Supporting Information

Electrochemical Synthesis of Glycosyl Fluorides Using Sulfur(VI) Hexafluoride as the Fluorinating Agent

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I. General Information

Methods and Reagents:

Unless otherwise stated, all reagents were purchased from commercial suppliers and used without further purification. All reactions were carried out in flame- or oven-dried glassware with magnetic stirring. Reactions were cooled using external cooling baths: ice water (0 °C). Deionized water was used in the preparation of all aqueous solutions and for all aqueous extractions. Solvents used for extraction and chromatography were ACS or HPLC grade. Purification of reaction mixtures was performed by flash column chromatography (FCC) using SiliCycle SilicaFlash P60 (230-400 mesh). Diastereomeric ratios were determined by ¹H NMR or ¹⁹F NMR analysis.

Instrumentation:

¹H NMR spectra were recorded on Varian vnmrs 700 (700 MHz), Varian vnmrs 600 (600 MHz), or Varian vnmrs 500 (500 MHz) spectrometers and chemical shifts (δ) are reported in parts per million (ppm) with solvent resonance as the internal standard (CDCl₃ at δ 7.26, CD₃CN at δ 1.94). Data are reported as (br = broad, s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet; coupling constant(s) in Hz; integration). Protondecoupled ¹³C NMR spectra were recorded on Varian vnmrs 700 (700 MHz), Varian vnmrs 600 (600 MHz), or Varian vnmrs 500 (500 MHz) spectrometers and chemical shifts (δ) are reported in ppm with solvent resonance as the internal standard (CDCl₃ at δ 77.0, CD₃CN at δ 118.7). ¹⁹F NMR spectra were recorded on Varian vnmrs 600 (600 MHz), Varian vnmrs 500 (500 MHz), or Varian vnmrs 400 (400 MHz) spectrometers and chemical shifts (δ) are reported in parts per million (ppm) and are referenced to CFCl₃ (δ 0.0). High resolution mass spectra (HRMS) were recorded on Micromass AutoSpec Ultima or VG (Micromass) 70-250-S Magnetic sector mass spectrometers in the University of Michigan mass spectrometry laboratory. Infrared (IR) spectra were recorded as thin films on NaCl plates on a Perkin Elmer Spectrum BX FT-IR spectrometer. Absorption peaks were reported in wavenumbers (cm⁻¹).

Information of the DC power supply equipment:

- a. Tekpower TP3005P
 - Output Voltage: 0-30 V
 - Output Current: 0-5 A
 - Setup Resolution: 10 mV / 1 mA
 - Setup Accuracy: ≤ 0.5 % + 20 mV / ≤ 0.5 % + 10 mA

Assignment of Stereochemistry:

Stereochemistry of glycosyl fluoride products (except mannose and rhamnose) are characterized by: 1. One-bond C-F coupling of anomeric carbon in H-coupled ¹³C NMR; α -glycosyl fluoride, ¹J_{C-F} = 224 - 228 Hz and β -glycosyl fluoride, ¹J_{C-F} = 213 - 219 Hz.

2. Three-bond H-H coupling between anomeric hydrogen and C2-hydrogen in ¹H NMR; α -glycosyl fluoride, ³J_{H-H} = < 3 Hz, and β -glycosyl fluoride, ³J_{H-H} = 5.6 – 7.8 Hz.

For D-mannosyl fluorides and L-rhamnosyl fluorides:

1. One-bond C-F coupling of anomeric carbon in H-coupled ^{13}C NMR; α -anomers, $^1J_{C-F}$ = 221 - 224 Hz and β -anomers, $^1J_{C-F}$ = 213-218 Hz

When previously reported NMR spectra were available, the stereochemistry of the product was confirmed again by comparing with them.

II. Optimization

a. Reaction Condition Optimization

Table S1. Solvent Screening

	Bn(-	BnO ^{OBn} S BnO ^M OH 1a 0.1 mmol	F ₆ (1 atm), DIPEA (5 eq.) CIO ₄ (0.2 M), solvent (3 m (+)Pt/(-)Pt, 15 mA	BnO OBn BnO F 2a
Entry	Solvent	Reaction Time	¹⁹ F NMR %Yield	Comments
1	MeCN	2 h 30 min	28%	3 eq. of DIPEA instead of 5 eq.
2	DCE	1h	32%	-
3	DCM	20 min	16%	Cathode was coated with brown chunks
4	DMF	1h	0%	-
5	THF	45 min	35%	¹ H NMR shows 67% consumption of starting material ¹⁹ F NMR shows fluorinated side products

Table S2. Electrode Screening

	BnO BnO	OBn SF ₆ (1 atm), I TBACIO ₄ (0.2 N	DIPEA (5 eq.) M), THF (3 mL)		
	0	BnÒ [°] OH (+)An/(-)Cat, 1 1a .1 mmol	15 mA, 30 min	BnO [°] F 2a	
Entry	(+)An/(-)Cat	Comment	Current Density	Isolated Yield	BRSM ^a
1	(+)Zn/(-)Pt	Worked up ~1h after the reaction was stopped	56 mA/cm ²	94% (51 mg) α : β = 1 : 1.3	95%
2	(+)Zn/(-)Pt	Worked up right after the reaction was stopped	56 mA/cm ²	90% (48.9 mg)	-
3	(+)Zn/(-)Pt	reaction time = 20min	56 mA/cm ²	80% (43.5 mg)	93.2%
4	(+)Zn/(-)Sn	soldering wire (98% Sn / 2% Ag and Cu)	13 mA/cm ²	84% (45.6 mg) α : β = 1 : 1.5	-
5	(+)Zn/(-)C	-	3.3 mA/cm ²	22% (11.7 mg) α : β = 1 : 1.3	-
6	(+)Mg/(-)Sn	Mg plate (from Amazon) + soldering wire	13 mA/cm ²	66% (36 mg) α : β = 1 : 1.2	-

^aYield based on the recovered starting material.

	BnO OBn BnO BnO OH	SF ₆ (1 atm), DIPEA (X eq.) TBACIO ₄ (0.2 M), THF (3 mL) (+)Zn/(-)Sn, 15 mA, 30 min	BnO OBn BnO F 2a
Entry	0.1 mmol DIPEA (X eg.)	Deviation	Isolated Yield
1	5	-	84% (43.5 mg, α : β = 1 : 1.5)
2	3	-	77% (42 mg, α : β = 1 : 1.4)
3	3	New Zn electrode	77.6% (42.1 mg, α : β = 1 : 1.4)
4	3	0.6 mm soldering wire	81% (43.7 mg, α : β = 1 : 1.6)
5	3	0.6 mm soldering wire, 45 min	86% (46.9 mg, $\alpha : \beta = 1 : 1.5$)
6	3	0.6 mm soldering wire, 1 h	86% (46.7 mg, $\alpha : \beta = 1 : 1.4$)
7	3	Reaction time = 45 min	88% (47.6 mg, α : β = 1 : 1.3)
8	3	Reaction time = 1 h	90% (49 mg, α : β = 1 : 1.3)
9	1	New Zn electrode	55% (30.1 mg, α : β = 1 : 1.2)
10	1	Reaction time = 1 h	73% (39.5 mg, α : β = 1 : 1.3)
11	1	Reaction time = 1.5 h	67% (36.3 mg, α : β = 1 : 1.2) Decomposition
12	0	Reaction time = 16 min (Voltage went above 30V)	57% (31 mg, α : β = 1 : 1.2)
13	0	New Zn electrode 15 mA 15.4V \rightarrow 8 mA 30V	67% (36.1 mg, α : β = 1 : 1.2)

Table S3. DIPEA Loading and Reaction Time Screening



Figure S1. Amine base prevents the electrode passivation.

	В		SF ₆ (1 atm), DIPE electrolyte (X M), T	EA (3 eq.) THF (3 mL)	
	BnC	BnO OH	<mark>(+)</mark> Zn/(-)Sn, 15	mA, 1h BnO BnO	F
		1a 0.1 mmol		2a	
Entry	Electrolyte	Concentration	Monitored V ^a	Comment	Yield (%)
1	TBACIO ₄	0.2 M	15-16 V	-	90% (49 mg) α : β = 1 : 1.3
2	TBACIO ₄	0.15 M	22-23 V	-	93% (50.5 mg) α : β = 1 : 1.3
3	TBACIO ₄	0.1 M	30 V (13-14 mA)	-	74% (40 mg) α : β = 1 : 1.3
4	TBAOAc	0.2 M	30 V (0 mA)	electrolytes not dissolving	-
5	TBAPF ₆	0.2 M	9-10 V	electrolytes extracted in organic layer	75% (40.9 mg) α : β = 1 : 1.4

Table S4. Electrolyte Screening

^aVoltage applied to the electrolytic cell by DC power supply.

Table S5. Amine Base Screening

	Br	BnO 0 0 0 0 0 0 0 0 0 0 0 0 0	SF ₆ (1 atm), Base (X eq.) TBACIO₄ (0.15 M), THF (3 mL) (+)Zn/(-)Sn, 15 mA, 1h	BnO OBn BnO F 2a
Entry	Base	Equiv. (X)	Comment	Yield (%)
1	DIPEA	3	-	93% (50.5 mg), α : β = 1 : 1.3
2	TEA	3	22 V \rightarrow 30V (13 mA)	93% (50.2 mg), α : β = 1 : 1.4
3	Pyridine	3	30 V / 14 mA \rightarrow 30 V / 1-2 mA	Crude NMR shows products and byproduct

Table S6. TEA vs DIPEA

	RO RO RO OH 0.1 mmol	SF ₆ (1 atm), Base (3 eq.) BACIO₄ (0.15 M), THF (3 mL) (+)Zn/(-)Sn, 15 mA, 1h	RO F
Base	R = Bn (1a)	R = Ac (1k)	R = PMB (1h)
TEA	93% (50.2 mg, α : β = 1 : 1.4)	69% (24.3 mg, α : β = 1.6 : 1)	97% (64.1 mg, α : β = 1 : 1.2)
DIPEA	93% (50.5 mg, α : β = 1 : 1.3)	59% (20.8 mg, $\alpha : \beta = 1.4 : 1$)	87% (57.8 mg, α : β = 1 : 1.3)

III. Experimental Procedure

a. Synthesis of Substrates











BnC

ÓН



1m







10



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Ph





Preparation of 4-Methoxybenzyl bromide (S1)

B





4-Methoxybenzyl bromide (S1)

In a dry round-bottom flask, 4-methoxybenzyl alcohol (112 mmol, 2.8 equiv.) was dissolved with Et_2O (40 mL). After cooling the mixture by placing the flask in an ice bath, PBr₃ (40 mmol, 1 equiv.) was added dropwise into the reaction mixture. The reaction was

stirred for 5 hours while it was gradually warmed up to room temperature and quenched by adding aqueous NaCl solution (50 mL). The crude mixture was washed with aqueous NaHCO₃ (saturated, 50 mL) and then aqueous NaCl solution (saturated, 50 mL). The organic layer is concentrated *in vacuo* to yield 4-methoxybenzyl bromide **S1** as colorless liquid. The product was confirmed by ¹H NMR and was subjected to the next synthesis without further purification. ¹H NMR (700 MHz, Chloroform-*d*) δ 7.33 (d, *J* = 8.6 Hz, 2H), 6.87 (d, *J* = 8.6 Hz, 2H), 4.51 (s, 2H), 3.81 (s, 3H).

Preparation of 2,3,4,6-tetrakis-O-(phenylmethyl)-D-galactopyranoside (1a)

Purchased from CarboSynth. Lot#: MT040861801.

OBn BnC BnO BnÒ OH 1a

Preparation of 2,3,4,6-Tetrakis-O-(phenylmethyl)-D-glucopyranoside (1b)

Purchased from CarboSynth. Batch#: MT066911502.

BnC BnO BnO BnÒ 1b



Preparation of 2,3,4,6-tetrakis-O-(phenylmethyl)-D-Mannopyranoside (1c)

1,2,3,4,6-Penta-O-acetyl-D-mannopyranoside (S2)



In an oven dried round-bottom flask, D-mannose (27.8 mmol, 1.0 equiv.) and pyridine (40 mL) were added. The reaction mixture was stirred until the solution became homogeneous, and then the round-bottom flask was placed in an ice-bath. After the

reaction was cooled down, acetic anhydride (166.8 mmol, 6.0 equiv.) and DMAP (1.35 mmol, 0.05 equiv.) were added in that order. The reaction was stirred overnight while it was gradually warmed up to the room temperature. The reaction was quenched by diluting with DCM (200 mL) followed by washing with 1N HCl (200 mL x 2), water (200 mL x 2), and then brine (200 mL). Extracted organic layer was dried with Na₂SO₄ and was concentrated *in vacuo* to yield 1,2,3,4,6-penta-*O*-acetyl-D-mannopyranoside **S2**. The product was subjected to the next synthesis without further purification. ¹H NMR (500 MHz, CDCl₃) δ 6.09 (d, *J* = 1.9 Hz, 1H (α -C₁-H)), 5.35 (d, *J* = 6.2 Hz, 2H), 5.30 (s, 1H), 5.27 – 5.26 (m, 1H), 4.29 (dd, *J* = 12.4, 4.9 Hz, 1H), 4.11 (dd, *J* = 12.4, 2.5 Hz, 1H), 4.06 (q, *J* = 6.9, 6.2 Hz, 1H), 2.18 (s, 3H), 2.17 (s, 3H), 2.10 (s, 3H), 2.06 (s, 3H), 2.01 (s, 3H).

AcO AcO AcO S3

Phenyl 2,3,4,6-tetra-O-acetyl-1-thio-α-D-mannopyranoside (S3)

In an oven dried round-bottom flask, **S2** (27.8 mmol, 1.0 equiv.) and DCM (30 mL) were added. After the addition of thiophenol (41.7 mmol, 1.5 equiv.), the round-bottom flask was sealed with a rubber-septum cap and purged with nitrogen gas. The reaction mixture was

stirred until it became homogeneous, and then BF₃·OEt₂ (83.4 mmol, 3.0 equiv.) was slowly injected into the flask. The reaction was stirred overnight at room temperature and quenched with aqueous NaHCO₃ (saturated, 50 mL). After extracting the mixture with DCM (100mL x 3), the combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane \rightarrow 40% EtOAc in hexane) to afford phenyl 2,3,4,6-tetra-*O*-acetyl-1-thio-D-mannopyranoside **S3**.



Phenyl 1-thio-α-D-mannopyranoside (S4)

A dry round-bottom flask was filled with **S3** (26.5 mmol, 1.0 equiv.) and anhydrous methanol (50 mL). After the reaction mixture became homogeneous, a chip of sodium metal (2.7 mmol, 0.1 equiv.) was added. The reaction was stirred overnight at room temperature,

and then quenched by adding acetic acid until the pH of the reaction became neutral. The crude mixture was concentrated *in vacuo* to yield S4 as yellowish white solid. The product was subjected to the next synthesis without further purification.

Phenyl 2,3,4,6-tetrakis-O-(phenylmethyl)-1-thio-D-mannopyranoside (S5)



An oven dried and nitrogen filled round-bottom flask was charged with **S4** (11 mmol, 1.0 equiv.) and DMF (30 mL). After cooling down the reaction in an ice bath, NaH (88 mmol, 8.0 equiv.) was added portion wise. Once H_2 gas stops generating, PMBBr (88

mmol, 8.0 equiv.) was added followed by TBAI (0.55 mmol, 0.05 equiv.). The reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (100 mL x 3), and the organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane \rightarrow 30% EtOAc in hexane) to afford phenyl 2,3,4,6-tetrakis-*O*-(phenylmethyl)-1-thio-D-mannopyranoside **S5** as product

2,3,4,6-Tetrakis-O-(phenylmethyl)-D-mannopyranoside (1c)



In a clean round-bottom flask, **S5** (10.9 mmol, 1.0 equiv.) was dissolved with acetone/water mixture (9:1, 55.5 mL). After cooling down the mixture in an ice bath,

NBS (22.9 mmol, 2.1 equiv.) was added. The reaction was stirred overnight while it was gradually warmed up to the room temperature. Water (50 mL) was added to quench the reaction followed by the extraction with EtOAc (120 mL x 3). The combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 40% EtOAc in hexane \rightarrow 60% EtOAc in hexane) to afford 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-mannopyranoside **1c** as yellowish syrup.



Preparation of 2,3,4-tris-O-(phenylmethyl)-D-xylopyranoside (1d)



AcO

1,2,3,4-Tetra-O-acetyl-D-xylopyranoside (S6)

^hOAc In an oven dried round bottom flask, D-xylopyranose (13 mmol, 1.0 equiv.) and anhydrous pyridine (40 mL) were added. After stirring the mixture until it became

1d

homogeneous, the round bottom flask was placed in an ice bath and was cooled down to 0°C. Into the cooled down reaction, acetic anhydride (65 mmol, 5.0 equiv.) and DMAP (0.65 mmol, 0.05 equiv.) were added in that order. The reaction was slowly warmed up to room temperature with stirring until starting materials are gone. The mixture was then diluted with DCM (150 mL) and washed with 1N HCl solution (2 x 120 mL). The organic layer was washed with water (2 x 120 mL) and then saturated brine (2 x 120 mL). The extracted organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The product 1,2,3,4-Tetra-*O*-acetyl-D-xylopyranose **S6** is subjected to the next synthesis without further purification.

Phenyl 2,3,4-tri-O-acetyl-1-thio-D-xylopyranoside (S7)

 AcO_{AcO} SPh S7 In an oven dried round-bottom flask, S6 (13 mmol, 1.0 equiv.) and DCM (15 mL) were added. After the addition of thiophenol (19.5 mmol, 1.5 equiv.), the round-bottom flask was sealed with a rubber-septum cap and purged with nitrogen gas. The reaction mixture was stirred until it became homogeneous, and then BF₃·OEt₂ (33 mmol, 3.0 equiv.) was slowly injected into the flask. The reaction was stirred overnight at room temperature and quenched with saturated NaHCO₃ solution (50 mL). After extracting the mixture with DCM (3 x 70mL), the combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane) to afford phenyl 2,3,4-tri-O-acetyl-1-thio-D-xylopyranose S7 as yellowish oil.



Phenyl 1-thio-D-xylopyranoside (S8)

A dry round-bottom flask was filled with S7 (13 mmol, 1.0 equiv.) and anhydrous methanol (40 mL). After the reaction mixture became homogeneous, a chip of sodium **S**8 metal (1.3 mmol, 0.1 equiv.) was added. The reaction was stirred overnight at room temperature, and then quenched by adding acetic acid until the pH of the reaction became neutral. The crude mixture was concentrated in vacuo to yield yellowish white solid. The product phenyl 1-thio-D-xylopyranose S8 was subjected to the next synthesis without further purification.

Phenyl 2,3,4-tris-O-(phenylmethyl)-1-thio-D-xylopyranoside (S9)

BnO An oven dried and nitrogen filled round-bottom flask was charged with S8 (13 mmol, BnÒ 1.0 equiv.) and DMF (60 mL). After cooling down the reaction in an ice bath, NaH (78 59 mmol, 6.0 equiv.) was added portion wise. Once H₂ gas stops generating, BnBr (78 mmol, 6.0 equiv.) was added followed by TBAI (0.65 mmol, 0.05 equiv.). The reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 100 mL), and the organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane) to afford phenyl 2,3,4-tris-O-(phenylmethyl)-1-thio-D-xylopyranoside **S9** as yellowish oil.



BnO

2,3,4-tris-O-(phenylmethyl)-D-xylopyranoside (1d)

In a clean round-bottom flask, S9 (13 mmol, 1 equiv.) was dissolved with acetone/water mixture (9:1, 50 mL). After cooling down the mixture in an ice bath, NBS (26 mmol, 2

equiv.) was added. The reaction was stirred overnight while it was gradually warmed up to the room temperature. Water (50 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 120 mL). The combined organic layer was dried with Na₂SO₄ and concentrated in vacuo. The concentrated crude mixture was subjected to FCC (SiO₂, 10% EtOAc in DCM) to afford 2,3,4-tris-O-(phenylmethyl)-D-xylopyranose 1d as white solid, which is confirmed by comparing with literature NMR spectra.¹

¹Zhang, J.; Fu, J.; Si, W.; Wang, X.; Wang, Z.; Tang, J. Carbohydrate Research. 2011, 346, 2290-2293

Preparation of 2,3,4-tris-O-(phenylmethyl)-L-fucopyranoside (1e)



1,2,3,4-Tetra-O-acetyl-L-fucopyranoside (S10)



In an oven dried round bottom flask, L-fucose (3.05 mmol, 1.0 equiv.) and anhydrous pyridine (10 mL) were added. After stirring the mixture until it became homogeneous, the round bottom flask was placed in an ice bath and was cooled down to 0°C. Into the cooled

down reaction, acetic anhydride (18.3 mmol, 6.0 equiv.) and DMAP (0.15 mmol, 0.05 equiv.) were added in that order. The reaction was slowly warmed up to room temperature with stirring until starting materials are gone. The mixture was then diluted with DCM (100 mL) and washed with 1N HCl solution (2 x 80 mL). The organic layer was washed with water (2 x 80 mL) and then saturated brine (2 x 80 mL). The extracted organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The product 1,2,3,4-tetra-*O*-acetyl-L-fucopyranoside **S10** is subjected to the next synthesis without further purification.

Phenyl 2,3,4-tri-O-acetyl-1-thio-L-fucopyranoside (S11)



In an oven dried round bottom flask, **S8** (3.0 mmol, 1.0 equiv.) and anhydrous DCM (5 mL) were added. After stirring until the mixture became homogeneous, PhSH (4.5 mmol, 1.5 equiv.) and BF₃·OEt₂ (9.0 mmol, 3.0 equiv.) were added. Then, the reaction mixture was

stirred overnight and quenched with saturated NaHCO₃ solution (20 mL). The aqueous layer was extracted with DCM (3 x 35 mL) and dried with Na₂SO₄. The extracted organic layer was concentrated in vacuo and subjected to FCC (SiO₂, 30% EtOAc in hexane \rightarrow 50% EtOAc in hexane) to afford phenyl 2,3,4-tri-*O*-acetyl-1-thio-L-fucopyranoside **S11**.



Phenyl 1-thio-L-fucopyranoside (S12)

In an oven dried round bottom flask, **S9** (2.9 mmol, 1 equiv.) and MeOH (20 mL) were added. After stirring until the mixture became homogeneous, Na metal (0.29 mmol, 0.1 equiv.) was added and stirred at room temperature until all starting materials were

consumed. Once all starting materials are gone, the reaction was neutralized with acetic acid and was concentrated *in vacuo* to afford phenyl 1-thio-L-fucopyranoside **S10** as yellowish syrup. The product was subjected to next reaction without further purification.

Phenyl 2,3,4-tri-O-(phenylmethyl)-1-thio-L-fucopyranoside (S13)



An oven dried and nitrogen filled round-bottom flask was charged with **S10** (3.0 mmol, 1.0 equiv.) and DMF (23 mL). After cooling down the reaction in an ice bath, NaH (18 mmol, 6.0 equiv.) was added portion wise. Once H_2 gas stops generating, BnBr (18 mmol,

6.0 equiv.) was added followed by TBAI (0.15 mmol, 0.05 equiv.). The reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 100 mL), and the organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane) to afford phenyl 2,3,4-tris-*O*-(phenylmethyl)-1-thio-D-fucopyranoside **S11** as yellowish oil.

Me OBn OBn BnO 1e

2,3,4-Tris-O-(phenylmethyl)-L-fucopyranoside (1e)

In a clean round-bottom flask, **S11** (2.6 mmol, 1 equiv.) was dissolved with acetone/water mixture (9:1, 25 mL). After cooling down the mixture in an ice bath, NBS (5.2 mmol, 2 equiv.) was added. The reaction was stirred overnight while it was gradually warmed up to

the room temperature. Water (20 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 100 mL). The combined organic layer was dried with Na_2SO_4 and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, 30% EtOAc in hexane) to afford **1e** as white solid, which is confirmed by comparing with literature NMR spectra.²

²Nishi, Y.; Tanimoto, T. Biosci. Biotechnol. Biochem. 2009, 73 (3), 562–569

SPh Me BnNH Me AcO ÓAc Pyridine, 0° ÓAc ÓAc ÓAc S14 S15 NaH, BnBr SPh SPh NaOMe TBAI Me Me MeOH, r.t DMF, 0°→r.t ÓВп ċ⊢ ÓВп S16 S17 OH NBS Acetone/H₂O (9 0°→r ÒBn 1f

Preparation of 2,3,4-tri-O-(phenylmethyl)-α-L-rhamnopyranoside (1f)





An oven dried round bottom flask was charged with L-rhamnose (3.0 mmol, 1.0 equiv.) and anhydrous pyridine (10 mL). After stirring the mixture until it became homogeneous, the round bottom flask was placed in an ice bath and was cooled down to 0°C. Into the

cooled down mixture, acetic anhydride (18 mmol, 6.0 equiv.) and DMAP (0.15 mmol, 0.05 equiv.) were added in that order. The reaction was slowly warmed up to room temperature with stirring until all starting materials were gone. The mixture was then diluted with DCM (100 mL) and washed with 1N HCl solution (2 x 80 mL). The organic layer was washed with water (2 x 80 mL) and then saturated brine (2 x 80 mL). The extracted organic layer was dried with Na₂SO₄ and concentrated *in vacuo* to afford 1,2,3,4-tetra-*O*-acetyl-L-rhamnopyranoside **S14** as yellowish syrup. The product was then subjected to the next synthesis without further purification.

Phenyl 2,3,4-tri-O-acetyl-1-thio-L-rhamnopyranoside (S15)



In an oven dried round bottom flask, **S14** (3.0 mmol, 1.0 equiv.) and anhydrous DCM (5 mL) were added. After stirring until the mixture became homogeneous, PhSH (4.5 mmol, 1.5 equiv.) and BF₃·OEt₂ (9.0 mmol, 3.0 equiv.) were added. The reaction mixture was

stirred overnight and quenched with saturated NaHCO₃ solution (20 mL). The aqueous layer was extracted with DCM (3 x 35 mL) and dried with Na₂SO₄. The extracted organic layer was concentrated *in vacuo* and subjected to FCC (SiO₂, 30% EtOAc in hexane) to afford phenyl 2,3,4-tri-*O*-acetyl-1-thio-L-rhamnopyranoside **S15** as white solid.



Phenyl 1-thio-L-rhamnopyranoside (S16)

In an oven dried round bottom flask, **S15** (3.0 mmol, 1 equiv.) and MeOH (20 mL) were added. After stirring until the mixture became homogeneous, a chip of sodium metal (0.3 mmol, 0.1 equiv.) was added and stirred at room temperature until all starting materials

were consumed. Once all starting materials are gone, the reaction was neutralized with acetic acid and was concentrated *in vacuo* to afford phenyl 1-thio-L-rhamnopyranoside **S16** as white syrup. The product was then subjected to next reaction without further purification.

Phenyl 2,3,4-tri-O-(phenylmethyl)-1-thio-L-rhamnopyranoside (S17)



An oven dried and nitrogen filled round-bottom flask was charged with **S16** (3.0 mmol, 1.0 equiv.) and DMF (20 mL). After cooling down the reaction in an ice bath, NaH (18 mmol, 6.0 equiv.) was added portion wise. Once H_2 gas stops generating, BnBr (18 mmol,

6.0 equiv.) was added followed by TBAI (0.15 mmol, 0.05 equiv.). The reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 60 mL), and the organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 5% EtOAc in hexane) to afford phenyl 2,3,4-tri-*O*-(phenylmethyl)-1-thio-L-rhamnopyranoside **S17** as yellowish oil.

2,3,4-tri-*O*-(phenylmethyl)-α-L-rhamnopyranoside (1f)



In a clean round-bottom flask, **S7** (2.5 mmol, 1 equiv.) was dissolved with acetone/water mixture (9:1, 22.2 mL). After cooling down the mixture in an ice bath, NBS (5.0 mmol, 2.0 equiv.) was added. The reaction was stirred overnight while it was gradually warmed up to

the room temperature. Water (20 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 70 mL). The combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, 30% EtOAc in hexane) to afford 2,3,4-tri-O-(phenylmethyl)- α -L-rhamnopyranoside **1f** as white solid, which is confirmed by comparing with literature NMR spectra.³

Preparation of 2,3,5-Tris-O-(phenylmethyl)-β-D-ribofuranose (1g)

Purchased from Combi-Blocks. Batch#: A98634.

BnC BnC

³Zhang, J.; Fu, J.; Si, W.; Wang, X.; Wang, Z.; Tang, J. Carbohydrate Research. 2011, 346, 2290-2293



Preparation of 2,3,4,6-tetrakis-O-[(4-methoxyphenyl)methyl]-D-galactopyranose (1h)



1,2,3,4,6-Penta-O-acetyl-α-D-galactopyranoside (S18)

In an oven dried round-bottom flask, D-galactose (5.55 mmol, 1.0 equiv.) and pyridine (20 mL) were added. The reaction mixture was stirred until the solution became homogeneous, and then the round-bottom flask was placed in an ice-bath. After the reaction was cooled

down, acetic anhydride (33.3 mmol, 6.0 equiv.) and DMAP (0.278 mmol, 0.05 equiv.) were added in that order. The reaction was stirred overnight while it was gradually warmed up to the room temperature. The reaction was quenched by diluting with DCM (150 mL) followed by washing with 1N HCl (2 x 150 mL), water (2 x 150 mL), and then brine (150 mL). Extracted organic layer was dried with Na₂SO₄ and was concentrated *in vacuo* to yield 1,2,3,4,6-penta-*O*-acetyl- α -D-galactopyranoside **S18** as yellowish syrup. The product was subjected to the next synthesis without further purification. ¹H NMR (500 MHz, Chloroform-*d*) δ 6.38 (d, *J* = 1.8 Hz, 1H), 5.50 (d, *J* = 1.5 Hz, 1H), 5.36 – 5.32 (m, 2H), 4.34 (td, *J* = 6.6, 1.4 Hz, 1H), 4.14 – 4.05 (m, 2H), 2.16 (s, 3H), 2.04 (s, 3H), 2.02 (s, 3H), 2.01 (s, 3H).

Phenyl 2,3,4,6-tetra-O-acetyl-1-thio-D-galactopyranoside (S19)



In an oven dried round-bottom flask, **S18** (11 mmol, 1.0 equiv.) and DCM (15 mL) were added. After the addition of thiophenol (16.5 mmol, 1.5 equiv.), the round-bottom flask was sealed with a rubber-septum cap and purged with nitrogen gas. The reaction mixture

was stirred until it became homogeneous, and then BF₃·OEt₂ (33 mmol, 3.0 equiv.) was slowly injected into the flask. The reaction was stirred overnight at room temperature and quenched with saturated NaHCO₃ solution (50 mL). After extracting the mixture with DCM (3 x 70mL), the combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane) \rightarrow 40% EtOAc in hexane) to afford phenyl 2,3,4,6-tetra-*O*-acetyl-1-thio-D-galactopyranoside **S19** as product. ¹H **NMR** (500 MHz, Chloroform-*d*) δ 7.54 – 7.49 (m, 2H), 7.34 – 7.30 (m, 3H), 5.42 (dd, *J* = 3.4, 1.1 Hz, 1H), 5.24 (t, *J* = 10.0 Hz, 1H), 5.05 (dd, *J* = 9.9, 3.3 Hz, 1H), 4.72 (d, *J* = 10.0 Hz, 1H), 4.19 (dd, *J* = 11.3, 7.0 Hz, 1H), 4.12

(dd, J = 11.4, 6.3 Hz, 2H), 3.96 – 3.92 (m, 1H), 2.12 (s, 3H), 2.10 (s, 3H), 2.04 (s, 3H), 1.98 (s, 3H).



Phenyl 1-thio-D-galactopyranoside (S20)

A dry round-bottom flask was filled with **S19** (11 mmol, 1.0 equiv.) and dry methanol (45 mL). After the reaction mixture became homogeneous, a chip of sodium metal (1.1 mmol, 0.1 equiv.) was added. The reaction was stirred overnight at room temperature, and then

quenched by adding acetic acid until the pH of the reaction became neutral. The crude mixture was concentrated *in vacuo* to afford phenyl 1-thio-D-galactopyranoside **S20** as yellowish white solid. The product was subjected to the subsequent reaction without further purification.

Phenyl 2,3,4,6-tetrakis-O-[(4-methoxyphenyl)methyl]-1-thio-D-galactopyranoside (S21)



An oven dried and nitrogen filled round-bottom flask was charged with **S20** (11 mmol, 1.0 equiv.) and DMF (60 mL). After cooling down the reaction in an ice bath, NaH (88 mmol, 8.0 equiv.) was added portion wise. Once H₂ gas stops generating, PMBBr (88 mmol, 8.0 equiv.) was added followed by TBAI (1.1 mmol, 0.1 equiv.). The reaction

mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 100 mL), and the organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 20% EtOAc in hexane \rightarrow 30% EtOAc in hexane) to afford phenyl 2,3,4,6-tetrakis-O-[(4-methoxyphenyl)methyl]-1-thio-D-galactopyranoside **S21** as yellowish syrup.



2,3,4,6-tetrakis-O-[(4-methoxyphenyl)methyl]-D-galactopyranoside (1h)

In a clean round-bottom flask, **S21** (10 mmol, 1.0 equiv.) was dissolved with acetone/water mixture (9:1, 55.5 mL). After cooling down the mixture in an ice bath, NBS (21 mmol, 2.1 equiv.) was added. The reaction was stirred overnight while it was

gradually warmed up to the room temperature. Water (50 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 120 mL). The combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO2, 30% EtOAc in hexane \rightarrow 40% EtOAc in hexane) to afford 2,3,4,6-tetrakis-*O*-[(4-methoxyphenyl)methyl]-D-galactopyranoside **1h** as white solid product, which is confirmed by comparing with literature NMR spectra.⁴

⁴Whalen, L. J.; Halcomb, R. L. Org. Lett. 2004, 6, 19, 3221-3224



Preparation of 2,3,4,6-tetrakis-O-[(4-methoxyphenyl)methyl]-α-D-glucopyranoside (1i)

Phenyl 2,3,4,6-tetra-O-acetyl-1-thio-D-glucopyranoside (S22)



In an oven dried round-bottom flask, 1,2,3,4,6-penta-*O*-acetyl-D-glucopyranoside (5.0 mmol, 1.0 equiv.) and DCM (5 mL) were added. After the addition of thiophenol (7.5 mmol, 1.5 equiv.), the round-bottom flask was sealed with a rubber-septum cap and

purged with nitrogen gas. The reaction mixture was stirred until it became homogeneous, and then BF₃·OEt₂ (15 mmol, 3.0 equiv.) was slowly injected into the flask. The reaction was stirred overnight at room temperature and quenched with saturated NaHCO₃ solution (10 mL). After extracting the mixture with DCM (3 x 50mL), the combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 30% EtOAc in hexane) to afford phenyl 2,3,4,6-tetra-*O*-acetyl-1-thio-D-glucopyranoside **S22** as white solid.



Phenyl 1-thio-D-glucopyranoside (S23)

A dry round-bottom flask was filled with **S22** (5.0 mmol, 1.0 equiv.) and dry methanol (30 mL). After the reaction mixture became homogeneous, a chip of sodium metal (0.5 mmol, 0.1 equiv.) was added. The reaction was stirred overnight at room temperature,

and then quenched by adding acetic acid until the pH of the reaction became neutral. The crude mixture was concentrated *in vacuo* to yield **S23** as yellowish white solid. The product was subjected to the subsequent reaction without further purification.

Phenyl 2,3,4,6-tetrakis-O-[(4-methoxyphenyl)methyl]-1-thio-D-glucopyranoside (S24)



An oven dried and nitrogen filled round-bottom flask was charged with **S23** (5.0 mmol, 1.0 equiv.) and DMF (30 mL). After cooling down the reaction in an ice bath, NaH (25 mmol, 5.0 equiv.) was added portion wise. Once H₂ gas stops generating, PMBBr (25 mmol, 5.0 equiv.) was added followed by TBAI (0.5 mmol, 0.1 equiv.). The

reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 100 mL), and the organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 40% EtOAc in hexane) to afford phenyl 2,3,4,6-tetrakis-*O*-[(4-methoxyphenyl)methyl]-1-thio-D-glucopyranoside **S24** as yellowish white syrup.

2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-α-D-glucopyranoside (1i)



In a clean round-bottom flask, **S24** (5.0 mmol, 1.0 equiv.) was dissolved with acetone/water mixture (9:1, 33.3 mL). After cooling down the mixture in an ice bath, NBS (10 mmol, 2.0 equiv.) was added. The reaction was stirred overnight while it was

gradually warmed up to the room temperature. Water (35 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 100 mL). The combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 40% EtOAc in hexane) to afford 2,3,4,6-tetrakis-O-[(4-methoxyphenyl)methyl]- α -D-glucopyranoside **1i** as white solid, which is confirmed by comparing with literature NMR spectra.⁵

⁵Lucchetti, N.; Gilmour, R. Chem. Eur. J. 2018, 24,16266 –16270

Preparation of 2,3,4,6-tetrakis-O-[(4-methoxyphenyl)methyl]-α-D-mannopyranoside (1j)



Phenyl 2,3,4,6-tetrakis-O-[(4-methoxyphenyl)methyl]-1-thio-α-D-mannopyranoside (S25)



An oven dried and nitrogen filled round-bottom flask was charged with S4 (9.5 mmol, 1.0 equiv.) and DMF (50 mL). After cooling down the reaction in an ice bath, NaH (52 mmol, 5.5 equiv.) was added portion wise. Once H₂ gas stops generating, PMBBr (47.5 mmol, 5.0 equiv.) was added followed by TBAI (0.95 mmol, 0.1 equiv.). The reaction

mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 100 mL), and the organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 40% EtOAc in hexane) to afford phenyl 2,3,4,6-tetrakis-*O*-[(4-methoxyphenyl)methyl]-1-thio- α -D-mannopyranoside **S25**.

2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-α-D-mannopyranoside (1j)



In a clean round-bottom flask, **S25** (9.5 mmol, 1.0 equiv.) was dissolved with acetone/water mixture (9:1, 55.5 mL). After cooling down the mixture in an ice bath, NBS (20 mmol, 2.1 equiv.) was added. The reaction was stirred overnight while it was gradually

warmed up to the room temperature. Water (50 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 120 mL). The combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 50% EtOAc in hexane \rightarrow 70% EtOAc in hexane) to afford 2,3,4,6-Tetrakis-*O*-[(4-methoxyphenyl)methyl]- α -D-mannopyranoside **1m** as white solid. ¹**H NMR** (700 MHz, CDCl₃) δ 7.27 (d, *J* = 10.2 Hz, 6H), 7.06 (dd, *J* = 8.7, 2.6 Hz, 2H), 6.90 – 6.78 (m, 8H), 5.21 (d, *J* = 1.9 Hz, 1H), 4.77 (d, *J* = 10.4 Hz, 1H), 4.66 (d, *J* = 20.2 Hz, 2H), 4.55 (d, *J* = 3.8 Hz, 2H), 4.52 (d, *J* = 11.9 Hz, 1H), 4.48 (d, *J* = 11.9 Hz, 1H), 4.39 (d, *J* = 10.5 Hz, 1H), 3.96 (ddd, *J* = 8.9, 6.3, 2.0 Hz, 1H), 3.90 (dd, *J* = 9.5, 3.0 Hz, 1H), 3.81 (s, 3H), 3.80 (s, 3H), 3.79 (s, 3H), 3.79 (s, 3H), 3.76 (t, *J* = 2.5 Hz, 1H), 3.68 – 3.66 (m, 1H), 3.65 (d, *J* = 1.9 Hz, 1H), 3.62 (dd, *J* = 10.5, 6.4 Hz, 1H); ¹³**C NMR** (176 MHz, CDCl₃) δ 159.1, 159.1, 159.1, 159.1, 130.7, 130.6, 130.4, 130.1, 129.8, 129.7, 129.6, 129.6, 129.6, 129.5, 129.2, 129.2, 113.7, 113.7, 113.7, 113.6, 92.7, 79.4, 74.9, 74.6, 74.3, 72.8, 72.2, 71.8, 71.4, 69.1, 55.2, 55.2.

Preparation of 2,3,4,6-tetra-O-acetyl-α-D-galactopyranose (1k)







In an oven dried round-bottom flask, **S18** (5.55 mmol, 1.0 equiv.) and THF (20 mL) were added. The mixture was stirred until the solution became homogeneous, and then benzylamine (8.33 mmol, 1.5 equiv.) was added. The reaction was stirred at room

temperate for 20 hours and quenched by evaporating the solvent under reduced pressure. The crude mixture was diluted with DCM (150 mL) and washed with 1N HCl (100 mL x 2), saturated NaHCO₃ solution (100 mL), and then water (100 mL). The organic layer was dried with Na₂SO₄ and was concentrated *in vacuo* to give ambercolored oil. This oil was subjected to FCC (SiO₂, 20% EtOAc in hexane \rightarrow 60% EtOAc in hexane) to afford 2,3,4,6-tetra-*O*-acetyl-D-galactopyranoside **1k** as yellowish solid, which was confirmed by comparing with reported NMR spectra.⁶

⁶Cai, T. B.; Lu, D.; Tang, X.; Zhang, Y.; Landerholm, M.; Wang, P. G. J. Org. Chem. 2005, 70, 9, 3518-3524

Preparation of 2,3,4,6-tetra-O-acetyl-D-glucopyranoside (11)



2,3,4,6-Tetra-O-acetyl-D-glucopyranoside (11)



In an oven dried round-bottom flask, 1,2,3,4,6-penta-*O*-acetyl- α -D-glucopyranoside (5.12 mmol, 1.0 equiv.) and THF (20 mL) were added. The mixture was stirred until the solution became homogeneous, and then benzylamine (7.69 mmol, 1.5 equiv.) was added.

The reaction was stirred at room temperate for 24 hours and quenched by evaporating the solvent under reduced pressure. The crude mixture was diluted with DCM (150 mL) and washed with 1N HCl (2 x 100 mL), saturated NaHCO₃ solution (100 mL), and then water (100 mL). The organic layer was dried with Na₂SO₄ and was concentrated *in vacuo* to give amber-colored oil. This oil was subjected to FCC (SiO₂, 60% EtOAc in hexane) to afford 2,3,4,6-tetra-*O*-acetyl-D-glucopyranoside **11** as yellowish solid, which is confirmed by comparing with literature NMR spectra.⁷

⁷Ikeda, K.; Morimoto, T.; Kakiuchi, K. J. Org. Chem. 2010, 75, 18, 6279-6282

Preparation of 2,3,4,6-tetra-O-acetyl-D-mannopyranose (1m)



2,3,4,6-Tetra-O-acetyl-a-D-mannopyranoside (1m)



In an oven dried round-bottom flask, **S2** (2.17 g, 5.55 mmol) and THF (20 mL) were added. The mixture was stirred until the solution became homogeneous, and then benzylamine (0.91 mL, 8.33 mmol) was added. The reaction was stirred at room temperate for 20 hours

and quenched by evaporating the solvent under reduced pressure. The crude mixture was diluted with DCM (150 mL) and washed with HCl (1.0 M, 100 mL x 2), NaHCO₃ (saturated, 100 mL), and then water (100 mL). The organic layer was dried with Na₂SO₄ and was concentrated *in vacuo* to give amber-colored oil. This oil was subjected to FCC (SiO₂, 20% EtOAc in hexane \rightarrow 60% EtOAc in hexane) to afford 2,3,4,6-tetra-O-acetyl-D-mannopyranoside **1m** as yellowish solid, which was confirmed by comparing with reported NMR spectra.⁸ **1H NMR** (700 MHz, Chloroform-*d*) δ 5.42 (dd, *J* = 10.1, 3.4 Hz, 1H), 5.30 (t, *J* = 9.8 Hz, 1H), 5.28 – 5.24 (m, 2H), 4.27 – 4.22 (m, 2H), 4.12 (q, *J* = 7.4 Hz, 2H), 3.19 (br, 1H), 2.16 (s, 3H), 2.11 (s, 3H), 2.04 (s, 3H), 2.00 (s, 3H).

⁸Carcabal, P.; Hunig, I.; Gamblin, D. P.; Liu, B.; Jockusch, R. A.; Kroemer, R. T.; Snoek, L. C.; Fairbanks, A. J.; Davis, B. G.; Simons, J. P. *J. Am. Chem. Soc.* **2006**, 128, 6, 1976-1981

Preparation of 2,3,4-tri-O-acetyl-α-L-rhamnopyranoside (1n)



2,3,4-Tri-O-acetyl-α-L-rhamnopyranoside (1n)



In an oven dried round-bottom flask, **S14** (3.0 mmol, 1.0 equiv.) and THF (15 mL) were added. The mixture was stirred until the solution became homogeneous, and then benzylamine (4.5 mmol, 1.5 equiv.) was added. The reaction was stirred at room temperate

for 20 hours and quenched by evaporating the solvent under reduced pressure. The crude mixture was diluted with DCM (70 mL) and washed with 1N HCl (2 x 50 mL), saturated NaHCO₃ solution (50 mL), and then water (50 mL). The organic layer was dried with Na₂SO₄ and was concentrated *in vacuo* to give amber-colored oil. This oil was subjected to FCC (SiO₂, 20% EtOAc in hexane \rightarrow 50% EtOAc in hexane) to afford 2,3,4-tri-*O*-acetyl- α -L-rhamnopyranoside **1n** as yellowish solid, which is confirmed by comparing with literature NMR spectra.⁹

⁹Donahue, M. G.; Johnston, J. N. Bioorg. Med. Chem. Lett. 2006, 16, 5602–5604

Preparation of 2,3,4-tri-O-benzoyl-L-rhamnopyranoside (10)



1,2,3,4-Tetra-O-benzoyl-L-rhamnopyranoside (S26)



An oven dried round bottom flask was charged with L-rhamnose (5.5 mmol, 1.0 equiv.) and anhydrous pyridine (12 mL). After stirring the mixture until it became homogeneous, the round bottom flask was placed in an ice bath and was cooled down

to 0°C. Into the cooled down mixture, benzoyl chloride (33 mmol, 6.0 equiv.) was added, and then the reaction was slowly warmed up to room temperature with stirring until all starting materials were gone. The mixture was then diluted with DCM (100 mL) and washed with 1N HCl solution (2 x 80 mL). The organic layer was washed with water (2 x 80 mL) and then saturated brine (2 x 80 mL). The extracted organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The worked-up mixture was then subjected to FCC (SiO₂, 20% EtOAc in hexane) to afford 1,2,3,4-tetra-*O*-benzoyl-L-rhamnopyranoside **S26** as white solid.

2,3,4-tri-O-benzoyl-L-rhamnopyranoside (10)



An oven dried round bottom flask was charged with 1,2,3,4-tetra-*O*-benzoyl-L-rhamnopyranoside **S26** (3.7 mmol, 1.0 equiv.) and THF:MeOH solution (7:3). After stirring the mixture until it became homogeneous, the round bottom flask was placed in

an ice bath and was cooled down to 0°C. Into the cooled down mixture, ammonia gas was bubbled for 1 minute, and then the reaction was stirred for 10 hours while slowly warmed up to room temperature. The mixture was then concentrated *in vacuo*, and then was subjected to FCC (SiO₂, 30% EtOAc in hexane) to afford 2,3,4-tri-*O*-benzoyl-L-rhamnopyranoside **10** as white solid.

Preparation of 2,3-bis-O-(phenylmethyl)-4,6-O-[(R)-phenylmethylene]-α-D-mannopyranoside (1p)





Phenyl 4,6-O-[(R)-phenylmethylene]-1-thio-α-D-manopyranoside (S27)

An oven dried and nitrogen filled round-bottom flask was charged with **S4** (4.5 mmol, 1.0 equiv.) and anhydrous acetonitrile (26 mL). PhCH(OMe)₂ (6.75 mmol, 1.5 equiv.) and p-TsOH (0.27 mmol, 0.06 equiv.) were added and the mixture was refluxed for 3

hours. Once all starting materials are consumed, the reaction was quenched with Et_3N (3 mL), diluted with EtOAc (50 mL), and then washed with brine. Collected organic layer was dried with NaSO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 5% EtOAc in hexane) to afford phenyl 4,6-*O*-[(R)-phenylmethylene]-1-thio- α -D-manopyranoside **S18** as white solid.



Phenyl 2,3-bis-*O*-(phenylmethyl)-4,6-*O*-[(R)-phenylmethylene]-1-thio-α-Dmannopyranoside (S28)

An oven dried and nitrogen filled round-bottom flask was charged with **S27** (3.0 mmol, 1.0 equiv.) and DMF (20 mL). After cooling down the reaction in an ice bath, NaH (18

mmol, 6.0 equiv.) was added portion wise. Once H_2 gas stops generating, BnBr (12 mmol, 4 equiv.) was added followed by TBAI (0.15 mmol, 0.05 equiv.). The reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 60 mL), and the organic layer was dried with Na₂SO₄ and concentrated *in vacuo*.



2,3-Bis-O-(phenylmethyl)-4,6-O-[(R)-phenylmethylene]-α-D-mannopyranoside (1p)

In a clean round-bottom flask, **S19** (3.0 mmol, 1.0 equiv.) was dissolved with acetone/water mixture (9:1, 22.2 mL). After cooling down the mixture in an ice bath, NBS (6.3 mmol, 2.1 equiv.) was added. The reaction was stirred overnight while it was

gradually warmed up to the room temperature. Water (20 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 60 mL). The combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 40% EtOAc in hexane) to afford 2,3-bis-*O*-(phenylmethyl)-4,6-*O*-[(R)-phenylmethylene]- α -D-mannopyranoside **1p** as white solid, which is confirmed by comparing with literature NMR spectra.¹⁰

¹⁰Codée, J. D. C.; Hossain, L. H.; Seeberger, P. H. Org. Lett. 2005, 7, 15, 3251-3254

Preparation of 2,3,6-tris-*O*-(phenylmethyl)-4-*O*-[2,3,4,6-tetrakis-*O*-(phenylmethyl)-β-D-glucopyranosyl]-D-glucopyranoside (1q)



Phenyl 4-O-(2,3,4,6-tetra-O-acetyl-β-D-glucopyranosyl)-1-thio-D-glucopyranoside 2,3,6-triacetate (829)



In an oven dried round-bottom flask, α -D-cellobiose octaacetate (3.0 mmol, 1.0 equiv.) and DCM (10 mL) were added. After the addition of thiophenol (4.5 mmol, 1.5 equiv.), the round-bottom flask was sealed with a rubber-septum cap and purged with nitrogen gas. The reaction mixture was stirred until it became homogeneous, and then BF₃·OEt₂ (9.0 mmol, 3.0 equiv.) was slowly injected into the flask. The reaction was stirred overnight at room temperature and quenched

with saturated NaHCO₃ solution (10 mL). After extracting the mixture with DCM (3 x 70mL), the combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, 50% EtOAc in hexane) to afford phenyl 4-*O*-(2,3,4,6-tetra-*O*-acetyl-β-D-glucopyranosyl)-1-thio-D-glucopyranoside 2,3,6-triacetate **S29** as fluffy white solid.

Phenyl 4-O-β-D-glucopyranosyl-1-thio-D-glucopyranoside (S30)



A dry round-bottom flask was filled with **S29** (3.0 mmol, 1.0 equiv.) and dry methanol (20 mL). After the reaction mixture became homogeneous, a chip of sodium metal (0.3 mmol, 0.1 equiv.) was added. The reaction was stirred overnight at room temperature, and then quenched by adding acetic acid until the pH of the reaction became neutral. The crude mixture was concentrated *in vacuo* to yield **S30**

as yellowish white solid. The product was subjected to the subsequent reaction without further purification.

Phenyl 2,3,6-tris-*O*-(phenylmethyl)-4-*O*-[2,3,4,6-tetrakis-*O*-(phenylmethyl)-β-D-glucopyranosyl]-1-thio-D-glucopyranoside (S31)



An oven dried and nitrogen filled round-bottom flask was charged with **S30** (3.0 mmol, 1.0 equiv.) and DMF (30 mL). After cooling down the reaction in an ice bath, NaH (48 mmol, 16 equiv.) was added portion wise. Once H₂ gas stops generating, BnBr (42 mmol, 14 equiv.) was added followed by TBAI (0.15 mmol, 0.05 equiv.). The reaction mixture was stirred at room temperature overnight and quenched by adding water. The mixture was extracted with EtOAc (3 x 80 mL), and the organic layer was dried with

Na₂SO₄ and concentrated *in vacuo*. The concentrated crude mixture was subjected to FCC (SiO₂, hexane \rightarrow 20% EtOAc in hexane) to afford phenyl 2,3,6-tris-*O*-(phenylmethyl)-4-*O*-[2,3,4,6-tetrakis-*O*-(phenylmethyl)- β -D-glucopyranosyl]-1-thio-D-glucopyranoside **S31** as yellowish white solid.

2,3,6-Tris-O-(phenylmethyl)-4-O-[2,3,4,6-tetrakis-O-(phenylmethyl)-β-D-glucopyranosyl]-D-

glucopyranoside (1q)



In a clean round-bottom flask, **S31** (2.0 mmol, 1.0 equiv.) was dissolved with acetone/water mixture (9:1, 22 mL). After cooling down the mixture in an ice bath, NBS (4.2 mmol, 2.1 equiv.) was added. The reaction was stirred overnight while it was gradually warmed up to the room temperature. Water (20 mL) was added to quench the reaction followed by the extraction with EtOAc (3 x 70 mL). The combined organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The

concentrated crude mixture was subjected to FCC (SiO2, 30% EtOAc in hexane) to afford 2,3,6-tris-O-(phenylmethyl)-4-O-[2,3,4,6-tetrakis-O-(phenylmethyl)- β -D-glucopyranosyl]-D-glucopyranoside 1q as white solid, which is confirmed by comparing with literature NMR spectra.¹¹

¹¹Liu, B.; Mechelen, J.; Berg, R.; Nieuwendijk, A.; Aerts, J.; Marel, G.; Codée, J. Overkleeft, H. S. *Eur. J. Org. Chem.* 2019, 118–129

b. List of components for the experimental setup



Figure S2. Components of experimental setup.

A) Mudder soldering tin wire (1.5 mm diameter) – Amazon Content parameter: Lead-free, Sn 99% / Ag 0.3% / Cu 0.7%

Flux 2% (washed away by sonicating in acetone)

B) Zinc wire (2.0 mm diameter) - Beantown Chemical

C) 5 cm zinc and tin wire

- D) 1 dram vial Sigma and PTFE/Silicon seal with a cap Fisher Scientific
- E) Hamilton MTB valve Thomas Scientific

F) Balloon for SF₆

c. Synthesis of glycosyl fluorides

General Procedure

An oven-dried 1 dram vial was charged with a glycosyl substrate (0.1 mmol, 1 equiv.) and tetrabutylammonium perchlorate (0.45 mmol, 0.15 M). The vial containing the mixture was sealed with PTFE/Silicone septum cap and was further dried under high vacuum for 5 minutes. Into the sealed vial, THF (3 mL) and triethylamine (0.3 mmol, 3 equiv.) were added, and the vial was stirred until the mixture became homogeneous. After plugging in the electrodes (2.3 cm submerged in the solution), the mixture was sparged with a balloon containing SF₆ gas for 30 seconds. Once the sparging was done, the valve of the balloon was adjusted to sparge small bubbles and left open until the reaction was stopped.

The reaction was stirred at high rpm while the electric current was applied to cathode and anode for 1 hour. The reaction mixture was then diluted with 30 mL of ethyl acetate, and then washed with 1N HCl (25 mL x 3), saturated NaHCO₃ (25 mL), water (25 mL), and brine (25 mL). The extracted organic layer was dried with Na₂SO₄ and concentrated *in vacuo*. The worked-up reaction mixture was then subjected to FCC (SiO₂) for purification.



Figure S3. Experimental setup for substrate screening

Cleaning Procedure for the Electrodes

Before each use, the surface of electrodes is polished with 1000 grit sandpaper and sonicated in acetone. After wiping the surface with a paper towel, the electrodes are dried and kept under high vacuum.







2c



2d

2a

Me



OBn | OBn BnO 2e











2j



2k



AcO-



2m



2n



20



2p



2q



Synthesis of 2,3,4,6-tetrakis-O-(phenylmethyl)-D-galactopyranosyl fluoride (2a)

The product was synthesized employing the general procedure. After a work-up, the reaction mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 20% EtOAc in hexane) to afford **2a** in average of 94% yield as yellowish cloudy syrup (R_{f- α} = 0.36 (20% EtOAc/Hex), R_{f- β} = 0.30 (20% EtOAc/Hex)).

	1 st Run	2 nd Run
Isolated Vield	93% (50.5 mg)	94% (51.3 mg)
Isolated Fleid	$\alpha:\beta=1:1.6$	$\alpha:\beta=1:1.6$

2,3,4,6-Tetrakis-O-(phenylmethyl)-α-D-galactopyranosyl fluoride (minor product).

BnO OBn BnO BnO F

OBn

BnC

BnO

¹**H NMR** (700 MHz, CDCl₃) δ 7.39 – 7.26 (m, 20H), 5.59 (dd, J = 53.7, 2.7 Hz, 1H), 4.94 (d, J = 11.3 Hz, 1H), 4.83 (dd, J = 23.1, 11.7 Hz, 2H), 4.74 (dd, J = 20.8, 11.8 Hz, 2H), 4.57 (d, J = 11.3 Hz, 1H), 4.48 (d, J = 11.8 Hz, 1H), 4.42 (d, J = 11.8 Hz, 1H), 4.13 – 4.09 (m, 1H), 4.08

- 4.01 (m, 2H), 3.95 (dd, J = 10.1, 2.7 Hz, 1H), 3.55 (d, J = 5.9 Hz, 2H); ¹³C NMR (176 MHz, CDCl₃) δ 138.4, 138.3, 138.0, 137.7, 128.4, 128.4, 128.3, 128.1, 128.0, 127.9, 127.8, 127.8, 127.7, 127.6, 127.5, 106.2 (d, J =226.1 Hz, α-C₁), 78.4, 75.7 (d, J = 23.8 Hz), 74.8, 74.3, 73.7, 73.5, 73.1, 71.7 (d, J = 2.8 Hz), 68.2; ¹⁹F NMR (564 MHz, CDCl₃) δ -150.13 (dd, J = 53.7, 25.3 Hz); **IR** (thin film, cm⁻¹): 3030, 2870, 1496, 1453, 1363, 1302, 1208, 1099, 1053, 1027, 910, 820, 732, 695; **HRMS (ESI+)** (m/z): [M+Na]⁺ calcd for C₃₄H₃₅FNaO₅ 565.2392, found 565.2353; **[α]** $p^{23} = +26.2$ (c = 0.4, CH₂Cl₂)

2,3,4,6-Tetrakis-O-(phenylmethyl)-β-D-galactopyranosyl fluoride (major product).

¹**H** NMR (700 MHz, CDCl₃) δ 7.39 – 7.27 (m, 20H), 5.19 (dd, J = 53.1, 7.0 Hz, 1H), 4.95 (d, J = 11.5 Hz, 1H), 4.86 (d, J = 11.0 Hz, 1H), 4.80 – 4.71 (m, 3H), 4.61 (d, J = 11.5 Hz,

2a. 1H), 4.50 (d, J = 11.8 Hz, 1H), 4.43 (d, J = 11.8 Hz, 1H), 3.99 – 3.91 (m, 2H), 3.69 – 3.61 (m, 3H), 3.56 (ddd, J = 9.6, 2.9, 0.9 Hz, 1H); ¹³C NMR (176 MHz, CDCl₃) δ 138.3, 138.1, 138.0, 137.6, 128.4, 128.4, 128.3, 128.2, 128.1, 128.1, 127.9, 127.9, 127.7, 127.7, 127.6, 127.5, 110.2 (d, J = 215.4 Hz, β-C₁), 81.0 (d, J = 11.3 Hz), 79.1 (d, J = 20.8 Hz), 75.0, 75.0, 74.6, 73.6, 73.6, 73.1, 73.0, 68.3; ¹⁹F NMR (564 MHz, CDCl₃) δ -138.91 (dd, J = 54.0, 12.8 Hz); **IR** (thin film, cm⁻¹): 3030, 2870, 1496, 1453, 1363, 1302, 1208, 1099, 1053, 1027, 910, 820, 732, 695; **HRMS (ESI+)** (m/z): [M+Na]⁺ calcd for C₃₄H₃₅FNaO₅ 565.2392, found 565.2353; **[α]**_D²³ = +29.8 (c = 1.1, CH₂Cl₂)



Synthesis of 2,3,4,6-tetrakis-O-(phenylmethyl)-D-glucopyranosyl fluoride (2b)

The product was synthesized employing the general procedure. After a work-up, the reaction mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 20% EtOAc in hexane) to afford **2b** as yellowish cloudy syrup in average of 98% yield (R_f = 0.38 (20% EtOAc/Hex)). The NMR spectrum of the product is in a good agreement with our previously published NMR data.¹²

	1 st Run	2 nd Run
Isolated Vield	96% (52.3 mg)	99% (54 mg)
Isolated Held	$\alpha: \beta = 1: 1.4$	$\alpha:\beta=1:1.5$

2,3,4,6-Tetrakis-O-(phenylmethyl)-D-glucopyranosyl fluoride



(Isolated as α : β = 1 : 1.4) ¹**H** NMR ¹H NMR (600 MHz, Chloroform-*d*) δ 7.42 – 7.28 (m, 41.4H), 7.21 – 7.14 (m, 4.8H), 5.59 (dd, *J* = 53.2, 2.6 Hz, 1H, C₁-H_{α}), 5.29 (dd, *J* = 52.8, 6.8 Hz, 1.4H, C₁-H_{β}), 4.99 (d, *J* = 10.9 Hz, 1H), 4.93 (d, *J* = 11.0 Hz, 1.4H), 4.91 – 4.80

(m, 7.3H), 4.74 (d, J = 11.3 Hz, 2.5H), 4.64 (dd, J = 16.1, 12.1 Hz, 2.5H), 4.60 – 4.54 (m, 3.4H), 4.54 – 4.48 (m, 1.6H), 4.02 (t, J = 9.4 Hz, 1H), 3.98 (dt, J = 10.2, 2.6 Hz, 1H), 3.81 – 3.73 (m, 6.3H), 3.73 – 3.67 (m, 2.5H), 3.66 – 3.57 (m, 3.9H); ¹⁹F NMR (564 MHz, CDCl₃) δ -138.06 (dd, J = 52.9, 11.9 Hz, C₁-F_{β}), -149.49 (dd, J = 53.1, 25.6 Hz, C₁-F_{α}); **IR** (thin film, cm⁻¹): 3030, 2867, 1496, 1453, 1360, 1308, 1208, 1155, 1088, 1062, 909, 821, 733, 695; **HRMS (ESI+**) (m/z): [M+Na]⁺ calcd for C₃₄H₃₅FNaO₅ 565.2392, found 565.2353

¹²Kim, S.; Khomutnyk, Y.; Bannykh, A.; Nagorny, P. Org. Lett. 2021, 23, 1, 190-194


Synthesis of 2,3,4,6-tetraki-O-(phenylmethyl)-D-mannopyranosyl fluoride (2c)

The product was synthesized employing the general procedure. After a work-up, the reaction mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 30% EtOAc in hexane) to afford **2c** as cloudy syrup in average of 95% yield (R_{f-a} = 0.57 (30% EtOAc/Hex), R_{f-β} = 0.5 (30% EtOAc/Hex)). Stereochemistry of the product was confirmed by comparing with previously reported NMR spectra.¹³

	1 st Run	2 nd Run
Isolated Vield	92% (50.1 mg)	97% (52.6 mg)
Isolated Held	$\alpha:\beta=23:1$	$\alpha:\beta=20:1$

2,3,4,6-Tetrakis-O-(phenylmethyl)-α-D-Mannopyranosyl fluoride (major product)



¹**H NMR** (700 MHz, CDCl₃) δ 7.38 – 7.27 (m, 18H), 7.20 – 7.17 (m, 2H), 5.61 (d, J = 50.8 Hz, 1H), 4.89 (d, J = 10.8 Hz, 1H), 4.81 (d, J = 12.3 Hz, 1H), 4.72 – 4.64 (m, 4H), 4.55 (d, J = 12.4 Hz, 2H), 4.09 (t, J = 9.5 Hz, 1H), 3.93 (ddd, J = 10.1, 4.7, 1.8 Hz, 1H), 3.89 (d, J =

8.9 Hz, 2H), 3.79 (dd, J = 11.0, 4.7 Hz, 1H), 3.73 (dd, J = 11.1, 1.9 Hz, 1H); ¹³C NMR (176 MHz, CDCl₃) δ 138.2, 138.1, 138.1, 137.8, 128.4, 128.3, 128.3, 127.9, 127.9, 127.8, 127.7, 127.6, 106.4 (d, J = 222.0 Hz), 79.2, 75.1, 74.2, 74.2, 74.0, 73.6, 73.4, 73.4, 73.2, 72.6, 68.6; ¹⁹F NMR (564 MHz, CDCl₃) δ -138.01 (d, J = 50.5 Hz); IR (thin film, cm⁻¹): 3030, 2863, 1496, 1453, 1363, 1311, 1183, 1094, 1026, 952, 801, 733, 695; HRMS (ESI+) (m/z): [M+Na]⁺ calcd for C₃₄H₃₅FNaO₅ 565.2366, found 565.2358; [α] $p^{24} = +12.0$ (c = 1.1, CH₂Cl₂)

¹³Zeng, J.; Vedachalam, S.; Xiang, S.; Liu, X-W. Org. Lett. 2011, 13, 1, 42-45



Synthesis of 2,3,4-tris-O-(phenylmethyl)-D-xylopyranosyl fluoride (2d)

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane $\rightarrow 20\%$ EtOAc in hexane) to afford 2d as yellowish white solid in average of 98% yield (R_f = 0.45(20% EtOAc/Hex)). The NMR spectrum of the product is in a good agreement with our previously published NMR data.14

	1 st Run	2 nd Run
Isolated Vield	96% (40.4 mg)	100% (42.1 mg)
Isolated Held	$\alpha:\beta=1:2.4$	$\alpha:\beta=1:2.7$



2,3,4-Tris-O-(phenylmethyl)-D-xylopyranosyl fluoride

(Isolated as α : $\beta = 1$: 2.4 mixture) ¹**H NMR** ¹**H NMR** (600 MHz, CDCl₃) δ 7.43 – 7.28 (m, 46.8H), 5.48 (dd, J = 53.1, 2.7 Hz, 1H, C₁-H_a), 5.35 (dd, J = 53.9, 5.6 Hz, 2.4H, C₁-2d H_{β} , 4.94 (s, 2H), 4.86 – 4.76 (m, 8.9H), 4.75 – 4.60 (m, 9.5H), 4.10 – 4.03 (m, 2.4H), 3.94 (t, J = 9.2 Hz, 1H), 3.80 (dd, J = 11.1, 5.6 Hz, 1H), 3.75 – 3.61 (m, 6.7H), 3.61 – 3.44 (m, 5.6H); ¹⁹F NMR (564 MHz, CDCl₃) δ -134.41 (dd, J = 54.0, 12.6 Hz, 2.4F, C₁-F_{β}), -151.04 (dd, J = 53.1, 25.7 Hz, 1F, C₁-F_{α}); **IR** (thin film, cm⁻¹): 3030, 2869, 1496, 1453, 1361, 1257, 1208, 1162, 1088. 1027, 942, 733, 695; HRMS (ESI+) (m/z): [M+Na]⁺ calcd for $C_{26}H_{27}FNaO_{4}\ 445.1792,\ found\ 445.1780.$

¹⁴Kim, S.; Khomutnyk, Y.; Bannykh, A.; Nagorny, P. Org. Lett. 2021, 23, 1, 190-194



Synthesis of 2,3,4-tris-O-(phenylmethyl)-L-fucopyranosyl fluoride (2e)

Me

Me

l ÓBn BnO

2e-

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane \rightarrow 30% EtOAc in hexane) to afford **2e** as cloudy syrup in average of 91% yield (R_{f-a} = 0.6 (30% EtOAc/Hex)), R_{f-β} = 0.53 (30% EtOAc/Hex)).

	1 st Run	2 nd Run
Isolated Viold	90% (39.4 mg)	91% (39.8 mg)
	$\alpha:\beta=2.5:1$	$\alpha:\beta=2:1$

2,3,4-Tris-O-(phenylmethyl)-α-L-fucopyranosyl fluoride (major product).

¹H NMR (700 MHz, Chloroform-*d*) δ 7.41 – 7.27 (m, 15H), 5.57 (dd, *J* = 54.0, 2.8 Hz, 1H),
4.99 (d, *J* = 11.5 Hz, 1H), 4.85 (dd, *J* = 11.8, 3.9 Hz, 2H), 4.75 (dd, *J* = 19.1, 11.8 Hz, 2H),
4.66 (d, *J* = 11.5 Hz, 1H), 4.08 – 4.00 (m, 2H), 3.94 (dd, *J* = 10.1, 2.7 Hz, 1H), 3.70 (dd, *J*

= 2.8, 1.2 Hz, 1H), 1.17 (d, J = 6.5 Hz, 3H); ¹³C NMR (176 MHz, CDCl₃) δ 138.5, 138.3, 138.1, 128.5, 128.4, 128.4, 128.3, 128.0, 127.8, 127.7, 127.7, 127.5, 106.4 (d, J = 224.5 Hz, α-C₁), 78.8, 75.6 (d, J = 23.9 Hz), 74.9, 73.7, 73.2, 69.1 (d, J = 3.2 Hz), 16.5; ¹⁹F NMR (564 MHz, CDCl₃) δ -149.68 (dd, J = 54.0, 25.3 Hz); IR (thin film, cm⁻¹): 3030, 2872, 1496, 1453, 1361, 1307, 1208, 1100, 1056, 1027, 912, 801, 732, 695; HRMS (ESI+) (m/z): [M+Na]⁺ calcd for C₂₇H₂₉FNaO₄ 459.1892, found 459.1917; **[α]** p^{24} = -41.9 (c=0.5, CH₂Cl₂)

2,3,4-tris-O-(phenylmethyl)-β-L-fucopyranosyl fluoride (minor product).

^{OBn} ^{BnO} ^{2e-} ^{OBn} ¹H NMR (700 MHz, CDCl₃) δ 7.41 – 7.28 (m, 15H), 5.16 (dd, J = 53.3, 7.1 Hz, 1H), 5.00 (d, J = 11.6 Hz, 1H), 4.87 (d, J = 10.9 Hz, 2H), 4.80 (d, J = 11.7 Hz, 1H), 4.79 (d, J = 10.8Hz, 1H), 4.74 (d, J = 11.8 Hz, 1H), 4.69 (d, J = 11.6 Hz, 1H), 3.94 (ddd, J = 13.3, 9.7, 7.0 Hz, 1H), 3.60 (m, 2H), 3.55 (dd, J = 9.8, 2.9 Hz, 1H), 1.26 (d, J = 6.4 Hz, 3H); ¹³C NMR (176 MHz, CDCl₃) δ 138.2, 138.2, 138.2, 128.4, 128.3, 128.3, 128.2, 128.1, 127.7, 127.7, 127.6, 110.3 (d, J = 213.5 Hz, β -C₁), 81.4 (d, J = 11.7 Hz), 79.0 (d, J = 20.6 Hz), 75.8, 74.9, 74.7, 73.3, 70.8 (d, J = 4.9 Hz), 16.6; ¹⁹F NMR (564 MHz, CDCl₃) δ -139.13 (dd, J = 53.4, 13.2 Hz); IR (thin film, cm⁻¹): 3030, 2872, 1496, 1453, 1361, 1307, 1208, 1100, 1056, 1027, 912, 801, 732, 695; HRMS (ESI+) (m/z): [M+Na]⁺ calcd for C₂₇H₂₉FNaO₄ 459.1892, found 459.1917; [**a**]**p**²⁴ = -54.2 (c=0.7, CH₂Cl₂)



Synthesis of 2,3,4-tris-O-(phenylmethyl)-L-rhamnopyranosyl fluoride (2f)

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane $\rightarrow 20\%$ EtOAc in hexane) to afford **2f** as yellowish syrup in average of 93% yield (R_{f- α} = 0.57 (20% EtOAc/Hex)), R_{f- β} = 0.43 (20% EtOAc/Hex)).

	1 st Run	2 nd Run
Isolated Vield	93% (40.8 mg)	93% (40.5 mg)
Isolated Held	$\alpha:\beta=20:1$	$\alpha:\beta=18:1$

2,3,4-Tris-O-(phenylmethyl)-α-L-rhamnopyranosyl fluoride (Major product)

Me O BnO OBn OBn

¹**H NMR** (700 MHz, CDCl₃) δ 7.38 – 7.29 (m, 15H), 5.49 (dd, J = 50.6, 2.0 Hz, 1H), 4.95 (d, J = 10.9 Hz, 1H), 4.81 (d, J = 12.2 Hz, 1H), 4.72 – 4.64 (m, 4H), 3.89 – 3.82 (m, 3H), 3.67 (t, J = 9.4 Hz, 1H), 1.36 (d, J = 6.2 Hz, 3H); ¹³**C NMR** (176 MHz, CDCl₃) δ 138.3,

138.2, 137.8, 128.5, 128.4, 128.4, 128.0, 128.0, 127.9, 127.7, 127.7, 127.7, 106.4 (d, J = 221.1 Hz), 79.6, 79.2, 75.4, 73.7 (d, J = 35.2 Hz), 73.4, 72.6, 70.6 (d, J = 2.3 Hz), 17.9; ¹⁹F NMR (471 MHz, CDCl₃) δ -137.07 (d, J = 50.6 Hz); **IR** (thin film, cm⁻¹): 3030, 2922, 1721, 1496, 1453, 1363, 1248, 1207, 1184, 1074, 1027, 911, 841, 734, 695; **HRMS (ESI+)** (m/z): [M+Na]⁺ calcd for C₂₇H₂₉FNaO₄ 459.1948, found 459.1948; **[a]**_D^{23.6} = +7.66 (c = 1.1, CH₂Cl₂)



Synthesis of 2,3,5-Tris-O-(phenylmethyl)-D-ribofuranosyl fluoride (2g)

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane \rightarrow 30% EtOAc in hexane) to afford **2g** as yellowish oil in average of 94% yield (R_{f- α} = 0.31 (20% EtOAc/Hex)), R_{f- β} = 0.43 (20% EtOAc/Hex)).

	1 st Run	2 nd Run
Isolated Vield	90% (37.9 mg)	98% (41.3 mg)
Isolated Held	$\alpha:\beta=1:17$	$\alpha:\beta=1:16$

2,3,5-Tris-O-(phenylmethyl)-β-D-ribofuranosyl fluoride (Major porduct).



¹**H NMR** (700 MHz, CDCl₃) δ 7.38 – 7.27 (m, 15H), 5.69 (d, J = 63.4 Hz, 1H, β-C₁-H), n 4.67 (s, 2H), 4.61 (d, J = 12.1 Hz, 1H), 4.56 (d, J = 13.6 Hz, 2H), 4.48 (d, J = 11.8 Hz, 1H), 4.44 (dtd, J = 8.1, 5.1, 2.7 Hz, 1H), 4.14 (ddd, J = 7.4, 4.5, 2.4 Hz, 1H), 4.00 (t, J = 4.2 Hz,

1H), 3.70 (dd, J = 11.0, 3.4 Hz, 1H), 3.60 (dd, J = 11.0, 5.3 Hz, 1H); ¹³C NMR (176 MHz, CDCl₃) δ 138.1, 137.4, 137.3, 128.5, 128.4, 128.3, 128.1, 128.0, 127.9, 127.6, 127.6, 112.5 (d, J = 224.4 Hz, β -C₁), 82.3 (d, J = 2.6 Hz), 78.9 (d, J = 30.1 Hz), 73.3, 72.8, 72.8, 70.2; ¹⁹F NMR (471 MHz, CDCl₃) δ -115.29 (dt, J = 63.1, 5.5 Hz, β -C₁-F); **IR** (thin film, cm⁻¹): 3030, 2862, 1726, 1496, 1454, 1358, 1308, 1257, 1208, 1096, 1026, 784, 735, 696; **HRMS** (**ESI+**) (m/z): [M+Na]⁺ calcd for C₂₆H₂₇FNaO₄ 445.1792, found 445.1782; **[a]**_D²⁴ = +63.9 (c=0.7, CH₂Cl₂)



Synthesis of 2,3,4,6-tetrakis-O-[(4-methoxyphenyl)methyl]-D-galactopyranosyl fluoride (2h)

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane \rightarrow 30% EtOAc in hexane) to afford **2h** as cloudy oil in average of 96% yield (R_{f-a} = 0.35 (30% EtOAc/Hex)), R_{f-β} = 0.33 (30% EtOAc/Hex)).

	1 st Run	2 nd Run
Isolated Vield	97% (64.1 mg)	96% (63.7 mg)
Isolated Held	$\alpha:\beta=1:1.2$	$\alpha: \beta = 1: 1.1$

2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-α-D-galactopyranosyl fluoride

РМВО ОРМВ

(minor product). ¹H NMR (700 MHz, CDCl₃) δ 7.32 – 7.26 (m, 4H), 7.22 – 7.15 (m, 4H),



2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-β-D-galactopyranosyl fluoride

(major product). ¹**H NMR** (700 MHz, CDCl₃) δ 7.28 (d, *J* = 8.6 Hz, 2H), 7.26 (d, *J* = 6.1 Hz, 2H), 7.20 (dd, *J* = 8.5, 5.5 Hz, 4H), 6.87 – 6.85 (m, 6H), 6.83 (d, *J* = 8.6 Hz, 2H), 5.13 (dd, *J* = 53.2, 7.0 Hz, 1H), 4.83 (d, *J* = 11.2 Hz, 1H), 4.76 (d, *J* = 10.7 Hz, 1H), 4.69 (d, *J*

= 10.6 Hz, 1H), 4.64 (q, J = 11.4 Hz, 2H), 4.52 (d, J = 11.3 Hz, 1H), 4.42 (d, J = 11.4 Hz, 1H), 4.34 (d, J = 11.4 Hz, 1H), 3.91 – 3.85 (m, 1H), 3.82 (d, J = 2.7 Hz, 1H), 3.81 (s, 3H), 3.80 (s, 3H), 3.80 (s, 3H), 3.79 (s, 3H), 3.61 – 3.57 (m, 2H), 3.56 – 3.52 (m, 1H), 3.49 – 3.46 (m, 1H); ¹³C NMR (176 MHz, CDCl₃) δ 159.4, 159.3, 159.2, 159.2, 130.5, 130.3, 130.3, 129.8, 129.8, 129.6, 129.2, 113.8, 113.8, 113.6, 110.3 (d, J = 215.3 Hz, β-Cl), 80.7 (d, J = 11.3 Hz), 78.8 (d, J = 20.7 Hz), 74.7, 74.6, 74.1, 73.7 (d, J = 4.8 Hz), 73.2, 72.7, 72.5, 68.1, 55.3, 55.2; ¹⁹F NMR (471 MHz, CDCl₃) δ -138.96 (dd, J = 53.0, 13.4 Hz); **IR** (thin film, cm⁻¹): 2912, 2835, 1611, 1585, 1512, 1463, 1442, 1422, 1362, 1301, 1244, 1172, 1096, 1031, 817, 756, 710, 664; **HRMS (ESI+)** (m/z): [M+NH₄]⁺ calcd for C₃₈H₄₇FNO₉ 680.3229, found 680.3281; **[α]** \mathbf{p}^{24} = +39.8 (c = 0.19, CH₂Cl₂).



Synthesis of 2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-D-glucopyranosyl fluoride (2i)

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane \rightarrow 30% EtOAc in hexane) to afford **2i** as clear oil in average of 96% yield (R_f = 0.33 (30% EtOAc/Hex)). The NMR spectrum of the product is in a good agreement with our previously published NMR data.¹⁵

	1 st Run	2 nd Run
Isolated Vield	99% (65.4 mg)	94% (62.3 mg)
	$\alpha:\beta=1:1.4$	$\alpha:\beta=1:1.4$



 $2,3,4,6-Tetrak is \textit{-}O-[(4-methoxyphenyl)methyl]-\beta-D-glucopyranosyl fluoride$

(Major product, Isolated as α : $\beta = 1$: 1.4 mixture).

PMBO 2i-¹H NMR (600 MHz, CDCl₃) δ 7.35 – 7.20 (m, 14H), 7.10 – 7.01 (m, 4.6H), 6.93 – 6.77 (m, 18.3H), 5.49 (dd, J = 53.2, 2.6 Hz, 1H, C₁-H_α), 5.23 (dd, J = 52.8, 6.8 Hz, 1.4H, C₁-H_β), 4.88 (d, J = 10.5 Hz, 1H), 4.83 (d, J = 10.7 Hz, 1.4H), 4.81 – 4.70 (m, 7H), 4.65 (dd, J = 11.1, 4.8 Hz, 2.4H), 4.58 (dd, J = 11.8, 3.3 Hz, 2.4H), 4.48 (d, J = 11.8 Hz, 1.4H), 4.45 – 4.37 (m, 3.3H), 3.96 – 3.88 (m, 2.4H), 3.84 – 3.77 (m, 27.6H), 3.73 – 3.59 (m, 8.6H), 3.59 – 3.49 (m, 3.7H); ¹⁹F NMR (564 MHz, CDCl₃) δ – 138.29 (dd, J = 53.0, 12.0 Hz, 1.4F, C₁-F_β), -149.54 (dd, J = 53.3, 25.7 Hz, 1F, C₁-F_α); IR (thin film, cm⁻¹): 2907, 2835, 1611, 1585, 1512, 1463, 1441, 1359, 1302, 1244, 1172, 1083, 1031, 816, 758, 712, 637; HRMS (ESI+) (m/z): [M+Na]⁺ calcd for C₃₈H₄₃FNaO₉ 685.2783, found 685.2772; [**a**]p²⁴ = +20.7 (c = 0.34, CH₂Cl₂).

¹⁵Kim, S.; Khomutnyk, Y.; Bannykh, A.; Nagorny, P. Org. Lett. 2021, 23, 1, 190-194



Synthesis of 2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-D-mannopyranosyl fluoride (2j)

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane $\rightarrow 10\%$ EtOAc in hexane) to afford **2**_j as cloudy oil in average of 96% yield (R_{f- $\alpha}$} = 0.53 (40% EtOAc/Hex), $R_{f-\beta} = 0.45 (40\% \text{ EtOAc/Hex})).$

	1 st Run	2 nd Run
Isolated Vield	97% (64.6 mg)	95% (62.9 mg)
Isolated Held	$\alpha:\beta=25:1$	$\alpha:\beta=23:1$



2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-α-D-mannopyranosyl fluoride

(Major Product) ¹**H NMR** (500 MHz, CDCl₃) δ 7.29 – 7.24 (m, 6H), 7.06 (d, J = 8.6 Hz, 2H), 6.90 – 6.79 (m, 8H), 5.53 (dd, *J* = 50.6, 1.6 Hz, 1H), 4.75 (dd, *J* = 24.8, 11.2 Hz, 2H), 2j-4.64 - 4.53 (m, 4H), 4.44 (dd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 1H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 1H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 1H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 1H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 1H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 2H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 2H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 2H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 2H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 2H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 2H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 2H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 2H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2H), 4.00 - 3.95 (m, 2H), 3.86 (ddd, J = 21.7, 11.1 Hz, 2.1, 11.1 Hz, 2.1 Hz, 2.1 Hz, 2.1 Hz, 2.1 Hz, 2.1 Hz, 210.0, 4.8, 1.8 Hz, 1H), 3.80 (dd, J = 10.6, 4.1 Hz, 14H), 3.71 (dd, J = 11.0, 4.7 Hz, 1H), 3.66 (dd, J = 10.9, 2.0 Hz, 1H); ¹³C NMR (126 MHz, cdcl₃) δ 159.4, 159.2, 159.2, 159.2, 130.4, 130.4, 130.2, 129.9, 129.6, 129.6, 129.3, 113.8, 113.7, 106.5 (d, *J* = 221.9 Hz), 78.8, 74.7, 74.2, 74.2, 73.7, 73.1, 73.0, 72.9, 72.8, 72.2, 68.2, 55.3, 55.2, 55.2; ¹⁹F NMR (471 MHz, CDCl₃) δ -137.88 (d, J = 50.6 Hz); IR (thin film, cm⁻¹): 2908, 2835, 1611, 1585, 1512,

1463, 1442, 1422, 1361, 1301, 1244, 1172, 1090, 1030, 950, 816, 756, 711, 667; **HRMS (ESI+)** (m/z): [M+Na]⁺ calcd for $C_{38}H_{43}FNaO_9$ 685.2783, found 685.2753; [α] p^{24} = +7.45 (c = 0.88, CH₂Cl₂).



Synthesis of 2,3,4,6-Tetra-O-acetyl-D-galactopyranosyl fluoride (2k)

AcO

OAc

AcÒ

2k-

AcO

AcC

AcO

The product was synthesized employing a procedure for acetylated glycosides. After a work-up, the reaction mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 40% EtOAc in hexane) to afford **2k** as yellowish oil in average of 73% yield (R_{f-\alpha} = 0.38 (40% EtOAc/Hex), R_{f-\beta} = 0.30 (40% EtOAc/Hex)).

	1 st Run	2 nd Run
Isolated Viold	73% (25.6 mg)	73% (25.6 mg)
	$\alpha:\beta=1.9:1$	$\alpha:\beta=1.8:1$

2,3,4,6-Tetra-O-acetyl-a-D-galactopyranosyl fluoride (Minor product).

¹**H NMR** (700 MHz, Chloroform-*d*) δ 5.80 (d, *J* = 53.3 Hz, 1H), 5.53 (t, *J* = 2.5 Hz, 1H), 5.37 (dt, *J* = 10.8, 2.8 Hz, 1H), 5.19 (ddt, *J* = 23.8, 10.9, 2.6 Hz, 1H), 4.41 (t, *J* = 6.7 Hz, 1H), 4.18 – 4.09 (m, 2H), 2.16 (s, 3H), 2.12 (s, 3H), 2.06 (s, 3H), 2.01 (s, 3H); ¹³C NMR (176 MHz,

2k- - 4.09 (m, 2H), 2.16 (s, 3H), 2.12 (s, 3H), 2.06 (s, 3H), 2.01 (s, 3H); ¹³C NMR (1/6 MHz, cdcl₃) δ 170.3, 170.2, 170.0, 169.9, 104.3 (d, J = 228.4 Hz, α-C₁), 68.9 (d, J = 3.5 Hz), 67.5, 67.3, 67.0, 61.3, 20.7, 20.6, 20.6; ¹⁹F NMR (564 MHz, CDCl₃) δ -150.78 (dd, J = 53.3, 23.9 Hz); **IR** (thin film, cm⁻¹): 2921, 2851, 1744, 1369, 1212, 1167, 1104, 1042, 951, 899; **HRMS (ESI+)** (m/z): [M+Na]+ calcd for C₁₄H₁₉FNaO₉ 373.0911, found 373.0905; **[a]** $p^{25} = +28.1$ (c = 0.4, CH₂Cl₂)

2,3,4,6-Tetra-O-acetyl-β-D-galactopyranosyl fluoride (Major product).

¹**H NMR** (700 MHz, Chloroform-*d*) δ 5.44 – 5.40 (m, 1H), 5.35 – 5.21 (m, 2H), 5.04 (dt, J = 10.5, 2.8 Hz, 1H), 4.24 – 4.18 (m, 2H), 4.05 (t, J = 6.6 Hz, 1H), 2.17 (s, 3H), 2.10 (s, 3H), 2.06 (s, 3H), 2.01 (s, 3H); ¹³**C NMR** (176 MHz, CDCl₃) δ 170.6, 170.0, 170.0, 169.3,

107.1 (d, J = 218.6 Hz, β -C₁), 71.2 (d, J = 4.7 Hz), 69.9 (d, J = 10.6 Hz), 68.8 (d, J = 24.9 Hz), 66.4, 61.3, 20.7, 20.6, 20.5; ¹⁹F NMR (564 MHz, CDCl₃) δ -141.49 (dd, J = 51.0, 12.9 Hz); **IR** (thin film, cm⁻¹): 2921, 2851, 1744, 1369, 1212, 1167, 1104, 1042, 951, 899; **HRMS (ESI+)** (m/z): [M+Na]+ calcd for C₁₄H₁₉FNaO₉ 373.0911, found 373.0905; **[a]**_D²⁵ = +6.0 (c = 0.4, CH₂Cl₂)



Synthesis of 2,3,4,6-Tetra-O-acetyl-D-glucopyranosyl fluoride (21)

The product was synthesized employing a procedure for acetylated glycosides. After a work-up, the reaction mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 50% EtOAc in hexane) to afford **2l** as colorless oil in average of 76% yield (R_{f- α} = 0.47 (50% EtOAc/Hex), R_{f- β} = 0.43 (50% EtOAc/Hex)).

	1 st Run	2 nd Run
Isolated Yield	76% (26.7 mg)	75% (26.3 mg)
	$\alpha : \beta = 1 : 2.3$	$\alpha : p = 1 : 2.5$

2,3,4,6-Tetra-O-acetyl-α-D-glucopyranosyl fluoride (Minor product)

¹**H NMR** (700 MHz, CDCl₃) δ 5.75 (dd, J = 52.8, 2.8 Hz, 1H), 5.50 (t, J = 9.9 Hz, 1H), 5.16 (t, J = 10.0 Hz, 1H), 4.96 (ddd, J = 24.2, 10.2, 2.8 Hz, 1H), 4.29 (dd, J = 12.5, 4.1 Hz, 1H), 4.19 (ddd, J = 10.3, 4.2, 2.2 Hz, 1H), 4.15 (dd, J = 12.5, 2.2 Hz, 1H), 2.11 (s,

3H), 2.10 (s, 6H), 2.05 (s, 3H), 2.03 (s, 3H); ¹³**C NMR** (176 MHz, CDCl₃) δ 170.5, 170.0, 169.9, 169.4, 103.7 (d, J = 229.5 Hz, α -C₁), 70.2 (d, J = 24.5 Hz), 69.8 (d, J = 4.3 Hz), 69.4, 67.3, 61.2, 20.7, 20.6, 20.5; ¹⁹**F NMR** (564 MHz, CDCl₃) δ -149.76 (dd, J = 52.8, 24.2 Hz, α -C₁-F); **IR** (thin film, cm⁻¹): 2918, 1743, 1434, 1367, 1208, 1160, 1103, 1033, 904, 779; **HRMS (ESI+)** (m/z): [M+Na]⁺ calcd for C₁4H₁₉FNaO₉ 373.0892, found 373.0899

2,3,4,6-Tetra-O-acetyl-β-D-glucopyranosyl fluoride (Major product)



.OAc

AcÒ

21-

AcO⁻ AcC

¹**H** NMR (700 MHz, CDCl₃) δ 5.36 (ddd, J = 52.0, 6.2, 1.8 Hz, 1H), 5.20 (tt, J = 9.4, 4.8 Hz, 2H), 5.12 – 5.08 (m, 1H), 4.29 – 4.24 (m, 1H), 4.24 – 4.20 (m, 1H), 3.90 (ddd, J = 7.5, 4.9, 2.5 Hz, 1H), 2.10 (s, 3H), 2.10 (s, 3H), 2.04 (s, 3H), 2.03 (s, 3H); ¹³C NMR

(176 MHz, CDCl₃) δ 170.6, 170.0, 169.3, 169.1, 106.2 (d, J = 219.6 Hz, β -C₁), 72.0 (d, J = 3.8 Hz), 71.8 (d, J = 8.3 Hz), 71.2 (d, J = 28.8 Hz), 67.4, 61.7, 20.7, 20.6; ¹⁹F NMR (564 MHz, CDCl₃) δ -137.24 (dd, J = 52.0, 10.4 Hz, β -C₁-F), -149.76 (dd, J = 52.8, 24.2 Hz).; **IR** (thin film, cm⁻¹): 2918, 1743, 1434, 1367, 1208, 1160, 1103, 1033, 904, 779; **HRMS (ESI+)** (m/z): [M+Na]⁺ calcd for C₁₄H₁₉FNaO₉ 373.0892, found 373.0899; **[a]** $_{D}^{24} = +13.6$ (c = 0.5, CH₂Cl₂)



Synthesis of 2,3,4,6-Tetra-O-acetyl-D-mannopyranosyl fluoride (2m)

The product was synthesized employing the general procedure. After a work-up, the reaction mixture was subjected to FCC (SiO₂, pure hexane \rightarrow 40% EtOAc in hexane) to afford pure **2m** in average of 81% yield as cloudy oil (Rf = 0.33 (40% EtOAc/Hex)). Stereochemistry of the products was confirmed by comparing with previously reported NMR spectra.¹⁶

	1 st Run	2 nd Run
Isolated Vield	81% (28.4 mg)	80% (28.1 mg)
Isolated Held	$\alpha:\beta=20:1$	$\alpha:\beta=20:1$

2,3,4,6-Tetra-O-acetyl-a-D-mannopyranosyl fluoride (Major product).



¹**H NMR** (700 MHz, Chloroform-*d*) δ 5.58 (d, *J* = 48.6 Hz, 1H), 5.40 (s, 1H), 5.35 (d, *J* = 5.0 Hz, 2H), 4.32 – 4.27 (m, 1H), 4.16 (dd, *J* = 14.7, 4.4 Hz, 2H), 2.17 (s, 3H), 2.11 (s, 3H), 2.06 (s, 3H), 2.01 (s, 3H); ¹³C NMR (176 MHz, CDCl₃) δ 170.5, 169.7, 169.6, 169.5, 104.7 É $(d, J = 223.7 \text{ Hz}, \beta$ -C₁), 70.9 (d, J = 2.7 Hz), 68.2, 67.7 (d, J = 39.5 Hz), 65.0, 61.8, 20.7, 20.6, 20.6; ¹⁹F NMR $(564 \text{ MHz}, \text{CDCl}_3) \delta - 138.38 \text{ (d}, J = 48.4 \text{ Hz}); \text{ IR (thin film, cm}^{-1}): 2960, 1742, 1434, 1368, 1211, 1172, 1087,$

1051, 1019, 974, 916, 795, 685; HRMS (ESI+) (m/z): [M+Na]⁺ calcd for C₁₄H₁₉FNaO₉ 373.0911, found 373.0905; $[\alpha]_D^{24} = +11.2 (c = 0.5, CH_2Cl_2).$

¹⁶ Chambers, R. D.; Sandford, G.; Sparrowhawk, M. E.; Atherton, M. J. J. Chem. Soc., Perkin Trans. 1, 1996, 1941-1944



Synthesis of 2,3,4-tri-O-acetyl-L-rhamnopyranosyl fluoride (2n)

The product was synthesized employing the general procedure. After a work-up, the reaction mixture was purified by FCC (SiO₂, pure hexane \rightarrow 40% EtOAc in hexane) to afford **2n** as colorless oil in average of 79% yield (R_f = 0.37(30% EtOAc/Hex)). Stereochemistry of the products was confirmed by comparing with previously reported NMR spectra.¹⁷

	1 st Run	2 nd Run
Isolated Vield	82% (23.8 mg)	77% (22.5 mg)
Isolated Held	$\alpha: \beta = 22: 1$	$\alpha:\beta=18:1$

2,3,4-Tri-O-acetyl-a-L-rhamnopyranosyl fluoride (Major product)



¹**H NMR** (500 MHz, CDCl₃) δ 5.49 (dd, J = 48.6, 1.9 Hz, 1H), 5.38 (ddd, J = 3.2, 1.9, 0.9 Hz, 1H), 5.29 (ddd, J = 10.2, 3.6, 1.7 Hz, 1H), 5.12 (t, J = 10.0 Hz, 1H), 4.04 (dq, J = 9.9, 6.2 Hz, 1H), 2.16 (s, 3H), 2.06 (s, 3H), 2.00 (s, 3H), 1.27 (d, J = 6.3 Hz, 3H); ¹³C NMR (126

MHz, CDCl₃) δ 169.8, 169.8, 169.7, 104.8 (d, J = 221.9 Hz), 70.0, 68.9 (d, J = 3.0 Hz), 68.2 (d, J = 1.7 Hz), 68.0 (d, J = 40.3 Hz), 20.7, 20.7, 20.6, 17.3; ¹⁹F NMR (564 MHz, CDCl₃) δ -137.49 (d, J = 49.1 Hz); **IR** (thin film, cm⁻¹): 2925, 1748, 1605, 1509, 1457, 1369, 1293, 1242, 1214, 1173, 1038, 974, 929, 893, 830, 813, 793, 778, 684

¹⁷Nishiyama, K.; Esaki, S.; Deguchi, I.; Sugiyama, N.; Kamiya, S. Biosci. Biotechnol. Biochem. 1993, 57, 107–114



Synthesis of 2,3,4-tri-O-benzoyl-L-rhamnopyranosyl fluoride (20)

The product was synthesized employing the general procedure. After a work-up, the reaction mixture was purified by FCC (SiO₂, pure hexane \rightarrow 30% EtOAc in hexane) to afford **20** as cloudy syrup in average of 84% yield (R_{f-a} = 0.52 (30% EtOAc/Hex), $R_{f-\beta} = 0.46$ (30% EtOAc/Hex)).

	1 st Run	2 nd Run
Isolated Yield	86% (41.2 mg)	82% (39.1 mg)
	$\alpha:\beta=20:1$	$\alpha:\beta=21:1$

2,3,4-Tri-O-benzoyl-a-L-rhamnopyranosyl fluoride



¹**H NMR** (700 MHz, CDCl₃) δ 8.10 (dd, J = 8.3, 1.3 Hz, 2H), 8.00 – 7.97 (m, 2H), 7.83 (dd, *J* = 8.4, 1.4 Hz, 2H), 7.63 (tt, *J* = 8.8, 1.3 Hz, 1H), 7.55 – 7.52 (m, 3H), 7.52 – 7.49 (m, 2H), 7.43 (tt, J = 7.4, 1.3 Hz, 1H), 7.42 – 7.38 (m, 2H), 5.87 – 5.81 (m, 2.5H, C₄-H + C₂-H + 1/2 C₁-H), 5.76 (d, J = 1.8 Hz, 0.5H, 1/2 C₁-H), 5.74 (t, J = 9.9 Hz, 1H, C₃-H), 4.39 (dq, J = 9.9, 6.2 Hz, 1H), 1.43 (d, J = 6.2 Hz, 3H); ¹³C NMR (176 MHz, cdcl₃) δ 165.6, 165.4, 165.3, 133.7, 133.5, 133.3, 130.0, 129.8, 129.7, 129.0, 128.9, 128.9, 128.7, 128.5, 128.3, 104.9 (d, *J* = 222.4 Hz, α-C₁-F), 70.7, 69.2 (d, *J* = 2.7 Hz), 69.1 (d, *J* = 1.0 Hz), 68.9 (d, J = 40.6 Hz), 17.6; ¹⁹F NMR (564 MHz, CDCl₃) δ -137.24 (d, J = 48.8 Hz); IR (thin film, cm⁻ ¹): 2985, 1724, 1601, 1584, 1451, 1315, 1276, 1258, 1173, 1091, 1068, 1026, 973, 910, 851, 793, 705, 685; **HRMS** (ESI+) (m/z): $[M-F]^+$ calcd for C₂₇H₂₃O₇⁺ 459.1438, found 459.1427; $[\alpha]_{D}^{23.9} = +166.9$ (c (w/v%) = 4.15, CH₂Cl₂)



Synthesis of 2,3-bis-O-(phenylmethyl)-4,6-O-[(R)-phenylmethylene]-D-mannopyranosyl fluoride (2p) The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane $\rightarrow 20\%$ EtOAc in hexane) to afford 2p as cloudy oil in average of 96% yield (R_{f-a} = 0.44 (20% EtOAc/Hex)). Stereochemistry of the products was determined by comparing with previously reported NMR spectra.¹⁸

	1 st Run	2 nd Run
Isolated Yield	97% (43.9 mg)	94% (42.2 mg)
	lpha:eta=24:1	$\alpha:\beta=28:1$

BnO

BnC

2p

2,3-Bis-*O***-(phenylmethyl)-4,6-***O***-[(R)-phenylmethylene]-***α***-D-mannopyranosyl fluoride** (major product).

¹**H NMR** (700 MHz, CDCl₃) δ 7.51 (d, *J* = 7.1 Hz, 2H), 7.41 – 7.28 (m, 13H), 5.65 (s, 1H), 5.50 (d, *J* = 49.6 Hz, 1H), 4.89 (d, *J* = 12.0 Hz, 2H), 4.74 – 4.68 (m, 2H), 4.33 –

4.27 (m, 2H), 3.99 - 3.92 (m, 3H), 3.87 (td, J = 10.3, 2.0 Hz, 1H); ¹³C NMR (176 MHz, CDCl₃) δ 138.3, 137.6, 137.4, 128.9, 128.5, 128.4, 128.2, 128.1, 128.1, 127.7, 127.6, 126.1, 107.0 (d, J = 223.7 Hz), 101.6, 78.3, 75.5 (d, J = 2.5 Hz), 75.0, 74.8, 74.2, 73.5, 68.3, 66.1 (d, J = 2.4 Hz); ¹⁹F NMR (471 MHz, CDCl₃) δ -137.06 (d, J = 49.7 Hz); IR (thin film, cm⁻¹): 2919, 1454, 1372, 1313, 1214, 1166, 1099, 1054, 1006, 965, 925, 799, 735, 696; HRMS (ESI+) (m/z): [M+Na]⁺ calcd for C₂₇H₂₇FNaO₅ 473.1692, found 473.1729; [α] $p^{23} = -17.4$ (c = 0.6, CH₂Cl₂).

¹⁸Lee, Y. J.; Baek, J. Y.; Lee, B-Y.; Kang, S. S.; Park, H-S.; Jeon, H. B.; Kim, K. S. Carbohydrate Research. **2006**. 341(10). 1708-1716



Synthesis of 2,3,6-Tris-O-(phenylmethyl)-4-O-[2,3,4,6-tetrakis-O-(phenylmethyl)-β-D-glucopyranosyl]-D-glucopyranosyl fluoride (2q)

The product was synthesized employing the general procedure. The product was purified by FCC (SiO₂, pure hexane \rightarrow 30% EtOAc in hexane) to afford **2q** as white solid in average of 84% yield (R_{f-a} = 0.58 (30% EtOAc/Hex)), R_{f-β} = 0.57 (30% EtOAc/Hex)).

	1 st Run	2 nd Run
Isolated Yield	80% (78.1 mg)	88% (85.8 mg)
	$\alpha:\beta=1:3.4$	$\alpha:\beta=1:3.2$

2.8 Hz, 3H), 4.07 - 4.03 (m, 1H), 3.90 - 3.84 (m, 2H), 3.80 (dd, J = 10.1, 2.2 Hz, 1H), 3.73 (d, J = 10.9 Hz, 1H), 3.61 (t, J = 8.8 Hz, 1H), 3.58 - 3.55 (m, 1H), 3.48 (ddd, J = 25.5, 19.6, 10.3 Hz, 4H), 3.39 - 3.35 (m, 1H), 3.33 - 3.29 (m, 1H); ¹³**C** NMR (176 MHz, CDCl₃) δ 139.2, 138.6, 138.5, 138.3, 138.3, 137.9, 137.6, 128.5, 128.4, 128.4, 128.4, 128.3, 128.2, 128.1, 128.0, 128.0, 127.9, 127.8, 127.8, 127.8, 127.8, 127.7, 127.6, 127.6, 127.3, 127.2, 105.8 (d, J = 226.8 Hz, α -C₁), 102.5, 84.9, 82.7, 79.8, 78.2 (d, J = 24.3 Hz), 78.0, 75.6, 75.6, 75.4, 75.2, 75.0, 74.8, 73.8, 73.3, 72.6 (d, J = 4.2 Hz), 69.0, 69.0, 67.1; ¹⁹F NMR (471 MHz, CDCl₃) δ -149.55 (dd, J = 53.4, 25.8 Hz); **IR** (thin film, cm⁻¹): 1497, 1453, 1361, 1072, 1029, 1005, 908, 753, 731, 695, 665; **HRMS (ESI+)** (m/z): [M+NH₄]⁺ calcd for C₆₁H₆₇FNO₁₀ 992.4738, found 992.4733; **[a]**_D^{23.6} = +23.1 (c = 0.23, CH₂Cl₂)

BnO OBn OBn OBn OBn OBn 2q-OBn OBn 2,3,6-Tris-O-(phenylmethyl)-4-O-[2,3,4,6-tetrakis-O-(phenylmethyl)-β-Dglucopyranosyl]-β-D-glucopyranosyl fluoride (major product).

¹**H NMR** (700 MHz, CDCl₃) δ 7.37 (d, J = 5.2 Hz, 2H), 7.34 – 7.22 (m, 31H), 7.19 (d, J = 7.3 Hz, 2H), 5.25 (dd, J = 53.5, 6.5 Hz, 1H, β-C₁-H), 5.05 (d, J = 11.4 Hz, 1H), 4.89 (d, J = 10.9 Hz, 1H), 4.81 (d, J = 10.6 Hz, 2H), 4.75 (dd, J = 17.7, 9.3 Hz, 4H), 4.69 (d, J = 11.2 Hz, 1H), 4.58 (dd, J = 17.0, 11.4 Hz, 2H), 4.50 (d, J = 7.9 Hz, 1H),

4.47 – 4.39 (m, 3H), 4.13 (t, J = 8.9 Hz, 1H), 3.84 (dd, J = 11.1, 3.4 Hz, 1H), 3.72 – 3.48 (m, 8H), 3.39 (t, J = 8.5 Hz, 1H), 3.30 (dd, J = 10.4, 4.0 Hz, 1H); ¹³C NMR (176 MHz, CDCl₃) δ 138.9, 138.5, 138.4, 138.4, 138.2, 137.8, 137.8, 128.4, 128.4, 128.4, 128.3, 128.3, 128.2, 128.1, 128.1, 127.9, 127.8, 127.8, 127.8, 127.8, 127.8, 127.7, 127.6, 127.6, 127.5, 127.4, 127.3, 109.5 (d, J = 215.9 Hz, β-C₁), 102.3, 84.9, 82.7, 81.3 (d, J = 10.3 Hz), 80.5 (d, J = 23.0 Hz), 77.9, 75.8, 75.6, 75.0, 74.9 (d, J = 3.7 Hz), 74.8, 74.7, 74.3, 73.4, 73.3, 68.9, 67.7; ¹⁹F NMR (471 MHz, CDCl₃) δ -135.75 (dd, J = 53.5, 12.3 Hz); **IR** (thin film, cm⁻¹): 1497, 1453, 1361, 1072, 1029, 1005, 908, 753, 731, 695, 665; **HRMS (ESI+)** (m/z): [M+NH₄]⁺ calcd for C₆₁H₆₇FNO₁₀ 992.4738, found 992.4733; **[a]p²³** = +25.9 (c = 1.1, CH₂Cl₂)

d. Summary of α:β ratio for 1a-q and 2a-q

Substrate Initial dr of 1a-q determined by ¹ H NMR		dr of 2a-q determined by ¹⁹ F NMR	
a	alpha only	1:1.6	
b	4.8:1	1:1.5	
c	alpha only	22:1	
d	5.5:1	1:2.6	
e	2.7:1	2.3:1	
f	alpha only	19:1	
g	beta only	1:17	
h	3.4:1	1:1.2	
i	3:1	1:1.4	
j	alpha only	24:1	
k	4.2:1	1.9:1	
l	5:1	1:2.4	
m	alpha only	20:1	
n	alpha only	20:1	
0	6.8:1	21:1	
р	alpha only	26:1	
q	1.9:1 1:3.3		

Table S7. Initial and final α : β ratios of the starting materials 1a-q and products 2a-q

Product	Our method	DAST method		
		α:β of SM	Yield	
2a	94% (α:β = 1:1.6)	α:β = 1.7:1	91% (α:β = 3:5)	
2b	98% (α:β = 1:1.5)	α:β = 1.7:1	89% (α:β = 5:1)	
2c	95% (α:β = 22:1)	α only	91% (α only)	
2d	98% (α:β = 1:2.6)	-	-	
2e	95% (α:β = 2.3:1)	α:β mixture	100% (α:β mixture)	
2f	93% (α:β = 19:1)	-	-	
2g	94% (α:β = 1:17)	α:β mixture	95% (α:β = 1:19 to 1:99)	
2h	96% (α:β = 1:1.2)	α:β mixture	93% (α:β mixture)	
2i	96% (α:β = 1:1.4)	-	-	
2j	96% (α:β = 24:1)	-	-	
2k	73% (α:β = 1.9:1)	-	-	
21	76% (α:β = 1:2.4)	α:β = 1:2	92% (α:β = 3:1)	
2m	81% (α:β = 20:1)	-	-	
2n	79% (α:β = 3:1)	-	-	
20	84% (α:β = 21:1)	-	-	
2р	96% (α:β = 26:1)	-	-	
2q	84% (α:β = 1:3.3)	-	-	

e. Comparison with DAST method

References:

2a, 2b, 2c: Zeng, J.; Vedachalam, S.; Xiang, S.; Liu, X.-W. Org. Lett. 2011, 13, 1, 42-45

2e: Guan, X.; Chaffey, P. K.; Zeng, C.; Greene, E. R.; Chen, L.; Drake, M. R.; Chen, C.; Groobman, A.; Resch, M. G.; Himmel, M. E.; Beckham, G. T.; Tan, Z. *Chem. Sci.* **2015**, 6, 7185

2g: Ginisty, M.; Gravier-Pelletier, C.; Merrer, Y. L. Tetrahedron: Asymmetry. 2006, 17, 142-150

2h: Li, X.; Wu, P.; Cheng, S.; Lv, X. J. Med. Chem. 2021, 55, 6, 2702-2710

2I: Zagrobelny, M.; Olsen, C. E.; Pentzold, S.; Fürstenberg-Hägg, J.; Jørgensen, K.; Bak, S.; Møller, B. L.; Motawia, M. S. *Insect Biochemistry and Molecular Biology.* **2014**, 44, 44-53

f. Large Scale Batch Synthesis

5 Gram Scale Synthesis



Figure S4. Experimental setup adopted from Baran Group's Work¹⁹



Figure S5. (A) Reaction setup (B) Reaction progress monitored by TLC (C) Electrodes after the reaction (D) Isolated product

¹⁹Hu, P.; Peters, B. K.; Malapit, C. A.; Vantourout, J. C.; Wang, P.; Li, J.; Mele, L.; Echeverria, P.-G.; Minteer, S. D.; Baran, P. S. *J. Am. Chem. Soc.* **2020**, 142, 50, 20979-20986

BnO BnO 5 g	SF ₆ (1 atm), TEA (3 eq.) TBACIO₄ (0.3 M) , THF (278 mL) (+)Zn/(-)Sn, 730 mA, 5 h	BnO BnO BnO BnO BnO F 98% (4.92 g, : = 1:2.3)
Applied Current	Surface Area	Current Density
730 mA	58 cm ²	12.6 mA/cm^2

To achieve high current density, the experimental setup was adopted from Baran group's large scale setup (**Fig. SI-4**).¹⁹ Electrodes were prepared by cutting off tin plate (3.2 mm thickness, 5.8 cm x 10 cm x 2) and zinc plate (0.5 mm thickness, 6.0 cm x 10 cm) and 5.0 cm height of the electrodes were submerged in the solution. Due to the voltage limit (31 V) of the DC power supply, the concentration of electrolyte was doubled (0.3 M). Into an oven dried glass jar containing a stir bar, 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranoside **1a** (9.25 mmol, 1.0 equiv.), tetrabutylammonium perchlorate (83.4 mmol, 0.3 M) were dissolved in dry THF (278 mL). After the addition of triethylamine (27.7 mmol, 3.0 equiv.), the reaction was placed inside a water bath (to relieve the heat generated during electrolysis) and stirred until the solution became homogeneous. The reaction was then sparged with SF₆ balloons for 1 minute before the electric current was applied, and then the rest of the reaction. The electric current was applied for 6 hours, and then the reaction mixture was diluted with 300 mL of ethyl acetate. The diluted mixture was then washed with 1N HCl (250 mL x 3), NaHCO₃ (250 mL x 1), water (250 mL x 5), and brine (250 mL x 1). The extracted organic layer was dried with Na₂SO₄ and concentrated *in vacuo*, which was subjected to FCC to afford 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranosyl fluoride **2a** in 98 % yield (4.92 g, α : $\beta = 1 : 2.3$) as yellow syrup.

IV. Cyclic Voltammetry Studies

General information

Cyclic voltammetry studies were performed with the Pine WaveNow^{XV} Potentiostat/Galvanostat. All experiments were measured under slight argon stream. SF₆ was added by sparging SF₆ gas to the solution for 1 minute. Electrodes were polished with 1000-grit sandpaper and cleaned by sonicating in acetone, and they were dried under high vacuum before use. All electrolytes were dried under high vacuum before use. THF was prepared from Innovative Technology PureSolv solvent drying system and dry acetonitrile was purchased from Alfa Aesar. All solvents were degassed with argon before use.

a. Identifying half reduction potential of SF₆

Scan range: -3.0 V to 0.0 V (-3.1 V to -0.5 V is shown on the graph after voltage reference with Fc^+/Fc) Scan rates: 20 mV/s Electrolyte: TBAClO₄ (0.1M) / Solvent: acetonitrile (12 mL) Working electrode: 0.635 mm diameter Pt wire (0.6 cm submerged)

Counter electrode: Pt plate (0.8 cm^2)

Reference electrode: Ag/AgCl (voltage is referenced to ferrocene [0.001 M])



Figure S6. Blank is shown in black line and the reduction of SF₆ is shown in red line. Voltage was corrected by ferrocene (0.001 M)

We first attempted to measure the half reduction potential of SF₆ by adopting the CV experimental set-up from Magnier group.²⁰ The voltammogram shows $E_{red1/2} = -2.44$ V (vs. Fc⁺/Fc).

Half reduction potential of SF₆ reference to SCE: -1.99 V vs. SCE

²⁰Bouvet, S.; Pegot, B.; Sengmany, S.; Gall, E. L.; Leonel, E.; Goncalves, A.-M.; Magnier, E. *Beilstein J. Org. Chem.* **2020**, 16, 2948-2953

b. SF₆ at multiple scan rate

Scan range: -2.0 V to 0.0 V

Scan rates (in chronological order): 1) 50 mV/s 2) 100 mV/s 3) 25 mV/s 4) 75 mV/s 5) 200 mV/s Electrolyte: TBAClO₄ (0.15M) / Solvent: THF (12 mL) Working electrode: 1.5 mm diameter Sn wire (0.6 cm submerged) Counter electrode: 2.0 mm diameter Zn wire (1.5 cm submerged) Reference electrode: Ag/AgCl



Figure S7. Reduction of SF₆ with Zn as sacrificial anode at multiple scan rates.

The voltammogram shows the decrease of cathodic current in the chronological order of the scans. It suggests the increased internal resistance from either a) electrode passivation or b) accumulation of salt formed from solvated Zn cation.

Scan range: -2.0 V to 0.0 V Scan rates (in chronological order): 1) 50 mV/s 2) 100 mV/s 3) 25 mV/s 4) 75 mV/s 5) 200 mV/s Electrolyte: TBAClO₄ (0.15M) / Solvent: THF (12 mL) Working electrode: 1.5 mm diameter Sn wire (0.6 cm submerged) Counter electrode: Pt plate (0.8 cm²) Reference electrode: Ag/AgCl



Figure S8. Reduction of SF₆ with Pt as anode at multiple scan rates.

Pt anode is resistant to galvanic corrosion and does not dissociate Pt cation under electric current. As expected with Pt as anode, the voltammogram shows consistent current throughout the entire CV experiment. This result supports that the decrease of cathodic current in previous CV experiment is caused by Zn sacrificial anode.

c. SF₆ + NEt₃ at multiple scan rate

Scan range: -2.0 V to 0.0 V

Scan rates (in chronological order): 1) 50 mV/s 2) 100 mV/s 3) 25 mV/s 4) 75 mV/s 5) 200 mV/s Electrolyte: TBAClO₄ (0.15M) / Solvent: THF (12 mL) Working electrode: 1.5 mm diameter Sn wire (0.6 cm submerged) Counter electrode: 2.0 mm diameter Zn wire (1.5 cm submerged) Reference electrode: Ag/AgCl



Figure S9. Reduction of SF₆ in the presence of NEt₃ with Zn as sacrificial anode at multiple scan rates. [NEt₃] = 99 mM

In the presence of triethylamine, the cathodic current is now dependent on the scan rate. Slower scan rate allows more diffusion time for amine base to either scavenge the reduced SF_6 material or promote the solvation of Zn cation. This maintains the internal resistance fairly stable, which enhances the cathodic current compared to faster scan rate.

Scan range: -2.0 V to 0.0 V Scan rates (in chronological order): 1) 50 mV/s 2) 100 mV/s 3) 25 mV/s 4) 75 mV/s 5) 200 mV/s Electrolyte: TBAClO₄ (0.15M) / Solvent: THF (12 mL) Working electrode: 1.5 mm diameter Sn wire (0.6 cm submerged) Counter electrode: Pt plate (0.8 cm²) Reference electrode: Ag/AgCl



Figure S10. Reduction of SF₆ in the presence of NEt₃ with Pt as anode at multiple scan rates. $[NEt_3] = 99 \text{ mM}$

With Pt as anode, the voltammogram still shows enhanced cathodic current under slower scan rate. The result supports that the amine base scavenges reduced SF_6 material (S_nF_m or F-), which can maintain the internal resistance stable.

d. SF₆ + glycoside at multiple scan rate

Glycoside: 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranoside (**1b**) Scan range: -2.0 V to 0.0 V Scan rates (in chronological order): 1) 50 mV/s 2) 100 mV/s 3) 25 mV/s 4) 75 mV/s 5) 200 mV/s Electrolyte: TBAClO₄ (0.15M) / Solvent: THF (12 mL) Working electrode: 1.5 mm diameter Sn wire (0.6 cm submerged)

Counter electrode: 2.0 mm diameter Zn wire (1.5 cm submerged)

Reference electrode: Ag/AgCl



Figure S11. Reduction of SF_6 in the presence of glycoside substrate with Zn as sacrificial anode. [1b] = 33 mM.

In the presence of SF_6 and glycoside substrate, the voltammogram is consistently showing the decrease of cathodic current in chronological order of the scanning. This result suggests that deoxofluorination of glycoside is a chemical reaction process which is not catalyzed by electrolysis.

e. SF₆ + glycoside +NEt₃ at multiple scan rate

Glycoside: 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranoside (**1b**) Scan range: -2.0 V to 0.0 V Scan rates (in chronological order): 1) 50 mV/s 2) 100 mV/s 3) 25 mV/s 4) 75 mV/s 5) 200 mV/s Electrolyte: TBACIO4 (0.15M) / Solvent: THF (12 mL)

Working electrode: 1.5 mm diameter Sn wire (0.6 cm submerged)

Counter electrode: 2.0 mm diameter Zn wire (1.5 cm submerged)

Reference electrode: Ag/AgCl



Figure S12. Reduction of SF₆ under standard reaction condition. [1b] = 33 mM, [NEt3] = 99 mM

The CV experiment under standard reaction condition shows the enhancement of cathodic current under slower scan rate again because the amine base is present in the solution.

V. Additional Experiments

a. Electric Current On/Off Experiment



Figure S13. Electric current On/Off experiment

An oven-dried 1 dram vial was charged with 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranoside **1a** (0.1 mmol, 1 equiv.) and tetrabutylammonium perchlorate (0.45 mmol, 0.15 M). The vial containing the mixture was sealed with PTFE/Silicone septum cap and was further dried under high vacuum for 5 minutes. Into the sealed vial, THF (3 mL), triethylamine (0.3 mmol, 3 equiv.), and α,α,α -trifluorotoluene (5 microL) were added, and the vial was stirred until the mixture became homogeneous. The mixture was sparged with a balloon containing SF₆ gas for 30 seconds. After the sparging was done, the valve of the balloon was adjusted to sparge small bubbles and left open until the reaction was stopped. Cathode and anode were connected to DC power supply and the reaction was stirred at high rpm. The reaction progress was monitored by taking an aliquot of sample (0.2 mL) every 15 minutes.

30 minutes after the start of reaction, the electric current was shut off and then a sample was taken out. 15 minutes after, another aliquot of sample was taken out, and then the electric current was applied again.

The result is showing that the reaction proceeds only when the electric current is applied, which means that the active fluorinating agent formed *in situ* is consumed right away and does not accumulate.

b. Control Experiment with ZnF₂



An oven-dried 1 dram vial was charged with 2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranoside **1a** (0.1 mmol, 1 equiv.), zinc(II) fluoride (0.1 mmol, 1 equiv.), and tetrabutylammonium perchlorate (0.45 mmol, 0.15 M). The vial containing the mixture was sealed with PTFE/Silicone septum cap and was further dried under high vacuum for 5 minutes. Into the sealed vial, THF (3 mL) and triethylamine (0.3 mmol, 3 equiv.) were added. The mixture was stirred until everything, except ZnF₂, became homogeneous. Sn cathode and Zn anode were connected to DC power supply and the reaction was stirred at high rpm for 1 hour. The color of the reaction turned from clear to black, and the electric current dropped from 15 mA to 3 mA after the voltage reached the limit of DC power supply (31 V). ¹H and ¹⁹F NMR of crude and worked-up reaction did not show any formation of fluorinated product.

c. PPh₃ Quenching Experiment



2,3,4,6-tetrakis-*O*-(phenylmethyl)-D-glucopyranoside **1a** was subjected to a standard reaction. After the electric current was shut off, an aliquot of crude reaction was taken out, and then PPh₃ solution (0.1 mmol in 0.5 mL of THF) was injected into the reaction vial. The mixture was then stirred for 4 hours, and then the samples were taken out. All samples were subjected to ¹⁹F and ³¹P NMR experiments and were compared with each other and with the NMR spectra from our photocatalytic reaction.²¹ Careful examination of ¹⁹F and ³¹P NMR spectrum of electrocatalytic reaction showed the formation of S=PPh₃ and O=PPh₃ just like photocatalytic reaction. However, F₂PPh₃ is formed in trace amount unlike photocatalytic reaction, which suggests that elemental fluorine or fluoride anion is scavenged by triethylamine or zinc cations.

²¹Kim, S.; Khomutnyk, Y.; Bannykh, A.; Nagorny, P. Org. Lett. 2021, 23, 1, 190-194



Figure S14. ³¹P NMR spectra of PPh₃ quenching experiment.



Figure S15. ¹⁹F NMR spectra of PPh₃ quenching experiment.

¹H NMR (CDCl₃, 700MHz) ,OBn BnO BnO BnÒ **2a-**β 8288888888888888 (Major) - -رتدر Ŵ 3,90 3,70 3,65 f1 (ppm) 3,95 4,00 3,60 3,55 3,50 *** ሣ ሥሥ Ч 575 9:36J Ч 8 883 228 8.0 83 oi. 7,5 7,0 6,5 6,0 5,5 4,0 3,5 3,0 2,5 2,0 5,0 4,5 1,5 1,0 'n 0,

2,3,4,6-Tetrakis-O-(phenylmethyl)-β-D-galactopyranosyl fluoride (2a-β, Major diastereomer)

f1 (ppm)





¹H NMR (CDCl₃, 700MHz) BnO _OBn BnO-BnÒ, **2a-**α (Minor) 55588 17 \7 95556 111 1111-00000 8 8 4,15 4,10 4,05 4,00 3,95 3,90 f1 (ppm) 20.10-ペヤヤ * 77 ላሌ ት ٣ 4 2.09 8 958 ទទទ 888 0.01 7,5 6,5 5,5 3,5 7,0 4,5 4,0 3,0 2,5 2,0 1,5 6,0 5,0 1,0 0, f1 (ppm)

2,3,4,6-Tetrakis-O-(phenylmethyl)-α-D-galactopyranosyl fluoride (2a-α, Minor diastereomer)










2,3,4,6-Tetrakis-O-(phenylmethyl)-D-glucopyranosyl fluoride (2b, Isolated as α : β = 1 : 1.4)









4,6-Tetrakis-*O*-(phenylmethyl)-α-D-Mannopyranosyl fluoride (2c-α, Major diastereomer)













2,3,4-tris-O-(phenylmethyl)-D-xylopyranosyl fluoride (2d, Isolated as α : β = 1 : 2.4 mixture)





<111 116 116 116 ¹H NMR (CDCl₃, 700MHz) Me OBn l OBn BnO **2e-**α (Major) 55 55 61 61 1477 1474 1747 1747 1747 1747 8:8:8 3:8:8:8 1/// 4.67 22888888888892 1111111 5.50 5.00 4.95 4.90 4.85 4.80 4.75 4.70 4.65 f1 (ppm) 5,60 f1 (ppm) 4,05 3,95 4,00 f1 (ppm) ተ ሣምሥ ተተተ н Ч н 8 8 888 1.96 10 8 7,5 7,0 6,5 5,5 2,5 1,0 6,0 4,5 4,0 3,5 3,0 2,0 1,5 5,0 f1 (ppm)

4-Tris-*O*-(phenylmethyl)-α-L-fucopyranosyl fluoride (2e-α, Major diastereomer)







2,3,4-Tris-O-(phenylmethyl)-β-L-fucopyranosyl fluoride (2e-β, Minor diastereomer)

























2,3,5-Tris-O-(phenylmethyl)- β -D-ribofuranosyl fluoride (2g- β , Major diastereomer)










2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-β-D-galactopyranosyl fluoride (2h-β, Major diastereomer)









2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-α-D-galactopyranosyl fluoride (2h-α, Minor diastereomer)













2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]- β -D-glucopyranosyl fluoride (2i, Isolated as α : β = 1 : 1.4 mixture)







2,3,4,6-Tetrakis-O-[(4-methoxyphenyl)methyl]-α-D-mannopyranosyl fluoride (2j- α, Major diastereomer)













2,3,4,6-Tetra-*O*-acetyl-α-D-galactopyranosyl fluoride (2k-α, Major diastereomer)







2,3,4,6-Tetra-O-acetyl-β-D-galactopyranosyl fluoride (2k-β, Minor diastereomer)











2,3,4,6-Tetra-O-acetyl-β-D-glucopyranosyl fluoride (2I-β, Major diastereomer)









2,3,4,6-Tetra-O-acetyl-α-D-glucopyranosyl fluoride (2I-α, Minor diastereomer)












2,3,4,6-Tetra-O-acetyl-α-D-mannopyranosyl fluoride (2m-α, Major diastereomer)



















2,3,4-Tri-O-benzoyI-α-L-rhamnopyranosyl fluoride (20-α, Major diastereomer)













2,3-Bis-*O*-(phenylmethyl)-4,6-*O*-[(R)-phenylmethylene]-α-D-mannopyranosyl fluoride (2p-α, Major diastereomer)







2,3-Bis-O-(phenylmethyl)-4,6-O-[(R)-phenylmethylene]-β-D-mannopyranosyl fluoride (2p-β, Minor diastereomer)











2,3,6-Tris-O-(phenylmethyl)-4-O-[2,3,4,6-tetrakis-O-(phenylmethyl)-β-D-glucopyranosyl]-β-D-glucopyranosyl fluoride (2q-β, Major diastereomer)







2,3,6-Tris-*O*-(phenylmethyl)-4-*O*-[2,3,4,6-tetrakis-*O*-(phenylmethyl)-β-D-glucopyranosyl]-α-D-glucopyranosyl fluoride (2q-α, Minor diastereomer)









