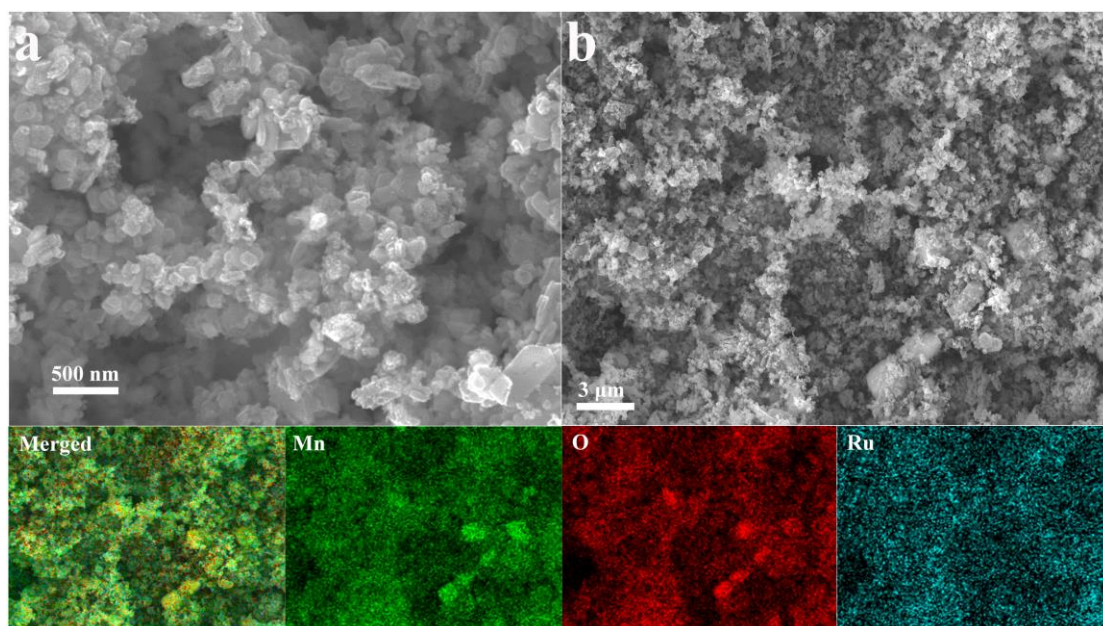


Supporting Information

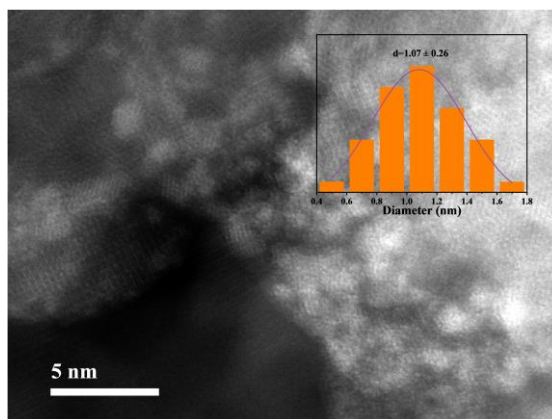
Photo-thermal coupling to enhance CO₂ hydrogenation toward CH₄ over Ru/MnO/Mn₃O₄

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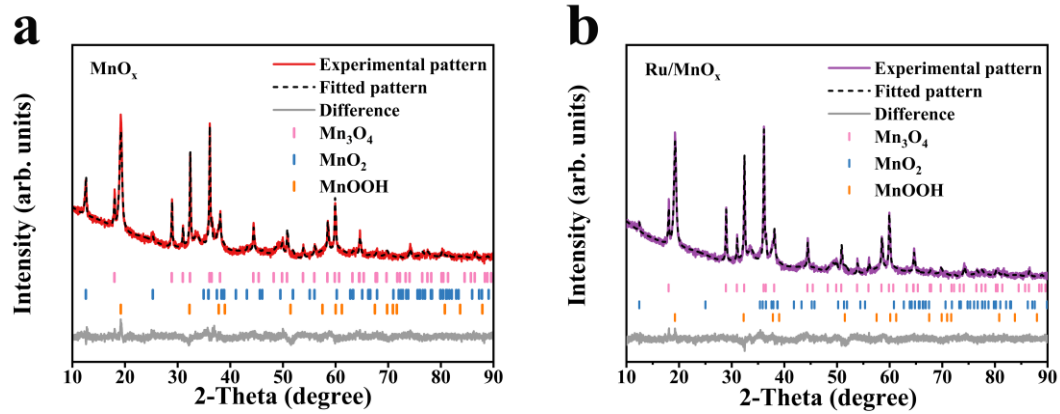
1. Shanghai Key Laboratory of Green Chemistry and Chemical Processes, State Key Laboratory of Petroleum Molecular & Process Engineering, School of Chemistry and Molecular Engineering, East China Normal University, Shanghai 200062, China
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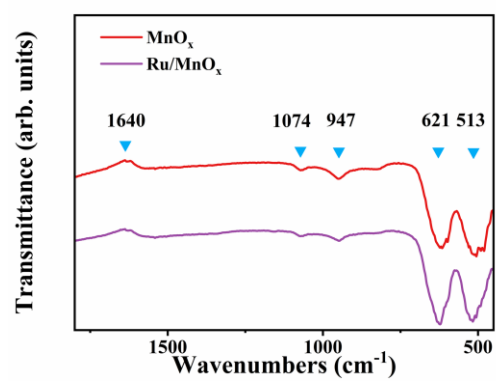
Supplementary Fig. 1 SEM image of (a) Ru/MnO_x; (b) The elemental mapping of Ru/MnO_x.



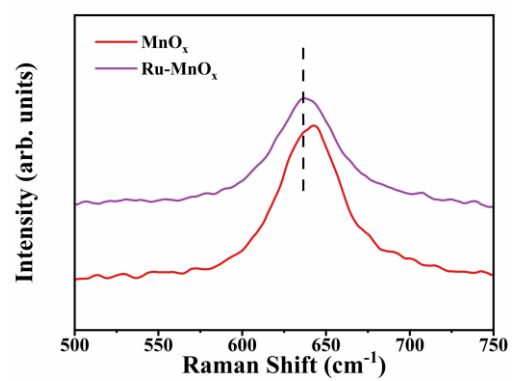
Supplementary Fig. 2 The high angle annular dark-field scanning transmission electron microscope (HAADF-STEM) image of the Ru/MnO_x catalyst.



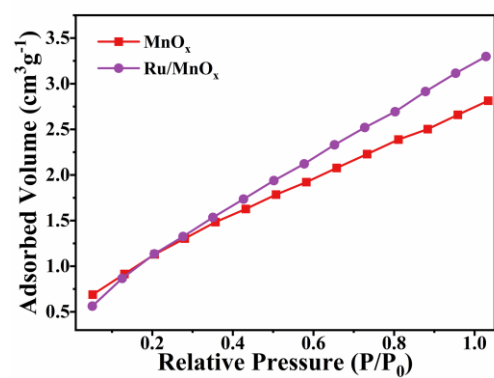
Supplementary Fig. 3 Rietveld refinement result of XRD patterns: (a) MnO_x ; (b) Ru/MnO_x .



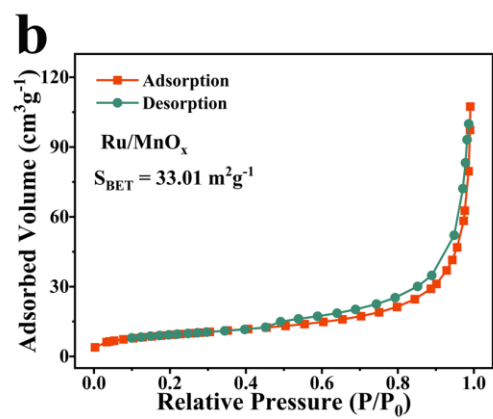
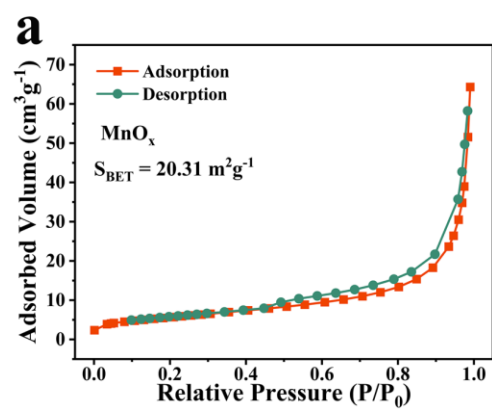
Supplementary Fig. 4 FT-IR spectra of MnO_x and Ru/MnO_x.



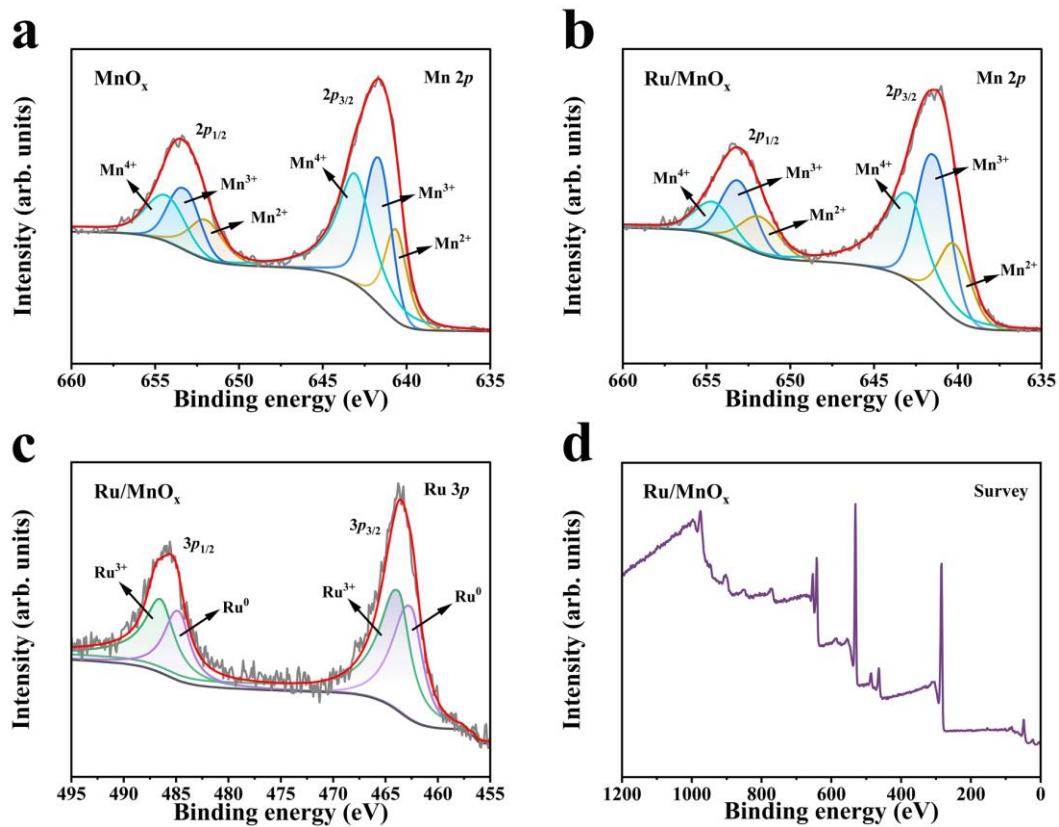
Supplementary Fig. 5 Raman spectra of MnO_x and Ru/MnO_x.



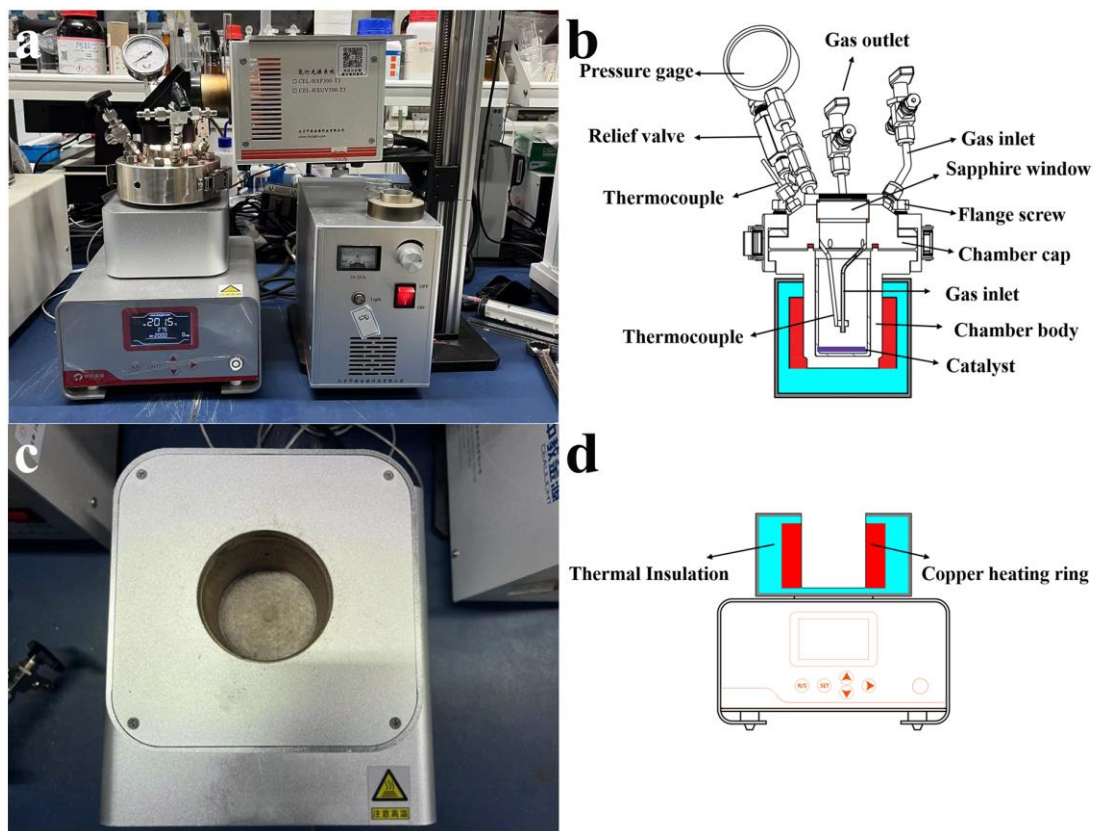
Supplementary Fig. 6 CO₂ adsorption isotherms of MnO_x and Ru/MnO_x.



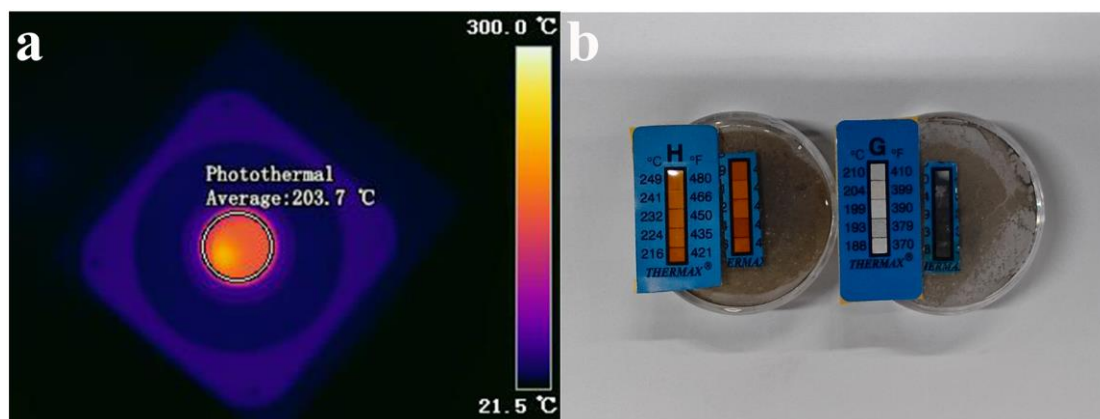
Supplementary Fig. 7 N_2 adsorption-desorption isotherms of (a) MnO_x ; (b) Ru/MnO_x .



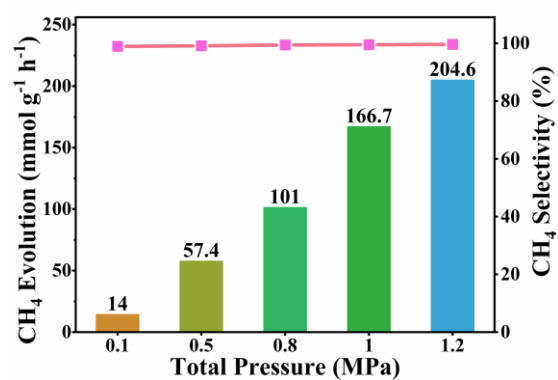
Supplementary Fig. 8 (a-b) High-resolution Mn 2p XPS spectra of MnO_x and Ru/MnO_x ; (c) High-resolution Ru 3p XPS spectra of Ru/MnO_x ; (d) XPS survey spectrum of Ru/MnO_x .



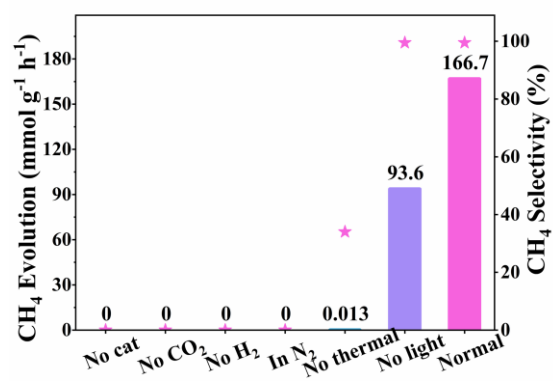
Supplementary Fig. 9 (a) Photograph of the apparatus setup for photo-thermal CO₂ experiments in the batch reactor; (b) Schematic illustration of the photo-thermal reactor; (c) and (d) Schematic illustration of the heating system.



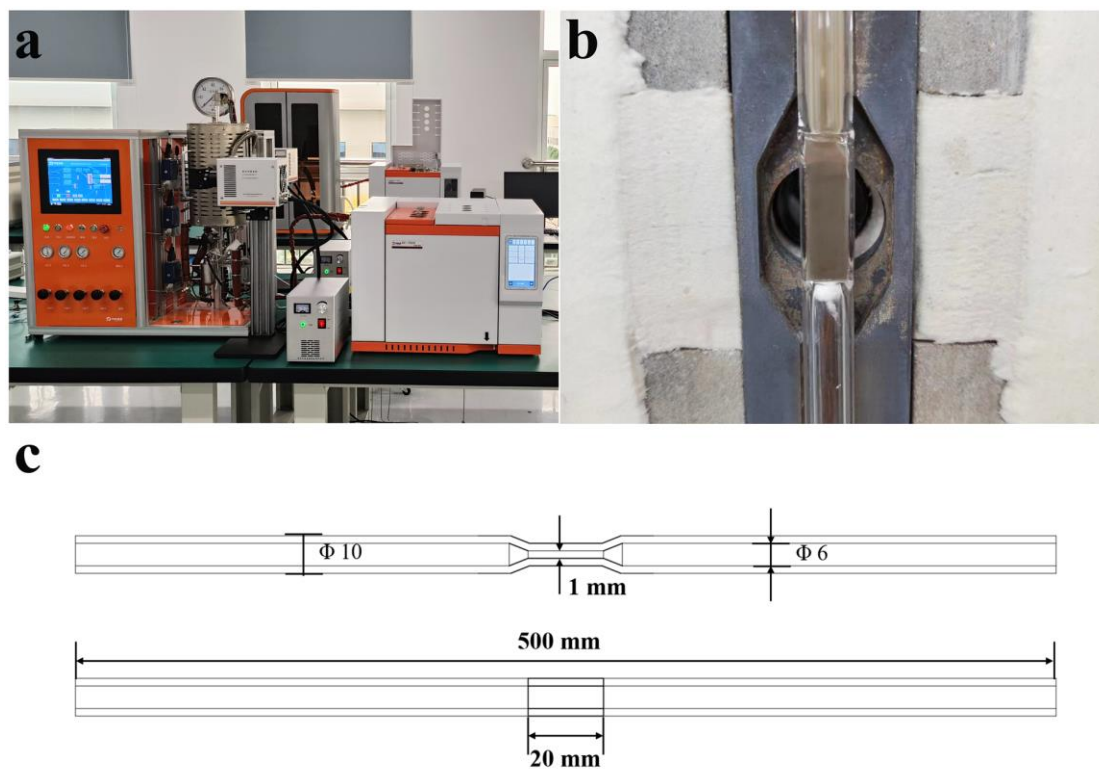
Supplementary Fig. 10 (a) Infrared thermal images captured for the catalyst surface temperature under 2.5 W cm^{-2} irradiation, 0.1 MPa and external heating (Set temperature: $200 \text{ }^{\circ}\text{C}$); (b) The temperature at the bottom of the catalyst, measured using a commercially available thermochromic temperature indicator.



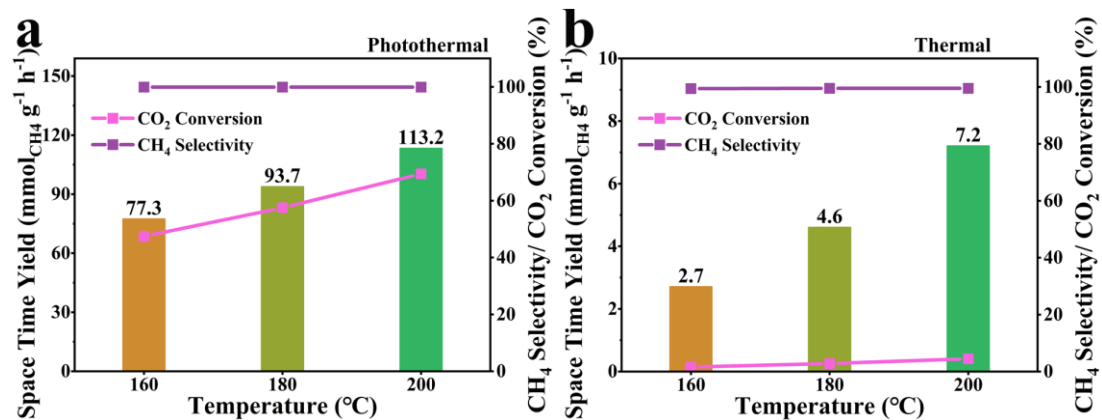
Supplementary Fig. 11 Influence of total pressure on CH₄ evolution rate over Ru/MnO_x; Reaction conditions: 15 mg of catalyst, full-arc 300 W UV-xenon lamp, 2.5 W cm⁻², 200 °C, irradiation time 4 hours, H₂/CO₂=4/1.



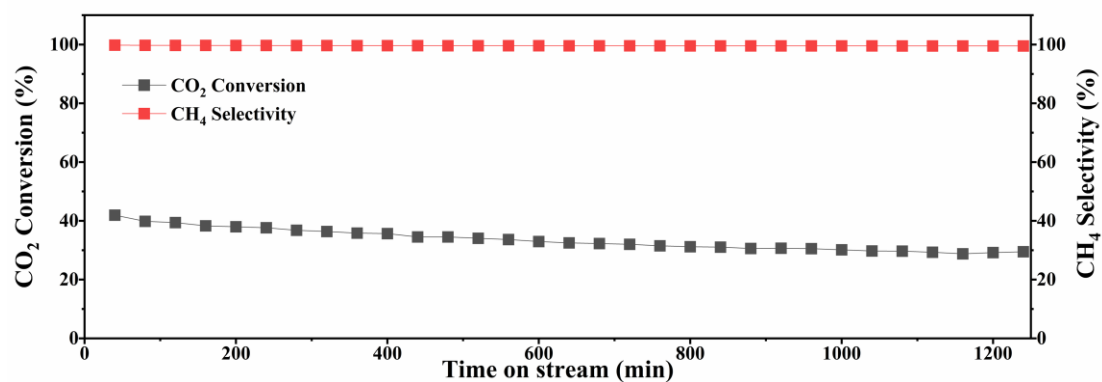
Supplementary Fig. 12 Control experiments for Ru/MnO_x under various conditions. Reaction conditions: 15 mg of catalyst, full-arc 300 W UV-xenon lamp, 2.5 W cm⁻², 200 °C, irradiation time 4 hours, initial pressure 1 MPa (H₂/CO₂=4/1).



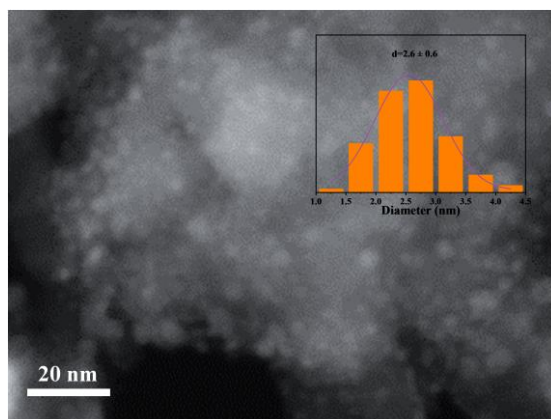
Supplementary Fig. 13 The images of (a) the photo-thermal catalytic performance evaluation process carried out in the flow reaction system and (b) the fixed-bed quartz tube reactor. (c) Dimensions of the fixed-bed quartz tube reactor.



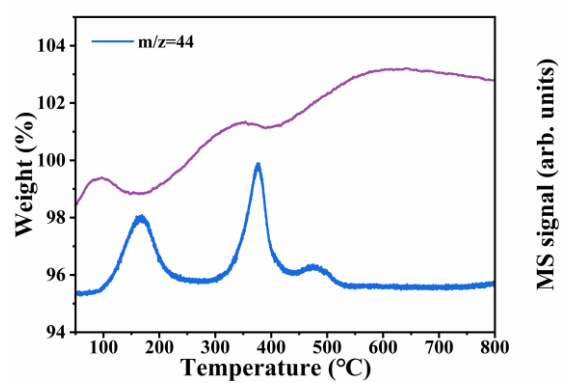
Supplementary Fig. 14 Temperature-dependent space time yield of CH₄ over Ru/MnO_x under photothermal (a) and thermal (b) conditions. Reaction conditions: 150 mg of catalyst, full-arc 300 W UV-xenon lamp, 2.5 W cm⁻², initial pressure 0.1 MPa, CO₂/H₂ mixture flow (10 mL min⁻¹/40 mL min⁻¹).



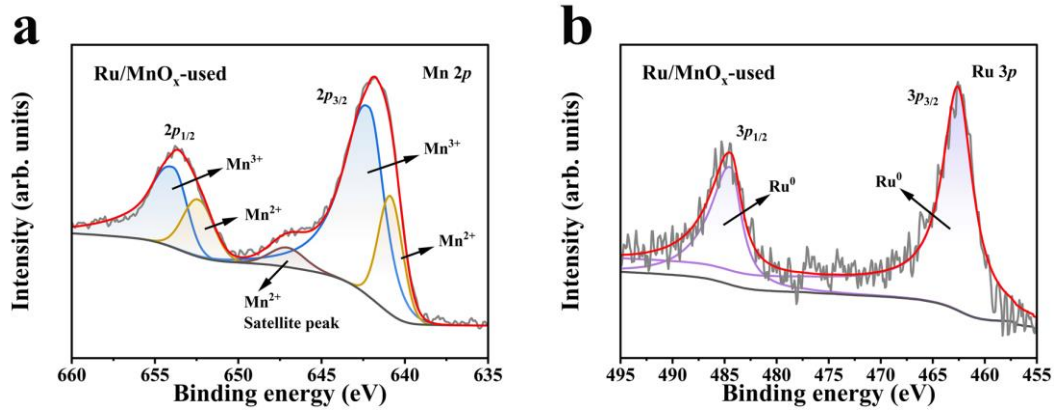
Supplementary Fig. 15 The photothermal catalytic performance of Ru/MnO_x catalyst in a fixed-bed reactor. Reaction conditions: 150 mg of catalyst, full-arc 300 W UV-xenon lamp, 2.5 W cm⁻², 200 °C, initial pressure 0.1 MPa, CO₂/H₂ mixture flow (20 mL min⁻¹/80 mL min⁻¹).



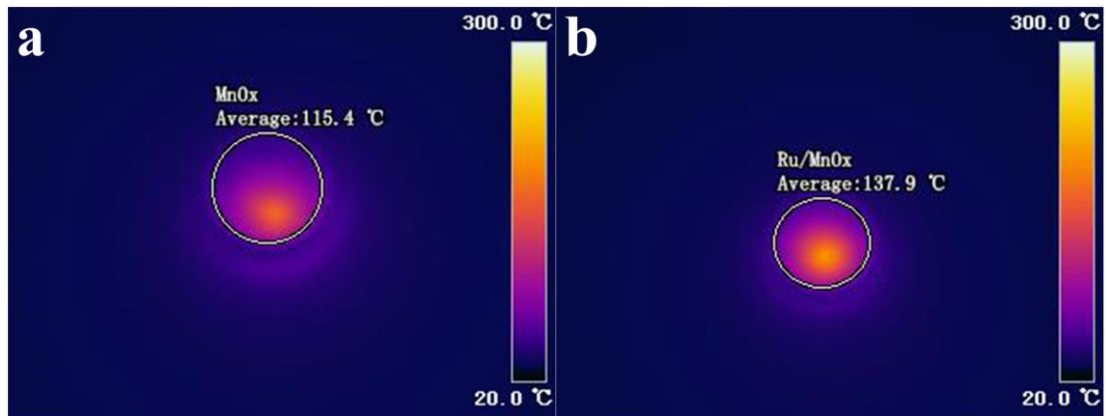
Supplementary Fig. 16 TEM image of Ru/MnO_x after reaction of 20 h at 200 °C under photothermal condition in the fixed-bed reactor.



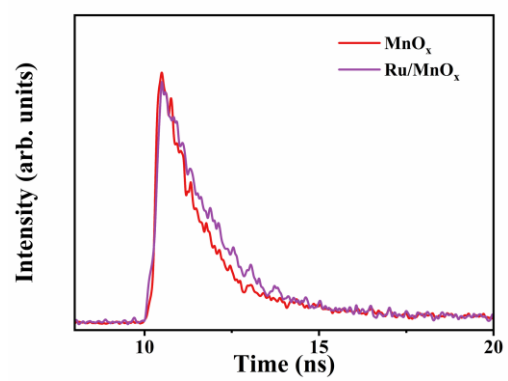
Supplementary Fig. 17 TG-MS analysis of Ru/MnO_x after reaction of 20 h at 200 °C under photothermal condition in the fixed-bed reactor.



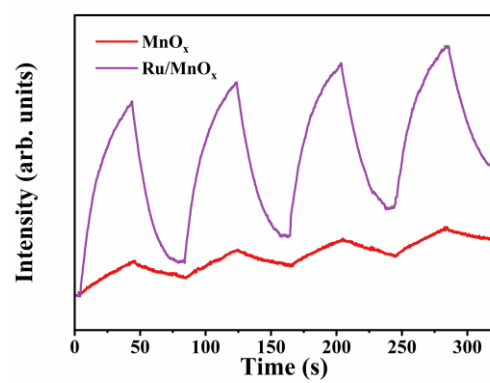
Supplementary Fig. 18 XPS spectra of Ru/MnO_x after reaction in 4 h at 200 °C in the batch reactor: (a) High-resolution of Mn 2*p* XPS spectra; (b) High-resolution of Ru 3*p* XPS spectra.



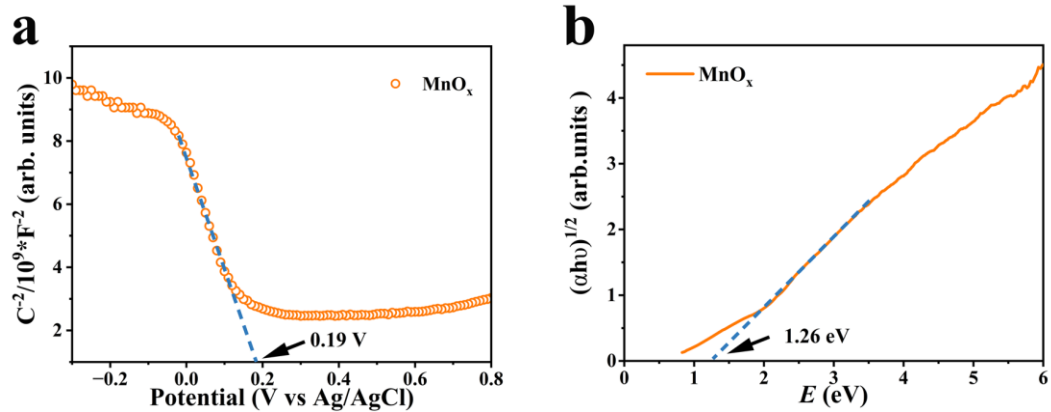
Supplementary Fig. 19 Infrared thermal images captured for (a) MnO_x and (b) Ru/MnO_x under 2.5 W cm^{-2} illumination.



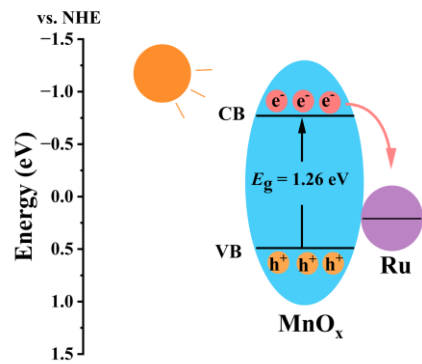
Supplementary Fig. 20 TRPL spectra of MnO_x and Ru/MnO_x .



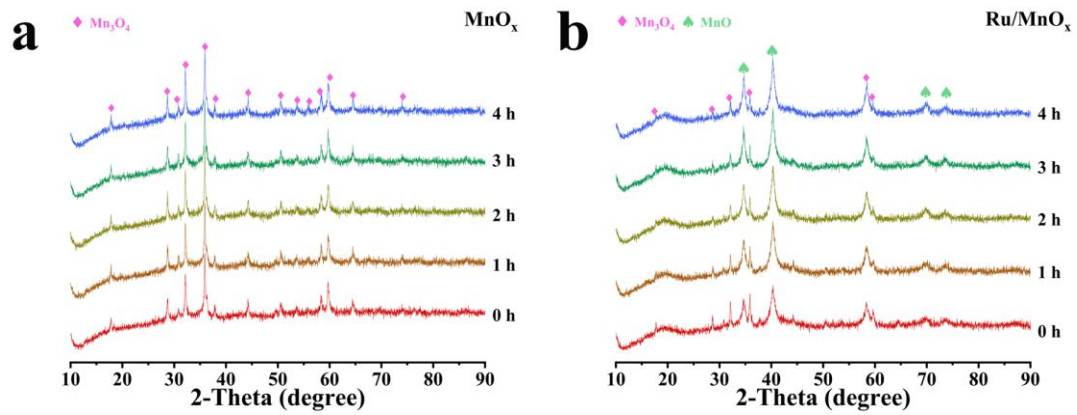
Supplementary Fig. 21 The periodic on/off photocurrent response spectra of MnO_x and Ru/MnO_x.



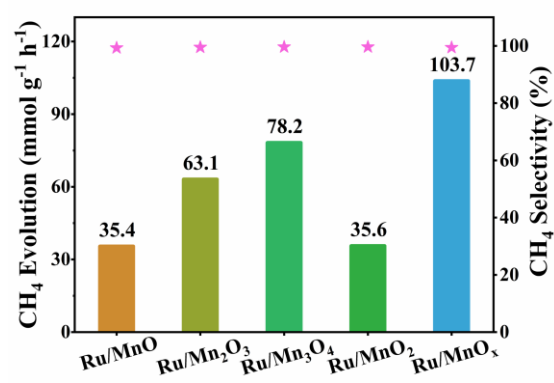
Supplementary Fig. 22 (a) Mott–Schottky plots of the MnO_x; (b) The bandgap value of the MnO_x.



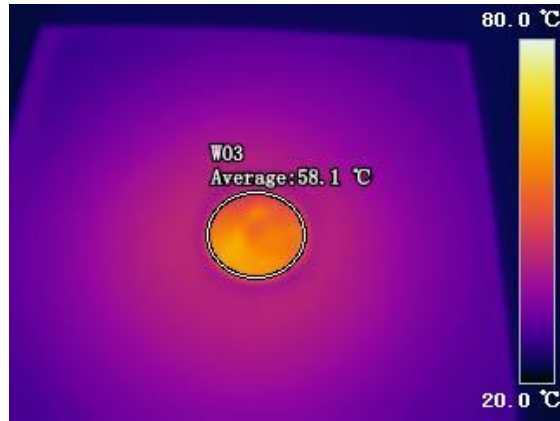
Supplementary Fig. 23 The work function of Ru and band structures of MnO_x .



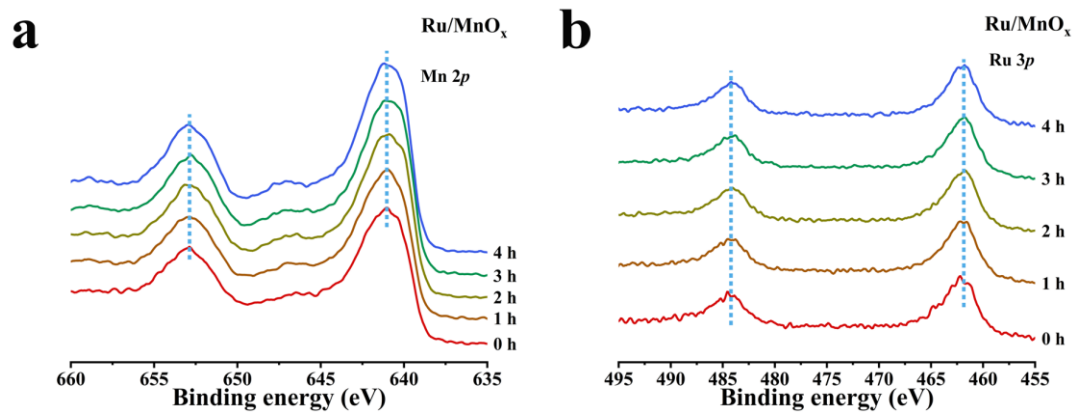
Supplementary Fig. 24 The different time of variable temperature XRD results in 20% CO_2/H_2 atmosphere at 200 °C: (a) MnO_x ; (b) Ru/MnO_x .



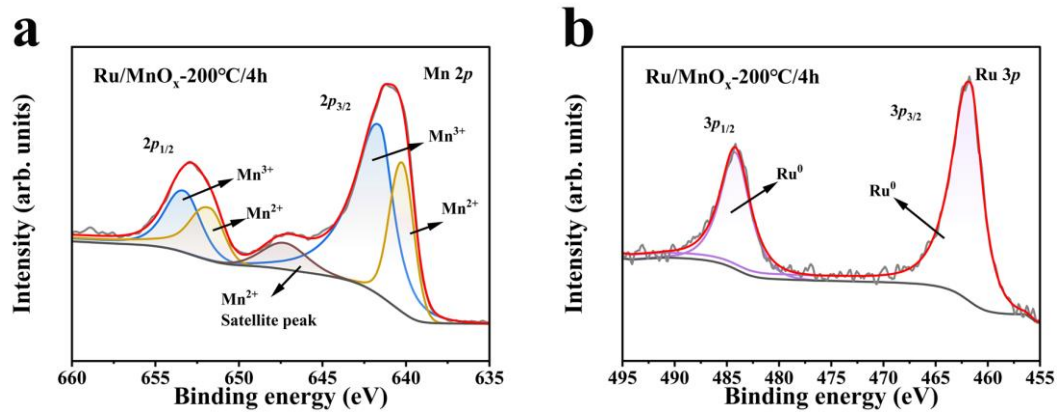
Supplementary Fig. 25 Influence of various manganese oxide on CH₄ evolution rate. Reaction conditions: 15 mg of catalyst, full-arc 300 W UV-xenon lamp, 2.5 W cm⁻², 200 °C, irradiation time 4 hours, initial pressure 1 MPa (H₂/CO₂ = 1/1).



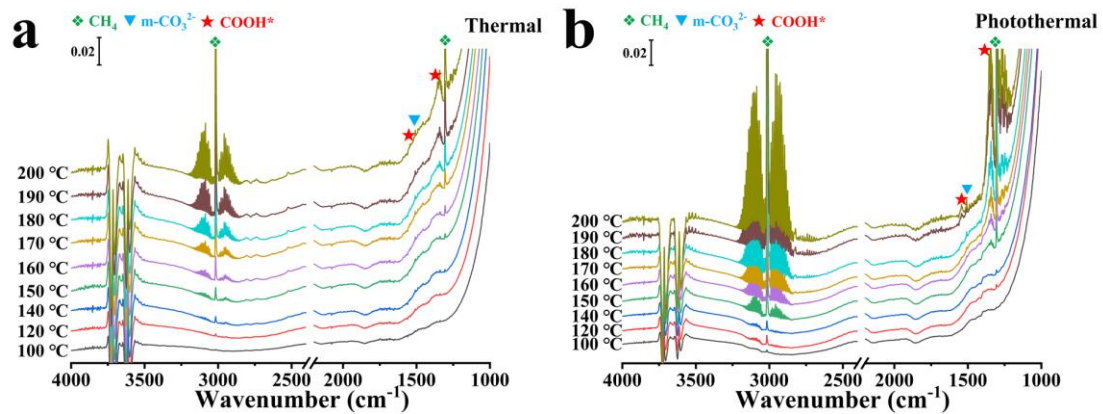
Supplementary Fig. 26 Infrared thermal images captured for WO_3 under 0.3 W cm^{-2} illumination.



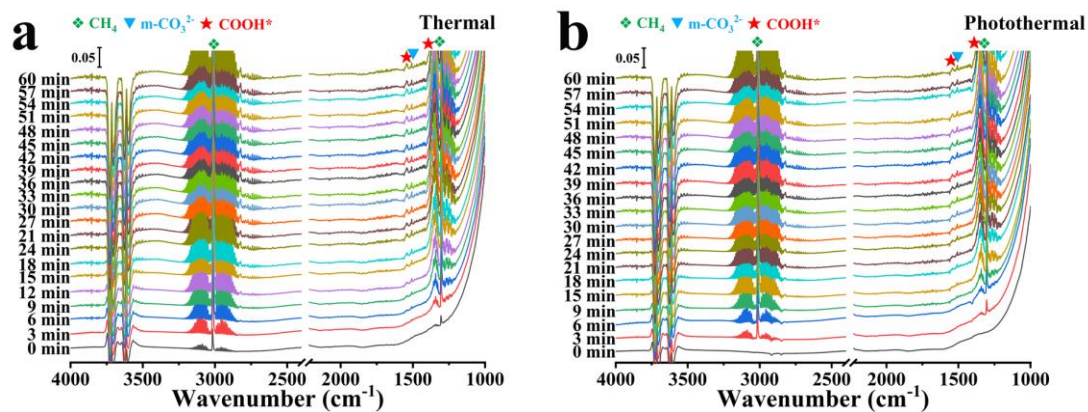
Supplementary Fig. 27 XPS spectra of Ru/MnO_x in 20% CO₂/H₂ atmosphere under variable time at 200 °C: (a) High-resolution of Mn 2p XPS spectra; (b) High-resolution of Ru 3p XPS spectra.



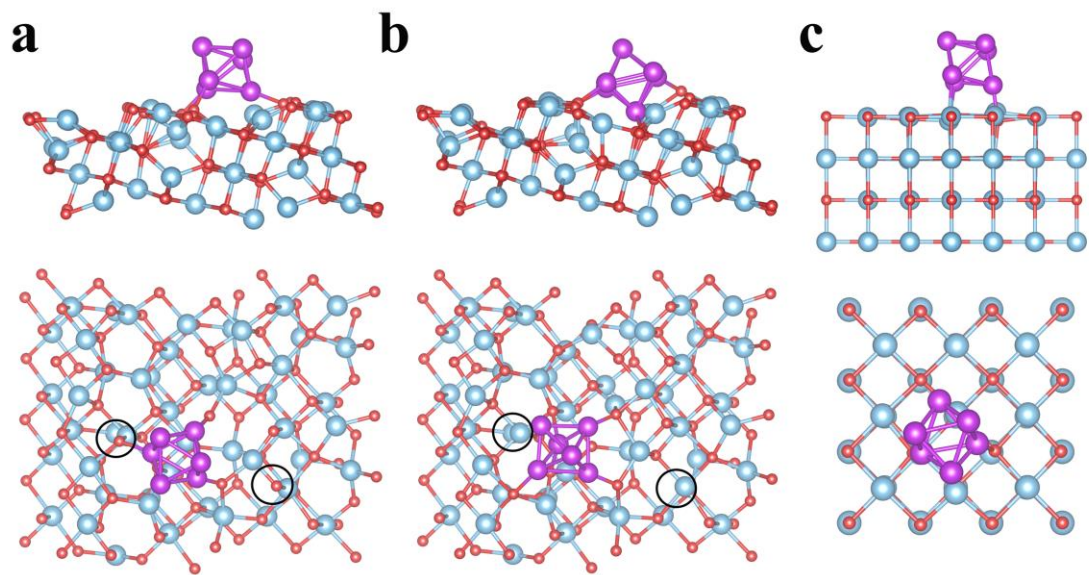
Supplementary Fig. 28 XPS spectra of Ru/MnO_x after reacting at 200 °C for 4 h in a 20% CO₂/H₂ atmosphere: (a) High-resolution of Mn 2p XPS spectra; (b) High-resolution of Ru 3p XPS spectra.



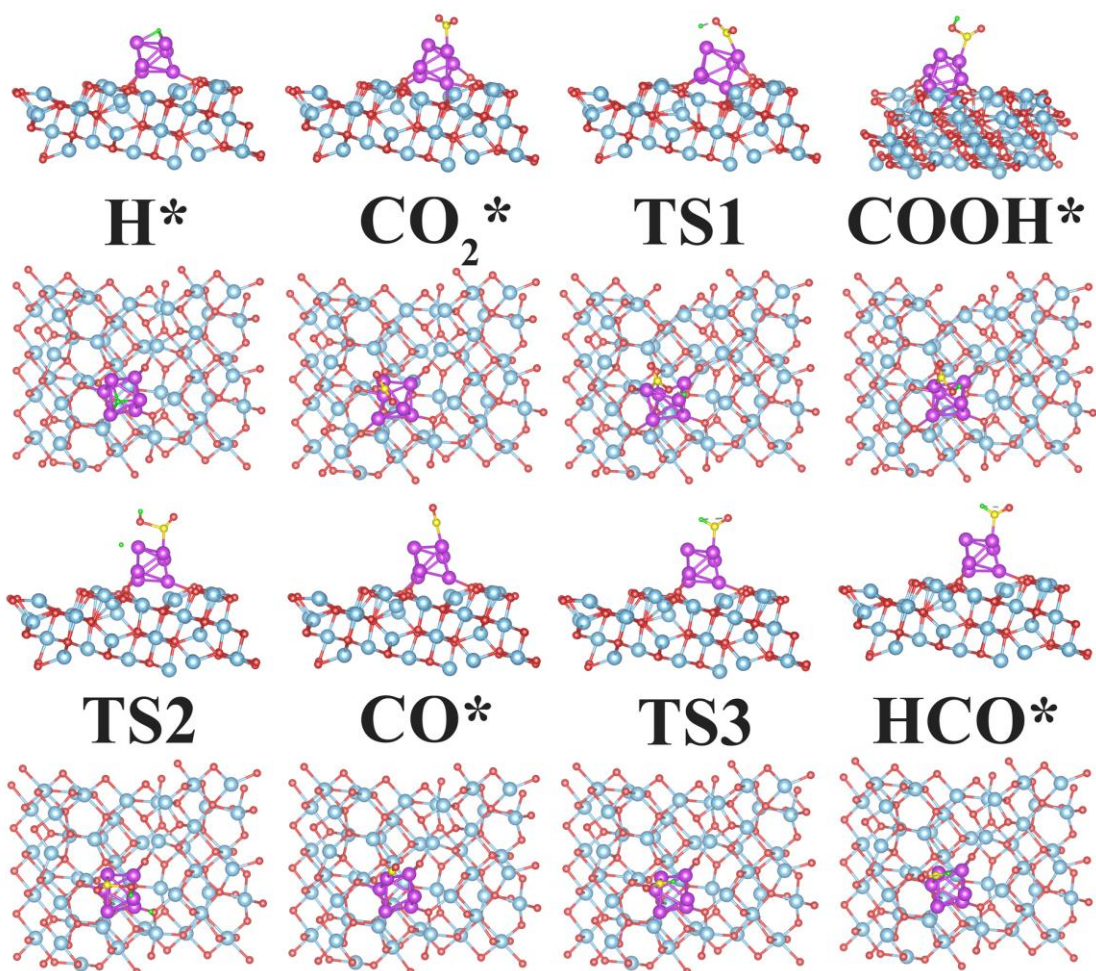
Supplementary Fig. 29 Spectra of FT-IR study of Ru/MnO_x at different conditions: (a) Effect of different temperature under thermal condition; (b) Effect of different temperature under photothermal conditions.



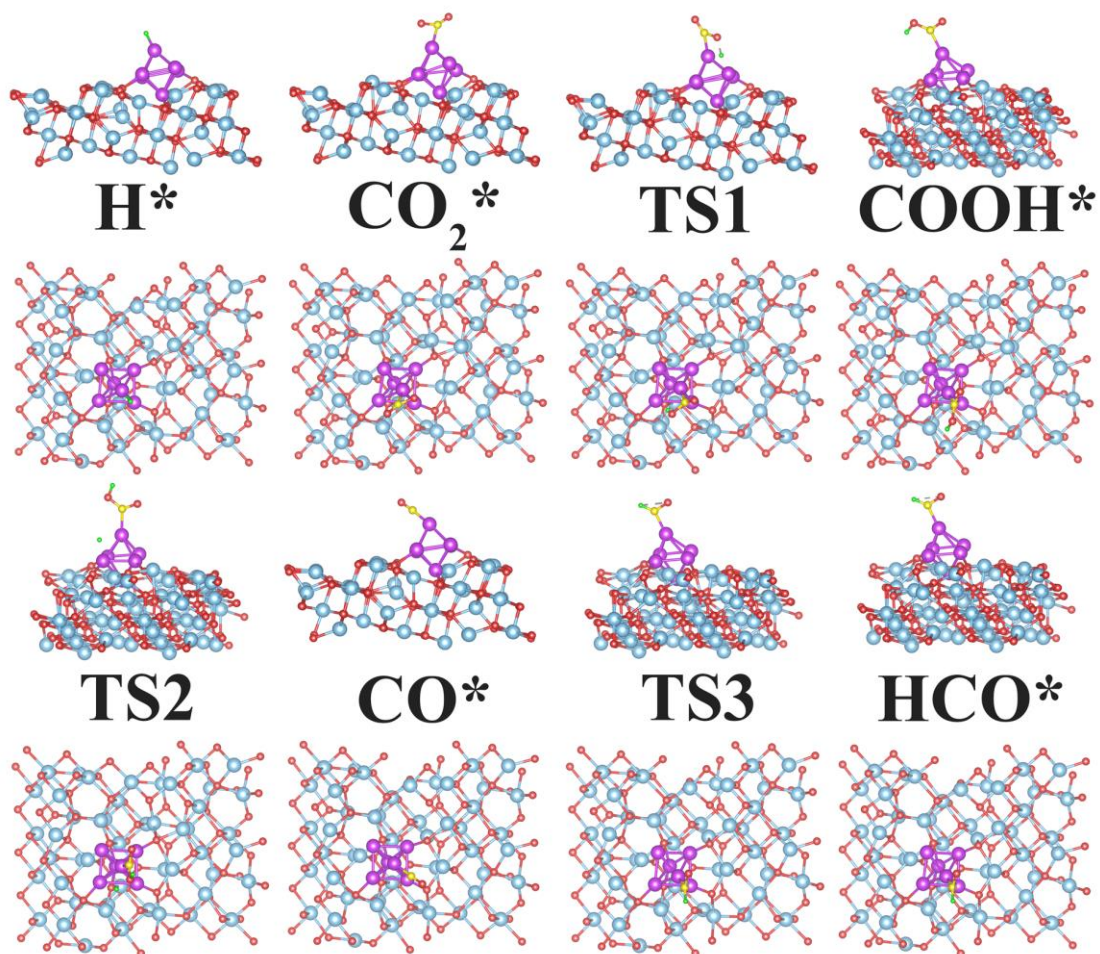
Supplementary Fig. 30 Spectra of FT-IR study of Ru/MnO_x at different conditions: (a) Effect of different time at 200 °C under thermal condition; (b) Effect of different time at 200 °C under photothermal conditions.



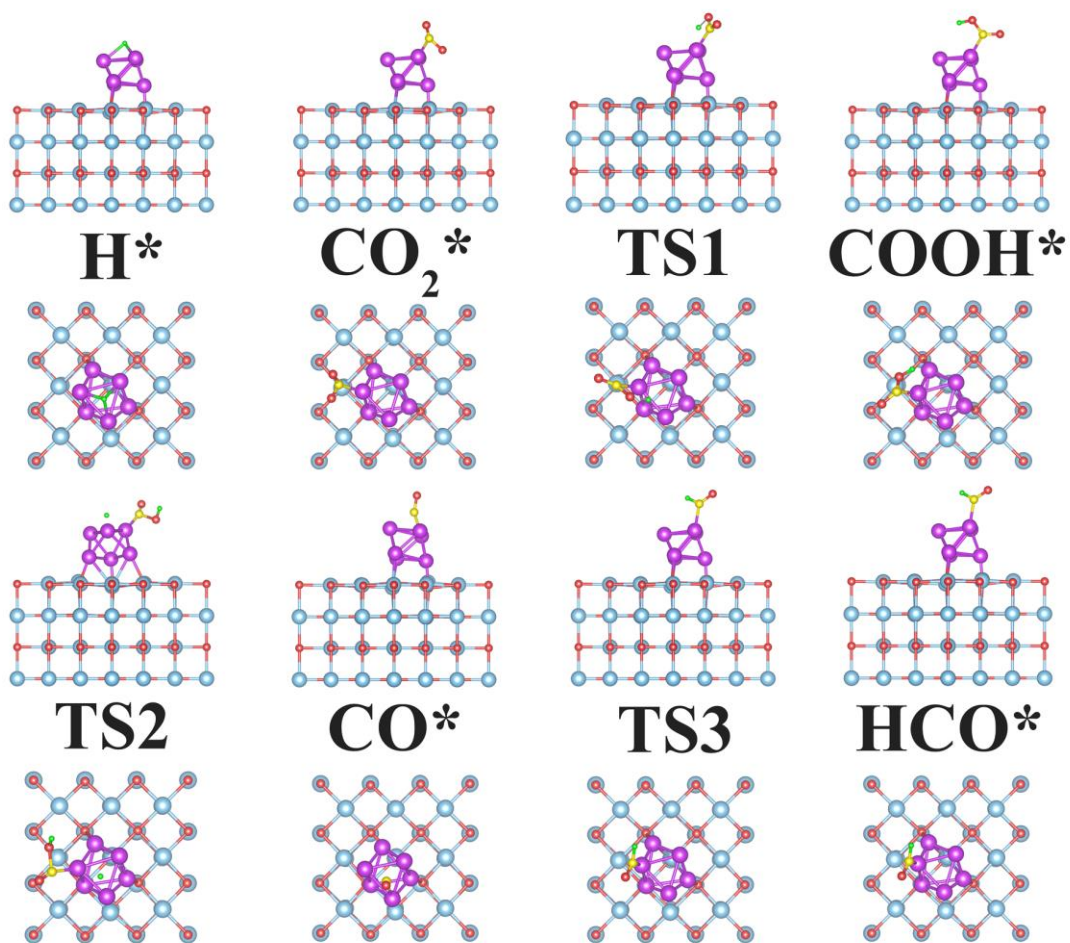
Supplementary Fig. 31 Calculation model of (a) Ru/Mn₃O₄ (321), (b) Ru/Mn₃O_{4-x} (321) and (c) Ru/MnO (200).



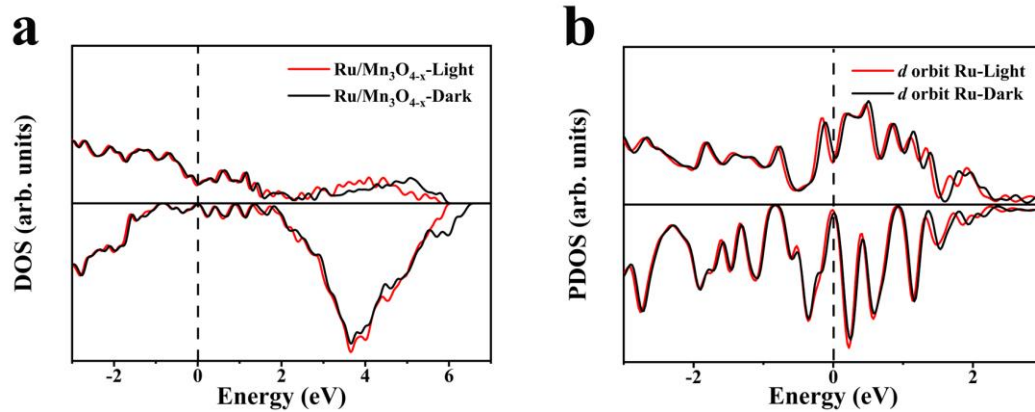
Supplementary Fig. 32 Adsorption configurations of all the involved species on Ru/Mn₃O₄ (321). The blue, red, purple, yellow, and green spheres represent the Mn, O, Ru, C, and H atoms, respectively.



Supplementary Fig. 33 Adsorption configurations of all the involved species on Ru/Mn₃O_{4-x} (321). The blue, red, purple, yellow, and green spheres represent the Mn, O, Ru, C, and H atoms, respectively.



Supplementary Fig. 34 Adsorption configurations of all the involved species on Ru/MnO (200). The blue, red, purple, yellow, and green spheres represent the Mn, O, Ru, C, and H atoms, respectively.



Supplementary Fig. 35 The calculated (a) densities of states and (b) projected densities of states for Ru/Mn₃O_{4-x} under dark and light conditions. Fermi levels are at 0 eV.

Supplementary Table 1 ICP-OES analysis of Ru/MnO_x.

Catalyst	Ru (wt%)	RuCl ₃ ·3H ₂ O addition (mmol)
1.4%Ru/MnO _x	1.4	0.02
2.8%Ru/MnO _x	2.8	0.04
4.0%Ru/MnO _x	4.0	0.06
5.8%Ru/MnO _x	5.8	0.08
7.3%Ru/MnO _x	7.3	0.1
8.1%Ru/MnO _x	8.1	0.12

Supplementary Table 2 Crystal parameters and reliability factors of the refinement for MnO_x and Ru/MnO_x.

Sample	MnO _x			Ru/MnO _x		
	Phase	Mn ₃ O ₄	MnOOH	MnO ₂	Mn ₃ O ₄	MnOOH
Abundance (%)	69.023	25.976	5.002	72.528	27.544	0.928
Space group	I41/amd	P-3m1	C12/m1	I41/amd	P-3m1	C12/m1
a (Å)	5.7702(4)	3.2031(16)	5.1657(61)	5.7698(2)	3.2016(17)	5.1657(61)
b(Å)	5.7702(4)	3.2031(16)	2.8645(61)	5.7698(2)	3.2016(17)	2.8645(33)
c(Å)	9.4544(9)	4.6199(9)	7.0860(32)	9.4490(5)	4.6141(7)	7.0860(32)
Volume(Å ³)	314.796(62)	41.050(42)	104.17(18)	314.569(31)	40.959(44)	104.17(18)
R _{wp}		1.62%			1.71%	
R _p		1.25%			1.33%	
GOF		1.34			1.33	

Supplementary Table 3 The summarized CH₄ yields for recently reported photo-thermo-catalysts.

Catalysts	Metal loading (wt%)	H ₂ :CO ₂ ratio	Pressure (Mpa)	Light sources	Light intensity (W cm ⁻²)	Temperature (°C)	CH ₄ production rate (mmol g ⁻¹ h ⁻¹)	CO ₂ conversion (%)	CH ₄ selectivity (%)	TOF (h ⁻¹)	Ref
Ru/MnO _x	7.3	4:1	1	300 W Xe lamp 200-1100 nm	2.5	200 (external heater)	166.7	66.8	99.5	232	This work
Co ₇ Cu ₁ Mn ₁ O _x (200)	—	3:1	0.1	300 W Xe lamp 300-1100 nm	0.234	200 (external heater)	14.5	27.45	85.3	—	1
Ru/Al ₂ O ₃	2.4	4:1	0.08	1000 W Xe lamp	—	396	115	0.95	99.2	484	2
Cu ₂ O/Graphene	—	4:1	0.13	300 W Xe lamp	0.2	250 (external heater)	14.93 (Cu)	2.84	99	0.256	3
Ru@Ni ₂ V ₂ O ₇	0.35	4:1	0.067	300 W Xe lamp	2	350	114.9	96.3	99.3	3340	4
Ru/Mg(OH) ₂	11.5	1:1	0.1	300 W Xe lamp	1.8	—	44.85	1.68	69.5	56.7	5
Rh/Al	5	3:1	1.5	300 W Xe lamp	11.3	200 (external heater)	550	—	99	1132	6
21%Co/Al ₂ O ₃	0.21	4:1	0.1	300 W Xe lamp 200-1100 nm	1.3	292	6.04	—	97.7	1.74	7
Ru-TiO _x	1.77	4:1	—	300 W Xe lamp	2	276	22.35	—	99.99	12.76	8
Ir@UiO66	0.14	4:1	0.1	300 W Xe lamp	2.3	250 (external heater)	19.9 (Flow reactor)	9.3	95	2876	9
8 % Ru/SiO ₂	0.8	6:1	—	300 W Xe lamp	0.063	300 (external heater)	55.44 (Flow reactor)	51.8	99	70	10
Ru-Al ₂ O _{3-x} -L	0.7	4:1	0.1	300 W Xe lamp	2.27	236	0.84	86.47	99	1248	11
Ru/H _x MoO _{3-y}	4	1:1	—	300 W Xe lamp Vis-IR	0.75	140 (external heater)	20.8 (Flow reactor)	—	99	52.6	12

Supplementary References

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