

## The Acetylene - Ethylene Assay for N<sub>2</sub> Fixation: Laboratory and Field Evaluation<sup>1</sup>

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**Abstract.** The methodology, characteristics and application of the sensitive C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> assay for N<sub>2</sub> fixation by nitrogenase preparations and bacterial cultures in the laboratory and by legumes and free-living bacteria *in situ* is presented in this comprehensive report. This assay is based on the N<sub>2</sub>ase-catalyzed reduction of C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub>, gas chromatographic isolation of C<sub>2</sub>H<sub>2</sub> and C<sub>2</sub>H<sub>4</sub>, and quantitative measurement with a H<sub>2</sub>-flame analyzer. As little as 1 μmole C<sub>2</sub>H<sub>4</sub> can be detected, providing a sensitivity 10<sup>3</sup>-fold greater than is possible with <sup>15</sup>N analysis.

A simple, rapid and effective procedure utilizing syringe-type assay chambers is described for the analysis of C<sub>2</sub>H<sub>2</sub>-reducing activity in the field. Applications to field samples included an evaluation of N<sub>2</sub> fixation by commercially grown soybeans based on over 2000 analyses made during the course of the growing season. Assay values reflected the degree of nodulation of soybean plants and indicated a calculated seasonal N<sub>2</sub> fixation rate of 30 to 33 kg N<sub>2</sub> fixed per acre, in good agreement with literature estimates based on Kjeldahl analyses. The assay was successfully applied to measurements of N<sub>2</sub> fixation by other symbionts and by free living soil microorganisms, and was also used to assess the effects of light and temperature on the N<sub>2</sub> fixing activity of soybeans. The validity of measuring N<sub>2</sub> fixation in terms of C<sub>2</sub>H<sub>2</sub> reduction was established through extensive comparisons of these activities using defined systems, including purified N<sub>2</sub>ase preparations and pure cultures of N<sub>2</sub>-fixing bacteria.

With this assay it now becomes possible and practicable to conduct comprehensive surveys of N<sub>2</sub> fixation, to make detailed comparisons among different N<sub>2</sub>-fixing symbionts, and to rapidly evaluate the effects of cultural practices and environmental factors on N<sub>2</sub> fixation. The knowledge obtained through extensive application of this assay should provide the basis for efforts leading to the maximum agricultural exploitation of the N<sub>2</sub> fixation reaction.

To meet the imminent crisis in the world food supply (38) it is imperative that the resources of this planet be mobilized as rapidly and effectively as possible. Basic to such mobilization is a knowledge of the magnitude of the dynamic processes in the biosphere which affect the availability of nitrogen, the one element most often limiting in the production of foodstuffs (37). Of paramount importance in this context is the process of biological nitrogen fixation. Just as photosynthesis utilizes the freely available CO<sub>2</sub> of the atmosphere, nitrogen fixation draws on the unlimited supply of atmospheric nitrogen, and its potential role in increasing nitrogen availability has long been recognized. In spite of the importance of N<sub>2</sub> fixation very little accurate information is available to define the quantitative extent to which it occurs in the biosphere (35),

and virtually nothing is known concerning the effects of various field practices on N<sub>2</sub> fixation. These gaps in our knowledge are attributable to the absence of effective methods for quantitative measurement of N<sub>2</sub> fixation *in situ*.

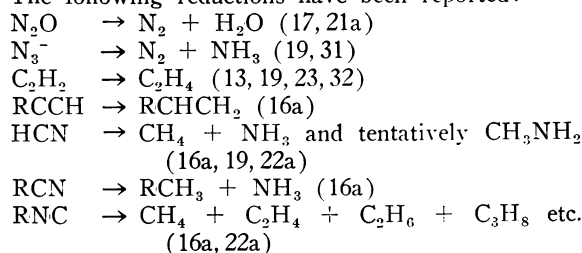
In the laboratory, N<sub>2</sub> fixation by living organisms has been measured by Kjeldahl analysis (7), <sup>15</sup>N-enrichment assayed by mass spectrometry (7), and <sup>13</sup>N-incorporation assayed by radioactive counting (8,30); N<sub>2</sub> fixation by nitrogenase (N<sub>2</sub>ase)<sup>2</sup> in cell-free extracts has been measured by <sup>15</sup>N-enrichment (9), <sup>13</sup>N-incorporation (8,30), micro-Conway diffusion technique coupled with titrimetric (26) or colorimetric analysis of NH<sub>3</sub> (14), and N<sub>2</sub>-H<sub>2</sub> uptake (27) or H<sub>2</sub> evolution (5) assayed manometrically. These methods are relatively insensitive except for the <sup>13</sup>N method, and its application is extremely limited because of its short half-life (10 minutes). Of these procedures only Kjeldahl analysis has been used to an appreciable extent for estimating N<sub>2</sub> fixation in field samples (35), but the method is insensitive and time-consuming. Although isotopic analysis of samples exposed to <sup>15</sup>N<sub>2</sub> in the field has been used to demon-

<sup>1</sup> Contribution No. 1451.

<sup>2</sup> The following abbreviations are used: N<sub>2</sub>ase for nitrogenase; DTT for dithiothreitol; TES for N-tris (hydroxymethyl)-methyl-2-amino-ethanesulfonic acid; and MES for 2-(N-morpholino)ethanesulfonic acid; DEAE for diethylaminoethyl.

strate  $N_2$  fixation *in situ*, the complexity and expense of this method have limited widespread application to field studies.

The opportunity for a novel approach to  $N_2$  fixation analysis arose from demonstrations that  $N_2$ ase is a most versatile reducing catalyst (15, 20). The following reductions have been reported:



Schollhorn and Burris (31) and Dilworth (13) independently observed inhibition of  $N_2$  fixation by  $C_2H_2$  with extracts of *C. pasteurianum*: Schollhorn and Burris (32) established the competitive nature of this inhibition and Dilworth found  $C_2H_2$  to be reduced to  $C_2H_4$  in a reaction analogous to the reduction of  $N_2$  to  $NH_3$ . The application of this reaction to a sensitive assay procedure for  $N_2$ -fixing activity was proposed by Hardy and Knight (19): "Utilization of the reduction of HCN to  $CH_4$ , of  $H^{14}CN$  to  $^{14}CH_3NH_2$ , or of  $C_2H_2$  to  $C_2H_4$ , and detection of  $CH_4$  and  $C_2H_4$  by hydrogen flame ionization after gas chromatography or detection of  $^{14}CH_3NH_2$  may provide a sensitive new assay for detection of the  $N_2$ -fixing system. The gas chromatographic determination makes possible a range of about 10,000 times between minimum and maximum, in contrast to a 20-fold range with the  $NH_3$  assay.  $C_2H_2$  is the preferred assay substrate, since more product is formed because of its requirement for 2 electrons *versus* 6 electrons for HCN". Subsequently, Koch, Evans, and Russell (24, 25), Silver (33), Sloger and Silver (34) and Stewart, Fitzgerald, and Burris (36) have successfully employed  $C_2H_2$  reduction coupled with  $C_2H_4$  detection by  $H_2$  flame ionization as an assay for  $N_2$ ase activity.

Since the original proposal by Hardy and Knight (19), the  $C_2H_2$ - $C_2H_4$  assay of  $N_2$ -fixing activity has undergone extensive development in this laboratory. This paper reports: 1) methodology of  $C_2H_2$  and  $C_2H_4$  analyses; 2) methodology of the  $C_2H_2$ - $C_2H_4$  assay of  $N_2$ -fixing activity *in situ*, and 3) characteristics of  $C_2H_2$  reduction by  $N_2$ ase *in vitro* by cultures of  $N_2$ -fixing bacteria and by samples of the biosphere *in situ*. The results indicate that the  $C_2H_2$ - $C_2H_4$  assay of  $N_2$ -fixing activity is sensitive, universal, specific, rapid, simple, economical, and quantitative. Since this assay has the potential to promote revolutionary fundamental and practical advances, we believe that the  $C_2H_2$ - $C_2H_4$  assay of  $N_2$  fixation represents one of the most important developments in  $N_2$  fixation research. It is emphasized that the adoption of a consistent procedure by the various disciplines, *e.g.*, soil science, agronomy, marine biology, plant biology, microbiology, and

biochemistry, which will utilize this method is essential if valid comparisons are to be made among results obtained from various sources.

## Methods

*Growth of Cells and Preparation of Extracts.* *Azotobacter vinelandii*, ATCC 12518, was grown on nitrogen-free or urea media. Optical densities of bacterial cultures were determined in Will colorimeter tubes with a 650 m $\mu$  filter in a Lumetron colorimeter. Cells were broken and  $N_2$ ase was purified as previously described for  $N_2$ ,  $N_2O$ ,  $N_3^-$ , or HCN reduction experiments (5, 17, 18, 19).  $N_2$ ase was fractionated into the Mo-Fe protein fraction (also called Enzyme I) and Fe protein fraction (also called Enzyme II) (15) by published procedures (4, 22) modified to produce a discrete fractionation of the 2 components. Fractions were designated as follows: 1) crude extract—supernatant after 35,000  $\times g$  for 30 minutes; 2) heated extract—crude extract heated for 10 minutes at 60° under 0.5 atm of  $H_2$  and centrifuged at 35,000  $\times g$  for 1 hour; 3) pre-protamine sulfate precipitate—phosphocellulose resolubilized precipitate between 0.0 to 0.1 mg protamine sulfate per mg protein of heated extract; 4) protamine sulfate precipitate—phosphocellulose resolubilized precipitate between 0.100 to 0.125 mg protamine sulfate per mg protein of heated extract; 5) protamine sulfate supernatant—supernatant after 0.125 mg protamine sulfate per mg protein of heated extract; 6) Mo-Fe protein fraction—fraction of protamine sulfate precipitate eluted from a DEAE-cellulose column in an anaerobic chamber by 0.20 M NaCl + 0.02 M  $MgCl_2$  +  $5 \times 10^{-4}$  M DTT in 0.02 M tris-HCl at pH 7.0 following pre-elution with 0.15 M NaCl + 0.02 M  $MgCl_2$  +  $5 \times 10^{-4}$  M DTT in 0.02 M tris-HCl at pH 7.0; and 7) Fe protein fraction—fraction eluted from above column by 0.35 M NaCl + 0.02 M  $MgCl_2$  +  $5 \times 10^{-4}$  M DTT in 0.02 M tris-HCl at pH 7.0; All fractions were stored anaerobically with fractions 4 and 7 stored at room temperature. Fraction 6 contained Fe and Mo and fraction 7 contained Fe as reported by others (4). Neither fraction was homogeneous by gel electrophoresis.

*Clostridium pasteurianum*, ATCC 6013, was grown on nitrogen-free or  $NH_4Cl$  media (9). Dried cells were broken by autolysis as previously described for  $N_2$ ,  $N_2O$ ,  $N_3^-$ , or HCN reduction experiments (9, 17, 19). An extremely sharp fractionation of  $N_2$ ase into its 2 components was produced by modification of an old procedure (29). Fractions were designated as follows: 1) crude extract—supernatant after 35,000  $\times g$  for 30 minutes of autolysate; 2) negative phosphate gel preparation—supernatant from crude extract treated with protamine sulfate to remove nucleic acids and with calcium phosphate gel to remove inactive protein (29); 3) Mo-Fe protein fraction—fraction eluted from DEAE-cellulose

between 0.1 to 0.3 M NaCl in 0.01 M  $KPO_4$ , pH 7.0, after 15 minutes anaerobic batch treatment of a phosphate gel preparation with 8 mg damp DEAE-cellulose per mg protein. This fraction contains Mo and Fe and is stable for months when stored anaerobically at 4°. In some cases it was further purified by anaerobic chromatography on DEAE-cellulose with elution by 0.29 M NaCl in 0.01 M tris·HCl, pH 7.0. This fraction is not homogeneous by gel electrophoresis. 4) Fe-protein fraction—crude extract heated for 10 minutes at 60° under 0.5 atm  $H_2$  and centrifuged at  $35,000 \times g$  for 30 minutes; this fraction was stored under anaerobic conditions at room temperature. Fraction 3 or 4 could only be obtained from  $N_2$ -grown cells.

*Clostridium butyricum*, *Lactobacillus leichmanii*, *Bacillus subtilis*, *Serratia marcescens*, *Escherichia coli*, *Streptococcus lactis*, *Saccharomyces cerevisiae*, *Bacillus cereus* var. *mycoides*, *Pseudomonas aeruginosa*, *Pseudomonas fluorescens*, *Aerobacter aerogenes*, *Staphylococcus epidermidis*, *Sarcina lutea*, *Spirillum itersonii*, *Proteus vulgaris*, *Alcaligenes faecalis*, and *Rhodospirillum rubrum* were the kind gifts of Dr. R. Bailey of the University of Delaware and were grown on appropriate media containing fixed nitrogen. *Rhizobium japonicum*, ATCC 10324, *R. meliloti*, ATCC 10312, *R. leguminosarum*, ATCC 10004, *Rhizobium sp.*, ATCC 10317, and *R. trifolii*, ATCC 10328, were grown on a medium containing in g per liter:  $K_2HPO_4$ , 1.0;  $KH_2PO_4$ , 1.0;  $MgSO_4 \cdot 7H_2O$ , 0.36;  $CaSO_4 \cdot 2H_2O$ , 0.17;  $FeCl_3 \cdot 6H_2O$ , 0.005;  $KNO_3$ , 0.7; yeast extract, 1.0; and mannitol, 3.0 (10).

*Growth of Legumes.* Field-grown soybeans (*Glycine max* Merr. var. Wayne) were sown on May 15, 1967, in 38-inch rows by a commercial grower in Chester County, Pennsylvania. Standard agricultural practices including seed inoculation with commercial inoculum and recommended additions of phosphorus and potassium but no nitrogen were used. In addition, soybeans and other legumes (*Phaseolus vulgaris*, *Medicago sativa*, *Arachis hypogaea*, and *Pisum sativum*) were grown in sterilized Perlite using a nitrogen-free nutrient solution (1) in a greenhouse or in controlled environment growth chambers. A normal day-night regime of 16 hours, 24°, and 8 hours, 18° was maintained in Sherer-Gillett Model CEI,255-6 chambers operated at maximum light intensity during the light period. Other samples of legumes and soils collected within a radius of 200 miles of Wilmington include samples from the Jordan Fertility Plots through the courtesy of Professors A. Richer and E. S. Lindstrom of Pennsylvania State University, and from the Georgetown Experimental Station through the courtesy of Professor R. Cole of the University of Delaware. Indicated times are Eastern Daylight Saving Time. The date of bud opening was recorded as the date of flowering, and the first indication of leaf yellowing was recorded as the onset of senescence.

*Assays.* Reductions of  $N_2$  or  $C_2H_2$  by  $N_2$ ase preparations or cultures of bacteria were performed in 40 ml incubation vessels sealed with serum caps. For  $N_2$ ase preparations dithionite was dissolved in  $O_2$ -free water containing a pre-determined quantity of acid or base to produce a final pH of 7. The energy source and reductant were placed in the sidearm, the extract and other components were placed in the main compartment, and the incubation flask was immediately evacuated. After repeated flushing with the indicated gas, the contents of the sidearm were tipped in to initiate the reaction. For cultures of bacteria, the incubation was initiated by the aseptic addition of the bacteria to the sealed incubation vessel containing appropriate medium and gas phase. The reaction mixture or culture was incubated on a rotary shaker at 30° for the indicated time, and the incubation was stopped by the addition of 0.5 ml of 6 N  $H_2SO_4$ . Samples of gas phase were analyzed with a mass spectrometer utilizing the initial gas phase as an internal standard or with a  $H_2$ -flame ionization detector after gas chromatographic separation (see below). Nitrogen fixation by  $N_2$ ase in extracts was measured by titration of  $NH_3$  after micro-diffusion (26), and  $N_2$  fixation by cultures was measured by Kjeldahl analysis of 5 ml aliquots. Deuterated ethylenes were analyzed in a Perkin-Elmer Model 21 infrared spectrophotometer using a 3.3 cm micro gas cell.

*Assay of Acetylene and Ethylene.* In early work an activated alumina column at 150° and a Perkin-Elmer 880 or 800 gas chromatograph with a dual  $H_2$ -flame ionization detector were used (19). Subsequently, a one-eighth inch  $\times$  10 foot column containing 20% ethyl, N',N'-dimethyl oxalamide on 100 to 120 mesh acid-washed firebrick at 0° with a He flow rate of 30 ml/minute has been found to be most effective for gas chromatographic separation of acetylene and ethylene as well as other saturated and unsaturated hydrocarbons containing up to 4 carbons. Modified Perkin-Elmer F-11 gas chromatographs equipped with  $H_2$ -flame ionization detectors are utilized. Representative retention times in minutes are: methane, 0.8; ethane, 1.0; ethylene, 1.1; propane, 1.4; propylene, 1.9; isobutane, 2.1; butane, 2.8; acetylene, 3.8; 1-butene, 4.4; isobutylene, 4.5; allene, 4.8; trans-2-butene, 5.2; cis-2-butene, 6.2; methylacetylene, 10.4. A typical chromatogram of a standard mixture of  $C_2H_2$  (0.1 atmosphere) and  $C_2H_4$  ( $2.5 \times 10^{-4}$  atmosphere) is shown in figure 1a, and a chromatogram of  $C_2H_2$  (0.1 atmosphere initial pressure) and  $C_2H_4$  produced by a culture of  $N_2$ -grown *Clostridium pasteurianum* in figure 1b. The symmetry of  $C_2H_2$  and  $C_2H_4$  peaks and the absence of other components are indicated in figure 1b. A standard curve of peak height vs.  $C_2H_4$  or  $C_2H_2$  content of injected sample (fig 2) demonstrates the linear response and sensitivity of the assay. Less than  $10^{-12}$  moles of  $C_2H_4$  can be detected per injected sample of 200  $\mu$ l. Ethylene content can be calculated from this standard curve, or

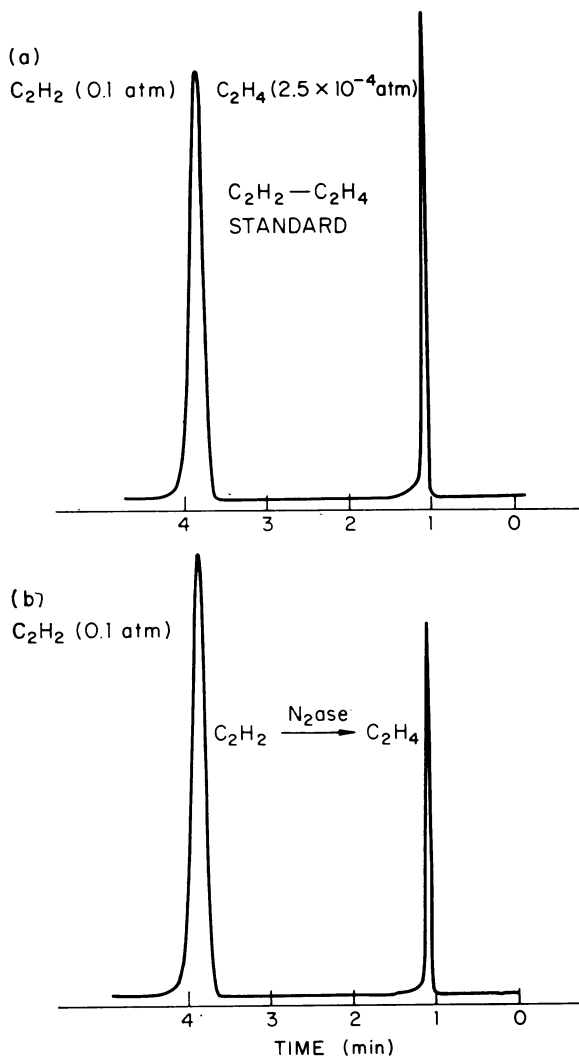


FIG. 1. Typical chromatograms of a) a known mixture of 200  $\mu$ l of  $C_2H_2$  (0.1 atm)  $C_2H_4$  ( $2.5 \times 10^{-4}$  atm) and He to 1 atm, and b) 200  $\mu$ l of the gas phase of an incubation after  $N_2ase$ -catalyzed reduction of 0.1 atm  $C_2H_2$ . An ester-amide gas chromatographic column was used and detection was by hydrogen flame ionization (see under Methods).

alternatively the "built-in" internal standard,  $C_2H_2$ , can be used since both  $C_2H_2$  and  $C_2H_4$  are determined. Our broad experience with the  $C_2H_2$ - $C_2H_4$  assay (over 2000 samples assayed) indicates that  $C_2H_2$  is a valid and useful internal standard, since with the exception of large nodulated plant roots less than 2% of the initial  $C_2H_2$  (0.1 atmosphere) is converted to  $C_2H_4$  during a 1 hour incubation.

**Reagents.** ATP, GTP, CTP, UTP, creatine phosphate, creatine kinase (ATP:creatine phosphotransferase, EC 2.7.1.40), and protamine sulfate were obtained from Sigma Chemical Company;  $Na_2S_2O_4$ , reagent grade, from Fisher Scientific Company; He, A,  $N_2$ , CO, and  $C_2H_2$  as highest purity available

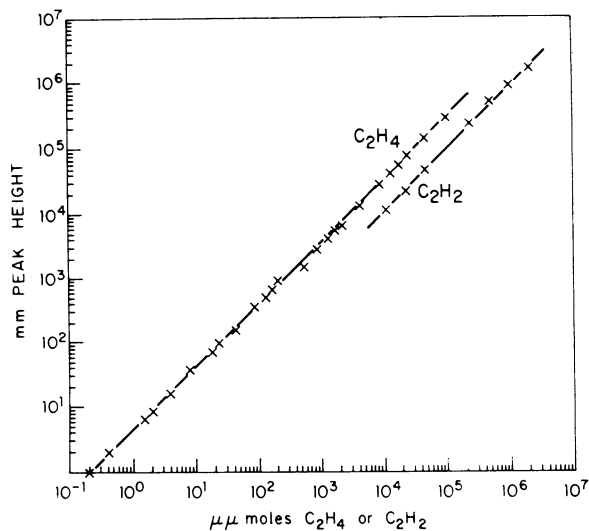


FIG. 2. A standard curve of peak heights of  $C_2H_2$  and  $C_2H_4$  determined with the gas chromatographic system of figure 1.

from The Matheson Company. Acetone was removed from  $C_2H_2$  by a concentrated  $H_2SO_4$ -scrubber and corrections were made for the  $C_2H_4$  content of  $C_2H_2$ . The  $C_2H_4$  impurity in  $C_2H_2$  from a given cylinder must be determined daily since it varies inversely with the pressure in the cylinder.

## Results

**$C_2H_2$  Reduction by  $N_2ase$  in vitro.** Reduction of  $C_2H_2$  to  $C_2H_4$  by  $N_2ase$  of cell-free extracts of *A. vinelandii* was examined with respect to a wide variety of characteristics, and the striking similarities between  $N_2$  fixation and  $C_2H_2$  reduction are reported in this section.

**Requirements and Products of  $C_2H_2$  Reduction.** Reduction of  $C_2H_2$  to  $C_2H_4$ , like reduction of  $N_2$  to 2  $NH_3$  (3, 12, 16, 25, 27), requires an enzyme extract containing  $N_2ase$ , an energy source, and a reductant (table I). A similar energy and reductant requirement has been reported for reduction of acetylene by extracts of *C. pasteurianum* and soybean bacteroids (13, 25, 29a, 32). Extracts of urea-grown cells do not have  $N_2ase$  activity and do not have  $C_2H_2$ -reducing activity. No  $C_2H_4$  formation is found in the absence of  $C_2H_2$ . Detectable amounts of  $C_2H_6$  or  $CH_4$  are not formed by the complete system capable of reducing  $C_2H_2$  (sensitive analyses indicate that  $C_2H_6$  can be no more than 0.01% as abundant as  $C_2H_4$ ); furthermore, ethylene is not reduced to  $C_2H_6$  or  $CH_4$  by *Azotobacter*  $N_2ase$  in a complete system. Thus, the  $N_2ase$ -catalyzed reaction appears to be quite specific for the reduction of  $C_2H_2$  only to  $C_2H_4$ .

**Specific Requirement for ATP.** The sensitivity of the  $C_2H_2$ - $C_2H_4$  assay of  $N_2ase$  permits a deter-

Table I. *Requirements and Product of C<sub>2</sub>H<sub>2</sub> Reduction by Azotobacter N<sub>2</sub>ase*

Complete system contained per ml in  $\mu$ moles: tris $\cdot$ HCl, 50; creatine phosphate (CrP), 56; ATP, 5; Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub>, 20 (all at pH 7.0); and MgCl<sub>2</sub>, 5; and in mg proteins: heated extract of N<sub>2</sub>-grown or NH<sub>3</sub>-grown *A. vinelandii*, as indicated, 4; and creatine kinase (CrK), 0.2. Gas volume 36 ml; liquid volume, 4 ml; incubation time, 30 min; temperature, 30°. Aliquots of gas phase assayed by gas chromatography on alumina column.

Incubation system	Gas phase	C <sub>2</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>6</sub>	CH <sub>4</sub>
Requirements complete, N <sub>2</sub> -grown	0.05 atm C <sub>2</sub> H <sub>2</sub> 0.95 atm He	24	$\mu$ moles/incubation <0.001	<0.002
Minus enzyme or CRP, CrK, ATP or Na <sub>2</sub> S <sub>2</sub> O <sub>4</sub>	0.05 atm C <sub>2</sub> H <sub>2</sub> 0.95 atm He	<0.001	<0.001	<0.002
Minus C <sub>2</sub> H <sub>2</sub>	1.0 atm He	<0.001	<0.001	<0.002
Specificity for N <sub>2</sub> -grown cells complete, N <sub>2</sub> -grown	0.1 atm C <sub>2</sub> H <sub>2</sub>	20.2	<0.001	<0.002
complete, NH <sub>3</sub> -grown	0.1 atm C <sub>2</sub> H <sub>2</sub>	<0.001	<0.001	<0.002
Stability of C <sub>2</sub> H <sub>4</sub> complete, N <sub>2</sub> -grown	0.1 atm C <sub>2</sub> H <sub>4</sub> 0.9 atm He	150	<0.001	<0.002

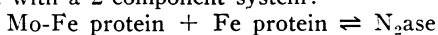
Table II. *Specificity of Phosphagen Requirement for C<sub>2</sub>H<sub>2</sub> Reduction by Azotobacter N<sub>2</sub>ase*

Complete incubation system, table I, except replacement of creatine phosphate, and creatine kinase by 10 mM ATP, GTP, CTP, or UTP, pH 7.0; 2.2 mg protein of a protamine ppt of N<sub>2</sub>ase of *A. vinelandii*. Liquid volume, 2 ml; gas volume, 38 ml; incubation time, 5 min; gas phase, 0.1 atm C<sub>2</sub>H<sub>2</sub>, 0.9 atm He. Aliquots of gas phase assayed by gas chromatography on esteramide column.

Phosphagen	$\mu$ moles C <sub>2</sub> H <sub>4</sub> /incubation
ATP	1440
CTP	0.8
GTP	<0.5
UTP	<0.5

mination of the specificity of its requirement for nucleoside triphosphate. An *Azotobacter* N<sub>2</sub>ase preparation was incubated with C<sub>2</sub>H<sub>2</sub>, reductant and ATP, UTP, CTP, or GTP as the sole energy source (table II). Only ATP supported C<sub>2</sub>H<sub>2</sub> reduction indicating that the energy requirement of N<sub>2</sub>ase is very specific for ATP. Burns has observed similar phosphagen specificity for the energy-dependent H<sub>2</sub>-evolution activity of N<sub>2</sub>ase (15).

**Enzyme Level.** The rate of C<sub>2</sub>H<sub>2</sub> reduction is related to N<sub>2</sub>ase concentration in a sigmoidal fashion (fig 3a); the plot of the rate of N<sub>2</sub> fixation *vs.* enzyme concentration does not extrapolate linearly to zero enzyme (3, 19) and might show a sigmoidal relationship if the NH<sub>3</sub> assay were sufficiently sensitive to measure NH<sub>3</sub> formation at the lower limits of enzyme concentration. The plot of the rate of C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub> reduction *vs.* enzyme is consistent with a 2-component system:



**Time Course.** The rate of C<sub>2</sub>H<sub>2</sub> reduction *vs.* time is linear for about 45 minutes, and the reaction

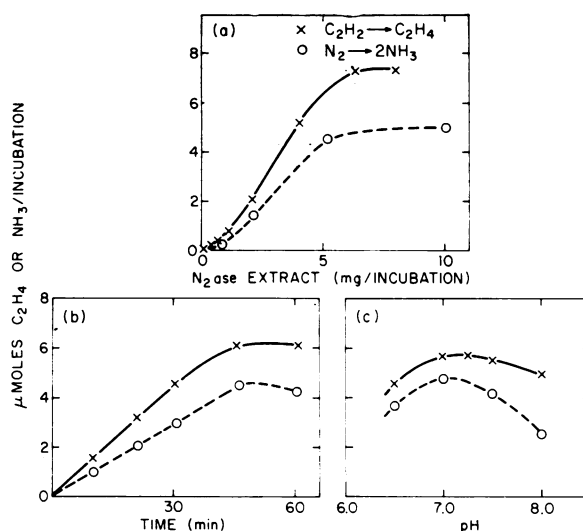


FIG. 3. Reduction by *Azotobacter* N<sub>2</sub>ase of C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub> and N<sub>2</sub> to 2NH<sub>3</sub> versus a) enzyme level, b) time, and c) pH. Complete incubation system, table I; liquid volume, 2 ml; gas volume, 38 ml; time, 30 min unless otherwise indicated; gas phase, 0.1 atm C<sub>2</sub>H<sub>2</sub> + 0.9 atm He for C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub> and 1 atm of N<sub>2</sub> with 1 atm He as control for N<sub>2</sub> to 2NH<sub>3</sub>. Mixture of TES and MES used to vary pH from 6.5 to 8.0.

stops when ATP is exhausted, in complete analogy to the N<sub>2</sub> reduction reaction (fig 3b).

**pH Maximum.** Nitrogenase-catalyzed C<sub>2</sub>H<sub>2</sub> reduction was examined from pH 6.5 to pH 8.0 (fig 3c). Maximum activity occurred over a wide range near pH 7.0, which is similar to that observed for N<sub>2</sub> (3) or other reducible substrates of N<sub>2</sub>ase (21).

**K<sub>m</sub> of C<sub>2</sub>H<sub>2</sub>.** Nitrogenase is saturated by 0.03 to 0.10 atmosphere C<sub>2</sub>H<sub>2</sub> (fig 4a) while 0.5 atmosphere has been observed to inhibit C<sub>2</sub>H<sub>2</sub> reduction

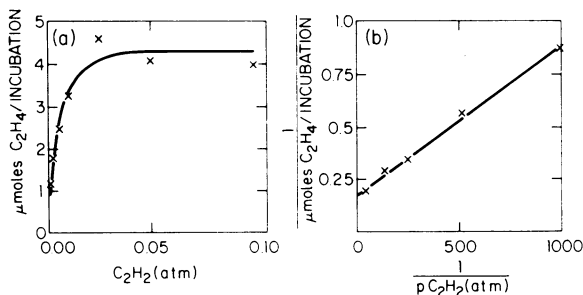


FIG. 4. a) Ethylene formation from C<sub>2</sub>H<sub>2</sub> as a function of pC<sub>2</sub>H<sub>2</sub> by *Azotobacter* N<sub>2</sub>ase, and b) plot of reciprocal velocity versus reciprocal pC<sub>2</sub>H<sub>2</sub> for determination of K<sub>m</sub> of C<sub>2</sub>H<sub>2</sub>. Complete incubation system, table I; liquid volume, 2 ml; gas volume, 38 ml; time, 30 min; gas phase, indicated pC<sub>2</sub>H<sub>2</sub> plus He to 1 atm.

and reductant-dependent ATPase activity of N<sub>2</sub>ase. This inhibition at 0.5 atmosphere C<sub>2</sub>H<sub>2</sub> may be due to C<sub>2</sub>H<sub>2</sub> or to possible trace impurities in C<sub>2</sub>H<sub>2</sub>. A typical plot of reciprocal velocity of C<sub>2</sub>H<sub>2</sub> reduction vs. reciprocal pC<sub>2</sub>H<sub>2</sub> is shown in figure 4b. Estimated Michaelis constants of 0.002 to 0.009 atmosphere of C<sub>2</sub>H<sub>2</sub>, with an average value of 0.004 atmosphere, have been obtained. A tentative K<sub>m</sub> of 0.01 atm has been reported for clostridial

Table III. Inhibition of ATP-Dependent H<sub>2</sub> Evolution by N<sub>2</sub> and C<sub>2</sub>H<sub>2</sub>

Complete system, table I; liquid volume 2 ml; gas volume 38 ml; N<sub>2</sub>ase preparation, heated extract of *A. vinelandii*, 7.5 mg; gas phase, as indicated plus He to 1 atm. Hydrogen determined by mass spectrometric analysis of gas phase with He as an internal standard.

Added substrate	H <sub>2</sub>	ΔH <sub>2</sub>	% Inhibition of H <sub>2</sub> evolution
	μmoles per min per mg protein		
None	71	0	0
N <sub>2</sub> , 0.5 atm	18	53	75
C <sub>2</sub> H <sub>2</sub> , 0.1 atm	11	61	85

Table IV. Stoichiometry of C<sub>2</sub>H<sub>2</sub> Reduction by *Azotobacter* N<sub>2</sub>ase

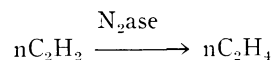
Complete system, table I; liquid volume, 4 ml; gas volume, 36 ml; incubation time, 30 min; aliquots of gas phase assayed by mass spectrometry.

Incubation system	Gas phase	C <sub>2</sub> H <sub>2</sub>	C <sub>2</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>6</sub>	CH <sub>4</sub>	H <sub>2</sub>
		μmoles/incubation				
Complete	0.02 atm C <sub>2</sub> H <sub>2</sub> 0.18 atm He	15.5	14.5	0.0	0.0	11.0
Minus enzyme or CrP, CrK, ATP, or Na <sub>2</sub> S <sub>2</sub> O <sub>4</sub>	0.02 atm C <sub>2</sub> H <sub>2</sub> 0.18 atm He	29.3	0.0	0.0	0.0	0.0
Minus C <sub>2</sub> H <sub>2</sub>	0.2 atm He	0.0	0.0	0.0	0.0	23.3
Complete	0.02 atm C <sub>2</sub> H <sub>2</sub> 0.18 atm CO	25.0	0.0	0.0	0.0	24.7

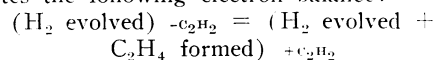
N<sub>2</sub>ase (13). Michaelis constants of 0.05 to 0.17 atmosphere of N<sub>2</sub> have been reported for N<sub>2</sub>ase (14, 19, 25, 27). Based on partial pressures, the estimated K<sub>m</sub> of C<sub>2</sub>H<sub>2</sub> is only about 5% that of N<sub>2</sub>. Based on the calculated concentrations of C<sub>2</sub>H<sub>2</sub> and N<sub>2</sub> in an aqueous solution, the estimated K<sub>m</sub> of C<sub>2</sub>H<sub>2</sub>, 0.1 to 0.3 mM, is similar to that of N<sub>2</sub>, 0.03 to 0.1 mM.

**Inhibition of H<sub>2</sub> Evolution.** The ATP-dependent H<sub>2</sub>-evolving activity of N<sub>2</sub>ase is decreased by N<sub>2</sub>, N<sub>2</sub>O, or N<sub>2</sub>O reduction, and the decrease in H<sub>2</sub> evolved is equivalent in electrons to those required for reduction of N<sub>2</sub> to 2 NH<sub>3</sub>, N<sub>2</sub> to N<sub>2</sub> + NH<sub>3</sub>, or N<sub>2</sub>O to N<sub>2</sub> + H<sub>2</sub>O (4, 17, 19, 25). Reduction of C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub> also inhibits H<sub>2</sub> evolution by clostridial (13, 32) and *Azotobacter* N<sub>2</sub>ase (21). Inhibition by a saturating level of C<sub>2</sub>H<sub>2</sub> may be greater than by that of N<sub>2</sub>, e.g., 85% for C<sub>2</sub>H<sub>2</sub> and 75% for N<sub>2</sub> (table III).

**Stoichiometry of C<sub>2</sub>H<sub>2</sub> Reduction.** An excellent balance exists between the concomitant decrease in C<sub>2</sub>H<sub>2</sub>, increase in C<sub>2</sub>H<sub>4</sub>, and decrease in H<sub>2</sub> evolution (table IV) during N<sub>2</sub>ase-catalyzed reduction of C<sub>2</sub>H<sub>2</sub>. Thus, reduction of C<sub>2</sub>H<sub>2</sub> decreased C<sub>2</sub>H<sub>2</sub> by 13.8 μmoles and increased C<sub>2</sub>H<sub>4</sub> by 14.5 μmoles, supporting the following relationship:



The decrease of 12.3 μmoles in H<sub>2</sub> evolution (equivalent to 24.6 μmoles of electrons) produced by C<sub>2</sub>H<sub>2</sub> reduction corresponds to the formation of 14.5 μmoles of C<sub>2</sub>H<sub>4</sub> (equivalent to 29 μmoles of electrons) and indicates the following electron balance:



Addition of 0.18 atmosphere CO inhibited C<sub>2</sub>H<sub>2</sub> reduction and restored H<sub>2</sub> evolution. Since the loss of C<sub>2</sub>H<sub>2</sub> can be accounted for as C<sub>2</sub>H<sub>4</sub> and the loss of electrons evolved as H<sub>2</sub> can be accounted for as the electrons required for C<sub>2</sub>H<sub>2</sub> reduction, no significant product of C<sub>2</sub>H<sub>2</sub> reduction in addition to C<sub>2</sub>H<sub>4</sub> is indicated. Furthermore, the equivalence between the decrease in electrons evolved as H<sub>2</sub> and

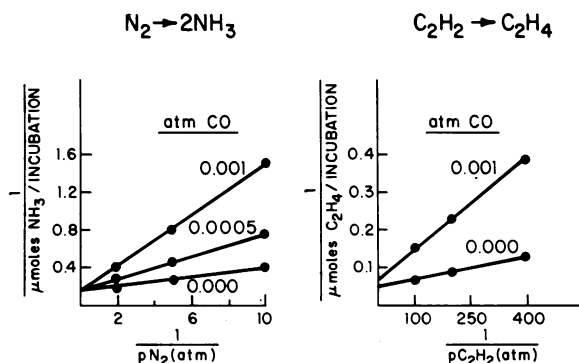


FIG. 5. Competitive inhibition of  $C_2H_2 \rightarrow C_2H_4$  and  $N_2 \rightarrow 2NH_3$  by CO using *Azotobacter*  $N_2$ ase. Complete incubation system, table I; liquid volume, 2 ml; gas volume, 38 ml; time, 30 min; gas phase, indicated  $pC_2H_2$  or  $pN_2$  plus indicated  $pCO$  and He or A to 1 atm.  $N_2$  is replaced by A as control for  $N_2 \rightarrow 2NH_3$ .

the electrons used for ethylene formation indicates that at least the proposed electron-activating site of  $N_2$ ase (15,20) is involved in  $C_2H_2$  reduction, as has been proposed for  $N_2O$  and  $N_3^-$  reductions (17, 19, 21a).

**Inhibition of  $C_2H_2$  Reduction by CO.** Carbon monoxide is a competitive inhibitor of  $N_2$  fixation (15, 20, 25a). Figure 5 demonstrates that CO is also a competitive inhibitor of  $C_2H_2$  reduction by *Azotobacter*  $N_2$ ase. Furthermore, the similar CO inhibition constants of  $2.9 \times 10^{-4}$  and  $3.1 \times 10^{-4}$  atmosphere for  $N_2$  fixation and  $C_2H_2$  reduction, respectively, provide indirect support that the substrate-complexing site of  $N_2$ ase (15, 20) is involved in the reduction of  $C_2H_2$  as well as  $N_2$ .

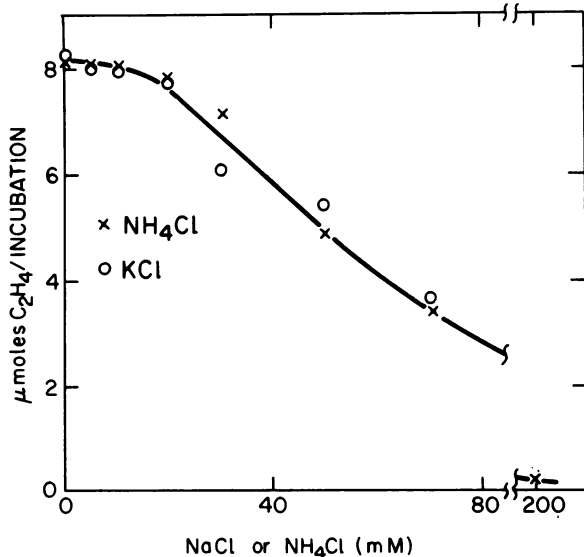


FIG. 6. Inhibition of  $C_2H_2 \rightarrow C_2H_4$  by  $NH_4Cl$  or NaCl. Complete incubation system with *Azotobacter*  $N_2$ ase, table I; liquid volume, 2 ml; gas volume, 38 ml; time, 30 min; gas phase, 0.1 atm  $C_2H_2$ , 0.9 atm He;  $NH_4Cl$  or KCl as indicated.

**Effect of  $NH_4^+$  and  $Na^+$  on  $C_2H_2 \rightarrow C_2H_4$  Reduction.** Ammonia is the product of  $N_2$  fixation; however,  $N_2$  fixation appears to be relatively insensitive to added  $NH_4^+$  (9). This insensitivity suggests that the product of  $N_2$  fixation does not effectively compete with  $N_2$  for the substrate-complexing site of  $N_2$ ase and that  $NH_4^+$  does not control activities of  $N_2$ ase associated with electron-activation. Reduction of  $C_2H_2$  to  $C_2H_4$  provides an opportunity to determine if there is a specific effect of  $NH_4^+$  on other  $N_2$ ase-catalyzed reductions. Figure 6 indicates no specific inhibition by  $NH_4^+$ , since  $C_2H_2$  reduction is equally sensitive to  $NH_4^+$  or  $Na^+$  with 50% inhibition produced by 50 to 70 mM  $NH_4Cl$  or NaCl.

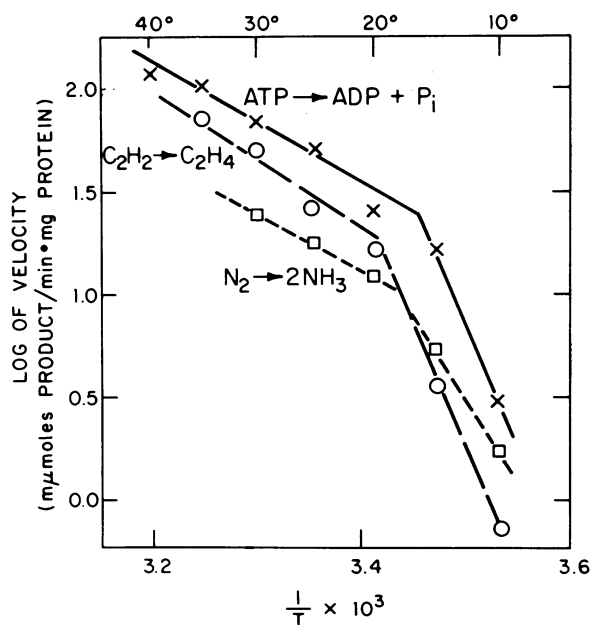


FIG. 7. Arrhenius plots of  $C_2H_2 \rightarrow C_2H_4$ ,  $N_2 \rightarrow 2NH_3$  and  $ATP \rightarrow ADP + P_i$  by *Azotobacter*  $N_2$ ase in the range  $10^\circ$  to  $40^\circ$ . Complete incubation system, table I; liquid volume, 2 ml; gas volume, 38 ml; time, 30 min for  $C_2H_2$  or  $N_2$  reduction, 15 min for ATP hydrolysis; gas phase, 0.1 atm  $C_2H_2$  plus He to 1 atm for  $C_2H_2 \rightarrow C_2H_4$ , 1 atm  $N_2$  with 1 atm He as control for  $N_2 \rightarrow 2NH_3$ , and 1 atm He for  $ATP \rightarrow ADP + P_i$ . Points represent averages of 3 samples.

**Activation Energy of  $C_2H_2$  Reduction.** The activation energies for reduction of  $N_2$  and other reactions of  $N_2$ ase, including ATP-dependent  $H_2$  evolution and reductant-dependent ATPase, have been recently determined (6, 21). A break in the Arrhenius plots for all these activities is observed near  $20^\circ$  with similar but lower activation energies above (13–15 kcal/mole) and similar but higher activation energies below this point (35–50 kcal/mole). Arrhenius plots of  $C_2H_2$  reduction also show a similar break and similar activation energy (fig 7).

*Stereochemistry of C<sub>2</sub>H<sub>2</sub> Reduction.* Ethylene formed from C<sub>2</sub>H<sub>2</sub> by reduction by *Azotobacter* N<sub>2</sub>ase in a 99.8% D<sub>2</sub>O system was examined by infrared spectrophotometry in order to identify the deuterated species (fig 8). *cis*-1,2-Dideuteroethylene (843 cm<sup>-1</sup>) is the major product, as reported for clostridial N<sub>2</sub>ase (13). A small amount of mono-

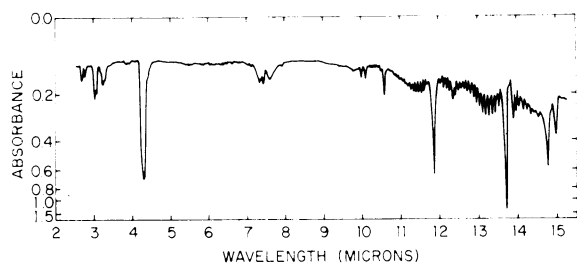


FIG. 8. Infrared spectrum of deuterated ethylenes produced by reduction of C<sub>2</sub>H<sub>2</sub> by *Azotobacter* N<sub>2</sub>ase. Complete incubation system, table I; liquid volume, 10 ml; gas volume, 30 ml; time, 30 min; gas phase 0.1 atm C<sub>2</sub>H<sub>2</sub>; all reagents in 99.8% D<sub>2</sub>O and protamine sulfate ppt of N<sub>2</sub>ase resuspended in 99.8% D<sub>2</sub>O.

deuteroethylene (1000 cm<sup>-1</sup>) and a possible trace of *trans*-1,2-dideuteroethylene (988 cm<sup>-1</sup>) were found. These results indicate that neither of the original hydrogens of acetylene is replaced during reduction and that acetylene may be complexed to the substrate-complexing site of N<sub>2</sub>ase *via* a "side-on" orientation.

*Fractionation of N<sub>2</sub>ase and N<sub>2</sub>-Fixing and C<sub>2</sub>H<sub>2</sub>-Reducing Activities.* Nitrogen-fixing extracts of *A. vinelandii* were fractionated according to the established procedure in this laboratory for N<sub>2</sub>ase purification. The N<sub>2</sub>- and C<sub>2</sub>H<sub>2</sub>-reducing activities paralleled each other, and the ratio of C<sub>2</sub>H<sub>4</sub> formed to N<sub>2</sub> fixed was found to be in the range of 3 to 4.5 (table V).

*Recombination of Mo-Fe Protein and Fe Protein Fractions of N<sub>2</sub>ase.* Nitrogenase can be separated into 2 protein fractions (4,22). One contains Fe and Mo and is called the Mo-Fe protein fraction; the other contains Fe and is called the Fe protein fraction. Neither individual protein fraction has biological activity, but N<sub>2</sub>ase, the complex formed by the protein fractions, is active for N<sub>2</sub> reduction.

Table V. C<sub>2</sub>H<sub>2</sub> and N<sub>2</sub> Reduction by *Azotobacter* N<sub>2</sub>ase Preparations

Complete incubation system, table I; liquid volume, 2 ml; gas volume, 38 ml; N<sub>2</sub>ase preparation, as indicated; gas phase, 0.1 atm C<sub>2</sub>H<sub>2</sub>, 0.9 atm He for C<sub>2</sub>H<sub>2</sub> reduction, 1 atm N<sub>2</sub> with 1 atm He as control for N<sub>2</sub> fixation. C<sub>2</sub>H<sub>4</sub> assayed gas chromatographically; NH<sub>3</sub> assayed titrimetrically.

N <sub>2</sub> ase preparation	Protein mg/incubation	C <sub>2</sub> H <sub>2</sub> →C <sub>2</sub> H <sub>4</sub> μmoles/incubation	N <sub>2</sub> →2NH <sub>3</sub> μmoles/incubation	C <sub>2</sub> H <sub>2</sub> →C <sub>2</sub> H <sub>4</sub> / N <sub>2</sub> →2NH <sub>3</sub>
Heated extract	4.3	5.00	1.35	3.7
Pre-protamine precipitate	4.0	<0.002	0.00	
Protamine precipitate	0.91	6.02	1.43	4.2
Protamine supernatant	3.6	<0.002	0.00	

Table VI. C<sub>2</sub>H<sub>2</sub> Reduction by Mo-Fe Protein and Fe Protein Fractions of *Azotobacter* and *Clostridial* N<sub>2</sub>ase

Complete incubation system, table I; liquid volume, 2 ml; gas volume, 38 ml; incubation time, 30 min; gas phase, 0.1 atm C<sub>2</sub>H<sub>2</sub>, 0.9 atm He; protein fraction as indicated (see under Methods for preparation and designation of fractions). Data from 2 experiments are shown. C<sub>2</sub>H<sub>4</sub> assayed gas chromatographically on ester-amide column.

Fraction	Organism	μmoles C <sub>2</sub> H <sub>4</sub> /incubation	mg Protein/incubation
Crude extract	<i>Clostridium</i>	1.38	8.7
Mo-Fe protein	"	0.0007	1.5
Fe-protein	"	0.0073	5.0-8.5
Mo-Fe protein + Fe protein	"	1.72	1.5
Protamine ppt.	<i>Azotobacter</i>	7.10	5.0
Mo-Fe protein	"	0.152	4.2
Fe protein	"	0.032	0.0008
Mo-Fe protein + Fe protein	"	4.55	0.0214
Mo-Fe protein + Fe protein	"		2.73
Mo-Fe protein + Fe protein	<i>Azotobacter</i>	...	2.4-4.5
Mo-Fe protein + Fe protein	<i>Clostridium</i>		2.2-2.9
Mo-Fe protein + Fe protein	<i>Clostridium</i>	0.102	2.4-4.5
Mo-Fe protein + Fe protein	<i>Azotobacter</i>		2.2-2.9



ATP-dependent H<sub>2</sub> evolution, and reductant-dependent ATPase (4,22). Nitrogenases of *C. pasteurianum* and *A. vinelandii* were separated into their Mo-Fe protein and Fe protein fractions as described under Methods. Acetylene reduction was determined with the individual fractions, the recombination of the individual fractions from the same species and the cross-combination of the proteins from different species (table VI). No nitrogen-fixing activity remained in the individual fractions, but it was restored by recombination of the fractions. Neither the Mo-Fe protein fraction nor the Fe protein fraction has appreciable C<sub>2</sub>H<sub>2</sub>-reducing activity, and thus represent the "lowest activity" fractions that have been reported, e.g., our best preparations of Mo-Fe protein (*Clostridium*), Fe protein (*Clostridium*), Mo-Fe protein (*Azotobacter*), and Fe protein (*Azotobacter*) have less than 0.04, 0.02, 0.03, and 0.7%, respectively, of their recombined activities. Recombination of the 2 protein fractions of *Azotobacter* produces stimulations up to 123-fold, and of *Clostridium* up to 1080-fold. A recent report indicates an enhancement of activity of 5.5-fold by recombination of *Azotobacter* fractions (22). Cross-combination of Mo-Fe protein (*Azotobacter*) + Fe protein (*Clostridium*) or Mo-Fe protein (*Clostridium*) + Fe protein (*Azotobacter*) produces <~5% of the C<sub>2</sub>H<sub>2</sub>-reducing activity found in the recombination within species experiment. The above cross-combinations have been reported to produce no N<sub>2</sub>-fixing activity (11).

These experiments were designed to demonstrate the absence of C<sub>2</sub>H<sub>2</sub>-reducing activity in each fraction and the presence of this activity in recombined fractions from the same species. Specific activities were not maximized by the addition of an excess of one fraction to a limiting amount of the fraction whose activity is to be maximized.

**C<sub>2</sub>H<sub>2</sub> Reduction by Bacterial Cells.** Characteristics of the reduction of C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub> by N<sub>2</sub>-fixing cultures of *A. vinelandii* and *C. pasteurianum* are

reported in this section. These results indicate the validity of the C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> assay of N<sub>2</sub> fixation with living organisms and complement results reported in the previous section with N<sub>2</sub>ase preparations from *Azotobacter* and *Clostridium*.

**Requirements and Products of C<sub>2</sub>H<sub>2</sub> Reduction.** Acetylene is reduced to C<sub>2</sub>H<sub>4</sub> by N<sub>2</sub>-grown cells of *A. vinelandii* and *C. pasteurianum* (table VII). Extracts from these cells contain N<sub>2</sub>ase and reduce C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub>. Control cultures grown on fixed nitrogen sources reduced less than 0.1% (*Azotobacter*) and less than 5% (*Clostridium*) the amount of C<sub>2</sub>H<sub>2</sub> reduced by the N<sub>2</sub>-grown cultures. Extracts from these cells contain little or no N<sub>2</sub>ase and do not reduce C<sub>2</sub>H<sub>2</sub>. Neither ethane nor methane is detected as a product of acetylene reduction; ethylene is not reduced to ethane or methane. The decrease in acetylene during reduction equals the increase in ethylene. The aerobic *Azotobacter* requires aerobic conditions for C<sub>2</sub>H<sub>2</sub> reduction, while the anaerobe *Clostridium* reduces acetylene anaerobically. Negligible ethylene is formed in the absence of C<sub>2</sub>H<sub>2</sub>. Thus, no correction is required for background C<sub>2</sub>H<sub>4</sub>.

**Time Course.** Time courses of acetylene reduction by N<sub>2</sub>-grown *Azotobacter* and *Clostridium* are shown in figures 9, 10 and 13. The rate of C<sub>2</sub>H<sub>2</sub> reduction is constant up to 18 to 20 hours for *Azotobacter* and 6 hours for *Clostridium*. An initial lag is often observed with *Azotobacter*, presumably because of the effect of transfer and dilution. Addition of 40 mM NH<sub>4</sub>Cl to N<sub>2</sub>-grown cells decreases the rate of C<sub>2</sub>H<sub>2</sub> reduction by 95% after 4 hours (fig 10). This inhibition is in contrast to the effect on N<sub>2</sub>ase *in vitro*. The C<sub>2</sub>H<sub>2</sub> reduction assay offers a potent method to further define the relationship of N<sub>2</sub> and fixed nitrogen compounds to induction and repression of N<sub>2</sub>ase. The results with *Azotobacter* in figure 9 permit calculation of a correlation between C<sub>2</sub>H<sub>2</sub> reduction and N<sub>2</sub> fixation. Ethylene formation stops when 0.2 moles of C<sub>2</sub>H<sub>4</sub> have been formed for each mole of O<sub>2</sub> initially

Table VII. *Distribution of and Requirements for C<sub>2</sub>H<sub>2</sub> Reduction by Cultures of Azotobacter and Clostridium*

One ml of culture in early log phase was aseptically added to 4 ml of its nitrogen-free growth media for N<sub>2</sub>-grown bacteria and its nitrogen supplemented media for NH<sub>3</sub>- or urea-grown bacteria in a sealed incubation vessel of 40 ml total volume containing the indicated gas phase. Incubation time, 16 hr, temperature, 30°. Gas phase assayed chromatographically on ester-amide column.

Organism	Cells (× 10 <sup>-6</sup> )/incubation	Gas phase (atm)				μmoles C <sub>2</sub> H <sub>4</sub> /hr
		A	O <sub>2</sub>	C <sub>2</sub> H <sub>2</sub>	He	
<i>Azotobacter</i> , N <sub>2</sub> -grown	85	0.8	0.2			0.000074
<i>Azotobacter</i> , N <sub>2</sub> -grown	85			0.1	0.9	0.030
<i>Azotobacter</i> , N <sub>2</sub> -grown	85	0.7	0.2	0.1		1.42
<i>Clostridium</i> , N <sub>2</sub> -grown	88				1.0	<0.000001
<i>Clostridium</i> , N <sub>2</sub> -grown	88			0.1	0.9	1.05
<i>Azotobacter</i> , N <sub>2</sub> -grown	125	0.7	0.2	0.1		1.025
<i>Azotobacter</i> , Urea-grown	240	0.7	0.2	0.1		0.0009
<i>Clostridium</i> , N <sub>2</sub> -grown	115			0.1	0.9	2.33
<i>Clostridium</i> , NH <sub>3</sub> -grown	210			0.1	0.9	0.12

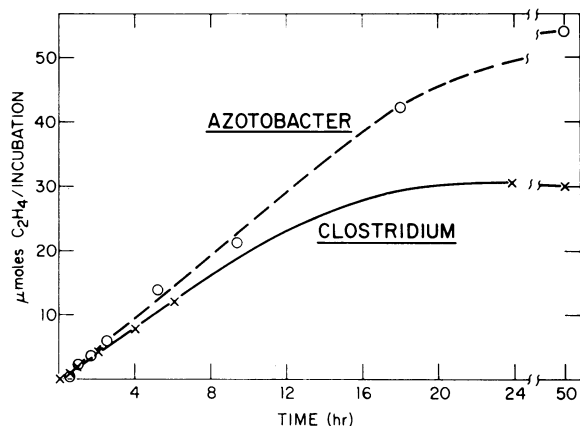


FIG. 9. Time course of  $C_2H_2 \rightarrow C_2H_4$  reduction by cultures of  $N_2$ -grown *A. vinelandii* and *C. pasteurianum*. Incubation system, table VII; liquid volume, 5 ml; gas volume, 35 ml; temperature,  $30^\circ$ ; gas phase, 0.1 atm  $C_2H_2$ , 0.2 atm  $O_2$ , 0.7 atm A for *Azotobacter* and 0.1 atm  $C_2H_2$ , 0.9 atm He for *Clostridium*; *Azotobacter*,  $250 \times 10^6$  cells; *Clostridium*,  $110 \times 10^6$  cells. Points represent averages of 3 samples.

present. Based on aerobic oxidation of glucose, 1.25 moles of ethylene are formed per mole of glucose oxidized. On the basis of one mole of  $N_2$  fixed per 3 to 4 moles of  $C_2H_2$  reduced, 0.3 to 0.4 mole of  $N_2$  would be fixed per mole of glucose oxidized. This calculated ratio of  $N_2$  fixed per glucose oxidized, based on  $C_2H_2$  reduction, is in reasonable agreement with reported experimental values based on direct measurements of  $N_2$  fixation (35).

**Cell Number.** A linear relationship exists between cell number and acetylene reduced (table VIII). Ethylene formation measured after 16 hours

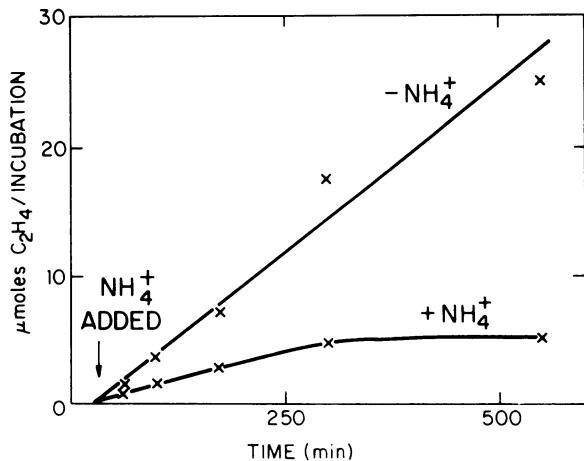


FIG. 10. Effect of  $NH_4Cl$  on time course of  $C_2H_2 \rightarrow C_2H_4$  reduction by a culture of  $N_2$ -grown *A. vinelandii*. Incubation system, figure 9;  $300 \times 10^6$  cells;  $NH_4Cl$  added to 3 flasks at 30 min to produce 40 mM  $NH_4^+$  while no addition was made to 3 control flasks.

Table VIII. Cell Dilution and  $C_2H_2$  Reduction by *Azotobacter* Culture

Incubation system, table VII, for  $N_2$ -grown *Azotobacter*; incubation time, 16 hr; temperature,  $30^\circ$ ; gas phase, 0.1 atm  $C_2H_2$ , and A: $O_2$  (0.8:0.2) to 1 atm. Culture serially diluted and cells in original culture counted in a hemocytometer. Ethylene assayed gas chromatographically on ester-amide column.

Cells/incubation	$\mu\text{moles } C_2H_4/\text{hr}$ incubation	$\mu\text{moles } C_2H_4/\text{hr}\cdot\text{cell}$
9,250,000	183	0.020
1,550,000	34	0.022
255,000	6.0	0.024
43,000	0.875	0.021
7,000	0.106	0.015

of incubation at  $30^\circ$  is 0.02  $\mu\text{mole}$  per hour per *Azotobacter* cell over a 1000-fold range of cell concentration. The extreme sensitivity of the  $C_2H_2-C_2H_4$  assay is indicated; theoretically, as few as 2 to 3 cells produce sufficient  $C_2H_4$  for detection by the  $H_2$ -flame ionization system.

**$K_m$  of  $C_2H_2$ .** Acetylene saturation of  $N_2$ -grown clostridial cells occurs between 0.025 and 0.1 atmosphere (fig 11a), and even 0.5 atmosphere is not inhibitory (fig 11b). The saturation concentration is similar for  $N_2$ -grown *Azotobacter*, but these cells

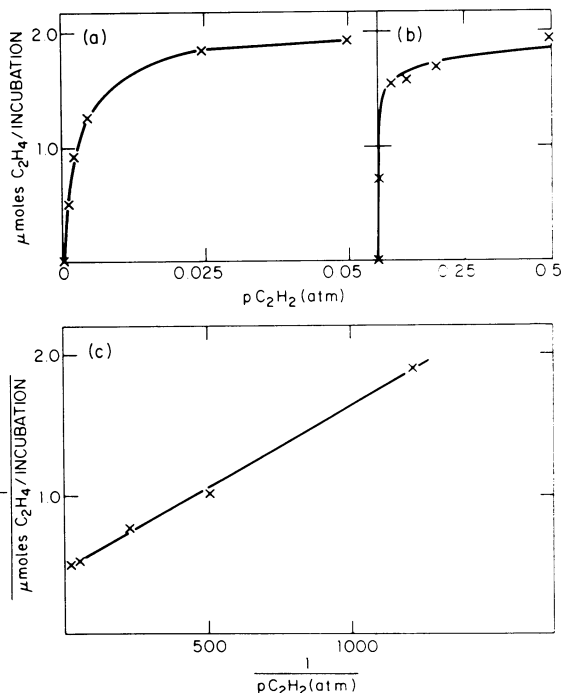


FIG. 11. a) and b). Ethylene formation from  $C_2H_2$  as a function of  $pC_2H_2$  by culture of  $N_2$ -grown *C. pasteurianum*, and c) plot of reciprocal velocity versus reciprocal  $pC_2H_2$  for determination of  $K_m$  of  $C_2H_2$ . Incubation system, table VII; liquid volume, 5 ml; gas phase, indicated  $pC_2H_2$  plus He to 1 atm; points represent averages of 3 incubations.

show an as yet unexplained increase in rate of C<sub>2</sub>H<sub>2</sub> reduction at 0.2 and 0.5 atmosphere of C<sub>2</sub>H<sub>2</sub>. A plot of reciprocal pC<sub>2</sub>H<sub>2</sub> vs. reciprocal rate of C<sub>2</sub>H<sub>2</sub> reduction by clostridial cells is shown in figure 11c. The range of Michaelis constants with *Clostridium* is 0.003 to 0.008 atmosphere C<sub>2</sub>H<sub>2</sub> with an average of 0.006, while that for *A. vinelandii* incubated at 0.1 atmosphere or less C<sub>2</sub>H<sub>2</sub> is 0.003 to 0.006 atmosphere with an average of 0.005.

**Activation Energy of C<sub>2</sub>H<sub>2</sub> Reduction.** The effect of incubation temperature on C<sub>2</sub>H<sub>2</sub> reduction by clostridial cells was determined over the range 10° to 35°. A close analogy with the *in vitro* results on N<sub>2</sub>ase is observed which suggests that the limiting factor in growth may be N<sub>2</sub>ase activity, and furthermore that this is related specifically to a property of the N<sub>2</sub>ase enzyme *per se*, rather than to reactions which furnish energy or reductant to the enzyme. An Arrhenius plot of the cellular activities is shown in figure 12. Results from 20° to 35° form a linear plot with a calculated activation energy of 13 to 15 kcal/mole; results from 10° to 20° are not co-linear with those from 20° to 35°, and a much higher activation energy, approximately 50 kcal/mole, is estimated for the lower temperatures.

**C<sub>2</sub>H<sub>2</sub> Reduction and N<sub>2</sub> Fixation.** Acetylene reduction by N<sub>2</sub>-grown *Azotobacter* cultures was compared with N<sub>2</sub> fixation and increase in OD by identical cultures incubated with air under the same conditions. In all cases the 3 parameters measured showed parallel increases with time of incubation. The late-log phase culture (fig 13) did not reduce

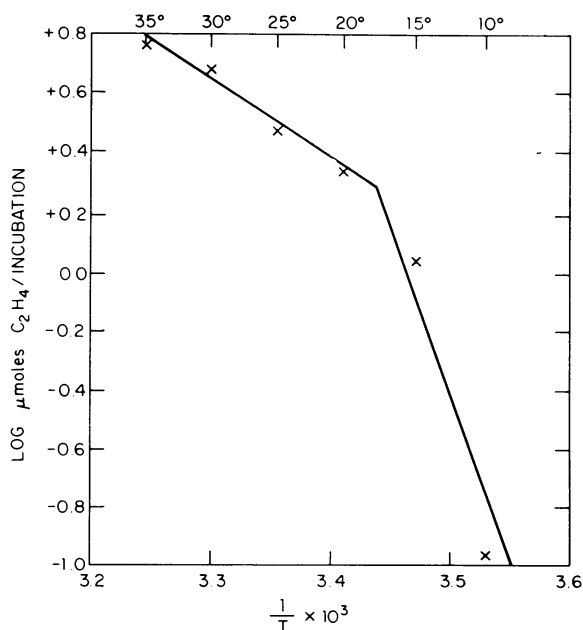


FIG. 12. Arrhenius plot of C<sub>2</sub>H<sub>2</sub>→C<sub>2</sub>H<sub>4</sub> by culture of N<sub>2</sub>-grown *C. pasteurianum* in the range of 10° to 35°. Incubation system, table VII; liquid volume, 5 ml; gas volume, 35 ml; temperature as indicated; time, 1 hr; gas phase, 0.1 atm C<sub>2</sub>H<sub>2</sub>, 0.9 atm He.

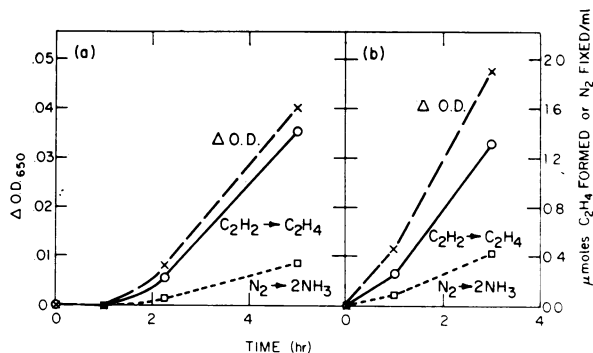


FIG. 13. Acetylene reduction, ΔOD<sub>650</sub> and N<sub>2</sub> fixation by culture of N<sub>2</sub>-grown *A. vinelandii* in a) late log phase, and b) early log phase of growth. Incubation system, table VII; liquid volume, 40 ml; gas volume, 290 ml; temperature, 30°; gas phase, 0.1 atm C<sub>2</sub>H<sub>2</sub>, 0.2 atm O<sub>2</sub>, 0.7 atm A. Initial OD<sub>650</sub> of late log phase culture after dilution with fresh media was 0.087, while that of early log phase culture was 0.112. Samples of gas phase and culture were analyzed for C<sub>2</sub>H<sub>2</sub> and C<sub>2</sub>H<sub>4</sub> by gas chromatography and for fixed nitrogen by Kjeldahl analysis at indicated times.

C<sub>2</sub>H<sub>2</sub>, fix N<sub>2</sub>, nor increase in optical density during the initial hour of incubation; in contrast the early log phase culture (fig 13) showed uniformly positive responses during the initial hour. The ratio of moles of N<sub>2</sub> fixed to moles of C<sub>2</sub>H<sub>4</sub> formed is 3 to 4.5.

**Distribution of C<sub>2</sub>H<sub>2</sub>-Reducing Activity.** The absence of significant C<sub>2</sub>H<sub>2</sub>-reducing activity in a variety of organisms grown under non-N<sub>2</sub>-fixing conditions further establishes the validity of the relationship between C<sub>2</sub>H<sub>2</sub>-reducing and N<sub>2</sub>-fixing ability. Organisms tested included *Clostridium butyricum* under anaerobic conditions on complete

Table IX. C<sub>2</sub>H<sub>2</sub> Reduction by Selected N<sub>2</sub>-Fixing and non-N<sub>2</sub>-Fixing Bacterial Cultures

One ml of indicated culture in log phase of growth added to 4 ml of its respective media. Total volume of incubation vessel, 40 ml; gas phase, 0.1 atm C<sub>2</sub>H<sub>2</sub>, 0.9 atm He for anaerobes, 0.1 atm C<sub>2</sub>H<sub>2</sub>, 0.2 atm O<sub>2</sub>, 0.7 atm A for aerobes; incubation time, 60 min; temperature, 30°. Ethylene assayed gas chromatographically on esteramide column. Similar results obtained with 0.01 atm C<sub>2</sub>H<sub>2</sub>.

Organisms	OD <sub>650</sub> mμ of culture	μmoles C <sub>2</sub> H <sub>4</sub> /hr·incubation	N <sub>2</sub> -fixing ability
<i>Azotobacter vinelandii</i>	0.30	1.58	+
<i>Clostridium pasteurianum</i>	0.19	1.62	+
<i>Rhizobium japonicum</i>	0.3	0.0002	—
<i>R. melliloti</i>	0.3	<0.00004	—
<i>R. leguminosarum</i>	0.3	<0.00004	—
<i>R. sp. (ATCC 10317)</i>	0.3	<0.0002	—
<i>R. trifolii</i>	0.3	<0.00004	—

medium and *Lactobacillus leichmanii*, *Bacillus subtilis*, *Bacillus cereus* var. *mycoides*, *Serratia marcescens*, *Escherichia coli*, *Streptococcus lactis*, *Saccharomyces cerevisiae*, *Pseudomonas aeruginosa*, *Pseudomonas fluorescens*, *Aerobacter aerogenes*, *Staphylococcus epidermidis*, *Sarcina lutea*, *Spirillum itersonii*, *Proteus vulgaris*, *Alcaligenes faecalis*, and *Rhodospirillum rubrum* under aerobic conditions on complete medium. All had <0.1% of the  $C_2H_2$ -reducing activity of  $N_2$ -grown *A. vinelandii* or *C. pasteurianum*. The absence of  $C_2H_2$ -reducing activity (<0.02% of the *Azotobacter* or clostridial activity) among the various *Rhizobia* grown on the indicated medium (see under Methods) is specifically emphasized, since these data are the most sensitive indication of the inability of cultures of these organisms to fix  $N_2$  (table IX).

*C<sub>2</sub>H<sub>2</sub> Reduction by Selected Biosphere Samples Assayed in situ.* An effective procedure for the *in situ* assay of  $N_2$ -fixing activity of the biosphere via the  $C_2H_2$ - $C_2H_4$  assay is described in this section. The process is used: 1) to assess requirements, reproducibility, sample variability, and environmental factors that influence  $C_2H_2$ -reducing activity of field-grown soybeans, 2) to quantitate  $C_2H_2$ -reducing activity of field-grown soybeans through one growing season, and 3) to determine  $C_2H_2$ -reducing activities of free-living organisms in the soil and in the hydrosphere. These results represent the first report of the systematic use of  $C_2H_2$  reduction to estimate  $N_2$  fixation during an entire growing season. Since results were obtained during a single season, it is obvious that assays in future seasons will be required to firmly define patterns of  $N_2$  fixation and to establish the effects of variable environmental conditions. Field-grown soybeans were selected for this test because of their agricultural significance. Other biosphere samples were selected to test the universality of the  $C_2H_2$ - $C_2H_4$  assay for determination of  $N_2$  fixation.

*Process of in situ C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> Assay.* The 10 steps involved in our *in situ*  $C_2H_2$ - $C_2H_4$  assay process are outlined in figure 14. All the steps are designed to minimize sample alteration and to establish in the assay chamber a micro-ecosystem that is identical in temperature, moisture, porosity, etc. to the macro-ecosystem from which the sample is selected. Samples may consist of nodulated plant roots, plant root soil bores, soil bores, or hydrosphere.

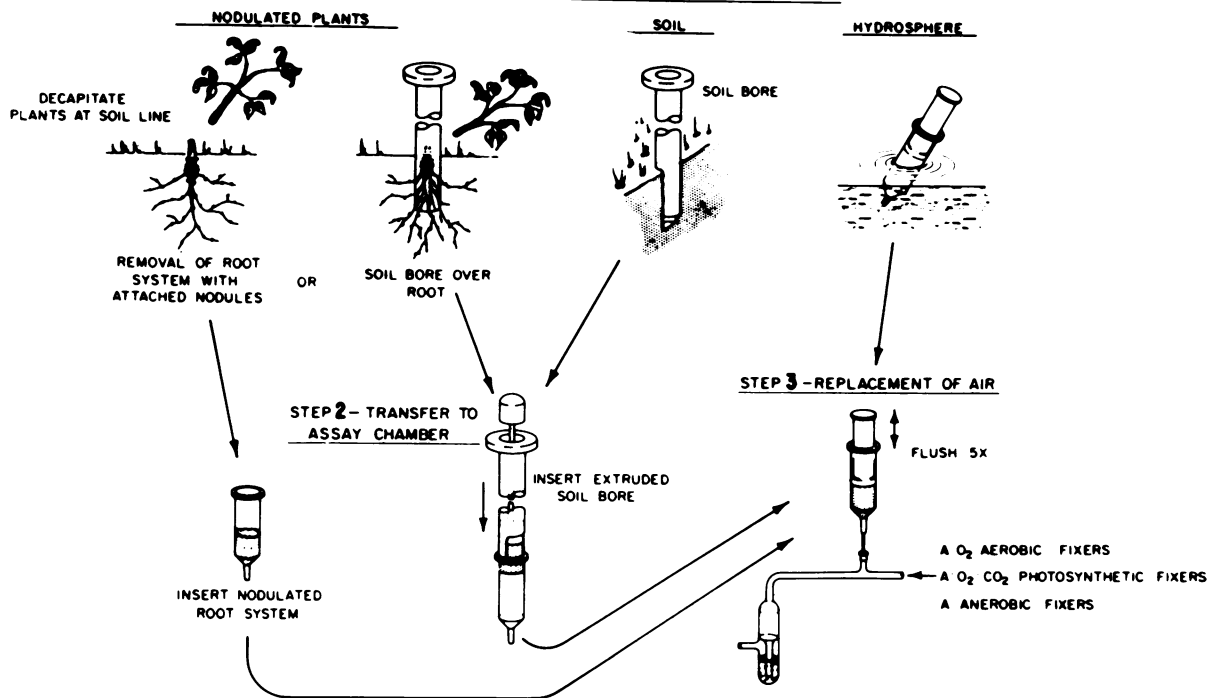
Nodulated legumes, after decapitation to eliminate possible background ethylene formation and to decrease the total sample size, were sampled by removal of the complete root system with attached nodules or by a soil bore directly over the tap root. Detached nodules have been used in some preliminary work (23, 33, 34, 36), but they are less active and their collection is time-consuming, particularly in the case of legumes with small nodules, e.g., *Medicago sativa*. We recommend the nodulated root system

over the root soil bore system because of: 1) the heterogeneous distribution of nodules per plant and of nodulated plants sown in rows, 2) the injury to nodules by the soil borer, and 3) the lower activity observed with the soil bore technique (fig 18c and 19c). Soil for free-living  $N_2$  fixation is sampled with a soil borer (1-inch  $\times$  15-inch graduated in 3-inch steps). Immediately after collection, samples (45 ml) are transferred to the assay chamber, a 50 ml syringe. The diameter of the assay chamber matches the diameter of the soil bore to promote effective gas exchange, addition and removal. Advantages of this assay chamber include: 1) sufficient size to accommodate a representative sample of a nodulated root or 45 ml soil bore plus gas phase, 2) effective and rapid replacement of air with 20 ml additions of desired aerobic [ $A:O_2(0.8:0.2)$ ], photosynthetic [ $A:O_2:CO_2(0.8:0.2:0.001)$ ] or anaerobic (A or He) gas phase (step 3), 3) uniform and rapid mixing of 20 ml of  $C_2H_2$  gas mixture (0.2 atm  $C_2H_2$  plus aerobic or anaerobic gas to 1 atm) with sample (steps 5 and 6), and 4) removal and mixing of gas phase at the end of the incubation (step 8), which precludes the requirement for sample inactivation. The determination of  $C_2H_2$  as well as  $C_2H_4$  can provide a useful internal standard for detection and correction of gas leaks, elimination of faulty assays and determination of gas volume of sample in the case of soil bores. Gas volume of soil bores was estimated as 50% of the volume of the soil bore in the results reported here. It is emphasized that  $C_2H_2$  is a very explosive gas and strict safety measures must be taken to eliminate the possibility of ignition (no smoking) during all steps involving  $C_2H_2$ . Observed  $C_2H_2$ -reducing activity is converted to a calculated  $N_2$ -fixing activity on the basis of the ratio of their electron requirements, i.e., 2 electrons for  $C_2H_2$  reduction and 6 electrons for  $N_2$  reduction. A ratio in the area of 3 to 4.5 is compatible with results of  $C_2H_2$  reduction and  $N_2$  fixation reported in the previous sections. The process described appears to be effective, simple and rapid. More than 450 samples per day have been routinely collected and assayed in the field (steps 1-8) and over 80 samples per hour have been incubated (steps 2-8).

*C<sub>2</sub>H<sub>2</sub> Reduction by Field-Grown Soybeans—Requirements, Distribution, and Product.* Root soil bores or nodulated roots of field-grown soybeans reduced  $C_2H_2$  to  $C_2H_4$  (table X). Neither  $C_2H_6$  nor  $CH_4$  was detected and  $C_2H_4$  was not reduced to  $C_2H_6$  or  $CH_4$ . Negligible  $C_2H_4$  (<0.001% of that with  $C_2H_2$ ) was formed in the absence of  $C_2H_2$ . Root soil bores containing no nodules possessed only a very low level of  $C_2H_2$ -reducing activity (0.01-0.02% of bores containing nodules), equivalent to that of soil bores made 19 inches from the soybean rows or of soil collected around the nodulated roots. Samples containing nodules but not possessing  $C_2H_2$ -reducing activity were always found to contain only white nodules. Aerobic conditions are required

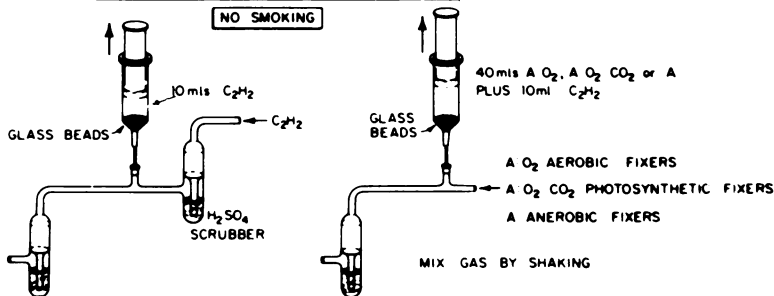
**THE C<sub>2</sub>H<sub>2</sub> → C<sub>2</sub>H<sub>4</sub> ASSAY FOR N<sub>2</sub>-FIXING ACTIVITY**

**STEP 1 - BIOSPHERE SAMPLE PREPARATION**



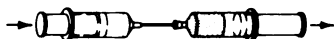
**STEP 5 - DILUTION AND MIXING OF C<sub>2</sub>H<sub>2</sub>**

**STEP 4 - SEALING ASSAY CHAMBER**



**STEP 6 - ADDITION OF C<sub>2</sub>H<sub>2</sub> MIXTURE**

**NO SMOKING**



ADD 20 ml C<sub>2</sub>H<sub>2</sub> MIXTURE TO ASSAY CHAMBER AND FLUSH  
ADD ANOTHER 20 ml C<sub>2</sub>H<sub>2</sub> MIXTURE AND INCUBATE

**STEP 7 - INCUBATION**

**NO SMOKING**



INCUBATE FOR ONE HOUR UNDER *IN SITU* CONDITIONS

**STEP 8 - REMOVAL AND MIXING OF GAS PHASE**

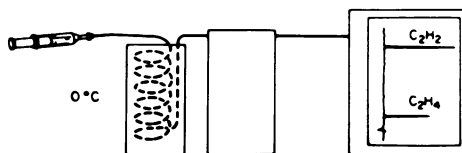
**NO SMOKING**



GAS IS FORCED FROM ASSAY CHAMBER INTO GAS RECEIVER

**STEP 9 - C<sub>2</sub>H<sub>2</sub> - C<sub>2</sub>H<sub>4</sub> ASSAY BY FLAME IONIZATION AFTER GAS CHROMATOGRAPHY**

**NO SMOKING**



200 μl FROM THE GAS RECEIVER

**STEP 10 - CONVERSION OF C<sub>2</sub>H<sub>2</sub> → C<sub>2</sub>H<sub>4</sub> ACTIVITY TO N<sub>2</sub> → 2NH<sub>3</sub> ACTIVITY**

FIG. 14. Steps in C<sub>2</sub>H<sub>2</sub> → C<sub>2</sub>H<sub>4</sub> assay for N<sub>2</sub>-fixing activity of the biosphere, including samples of nodulated plants, soil, or hydrosphere.

Table X. *Distribution of C<sub>2</sub>H<sub>2</sub>-Reducing Activity in Soybean Field*

Samples collected between 8 to 9 AM and immediately assayed as outlined in figure 14. Gas phase, 0.1 to 0.2 atm C<sub>2</sub>H<sub>2</sub> plus indicated gases A:O<sub>2</sub> (0.8:O<sub>2</sub>), or He to 1 a tm; root soil bore volume, 45 ml; total gas phase volume, 40 ml; incubation time, 1 hr.

Sample	Gas phase	$\mu\text{moles C}_2\text{H}_4/\text{day}\cdot\text{sample}$	$m\mu\text{moles C}_2\text{H}_4/\text{day}\cdot\text{mg fr wt nodule}$	mg fr wt nodule
Root soil bore	A:O <sub>2</sub>	0.001	0.001	1030
Root soil bore	He:C <sub>2</sub> H <sub>2</sub>	1.87	3.74	503
Root soil bore	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	0.026	...	1030
Root soil bore	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	171	168	0
Nodulated root	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	195	172	1114
Soil around nodulated root	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	0.035	...	0
Soil bore between rows 0-4" deep	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	0.033	...	0
4-8" deep	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	0.030	...	0
8-12" deep	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	0.028	...	0

for C<sub>2</sub>H<sub>2</sub> reduction (table X, ref. 23). Failure to replace air (N<sub>2</sub>) with A:O<sub>2</sub> results in a 10 to 20 % decrease in C<sub>2</sub>H<sub>2</sub> reduction (table XI).

*Time Course.* The rate of C<sub>2</sub>H<sub>2</sub> reduction by nodulated roots or root soil bores of soybeans is constant up to 60 minutes (fig 15) with the standard system described in figure 14. However, it is recommended that heavily nodulated roots be assayed for a shorter time (30 min) since the rate for such

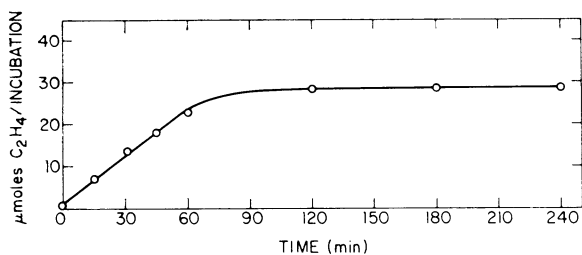


FIG. 15. Time course of C<sub>2</sub>H<sub>2</sub>→C<sub>2</sub>H<sub>4</sub> reduction by nodulated soybean roots. Incubation system, figure 14, except that analyses were made from 0 to 240 min. Glass beads were added to assay chambers to facilitate mixing of gas phase.

samples decreases shortly after 60 minutes. This decrease is presumably due to O<sub>2</sub> depletion, since samples which were reflushed with A:O<sub>2</sub> and re-gassed with the C<sub>2</sub>H<sub>2</sub> mixture showed activities during a 1 to 2 hour incubation that were comparable to those observed during a 0 to 1 hour incubation (tables XI, XIII).

*Interval Between Sampling and Assay.* Nodulated roots and root soil bores of soybeans were assayed at 0, 2, 6.5, and 13 hours after sampling (fig 16). Values are expressed on the basis of nodular efficiency,  $m\mu\text{moles C}_2\text{H}_4$  per mg fr wt nodule per day, in order to compensate for the variable nodulation of samples. The results indicate the importance of a minimum interval (0-2 hr) between sampling and assay in order to obtain values which are representative of the *in situ* activity. Consequently, all the results reported in this paper unless otherwise indicated were obtained from assays initiated within 30 minutes after collection.

*Sample Variability and Reproducibility.* The sample variability with respect to C<sub>2</sub>H<sub>2</sub>-reducing activity of field-grown soybeans is shown in table

Table XI. *Effect of Air on C<sub>2</sub>H<sub>2</sub>-Reducing Activity of Soybeans*

Samples were collected between 8 to 9 AM and assayed from 0 to 1 and again from 1 to 2 hr as outlined in figure 14. Air (N<sub>2</sub>) replaced A:O<sub>2</sub> where indicated.

Sample	Gas phase	$\mu\text{moles C}_2\text{H}_4/\text{day}\cdot\text{sample}$	$m\mu\text{moles C}_2\text{H}_4/\text{day}\cdot\text{mg fr wt nodule}$	mg fr wt nodule
Nodulated root				
0-1 hr	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	659	197	3333
1-2 hr	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	750	225	3333
0-1 hr	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	410	91	4511
1-2 hr	Air:C <sub>2</sub> H <sub>2</sub>	328	73	4511
Root soil bore				
0-1 hr	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	185	79	2340
1-2 hr	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	190	81	2340
0-1 hr	A:O <sub>2</sub> :C <sub>2</sub> H <sub>2</sub>	289	142	2047
1-2 hr	Air:C <sub>2</sub> H <sub>2</sub>	266	130	2047

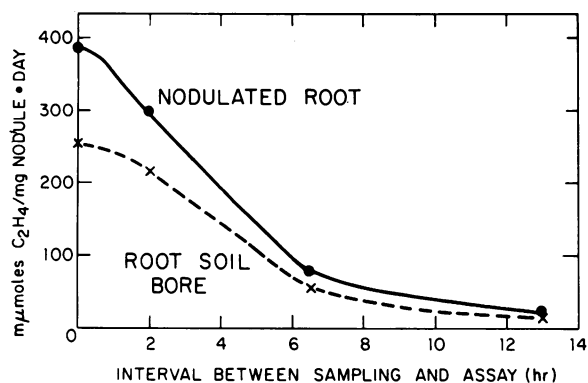


FIG. 16. Reduction of C<sub>2</sub>H<sub>2</sub>→C<sub>2</sub>H<sub>4</sub> by nodulated soybean root or soybean root soil bore assayed at various times after collection between 8 to 9 AM. Incubation system, figure 14. Each point represents the average of 5 individual samples.

XII. The standard deviation of C<sub>2</sub>H<sub>2</sub>-reducing activity of root soil bores or nodulated roots collected at 15, 25, 29, and 44 days post-flowering is 25 to 35 % while that of nodular efficiency is 17 to 22 % for root soil bores and 5 to 10 % for nodulated roots. Thus, expression of activity on the basis of nodular efficiency substantially decreases the variability due to differences in nodule weights.

The excellent reproducibility of C<sub>2</sub>H<sub>2</sub>-reducing activity of field-grown soybeans is shown in table XIII. Samples collected at various times after flowering show C<sub>2</sub>H<sub>2</sub>-reducing activities during a 1 to 2 hour assay that are 89 to 107 % of the respective activities determined during a 0 to 1 hour assay.

Km of C<sub>2</sub>H<sub>2</sub>. Nodulated roots of soybeans are saturated by 0.025 to 0.2 atmosphere C<sub>2</sub>H<sub>2</sub> (fig

Table XII. *Sample Variability of Field-Grown Soybeans for C<sub>2</sub>H<sub>2</sub> Reduction*

All samples collected from different areas of the same field between 8 to 9 AM on indicated day and assayed immediately as described in figure 14. Each value represents average of 5 samples.

Sampling time (days post-flowering)	μmoles C <sub>2</sub> H <sub>4</sub> / day•sample	mμmoles C <sub>2</sub> H <sub>4</sub> / mg fr wt nodule•day	mg fr wt nodule/ sample
Root soil bores			
15	33	94	355
	47	72	659
	68	84	812
	43	62	701
Avg	48	79	632
25	105	101	1035
	190	153	1238
	90	180	501
	116	149	776
Avg	125	146	888
Nodulated root			
29	513	176	2908
	514	144	3577
	785	166	4731
Avg	604	162	3739
44	751	226	3319
	412	242	1697
Avg	582	234	2508

Table XIII. *Reproducibility of C<sub>2</sub>H<sub>2</sub>-Reducing Activity of Field-Grown Soybeans*

Soybean root soil bores or soybean nodulated roots collected between 8 to 9 AM and assayed as in figure 14, from 0 to 1 hr and again from 1 to 2 hr after collection. Each value is average of 5 individual samples.

Sampling time (days post-flowering)	μmoles C <sub>2</sub> H <sub>4</sub> /sample•day		mμmoles C <sub>2</sub> H <sub>4</sub> /mg fr wt nodule•day		mg fr wt nodule/ sample
	0-1 hr	1-2 hr	0-1 hr	1-2 hr	
Root soil bore					
0	12	12	74	79	158
0	8	9	71	83	113
2	12	11	65	58	191
44	177	166	56	53	3140
Nodulated root					
44	752	742	226	223	3319

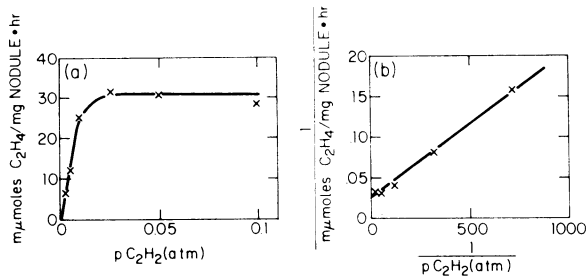


FIG. 17. a) Ethylene formation from  $C_2H_2$  as a function of  $pC_2H_2$  by nodulated roots of soybeans, and b) plot of reciprocal velocity *versus* reciprocal  $pC_2H_2$  for determination of  $K_m$  of  $C_2H_2$ . Incubation system, figure 14.

17a); soybean root soil bores showed identical activities when incubated with a  $C_2H_2$  gas mixture of 0.1 to 0.4 atmosphere  $C_2H_2$ . Activity was decreased by 0.5 atmosphere  $C_2H_2$ . This decrease might be due to direct inhibition by  $C_2H_2$  or to indirect inhibition by the concomitant depletion of  $O_2$ . An average  $K_m$  of 0.007 atmosphere  $C_2H_2$  is obtained from a plot of reciprocal  $pC_2H_2$  *vs.* reciprocal rate of  $C_2H_4$  formation. This value is somewhat lower than that reported for excised soybean nodules (23).

**Temperature.** The effect of temperature of incubation on  $C_2H_2$  reduction by nodulated roots or root soil bores of soybeans was less pronounced than observed with either  $N_2ase$  preparations or bacterial cultures. Activity was lower at  $10^\circ$  to  $15^\circ$  and possibly at  $35^\circ$  than at  $20^\circ$  to  $30^\circ$  (table XIV). In contrast, preliminary results suggest that the temperature of growth has a more marked effect on  $C_2H_2$ -reducing activity. Nodulated roots from plants maintained in growth cabinets for periods of 1 to 14 days at  $30^\circ$  had only 10 to 20% of the  $C_2H_2$ -reducing activity of those at  $20^\circ$ .

**$C_2H_2$ -Reducing Activity and Calculated  $N_2$  Fixation During One Growing Season.** Acetylene-reducing activity of field-grown soybeans was determined during a complete growth and maturation

Table XIV. *Effect of Temperature on  $C_2H_2$  Reduction by Soybeans*

Soybean root soil bores or nodulated roots collected between 8 to 9 AM, flushed with A: $O_2$  as described in figure 14, equilibrated for 10 min in a water bath at indicated temperature, then incubated for 1 hr after addition of  $C_2H_2$  mixture. Values are the averages of 3 nodulated roots or 5 root soil bores.

Incubation temp	Nodulate root	Root soil bore
deg	$m\mu\text{moles } C_2H_4/\text{mg fr wt nodule}\cdot\text{day}$	
10	29	...
15	86	...
20	214	156
25	176	206
30	250	188
35	151	151

cycle. Both the root soil bore (fig 18) and nodulated root (fig 19) techniques were used. Samples were collected on 41 different days for the root soil bore assays and on 27 different days for the nodulated root analyses. The  $C_2H_2$ -reducing activity was determined as described in figure 14 utilizing 1 hour incubations. Results are expressed on a 24 hour basis (fig 18a and 19a). The fresh weight of nodules per sample was tabulated (fig 18b and 19b), and the nodular efficiency of  $C_2H_2$  reduction, expressed as  $m\mu\text{moles } C_2H_4$  formed per mg fresh weight nodule per day, was calculated (fig 18c and 19c). The averages of all analyses on an individual day and during each week are shown.

Nitrogen fixation, as measured by the  $C_2H_2$ - $C_2H_4$  assay, was found to parallel the nitrogen demand of

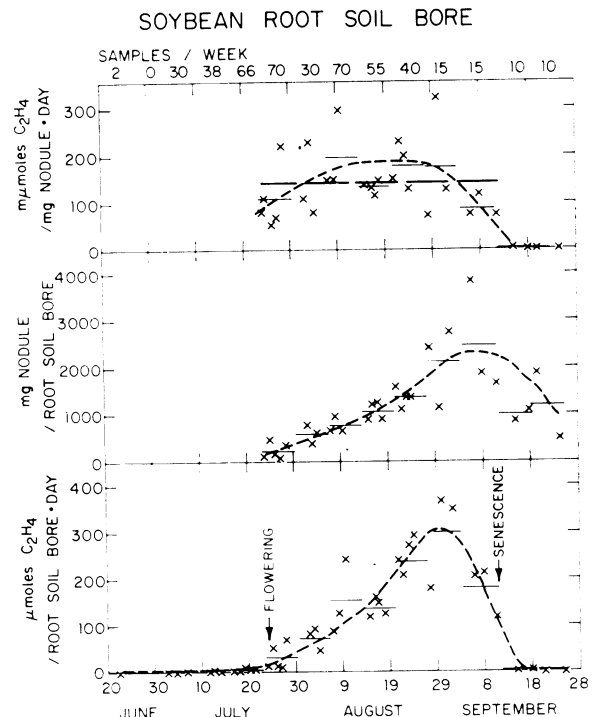


FIG. 18. Summary of a)  $C_2H_2 \rightarrow C_2H_4$  reducing activity, b) mg fr wt nodule, and c)  $m\mu\text{moles } C_2H_2 \rightarrow C_2H_4$  per mg fr wt nodule per day by soybean root soil bores at various stages from initiation of activity through flowering and maturation to loss of activity. Incubation system, figure 14; root soil bore volume, 45 ml; gas phase volume, 40 ml. Flowering indicated by initial bud opening and senescence indicated by initial yellowing of leaves. The average of all samples assayed each day (x), during each week — and the average of  $m\mu\text{moles } C_2H_2 \rightarrow C_2H_4$  per mg fr wt nodule per day for the period from flowering to senescence (— — —) are recorded. The number of soil bores of soybean roots assayed each week is recorded. Samples were collected between 8 to 9 AM except for July 27 to 28 and August 9 to 10 when samples were collected at various times as specified in figure 20. Samples were assayed immediately under *in situ* conditions of temperature and moisture. Assay time, 1 hr; results expressed on 24 hr basis.



NODULATED SOYBEAN ROOT

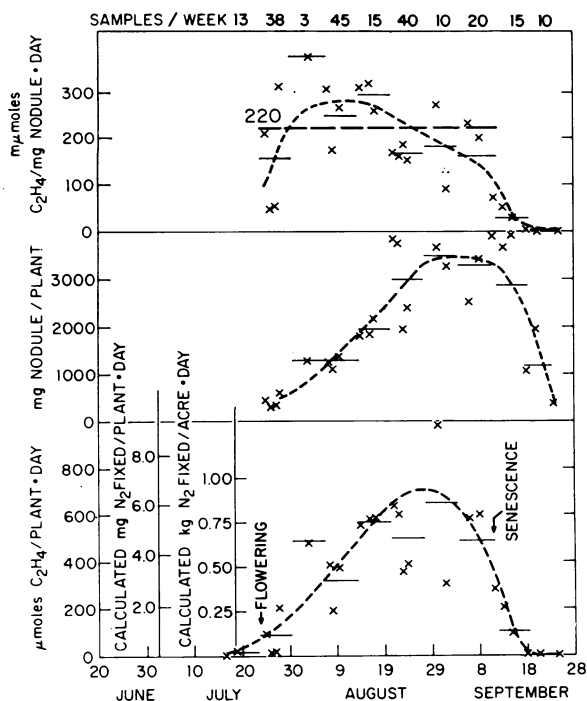


FIG. 19. Summary of a) C<sub>2</sub>H<sub>2</sub>→C<sub>2</sub>H<sub>4</sub> reducing activity, b) mg fr wt nodule and c) mμmoles C<sub>2</sub>H<sub>2</sub>→C<sub>2</sub>H<sub>4</sub> per mg fr wt nodule per day by nodulated roots of field-grown soybean plants at stages of development similar to those in figure 18. Incubation system, figure 14; gas phase volume, 40 ml. The average of all samples assayed each day (x), each week (—), and the average mμmoles C<sub>2</sub>H<sub>2</sub>→C<sub>2</sub>H<sub>4</sub> per mg fr wt nodule per day for the period from flowering to senescence (---) are shown. The number of nodulated soybean roots assayed each week is recorded. Sample collection and assay conditions as described in figure 14. Mg N<sub>2</sub> fixed per plant per day and kg N<sub>2</sub> fixed per acre per day are calculated on the theoretical basis of one-third N<sub>2</sub> reduced per C<sub>2</sub>H<sub>2</sub> reduced.

the plant. Low C<sub>2</sub>H<sub>2</sub>-reducing activity occurred until macroscopic flowering was observed, although activity could be detected as early as 32 days before this time; the utilization of residual nitrogen fertilizer during this period may have suppressed N<sub>2</sub>-fixing activity to some extent. Following flowering the C<sub>2</sub>H<sub>2</sub>-reducing activity increased continuously, reflecting the increasing nitrogen requirement for pod formation and filling. Average weekly activity increased from 30 to 299 μmoles C<sub>2</sub>H<sub>4</sub> formed per root soil bore per day or 84 to 650 μmoles C<sub>2</sub>H<sub>4</sub> formed per plant root per day. The average weekly nodule weight increased from 244 to 2453 mg per root soil bore or 457 to 3478 mg per nodulated plant root. After pod filling was complete the C<sub>2</sub>H<sub>2</sub>-reducing activity rapidly declined. The decline in nodule weight following senescence lagged behind that of C<sub>2</sub>H<sub>2</sub>-reducing activity.

The nodule efficiency was relatively constant during the period from flowering to senescence (fig 18c and 19c). The average weekly efficiencies varied from 88 to 196 mμmoles C<sub>2</sub>H<sub>4</sub> formed per mg fresh weight nodule per day for root soil bores and from 156 to 378 for nodulated roots. The average nodule efficiency from flowering to senescence was 142 mμmoles C<sub>2</sub>H<sub>4</sub> formed per mg fresh weight nodule per day for root soil bores and 220 for nodulated roots.

Summation of the weekly averages of C<sub>2</sub>H<sub>2</sub>-reducing activity indicates that 22.5 mmoles of C<sub>2</sub>H<sub>4</sub> could be formed per plant per season (fig 19a). The calculated N<sub>2</sub> fixation per acre per season is 30 to 33 kg of nitrogen based on: 1) the C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> assay, 2) 142,000 plants per acre, and 3) a theoretical conversion factor of one-third N<sub>2</sub> fixed for each C<sub>2</sub>H<sub>4</sub> formed.

$$\frac{22.5 \text{ to } 25}{3} \times \frac{28}{10^6} \times$$

$$142,000 = 30 \text{ to } 33 \text{ kg N}_2 \text{ fixed per acre per season}$$

This calculated value was determined during development of the C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> assay and with samples (except for 2 occasions) collected between 8 to 9 AM, a period which may represent less than maximal activity (fig 20). However, this value is in excellent agreement with the average value of 38 kg

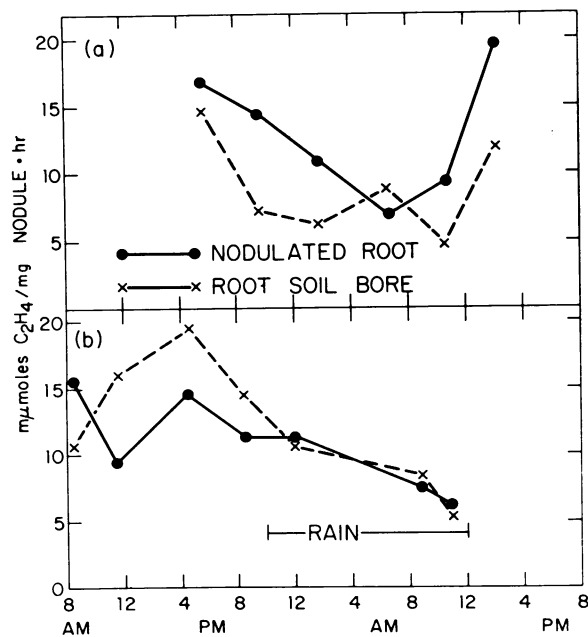


FIG. 20. Diurnal variation of C<sub>2</sub>H<sub>2</sub>→C<sub>2</sub>H<sub>4</sub> reduction by nodulated soybean roots and soybean root soil bores collected in the field at indicated times and assayed immediately. a) Samples 3 to 4 days post-flowering, and b) 16 to 17 days post-flowering. Incubation system, figure 14. Each point represents the average of 5 samples. Heavy rainfall occurred during the night and second day of experiment in b).

of nitrogen fixed per acre reported for Kjeldahl and  $^{15}\text{N}$  analyses of  $\text{N}_2$  fixation by soybeans (35). This correlation provides support for the quantitative reliability of the  $\text{C}_2\text{H}_2$ - $\text{C}_2\text{H}_4$  assay performed as outlined in figure 14.

**Diurnal Variation.** The diurnal variation of  $\text{C}_2\text{H}_2$ -reducing activity of field-grown soybeans was determined at 3 to 4 (fig 20a) and 16 to 17 (fig 20b) days post-flowering. Activities expressed as  $\mu\text{moles C}_2\text{H}_4$  formed per mg fr wt nodule per hour appear to be maximal for samples collected from noon to 8 PM and minimal for those collected from midnight to 8 AM. Thus, a close relationship between light and  $\text{N}_2$ -fixing activity is suggested. The effect of light on  $\text{C}_2\text{H}_2$ -reducing activity was further demonstrated with nodulated roots of soybean plants in growth cabinets (fig 21). Control plants maintained on a 16 hour light and 8 hour dark cycle did not show a marked diurnal variation, while experimental plants showed a rapid decline to 30% of control activity after 17 hours of total darkness, but still had 15% of control activity after 64 hours of darkness. The initial decline may reflect the depletion of photosynthate, while the residual activity may represent utilization of storage products.

Heavy rainfall eliminated the normal diurnal variation (fig 20b). This effect might be due to increased soil moisture and/or decreased light intensity. Saturation of soybean root soil bores with water decreased their  $\text{C}_2\text{H}_2$ -reducing activities from 170 to 50  $\mu\text{moles C}_2\text{H}_4$  per mg fresh weight nodule per day.

**Leaf or Pod Removal.** Removal of leaves decreased  $\text{C}_2\text{H}_2$ -reducing activity to 12% of control

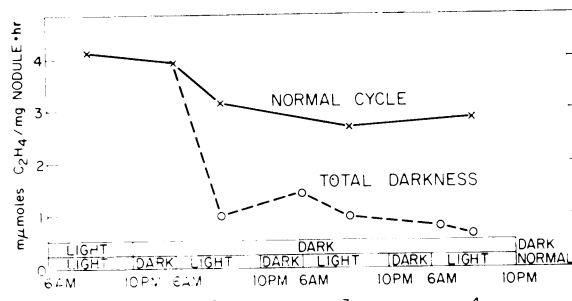


FIG. 21. Effect of light and darkness on  $\text{C}_2\text{H}_2 \rightarrow \text{C}_2\text{H}_4$  reduction by nodulated soybean roots from plants maintained in plant growth chamber. Incubation system, figure 14. Each point represents the average of 5 individual samples. Control cycle, 6 AM to 10 PM, 24°, with maximum light (see under Methods), and 10 PM to 6 AM, 18°, total darkness. Experimental sample was placed in total darkness with normal temperature cycle at 10 PM of first day of experiment.

after 1 day, and this activity was still only 14% of control at 10 days after leaf removal.

Removal of pods at 19 days post-flowering did not alter  $\text{C}_2\text{H}_2$ -reducing activity per plant (fig 22a) or mg fresh weight nodule per plant (fig 22b) during the following 10 days. However, during the same period control plants increased both  $\text{C}_2\text{H}_2$ -reducing activity per plant and mg fresh weight nodules per plant. Thus, the magnitude of  $\text{N}_2$  fixation reflects the demands of the plant, specifically the pod in this case, for nitrogen.

**Varieties.** Acetylene reduction provides a technique for the determination of differences in the  $\text{N}_2$ -fixing activities of different varieties of legumes. A single exploratory experiment was conducted with

Table XV.  $\text{C}_2\text{H}_2$  Reduction by Different Varieties of Field-Grown Soybeans

Nodulated roots of soybean varieties collected between 2 to 3 PM on the same day (114 days after planting) were assayed immediately for  $\text{C}_2\text{H}_2$  reduction as indicated in figure 14. Soil temperature, 19 to 20°. C1311-Wabash X C1069-Clark X C1069; and UD 61-1806-EC33243 X D49-249L. Each value represents 5 individual roots.

Variety	Stage of development	$\mu\text{moles C}_2\text{H}_4/\text{plant}\cdot\text{day}$	$\mu\text{moles C}_2\text{H}_4/\text{mg fr wt nodule day}$
Early-maturing:			
Verde	Yellow, loss of leaves	1.5	1.1
Adelphia	"	1.1	1.5
Clark	"	1.0	1.8
Intermediate-maturing:			
C1311	" + nodule decay	0.8	4.8
C1278	Initial yellowing	7.0	7.9
Kent	"	22.7	17
Delmar	"	30.2	39
Late-maturing:			
Dare	Green, pods still filling	35.5	57
York	"	59.2	98
UD61-1806	"	136	104
Hill	"	162	119

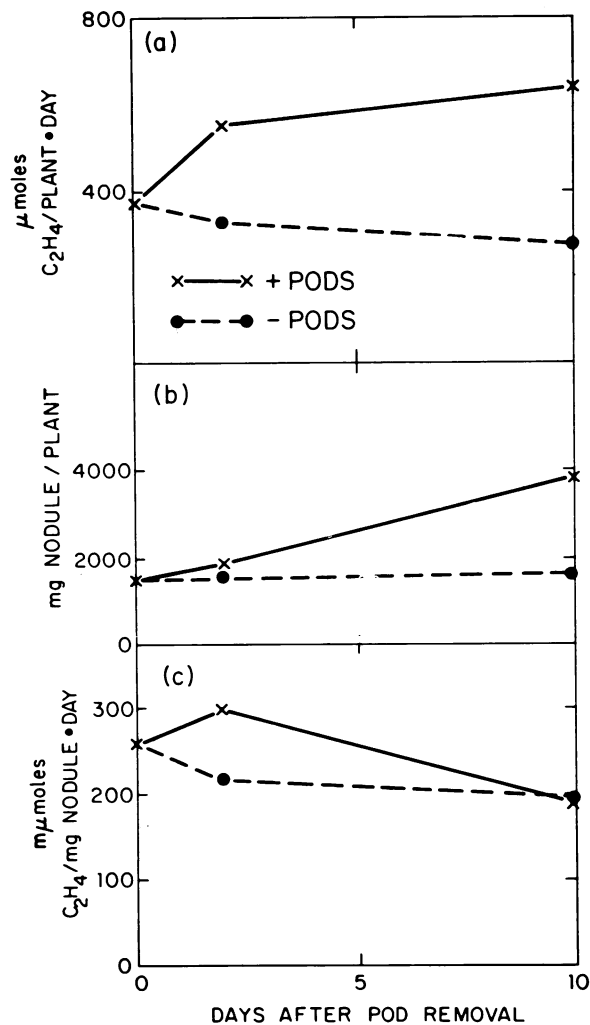


FIG. 22. a to c). Effect of pod removal on C<sub>2</sub>H<sub>2</sub> → C<sub>2</sub>H<sub>4</sub> reduction by nodulated soybean root or soybean soil bore. Incubation system, figure 14. Pods were removed 19 days post-flowering.

varieties of soybeans representing different maturation dates. Marked differences found in C<sub>2</sub>H<sub>2</sub>-reducing activity (table XV) correlated with differences in the stage of maturity at analysis. Varietal differences will be further investigated.

**C<sub>2</sub>H<sub>2</sub> Reduction by Selected Legumes.** Nodulated roots of *Phaseolus vulgaris*, *Medicago sativa*, *Arachis hypogea*, and *Pisum sativum* as well as *Glycine max* reduce C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub> (table XVI). The nodular efficiencies, mμmoles C<sub>2</sub>H<sub>4</sub> formed per mg fresh weight nodule per day, of these legumes were in the range of 106 to 402, similar to those reported by Koch and Evans (23) for excised soybean nodules, but substantially higher than those reported by Sloger and Silver (34) and subsequently by Stewart, Fitzgerald, and Burris (36) for a variety of excised legume and non-legume nodules.

Table XVI. C<sub>2</sub>H<sub>2</sub> Reduction by Selected Legumes

Nodulated roots of the indicated legumes collected between 8 to 10 AM and immediately assayed as described in figure 14.

Legume	mμmoles C <sub>2</sub> H <sub>4</sub> /mg fr wt nodule·day	Calculated kg N <sub>2</sub> fixed/acre·day
<i>Glycine max</i> <sup>1</sup>	220	0.60
<i>Phaseolus vulgaris</i> <sup>2</sup>	106	...
<i>Medicago sativa</i> <sup>2</sup>	161	0.47
<i>Arachis hypogea</i> <sup>2</sup>	402	...
<i>Pisum sativum</i> <sup>2</sup>	304	...

<sup>1</sup> Average value from figure 19 for flowering to senescence period.

<sup>2</sup> Samples assayed at flowering.

**C<sub>2</sub>H<sub>2</sub> Reduction by Free-Living Bacteria in Soil.** Nitrogen-fixing activities calculated from the C<sub>2</sub>H<sub>2</sub>-reducing activities of selected soil samples are tabulated in table XVII. The activities varied over 500-fold between locations and suggest that N<sub>2</sub>-fixation by free-living soil bacteria varies from negligible to highly significant. Activities under both aerobic and anaerobic conditions were found; the anaerobic activity was higher in the lower soil levels.

Variations in C<sub>2</sub>H<sub>2</sub>-reducing activity of the Jordan Fertility Plot samples correlated with nutrient applications. The 4 plots receiving recommended additions of N, P, and K showed similar activities, and the average calculated value of N<sub>2</sub> fixation is 0.51 kg of N<sub>2</sub> fixed per acre per day. Plots receiving less or no nitrogen showed, in general, greater activities. The calculated N<sub>2</sub> fixation of the plot receiving only K is 1.52 kg N<sub>2</sub> fixed per acre per day. This is the highest calculated value of N<sub>2</sub> fixation that we have obtained for soil samples analyzed with the C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> assay. Samples from all the limed plots had double the activity of those from all the unlimed plots. Initial experiments with soil to which bacterial cultures were added (*Azotobacter* and *Clostridium*) showed negligible increases in C<sub>2</sub>H<sub>2</sub>-reducing activity.

**Miscellaneous Examples of C<sub>2</sub>H<sub>2</sub> Reduction.** Nitrogen-fixing activity of a variety of miscellaneous samples of the biosphere, including pond water and rumen contents were assayed by C<sub>2</sub>H<sub>2</sub> reduction. Samples of pond water reduced C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub> when they contained blue-green algae. Negligible C<sub>2</sub>H<sub>4</sub> was formed in the absence of C<sub>2</sub>H<sub>2</sub>. More extensive studies of C<sub>2</sub>H<sub>2</sub> reduction by blue-green algae have been communicated recently (36).

Rumen contents from a fistulated steer reduced C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub>. Ethylene formation was 10-fold greater under anaerobic than aerobic conditions. Methane formation was markedly decreased in the presence of acetylene. The N<sub>2</sub>-fixing activity of a rumen calculated on the basis of the anaerobic results was 10 mg N<sub>2</sub> fixed per rumen per day.

Table XVII. *C<sub>2</sub>H<sub>2</sub>-Reducing Activity and Calculated N<sub>2</sub> Fixation by Soil Samples*

Soil bores were collected from 12 to 4 PM and immediately assayed as indicated in figure 14. Each value represents the average of 6 samples. Soil bores, 2 × 45 ml for 0 to 6" and 1 × 45 ml for 0 to 3" and 6 to 9".

Sample level	Special treatment	mμmoles C <sub>2</sub> H <sub>4</sub> /sample·day		Calculated kg N <sub>2</sub> fixed /3" or 6" acre·day	
		Aerobic	Anaerobic	Aerobic	Anaerobic
Chester Co., Pa.					
0-3"	...	31	67	0.0020	0.0043
3-6"	...	27	205	0.0018	0.0130
6-9"	...	27	75	0.0018	0.0048
Georgetown, Del.					
0-3"	...	310	155	0.020	0.010
6-9"	...	235	940	0.015	0.106
Jordan fertility plots					
0-6"	+K <sup>1</sup>	11,900	...	1.52	...
0-6"	+N <sup>1</sup>	11,500	...	1.47	...
0-6"	+P <sup>1</sup>	8,700	...	1.12	...
0-6"	No fertilizer	8,600	...	1.10	...
0-6"	+P,K	8,740	...	1.02	...
0-6"	+N,P	7,400	...	0.95	...
0-6"	+N,K	5,850	...	0.75	...
0-6"	+N,P,K	5,700	...	0.73	...
0-6"	Recommended treatment <sup>2</sup>	4,600	...	0.59	...
0-6"	"	4,200	...	0.54	...
0-6"	"	4,100	...	0.53	...
0-6"	"	2,900	...	0.37	...
0-6"	+3(N), as NaNO <sub>3</sub> , P,K	2,800	...	0.36	...
0-6"	Excess burned lime	2,260	...	0.29	...
0-6"	+N,P,K	1,400	...	0.27	...
0-6"	Ground bone	1,250	...	0.18	...
0-6"	+3(N) as (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , P,K	1,100	...	0.16	...
0-6"	+Lime	5,850	...	0.75	...
0-6"	-Lime	2,650	...	0.34	...

<sup>1</sup> +K = 100 lb K<sub>2</sub>O per acre; +N = 24 lb N per acre; +P = 48 lb P<sub>2</sub>O<sub>5</sub> per acre.

<sup>2</sup> Recommended treatment = 115 lb N, 130 lb P<sub>2</sub>O<sub>5</sub>, 130 lb K<sub>2</sub>O per acre.

## Discussion

The results reported here support the validity of the C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> assay as a sensitive and universal analysis for N<sub>2</sub>-fixing activity. The advantages of this assay indicate significant broad applications for measurement of N<sub>2</sub> fixation in both laboratory and field investigations. The essential relationship between C<sub>2</sub>H<sub>2</sub>-reducing activity and N<sub>2</sub>-fixing activity is supported and extended to a most convincing degree by studies within and among cell-free extracts, bacterial cultures, and symbionts.

This relationship is firmly established by detailed experiments with *Azotobacter* preparations which lead to the conclusion that C<sub>2</sub>H<sub>2</sub> and N<sub>2</sub> evoke identical responses from N<sub>2</sub>ase. Convincingly parallel responses are observed with respect to: requirement for ATP and reductant, linear time course, optimum pH, sigmoidal relationship between rate and enzyme concentration, inhibition of H<sub>2</sub> evolution in an amount equivalent in electrons to those used for C<sub>2</sub>H<sub>2</sub> or N<sub>2</sub> reduction, competitive inhibition by CO, relative insensitivity to NH<sub>4</sub><sup>+</sup>, activation energy of 13 to 15 kcal/mole above 20° and 35 to 50 kcal/mole below 20°, activity in extracts from

N<sub>2</sub>- but not urea-grown cells, distribution of activity during fractionation, requirement for both the Mo-Fe and Fe protein fractions of N<sub>2</sub>ase, and relative inactivity of the clostridial-*Azotobacter* interspecies recombination of the Mo-Fe and Fe protein fractions. These results provide the most complete correlation between N<sub>2</sub> fixation and C<sub>2</sub>H<sub>2</sub> reduction to date. Some of these similarities have been reported for cell-free extracts of bacteroids (24, 25) and *C. pasteurianum* (13, 32). The inhibition of ATP-dependent H<sub>2</sub> evolution by *Azotobacter* N<sub>2</sub>ase in an amount equivalent to the formation of ethylene establishes the electron-activating reaction of N<sub>2</sub>ase as the source of electrons for C<sub>2</sub>H<sub>2</sub> reduction, just as for N<sub>2</sub>, N<sub>2</sub>O and N<sub>3</sub><sup>-</sup> reductions (15, 20). The similar competitive inhibitions of N<sub>2</sub> fixation and C<sub>2</sub>H<sub>2</sub> reduction by CO provide strong indirect support for the role of the substrate-complexing site of *Azotobacter* N<sub>2</sub>ase for both C<sub>2</sub>H<sub>2</sub> and N<sub>2</sub> reduction.

Whole cell experiments are completely consistent with the results obtained *in vitro* and demonstrate the parallel C<sub>2</sub>H<sub>2</sub>-N<sub>2</sub> relationship in *in vivo* analyses. Thus, cultures of *Azotobacter* or *Clostridium* reduce C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub> with characteristics similar to fixation of N<sub>2</sub>, which include: anaerobic requirement

for *Clostridium* and aerobic requirement for *Azotobacter*, activity in N<sub>2</sub> but not NH<sub>3</sub>- or urea-grown cells, time course, and ratio of C<sub>2</sub>H<sub>2</sub> reduced to N<sub>2</sub> fixed of 3 to 4.5. In addition, bacterial species without N<sub>2</sub>-fixing activity do not possess significant C<sub>2</sub>H<sub>2</sub>-reducing activity.

Experiments with symbionts establish that the C<sub>2</sub>H<sub>2</sub>-N<sub>2</sub> correlation is consistently applicable to even these most complex natural N<sub>2</sub>-fixing systems. Thus, various legumes also reduce C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub> with characteristics similar to fixation of N<sub>2</sub>. These include: aerobic requirement, activity only in nodules possessing leghemoglobin, absence of activity in either the root or infecting bacteria, and similar rate of C<sub>2</sub>H<sub>4</sub> formation or N<sub>2</sub> fixation per season based on electron requirement.

The characteristics of N<sub>2</sub>ase activity, as exemplified by C<sub>2</sub>H<sub>2</sub> reduction, are consistent through the entire range of organization studied; results obtained with the most defined *in vitro* system apply with equal validity to even the most complex symbiotic systems. Thus, all systems reduce C<sub>2</sub>H<sub>2</sub> to a single significant product, C<sub>2</sub>H<sub>4</sub>; none reduce substrate C<sub>2</sub>H<sub>4</sub>; all are saturated by 0.02 to 0.2 atmosphere of C<sub>2</sub>H<sub>2</sub>; the *K<sub>m</sub>* values are 0.002 to 0.009 atmosphere, and the calculated activation energies are similar for both N<sub>2</sub>ase preparations and bacterial cells with a break in the Arrhenius plots near 20°. A common enzyme, N<sub>2</sub>ase, appears to be responsible for C<sub>2</sub>H<sub>2</sub> reduction by these diverse systems.

The advantages and disadvantages of the C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> assay for N<sub>2</sub> fixation are summarized in table XVIII. The advantages emphasize the superior attributes of this method relative to other assays of N<sub>2</sub> fixation.

Sensitivity of C<sub>2</sub>H<sub>4</sub> detection by flame ionization as initially indicated by Hardy and Knight (19) is the critical advantage of the C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> assay. The sensitivity of this method is related to other methods of N<sub>2</sub> fixation by the following comparison: the C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> method is to the <sup>15</sup>N method as the <sup>15</sup>N method was to the Kjeldahl method. This sensitivity of the C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> assay makes it possible to detect low levels of N<sub>2</sub>-fixing activity in biosphere samples, bacterial cultures, or N<sub>2</sub>ase preparations, and to investigate variations in N<sub>2</sub>-fixing activity with short-term incubations. It is anticipated that the list of N<sub>2</sub>-fixing organisms will undergo additions and possibly some deletions because of the C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> assay. The only other method with equivalent or greater sensitivity is <sup>15</sup>N-incorporation; however, the complex production facilities and short half-life restrict the application of <sup>15</sup>N to limited laboratory investigations.

Disadvantages of the C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> assay include the indirect nature of the reaction. Although no defined sample has been found with significant C<sub>2</sub>H<sub>2</sub>-reducing activity and without N<sub>2</sub>-fixing activity, the possibility exists for a non-N<sub>2</sub>ase-dependent catalysis by biosphere samples of the reduction

of C<sub>2</sub>H<sub>2</sub> to C<sub>2</sub>H<sub>4</sub>. The explosive nature of C<sub>2</sub>H<sub>2</sub> is emphasized. Laboratories accustomed to utilizing the relatively inert N<sub>2</sub> for <sup>15</sup>N or Kjeldahl analysis are cautioned to employ safe practices with C<sub>2</sub>H<sub>2</sub>.

As recently as 3 years ago Allison (2) found it necessary to state "During the past 50 years much effort has been devoted to the economic evaluation

Table XVIII. The C<sub>2</sub>H<sub>2</sub>-C<sub>2</sub>H<sub>4</sub> Assay for N<sub>2</sub> Fixation

<i>Advantages</i>
A) Analytical
Sensitivity. <1 μmole C <sub>2</sub> H <sub>4</sub> per sample is detectable; 10 <sup>3</sup> times as sensitive as <sup>15</sup> N and 10 <sup>6</sup> times as sensitive as Kjeldahl analyses.
Facility. No chemical treatment or manual manipulation of product required, <i>c.f.</i> <sup>15</sup> N and Kjeldahl analyses which require several time-consuming chemical conversions. Both product and unused reactant are simultaneously recovered in a single, easily secured sample of the gas phase which is directly analyzed.
Specificity. C <sub>2</sub> H <sub>4</sub> is separated from CH <sub>4</sub> , C <sub>2</sub> H <sub>2</sub> , C <sub>2</sub> H <sub>6</sub> .
Internal Standard. C <sub>2</sub> H <sub>2</sub> is a natural internal standard and its measurement requires no additional steps.
Storage Stability of Product. C <sub>2</sub> H <sub>4</sub> stable indefinitely in simple containers at ambient temperature.
Simplicity. <i>In situ</i> assay process and gas chromatography are simple techniques and can be performed by technicians <i>c.f.</i> <sup>15</sup> N mass spectrometric analysis. Few and simple calculations convert raw data to activity values.
Economy. Substrate, assay hardware, and gas chromatographic equipment are inexpensive <i>c.f.</i> mass spectrometric equipment and <sup>15</sup> N <sub>2</sub> cost.
Rapidity. Ten complete assays (fig 1) per man hour in the field. Twenty complete gas chromatographic analyses per man hour in the laboratory.
Mobility. All required equipment sufficiently rugged and portable to permit use of a mobile analytical unit.
Sample Conservation. The test material is not sacrificed by atmospheric sampling and sequential samples can be obtained from a single incubation.
B) Biological
Characteristics of C <sub>2</sub> H <sub>2</sub> →C <sub>2</sub> H <sub>4</sub> parallel those of N <sub>2</sub> →2NH <sub>3</sub> .
Universality. N <sub>2</sub> ase preparations, and N <sub>2</sub> -fixing bacteria, blue-green algae and symbionts reduce C <sub>2</sub> H <sub>2</sub> →C <sub>2</sub> H <sub>4</sub> , while non-N <sub>2</sub> -fixing preparations or organisms do not.
Quantitative Relationship of C <sub>2</sub> H <sub>2</sub> reduced to N <sub>2</sub> fixed of 3 to 4.
Saturation by a low pC <sub>2</sub> H <sub>2</sub> - <i>K<sub>m</sub></i> (C <sub>2</sub> H <sub>2</sub> ) of 0.003-0.008 atm <i>c.f.</i> <i>K<sub>m</sub></i> (N <sub>2</sub> ) of 0.02 to 0.16 atm.
Specificity of Reaction Product. No significant product other than C <sub>2</sub> H <sub>4</sub> .
Low Background. Negligible C <sub>2</sub> H <sub>4</sub> formed in the absence of C <sub>2</sub> H <sub>2</sub> <i>c.f.</i> the natural background of <sup>15</sup> N.
Metabolic Stability of Product. C <sub>2</sub> H <sub>4</sub> is not metabolized <i>c.f.</i> the metabolic conversions of inorganic and organic forms of nitrogen.
<i>Disadvantages</i>
Indirect Nature of Reaction. The possibility exists for non-N <sub>2</sub> ase catalysis of this reduction by samples of the biosphere.
Explosive Nature of C <sub>2</sub> H <sub>2</sub> . C <sub>2</sub> H <sub>2</sub> is a highly explosive gas <i>c.f.</i> N <sub>2</sub> which is relatively inert.

of free-living nitrogen-fixing micro-organisms in soils, but we are nearly as much in the dark now---. There is no sound base upon which to make direct estimate---", while only a year ago the Subcommittee of Production Processes of the United States National Committee for the International Biological Program (37) suggested the following immediate action with respect to nitrogen fixation: "Providing means for the use of available specialized instruments (*e.g.*, mass spectrometers) in the accomplishment of routine analytical determinations." With the advent of the  $C_2H_2-C_2H_4$  assay procedure, these statements are no longer relevant. The  $C_2H_2-C_2H_4$  assay is as crucial for the optimization of biological nitrogen fixation as soil analyses were fundamental to the development of agricultural fertilizer use.

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