# **Science Advances NAAAS**

# Supplementary Materials for

### **A 10-micrometer-thick nanomesh-reinforced gas-permeable hydrogel skin sensor for long-term electrophysiological monitoring**

Zongman Zhang *et al.*

Corresponding author: Chunya Wang, chunyawang@cup.edu.cn; Takao Someya, someya@ee.t.u-tokyo.ac.jp; Yan Wang, yan.wang@gtiit.edu.cn

> *Sci. Adv.* **10**, eadj5389 (2024) DOI: 10.1126/sciadv.adj5389

#### **The PDF file includes:**

Figs. S1 to S30 Tables S1 and S2 Legends for movies S1 to S6

#### **Other Supplementary Material for this manuscript includes the following:**

Movies S1 to S6



**Fig. S1. The morphologies of PU nanomeshes with different densities and diameter**  distribution. (A to C) PU nanomeshes with densities of (A) 0.3, (B) 0.6, and (C) 1.1 mg cm-2, respectively. Scale bar, 20 μm. **(D)** The statistical data of the diameter distribution obtained from SEM images using image analysis software (image J).  $N = 50$ . The average diameter of PU nanomeshes is  $0.9 \pm 0.1$  µm.



**Fig. S2. Schematic illustration of the fabrication process of the ultrathin hydrogels.** 

The ultrathin-reinforced hydrogel was achieved by dip-coating electrospun nanomeshes (density of 0.3 mg cm<sup>-2</sup>) into a diluted gelatin hydrogel solution at a high temperature (55– 75°C) and then gelating at a physiological temperature or under ambient environment.



**Fig. S3. Mechanical robustness of the ultrathin hydrogels.** The ultrathin hydrogel could support water with a volume of 31.0 mL. Scale bar, 1 cm.

In this experiment, the ultrathin hydrogel was fixed on a circular PET window with a diameter of 4 cm, and the water was dripped into the suspended ultrathin hydrogel dropwise.



**Fig. S4. Tensile test of PU nanomesh-reinforced hydrogels.** Tensile stress curves of the PU nanomesh-reinforced hydrogels with different borax content, the insert photographs showing the initial and stretched PU nanomesh-reinforced hydrogels with 1.5 g borax. Scale bar, 1 cm.



**Fig. S5. Influence of glycerol content on anti-drying property and intrinsic impedance of PU nanomesh-reinforced hydrogels. (A)** A photograph of PU nanomeshreinforced hydrogels with glycerol content of 5.0 g (PU-GG5.0BWNa) after being stored for 24 h under an ambient environment. Scale bar, 1 cm. **(B)** Microscope image of PU nanomesh-reinforced hydrogels obtained from (A), showing Na2SO4 salted out. Scale bar, 1 cm. **(C and D)** Photographs of PU nanomesh-reinforced hydrogels with glycerol content of (C) 8.0 g (PU-GG $_{8.0}$ BWNa) and (D) 13.0 g (PU-GG $_{13.0}$ BWNa) after being stored for 24 h, respectively. Scale bar, 1 cm. **(E and F)** Intrinsic impedance characterizations of PU nanomesh-reinforced hydrogels with different glycerol content at (E) initial state and (F) after 24 h. Fiber density: 0.3 mg cm-2.

There is a trade-off effect between anti-drying capability and intrinsic impedance. The anti-drying performance was enhanced with the increase of glycerol content, whereas resulting in higher intrinsic impedance. Large amounts of Na2SO4 salted out in PU-GG5.0BWNa after 24 h due to low glycerol content. Therefore, PU-GG8.0BWNa was chosen for further experiments.



**Fig. S6. Influence of Na2SO4 content on the intrinsic impedance of PU nanomeshreinforced hydrogels. (A)** The intrinsic impedance curves of PU nanomesh-reinforced hydrogels with different Na2SO4 content. **(B to D)** Photographs of the PU nanomeshreinforced hydrogels with Na2SO4 content of (B) 0.8 g, (C) 1.3 g, (D) 2.0 g after being stored for 12 h under an ambient environment, respectively. Scale bar, 1 cm. Fiber density:  $0.3 \text{ mg cm}^{-2}$ .

The intrinsic impedance of PU nanomesh-reinforced hydrogels decreased with the increase of Na2SO4 content (Fig. S6A). However, as the Na2SO4 content increased to 2.0 g, a large amount of Na2SO4 dissolved out from PU-GGBWNa2.0 after being stored for 12 h (Fig. S6B to S6D). Therefore, PU-GGBWNa1.3 was chosen for further experiments.



**Fig. S7. Rheological test. (A)** Rheological characterization of hydrogels solution with different recipes at a temperature sweep from 10 to 75 ℃, showing their transition between viscoelastic gel state and viscous liquid state. **(B)** The corresponding tanδ (G″/G′) calculated from (A).



**Fig. S8. A photograph demonstrating the large-scale fabrication capacity of the ultrathin hydrogels**. Scale bar, 1 cm.



**Fig. S9. Comparison of WVTR for PU nanomesh-reinforced hydrogels with different thickness.** Fiber density: 0.3 mg cm<sup>-2</sup>. Error bars represent the SD of the measured values (n = 3).



**Fig. S10. Anti-drying performance of the ultrathin hydrogels without glycerol.**



**Fig. S11. Cross-sectional SEM images of PU nanomesh-reinforced hydrogels. (A)** PU0.6-GGBWNa. **(B)** PU1.1-GGBWNa. Scale bar, 20 μm.



**Fig. S12. Schematic illustration of structural changes of the ultrathin hydrogels during reversible stretching.** 



Fig. S13. Tensile stress curves of PU nanomesh-reinforced hydrogels. (A) PU<sub>0.3</sub>-GGBWNa. **(B)** PU0.6-GGBWNa. **(C)** PU1.1-GGBWNa.



**Fig. S14. Mechanical properties of hydrogels of PU nanomesh reinforced hydrogels**  with different thicknesses. (A) Tensile stress curves of PU nanomesh-reinforced hydrogels with different thicknesses. **(B)** Comparison of Young's modulus and toughness for PU nanomesh-reinforced hydrogels from A. Fiber density: 0.3 mg cm<sup>-2</sup>. Error bars represent the SD of the measured values ( $n = 3$ ).



**Fig. S15. Comparison of adhesion performance of reinforced hydrogels with different thicknesses and reinforced PDMS films. (A)** Force stroke curves of the reinforced hydrogel with different thicknesses and reinforced PDMS films. **(B)** Comparison of the area adhesion energy of the reinforced hydrogel with different thicknesses reinforced PDMS films. Fiber density: 0.3 mg cm-2. Error bars represent the SD of the measured values  $(n = 3)$ .



**Fig. S16. Photographs of the ultrathin hydrogels attached to different substrates. (A)**  Attached to rubber glove. **(B)** Attached to metal. **(C)** Attached to glass. **(D)** Attached to paper. Scale bar, 1 cm.



**Fig. S17. Optical microscopic image of the peeled-off ultrathin hydrogel surface from the human skin**. Scale bar, 500 μm.



**Fig. S18. High skin conformability of the ultrathin hydrogels.** The ultrathin hydrogel moves along with natural skin motions during compressing and stretching. Scale bar, 1 cm.



**Fig. S19. Adhesion performance of the ultrathin hydrogels and commercial baby-safe ECG gels. (A)** Force stroke curves of the ultrathin hydrogels and commercial baby-safe ECG gels. **(B)** Comparison of the area adhesion energy and thickness of the ultrathin hydrogels and commercial baby-safe ECG gels. Error bars represent the SD of the measured values  $(n = 3)$ .



**Fig. S20. Evaluation of negative sensations/feelings** (e.g., skin irritation, itches, and inflammation) of the ultrathin hydrogel after 24 h of continuous wearing and peeling off. Photographs of **(A)** pristine human skin, **(B)** hydrogel sensor attached to the skin, **(C)** hydrogel attached to the skin for 1 day, and **(D)** after removal.



**Fig. S21. Skin contact impedance analysis during 24 h of continuous attachment.**



**Fig. S22. Photographs of the measurement setup. (A)** A photograph demonstrating the wireless module and the design of the ultrathin hydrogel electrode. **(B)** A photograph of the measuring set attached on the human chest. The white dotted line demonstrates the ultrathin hydrogel electrodes. Circular-shaped hydrogel sensors were connected to the contacts of the wireless module that was fixed on the subject's chest. Scale bar, 1 cm.



**Fig. S23. ECG and PPG signals recorded by the ultrathin hydrogels and commercial PPG ring.** 



**Fig. S24. Wireless ECG recording before (top) and after (bottom) strenuous exercise captured by the ultrathin hydrogels.** 



**Fig. S25. The SNR of wireless ECG signals continuously monitored by the ultrathin hydrogels for 8 days.** 



**Fig. S26. Long-term skin contact impedance and wireless ECG signals monitoring by the ultrathin hydrogels over 12-day continuous wear under daily life conditions. (A)** Skin contact impedance results over 12 days. **(B)** Skin contact impedance at 100 Hz over 12 days. **(C)** ECG signals recorded by the ultrathin hydrogels over 12 days.



**Fig. S27. 21-day ECG signals monitoring by the ultrathin hydrogels. (A)** ECG signals recorded by the ultrathin hydrogels after storage for 1, 3, 7. 14, and 21 days, respectively. **(B)** High and stable SNR values by the ultrathin hydrogels after various storage days.



**Fig. S28. Long-term, continuous, high-quality EMG monitoring by the ultrathin hydrogels. (A)** Schematic illustration of the experimental setup for measuring EMG signals. **(B)** Comparison of EMG signals recorded by commercial gels and ultrathin hydrogels when the subject opened and clenched fist at the initial state and after 24 h of continuous wearing.



**Fig. S29. Strain sensing applications by the ultrathin hydrogels. (A)** Relative resistance variation versus applied strain. The slope of the fitted solid line represents the sensitivity of the ultrathin hydrogel. **(B)** Resistance response curves with stepwise increasing tensile strain from 50 to 300% strain. **(C)** Durability test of cyclic stretching/releasing for 500 cycles at 60% strain. **(D)** The response time of the ultrathin hydrogel. **(E)** Relative resistance changes of the ultrathin hydrogel upon bending finger to different angles.

Meanwhile, the ultrathin hydrogel could also be utilized as an on-skin strain gauge. A distinctly nonlinear segmentation trend of the resistance-strain curve is displayed in fig. S29A, suggesting a sensitivity of 2.5 and 4.3 under 0–200% and 200–350% strain, respectively, which are sufficiently large to realize the detection of full body range of motions with high resolution. Fig. S29B presents reproducible dynamic electrical signals during 50–300% strain. The durability test of 500 cyclic stretching and releasing under 60% strain is demonstrated in fig. S29C. Additionally, the ultrathin hydrogel strain gauge exhibits a response time of 0.77 s and a recovery time of 0.75 s (fig. S29D). Finally, realtime finger bending monitoring was demonstrated by the ultrathin hydrogel strain gauge (fig. S29E).



**Fig. S30. Photographs demonstrating easy removal of the hydrogel materials from the hairy scalp, taking advantage of the temperature-controlled phase change property. (A)** Hydrogels painted on the hairy scalp in a gel state. **(B)** Hydrogel turned into a liquid state after being heated by a hair dryer. **(C)** Clean hairy scalp after hydrogel removal. Scale bar, 1 cm.

**Table S1. Comparison of the ultrathin hydrogel with recently reported gaspermeable and/or anti-drying hydrogels in skin bioelectronics on various features, such as thickness, gas permeability, anti-drying, skin adhesion, conductivity, application, and skin-wear period for long-term, continuous health monitoring under daily life conditions.** 



	<b>PU</b> nanomeshes (Density)	Gelatin	Sodium sulfate	<b>Borax</b>	Glycerol	Water
<b>GWNa</b>		2.5 g	1.3 <sub>g</sub>	$\sqrt{2}$	$\sqrt{2}$	$28.0$ mL
<b>GGWNa</b>	7	2.5 g	1.3 <sub>g</sub>	$\sqrt{2}$	8.0 g	28.0 mL
<b>GGBWNa</b>	$\overline{1}$	2.5 g	1.3 g	1.5 <sub>g</sub>	8.0 g	$28.0$ mL
$PU-GG5.0WBNa$	$0.3 \text{ mg/cm}^2$	2.5 g	1.3 <sub>g</sub>	1.5 <sub>g</sub>	5.0 g	28.0 mL
PU-GG <sub>8.0</sub> WBNa	$0.3 \text{ mg/cm}^2$	2.5 g	1.3 <sub>g</sub>	1.5 <sub>g</sub>	8.0 g	$28.0$ mL
$PU-GG13.0WBNa$	$0.3 \text{ mg/cm}^2$	2.5 g	1.3 <sub>g</sub>	1.5 <sub>g</sub>	13.0 g	$28.0$ mL
$PU-GGWB1.0Na$	$0.3 \text{ mg/cm}^2$	2.5 g	1.3 <sub>g</sub>	1.0 <sub>g</sub>	8.0 g	$28.0$ mL
$PU-GGWB_{1.5}Na$	$0.3 \text{ mg/cm}^2$	2.5 g	1.3 g	1.5 <sub>g</sub>	8.0 g	$28.0$ mL
$PU-GGWB2.0Na$	$0.3 \text{ mg/cm}^2$	2.5 g	1.3 <sub>g</sub>	2.0 g	8.0 g	$28.0$ mL
$PU-GGWBNa_{0.8}$	$0.3 \text{ mg/cm}^2$	2.5 g	0.8 g	1.0 <sub>g</sub>	8.0 g	$28.0$ mL
PU-GGWBNa <sub>1.3</sub>	$0.3 \text{ mg/cm}^2$	2.5 g	1.3 <sub>g</sub>	1.5 <sub>g</sub>	8.0 g	$28.0$ mL
PU-GGWBNa2.0	$0.3 \text{ mg/cm}^2$	2.5 g	2.0 g	2.0 g	8.0 g	$28.0$ mL
$PU0.3$ -GGWBNa	$0.3 \text{ mg/cm}^2$	2.5 g	1.3 <sub>g</sub>	1.5 <sub>g</sub>	8.0 g	28.0 mL
$PU0.6$ -GGWBNa	$0.6 \text{ mg/cm}^2$	2.5 g	1.3 <sub>g</sub>	1.5 <sub>g</sub>	8.0 g	$28.0$ mL
$PU1.1$ -GGWBNa	1.1 mg/cm <sup>2</sup>	2.5 g	1.3 <sub>g</sub>	1.5 <sub>g</sub>	8.0 <sub>g</sub>	28.0 mL

**Table S2. Hydrogels recipes with varying components.**

### **Movie captions:**

- **Movie S1.** Ultrathin hydrogel supporting liquid.
- **Movie S2.** Dynamic adhesion to the human skin.
- **Movie S3.** Tensile test.
- **Movie S4.** Adhesion evaluation test.
- **Movie S5.** Dynamic compliance to the human skin.
- **Movie S6.** Application of electrical muscle stimulation.