Supplementary Information

Tuning oxidant and antioxidant activities of ceria by anchoring copper single-site for antibacterial application

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and Yuguang Wang^{2*}

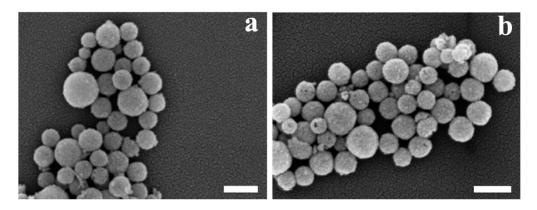
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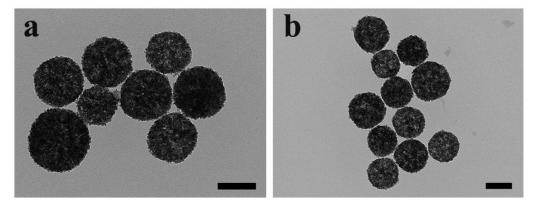
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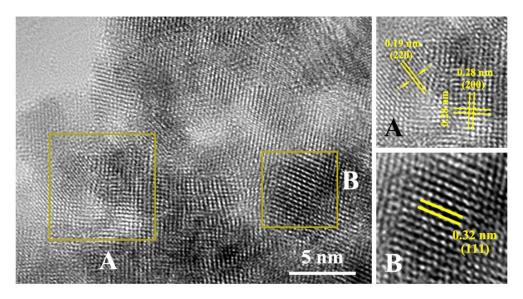
wangyuguang@bjmu.edu.cn.



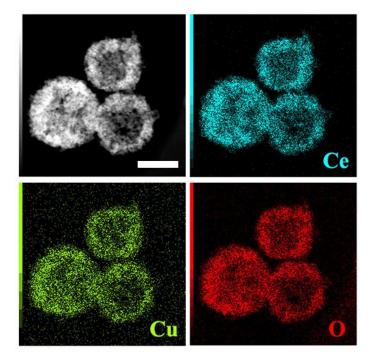
Supplementary Fig. 1. Morphology characterization of CeO₂ and Cu-CeO₂-air catalyst. Representative SEM images of (a) CeO₂ and (b) Cu-CeO₂-air catalyst (scale bar: 200 nm). A representative image of three replicates from each group is shown.



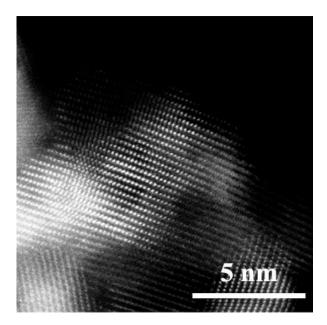
Supplementary Fig. 2. Morphology characterization of Cu-CeO₂ precursor and Cu-CeO₂-air catalyst. Representative TEM images of (a) Cu-CeO₂ precursor and (b) Cu-CeO₂-air catalyst (scale bar: 100 nm). The experiments were repeated three times with similar results.



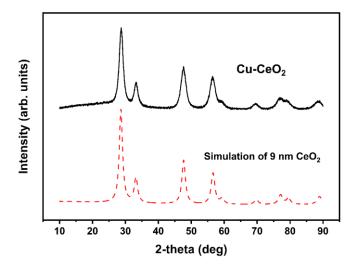
Supplementary Fig. 3. HRTEM images of Cu-CeO₂ precursor. A representative image of three replicates from each group is shown.



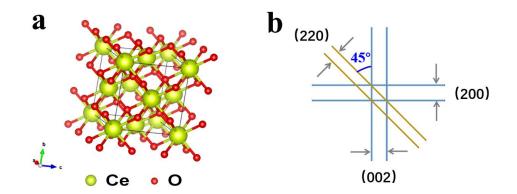
Supplementary Fig. 4. The elemental composition imaging analysis of Cu-CeO₂ precursor (scale bar: 50 nm). A representative image of three replicates from each group is shown.



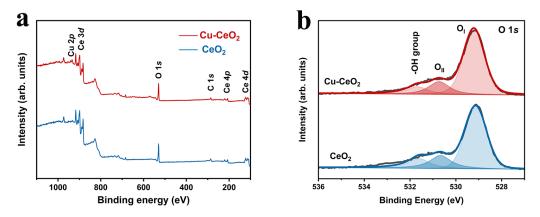
Supplementary Fig. 5. AC-HAADF-STEM image of Cu-CeO₂ precursor. A representative image of three replicates from each group is shown.



Supplementary Fig. 6. XRD fitting pattern of 9 nm Cu-CeO₂ sample. Source data are provided as a Source Data file.



Supplementary Fig. 7. Structural illustration of cubic CeO₂**.** The model diagram of the crystal structure of cubic CeO₂ (a), the relationship between the spacing and the angle between the different crystal planes of the cubic lattice (b).

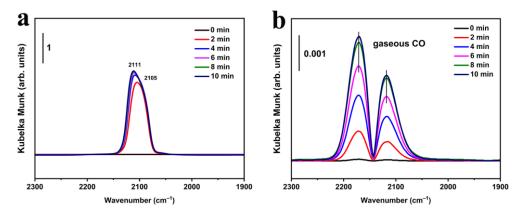


Supplementary Fig. 8. XPS spectra of Cu-CeO₂ and reference material. (a) XPS spectra of CeO₂ and Cu-CeO₂ samples, (b) O 1*s* photoelectron profiles of CeO₂ and Cu-CeO₂ catalysts. Source data are provided as a Source Data file.

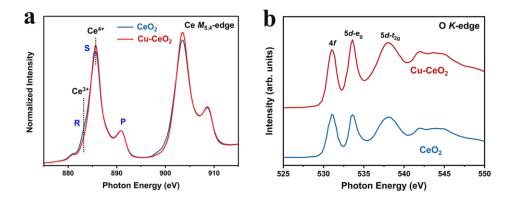
Supplementary Table 1. The area ratios between Ce³⁺ and Ce⁴⁺ species of different samples in Figure

2c.

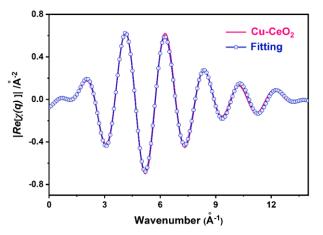
Sample	Ce ³⁺ /(Ce ³⁺ +Ce ⁴⁺)		
CeO ₂	0.23		
Cu-CeO ₂	0.18		



Supplementary Fig. 9. Study on Adsorption of CO on Cu-CeO₂ and CeO₂ samples. In situ DRIFTS study of CO adsorption on Cu-CeO₂ (a) and CeO₂ (b) samples. (2% CO/Ar flow rate, 30 mL min⁻¹; catalyst mass, 100 mg; temperature, 30 °C) Source data are provided as a Source Data file.



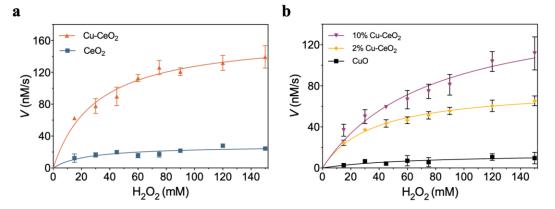
Supplementary Fig. 10. XANES spectra of CeO₂ and Cu-CeO₂ catalysts. (a) Soft XAS spectra of Ce $M_{4,5}$ absorption edges for CeO₂ and Cu-CeO₂ samples, (b) O K-edge XANES spectra for CeO₂ and Cu-CeO₂ catalysts. Source data are provided as a Source Data file.



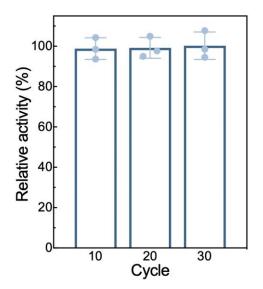
Supplementary Fig. 11. The q space fitting curve and experimental data of Cu-CeO₂. Source data are provided as a Source Data file.

sample	Scattering path	Distance (Å)	C. N.	σ² (Ų)	ΔE₀ (eV)	R-factor
Cu-CeO₂	Cu-O	1.94	3.2	0.004	-3.8	0.004

Supplementary Table 2. Fitting results of FT-EXAFS spectra of Cu-CeO₂ sample at the Cu K-edge.



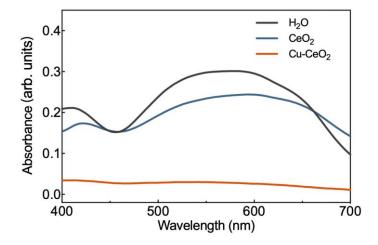
Supplementary Fig. 12. Steady-state kinetic assay of POD-like catalytic activity of CeO₂ and Cu-CeO₂ nanozymes with different Cu contents. Michaelis-Menten plot with different concentrations of H_2O_2 of (a) CeO₂, Cu-CeO₂ and (b) CuO, 2%, and 10% Cu-CeO₂. Data are presented as mean values +/- standard error of the mean, n = 3 independent replicates. Source data are provided as a Source Data file.



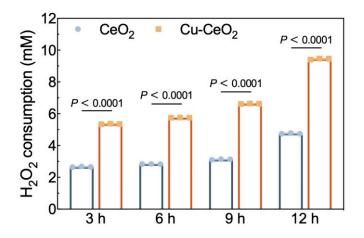
Supplementary Fig. 13. Relative cyclic stability of Cu-CeO₂. Data are presented as mean values +/- standard deviation, n = 3 independent replicates. Source data are provided as a Source Data file.

Catalyst	V _{max} (nM/s)	K _m (mM)	<i>E</i> (M)	Turnover rate(/s)
CeO ₂	28.05	24.34	-	-
CuO	14.15	65.86	6.24×10 ⁻⁴	2.27×10 ⁻⁵
2% Cu-CeO ₂	78.34	35.94	1.47×10 ⁻⁵	5.33×10 ⁻³
5% Cu-CeO ₂	166.7	30.76	3.46×10 ⁻⁵	4.82×10 ⁻³
10% Cu-CeO ₂	161.4	75.83	7.03×10 ⁻⁵	2.30×10 ⁻³

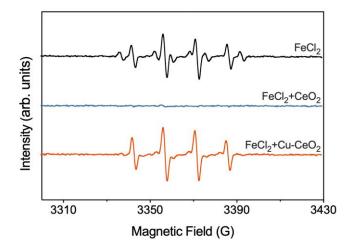
Supplementary Table 3. POD-like kinetic parameters of CeO₂ and Cu-CeO₂. (*K_m*: Michaelis-Menten constant, *V_{max}*: maximal reaction velocity)



Supplementary Fig. 14. SOD-like catalytic activity of CeO₂ and Cu-CeO₂. Source data are provided as a Source Data file.



Supplementary Fig. 15. H₂O₂ consumption of CeO₂ and Cu-CeO₂. Data are presented as mean values +/- standard deviation, n = 3 independent replicates. Significance was calculated by two-sided Student's t-test. Source data are provided as a Source Data file.



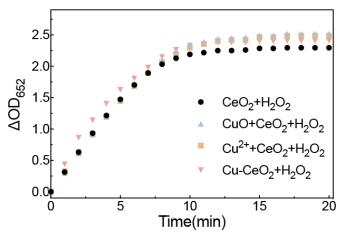
Supplementary Fig. 16. EPR spectrum of HORAC reactions of CeO₂ and Cu-CeO₂. Source data are provided as a Source Data file.

Supplementary Table 4. Qua	intitive countings of ERP	P spins of HORAC reactions of	CeO ₂ and Cu-CeO ₂ .
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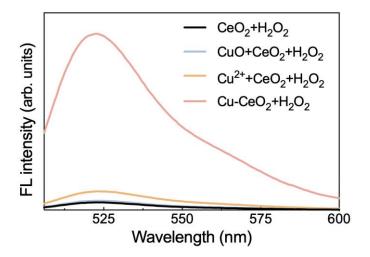
Group	Spins	Spin concentration (M)
FeCl ₂	3.91×10 ¹³	3.64×10⁻ ⁶
FeCl ₂ +CeO ₂	1.07×10 ¹²	9.98×10⁻ ⁸
FeCl ₂ +Cu-CeO ₂	6.20×10 ¹³	5.77×10⁻ ⁸

Supplementary Table 5. DFT calculated reaction energy (in eV) of POD-like (P1-P4) and HORAC (H1-H6) processes on different reaction sites.

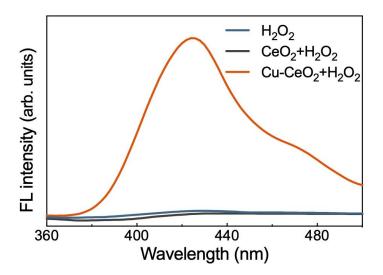
	Reaction Sites					
Reaction	Ce7c	Ce7c@	Cu@	Cu@	diCe6c@	Ce7c@
	Cert	Cu-ad	Cu-ad	Cu-sub	Cu-sub	Cu-sub
P1	-0.952	-1.046	-0.986	-1.442	-1.123	-0.760
P2	2.591	1.792	1.358	2.479	0.835	1.556
Р3	-1.064	-0.373	-0.010	-0.617	0.597	-0.082
P4	0.704	0.906	0.917	0.860	0.970	0.566
H1	-1.416	-2.309	-2.683	-2.019	-3.343	-2.234
H2	-3.292	-1.748	-2.574	-1.674	-2.265	-1.542
H3	0.909	1.258	1.110	0.652	0.746	0.924
H4	-4.711	-6.359	-4.946	-6.112	-4.888	-6.296
H5	0.299	0.914	0.737	0.606	0.700	0.663
H6	0.114	0.148	0.261	0.450	0.954	0.390



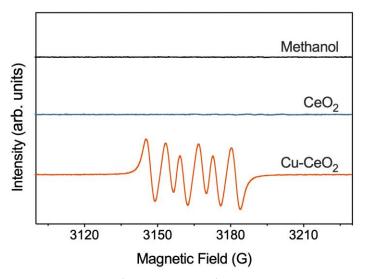
Supplementary Fig. 17. Time-dependent optical density change at 652 nm of TMB in POD reactions. Source data are provided as a Source Data file.



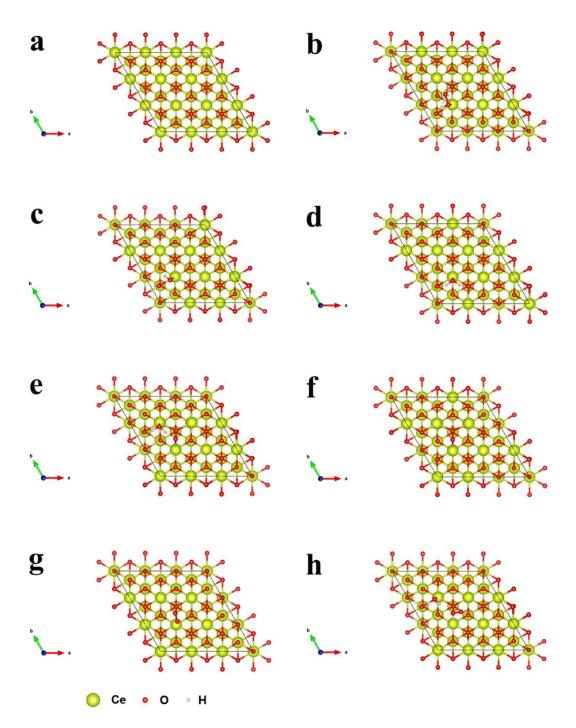
Supplementary Fig. 18. Fluorescent spectra of DCFH after 10 min reaction with H₂O₂ and different nanozymes. Source data are provided as a Source Data file.



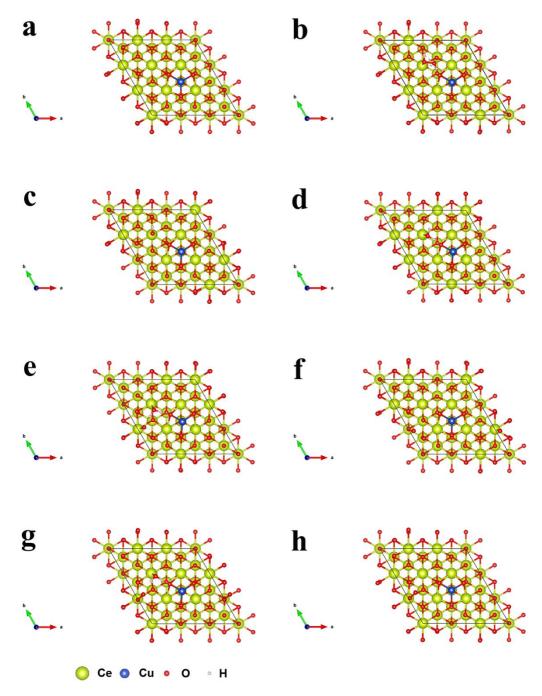
Supplementary Fig. 19. Fluorescent spectra of TA after 30 min reaction with H_2O_2 and different nanozymes. Source data are provided as a Source Data file.



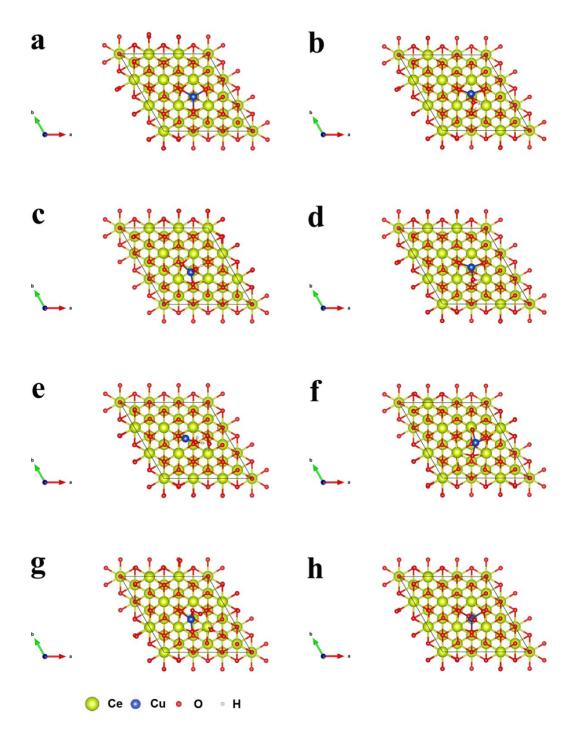
Supplementary Fig. 20. EPR spectrum of OXD reactions of CeO₂ and Cu-CeO₂. Source data are provided as a Source Data file.



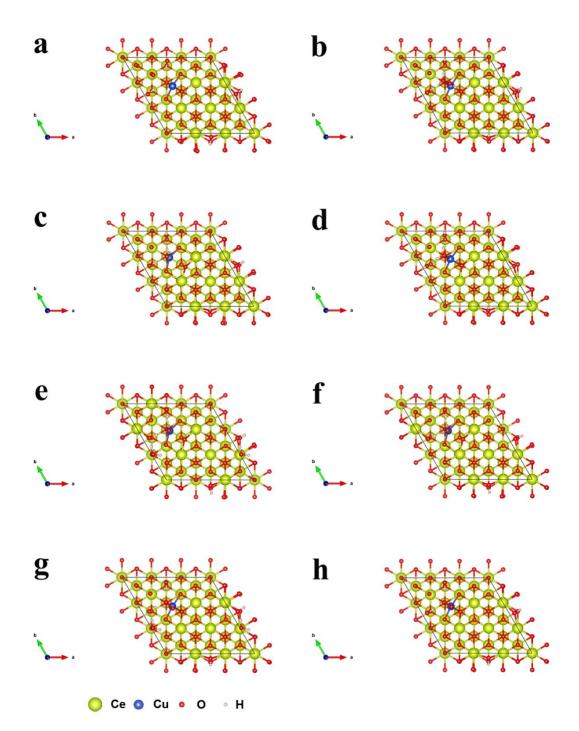
Supplementary Fig. 21. The DFT optimized geometry of reaction intermediates on Ce7c site. (a) clean slab, (b) H_2O_2 , (c) OH, (d) H_2O , (e) $H_2O + O$, (f) O_2 , (h) $H_2O + O_2$. Source data are provided as a Supplementary Data 1 file.



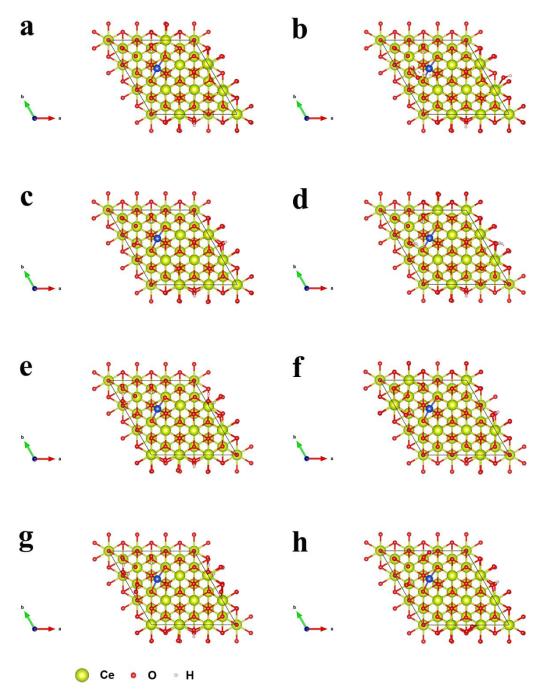
Supplementary Fig. 22. The DFT optimized geometry of reaction intermediates on Ce7c@Cu-ad site. (a) clean slab, (b) H_2O_2 , (c) OH, (d) H_2O , (e) $H_2O + O$, (f) O_2 , (g) $H_2O + O_2$, (h) O_2 . Source data are provided as a Supplementary Data 1 file.



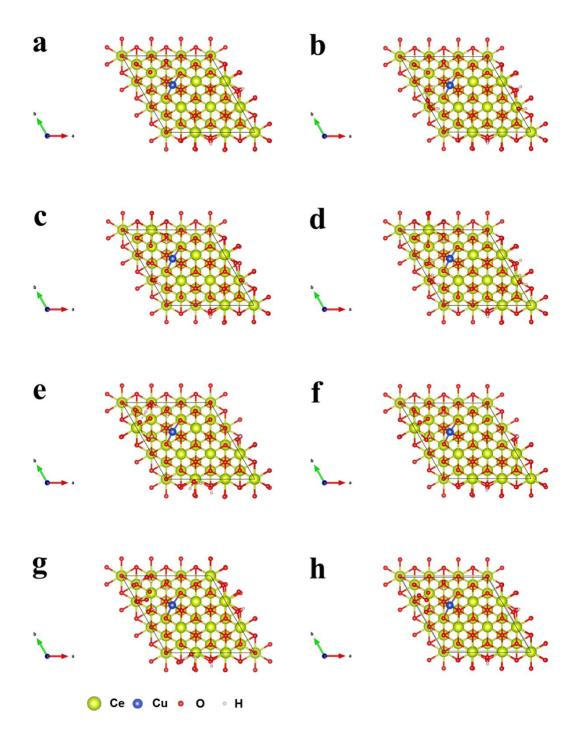
Supplementary Fig. 23. The DFT optimized geometry of reaction intermediates on Cu@Cu-ad. (a) clean slab, (b) $*H_2O_2$, (c) *OH, (d) $*H_2O$, (e) $*H_2O + *O$, (f) *O, (g) $*H_2O + *O_2$, (h) $*O_2$. Source data are provided as a Supplementary Data 1 file.



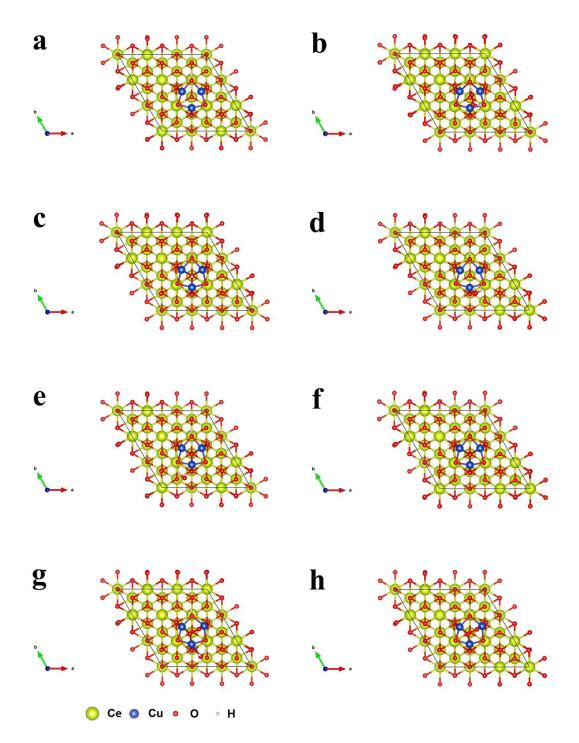
Supplementary Fig. 24. The DFT optimized geometry of reaction intermediates on Cu@Cu-sub. (a) clean slab, (b) $*H_2O_2$, (c) *OH, (d) $*H_2O$, (e) $*H_2O + *O$, (f) *O, (g) $*H_2O + *O_2$, (h) $*O_2$. Source data are provided as a Supplementary Data 1 file.



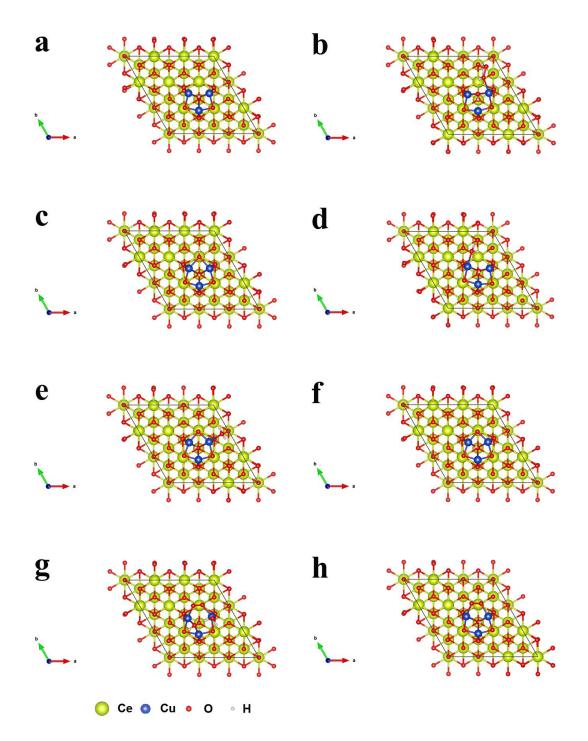
Supplementary Fig. 25. The DFT optimized geometry of reaction intermediates on diCe6c@Cu-sub. (a) clean slab, (b) $*H_2O_2$, (c) *OH, (d) $*H_2O$, (e) $*H_2O + *O$, (f) *O, (g) $*H_2O + *O_2$, (h) $*O_2$. Source data are provided as a Supplementary Data 1 file.



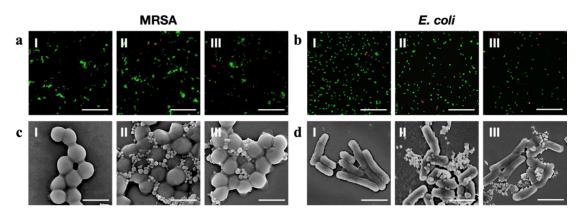
Supplementary Fig. 26. The DFT optimized geometry of reaction intermediates on Ce7c@Cu-sub. (a) clean slab, (b) $*H_2O_2$, (c) *OH, (d) $*H_2O$, (e) $*H_2O + *O$, (f) *O, (g) $*H_2O + *O_2$, (h) $*O_2$. Source data are provided as a Supplementary Data 1 file.



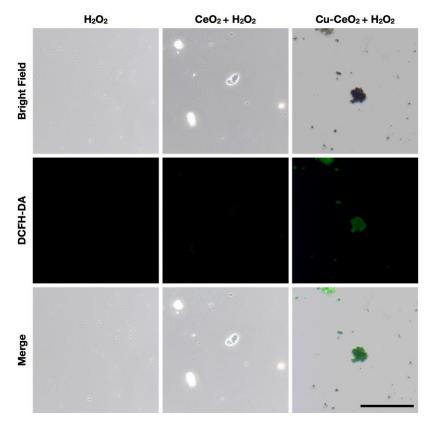
Supplementary Fig. 27. The DFT optimized geometry of reaction intermediates on Cu₃ site. (a) clean slab, (b) H_2O_2 , (c) OH, (d) H_2O , (e) $H_2O + O$, (f) O, (g) O_2 , (h) $H_2O + O_2$. Source data are provided as a Supplementary Data 1 file.



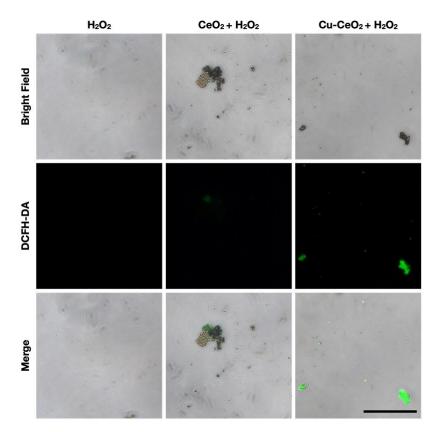
Supplementary Fig. 28. The DFT optimized geometry of reaction intermediates on Cu₂Ce site. (a) clean slab, (b) H_2O_2 , (c) OH, (d) H_2O , (e) $H_2O + O$, (f) O, (g) O_2 , (h) $H_2O + O_2$. Source data are provided as a Supplementary Data 1 file.



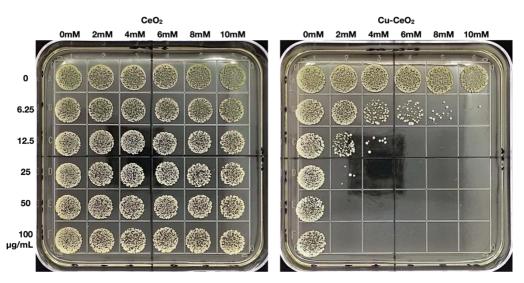
Supplementary Fig. 29. In vitro antibacterial performance of the nanozymes. Fluorescent images (scale bar: 50 μ m) and SEM images (scale bar: 1 μ m) of MRSA (a, c) and *E. coli* (b, d) after grouped treatment (I: PBS, II: CeO₂, III: Cu-CeO₂). A representative image of three replicates from each group is shown.



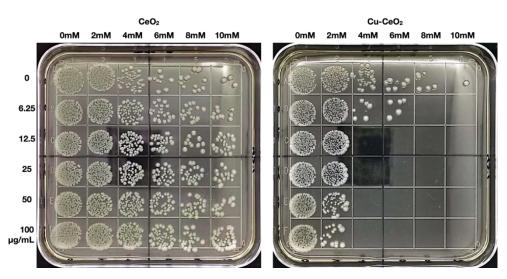
Supplementary Fig. 30. ROS fluorescent staining of MRSA after different treatments. Scale bar: 50 μ m. A representative image of three replicates from each group is shown.



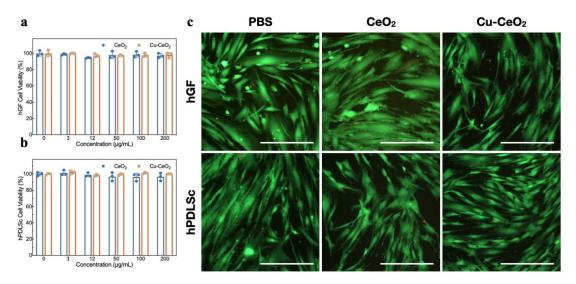
Supplementary Fig. 31. ROS fluorescent staining of *E. coli* after different treatments. Scale bar: 50 μ m. A representative image of three replicates from each group is shown.



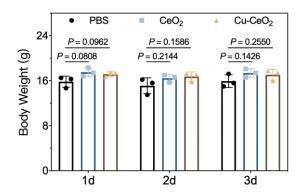
Supplementary Fig. 32. Bacterial colonies of MRSA after treatment with different concentrations of CeO₂, Cu-CeO₂, and H₂O₂. A representative image of three replicates from each group is shown.



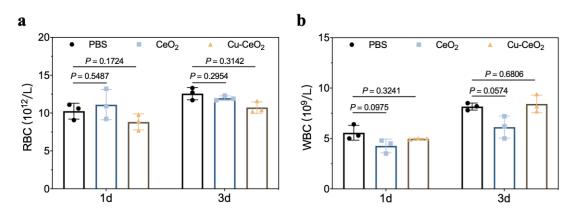
Supplementary Fig. 33. Bacterial colonies of *E. coli* after treatment with different concentrations of CeO₂, Cu-CeO₂ and H₂O₂. A representative image of three replicates from each group is shown.



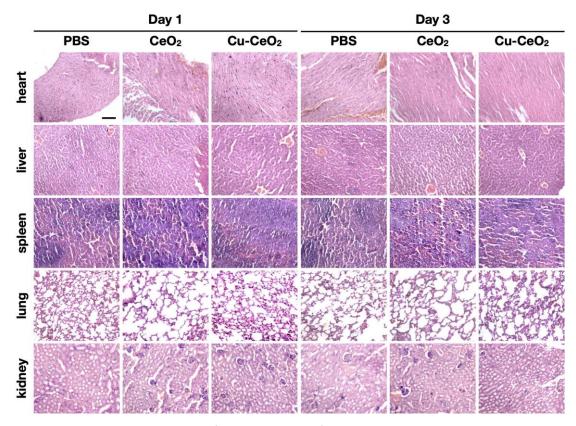
Supplementary Fig. 34. Biocompatibility of CeO₂ and Cu-CeO₂. Relative cell viability of hGF (a) and hPDLSc (b) after treatment with different concentrations of nanozymes. (c) Fluorescent images of hGF and hPDLSc after grouped treatment. Scale bar: 200 μ m. A representative image of three replicates from each group is shown. Data are presented as mean values +/- standard deviation, n = 3 biologically independent replicates. Source data are provided as a Source Data file.



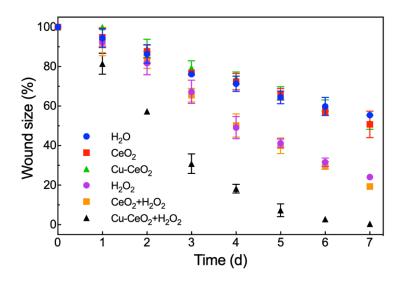
Supplementary Fig. 35. Body weights of balb/c mice after intravenous injection with PBS, CeO₂ and Cu-CeO₂ (n.s., P > 0.05, n=3). Data are presented as mean values +/- standard deviation, n = 3 biologically independent replicates. Significance was calculated by two-sided Student's t-test. Source data are provided as a Source Data file.



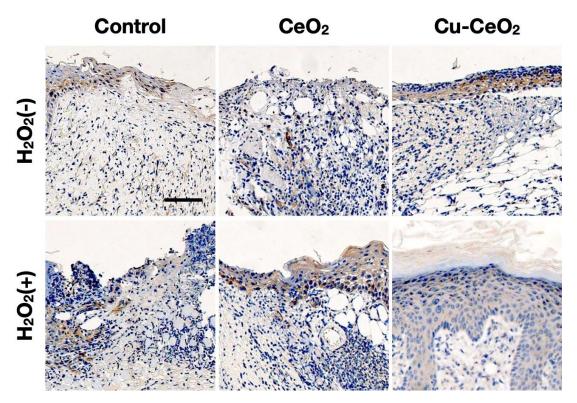
Supplementary Fig. 36. Blood routine results of balb/c mice after intravenous injection with PBS, CeO_2 and $Cu-CeO_2$. (a) Red blood cells. (b) White blood cells. Data are presented as mean values +/- standard deviation, n = 3 biologically independent replicates. Significance was calculated by two-sided Student's t-test. Source data are provided as a Source Data file.



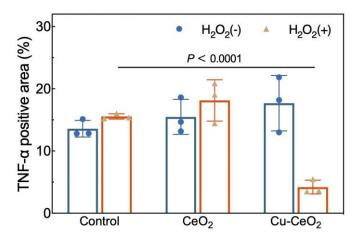
Supplementary Fig. 37. HE staining of the main organs after intravenous injection with PBS, CeO₂ and Cu-CeO₂. Scale bar: 500 μm. A representative image of three replicates from each group is shown.



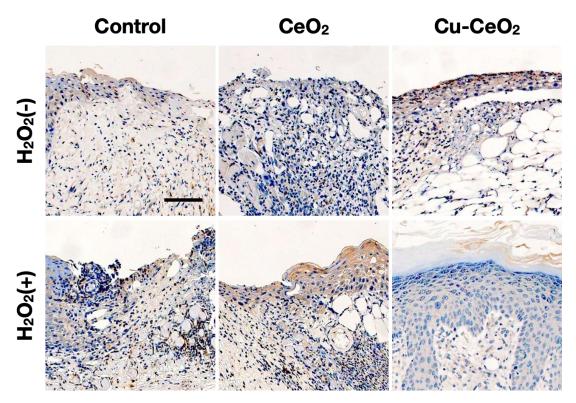
Supplementary Fig. 38. Time-dependent relative wound size after different treatments. Data are presented as mean values +/- standard deviation, n = 3 biologically independent replicates. Source data are provided as a Source Data file.



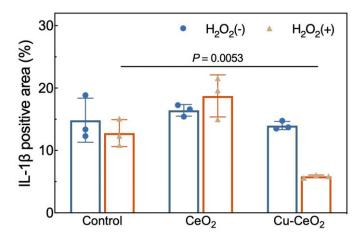
Supplementary Fig. 39. Immunohistochemsistry staining for TNF- α expression in the wound tissues of different groups. Scale bar: 500 μ m. A representative image of three replicates from each group is shown.



Supplementary Fig. 40. Quantitive analysis of TNF- α expression of different groups. Data are presented as mean values +/- standard deviation, n = 3 biologically independent replicates. Significance was calculated by two-sided Student's t-test. Source data are provided as a Source Data file.



Supplementary Fig. 41. Immunohistochemsistry staining for IL-1 β expression in the wound tissues of different groups. Scale bar: 500 μ m. A representative image of three replicates from each group is shown.



Supplementary Fig. 42. Quantitive analysis of IL-1 β expression of different groups. Data are presented as mean values +/- standard deviation, n = 3 biologically independent replicates. Significance was calculated by two-sided Student's t-test. Source data are provided as a Source Data file.