Supporting Information

Boron complexes with propiolamidinato ligands: synthesis, structure and photophysical properties

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1.- General Methods and Materials

All manipulations were performed under a nitrogen atmosphere using standard Schlenk and glove-box techniques. The solvents used for the syntheses and for the NMR experiments were all anhydrous. All reagents were obtained from commercial suppliers (Sigma Aldrich) and used without further purification, with the exception of N,N'-di-*p*-tolylcarbodiimide, which was synthesized as described in the literature. [1]

NMR spectra were recorded on Bruker Avance Neo 400 and 500 spectrometers at 298 K, using standard TOPSPIN 4.0 software. Chemical shifts (δ) are given in ppm and coupling constants (*J*) in Hz. The chemical shifts were assigned based on homo- and heteronuclear 2D NMR experiments (COSY, HSQC, HMBC). Characterisation details, including ¹H, ¹³C{¹H}, ¹⁹F{¹H} (when necessary) and ¹¹B{¹H} (when necessary) NMR spectra for compounds **3a-e**, **5a-e**, **4a-e** and **6a-e** are included in the following sections of this Supporting Information. Melting points were measured with a Büchi M-56 apparatus and are uncorrected. Elemental analysis data were recorded on a Foss-Heraeus CHNO-Rapid analyzer. Infrared spectra were recorded on a Bruker Tensor 27 spectrometer, using an ATR accessory. X-Ray diffraction studies were performanced on a Bruker APEX II CCD-based diffractometer equipped with a graphite monochromated MoK α radiation source (λ =0.71073 Å).

1.1.- Photophysical properties.

UV-visible absorption spectra were acquired on a Cary 100 (Varian) spectrophotometer at room temperature, using a slit width of 0.4 nm and a scan rate of 600 nm min-1. Steady state fluorescence (SSF) and time resolved fluorescence (TRF) spectra were recorded on an FLS920 spectrofluorometer (Edinburgh Instruments) equipped with a Microchannel plate-photomultiplier tube (MCP-PMT) detector (R3809) and a time correlated single photon counting (TCSPC) data acquisition card (TCC900model). Ten-millimetre quartz hermetic cuvettes (Hellma Analytics) were employed for all spectroscopic measurements. For SSF spectra, a 450 W Xe lamp was

used, and the light source and the excitation and emission slits were fixed at different size to 1-10 nm. For TRF experiments, an EPLED 291 and 368 sub-nanosecond pulsed light emitting diode (Edinburgh Photonics) was employed as light source.

The fluorescence intensity decays, I(t), were fitted by using an iterative least-squares fit method with the following multiexponential function (Eq. S1):

Equation S1
$$T(t) = \sum_{i=1}^{n} \alpha_i \exp\left(-\frac{t}{\tau_i}\right)$$

where α i and τ i are the amplitude and lifetime for each ith term. The average decay lifetime (τ m) was then calculated using Equation S2:

Equation S2
$$\tau_m = \frac{\sum_{i=1}^n \alpha_i \tau_i^2}{\sum_{i=1}^n \alpha_i \tau_i}$$

A sample concentration of 10 μ M was generally employed for UV–vis absorption and fluorescence spectroscopy experiments.

Quantum yields were measured in a FS5 spectrometer (Edinburgh Instruments) equipped with an integrating sphere, a 150W Xe lamp as the light source and a photomultiplier tube (PMT) detector (R928P model). Quantum calculations were carried out using the F980 Software of Edinburgh Instruments.



Figure S1. Lifetime decay profiles of compounds **6a-e** in acetonitrile (A) and Dichlorometane (B). Lifetime decay profiles of compounds **4a-d-e** in acetonitrile (C) and compounds **4a-b-d-e** in dichlorometane (D).



Figure S2. (A) Samples of the compounds 3a, 3d, 3b, 3c, 3e, 5a, 5d, 5b, 5c, 5e (from left to right) in CH_2Cl_2 , under UV light (top, handheld UV lamp, 365 nm) and under natural light (bottom). (B) Emission spectra of compounds 3d (left) and 3e (right) in $CH_2Cl_2(10 \ \mu\text{M})$.

1.2.- DFT calculations

Density functional theory (DFT) calculations were carried out to optimize the molecular structure of compounds. m062x as implemented in Gaussian16 (revision C.01), [2] along with the 6-31G* basis sets were used for the molecular structure optimization.

The electronic vertical transitions were calculated at the time-dependent (TD)-m062x/6-31g*.

2.- Experimental procedure and characterisation details

2.1.- General procedure for the obtaining of propiolamidines by catalytic methods, compounds 3a-e.

Syntheses of compounds **3a-e** were performed under N₂ atmosphere. In a schlenk, 1 equivalent of the appropriate alkyne (**1a-e**, $3 \cdot 10^{-3}$ mol) was dissolved in 5 mL of toluene, followed by the addition of 3 mol% ZnEt₂ (1 M in hexanes) ($9 \cdot 10^{-5}$ mol). Subsequently, 1 equivalent of N,N²-diisopropylcarbodiimide (**2a**, $3 \cdot 10^{-3}$ mol) was added and the solution was stirred for 4h at 120°C (6h in the case of **3e** due to its lack of solubility). The solvent was eliminated in vacuo. Then, 5 mL of pentane were added, and the mixture was placed in a refrigerator at -20 °C for 16 h, affording colorless powder which was filtrated and dried under reduced pressure. Compounds **3a-e** were obtained as yellowish microcrystalline solids with excellent yields (91–98%).

2.2.- General procedure for the obtaining of propiolamidines by stoichiometric methods, compounds 5a-e.

Syntheses of compounds **5a-e** were performed under air and at room temperature. In a Schlenk flask, 1 equivalent of the appropriate alkyne (**1a-e**, $3 \cdot 10^{-3}$ mol) was dissolved in 5 mL of 2-MeTHF, followed by the addition (under stirring) of 1 equivalent of nBuLi (1.6 M in hexanes) ($3 \cdot 10^{-3}$ mol) (addition time = 5 seconds) which gave rise to the lithium amidinate intermediate. Subsequently, 1 equivalent of N,N'-di-*p*-tolylcarbodiimide (**2b**, $3 \cdot 10^{-3}$ mol) was added and the solution was stirred for 60 seconds before quenching with 5 mL of a saturated solution of NH₄Cl. The mixture was extracted with 2-MeTHF (3×5 mL). The combined organic phases were dried over anhydrous MgSO₄ and the solvent was concentrated in vacuo. Compounds **5a-e** were obtained as yellow or orangish microcrystalline solids with good to excellent yields (84-99%).

2.3.- General procedure for the obtaining of boron complexes, compounds 4a-e and 6a-e.

Syntheses of compounds **4a-e** and **6a-e** were performed under N_2 atmosphere. In a Schlenk flask, 1 equivalent of the appropriate amidine (**3a-e** or **5a-e**, $1 \cdot 10^{-3}$ mol) was

dissolved in 5 mL of toluene, followed by the addition of 1 equivalent of BPh₃ ($1 \cdot 10^{-3}$ mol). The solution was stirred for 4h at 120°C (series **4a-e**) and for 5h at 120°C (series **6a-e**). The solvent was eliminated in vacuo. Then, 5 mL of pentane were added, and the mixture was placed in a refrigerator at -20 °C for 16 h, affording crystalline solid which was filtrated and dried under reduced pressure. Compounds **4a-e** were obtained as yellowish microcrystalline solids with fantastic yields (93–98%) while compounds **6a-e** were obtained as orange to reddish microcrystalline solids with excellent yields (90–96%).

2.4.- Characterisation details



3a: FT-IR (cm⁻¹) = NH 3382, C=C 2227, C=N 1568. ¹H NMR (C₆D₆) δ (ppm) = 0.99 (d, *J* = 6.3 Hz, 6H, CH₃), 1.39 (d, *J* = 5.3 Hz, 6H, CH₃), 3.81 (bs, 1H, NH), 4.21-4.27 (m, 2H, CH), 6.96-6.98 (m, 3Harom), 7.38-7.40 (m, 2Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 22.7 (2C), 25.5 (2C), 42.7,

53.2, 80.9, 90.4, 122.2, 128.8 (2C), 129.4, 132.2 (2C), 140.1. Elem. anal. calc for [C₁₅H₂₀N₂]: C, 78.90; H, 8.83; N, 12.27. Found: C, 78.83; H, 8.92; N, 12.35.



3b: FT-IR (cm⁻¹) = NH 3390, C=C 2223, C=N 1565. ¹H NMR (C₆D₆) δ (ppm) = 1.04 (bs, 6H, CH₃), 1.41 (bs, 6H, CH₃), 3.17 (s, 3H, CH₃O), 3.81 (bs, 1H, NH), 4.26-4.34 (m, 2H, CH), 6.57-6.59 (m, 2Harom), 7.36-7.38 (m, 2Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 24.0 (2C), 25.3 (2C), 49.2, 53.1, 54.9, 79.9, 90.8,

114.1, 114.6 (2C), 133.8 (2C), 140.6, 160.8. Elem. anal. calc for [C₁₆H₂₂N₂O]: C, 74.38; H, 8.58; N, 10.84. Found: C, 74.44; H, 8.68; N, 10.92.



3c: FT-IR (cm⁻¹) = NH 3388, C=C 2220, C=N 1562. ¹H NMR (C₆D₆) δ (ppm) = 0.99 (d, *J* = 6.4 Hz, 6H, CH₃), 1.39 (d, *J* = 6.2 Hz, 6H, CH₃), 3.77 (bs, 1H, NH), 4.17-4.26 (m, 2H, CH), 6.56-6.60 (m, 2Harom), 7.10-7.15 (m, 2Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 22.7 (2C), 25.5 (2C), 42.7, 53.2, 80.6, 89.2, 115.9, 116.2, 118.1, 134.2, 139.9, 162.1, 164.6. ¹⁹F{¹H} NMR (C_6D_6) δ (ppm) = -109.14. Elem. anal. calc for [$C_{15}H_{19}FN_2$]: C, 73.14; H, 7.77; N, 11.37. Found: C, 73.24; H, 7.90; N, 11.42.



3d: FT-IR (cm⁻¹) = NH 3391, C=C 2223, C=N 1563. ¹H NMR (C₆D₆) δ (ppm) = 1.02 (d, *J* = 6.4 Hz, 6H, CH₃), 1.46 (d, *J* = 6.2 Hz, 6H, CH₃), 2.34 (s, 6H, N(CH₃)₂), 3.83 (bs, 1H, NH), 4.28-4.45 (m, 2H, CH), 6.32-6.35 (m, 2Harom), 7.49-7.52 (m, 2Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 22.8 (2C), 25.6 (2C), 39.6 (2C), 42.6,

53.1, 79.5, 92.3, 108.7, 112.1 (2C), 133.5 (2C), 141.1, 151.0. Elem. anal. calc for [C₁₇H₂₅N₃]: C, 75.23; H, 9.28; N, 15.48. Found: C, 75.35; H, 9.33; N, 15.52.



3e: FT-IR (cm⁻¹) = NH 3378, C=C 2215, C=N 1565. ¹H NMR (C₆D₆) δ (ppm) = 1.06 (d, *J* = 6.5 Hz, 6H, CH₃), 1.49 (d, *J* = 6.1 Hz, 6H, CH₃), 3.91 (bs, 1H, NH), 4.33-4.37 (m, 1H, CH), 4.44-4.49 (m, 21H, CH), 7.27-7.30 (m, 1Harom), 7.34-7.41 (m, 2Harom), 7.44-7.51 (m, 2Harom), 7.96 (s, 1Harom), 8.33 (d, *J* = 8.3 Hz,

1Harom), 8.39 (d, J = 8.2 Hz, 1Harom), 8.65 (d, J = 8.1 Hz, 1Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 24.2 (4C), 48.2 (2C), 85.1, 88.9, 118.6, 123.0, 123.4, 127.1, 127.3, 127.6, 127.7, 128.2, 129.1, 130.6, 131.1, 131.3, 131.4, 133.6, 140.4. Elem. anal. calc for [C₂₃H₂₄N₂]: C, 84.11; H, 7.37; N, 8.53. Found: C, 84.01; H, 7.25; N, 8.61.



5a: FT-IR (cm⁻¹) = NH 3378, C=C 2220, C=N 1570. ¹H NMR (C₆D₆) δ (ppm) = 2.13 (s, 6H, CH₃), 6.85-6.89 (m, 3Harom), 7.01-7.03 (m, 4Harom), 7.20-7.22 (m, 2Harom), 7.37-7.39 (m, 4Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 20.9 (2C), 81.7, 92.2, 121.5, 121.7 (4C), 128.6 (2C), 129.5 (4C), 129.7 (2C), 132.3 (2C), 132.5 (2C), 139.5, 143.5. Elem. anal. calc for [C₂₃H₂₀N₂]: C, 85.15; H, 6.21; N, 8.63. Found: C, 85.10; H, 6.17; N, 8.72.



5b: FT-IR (cm⁻¹) = NH 3401, C=C 2215, C=N 1560. ¹H NMR (C₆D₆) δ (ppm) = 2.15 (s, 6H, CH₃), 3.11 (s, 3H, CH₃O), 6.46 (d, *J* = 8.9 Hz, 2Harom), 7.04 (d, *J* = 8.2 Hz, 4Harom), 7.18 (d, *J* = 8.9 Hz, 2Harom), 7.42 (m, *J* = 8.1 Hz, 4Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 20.9 (2C), 54.7, 80.9, 92.9, 113.4, 114.5 (2C), 121.8 (4C), 129.5 (4C), 132.4 (2C), 134.1 (2C), 139.9, 161.1. Elem. anal. calc for [C₂₄H₂₂N₂O]: C, 81.33; H,

6.26; N, 7.90. Found: C, 81.44; H, 6.32; N, 7.96.



5c: FT-IR (cm⁻¹) = NH 3399, C=C 2223, C=N 1566. ¹H NMR (C₆D₆) δ (ppm) = 2.14 (s, 6H, CH₃), 6.44-6.48 (m, 2Harom), 6.93-6.97 (m, 2Harom), 7.02 (d, *J* = 8.1 Hz, 4Harom), 7.36 (d, *J* = 8.3 Hz, 4Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 20.9 (2C), 81.4, 91.4, 115.8 (2C), 116.1 (2C), 117.4, 121.8 (2C), 129.5 (4C), 132.7 (2C), 134.5 (2), 139.6, 143.3, 162.2, 164.7. ¹⁹F{¹H} NMR (C₆D₆) δ (ppm) = -108.24. Elem. anal. calc for [C₂₃H₁₉FN₂]: C,

80.68; H, 5.59; N, 8.18. Found: C, 80.73; H, 5.68; N, 8.29.



5d: FT-IR (cm⁻¹) = NH 3378, C=C 2220, C=N 1570. ¹H NMR (C₆D₆) δ (ppm) = 2.16 (s, 6H, CH₃), 2.29 (s, 6H, N(CH₃)₂), 6.20 (d, *J* = 8.9 Hz, 2Harom), 7.05 (d, *J* = 8.1 Hz, 4Harom), 7.29 (d, *J* = 8.7 Hz, 2Harom), 7.45 (d, *J* = 8.1 Hz, 4Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 20.9 (2C), 39.4 (2C), 80.7, 94.9, 107.7 (2C), 111.9 (2C), 121.8 (4C), 129.4 (4C), 132.1 (2C), 133.8 (2C), 140.4, 143.9, 151.0. Elem. anal. calc for [C₂₅H₂₅N₃]: C, 81.71; H, 6.86;

N, 11.43. Found: C, 81.79; H, 6.93; N, 11.50.



5e: FT-IR (cm⁻¹) = NH 3384, C=C 2218, C=N 1571. ¹H NMR (C₆D₆) δ (ppm) = 2.18 (s, 6H, CH₃), 7.07 (d, *J* = 8.0 Hz, 4Harom), 7.23-7.26 (m, 1Harom), 7.31-7.34 (m, 1Harom), 7.39-7.46 (m, 7Harom), 7.78 (s, 1Harom), 7.87-7.88 (m, 1Harom), 8.28-8.29 (m, 1Harom), 8.31-8.33 (m, 1Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 20.9 (2C), 85.8, 90.8, 118.2, 121.6 (2C), 123.0 (2C), 127.2, 127.4, 127.5, 127.6, 127.9, 128.2, 129.1, 129.8 (4C), 130.4, 131.2, 131.3, 132.6, 133.9, 139.8. Elem. anal. calc for

[C₃₁H₂₄N₂]: C, 87.70; H, 5.70; N, 6.60. Found: C, 87.73; H, 5.74; N, 6.69.



4a: FT-IR (cm⁻¹) = C=C 2222, C=N 1638. ¹H NMR (C₆D₆) δ (ppm) = 1.09 (d, *J* = 6.6 Hz, 12H, CH₃), 3.85-3.94 (m, 2H, CH), 6.87-6.98 (m, 3Harom), 7.27-7.41 (m, 8Harom), 7.85-7.87 (m, 4Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 23.5 (4C), 47.9 (2C), 76.1, 97.6, 120.4, 127.0 (2C), 127.7 (4C), 128.9 (2C), 130.6, 132.4 (2C), 134.4 (4C), 149.9. ¹¹B{¹H}

NMR (C₆D₆) δ (ppm) = 9.73. Elem. anal. calc for [C₂₇H₂₉BN₂]: C, 82.65; H, 7.45; N, 7.14. Found: C, 82.71; H, 7.49; N, 7.25. m.p. 66–69 °C.



4b: FT-IR (cm⁻¹) = C=C 2219, C=N 1637. ¹H NMR (C₆D₆) δ (ppm) = 1.11 (d, *J* = 6.6 Hz, 12H, CH₃), 3.13 (s, 3H, CH₃O), 3.91-3.98 (m, 2H, CH), 6.52-6.54 (m, 2Harom), 7.27-7.34 (m, 4Harom), 7.38-7.42 (m, 4Harom), 7.87-7.90 (m, 4Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 23.5 (4C), 47.8 (2C), 54.9, 75.5, 98.4, 112.2, 114.7 (2C), 127.0

(2C), 127.7 (4C), 134.3 (2C), 134.4 (4C), 138.9, 150.4, 161.8. ¹¹B{¹H} NMR (C₆D₆) δ (ppm) = 9.49. Elem. anal. calc for [C₂₈H₃₁BN₂O]: C, 79.62; H, 7.40; N, 6.63. Found: C, 79.71; H, 7.49; N, 6.74. m.p. 68–71 °C.



4c: FT-IR (cm⁻¹) = C=C 2220, C=N 1637. ¹H NMR (C₆D₆) δ (ppm) = 1.09 (d, *J* = 6.7 Hz, 12H, CH₃), 3.83-3.93 (m, 2H, CH), 6.50-6.54 (m, 2Harom), 6.99-7.03 (m, 2Harom), 7.29-7.42 (m, 6Harom), 7.84-7.87 (m, 4Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 23.5 (4C), 47.9 (2C), 75.8, 96.5, 116.2, 116.4, 127.1 (2C), 134.4 (4C), 134.6,

134.7, 149.7, 162.7, 165.3. ${}^{19}F{}^{1}H$ NMR (C₆D₆) δ (ppm) = -106.53. ${}^{11}B{}^{1}H$ NMR (C₆D₆) δ (ppm) = 9.88. Elem. anal. calc for [C₂₇H₂₈BFN₂]: C, 79.03; H, 6.88; N, 6.83. Found: C, 79.11; H, 6.91; N, 6.90. m.p. 70–73 °C.



4d: FT-IR (cm⁻¹) = C=C 2221, C=N 1638. ¹H NMR (C₆D₆) δ (ppm) = 1.14 (d, *J* = 6.6 Hz, 12H, CH₃), 2.30 (s, 6H, N(CH₃)₂), 3.96-4.03 (m, 2H, CH), 6.25-6.28 (m, 2Harom), 7.29-7.44 (m, 9Harom), 7.90-7.92 (m, 4Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 23.6 (4C), 39.3 (2C), 47.8 (2C), 75.5, 100.8, 106.4, 111.9 (2C), 126.9

(2C), 127.7 (4C), 134.1 (2C), 134.5 (4C), 151.2, 151.6. ¹¹B{¹H} NMR (C₆D₆) δ (ppm) = 9.39. Elem. anal. calc for [C₂₉H₃₄BN₃]: C, 80.00; H, 7.87; N, 9.65. Found: C, 80.08; H, 7.81; N, 9.75. m.p. 65–70 °C.



4e: FT-IR (cm⁻¹) = C=C 2219, C=N 1637. ¹H NMR (C₆D₆) δ (ppm) = 1.18 (d, *J* = 6.6 Hz, 12H, CH₃), 4.03-4.11 (m, 2H, CH), 7.26-7.29 (m, 2Harom), 7.32-7.39 (m, 4Harom), 7.41-7.46 (m, 6Harom), 7.93 (d, *J* = 8.1 Hz, 4Harom), 8.29 (d, *J* = 8.2 Hz, 1Harom), 8.35 (d, *J* = 8.1 Hz, 1Harom), 8.51-8.53 (m, 1Harom). ¹³C{¹H} NMR (C₆D₆) δ

 $(ppm) = 23.7 (4C), 48.1 (2C), 80.0, 96.5, 116.9, 123.0, 123.1, 123.5, 126.5, 127.1, 127.5, 127.7, 127.9, 128.9, 129.3, 130.6, 130.8, 130.9, 131.5, 131.6, 132.4, 132.5, 134.5, 135.1, 136.2, 146.0, 150.0. {}^{11}B{}^{1}H{} NMR (C_6D_6) \delta (ppm) = 10.01. Elem. anal. calc for$

[C₃₅H₃₃BN₂.C₅H₁₂]: C, 85.09; H, 8.03; N, 4.96. Found: C, 85.00; H, 7.96; N, 5.01. m.p. 72–75 °C.



6a: FT-IR (cm⁻¹) = C=C 2220, C=N 1639. ¹H NMR (C₆D₆) δ (ppm) = 2.00 (s, 6H, CH₃), 6.81-6.83 (m, 4Harom), 6.86- 6.95 (m, 3Harom), 7.22-7.26 (m, 2Harom), 7.29-7.35 (m, 6Harom), 7.52-7.54 (m, 4Harom), 7.94-7.96 (m, 4Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 20.9 (2C), 78.6, 99.8, 119.6 (4C), 119.9, 127.6 (2C), 128.2 (4C), 128.9 (2C), 130.1 (4C), 131.1 (2C), 132.7 (2C), 133.8 (2C), 134.2 (4C), 136.3, 137.9 (2C), 141.2. ¹¹B{¹H} NMR (C₆D₆) δ (ppm) = 9.60. Elem. anal.

calc for [C₃₅H₂₉BN₂]: C, 86.07; H, 5.98; N, 5.74. Found: C, 86.14; H, 6.05; N, 5.82. m.p. 65–68 °C.



6b: FT-IR (cm⁻¹) = C=C 2221, C=N 1639. ¹H NMR (C₆D₆) δ (ppm) = 2.02 (s, 6H, CH₃), 3.11 (s, 3H, CH₃O), 6.50-6.52 (m, 2Harom), 6.84- 6.86 (m, 4Harom), 7.23-7.27 (m, 2Harom), 7.29-7.35 (m, 6Harom), 7.55-7.58 (m, 4Harom), 7.96-7.98 (m, 4Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 20.9 (2C), 54.9, 78.2, 100.9, 111.7, 114.8 (2C), 119.6 (4C), 127.6 (2C), 127.9 (4C), 128.2 (2C), 130.0 (4C), 133.6 (2C), 134.2 (4C), 134.7 (2C), 138.1 (2C), 141.7, 162.2. ¹¹B{¹H} NMR (C₆D₆) δ (ppm)

= 9.88. Elem. anal. calc for [C₃₆H₃₁BN₂O]: C, 83.40; H, 6.03; N, 5.40. Found: C, 83.52; H, 6.11; N, 5.45. m.p. 69–73 °C.



6c: FT-IR (cm⁻¹) = C=C 2219, C=N 1640. ¹H NMR (C₆D₆) δ (ppm) = 2.02 (s, 6H, CH₃), 6.49-6.53 (m, 2Harom), 6.83- 6.85 (m, 4Harom), 7.02-7.05 (m, 2Harom), 7.24-7.26 (m, 2Harom), 7.32-7.36 (m, 4Harom), 7.48-7.50 (m, 4Harom), 7.94-7.96 (m, 4Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 20.9 (2C), 78.3, 98.6, 116.3 (2C), 116.5 (2C), 119.6 (4C), 127.7 (4C), 129.3 (2C), 130.1 (4C), 133.9 (2C), 134.1 (4C), 134.9 (2C), 137.8 (2C), 141.1, 163.3, 165.3. ¹⁹F{¹H}

NMR (C_6D_6) δ (ppm) = -105.36. ¹¹B{¹H} NMR (C_6D_6) δ (ppm) = 9.96. Elem. anal. calc for [$C_{35}H_{28}BFN_2$]: C, 83.01; H, 5.57; N, 5.53. Found: C, 82.91; H, 5.45; N, 5.59. m.p. 69–73 °C.



6d: FT-IR (cm⁻¹) = C=C 2223, C=N 1641. ¹H NMR (C₆D₆) δ (ppm) = 2.03 (s, 6H, CH₃), 2.27 (s, 6H, N(CH₃)₂), 6.21-6.24 (m, 2Harom), 6.86- 6.88 (m, 4Harom), 7.23-7.27 (m, 2Harom), 7.32-7.35 (m, 4Harom), 7.45-7.47 (m, 2Harom), 7.63-7.65 (m, 4Harom), 7.98-8.01 (m, 4Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 20.9 (2C), 39.2 (2C), 78.8, 86.2, 103.9, 105.8 (2C), 111.9 (2C), 119.5 (4C), 127.4 (2C), 128.6 (4C), 129.9 (4C), 133.2 (2C), 134.3 (4C), 134.5

(2C), 138.4 (2C), 142.3, 151.9. ¹¹B{¹H} NMR (C₆D₆) δ (ppm) = 10.49. Elem. anal. calc for [C₃₇H₃₄N₃.CH₂Cl₂]: C, 74.04; H, 5.89; N, 6.82. Found: C, 73.90; H, 5.79; N, 6.77. m.p. 68–70 °C.



6e: FT-IR (cm⁻¹) = C=C 2221, C=N 1641. ¹H NMR (C₆D₆) δ (ppm) = 2.03 (s, 6H, CH₃), 6.87-6.89 (m, 5Harom), 7.26-7.28 (m, 3Harom), 7.35-7.45 (m, 7Harom), 7.66-7.67 (m, 4Harom), 7.90 (s, 1Harom), 8.01-8.03 (m, 4Harom), 8.27-8.29 (m, 1Harom), 8.33-8.34 (m, 1Harom), 8.47-8.48 (m, 1Harom). ¹³C{¹H} NMR (C₆D₆) δ (ppm) = 20.9 (2C), 82.3, 98.9, 116.7, 119.9 (4C), 123.1, 123.4, 126.9, 127.4, 127.5, 127.7 (2C), 127.9 (4C), 128.2 (2C), 129.2,

129.5, 129.8, 130.1 (4C), 130.5, 130.8, 130.9, 131.8, 133.9 (2C), 134.2 (4C), 134.4, 136.1, 138.0, 141.4. ¹¹B{¹H} NMR (C₆D₆) δ (ppm) = 9.05. Elem. anal. calc for [C₄₃H₃₃BN₂]: C, 87.75; H, 5.65; N, 4.76. Found: C, 87.81; H, 5.67; N, 4.80. m.p. 70–72 °C.



3.- ¹H and ¹³C NMR spectra

Figure S3. ¹H-NMR full chart for 3a in C₆D₆.



Figure S5. ¹H-NMR full chart for **3b** in C₆D₆.



Figure S6. ${}^{13}C{}^{1}H$ -NMR full chart for 3b in C₆D₆.



Figure S7. ¹H-NMR full chart for 3c in C₆D₆.



Figure S8. ${}^{13}C{}^{1}H$ -NMR full chart for 3c in C₆D₆.



Figure S9. $^{19}\mathrm{F}\{^{1}\mathrm{H}\}\text{-}\mathrm{NMR}$ full chart for 3c in C₆D₆.



Figure S11. ${}^{13}C{}^{1}H$ -NMR full chart for 3d in C₆D₆.



Figure S12. ¹H-NMR full chart for 3e in C₆D₆.



Figure S13. ${}^{13}C{}^{1}H$ -NMR full chart for 3e in C₆D₆.



Figure S14. ¹H-NMR full chart for 5a in C_6D_6 .



Figure S15. ${}^{13}C{}^{1}H$ -NMR full chart for 5a in C₆D₆.



Figure S16. ¹H-NMR full chart for 5b in C_6D_6 .



Figure S17. ${}^{13}C{}^{1}H$ -NMR full chart for 5b in C₆D₆.



Figure S19. ${}^{13}C{}^{1}H$ -NMR full chart for 5c in C₆D₆.



Figure S20. ${}^{19}F{}^{1}H$ -NMR full chart for 5c in C₆D₆.



Figure S21. ¹H-NMR full chart for 5d in C₆D₆.



Figure S22. ${}^{13}C{}^{1}H$ -NMR full chart for 5d in C₆D₆.



Figure S23. ¹H-NMR full chart for 5e in C₆D₆.



Figure S24. ${}^{13}C{}^{1}H$ -NMR full chart for 5e in C₆D₆.



Figure S25. ¹H-NMR full chart for 4a in C₆D₆.



Figure S26. ${}^{13}C{}^{1}H$ -NMR full chart for 4a in C₆D₆.



Figure S27. ${}^{11}B{}^{1H}$ -NMR full chart for 4a in C₆D₆.



Figure S28. ¹H-NMR full chart for 4b in C₆D₆.



Figure S29. ${}^{13}C{}^{1}H$ -NMR full chart for 4b in C₆D₆.



Figure S30. ${}^{11}B{}^{1}H$ -NMR full chart for 4b in C₆D₆.



Figure S31. ¹H-NMR full chart for 4c in C₆D₆.







Figure S34. ¹¹B{¹H}-NMR full chart for 4c in C_6D_6 .



Figure S35. ¹H-NMR full chart for 4d in C₆D₆.



Figure S36. ${}^{13}C{}^{1}H$ -NMR full chart for 4d in C₆D₆.



Figure S37. ¹¹B $\{^{1}H\}$ -NMR full chart for 4d in C₆D₆.



Figure S38. ¹H-NMR full chart for 4e in C₆D₆.



Figure S39. ${}^{13}C{}^{1}H$ -NMR full chart for 4e in C₆D₆.



Figure S40. $^{11}B{^{1}H}$ -NMR full chart for 4e in C₆D₆.



Figure S41. ¹H-NMR full chart for 6a in C₆D₆.



Figure S42. ${}^{13}C{}^{1}H$ -NMR full chart for 6a in C₆D₆.



Figure S43. ${}^{11}B{}^{1H}$ -NMR full chart for 6a in C₆D₆.



Figure S44. ¹H-NMR full chart for 6b in C₆D₆.



Figure S45. ${}^{13}C{}^{1}H$ -NMR full chart for 6b in C₆D₆.



Figure S46. ${}^{11}B{}^{1}H$ -NMR full chart for 6b in C₆D₆.



Figure S47. ¹H-NMR full chart for 6c in C₆D₆.



Figure S48. ${}^{13}C{}^{1}H$ -NMR full chart for 6c in C₆D₆.



Figure S49. ${}^{19}F{}^{1}H$ -NMR full chart for 6c in C₆D₆.



Figure S50. $^{11}B{^1H}$ -NMR full chart for 6c in C₆D₆.



Figure S51. ¹H-NMR full chart for 6d in C₆D₆.



Figure S52. ${}^{13}C{}^{1}H$ -NMR full chart for 6d in C₆D₆.



Figure S53. ¹¹B $\{^{1}H\}$ -NMR full chart for **6d** in C₆D₆.



Figure S54. ¹H-NMR full chart for 6e in C₆D₆.



Figure S55. ${}^{13}C{}^{1}H$ -NMR full chart for 6e in C₆D₆.



Figure S56. ¹¹B $\{^{1}H\}$ -NMR full chart for 6e in C₆D₆.



.27 3.26 3.25 3.24 3.23 3.22 3.21 3.20 3.19 3.18 3.17 3.16 3.15 3.14 3.13 3.12 3.11 3.10 3.09 3.08 3.07 3.06 3.05 3.04 3.03 3.02 3.01 3.00 2.99 f1 (ppm)

Figure S57. ¹H NMR spectra of **3d** after reaction with triflic acid (up), **4d** after reaction with triflic acid (middle) and **4d** (bottom), in CD₃CN, in the region of the NMe₂ group.

4.- X-ray structural analyses

Crystals were mounted on a glass fibre and cooled to the 100 K for compound 4a, 150K for compounds 4e and 6a, and room temperature for compounds 4b, 4c, 6e and 6d. Data were collected on a Bruker APEX II CCD-based diffractometer equipped with a graphite monochromated MoK α radiation source (λ =0.71073 Å). Intensities were integrated in SAINT [3] and absorption corrections based on equivalent reflections were applied using SADABS. [4] Structures were solved using ShelXT, [5] all of the structures were refined by full matrix least squares against F^2 in ShelXL [6] using Olex2. [7] All of the non-hydrogen atoms were refined anisotropically, while all of the hydrogen atoms were located geometrically and refined using a riding model. Compounds 4a, 4b, 4c, 6e and 6d show some disordered fragments and the occupancies of the disordered group were refined with their sum set to equal 1 and subsequently fixed at the refined values. Restraints were applied to maintain sensible thermal and geometric parameters. The structure of compound 6d was also refined as a two-component twin (-1 0 0 0 -1 0 0 0 1), with twin fraction 0.496(2). Moreover, in compound 4b due to the quality of the crystals some data was rejected as poor during integration and scaling, being the origin of the alerts B during checkcif test. The X-ray crystallographic coordinates for structures reported in this study have been deposited at the Cambridge Crystallographic Data Centre (CCDC) under deposition numbers 2307608-2307613 for 4a, 4b, 4c, 4e, 6a, 6d and 6e respectively. These data can be obtained free of charge via http://www.ccdc.cam.ac.uk/data request/cif

Identification code	4a	4b	4c	4 e
Empirical formula	$C_{27}H_{29}BN_2$	C ₂₈ H ₃₁ BN ₂ O	C ₂₇ H ₂₈ BFN ₂	$C_{75}H_{78}B_2N_4$
Formula weight	392.33	422.36	410.32	1057.03
Temperature/K	100.15	296.15	296.15	150.16
Crystal system	monoclinic	monoclinic	monoclinic	triclinic
Space group	$P2_1/n$	$P2_1/n$	P2 ₁ /n	P-1
a/Å	9.5576(3)	9.823(16)	9.716(15)	9.2936(4)
b/Å	18.0239(8)	18.37(3)	18.14(3)	11.0040(6)
c/Å	13.8891(4)	14.50(3)	14.07(2)	15.8337(9)
α/°	90	90	90	98.173(2)
β/°	101.516(3)	101.37(3)	101.32(2)	106.630(2)
γ/°	90	90	90	95.270(2)

Table S1. Crystal data and structure refinement for 4a, 4b, 4c and 4e.

Volume/Å ³	2344.44(15)	2566(8)	2432(6)	1520.70(14)
Ζ	4	4	4	1
$\rho_{calc}g/cm^3$	1.112	1.093	1.121	1.154
μ/mm ⁻¹	0.064	0.065	0.070	0.066
F(000)	840.0	904.0	872.0	566.0
Crystal size/mm ³	0.313 × 0.293 ×	$0.614 \times 0.360 \times$	0.421× 0.205 ×	0.271×0.271
	0.284	0.201	0.234	× 0.086
Radiation	MoK α (λ =	MoK α (λ =	ΜοΚα (λ =	MoK α (λ =
Radiation	0.71073)	0.71073)	0.71073)	0.71073)
2⊖ range for data collection/°	5.422 to 50.788	4.776 to 54.034	5.204 to 55.476	3.776 to 52.934
	$-10 \le h \le 11, -$	$-12 \le h \le 9, -22$	$-12 \le h \le 9, -23$	$-11 \le h \le 11, -$
Index ranges	$21 \le k \le 21, -16$	\leq k \leq 21, -17 \leq l	\leq k \leq 23, -18 \leq l	$13 \le k \le 13, -$
	$\leq l \leq 15$	≤ 18	≤ 18	$19 \le l \le 19$
Reflections collected	21988	12776	18874	92440
Independent reflections	4298 [R _{int} = 0.0296, R _{sigma} = 0.0251]	5228 [R _{int} = 0.0765, R _{sigma} = 0.0957]	5685 [R _{int} = 0.0547, R _{sigma} = 0.0679]	$6244 [R_{int} = 0.0645, R_{sigma} = 0.0305]$
Data/restraints/para meters	4298/14/296	5228/1036/516	5685/50/305	6244/14/394
Goodness-of-fit on F ²	1.032	1.247	1.049	1.060
Final R indexes	$R_1 = 0.0408,$	$R_1 = 0.1722,$	$R_1 = 0.0807,$	$R_1 = 0.0547,$
[I>=2σ (I)]	$wR_2 = 0.1021$	$wR_2 = 0.3835$	$wR_2 = 0.1684$	$wR_2 = 0.1464$
Final R indexes [all	$R_1 = 0.0545,$	$R_1 = 0.2207,$	$R_1 = 0.1529,$	$R_1 = 0.0852,$
data]	$wR_2 = 0.1110$	$wR_2 = 0.4223$	$wR_2 = 0.2001$	$wR_2 = 0.1691$
Largest diff. peak/hole / e Å ⁻³	0.24/-0.29	0.62/-0.56	0.18/-0.22	0.37/-0.41

 Table S2. Crystal data and structure refinement for 6a, 6d and 6e.

Identification code	6a	6e	6d
Empirical formula	$C_{35}H_{29}BN_2$	C ₃₇ H ₃₄ BN ₃	$C_{44}H_{35}BCl_2N_2$
Formula weight	488.41	531.48	673.45
Temperature/K	150.01	296.15	293
Crystal system	triclinic	triclinic	monoclinic
Space group	P-1	P-1	$P2_1/c$
a/Å	8.4512(5)	11.090(4)	9.689(4)
b/Å	11.5034(5)	11.227(4)	34.583(14)
c/Å	14.4397(8)	13.757(4)	10.707(5)

α/°	83.230(2)	104.370(4)	90
β/°	86.245(3)	92.473(4)	90.005(4)
γ/°	73.563(2)	111.352(5)	90
Volume/Å ³	1336.27(12)	1528.3(8)	3588(3)
Ζ	2	2	4
$\rho_{calc}g/cm^3$	1.214	1.155	1.247
µ/mm ⁻¹	0.070	0.067	0.215
F(000)	516.0	564.0	1408.0
Cravatal aiza/mm ³	$0.201 \times 0.084 \times$	$0.301 \times 0.191 \times$	$0.112 \times 0.101 \times$
	0.038	0.189	0.043
Padiation	MoK α (λ =	ΜοΚα (λ =	MoK α (λ =
	0.71073)	0.71073)	0.71073)
2⊖ range for data collection/°	7.398 to 50.696	5.092 to 53.852	3.982 to 49.996
	$-10 \le h \le 10, -$	$-14 \le h \le 14, -$	$-11 \le h \le 8, -38$
Index ranges	$13 \le k \le 13, -17$	$14 \le k \le 13, -17$	\leq k \leq 36, -12 \leq 1
	$\leq l \leq 17$	$\leq l \leq 17$	≤ 8
Reflections collected	23641	11946	11063
	4827 [R _{int} =	6426 [R _{int} =	5853 [R _{int} =
Independent reflections	$0.0530, R_{sigma} =$	$0.0367, R_{sigma} =$	$0.0927, R_{sigma} =$
	0.0464]	0.0712]	0.1467]
Data/restraints/parameter s	4827/0/345	6426/87/404	5853/144/511
Goodness-of-fit on F ²	1.026	0.967	0.970
Final R indexes [I>=2 σ	$R_1 = 0.0457,$	$R_1 = 0.0594,$	$R_1 = 0.0681,$
(I)]	$wR_2 = 0.1043$	$wR_2 = 0.1314$	$wR_2 = 0.1410$
Final D indover [all d-t-]	$R_1 = 0.0775,$	$R_1 = 0.1588,$	$R_1 = 0.1577,$
	$wR_2 = 0.1187$	$wR_2 = 0.1810$	$wR_2 = 0.1864$
Largest diff. peak/hole / e Å ⁻³	0.28/-0.24	0.14/-0.20	0.24/-0.24

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