = 12.3 Hz, 1H), 4.72 – 4.64 (m, 4H), 4.55 (d, *J*

= 12.4 Hz, 2H), 4.09 (t, *J* = 9.5 Hz, 1H), 3.93 (ddd, *J* = 10.1, 4.7, 1.8 Hz, 1H), 3.89 (d, *J* = 8.9 Hz, 2H), 3.79 (dd, *J* = 11.0, 4.7 Hz, 1H), 3.73 (dd, *J* = 11.1, 1.9 Hz, 1H); **13C NMR** (176 MHz, CDCl3) δ 138.2, 138.1, 138.1, 137.8, 128.4, 128.3, 128.3, 127.9, 127.9, 127.8, 127.7, 127.7, 127.6, 106.4 (d, *J* = 222.0 Hz), 79.2, 75.1, 74.2, 74.2, 74.0, 73.6, 73.4, 73.4, 73.2, 72.6, 68.6; **19F NMR** (564 MHz, CDCl3) δ -138.01 (d, *J* = 50.5 Hz); **IR** (thin film, cm-1): 3030, 2863, 1496, 1453, 1363, 1311, 1183, 1094, 1026, 952, 801, 733, 695; **HRMS (ESI+)** (m/z): $[M+Na]^+$ calcd for $C_{34}H_{35}FNaO_5$ 565.2366, found 565.2358; $\left[\alpha\right]n^{24} = +12.0$ (c = 1.1, CH₂Cl₂)

¹³Zeng, J.; Vedachalam, S.; Xiang, S.; Liu, X-W. *Org. Lett.* **2011**, 13, 1, 42-45

Synthesis of 2,3,4-tris-*O***-(phenylmethyl)-D-xylopyranosyl fluoride (2d)**

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane \rightarrow 20% EtOAc in hexane) to afford 2d as yellowish white solid in average of 98% yield (R_f = 0.45(20%) EtOAc/Hex)). The NMR spectrum of the product is in a good agreement with our previously published NMR data.14

2,3,4-Tris-*O***-(phenylmethyl)-D-xylopyranosyl fluoride**

(Isolated as α : β = 1 : 2.4 mixture) ¹H NMR¹H NMR (600 MHz, CDCl₃) δ 7.43 – 7.28 (m, 46.8H), 5.48 (dd, *J* = 53.1, 2.7 Hz, 1H, C1-Hα), 5.35 (dd, *J* = 53.9, 5.6 Hz, 2.4H, C1- $2d$ Hβ), 4.94 (s, 2H), 4.86 – 4.76 (m, 8.9H), 4.75 – 4.60 (m, 9.5H), 4.10 – 4.03 (m, 2.4H), 3.94 (t, *J* = 9.2 Hz, 1H), 3.80 (dd, *J* = 11.1, 5.6 Hz, 1H), 3.75 – 3.61 (m, 6.7H), 3.61 – 3.44 (m, 5.6H); **19F NMR** (564 MHz, CDCl3) δ - 134.41 (dd, *J* = 54.0, 12.6 Hz, 2.4F, C1-Fβ), -151.04 (dd, *J* = 53.1, 25.7 Hz, 1F, C1-Fα); **IR** (thin film, cm-1): 3030, 2869, 1496, 1453, 1361, 1257, 1208, 1162, 1088. 1027, 942, 733, 695; **HRMS (ESI+)** (m/z): [M+Na] ⁺ calcd for C26H27FNaO4 445.1792, found 445.1780.

¹⁴Kim, S.; Khomutnyk, Y.; Bannykh, A.; Nagorny, P. *Org. Lett*. **2021**, 23, 1, 190-194

Synthesis of 2,3,4-tris-*O***-(phenylmethyl)-L-fucopyranosyl fluoride (2e)**

BnĊ

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane → 30% EtOAc in hexane) to afford **2e** as cloudy syrup in average of 91% yield (Rf-α = 0.6 (30% EtOAc/Hex), $R_{f-\beta} = 0.53$ (30% EtOAc/Hex)).

2,3,4-Tris-*O***-(phenylmethyl)-α-L-fucopyranosyl fluoride** (major product).

1 H NMR (700 MHz, Chloroform-*d*) δ 7.41 – 7.27 (m, 15H), 5.57 (dd, *J* = 54.0, 2.8 Hz, 1H), 4.99 (d, *J* = 11.5 Hz, 1H), 4.85 (dd, *J* = 11.8, 3.9 Hz, 2H), 4.75 (dd, *J* = 19.1, 11.8 Hz, 2H), 4.66 (d, *J* = 11.5 Hz, 1H), 4.08 – 4.00 (m, 2H), 3.94 (dd, *J* = 10.1, 2.7 Hz, 1H), 3.70 (dd, *J*

= 2.8, 1.2 Hz, 1H), 1.17 (d, *J* = 6.5 Hz, 3H); **¹³ C NMR** (176 MHz, CDCl3) δ 138.5, 138.3, 138.1, 128.5, 128.4, 128.4, 128.3, 128.0, 127.8, 127.7, 127.7, 127.5, 106.4 (d, *J* = 224.5 Hz, α-C1), 78.8, 75.6 (d, *J* = 23.9 Hz), 74.9, 73.7, 73.2, 69.1 (d, *J* = 3.2 Hz), 16.5; **19F NMR** (564 MHz, CDCl3) δ -149.68 (dd, *J* = 54.0, 25.3 Hz); **IR** (thin film, cm-1): 3030, 2872, 1496, 1453, 1361, 1307, 1208, 1100, 1056, 1027, 912, 801, 732, 695; **HRMS (ESI+)** (m/z) : $[M+Na]^+$ calcd for C₂₇H₂₉FNaO₄ 459.1892, found 459.1917; α _D²⁴ = -41.9 (c=0.5, CH₂Cl₂)

2,3,4-tris-*O***-(phenylmethyl)-β-L-fucopyranosyl fluoride** (minor product).

Me **1 H NMR** (700 MHz, CDCl3) δ 7.41 – 7.28 (m, 15H), 5.16 (dd, *J* = 53.3, 7.1 Hz, 1H), 5.00 OBn **BnÒ** $(d, J = 11.6 \text{ Hz}, 1H)$, 4.87 $(d, J = 10.9 \text{ Hz}, 2H)$, 4.80 $(d, J = 11.7 \text{ Hz}, 1H)$, 4.79 $(d, J = 10.8 \text{ Hz})$ $2e-$ Hz, 1H), 4.74 (d, *J* = 11.8 Hz, 1H), 4.69 (d, *J* = 11.6 Hz, 1H), 3.94 (ddd, *J* = 13.3, 9.7, 7.0 Hz, 1H), 3.60 (m, 2H), 3.55 (dd, *J* = 9.8, 2.9 Hz, 1H), 1.26 (d, *J* = 6.4 Hz, 3H); **13C NMR** (176 MHz, CDCl3) δ 138.2, 138.2, 138.2, 128.4, 128.3, 128.3, 128.2, 128.1, 127.7, 127.7, 127.6, 110.3 (d, *J* = 213.5 Hz, β-C1), 81.4 (d, *J* = 11.7 Hz), 79.0 (d, *J* = 20.6 Hz), 75.8, 74.9, 74.7, 73.3, 70.8 (d, *J* = 4.9 Hz), 16.6; **19F NMR** (564 MHz, CDCl3) δ -139.13 (dd, *J* = 53.4, 13.2 Hz); **IR** (thin film, cm⁻¹): 3030, 2872, 1496, 1453, 1361, 1307, 1208, 1100, 1056, 1027, 912, 801, 732, 695; **HRMS (ESI**+) (m/z): [M+Na]⁺ calcd for C₂₇H₂₉FNaO₄ 459.1892, found 459.1917; $[\alpha]_D^{24} = -54.2$ (c=0.7, CH₂Cl₂)

Synthesis of 2,3,4-tris-*O***-(phenylmethyl)-L-rhamnopyranosyl fluoride (2f)**

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane → 20% EtOAc in hexane) to afford 2f as yellowish syrup in average of 93% yield (R_{f-α} = 0.57 (20%) EtOAc/Hex), $R_{f-β} = 0.43$ (20% EtOAc/Hex)).

2,3,4-Tris-*O***-(phenylmethyl)-α-L-rhamnopyranosyl fluoride** (Major product)

BnO

1 H NMR (700 MHz, CDCl3) δ 7.38 – 7.29 (m, 15H), 5.49 (dd, *J* = 50.6, 2.0 Hz, 1H), 4.95 (d, *J* = 10.9 Hz, 1H), 4.81 (d, *J* = 12.2 Hz, 1H), 4.72 – 4.64 (m, 4H), 3.89 – 3.82 (m, 3H), 3.67 (t, *J* = 9.4 Hz, 1H), 1.36 (d, *J* = 6.2 Hz, 3H); **13C NMR** (176 MHz, CDCl3) δ 138.3,

138.2, 137.8, 128.5, 128.4, 128.4, 128.0, 128.0, 127.9, 127.7, 127.7, 127.7, 106.4 (d, *J* = 221.1 Hz), 79.6, 79.2, 75.4, 73.7 (d, *J* = 35.2 Hz), 73.4, 72.6, 70.6 (d, *J* = 2.3 Hz), 17.9; **19F NMR** (471 MHz, CDCl3) δ -137.07 (d, *J* = 50.6 Hz); **IR** (thin film, cm-1): 3030, 2922, 1721, 1496, 1453, 1363, 1248, 1207, 1184, 1074, 1027, 911, 841, 734, 695; **HRMS (ESI**+) (m/z): [M+Na]⁺ calcd for C₂₇H₂₉FNaO₄ 459.1948, found 459.1948; $\alpha \ln^{23.6} = +7.66$ (c = 1.1, $CH₂Cl₂$)

Synthesis of 2,3,5-Tris-*O***-(phenylmethyl)-D-ribofuranosyl fluoride (2g)**

The product was synthesized employing the general procedure. The product was purified by FCC ($SiO₂$, pure hexane \rightarrow 30% EtOAc in hexane) to afford 2g as yellowish oil in average of 94% yield (R_{f-α} = 0.31 (20%) EtOAc/Hex), $R_{f-\beta} = 0.43$ (20% EtOAc/Hex)).

2,3,5-Tris-*O***-(phenylmethyl)-β-D-ribofuranosyl fluoride** (Major porduct).

1 H NMR (700 MHz, CDCl3) δ 7.38 – 7.27 (m, 15H), 5.69 (d, *J* = 63.4 Hz, 1H, β-C1-H), 4.67 (s, 2H), 4.61 (d, *J* = 12.1 Hz, 1H), 4.56 (d, *J* = 13.6 Hz, 2H), 4.48 (d, *J* = 11.8 Hz, 1H), 4.44 (dtd, *J* = 8.1, 5.1, 2.7 Hz, 1H), 4.14 (ddd, *J* = 7.4, 4.5, 2.4 Hz, 1H), 4.00 (t, *J* = 4.2 Hz,

1H), 3.70 (dd, *J* = 11.0, 3.4 Hz, 1H), 3.60 (dd, *J* = 11.0, 5.3 Hz, 1H); **13C NMR** (176 MHz, CDCl3) δ 138.1, 137.4, 137.3, 128.5, 128.4, 128.3, 128.1, 128.0, 127.9, 127.6, 127.6, 112.5 (d, *J* = 224.4 Hz, β-C1), 82.3 (d, *J* = 2.6 Hz), 78.9 (d, *J* = 30.1 Hz), 73.3, 72.8, 72.8, 70.2; **19F NMR** (471 MHz, CDCl3) δ -115.29 (dt, *J* = 63.1, 5.5 Hz, β-C1- F); **IR** (thin film, cm-1): 3030, 2862, 1726, 1496, 1454, 1358, 1308, 1257, 1208, 1096, 1026, 784, 735, 696; **HRMS (ESI+)** (m/z): [M+Na]⁺ calcd for C₂₆H₂₇FNaO₄ 445.1792, found 445.1782; [α] $p^{24} = +63.9$ (c=0.7, CH₂Cl₂)

Synthesis of 2,3,4,6-tetrakis-*O***-[(4-methoxyphenyl)methyl]-D-galactopyranosyl fluoride (2h)**

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane → 30% EtOAc in hexane) to afford **2h** as cloudy oil in average of 96% yield (R_{f-α} = 0.35 (30% EtOAc/Hex), $R_{f-6} = 0.33$ (30% EtOAc/Hex)).

OPMB PMBO **PMRO PMF**

2,3,4,6-Tetrakis-*O***-[(4-methoxyphenyl)methyl]-α-D-galactopyranosyl fluoride**

(minor product). **¹ H NMR** (700 MHz, CDCl3) δ 7.32 – 7.26 (m, 4H), 7.22 – 7.15 (m, 4H), $6.92 - 6.79$ (m, 8H), 5.51 (dd, J = 53.7, 2.8 Hz, 1H), 4.84 (d, J = 11.0 Hz, 1H), 4.77 (d, J = 11.5 Hz, 1H), 4.72 (d, J = 11.3 Hz, 1H), 4.65 (dd, J = 13.8, 11.4 Hz, 2H), 4.49 (d, J = 11.0

 2_h Hz, 1H), 4.42 (d, J = 11.5 Hz, 1H), 4.34 (d, J = 11.5 Hz, 1H), 4.04 (t, J = 6.5 Hz, 1H), 4.01 – 3.93 (m, 2H), 3.88 (dd, J = 10.1, 2.7 Hz, 1H), 3.82 (s, 3H), 3.80 (s, 3H), 3.80 (s, 3H), 3.79 (s, 3H), 3.51 – 3.44 (m, 2H); **13C NMR** (176 MHz, CDCl3) δ 159.3, 159.3, 159.2, 159.2, 130.6, 130.6, 130.2, 129.8, 129.6, 129.5, 129.2, 113.8, 113.7, 106.4 (d, *J* = 225.8 Hz), 78.2, 75.3 (d, *J* = 23.5 Hz), 74.4, 73.8, 73.4, 73.1, 72.7, 71.8 (d, *J* = 2.7 Hz), 68.0, 55.3, 55.3, 55.2; **19F NMR** (564 MHz, CDCl3) δ -150.14 (dd, *J* = 53.7, 25.5 Hz);**IR** (thin film, cm-1): 2912, 2835, 1611, 1585, 1512, 1463, 1442, 1422, 1362, 1301, 1244, 1172, 1096, 1031, 817, 757, 710, 665; **HRMS (ESI+)** (m/z): $[M+NH_4]^+$ calcd for C₃₈H₄₇FNO₉ 680.3229, found 680.3281; $[\alpha]_D^{23.3} = +26.9$ (c = 0.13, CH₂Cl₂).

2,3,4,6-Tetrakis-*O***-[(4-methoxyphenyl)methyl]-β-D-galactopyranosyl fluoride**

(major product). **¹ H NMR** (700 MHz, CDCl3) δ 7.28 (d, *J* = 8.6 Hz, 2H), 7.26 (d, *J* = 6.1 Hz, 2H), 7.20 (dd, *J* = 8.5, 5.5 Hz, 4H), 6.87 – 6.85 (m, 6H), 6.83 (d, *J* = 8.6 Hz, 2H), 5.13 (dd, *J* = 53.2, 7.0 Hz, 1H), 4.83 (d, *J* = 11.2 Hz, 1H), 4.76 (d, *J* = 10.7 Hz, 1H), 4.69 (d, *J*

= 10.6 Hz, 1H), 4.64 (q, *J* = 11.4 Hz, 2H), 4.52 (d, *J* = 11.3 Hz, 1H), 4.42 (d, *J* = 11.4 Hz, 1H), 4.34 (d, *J* = 11.4 Hz, 1H), 3.91 – 3.85 (m, 1H), 3.82 (d, *J* = 2.7 Hz, 1H), 3.81 (s, 3H), 3.80 (s, 3H), 3.80 (s, 3H), 3.79 (s, 3H), 3.61 – 3.57 (m, 2H), 3.56 – 3.52 (m, 1H), 3.49 – 3.46 (m, 1H); **13C NMR** (176 MHz, CDCl3) δ 159.4, 159.3, 159.2, 159.2, 130.5, 130.3, 130.3, 129.8, 129.8, 129.6, 129.2, 113.8, 113.8, 113.8, 113.6, 110.3 (d, *J* = 215.3 Hz, β-C1), 80.7 (d, *J* = 11.3 Hz), 78.8 (d, *J* = 20.7 Hz), 74.7, 74.6, 74.1, 73.7 (d, *J* = 4.8 Hz), 73.2, 72.7, 72.5, 68.1, 55.3, 55.3, 55.2; **19F NMR** (471 MHz, CDCl3) δ -138.96 (dd, *J* = 53.0, 13.4 Hz); **IR** (thin film, cm-1): 2912, 2835, 1611, 1585, 1512, 1463, 1442, 1422, 1362, 1301, 1244, 1172, 1096, 1031, 817, 756, 710, 664; **HRMS (ESI+)** (m/z): $[M+NH_4]^+$ calcd for C₃₈H₄₇FNO₉ 680.3229, found 680.3281; $[\alpha]_D^{24} = +39.8$ (c = 0.19, CH₂Cl₂).

Synthesis of 2,3,4,6-Tetrakis-*O***-[(4-methoxyphenyl)methyl]-D-glucopyranosyl fluoride (2i)**

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane \rightarrow 30% EtOAc in hexane) to afford 2i as clear oil in average of 96% yield (R_f = 0.33 (30% EtOAc/Hex)). The NMR spectrum of the product is in a good agreement with our previously published NMR data.¹⁵

	$1st$ Run	$2nd$ Run
Isolated Yield	$99\% (65.4 mg)$	$94\% (62.3 mg)$
	α : β = 1 : 1.4	α : β = 1 : 1.4

2,3,4,6-Tetrakis-*O***-[(4-methoxyphenyl)methyl]-β-D-glucopyranosyl fluoride**

(Major product, Isolated as α : β = 1 : 1.4 mixture).

1 H NMR (600 MHz, CDCl3) δ 7.35 – 7.20 (m, 14H), 7.10 – 7.01 (m, 4.6H), 6.93 – 6.77 (m, 18.3H), 5.49 (dd, $J = 53.2$, 2.6 Hz, 1H, C₁-H_a), 5.23 (dd, $J = 52.8$, 6.8 Hz, $2i-$ 1.4H, C1-Hβ), 4.88 (d, *J* = 10.5 Hz, 1H), 4.83 (d, *J* = 10.7 Hz, 1.4H), 4.81 – 4.70 (m, 7H), 4.65 (dd, *J* = 11.1, 4.8 Hz, 2.4H), 4.58 (dd, *J* = 11.8, 3.3 Hz, 2.4H), 4.48 (d, *J* = 11.8 Hz, 1.4H), 4.45 – 4.37 (m, 3.3H), 3.96 – 3.88 (m, 2.4H), 3.84 – 3.77 (m, 27.6H), 3.73 – 3.59 (m, 8.6H), 3.59 – 3.49 (m, 3.7H); **19F NMR** (564 MHz, CDCl3) δ - 138.29 (dd, *J* = 53.0, 12.0 Hz, 1.4F, C1-Fβ), -149.54 (dd, *J* = 53.3, 25.7 Hz, 1F, C1-Fα); **IR** (thin film, cm-1): 2907, 2835, 1611, 1585, 1512, 1463, 1441, 1359, 1302, 1244, 1172, 1083, 1031, 816, 758, 712, 637; **HRMS (ESI+)** (m/z) : $[M+Na]^+$ calcd for C₃₈H₄₃FNaO₉ 685.2783, found 685.2772; $[a]_p^{24} = +20.7$ (c = 0.34, CH₂Cl₂).

¹⁵Kim, S.; Khomutnyk, Y.; Bannykh, A.; Nagorny, P. *Org. Lett*. **2021**, 23, 1, 190-194

Synthesis of 2,3,4,6-Tetrakis-*O***-[(4-methoxyphenyl)methyl]-D-mannopyranosyl fluoride (2j)**

The product was synthesized employing the general procedure. The product was purified by $FCC(SiO₂, pure)$ hexane → 10% EtOAc in hexane) to afford 2**j** as cloudy oil in average of 96% yield (R_{f-α} = 0.53 (40% EtOAc/Hex), $R_{f-6} = 0.45$ (40% EtOAc/Hex)).

2,3,4,6-Tetrakis-*O***-[(4-methoxyphenyl)methyl]-α-D-mannopyranosyl fluoride**

(Major Product) **¹ H NMR** (500 MHz, CDCl3) δ 7.29 – 7.24 (m, 6H), 7.06 (d, *J* = 8.6 Hz, 2H), 6.90 – 6.79 (m, 8H), 5.53 (dd, *J* = 50.6, 1.6 Hz, 1H), 4.75 (dd, *J* = 24.8, 11.2 Hz, 2H), $2j-$ 4.64 – 4.53 (m, 4H), 4.44 (dd, *J* = 21.7, 11.1 Hz, 2H), 4.00 – 3.95 (m, 1H), 3.86 (ddd, *J* = 10.0, 4.8, 1.8 Hz, 1H), 3.80 (dd, *J* = 10.6, 4.1 Hz, 14H), 3.71 (dd, *J* = 11.0, 4.7 Hz, 1H), 3.66 (dd, *J* = 10.9, 2.0 Hz, 1H); ¹³C NMR (126 MHz, cdcl₃) δ 159.4, 159.2, 159.2, 159.2, 130.4, 130.4, 130.2, 129.9, 129.6, 129.6, 129.3, 113.8, 113.7, 106.5 (d, *J* = 221.9 Hz), 78.8, 74.7, 74.2, 74.2, 73.7, 73.1, 73.0, 72.9, 72.8, 72.2, 68.2, 55.3, 55.2, 55.2; **19F NMR** (471 MHz, CDCl3) δ -137.88 (d, *J* = 50.6 Hz); **IR** (thin film, cm-1): 2908, 2835, 1611, 1585, 1512, 1463, 1442, 1422, 1361, 1301, 1244, 1172, 1090, 1030, 950, 816, 756, 711, 667; **HRMS (ESI+)** (m/z): [M+Na]+

calcd for C₃₈H₄₃FNaO₉ 685.2783, found 685.2753; α _D²⁴ = +7.45 (c = 0.88, CH₂Cl₂).

Synthesis of 2,3,4,6-Tetra-*O***-acetyl-D-galactopyranosyl fluoride (2k)**

 ACC

OAc

∆ഹ് $2k -$

AcO

 AcC

The product was synthesized employing a procedure for acetylated glycosides. After a work-up, the reaction mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 40% EtOAc in hexane) to afford 2k as yellowish oil in average of 73% yield ($R_{f-a} = 0.38$ (40% EtOAc/Hex), $R_{f-a} = 0.30$ (40% EtOAc/Hex)).

2,3,4,6-Tetra-*O***-acetyl-α-D-galactopyranosyl fluoride** (Minor product).

1 H NMR (700 MHz, Chloroform-*d*) δ 5.80 (d, *J* = 53.3 Hz, 1H), 5.53 (t, *J* = 2.5 Hz, 1H), 5.37 (dt, *J* = 10.8, 2.8 Hz, 1H), 5.19 (ddt, *J* = 23.8, 10.9, 2.6 Hz, 1H), 4.41 (t, *J* = 6.7 Hz, 1H), 4.18 – 4.09 (m, 2H), 2.16 (s, 3H), 2.12 (s, 3H), 2.06 (s, 3H), 2.01 (s, 3H); **13C NMR** (176 MHz,

 $2k$ cdcl3) δ 170.3, 170.2, 170.0, 169.9, 104.3 (d, *J* = 228.4 Hz, α-C1), 68.9 (d, *J* = 3.5 Hz), 67.5, 67.3, 67.0, 61.3, 20.7, 20.6, 20.6; **19F NMR** (564 MHz, CDCl3) δ -150.78 (dd, *J* = 53.3, 23.9 Hz); **IR** (thin film, cm-1): 2921, 2851, 1744, 1369, 1212, 1167, 1104, 1042, 951, 899; **HRMS (ESI+)** (m/z): [M+Na]+ calcd for C14H19FNaO9 373.0911, found 373.0905; α b^{25} = +28.1 (c = 0.4, CH₂Cl₂)

2,3,4,6-Tetra-*O***-acetyl-β-D-galactopyranosyl fluoride** (Major product).

1 H NMR (700 MHz, Chloroform-*d*) δ 5.44 – 5.40 (m, 1H), 5.35 – 5.21 (m, 2H), 5.04 (dt, *J* = 10.5, 2.8 Hz, 1H), 4.24 – 4.18 (m, 2H), 4.05 (t, *J* = 6.6 Hz, 1H), 2.17 (s, 3H), 2.10 (s, 3H), 2.06 (s, 3H), 2.01 (s, 3H); **13C NMR** (176 MHz, CDCl3) δ 170.6, 170.0, 170.0, 169.3,

107.1 (d, *J* = 218.6 Hz, β-C1), 71.2 (d, *J* = 4.7 Hz), 69.9 (d, *J* = 10.6 Hz), 68.8 (d, *J* = 24.9 Hz), 66.4, 61.3, 20.7, 20.6, 20.5; **19F NMR** (564 MHz, CDCl3) δ -141.49 (dd, *J* = 51.0, 12.9 Hz); **IR** (thin film, cm-1): 2921, 2851, 1744, 1369, 1212, 1167, 1104, 1042, 951, 899; **HRMS (ESI+)** (m/z): [M+Na]+ calcd for C14H19FNaO9 373.0911, found 373.0905; $\left[\alpha\right]p^{25} = +6.0$ (c = 0.4, CH₂Cl₂)

Synthesis of 2,3,4,6-Tetra-*O***-acetyl-D-glucopyranosyl fluoride (2l)**

The product was synthesized employing a procedure for acetylated glycosides. After a work-up, the reaction mixture was subjected to FCC (SiO2, pure hexane → 50% EtOAc in hexane) to afford **2l** as colorless oil in average of 76% yield ($R_{f-a} = 0.47$ (50% EtOAc/Hex), $R_{f-a} = 0.43$ (50% EtOAc/Hex)).

2,3,4,6-Tetra-*O***-acetyl-α-D-glucopyranosyl fluoride** (Minor product)

1 H NMR (700 MHz, CDCl3) δ 5.75 (dd, *J* = 52.8, 2.8 Hz, 1H), 5.50 (t, *J* = 9.9 Hz, 1H), 5.16 (t, *J* = 10.0 Hz, 1H), 4.96 (ddd, *J* = 24.2, 10.2, 2.8 Hz, 1H), 4.29 (dd, *J* = 12.5, 4.1 Hz, 1H), 4.19 (ddd, *J* = 10.3, 4.2, 2.2 Hz, 1H), 4.15 (dd, *J* = 12.5, 2.2 Hz, 1H), 2.11 (s,

3H), 2.10 (s, 6H), 2.05 (s, 3H), 2.03 (s, 3H); **13C NMR** (176 MHz, CDCl3) δ 170.5, 170.0, 169.9, 169.4, 103.7 (d, *J* = 229.5 Hz, α-C1), 70.2 (d, *J* = 24.5 Hz), 69.8 (d, *J* = 4.3 Hz), 69.4, 67.3, 61.2, 20.7, 20.6, 20.5; **19F NMR** (564 MHz, CDCl₃) δ -149.76 (dd, *J* = 52.8, 24.2 Hz, α-C₁-F); **IR** (thin film, cm⁻¹): 2918, 1743, 1434, 1367, 1208, 1160, 1103, 1033, 904, 779; **HRMS (ESI+)** (m/z): [M+Na]+ calcd for C14H19FNaO9 373.0892, found 373.0899

2,3,4,6-Tetra-*O***-acetyl-β-D-glucopyranosyl fluoride** (Major product)

OAc

 Ac $21-$

 $AccC$ AcC

> **1 H NMR** (700 MHz, CDCl3) δ 5.36 (ddd, *J* = 52.0, 6.2, 1.8 Hz, 1H), 5.20 (tt, *J* = 9.4, 4.8 Hz, 2H), 5.12 – 5.08 (m, 1H), 4.29 – 4.24 (m, 1H), 4.24 – 4.20 (m, 1H), 3.90 (ddd, *J* = 7.5, 4.9, 2.5 Hz, 1H), 2.10 (s, 3H), 2.10 (s, 3H), 2.04 (s, 3H), 2.03 (s, 3H); **13C NMR**

(176 MHz, CDCl3) δ 170.6, 170.0, 169.3, 169.1, 106.2 (d, *J* = 219.6 Hz, β-C1), 72.0 (d, *J* = 3.8 Hz), 71.8 (d, *J* = 8.3 Hz), 71.2 (d, *J* = 28.8 Hz), 67.4, 61.7, 20.7, 20.6; **19F NMR** (564 MHz, CDCl3) δ -137.24 (dd, *J* = 52.0, 10.4 Hz, β-C1-F), -149.76 (dd, *J* = 52.8, 24.2 Hz).; **IR** (thin film, cm-1): 2918, 1743, 1434, 1367, 1208, 1160, 1103, 1033, 904, 779; **HRMS** (ESI+) (m/z): [M+Na]⁺ calcd for C₁₄H₁₉FNaO₉ 373.0892, found 373.0899; $\left[\alpha\right]_0^{24}$ = +13.6 $(c = 0.5, CH_2Cl_2)$

Synthesis of 2,3,4,6-Tetra-*O***-acetyl-D-mannopyranosyl fluoride (2m)**

The product was synthesized employing the general procedure. After a work-up, the reaction mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 40% EtOAc in hexane) to afford pure **2m** in average of 81% yield as cloudy oil ($Rf = 0.33$ (40% EtOAc/Hex)). Stereochemistry of the products was confirmed by comparing with previously reported NMR spectra.16

2,3,4,6-Tetra-*O***-acetyl-α-D-mannopyranosyl fluoride** (Major product).

1 H NMR (700 MHz, Chloroform-*d*) δ 5.58 (d, *J* = 48.6 Hz, 1H), 5.40 (s, 1H), 5.35 (d, *J* = 5.0 Hz, 2H), 4.32 – 4.27 (m, 1H), 4.16 (dd, *J* = 14.7, 4.4 Hz, 2H), 2.17 (s, 3H), 2.11 (s, 3H), 2.06 (s, 3H), 2.01 (s, 3H); **13C NMR** (176 MHz, CDCl3) δ 170.5, 169.7, 169.6, 169.5, 104.7

(d, *J* = 223.7 Hz, β-C1), 70.9 (d, *J* = 2.7 Hz), 68.2, 67.7 (d, *J* = 39.5 Hz), 65.0, 61.8, 20.7, 20.6, 20.6; **19F NMR** (564 MHz, CDCl3) δ -138.38 (d, *J* = 48.4 Hz); **IR** (thin film, cm-1): 2960, 1742, 1434, 1368, 1211, 1172, 1087, 1051, 1019, 974, 916, 795, 685; **HRMS** (ESI+) (m/z): [M+Na]+ calcd for C14H19FNaO9 373.0911, found 373.0905; $[\alpha]_D^{24} = +11.2$ (c = 0.5, CH₂Cl₂).

¹⁶ Chambers, R. D.; Sandford, G.; Sparrowhawk, M. E.; Atherton, M. J. *J. Chem. Soc., Perkin Trans. 1*, **1996**, 1941-1944

Synthesis of 2,3,4-tri-*O***-acetyl-L-rhamnopyranosyl fluoride (2n)**

The product was synthesized employing the general procedure. After a work-up, the reaction mixture was purified by FCC (SiO₂, pure hexane \rightarrow 40% EtOAc in hexane) to afford 2n as colorless oil in average of 79% yield (R_f = 0.37(30% EtOAc/Hex)). Stereochemistry of the products was confirmed by comparing with previously reported NMR spectra.¹⁷

2,3,4-Tri-*O***-acetyl-α-L-rhamnopyranosyl fluoride** (Major product)

1 H NMR (500 MHz, CDCl3) δ 5.49 (dd, *J* = 48.6, 1.9 Hz, 1H), 5.38 (ddd, *J* = 3.2, 1.9, 0.9 Hz, 1H), 5.29 (ddd, *J* = 10.2, 3.6, 1.7 Hz, 1H), 5.12 (t, *J* = 10.0 Hz, 1H), 4.04 (dq, *J* = 9.9, 6.2 Hz, 1H), 2.16 (s, 3H), 2.06 (s, 3H), 2.00 (s, 3H), 1.27 (d, *J* = 6.3 Hz, 3H); **13C NMR** (126

MHz, CDCl3) δ 169.8, 169.8, 169.7, 104.8 (d, *J* = 221.9 Hz), 70.0, 68.9 (d, *J* = 3.0 Hz), 68.2 (d, *J* = 1.7 Hz), 68.0 (d, $J = 40.3$ Hz), 20.7, 20.7, 20.6, 17.3; ¹⁹F NMR (564 MHz, CDCl₃) δ -137.49 (d, $J = 49.1$ Hz); **IR** (thin film, cm-1): 2925, 1748, 1605, 1509, 1457, 1369, 1293, 1242, 1214, 1173, 1038, 974, 929, 893, 830, 813, 793, 778, 684

¹⁷Nishiyama, K.; Esaki, S.; Deguchi, I.; Sugiyama, N.; Kamiya, S. *Biosci. Biotechnol. Biochem.* **1993**, 57, 107–114

Synthesis of 2,3,4-tri-*O***-benzoyl-L-rhamnopyranosyl fluoride (2o)**

The product was synthesized employing the general procedure. After a work-up, the reaction mixture was purified by FCC (SiO2, pure hexane → 30% EtOAc in hexane) to afford **2o** as cloudy syrup in average of 84% yield (Rf-α $= 0.52$ (30% EtOAc/Hex), R_{f-β} = 0.46 (30% EtOAc/Hex)).

2,3,4-Tri-*O***-benzoyl-α-L-rhamnopyranosyl fluoride**

1 H NMR (700 MHz, CDCl3) δ 8.10 (dd, *J* = 8.3, 1.3 Hz, 2H), 8.00 – 7.97 (m, 2H), 7.83 (dd, *J* = 8.4, 1.4 Hz, 2H), 7.63 (tt, *J* = 8.8, 1.3 Hz, 1H), 7.55 – 7.52 (m, 3H), 7.52 – 7.49 (m, 2H), 7.43 (tt, *J* = 7.4, 1.3 Hz, 1H), 7.42 – 7.38 (m, 2H), 5.87 – 5.81 (m, 2.5H, C4-H + C2-H + 1/2 C_1-H), 5.76 (d, $J = 1.8$ Hz, 0.5H, $1/2$ C₁-H), 5.74 (t, $J = 9.9$ Hz, 1H, C₃-H), 4.39 (dq, $J = 9.9$, 6.2 Hz, 1H), 1.43 (d, *J* = 6.2 Hz, 3H); **13C NMR** (176 MHz, cdcl3) δ 165.6, 165.4, 165.3, 133.7, 133.5, 133.3, 130.0, 129.8, 129.7, 129.0, 128.9, 128.9, 128.7, 128.5, 128.3, 104.9 (d, *J* = 222.4 Hz, α-C1-F), 70.7, 69.2 (d, *J* = 2.7 Hz), 69.1 (d, *J* = 1.0 Hz), 68.9 (d, *J* = 40.6 Hz), 17.6; **19F NMR** (564 MHz, CDCl3) δ -137.24 (d, *J* = 48.8 Hz); **IR** (thin film, cm-1): 2985, 1724, 1601, 1584, 1451, 1315, 1276, 1258, 1173, 1091, 1068, 1026, 973, 910, 851, 793, 705, 685; **HRMS (ESI+)** (m/z): $[M-F]^+$ calcd for C₂₇H₂₃O₇⁺ 459.1438, found 459.1427; $[a]p^{23.9} = +166.9$ (c (w/v%) = 4.15, CH₂Cl₂)

Synthesis of 2,3-bis-*O***-(phenylmethyl)-4,6-***O***-[(R)-phenylmethylene]-D-mannopyranosyl fluoride (2p)** The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane \rightarrow 20% EtOAc in hexane) to afford 2p as cloudy oil in average of 96% yield (R_{f-α} = 0.44 (20%) EtOAc/Hex)). Stereochemistry of the products was determined by comparing with previously reported NMR spectra.¹⁸

 RnC

 $2p$

2,3-Bis-*O***-(phenylmethyl)-4,6-***O***-[(R)-phenylmethylene]-α-D-mannopyranosyl fluoride** (major product).

1 H NMR (700 MHz, CDCl3) δ 7.51 (d, *J* = 7.1 Hz, 2H), 7.41 – 7.28 (m, 13H), 5.65 (s, 1H), 5.50 (d, *J* = 49.6 Hz, 1H), 4.89 (d, *J* = 12.0 Hz, 2H), 4.74 – 4.68 (m, 2H), 4.33 –

4.27 (m, 2H), 3.99 – 3.92 (m, 3H), 3.87 (td, *J* = 10.3, 2.0 Hz, 1H); **13C NMR** (176 MHz, CDCl3) δ 138.3, 137.6, 137.4, 128.9, 128.5, 128.4, 128.2, 128.1, 128.1, 127.7, 127.6, 126.1, 107.0 (d, *J* = 223.7 Hz), 101.6, 78.3, 75.5 (d, *J* = 2.5 Hz), 75.0, 74.8, 74.2, 73.5, 68.3, 66.1 (d, *J* = 2.4 Hz); **19F NMR** (471 MHz, CDCl3) δ -137.06 (d, *J* = 49.7 Hz); **IR** (thin film, cm-1): 2919, 1454, 1372, 1313, 1214, 1166, 1099, 1054, 1006, 965, 925, 799, 735, 696; **HRMS (ESI+)** (m/z): $[M+Na]^+$ calcd for C₂₇H₂₇FNaO₅ 473.1692, found 473.1729; $\left[\alpha\right]p^{23} = -17.4$ (c = 0.6, CH₂Cl₂).

¹⁸Lee, Y. J.; Baek, J. Y.; Lee, B-Y.; Kang, S. S.; Park, H-S.; Jeon, H. B.; Kim, K. S. Carbohydrate Research. **2006**. 341(10). 1708-1716

Synthesis of 2,3,6-Tris-O-(phenylmethyl)-4-O-[2,3,4,6-tetrakis-O-(phenylmethyl)-β-D-glucopyranosyl]-Dglucopyranosyl fluoride (2q)

The product was synthesized employing the general procedure. The product was purified by FCC ($SiO₂$, pure hexane → 30% EtOAc in hexane) to afford **2q** as white solid in average of 84% yield (Rf-α = 0.58 (30% EtOAc/Hex), $R_{f-\beta} = 0.57$ (30% EtOAc/Hex)).

2,3,6-Tris-O-(phenylmethyl)-4-O-[2,3,4,6-tetrakis-O-(phenylmethyl)-β-D-BnO OBn **glucopyranosyl]-α-D-glucopyranosyl fluoride** (minor product). Ċ. **1 H NMR** (700 MHz, CDCl3) δ 7.40 (d, *J* = 6.5 Hz, 2H), 7.33 – 7.20 (m, 31H), 7.19 BnO. (d, $J = 7.4$ Hz, 2H), 5.48 (dd, $J = 53.4$, 2.7 Hz, 1H, α -C₁-H), 5.12 (dd, $J = 11.4$, 2.0 ÓBn OBn Hz, 1H), 4.89 – 4.86 (m, 1H), 4.83 – 4.72 (m, 6H), 4.63 (dd, *J* = 11.9, 2.0 Hz, 1H), ٠o $2q -$ 4.56 (d, *J* = 11.5 Hz, 2H), 4.44 (dd, *J* = 12.3, 2.0 Hz, 1H), 4.39 (ddd, *J* = 11.7, 6.9, \overline{ORn}

2.8 Hz, 3H), 4.07 – 4.03 (m, 1H), 3.90 – 3.84 (m, 2H), 3.80 (dd, *J* = 10.1, 2.2 Hz, 1H), 3.73 (d, *J* = 10.9 Hz, 1H), 3.61 (t, *J* = 8.8 Hz, 1H), 3.58 – 3.55 (m, 1H), 3.48 (ddd, *J* = 25.5, 19.6, 10.3 Hz, 4H), 3.39 – 3.35 (m, 1H), 3.33 – 3.29 (m, 1H); **13C NMR** (176 MHz, CDCl3) δ 139.2, 138.6, 138.5, 138.3, 138.3, 137.9, 137.6, 128.5, 128.4, 128.4, 128.4, 128.3, 128.2, 128.1, 128.0, 128.0, 127.9, 127.8, 127.8, 127.8, 127.8, 127.7, 127.6, 127.6, 127.3, 127.2, 105.8 (d, *J* = 226.8 Hz, α-C1), 102.5, 84.9, 82.7, 79.8, 78.2 (d, *J* = 24.3 Hz), 78.0, 75.6, 75.6, 75.4, 75.2, 75.0, 74.8, 73.8, 73.3, 72.6 (d, *J* = 4.2 Hz), 69.0, 69.0, 67.1; **19F NMR** (471 MHz, CDCl3) δ -149.55 (dd, *J* = 53.4, 25.8 Hz); **IR** (thin film, cm-1): 1497, 1453, 1361, 1072, 1029, 1005, 908, 753, 731, 695, 665; **HRMS (ESI+)** (m/z): $[M+NH_4]^+$ calcd for $C_{61}H_{67}FNO_{10}$ 992.4738, found 992.4733; $[\alpha]_D^{23.6} = +23.1$ (c = 0.23, CH₂Cl₂)

BnO. ORn **BnO OBn** OBn Ω $2q -$ OB_n ÓBn

2,3,6-Tris-O-(phenylmethyl)-4-O-[2,3,4,6-tetrakis-O-(phenylmethyl)-β-Dglucopyranosyl]-β-D-glucopyranosyl fluoride (major product).

1 H NMR (700 MHz, CDCl3) δ 7.37 (d, *J* = 5.2 Hz, 2H), 7.34 – 7.22 (m, 31H), 7.19 (d, *J* = 7.3 Hz, 2H), 5.25 (dd, *J* = 53.5, 6.5 Hz, 1H, β-C1-H), 5.05 (d, *J* = 11.4 Hz, 1H), 4.89 (d, *J* = 10.9 Hz, 1H), 4.81 (d, *J* = 10.6 Hz, 2H), 4.75 (dd, *J* = 17.7, 9.3 Hz, 4H), 4.69 (d, *J* = 11.2 Hz, 1H), 4.58 (dd, *J* = 17.0, 11.4 Hz, 2H), 4.50 (d, *J* = 7.9 Hz, 1H),

4.47 – 4.39 (m, 3H), 4.13 (t, *J* = 8.9 Hz, 1H), 3.84 (dd, *J* = 11.1, 3.4 Hz, 1H), 3.72 – 3.48 (m, 8H), 3.39 (t, *J* = 8.5 Hz, 1H), 3.30 (dd, *J* = 10.4, 4.0 Hz, 1H); **13C NMR** (176 MHz, CDCl3) δ 138.9, 138.5, 138.4, 138.4, 138.2, 137.8, 137.8, 128.4, 128.4, 128.4, 128.3, 128.3, 128.2, 128.1, 128.1, 127.9, 127.8, 127.8, 127.8, 127.8, 127.8, 127.7, 127.7, 127.6, 127.6, 127.5, 127.4, 127.3, 109.5 (d, *J* = 215.9 Hz, β-C1), 102.3, 84.9, 82.7, 81.3 (d, *J* = 10.3 Hz), 80.5 (d, *J* = 23.0 Hz), 77.9, 75.8, 75.6, 75.0, 74.9 (d, *J* = 3.7 Hz), 74.8, 74.7, 74.3, 73.4, 73.3, 68.9, 67.7; **19F NMR** (471 MHz, CDCl3) δ -135.75 (dd, *J* = 53.5, 12.3 Hz); **IR** (thin film, cm-1): 1497, 1453, 1361, 1072, 1029, 1005, 908, 753, 731, 695, 665; **HRMS (ESI+)** (m/z): [M+NH4] ⁺ calcd for C61H67FNO10 992.4738, found 992.4733; $\begin{bmatrix} \alpha \vert p^{23} \end{bmatrix}$ = +25.9 (c = 1.1, CH₂Cl₂)

d. Summary of α:β ratio for 1a-q and 2a-q

Table S7. Initial and final α:β ratios of the starting materials 1a-q and products 2a-q

e. Comparison with DAST method

References:

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2e: Guan, X.; Chaffey, P. K.; Zeng, C.; Greene, E. R.; Chen, L.; Drake, M. R.; Chen, C.; Groobman, A.; Resch, M. G.; Himmel, M. E.; Beckham, G. T.; Tan, Z. *Chem. Sci.* **2015**, 6, 7185

2g: Ginisty, M.; Gravier-Pelletier, C.; Merrer, Y. L. *Tetrahedron: Asymmetry.* **2006**, 17, 142-150

2h: Li, X.; Wu, P.; Cheng, S.; Lv, X. *J. Med. Chem.* **2021**, 55, 6, 2702-2710

2l: Zagrobelny, M.; Olsen, C. E.; Pentzold, S.; Fürstenberg-Hägg, J.; Jørgensen, K.; Bak, S.; Møller, B. L.; Motawia, M. S. *Insect Biochemistry and Molecular Biology.* **2014**, 44, 44-53

f. Large Scale Batch Synthesis

5 Gram Scale Synthesis

Figure S4. Experimental setup adopted from Baran Group's Work¹⁹

Figure S5. (A) Reaction setup (B) Reaction progress monitored by TLC (C) Electrodes after the reaction (D) Isolated product

¹⁹Hu, P.; Peters, B. K.; Malapit, C. A.; Vantourout, J. C.; Wang, P.; Li, J.; Mele, L.; Echeverria, P.-G.; Minteer, S. D.; Baran, P. S. *J. Am. Chem. Soc.* **2020**, 142, 50, 20979-20986

To achieve high current density, the experimental setup was adopted from Baran group's large scale setup (**Fig. SI-4**).19 Electrodes were prepared by cutting off tin plate (3.2 mm thickness, 5.8 cm x 10 cm x 2) and zinc plate (0.5 mm thickness, 6.0 cm x 10 cm) and 5.0 cm height of the electrodes were submerged in the solution. Due to the voltage limit (31 V) of the DC power supply, the concentration of electrolyte was doubled (0.3 M). Into an oven dried glass jar containing a stir bar, 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranoside **1a** (9.25 mmol, 1.0 equiv.), tetrabutylammonium perchlorate (83.4 mmol, 0.3 M) were dissolved in dry THF (278 mL). After the addition of triethylamine (27.7 mmol, 3.0 equiv.), the reaction was placed inside a water bath (to relieve the heat generated during electrolysis) and stirred until the solution became homogeneous. The reaction was then sparged with $SF₆$ balloons for 1 minute before the electric current was applied, and then the rest of the reaction. The electric current was applied for 6 hours, and then the reaction mixture was diluted with 300 mL of ethyl acetate. The diluted mixture was then washed with 1N HCl (250 mL x 3), NaHCO₃ (250 mL x 1), water (250 mL x 5), and brine (250 mL x 1). The extracted organic layer was dried with Na2SO4 and concentrated *in vacuo*, which was subjected to FCC to afford 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranosyl fluoride **2a** in 98 % yield (4.92 g, α : β = 1 : 2.3) as yellow syrup.

IV. Cyclic Voltammetry Studies

General information

Cyclic voltammetry studies were performed with the Pine WaveNow^{XV} Potentiostat/Galvanostat. All experiments were measured under slight argon stream. SF_6 was added by sparging SF_6 gas to the solution for 1 minute. Electrodes were polished with 1000-grit sandpaper and cleaned by sonicating in acetone, and they were dried under high vacuum before use. All electrolytes were dried under high vacuum before use. THF was prepared from Innovative Technology PureSolv solvent drying system and dry acetonitrile was purchased from Alfa Aesar. All solvents were degassed with argon before use.

a. Identifying half reduction potential of SF_6

Scan range: -3.0 V to 0.0 V (-3.1 V to -0.5 V is shown on the graph after voltage reference with $Fc^+/Fc)$ Scan rates: 20 mV/s Electrolyte: TBAClO4 (0.1M) / Solvent: acetonitrile (12 mL)

Working electrode: 0.635 mm diameter Pt wire (0.6 cm submerged)

Counter electrode: Pt plate (0.8 cm^2)

Reference electrode: Ag/AgCl (voltage is referenced to ferrocene [0.001 M])

Figure S6. Blank is shown in black line and the reduction of SF_6 is shown in red line. Voltage was corrected by ferrocene (0.001 M)

We first attempted to measure the half reduction potential of $SF₆$ by adopting the CV experimental set-up from Magnier group.²⁰ The voltammogram shows $E_{\text{red1/2}} = -2.44 \text{ V (vs. } Fc^{\dagger}/Fc)$.

Half reduction potential of SF_6 reference to SCE: -1.99 V vs. SCE

²⁰Bouvet, S.; Pegot, B.; Sengmany, S.; Gall, E. L.; Leonel, E.; Goncalves, A.-M.; Magnier, E. *Beilstein J. Org. Chem.* **2020**, 16, 2948-2953

b. $SF₆$ at multiple scan rate

Scan range: -2.0 V to 0.0 V

Scan rates (in chronological order): 1) 50 mV/s 2) 100 mV/s 3) 25 mV/s 4) 75 mV/s 5) 200 mV/s Electrolyte: TBAClO4 (0.15M) / Solvent: THF (12 mL) Working electrode: 1.5 mm diameter Sn wire (0.6 cm submerged) Counter electrode: 2.0 mm diameter Zn wire (1.5 cm submerged) Reference electrode: Ag/AgCl

Figure S7. Reduction of SF₆ with Zn as sacrificial anode at multiple scan rates.

The voltammogram shows the decrease of cathodic current in the chronological order of the scans. It suggests the increased internal resistance from either a) electrode passivation or b) accumulation of salt formed from solvated Zn cation.

Scan range: -2.0 V to 0.0 V Scan rates (in chronological order): 1) 50 mV/s 2) 100 mV/s 3) 25 mV/s 4) 75 mV/s 5) 200 mV/s Electrolyte: TBAClO4 (0.15M) / Solvent: THF (12 mL) Working electrode: 1.5 mm diameter Sn wire (0.6 cm submerged) Counter electrode: Pt plate (0.8 cm²) Reference electrode: Ag/AgCl

Figure S8. Reduction of SF_6 with Pt as anode at multiple scan rates.

Pt anode is resistant to galvanic corrosion and does not dissociate Pt cation under electric current. As expected with Pt as anode, the voltammogram shows consistent current throughout the entire CV experiment. This result supports that the decrease of cathodic current in previous CV experiment is caused by Zn sacrificial anode.

c. SF6 + NEt3 at multiple scan rate

Scan range: -2.0 V to 0.0 V Scan rates (in chronological order): 1) 50 mV/s 2) 100 mV/s 3) 25 mV/s 4) 75 mV/s 5) 200 mV/s Electrolyte: TBAClO4 (0.15M) / Solvent: THF (12 mL) Working electrode: 1.5 mm diameter Sn wire (0.6 cm submerged) Counter electrode: 2.0 mm diameter Zn wire (1.5 cm submerged) Reference electrode: Ag/AgCl

Figure S9. Reduction of SF_6 in the presence of NEt₃ with Zn as sacrificial anode at multiple scan rates. $[NEt_3] = 99$ mM

In the presence of triethylamine, the cathodic current is now dependent on the scan rate. Slower scan rate allows more diffusion time for amine base to either scavenge the reduced $SF₆$ material or promote the solvation of Zn cation. This maintains the internal resistance fairly stable, which enhances the cathodic current compared to faster scan rate.

Scan range: -2.0 V to 0.0 V Scan rates (in chronological order): 1) 50 mV/s 2) 100 mV/s 3) 25 mV/s 4) 75 mV/s 5) 200 mV/s Electrolyte: TBAClO4 (0.15M) / Solvent: THF (12 mL) Working electrode: 1.5 mm diameter Sn wire (0.6 cm submerged) Counter electrode: Pt plate (0.8 cm²) Reference electrode: Ag/AgCl

Figure S10. Reduction of SF_6 in the presence of NEt₃ with Pt as anode at multiple scan rates. [NEt₃] = 99 mM

With Pt as anode, the voltammogram still shows enhanced cathodic current under slower scan rate. The result supports that the amine base scavenges reduced SF_6 material (S_nF_m or F-), which can maintain the internal resistance stable.

d. SF6 + glycoside at multiple scan rate

Glycoside: 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranoside (**1b**) Scan range: -2.0 V to 0.0 V Scan rates (in chronological order): 1) 50 mV/s 2) 100 mV/s 3) 25 mV/s 4) 75 mV/s 5) 200 mV/s Electrolyte: TBAClO4 (0.15M) / Solvent: THF (12 mL)

Working electrode: 1.5 mm diameter Sn wire (0.6 cm submerged)

Counter electrode: 2.0 mm diameter Zn wire (1.5 cm submerged)

Reference electrode: Ag/AgCl

Figure S11. Reduction of SF_6 in the presence of glycoside substrate with Zn as sacrificial anode. $[1b] = 33$ mM.

In the presence of SF_6 and glycoside substrate, the voltammogram is consistently showing the decrease of cathodic current in chronological order of the scanning. This result suggests that deoxofluorination of glycoside is a chemical reaction process which is not catalyzed by electrolysis.

e. SF6 + glycoside +NEt3 at multiple scan rate

Glycoside: 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranoside (**1b**) Scan range: -2.0 V to 0.0 V Scan rates (in chronological order): 1) 50 mV/s 2) 100 mV/s 3) 25 mV/s 4) 75 mV/s 5) 200 mV/s

Electrolyte: TBAClO4 (0.15M) / Solvent: THF (12 mL) Working electrode: 1.5 mm diameter Sn wire (0.6 cm submerged) Counter electrode: 2.0 mm diameter Zn wire (1.5 cm submerged)

Reference electrode: Ag/AgCl

Figure S12. Reduction of SF_6 under standard reaction condition. [1b] = 33 mM, [NEt3] = 99 mM

The CV experiment under standard reaction condition shows the enhancement of cathodic current under slower scan rate again because the amine base is present in the solution.

V. Additional Experiments

a. Electric Current On/Off Experiment

Figure S13. Electric current On/Off experiment

An oven-dried 1 dram vial was charged with 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranoside **1a** (0.1 mmol, 1 equiv.) and tetrabutylammonium perchlorate (0.45 mmol, 0.15 M). The vial containing the mixture was sealed with PTFE/Silicone septum cap and was further dried under high vacuum for 5 minutes. Into the sealed vial, THF (3 mL), triethylamine (0.3 mmol, 3 equiv.), and $α, α$ -trifluorotoluene (5 microL) were added, and the vial was stirred until the mixture became homogeneous. The mixture was sparged with a balloon containing SF6 gas for 30 seconds. After the sparging was done, the valve of the balloon was adjusted to sparge small bubbles and left open until the reaction was stopped. Cathode and anode were connected to DC power supply and the reaction was stirred at high rpm. The reaction progress was monitored by taking an aliquot of sample (0.2 mL) every 15 minutes.

30 minutes after the start of reaction, the electric current was shut off and then a sample was taken out. 15 minutes after, another aliquot of sample was taken out, and then the electric current was applied again.

The result is showing that the reaction proceeds only when the electric current is applied, which means that the active fluorinating agent formed *in situ* is consumed right away and does not accumulate.

b. Control Experiment with ZnF2

An oven-dried 1 dram vial was charged with 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranoside **1a** (0.1 mmol, 1 equiv.), zinc(II) fluoride (0.1 mmol, 1 equiv.), and tetrabutylammonium perchlorate (0.45 mmol, 0.15 M). The vial containing the mixture was sealed with PTFE/Silicone septum cap and was further dried under high vacuum for 5 minutes. Into the sealed vial, THF (3 mL) and triethylamine (0.3 mmol, 3 equiv.) were added. The mixture was stirred until everything, except ZnF2, became homogeneous. Sn cathode and Zn anode were connected to DC power supply and the reaction was stirred at high rpm for 1 hour. The color of the reaction turned from clear to black, and the electric current dropped from 15 mA to 3 mA after the voltage reached the limit of DC power supply (31 V). ¹H and ¹⁹F NMR of crude and worked-up reaction did not show any formation of fluorinated product.

c. PPh3 Quenching Experiment

2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranoside **1a** was subjected to a standard reaction. After the electric current was shut off, an aliquot of crude reaction was taken out, and then PPh₃ solution (0.1 mmol in 0.5 mL of THF) was injected into the reaction vial. The mixture was then stirred for 4 hours, and then the samples were taken out. All samples were subjected to ¹⁹F and ³¹P NMR experiments and were compared with each other and with the NMR spectra from our photocatalytic reaction.²¹ Careful examination of ¹⁹F and ³¹P NMR spectrum of electrocatalytic reaction showed the formation of S=PPh3 and O=PPh3 just like photocatalytic reaction. However, F₂PPh₃ is formed in trace amount unlike photocatalytic reaction, which suggests that elemental fluorine or fluoride anion is scavenged by triethylamine or zinc cations.

²¹Kim, S.; Khomutnyk, Y.; Bannykh, A.; Nagorny, P. *Org. Lett*. **2021**, 23, 1, 190-194

Figure S14. ³¹P NMR spectra of PPh₃ quenching experiment.

Figure S15. ¹⁹F NMR spectra of PPh₃ quenching experiment.

$1H$ NMR (CDCl₃, 700MHz) OBn **BnO BnO BnO** $2a-\beta$ 85888838888 8558883338858888888 (Major) $\frac{1}{2}$ تشفقف ננו шt MМ $\frac{1}{3,90}$ $\frac{1}{3,70}$ $\frac{1}{3,65}$
f1 (ppm) $3,95$ 4,00 ՝ 3,60 $\frac{1}{3.55}$ ່ 3,50 $\overline{}$ \top セット - 번 번 번 Ŧ H^{\dagger} 558 $^{8.00}_{2.00}$ 8 883 88 경 Ξm κi $7,5$ $7,0$ $6,5$ $6,0$ $5,5$ $5,0$ 4.5 $4,0$ $3,5$ $3,0$ $2,5$ $2,0$ $1,5$ $1,0$ $0,$ ΪO. $f1$ (ppm)

2,3,4,6-Tetrakis-*O***-(phenylmethyl)-β-D-galactopyranosyl fluoride (2a-β, Major diastereomer)**

70

¹H NMR (CDCl₃, 700MHz)
BnO \mathcal{L}^{OBn} BnO- BnO $2a-\alpha$ (Minor) $\frac{8833}{1777}$ NEERR 66883888888 रानर 1111111111 -8 g $4, 15$ $\frac{1}{4}$ 10 4.05 $4,00$ $3,95$ $3,90$ $f1$ (ppm) -77.77 44.4 $20.10 -$ יליד אי ቸ \top 8 $rac{888}{365}$ 8.08 852 888 ∸வ்வ щ. $7,5$ $7,0$ $6,5$ $6,0$ $5,5$ $5,0$ $4,5$ $4,0$ $3,5$ $3,0$ $2,5$ $2,0$ $1,5$ $1,0$ $\overline{0}$.

2,3,4,6-Tetrakis-*O***-(phenylmethyl)-α-D-galactopyranosyl fluoride (2a-α, Minor diastereomer)**

 $f1$ (ppm)

4,6-Tetrakis-*O***-(phenylmethyl)-α-D-Mannopyranosyl fluoride (2c-α, Major diastereomer)**

2,3,4-tris-O-(phenylmethyl)-D-xylopyranosyl fluoride (2d, Isolated as α : β = 1 : 2.4 mixture)

$rac{17}{\sqrt{116}}$ ¹H NMR (CDCl₃, 700MHz) Me OBn \overline{BD} 2e- α (Major) $\frac{55}{100}$ $\begin{matrix} 58 \\ 66 \\ 7 \end{matrix}$ 58883
1117
117 ERER
1111 $\frac{8833}{377}$ -5.38 -4.67
 -4.65 $5,60^{27}$
f1 (ppm) 3.95 4.05 $\frac{4,00}{f1(ppm)}$ ቸ ጘሦፇ ተዛ Ť Н H 8 8 န္တမ္မမွ 86 5 ဖွ ∸ஷ் $7,5$ $6,5$ $5,5$ $7,0$ $4,5$ $4,0$ $\overline{3,5}$ $2,5$ $\overline{2,0}$ $1,5$ $1,0$ $6,0$ $5,0$ $3,0$ $f1$ (ppm)

4-Tris-*O***-(phenylmethyl)-α-L-fucopyranosyl fluoride (2e-α, Major diastereomer)**

2,3,4-Tris-O-(phenylmethyl)-β-L-fucopyranosyl fluoride (2e-β, Minor diastereomer)

2,3,5-Tris-O-(phenylmethyl)-β-D-ribofuranosyl fluoride (2g- β, Major diastereomer)

2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-β-D-galactopyranosyl fluoride (2h-β, Major diastereomer)

2,3,4,6-Tetrakis-*O***-[(4-methoxyphenyl)methyl]-α-D-galactopyranosyl fluoride (2h-α, Minor diastereomer)**

2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-β-D-glucopyranosyl fluoride (2i, Isolated as α : β = 1 : 1.4 mixture)

$\overset{\frac{55}{500}}{\text{V}}\overset{\frac{35}{40}}{\text{V}}$ ¹H NMR (CDCl₃, 700MHz)
PMBO_> OPMP OPMB Ω PMBO⁻ PMBO $2j-\alpha$ (Major) 讚 $\frac{1}{2}$ $\frac{3558}{111}$ **ក្ដីអ្នកមានខេត្ត** $0.00 - 0.00 - 4.00$ က်တံတံတံတံတံတံတ titud $777 - 122$ 从现 -8 $4,00 - 3,95 - 3,90 - 3,85 - 3,80$ $\sqrt{3.75}$ $\sqrt{3.70}$ 3.65 $\sqrt{4.8}$ ख $4\overline{6}$ "Ni 4.7 4.5 4.4 $f1$ (ppm) $f1$ (ppm) ۆۋتى
بېرىلېلى
بېرىلېلى $\overline{}$ $+$ H $\overline{}$ بلبريليا g 8 $\frac{1}{2}$ 8 g $\frac{5}{4}$ 5 4ń κi m نم نم $\overline{5.5}$ $2,5$ $\overline{2,0}$ $\overline{1,5}$ $7,5$ 7.0 $6,5$ $6,0$ $5,0$ $4₀$ 3.5 $3,0$ $1,0$ 4.5 $f1$ (ppm)

2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-α-D-mannopyranosyl fluoride (2j- α, Major diastereomer)

2,3,4,6-Tetra-*O***-acetyl-α-D-galactopyranosyl fluoride (2k-α, Major diastereomer)**

2,3,4,6-Tetra-*O***-acetyl-β-D-galactopyranosyl fluoride (2k-β, Minor diastereomer)**

2,3,4,6-Tetra-*O***-acetyl-β-D-glucopyranosyl fluoride (2l-β, Major diastereomer)**

2,3,4,6-Tetra-*O***-acetyl-α-D-glucopyranosyl fluoride (2l-α, Minor diastereomer)**

2,3,4,6-Tetra-*O***-acetyl-α-D-mannopyranosyl fluoride (2m-α, Major diastereomer)**

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2,3-Bis-*O***-(phenylmethyl)-4,6-***O***-[(R)-phenylmethylene]-α-D-mannopyranosyl fluoride (2p-α, Major diastereomer)**

2,3-Bis-*O***-(phenylmethyl)-4,6-***O***-[(R)-phenylmethylene]-β-D-mannopyranosyl fluoride (2p-β, Minor diastereomer)**

2,3,6-Tris-*O***-(phenylmethyl)-4-***O***-[2,3,4,6-tetrakis-***O***-(phenylmethyl)-β-D-glucopyranosyl]-β-D-glucopyranosyl fluoride (2q-β, Major diastereomer)**

2,3,6-Tris-*O***-(phenylmethyl)-4-***O***-[2,3,4,6-tetrakis-***O***-(phenylmethyl)-β-D-glucopyranosyl]-α-D-glucopyranosyl fluoride (2q-α, Minor diastereomer)**

