

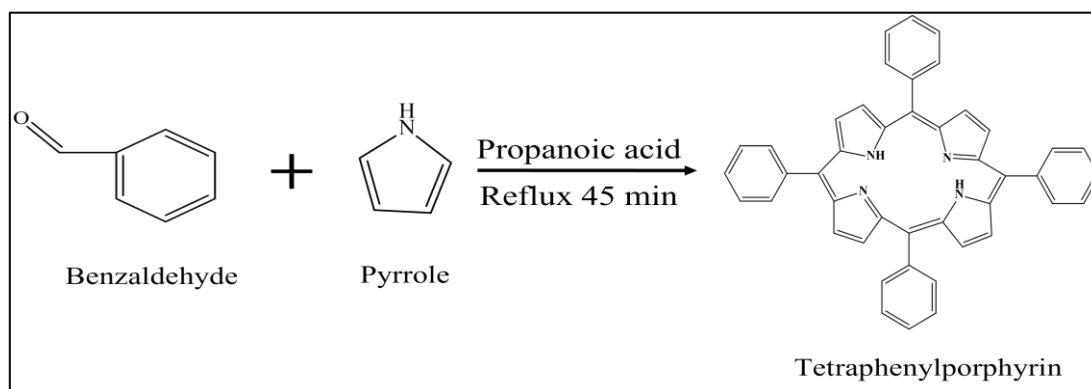
# Fabrication of Mn-TPP/RGO Tailored Glassy Carbon Electrode for Doxorubicin Sensing

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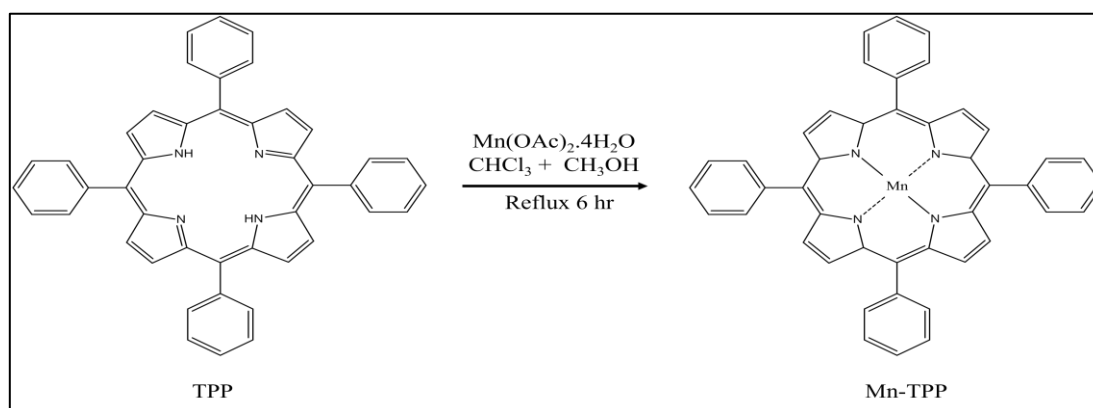
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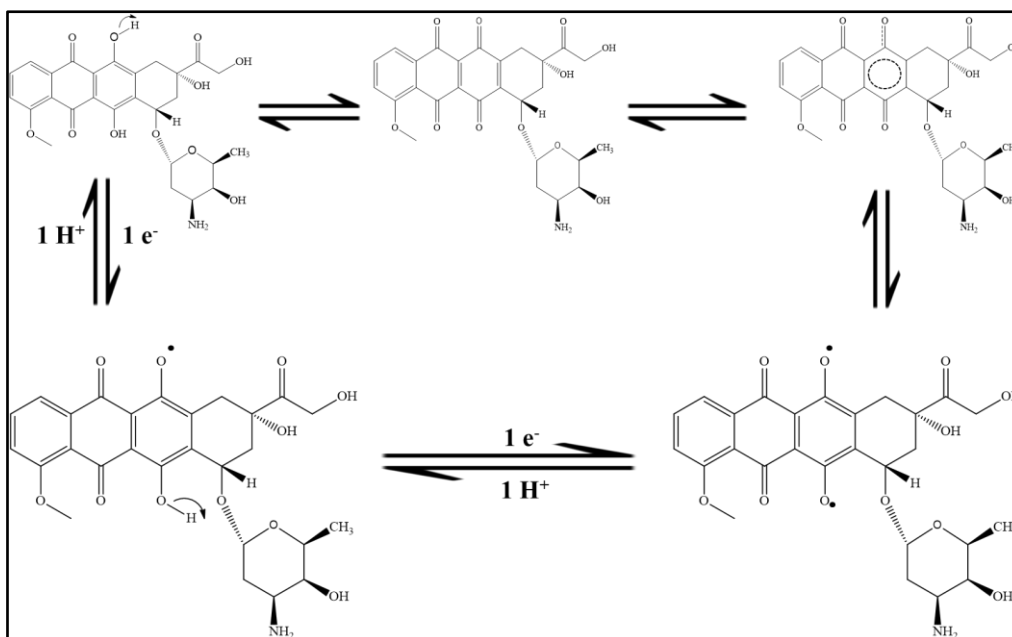
## Schemes



**Scheme S1.** Synthesis scheme of TPP.



**Scheme S2.** Synthesis scheme of Mn-TPP.



**Scheme S3.** Electrochemical oxidation of DOX.

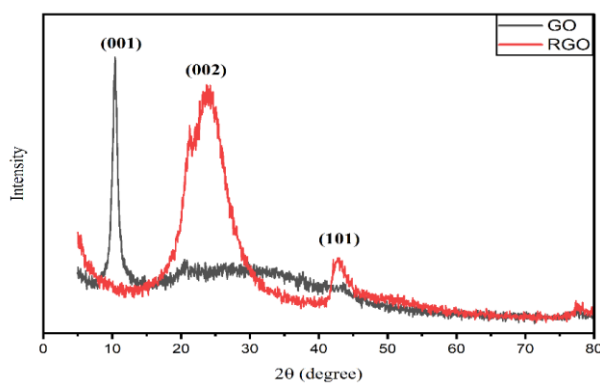
## Figures

### *Electrode Fabrication*

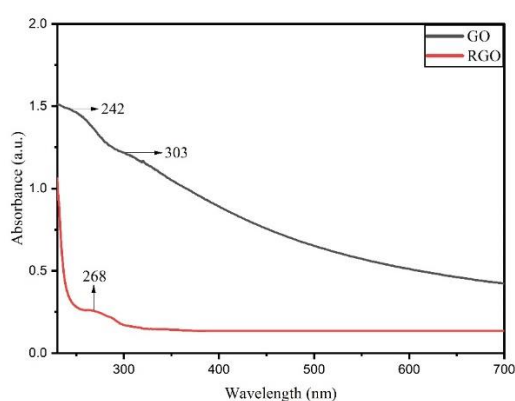
The electrode was fabricated by following the procedure mentioned in literature <sup>1</sup>. The diameter of GCE was 3mm with geometrical surface area of 0.07 cm<sup>2</sup>. Following is the image of fabricated electrode in Figure S1.



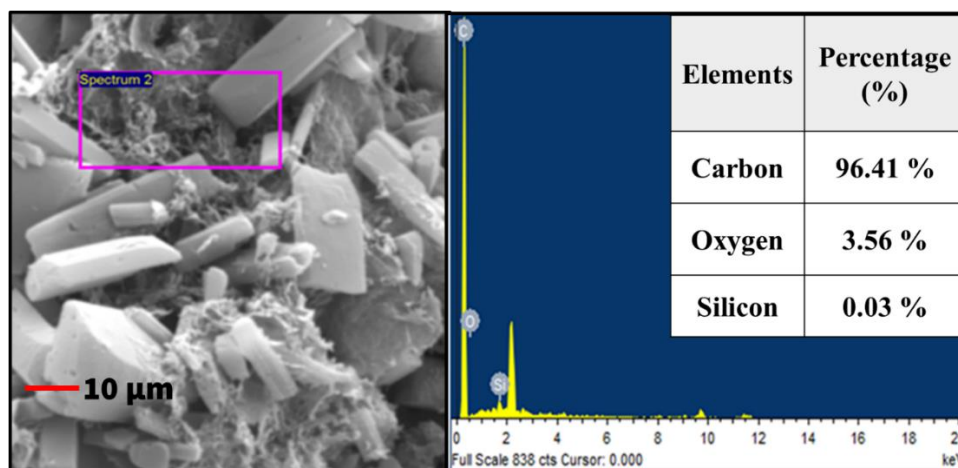
**Figure S1.** Optical image of Mn-TPP/RGO/GCE fabricated electrode with geometrical surface area of 0.07 cm<sup>2</sup>.



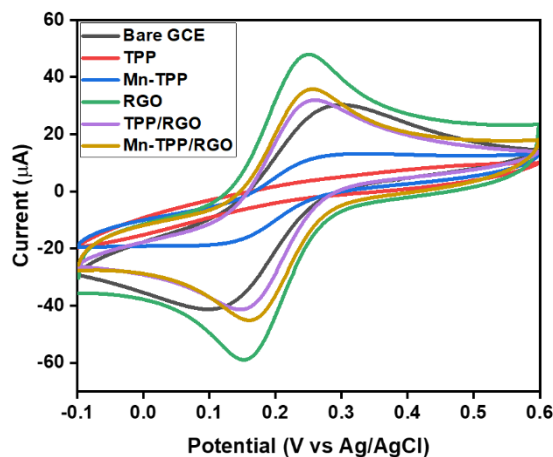
**Figure S2.** XRD Spectra of GO and RGO.



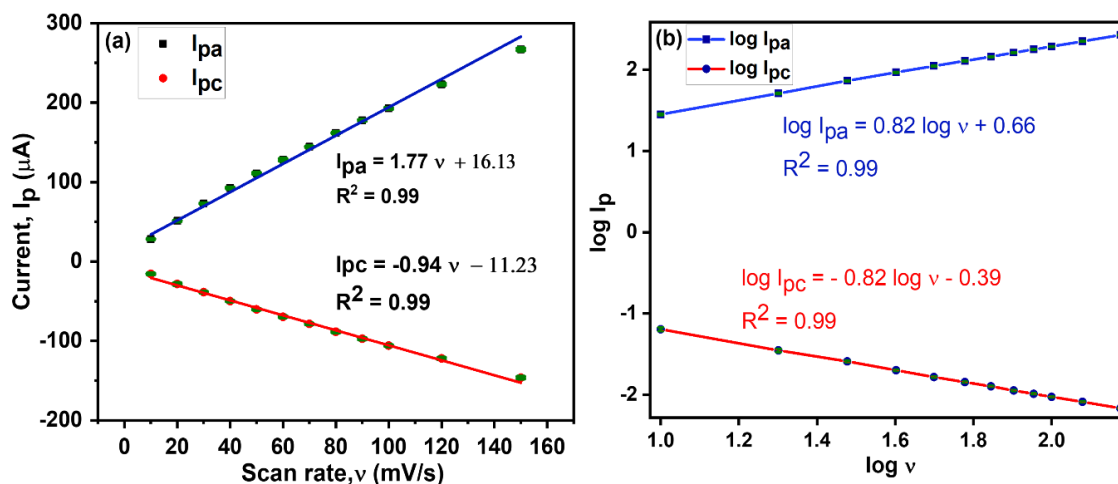
**Figure S3.** UV/vis analysis of GO and RGO.



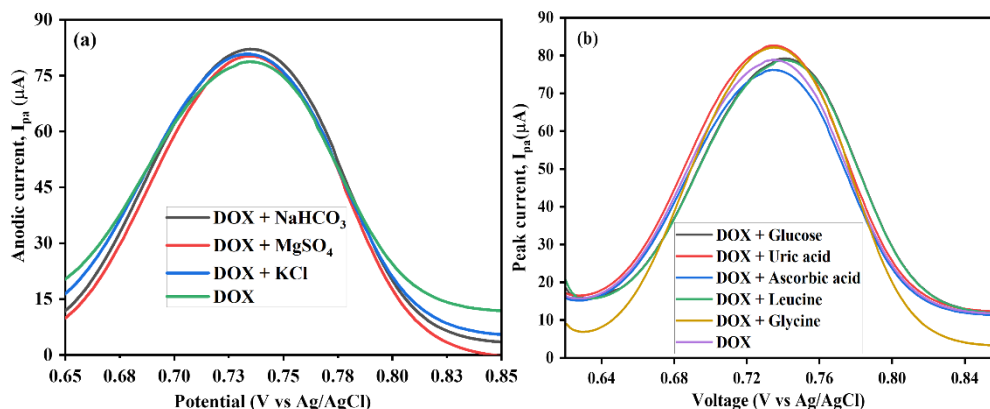
**Figure S4.** SEM and EDS analysis of TPP/RGO.



**Figure S5.** Cyclic voltammetric analysis of Bare GCE, TPP, Mn-TPP, RGO, TPP/RGO and Mn-TPP/RGO in 0.5mM potassium ferricyanide solution.



**Figure S6.** Calibration graphs of (a) scan rate vs. redox current and (b) logarithmic relationship of scan rate versus anodic peak current.



**Figure S7.** DPV studies of selectivity of Mn-TPP/RGO/GCE towards DOX in the presence (a) inorganic compounds (b) biomolecules commonly present in the blood serum.

## Tables

**Table S1.** Statistical data of EIS analysis of modified electrodes in 0.5 mM potassium ferricyanide solution.

Sr. No	Modified electrodes	Charge Transfer Resistance ( $R_{ct}$ )	Solution Resistance ( $R_u$ )	Interfacial Capacitance ( $C_{dl}$ )
1	TPP	40.70 k $\Omega$	149.7 $\Omega$	709.5 x 10 <sup>-9</sup> F
2	Bare GCE	29.43 k $\Omega$	308.8 $\Omega$	13.06 x 10 <sup>-6</sup> F
3	Mn-TPP	25.34 k $\Omega$	169.1 $\Omega$	22.59 x 10 <sup>-6</sup> F
4	RGO	3.905 k $\Omega$	137.5 $\Omega$	325.9 x 10 <sup>-6</sup> F
5	TPP/RGO	984.5 $\Omega$	149.4 $\Omega$	9.37 x 10 <sup>-6</sup> F
6	Mn-TPP/RGO	722.9 $\Omega$	212.0 $\Omega$	101.2 x 10 <sup>-6</sup> F

**Table S2.** Data of selectivity studies for Mn-TPP/RGO/GCE towards DOX.

Interfering agents	Current ( $\mu$ A)	Tolerance (%)
DOX	78.57	--
DOX + KCl	80.65	-2.7%
DOX + NaHCO <sub>3</sub>	82.09	-4.5%
DOX + MgSO <sub>4</sub>	80.11	-1.9%
DOX + Glucose	78.74	-0.21
DOX + Ascorbic acid	75.68	+3.7%
DOX + Leucine	78.91	-0.43%

<b>DOX + Glycine</b>	81.97	-4.3%
<b>DOX + Uric acid</b>	82.61	-5.1%

**Table S3.** Comparison of working ability of Mn-TPP/RGO/GCE with electrodes present in literature for the detection of DOX.

<b>Sr No.</b>	<b>Sensors</b>	<b>Method</b>	<b>LOD</b>	<b>Sensitivity</b>	<b>Linear range</b>	<b>References</b>
1.	Nano-TiO <sub>2</sub> /nafion/GCE	CV	1 nM/L	207.3 $\mu\text{A}\mu\text{M}^{-1}\text{cm}^{-2}$	5 – 2 nM/L	<sup>2</sup>
2.	RGO/Au nanoparticles/Ppy/GCE	CV, EIS	0.02 $\mu\text{M}$	185 $\mu\text{A}\text{mM}^{-1}\text{cm}^{-2}$	0.02 $\mu\text{M}$ – 25 mM	<sup>3</sup>
3.	FeV/SCNF	LSV	5 nM	46.04 $\mu\text{A}\text{mM}^{-1}\text{cm}^{-2}$	20 nM – 542.5 $\mu\text{M}$	<sup>4</sup>
4.	VMSF/p-GCE	CV, DPV	0.2 nM	23.94 $\mu\text{A}\text{mM}^{-1}\text{cm}^{-2}$	0.5 nM to 23 $\mu\text{M}$	<sup>5</sup>
<b>5.</b>	<b>Mn-TPP/RGO/GCE</b>	<b>CV</b>	<b>63.5 <math>\mu\text{M}</math></b>	<b>112.09 <math>\mu\text{A}\text{mM}^{-1}\text{cm}^{-2}</math></b>	<b>0.1-0.6 mM</b>	<b>This work</b>
<b>6.</b>	<b>Mn-TPP/RGO/GCE</b>	<b>DPV</b>	<b>27.0 <math>\mu\text{M}</math></b>	<b>0.174 <math>\mu\text{A}\mu\text{M}^{-1}\text{cm}^{-2}</math></b>	<b>0.1-0.6 mM</b>	<b>This work</b>

## References

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