

Supporting Information

Impact of oligoether side-chain length on the thermoelectric properties of a polar polythiophene

Mariavittoria Craighero¹, Jiali Guo², Sepideh Zokaei¹, Sophie Griggs³, Junfu Tian³, Jesika Asatryan⁴, Joost Kimpel¹, Renee Kroon,⁵ Kai Xu², Juan Sebastian Reparaz², Jaime Martín^{4,6}, Iain McCulloch³, Mariano Campoy-Quiles², Christian Müller^{1*}

¹Department of Chemistry and Chemical Engineering, Chalmers University of Technology, Goteborg 41296, Sweden

*e-mail: christian.muller@chalmers.se

²Materials Science Institute of Barcelona, ICMAB-CSIC, Campus UAB, 08193 Bellaterra, Spain

³Department of Chemistry, University of Oxford, Chemistry Research Laboratory, 12 Mansfield Road, Oxford OX1 3TA, United Kingdom

⁴Universidade da Coruña, Campus Industrial de Ferrol, CITENI, Esteiro, 15403, Ferrol, Spain

⁵Laboratory of Organic Electronics, Linköping University, 60174, Norrköping, Sweden

⁶POLYMAT, Paseo Manuel de Lardizabal 3, 20018, Donostia-San Sebastián, Spain

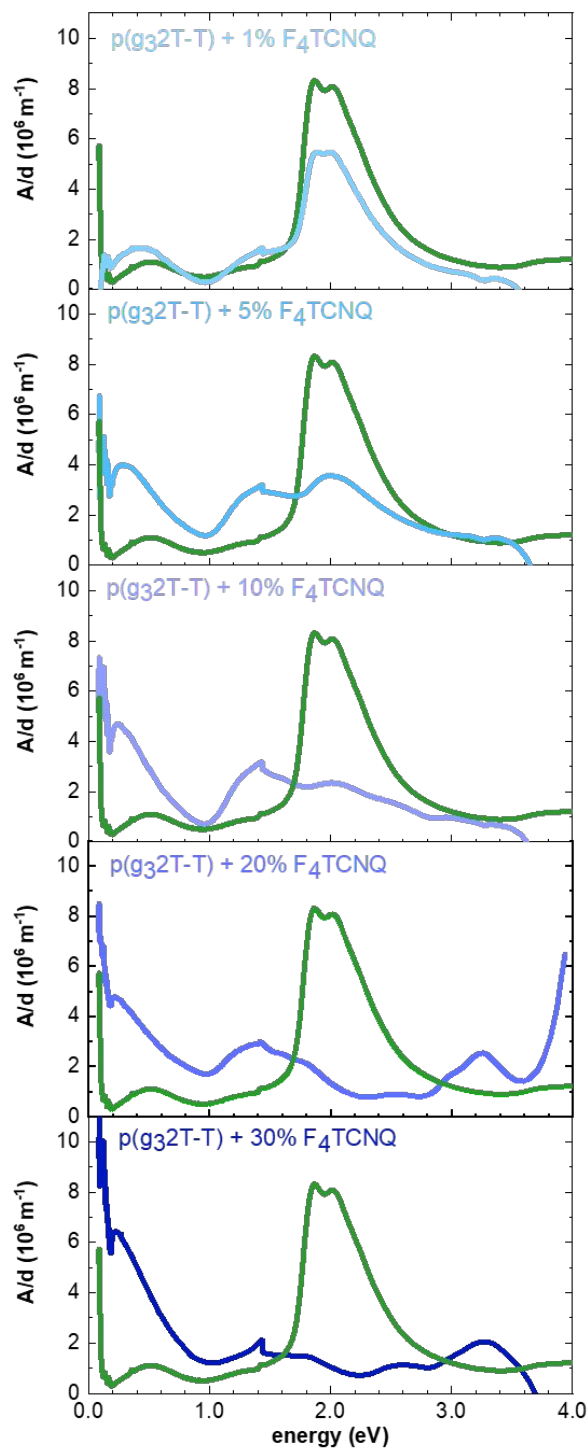


Figure S1. UV-vis absorbance spectra, with the absorbance A normalized by the film thickness d , of p(g₃2T-T) before (green) and after co-processing with 1 mol%, 5 mol%, 10 mol%, 20 mol% and 30 mol% F₄TCNQ (blue).

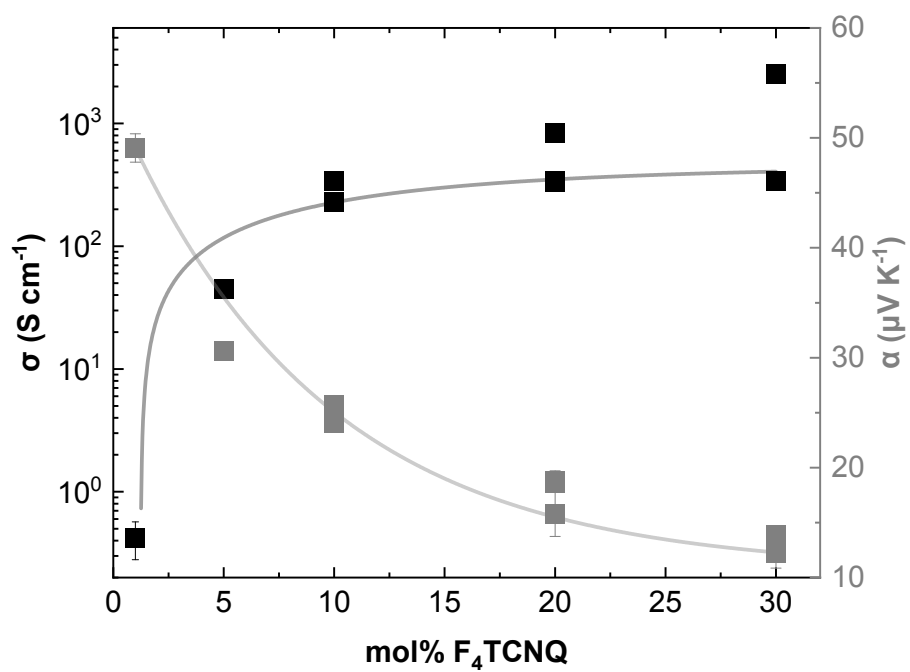


Figure S2. Thermoelectric properties of p(g₃2T-T) co-processed with 1, 5, 10, 20 and 30 mol% F₄TCNQ.

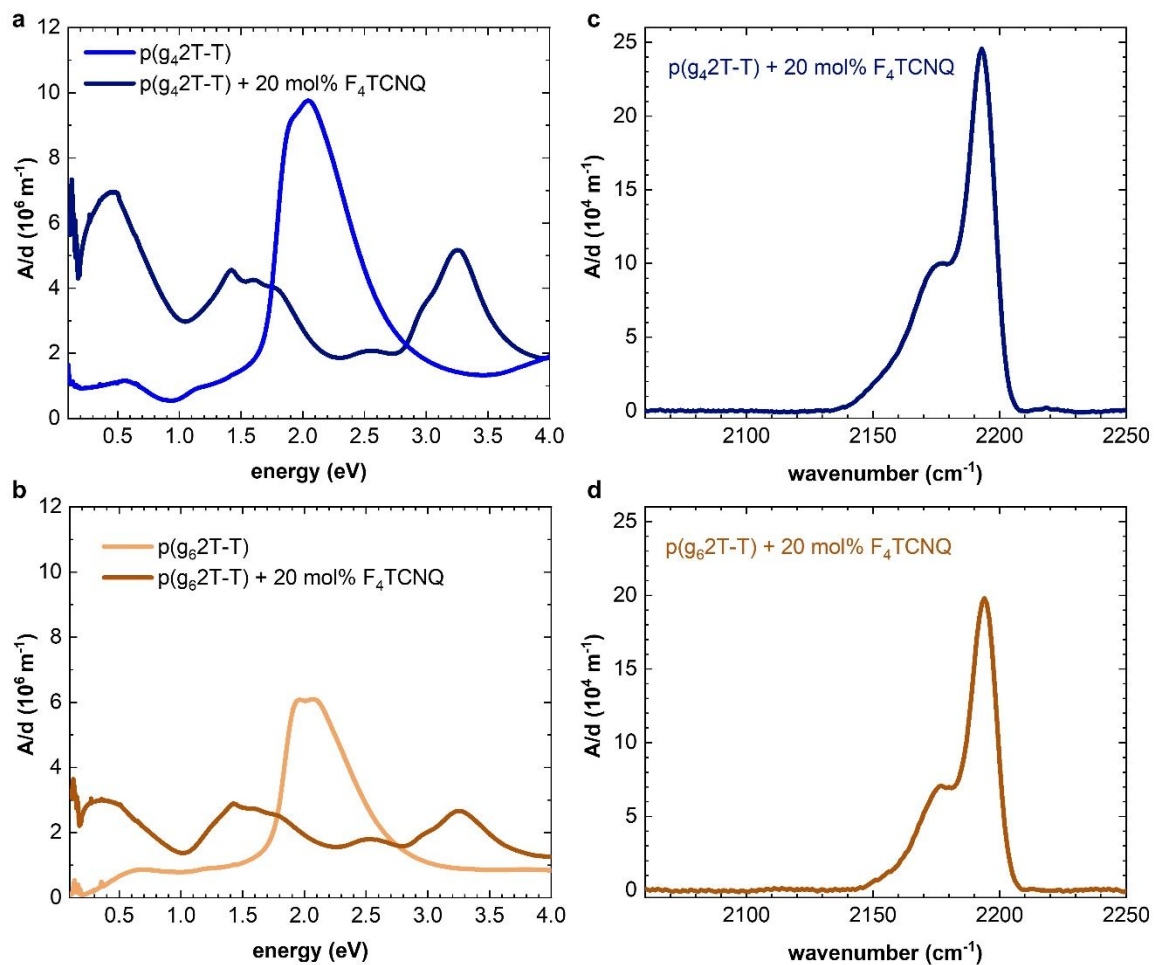


Figure S3. UV-vis (a,b) and transmission FTIR (c,d) absorbance spectra, with the absorbance A normalized by the film thickness d , of $p(g_x2T-T)$ before (light color) and after co-processing with 20 mol% F_4TCNQ (dark color).

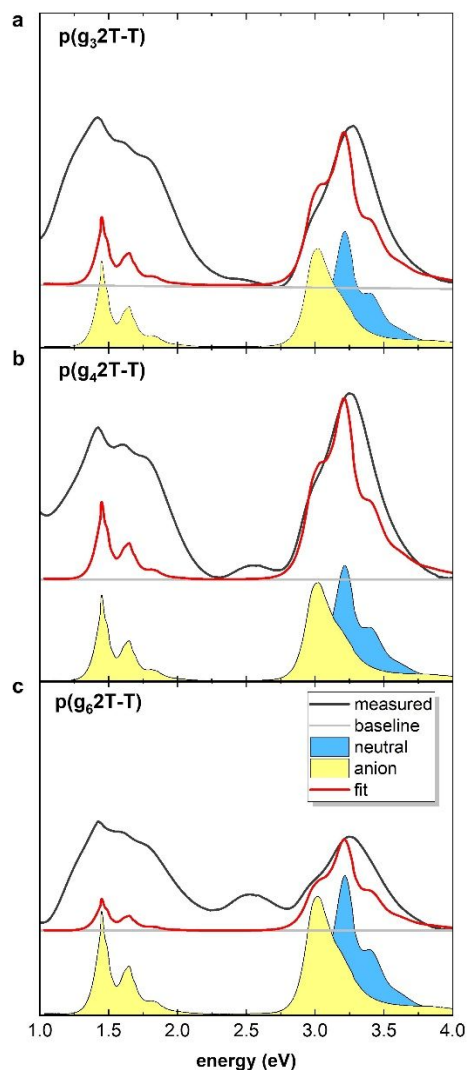


Figure S4. UV-vis spectra of p(g₃2T-T) (a), p(g₄2T-T) (b) and p(g₆2T-T) (c) films co-processed with 20 mol% F₄TCNQ (dark grey), of neutral F₄TCNQ and its anion in acetonitrile from Kiefer et al., Nature Mater. 2019, 18, 149-155 (blue and yellow filled curves) and fit of the measured spectra in the UV region composed of the spectra of neutral F₄TCNQ and its anion as well as a vertical offset (red). The experimental spectra were fitted with a superposition of a horizontal baseline (grey) and the spectra of neutral F₄TCNQ and its anion (blue and yellow). The ratio of ionized and neutral dopant molecules f was calculated by comparing the contributions of their respective UV-vis spectra to the fit.

Calculation of number of charge carriers (polarons) per unit volume

The number of charge carriers (polarons) N_p^{UVvis} per unit volume was estimated according to:

$$N_p^{UVvis} = \eta_{ion} \cdot x_d \left(\frac{M_{g_x2T-T}}{3 \cdot \rho_{p(g_x2T-T)}} (100 - x_d) + \frac{M_{F_4TCNQ}}{\rho_{F_4TCNQ}} x_d \right)^{-1} \quad (S1)$$

where x_d is the dopant molar concentration (i.e. 20 mol% F₄TCNQ), $\rho_{p(g_x2T-T)} = 1 \text{ g cm}^{-3}$ and $\rho_{F_4TCNQ} = 1.6 \text{ g cm}^{-3}$ are the assumed density of p(g_x2T-T) and F₄TCNQ, and $M_{g_x2T-T} = 570, 659$ or 835 g mol^{-1} and $M_{F_4TCNQ} = 276 \text{ g mol}^{-1}$ are the molecular weight of the g_x2T-T repeat unit and F₄TCNQ.

Table S1. Electrical properties of thin films co-processed with 20 mol% F₄TCNQ per thiophene ring: polymer, number of polarons N_p^{FTIR} per unit volume (estimated error of 30% from the thickness measurement and the applied fitting procedure) and ionization efficiency η_{ion}^{FTIR} from analysis of FTIR spectra, electrical conductivity σ (error represents the standard deviation of five measurements on the same sample), charge mobility μ .

polymer	N_p^{FTIR} (10^{26} m^{-3})	η_{ion}^{FTIR} (%)	σ (S cm^{-1})	μ ($\text{cm}^2 \text{ V}^{-1} \text{ s}^{-1}$)
p(g ₃ 2T-T)	2.2 ± 0.7	31 ± 9	830 ± 15	23.9 ± 7.2
p(g ₄ 2T-T)	4.2 ± 1.3	60 ± 18	56 ± 3	0.8 ± 0.3
p(g ₆ 2T-T)	3.3 ± 1.0	47 ± 14	51 ± 4	1.0 ± 0.3

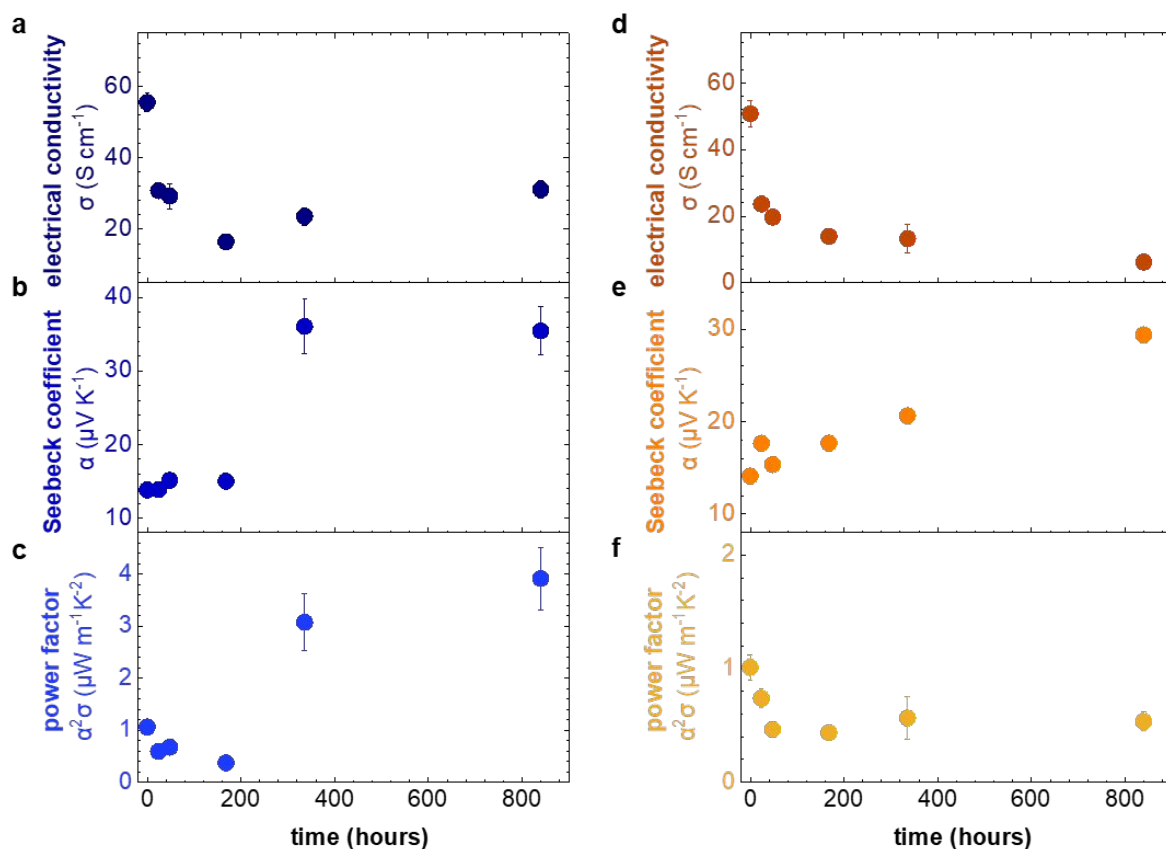


Figure S5. Electrical conductivity (a,d), Seebeck coefficient (b,e) and power factor (c,f) vs aging time of p(g₄2T-T) and p(g₆2T-T) co-processed with 20 mol% of F₄TCNQ as-cast and aged at ambient conditions.

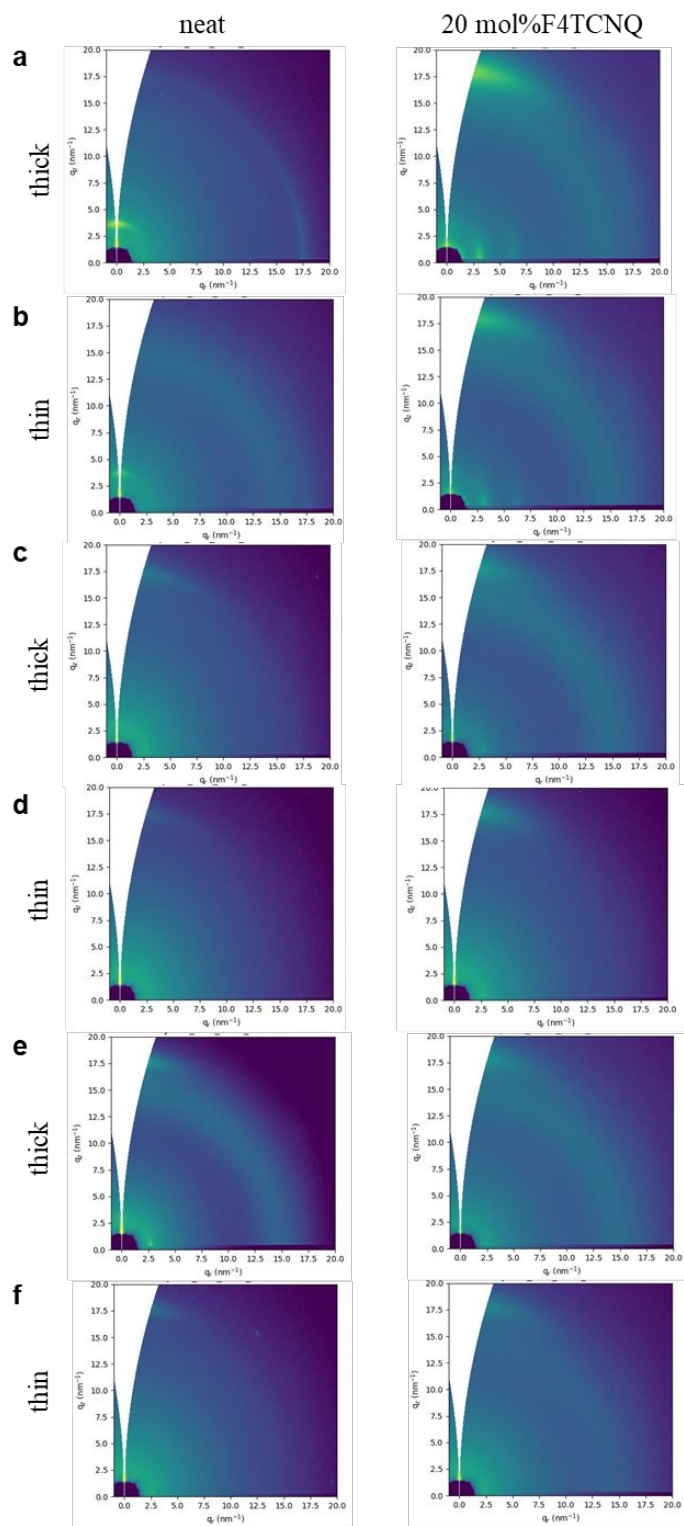


Figure S6. GIWAXS patterns of p(g₂T-T) (a,b), p(g₄2T-T) (c,d) and p(g₆2T-T) (e,f) neat and co-processed with 20 mol% F₄TCNQ.

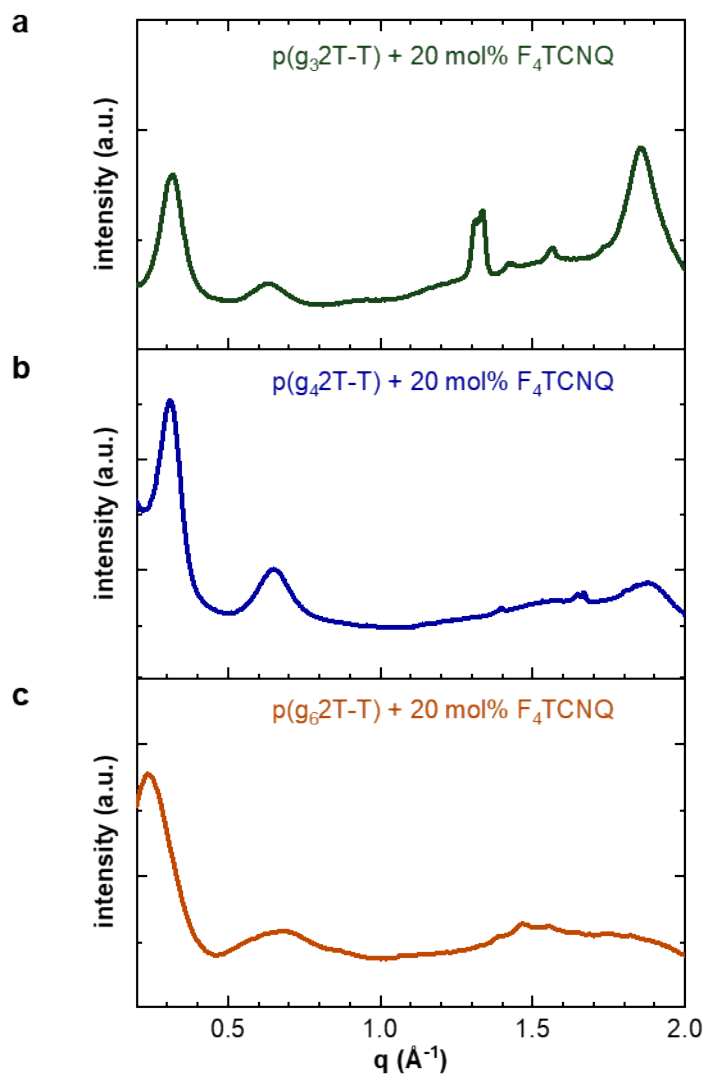


Figure S7. Transmission WAXS diffractograms of (a) p(g₃2T-T), (b) p(g₄2T-T) and (c) p(g₆2T-T) (c) co-processed with 20 mol% F₄TCNQ. The additional peaks between 1.3 and 1.6 \AA^{-1} likely arise because of the presence of excess F₄TCNQ.

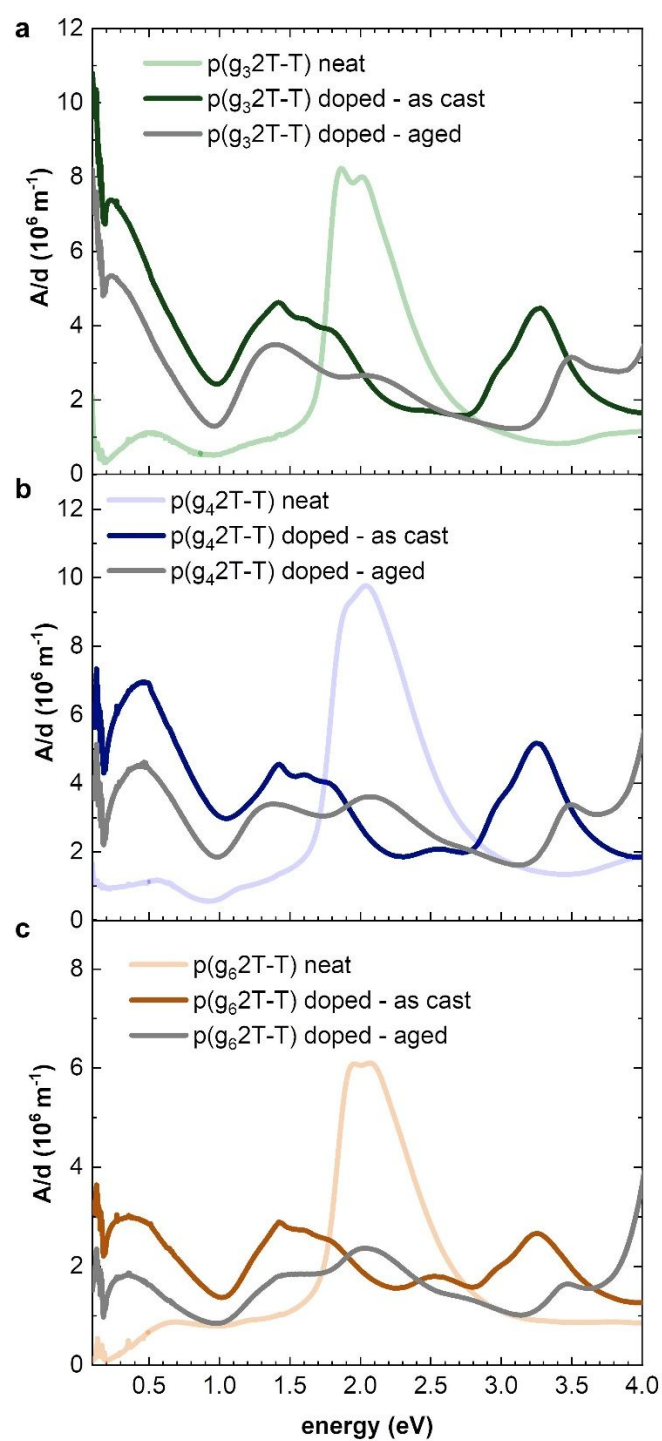


Figure S8. UV-vis and transmission FTIR absorbance spectra, with the absorbance A normalized by the film thickness d , of $p(g_32T-T)$ (a), $p(g_42T-T)$ (b) and $p(g_62T-T)$ (c) neat and doped with 20 mol% F_4TCNQ samples as-cast and aged at ambient conditions for 3 months.

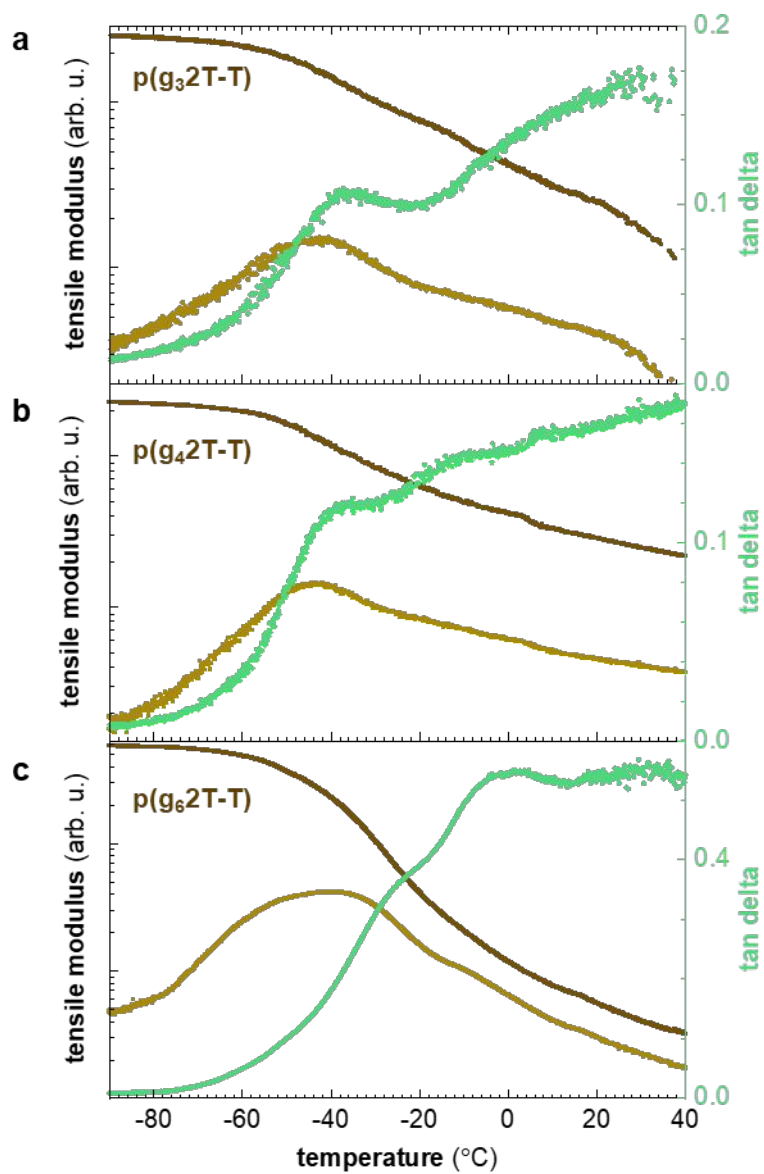


Figure S9. Tensile storage (dark brown) and loss (light brown) modulus, E' and E'' , and $\tan \delta = E''/E'$ (green) of neat $p(g_32T-T)$ (a), $p(g_42T-T)$ (b) and $p(g_62T-T)$ (c) recorded as a function of temperature on samples supported by a glass fiber mesh.

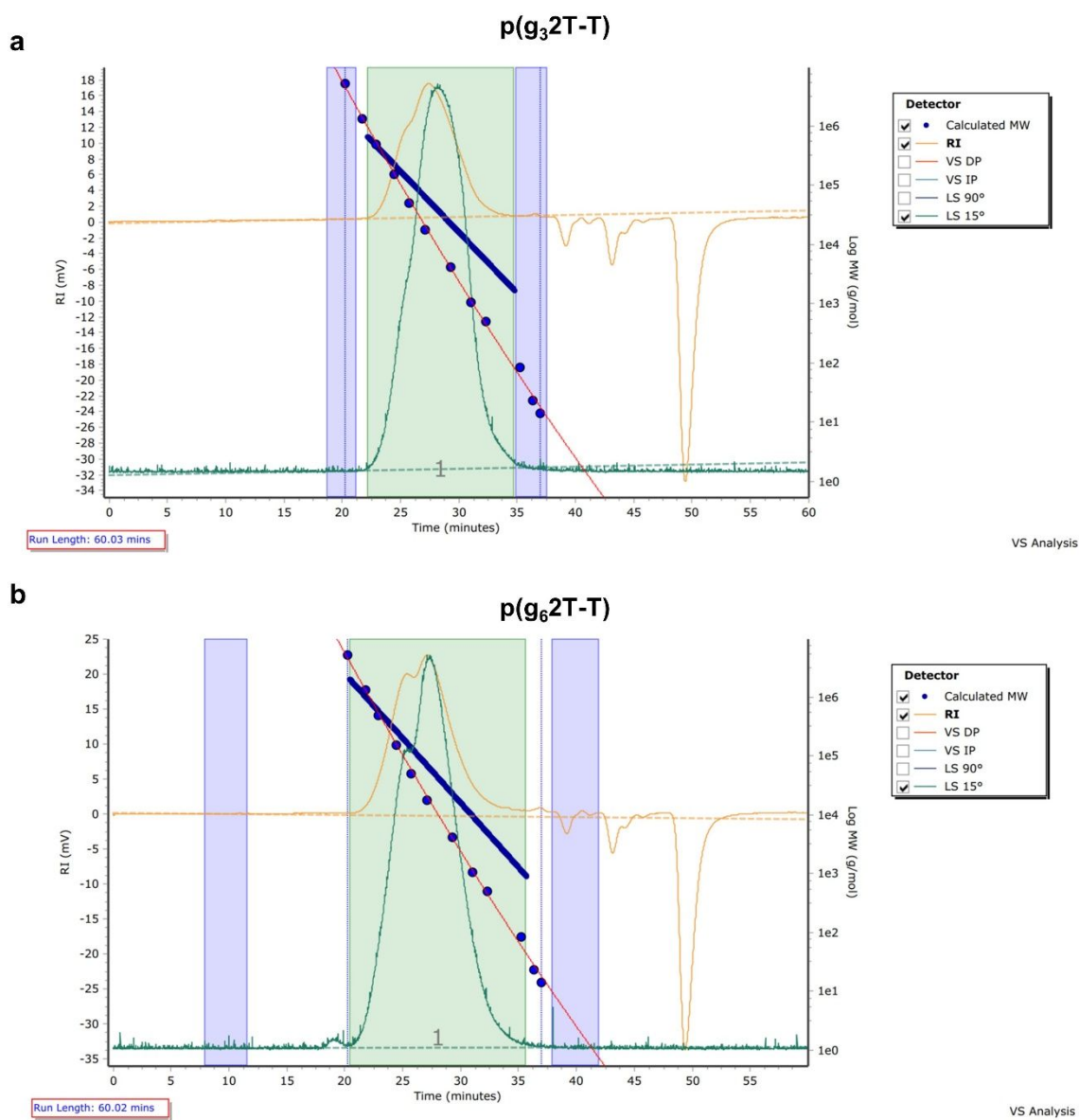


Figure S10. GPC traces of $p(g_32T-T)$ (a) and $p(g_62T-T)$ (b) at 343 K in DMF with 0.1 wt% LiBr at a polymer concentration of 1 mg mL^{-1} . Detectors showcased in traces are refractive index (orange) and light scattering (green).