# **Supporting Information**

# Difluoromethyl-1,3,4-oxadiazoles are selective, mechanism-based, and essentially irreversible inhibitors of histone deacetylase 6

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#### 1 Supplementary Schemes, Figures, Equations and Tables



Scheme S1. Synthesis of HDAC6 inhibitor fragments. a) i: NaN<sub>3</sub>, NH<sub>4</sub>Cl, LiCl·H<sub>2</sub>O, DMF, 100 °C, 150 W, 24 h; ii: DFAA, DCM, rt., 24 h (1, 3); b) i: NaN<sub>3</sub>, NH<sub>4</sub>Cl, LiCl, DMF, 100 °C, 18 h; ii: difluoroacetic anhydride (DFAA), toluene, 70 °C, 18 h (2).



**Scheme S2.** Synthesis of the acylhydrazide **13**. a) Methyl 2-chlorpyrimidine-5-carboxylate, DIPEA, EtOH, 90 °C, 18 h; b) i: hydrazine monohydrate, MeOH, 70 °C, 3 h; ii: DFAA, DMF, 70 °C, 1 h.



Scheme S3. Synthesis of the trifluoromethyl-1,3,4-oxadiazole (17) and methyl-1,3,4-oxadiazole (15) analogs. a) i: NaN<sub>3</sub>, NH<sub>4</sub>Cl, LiCl, DMF, 100 °C, 18 h; ii: trifluoroacetic anhydride, toluene, 70 °C, 18 h (14); b) i: NaN<sub>3</sub>, NH<sub>4</sub>Cl, LiCl, DMF, 100 °C, 18 h; ii: acetic anhydride, toluene, 70 °C, 18 h; iii: K<sub>2</sub>CO<sub>3</sub>, MeOH/H<sub>2</sub>O (15).



**Scheme S4.** Synthesis of hydrazide **14** and monofluoromethyl-1,3,4-oxadiazole **16**. a) hydrazine monohydrate, MeOH, 70 °C, 3 h; b) i: monofluroacetic acid, DMF, 70 °C, 3 h; ii: Burgess reagent, THF, 60 °C, 18 h.



**Figure S1.** Representative examples of different kinetic mechanisms of enzyme inhibition, including the relationships between the respective association and dissociation rate constants (e. g.,  $k_1 \& k_{-1}$ ) and the related equilibrium dissociation constant  $K_i$ . **A**) Fast-on/fast-off binding kinetics. For competitive fast-on/fast-off inhibitors the half maximum inhibitory concentration (IC<sub>50</sub>) and the  $K_i$  are directly related by the Cheng-Prusoff equation<sup>1</sup>; **B**) slow-binding Mechanism I: single-step slow binding,  $k_1 \& k_{-1}$  are inherently slow; **C**) slow-binding Mechanism II: two-step slow binding. Initially, inhibitor and enzyme form an encounter complex [EI] that subsequently slowly undergoes isomerization to a binary enzyme inhibitor complex [E\*I].<sup>2</sup>



**Figure S2.** Michaelis-Menten constant  $K_M$  determination for HDAC6 using a series of substrate concentrations. Steady-state velocities  $[\mu M^*s^{-1}]$  (mean  $\pm$  SD) were plotted against the corresponding substrate concentrations  $[\mu M]$  and fitted to the Michaelis-Menten equation yielding the Michaelis-Menten constant:  $K_M$  HDAC6 = 19.27  $\mu$ M. Experiment was performed in triplicates.



**Figure S3.** Quantified hydrolysis products from LC-UV-MS analysis after over night incubation of the respective compound (100  $\mu$ M) with HDAC6. Experiments were performed in triplicates. DFMO: difluoromethyl-1,3,4-oxadiazole; TFMO: trifluoromethyl-1,3,4-oxadiazole. n.d.: not determined.



**Figure S4**. Representative UV and related mass traces from two independent LC-UV-MS experiments. **A**: Compound **6** was incubated with HDAC6 overnight; **B**: Compound **17** was incubated with HDAC6 overnight; x axis: retention time in mins (chromatogram), m/z ration (mass spectras), y axis: intensity in Absorbance Units (AU). Experiments were performed in triplicates.



**Figure S5**. Representative UV and related mass traces of LC-UV-MS experiments. **A**: Compound **6** was incubated with HDAC6 overnight in  $H_2^{16}O$  water; **B**: Compound **6** incubated with HDAC6 overnight in  $H_2^{18}O$  water; x axis: retention time in mins (chromatogram), m/z ration (mass spectras), y axis: intensity in Absorbance Units (AU). Experiments were performed in triplicates.



HDACC	Hydrolysis products from compound 6									
HDACO	Acylhydrazide 13	Hydrazide 14								
WT	22.5 ± 1.11 μM	9.0 ± 1.3 μM								
H573A	10.2 ± 3.2 μM	n.o.								
H574A	6.9 ± 2.3 μM	n.o.								
Y745F	14.9 ± 1.9 μM	26.2 ± 6.8 μM								
	1	1								

HDACE	Hydrolysis products from compound 17								
HDAC6	Acylhydrazide 20	Hydrazide 14							
WT	n.d.	18.6 μM ± 1.2 μM							
H573A	n.d.	n.o.							
H574A	n.d.	n.o.							
Y745F	n.d.	64.4 ± 8.9 μM							

**Figure S6.** Quantified hydrolysis products from LC-UV-MS analysis after over night incubation of the respective compound (100  $\mu$ M) with various HDAC6 mutants (wild-type (WT), H573A, H574A, Y745F). Experiments were performed in triplicates. n.d.: not determined; n.o.: not observed.

$$[P] = v_{ss}t + \frac{v_{in} - v_{ss}}{k_{obs}} (1 - e^{-k_{obs}t})$$
(Eq. 1)

**Equation 1.** Time-dependent product formation for inhibitors showing slow-binding Mechanism I&II. [P]: amount of generated AMC;  $v_{ss}$ : steady-state velocity (product formation); t: time;  $v_{in}$ : initial velocity (product formation);  $k_{obs}$ : apparent first-order rate constant for the conversion from  $v_{in}$  to  $v_{ss}$ .

$$k_{obs} = k_{-1} + k_1 \left( 1 + \frac{[S]}{K_M} \right) [I]$$
 (Eq. 2)

**Equation 2.** The single-step slow-binding Mechanism I results in a linear relationship between  $k_{obs}$  and inhibitor concentration.  $k_{-1}$ : dissociation rate constant;  $k_1$ : association rate constant; [S]: substrate concentration;  $K_M$ : Michelis-Menten constant; [I]: inhibitor concentration.

$$k_{obs} = k_{-2} + \frac{k_2}{[I] + K_{i,1} \left(1 + \frac{[S]}{K_M}\right)} [I]$$
(Eq. 3)

**Equation 3.** The two-step slow-binding Mechanism II results in a hyperbolic relationship between  $k_{obs}$  and inhibitor concentration.  $k_{-2}$ : secondary dissociation rate constant;  $k_2$ : secondary association rate constant;  $K_{i,1}$ : equilibrium dissociation constant of the enzyme inhibitor encounter complex [EI].

	HDAC6 CD2– <b>13</b> Complex
Space group	$P2_{1}2_{1}2_{1}$
a,b,c (Å)	74.60, 92.30, 96.60
α, β, γ (°)	90.00, 90.00, 90.00
R <sub>merge</sub> <sup>b</sup>	0.210 (0.706)
$\mathbf{R}_{\mathrm{pim}}^{\mathrm{c}}$	0.084(0.289)
$CC_{1/2}^{d}$	0.993(0.851)
Redundancy	1.9
Completeness (%)	99.5(94.3)
I/σ	7.3(2.5)
Re	efinement
Resolution (Å)	36.216-2.00 (2.07-2.00)
No. reflections	90179 (8857)
R <sub>work</sub> /R <sub>free</sub> <sup>e</sup>	0.185/0.223
	(0.228/0.0.266)
Numb	per of Atoms <sup>f</sup>
Protein	5469
Ligand	52
Solvent	424
Average	e B factors (Å <sup>2</sup> )
Protein	15
Ligand	20
Solvent	20
RMS	5 Deviations
Bond lengths (Å)	0.03
Bond angles (°)	1.4
Ramac	chandran Plot <sup>g</sup>
Favored	97.01
Allowed	2.99
Outliers	0.00

Table S1: Data collection	and refinement	statistics
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<sup>a</sup>Values in parentheses refer to the highestresolution shell of data.

<sup>b</sup>R<sub>merge</sub> =  $\sum_{h}\sum_{i}/I_{i,h} - \langle I \rangle_{h} | / \sum_{h}\sum_{i}I_{i,h}$ , where  $\langle I \rangle_{h}$  is the average intensity calculated for reflection *h* from *i* replicate measurements.

<sup>c</sup>**R**<sub>p.i.m.</sub> =  $(\sum_{h}(1/(N-1))^{1/2}\sum_{i}|\mathbf{I}_{i,h}-\langle \mathbf{I}\rangle_{h}|)/\sum_{h}\sum_{i}\mathbf{I}_{i,h}$ , where N is the number of reflections and  $\langle \mathbf{I}\rangle_{h}$  is the average intensity calculated for reflection *h* from replicate measurements.

<sup>d</sup>Pearson correlation coefficient between random half-datasets.

 ${}^{e}R_{work} = \sum ||F_o| - |F_c|| / \sum |F_o|$  for reflections contained in the working set.  $|F_o|$  and  $|F_c|$  are the observed and calculated structure factor amplitudes, respectively.  $R_{free}$  is calculated using the same expression for reflections contained in the test set held aside during refinement.

<sup>f</sup>Per asymmetric unit.

<sup>g</sup>Calculated with MolProbity.

#### 2 NMR Data of synthesized compounds

<sup>1</sup>H NMR spectrum of 1 (300 MHz, CDCl<sub>3</sub>)



# <sup>19</sup>F NMR spectrum of 1 (377 MHz, CDCl<sub>3</sub>)

70 60 50 40 30 20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210 -220 -230 -240 -250 -250 -270

-127.0 -127.1

# <sup>1</sup>**H NMR** spectrum of **2** (600 MHz, DMSO- $d_6$ )



<sup>19</sup>F NMR spectrum of 2 (565 MHz, DMSO- $d_6$ )



10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210

# <sup>1</sup>H NMR spectrum of **3** (400 MHz, CDCl<sub>3</sub>)





<sup>19</sup>F NMR spectrum of **3** (377 MHz, CDCl<sub>3</sub>)



	_				_	_		 					 _	 			· · ·		-					 _				 _				
70	60	50	40	30	20	10	0	-20	-4	ю	-	50	-80	-100	-1	20		140		-16	0	- 1	180	-200	)	-2	20	-240	)	-2	260	

# <sup>1</sup>**H NMR** spectrum of **4** (600 MHz, DMSO- $d_6$ )



# <sup>1</sup>H NMR spectrum of **5** (600 MHz, CDCl<sub>3</sub>)



# <sup>1</sup>H NMR spectrum of 6 (600 MHz, DMSO-*d*<sub>6</sub>)



<sup>19</sup>F NMR spectrum of 6 (565 MHz, DMSO-*d*<sub>6</sub>)

 $<^{^{121.1}}_{^{-121.2}}$ 

10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210

# <sup>1</sup>**H NMR** spectrum of **7** (600 MHz, DMSO- $d_6$ )



<sup>19</sup>**F NMR** spectrum of **7** (565 MHz, DMSO-*d*<sub>6</sub>)

 $<^{^{121.1}}_{^{-121.2}}$ 

10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210

# <sup>1</sup>**H NMR** spectrum of **8** (600 MHz, DMSO-*d*<sub>6</sub>)



<sup>1</sup>**H NMR** spectrum of **9** (600 MHz, DMSO-*d*<sub>6</sub>)



<sup>19</sup>F NMR spectrum of 9 (565 MHz, DMSO-*d*<sub>6</sub>)



#### 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210

# <sup>1</sup>H NMR spectrum of **10** (600 MHz, CDCl<sub>3</sub>)







10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210

# <sup>1</sup>**H NMR** spectrum of **12** (600 MHz, DMSO-*d*<sub>6</sub>)



<sup>19</sup>F NMR spectrum of **12** (565 MHz, DMSO-*d*<sub>6</sub>)



10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210

<sup>1</sup>**H NMR** spectrum of **13** (600 MHz, DMSO-*d*<sub>6</sub>)



<sup>19</sup>F NMR spectrum of **13** (565 MHz, DMSO-*d*<sub>6</sub>)

10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210

 $<^{^{-127.1}}_{^{-127.2}}$ 

# <sup>1</sup>**H NMR** spectrum of **14** (600 MHz, DMSO-*d*<sub>6</sub>)





#### <sup>1</sup>**H NMR** spectrum of **15** (600 MHz, DMSO-*d*<sub>6</sub>)



<sup>13</sup>C NMR spectrum of 15 (126 MHz, DMSO-*d*<sub>6</sub>)



# <sup>1</sup>H NMR spectrum of 16 (500 MHz, DMSO-*d*<sub>6</sub>)





<sup>19</sup>**F NMR** spectrum of **16** (471 MHz, DMSO-*d*<sub>6</sub>)

 60
 50
 40
 30
 20
 10
 0
 -10
 -20
 -30
 -40
 -50
 -60
 -70
 -80
 -90
 -100
 -110
 -120
 -130
 -140
 -150
 -160
 -17

#### <sup>1</sup>**H NMR** spectrum of **17** (600 MHz, DMSO-*d*<sub>6</sub>)



<sup>19</sup>F NMR spectrum of **17** (565 MHz, DMSO-*d*<sub>6</sub>)

10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 -210

# <sup>1</sup>**H NMR** spectrum of **18** (600 MHz, DMSO-*d*<sub>6</sub>)



# <sup>1</sup>H NMR spectrum of **19** (600 MHz, CDCl<sub>3</sub>)



#### **3 HPLC Chromatograms**

HPLC chromatogram of **6**.



HPLC chromatogram of 9.





HPLC chromatogram of 12.





HPLC chromatogram of 14.





HPLC chromatogram of 16.







#### **4** References

- Yung-Chi, C.; Prusoff, W. H. Relationship between the Inhibition Constant (KI) and the Concentration of Inhibitor which causes 50 per cent Inhibition (I50) of an Enzymatic Reaction. *Biochem. Pharmacol.* 1973, 22, 3099–3108.
- Schäker-Hübner, L.; Haschemi, R.; Büch, T.; Kraft, F. B.; Brumme, B.; Schöler, A.; Jenke, R.; Meiler, J.; Aigner, A.; Bendas, G.; Hansen, F. K. Balancing Histone Deacetylase (HDAC) Inhibition and Drug-Likeness: Biological and Physicochemical Evaluation of Class I Selective HDAC Inhibitors. *ChemMedChem* 2022, *17*, e202100755.