

## Supporting Information

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Conformal 3D Li/Li<sub>13</sub>Sn<sub>5</sub> Scaffolds Anodes for High-Areal Energy Density Flexible Lithium Metal Batteries

*Xiaomei Huo, Xin Gong, Yuhang Liu, Yonghui Yan, Zhuzhu Du and Wei Ai\**

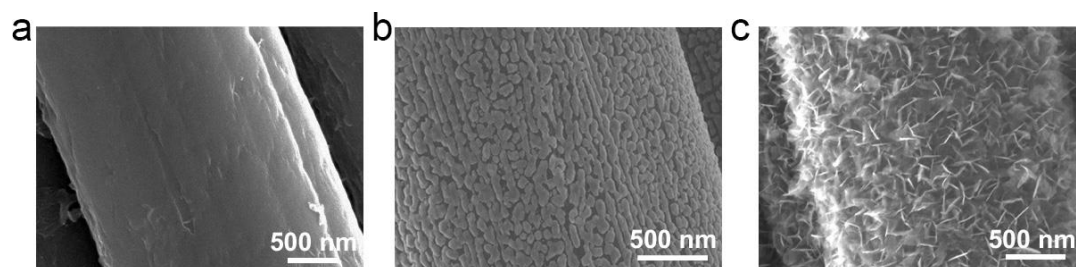
## Supporting Information

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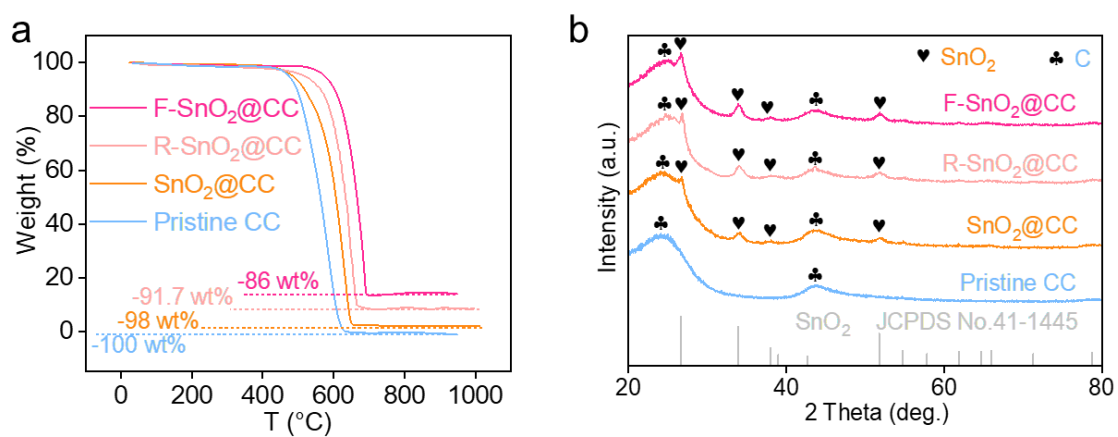
*Xiaomei Huo, Xin Gong, Yuhang Liu, Yonghui Yan, Zhuzhu Du and Wei Ai\**

Frontiers Science Center for Flexible Electronics & Xi'an Institute of Flexible Electronics,  
Northwestern Polytechnical University, Xi'an 710072, China.

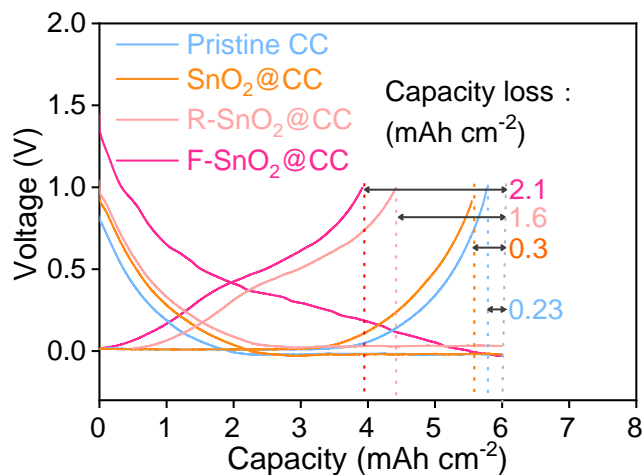
E-mail: iamwai@nwpu.edu.cn



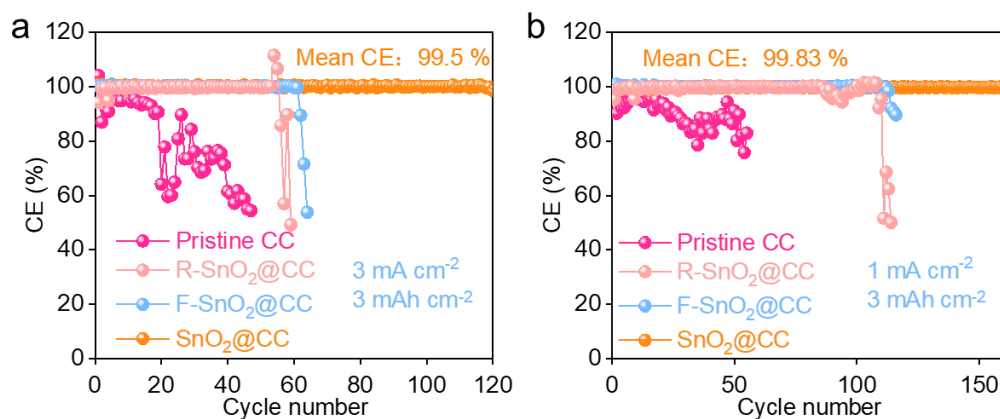
**Figure S1.** SEM images of (a) pristine CC, (b) R-SnO<sub>2</sub>@CC and (c) F-SnO<sub>2</sub>@CC.



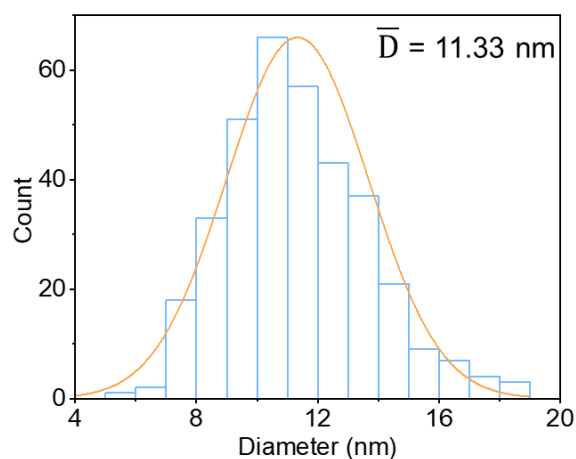
**Figure S2.** (a) TGA curves of pristine CC, SnO<sub>2</sub>@CC, R-SnO<sub>2</sub>@CC and F-SnO<sub>2</sub>@CC. (b) XRD patterns of pristine CC, SnO<sub>2</sub>@CC, R-SnO<sub>2</sub>@CC and F-SnO<sub>2</sub>@CC in comparison with the standard diffraction peaks for rutile SnO<sub>2</sub> (JCPDS No. 41-1445).



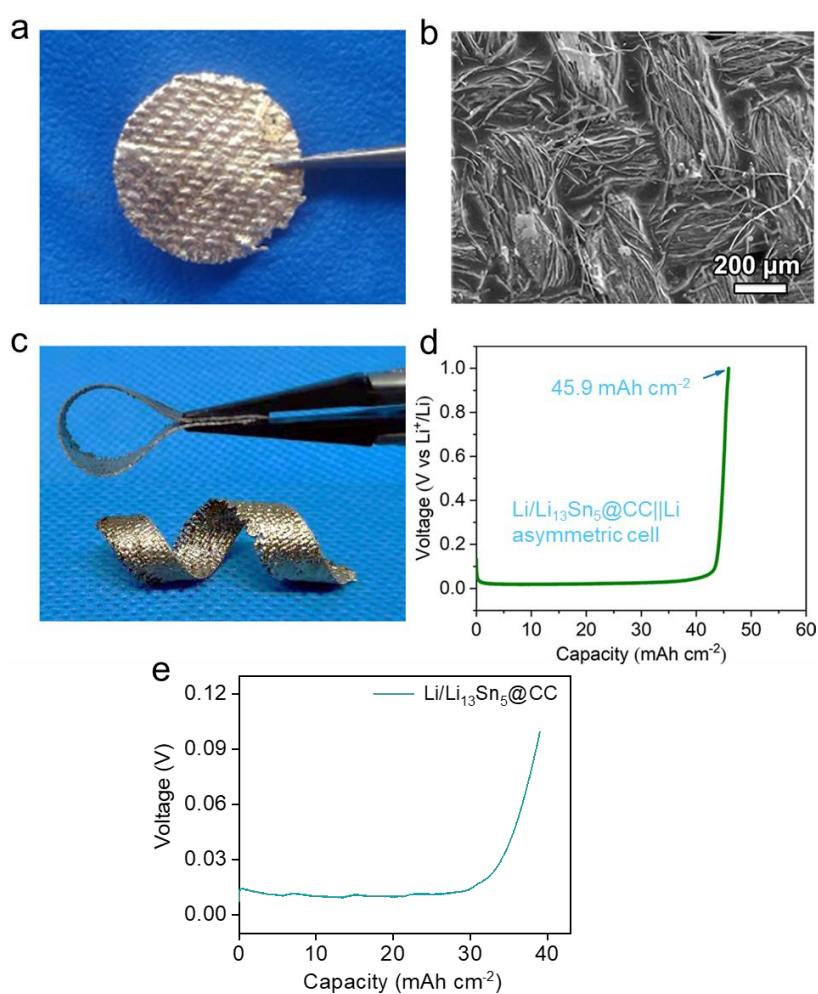
**Figure S3.** Capacity-voltage curves of pristine CC, SnO<sub>2</sub>@CC, R-SnO<sub>2</sub>@CC and F-SnO<sub>2</sub>@CC electrodes plated at 0.5 mA cm<sup>-2</sup> for 6 mAh cm<sup>-2</sup> and stripped to 1 V cut-off voltage.



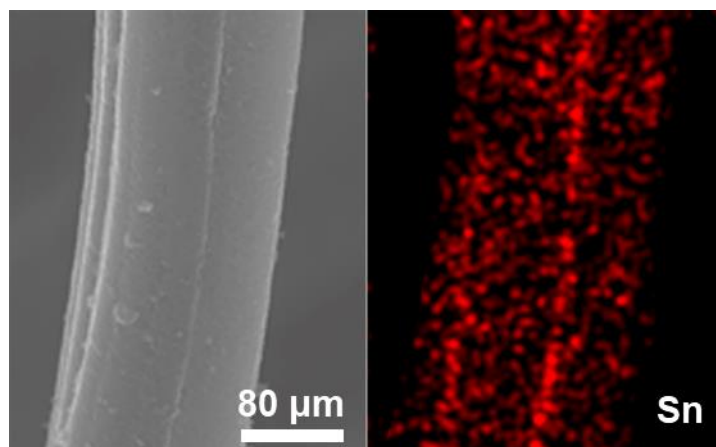
**Figure S4.** Coulombic efficiency of the pristine CC, SnO<sub>2</sub>@CC, R-SnO<sub>2</sub>@CC and F-SnO<sub>2</sub>@CC electrodes at (a) 3 mA cm<sup>-2</sup> for 3 mAh cm<sup>-2</sup> and (b) 1 mA cm<sup>-2</sup> for 3 mAh cm<sup>-2</sup>.



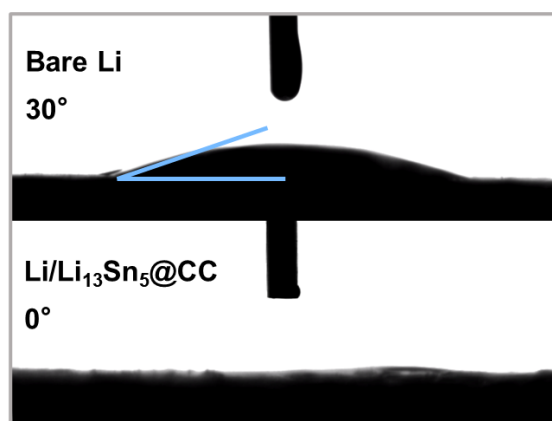
**Figure S5.** The particle size distribution curve of SnO<sub>2</sub> on CC.



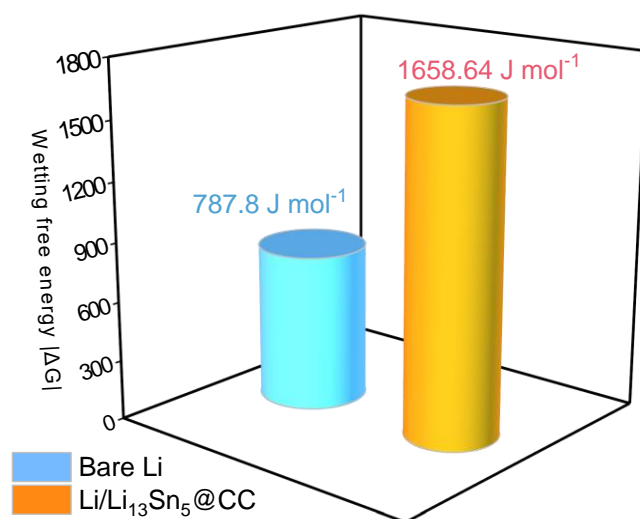
**Figure S6.** (a) Optical and (b) SEM images of Li/Li<sub>13</sub>Sn<sub>5</sub>@CC. (c) Optical images showing the twisting and bending of Li/Li<sub>13</sub>Sn<sub>5</sub>@CC. (d) Capacity-voltage curve of the Li/Li<sub>13</sub>Sn<sub>5</sub>@CC electrode after fully Li stripping. (e) Capacity-voltage curve of the Li/Li<sub>13</sub>Sn<sub>5</sub>@CC electrode after Li stripping to 0.1V.



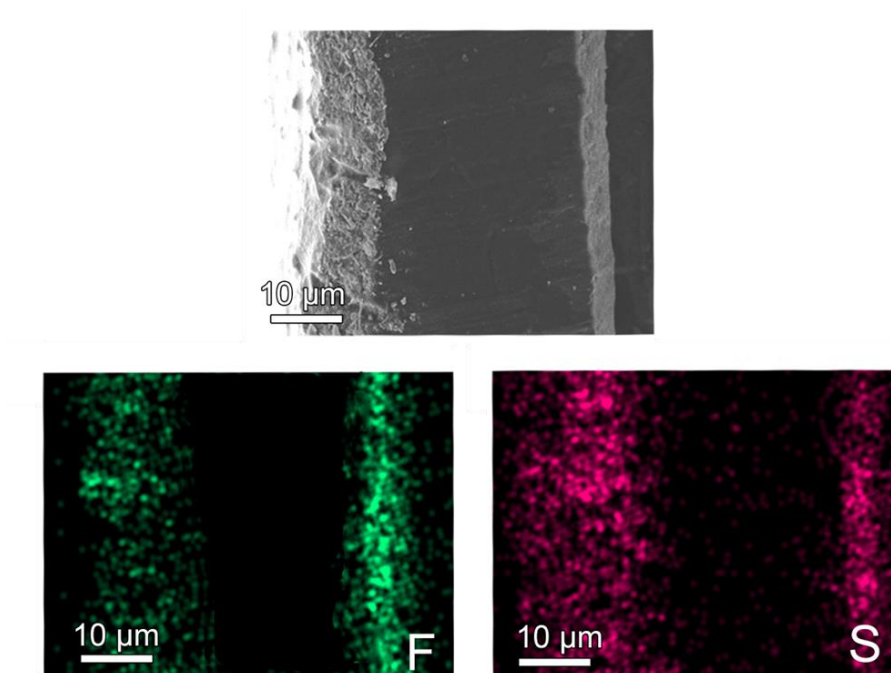
**Figure S7.** SEM and the Sn elemental mapping images of Li/Li<sub>13</sub>Sn<sub>5</sub>@CC electrode after Li stripping.



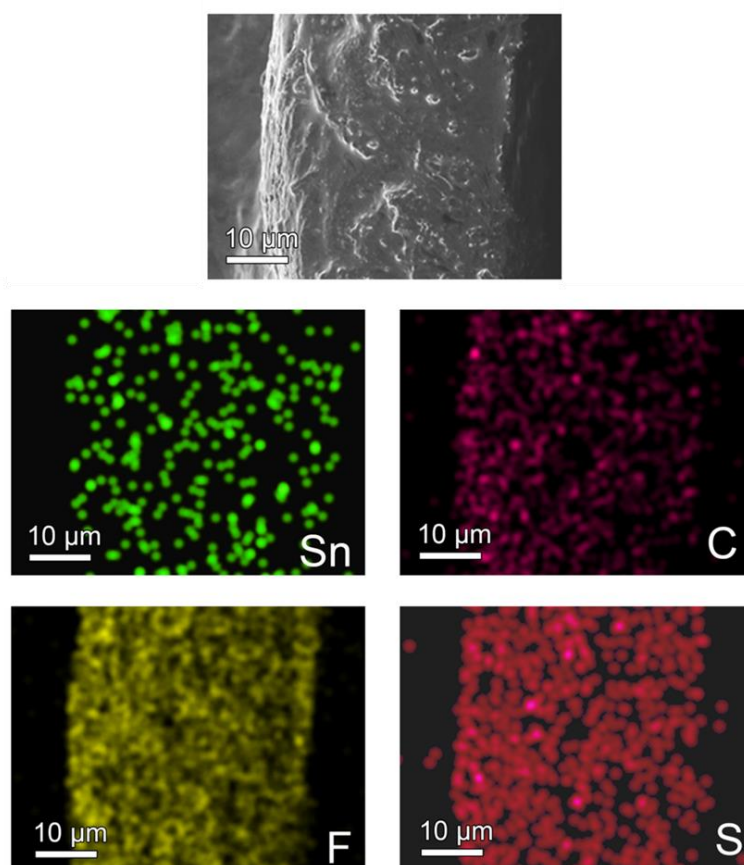
**Figure S8.** Contact angle test of bare Li and Li/Li<sub>13</sub>Sn<sub>5</sub>@CC electrode.



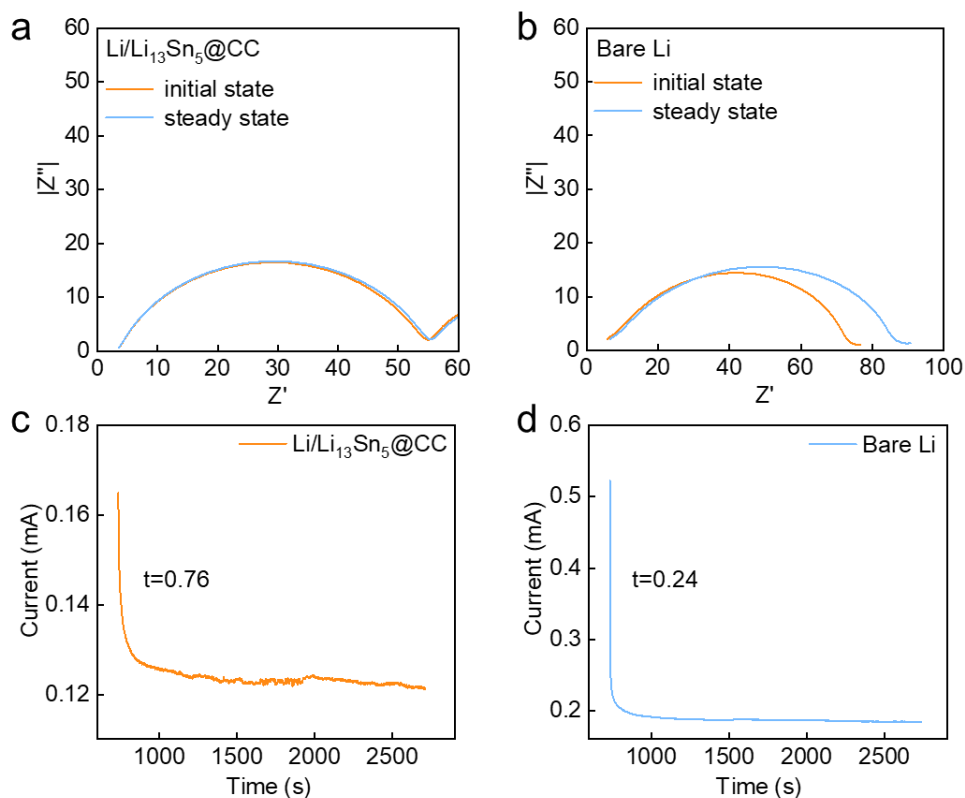
**Figure S9.** The wetting free energy of bare Li and Li/Li<sub>13</sub>Sn<sub>5</sub>@CC electrode.



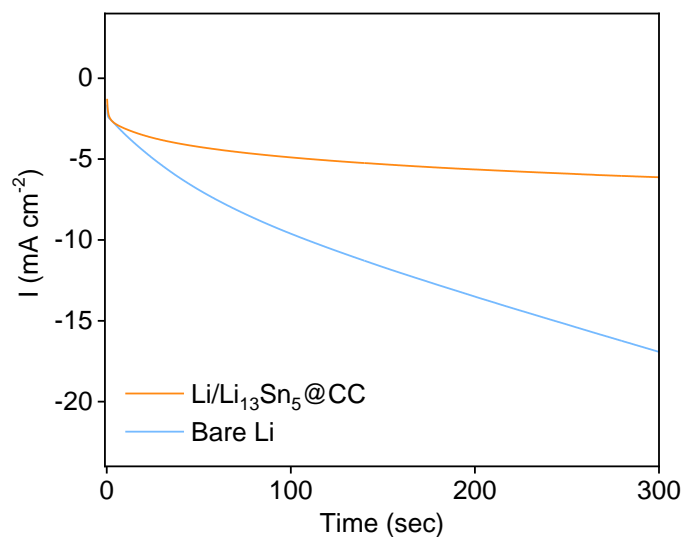
**Figure S10.** Cross-section SEM image of bare Li and its F and S elemental mappings.



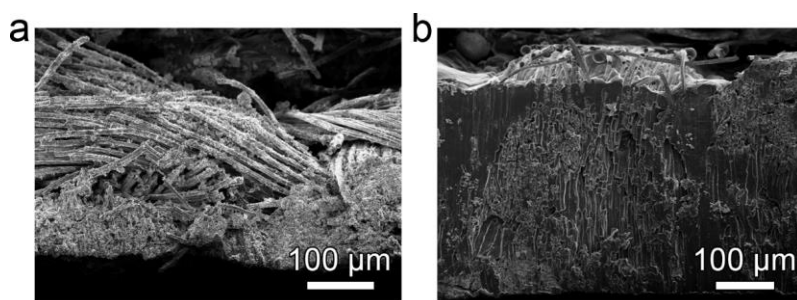
**Figure S11.** Cross-section SEM image of Li/Li<sub>13</sub>Sn<sub>5</sub>@CC and its Sn, C, F and S elemental mappings.



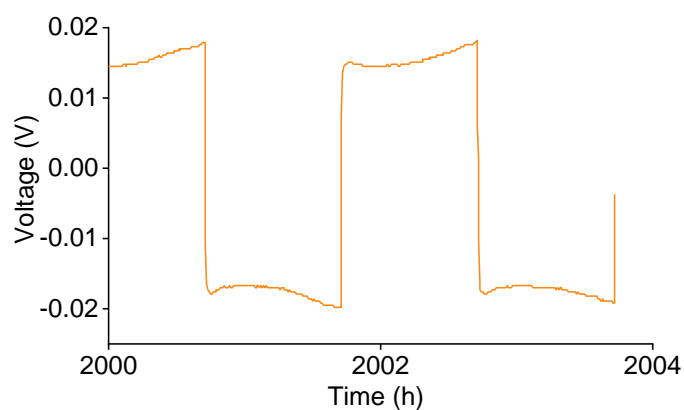
**Figure S12.** (a, b) EIS curves and (c, d) the associated current variation with time during polarization of Li/Li<sub>13</sub>Sn<sub>5</sub>@CC and bare Li symmetrical cells with applied potential difference of 10 mV.



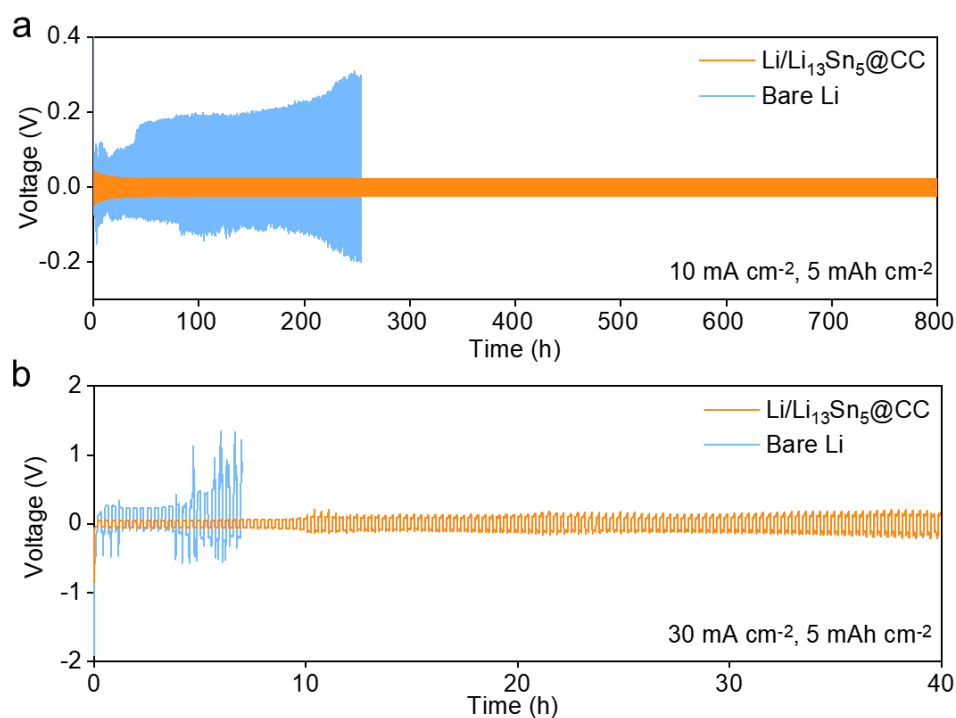
**Figure S13.** Chronoamperometry curves of Li/Li<sub>13</sub>Sn<sub>5</sub>@CC and bare Li at an overpotential of -150 mV.



**Figure S14.** Cross-section SEM image of Li/Li<sub>13</sub>Sn<sub>5</sub>@CC after (a) 40 mAh cm<sup>-2</sup> Li stripping and (b) 40 mAh cm<sup>-2</sup> Li replating (87.1% DOD).

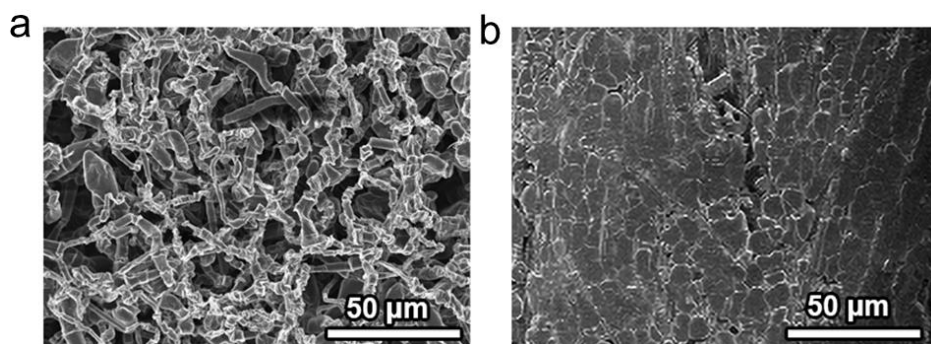


**Figure S15.** Enlarged voltage-time curves of cycling at 5 mA cm<sup>-2</sup> for 5 mAh cm<sup>-2</sup>.

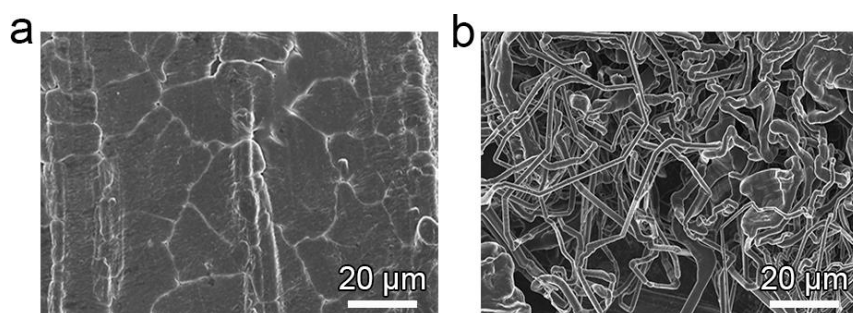


**Figure S16.** Cycling performance of Li/Li<sub>13</sub>Sn<sub>5</sub>@CC and bare Li symmetric cells at (a) 10 mA cm<sup>-2</sup> and (b) 30 mA cm<sup>-2</sup> with a fixed capacity of 5 mAh cm<sup>-2</sup>.

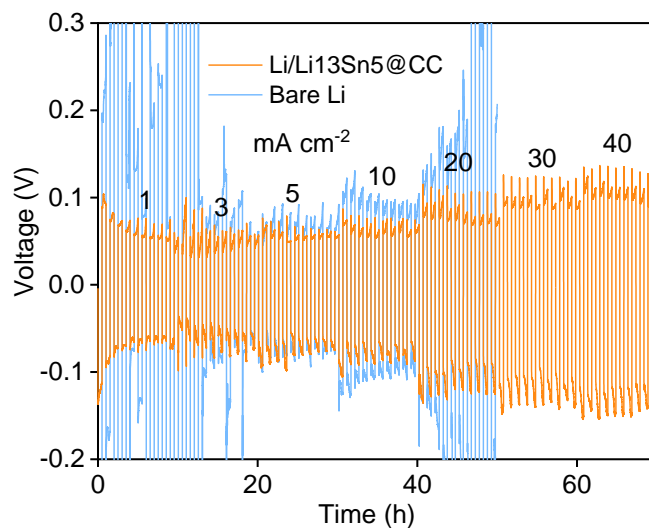




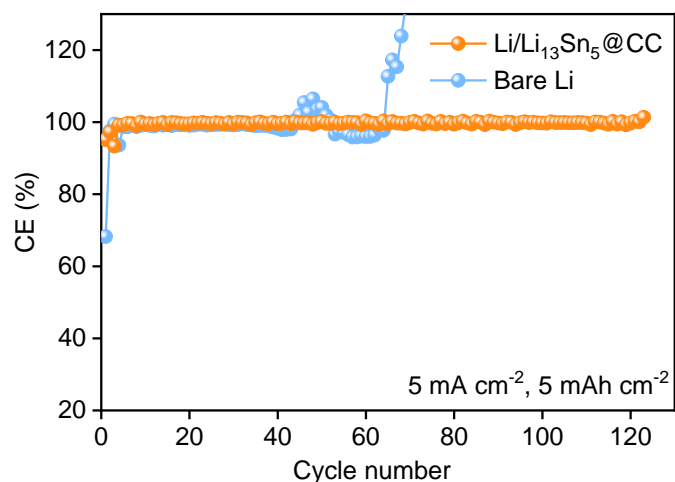
**Figure S17.** SEM images of (a) bare Li and (b) Li/Li<sub>13</sub>Sn<sub>5</sub>@CC after 50 cycles test at 30 mA cm<sup>-2</sup> and 5 mAh cm<sup>-2</sup>.



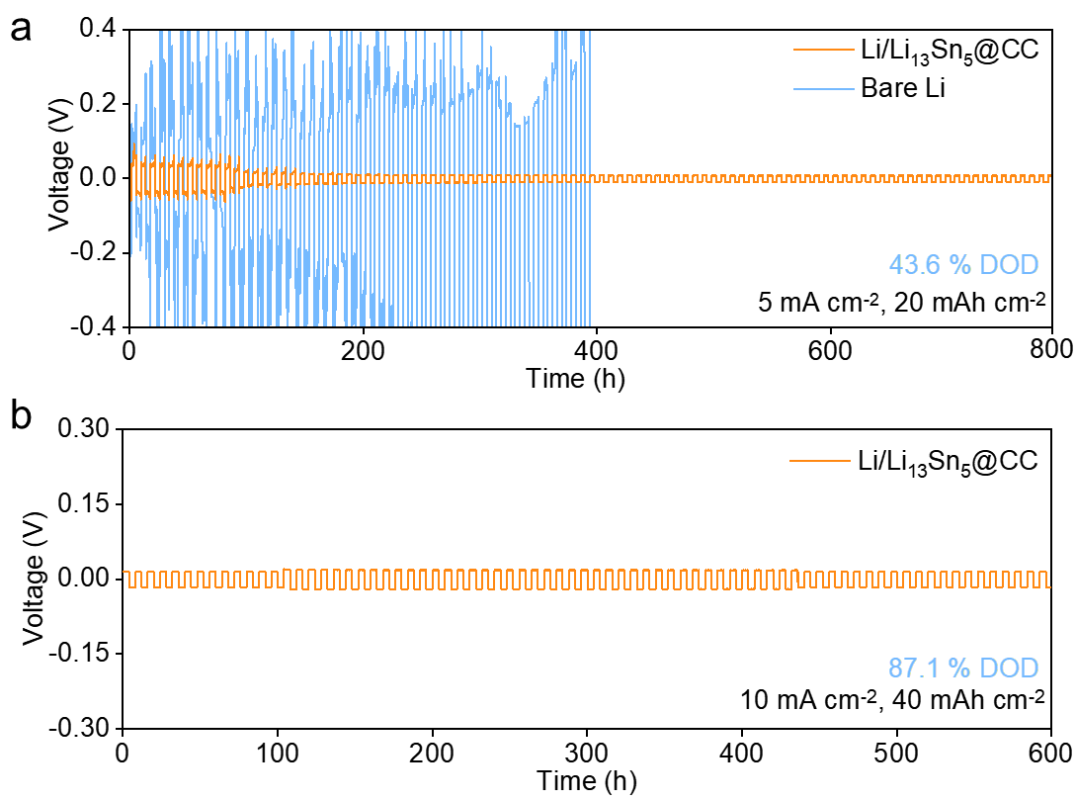
**Figure S18.** SEM images of (a) Li/Li<sub>13</sub>Sn<sub>5</sub>@CC after cycles of 2600 h and (b) bare Li after cycles of 700 h at 5 mA cm<sup>-2</sup> and 5 mAh cm<sup>-2</sup>.



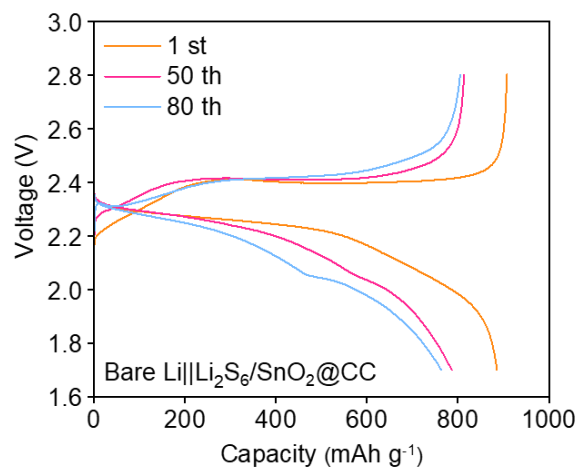
**Figure S19.** Rate performance of the symmetrical cells.



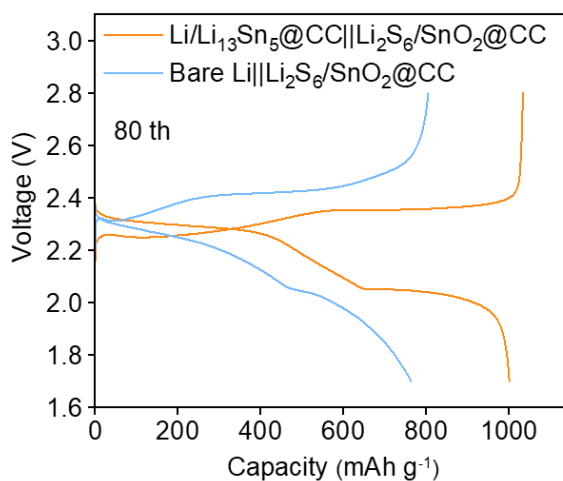
**Figure S20.** Coulombic efficiency of the Li/Li<sub>13</sub>Sn<sub>5</sub>@CC and bare Li electrodes at 5 mA cm<sup>-2</sup> for 5 mAh cm<sup>-2</sup>.



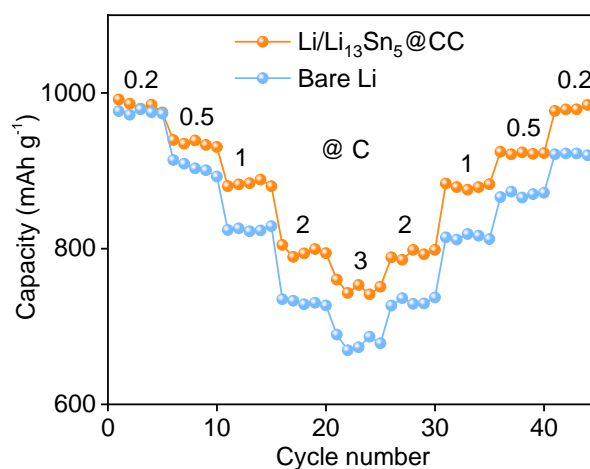
**Figure S21.** Cycling performance of Li/Li<sub>13</sub>Sn<sub>5</sub>@CC and bare Li symmetric cells at 5 mA cm<sup>-2</sup> with a fixed capacity of (a) 20 mAh cm<sup>-2</sup> and (b) 10 mA cm<sup>-2</sup> with a fixed capacity of 40 mAh cm<sup>-2</sup>.



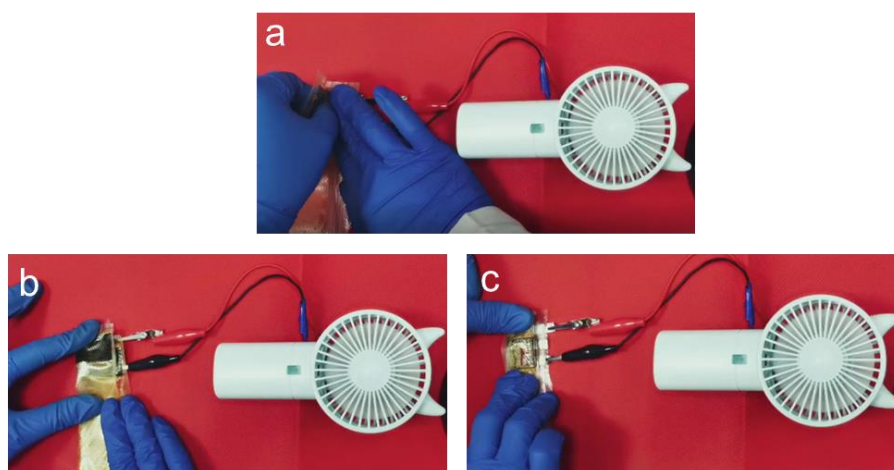
**Figure S22.** Charge-discharge voltage curves of bare Li||Li<sub>2</sub>S<sub>6</sub>/SnO<sub>2</sub>@CC cell at different cycles.



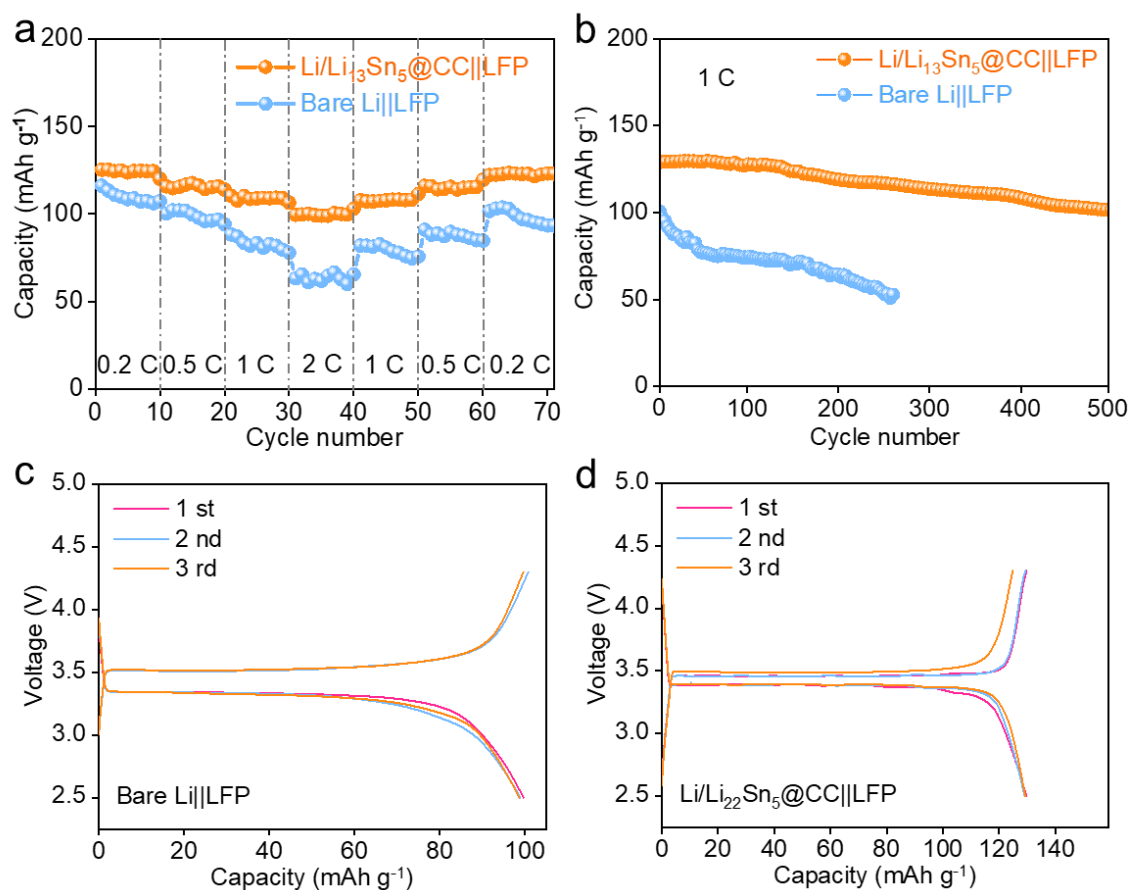
**Figure S23.** Comparison of Charge-discharge voltage curves of bare Li||Li<sub>2</sub>S<sub>6</sub>/SnO<sub>2</sub>@CC and Li/Li<sub>13</sub>Sn<sub>5</sub>@CC||Li<sub>2</sub>S<sub>6</sub>/SnO<sub>2</sub>@CC cells at 80th cycle.



**Figure S24.** Rate capacities of Li/Li<sub>13</sub>Sn<sub>5</sub>@CC||Li<sub>2</sub>S<sub>6</sub>/SnO<sub>2</sub>@CC and Li||Li<sub>2</sub>S<sub>6</sub>/SnO<sub>2</sub>@CC batteries at various current rates from 0.2 C to 3 C.



**Figure S25.** The Li/Li<sub>13</sub>Sn<sub>5</sub>@CC||Li<sub>2</sub>S<sub>6</sub>/SnO<sub>2</sub>@CC pouch cell powering electric fans under (a) 90° bending, (b) one folding and (c) two folding states, respectively.



**Figure S26.** Electrochemical performance of full cell. (a) Rate performance. (b) Cycling performance at 1 C. Charge-discharge voltage curves of (c) bare Li||LFP full cell and (d) Li/Li<sub>13</sub>Sn<sub>5</sub>@CC||LFP full cell at different cycles.

**Table S1.** The actual weight change after molten Li infusion.

	Pristine CC	SnO <sub>2</sub> @CC	R-SnO <sub>2</sub> @CC	F-SnO <sub>2</sub> @CC
Pre-lithiation capacity (mAh cm <sup>-2</sup> )	2.1	2.3	2.7	6
Irreversible capacity (mAh cm <sup>-2</sup> )	0.23	0.3	1.6	2.1

**Table S2.** Comparison of symmetric cell with the recently reported Sn-based LMAs.

Types	Electrode	Current density (mA cm <sup>-2</sup> )	Areal capacity (mAh cm <sup>-2</sup> )	Cycles	Voltage hysteresis (mV)	Sn-containing (%)	Ref.
3D-host materials	Li/Li <sub>13</sub> Sn <sub>5</sub> @CC	5	5	1000	31.6	2	This work
		30	5	100	192		
		5	20	100	21		
		10	30	100	19.2		
	Li-Li <sub>x</sub> Sn@CF	1	1	372	20	50	1
		3	1	180	160		
	Li-Li <sub>x</sub> Sn@CT	1	1	110	20	2.5	2
		3	1	55	40		
	Li-Li <sub>x</sub> Sn@CP	1	1	400	18	30	3
		2	1	250	20		
		5	1	100	120		
	Sn/C/Li	1	1	900	15	25	4
		2	1	1200	22		
		5	1	700	60		
	Li-Sn@NF	5	5	100	9.5	5	5
		10	5	200	27		
		3	1	300	100		
		10	1	1000	200		
	Li-Co@CS	1	1	400	50	/	6
		20	1	600	20		
1		3	120	280			
Li-Mn/G foam	1	1	800	30	/	7	
	2	1	300	12			
Li@Cu	1	1	800	12	/	8	
	1	6	200	15			
	5	1	150	60			
Li-N@CF	1	1	300	30	/	9	
	3	1	270	50			
Layered Li-rGO	1	1	100	100	/	10	
	3	1	46	200			
Protective layer	Li <sub>3</sub> Mg <sub>7</sub> @Li	0.5	1	1000	11	/	11
		2	1	900	15		

	1	5	85	20		
PTMEG@Li	1	1	500	10	/	12
	1	1	400	20		
Li <sub>x</sub> Sn@Li	3	1	150	50	/	13
	5	1	250	100		
Li <sub>2</sub> S@Li	5	2	937	20	/	14
	0.5	1	1000	150		
LiF@Li	1	1	280	200	/	15
	2	1	500	350		

**Table S3.** Comparison of Li/Li<sub>13</sub>Sn<sub>5</sub>@CC||Li<sub>2</sub>S<sub>6</sub>/SnO<sub>2</sub>@CC||Li/Li<sub>13</sub>Sn<sub>5</sub>@CC cell under various deformation states with previously reported Li||S cells

Cell	S loading (mg cm <sup>-2</sup> )	Current density (mA cm <sup>-2</sup> )	Areal capacity (mAh cm <sup>-2</sup> )	Areal energy density (mWh cm <sup>-2</sup> )	Ref.
Li/Li <sub>13</sub> Sn <sub>5</sub> @CC  Li <sub>2</sub> S <sub>6</sub> /SnO <sub>2</sub> @CC	<b>6.33</b>	<b>1.06</b>	<b>5.04</b>	<b>10.6</b>	<b>This work</b>
LV/Li-S	2	1.68	2.64	2.59	16
Li/CuCF-NSHG/S/NiCF	3.2	1	2	4.2	17
Li-HMSC	7.26	1	4	8.2	18
LiCSMF-S/CSMF	1.28	2	0.9	1.89	19
Li/CC-Graphene/S	5.1	4.27	4	8.2	20

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