Cobalt-catalyzed alkene hydrogenation by reductive turnover

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Contents

1. General Methods

All reactions were carried out under positive pressure of argon and glassware was oven-dried at 120 ˚C for a minimum of 12 hours or flame-dried with a propane torch under high vacuum unless otherwise stated. Anhydrous solvents were distilled over calcium hydride or purchased and used without further purification. Other commercially available reagents were also used without further purification. Reactions were monitored by thin layer chromatography (TLC) using silica gel plates purchased from EMD chemicals (TLC Silica gel 60 F_{254} , 250 μ m thickness), gas chromatography mass spectrometry (GCMS, Agilent 7820A/5975). Flash column chromatography was performed over Silica gel 60 (particle size 0.04–0.063 mm) from EMD chemicals.

NMR spectra were recorded on Varian-400, Bruker DPX-400, DRX-500 and DRX-600 (cryoprobe) spectrometers. The following abbreviations (or combinations thereof) were used to explain multiplicities: $s = singlet$, $d = doublet$, $t = triplet$, $q = quartet$, $m = multiplet$, $app = apparent$, $br = broad.$

2. Starting Materials

4-phenyl-1-butene, AlCl3, was purchased and used without further purification.

*N-*allylbenzamide was prepared according to literature procedure. Spectral data matched that reported in the literature.¹

*N-*allyl-4-methylbenzamide was prepared according to literature procedure. Spectral data matched that reported in the literature.²

N-allyl-4-fluorobenzamide was prepared according to literature procedure. Spectral data matched that reported in the literature.3

N-allyl-4-bromobenzamide was prepared according to literature procedure. Spectral data matched that reported in the literature.3

*N-*allyl-2-methoxybenzamide was prepared according to literature procedure. Spectral data matched that reported in the literature.⁴

N-allyl-3-chlorobenzamide was prepared according to literature procedure. Spectral data matched that reported in the literature.2

*N-*allyl-4-methylbenzenesulfonamide was prepared according to literature procedure. Spectral data matched that reported in the literature.⁵

N-allyl-3-phenylpropanamide was prepared according to literature procedure. Spectral data matched that reported in the literature.⁶

*N-*allyl-4-(trifluoromethyl)benzenesulfonamide was prepared according to literature procedure. Spectral data matched that reported in the literature.⁷

N-allyl-4-iodobenzamide was prepared according to literature procedure.³ Spectral data matched that reported in the literature.⁸

N-allyl-3-methylbenzofuran-2-carboxamide was prepared according to literature procedure.³ ¹H NMR (600 MHz, Chloroform-*d*) δ 7.62 (dt, *J* = 7.9, 1.0 Hz, 1H), 7.47 – 7.42 (m, 1H), 7.42 (ddd, *J* = 8.3, 6.8, 1.2 Hz, 1H), 7.30 (ddd, *J* = 8.0, 6.8, 1.3 Hz, 1H), 6.72 (s, 1H), 5.96 (ddt, *J* = 17.0, 10.2, 5.7 Hz, 1H), 5.31 (dq, *J* = 17.1, 1.6 Hz, 1H), 5.21 (dq, *J* = 10.2, 1.4 Hz, 1H), 4.11 (tt, *J* = 5.8, 1.6 Hz, 2H), 2.64 (s, 2H). 13C NMR (151 MHz, Chloroform-*d*) δ 159.70, 152.79, 142.17, 133.58, 129.32, 126.56, 122.63, 122.04, 120.49, 116.26, 111.01, 40.93, 8.45. LCMS: found 216.1 (M+H), calc'd 216.1 (M+H).

N-allyl-4-methoxybenzamide was prepared according to literature procedure. Spectral data matched that reported.9

N-allyl-4-butylbenzamide was prepared according to literature procedure.³ ¹H NMR (600 MHz, Chloroform-*d*) δ 7.72 – 7.67 (m, 2H), 7.26 – 7.21 (m, 2H), 6.16 (s, 1H), 5.94 (ddtd, *J* = 16.9, 10.2, 5.7, 1.0 Hz, 1H), 5.26 (dp, *J* = 17.1, 1.4 Hz, 1H), 5.18 (dp, *J* = 10.1, 1.4 Hz, 1H), 4.09 (tq, *J* = 5.8, 1.5 Hz, 2H), 2.68 – 2.62 (m, 2H), 1.64 – 1.56 (m, 2H), 1.35 (dq, *J* = 14.7, 7.4 Hz, 2H), 0.92 (t, *J* = 7.4 Hz, 3H). 13C NMR (151 MHz, Chloroform-*d*) δ 146.44, 133.84, 126.44, 116.17, 41.92, 35.06, 32.89, 21.83, 13.45. LCMS: found 218.1 (M+H), calc'd 218.2 (M+H)

3. Olefin Hydrogenation

General procedure with AlCl₃: This reaction may be carried out under ambient temperature and atmosphere. The olefin (0.2 mmol) (if solid), Mn powder (33 mg, 0.3 mmol), $Co(OAc)_2 \cdot 4H_2O$ $(14.2 \text{ mg}, 30 \text{ mol})$ and AlCl_3 $(26.6 \text{ mg}, 0.2 \text{ mmol})$ are weighed out and added to a 5 mL onedram vial equipped with a stirring bar. A cap is placed on the reaction vessel and *i*-PrOH (2 mL) is added. The reaction is stirred vigorously for 4-20 h at rt. The reaction initially turns blue and over time turns black. The mixture is diluted with CH_2Cl_2 , poured into a separating funnel and quenched with saturated aqueous $NaHCO₃$ solution. After phase separation the aqueous phase is extracted twice further with CH_2Cl_2 . The combined organic phases are washed with brine, dried over MgSO4 and filtered through a short plug of silica gel. The solvent is removed *in vacuo* to furnish the hydrogenated product without the need for further purification.

General procedure with HCl: This reaction may be carried out under ambient temperature and atmosphere. A solution of concentrated (12.5 M) HCl is diluted with *i*-PrOH to approximate concentration 1–2M. 0.10 mL of this HCl/*i-*PrOH is diluted with DI H2O. The resulting solution is titrated with a freshly prepared solution of \sim 1M NaOH (aqueous) to determine concentration, using pH strips (Cytiva Whatman pH Indicator Papers, pH range 0-14) to monitor for equivalence point. The olefin (0.1 mmol) (if solid), Mn powder (16.5 mg, 0.3 mmol), and $Co(OAc)_{2} \cdot 4H_{2}O$ (7.5 mg, 30 mol%) are weighed out and added to a 5 mL one-dram vial equipped with a stirring bar. A cap is placed on the reaction vessel and *i*-PrOH (2 mL), followed by HCl/*i*-PrOH solution (2 equiv) is added. The reaction is stirred vigorously for 4-20 h at rt. The reaction initially turns blue and over time turns grey. The mixture is diluted with CH₂Cl₂, poured into a separating funnel and quenched with saturated aqueous NaHCO₃ solution. After phase separation the aqueous phase is extracted twice further with CH_2Cl_2 . The combined organic phases are washed with brine, dried over $MgSO_4$ and filtered through a short plug of silica gel. The solvent is removed *in vacuo* to furnish the hydrogenated product without the need for further purification.

4. Olefin Hydrogenation products

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\bigcap_{\mathsf{Me}}\mathsf{Me}
$$

Butylbenzene. The spectral data was in agreement with the literature.¹⁰

$$
\bigcap_{\text{MeO}}\underbrace{\qquad \qquad }^{\text{Me}}
$$

1-methoxy-4-propylbenzene. The spectral data was in agreement with the literature.¹¹

Propoxybenzene. The spectral data was in agreement with the literature.¹²

3-methyl-*N*-propylbenzofuran-2-carboxamide. 1 H NMR (400 MHz) CDCl3: 7.64 (d, *J* = 7.8 Hz, 1H, Ar-H), 7.49-7.40 (m, 2H, Ar-H), 7.35-7.29 (m, 1H, Ar-H), 3.46 (app-q, 2H, CH2), 1.70 (apph, 2H, CH2), 1.04 (t, *J* = 7.4 Hz, 3H, CH3). 13C NMR (151 MHz, Chloroform-*d*) δ 159.70, 152.79, 142.17, 133.58, 129.32, 126.56, 122.63, 122.04, 120.49, 116.26, 111.01, 40.93, 8.45. LCMS: found 216.1 (M+H), Calc'd 216.1 (M+H)

4-bromo-*N*-propylbenzamide. The spectral data was in agreement with the literature.¹³

4-methyl-*N*-propylbenzamide. The spectral data was in agreement with the literature.14

2-methoxy-*N*-propylbenzamide. The spectral data was in agreement with the literature.15

3-chloro-*N*-propylbenzamide. The spectral data was in agreement with the literature.16

4-methyl-*N*-propylbenzenesulfonamide. The spectral data was in agreement with the literature.¹⁷

3-Phenyl-*N*-propylpropanamide. The spectral data was in agreement with the literature.18

N-propyl-4-(trifluoromethyl)benzenesulfonamide. ¹H NMR (400 MHz) CDCl₃: 8.03 (d, *J* = 8.2 Hz, 2H, Ar-H), 7.81 (d, *J* = 8.2 Hz, 2H, Ar-H), 4.95 (t, *J* = 5.9 Hz, 1H, NH), 2.97 (app-q, 2H, CH2), 1.52 (app-h, 2H, CH2), 0.90 (t, *J* = 7.4 Hz, 3H, CH3). 13C NMR (151 MHz, Chloroform-*d*) δ 143.30, 127.09, 125.85, 125.82, 125.80, 44.60, 22.59, 10.58. LCMS: found 268.0 (M+H), 268.1 calc'd (M+H).

*N-*propylbenzamide. The spectral data was in agreement with the literature. 19

4-Fluoro-*N*-propylbenzamide. The spectral data was in agreement with the literature.²⁰

4-iodo-*N*-propylbenzamide. The spectral data was in agreement with the literature. 21

4-methoxy-*N*-propylbenzenesulfonamide. The spectral data was in agreement with the literature.²²

4-butyl-*N*-propylbenzamide. ¹ H NMR (600 MHz, Chloroform-*d*) δ 7.70 – 7.64 (m, 2H), 7.26 – 7.21 (m, 2H), 6.07 (s, 1H), 3.42 (ddd, *J* = 7.8, 7.1, 5.9 Hz, 2H), 2.67 – 2.62 (m, 2H), 1.67 – 1.58 (m, 4H), 1.35 (dt, *J* = 14.9, 7.4 Hz, 2H), 0.99 (t, *J* = 7.4 Hz, 3H), 0.92 (t, *J* = 7.4 Hz, 3H). 13C NMR (151 MHz, Chloroform-*d*) δ 167.07, 146.19, 131.78, 128.12, 126.36, 41.24, 35.04, 32.91, 22.53, 21.83, 13.45, 10.99. LCMS: found 220.1 (M+H), calc'd 220.2 (M+H)

5. Unsuccessful Substrates

6. Mechanistic Studies

A. Cyclization study: This reaction may be carried out under ambient temperature and atmosphere. Mn powder (16.5 mg, 0.3 mmol), Co(OAc)₂•4H₂O (6.5 mg, 30 mol%) and AlCl₃ (13.3 mg, 0.1 mmol) are weighed out and added to a 5 mL one-dram vial equipped with a stirring bar. A septum cap is placed on the reaction vessel and *i*-PrOH (1 mL) is added. Allyl ether (12 µL, 0.1 mmol) is added via syringe through septum. The reaction is stirred vigorously for 4 h at rt. 10 µL of reaction solution is diluted with 2 mL EtOAc and filtered through a short plug of silica into a 2 mL vial. The sample is submitted for GC/MS analysis. Only propyl ether is observed, no cyclized product.

B. Reaction purge with argon: Mn powder (16.5 mg, 0.3 mmol), $Co(OAc)_{2} \cdot 4H_{2}O$ (6.5 mg, 30) mol%) and AlCl₃ (13.3 mg, 0.1 mmol) are weighed out and added to a 5 mL one-dram vial equipped with a stirring bar. A septum cap is placed on the reaction vessel and *i*-PrOH (1 mL) is added. Argon is bubbled through the solution slightly above atmospheric pressure (via balloon) for five minutes. 4-phenylbutene (15 μ L, 0.1 mmol) is added via syringe through septum. The reaction is stirred vigorously for 24 h at rt. 10 µL of reaction solution is diluted with 2 mL EtOAc and filtered through a short plug of silica. The sample is submitted for GC/MS analysis. Butylbenzene is detected. No starting material is observed in the chromatogram.

C. Catalytic Mn with hydrogen atmosphere. *N*-allyl-4-bromobenzamide (24 mg, 0.1 mmol), Mn powder (0.6 mg, 0.01 mmol), Co(OAc)2•4H2O (6.5 mg, 30 mol%) and AlCl3 (13.3 mg, 0.1 mmol) are weighed out and added to a 5 mL one-dram vial equipped with a stirring bar. A septum cap is placed on the reaction vessel and *i*-PrOH (1 mL) is added. Hydrogen is bubbled through the solution slightly above atmospheric pressure (via balloon) for five minutes. The hydrogen balloon is removed from solution but left inserted through septum to maintain a positive pressure of hydrogen above the reacting solution. The reaction is stirred vigorously for 24 h at rt. 10 µL of reaction solution is diluted with 2 mL MeOH and filtered through a short plug of silica. The sample is submitted for LC/MS analysis. No *N*-propyl-4-bromobenzamide is detected.

D. MnCl₂ and PhSiH₃: This reaction may be carried out under ambient temperature and atmosphere. *N*-allyl-4-bromobenzamide (24 mg, 0.1 mmol), MnCl₂•4H₂O powder (59.1 mg, 0.3 mmol), $Co(OAc)₂•4H₂O (6.5 mg, 30 mol%)$ are weighed out and added to a 5 mL one-dram vial equipped with a stirring bar. A septum cap is placed on the reaction vessel and *i*-PrOH (1 mL) is added. PhSiH₃ is added The reaction is stirred vigorously for 24 h at rt. 10 μ L of reaction solution is diluted with 2 mL MeOH and filtered through a short plug of silica. The sample is submitted for LC/MS analysis. No *N*-propyl-4-bromobenzamide is detected.

7. ¹H and ¹³C NMR Spectra of Novel Compounds

8. References

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