

# Mitochondrial Respiration Can Support $\text{NO}_3^-$ and $\text{NO}_2^-$ Reduction during Photosynthesis<sup>1</sup>

## Interactions between Photosynthesis, Respiration, and N Assimilation in the N-Limited Green Alga *Selenastrum minutum*

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### ABSTRACT

Mass spectrometric analysis shows that assimilation of inorganic nitrogen ( $\text{NH}_4^+$ ,  $\text{NO}_2^-$ ,  $\text{NO}_3^-$ ) by N-limited cells of *Selenastrum minutum* (Naeg.) Collins results in a stimulation of tricarboxylic acid cycle (TCA cycle)  $\text{CO}_2$  release in both the light and dark. In a previous study we have shown that TCA cycle reductant generated during  $\text{NH}_4^+$  assimilation is oxidized via the cytochrome electron transport chain, resulting in an increase in respiratory  $\text{O}_2$  consumption during photosynthesis (HG Weger, DG Birch, IR Elrifi, DH Turpin [1988] *Plant Physiol* 86: 688–692).  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation resulted in a larger stimulation of TCA cycle  $\text{CO}_2$  release than did  $\text{NH}_4^+$ , but a much smaller stimulation of mitochondrial  $\text{O}_2$  consumption.  $\text{NH}_4^+$  assimilation was the same in the light and dark and insensitive to DCMU, but was 82% inhibited by anaerobiosis in both the light and dark.  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation rates were maximal in the light, but assimilation could proceed at substantial rates in the light in the presence of DCMU and in the dark. Unlike  $\text{NH}_4^+$ ,  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation were relatively insensitive to anaerobiosis. These results indicated that operation of the mitochondrial electron transport chain was not required to maintain TCA cycle activity during  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation, suggesting an alternative sink for TCA cycle generated reductant. Evaluation of changes in gross  $\text{O}_2$  consumption during  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation suggest that TCA cycle reductant was exported to the chloroplast during photosynthesis and used to support  $\text{NO}_3^-$  and  $\text{NO}_2^-$  reduction.

Mitochondrial respiration consists of two interrelated processes: TCA<sup>2</sup> cycle carbon flow and mitochondrial electron transport. The role of mitochondrial respiration in the light has been the subject of considerable debate, but it is generally agreed that some TCA cycle activity is maintained to provide carbon skeletons for biosynthetic reactions (10, 24, 29). Use of radiolabeled TCA cycle intermediates has provided evi-

dence for the maintenance of TCA cycle carbon flow in photosynthesizing cells (4, 20). Additional work has shown that during the assimilation of  $\text{NH}_4^+$  and  $\text{NO}_3^-$  the flow of carbon through the TCA cycle increases to provide carbon skeletons for amino acid synthesis (9, 15, 18, 30).

Operation of the TCA cycle is dependent upon the oxidation of NADH and  $\text{FADH}_2$  (17). This may occur via the mitochondrial electron transport chain, which results in  $\text{O}_2$  consumption (22), or it may be coupled to biosynthetic reactions such as  $\text{NO}_3^-$  reduction (31). It is often assumed that during photosynthesis an increase in the ATP/ADP ratio would inhibit mitochondrial electron transport chain activity. Recent evidence, however, indicates that there is little difference in ATP/ADP ratios between light and dark (12, 27), and mass spectrometric analyses of  $\text{O}_2$  exchange in green algae have shown that there is considerable activity of the mitochondrial electron transport chain during photosynthesis (23, 29).

In previous studies, we have demonstrated that the assimilation of  $\text{NH}_4^+$  by photosynthesizing cells of *S. minutum* results in increased rates of both TCA cycle  $\text{CO}_2$  release (3, 29) and mitochondrial  $\text{O}_2$  consumption (29). This stimulation of TCA cycle carbon flow would result in an increase in the production of  $\alpha\text{KG}$ , needed for  $\text{NH}_4^+$  assimilation and net glutamate synthesis. The inhibition of  $\text{NH}_4^+$ -induced  $\text{CO}_2$  release by cyanide and anaerobiosis was consistent with the mitochondrial electron transport chain oxidizing the NADH and  $\text{FADH}_2$  produced during this increase in TCA cycle activity (29). In this study we present evidence which shows that  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation stimulate TCA cycle activity to a greater degree than does the assimilation of  $\text{NH}_4^+$ . We also show that a substantial portion of the reductant generated during this increase in TCA cycle activity appears to be exported from the mitochondrion to the chloroplast and used in support of  $\text{NO}_3^-$  and  $\text{NO}_2^-$  reduction both in the dark and during photosynthesis.

### MATERIALS AND METHODS

*Selenastrum minutum* (Naeg.) Collins (Chlorophyta) (UTEX 2459) was grown in chemostat culture under N-limitation as previously described (7). Gross  $\text{O}_2$  and  $\text{CO}_2$  exchange were measured by mass spectrometry as described by Weger *et al.* (29), except that mixing employed a magnetic stirrer and  $^{18}\text{O}_2$  was introduced as a bubble into the algal

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<sup>2</sup> Abbreviations: TCA cycle, tricarboxylic acid cycle; DIC, dissolved inorganic carbon; GS, glutamine synthetase; GOGAT, glutamine 2-oxoglutarate aminotransferase; RuBP, ribulose 1,5-bisphosphate; Rubisco, ribulose 1,5-bisphosphate carboxylase oxygenase;  $\alpha\text{KG}$ ,  $\alpha$ ketoglutarate; OAA, oxaloacetate.

suspension. Dark O<sub>2</sub> consumption was measured using an O<sub>2</sub> electrode (Hansatech, King's Lynn, England).

Ammonium, nitrite, and nitrate assimilation were measured as N disappearance from the medium. Assimilation was measured using chemostat cells concentrated 2.5 times to a Chl concentration of approximately 5.3  $\mu\text{g}\cdot\text{mL}^{-1}$ . Cells were placed in a water-jacketed (20°C) glass cuvette containing two serum stoppered sampling ports, one of which contained an O<sub>2</sub> electrode (YSI 5331, Yellow Springs, OH). Samples were withdrawn through the other port using 5 mL syringes. Samples were filtered through Whatman 934AH filters, and the filtrate was frozen until analysis. Ammonium was determined using an Orion 95-12 NH<sub>3</sub> electrode (Orion Instruments Inc., Cambridge, MA) for aerobic experiments. For anaerobic experiments the O<sub>2</sub> scavenging system (glucose/glucose oxidase/catalase) interfered with the operation of the NH<sub>3</sub> electrode; consequently, ammonium was determined colorimetrically according to Strickland and Parsons (27). Nitrate and nitrite were determined according to Strickland and Parsons (27), modified for flow through sample injection.

When required, anaerobiosis was achieved by adding glucose oxidase (final concentration 100  $\mu\text{g}\cdot\text{mL}^{-1}$ ) to a cell suspension containing 10 mM glucose and 20  $\mu\text{g}\cdot\text{mL}^{-1}$  catalase. Some anaerobic experiments in the light were conducted in the presence of DCMU to prevent O<sub>2</sub> production from water photolysis. DCMU was added to a final concentration of 20  $\mu\text{M}$ , which was found to have no effect on O<sub>2</sub> consumption in either the light or dark. All inhibitors and glucose oxidase were obtained from Sigma Chemical Co. (St. Louis, MO).

Illumination for mass spectrometric and N assimilation experiments was provided at photon flux densities of 80 or 250  $\mu\text{E}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$  (low and high light, respectively) by a tungsten light source. The higher photon flux density has been shown to be saturating for CO<sub>2</sub> fixation in this organism, while the lower rate is just subsaturating (7).

Steady state Chl *a* fluorescence was measured using a PAM fluorometer (Heinz Walz, Effeltrich FRG) as described by Turpin and Weger (28). Photochemical quenching ('Q quenching') was calculated according to Schreiber *et al.* (25). The effects of anaerobiosis on photochemical quenching were investigated by monitoring steady state fluorescence before and after adding glucose oxidase to an algal suspension (8  $\mu\text{g Chl}\cdot\text{mL}^{-1}$ ) containing glucose and catalase and illuminated at 110  $\mu\text{E}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ . These experiments were undertaken both in the presence and absence of 2 mM NaNO<sub>3</sub>.

Chl was measured by extraction in 100% methanol (7).

## RESULTS

### CO<sub>2</sub> Release

Addition of a nitrogen source (NH<sub>4</sub><sup>+</sup>, NO<sub>2</sub><sup>-</sup>, or NO<sub>3</sub><sup>-</sup>) to N-limited cells of *S. minutum* resulted in a dramatic increase in the rate of TCA cycle CO<sub>2</sub> release (Table I). The assimilation of NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> resulted in higher rates of TCA cycle CO<sub>2</sub> release than did the assimilation of NH<sub>4</sub><sup>+</sup>. In the dark the increase in TCA cycle CO<sub>2</sub> release during NO<sub>3</sub><sup>-</sup> or NO<sub>2</sub><sup>-</sup> assimilation is approximately 50% greater than that resulting from NH<sub>4</sub><sup>+</sup> assimilation. During photosynthesis the difference

**Table I.** Effects of N Assimilation on the Measured Rate of CO<sub>2</sub> Release

N sources were added to a concentration of 1 mM, and the rate of <sup>12</sup>CO<sub>2</sub> release into [<sup>13</sup>C]DIC labeled medium was monitored for 15 min. All rates are the means of at least three experiments.

| N Source                      | Dark | 80 $\mu\text{E}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$                                 |     | 250 $\mu\text{E}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ |  |
|-------------------------------|------|--|-----|---|--|
|                               |      | $\mu\text{mol CO}_2\text{ released}\cdot\text{mg}^{-1}\text{ Chl}\cdot\text{h}^{-1}$ |     |   |  |
| Control                       | 75   | 29   | 37  |   |  |
| +NH <sub>4</sub> <sup>+</sup> | 187  | 100  | 97  |   |  |
| +NO <sub>2</sub> <sup>-</sup> | 234  | 166  | 165 |   |  |
| +NO <sub>3</sub> <sup>-</sup> | 262  | 190  | 188 |   |  |

**Table II.** Dark O<sub>2</sub> Consumption in Response to Different N Sources Added to a Concentration of 1 mM

All rates are the means of at least three experiments.

| N Source                      | O <sub>2</sub> Consumption Rate                                     |     |
|-------------------------------|---|-----|
|                               | $\mu\text{mol O}_2\cdot\text{mg}^{-1}\text{ Chl}\cdot\text{h}^{-1}$ |     |
| Control                       | 98  |     |
| +NH <sub>4</sub> <sup>+</sup> | 223   | 125 |
| +NO <sub>2</sub> <sup>-</sup> | 132   | 34  |
| +NO <sub>3</sub> <sup>-</sup> | 113   | 15  |

is even more pronounced, with both NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> producing at least twice the increase in CO<sub>2</sub> release caused by NH<sub>4</sub><sup>+</sup>.

### O<sub>2</sub> Consumption

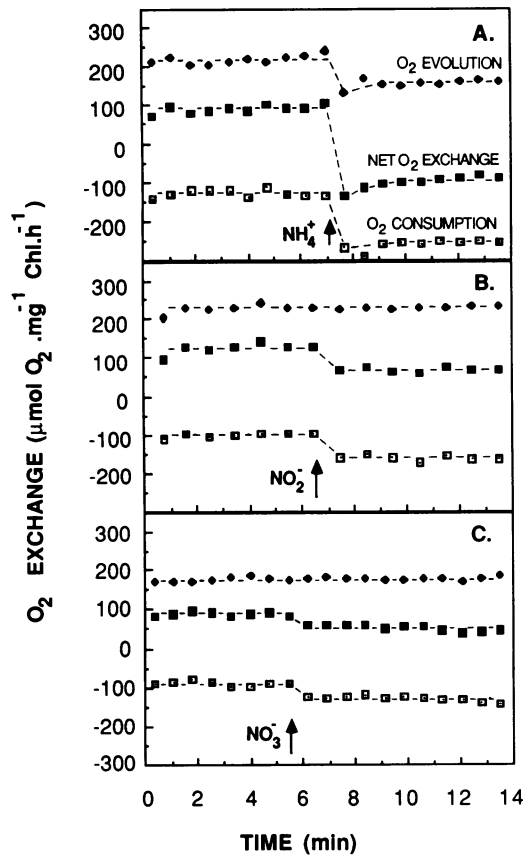
Dark O<sub>2</sub> consumption was stimulated by N assimilation, with NH<sub>4</sub><sup>+</sup> resulting in the largest stimulation, while NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> produced a relatively minor effect (Table II). These trends are also apparent during photosynthesis; NH<sub>4</sub><sup>+</sup> assimilation resulted in a large increase in the rate of gross O<sub>2</sub> consumption at both low and high light (Figs. 1A and 2A). NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> assimilation resulted in only a small stimulation of gross O<sub>2</sub> consumption at low light. At high light there is no effect of NO<sub>3</sub><sup>-</sup> addition on O<sub>2</sub> consumption and the stimulation of O<sub>2</sub> consumption by NO<sub>2</sub><sup>-</sup> was reduced (Fig. 1, B and C; Fig. 2, B and C).

### Photosynthetic O<sub>2</sub> Evolution

The various N sources also differed in their effects on gross O<sub>2</sub> evolution. NH<sub>4</sub><sup>+</sup> assimilation resulted in a rapid and substantial decline in the rate of water photolysis, at both low and high light (Figs. 1A and 2A). In contrast, NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> assimilation did not affect the rate of gross O<sub>2</sub> evolution at low light (Fig. 1, B and C), and resulted in a slight stimulation of O<sub>2</sub> evolution at high light (Fig. 2, B and C). The combined changes in gross O<sub>2</sub> evolution and gross consumption during N assimilation resulted in net O<sub>2</sub> consumption in the light during NH<sub>4</sub><sup>+</sup> assimilation, but only minor changes in net O<sub>2</sub> exchange during assimilation of NO<sub>3</sub><sup>-</sup> or NO<sub>2</sub><sup>-</sup>. These trends in net O<sub>2</sub> exchange confirm those previously reported using an O<sub>2</sub> electrode (7).

### N Assimilation

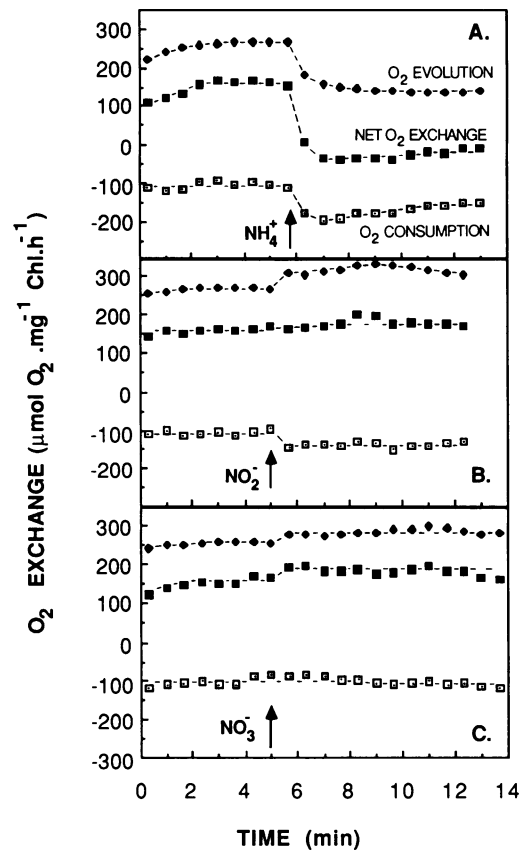
Culturing algae under N-limitation dramatically increases their capacity for N assimilation compared to N-sufficient



**Figure 1.** Effect of assimilation of different N sources on O<sub>2</sub> exchange at low light ( $80 \mu\text{E} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ ). The arrow indicates addition of 1 mM N. A,  $\text{NH}_4^+$ ; B,  $\text{NO}_2^-$ ; C,  $\text{NO}_3^-$ . (□), Gross O<sub>2</sub> consumption; (◆), gross O<sub>2</sub> evolution; (■), net O<sub>2</sub> exchange.

cells (7). Under aerobic conditions, the rate of  $\text{NH}_4^+$  assimilation by *S. minutum* is the same in the light and dark (approximately  $170 \mu\text{mol NH}_4^+ \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$ , Table III), and is unaffected by  $20 \mu\text{M}$  DCMU. Anaerobiosis in either the light or dark caused a 70 to 80% decline in the rate of  $\text{NH}_4^+$  assimilation. Under these conditions the presence of DCMU had only a slight inhibitory effect on  $\text{NH}_4^+$  assimilation, indicating that the O<sub>2</sub> scavenging system was capable of maintaining functionally anaerobic conditions even during photosynthesis at low light (Table III).

The rates of  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation were greatest at high light (Table III). Addition of DCMU to aerobic cell suspensions at low light resulted in approximately a 55% decrease in the rate of  $\text{NO}_3^-$  assimilation relative to that in the absence of DCMU. The resulting rate was equal to that observed aerobically in the dark. Placing DCMU-treated illuminated cells under anaerobic conditions had no further inhibitory effect on the rate of  $\text{NO}_3^-$  assimilation. The presence of an O<sub>2</sub> scavenging system alone (in the absence of DCMU) had no effect on the rate of  $\text{NO}_3^-$  assimilation. In contrast, in the dark, anaerobiosis produced a 50% decline in  $\text{NO}_3^-$  assimilation relative to the aerobic rate (Table III).  $\text{NO}_2^-$  release to the medium was found during  $\text{NO}_3^-$  assimilation under all conditions, similar to reports for several other microalgae (2, 5). The average  $\text{NO}_2^-$  release rate under illuminated conditions was  $34 \mu\text{mol NO}_2^- \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$ , while



**Figure 2.** The effect of assimilation of different N sources on O<sub>2</sub> exchange at high light ( $250 \mu\text{E} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ ). The arrow indicates addition of 1 mM N. A,  $\text{NH}_4^+$ ; B,  $\text{NO}_2^-$ ; C,  $\text{NO}_3^-$ . (□), Gross O<sub>2</sub> consumption; (◆), gross O<sub>2</sub> evolution; (■), net O<sub>2</sub> exchange.

**Table III.** Rates of N Assimilation under Various Conditions  
All rates are the means of at least two experiments.

| Photon Flux Density                                   | Treatment        | N Source  |                 |                 |
|---|------------------|---|-----------------|-----------------|
|   |                  | $\text{NH}_4^+$   | $\text{NO}_2^-$ | $\text{NO}_3^-$ |
| $\mu\text{E} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ |                  | $\mu\text{mol N assimilated} \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$ |                 |                 |
| 250   | Aerobic          | 170   | 160             | 155             |
| 80  | Aerobic          | 167   | 117             | 127             |
| 80  | Aerobic + DCMU   | 177   | 56              | 56              |
| 80  | Anaerobic        | 49  | 81              | 133             |
| 80  | Anaerobic + DCMU | 31  | 43              | 57              |
| Dark  | Aerobic          | 169   | 55              | 53              |
| Dark  | Anaerobic        | 24  | 11              | 27              |

in the dark the rate increased to  $63 \mu\text{mol NO}_2^- \cdot \text{mg}^{-1} \cdot \text{h}^{-1}$ . The rates of  $\text{NO}_3^-$  assimilation reported in Table III were calculated as  $\text{NO}_3^-$  disappearance minus  $\text{NO}_2^-$  release (*i.e.* net N assimilation).

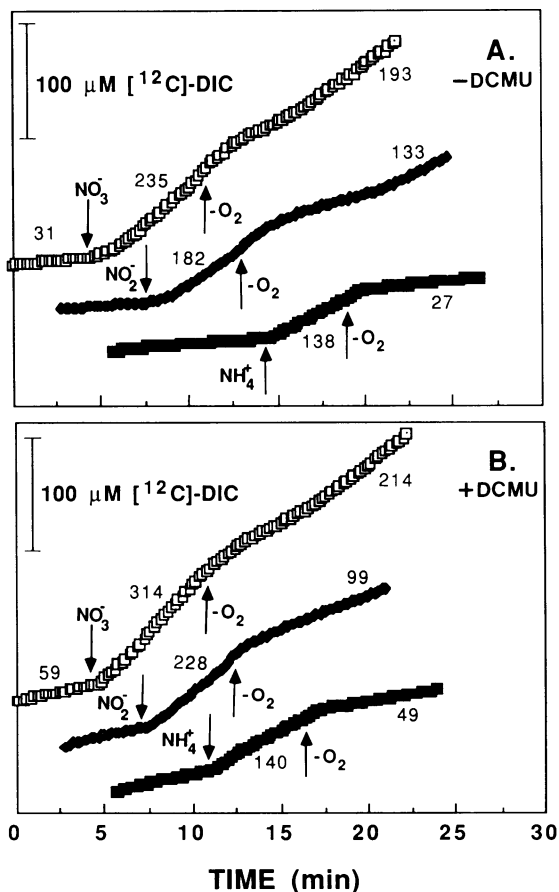
The effects of these treatments on  $\text{NO}_2^-$  assimilation were similar to those observed with  $\text{NO}_3^-$ . The addition of DCMU to cells illuminated with low light resulted in a 50% decline in the  $\text{NO}_2^-$  assimilation rate, to a level equal to the dark aerobic rate (Table III). Placing illuminated, DCMU-treated cells under anaerobic conditions resulted in only a 20% decrease in the rate of  $\text{NO}_2^-$  assimilation compared to the aerobic rate. The same treatment, in the absence of DCMU,

produced comparable results. In the dark, however, anaerobiosis resulted in a 75% inhibition of  $\text{NO}_2^-$  assimilation (Table III).

### Effects of Anaerobiosis on $\text{CO}_2$ Release during N Assimilation

Anaerobiosis reduced the control  $\text{CO}_2$  release rate at low light to  $6 \mu\text{mol CO}_2 \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$  (29) from control values of  $29 \mu\text{mol CO}_2 \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$  (Table I). As shown in Table I, N assimilation resulted in a stimulation of TCA cycle  $\text{CO}_2$  release. Subsequent anaerobiosis reduced the  $\text{NH}_4^+$ -stimulated rate to below aerobic control (Fig. 3A). In contrast, under anaerobic conditions, both  $\text{NO}_3^-$  and  $\text{NO}_2^-$  supported  $\text{CO}_2$  release rates more than 4 times greater than aerobic control (Fig. 3A).

These experiments were repeated in the presence of DCMU to prevent water photolysis and thus ensure strict anaerobic conditions (Fig. 3B). The results are similar to those described above.  $\text{NH}_4^+$ -stimulated  $\text{CO}_2$  release was reduced to below the aerobic control rate upon imposition of anaerobiosis, while both  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation supported much higher rates of anaerobic  $\text{CO}_2$  release.



**Figure 3.** Effects of different N sources and anaerobiosis on respiratory  $^{12}\text{CO}_2$  release during photosynthesis into medium containing  $2 \text{ mM } [^{13}\text{C}]\text{-DIC}$ . The first arrow indicates addition of  $1 \text{ mM N}$ , and the second arrow indicates the onset of anaerobiosis by addition of glucose oxidase;  $10 \text{ mM glucose}$  and  $20 \mu\text{g/mL catalase}$  were added prior to the experiment. A, In the absence of DCMU; B, in the presence of DCMU.

### Effect of Anaerobiosis on Photochemical Quenching

In control cells the onset of anaerobiosis produced a transient decline in photochemical quenching, which returned to aerobic control levels within 10 min. During  $\text{NO}_3^-$  assimilation, however, photochemical quenching declined from a mean aerobic level of 0.57 to 0.50 after 10 min of anaerobiosis. There was no recovery of photochemical quenching for the duration of the experiment. This indicates that the pool of reduced Q increases upon anaerobiosis during  $\text{NO}_3^-$  assimilation, suggesting a decrease in the requirements for photodriven electrons (16). This decline in photochemical quenching is not related to a decrease in the rate of  $\text{O}_2$  photoreduction (Mehler reaction) caused by the addition of glucose oxidase, as DCMU or light/dark/light transitions had no effect on short term gross  $\text{O}_2$  consumption (data not shown).

### DISCUSSION

In both higher plants and green algae the assimilation of inorganic nitrogen proceeds predominately via the GS-GOGAT pathway (21). Net synthesis of glutamate by this pathway requires the operation of a portion of the TCA cycle to provide  $\alpha\text{KG}$  for use in the GOGAT reaction. It is therefore not surprising that  $\text{NH}_4^+$  addition to photosynthesizing cells results in an increase in TCA cycle carbon flow (15, 18, 30). Growth of the green alga *S. minutum* under N-limitation causes an increase in the capacity for N assimilation (3, 7). As a result, resupply of  $\text{NH}_4^+$  produces a dramatic increase in TCA cycle carbon flow and  $\text{CO}_2$  release in both the light and dark (Table I; Refs. 3, 29). This increase in TCA cycle activity implies that the rate at which NADH and  $\text{FADH}_2$  are oxidized also increases. During  $\text{NH}_4^+$  assimilation by N-limited *S. minutum*, this oxidation is coupled to the mitochondrial electron transport chain both in the dark and during photosynthesis (29).

As with  $\text{NH}_4^+$  assimilation,  $\text{NO}_3^-$  and  $\text{NO}_2^-$  also increased the rate of TCA cycle carbon flow and  $\text{CO}_2$  release (Table I; Refs. 3, 8). Surprisingly, the rate of  $\text{CO}_2$  release observed during the assimilation of  $\text{NO}_3^-$  and  $\text{NO}_2^-$  was much greater than that observed with  $\text{NH}_4^+$  (Table I). In fact, the ratio of  $\text{CO}_2$  released/N assimilated was highest with  $\text{NO}_3^-$  and lowest with  $\text{NH}_4^+$ , implying greater TCA cycle activity during the assimilation of oxidized N sources (Table IV). This implied that the oxidation of TCA cycle reductant during  $\text{NO}_2^-$  and  $\text{NO}_3^-$  assimilation must occur at higher rates than during  $\text{NH}_4^+$  assimilation.

One potential mechanism for this oxidation would be the mitochondrial electron transport chain and the coupled reduction of  $\text{O}_2$  to  $\text{H}_2\text{O}$ . Unlike the case with  $\text{NH}_4^+$ , the assimilation of  $\text{NO}_3^-$  and  $\text{NO}_2^-$  resulted in only a small stimulation of  $\text{O}_2$  consumption either in the dark (Table II) or during photosynthesis (Fig. 1, B and C; Fig. 2, B and C). This suggests that the mitochondrial electron transport chain is not as important in the oxidation of TCA cycle reductant during  $\text{NO}_3^-$  or  $\text{NO}_2^-$  assimilation as it is during the assimilation of  $\text{NH}_4^+$ . In fact, the ratio of  $\text{CO}_2$  released/ $\text{O}_2$  consumed varies greatly depending upon the N source being assimilated (Table IV). During  $\text{NH}_4^+$  assimilation in the dark, the 'RQ' value was 0.84, a value within the theoretical range expected. When

**Table IV.** Stimulation of  $\text{CO}_2$  Release Relative to the Rates of N Assimilation (A) or, Relative to the Stimulation of  $\text{O}_2$  Consumption Resulting from the Assimilation of Different N Sources (B)

All ratios are expressed as  $\mu\text{mol CO}_2 \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1} / \mu\text{mol (N or O}_2) \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$ . The  $\text{CO}_2$  released/ $\text{O}_2$  consumed ratio (RQ) was determined using gross fluxes as calculated by mass spectrometry, and therefore differs from classical measures of RQ which employ net gas exchange measurements. The RQ value for  $\text{NO}_3^-$  at high light is undefined because no stimulation of gross  $\text{O}_2$  consumption was observed (see Fig. 2C). Note that RQ values determined in the light are 'apparent RQ's,' biased by photosynthetic refixation of respired  $\text{CO}_2$ .

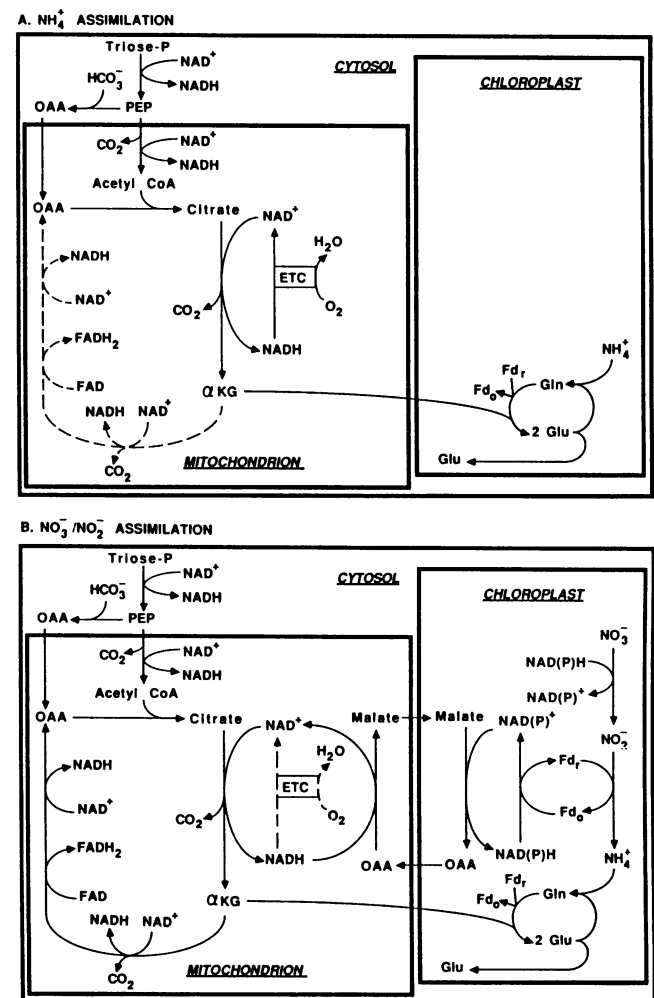
|  | Dark | $80 \mu\text{E} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ | $250 \mu\text{E} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ |
|--|------|--|---|
| <b>A. <math>\text{CO}_2</math> released/N assimilated (ratio: <math>\text{CO}_2/\text{N}</math>)</b>                           |      |  |   |
| $\text{NH}_4^+$  | 0.66 | 0.36   | 0.35  |
| $\text{NO}_2^-$  | 2.89 | 1.06   | 0.80  |
| $\text{NO}_3^-$  | 3.53 | 1.73   | 1.25  |
| <b>B. <math>\text{CO}_2</math> released/<math>\text{O}_2</math> consumed (RQ) (ratio: <math>\text{CO}_2/\text{O}_2</math>)</b> |      |  |   |
| Control  | 0.77 | 0.25   | 0.37  |
| $\text{NH}_4^+$  | 0.84 | 0.43   | 0.76  |
| $\text{NO}_2^-$  | 3.79 | 2.11   | 3.95  |
| $\text{NO}_3^-$  | 8.13 | 4.17   | Undefined   |

$\text{NO}_2^-$  and  $\text{NO}_3^-$  were assimilated, however, the RQ increased to 3.79 and 8.13, respectively. The most logical explanation of these extremely high values is that the oxidation of TCA cycle reductant during the assimilation of  $\text{NO}_2^-$  and  $\text{NO}_3^-$  did not result in concomitant  $\text{O}_2$  consumption. In other words, there must be an electron sink other than  $\text{O}_2$ , which is available during transient  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation, possibly  $\text{NO}_3^-$  and  $\text{NO}_2^-$  themselves. During photosynthesis, similar trends in the 'apparent RQ' were observed, once again suggesting the existence of alternative electron sinks.

During steady state photosynthesis and N assimilation, photogenerated reductant is thought to provide most or all of the reducing power for  $\text{NO}_3^-$  and  $\text{NO}_2^-$  reduction. However, examination of the rates of gross  $\text{O}_2$  evolution suggest that during transient N assimilation by N-limited *S. minutum*, photogenerated reductant alone was insufficient to meet the demands of  $\text{NO}_3^-$  and  $\text{NO}_2^-$  reduction, which exceed those of carbon fixation. For example, during  $\text{NO}_3^-$  assimilation at high light the rate of gross  $\text{O}_2$  evolution after  $\text{NO}_3^-$  addition (approximately  $300 \mu\text{mol O}_2 \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$ ; Fig. 2C) indicated a rate of photogenerated reductant production of  $600 \mu\text{mol } (2e^-) \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$ . The simultaneous rate of  $\text{NO}_3^-$  reduction to  $\text{NH}_4^+$  is  $155 \mu\text{mol NO}_3^- \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$  (Table III), requiring  $620 \mu\text{mol } (2e^-) \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$ . The  $\text{NO}_2^-$  release rate was  $34 \mu\text{mol NO}_2^- \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$ , indicating additional reductant demands of  $34 \mu\text{mol } (2e^-) \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$ . An additional  $155 \mu\text{mol } (2e^-) \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$  would be required for assimilation of  $\text{NH}_4^+$  via GOGAT. The total reductant requirements (not including photosynthetic carbon assimilation) are therefore  $809 \mu\text{mol } (2e^-) \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$ , a value far in excess of that available from the light reactions of photosynthesis ( $600 \mu\text{mol } (2e^-) \cdot \text{mg}^{-1} \text{Chl} \cdot \text{h}^{-1}$ ). The occurrence of photosynthetic carbon fixation makes the reductant deficit even larger. We hypothesize that this deficiency in photosynthetic reducing power is overcome by TCA cycle reductant. During  $\text{NH}_4^+$  assimilation, however, TCA cycle reductant is oxidized via the mitochondrial electron transport

chain. A diagrammatic representation of this hypothesis is presented in Figure 4.

This hypothesis can be tested in several ways. If the mitochondrial electron transport chain is responsible for oxidizing TCA cycle reductant during  $\text{NH}_4^+$  assimilation, but  $\text{NO}_3^-$  and  $\text{NO}_2^-$  fulfill this role during their assimilation (Fig. 4), then two predictions can be made. The first is that inhibition of the mitochondrial electron transport chain should result in a greater inhibition of TCA cycle  $\text{CO}_2$  release during  $\text{NH}_4^+$  assimilation than during the assimilation of  $\text{NO}_2^-$  or  $\text{NO}_3^-$ . Second, inhibition of the electron transport chain should result in a larger decrease in the rate of  $\text{NH}_4^+$  assimilation than for either  $\text{NO}_3^-$  and  $\text{NO}_2^-$ .



**Figure 4.** Proposed pathways of TCA cycle electron and carbon flow during transient  $\text{NH}_4^+$  assimilation (A) and transient  $\text{NO}_3^-/\text{NO}_2^-$  assimilation (B) in the light by N-limited *S. minutum*. Cytosolic triose phosphate would be provided by the chloroplast, from the Calvin cycle and from starch breakdown in the light (8, 9). The actual mechanism of reductant shuttling during inorganic N assimilation is unknown, and has been represented in (B) as malate/oxaloacetate exchange across the mitochondrial and chloroplast envelopes. The light reactions of photosynthesis would also provide photogenerated reductant (NADPH and Fd<sub>r</sub>), and therefore contribute to  $\text{NO}_3^-/\text{NO}_2^-$  reduction in the light (see "Discussion"). This is not illustrated. Abbreviations:  $\alpha\text{KG}$ ,  $\alpha$ ketoglutarate; OAA, oxaloacetate; Fd<sub>r</sub>, reduced ferredoxin; Fd<sub>o</sub>, oxidized ferredoxin.

Previous work with *S. minutum* has shown that inhibition of the mitochondrial electron transport chain via anaerobiosis inhibits both control and  $\text{NH}_4^+$ -stimulated  $\text{CO}_2$  release from the TCA cycle (29). Anaerobiosis, however, allowed much greater rates of TCA cycle  $\text{CO}_2$  release during  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation, both in the presence and absence of DCMU (Fig. 3, A and B). These results are consistent with oxidized N acting as an electron sink for TCA cycle reductant, thereby maintaining TCA cycle carbon flow and associated  $\text{CO}_2$  release.

Since anaerobiosis results in a much lower rate of TCA cycle carbon flow during  $\text{NH}_4^+$  assimilation than during the assimilation of oxidized N sources, this treatment should also substantially decrease  $\text{NH}_4^+$  assimilation due to a lack of carbon skeletons for the GOGAT reaction. Consistent with this prediction, imposition of anaerobiosis (in the absence of DCMU) resulted in a much larger decrease in the rate of  $\text{NH}_4^+$  assimilation than on the rates of  $\text{NO}_3^-$  or  $\text{NO}_2^-$  assimilation (Table III). The large decrease in the rate of  $\text{NH}_4^+$  assimilation is probably a direct effect of the decrease in TCA cycle carbon flow, as ATP can still be produced via both cyclic and noncyclic photophosphorylation (Fig. 3A).

The imposition of strict anaerobiosis in the light by the combination of an  $\text{O}_2$  scavenging system and 20  $\mu\text{M}$  DCMU resulted in the additional complication of preventing the production of photogenerated reductant. The decrease in the rate of oxidized N assimilation in the light caused by DCMU (Table III) can be ascribed to this inhibition. Assimilation of oxidized N requires a large amount of reductant, and nitrite reductase is thought to be intimately associated with the photosynthetic electron transport chain which supplies reduced ferredoxin (11). The observation that the assimilation of oxidized N occurs in the absence of photogenerated reductant (in illuminated cells treated with DCMU, or in the dark) indicates that other reductant sources can provide electrons to both nitrate reductase and nitrite reductase. In spite of the complication caused by DCMU, anaerobiosis in the presence of DCMU resulted in an 82% decrease in the rate of  $\text{NH}_4^+$  assimilation in the light, while  $\text{NO}_3^-$  assimilation was unaffected.  $\text{NO}_2^-$  assimilation was inhibited by only 20% (Table III).

Anaerobiosis in the dark greatly decreased the assimilation of all N sources (Table III). One possible explanation is the requirement for adequate ATP for N assimilation. In the dark, some mitochondrial electron transport chain activity is required for ATP generation, but in the light adequate ATP can be produced via cyclic photophosphorylation. The fact that  $\text{NH}_4^+$  assimilation is dependent upon mitochondrial electron transport chain activity, but  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation are not, suggests that the decrease in  $\text{NH}_4^+$  assimilation in the absence of mitochondrial electron transport chain activity during photosynthesis is not due to a limitation in ATP supply, but rather to a shortage of carbon skeletons for amino acid synthesis.

Interactions between the mitochondrion and the chloroplast during the assimilation of oxidized N should be detectable by monitoring changes in photochemical quenching of Chl *a* fluorescence. During  $\text{NO}_2^-$  assimilation the onset of anaerobiosis would remove one sink for TCA cycle reductant. Consequently, a diversion of mitochondrial reductant to

$\text{NO}_3^-$  and  $\text{NO}_2^-$  reduction in the chloroplast would be expected. Such a diversion should reduce the demand for photogenerated reductant, causing an increase in the pool of reduced Q and a decline in photochemical quenching. The data presented are consistent with this prediction.

An obvious characteristic of photosynthetic  $\text{NH}_4^+$  assimilation is the large decrease in the rate of gross  $\text{O}_2$  evolution (Figs. 1A and 2A). This occurs as a result of a decrease in RuBP concentration below the RuBP binding site density of Rubisco, causing RuBP limitation of photosynthetic carbon fixation (7, 9). The decreased rate of photosynthetic carbon fixation decreases the demand for photogenerated reductant and water photolysis. Suppression of carbon fixation also occurs during  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation (7, 9; HG Weger, DH Turpin, unpublished data), yet  $\text{O}_2$  evolution is relatively unaffected (Figs. 1 and 2). Apparently this is a result of a diversion of photogenerated reductant from  $\text{CO}_2$  fixation to the reduction of  $\text{NO}_3^-$  and  $\text{NO}_2^-$ , indicating that both photogenerated and TCA cycle reductant are utilized for the assimilation of oxidized N during photosynthesis. This is consistent with the partial inhibition of  $\text{NO}_3^-$  and  $\text{NO}_2^-$  reduction with DCMU.

In green algae both nitrite reductase and nitrate reductase are thought to be located in the chloroplast (11, 19). Since both the mitochondrial and chloroplastic inner membranes are impermeable to pyridine nucleotides, there must exist a shuttle mechanism that exports TCA cycle reductant from the mitochondrion to the chloroplast. Various dicarboxylate reductant shuttles have been proposed for both the mitochondrion (6, 14, 31) and the chloroplast (1, 13). However, in the current situation the shuttle would need to import reductant into the chloroplast during photosynthesis, the opposite of most of the proposed shuttle mechanisms. Such a system would allow the TCA cycle to operate both for the provision of the  $\alpha\text{KG}$  needed for net glutamate synthesis and as a source of reducing power in addition to that provided by the light reactions of photosynthesis (Fig. 4). Even if nitrate reductase was located in the cytosol, as is thought for higher plants, reductant import by the chloroplast for  $\text{NO}_2^-$  reduction (the bulk of the electron demand) in the light would still be occurring. The actual shuttle mechanism is unknown, but may be based upon an OAA/malate system.

## CONCLUSION

Evidence from this study suggests that in N-limited *S. minutum* TCA cycle reductant is exported from the mitochondrion to the chloroplast in support of transient  $\text{NO}_3^-$  and  $\text{NO}_2^-$  reduction both in the dark and during photosynthesis (Fig. 4). Assimilation of  $\text{NH}_4^+$  results in an increase in the rate of TCA cycle  $\text{CO}_2$  efflux in light and the dark, and a corresponding increase in mitochondrial  $\text{O}_2$  consumption.  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation result in a much larger stimulation of TCA cycle  $\text{CO}_2$  efflux than does  $\text{NH}_4^+$  assimilation, but the effects on mitochondrial  $\text{O}_2$  consumption are minor. We suggest that during  $\text{NH}_4^+$  assimilation the TCA cycle operates primarily to provide carbon skeletons for amino acid biosynthesis, and the mitochondrial electron transport chain then serves to oxidize the TCA cycle reductant. During photosynthetic  $\text{NO}_3^-$  and  $\text{NO}_2^-$  assimilation, the TCA cycle operates both to provide carbon skeletons for amino acid bio-

synthesis and to provide reducing power for  $\text{NO}_3^-$  and  $\text{NO}_2^-$  reduction (Fig. 4). These results imply that in this organism mitochondrial respiration can supply reducing power to the chloroplast during photosynthesis under conditions where the production of photogenerated reductant is inadequate to meet metabolic demands.

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