

Supplementary Information for

Growth of millimeter-sized 2D metal iodide crystals induced by ion-specific preference at water-air interfaces

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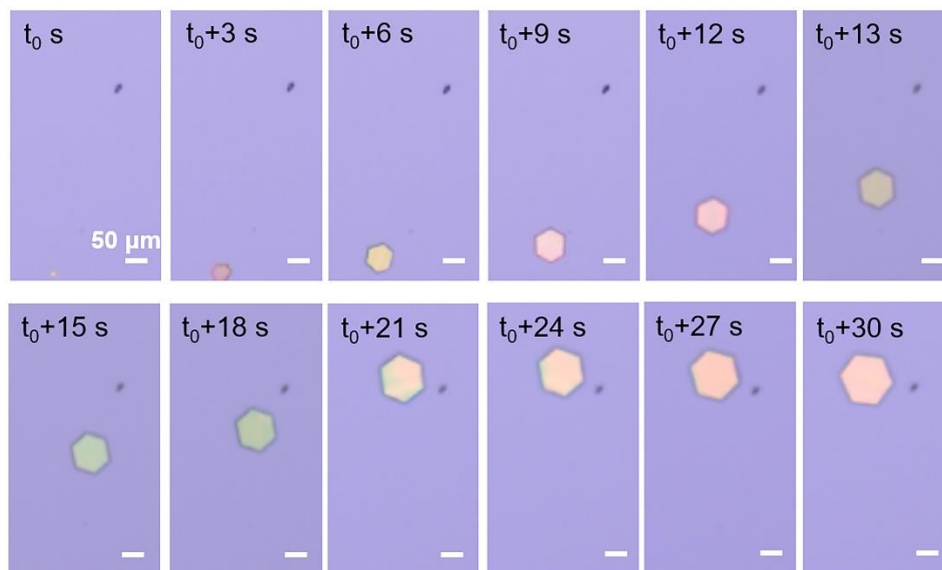
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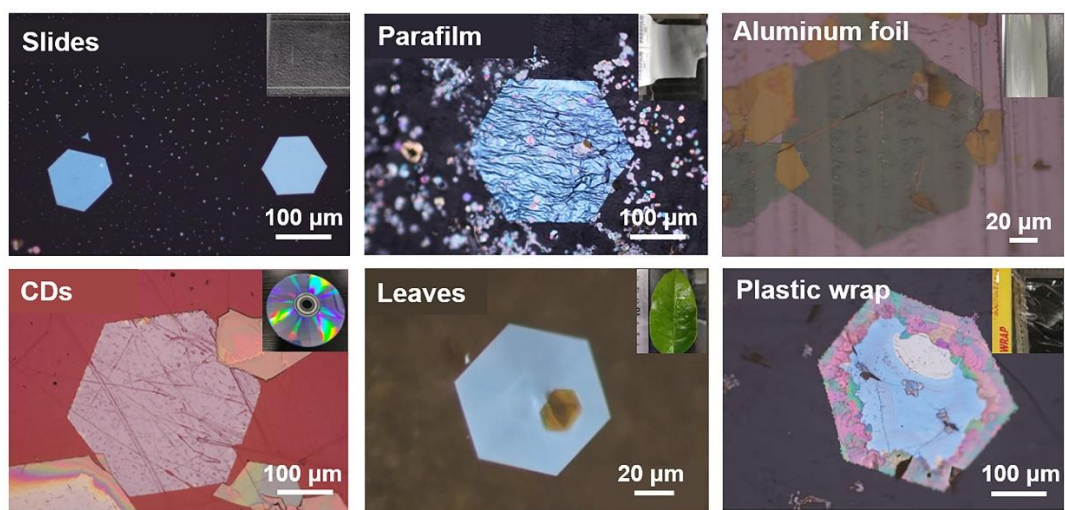
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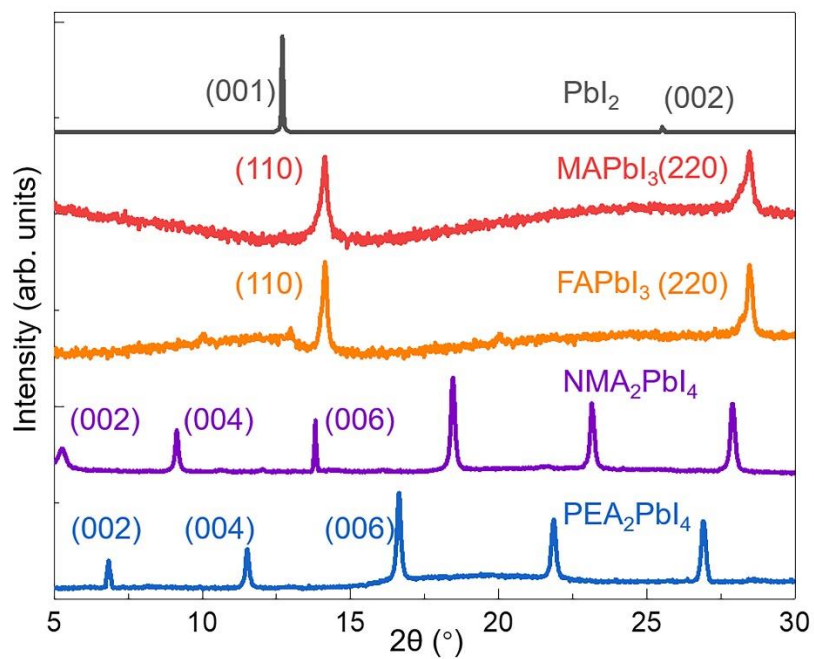
Supplementary Figures



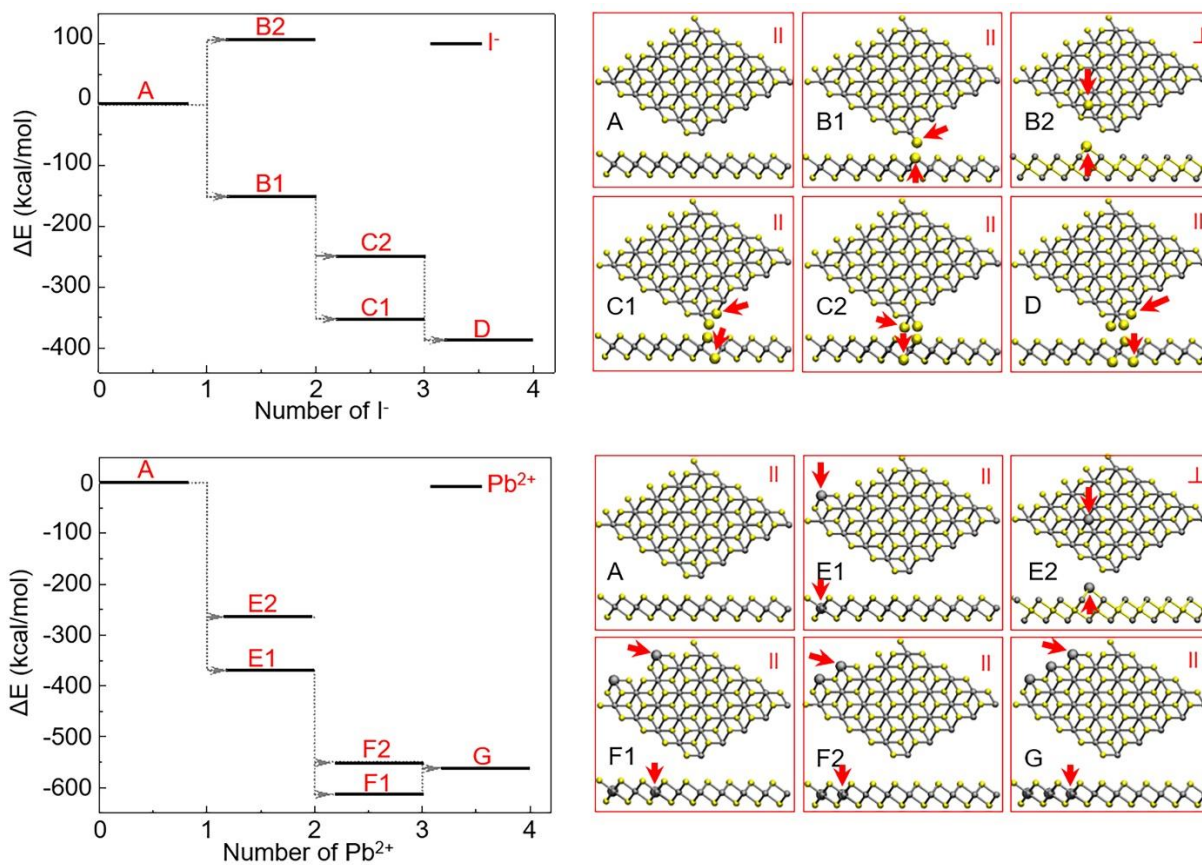
Supplementary Fig. 1 | Snapshots of PbI_2 growth at the water-air interface ($t_0 \approx 200$ s).



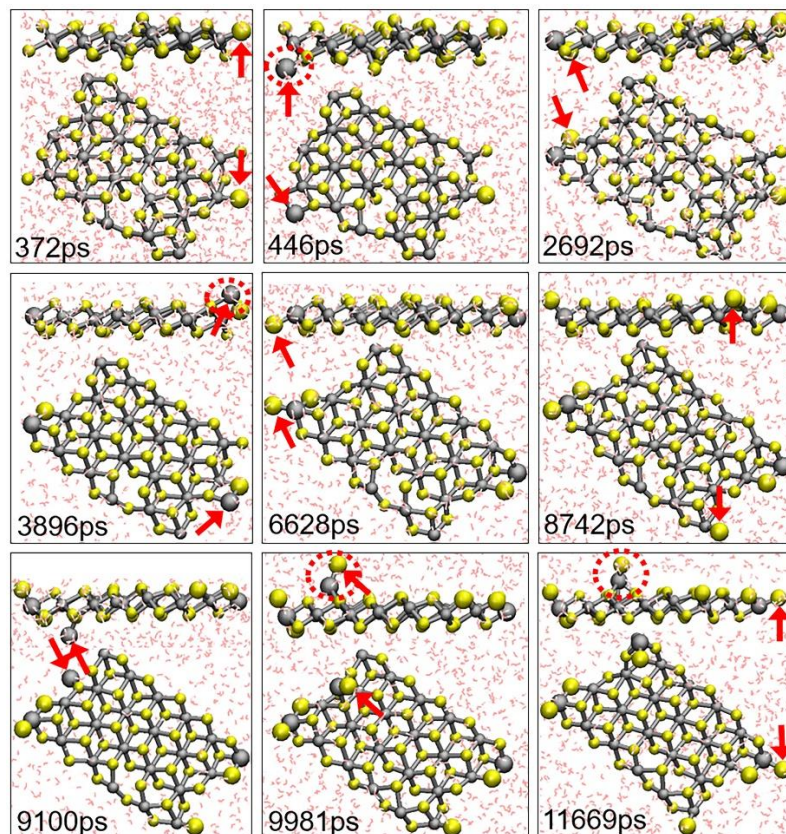
Supplementary Fig. 2 | PbI₂ crystals grown on the slides, parafilm, aluminum foil, CDs, leaves, and plastic wrap.



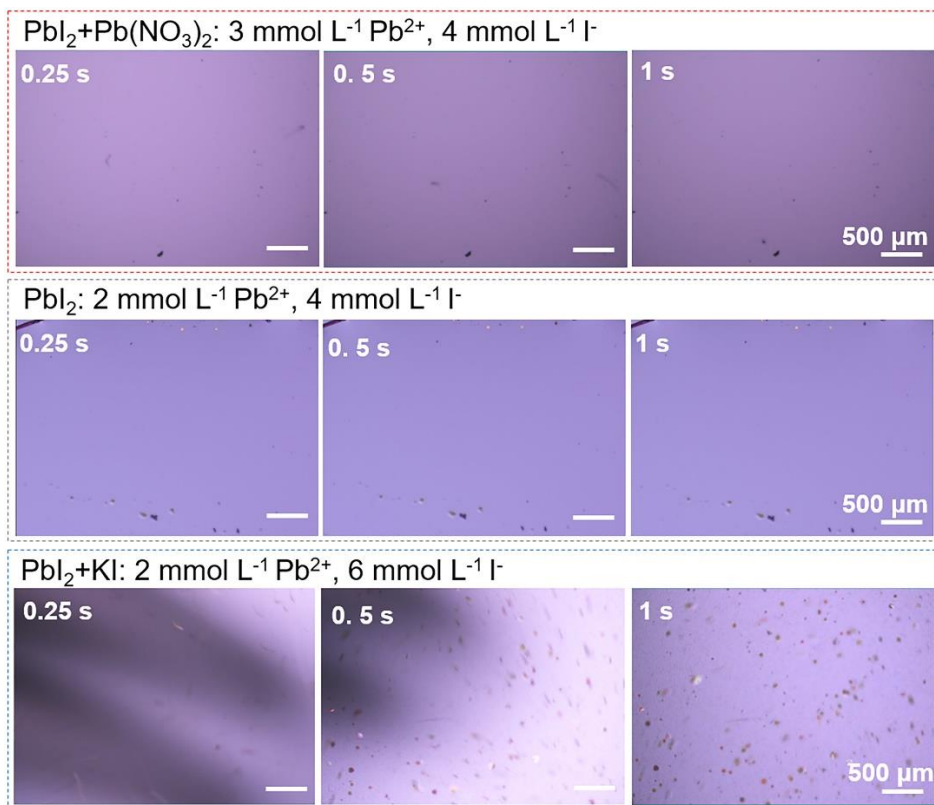
Supplementary Fig. 3 | X-ray diffraction pattern results of PbI_2 and various perovskite nanosheets (MAPbI_3 , FAPbI_3 , NMA_2PbI_4 , and PEA_2PbI_4)¹⁻³.



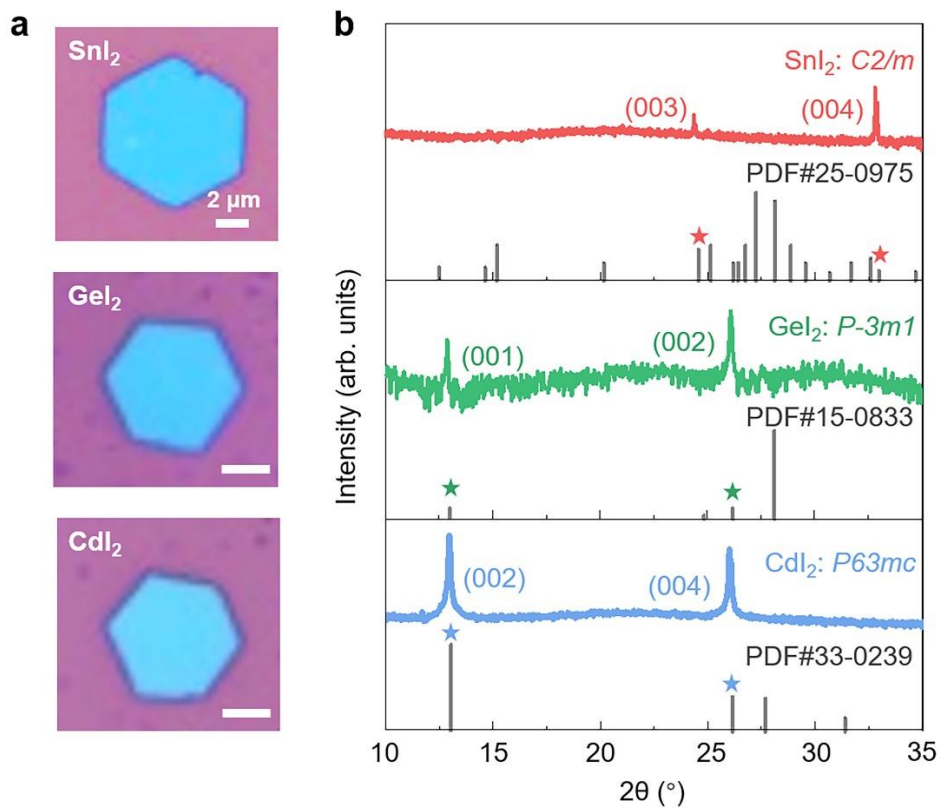
Supplementary Fig. 4 | Energy diagram of the stepwise growth of I^- ions on the nanosheet, where A represents the initial nanosheet. B represents the energy required for adding an I^- ion, with B1 indicating lateral growth and B2 vertical growth. C represents the continuous growth of an I^- ion based on the energy profile B. C1 and C2 correspond to two different lateral growth modes. D depicts the energy diagram for further growth of an I^- ion based on C.



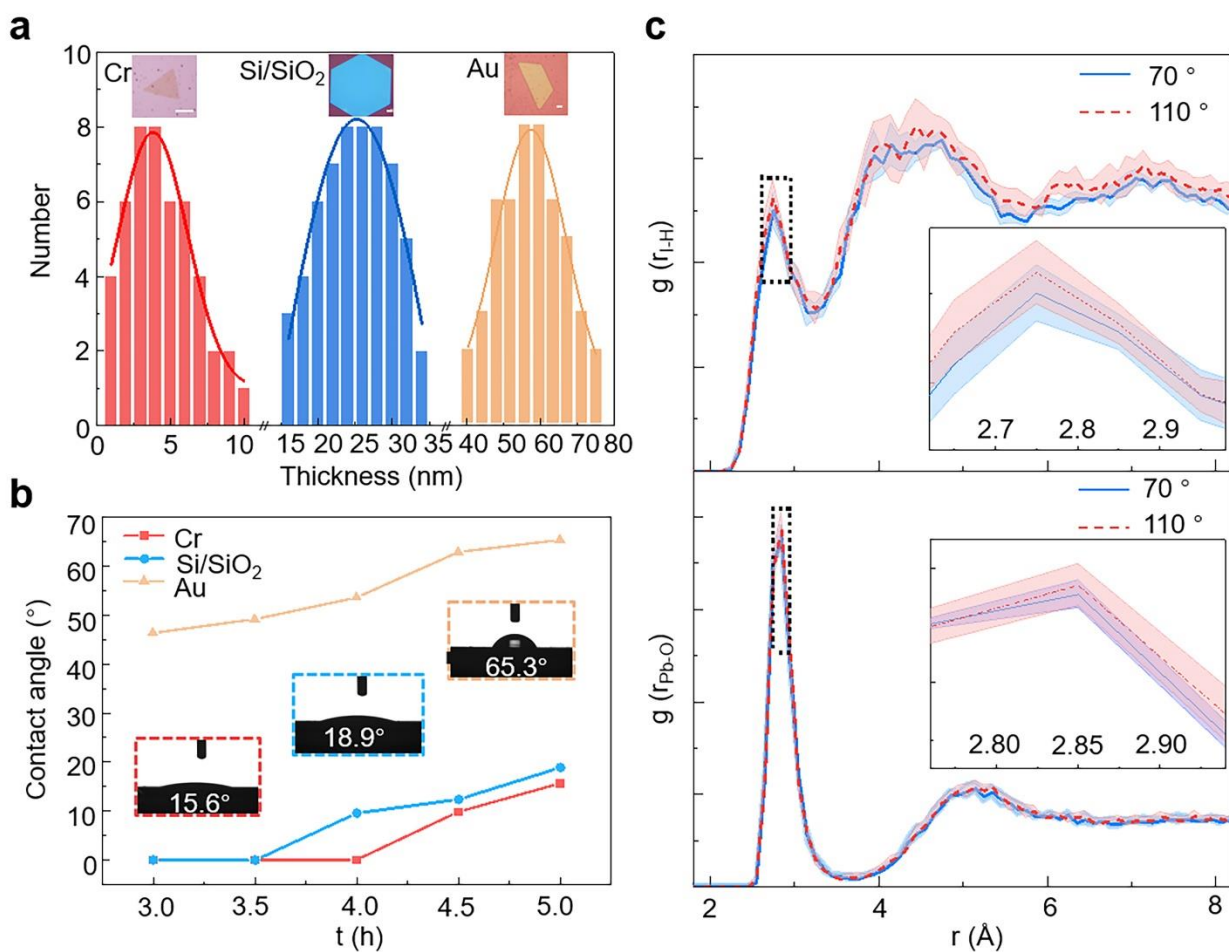
Supplementary Fig.5 | Time-lapse snapshots from molecular dynamics simulations illustrating the growth of the PbI_2 nanosheet in the bulk water, akin to Fig. 2f. The small grey and yellow spheres depict the Pb and I atoms of a pre-existing PbI_2 nanosheet, respectively. The large grey and yellow spheres represent the Pb and I ions initially dissolved in water and are marked with arrows, respectively. Dashed circles denote the vertical growth of ions in the water solution, while the remaining elements depict the horizontal growth of ions in the water solution.



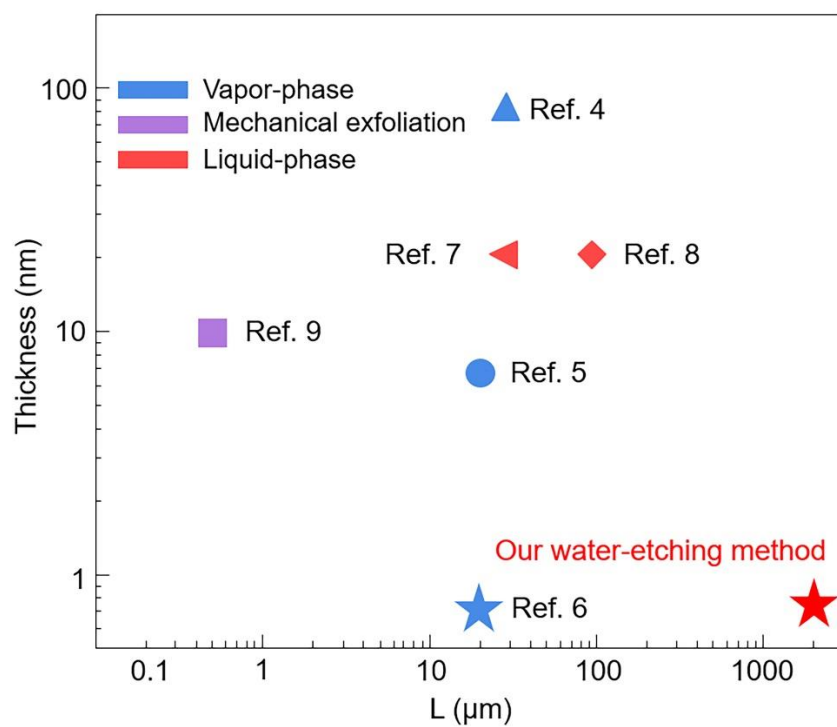
Supplementary Fig. 6 | Snapshots of PbI_2 growth with the addition of $\text{Pb}(\text{NO}_3)_2$ and KI.



Supplementary Fig. 7 | Water-air interfacial growth of other metal iodides. Optical images (a) and XRD results (b) of SnI_2 , GeI_2 , and CdI_2 .



Supplementary Fig. 8 | Effect of solution contact-angle on PbI₂ growth. (a) Statistical distribution of PbI₂ nanosheet thicknesses using different substrates. The solid lines are normal distribution fitting of the experimental data. (b) Variation in the contact angle of PbI₂ aqueous solution on different substrates over time. (c) Hydration numbers of Pb²⁺ and I⁻ ions in cases with different contact angles. By analyzing the integral area, a contact angle of 110° corresponds to a larger integral area, indicating a higher number of hydrations of Pb²⁺ and I⁻, and thus closer to the bulk phase. In contrast, a contact angle of 70° corresponds to a smaller integral area, indicating fewer hydrations and a closer proximity to the interface.



Supplementary Fig. 9 | Comparison of the lateral size (L) and thickness of PbI₂ single crystals prepared using previous methods and our methods ⁴⁻⁹.

Supplementary References

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