

Supplementary Information

Title: Iron photocatalysis via Brønsted acid-unlocked ligand-to-metal charge transfer

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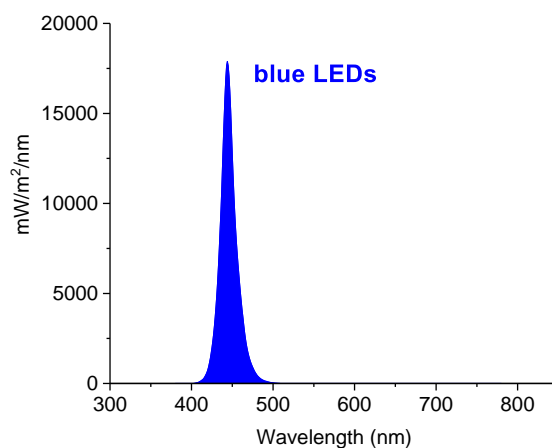
1. Supplementary Methods

1.1 General information

All manipulations were carried out by standard Schlenk techniques. Unless otherwise stated, analytical grade solvents and commercially available reagents were used to conduct the reactions. LED irradiation was accomplished using the photochemical reactors (Supplementary Figure 1), whose wavelength is 404 – 499 nm (Supplementary Figure 2). Thin layer chromatography (TLC) employed glass 0.20 – 0.25 mm silica gel plates. Purification of products was accomplished using flash chromatography on silica gel. All the new compounds were characterized by ^1H NMR, ^{13}C NMR, ^{19}F NMR and ESI-HRMS. The known compounds were characterized by ^1H NMR, ^{13}C NMR, ^{19}F NMR and ESI-HRMS. NMR spectra were recorded on a Bruker 400 instrument operating at 400, 101, and 376 MHz for ^1H NMR, ^{13}C NMR and ^{19}F NMR respectively. Chemical shifts (δ) were reported in ppm, and coupling constants (J) are in Hertz (Hz). Data was reported in ppm using CDCl_3 (^1H NMR δ 7.26, ^{13}C NMR δ 77.16) or $\text{DMSO-}d_6$ (^1H NMR δ 2.50, ^{13}C NMR δ 39.52) as solvent unless otherwise specified. The following abbreviations were used to explain the multiplicities: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet. High-resolution mass data was recorded on mass spectrometers in the ESI-HRMS mode.



Supplementary Figure 1. Photochemical reactors.

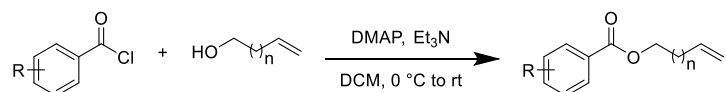


Supplementary Figure 2. Wavelength spectrum of photochemical reactors.

1.2 Preparation of substrates and catalysts

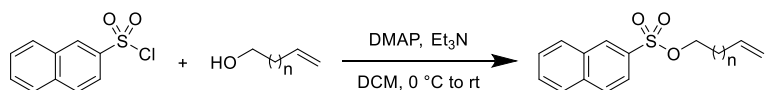
Methods of structural synthesis

To a 50 mL Schlenk tube, 4-dimethylaminopyridine (0.5 mmol), Et₃N (10.0 mmol) and alcohol (5.0 mmol) were dissolved in dichloromethane (12.0 mL). Acyl chloride (7.5 mmol) was added dropwise at 0 °C. The solution was stirred for 2 h at room temperature. Sat. NaHCO₃ aq. was poured into the reaction mixture. The organic layer was separated and extracted from water with EtOAc. The combined organic phase was washed with brine, dried over Na₂SO₄, filtered, and concentrated under vacuum. The residue was purified by silica gel column chromatography to afford the products¹.



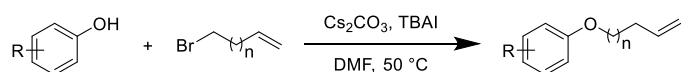
Supplementary Figure 3. Method A for the synthesis of non-activated alkenes.

To a 50 mL Schlenk tube, 4-dimethylaminopyridine (0.5 mmol), Et₃N (10.0 mmol) and alcohol (5.0 mmol) were dissolved in dichloromethane (12.0 mL). Sulfonyl chloride (7.5 mmol) was dropwise added at 0 °C. The reaction mixture was stirred for 2 h at room temperature. Sat. NaHCO₃ aq. was poured into the reaction mixture. The organic layer was separated and extracted from water with EtOAc. The combined organic phase was washed with brine, dried over Na₂SO₄, filtered, and concentrated under vacuum. The residue was purified by silica gel column chromatography to afford the products¹.



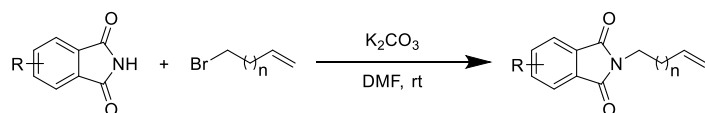
Supplementary Figure 4. Method B for the synthesis of non-activated alkenes.

To a 50 mL Schlenk tube, tetrabutylammonium iodide (0.5 mmol), Cs₂CO₃ (6.5 mmol) and phenol (6.0 mmol) were dissolved in dimethylformamide (12.0 mL). Alkenyl halide (5.0 mmol) was added and the mixture was stirred for 16 h at 50 °C. Diluted NaOH aq. was poured into the reaction mixture. The organic layer was separated and extracted from water with *tert*-Butyl methyl ether. The combined organic phase was washed with brine, dried over Na₂SO₄, filtered, and concentrated under vacuum. The residue was purified by silica gel column chromatography to afford the products¹.



Supplementary Figure 5. Method C for the synthesis of non-activated alkenes.

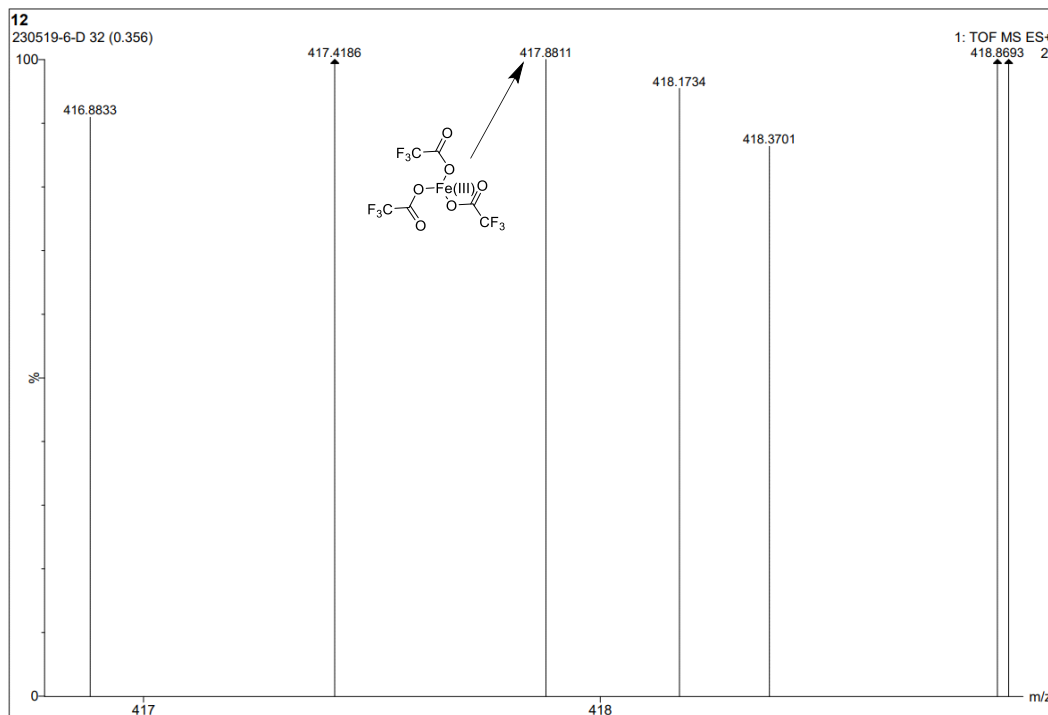
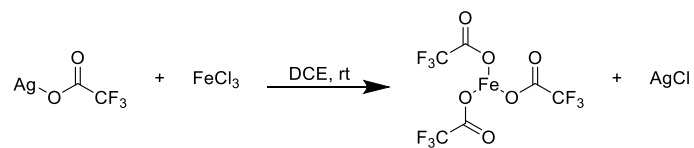
To a 100 mL Schlenk tube, isoindoline-1, 3-dione (10.0 mmol), alkenyl halide (13.0 mmol) and K₂CO₃ (13.0 mmol) were dissolved in dimethylformamide (30.0 mL). The reaction mixture was stirred at room temperature, and the progress of the reaction was monitored by the TLC analysis. After 6 h the reaction was completed, as judged by TLC. The reaction mixture was slowly poured into water (100.0 mL). The aqueous phase was extracted with CH₂Cl₂ (5 × 20.0 mL). The organic extract was washed with water (3 × 40.0 mL), dried over Na₂SO₄, filtered, and concentrated under vacuum. The residue was purified by silica gel column chromatography to afford the products².



Supplementary Figure 6. Method D for the synthesis of non-activated alkenes.

The synthesis of Fe(CF₃COO)₃: To a 50 mL Schlenk tube, iron(III) chloride (5.0 mmol) and silver trifluoroacetate (15.0 mmol) were dissolved and reacted in dichloromethane (15.0 mL) for 3 h at room temperature. The precipitate AgCl was filtered and the residue was washed with acetone. The corresponding filtrate was concentrated under reduced pressure and the resulting product was dried

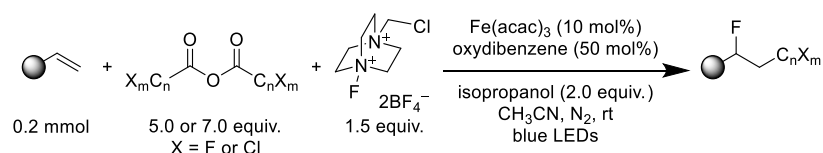
at 80 °C for 1 h. Iron(III) tri-fluoroacetate was obtained as red solid³. **ESI-HRMS** for $C_6F_9FeO_6Na^+$: 417.8793; found: 417.8811.



Supplementary Figure 7. Synthesis of $Fe(CF_3COO)_3$ and **ESI-HRMS** detection of $Fe(CF_3COO)_3$.

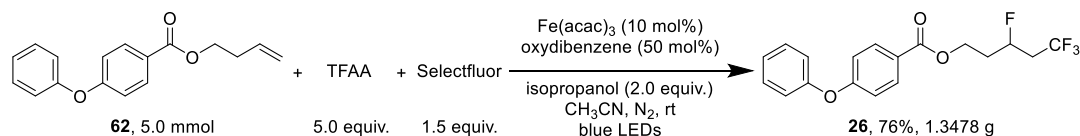
1.3 Experimental procedures for fluoro-polyhaloalkylation

General procedure for fluoro-polyhaloalkylation of alkenes: A 25 mL Schlenk flask equipped with a magnetic bar was charged with Fe(acac)₃ (7.1 mg, 0.02 mmol) and Selectfluor (106.3 mg, 0.3 mmol). The flask was evacuated and refilled with N₂ for three times. The vessel was then charged with extra dry CH₃CN (2.0 mL), alkene (0.2 mmol), (C_nX_mCO)₂O (X: F or Cl) (1.0 or 1.4 mmol), oxydibenzene (17.0 mg, 0.1 mmol) and isopropanol (24.0 mg, 0.4 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs for 12 or 24 or 36 h. After completion of the reaction, CO₂ was detected by TCD-GC. Then the reaction system was quenched by triethylamine, diluted with EtOAc. After concentrated under vacuum, the resulting residue was purified by silica gel flash column chromatography to give the products.



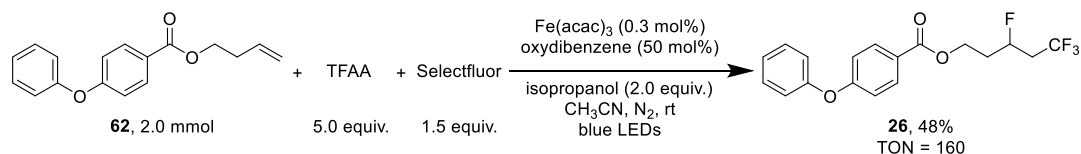
Supplementary Figure 8. Brønsted acid-unlocked iron LMCT photocatalysis for radical fluoro-polyhaloalkylation of alkenes.

General procedure for gram scale synthesis: A 100 mL Schlenk flask equipped with a magnetic bar was charged with Fe(acac)₃ (0.1766 g, 0.5 mmol) and Selectfluor (2.6570 g, 7.5 mmol). The flask was evacuated and refilled with N₂ for three times. The vessel was then charged with extra dry CH₃CN (50.0 mL), alkene (5.0 mmol), TFAA (5.2508 g, 25.0 mmol), oxydibenzene (0.4255 g, 2.5 mmol) and isopropanol (0.6010 g, 10.0 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs for 72 h. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. After concentrated under vacuum, the resulting residue was purified by silica gel flash column chromatography to give the product.



Supplementary Figure 9. Scale up reaction.

General procedure for TON experiment: A 100 mL Schlenk flask equipped with a magnetic bar was charged with Fe(acac)₃ (0.0021 g, 0.006 mmol) and Selectfluor (1.0628 g, 3.0 mmol). The flask was evacuated and refilled with N₂ for three times. The vessel was then charged with extra dry CH₃CN (20.0 mL), alkene (2.0 mmol), TFAA (2.1003 g, 10.0 mmol), oxydibenzene (0.1702 g, 1.0 mmol) and isopropanol (0.2406 g, 4.0 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs for 144 h. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. After concentrated under vacuum, the resulting residue was purified by silica gel flash column chromatography to give the product.

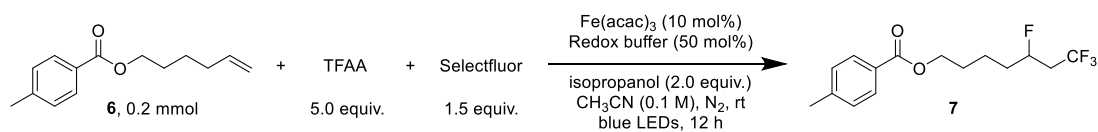


Supplementary Figure 10. TON experiment.

2. Supplementary Discussion

2.1 Investigation of conditions

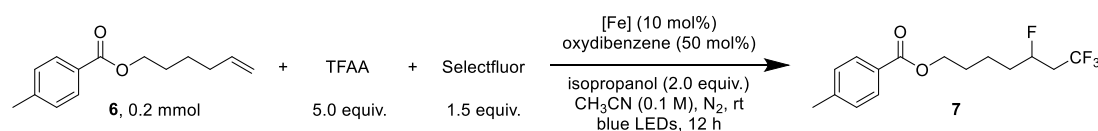
Investigation of anisoles



| Entry | Redox buffer | 7 , Yield |
|-------|----------------------------|------------------|
| 1 | anisole | 47% |
| 2 | methyl 4-methoxybenzoate | 43% |
| 3 | 1-methoxynaphthalene | Trace |
| 4 | <i>tert</i> -butoxybenzene | 15% |

Supplementary Table 1. Results of investigation of anisoles. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard.

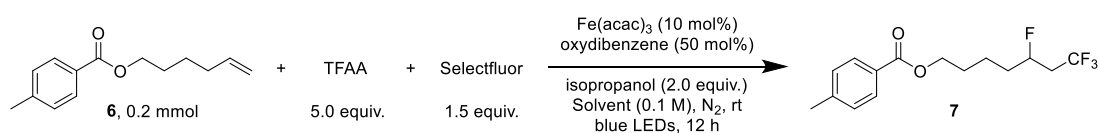
Investigation of iron salts



| Entry | [Fe] | 7 , Yield |
|-------|--|------------------|
| 1 | FeCl ₃ | 53% |
| 2 | Fe(CF ₃ COO) ₃ | 67% |
| 3 | Fe(OH)(OAc) ₂ | 55% |
| 4 | Fe(NO ₃) ₃ ·9H ₂ O | 43% |
| 5 | Fe(acac) ₂ | 64% |

Supplementary Table 2. Results of investigation of iron salts. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard.

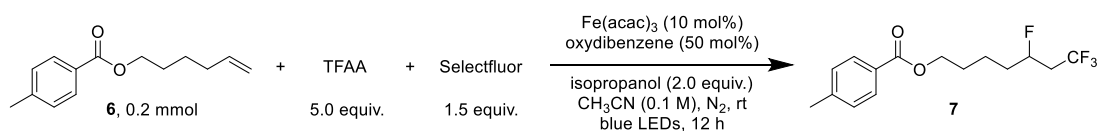
Investigation of solvents



| Entry | Solvent | 7, Yield |
|-------|-------------------------------------|----------|
| 1 | CH ₃ CN | 70% |
| 2 | Acetone | N.D. |
| 3 | DCE | N.D. |
| 4 | DCM | N.D. |
| 5 | EA | 42% |
| 6 | DMF | N.D. |
| 7 | DMSO | N.D. |
| 8 | THF | N.D. |
| 9 | CCl ₃ CN | N.D. |
| 10 | (CH ₃) ₃ CCN | 16% |
| 11 | N-methyl-2-pyrrolidone | N.D. |
| 12 | 1,4-Dioxane | N.D. |
| 13 | 1,2-Dimethoxyethane | N.D. |
| 14 | 2-Methoxy-2-methylpropane | N.D. |

Supplementary Table 3. Results of investigation of solvents. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard.

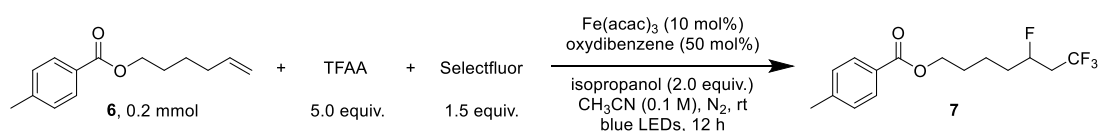
Deviation from standard condition



| Entry | Deviation from standard condition | 7, Yield |
|-------|---------------------------------------|----------|
| 1 | None | 70% |
| 2 | CF ₃ COONa instead of TFAA | N.D. |
| 3 | TFA instead of TFAA | 31% |
| 4 | NFSI as [F] source | 28% |
| 5 | NFPY-OTf as [F] source | N.D. |
| 6 | Green LEDs | N.D. |
| 7 | Heat to 60 °C or 80 °C (w/o light) | N.D. |

Supplementary Table 4. Results of deviation from standard condition. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard.

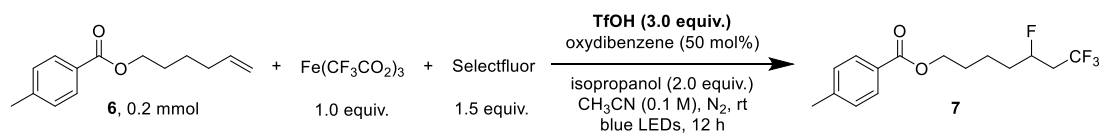
Controlled experiments for the fluorotrifluoromethylation of alkenes



| Entry | Deviation from the standard condition | 6 , Recovery | 7 , Yield |
|-------|--|---------------------|------------------|
| 1 | none | trace | 70% |
| 2 | w/o light | 85% | N.D. |
| 3 | w/o Fe(acac) ₃ | 65% | N.D. |
| 4 | w/o oxydibenzene | 0 | 15% |
| 5 | w/o light & oxydibenzene | 90% | N.D. |
| 6 | w/o Fe(acac) ₃ & oxydibenzene | 87% | N.D. |

Supplementary Table 5. Results of controlled experiments. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard.

Control experiments for the stoichiometric experiments

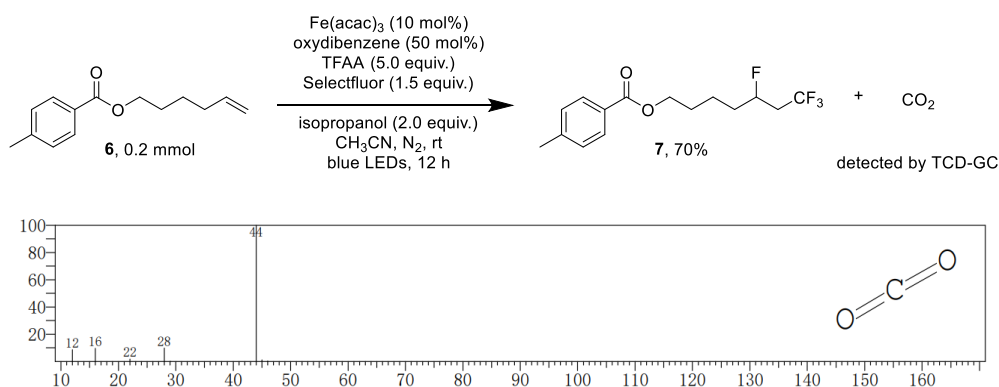


| Entry | Deviation from the standard conditions | 6 , Recovery | 7 , Yield |
|-------|---|---------------------|------------------|
| 1 | / | 10% | 41% |
| 2 | w/o isopropanol | 12% | 18% |
| 3 | w/o oxydibenzene | 0 | 21% |
| 4 | w/o oxydibenzene & isopropanol | 0 | 11% |
| 5 | lithium isopropoxide instead of isopropanol | 69% | N.D. |

Supplementary Table 6. Results of stoichiometric experiments. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard.

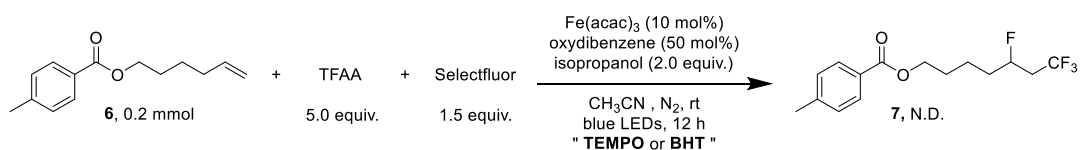
2.2 Mechanistic studies

General procedure for gas experimental analysis: A 25 mL Schlenk flask equipped with a magnetic bar was charged with Fe(acac)₃ (7.1 mg, 0.02 mmol) and Selectfluor (106.3 mg, 0.3 mmol). The flask was evacuated and refilled with N₂ for three times. The vessel was then charged with extra dry CH₃CN (2.0 mL), alkene (0.2 mmol), TFAA (210.0 mg, 1.0 mmol), oxydibenzene (17.0 mg, 0.1 mmol) and isopropanol (24.0 mg, 0.4 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs for 12 h. After completion of the reaction, CO₂ was detected by TCD-GC.



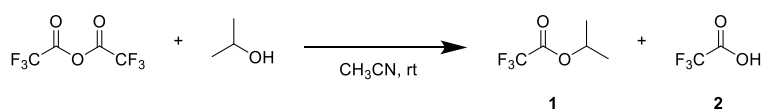
Supplementary Figure 11. Detection of CO₂.

General procedure for radical trapping experiments: A 25 mL Schlenk flask equipped with a magnetic bar was charged with Fe(acac)₃ (7.1 mg, 0.02 mmol), Selectfluor (106.3 mg, 0.3 mmol) and 2,2,6,6-tetramethyl-1-piperidinyloxy (**TEMPO**) (62.5 mg, 0.4 mmol) or butylated hydroxytoluene (**BHT**) (88.1 mg, 0.4 mmol). The flask was evacuated and refilled with N₂ for three times. The vessel was then charged with extra dry CH₃CN (2.0 mL), alkene (0.2 mmol), TFAA (210.0 mg, 1.0 mmol), oxydibenzene (17.0 mg, 0.1 mmol) and isopropanol (24.0 mg, 0.4 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs for 12 h. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. The reaction results were monitored by TLC and GC. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard. No desired product **7** was detected.



Supplementary Figure 12. Radical inhibition experiments.

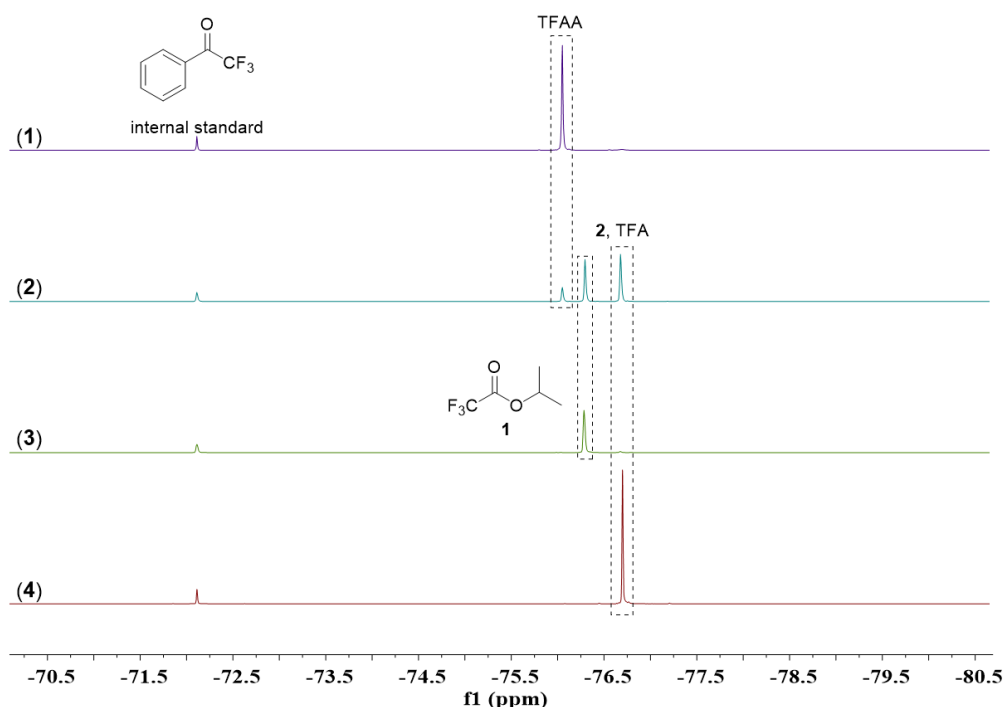
General Procedure for ¹⁹F NMR experiments: TFAA (0.5 mmol, 105.0 mg), isopropanol (0.4 mmol, 24.0 mg), trifluoroacetophenone (0.1 mmol, 17.4 mg) as internal standard, and CH₃CN (0.5 mL) were added in a vial, which was generated esters **1** and TFA **2**, the corresponding product was detected by ¹⁹F NMR.



Supplementary Figure 13. TFAA and alcohol *in situ* generating Brønsted acid.

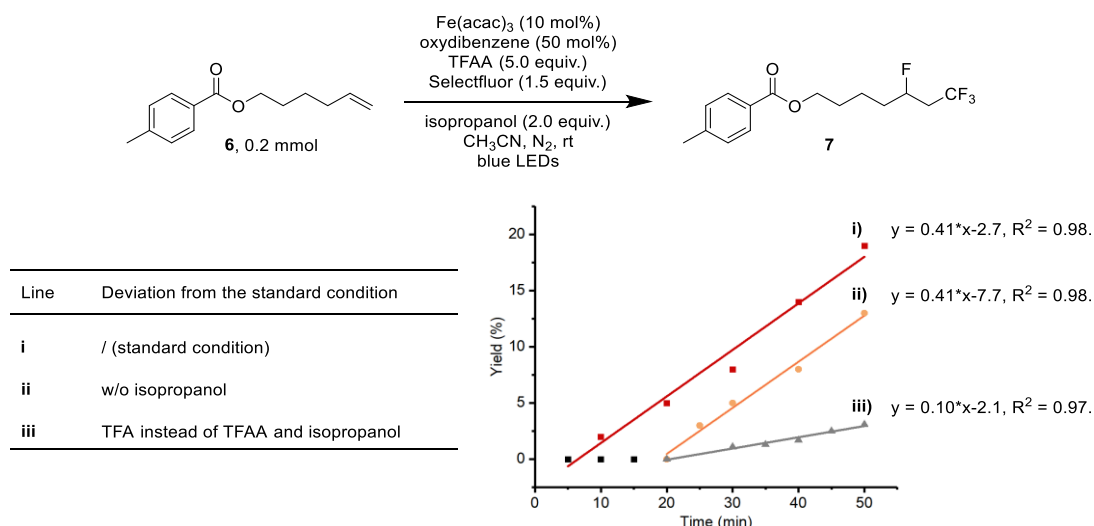
Reaction conditions:

- (1) TFAA (0.5 mmol) + internal standard (0.1 mmol) + CH₃CN (0.5 mL).
- (2) TFAA (0.5 mmol) + isopropanol (0.4 mmol) + internal standard (0.1 mmol) + CH₃CN (0.5 mL).
- (3) isopropyl 2,2,2-trifluoroacetate (0.5 mmol) + internal standard (0.1 mmol) + CH₃CN (0.5 mL).
- (4) TFA (0.5 mmol) + internal standard (0.1 mmol) + CH₃CN (0.5 mL).



Supplementary Figure 14. ^{19}F NMR analysis of TFAA and alcohol.

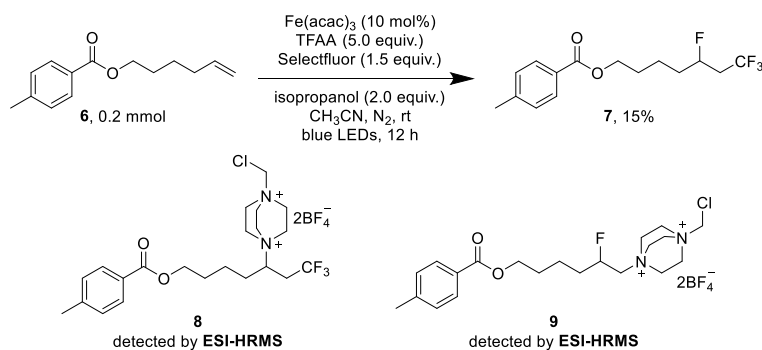
The advantage of using $(\text{CF}_3\text{CO})_2\text{O}$ and alcohol to *in situ* generate Brønsted acid: A 25 mL Schlenk flask equipped with a magnetic bar was charged with $\text{Fe}(\text{acac})_3$ (7.1 mg, 0.02 mmol) and Selectfluor (106.3 mg, 0.3 mmol). The flask was evacuated and refilled with N_2 for three times. The vessel was then charged with extra dry CH_3CN (2.0 mL), alkene (0.2 mmol), TFAA (210.0 mg, 1.0 mmol), oxydibenzene (17.0 mg, 0.1 mmol) and isopropanol (24.0 mg, 0.4 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard (Supplementary Figure 15 line i). In the absence of isopropanol under the standard condition (Supplementary Figure 15 line ii). When TFAA and isopropanol was replaced into TFA (Supplementary Figure 15 line iii). We found that line i works better than line ii and line iii.



Supplementary Figure 15. Initial reaction rates with TFAA or TFA as CF_3 source. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard.

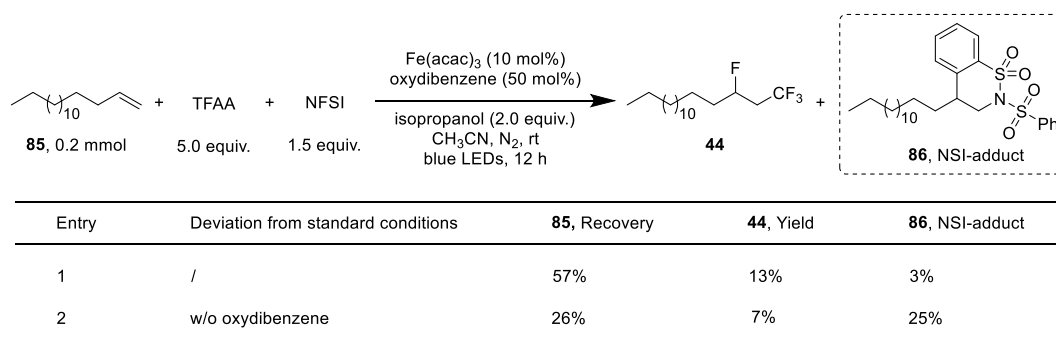
General procedure for aminotrifluoromethylation and fluoroamination of alkenes: A 25 mL Schlenk flask equipped with a magnetic bar was charged with $\text{Fe}(\text{acac})_3$ (7.1 mg, 0.02 mmol) and

Selectfluor (106.3 mg, 0.3 mmol). The flask was evacuated and refilled with N₂ for three times. The vessel was then charged with extra dry CH₃CN (2.0 mL), alkene (0.2 mmol), TFAA (1.0 mmol) and isopropanol (24.0 mg, 0.4 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs for 12 h. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. After concentrated under vacuum, the resulting residue was purified by silica gel flash column chromatography to give the product and by-products. Based on previous reports⁴, the role of oxydibenzene thus might serve as the redox buffer to timely quench the electrophilic *N* radical cation from the Selectfluor.



Supplementary Figure 16. Study of aminotrifluoromethylation and fluoroamination products.

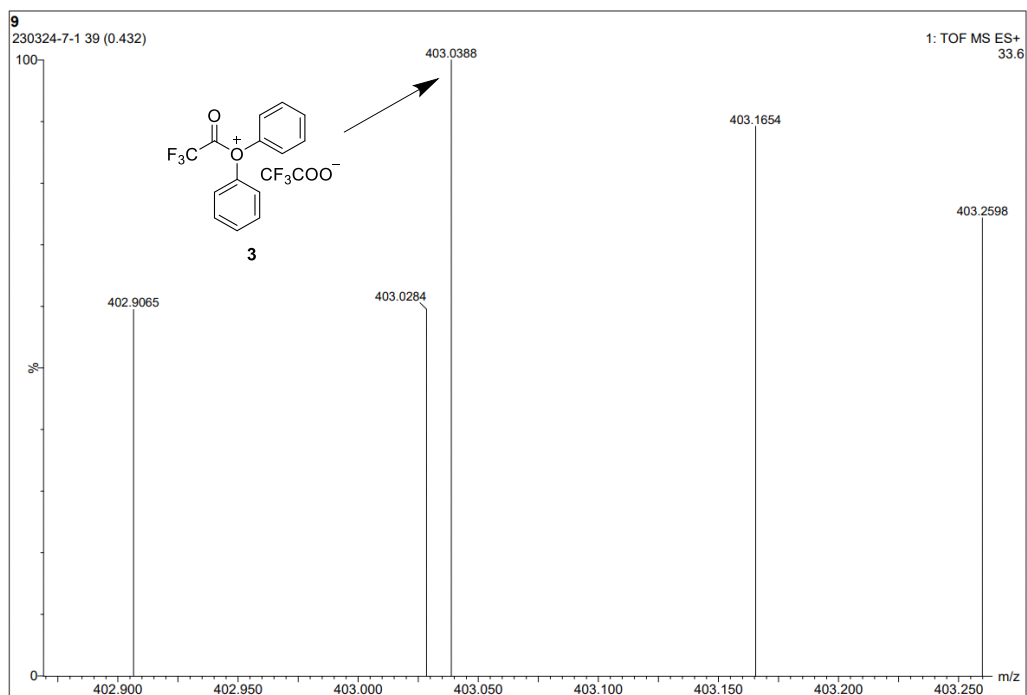
The role of the oxydibenzene in the reaction system: A 25 mL Schlenk flask equipped with a magnetic bar was charged with Fe(acac)₃ (7.1 mg, 0.02 mmol) and NFSI (94.6 mg, 0.3 mmol). The flask was evacuated and refilled with N₂ for three times. The vessel was then charged with extra dry CH₃CN (2.0 mL), alkene (0.2 mmol), TFAA (1.0 mmol), oxydibenzene (17.0 mg, 0.1 mmol) and isopropanol (24.0 mg, 0.4 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs for 12 h. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. After concentrated under vacuum, the resulting residue was purified by silica gel flash column chromatography to give the amination product, detected by NMR.



Supplementary Figure 17. Study of NFSI as fluorine source.

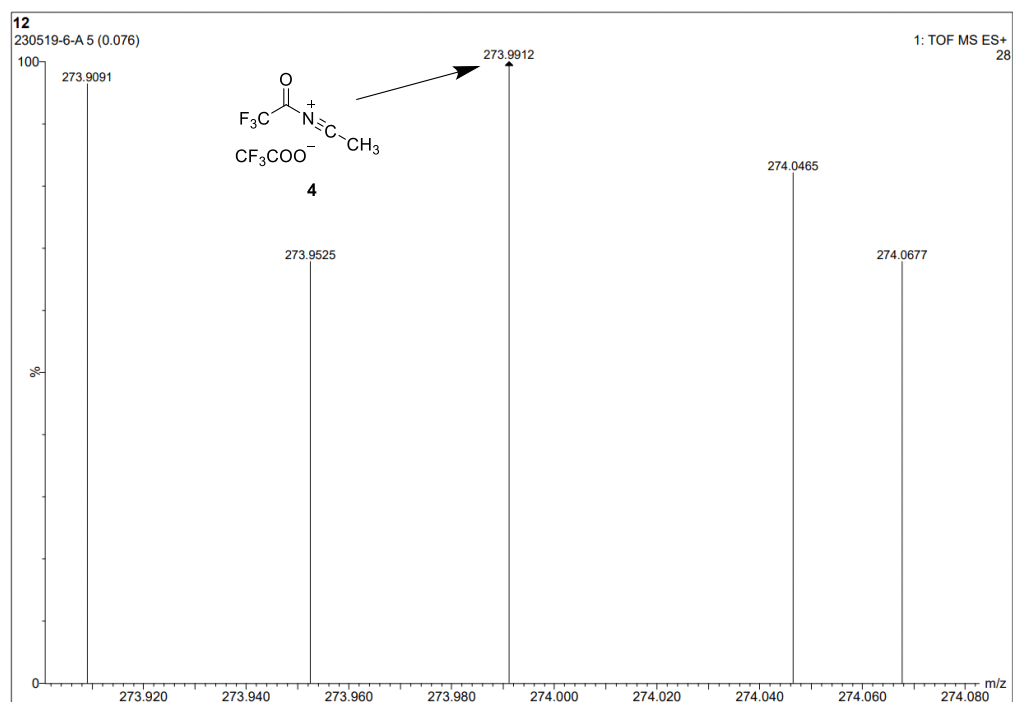
ESI-HRMS detection

ESI-HRMS for $C_{16}H_{10}F_6O_4Na^+$ (**3**): 403.0375; found: 403.0388.



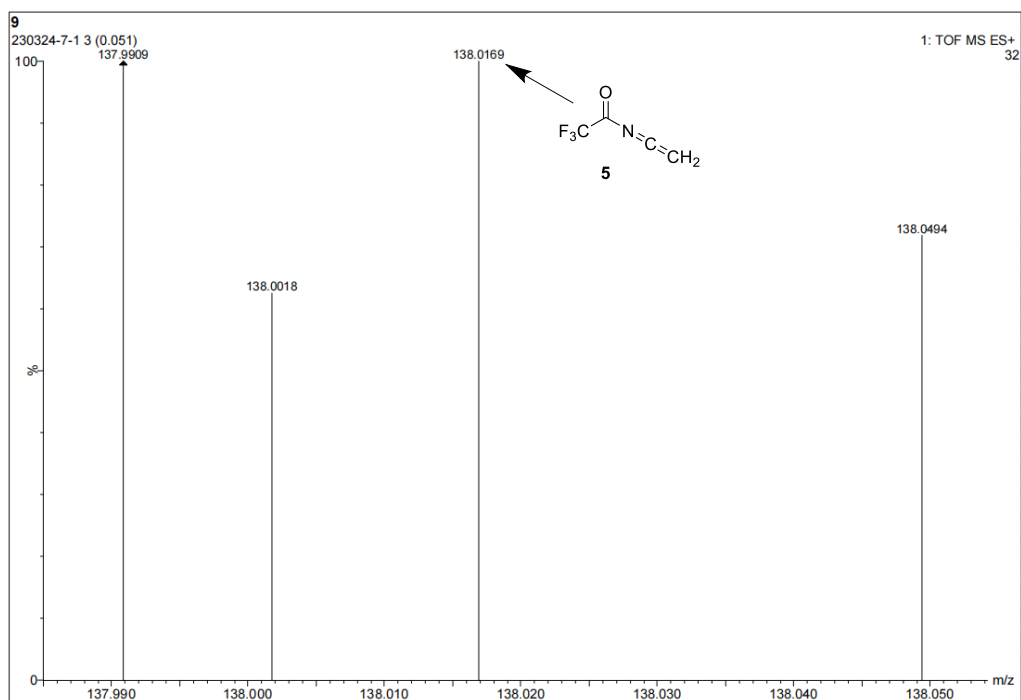
Supplementary Figure 18. ESI-HRMS detection of **3**.

ESI-HRMS for $C_6H_6F_6NO_3Na^+$ (**4**): 273.9909; found: 273.9912.



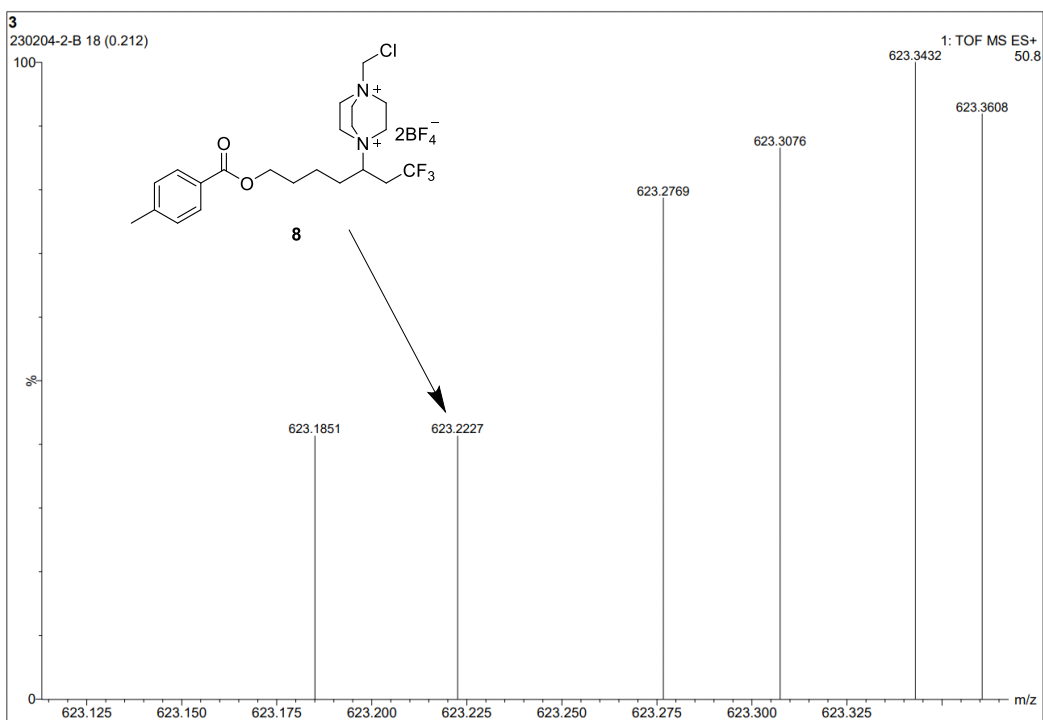
Supplementary Figure 19. ESI-HRMS detection of **4**.

ESI-HRMS for $C_4H_2F_3NOH^+$ (**5**): 138.0161; found: 138.0169.



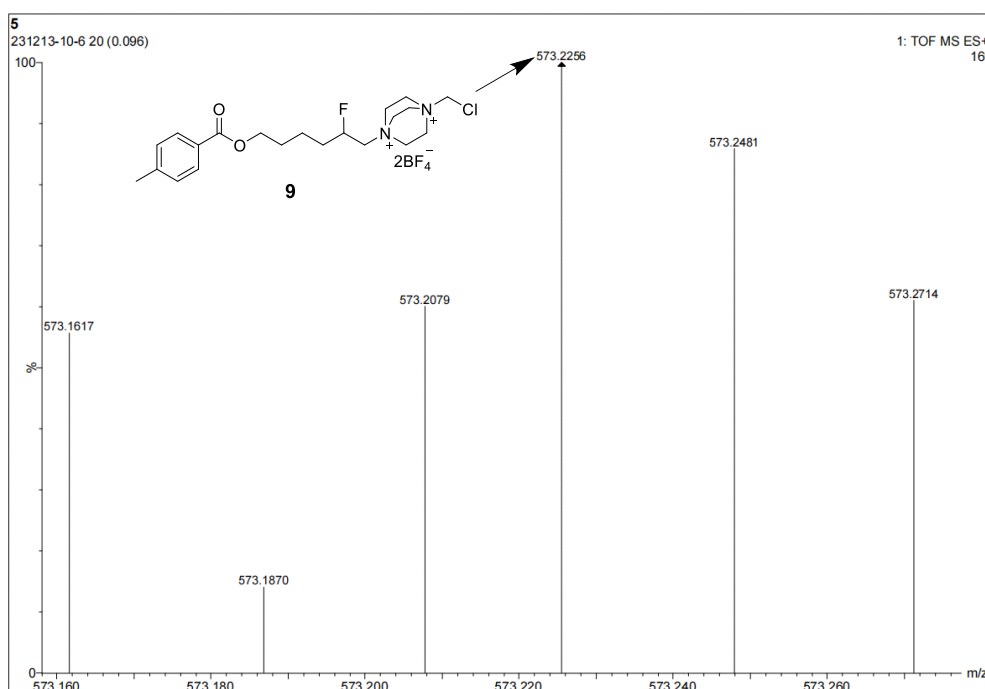
Supplementary Figure 20. ESI-HRMS detection of **5**.

ESI-HRMS for $C_{22}H_{32}B_2ClF_{11}N_2O_2H^+$ (**8**): 623.2236; found: 623.2227.



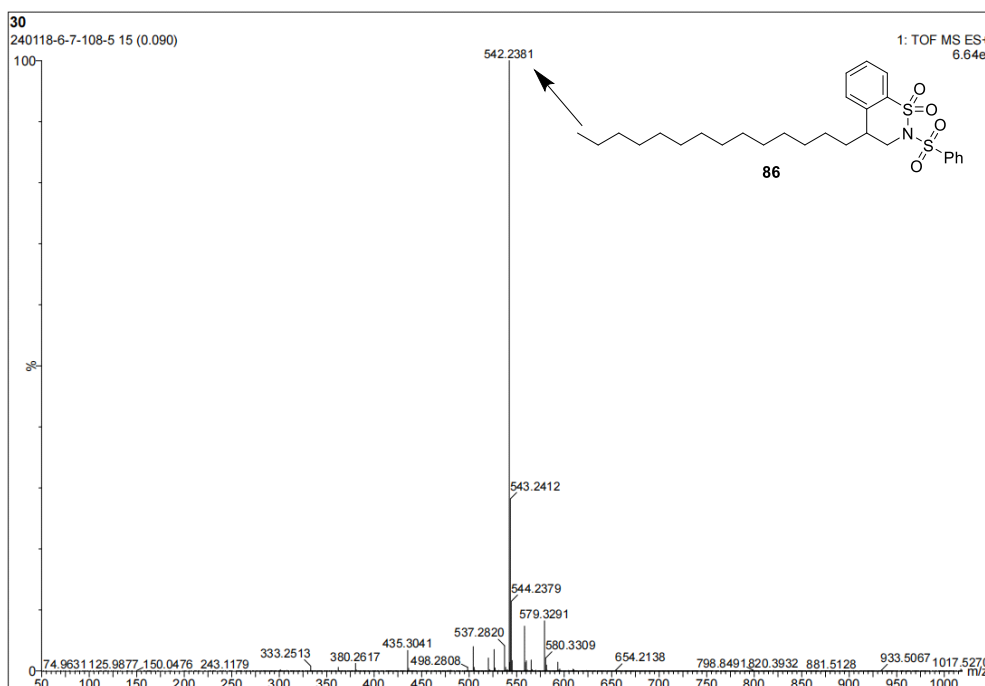
Supplementary Figure 21. ESI-HRMS detection of **8**.

ESI-HRMS for $C_{21}H_{32}B_2ClF_9N_2O_2H^+$ (**9**): 573.2267; found: 573.2256.



Supplementary Figure 22. ESI-HRMS detection of 9.

ESI-HRMS for $C_{28}H_{41}NO_4S_2Na^+$ (**86**): 542.2369; found: 542.2381.



Supplementary Figure 23. ESI-HRMS detection of 86.

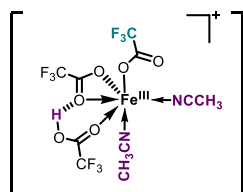
The study of the photoactive species (**12**)

a) $Fe(CF_3CO_2)_3$ + TFA (1:3) in CH_3CN , detected by **ESI-HRMS**, found: 477.9503, and 591.9427, consistent with theoretically exact mass of **12**: 477.9504, and **12** + $[CF_3CO_2]^-$ + $[H]^+$: 591.9433, respectively.

b) $Fe(CF_3CO_2)_3$ + TfOH (1:3) in CH_3CN , detected by **ESI-HRMS**, found: 477.9493, and 613.9250, consistent with theoretically exact mass of **12**: 477.9504, and **12** + $[CF_3CO_2]^-$ + $[Na]^+$: 613.9252,

respectively.

c) $\text{Fe}(\text{acac})_3 + \text{TFAA} + \text{isopropanol}$ (1:5:2) in CH_3CN , detected by **ESI-HRMS**, found: 477.9508, and 613.9256, consistent with theoretically exact mass of **12**: 477.9504, and **12** + $[\text{CF}_3\text{CO}_2^-] + [\text{Na}^+]$: 613.9252, respectively.



12: $\text{C}_{10}\text{H}_7\text{F}_9\text{FeN}_2\text{O}_6^+$, Exact mass: 477.9504.

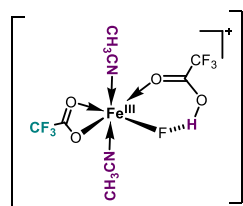
Supplementary Figure 24. Study of the photoactive species.

The study of the Fe-F species (**87**)

a) $\text{Fe}(\text{CF}_3\text{CO}_2)_3 + \text{TFA} + \text{TBAF}$ (1:3:3) in CH_3CN , detected by **ESI-HRMS**, found: 383.9360, and 519.9398, consistent with theoretically exact mass of **87**: 383.9638, and **87** + $[\text{CF}_3\text{CO}_2^-] + [\text{Na}^+]$: 519.9386, respectively.

b) $\text{Fe}(\text{CF}_3\text{CO}_2)_3 + \text{TfOH} + \text{TBAF}$ (1:3:3) in CH_3CN , detected by **ESI-HRMS**, found: 383.9305, and 497.9581, consistent with theoretically exact mass of **87**: 383.9638, and **87** + $[\text{CF}_3\text{CO}_2^-] + [\text{H}^+]$: 497.9567, respectively.

c) $\text{Fe}(\text{acac})_3 + \text{TFAA} + \text{isopropanol} + \text{TBAF}$ (1:5:2:3) in CH_3CN , detected by **ESI-HRMS**, found: 519.9399, consistent with theoretically exact mass of **87** + $[\text{CF}_3\text{CO}_2^-] + [\text{Na}^+]$: 519.9386.



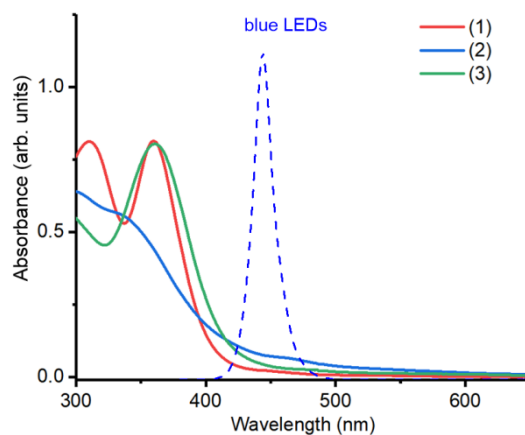
87: $\text{C}_8\text{H}_7\text{F}_7\text{FeN}_2\text{O}_4^+$, Exact mass: 383.9638.

Supplementary Figure 25. Study of the photoactive Fe-F species.

General procedure for UV-vis experiments:

- (1) $\text{FeCl}_3:\text{TFAA}:\text{isopropanol}:\text{dibenzene}$ (1:5:2:0.5).
- (2) $\text{Fe}(\text{CF}_3\text{COO})_3$ [0.25 mM].
- (3) $\text{Fe}(\text{CF}_3\text{COO})_3:\text{TfOH}$ (1:30).

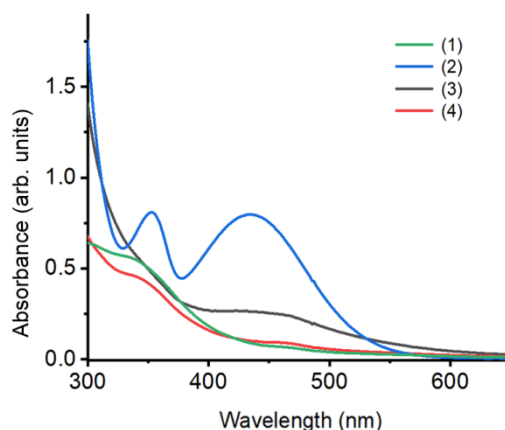
CH_3CN as solvent, the mixture was stirred 30 min in darkness, [iron] salt 0.25 mM.



Supplementary Figure 26. Absorbance profiles of different [iron] salt.

2. (1) $\text{Fe}(\text{CF}_3\text{CO}_2)_3$ under darkness [0.25 mM].
- (2) $\text{Fe}(\text{acac})_3$ under darkness [0.25 mM].
- (3) $\text{Fe}(\text{acac})_3$:TFA (1:3) under darkness.
- (4) $\text{Fe}(\text{acac})_3$:TFA (1:3) under blue LEDs.

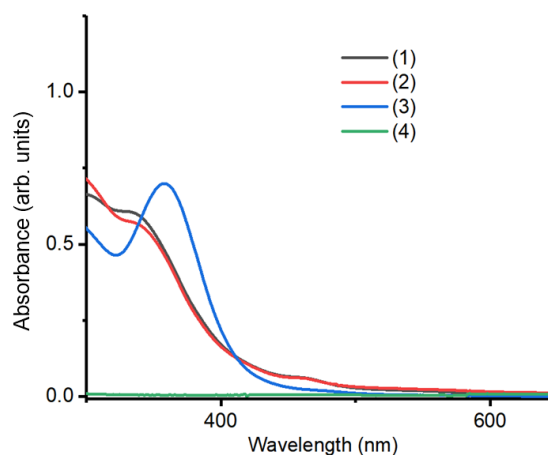
CH_3CN as solvent, the mixture was stirred 30 min, [iron] salt 0.25 mM.



Supplementary Figure 27. Absorbance profiles of [iron] salt and Brønsted acid.

3. (1) $\text{Fe}(\text{CF}_3\text{CO}_2)_3$ under darkness [0.25 mM].
- (2) $\text{Fe}(\text{CF}_3\text{CO}_2)_3$: $\text{CF}_3\text{CO}_2\text{Na}$ (1:3) under darkness.
- (3) $\text{Fe}(\text{CF}_3\text{CO}_2)_3$: $\text{CF}_3\text{CO}_2\text{Na}$:TfOH (1:3:6) under darkness.
- (4) $\text{Fe}(\text{CF}_3\text{CO}_2)_3$: $\text{CF}_3\text{CO}_2\text{Na}$:TfOH (1:3:6) under blue LEDs irradiation.

CH_3CN as solvent, the mixture was stirred 30 min, [iron] salt 0.25 mM.



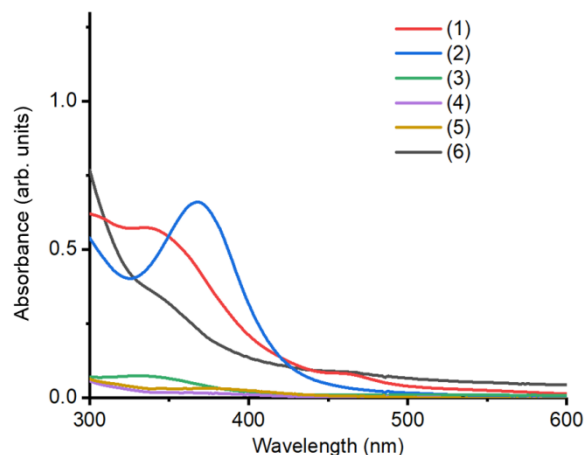
Supplementary Figure 28. Absorbance profiles of $\text{Fe}(\text{CF}_3\text{CO}_2)_3$ and $\text{CF}_3\text{CO}_2\text{Na}$.

4. The UV-vis monitorization of the oxidation of Fe(II) to Fe(III)

- (1) $\text{Fe}(\text{CF}_3\text{CO}_2)_3$ [0.25 mM].
- (2) $\text{Fe}(\text{CF}_3\text{CO}_2)_3$ + TfOH (1:30).
- (3) $\text{Fe}(\text{CF}_3\text{CO}_2)_3$ + TfOH (1:30) under blue LEDs irradiation.
- (4) $\text{Fe}(\text{CF}_3\text{CO}_2)_3$ + TfOH under blue LEDs irradiation, then, Selectfluor (1:30:15) added.
- (5) $\text{Fe}(\text{CF}_3\text{CO}_2)_3$ + TfOH under blue LEDs irradiation, then, Selectfluor + oxydibenzene (1:30:15:5) added.

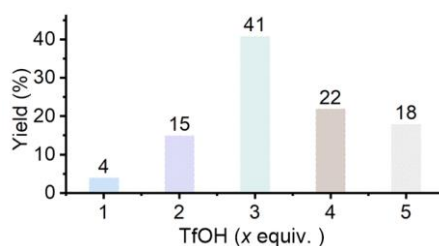
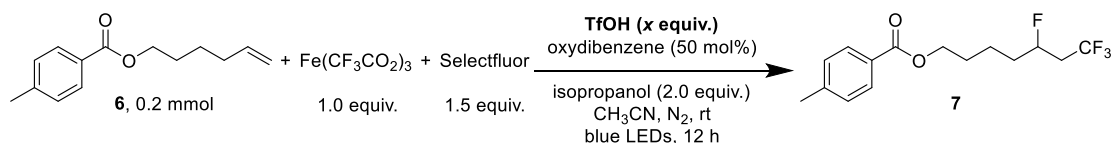
(6) $\text{Fe}(\text{CF}_3\text{CO}_2)_3$ + TfOH under blue LEDs irradiation, then, $\text{CF}_3\text{CO}_2\text{Na}$ + Selectfluor + oxydibenzene (1:30:10:15:5) added.

CH_3CN as solvent, the mixture was stirred 30 min, [iron] salt 0.25 mM.



Supplementary Figure 29. UV-vis monitorization of the oxidation of Fe(II) to Fe(III).

General procedure for Brønsted acid loading experiments: A 25 mL Schlenk flask equipped with a magnetic bar was charged with $\text{Fe}(\text{CF}_3\text{CO}_2)_3$ (78.8 mg, 0.2 mmol) and Selectfluor (106.3 mg, 0.3 mmol). The flask was evacuated and refilled with N_2 for three times. The vessel was then charged with extra dry CH_3CN (2.0 mL), alkene (0.2 mmol), oxydibenzene (17.0 mg, 0.1 mmol), isopropanol (24.0 mg, 0.4 mmol) and TfOH (0.2 mmol, 0.4 mmol, 0.6 mmol, 0.8 mmol, 1.0 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard.

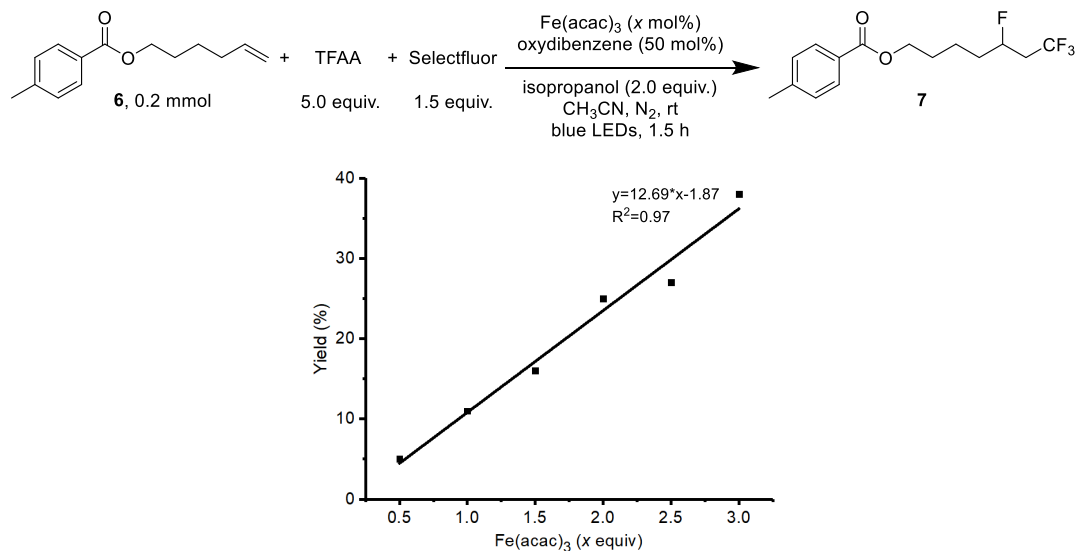


Supplementary Figure 30. Brønsted acid loading experiments.

General procedure for light on/off experiments: A 25 mL Schlenk flask equipped with a magnetic bar was charged with $\text{Fe}(\text{acac})_3$ (7.1 mg, 0.02 mmol) and Selectfluor (106.3 mg, 0.3 mmol). The flask was evacuated and refilled with N_2 for three times. The vessel was then charged with extra dry CH_3CN (2.0 mL), alkene (0.2 mmol), TFAA (210.0 mg, 1.0 mmol), oxydibenzene (17.0 mg, 0.1 mmol) and isopropanol (24.0 mg, 0.4 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard. Observed product 7 formation occurred only during periods of light irradiation, which ruled out a radical chain mechanism.

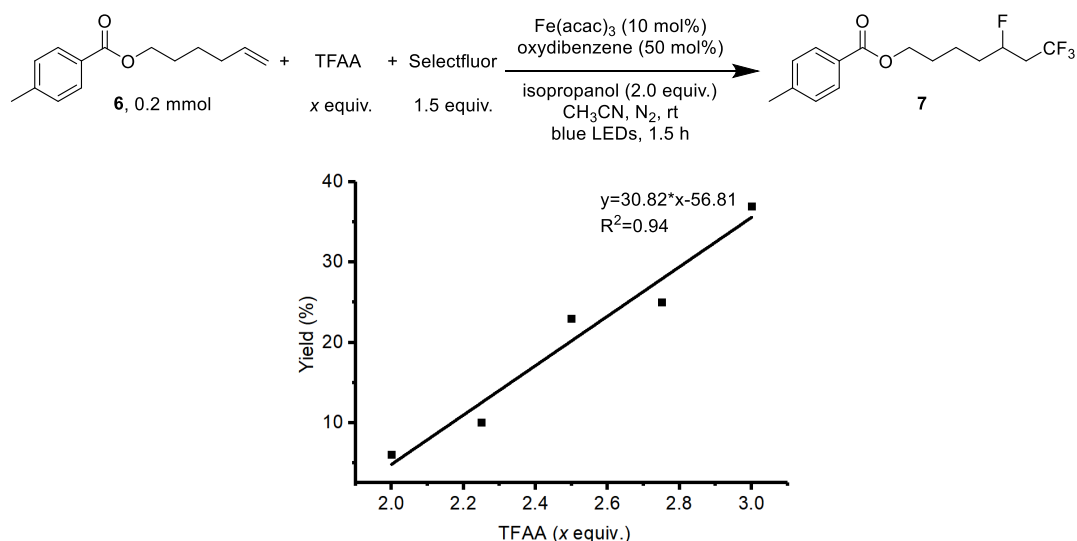
General procedure for kinetic curve of $\text{Fe}(\text{acac})_3$: A 25 mL Schlenk flask equipped with a magnetic bar was charged with $\text{Fe}(\text{acac})_3$ (0.001 mmol, 0.002 mmol, 0.003 mmol, 0.004 mmol, 0.005 mmol, 0.006 mmol) and Selectfluor (106.3 mg, 0.3 mmol). The flask was evacuated and

refilled with N₂ for three times. The vessel was then charged with extra dry CH₃CN (2.0 mL), alkene (0.2 mmol), TFAA (210.0 mg, 1.0 mmol), oxydibenzene (17.0 mg, 0.1 mmol) and isopropanol (24.0 mg, 0.4 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs for 1.5 h. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard.



Supplementary Figure 31. Kinetic studies for Fe(acac)₃.

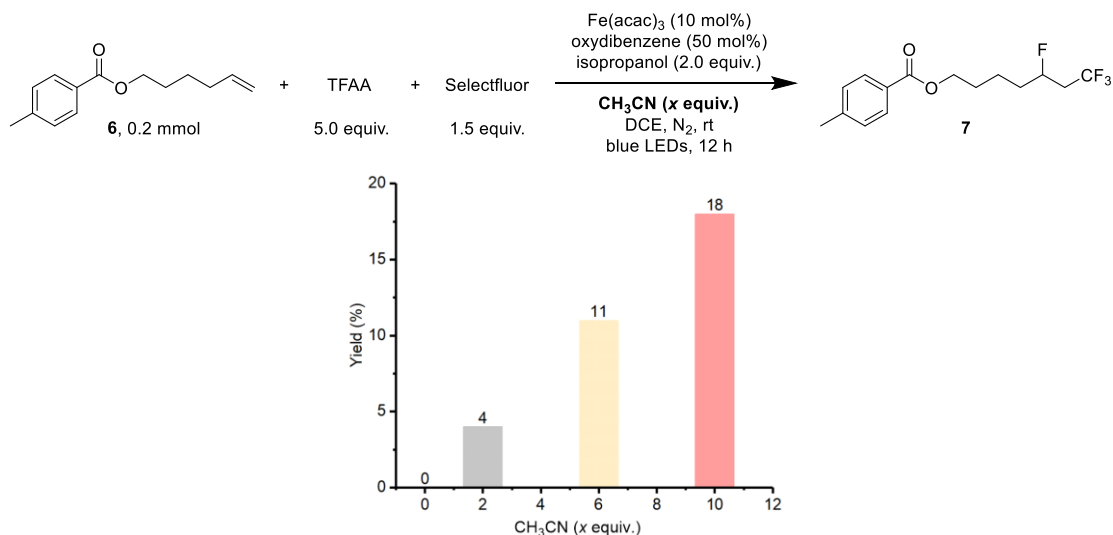
General procedure for kinetic curve of TFAA: A 25 mL Schlenk flask equipped with a magnetic bar was charged with Fe(acac)₃ (7.1 mg, 0.02 mmol) and Selectfluor (106.3 mg, 0.3 mmol). The flask was evacuated and refilled with N₂ for three times. The vessel was then charged with extra dry CH₃CN (2.0 mL), alkene (0.2 mmol), TFAA (0.4 mmol, 0.45 mmol, 0.5 mmol, 0.55 mmol, 0.6 mmol), oxydibenzene (17.0 mg, 0.1 mmol) and isopropanol (24.0 mg, 0.4 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs for 1.5 h. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard.



Supplementary Figure 32. Kinetic studies for TFAA.

General procedure for acetonitrile loading experiments: A 25 mL Schlenk flask equipped with a magnetic bar was charged with Fe(acac)₃ (7.1 mg, 0.02 mmol) and Selectfluor (106.3 mg, 0.3 mmol). The flask was evacuated and refilled with N₂ for three times. The vessel was then charged

with extra dry DCE (2.0 mL), alkene (0.2 mmol), TFAA (1.0 mmol), oxydibenzene (17.0 mg, 0.1 mmol), CH₃CN (0 mmol, 0.4 mmol, 0.12 mmol, 2.0 mmol) and isopropanol (24.0 mg, 0.4 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs for 12 h. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard.



Supplementary Figure 33. Acetonitrile loading experiments

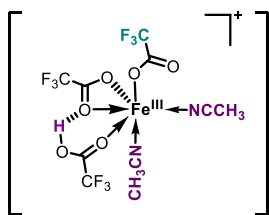
2.3 Density Functional Theory calculation

Computational Details

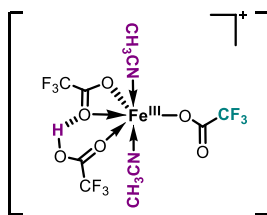
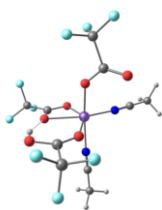
All the calculation have been performed using the Gaussian Program (Gaussian 16)⁵. The geometry optimization was conducted with the B3LYP functional and standard 6-31G(d, p) basis set. The solvent effect was considered by IEFPCM^{6,7}. Harmonic vibrational frequency calculations were performed for all stationary points to confirm whether they are local minima or transition state structures and to derive the thermochemical corrections for the enthalpies and free energies. Time-dependent density functional theory (TD-DFT) calculations were conducted at same level within the adiabatic approximation to predict the excitation energies^{8,9}.

The possible light-harvesting species in the reaction

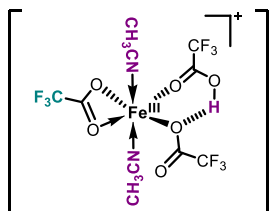
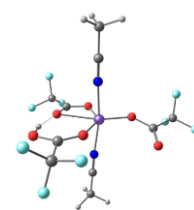
According to the experimental observations, the possible light-harvesting species was calculated by DFT. The free energy for **12** generation is lower than that of other possible structures in Supplementary Figure 34.



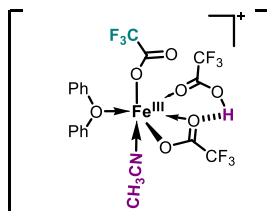
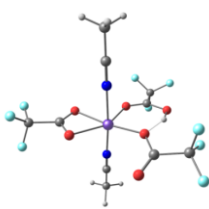
12 (0.0 kcal/mol)



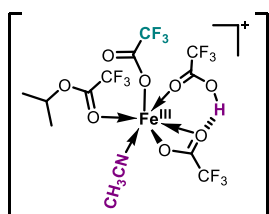
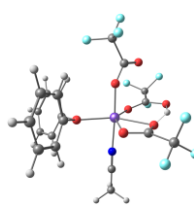
88 (0.3 kcal/mol)



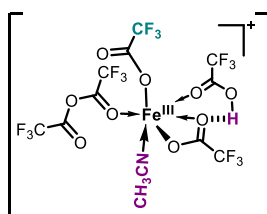
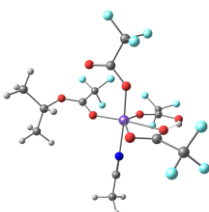
89 (1.2 kcal/mol)



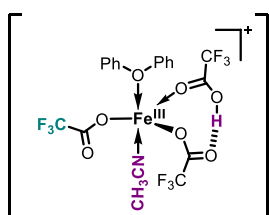
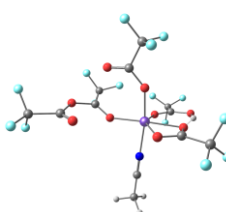
90 (11.2 kcal/mol)



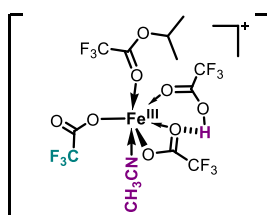
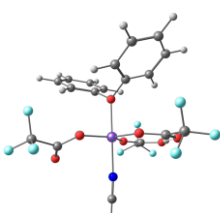
91 (7.0 kcal/mol)



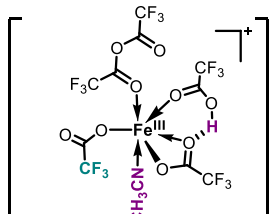
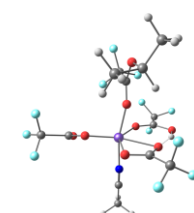
92 (11.0 kcal/mol)



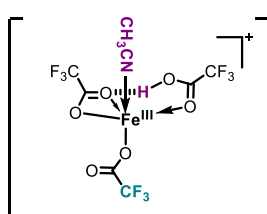
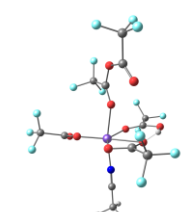
93 (7.7 kcal/mol)



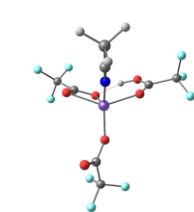
94 (7.6 kcal/mol)

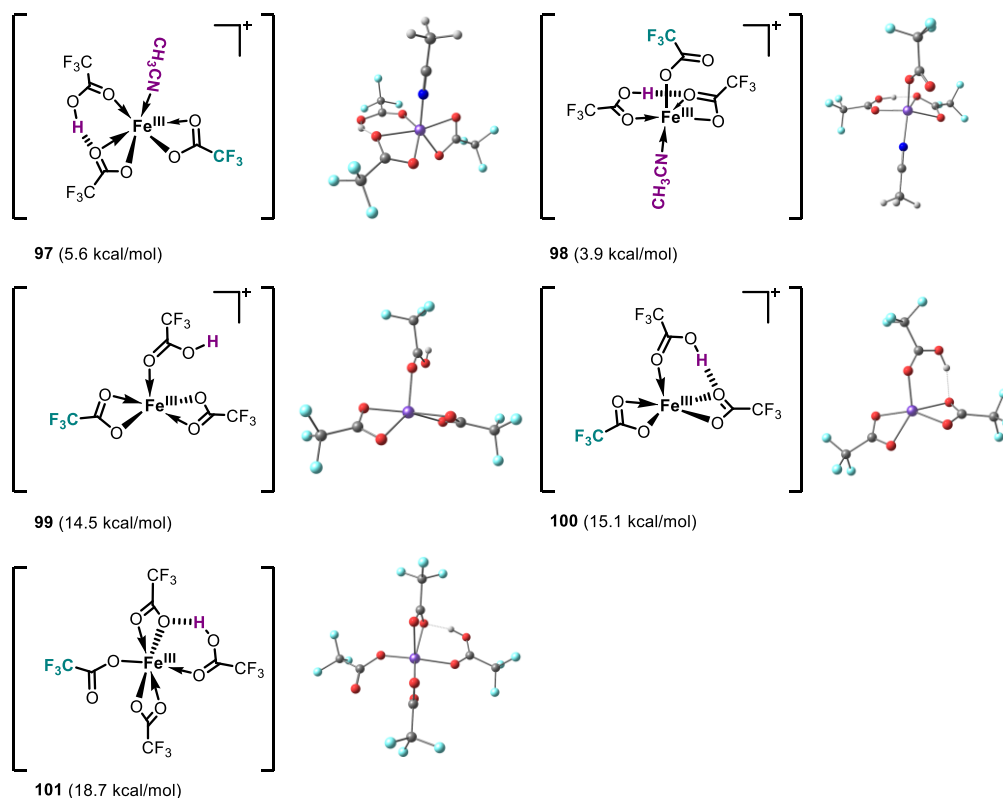


95 (12.5 kcal/mol)



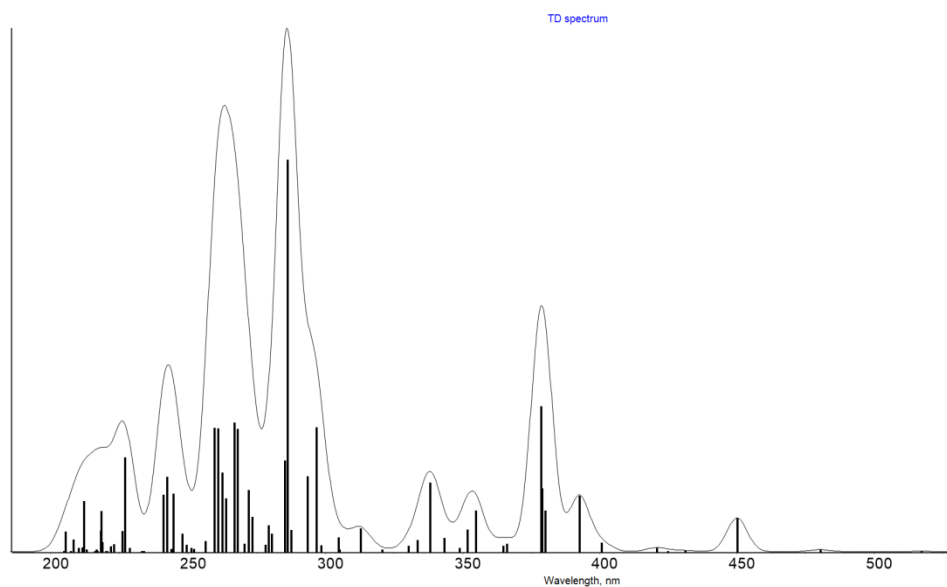
96 (3.3 kcal/mol)





Supplementary Figure 34. Computational investigations of iron and CF_3COO^- -based complexes.

Then, the TD-DFT calculations for the **12** were conducted to reveal the light-induced electron transfer process. As shown in the Supplementary Figure 35, 36, and 37, the absorption peak in 450.02 nm can be excited by our blue LEDs (λ : 404 – 499 nm).



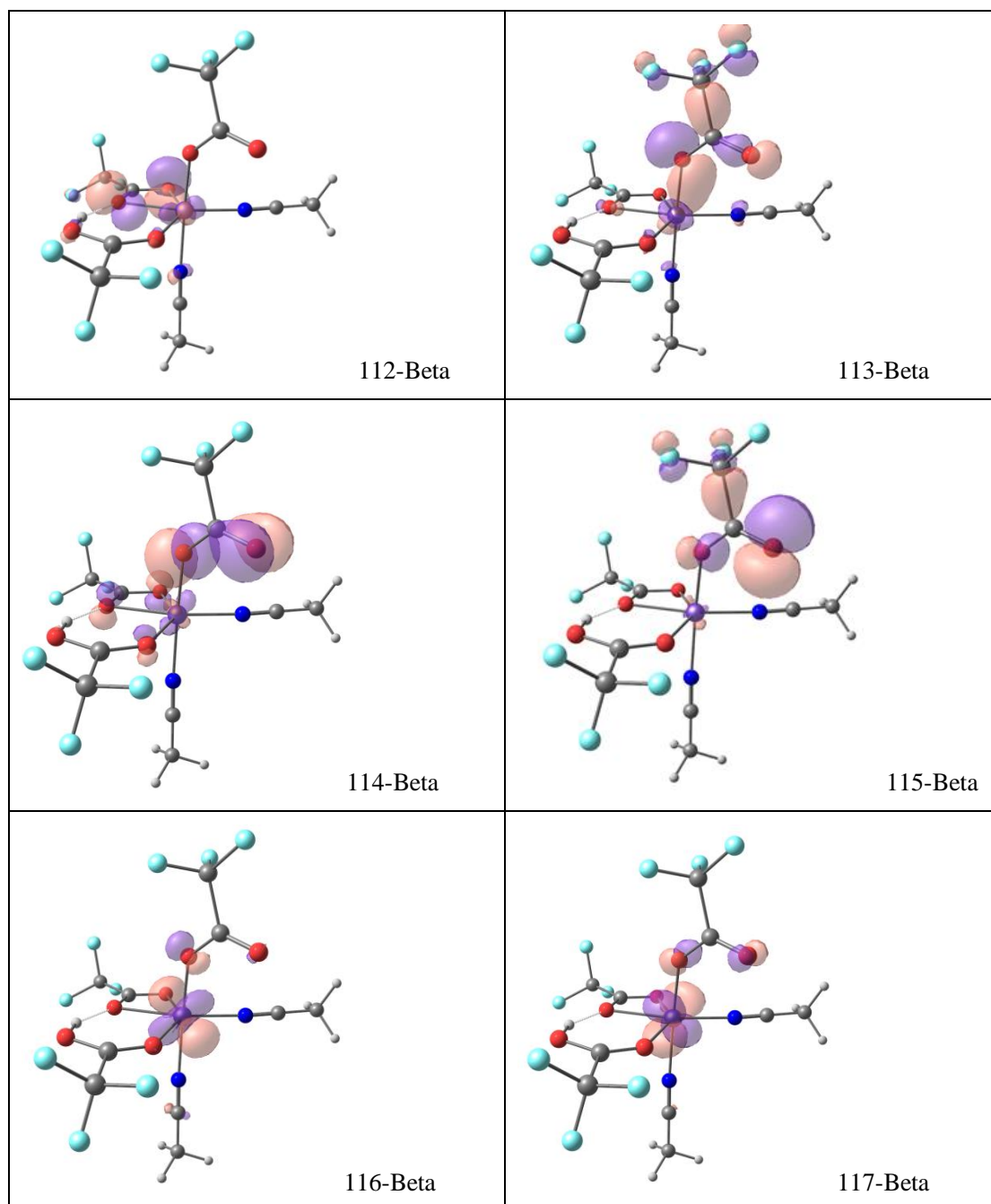
Supplementary Figure 35. UV-vis absorption predicted by the TD-DFT (**12**).

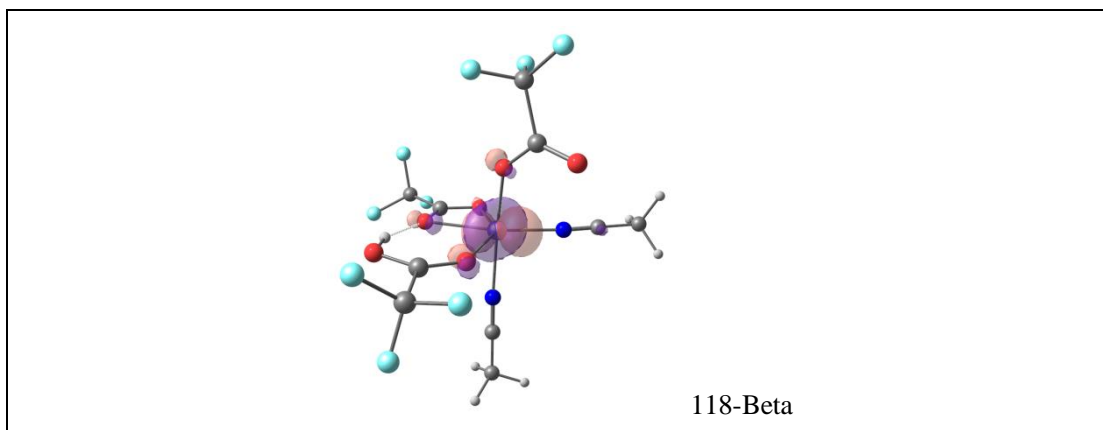
| transition | coefficient |
|-------------------|-------------|
| 112-Beta→117-Beta | 0.10650 |

| | |
|-------------------|----------|
| 113-Beta→116-Beta | -0.50854 |
| 113-Beta→118-Beta | 0.23940 |
| 114-Beta→117-Beta | 0.75302 |
| 115-Beta→116-Beta | -0.21293 |
| 115-Beta→118-Beta | 0.10251 |

Supplementary Figure 36. Major transition modes and coefficients (**12**).

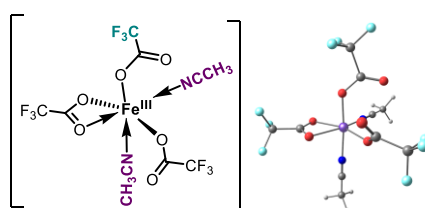
The orbital analysis shows that the electron in the trifluoroacetate of **12** could be excited to the iron center to achieve LMCT under our blue LEDs irradiation.



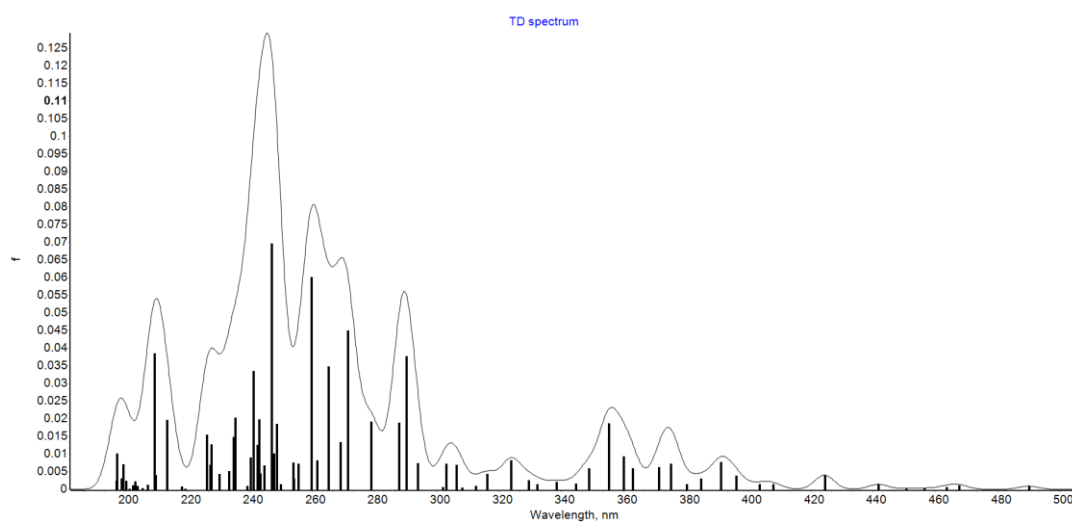


Supplementary Figure 37. Mentioned orbitals for the peak in 450.02 nm.

Light-harvesting species without H⁺



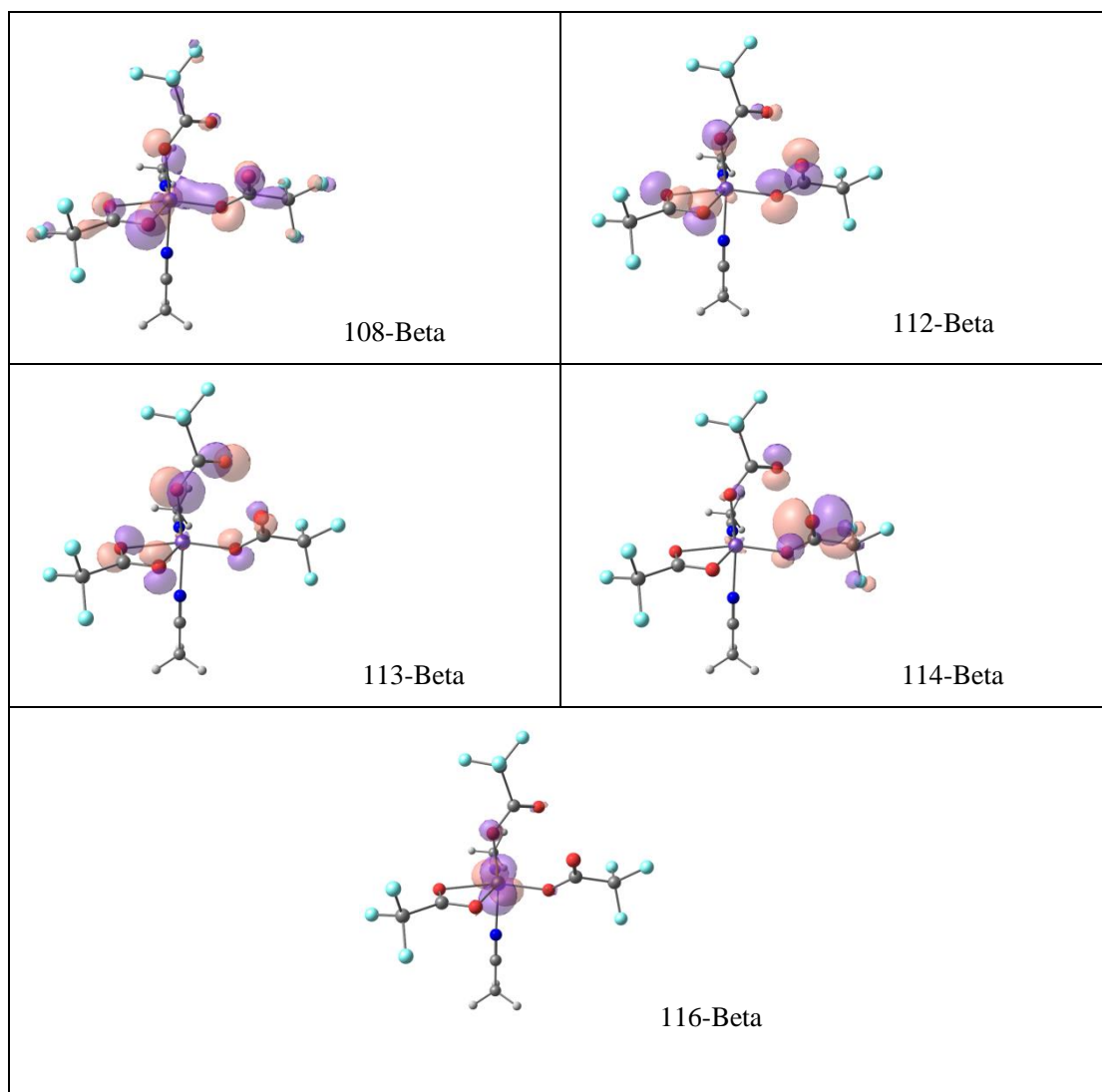
Supplementary Figure 38. 3D structure of 12'.



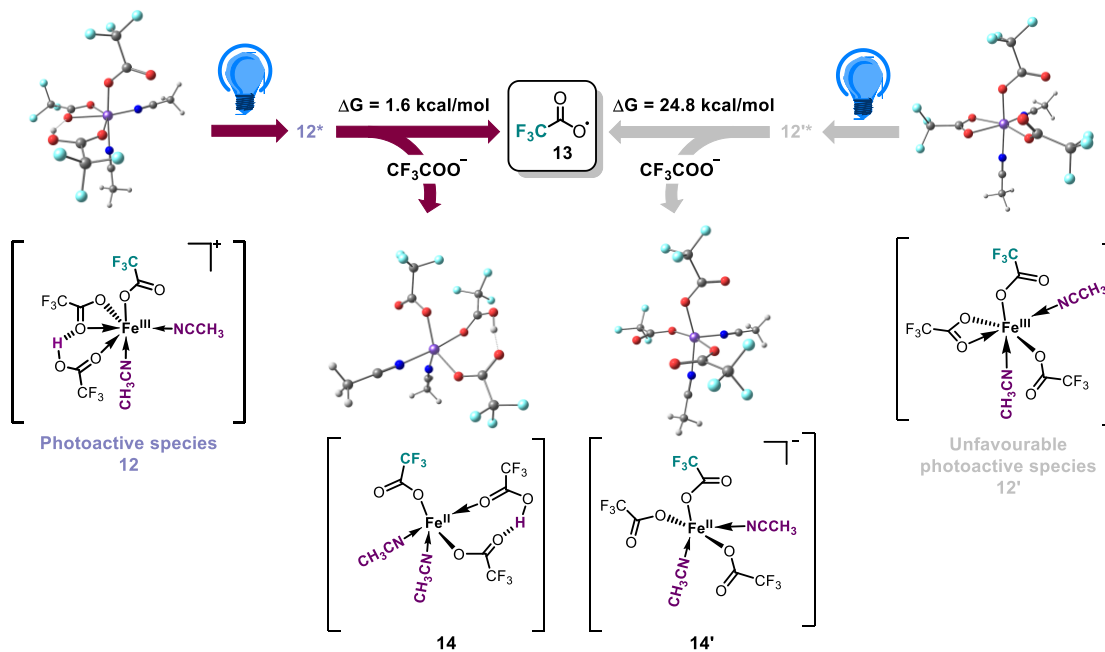
Supplementary Figure 39. UV-vis spectra predicted by the TD-DFT (12').

| transition | coefficient |
|-------------------|-------------|
| 108-Beta→116-Beta | 0.10802 |
| 112-Beta→116-Beta | -0.12540 |
| 113-Beta→116-Beta | 0.91231 |
| 114-Beta→116-Beta | -0.14856 |

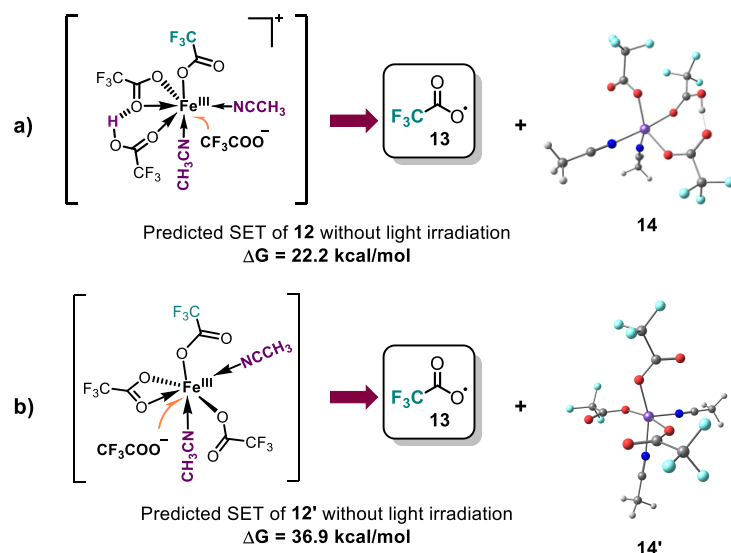
Supplementary Figure 40. Major transition modes and coefficients (12').



Supplementary Figure 41. Key orbitals of the most possible absorption peak (**12'**).



Supplementary Figure 42. Comparison between the **12** and **12'** under blue LEDs irradiation.



Supplementary Figure 43. SET comparison between **12** and **12'** without light irradiation.

We also compared the orbital energies of **12** and **12'** (Supplementary Figure 44). In the presence of intramolecular hydrogen bonding, the LUMO energy of iron significantly decreased by 0.86 eV, which is beneficial to the LMCT process.

| Species | E (arb. units) | Electron volt (eV) |
|-------------------|----------------|--------------------|
| 12 (LUMO) | -0.1892 | -5.1462 |
| 12' (LUMO) | -0.1575 | -4.2840 |

Supplementary Figure 44. LUMO energy comparison between **12** and **12'**.

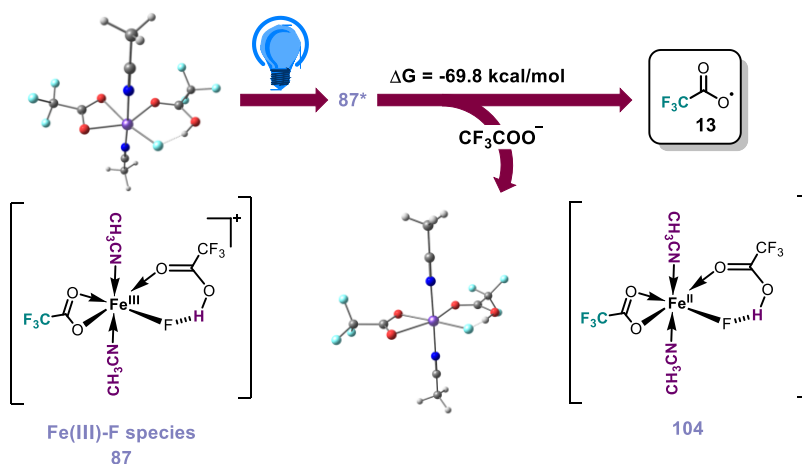
| Species | G (arb. units) |
|---------------------|---------------------|
| 12 (doublet) | -3108.130262 |
| 12 (quartet) | -3108.137950 |
| 12 (hextet) | -3108.161111 |
| 14 (singlet) | -3108.325434 |
| 14 (triplet) | -3108.350736 |
| 14 (quintet) | -3108.376685 |

Supplementary Figure 45. Spin states determination of **12** and **14**.

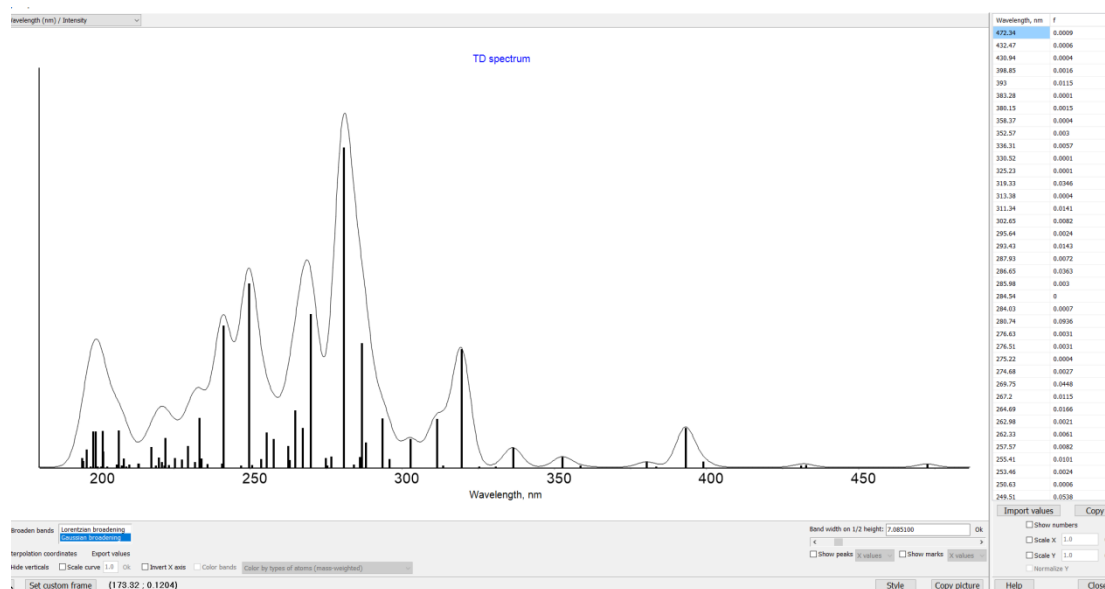
Light-harvesting species involved F^-

The formation of **87** requires the F^- generation from Selectfluor, while Selectfluor plays the role as an electrophilic fluorination reagent to capture the alkyl radical intermediate **15** in this system. Considering the fact that exogenous F^- shows the suppression effect for alkene fluorotrifluoromethylation (Supplementary Figure 51), we proposed that the concentration of F^- under the standard condition should be low. Although **87** was detected by ESI-HRMS spectrometry and DFT calculations also demonstrated its feasibility of generating CF_3 radical under blue light irradiation (Supplementary Figure 46), the low concentration of F^- in the process of reaction may indicate that **87** is not the first choice in comparison to **12**, but alternative. To further understand the role of possible species **87**, this protocol was extended to the trifluoromethylation of arene though directly replacing Selectfluor into sodium persulfate. As shown in Supplementary Figure 50, even

in the absence of fluorine source, this protocol is still feasible to produce CF₃ radical, while the exogenous F⁻ do not promote the trifluoromethylation of arene. It indicates that **87** is not necessary for this protocol and **12** is the first choice for effective iron-based light-harvesting species under this standard condition.



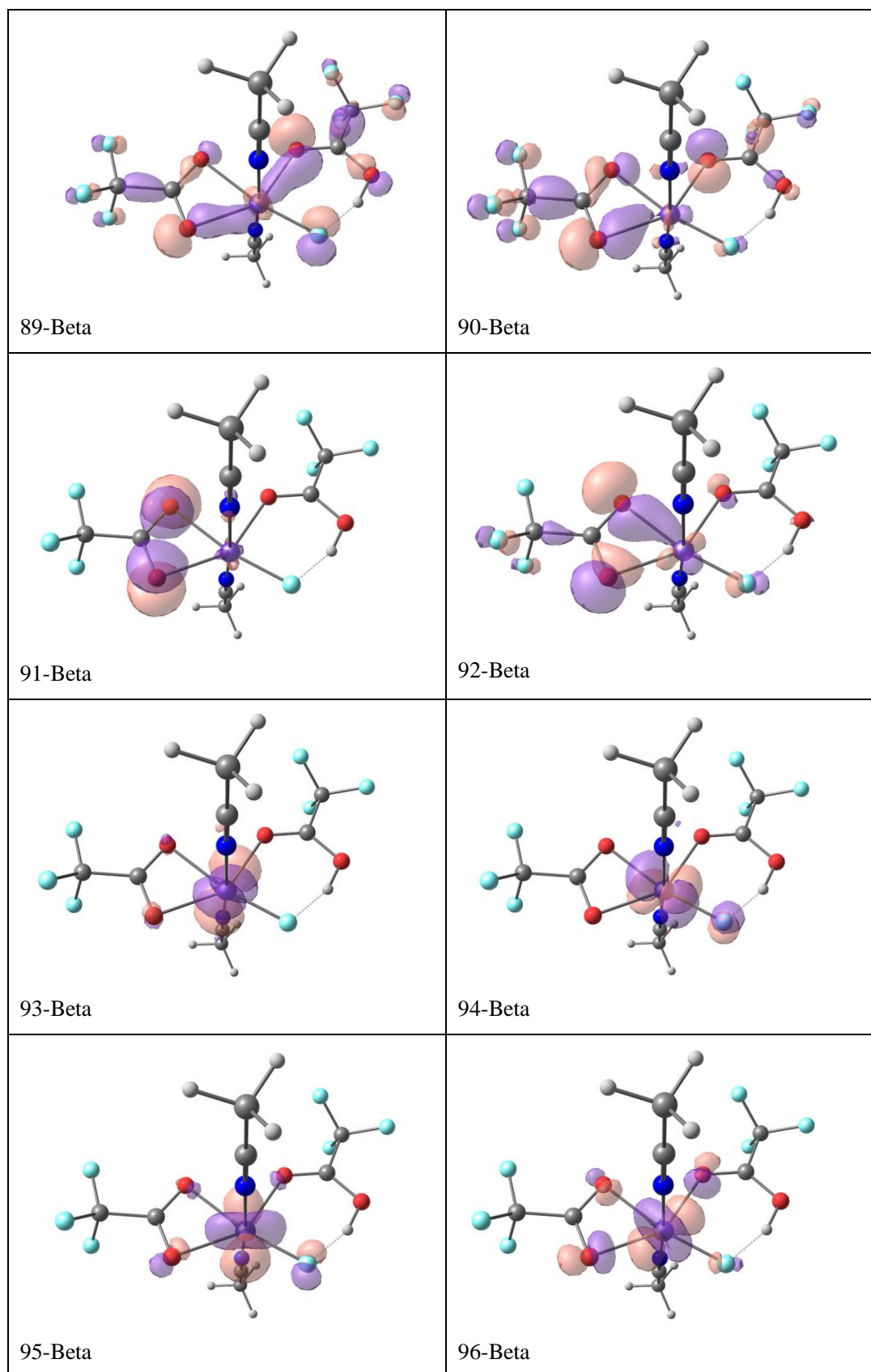
Supplementary Figure 46. DFT calculation of **87**.



Supplementary Figure 47. UV-vis spectra predicted by the TD-DFT (**87**).

| transition | coefficient |
|-----------------|-------------|
| 89-Beta→93-Beta | -0.18964 |
| 90-Beta→93-Beta | -0.15313 |
| 91-Beta→93-Beta | -0.26813 |
| 92-Beta→93-Beta | 0.92180 |

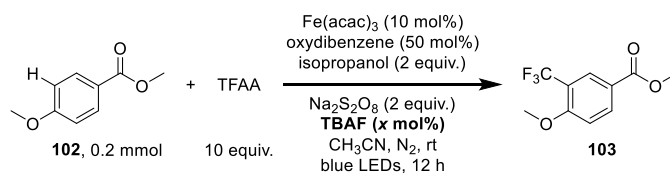
Supplementary Figure 48. Major transition modes and coefficients (**87**).



Supplementary Figure 49. Key orbitals of the most possible absorption peak (**87**).

General procedure for trifluoromethylation of arenes: A 25 mL Schlenk flask equipped with a magnetic bar was charged with $\text{Fe}(\text{acac})_3$ (0.02 mmol), $\text{Na}_2\text{S}_2\text{O}_8$ (0.4 mmol) and methyl anisate (102, 0.2 mmol). The flask was evacuated and refilled with N_2 for three times. The vessel was then

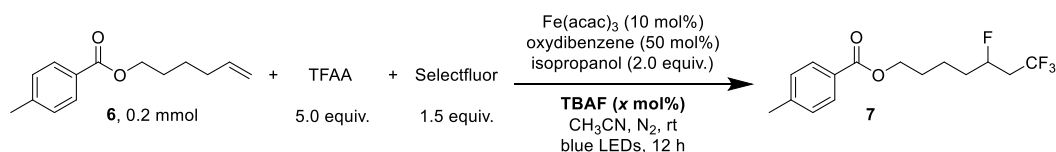
charged with extra dry CH₃CN (2.0 mL), oxydibenzene (17.0 mg, 0.1 mmol), isopropanol (24.0 mg, 0.4 mmol), TBAF (0 mmol, 0.08 mmol, 0.2 mmol) and TFAA (2.0 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs 12 h. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. Yields were determined by GC analysis of the crude reaction mixture, using naphthalene as an internal standard.



| Entry | TBAF (x mol%) | 103, Yield |
|-------|---------------|------------|
| 1 | 0 | 22% |
| 2 | 40 | 9% |
| 3 | 100 | 6% |

Supplementary Table 7. Investigation of exogenous F⁻ effect for trifluoromethylation of arenes.

General procedure for the fluorotrifluoromethylation of alkenes involved fluorion: A 25 mL Schlenk flask equipped with a magnetic bar was charged with Fe(acac)₃ (0.02 mmol), Selectfluor (0.3 mmol). The flask was evacuated and refilled with N₂ for three times. The vessel was then charged with extra dry CH₃CN (2.0 mL), oxydibenzene (17.0 mg, 0.1 mmol), isopropanol (24.0 mg, 0.4 mmol), TBAF (0 mmol, 0.1 mmol, 0.2 mmol) and TFAA (1.0 mmol). The reaction mixture was stirred under nitrogen atmosphere and irradiated by blue LEDs 12 h. After completion of the reaction. Then the reaction system was quenched by triethylamine, diluted with EtOAc. Yields were determined by GC analysis of the crude reaction mixture, using biphenyl as an internal standard.



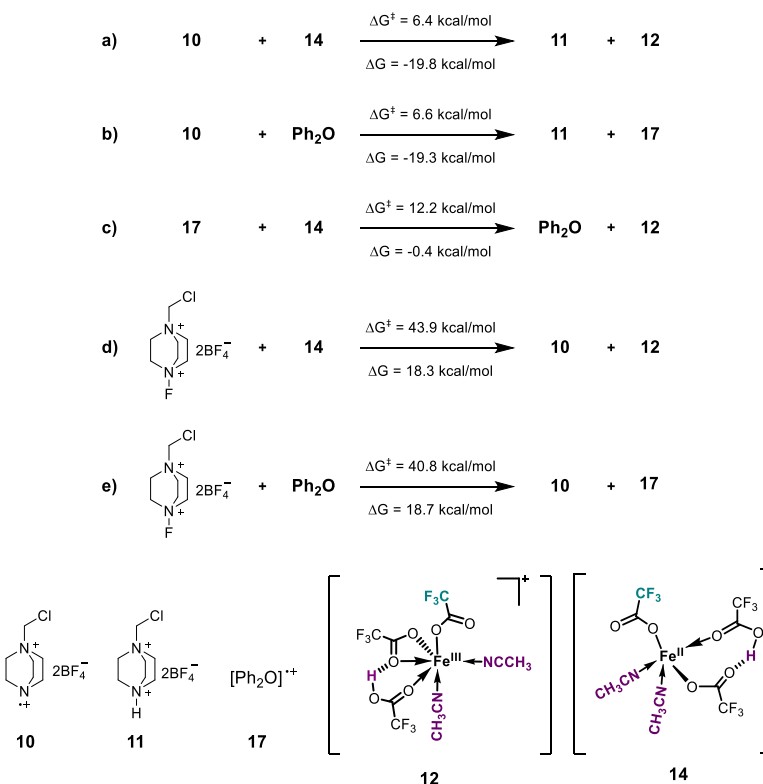
| Entry | TBAF (x mol%) | 7, Yield |
|-------|---------------|----------|
| 1 | / | 70% |
| 2 | 50 | 29% |
| 3 | 100 | N.D. |

Supplementary Table 8. Investigation of exogenous F⁻ effect for fluorotrifluoromethylation of alkenes.

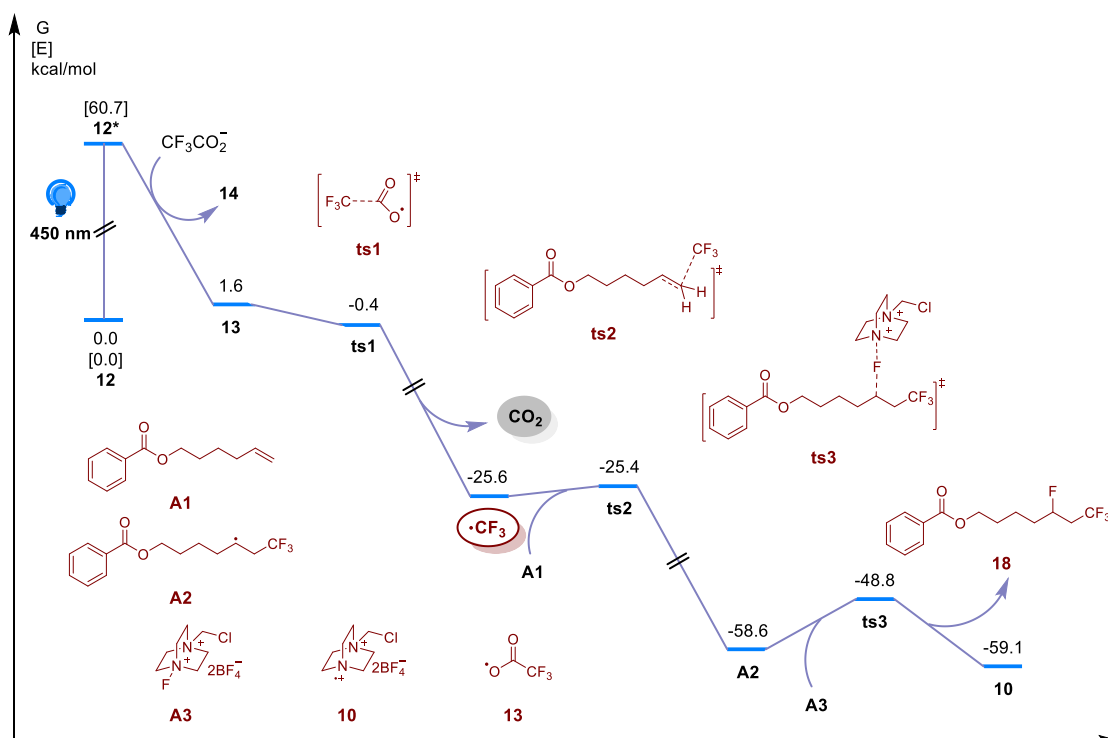
| Species | E (arb. units) |
|----------------------------|----------------|
| Selectfluor | -844.45939 |
| [Selectfluor] ⁺ | -844.70564 |
| Ph ₂ O | -538.53302 |
| 17 | -538.31759 |
| 12 | -3108.27344 |
| 14 | -3108.48845 |

| | |
|------------|-------------|
| 12' | -3107.85594 |
| 14' | -3108.02774 |

Supplementary Figure 50. Calculation for the possible single electron transfer.



Supplementary Figure 51. DFT calculation for SET process.

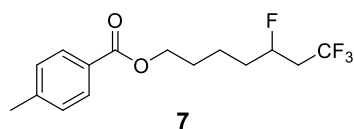


Supplementary Figure 52. Free energy profile for the generation of the product.

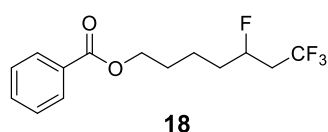
| Species | Gibbs free energies (arb. units) | Species | Gibbs free energies (arb. units) |
|---|----------------------------------|--|----------------------------------|
| 12 | -3108.161111 | 14 | -3108.376568 |
| 12' | -3107.757221 | 14' | -3107.935688 |
| 87 | -2681.839651 | 104 | -2682.16889 |
| 88 | -3108.160518 | 89 | -3108.159142 |
| 90 | -3513.783164 | 91 | -3620.063222 |
| 92 | -3952.512065 | 93 | -3513.788757 |
| 94 | -3620.062247 | 95 | -3952.509614 |
| 96 | -2975.409546 | 97 | -2975.410447 |
| 98 | -2975.406744 | 99 | -2842.58805 |
| 100 | -2842.64628 | 101 | -3369.04626 |
| CF₃CO₂⁻ | -526.333578 | 13 | -526.115524 |
| MeCN | -132.745395 | CF₃CO₂iPr | -644.658704 |
| Ph₂O | -538.385279 | (CF₃CO)₂O | -977.113857 |
| ts1 | -526.118672 | | |

Supplementary Figure 53. Corresponding free energies of the key species.

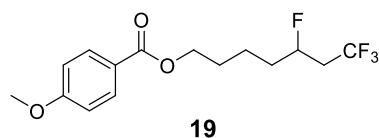
3. Supplementary Notes



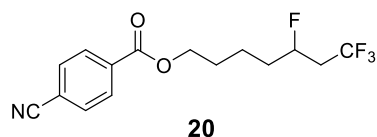
5,7,7,7-Tetrafluoroheptyl 4-methylbenzoate (7): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (42.7 mg) at 70% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.92 (d, J = 8.2 Hz, 2H), 7.23 (d, J = 8.0 Hz, 2H), 4.82 (dt, J = 48.7, 8.2, 3.3 Hz, 1H), 4.32 (t, J = 6.4 Hz, 2H), 2.62 – 2.19 (m, 2H), 2.40 (s, 3H), 1.88 – 1.48 (m, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 166.75, 143.73, 129.65, 129.19, 127.64, 125.54 (dq, J = 3.4, 276.8 Hz), 87.51 (dq, J = 172.4, 3.2 Hz), 64.39, 39.46 (qd, J = 28.5, 22.9 Hz), 34.71 (d, J = 20.8 Hz), 28.45, 21.71, 21.47 (d, J = 4.1 Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.12 (d, J = 7.6 Hz, 3F), -182.40 (q, J = 7.8 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{18}\text{F}_4\text{O}_2\text{Na}^+]$: 329.1135, found 329.1139.



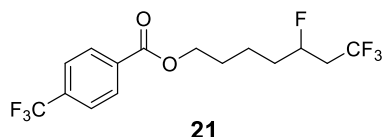
5,7,7,7-Tetrafluoroheptyl benzoate (18): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (29.2 mg) at 50% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.06 – 8.02 (m, 2H), 7.59 – 7.41 (m, 3H), 4.83 (dt, J = 48.8, 8.1, 3.5 Hz, 1H), 4.34 (t, J = 6.4 Hz, 2H), 2.63 – 2.20 (m, 2H), 1.88 – 1.54 (m, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 166.71, 133.07, 130.42, 129.65, 128.50, 125.54 (dq, J = 3.6, 276.8 Hz), 87.53 (dq, J = 172.5, 3.2 Hz), 64.60, 39.54 (qd, J = 28.5, 23.0 Hz), 34.75 (d, J = 20.9 Hz), 28.47, 21.50 (d, J = 4.1 Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.11 (d, J = 7.4 Hz, 3F), -182.39 (q, J = 7.4 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{14}\text{H}_{16}\text{F}_4\text{O}_2\text{Na}^+]$: 315.0979, found 315.0981.



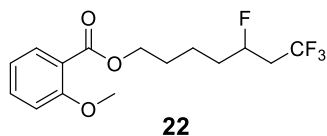
5,7,7,7-Tetrafluoroheptyl 4-methoxybenzoate (19): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (43.8 mg) at 68% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.99 – 7.95 (m, 2H), 6.93 – 6.80 (m, 2H), 4.82 (dt, J = 49.0, 8.2, 3.5 Hz, 1H), 4.30 (t, J = 6.4 Hz, 2H), 3.84 (s, 3H), 2.62 – 2.19 (m, 2H), 1.89 – 1.47 (m, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 166.40, 163.47, 131.62, 125.55 (dq, J = 3.4, 276.8 Hz), 122.80, 113.71, 87.51 (dq, J = 172.4, 3.2 Hz), 64.24, 55.47, 39.45 (qd, J = 28.5, 22.9 Hz), 34.70 (d, J = 20.9 Hz), 28.48, 21.46 (d, J = 4.1 Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.14 (d, J = 7.5 Hz, 3F), -182.44 (q, J = 7.4 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{18}\text{F}_4\text{O}_3\text{Na}^+]$: 345.1084, found 345.1092.



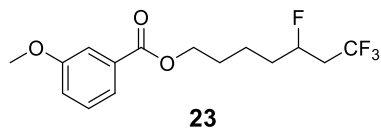
5,7,7,7-Tetrafluoroheptyl 4-cyanobenzoate (20): The compound was purified by column chromatography (petroleum ether/EtOAc = 30/1), resulting in a colorless liquid (39.3 mg) at 62% yield. $^1\text{H NMR}$ (400 MHz, DMSO-*d*₆) δ 8.11 – 7.95 (m, 4H), 4.95 – 4.73 (m, 1H), 4.30 (t, J = 6.4 Hz, 2H), 2.78 – 2.57 (m, 2H), 1.81 – 1.63 (m, 4H), 1.62 – 1.43 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, DMSO-*d*₆) δ 164.51, 133.72, 132.75, 129.76, 126.18 (q, J = 277.0 Hz), 118.03, 115.48, 87.90 (dq, J = 169.9, 3.2 Hz), 65.15, 37.91 (qd, J = 27.3, 21.1 Hz), 33.61 (d, J = 20.2 Hz), 27.69, 20.73 (d, J = 4.3 Hz). $^{19}\text{F NMR}$ (376 MHz, DMSO-*d*₆) δ -62.62 (d, J = 7.6 Hz, 3F), -182.36 (q, J = 7.6 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{15}\text{F}_4\text{NO}_2\text{Na}^+]$: 340.0931, found 340.0943.



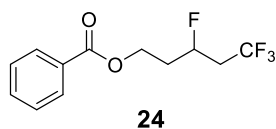
5,7,7,7-Tetrafluoroheptyl 4-(trifluoromethyl)benzoate (21): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (43.2 mg) at 60% yield. $^1\text{H NMR}$ (400 MHz, DMSO- d_6) δ 8.17 – 7.85 (m, 4H), 4.94 – 4.73 (m, 1H), 4.32 (t, J = 6.4 Hz, 2H), 2.78 – 2.58 (m, 2H), 1.82 – 1.63 (m, 4H), 1.62 – 1.43 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, DMSO- d_6) δ 164.67, 133.55 (d, J = 0.9 Hz), 132.86 (q, J = 32.0 Hz), 129.96, 126.19 (q, J = 277.0 Hz), 125.73 (q, J = 3.7 Hz), 123.72 (q, J = 272.7 Hz), 87.91 (dq, J = 169.8, 3.3 Hz), 65.03, 37.90 (qd, J = 27.3, 21.0 Hz), 33.60 (d, J = 20.0 Hz), 27.71, 20.73 (d, J = 4.2 Hz). $^{19}\text{F NMR}$ (376 MHz, DMSO- d_6) δ -61.84 (s, 3F), -62.61 (d, J = 7.7 Hz, 3F), -182.40 (q, J = 7.6 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{15}\text{F}_7\text{O}_2\text{Na}^+]$: 383.0852, found 383.0859.



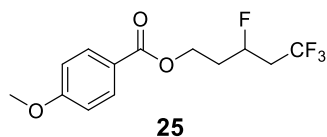
5,7,7,7-Tetrafluoroheptyl 2-methoxybenzoate (22): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (43.8 mg) at 68% yield. $^1\text{H NMR}$ (400 MHz, Chloroform- d) δ 7.79 – 7.75 (m, 1H), 7.5 – 7.43 (m, 1H), 7.01 – 6.94 (m, 2H), 4.92 – 4.72 (m, 1H), 4.31 (t, J = 6.3 Hz, 2H), 3.89 (s, 3H), 2.61 – 2.21 (m, 2H), 1.88 – 1.51 (m, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform- d) δ 166.44, 159.26, 133.60, 131.63, 125.55 (dq, J = 3.5, 276.9 Hz), 120.39, 120.26, 112.16, 87.57 (dq, J = 172.4, 3.2 Hz), 64.41, 56.04, 39.54 (qd, J = 28.7, 23.3 Hz), 34.73 (d, J = 20.8 Hz), 28.44, 21.46 (d, J = 4.2 Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform- d) δ -64.13 (d, J = 7.6 Hz, 3F), -182.23 – -182.41 (m, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{18}\text{F}_4\text{O}_3\text{Na}^+]$: 345.1084, found 345.1081.



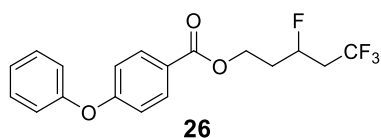
5,7,7,7-Tetrafluoroheptyl 3-methoxybenzoate (23): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (42.5 mg) at 66% yield. $^1\text{H NMR}$ (400 MHz, Chloroform- d) δ 7.64 – 7.53 (m, 2H), 7.36 – 7.30 (m, 1H), 7.11 – 7.06 (m, 1H), 4.91 – 4.71 (m, 1H), 4.32 (t, J = 6.5 Hz, 2H), 3.82 (s, 3H), 2.59 – 2.19 (m, 2H), 1.87 – 1.50 (m, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform- d) δ 166.45, 159.65, 131.67, 129.45, 125.54 (dq, J = 3.3, 276.8 Hz), 121.91, 119.27, 114.23, 87.45 (dq, J = 172.5, 3.2 Hz), 64.60, 55.38, 39.36 (qd, J = 28.5, 22.8 Hz), 34.62 (d, J = 20.9 Hz), 28.35, 21.38 (d, J = 4.1 Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform- d) δ -64.15 (d, J = 6.9 Hz, 3F), -182.50 (q, J = 7.5 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{18}\text{F}_4\text{O}_3\text{Na}^+]$: 345.1084, found 383.1089.



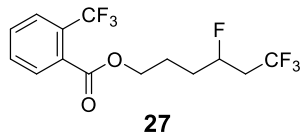
3,5,5,5-Tetrafluoropentyl benzoate (24): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (27.5 mg) at 52% yield. $^1\text{H NMR}$ (400 MHz, DMSO- d_6) δ 7.98 (d, J = 7.4 Hz, 2H), 7.69 – 7.62 (m, 1H), 7.56 – 7.50 (m, 2H), 5.16 – 4.94 (m, 1H), 4.48 – 4.34 (m, 2H), 2.92 – 2.69 (m, 2H), 2.22 – 2.07 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, DMSO- d_6) δ 165.65, 133.41, 129.66, 129.19, 128.78, 126.15 (q, J = 277.0 Hz), 85.76 (dq, J = 170.4, 3.4 Hz), 60.49 (d, J = 4.7 Hz), 37.85 (qd, J = 27.5, 20.6 Hz), 33.25 (d, J = 20.2 Hz). $^{19}\text{F NMR}$ (376 MHz, DMSO- d_6) δ -62.54 (d, J = 7.8 Hz, 3F), -184.51 (q, J = 8.2 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{12}\text{H}_{12}\text{F}_4\text{O}_2\text{Na}^+]$: 287.0666, found 287.0661.



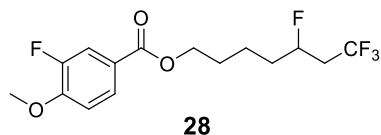
3,5,5,5-Tetrafluoropentyl 4-methoxybenzoate (25): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (41.2 mg) at 70% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.01 – 7.95 (m, 2H), 6.95 – 6.90 (m, 2H), 5.15 – 4.94 (m, 1H), 4.54 – 4.37 (m, 2H), 3.85 (s, 3H), 2.72 – 2.29 (m, 2H), 2.21 – 2.01 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 166.17, 163.69, 131.73, 125.37 (dq, $J = 3.6$, 276.8 Hz), 122.36, 113.85, 85.02 (dq, $J = 173.3$, 3.3 Hz), 59.98 (d, $J = 4.8$ Hz), 55.56, 39.57 (qd, $J = 28.8$, 22.7 Hz), 34.44 (d, $J = 20.9$ Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.02 (d, $J = 7.4$ Hz, 3F), -184.46 (q, $J = 7.4$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{13}\text{H}_{14}\text{F}_4\text{O}_3\text{Na}^+]$: 317.0771, found 317.0775.



3,5,5,5-Tetrafluoropentyl 4-phenoxybenzoate (26): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (58.4 mg) at 82% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.00 (d, $J = 8.8$ Hz, 2H), 7.42 – 7.36 (m, 2H), 7.23 – 6.97 (m, 5H), 5.04 (dt, $J = 48.8$, 8.2, 3.6 Hz, 1H), 4.54 – 4.41 (m, 2H), 2.71 – 2.31 (m, 2H), 2.20 – 2.03 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 165.91, 162.21, 155.64, 131.81, 130.17, 125.35 (dq, $J = 3.6$, 276.8 Hz), 124.72, 124.19, 120.27, 117.45, 84.98 (dq, $J = 173.5$, 3.3 Hz), 60.16 (d, $J = 4.7$ Hz), 39.56 (qd, $J = 28.8$, 22.7 Hz), 34.40 (d, $J = 20.9$ Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -63.98 (d, $J = 7.5$ Hz, 3F), -184.48 (q, $J = 7.7$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{18}\text{H}_{16}\text{F}_4\text{O}_3\text{H}^+]$: 357.1108, found 357.1111.

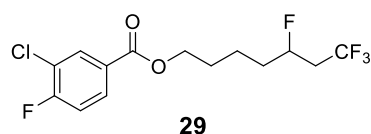


4,6,6,6-Tetrafluorohexyl 2-(trifluoromethyl)benzoate (27): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (44.5 mg) at 65% yield. $^1\text{H NMR}$ (400 MHz, DMSO-*d*₆) δ 7.89 – 7.74 (m, 4H), 5.00 – 4.78 (m, 1H), 4.32 (t, $J = 5.5$ Hz, 2H), 2.81 – 2.59 (m, 2H), 1.92 – 1.68 (m, 4H). $^{13}\text{C NMR}$ (101 MHz, DMSO-*d*₆) δ 166.27, 132.86, 131.86, 130.86 (q, $J = 2.0$ Hz), 129.99, 126.74 (q, $J = 31.8$ Hz), 126.71 (q, $J = 5.2$ Hz), 126.20 (q, $J = 277.8$ Hz), 124.85 (q, $J = 273.7$ Hz), 87.79 (dq, $J = 169.9$, 3.3 Hz), 65.34, 37.87 (qd, $J = 27.4$, 21.0 Hz), 30.63 (d, $J = 20.4$ Hz), 23.45 (d, $J = 4.1$ Hz). $^{19}\text{F NMR}$ (376 MHz, DMSO-*d*₆) δ -58.28 (s, 3F), -62.60 (d, $J = 7.7$ Hz, 3F), -182.74 (q, $J = 7.6$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{14}\text{H}_{13}\text{F}_7\text{O}_2\text{H}^+]$: 369.0696, found 369.0706.

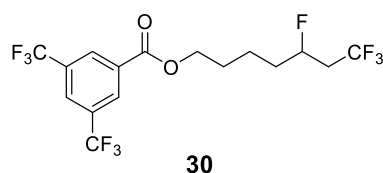


5,7,7,7-Tetrafluoroheptyl 3-fluoro-4-methoxybenzoate (28): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (41.5 mg) at 61% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.78 – 7.75 (m, 1H), 7.70 – 7.65 (m, 1H), 6.97 – 6.91 (m, 1H), 4.89 – 4.69 (m, 1H), 4.27 (t, $J = 6.4$ Hz, 2H), 3.90 – 3.87 (m, 3H), 2.59 – 2.17 (m, 2H), 1.84 – 1.47 (m, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 165.57 (d, $J = 2.8$ Hz), 151.83 (d, $J = 10.7$ Hz), 151.80 (d, $J = 246.8$ Hz), 126.75 (d, $J = 3.4$ Hz), 125.52 (dq, $J = 3.7$, 276.8 Hz), 123.13 (d, $J = 6.3$ Hz), 117.27 (d, $J = 19.6$ Hz), 112.45 (d, $J = 2.0$ Hz), 87.52 (dq, $J = 172.5$, 3.1 Hz), 64.68, 56.37, 39.53 (qd, $J = 28.6$, 23.1 Hz), 34.74 (d, $J = 20.9$ Hz), 28.46, 21.48 (d, $J = 4.1$ Hz). $^{19}\text{F NMR}$

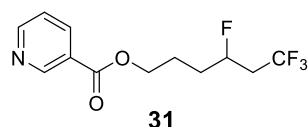
(376 MHz, Chloroform-*d*) δ -64.10 (d, J = 7.5 Hz, 3F), -134.65 (s, 1F), -182.40 (q, J = 7.9 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{17}\text{F}_5\text{O}_3\text{Na}^+]$: 363.0990, found 363.0997.



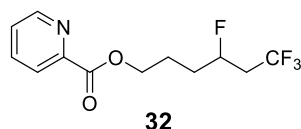
5,7,7,7-Tetrafluoroheptyl 3-chloro-4-fluorobenzoate (29): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (39.9 mg) at 58% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.10 – 8.05 (m, 1H), 7.95 – 7.89 (m, 1H), 7.22 – 7.15 (m, 1H), 4.83 (dt, J = 48.7, 8.3, 3.3 Hz, 1H), 4.33 (t, J = 6.5 Hz, 2H), 2.62 – 2.20 (m, 2H), 1.89 – 1.50 (m, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 164.67 (d, J = 0.5 Hz), 161.16 (d, J = 256.1 Hz), 132.42 (d, J = 1.1 Hz), 130.05 (d, J = 8.6 Hz), 127.61 (d, J = 3.9 Hz), 125.52 (dq, J = 3.7, 276.8 Hz), 121.55 (d, J = 18.3 Hz), 116.75 (d, J = 21.8 Hz), 87.48 (dq, J = 172.6, 3.5 Hz), 65.15, 39.49 (qd, J = 28.8, 22.8 Hz), 34.69 (d, J = 21.1 Hz), 28.37, 21.43 (d, J = 4.4 Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.14 (d, J = 7.5 Hz, 3F), -108.13 (s, 1F), -182.54 (q, J = 7.8 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{14}\text{H}_{14}\text{ClF}_5\text{O}_2\text{Na}^+]$: 367.0495, found 367.0502.



5,7,7,7-Tetrafluoroheptyl 3,4-bis(trifluoromethyl)benzoate (30): The compound was purified by column chromatography (petroleum ether/EtOAc = 50/1), resulting in a colorless liquid (52.2 mg) at 61% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.47 (s, 2H), 8.06 (s, 1H), 4.83 (dt, J = 48.9, 8.0, 3.6 Hz, 1H), 4.42 (t, J = 6.6 Hz, 2H), 2.63 – 2.22 (m, 2H), 1.94 – 1.54 (m, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 164.06, 132.57, 132.38 (q, J = 34.0 Hz), 129.94 – 129.75 (m), 126.61 – 126.31 (m), 125.49 (dq, J = 4.0, 276.7 Hz), 123.02 (q, J = 272.9 Hz), 87.46 (dq, J = 172.6, 3.1 Hz), 65.90, 39.59 (qd, J = 28.6, 23.2 Hz), 34.72 (d, J = 20.9 Hz), 28.37, 21.45 (d, J = 4.0 Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -63.03 (s, 3F), -63.04 (s, 3F), -64.16 (t, J = 7.1 Hz, 3F), -182.59 (q, J = 8.0 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{16}\text{H}_{14}\text{F}_{10}\text{O}_2\text{Na}^+]$: 451.0726, found 451.0739.

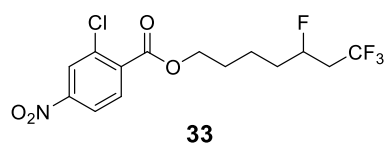


5,7,7,7-Tetrafluoroheptyl nicotinate (31): The compound was purified by column chromatography (petroleum ether/EtOAc = 30/1), resulting in a colorless liquid (41.4 mg) at 74% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 9.24 – 9.16 (m, 1H), 8.81 – 8.73 (m, 1H), 8.29 – 8.24 (m, 1H), 7.38 (dd, J = 8.0, 4.9 Hz, 1H), 4.86 (dt, J = 49.1, 7.9, 3.7 Hz, 1H), 4.48 – 4.33 (m, 2H), 2.64 – 2.22 (m, 2H), 2.07 – 1.70 (m, 4H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 165.25, 153.62, 150.93, 137.08, 126.07, 125.38 (dq, J = 3.9, 276.8 Hz), 123.42, 87.23 (dq, J = 173.0, 3.2 Hz), 64.64, 39.49 (qd, J = 28.6, 23.0 Hz), 31.74 (d, J = 21.2 Hz), 24.24 (d, J = 3.9 Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.12 (d, J = 7.3 Hz, 3F), -182.90 (q, J = 7.5 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{12}\text{H}_{13}\text{F}_4\text{NO}_2\text{H}^+]$: 280.0955, found 280.0956.

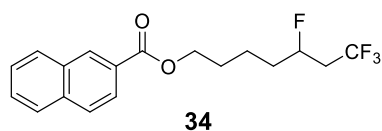


4,6,6,6-Tetrafluorohexyl picolinate (32): The compound was purified by column chromatography (petroleum ether/EtOAc = 30/1), resulting in a colorless liquid (40.2 mg) at 72% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.74 (d, J = 4.7 Hz, 1H), 8.10 (d, J = 7.8 Hz, 1H), 7.86 – 7.79 (m, 1H), 7.49 – 7.44 (m, 1H), 4.86 (dt, J = 49.2, 8.3, 3.6 Hz, 1H), 4.51 – 4.39 (m, 2H), 2.62 – 2.21 (m, 2H), 2.09 – 1.70 (m, 4H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 165.32, 150.07, 148.07, 137.18, 127.13,

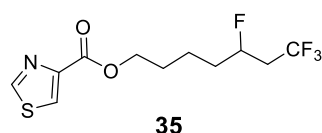
125.44 (dq, $J = 3.4$, 279.8 Hz), 125.31, 87.31 (dq, $J = 173.0$, 3.1 Hz), 65.18, 39.55 (qd, $J = 28.6$, 22.8 Hz), 31.80 (d, $J = 21.2$ Hz), 24.32 (d, $J = 4.0$ Hz). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -64.12 (d, $J = 7.5$ Hz, 3F), -182.81 (q, $J = 7.7$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{12}\text{H}_{13}\text{F}_4\text{NO}_2\text{Na}^+]$: 302.0775, found 302.0785.



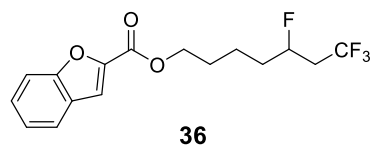
5,7,7,7-Tetrafluoroheptyl-2-chloro-4-nitrobenzoate (33): The compound was purified by column chromatography (petroleum ether/EtOAc = 30/1), resulting in a colorless liquid (23.8 mg) at 32% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 8.30 – 8.27 (m, 1H), 8.16 – 8.12 (m, 1H), 7.95 – 7.91 (m, 1H), 4.91 – 4.72 (m, 1H), 4.39 (t, $J = 6.5$ Hz, 2H), 2.63 – 2.21 (m, 2H), 1.90 – 1.50 (m, 6H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 164.38, 149.50, 136.12, 134.70, 132.06, 126.06, 125.48 (dq, $J = 3.6$, 276.8 Hz), 121.58, 87.44 (dq, $J = 172.5$, 3.1 Hz), 66.06, 39.43 (qd, $J = 28.6$, 23.0 Hz), 34.56 (d, $J = 20.9$ Hz), 28.17, 21.38 (d, $J = 4.0$ Hz). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -64.12 (d, $J = 7.1$ Hz, 3F), -182.59 (q, $J = 7.4$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{14}\text{H}_{14}\text{ClF}_4\text{O}_4\text{Na}^+]$: 394.0440, found 394.0446.



5,7,7,7-Tetrafluoroheptyl-2-naphthoate (34): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (32.8 mg) at 48% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 8.62 (s, 1H), 8.10 – 8.04 (m, 1H), 7.96 (d, $J = 8.0$ Hz, 1H), 7.92 – 7.84 (m, 2H), 7.42 – 7.36 (m, 2H), 4.93 – 4.73 (m, 1H), 4.41 (t, $J = 6.5$ Hz, 2H), 2.64 – 2.22 (m, 2H), 1.92 – 1.55 (m, 6H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 166.81, 135.62, 132.59, 131.07, 129.42, 128.35, 128.26, 127.85, 127.63, 126.76, 125.55 (dq, $J = 3.4$, 276.8 Hz), 125.27, 87.50 (dq, $J = 172.5$, 3.2 Hz), 64.72, 39.47 (qd, $J = 28.5$, 23.0 Hz), 34.71 (d, $J = 20.8$ Hz), 28.47, 21.48 (d, $J = 4.2$ Hz). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -64.04 (d, $J = 7.6$ Hz, 3F), -182.36 (q, $J = 8.0$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{18}\text{H}_{18}\text{F}_4\text{O}_2\text{Na}^+]$: 365.1135, found 365.1142.

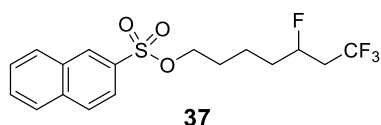


5,7,7,7-Tetrafluoroheptyl thiazole-4-carboxylate (35): The compound was purified by column chromatography (petroleum ether/EtOAc = 30/1), resulting in a white solid (42.5 mg) at 71% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 8.82 – 8.79 (m, 1H), 8.20 – 8.18 (m, 1H), 4.86 – 4.65 (m, 1H), 4.33 (t, $J = 6.6$ Hz, 2H), 2.56 – 2.14 (m, 2H), 1.85 – 1.44 (m, 6H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 161.26, 153.53, 147.94, 127.35, 125.42 (dq, $J = 3.2$, 276.8 Hz), 87.36 (dq, $J = 172.4$, 3.2 Hz), 64.98, 39.30 (qd, $J = 28.5$, 22.8 Hz), 34.51 (d, $J = 20.9$ Hz), 28.22, 21.24 (d, $J = 4.2$ Hz). ^{19}F NMR (376 MHz, DMSO-*d*₆) δ -62.42 (d, $J = 8.0$ Hz, 3F), -182.31 (q, $J = 8.1$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{11}\text{H}_{13}\text{F}_4\text{NO}_2\text{SNa}^+]$: 322.0495, found 322.0498.

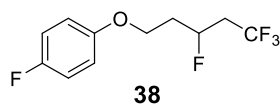


5,7,7,7-Tetrafluoroheptyl benzofuran-2-carboxylate (36): The compound was purified by column chromatography (petroleum ether/EtOAc = 50/1), resulting in a colorless liquid (31.2 mg) at 47% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 7.69 – 7.65 (m, 1H), 7.60 – 7.56 (m, 1H), 7.53 – 7.50 (m, 1H), 7.46 – 7.41 (m, 1H), 7.32 – 7.26 (m, 1H), 4.92 – 4.72 (m, 1H), 4.38 (t, $J = 6.5$ Hz, 2H), 2.62 – 2.20 (m, 2H), 1.88 – 1.50 (m, 6H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 159.67,

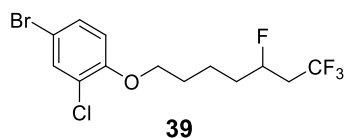
155.85, 145.59, 127.77, 127.03, 125.52 (dq, $J = 3.3, 276.6$ Hz), 123.92, 122.93, 114.03, 112.45, 87.48 (dq, $J = 172.5, 3.2$ Hz), 65.05, 39.49 (qd, $J = 28.6, 23.0$ Hz), 34.66 (d, $J = 20.9$ Hz), 28.38, 21.37 (d, $J = 4.1$ Hz). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -64.08 (d, $J = 7.5$ Hz, 3F), -182.44 (q, $J = 7.9$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{16}\text{H}_{16}\text{F}_4\text{O}_3\text{Na}^+]$: 355.0928, found 355.0930.



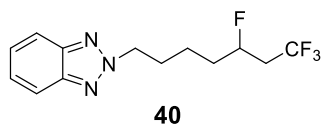
5,7,7,7-Tetrafluoroheptyl naphthalene-2-sulfonate (37): The compound was purified by column chromatography (petroleum ether/EtOAc = 10/1), resulting in a colorless liquid (34.0 mg) at 45% yield. ^1H NMR (400 MHz, DMSO-*d*₆) δ 8.68 – 8.61 (m, 1H), 8.27 – 8.18 (m, 2H), 8.10 (d, $J = 8.1$ Hz, 1H), 7.89 – 7.85 (m, 1H), 7.83 – 7.67 (m, 2H), 4.72 (dtt, $J = 50.1, 7.9, 4.2$ Hz, 1H), 4.09 (t, $J = 6.3$ Hz, 2H), 2.66 – 2.51 (m, 2H), 1.65 – 1.28 (m, 4H), 1.43 – 1.26 (m, 2H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 135.42, 132.91, 132.08, 129.86, 129.85, 129.57, 129.46, 128.15, 128.03, 125.43 (dq, $J = 3.7, 276.9$ Hz), 122.62, 87.35 (dq, $J = 172.8, 3.2$ Hz), 70.38, 39.42 (qd, $J = 28.6, 22.9$ Hz), 34.34 (d, $J = 20.9$ Hz), 28.54, 20.91 (d, $J = 4.1$ Hz). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -57.76 (d, $J = 7.5$ Hz, 3F), -177.66 (q, $J = 7.6$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{17}\text{H}_{18}\text{F}_4\text{O}_3\text{SNa}^+]$: 401.0805, found 401.0803.



1-Fluoro-4-((3,5,5,5-tetrafluoropentyl)oxy)benzene (38): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (37.1 mg) at 73% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 7.02 – 6.96 (m, 2H), 6.88 – 6.83 (m, 2H), 5.22 – 5.01 (m, 1H), 4.13 – 4.03 (m, 2H), 2.71 – 2.34 (m, 2H), 2.20 – 2.08 (m, 2H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 157.59 (d, $J = 238.5$ Hz), 154.77 (d, $J = 2.1$ Hz), 125.53 (dq, $J = 2.9, 276.8$ Hz), 116.01 (d, $J = 23.2$ Hz), 115.63 (d, $J = 8.0$ Hz), 84.98 (dq, $J = 172.5, 3.3$ Hz), 63.62 (d, $J = 4.9$ Hz), 39.60 (qd, $J = 28.7, 22.4$ Hz), 34.88 (d, $J = 20.7$ Hz). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -64.08 (d, $J = 7.6$ Hz, 3F), -123.62 (s, 1F), -184.82 (q, $J = 8.3, 7.5$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{11}\text{H}_{11}\text{F}_5\text{ONa}^+]$: 277.0622, found 277.0628.

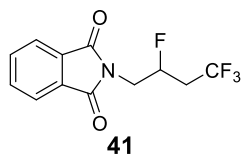


4-Bromo-2-chloro-1-((5,7,7,7-tetrafluoroheptyl)oxy)benzene (39): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (45.8 mg) at 61% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 7.51 – 7.48 (m, 1H), 7.33 – 7.28 (m, 1H), 6.77 (d, $J = 8.8$ Hz, 1H), 4.84 (dtt, $J = 48.3, 7.9, 3.2$ Hz, 1H), 4.01 (t, $J = 6.1$ Hz, 2H), 2.56 – 2.13 (m, 2H), 1.85 – 1.50 (m, 6H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 153.90, 132.82, 130.63, 125.57 (dq, $J = 3.4, 276.8$ Hz), 124.21, 114.62, 112.67, 87.59 (dq, $J = 172.4, 3.2$ Hz), 69.08, 39.52 (qd, $J = 28.5, 22.9$ Hz), 34.76 (d, $J = 20.8$ Hz), 28.67, 21.55 (d, $J = 4.3$ Hz). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -64.09 (d, $J = 7.5$ Hz, 3F), -182.33 (q, $J = 7.4$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{13}\text{H}_{14}\text{BrClF}_4\text{ONa}^+]$: 398.9745, found 398.9756.

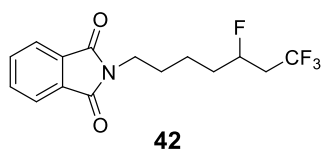


2-((5,7,7,7-Tetrafluoroheptyl)-2H-benzo[d][1,2,3]triazole (40): The compound was purified by column chromatography (petroleum ether/EtOAc = 20/1), resulting in a colorless liquid (28.3 mg) at 49% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 7.88 – 7.32 (m, 4H), 4.77 (dtt, $J = 12.1, 8.0, 3.5$ Hz, 1H), 4.75 (t, $J = 6.9$ Hz, 2H), 2.65 – 2.08 (m, 2H), 2.22 – 2.12 (m, 2H), 1.87 – 1.38 (m, 4H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 144.43, 126.43, 125.45 (dq, $J = 3.6, 276.9$ Hz), 118.06, 87.34

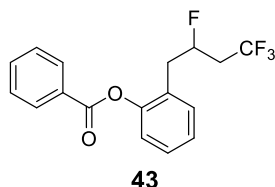
(dq, $J = 172.7, 3.2$ Hz), 56.21, 39.43 (qd, $J = 28.6, 22.9$ Hz), 34.41 (d, $J = 20.9$ Hz), 29.58, 21.90 (d, $J = 4.1$ Hz). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -64.12 (d, $J = 7.4$ Hz, 3F), -182.43 (q, $J = 7.4$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{13}\text{H}_{15}\text{F}_4\text{N}_3\text{Na}^+]$: 312.1094, found 312.1100.



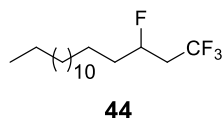
2-(1,3,3,3-Tetrafluoropropyl)isoindoline-1,3-dione (41): The compound was purified by column chromatography (petroleum ether/EtOAc = 30/1), resulting in a white solid (31.9 mg) at 58% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 7.87 – 7.70 (m, 4H), 7.72 (dd, $J = 5.5, 3.1$ Hz, 2H), 5.08 (dtt, $J = 48.7, 7.3, 3.7$ Hz, 1H), 4.08 – 3.96 (m, 1H), 3.92 – 3.77 (m, 1H), 2.65 – 2.38 (m, 2H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 167.92, 134.44, 131.77, 125.17 (dq, $J = 2.0, 276.8$ Hz), 123.66, 84.83 (dq, $J = 179.0, 3.2$ Hz), 41.02 (d, $J = 23.7$ Hz), 37.49 (qd, $J = 29.4, 21.7$ Hz). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -63.95 (d, $J = 7.3$ Hz, 3F), -187.53 (q, $J = 7.2$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{12}\text{H}_9\text{F}_4\text{NO}_2\text{Na}^+]$: 298.0462, found 298.0469.



2-(5,7,7,7-Tetrafluoroheptyl)isoindoline-1,3-dione (42): The compound was purified by column chromatography (petroleum ether/EtOAc = 30/1), resulting in a white solid (34.7 mg) at 55% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 7.83 – 7.66 (m, 4H), 7.68 (dd, $J = 5.5, 3.0$ Hz, 2H), 4.76 (dtt, $J = 49.1, 8.1, 3.6$ Hz, 1H), 3.67 (t, $J = 7.1$ Hz, 2H), 2.57 – 2.19 (m, 2H), 1.78 – 1.38 (m, 6H). ^{13}C NMR (101 MHz, DMSO-*d*₆) δ 167.94, 134.34, 131.61, 126.19 (q, $J = 277.0$ Hz), 122.98, 87.87 (dq, $J = 169.8, 3.3$ Hz), 37.83 (qd, $J = 27.2, 21.0$ Hz), 37.17, 33.52 (d, $J = 20.2$ Hz), 27.57, 21.44 (d, $J = 4.3$ Hz). ^{19}F NMR (376 MHz, DMSO-*d*₆) δ -62.49 (d, $J = 7.5$ Hz, 3F), -182.16 (q, $J = 7.6$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{15}\text{F}_4\text{NO}_2\text{Na}^+]$: 340.0931, found 340.0938.

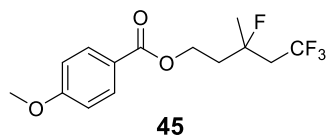


2-(2,4,4,4-Tetrafluorobutyl)phenyl benzoate (43): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (35.7 mg) at 55% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 8.21 – 8.17 (m, 2H), 7.68 – 7.62 (m, 1H), 7.55 – 7.49 (m, 2H), 7.38 – 7.31 (m, 2H), 7.28 – 7.18 (m, 2H), 5.17 – 4.16 (m, 1H), 3.08 – 2.88 (m, 2H), 2.57 – 2.19 (m, 2H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 165.07, 149.57, 134.06, 131.57, 130.24, 129.13, 128.91, 128.80, 127.88 (d, $J = 4.5$ Hz), 126.57, 125.45 (dq, $J = 3.1, 276.9$ Hz), 123.01, 87.11 (dq, $J = 175.8, 3.0$ Hz), 38.93 (qd, $J = 28.7, 22.6$ Hz), 35.81 (d, $J = 22.1$ Hz). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -63.98 (d, $J = 7.5$ Hz, 3F), -179.35 (q, $J = 8.0$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{17}\text{H}_{14}\text{F}_4\text{O}_2\text{Na}^+]$: 349.0822, found 349.0827.

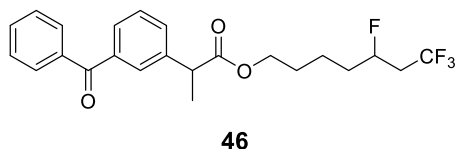


1,1,1,3-Tetrafluorooctane (44): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (40.5 mg) at 65% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 4.79 (dtt, $J = 49.0, 8.0, 3.7$ Hz, 1H), 2.60 – 2.19 (m, 2H), 1.80 – 1.56 (m, 2H), 1.50 – 1.24 (m, 24H), 0.88 (t, $J = 6.7$ Hz, 3H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 127.16 – 124.20 (m), 87.79 (dq, $J = 172.1, 3.2$ Hz), 39.56 (qd, $J = 28.5, 23.0$ Hz), 35.18 (d, $J = 20.6$

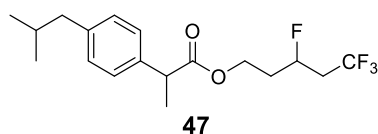
Hz), 32.09, 29.85, 29.83, 29.81, 29.81, 29.76, 29.66, 29.58, 29.52, 29.35, 24.73 (d, $J = 4.3$ Hz), 22.85, 14.27. ^{19}F NMR (376 MHz, Chloroform-*d*) δ -64.17 (d, $J = 7.3$ Hz, 3F), -181.69 (q, $J = 7.4$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{17}\text{H}_{32}\text{F}_4\text{H}^+]$: 313.2513, found 313.2526.



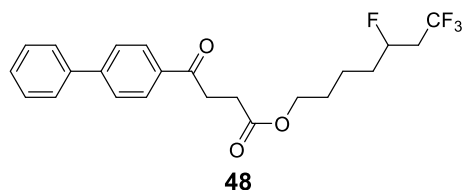
3,5,5,5-Tetrafluoro-3-methylpentyl 4-methoxybenzoate (45): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (22.2 mg) at 36% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 8.00 – 7.94 (m, 2H), 6.94 – 6.90 (m, 2H), 4.52 – 4.41 (m, 2H), 3.85 (s, 3H), 2.68 – 2.50 (m, 2H), 2.32 – 2.12 (m, 2H), 1.57 (d, $J = 22.1$ Hz, 3H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 166.19, 163.64, 131.69, 125.20 (dq, $J = 8.4$, 277.5 Hz), 122.43, 113.83, 92.39 (dq, $J = 172.8$, 2.2 Hz), 59.78 (d, $J = 6.2$ Hz), 55.54, 43.28 (qd, $J = 27.9$, 26.1 Hz), 38.74 (dd, $J = 22.5$, 1.6 Hz), 24.80 (dd, $J = 24.2$, 1.8 Hz). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -60.78 (d, $J = 8.9$ Hz, 3F), -143.12 (q, $J = 8.9$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{14}\text{H}_{16}\text{F}_4\text{O}_3\text{Na}^+]$: 331.0928, found 331.0938.



5,7,7,7-Tetrafluoroheptyl-2-(3-benzoylphenyl)propanoate (46): The compound was purified by column chromatography (petroleum ether/EtOAc = 50/1), resulting in a colorless liquid (60.2 mg) at 71% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 7.82 – 7.75 (m, 3H), 7.69 – 7.41 (m, 6H), 4.83 – 4.63 (m, 1H), 4.16 – 4.04 (m, 2H), 3.80 (q, $J = 7.2$ Hz, 1H), 2.55 – 2.37 (m, 1H), 2.35 – 2.15 (m, 1H), 1.75 – 1.57 (m, 4H), 1.54 (d, $J = 7.2$ Hz, 3H), 1.51 – 1.42 (m, 1H), 1.42 – 1.32 (m, 1H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 196.53, 174.12, 141.07, 114.05, 138.06, 137.59, 132.63, 131.56, 130.15, 129.23, 129.22, 129.13, 128.59, 128.41, 125.51 (dq, $J = 3.4$, 276.8 Hz), 87.43 (dq, $J = 172.6$, 3.0 Hz), 64.52, 45.55, 39.41 (qd, $J = 28.5$, 22.9 Hz), 34.56 (d, $J = 20.8$ Hz), 28.22, 21.20 (d, $J = 4.1$ Hz), 18.47. ^{19}F NMR (376 MHz, Chloroform-*d*) δ -64.08 (d, $J = 7.4$ Hz, 3F), -182.32 – -182.44 (m, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{23}\text{H}_{24}\text{F}_4\text{O}_3\text{Na}^+]$: 447.1554, found 447.1565.

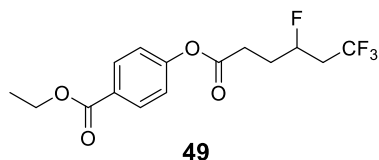


3,5,5,5-Tetrafluoropentyl-2-(4-isobutylphenyl)propanoate (47): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (36.9 mg) at 53% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 7.18 (d, $J = 8.1$ Hz, 2H), 7.10 (d, $J = 8.0$ Hz, 2H), 4.76 (dtt, $J = 48.5$, 8.1, 4.0 Hz, 1H), 4.34 – 4.09 (m, 2H), 3.70 (q, $J = 7.2$ Hz, 1H), 2.53 – 2.33 (m, 3H), 2.30 – 2.10 (m, 1H), 2.01 – 1.78 (m, 3H), 1.50 (dd, $J = 7.2$, 1.3 Hz, 3H), 0.90 (d, $J = 6.6$ Hz, 6H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 174.57, 174.57, 140.92, 140.91, 137.59, 137.55, 129.56, 129.55, 127.18, 127.17, 125.29 (dq, $J = 3.1$, 277.3 Hz), 85.76 – 83.78 (m), 59.92 – 59.72 (m), 45.30 – 45.17 (m), 45.13, 39.38 (qd, $J = 28.8$, 23.0 Hz), 34.12 (dd, $J = 21.0$, 5.0 Hz), 30.29, 22.49, 18.36. ^{19}F NMR (376 MHz, Chloroform-*d*) δ -64.09 – -64.18 (m, 3F), -184.63 (dq, $J = 68.4$, 7.4 Hz, 1F); **ESI-HRMS** exact mass calculated for $[\text{C}_{18}\text{H}_{24}\text{F}_4\text{O}_2\text{H}^+]$: 349.1785, found 349.1793.

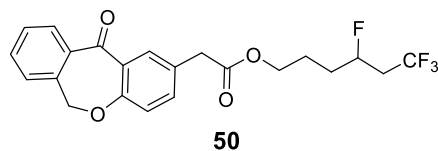


5,7,7,7-Tetrafluoroheptyl-4-([1,1'-biphenyl]-4-yl)-4-oxobutanoate (48): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a solid with color

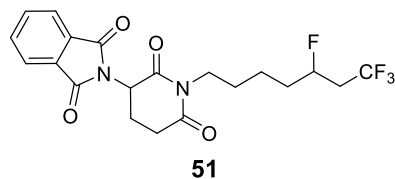
(36.5 mg) at 43% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.04 (d, $J = 8.4$ Hz, 2H), 7.67 (d, $J = 8.4$ Hz, 2H), 7.63 – 7.59 (m, 2H), 7.48 – 7.42 (m, 2H), 7.41 – 7.36 (m, 1H), 4.80 (dtt, $J = 49.2$, 8.1, 3.5 Hz, 1H), 4.13 (t, $J = 6.4$ Hz, 2H), 3.31 (t, $J = 6.5$ Hz, 2H), 2.77 (t, $J = 6.5$ Hz, 2H), 2.57 – 2.22 (m, 2H), 1.80 – 1.41 (m, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 197.56, 172.80, 145.71, 139.67, 135.21, 128.92, 128.56, 128.22, 127.16, 127.13, 125.53 (dq, $J = 2.8$, 276.9 Hz), 87.39 (dq, $J = 172.4$, 3.1 Hz), 64.06, 39.16 (qd, $J = 28.4$, 22.7 Hz), 34.43 (d, $J = 20.7$ Hz), 33.30, 28.19, 28.14, 21.10 (d, $J = 4.3$ Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -63.99 (d, $J = 7.7$ Hz, 3F), -182.25 (q, $J = 7.9$ Hz, 1F); **ESI-HRMS** exact mass calculated for $[\text{C}_{23}\text{H}_{24}\text{F}_4\text{O}_3\text{Na}^+]$: 477.1554, found 477.1561.



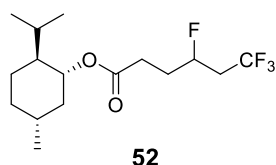
Ethyl 4-((4,6,6,6-tetrafluorohexanoyl)oxy)benzoate (49): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a white solid (26.9 mg) at 40% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.11 – 8.05 (m, 2H), 7.20 – 7.13 (m, 2H), 4.95 (dtt, $J = 49.1$, 8.1, 4.1 Hz, 1H), 4.38 (q, $J = 7.2$ Hz, 2H), 2.88 – 2.73 (m, 2H), 2.70 – 2.52 (m, 1H), 2.49 – 2.32 (m, 1H), 2.20 – 2.06 (m, 2H), 1.39 (t, $J = 7.1$ Hz, 3H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 170.66, 165.89, 154.12, 131.29, 128.38, 125.29 (dq, $J = 4.2$, 276.9 Hz), 121.55, 86.73 (dq, $J = 173.2$, 3.2 Hz), 61.26, 39.61 (qd, $J = 28.8$, 22.8 Hz), 30.16 (d, $J = 20.7$ Hz), 29.67 (d, $J = 4.1$ Hz), 14.45; $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.04 (d, $J = 7.4$ Hz, 3F), -184.67 (q, $J = 7.3$ Hz, 1F); **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{16}\text{F}_4\text{O}_4\text{Na}^+]$: 359.0877, found 359.0877.



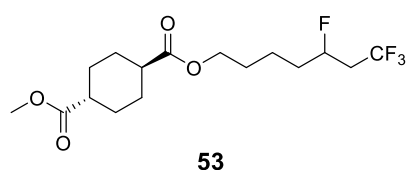
4,6,6,6-Tetrafluorohexyl-2-(11-oxo-6,11-dihydrodibenzo[b,e]oxepin-2-yl)acetate (50): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (41.6 mg) at 49% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.12 (d, $J = 2.4$ Hz, 1H), 7.91 – 7.86 (m, 1H), 7.59 – 7.34 (m, 4H), 7.06 – 7.01 (m, 1H), 5.19 (s, 2H), 4.81 (dtt, $J = 49.1$, 7.7, 3.7 Hz, 1H), 4.21 – 4.10 (m, 2H), 3.64 (s, 2H), 2.61 – 2.20 (m, 2H), 1.91 – 1.60 (m, 4H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 190.92, 171.43, 160.63, 140.51, 136.40, 135.68, 132.95, 132.52, 129.60, 129.41, 127.97, 127.85, 125.45 (dq, $J = 3.6$, 276.9 Hz), 125.30, 121.23, 87.27 (dq, $J = 172.9$, 3.2 Hz), 73.77, 64.23, 40.38, 39.50 (qd, $J = 28.6$, 22.9 Hz), 31.71 (d, $J = 21.0$ Hz), 24.16 (d, $J = 4.0$ Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.07 (d, $J = 7.5$ Hz, 3F), -182.91 (q, $J = 7.9$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{22}\text{H}_{20}\text{F}_4\text{O}_4\text{Na}^+]$: 447.1190, found 447.1193.



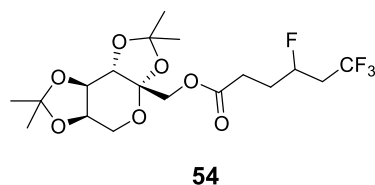
2-(2,6-Dioxo-1-(5,7,7,7-tetrafluoroheptyl)piperidin-3-yl)isoindoline-1,3-dione (51): The compound was purified by column chromatography (petroleum ether/EtOAc = 30/1), resulting in a solid with color (64.2 mg) at 75% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.86 – 7.77 (m, 2H), 7.76 – 7.67 (m, 2H), 5.02 – 4.93 (m, 1H), 4.86 – 4.63 (m, 1H), 3.81 – 3.74 (m, 2H), 2.97 – 2.86 (m, 1H), 2.80 – 2.66 (m, 2H), 2.54 – 2.16 (m, 2H), 2.14 – 2.03 (m, 1H), 1.79 – 1.30 (m, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 170.97, 168.73, 167.38, 134.44, 131.68, 129.77 – 121.37 (m), 123.64, 88.40 – 86.49 (m), 50.07, 40.08 – 39.84 (m), 39.70 – 38.40 (m), 34.39 (d, $J = 22.3$ Hz), 31.88, 27.29 (d, $J = 7.5$ Hz), 21.92, 21.61 (d, $J = 4.2$ Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.04 – -64.13 (m, 3F), -181.90 (dq, $J = 65.3$, 7.4 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{20}\text{H}_{20}\text{F}_4\text{N}_2\text{O}_4\text{Na}^+]$: 451.1251, found 451.1254.



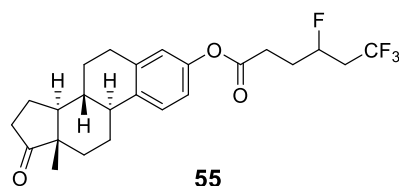
(1*R*,2*S*,5*R*)-2-isopropyl-5-methylcyclohexyl 4,6,6,6-tetrafluorohexanoate (52): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (40.4 mg) at 62% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 4.96 – 4.75 (m, 1H), 4.70 (td, $J = 10.9, 4.4$ Hz, 1H), 2.63 – 2.42 (m, 2H), 2.42 – 2.23 (m, 2H), 2.06 – 1.92 (m, 3H), 1.88 – 1.78 (m, 1H), 1.71 – 1.63 (m, 2H), 1.55 – 1.41 (m, 1H), 1.41 – 1.32 (m, 1H), 1.11 – 0.84 (m, 9H), 0.75 (d, $J = 7.0$ Hz, 3H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 172.13, 172.11, 125.40 (dq, $J = 3.7, 276.8$ Hz), 87.81 – 85.90 (m), 74.80, 74.78, 47.15, 41.05, 41.02, 39.59 (qd, $J = 28.8, 22.8$ Hz), 34.37, 31.54, 30.43 (d, $J = 21.0$ Hz), 29.88 – 29.65 (m), 26.50, 26.48, 23.60, 23.57, 21.12, 20.86, 16.43, 16.41; $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.14 – -64.18 (m, 3F), -184.41 (q, $J = 7.8$ Hz, 1F); **ESI-HRMS** exact mass calculated for $[\text{C}_{16}\text{H}_{26}\text{F}_4\text{O}_2\text{Na}^+]$: 349.1761, found 349.1770.



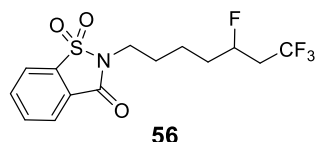
1-Methyl-4-(5,7,7,7-tetrafluoroheptyl) (1*r*,4*r*)-cyclohexane-1,4-dicarboxylate (53): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a solid with color (22.8 mg) at 32% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 4.78 (dt, $J = 48.9, 8.3, 3.5$ Hz, 1H), 4.06 (t, $J = 6.4$ Hz, 2H), 3.65 (s, 3H), 2.60 – 2.43 (m, 1H), 2.40 – 2.19 (m, 3H), 2.08 – 1.97 (m, 4H), 1.81 – 1.49 (m, 6H), 1.46 – 1.39 (m, 4H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 175.98, 175.55, 125.50 (dq, $J = 3.6, 276.8$ Hz), 87.46 (dq, $J = 172.5, 3.2$ Hz), 63.93, 51.69, 42.61, 42.50, 39.49 (qd, $J = 28.5, 23.1$ Hz), 34.66 (q, $J = 20.9$ Hz), 28.32, 28.14, 28.12, 21.31 (d, $J = 4.1$ Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.14 (d, $J = 7.5$ Hz, 3F), -182.45 (q, $J = 7.8$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{16}\text{H}_{24}\text{F}_4\text{O}_4\text{Na}^+]$: 379.1503, found 379.1512.



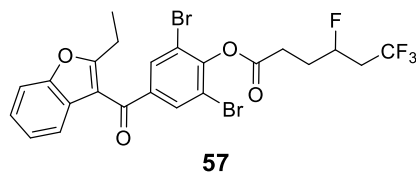
((3*aS*,5*aR*,8*aR*,8*bS*)-2,2,7,7-Tetramethyltetrahydro-3*aH*-bis([1,3]dioxolo)[4,5-*b*:4',5'-*d*]pyran-3*a*-yl)methyl 4,6,6,6-tetrafluorohexanoate (54): The compound was purified by column chromatography (petroleum ether/EtOAc = 10/1), resulting in a colorless liquid (38.7 mg) at 45% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 4.95 – 4.75 (m, 1H), 4.58 (dd, $J = 7.9, 2.6$ Hz, 1H), 4.42 (dd, $J = 11.6, 2.0$ Hz, 1H), 4.28 – 4.25 (m, 1H), 4.23 – 4.20 (m, 1H), 4.05 – 4.00 (m, 1H), 3.91 – 3.85 (m, 1H), 3.77 – 3.71 (m, 1H), 2.62 – 2.22 (m, 4H), 2.11 – 1.92 (m, 2H), 1.51 (s, 3H), 1.45 (s, 3H), 1.37 (s, 3H), 1.32 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 171.85, 125.33 (dq, $J = 3.4, 276.8$ Hz), 109.22, 108.86, 108.84, 101.53, 86.63 (dq, $J = 173.0, 3.2$ Hz), 70.83, 70.71, 70.13, 65.70, 65.65, 61.35, 40.10 – 38.86 (m), 30.08 (d, $J = 20.9$ Hz), 29.21 – 29.08 (m), 26.50, 25.95, 25.23, 24.11. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.06 – -64.12 (m, 3F), -184.85 (q, $J = 7.9$ Hz, 1F); **ESI-HRMS** exact mass calculated for $[\text{C}_{18}\text{H}_{26}\text{F}_4\text{O}_7\text{Na}^+]$: 453.1507, found 453.1514.



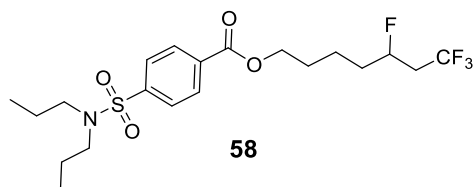
(8R,9S,13S,14S)-13-Methyl-17-oxo-7, 8, 9, 11, 12, 13, 14, 15, 16, 17-decahydro-6H-cyclopenta[a]phen-anthren-3-yl 5,7,7,7-tetra-fluoroheptanoate (55): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a white solid (44.2 mg) at 50% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.29 (d, $J = 8.5$ Hz, 1H), 6.85 (d, $J = 2.5$ Hz, 1H), 6.81 (s, 1H), 5.06 – 4.84 (m, 1H), 2.94 – 2.86 (m, 2H), 2.79 – 2.72 (m, 2H), 2.66 – 2.21 (m, 5H), 2.20 – 1.93 (m, 6H), 1.69 – 1.38 (m, 6H), 0.90 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 171.43, 148.45, 138.20, 137.66, 126.55, 125.31 (dq, $J = 3.7, 276.8$ Hz), 121.53, 118.69, 86.73 (dq, $J = 173.2, 3.3$ Hz), 50.49, 48.02, 44.23, 39.52 (qd, $J = 28.9, 22.7$ Hz), 38.06, 35.94, 31.63, 30.21 (d, $J = 20.9$ Hz), 29.55, 29.51, 29.49, 26.40, 25.83, 21.67, 13.90. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.03 (d, $J = 7.6$ Hz, 3F), -184.61 (q, $J = 7.6$ Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{24}\text{H}_{28}\text{F}_4\text{O}_3\text{Na}^+]$: 463.1867, found 463.1862.



2-(5,7,7,7-Tetrafluoroheptyl)benzo[d]isothiazol-3(2H)-one 1,1-dioxide (56): The compound was purified by column chromatography (petroleum ether/EtOAc = 30/1), resulting in a liquid with color (38.8 mg) at 55% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.07 – 8.03 (m, 1H), 7.94 – 7.90 (m, 1H), 7.89 – 7.79 (m, 2H), 4.81 (dt, $J = 48.8, 8.2, 3.5$ Hz, 1H), 3.79 (t, $J = 7.3$ Hz, 2H), 2.62 – 2.20 (m, 2H), 1.96 – 1.87 (m, 2H), 1.84 – 1.45 (m, 4H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 159.12, 137.79, 134.90, 134.48, 127.46, 125.53 (dq, $J = 3.4, 276.8$ Hz), 125.30, 121.05, 87.44 (dq, $J = 172.6, 3.2$ Hz), 39.46 (qd, $J = 28.5, 22.9$ Hz), 39.04, 34.47 (d, $J = 20.9$ Hz), 28.07, 21.99 (d, $J = 4.2$ Hz); $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -64.10 (d, $J = 7.5$ Hz, 3F), -182.30 (q, $J = 7.3$ Hz, 1F); **ESI-HRMS** exact mass calculated for $[\text{C}_{14}\text{H}_{15}\text{F}_4\text{NO}_3\text{SNa}^+]$: 376.0601, found 376.0604.

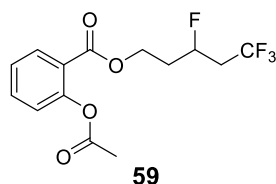


2,6-Dibromo-4-(2-ethylbenzofuran-3-carbonyl)phenyl 4,6,6,6-tetrafluorohexanoate (57): The compound was purified by column chromatography (petroleum ether/EtOAc = 50/1), resulting in a liquid with color (53.2 mg) at 45% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.03 (s, 2H), 7.51 (d, $J = 8.1$ Hz, 1H), 7.41 (d, $J = 7.7$ Hz, 1H), 7.36 – 7.22 (m, 2H), 5.10 – 4.90 (m, 1H), 3.01 – 2.87 (m, 4H), 2.69 – 2.32 (m, 2H), 2.28 – 2.11 (m, 2H), 1.37 (t, $J = 7.5$ Hz, 3H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 187.84, 168.58, 167.52, 153.82, 149.35, 139.22, 133.19, 126.34, 125.28 (dq, $J = 3.8, 276.9$ Hz), 124.95, 124.14, 121.12, 118.20, 115.31, 111.34, 86.45 (dq, $J = 173.6, 3.2$ Hz), 39.58 (qd, $J = 28.9, 22.7$ Hz), 30.02 (d, $J = 21.0$ Hz), 29.19 (d, $J = 4.3$ Hz), 22.20, 12.32. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -63.95 (d, $J = 7.3$ Hz, 3F), -184.91 (q, $J = 7.4$ Hz, 1F); **ESI-HRMS** exact mass calculated for $[\text{C}_{23}\text{H}_{18}\text{Br}_2\text{F}_4\text{O}_4\text{Na}^+]$: 614.9400, found 614.9404.

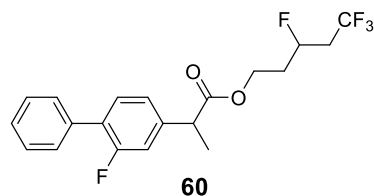


5,7,7,7-Tetrafluoroheptyl 4-(N,N-dipropylsulfamoyl)benzoate (58): The compound was purified by column chromatography (petroleum ether/EtOAc = 10/1), resulting in a colorless liquid (54.6 mg) at 60% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.14 – 7.78 (m, 4H), 4.77 (dt, $J = 49.1, 8.2, 3.4$ Hz, 1H), 4.31 (t, $J = 6.4$ Hz, 2H), 3.04 (t, $J = 8.0$ Hz, 4H), 2.57 – 2.18 (m, 2H), 1.83 – 1.43 (m, 10H), 0.80 (t, $J = 7.4$ Hz, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 165.14, 144.19, 133.55, 130.13, 126.95, 125.45 (dq, $J = 3.0, 276.8$ Hz), 87.39 (dq, $J = 172.4, 3.1$ Hz), 65.11, 49.90, 39.25 (qd, $J = 28.4, 22.7$ Hz), 34.47 (d, $J = 20.7$ Hz), 28.18, 21.88, 21.28 (d, $J = 4.1$ Hz), 11.04. $^{19}\text{F NMR}$

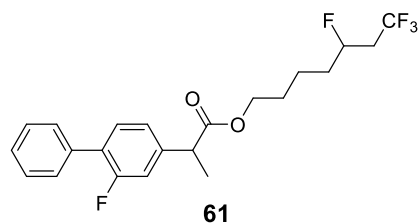
(376 MHz, Chloroform-*d*) δ -64.14 (d, $J = 7.1$ Hz, 3F), -182.74 (q, $J = 7.4$ Hz, 1F). **ESI-HRMS** exact mass calculated for [C₂₀H₂₉F₄NO₄Na⁺]: 478.1646, found 478.1653.



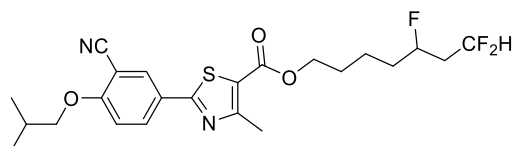
3,5,5,5-Tetrafluoropentyl 2-acetoxybenzoate (59): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (31.5 mg) at 49% yield. **¹H NMR** (400 MHz, Chloroform-*d*) δ 7.98 (dd, $J = 7.8, 1.6$ Hz, 1H), 7.63 – 7.52 (m, 1H), 7.35 – 7.29 (m, 1H), 7.15 – 7.05 (m, 1H), 5.12 – 4.90 (m, 1H), 4.54 – 4.34 (m, 2H), 2.68 – 2.28 (m, 5H), 2.18 – 1.99 (m, 2H). **¹³C NMR** (101 MHz, Chloroform-*d*) δ 169.74, 164.28, 150.84, 134.20, 131.62, 126.18, 125.30 (dq, $J = 3.4, 276.8$ Hz), 123.97, 123.08, 84.84 (dq, $J = 173.3, 3.3$ Hz), 60.39 (d, $J = 4.7$ Hz), 39.47 (qd, $J = 28.8, 22.5$ Hz), 34.22 (d, $J = 20.8$ Hz), 21.04. **¹⁹F NMR** (376 MHz, Chloroform-*d*) δ -63.97 (d, $J = 7.4$ Hz, 3F), -184.90 (q, $J = 7.4$ Hz, 1F). **ESI-HRMS** exact mass calculated for [C₁₄H₁₄F₄O₄Na⁺]: 345.0720, found 345.0726.



3,5,5,5-Tetrafluoropentyl 2-(2-fluoro-[1,1'-biphenyl]-4-yl)propanoate (60): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (45.3 mg) at 59% yield. **¹H NMR** (400 MHz, Chloroform-*d*) δ 7.56 (d, $J = 7.3$ Hz, 2H), 7.50 – 7.35 (m, 4H), 7.20 – 7.10 (m, 2H), 4.96 – 4.73 (m, 1H), 4.36 – 4.20 (m, 2H), 3.79 (q, $J = 7.1$ Hz, 1H), 2.61 – 2.16 (m, 2H), 2.08 – 1.85 (m, 2H), 1.60 – 1.52 (m, 3H). **¹³C NMR** (101 MHz, Chloroform-*d*) δ 173.77, 159.82 (d, $J = 248.5$ Hz), 141.68 (d, $J = 4.0$ Hz), 141.60 (d, $J = 4.0$ Hz), 135.50, 131.02 (d, $J = 3.8$ Hz), 129.03 (d, $J = 2.9$ Hz), 128.56, 128.11 (d, $J = 13.5$ Hz), 127.81, 125.29 (dq, $J = 3.5, 277.0$ Hz), 123.57 (d, $J = 3.3$ Hz), 85.74 – 83.75 (m), 60.30 – 60.10 (m), 45.12 – 45.00 (m), 39.35 (qd, $J = 28.7, 22.8$ Hz), 34.04 (dd, $J = 21.0, 5.1$ Hz), 18.25. **¹⁹F NMR** (376 MHz, Chloroform-*d*) δ -64.08 – -64.14 (m, 3F), -117.43 – -117.49 (m, 1F), -184.34 – -184.68 (m, 1F). **ESI-HRMS** exact mass calculated for [C₂₀H₁₉F₅O₂H⁺]: 387.1378, found 387.1385.



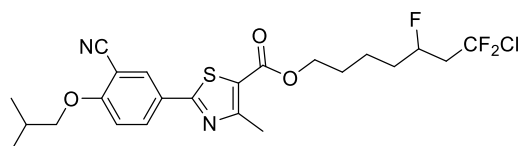
5,7,7,7-Tetrafluoroheptyl 2-(2-fluoro-[1,1'-biphenyl]-4-yl)propanoate (61): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (57.9 mg) at 70% yield. **¹H NMR** (400 MHz, Chloroform-*d*) δ 7.58 – 7.50 (m, 2H), 7.48 – 7.34 (m, 4H), 7.19 – 7.10 (m, 2H), 4.77 (dt, $J = 48.9, 7.6, 3.2$ Hz, 1H), 4.13 (t, $J = 6.4$ Hz, 2H), 3.77 (q, $J = 7.1$ Hz, 1H), 2.58 – 2.15 (m, 2H), 1.78 – 1.57 (m, 4H), 1.55 (d, $J = 7.2$ Hz, 3H), 1.53 – 1.34 (m, 2H). **¹³C NMR** (101 MHz, Chloroform-*d*) δ 174.07, 161.01, 158.55, 141.98 (d, $J = 7.7$ Hz), 135.54, 130.90 (d, $J = 3.9$ Hz), 129.02 (d, $J = 3.0$ Hz), 128.56, 127.93 (d, $J = 13.5$ Hz), 127.79, 125.51 (dq, $J = 3.5, 276.9$ Hz), 123.64 (d, $J = 3.3$ Hz), 87.44 (dq, $J = 172.7, 3.3$ Hz), 64.57, 45.17 (d, $J = 1.2$ Hz), 39.41 (qd, $J = 28.5, 23.0$ Hz), 34.57 (d, $J = 20.8$ Hz), 28.21, 21.22 (d, $J = 4.1$ Hz), 18.38. **¹⁹F NMR** (376 MHz, Chloroform-*d*) δ -64.09 (d, $J = 7.7$ Hz, 3F), -117.66 (s, 1F), -182.41 (q, $J = 7.7$ Hz, 1F). **ESI-HRMS** exact mass calculated for [C₂₂H₂₃F₅O₂H⁺]: 415.1691, found 415.1688.



63

5,7,7-Trifluoroheptyl 2-(3-cyano-4-isobutoxyphenyl)-4-methylthiazole-5-carboxylate (63):

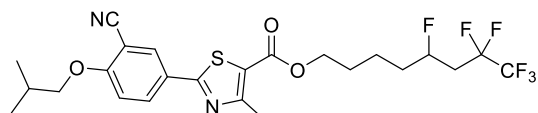
The compound was purified by column chromatography (petroleum ether/EtOAc = 50/1), resulting in a white solid (64.6 mg) at 69% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.15 – 8.12 (m, 1H), 8.08 – 8.03 (m, 1H), 6.99 (d, J = 8.9 Hz, 1H), 5.98 (tdd, J = 56.6, 6.7, 2.8 Hz, 1H), 4.84 – 4.61 (m, 1H), 4.29 (t, J = 6.4 Hz, 2H), 3.88 (d, J = 6.5 Hz, 2H), 2.73 (s, 3H), 2.29 – 1.94 (m, 3H), 1.85 – 1.48 (m, 6H), 1.07 (d, J = 6.7 Hz, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 167.27, 162.55, 162.04, 161.29, 132.61, 132.08, 126.00, 121.70, 115.44, 115.10 (dt, J = 4.9, 233.8 Hz), 112.70, 102.99, 88.92 – 88.02 (m), 75.75, 64.97, 40.04 (q, J = 21.3 Hz), 34.74 (d, J = 20.5 Hz), 28.38, 28.23, 21.52 (d, J = 4.3 Hz), 19.11, 17.55. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -116.06 – -118.24 (m, 2F), -183.71 (s, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{23}\text{H}_{27}\text{F}_3\text{N}_2\text{O}_3\text{SNa}^+]$: 491.1587, found 491.1592.



64

7-Chloro-5,7,7-trifluoroheptyl 2-(3-cyano-4-isobutoxyphenyl)-4-methylthiazole-5-carboxylate (64):

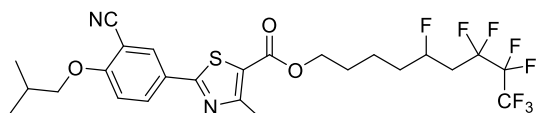
The compound was purified by column chromatography (petroleum ether/EtOAc = 50/1), resulting in a white solid (59.2 mg) at 59% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.14 – 8.11 (m, 1H), 8.07 – 8.02 (m, 1H), 6.98 (d, J = 8.9 Hz, 1H), 4.87 (dtt, J = 49.2, 8.0, 3.4 Hz, 1H), 4.29 (t, J = 6.4 Hz, 2H), 3.87 (d, J = 6.5 Hz, 2H), 2.73 (d, 3H), 2.82 – 2.40 (m, 2H), 2.23 – 2.11 (m, 1H), 1.86 – 1.45 (m, 6H), 1.07 (d, J = 6.7 Hz, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 167.25, 162.54, 162.00, 161.28, 132.59, 132.03, 127.76 (dt, J = 2.7, 292.6 Hz), 125.97, 121.66, 115.42, 112.69, 102.97, 88.21 (dt, J = 173.3, 2.9 Hz), 75.74, 64.89, 47.45 – 46.60 (m), 34.63 (d, J = 20.9 Hz), 28.32, 28.21, 21.37 (d, J = 4.2 Hz), 19.09, 17.53. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -48.57 – -50.31 (m, 2F), -181.86 (t, J = 9.0 Hz, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{23}\text{H}_{26}\text{ClF}_3\text{N}_2\text{O}_3\text{SNa}^+]$: 525.1197, found 525.1199.



65

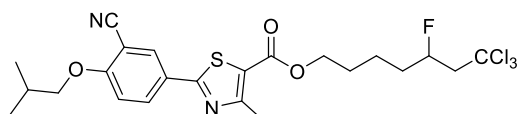
5,7,7,8,8,8-Hexafluorooctyl 2-(3-cyano-4-isobutoxyphenyl)-4-methylthiazole-5-carboxylate (65):

The compound was purified by column chromatography (petroleum ether/EtOAc = 50/1), resulting in a white solid (72.3 mg) at 68% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.15 (d, J = 2.2 Hz, 1H), 8.09 – 8.04 (m, 1H), 7.02 – 6.97 (m, 1H), 4.93 (dtt, J = 48.8, 8.0, 3.4 Hz, 1H), 4.31 (t, J = 6.4 Hz, 2H), 3.89 (d, J = 6.6 Hz, 2H), 2.75 (s, 3H), 2.59 – 2.12 (m, 3H), 1.93 – 1.54 (m, 6H), 1.08 (d, J = 6.7 Hz, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 167.35, 162.61, 162.08, 161.39, 132.64, 132.14, 126.05, 121.71, 120.80 – 117.10 (m), 115.46, 114.85 – 114.25 (m), 112.72, 103.07, 86.70 (dt, J = 173.2, 3.0 Hz), 75.81, 64.91, 36.45 (dt, J = 23.2, 21.4 Hz), 35.16 (d, J = 20.9 Hz), 28.36, 28.27, 21.44 (d, J = 4.1 Hz), 19.13, 17.58. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -85.79 – -85.85 (m, 3F), -116.95 – -117.02 (m, 2F), -180.70 – -180.78 (m, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{24}\text{H}_{26}\text{F}_6\text{N}_2\text{O}_3\text{SNa}^+]$: 559.1461, found 559.1466.



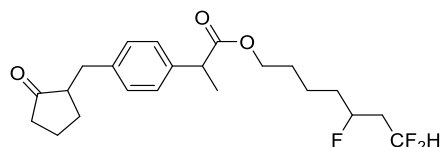
66

5,7,7,8,8,9,9,9-Octafluorononyl 2-(3-cyano-4-isobutoxyphenyl)-4-methylthiazole-5-carboxylate (66): The compound was purified by column chromatography (petroleum ether/EtOAc = 50/1), resulting in a white solid (83.2 mg) at 71% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.17 (d, $J = 2.0$ Hz, 1H), 8.11 – 8.06 (m, 1H), 7.00 (d, $J = 8.9$ Hz, 1H), 4.96 (dtt, $J = 48.9, 7.9, 3.4$ Hz, 1H), 4.32 (t, $J = 6.4$ Hz, 2H), 3.90 (d, $J = 6.5$ Hz, 2H), 2.76 (s, 3H), 2.66 – 2.13 (m, 3H), 1.93 – 1.53 (m, 6H), 1.09 (d, $J = 6.7$ Hz, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 167.18, 162.50, 161.94, 161.22, 132.53, 131.92, 125.91, 122.46 – 105.35 (m, $\text{CF}_3\text{CF}_2\text{CF}_2$), 121.62, 115.37, 112.64, 102.89, 86.58 (dt, $J = 173.6, 2.7$ Hz), 75.70, 64.84, 36.18 (q, $J = 21.9$ Hz), 35.11 (d, $J = 20.9$ Hz), 28.27, 28.18, 21.36 (d, $J = 4.1$ Hz), 18.99, 17.44. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -80.39 (t, $J = 11.3$ Hz, 3F), -114.00 – -114.24 (m, 2F), -127.80 – -127.92 (m, 2F), -180.48 – -180.57 (m, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{25}\text{H}_{26}\text{F}_8\text{N}_2\text{O}_3\text{SNa}^+]$: 609.1429, found 609.1439.



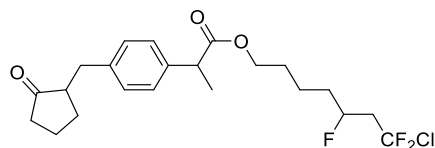
67

7,7,7-Trichloro-5-fluoroheptyl-2-(3-cyano-4-isobutoxyphenyl)-4-methylthiazole-5-carboxylate (67): The compound was purified by column chromatography (petroleum ether/EtOAc = 50/1), resulting in a white solid (57.7 mg) at 54% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.16 – 8.13 (m, 1H), 8.09 – 8.04 (m, 1H), 6.99 (d, $J = 8.9$ Hz, 1H), 4.34 – 4.24 (m, 3H), 3.88 (d, $J = 6.5$ Hz, 2H), 3.33 – 3.07 (m, 2H), 2.76 – 2.73 (m, 3H), 2.25 – 2.13 (m, 1H), 2.08 – 1.57 (m, 6H), 1.07 (d, $J = 6.7$ Hz, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 167.31, 162.57, 162.05, 161.34, 132.63, 132.12, 126.02, 121.72, 115.45, 112.71, 103.04, 96.89, 77.55 – 75.72 (m), 64.91, 62.28, 57.46, 38.52, 28.24, 28.04, 22.71, 19.15, 17.61. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -178.46 (s, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{23}\text{H}_{26}\text{Cl}_3\text{FN}_2\text{O}_3\text{SH}^+]$: 535.0787, found 535.0788.



68

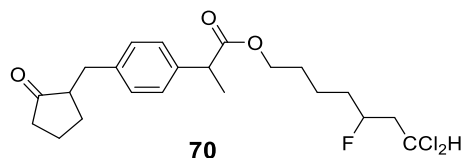
2-(4-(8,10,10-Trifluoro-3-oxodecan-2-yl)benzyl)cyclopentan-1-one (68): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (34.1 mg) at 45% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.23 – 7.09 (m, 4H), 5.95 (tdd, $J = 56.5, 6.8, 2.8$ Hz, 1H), 4.70 – 4.58 (m, 1H), 4.07 (t, $J = 6.3$ Hz, 2H), 3.68 (q, $J = 7.1$ Hz, 1H), 3.16 – 3.08 (m, 1H), 2.56 – 2.44 (m, 1H), 2.38 – 2.27 (m, 2H), 2.16 – 2.04 (m, 3H), 2.03 – 1.87 (m, 2H), 1.77 – 1.51 (m, 6H), 1.48 (d, $J = 7.2$ Hz, 3H), 1.44 – 1.28 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 220.30, 174.74, 139.00, 138.50, 129.21, 127.61, 115.13 (dt, $J = 4.9, 238.6$ Hz), 89.94 – 88.08 (m), 64.34, 51.07, 45.27, 40.40 – 39.60 (m), 38.29, 35.29, 34.69 (d, $J = 20.5$ Hz), 29.35, 28.26, 21.29 (d, $J = 4.5$ Hz), 20.65, 18.53. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -116.00 – -118.41 (m, 2F), -183.54 – -183.61 (m, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{22}\text{H}_{29}\text{F}_3\text{O}_3\text{Na}^+]$: 421.1961, found 421.1972.



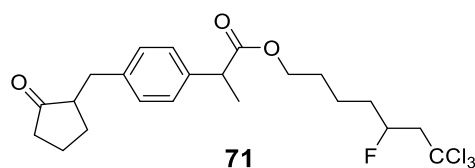
69

2-(4-(10-Chloro-8,10,10-trifluoro-3-oxodecan-2-yl)benzyl)cyclopentan-1-one (69): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a

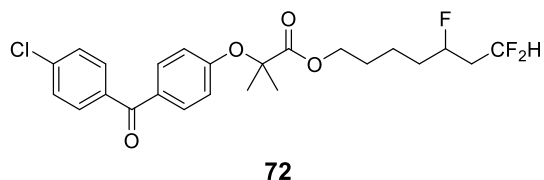
colorless liquid (47.1 mg) at 57% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.23 – 7.09 (m, 4H), 4.78 (dt, $J = 49.2, 7.3, 3.1$ Hz, 1H), 4.07 (t, $J = 6.3$ Hz, 2H), 3.68 (q, $J = 7.1$ Hz, 1H), 3.16 – 3.06 (m, 1H), 2.75 – 2.59 (m, 1H), 2.54 – 2.27 (m, 4H), 2.16 – 2.01 (m, 2H), 2.00 – 1.89 (m, 1H), 1.79 – 1.31 (m, 11H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 220.27, 174.73, 139.01, 138.49, 129.21, 127.80 (dt, $J = 2.8, 292.4$ Hz), 127.60, 88.23 (dt, $J = 173.3, 3.0$ Hz), 64.27, 51.06, 47.40 – 46.60 (m), 45.26, 38.28, 35.30, 34.59 (d, $J = 21.0$ Hz), 29.35, 28.20, 21.16 (d, $J = 4.3$ Hz), 20.64, 18.53. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -48.50 – -50.44 (m, 2F), -181.68 – -181.76 (m, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{22}\text{H}_{28}\text{ClF}_3\text{O}_3\text{Na}^+]$: 455.1571, found 455.1572.



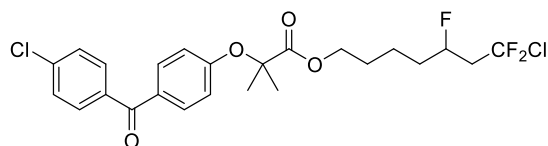
2-(4-(10,10-Dichloro-8-fluoro-3-oxodecan-2-yl)benzyl)cyclopentan-1-one (70): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (44.6 mg) at 54% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.23 – 7.09 (m, 4H), 5.93 – 5.81 (m, 1H), 4.76 – 4.56 (m, 1H), 4.07 (t, $J = 6.4$ Hz, 2H), 3.68 (q, $J = 7.1$ Hz, 1H), 3.17 – 3.08 (m, 1H), 2.54 – 2.45 (m, 2H), 2.37 – 2.20 (m, 3H), 2.15 – 2.04 (m, 2H), 2.00 – 1.90 (m, 1H), 1.80 – 1.51 (m, 6H), 1.48 (d, $J = 7.2$ Hz, 3H), 1.45 – 1.27 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 220.25, 174.73, 138.99, 138.49, 129.22, 127.60, 90.54 (d, $J = 169.7$ Hz), 69.69 (d, $J = 4.5$ Hz), 64.33, 51.07, 49.27 (d, $J = 20.8$ Hz), 45.26, 38.29, 35.32, 34.40 – 34.08 (m), 29.35, 28.29, 21.35 (d, $J = 4.5$ Hz), 20.66, 18.54. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -185.17 – -185.27 (m, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{22}\text{H}_{29}\text{Cl}_2\text{FO}_3\text{Na}^+]$: 453.1370, found 453.1374.



2-(4-(10,10,10-Trichloro-8-fluoro-3-oxodecan-2-yl)benzyl)cyclopentan-1-one (71): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (41.6 mg) at 46% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.23 – 7.09 (m, 4H), 4.98 – 4.78 (m, 1H), 4.08 (t, $J = 6.4$ Hz, 2H), 3.68 (q, $J = 7.1$ Hz, 1H), 3.17 – 3.04 (m, 2H), 2.89 – 2.78 (m, 1H), 2.55 – 2.45 (m, 1H), 2.38 – 2.28 (m, 2H), 2.16 – 2.02 (m, 2H), 2.01 – 1.90 (m, 1H), 1.79 – 1.68 (m, 2H), 1.65 – 1.50 (m, 4H), 1.48 (d, $J = 7.2$ Hz, 3H), 1.46 – 1.30 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 220.26, 174.73, 139.00, 138.49, 129.22, 127.61, 96.63, 90.29 (d, $J = 174.4$ Hz), 64.32, 59.56 (d, $J = 22.5$ Hz), 51.08, 45.27, 38.29, 35.32, 35.03 (d, $J = 21.3$ Hz), 29.36, 28.24, 21.21 (d, $J = 4.5$ Hz), 20.66, 18.56. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -178.32 – -178.36 (m, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{22}\text{H}_{28}\text{Cl}_3\text{FO}_3\text{Na}^+]$: 487.0980, found 487.0988.

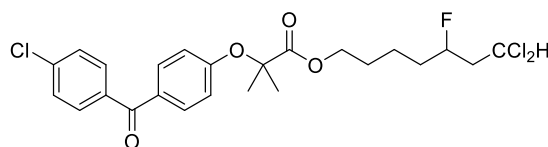


5,7,7-Trifluoroheptyl 2-(4-(4-chlorobenzoyl)phenoxy)-2-methylpropanoate (72): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (28.1 mg) at 30% yield. $^1\text{H NMR}$ (400 MHz, DMSO-*d*₆) δ 7.77 – 7.58 (m, 6H), 6.90 (d, $J = 8.7$ Hz, 2H), 6.18 – 6.06 (m, 1H), 4.68 – 4.54 (m, 1H), 4.13 (t, $J = 6.3$ Hz, 2H), 2.21 – 1.98 (m, 2H), 1.65 – 1.44 (m, 10H), 1.37 – 1.17 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 194.29, 173.87, 159.78, 138.53, 136.46, 132.16, 131.28, 130.49, 128.67, 117.21, 115.08 (dt, $J = 4.9, 238.7$ Hz), 89.86 – 87.96 (m), 79.52, 65.44, 40.35 – 39.60 (m), 34.68 (d, $J = 20.4$ Hz), 28.18, 25.56 (d, $J = 4.3$ Hz), 21.36 (d, $J = 4.3$ Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -115.96 – -118.38 (m, 2F), -183.71 (s, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{24}\text{H}_{26}\text{ClF}_3\text{O}_4\text{H}^+]$: 471.1544, found 471.1542.



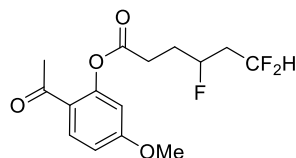
73

7-Chloro-5,7,7-trifluoroheptyl 2-(4-(4-chlorobenzoyl)phenoxy)-2-methylpropanoate (73): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (41.0 mg) at 41% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.77 – 7.65 (m, 4H), 7.47 – 7.40 (m, 2H), 6.89 – 6.81 (m, 2H), 4.80 – 4.68 (m, 1H), 4.17 (t, J = 6.4 Hz, 2H), 2.71–2.60 (m, 1H), 2.53 – 2.30 (m, 1H), 1.67 (s, 6H), 1.68 – 1.26 (m, 6H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 194.26, 173.84, 159.76, 138.51, 136.42, 132.13, 131.26, 130.48, 128.65, 127.72 (dt, J = 3.0, 292.5 Hz), 117.19, 88.17 (dt, J = 173.3, 3.0 Hz), 79.50, 65.35, 47.70 – 46.50 (m), 34.56 (d, J = 20.9 Hz), 28.12, 25.54 (d, J = 4.8 Hz), 21.22 (d, J = 4.1 Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -48.63 – -50.33 (m, 2F), -181.92 – -182.00 (m, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{24}\text{H}_{25}\text{Cl}_2\text{F}_3\text{O}_4\text{H}^+]$: 505.1155, found 505.1153.



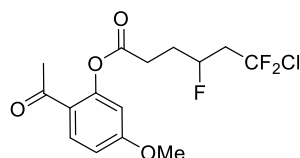
74

7,7-Dichloro-5-fluoroheptyl 2-(4-(4-chlorobenzoyl)phenoxy)-2-methylpropanoate (74): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (50.8 mg) at 51% yield. $^1\text{H NMR}$ (400 MHz, DMSO-*d*₆) δ 7.77 – 7.58 (m, 6H), 6.90 (d, J = 8.8 Hz, 2H), 6.33 – 6.21 (m, 1H), 4.67 – 4.45 (m, 1H), 4.13 (t, J = 6.2 Hz, 2H), 2.47 – 2.27 (m, 2H), 1.54 (m, 10H), 1.26 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, DMSO-*d*₆) δ 193.18, 172.77, 159.23, 137.07, 136.23, 131.93, 131.18, 129.57, 128.61, 117.05, 90.68 (d, J = 168.6 Hz), 79.11, 71.05 (d, J = 4.4 Hz), 65.04, 47.80 (d, J = 20.6 Hz), 33.26 (d, J = 19.8 Hz), 27.56, 25.11, 20.70 (d, J = 4.4 Hz). $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -178.15 (s, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{24}\text{H}_{26}\text{Cl}_3\text{FO}_4\text{H}^+]$: 503.0953, found 503.0953.



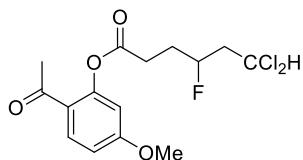
75

2-Acetyl-5-methoxyphenyl 4,6,6-trifluorohexanoate (75): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (19.8 mg) at 31% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.83 (d, J = 8.8 Hz, 1H), 6.86 – 6.80 (m, 1H), 6.60 (d, J = 2.5 Hz, 1H), 6.07 – 5.96 (m, 1H), 4.99 – 4.79 (m, 1H), 3.86 (s, 3H), 2.90 – 2.75 (m, 2H), 2.50 (s, 3H), 2.36 – 2.07 (m, 4H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 195.85, 171.37, 163.97, 151.39, 132.84, 122.78, 115.02 (dt, J = 5.0, 238.9 Hz), 111.82, 109.53, 89.31 – 87.42 (m), 55.89, 40.45 – 39.75 (m), 30.14 (d, J = 20.7 Hz), 29.78 (d, J = 4.4 Hz), 28.90. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -115.89 – -118.23 (m, 2F), -185.72 – -185.75 (m, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{17}\text{F}_3\text{O}_4\text{Na}^+]$: 341.0971, found 341.0988.



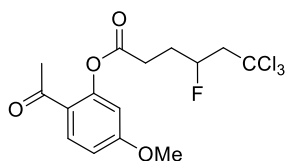
76

2-Acetyl-5-methoxyphenyl 6-chloro-4,6,6-trifluorohexanoate (76): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (36.2 mg) at 51% yield. $^1\text{H NMR}$ (400 MHz, DMSO- d_6) δ 7.93 (d, J = 8.8 Hz, 1H), 6.98 – 6.93 (m, 1H), 6.78 (d, J = 2.4 Hz, 1H), 5.11 – 4.91 (m, 1H), 3.82 (s, 3H), 3.05 – 2.87 (m, 2H), 2.76 (t, J = 7.5 Hz, 2H), 2.45 (s, 3H), 2.13 – 1.96 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, Chloroform- d) δ 195.82, 171.26, 163.97, 151.36, 132.86, 127.67 (dt, J = 2.6, 293.4 Hz), 122.69, 111.82, 109.52, 87.62 (dt, J = 173.5, 3.1 Hz), 55.87, 47.09 (td, J = 24.3, 22.2 Hz), 30.06 (d, J = 20.9 Hz), 29.61 (d, J = 4.2 Hz), 28.85. $^{19}\text{F NMR}$ (376 MHz, Chloroform- d) δ -48.56 – -50.43 (m, 2F), -183.95 – -184.03 (m, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{16}\text{ClF}_3\text{O}_4\text{Na}^+]$: 375.0581, found 375.0587.



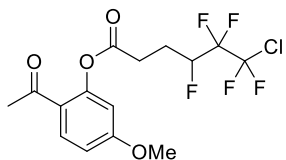
77

2-Acetyl-5-methoxyphenyl 6,6-dichloro-4-fluorohexanoate (77): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (30.1 mg) at 43% yield. $^1\text{H NMR}$ (400 MHz, Chloroform- d) δ 7.83 (d, J = 8.8 Hz, 1H), 6.87 – 6.79 (m, 1H), 6.60 (d, J = 2.5 Hz, 1H), 5.96 – 5.87 (m, 1H), 5.00 – 4.79 (m, 1H), 3.85 (s, 3H), 2.90 – 2.75 (m, 2H), 2.70 – 2.57 (m, 1H), 2.50 (s, 3H), 2.49 – 2.35 (m, 1H), 2.25 – 2.05 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, Chloroform- d) δ 195.80, 171.32, 163.94, 151.37, 132.83, 122.73, 111.78, 109.52, 89.91 (d, J = 170.4 Hz), 69.46 (d, J = 4.5 Hz), 55.87, 49.22 (d, J = 20.5 Hz), 29.79, 29.71 (d, J = 25.5 Hz), 28.88. $^{19}\text{F NMR}$ (376 MHz, Chloroform- d) δ -187.26 (s, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{17}\text{Cl}_2\text{FO}_4\text{Na}^+]$: 373.0380, found 373.0384.



78

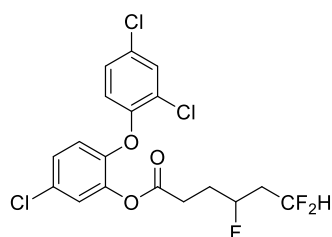
2-Acetyl-5-methoxyphenyl 6,6,6-trichloro-4-fluorohexanoate (78): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (40.8 mg) at 53% yield. $^1\text{H NMR}$ (400 MHz, Chloroform- d) δ 7.83 (d, J = 8.8 Hz, 1H), 6.88 – 6.78 (m, 1H), 6.61 (d, J = 2.5 Hz, 1H), 5.22 – 5.02 (m, 1H), 3.85 (s, 3H), 3.29 – 3.18 (m, 1H), 3.05 – 2.96 (m, 1H), 2.92 – 2.79 (m, 2H), 2.50 (s, 3H), 2.31 – 2.14 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, Chloroform- d) δ 195.79, 171.27, 163.96, 151.37, 132.84, 122.71, 111.81, 109.52, 96.35, 89.64 (d, J = 174.7 Hz), 59.54 (d, J = 22.2 Hz), 55.87, 30.46 (d, J = 21.4 Hz), 29.70 (d, J = 4.4 Hz), 28.88. $^{19}\text{F NMR}$ (376 MHz, Chloroform- d) δ -180.74 (s, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{15}\text{H}_{16}\text{Cl}_3\text{FO}_4\text{Na}^+]$: 406.9990, found 407.0002.



79

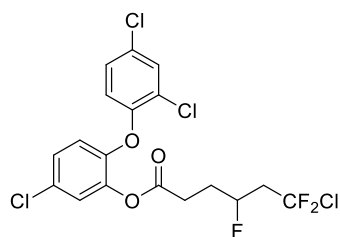
2-Acetyl-5-methoxyphenyl 7-chloro-4,6,6,7,7-pentafluoroheptanoate (79): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (33.7 mg) at 42% yield. $^1\text{H NMR}$ (400 MHz, Chloroform- d) δ 7.83 (d, J = 8.8 Hz, 1H), 6.88 – 6.78 (m, 1H), 6.61 (d, J = 2.5 Hz, 1H), 5.20 – 5.00 (m, 1H), 3.86 (s, 3H), 2.94 – 2.78 (m, 2H), 2.69 – 2.53 (m, 1H), 2.50 (s, 3H), 2.47 – 2.28 (m, 1H), 2.27 – 2.11 (m, 2H). $^{13}\text{C NMR}$ (101 MHz, Chloroform- d) δ 195.80, 171.24, 163.97, 151.36, 132.85, 123.45 (tt, J = 298.9, 37.1 Hz), 122.68, 118.99 – 113.10 (m), 111.81, 109.51, 86.40 (dt, J = 173.6, 2.7 Hz), 55.86, 36.70 – 35.95 (m), 30.54 (d, J = 20.9 Hz), 29.65 (d, J = 4.3 Hz), 28.82. $^{19}\text{F NMR}$ (376 MHz, Chloroform- d) δ -71.68 – -71.72

(m, 2F), -112.80 – -113.10 (m, 2F), -182.64 – -182.76 (m, 1F). **ESI-HRMS** exact mass calculated for $[C_{16}H_{16}ClF_5O_4Na^+]$: 425.0549, found 425.0551.



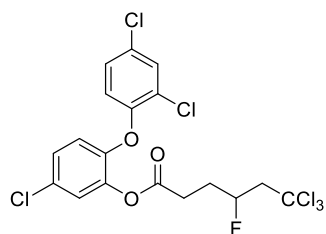
80

5-Chloro-2-(2,4-dichlorophenoxy)phenyl 4,6,6-trifluorohexanoate (80): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (28.5 mg) at 32% yield. 1H NMR (400 MHz, DMSO- d_6) δ 7.79 – 7.73 (m, 1H), 7.54 – 7.48 (m, 1H), 7.44 – 7.34 (m, 2H), 7.12 (d, J = 8.8 Hz, 1H), 6.97 (d, J = 8.8 Hz, 1H), 6.21 – 6.11 (m, 1H), 4.89 – 4.67 (m, 1H), 2.64 (t, J = 7.5 Hz, 2H), 2.32 – 2.06 (m, 2H), 2.02 – 1.83 (m, 2H). ^{13}C NMR (101 MHz, DMSO- d_6) δ 170.14, 150.65, 146.20, 141.46, 130.08, 128.80, 128.49, 128.27, 127.38, 124.62, 124.55, 121.19, 120.45, 115.78 (dt, J = 4.5, 237.3 Hz), 88.14 (dt, J = 167.5, 6.8 Hz), 38.94 – 38.24 (m), 29.49 (d, J = 20.5 Hz), 28.65 (d, J = 4.6 Hz). ^{19}F NMR (376 MHz, DMSO- d_6) δ -115.67 – -115.73 (m, 2F), -183.53 (t, J = 3.5 Hz, 1F). **ESI-HRMS** exact mass calculated for $[C_{18}H_{14}Cl_3F_3O_3Na^+]$: 462.9853, found 462.9852.



81

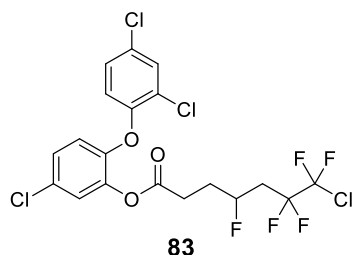
5-Chloro-2-(2,4-dichlorophenoxy)phenyl 6-chloro-4,6,6-trifluorohexanoate (81): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (55.8 mg) at 59% yield. 1H NMR (400 MHz, DMSO- d_6) δ 7.76 – 7.70 (m, 1H), 7.52 – 7.46 (m, 1H), 7.43 – 7.32 (m, 2H), 7.08 (d, J = 8.8 Hz, 1H), 6.96 (d, J = 8.8 Hz, 1H), 5.01 – 4.76 (m, 1H), 2.98 – 2.78 (m, 2H), 2.66 (t, J = 7.4 Hz, 2H), 2.06 – 1.83 (m, 2H). ^{13}C NMR (101 MHz, DMSO- d_6) δ 170.27, 150.72, 146.39, 141.54, 131.49 – 125.60 (m), 130.21, 128.92, 128.71, 128.39, 127.53, 124.85, 124.63, 121.16, 120.67, 87.79 (dt, J = 171.6, 3.3 Hz), 45.38 (td, J = 23.2, 21.0 Hz), 29.26 (d, J = 20.6 Hz), 28.58 (d, J = 4.6 Hz). ^{19}F NMR (376 MHz, DMSO- d_6) δ -46.48 – -48.87 (m, 2F), -183.61 (t, J = 9.9 Hz, 1F). **ESI-HRMS** exact mass calculated for $[C_{18}H_{13}Cl_4F_3O_3H^+]$: 474.9644, found 474.9652.



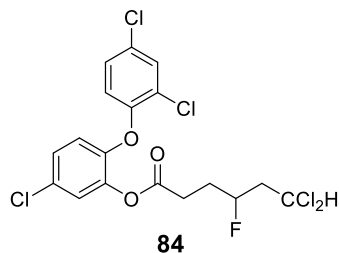
82

5-Chloro-2-(2,4-dichlorophenoxy)phenyl 6,6,6-trichloro-4-fluorohexanoate (82): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (50.3 mg) at 50% yield. 1H NMR (400 MHz, Chloroform- d) δ 7.49 – 7.43 (m, 1H), 7.23 – 7.14 (m, 3H), 6.92 – 6.80 (m, 2H), 5.11 – 4.92 (m, 1H), 3.23 – 3.13 (m, 1H), 2.97 – 2.86 (m, 1H), 2.73 (t, J = 7.1 Hz, 2H), 2.14 – 2.00 (m, 2H). ^{13}C NMR (101 MHz, Chloroform- d) δ 170.10, 150.92, 146.67, 141.59, 130.56, 129.74, 129.36, 128.31, 127.25, 126.02, 124.41, 120.51, 120.17, 96.12, 89.30 (d, J = 175.3 Hz), 59.41 (d, J = 22.4 Hz), 30.52 (d, J = 21.4 Hz), 29.18 (d, J = 4.4 Hz).

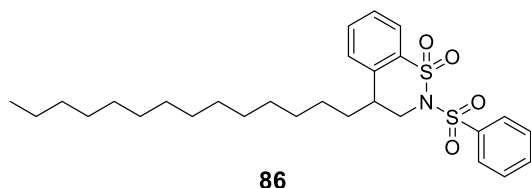
^{19}F NMR (376 MHz, Chloroform-*d*) δ -180.68 (s, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{18}\text{H}_{13}\text{Cl}_6\text{FO}_3\text{Na}^+]$: 528.8872, found 528.8871.



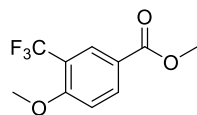
5-Chloro-2-(2,4-dichlorophenoxy)phenyl 7-chloro-4,6,6,7,7-pentafluoroheptanoate (83): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (68.5 mg) at 65% yield. ^1H NMR (400 MHz, DMSO-*d*₆) δ 7.77 – 7.71 (m, 1H), 7.53 – 7.47 (m, 1H), 7.44 – 7.33 (m, 2H), 7.09 (d, *J* = 8.8 Hz, 1H), 6.98 (d, *J* = 8.8 Hz, 1H), 5.09 – 4.86 (m, 1H), 2.79 – 2.53 (m, 4H), 2.13 – 1.88 (m, 2H). ^{13}C NMR (101 MHz, DMSO-*d*₆) δ 170.11, 150.61, 146.28, 141.44, 130.06, 128.76, 128.56, 128.24, 127.35, 126.50 – 119.66 (m), 124.73, 124.51, 121.02, 120.54, 119.35 – 113.48 (m), 87.46 – 85.55 (m), 34.69 (q, *J* = 21.0 Hz), 29.56 (d, *J* = 20.4 Hz), 28.54 (d, *J* = 4.4 Hz). ^{19}F NMR (376 MHz, DMSO-*d*₆) δ -71.28 – -71.34 (m, 2F), -110.53 – -113.46 (m, 2F), -182.25 – -182.40 (m, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{19}\text{H}_{13}\text{Cl}_4\text{F}_5\text{O}_3\text{H}^+]$: 524.9612, found 524.9628.



5-Chloro-2-(2,4-dichlorophenoxy)phenyl 6,6-dichloro-4-fluorohexanoate (84): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a colorless liquid (36.8 mg) at 39% yield. ^1H NMR (400 MHz, DMSO-*d*₆) δ 7.79 – 7.73 (m, 1H), 7.53 – 7.47 (m, 1H), 7.42–7.44 (m, 2H), 7.10 (d, *J* = 8.8 Hz, 1H), 6.98 (d, *J* = 8.8 Hz, 1H), 6.33 – 6.27 (m, 1H), 4.86 – 4.63 (m, 1H), 2.65 (t, *J* = 7.4 Hz, 2H), 2.61 – 2.39 (m, 2H), 2.04 – 1.82 (m, 2H). ^{13}C NMR (101 MHz, DMSO-*d*₆) δ 170.13, 150.62, 146.27, 141.41, 130.07, 128.80, 128.53, 128.21, 127.36, 124.72, 124.49, 121.06, 120.57, 89.78 (d, *J* = 169.2 Hz), 70.87 (d, *J* = 4.4 Hz), 47.56 (d, *J* = 20.2 Hz), 29.04 (d, *J* = 20.4 Hz), 28.63 (d, *J* = 4.6 Hz). ^{19}F NMR (376 MHz, DMSO-*d*₆) δ -184.75 (s, 1F). **ESI-HRMS** exact mass calculated for $[\text{C}_{18}\text{H}_{14}\text{Cl}_5\text{FO}_3\text{Na}^+]$: 494.9262, found 494.9261.



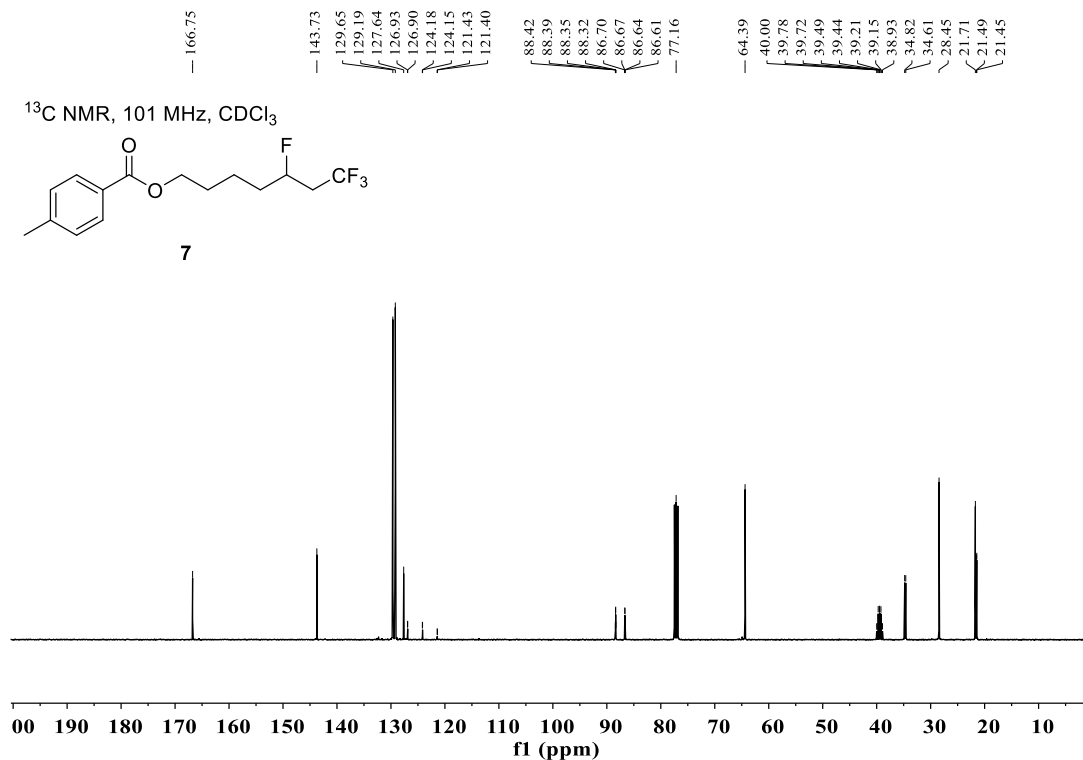
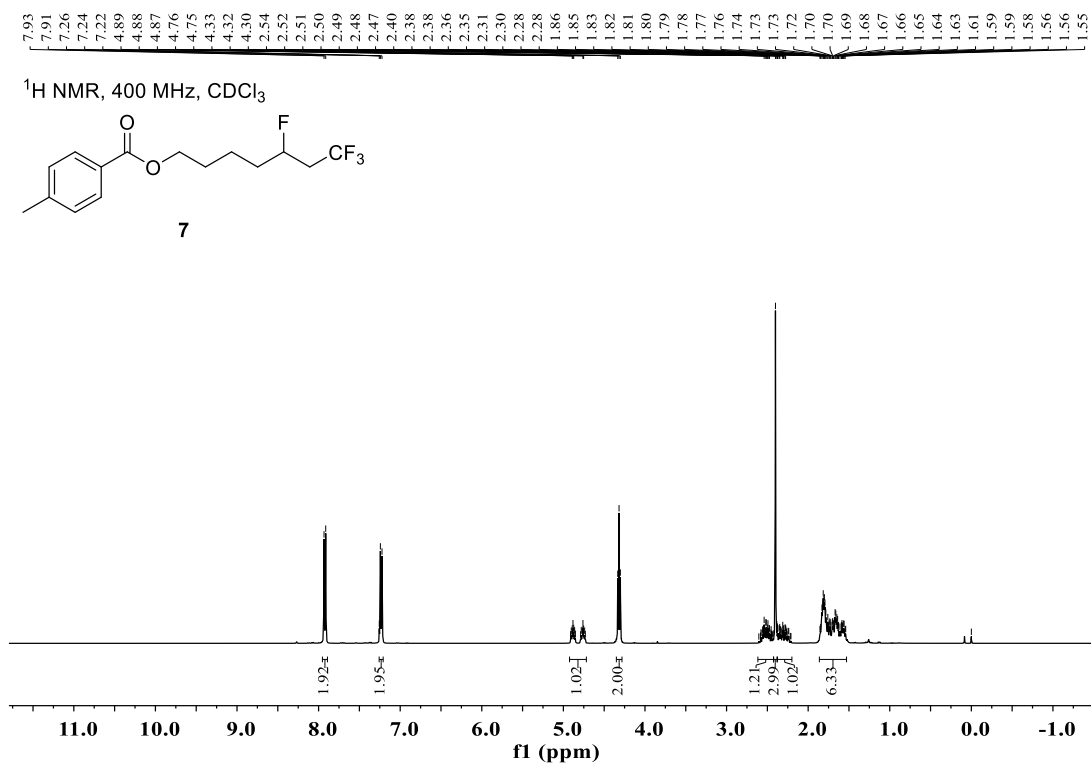
2-(phenylsulfonyl)-4-Tetradecyl-3,4-dihydro-2H-benzo[e][1,2]thiazine 1,1-dioxide (86): The compound was purified by column chromatography (petroleum ether/EtOAc = 60/1), resulting in a colorless liquid (27.0 mg) at 26% yield. ^1H NMR (400 MHz, Chloroform-*d*) δ 7.99 – 7.94 (m, 2H), 7.75 – 7.70 (m, 1H), 7.63 – 7.57 (m, 1H), 7.52 – 7.46 (m, 3H), 7.37 – 7.31 (m, 1H), 7.31 – 7.27 (m, 1H), 4.52 – 4.28 (m, 2H), 3.24 – 3.15 (m, 1H), 1.79 – 1.70 (m, 2H), 1.49 – 1.36 (m, 2H), 1.30 – 1.21 (m, 22H), 0.88 (t, *J* = 6.7 Hz, 3H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 139.26, 139.16, 137.21, 134.02, 132.94, 129.05, 129.00, 128.22, 127.32, 123.74, 50.02, 37.17, 34.24, 31.97, 29.75, 29.74, 29.71, 29.70, 29.68, 29.63, 29.62, 29.48, 29.41, 26.51, 22.74, 14.18. **ESI-HRMS** exact mass calculated for $[\text{C}_{28}\text{H}_{41}\text{NO}_4\text{S}_2\text{Na}^+]$: 542.2369; found: 542.2381.

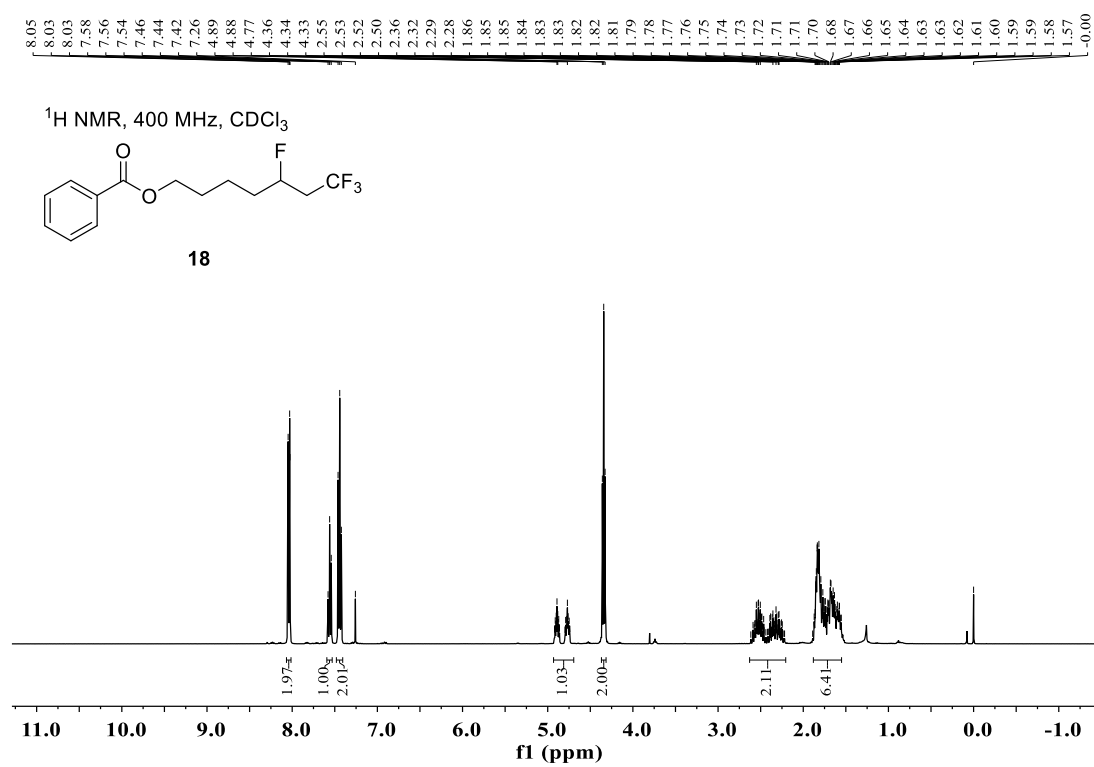
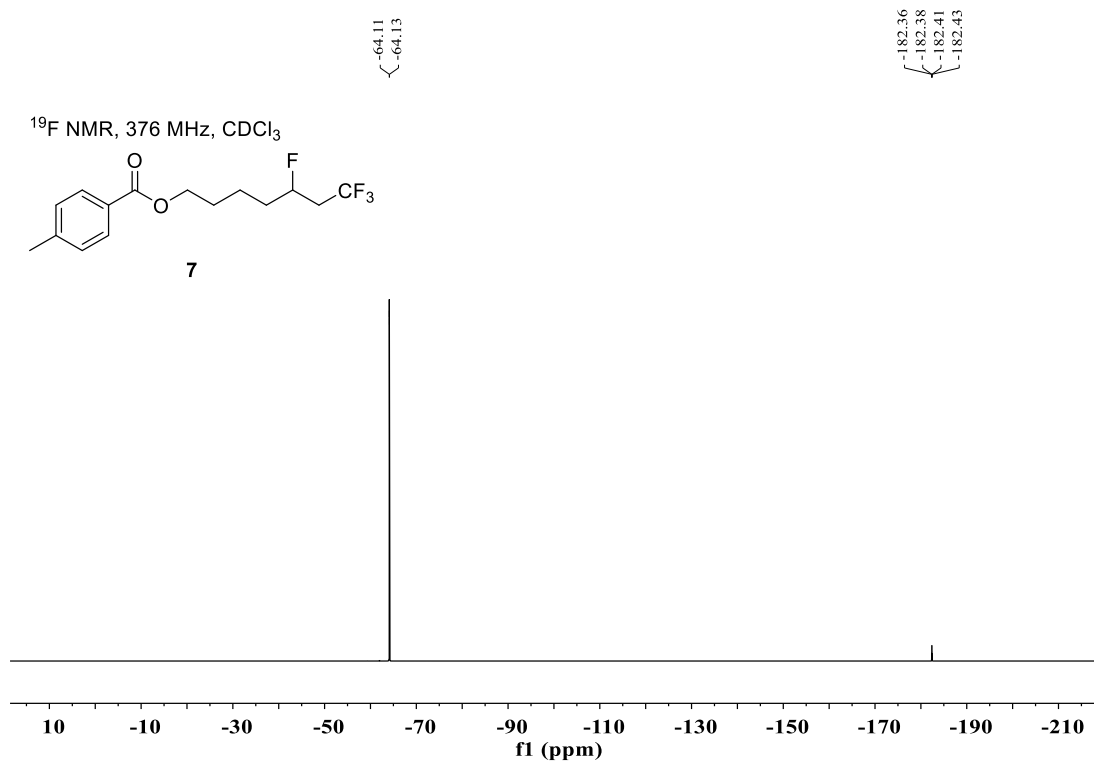


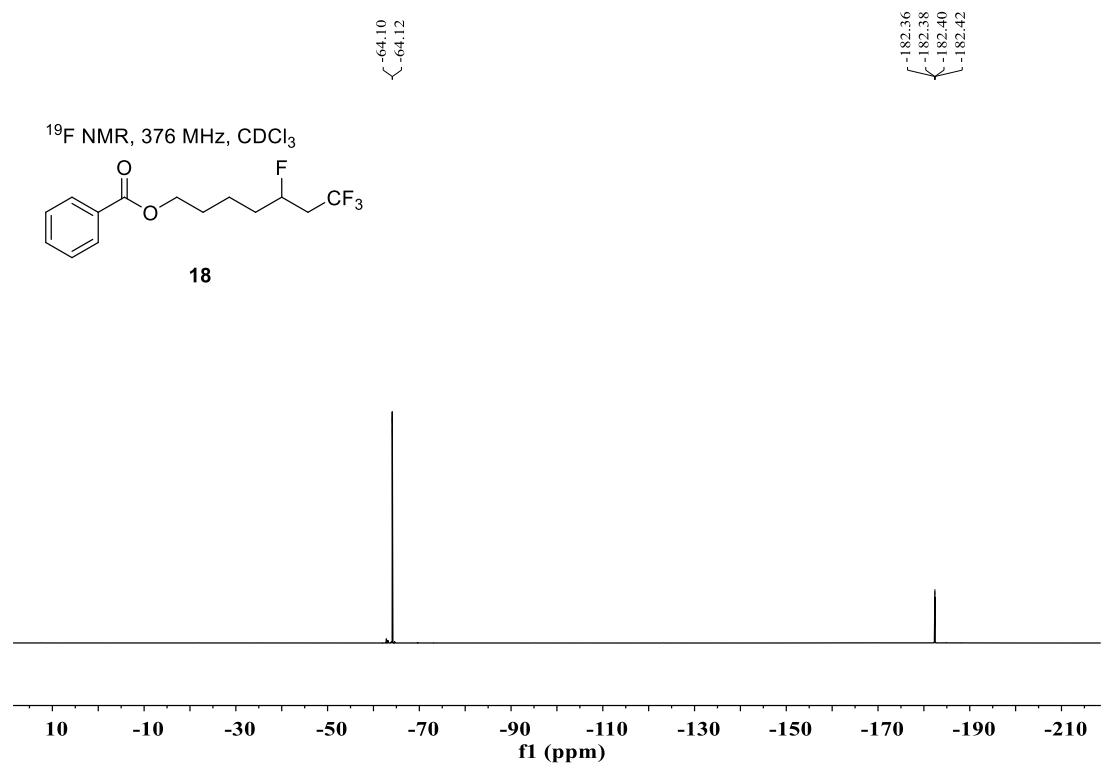
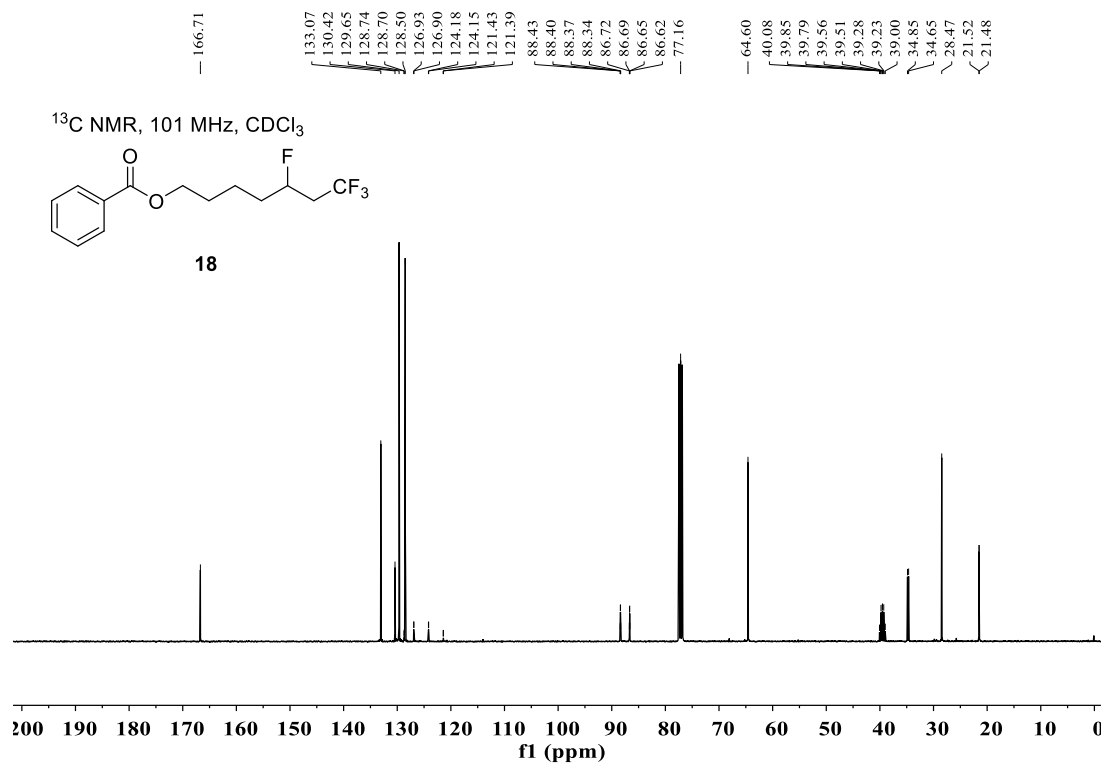
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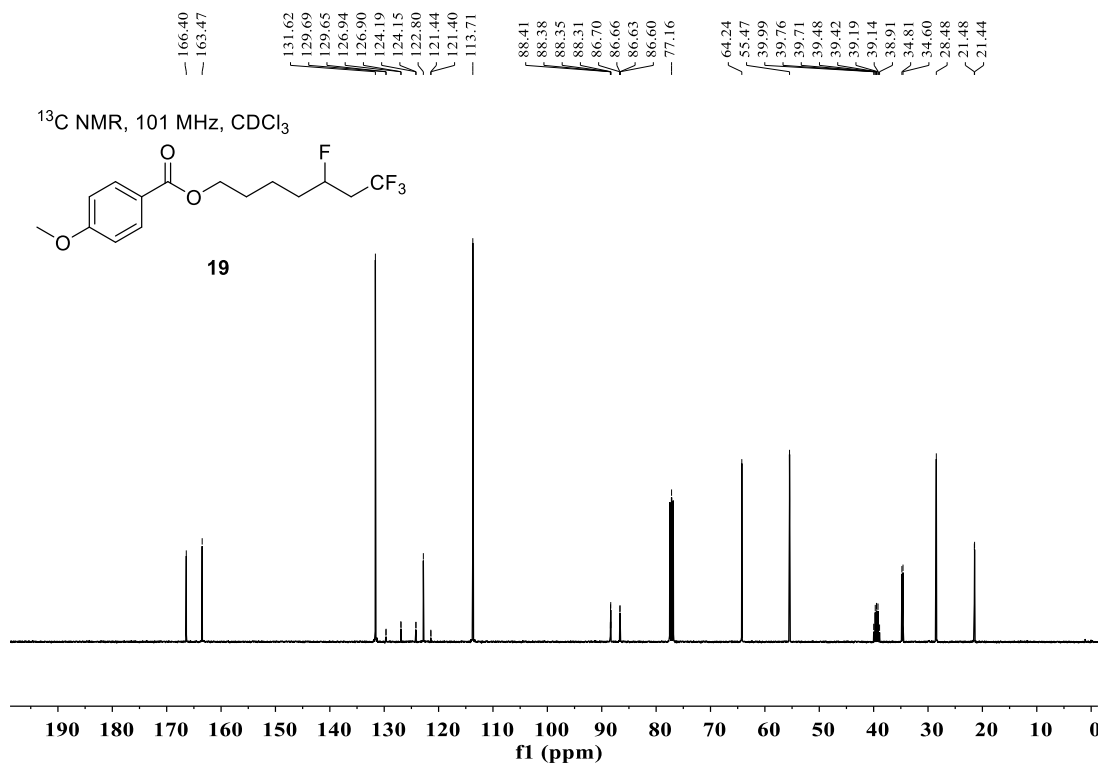
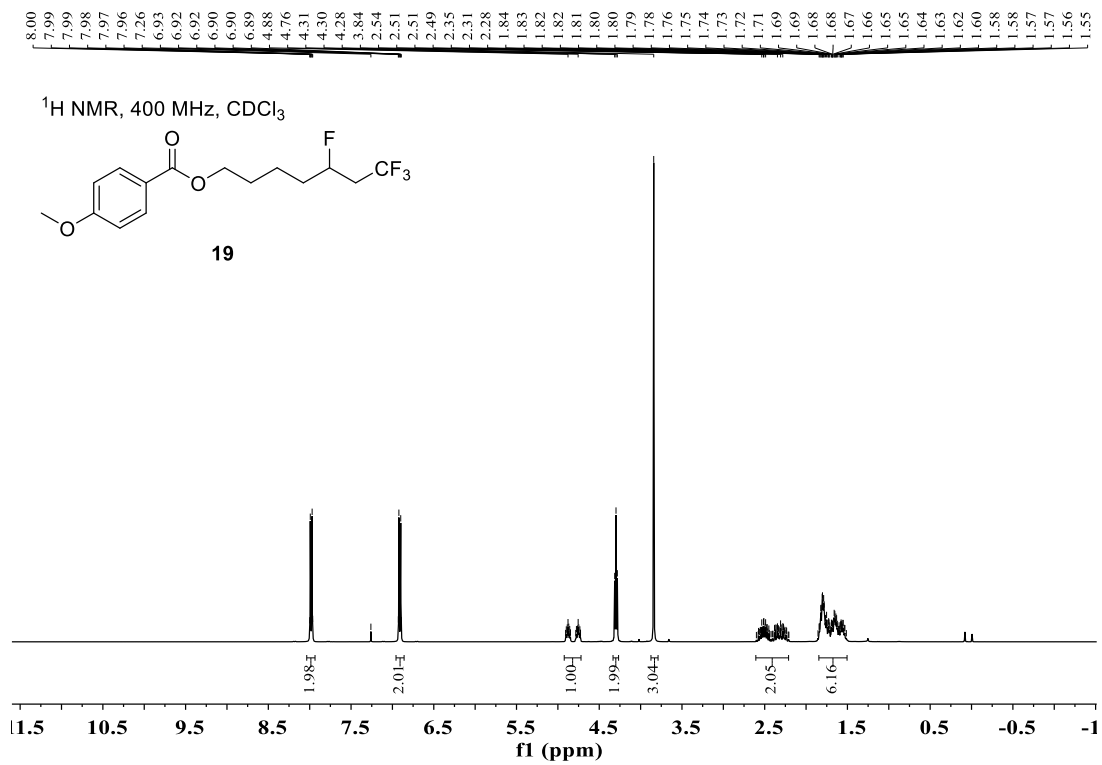
3-Methyl 4-methoxy-3-(trifluoromethyl)benzoate (103): The compound was purified by column chromatography (petroleum ether/EtOAc = 80/1), resulting in a solid with color (10.3 mg) at 22% yield. $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 8.2 – 8.25 (m, 1H), 8.21 – 8.17 (m, 1H), 7.03 (d, $J = 8.7$ Hz, 1H), 3.97 (s, 3H), 3.91 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, Chloroform-*d*) δ 165.92, 161.12 – 161.02 (m), 135.28, 129.16 (q, $J = 5.3$ Hz), 123.25 (q, $J = 272.5$ Hz), 122.29, 118.89 (q, $J = 31.6$ Hz), 111.66, 56.35, 52.34. $^{19}\text{F NMR}$ (376 MHz, Chloroform-*d*) δ -62.90 (s, 3F). **ESI-HRMS** exact mass calculated for $[\text{C}_{10}\text{H}_9\text{F}_3\text{O}_3\text{H}^+]$: 235.0577; found: 235.0577.

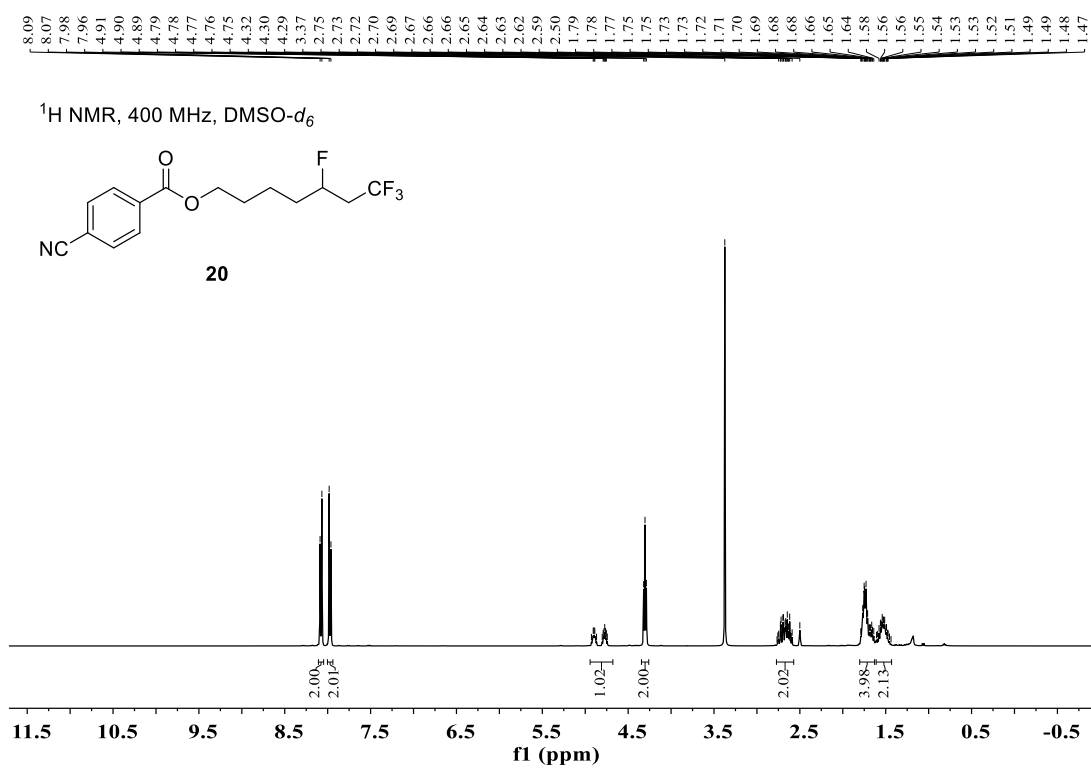
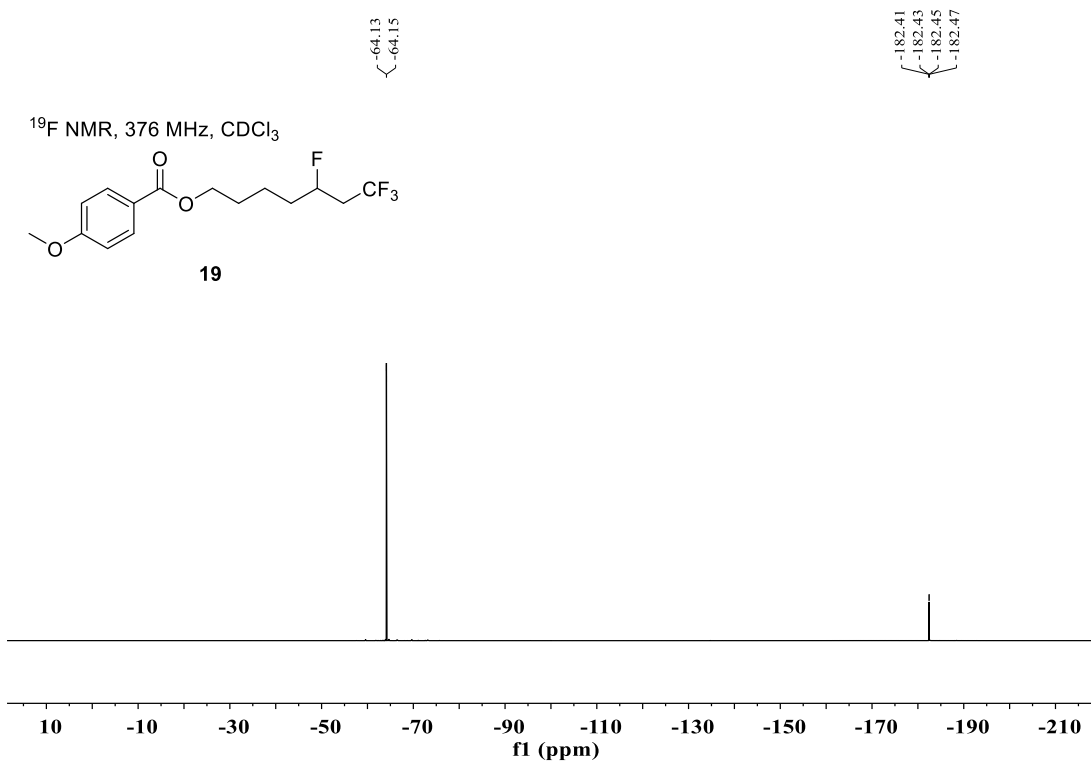
Copies of ^1H NMR, ^{13}C NMR and ^{19}F NMR spectra

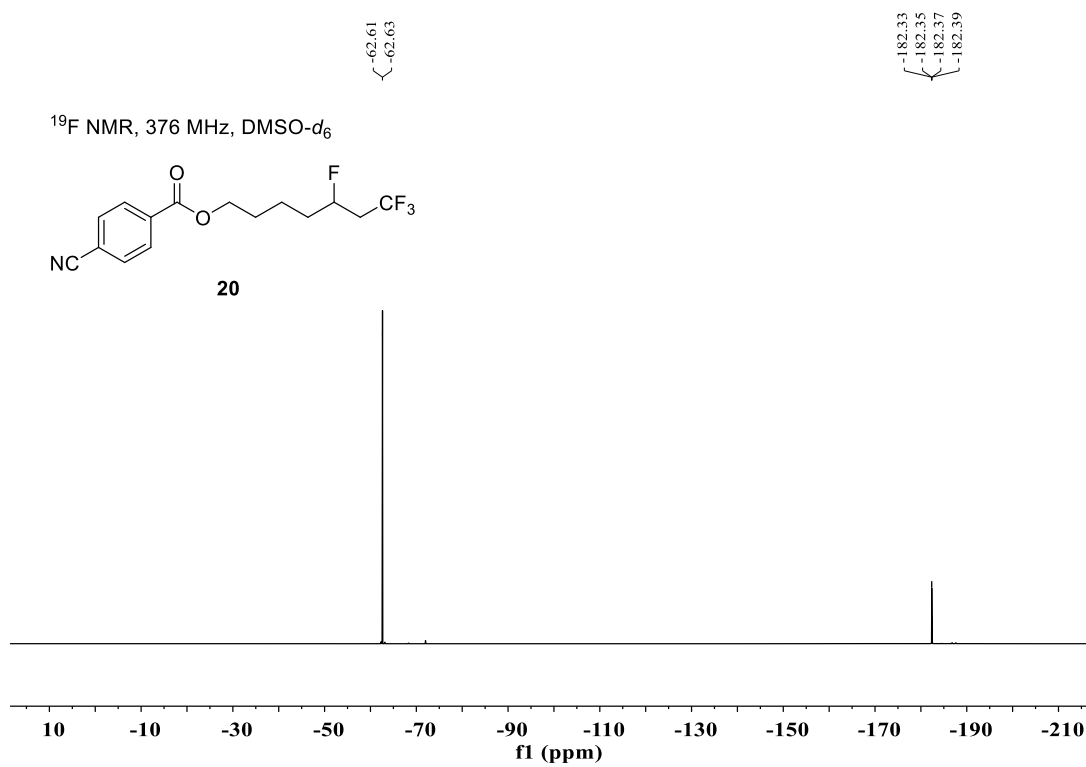
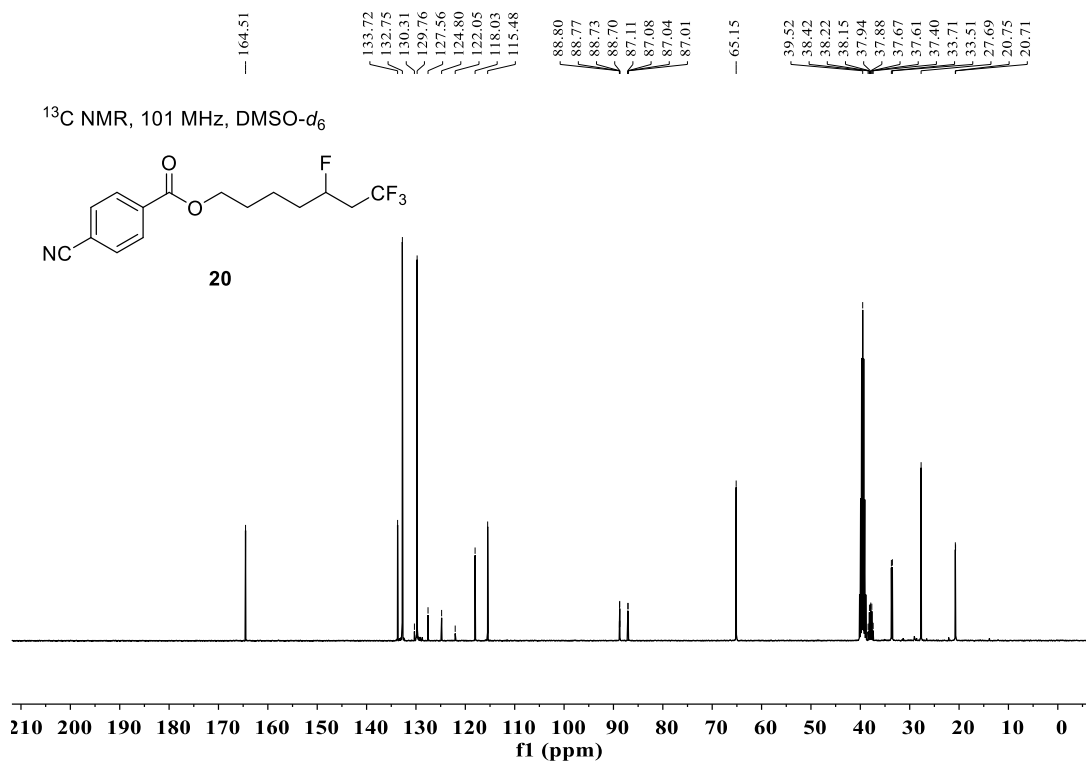


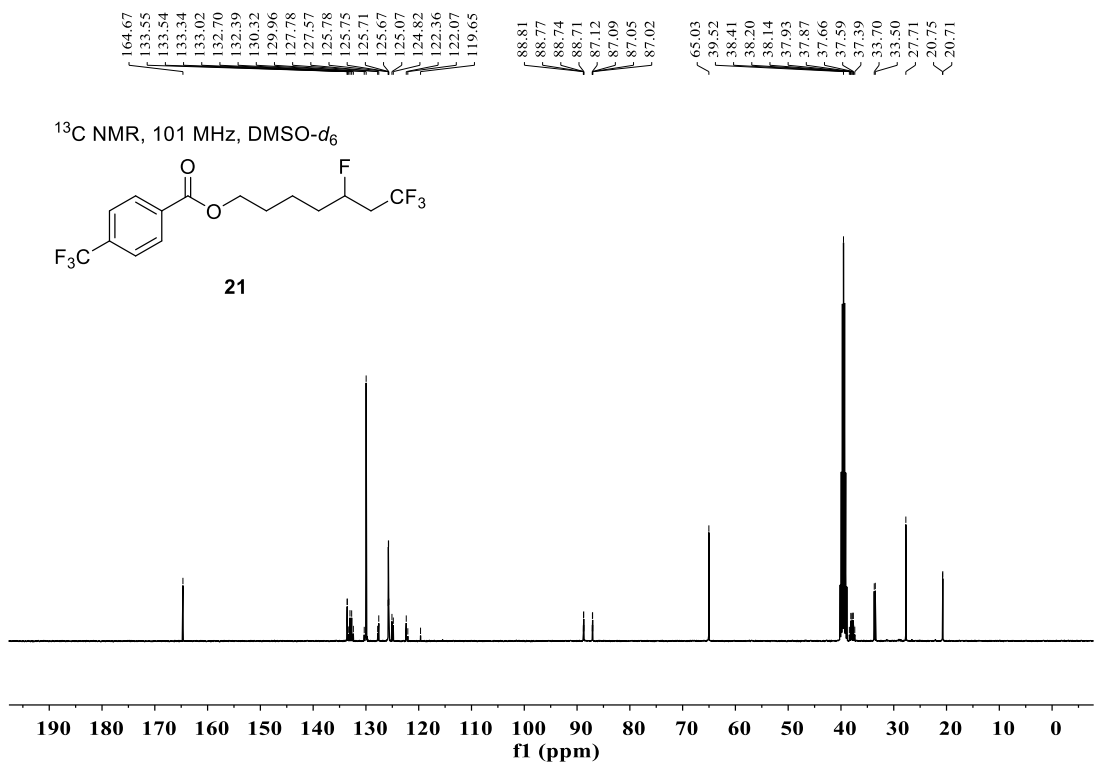
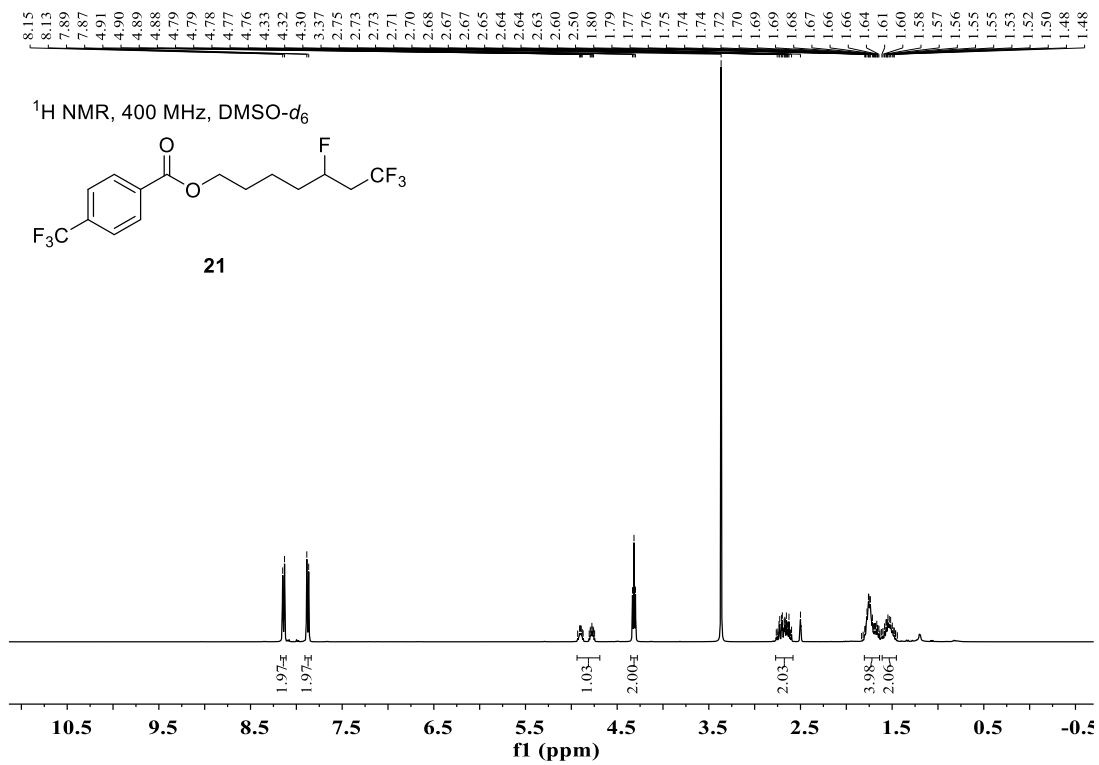


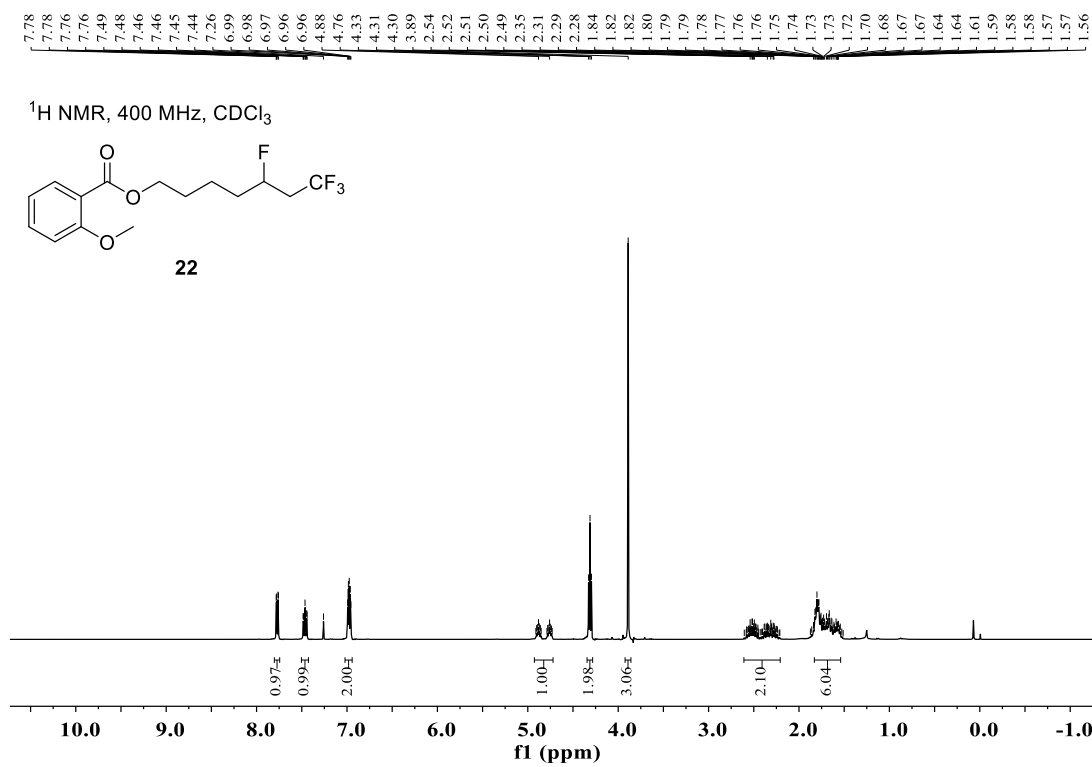
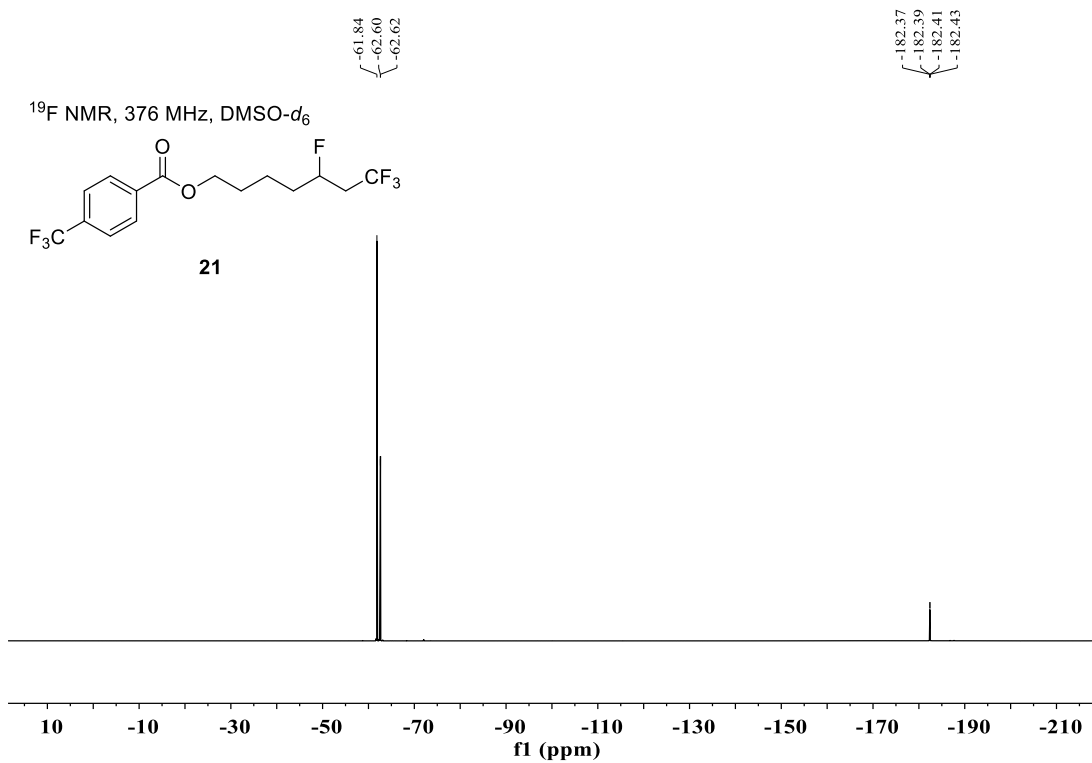


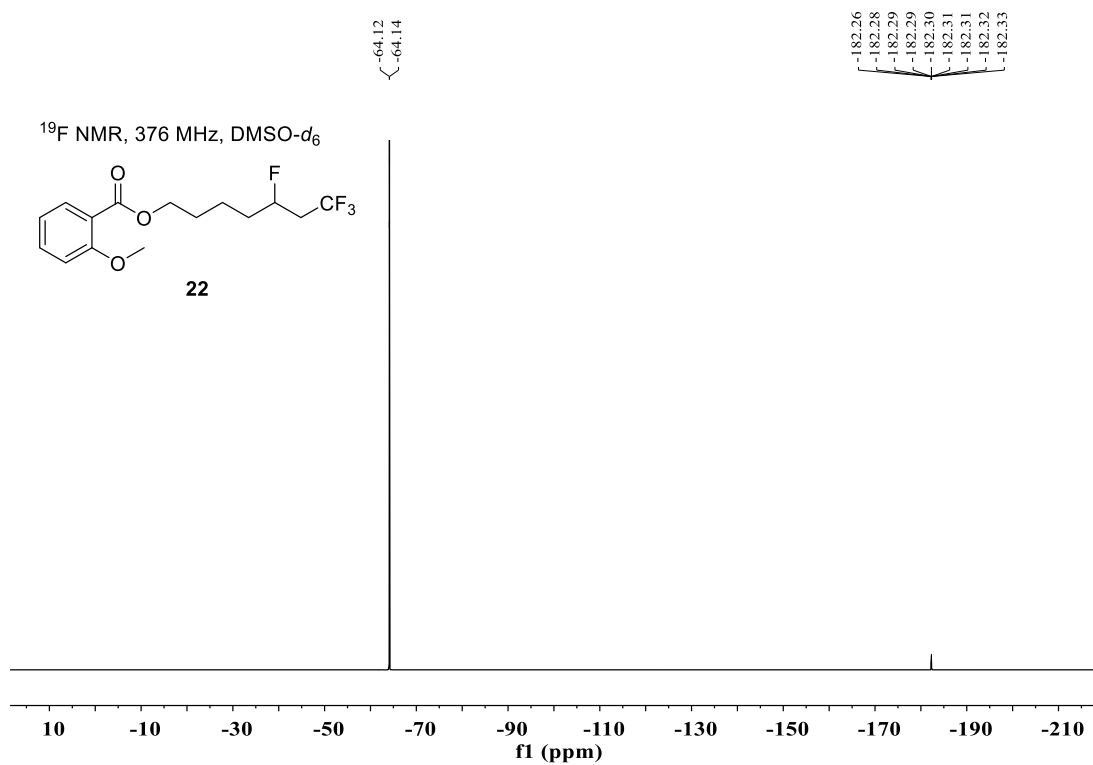
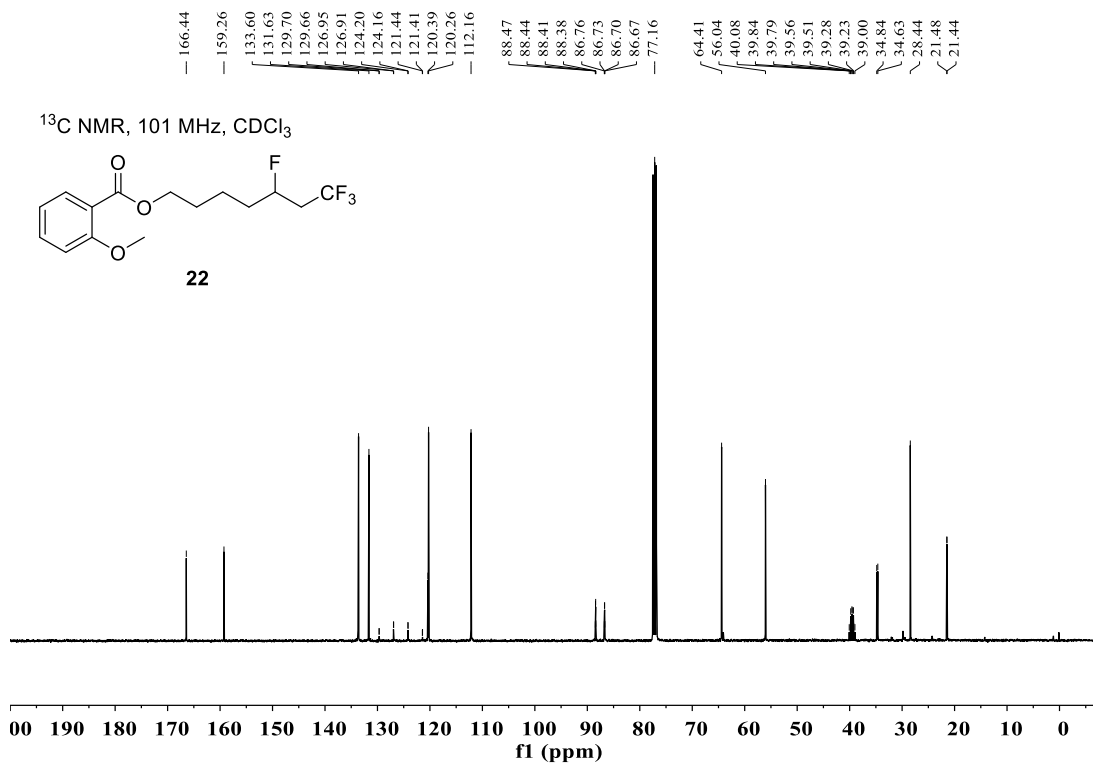


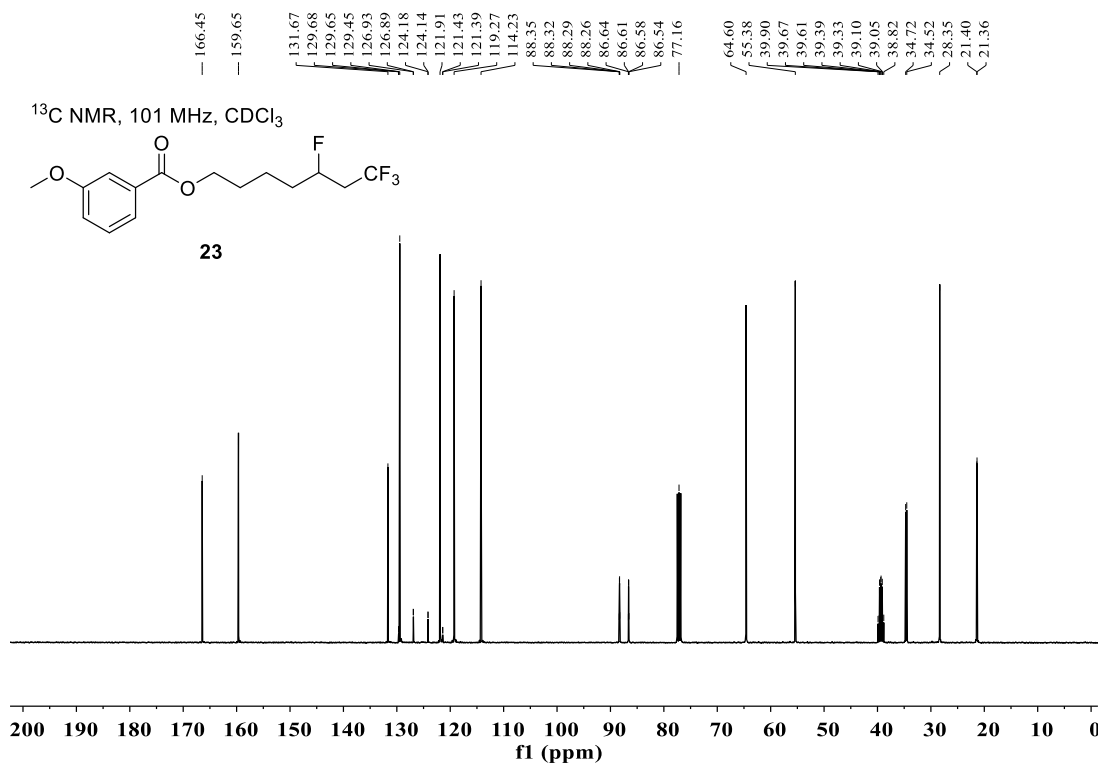
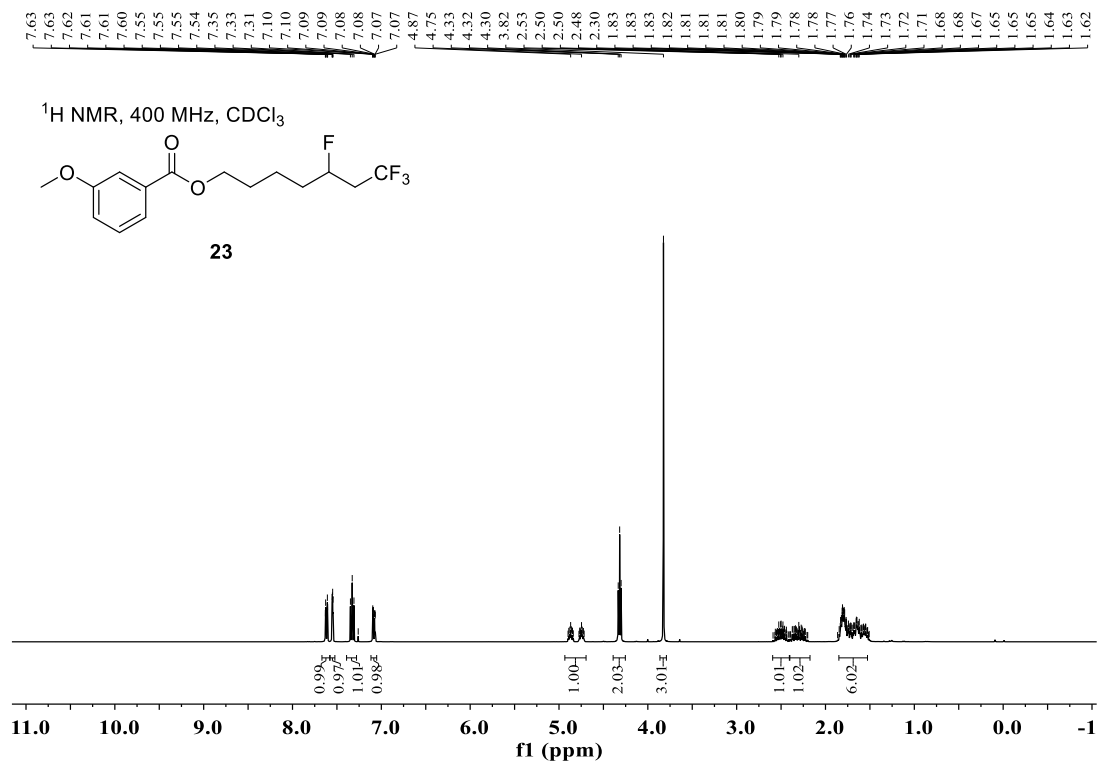




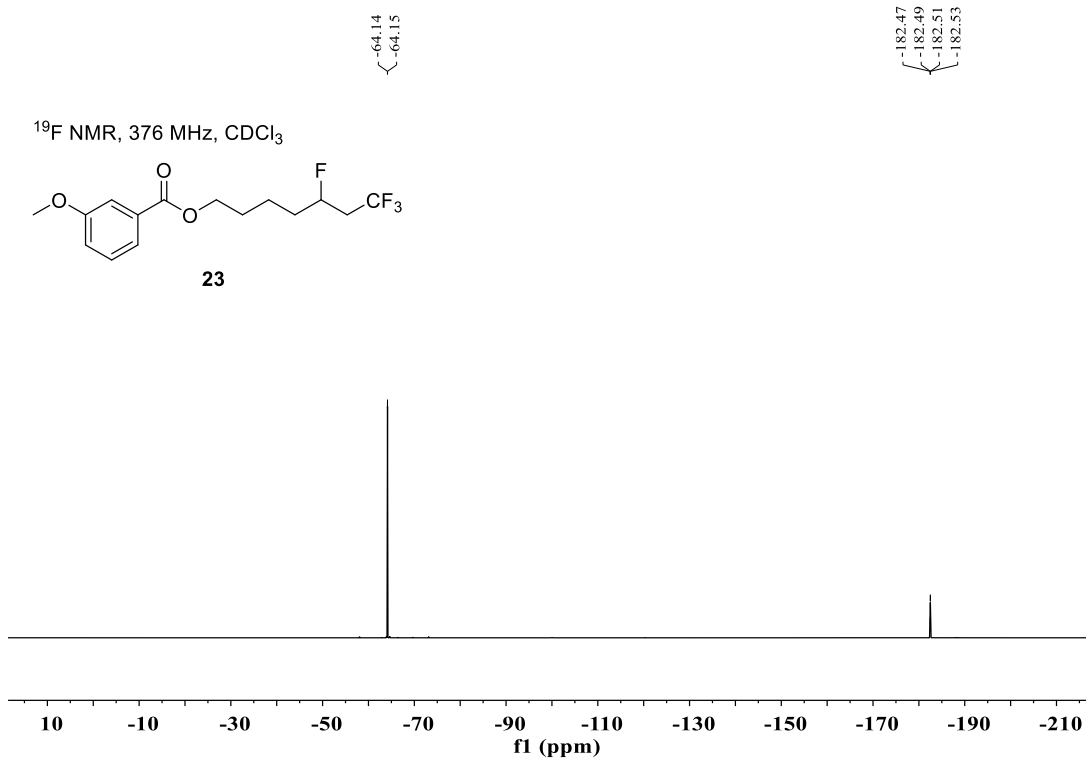
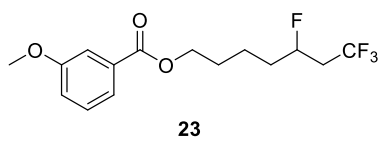




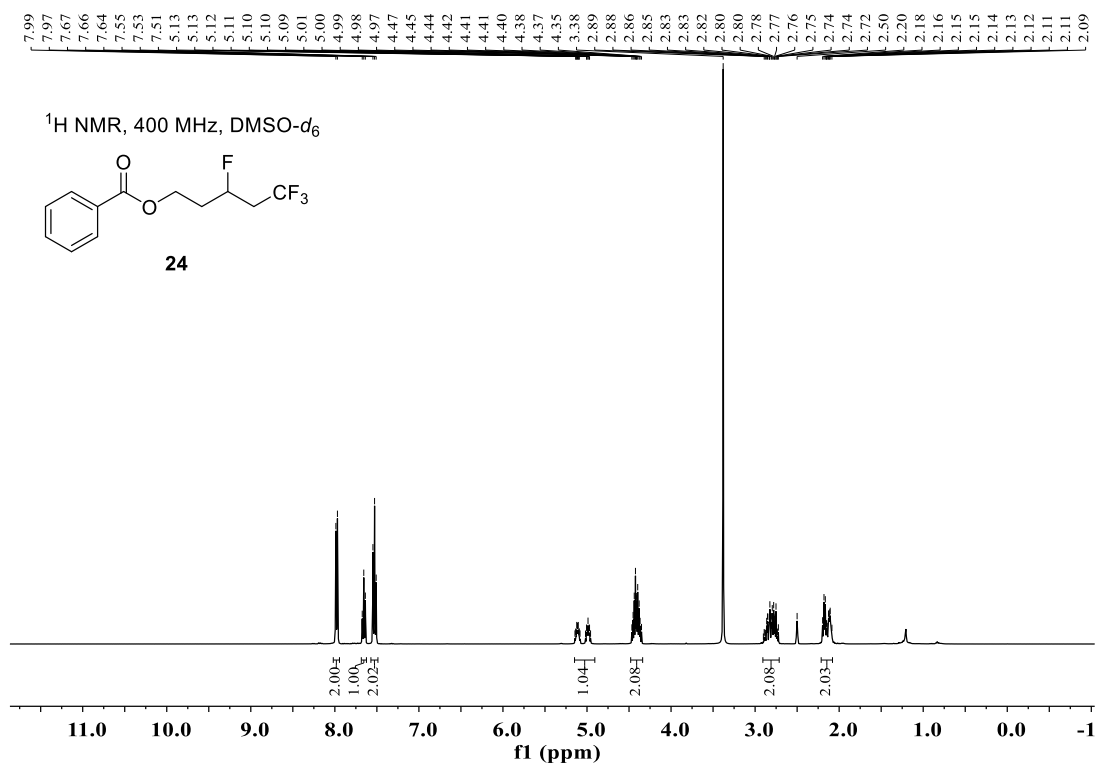
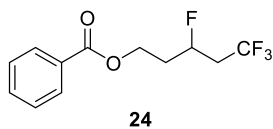


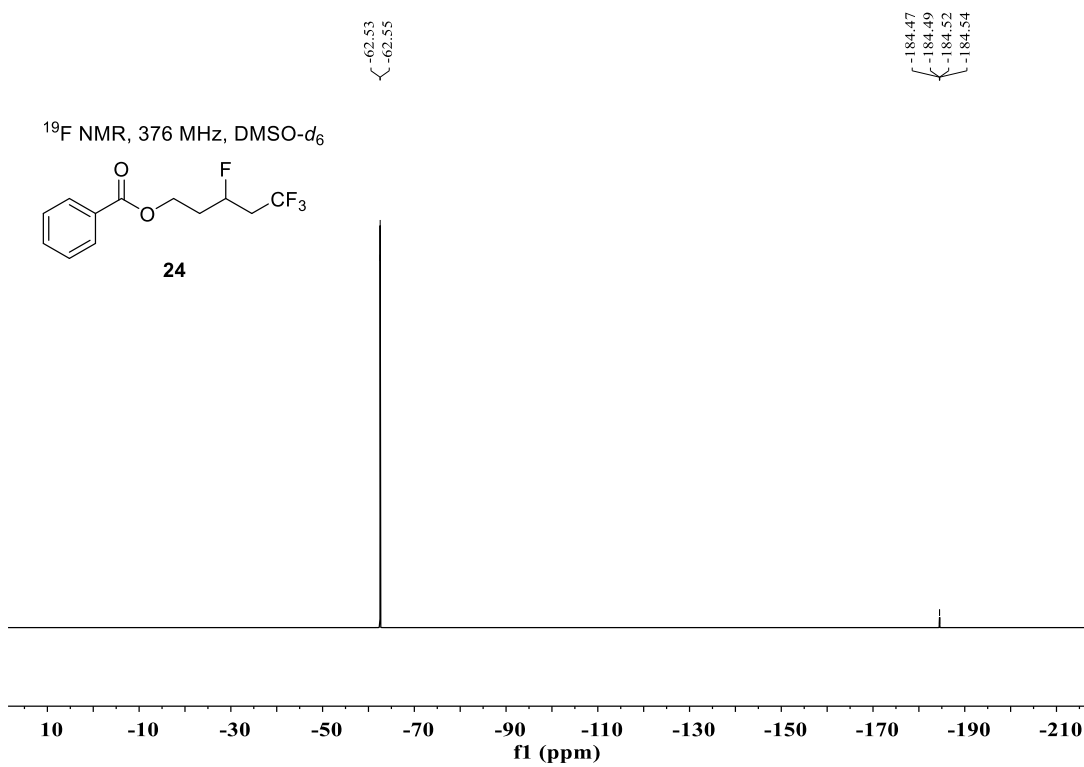
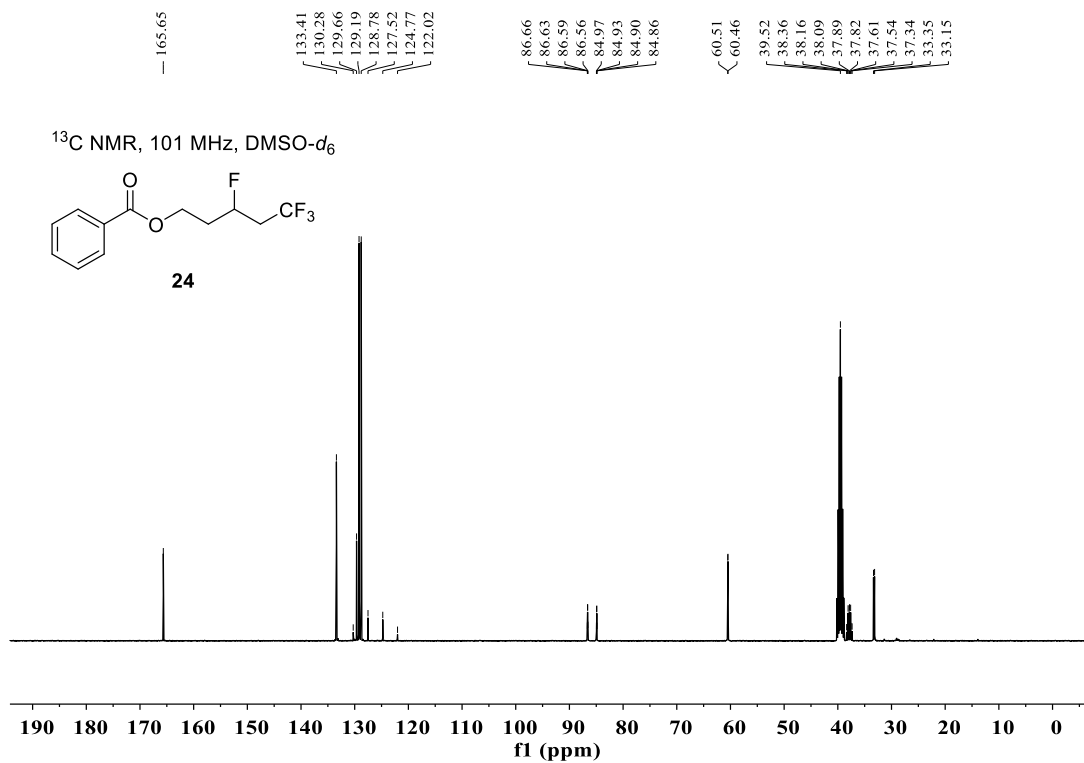


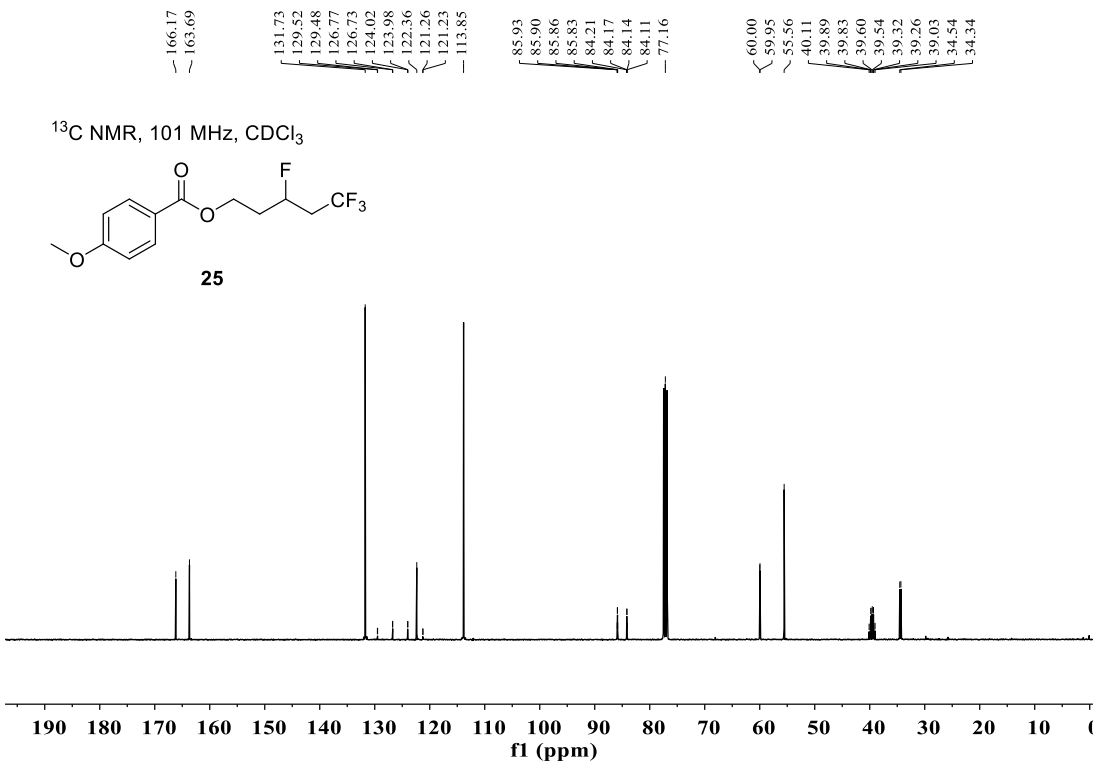
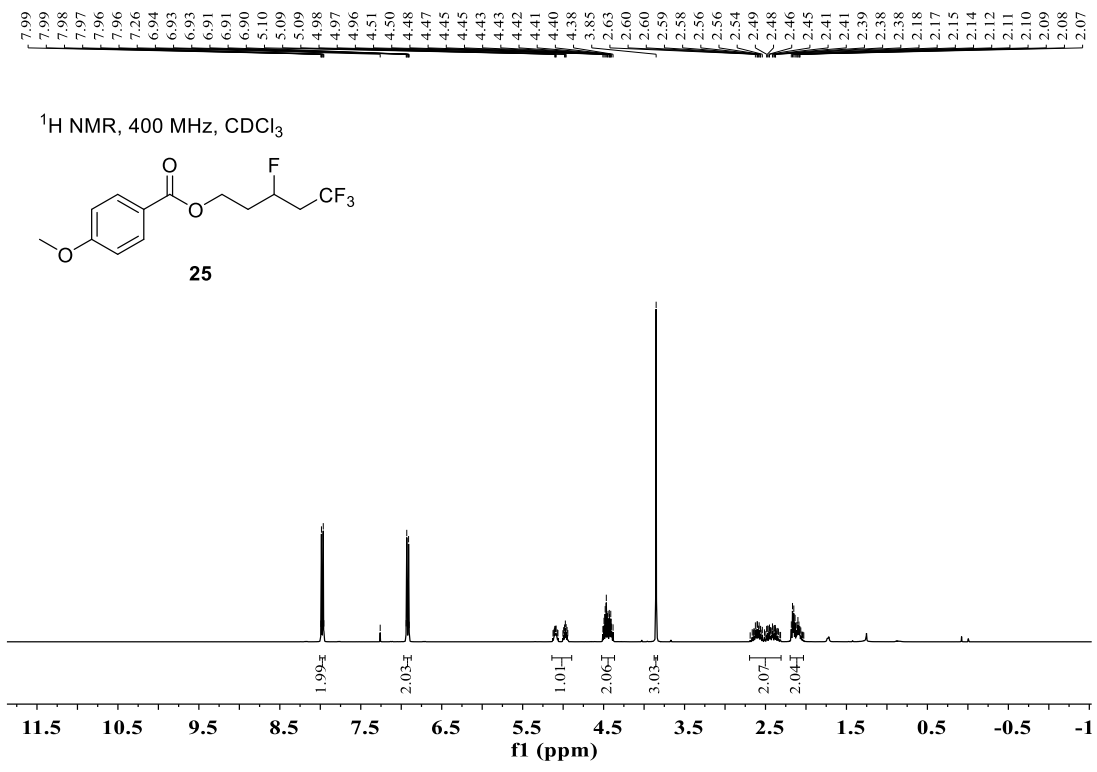
¹⁹F NMR, 376 MHz, CDCl₃

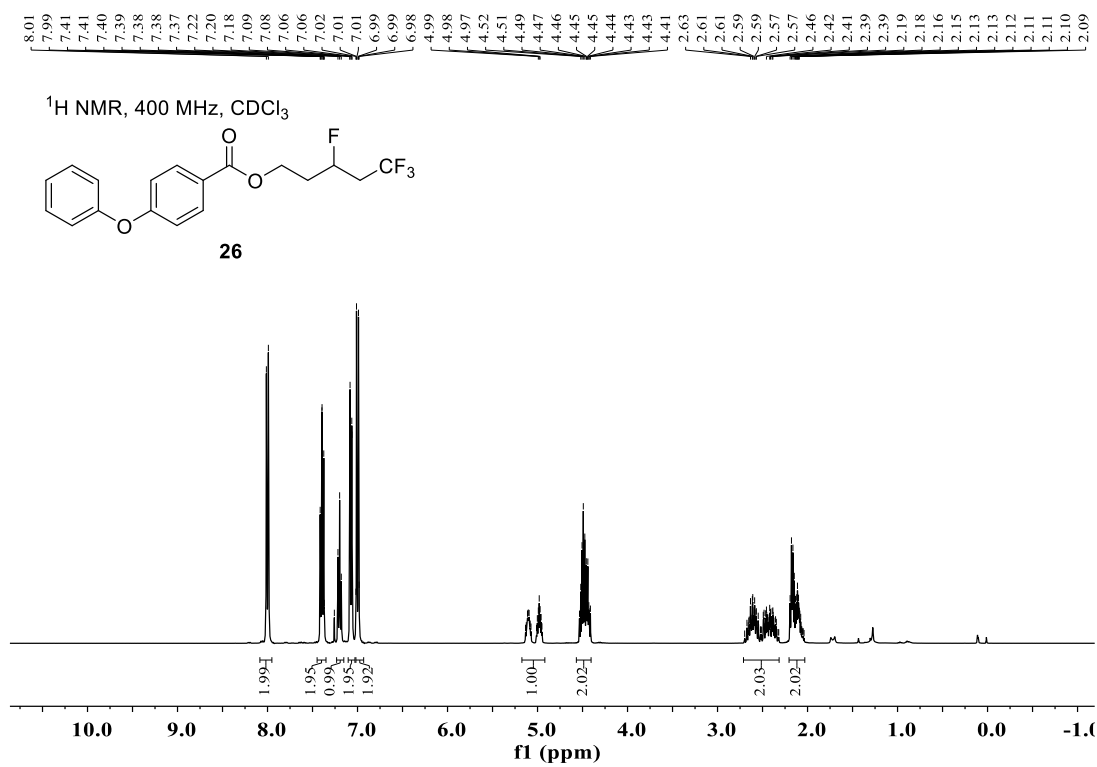
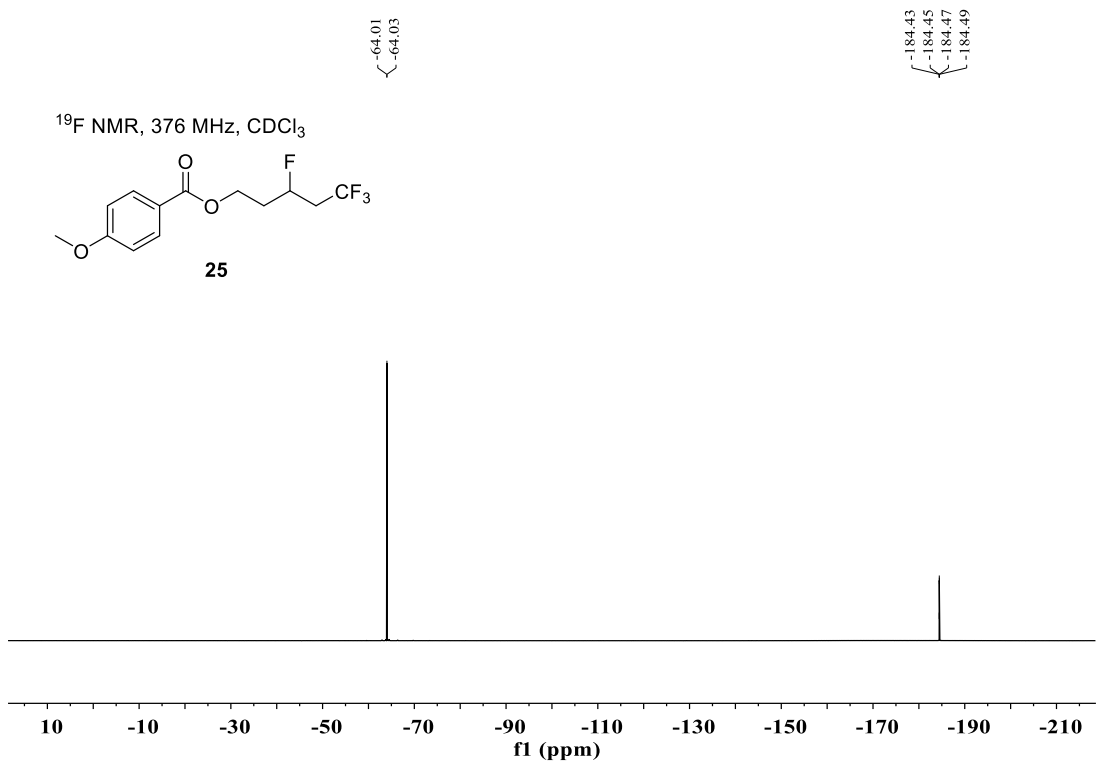


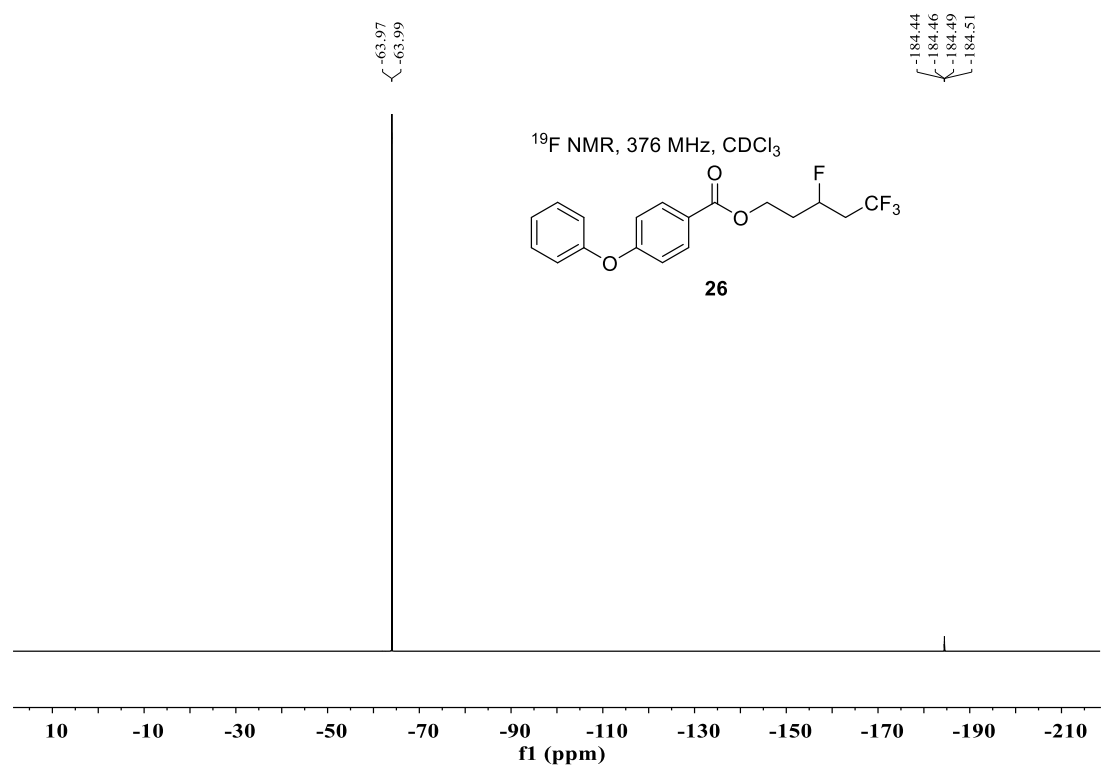
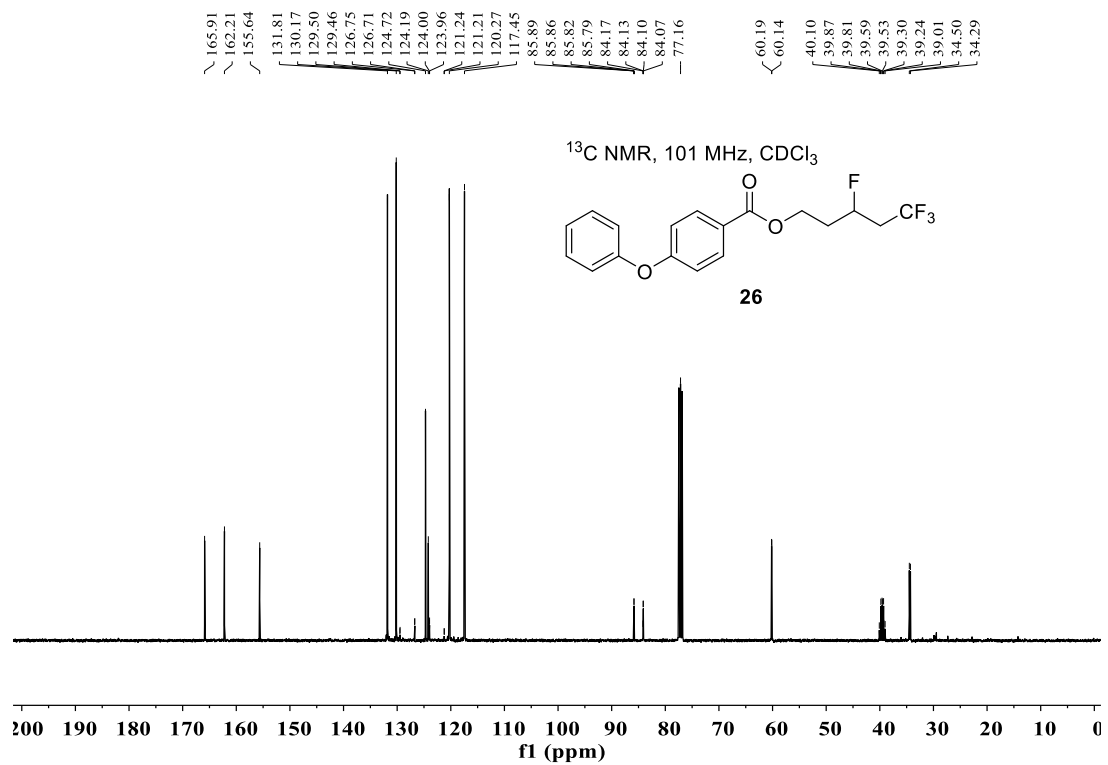
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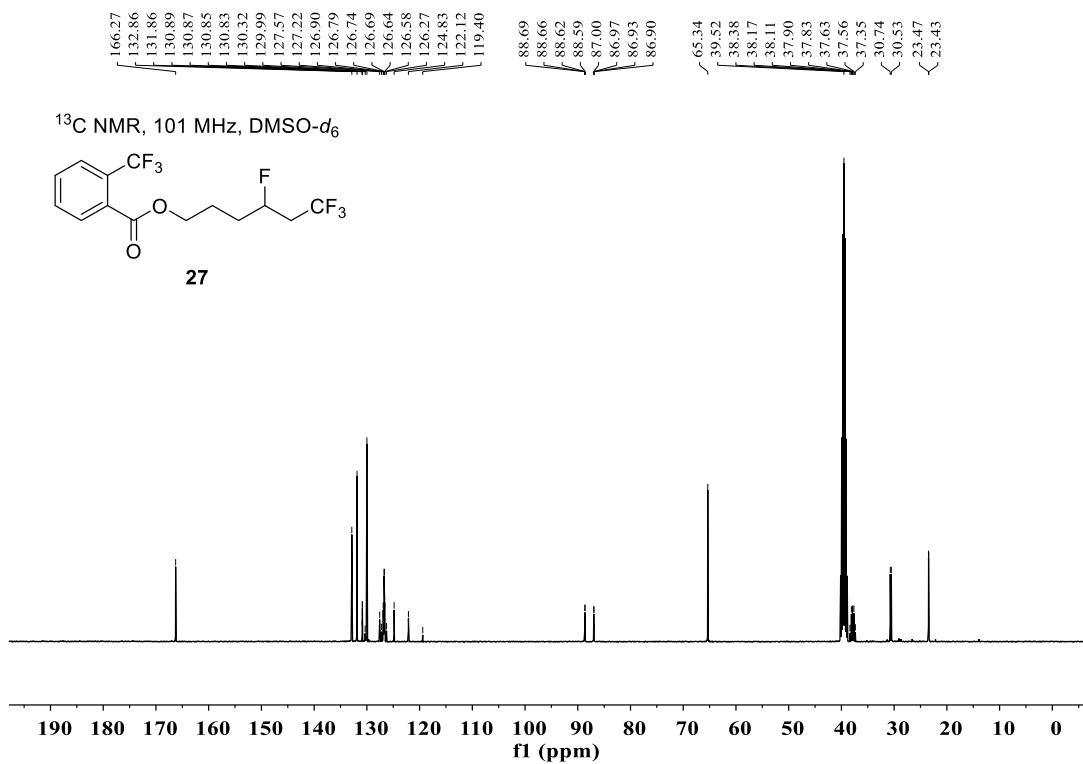
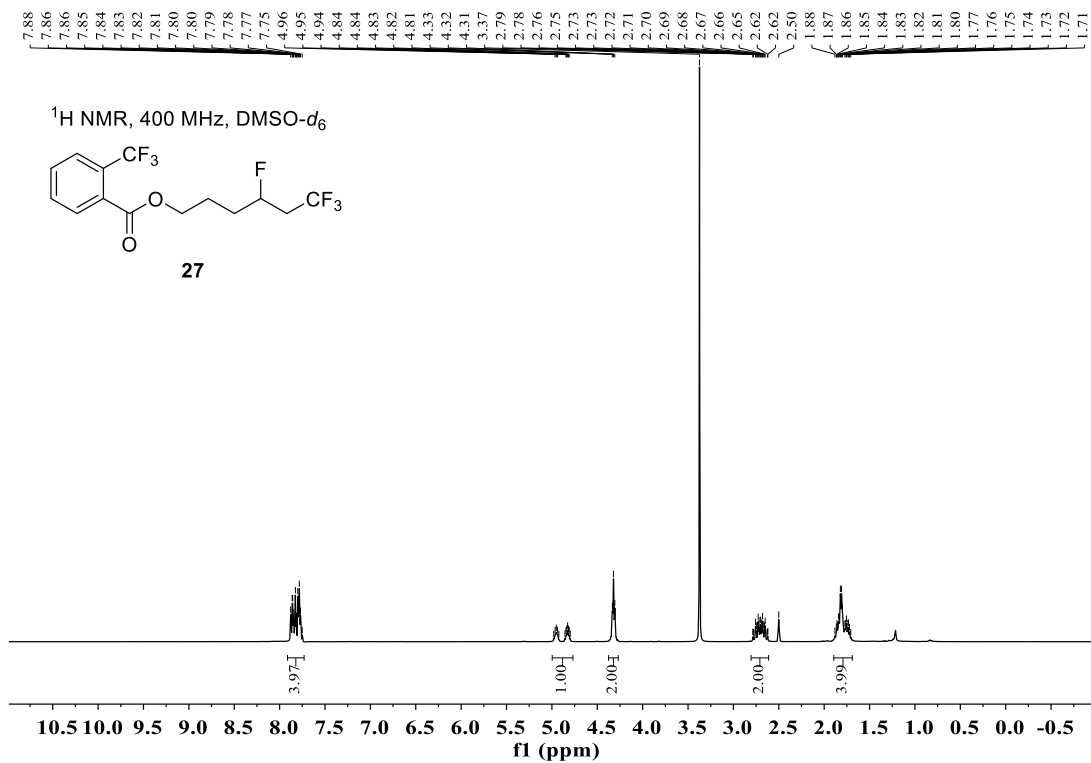


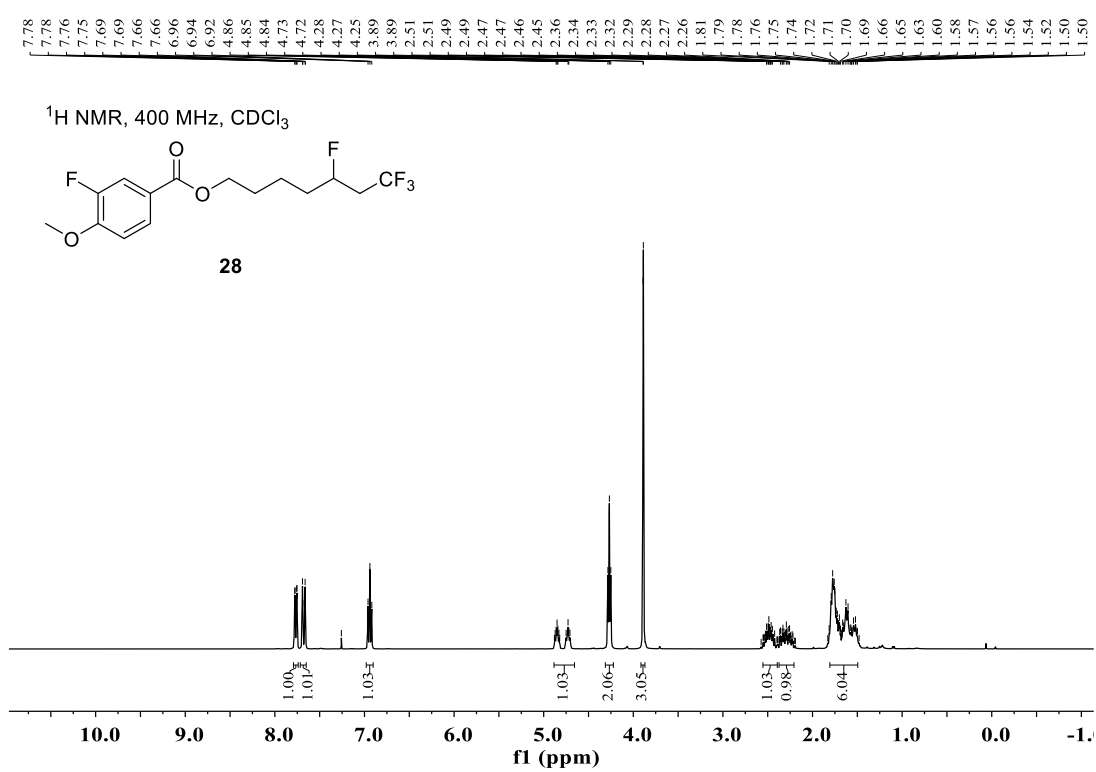
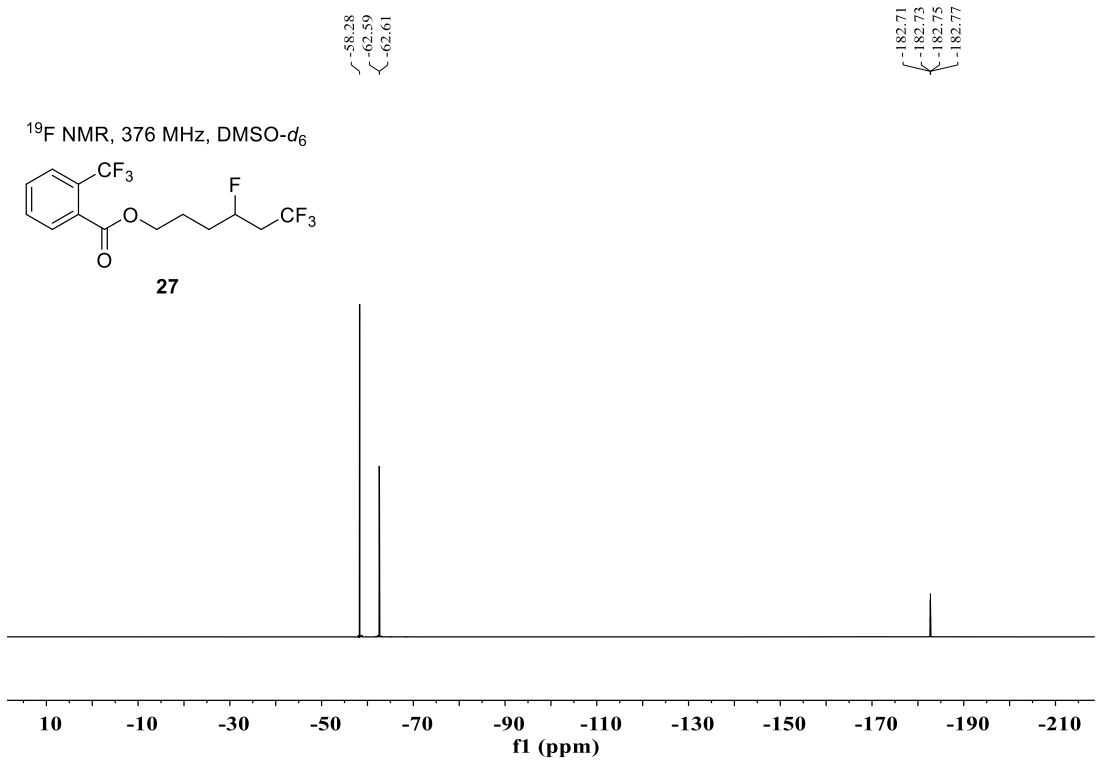


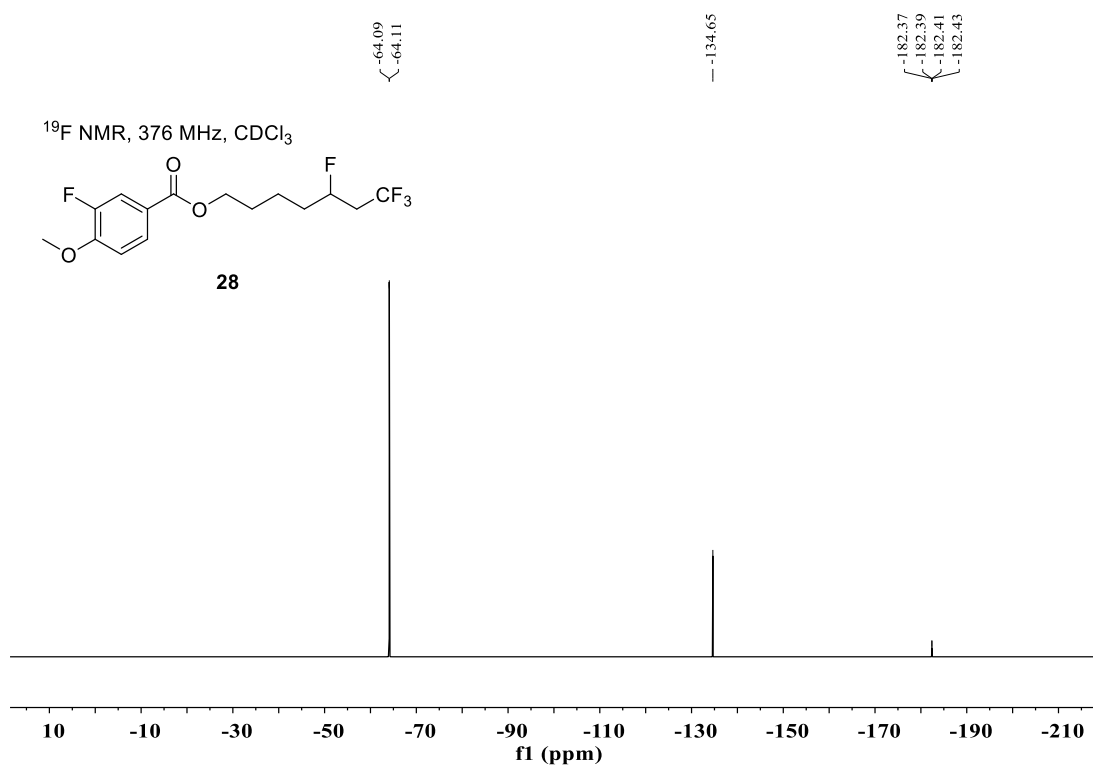
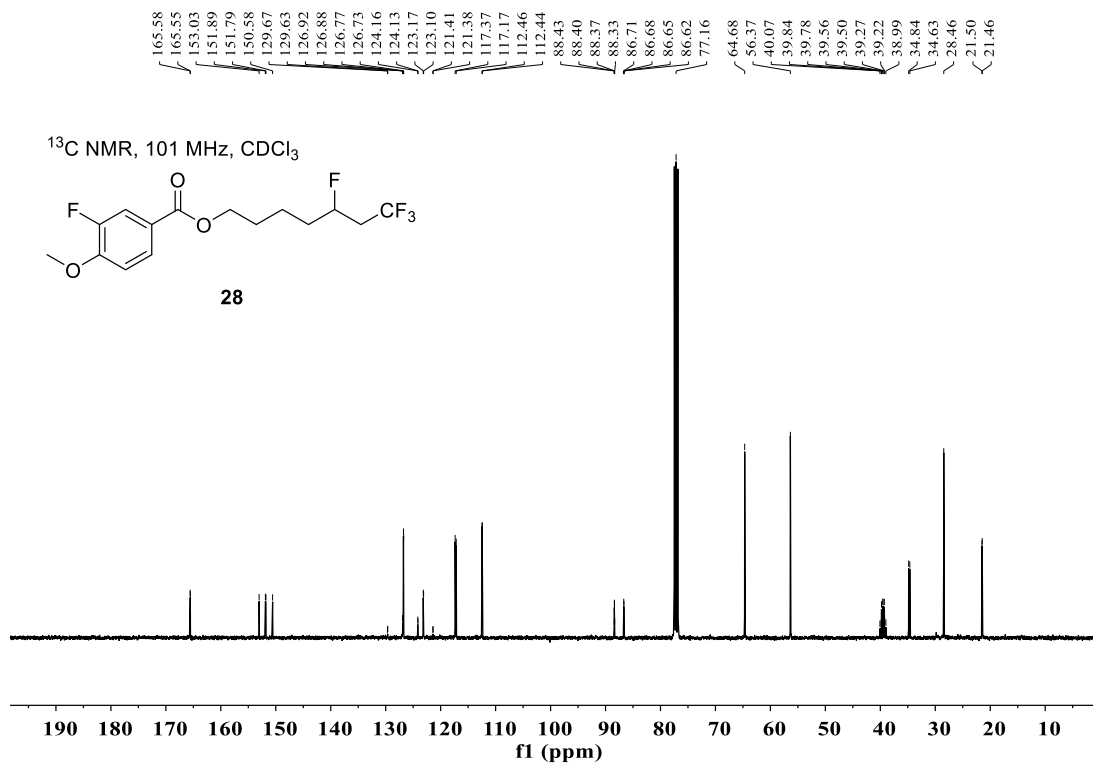


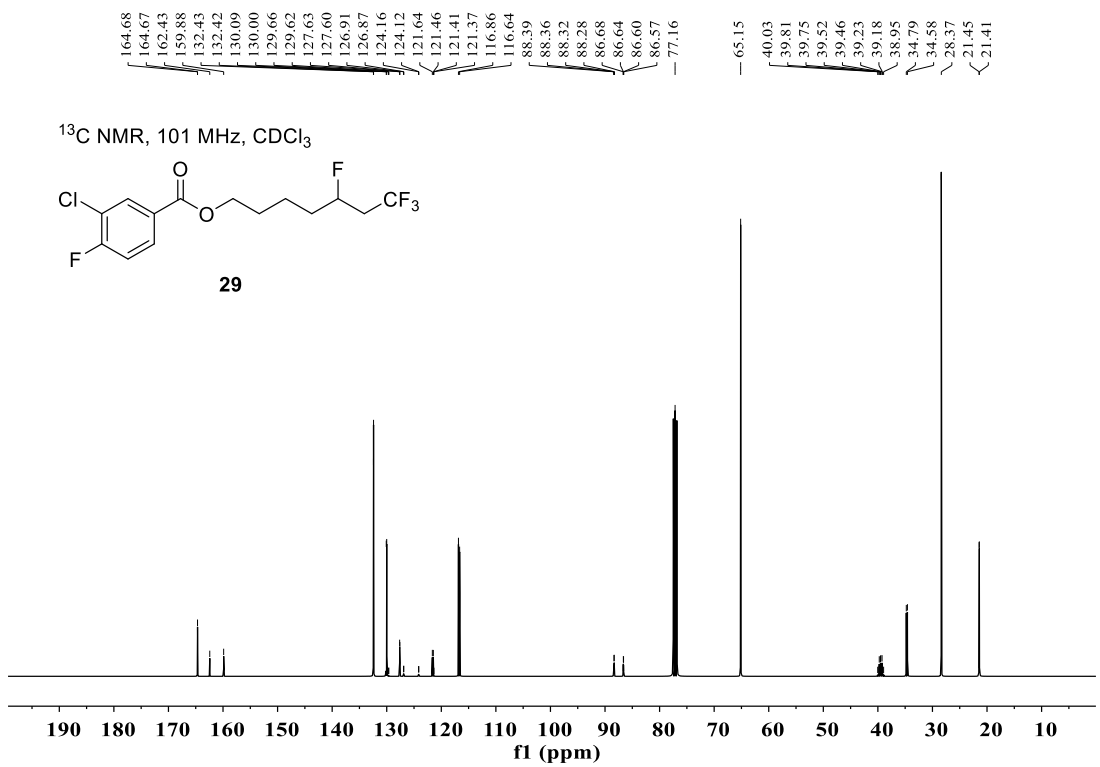
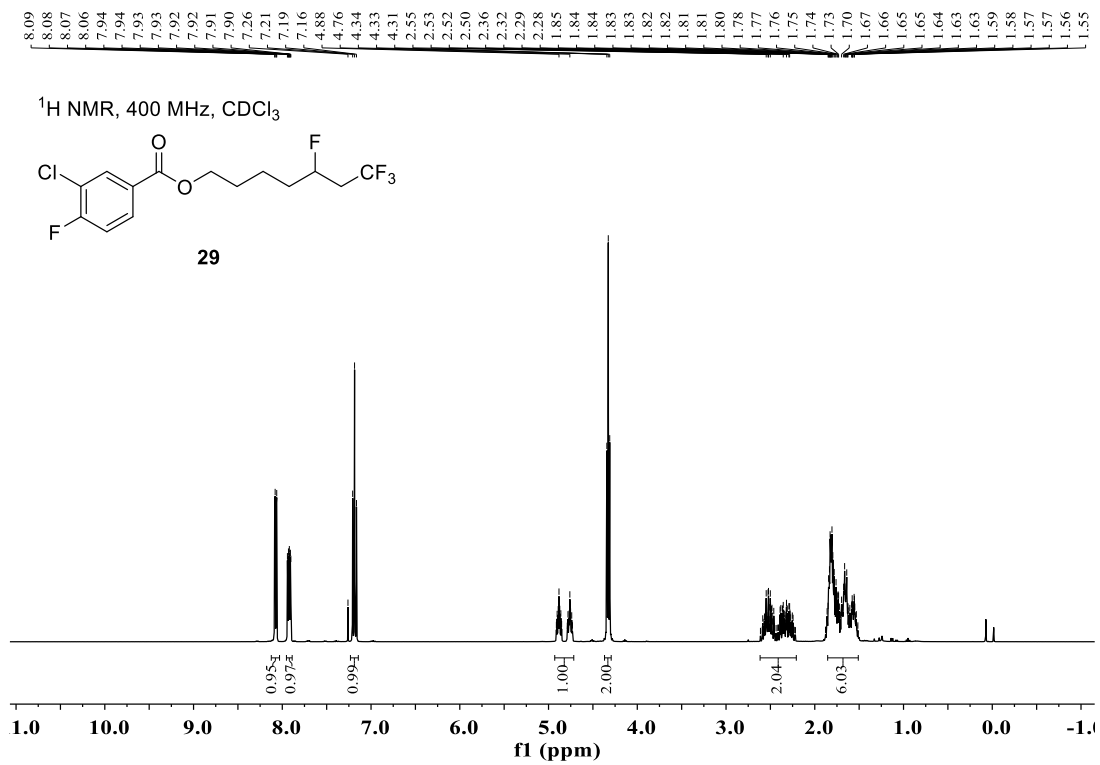


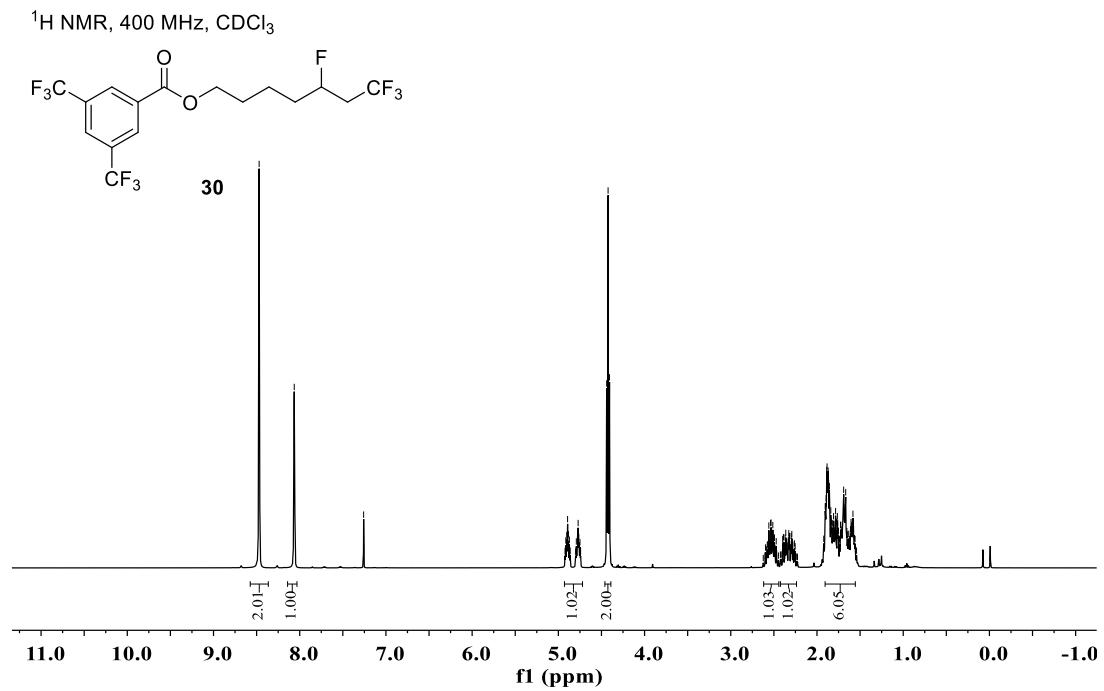
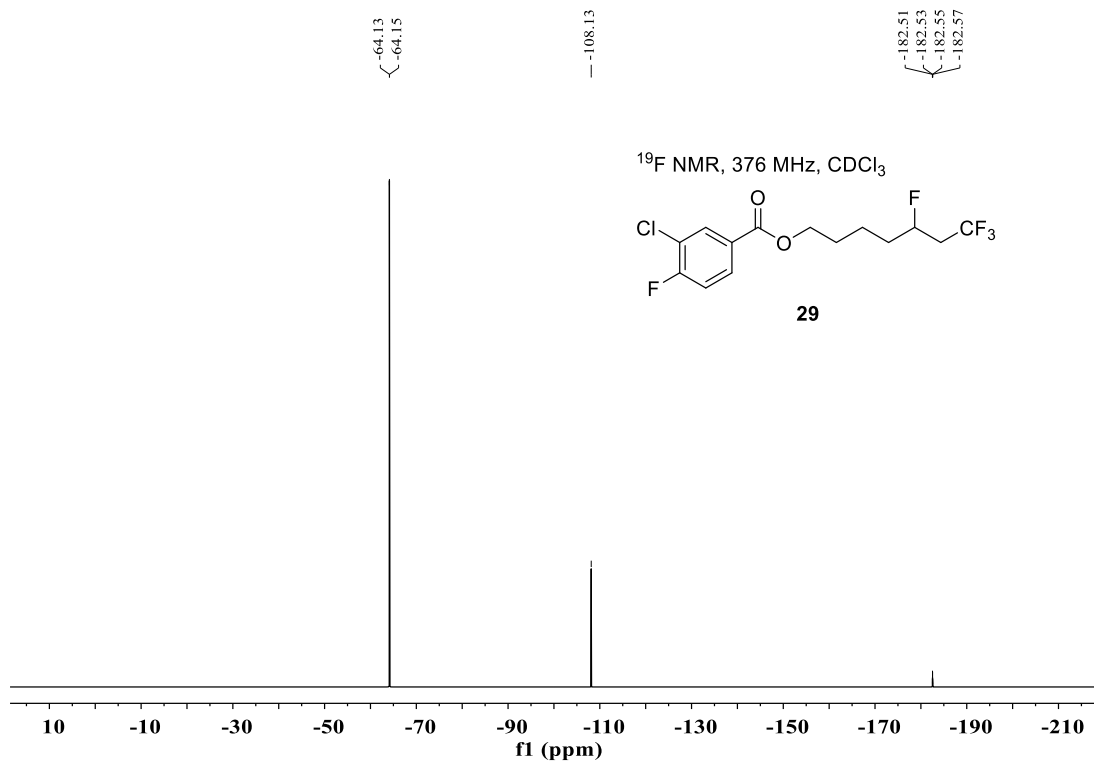


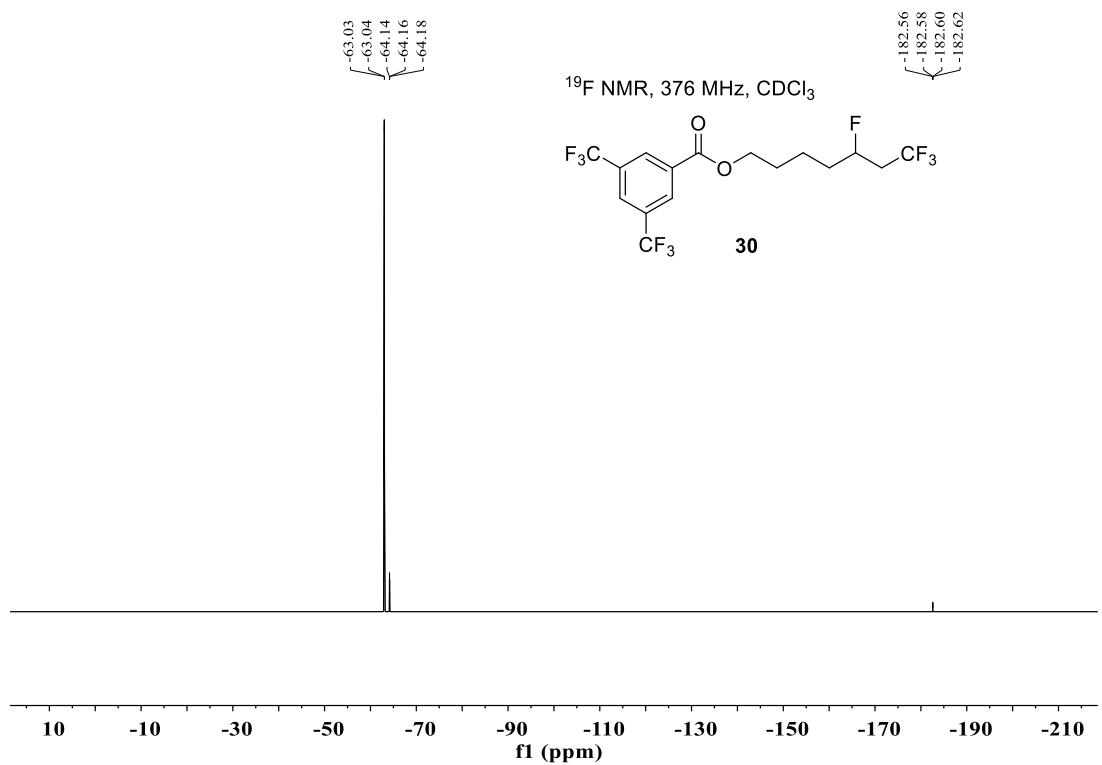
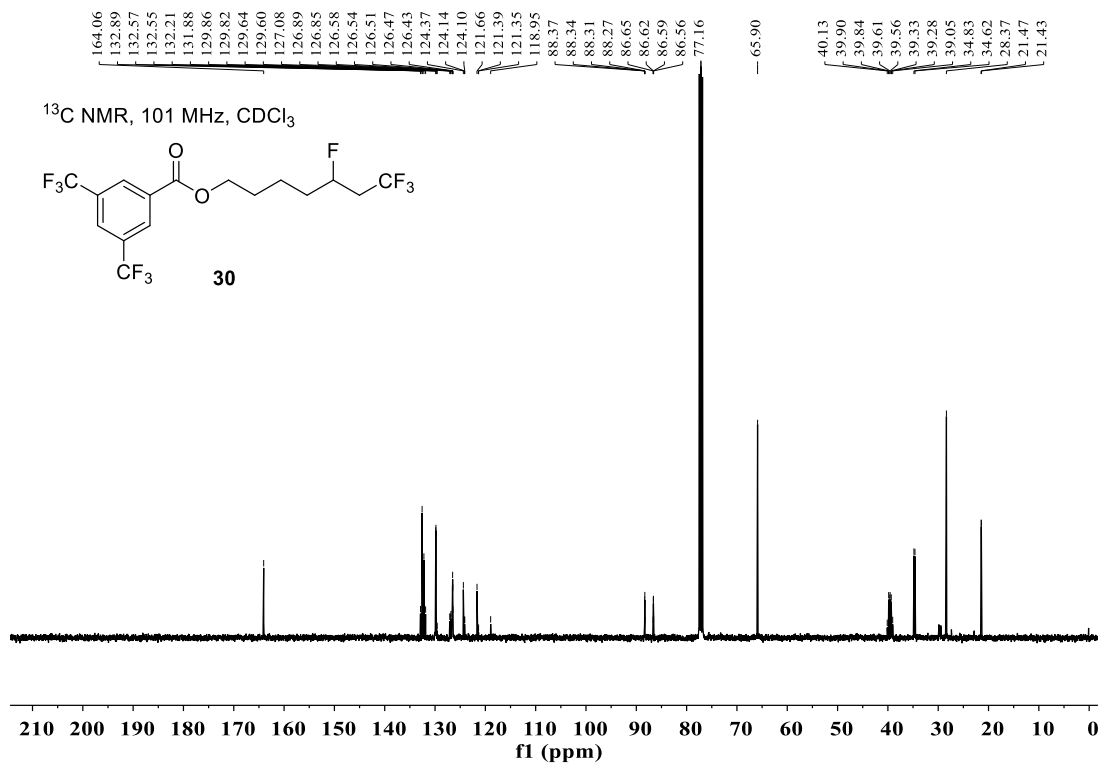


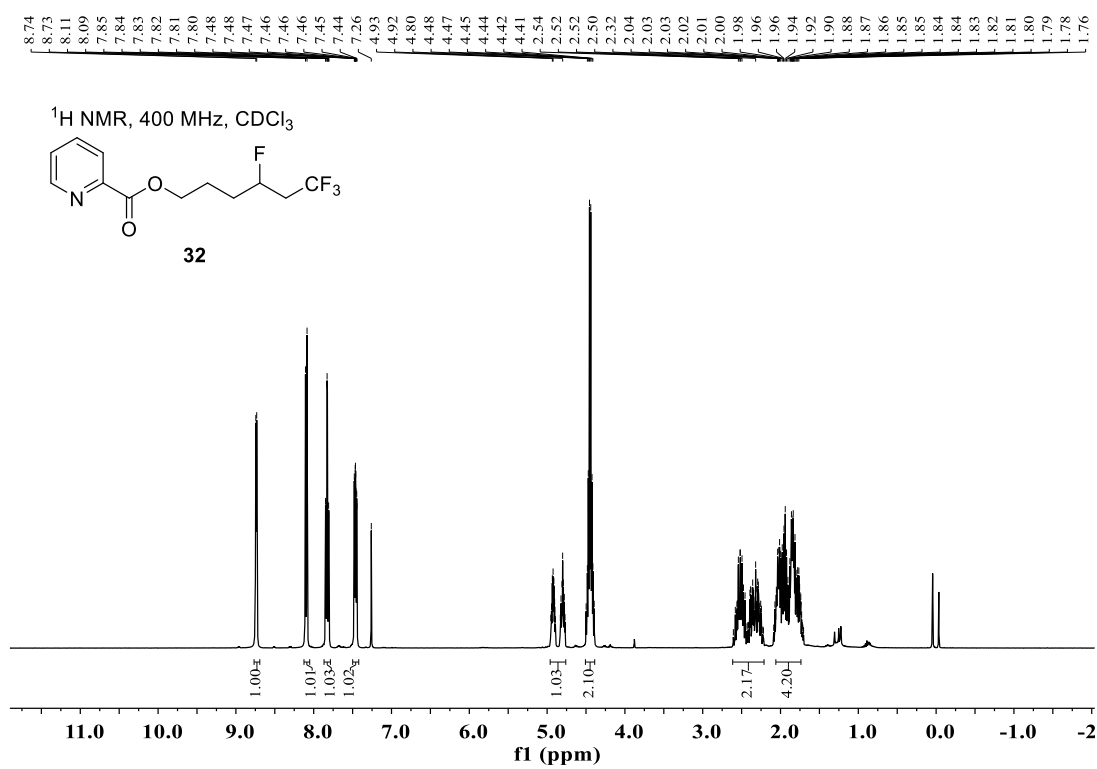
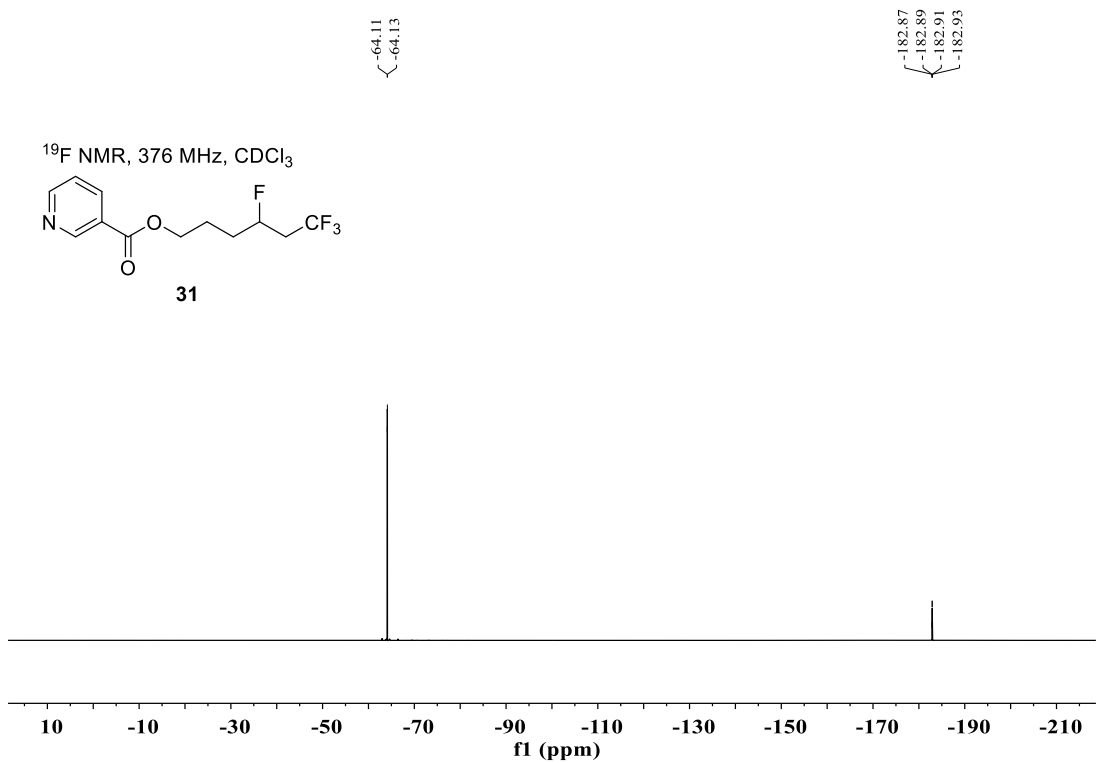


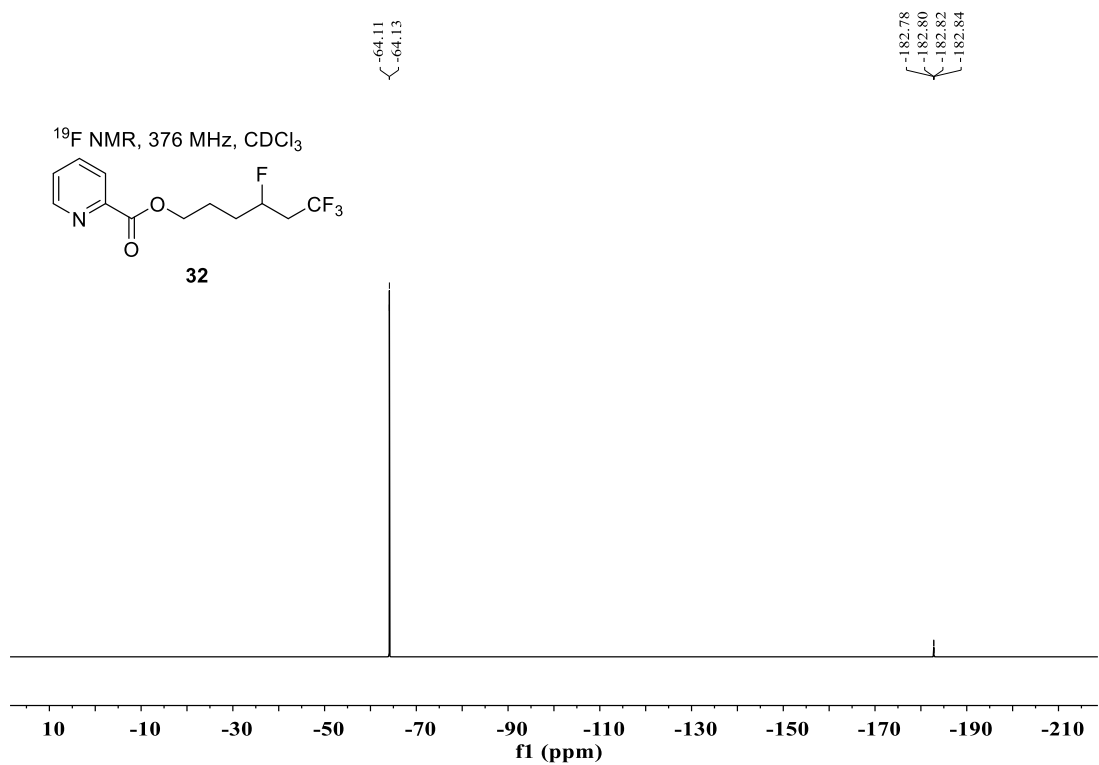
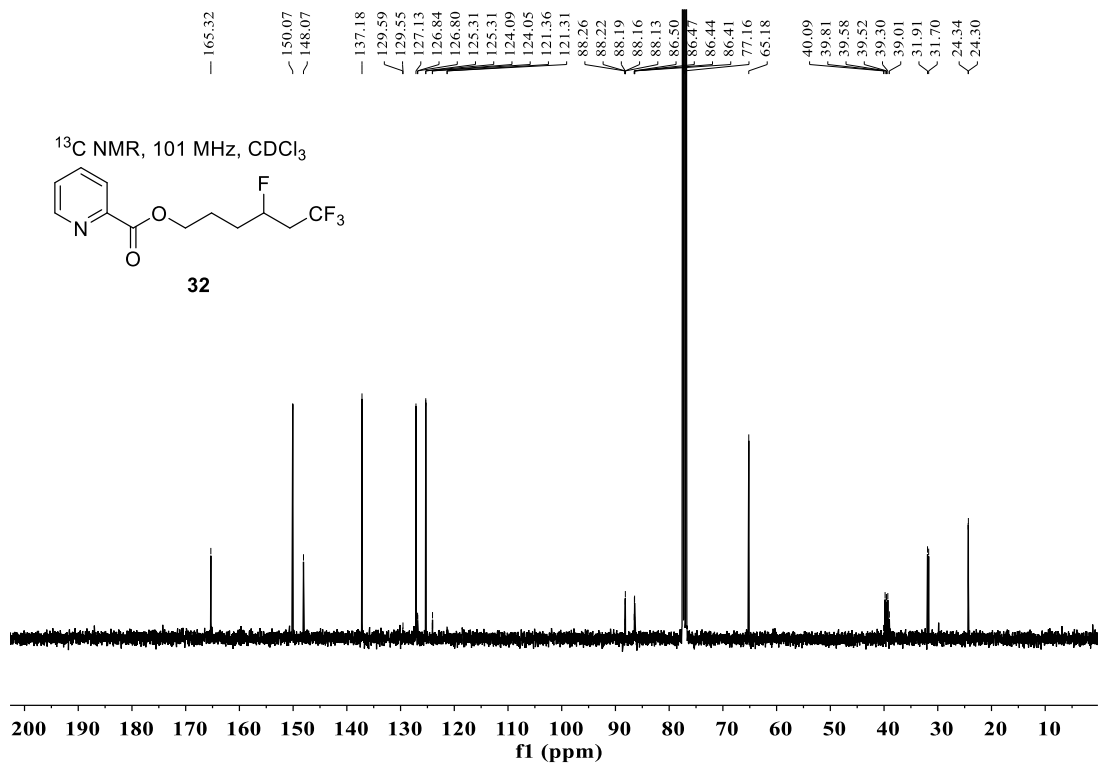


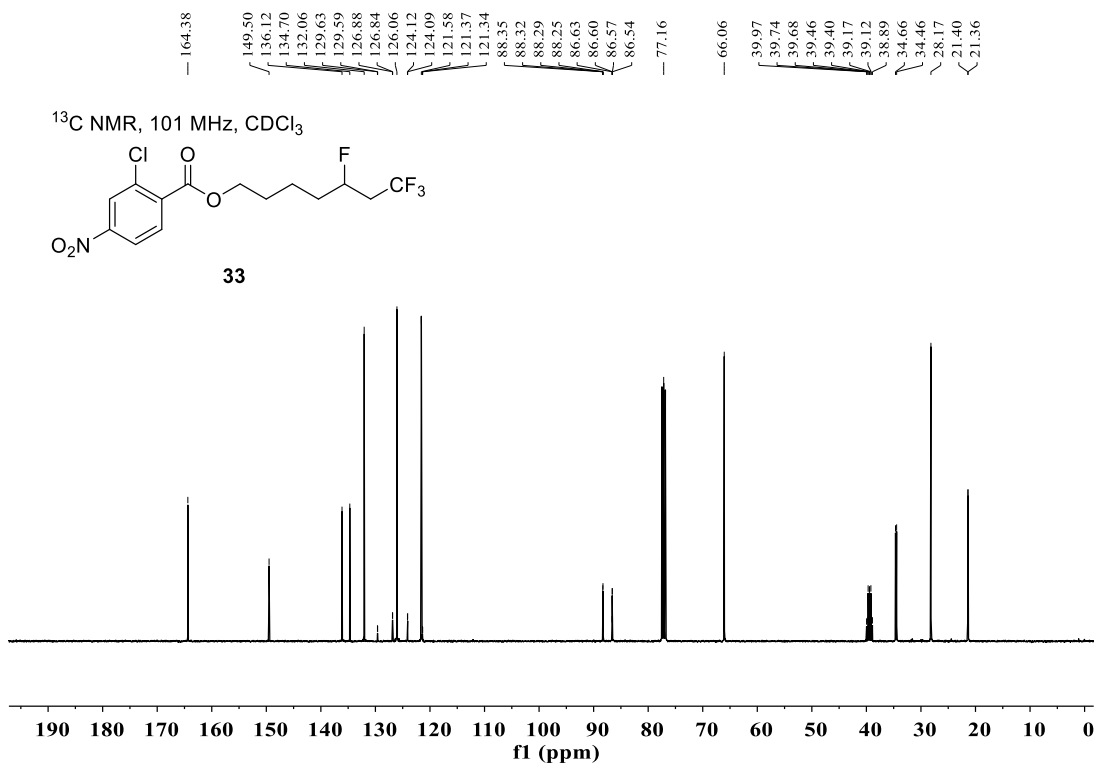
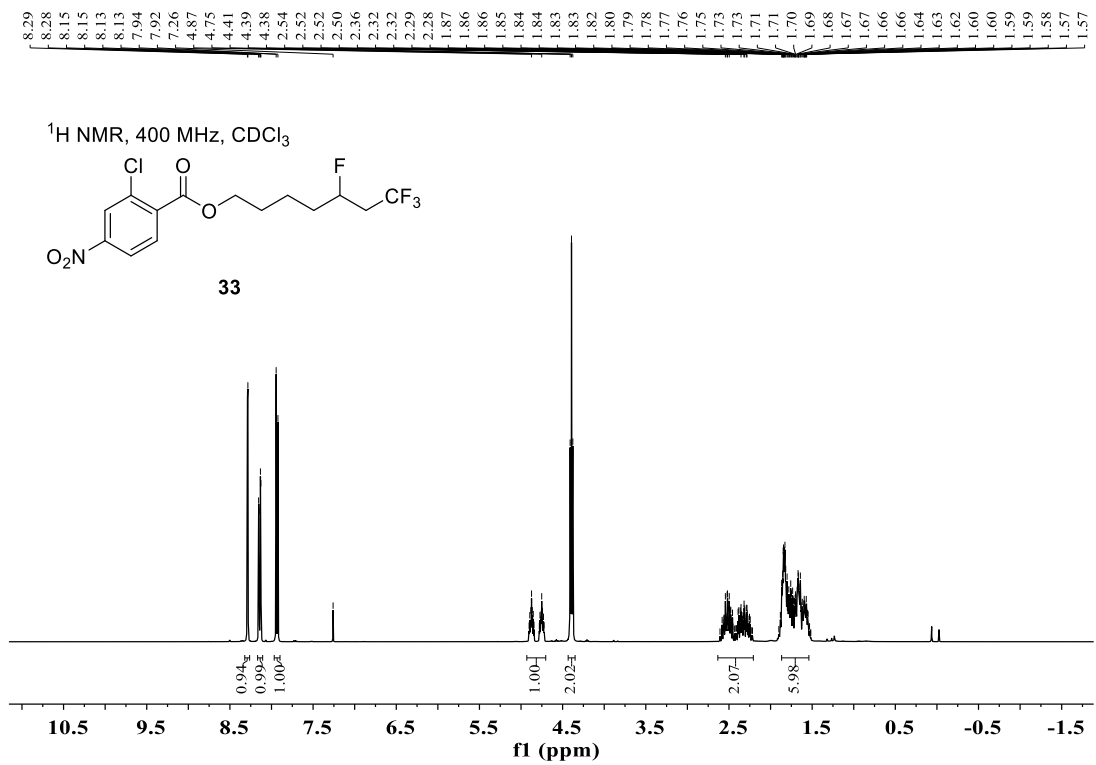


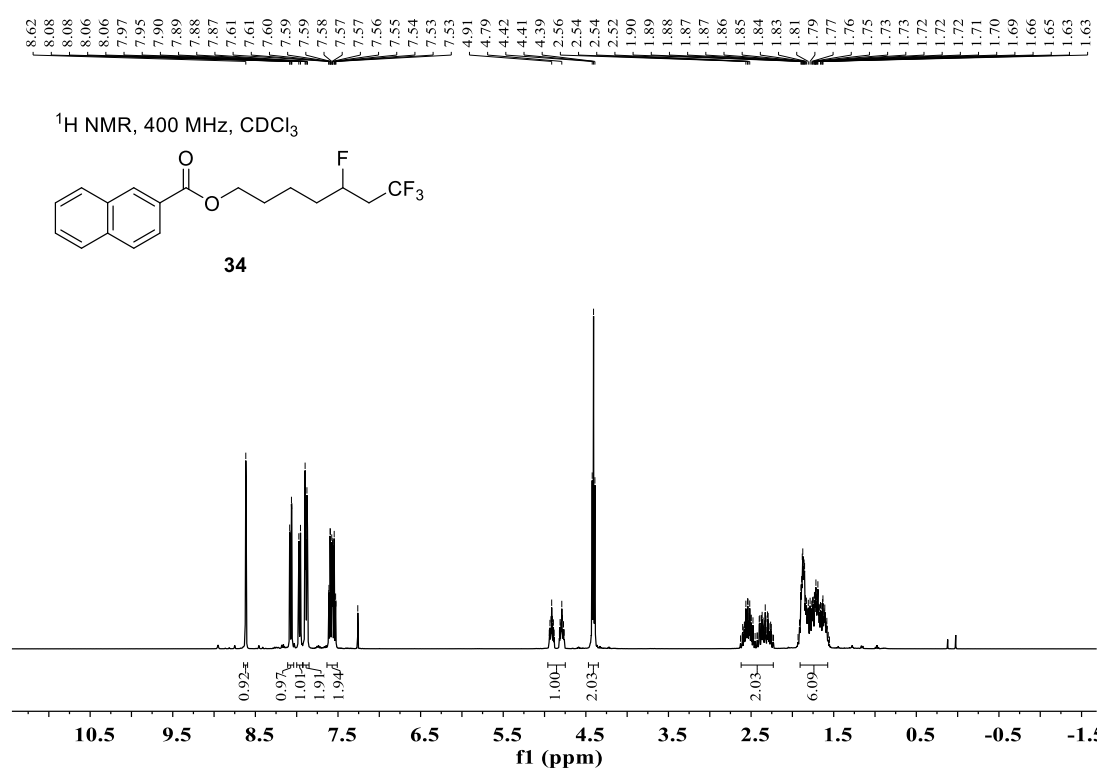
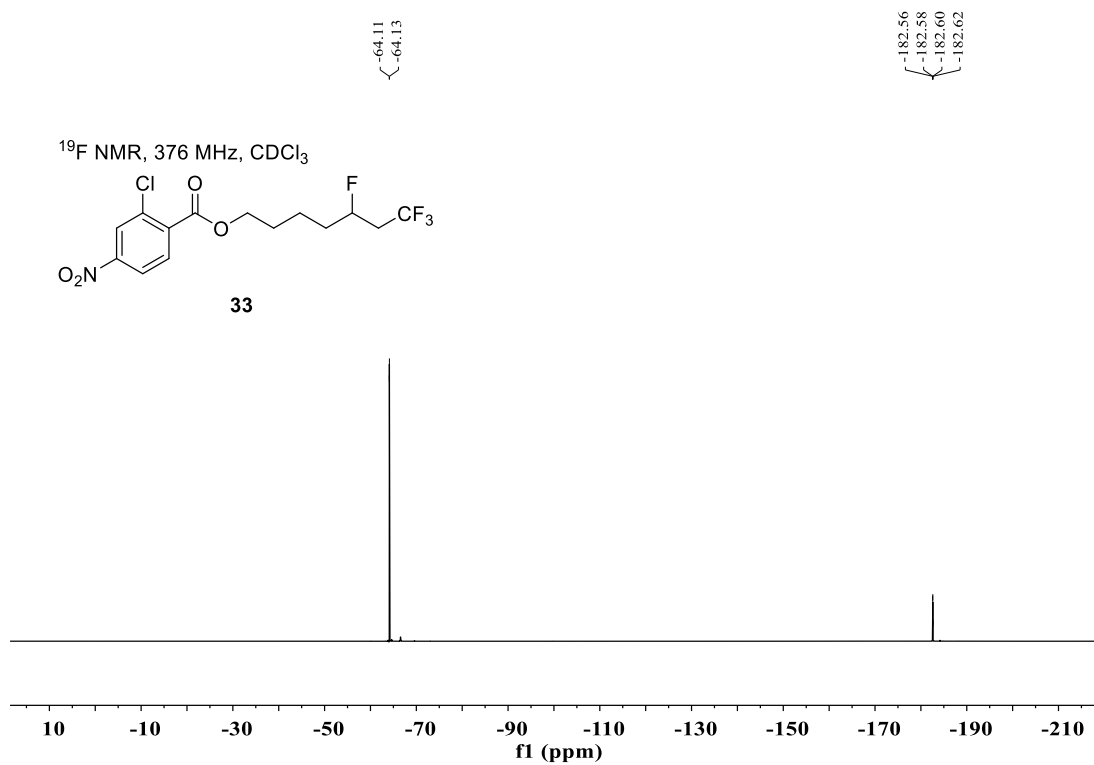


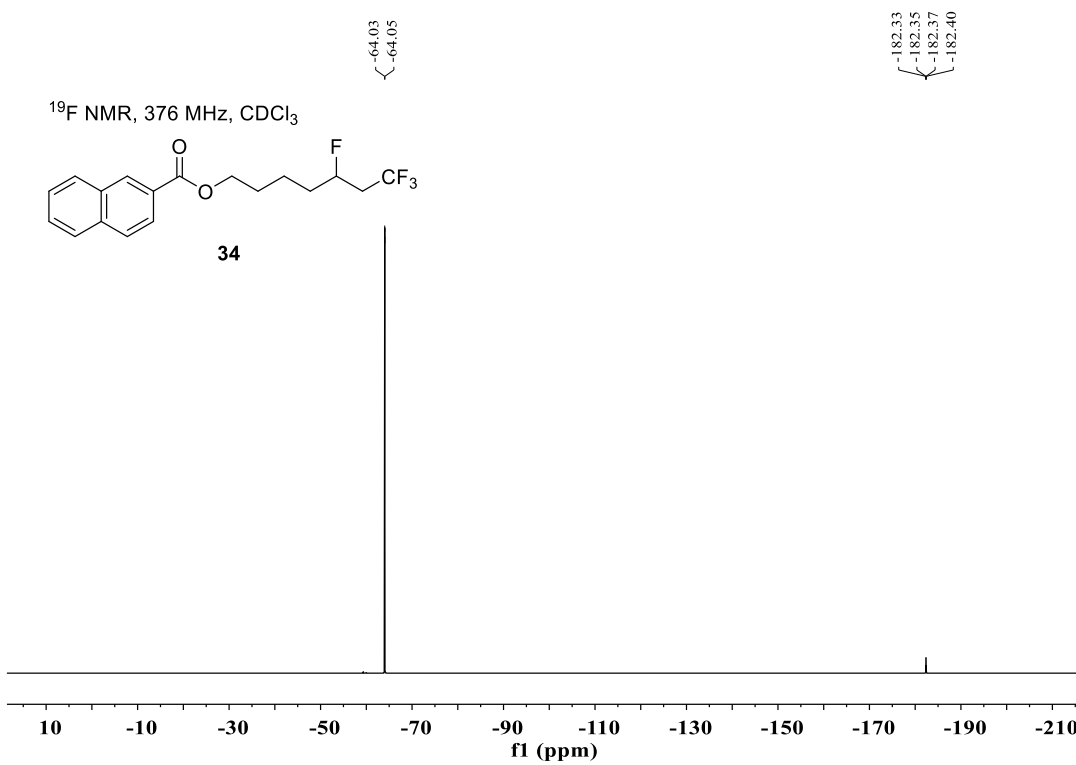
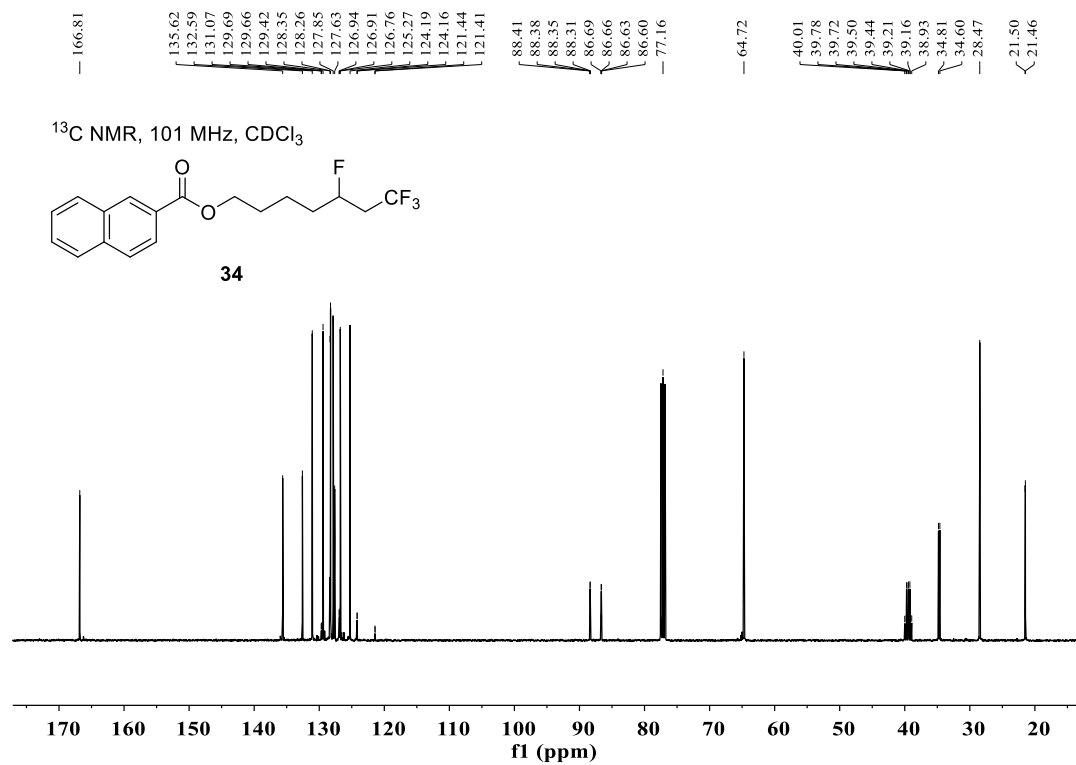


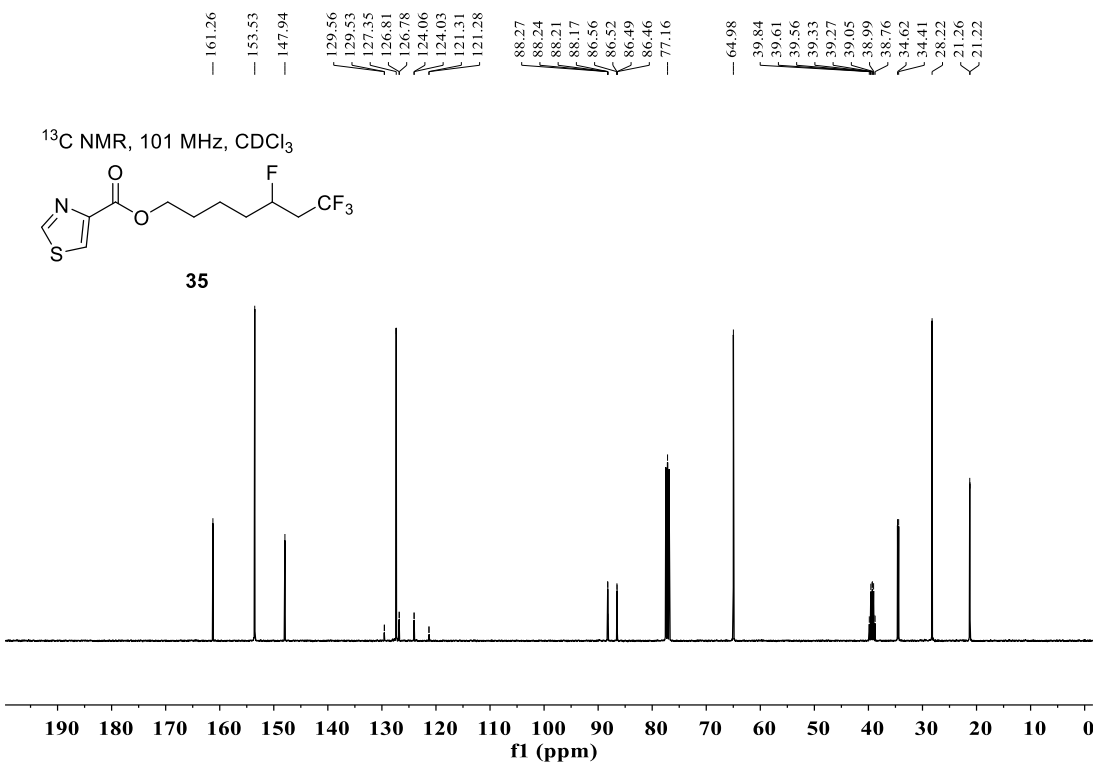
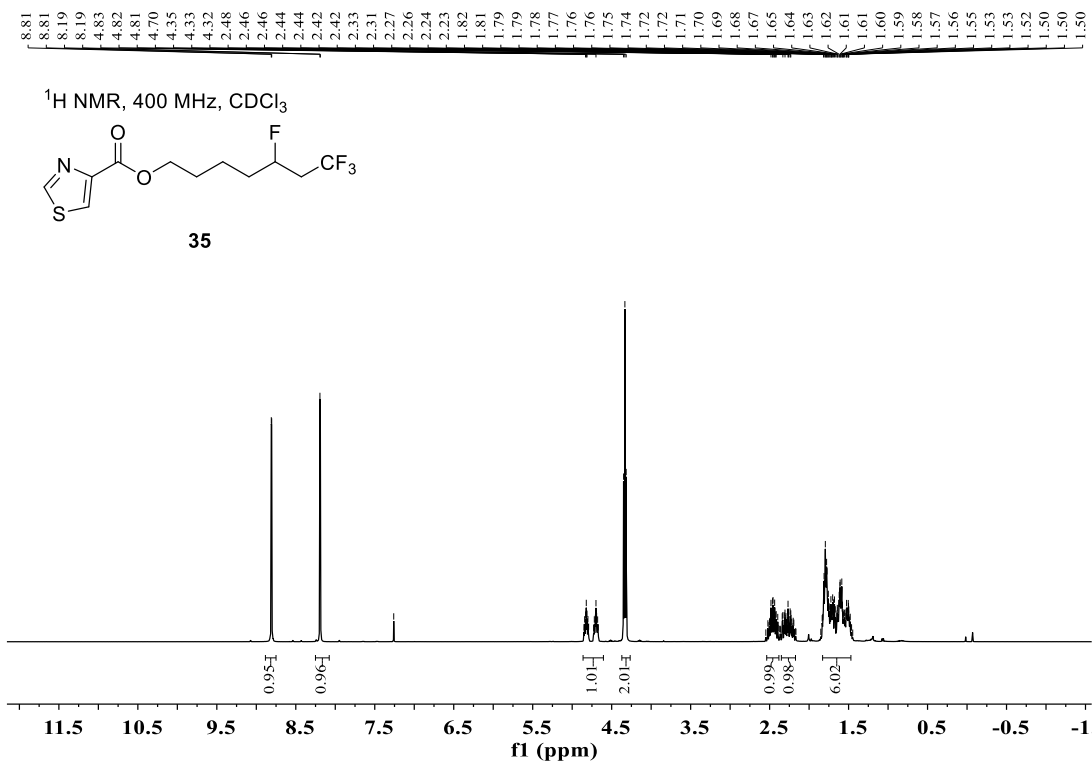




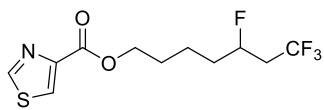




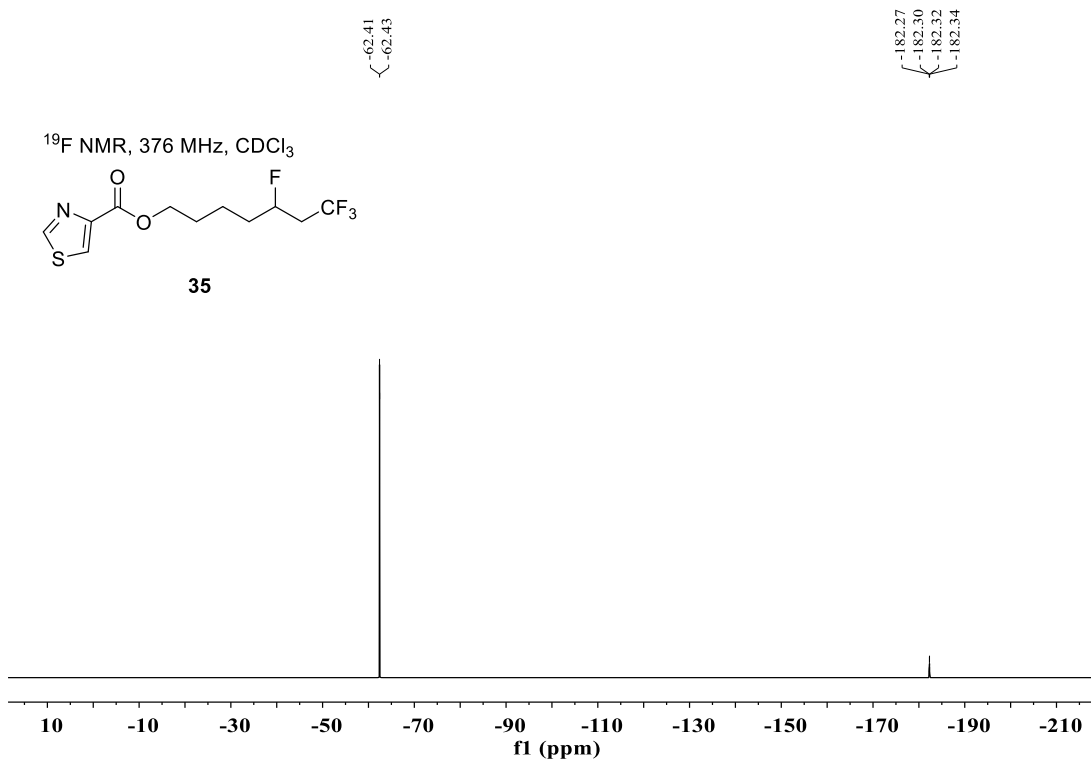




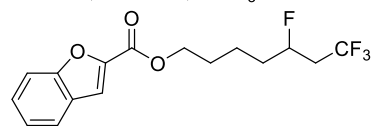
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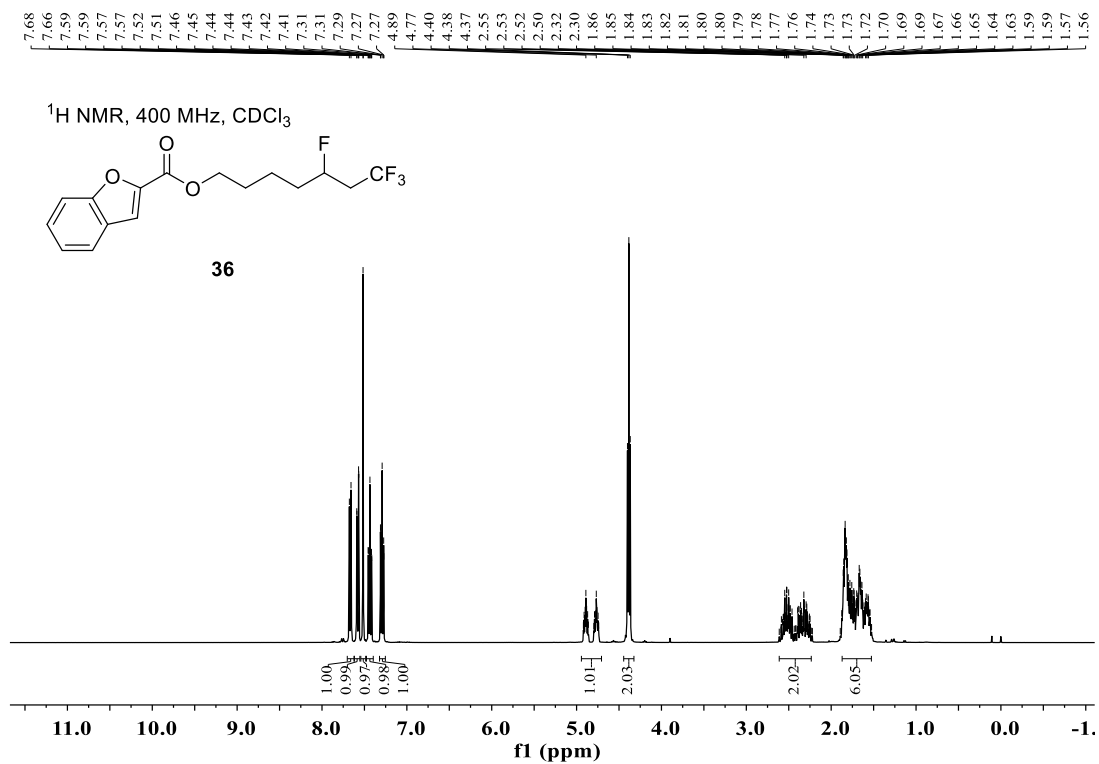
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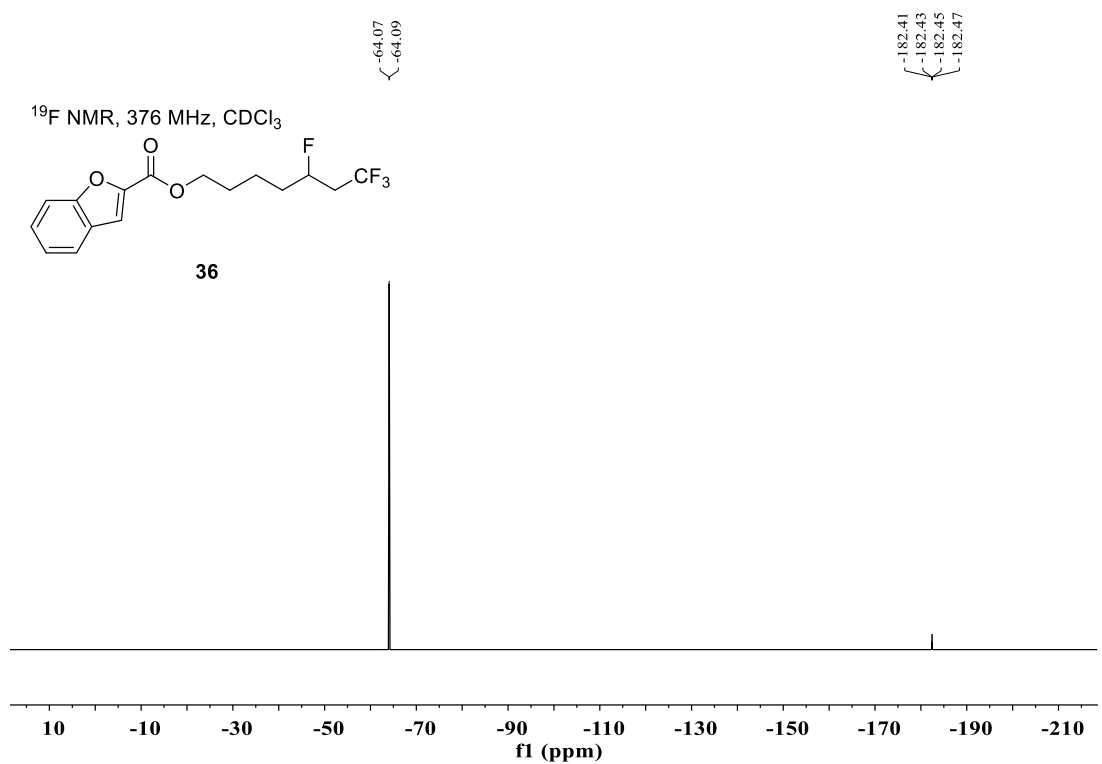
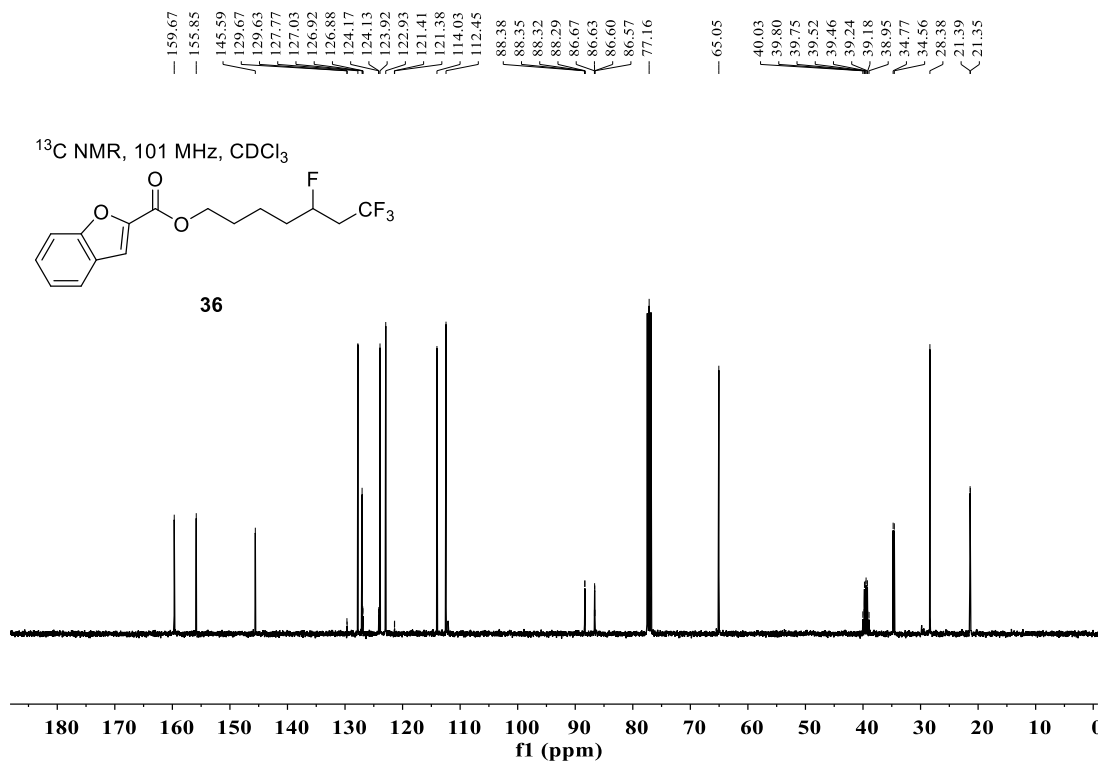


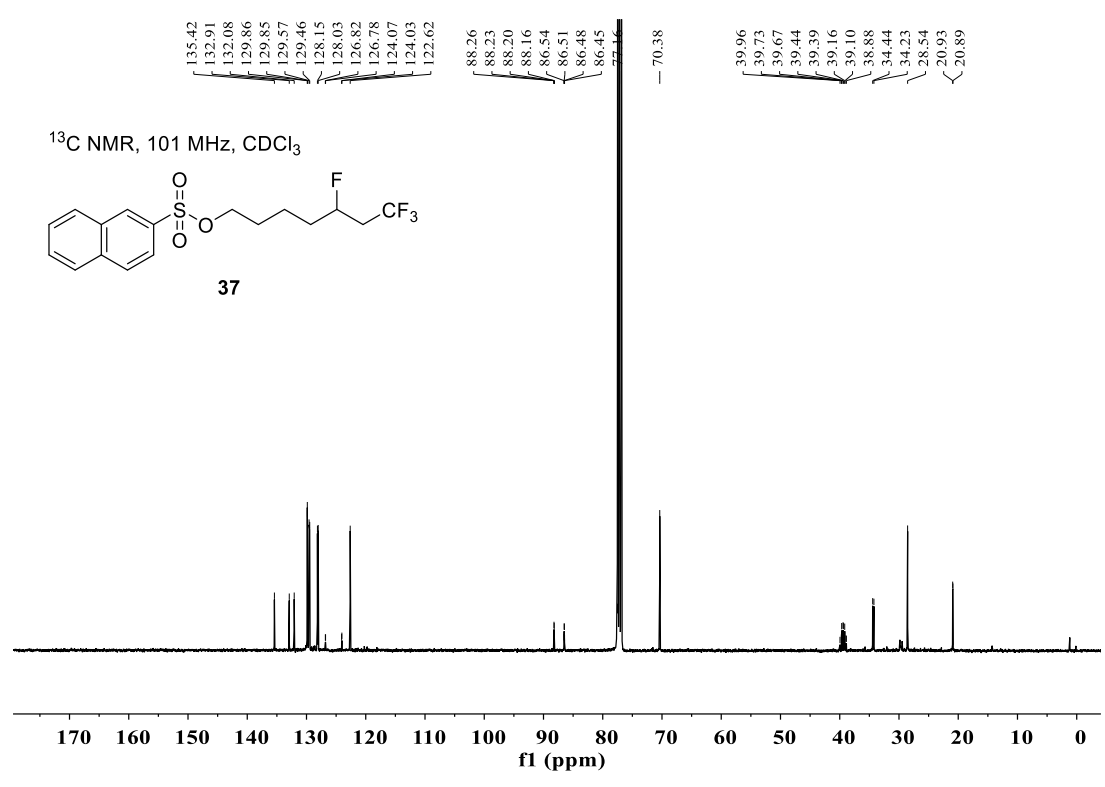
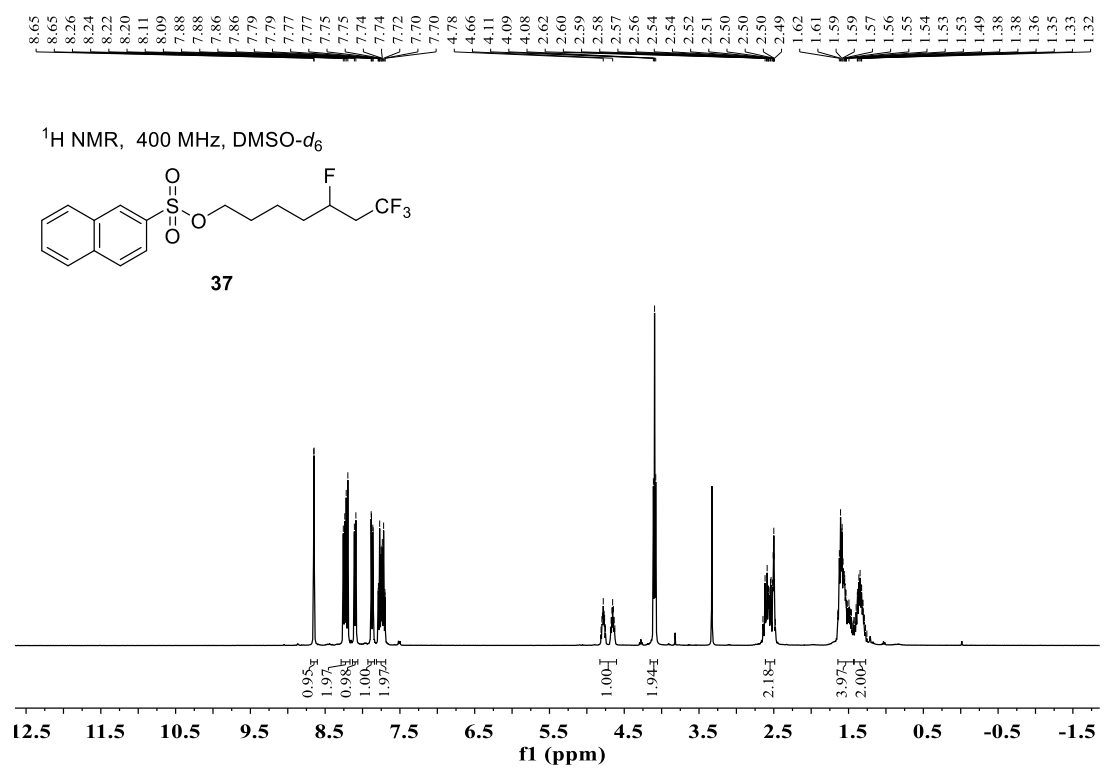
¹H NMR, 400 MHz, CDCl₃

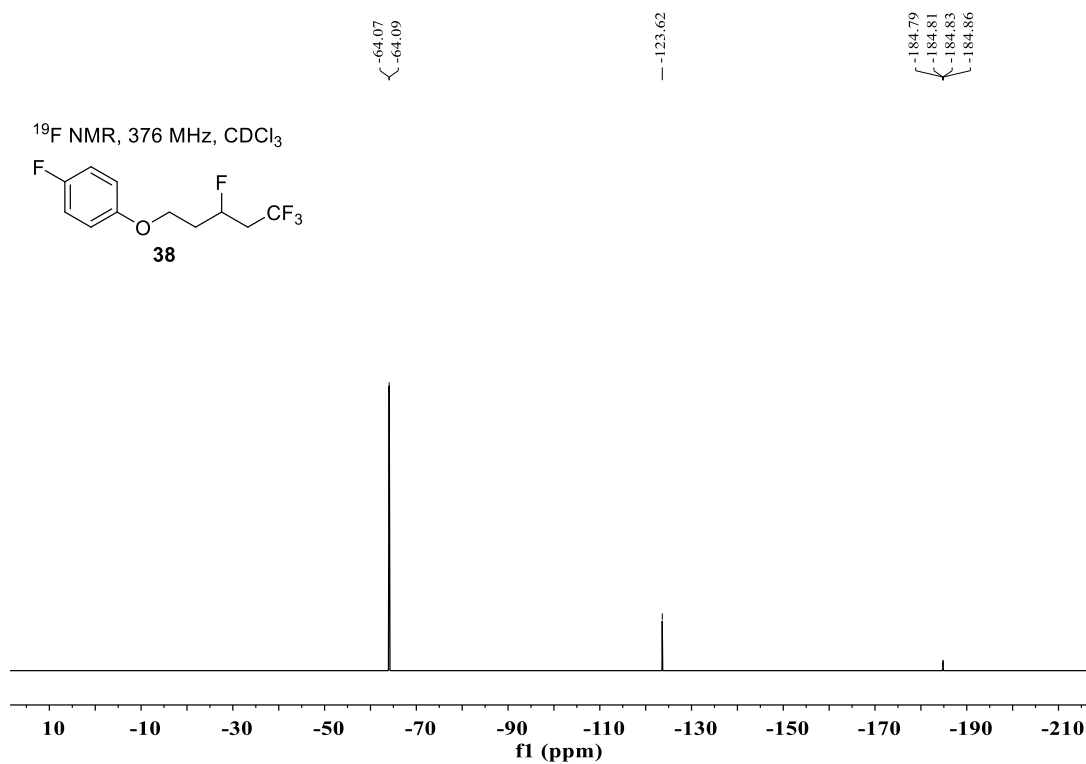
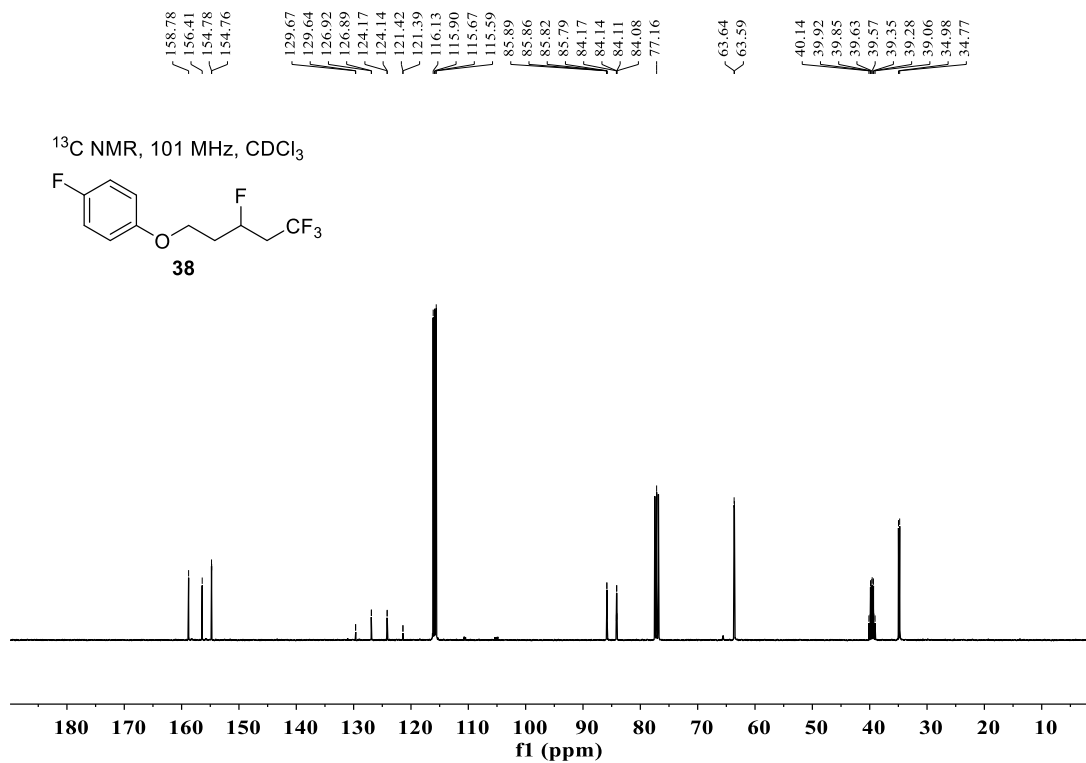


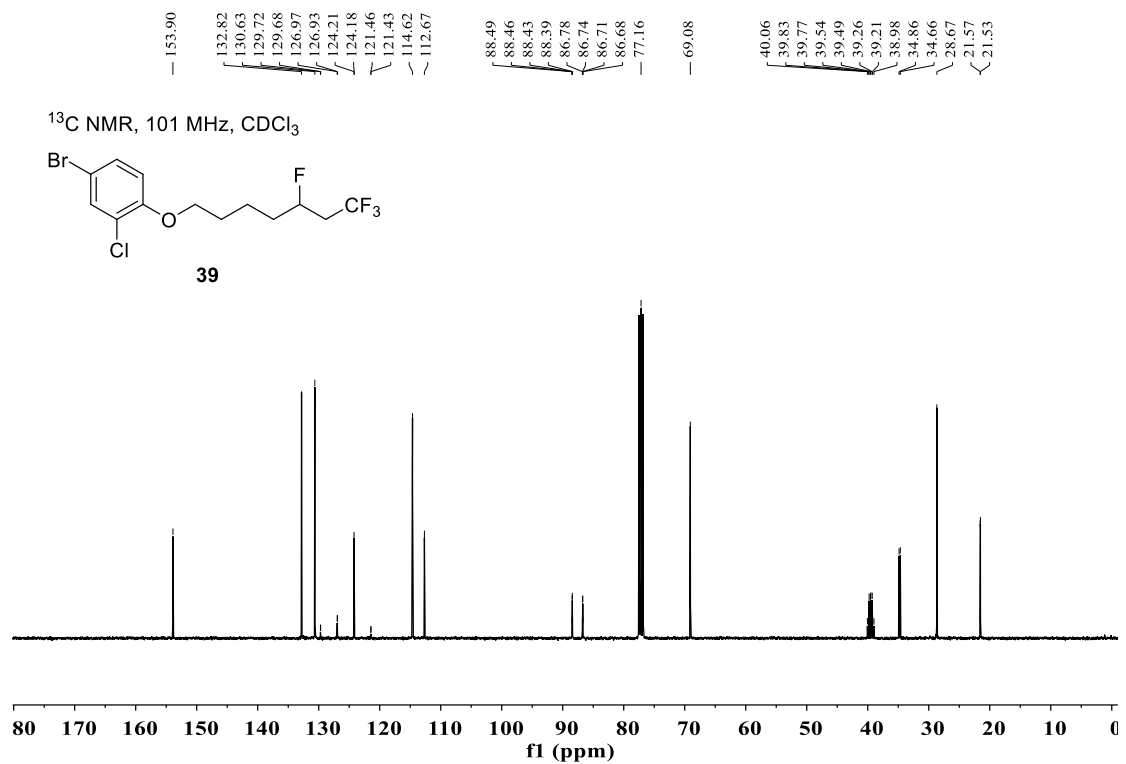
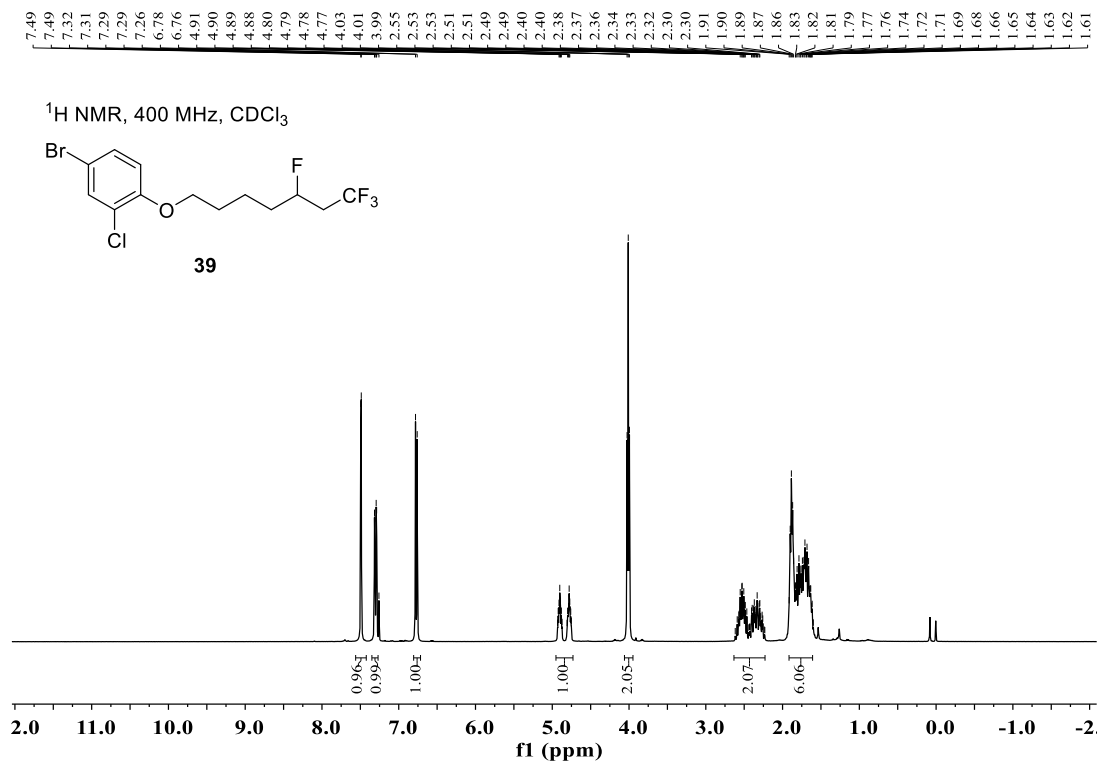
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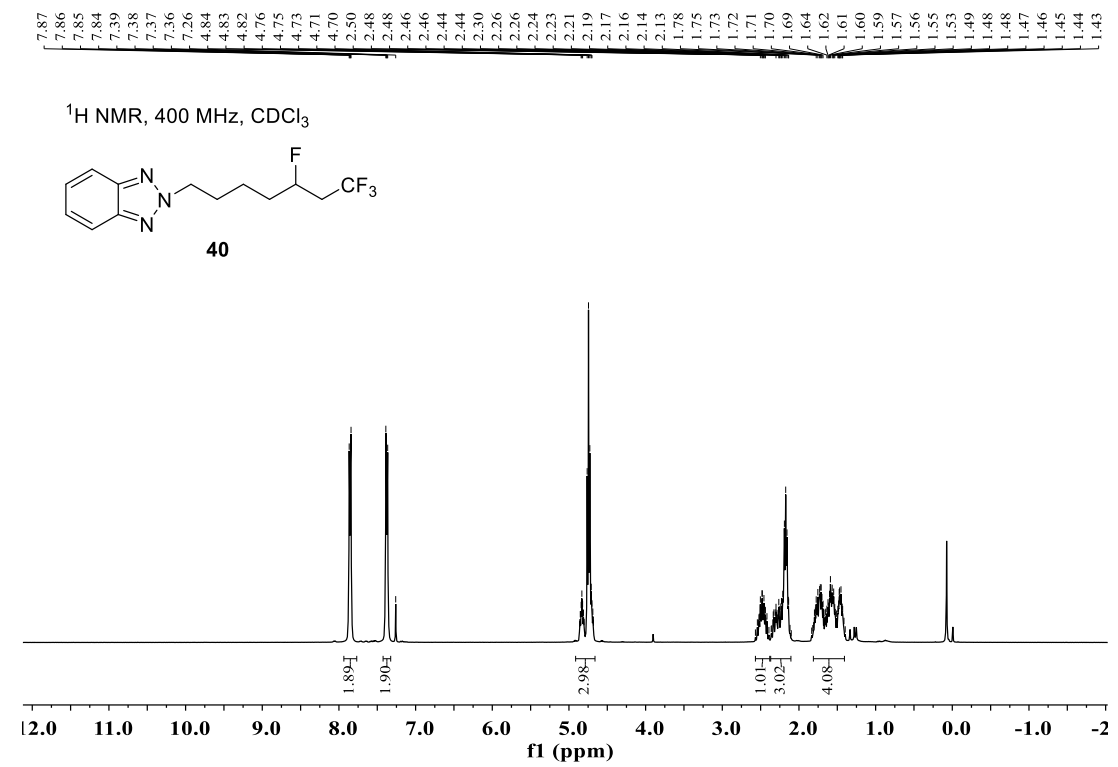
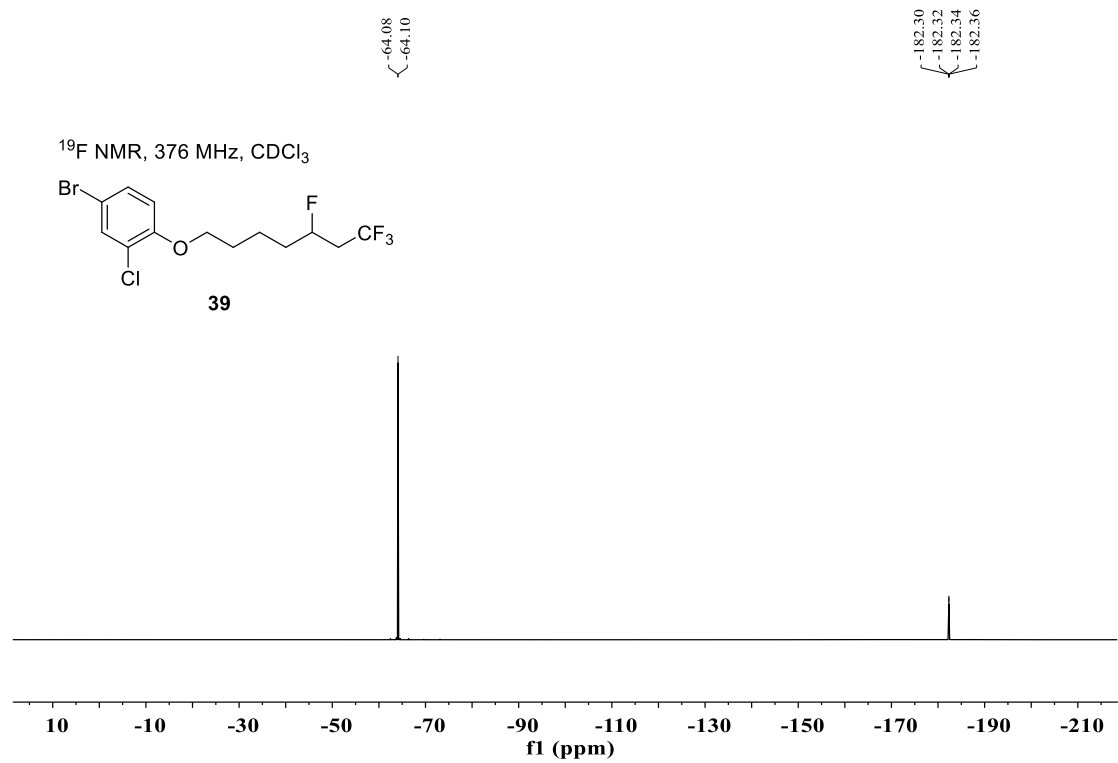


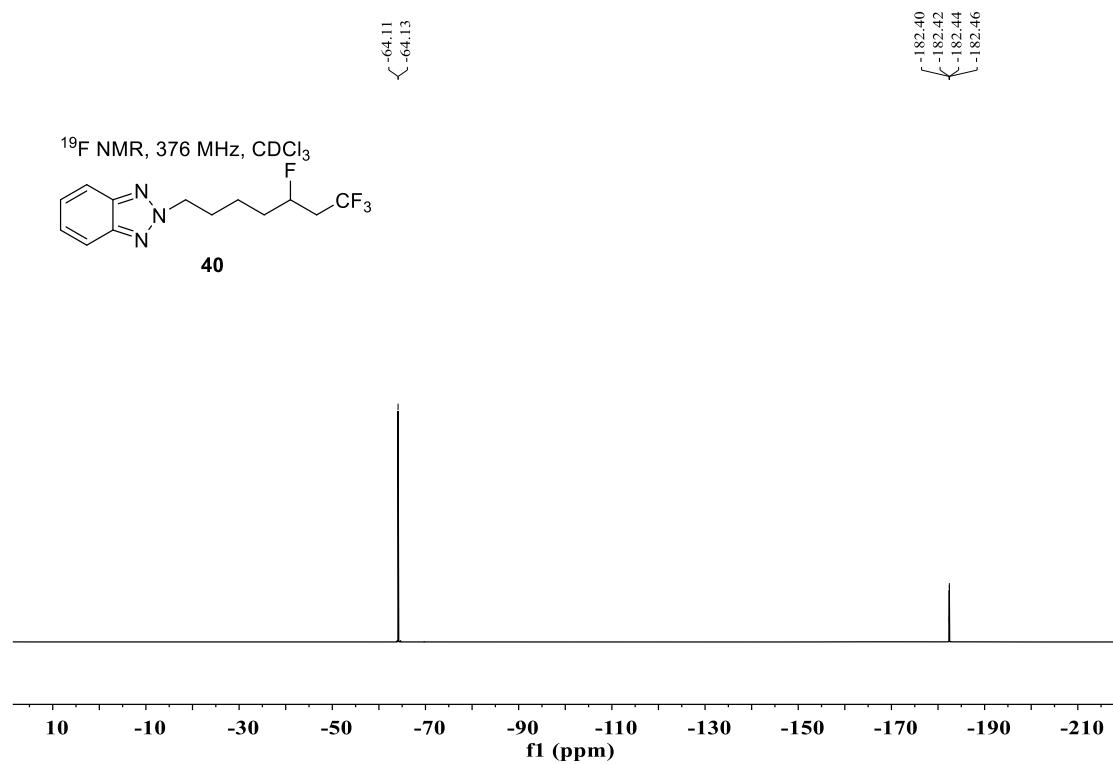
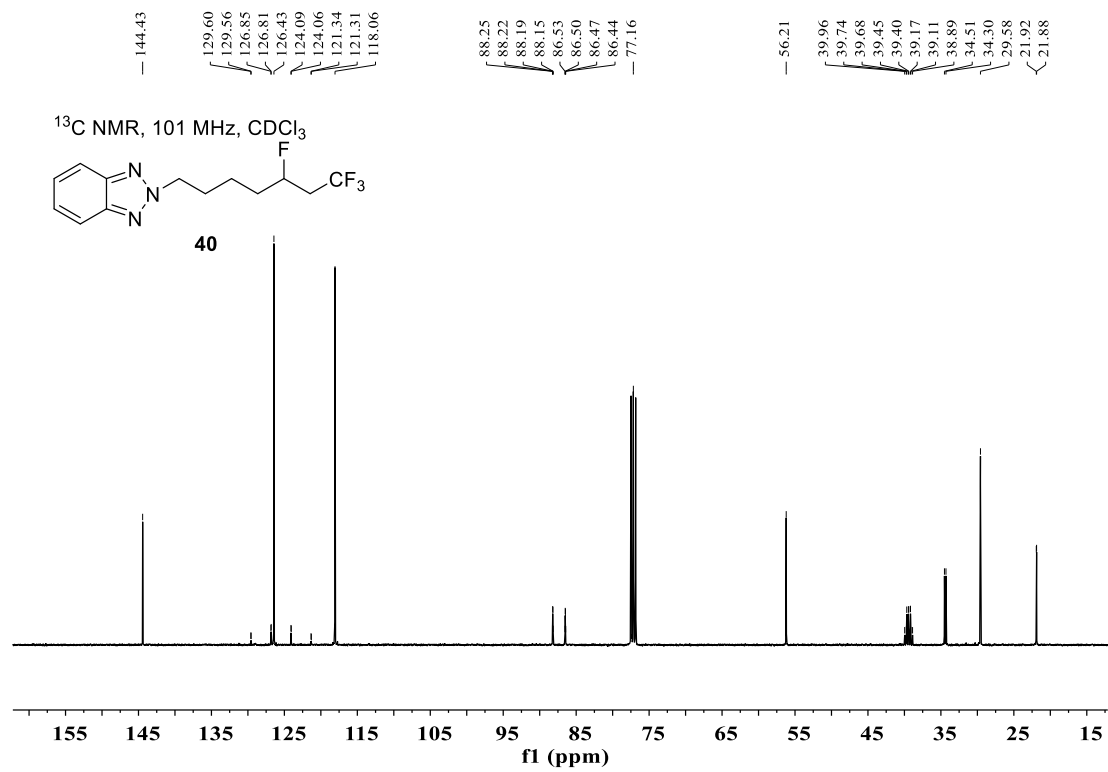


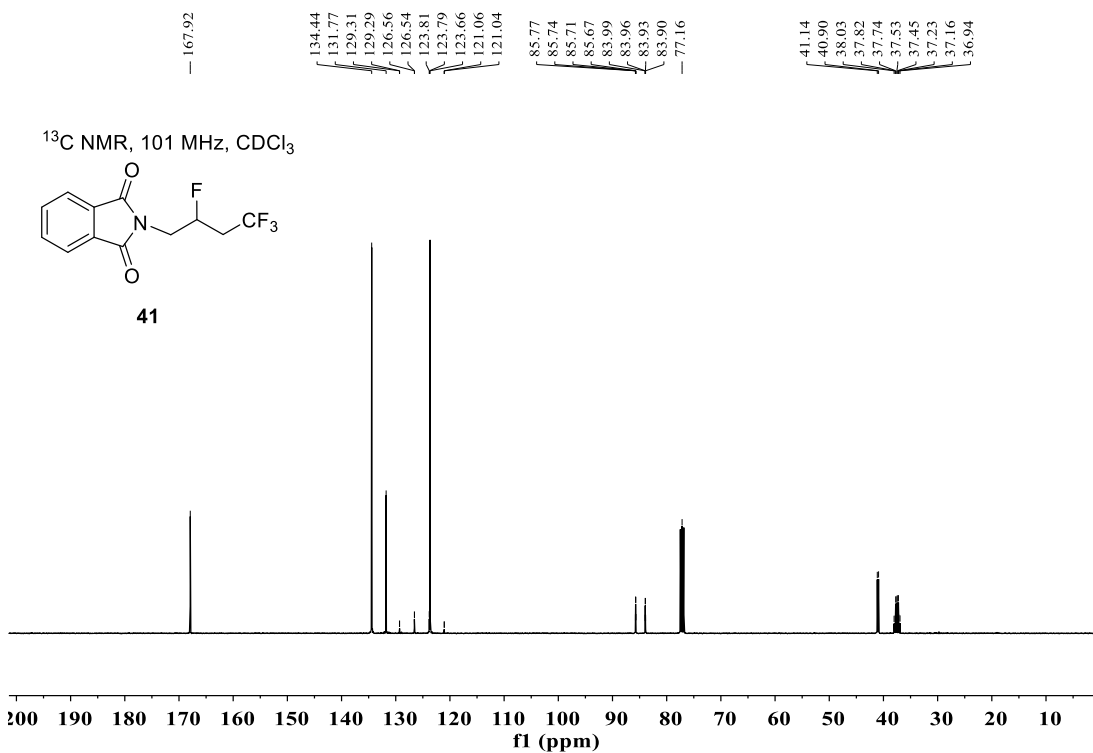
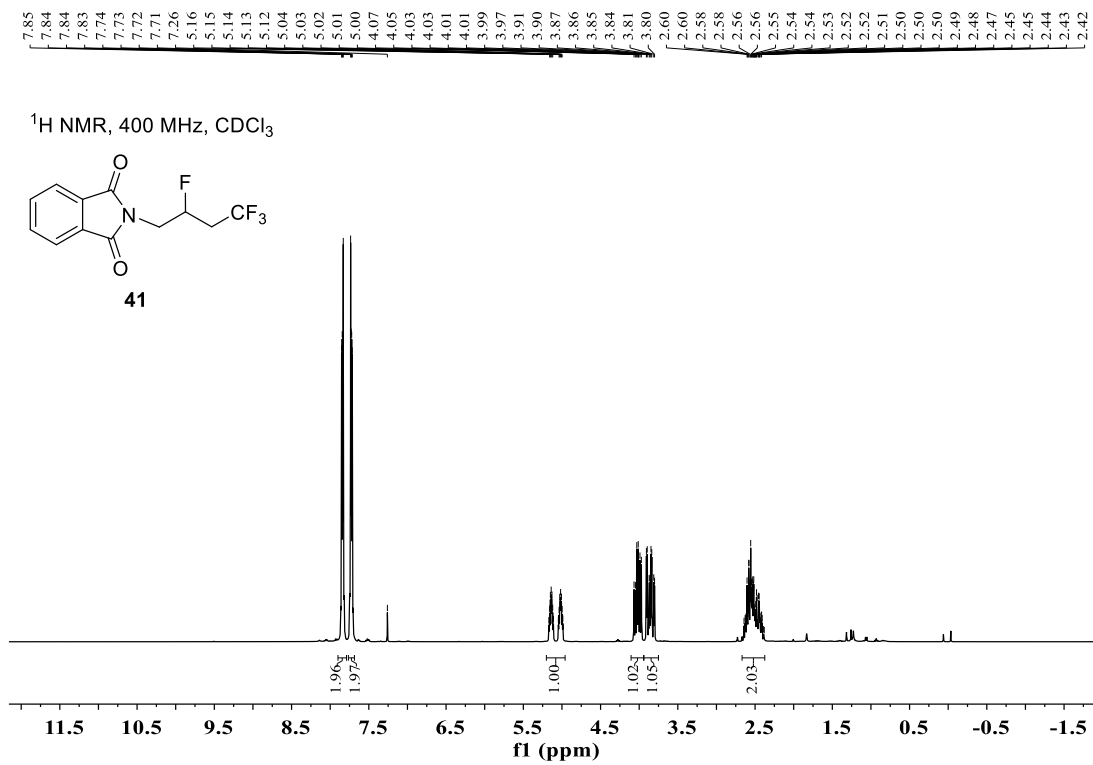


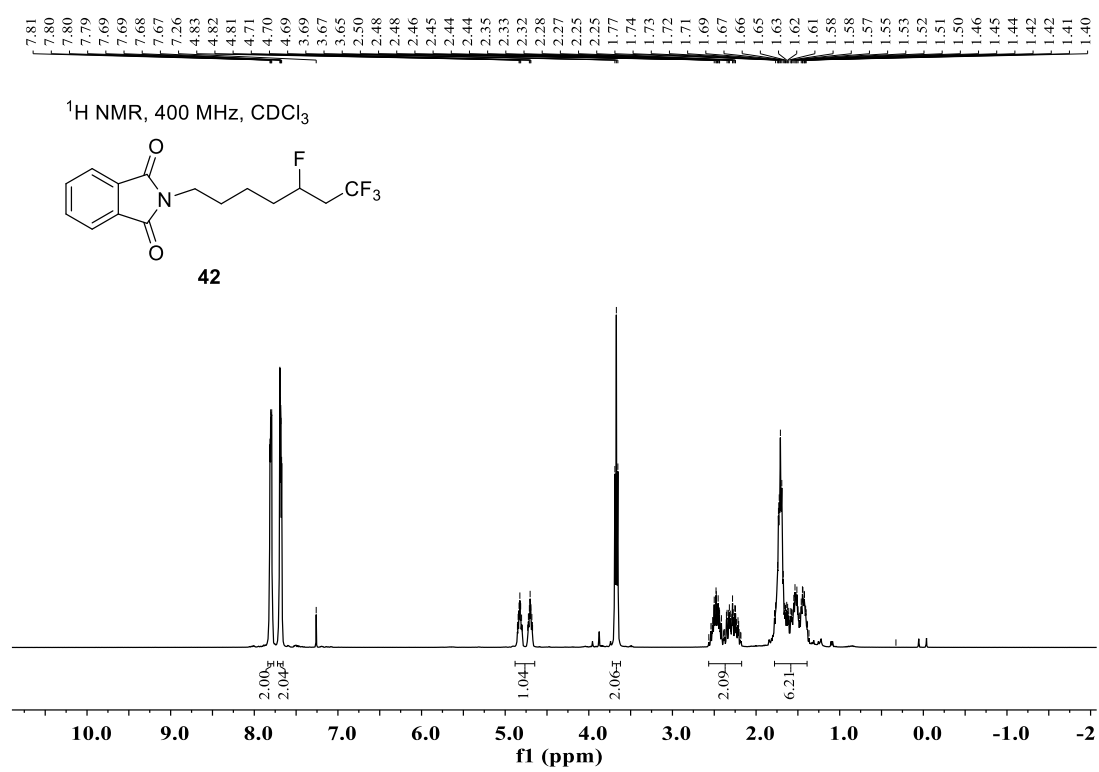
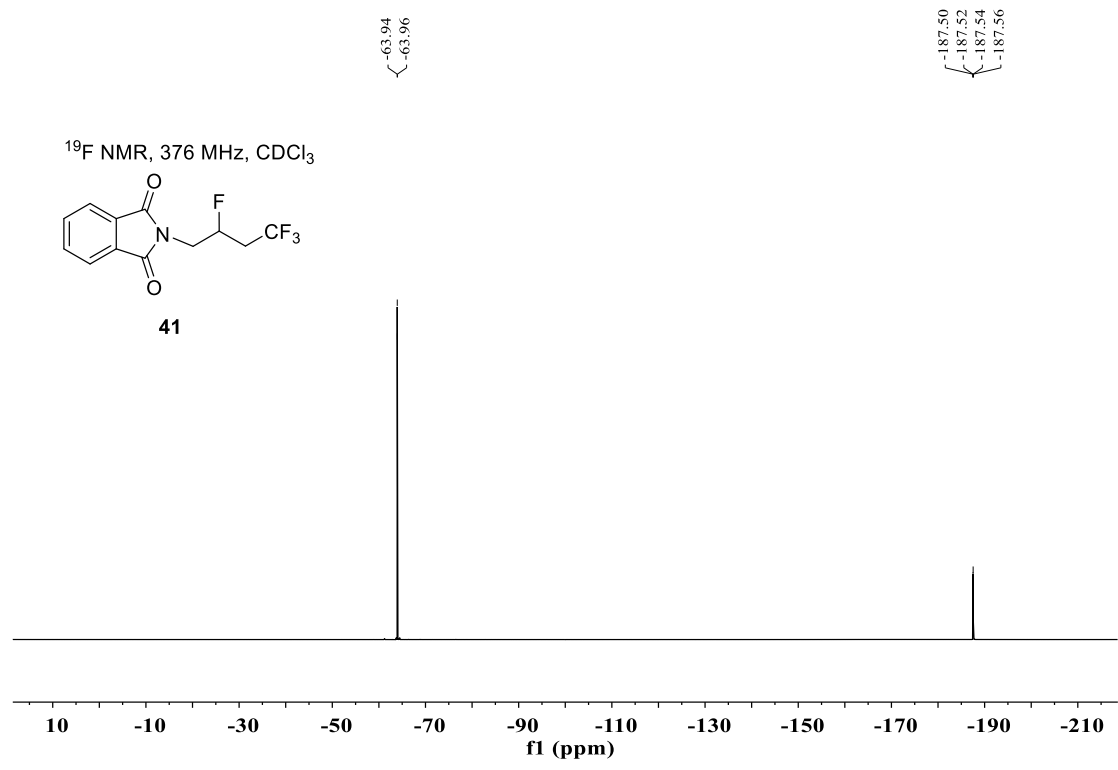


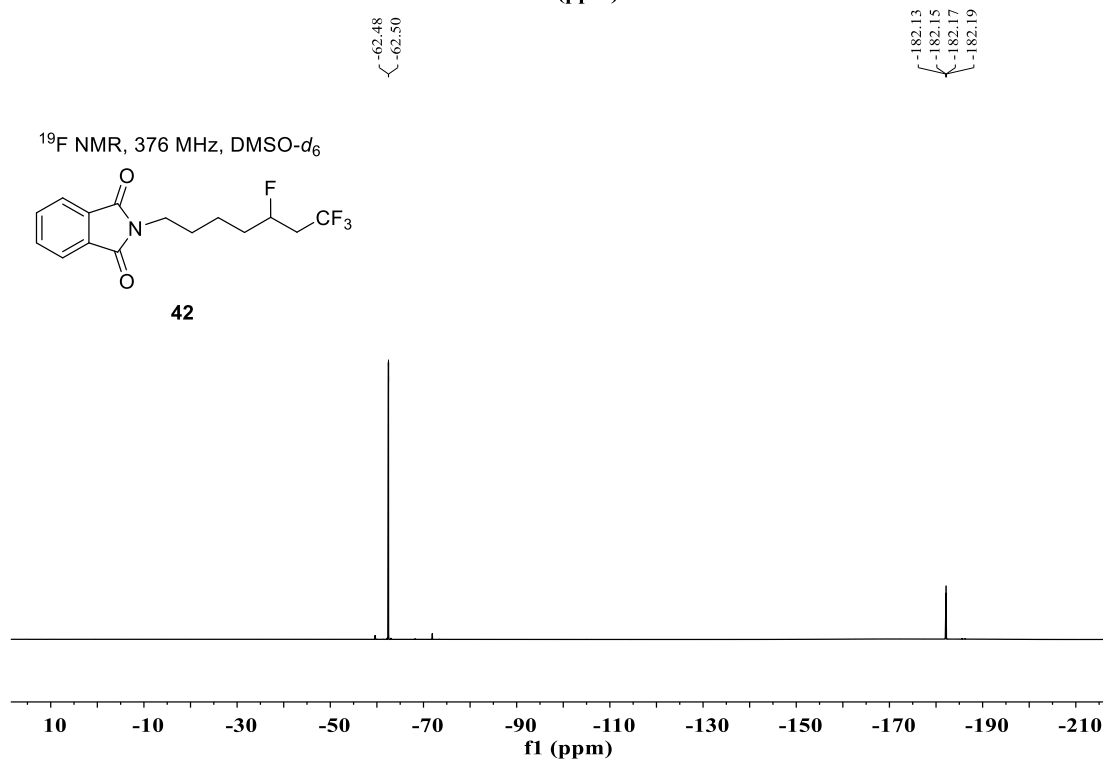
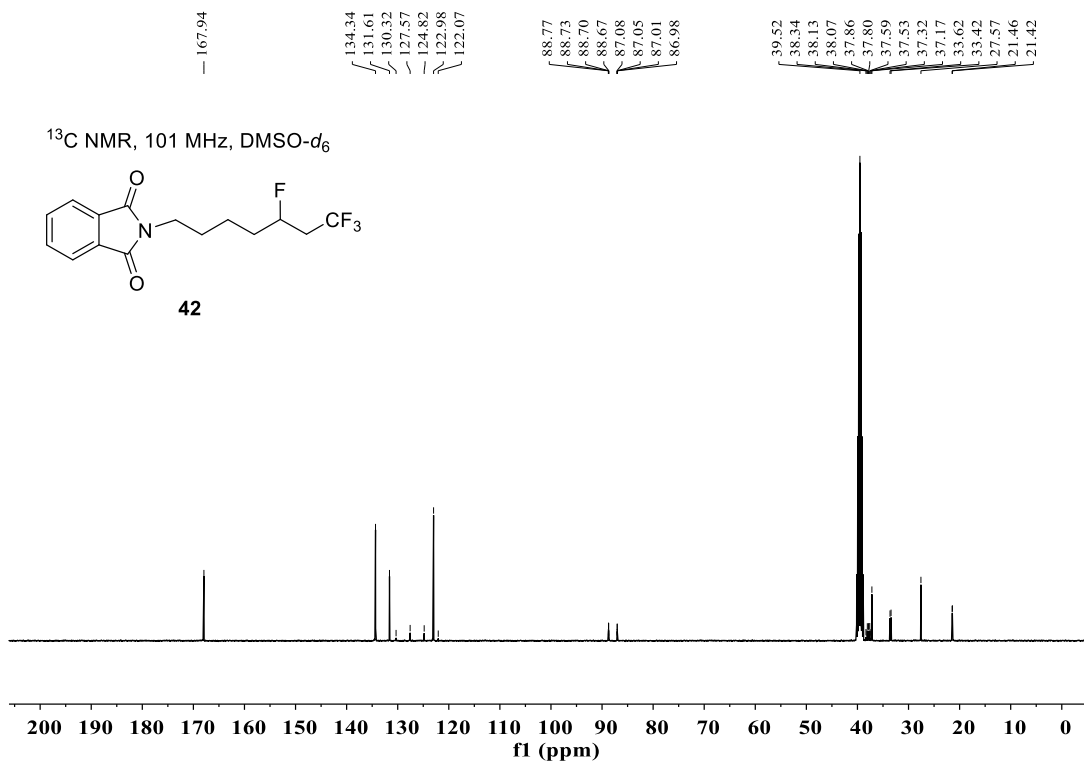


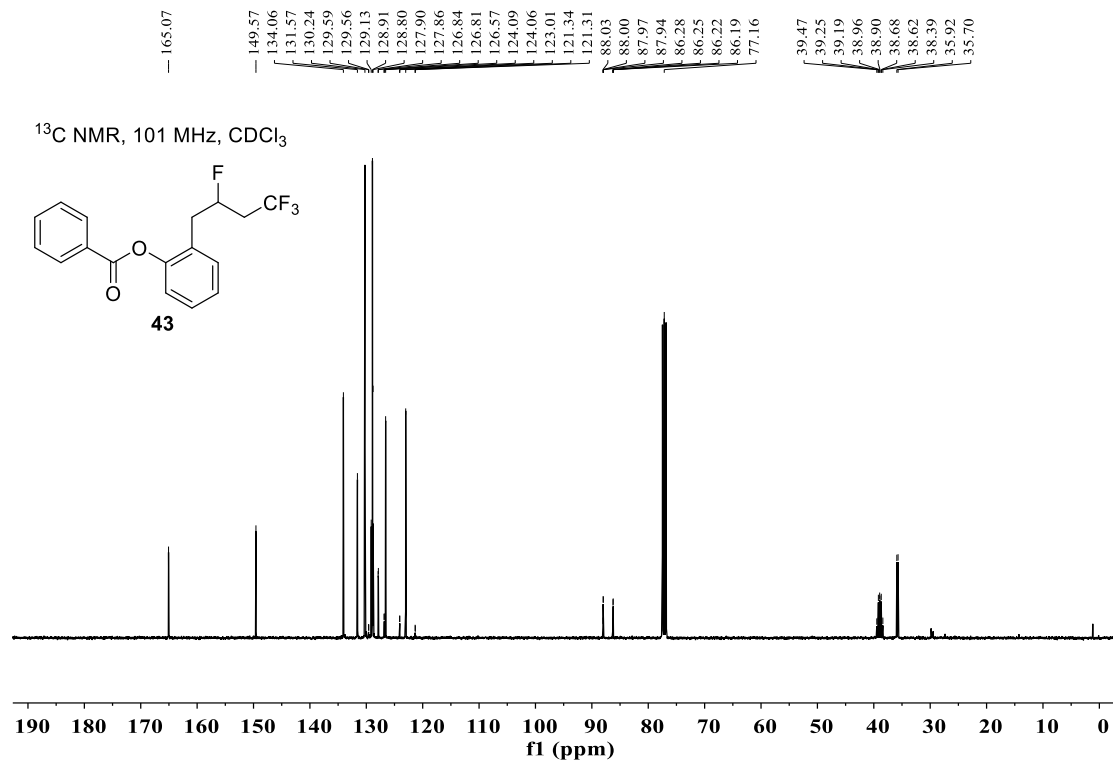
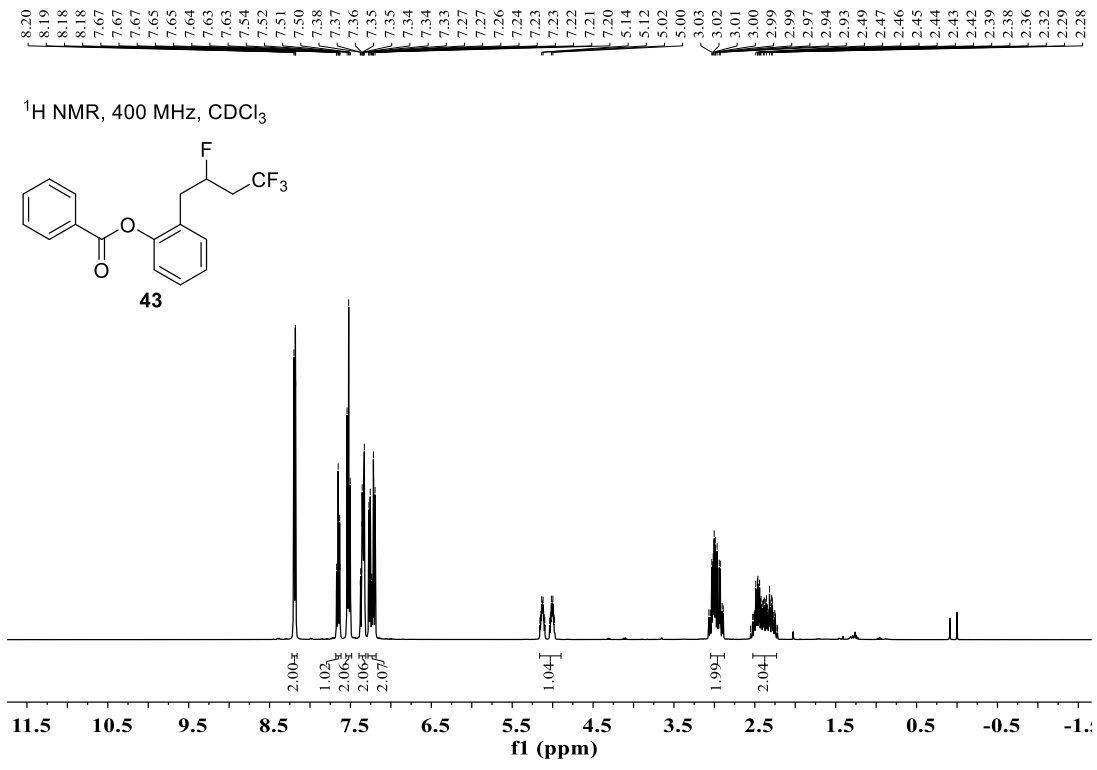


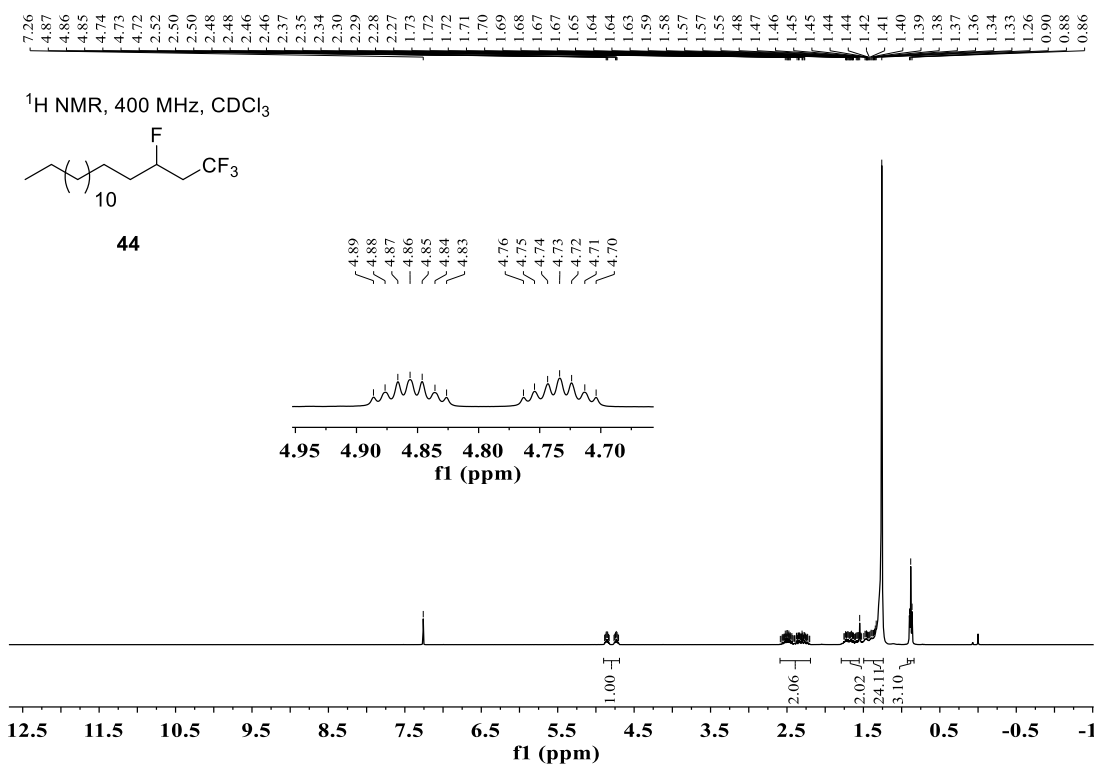
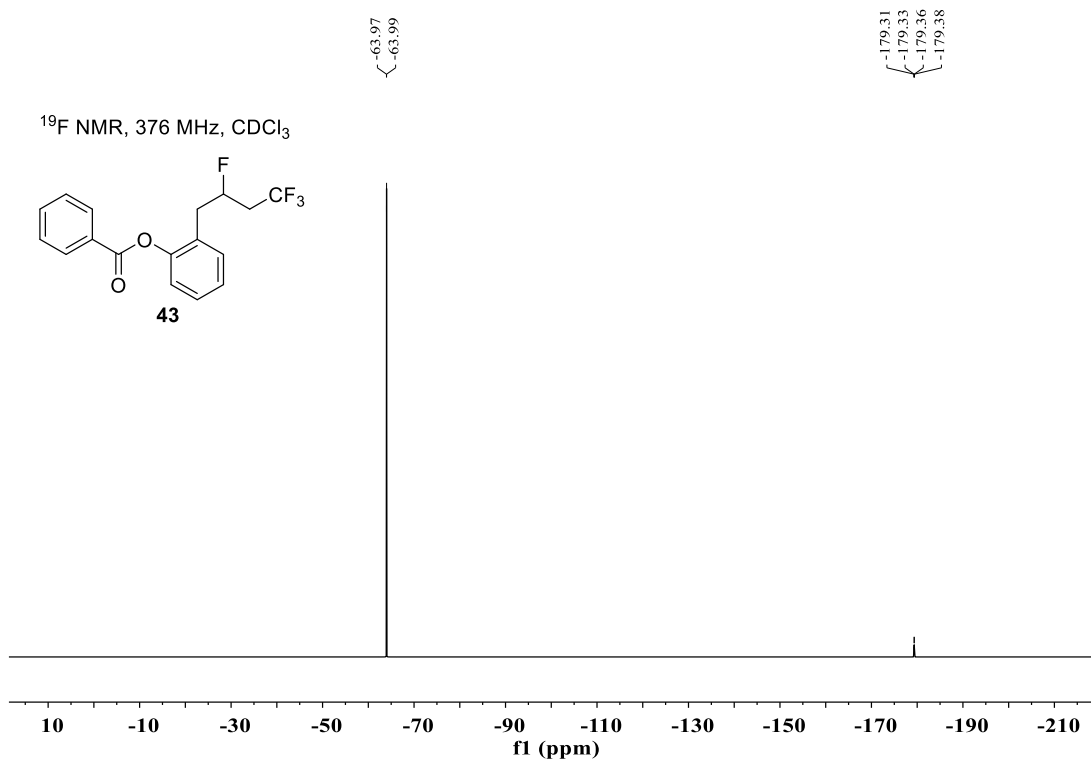


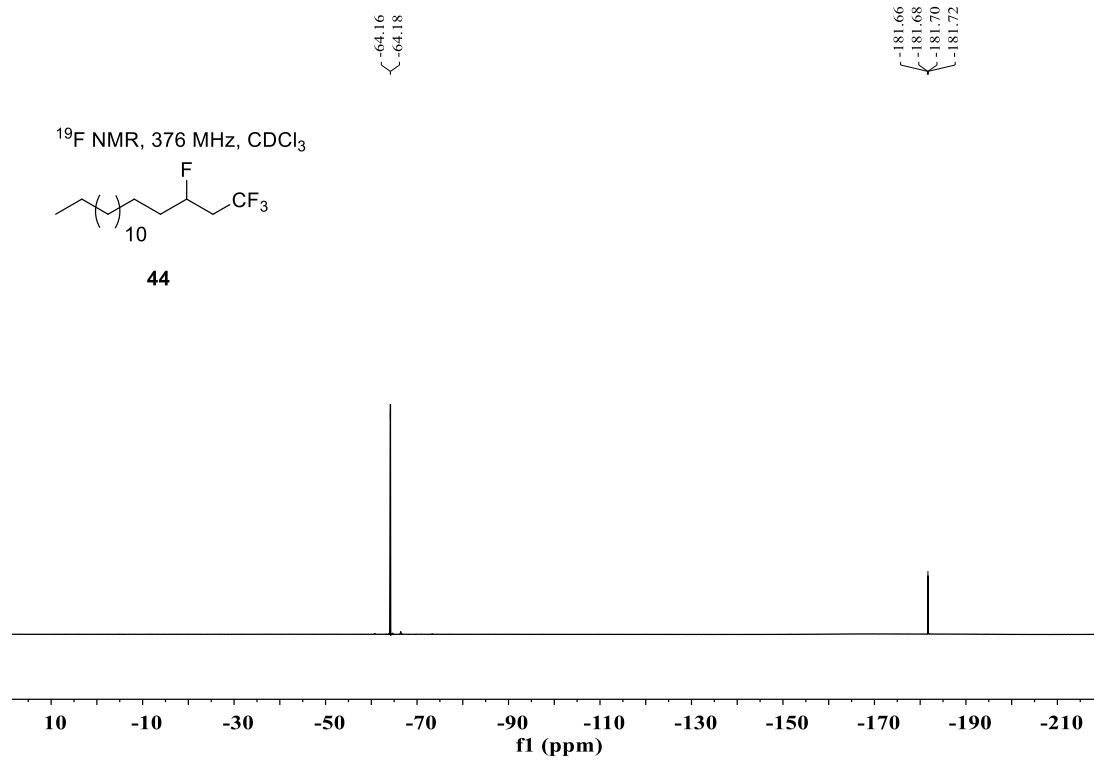
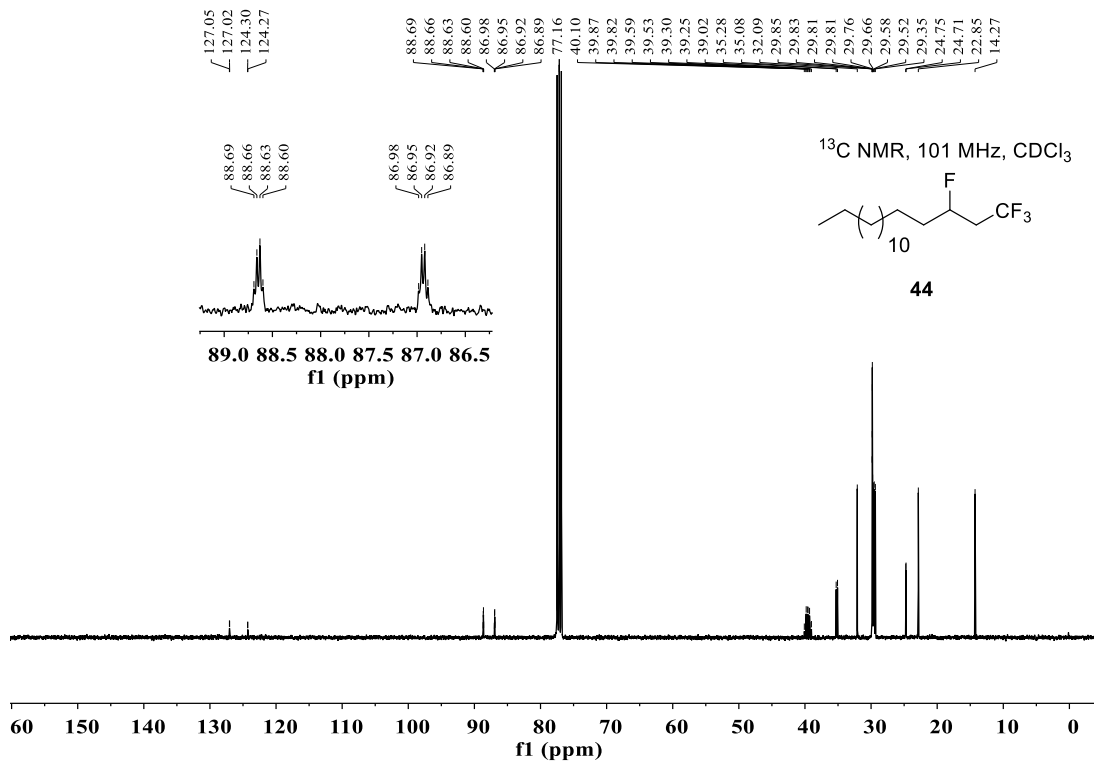


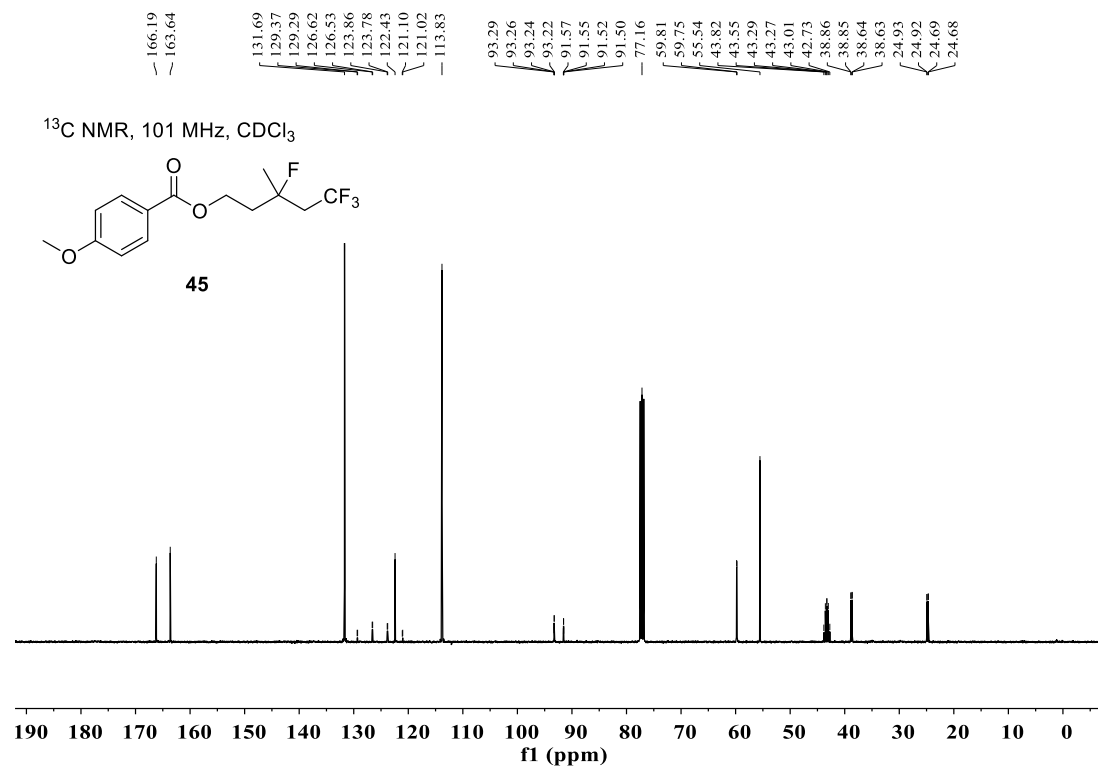
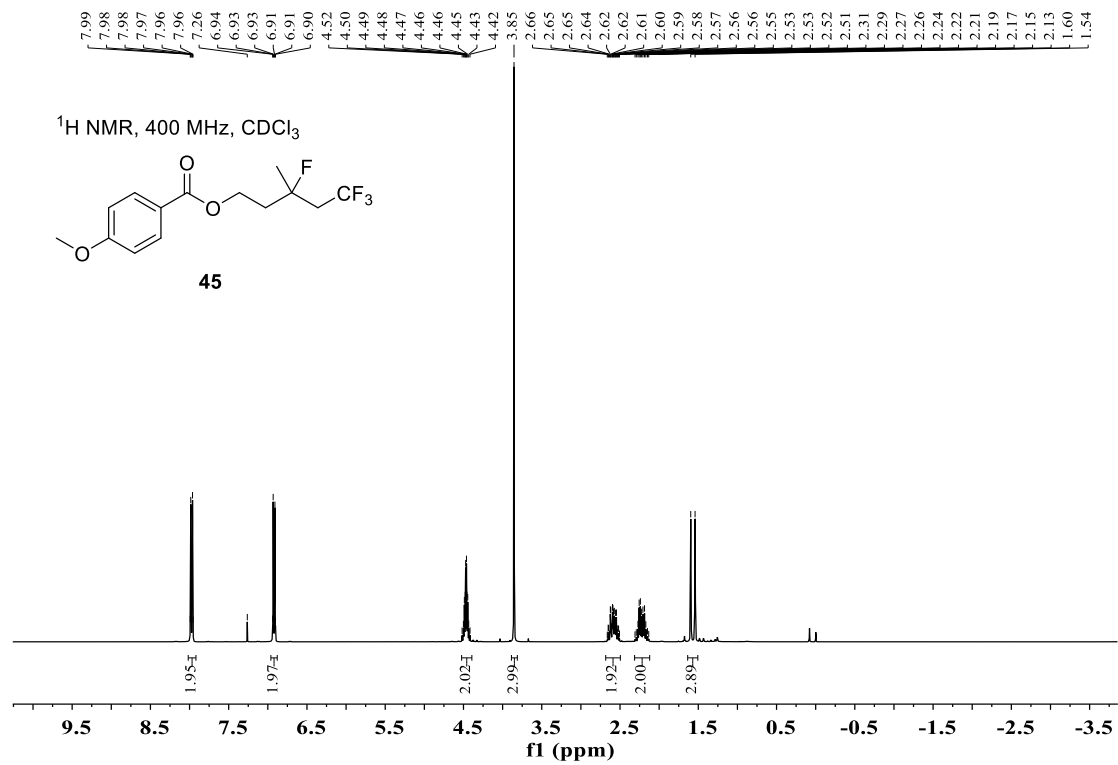


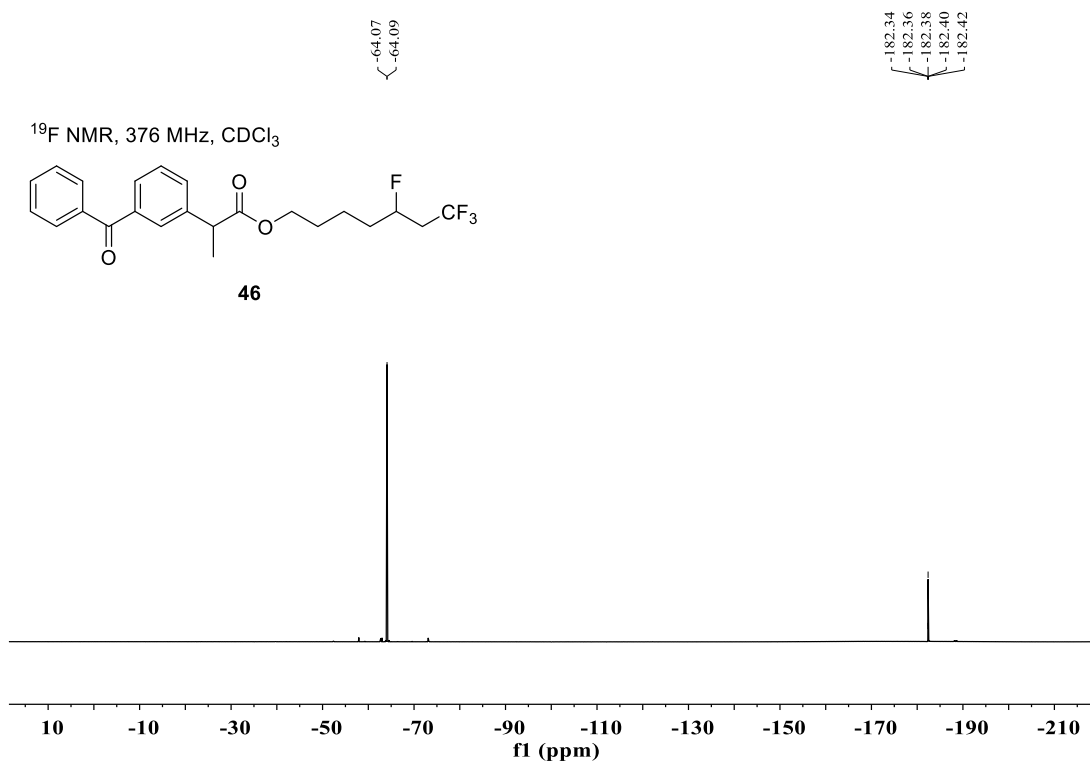
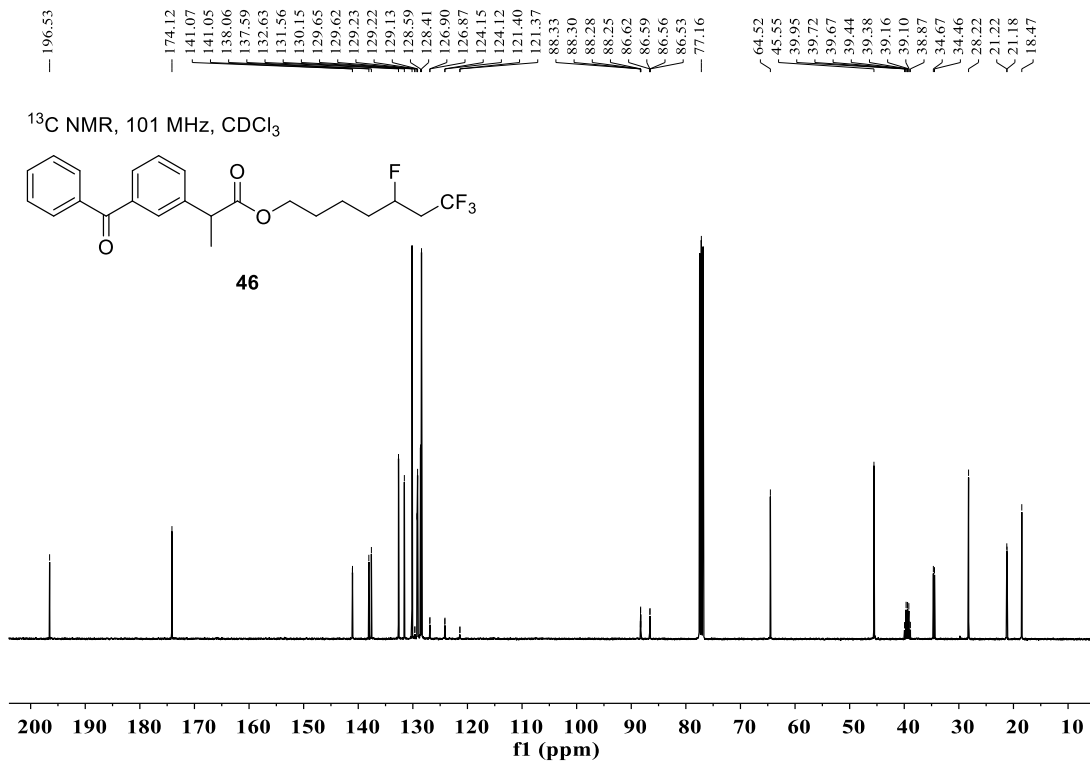


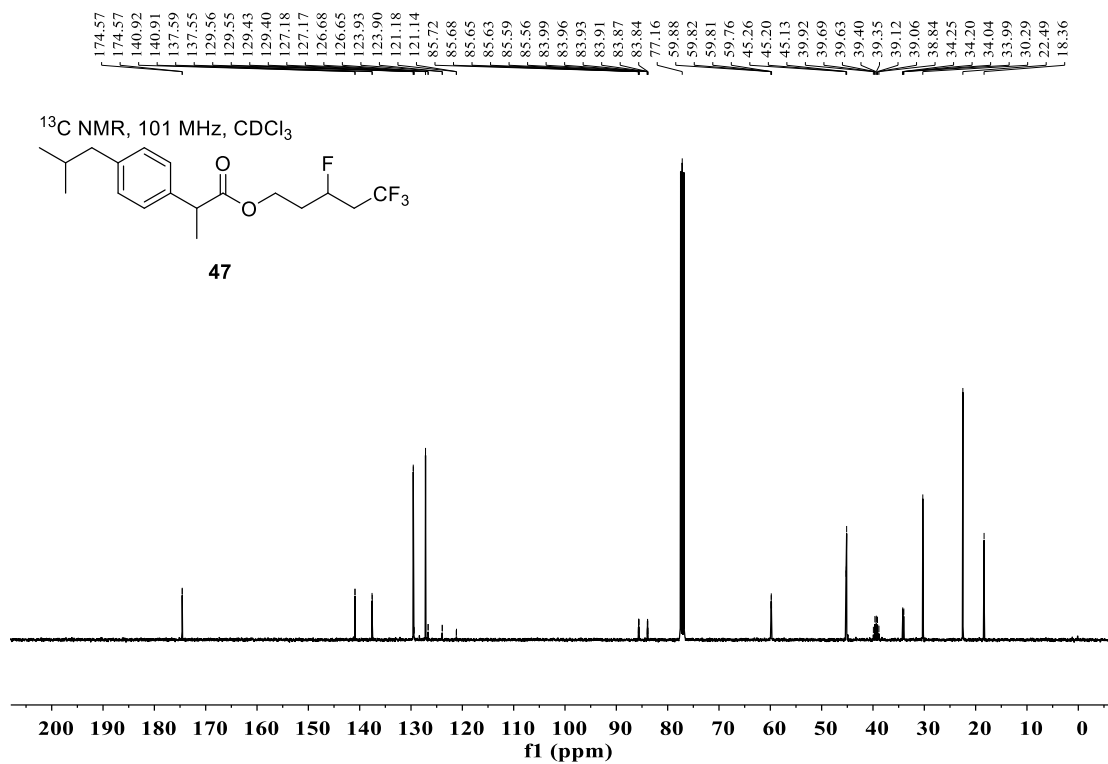
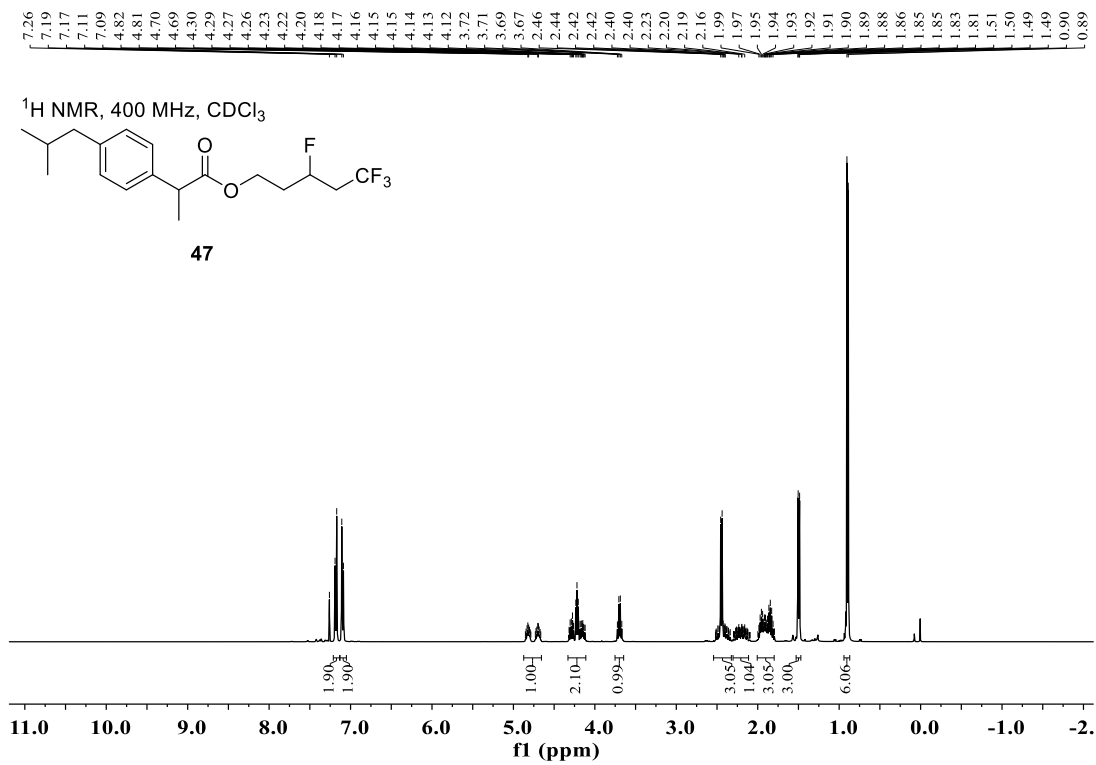


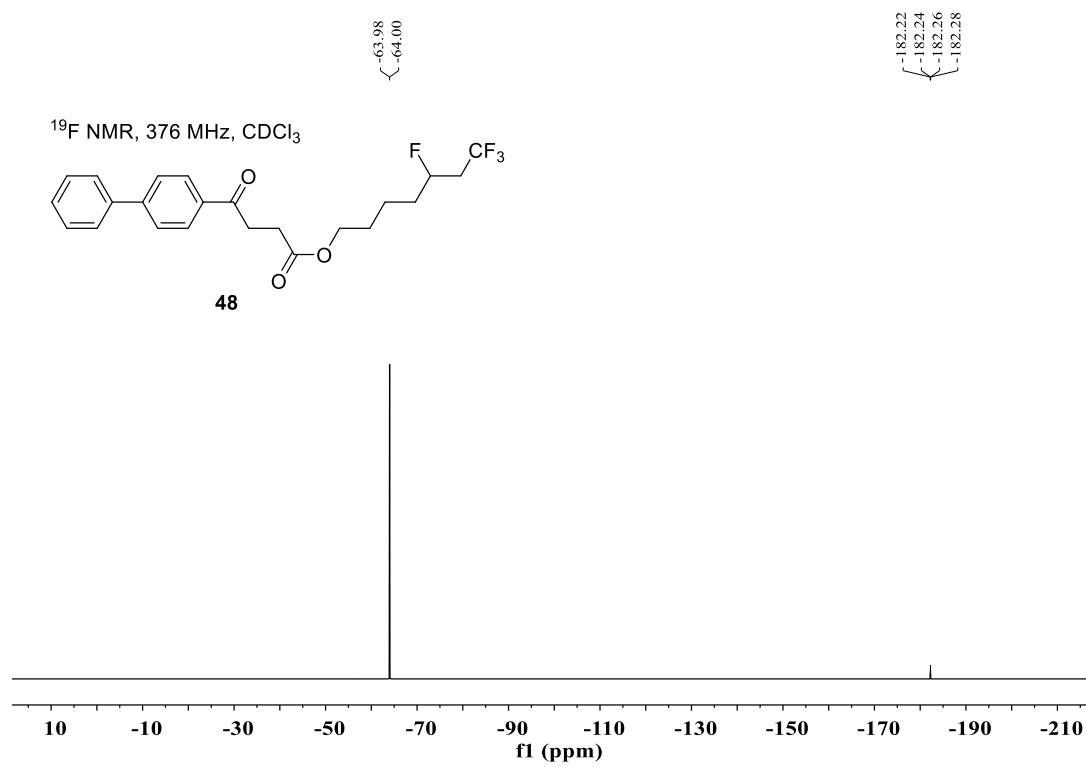
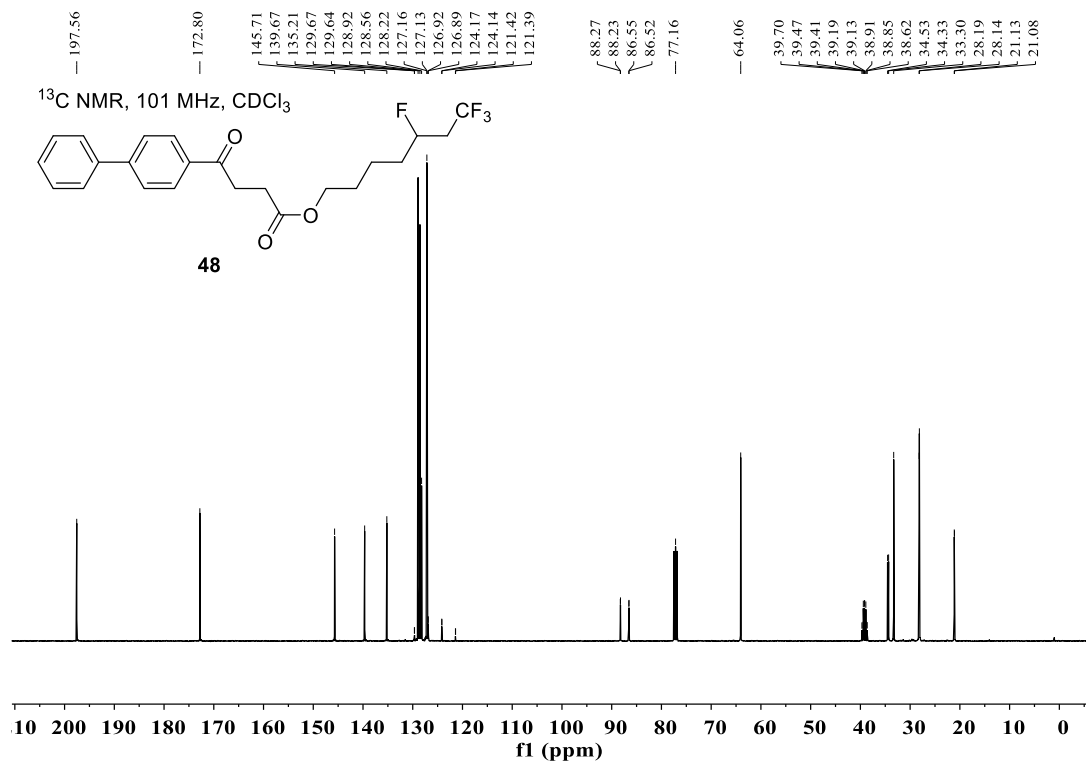


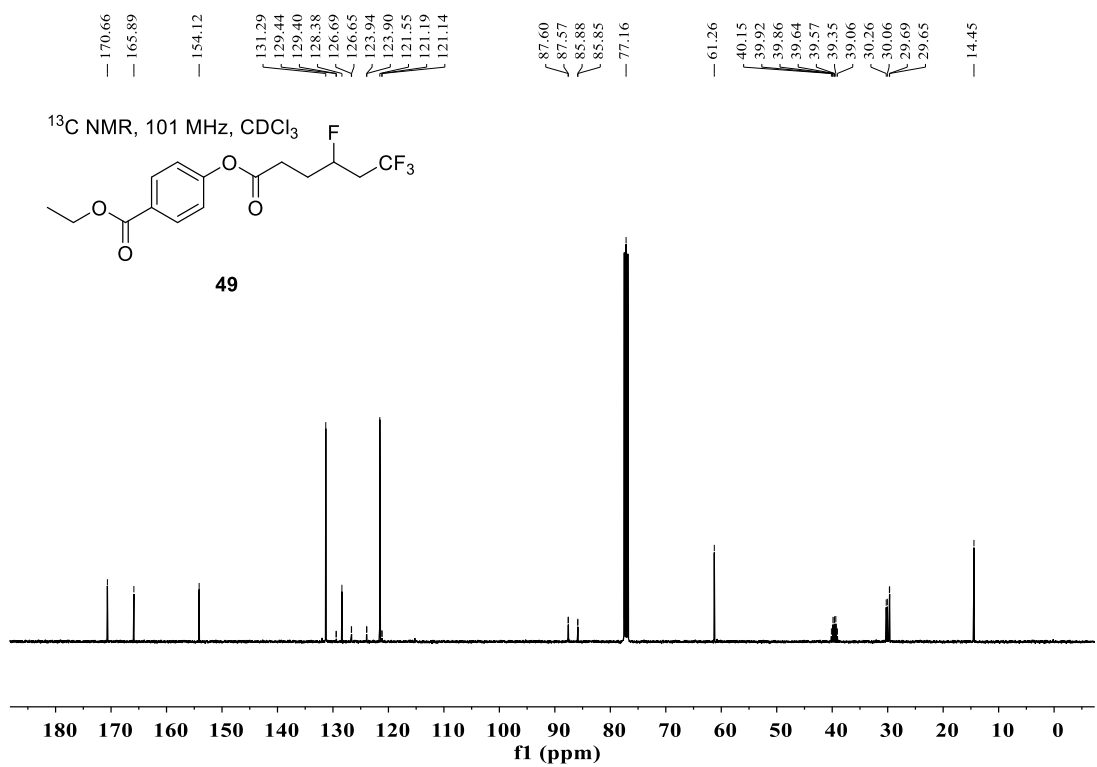
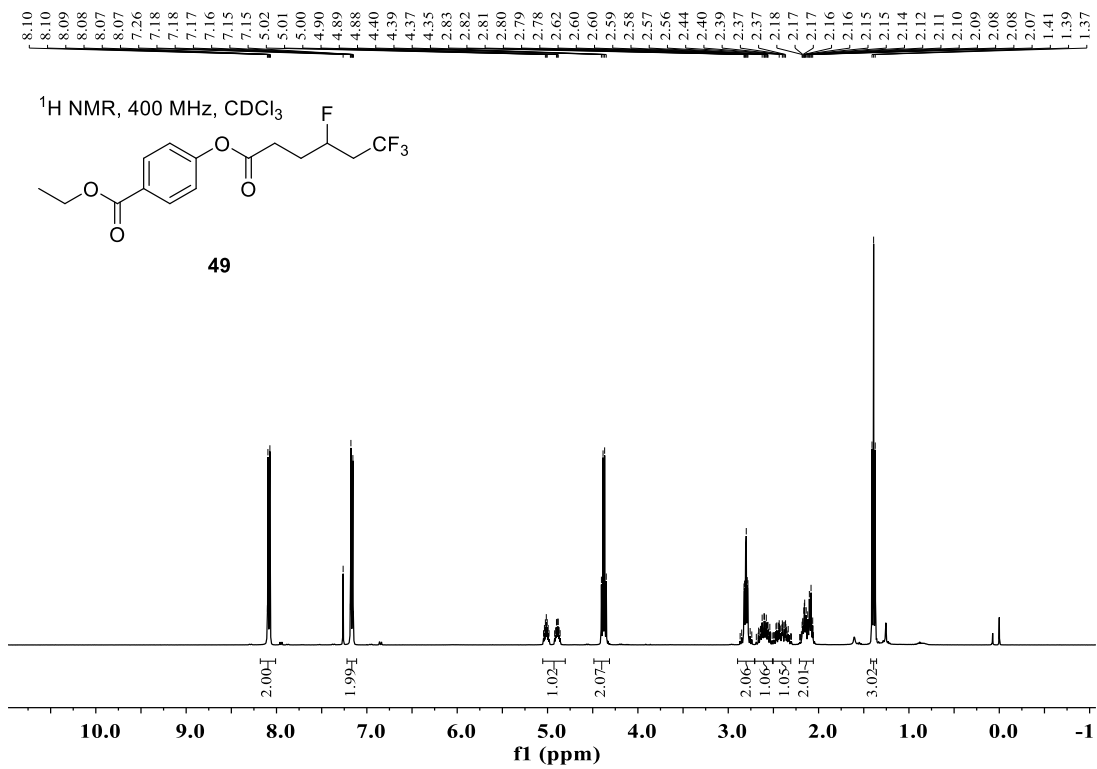


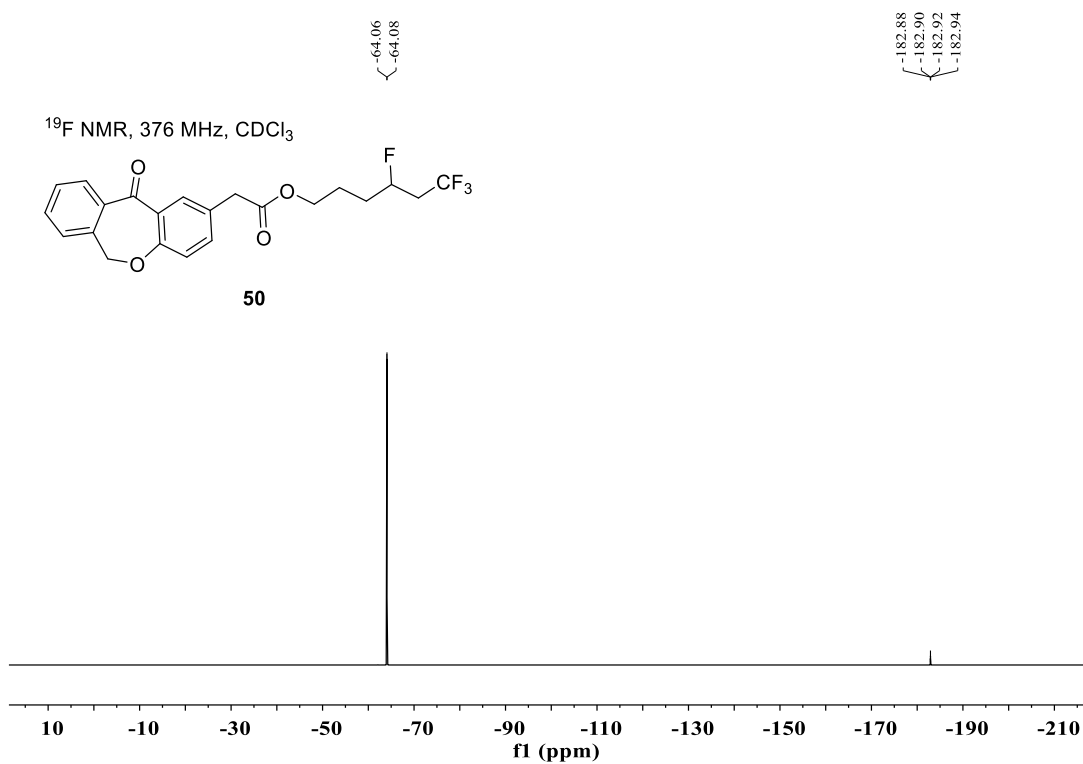
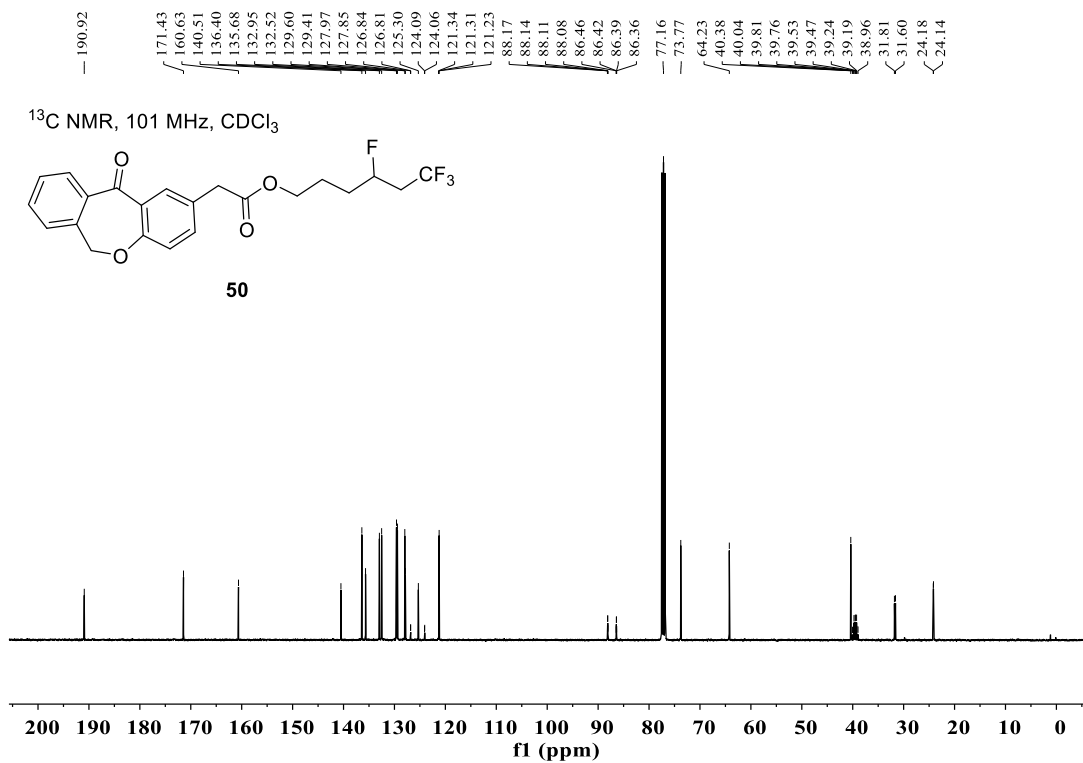


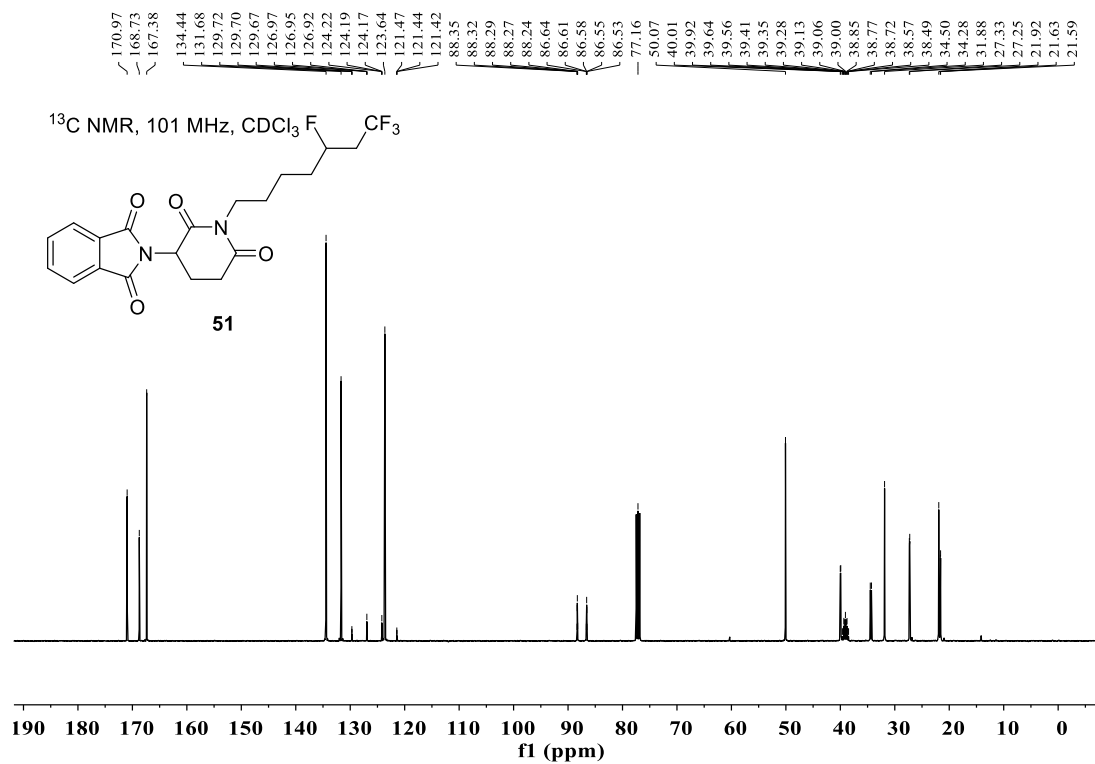
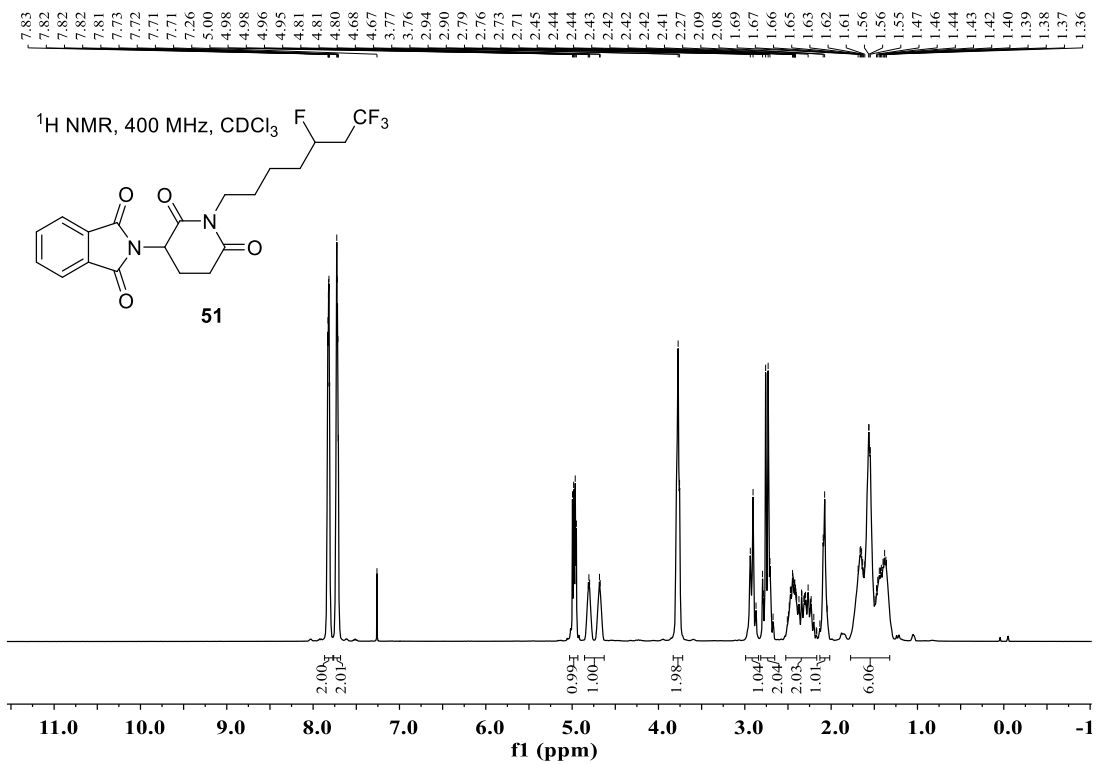


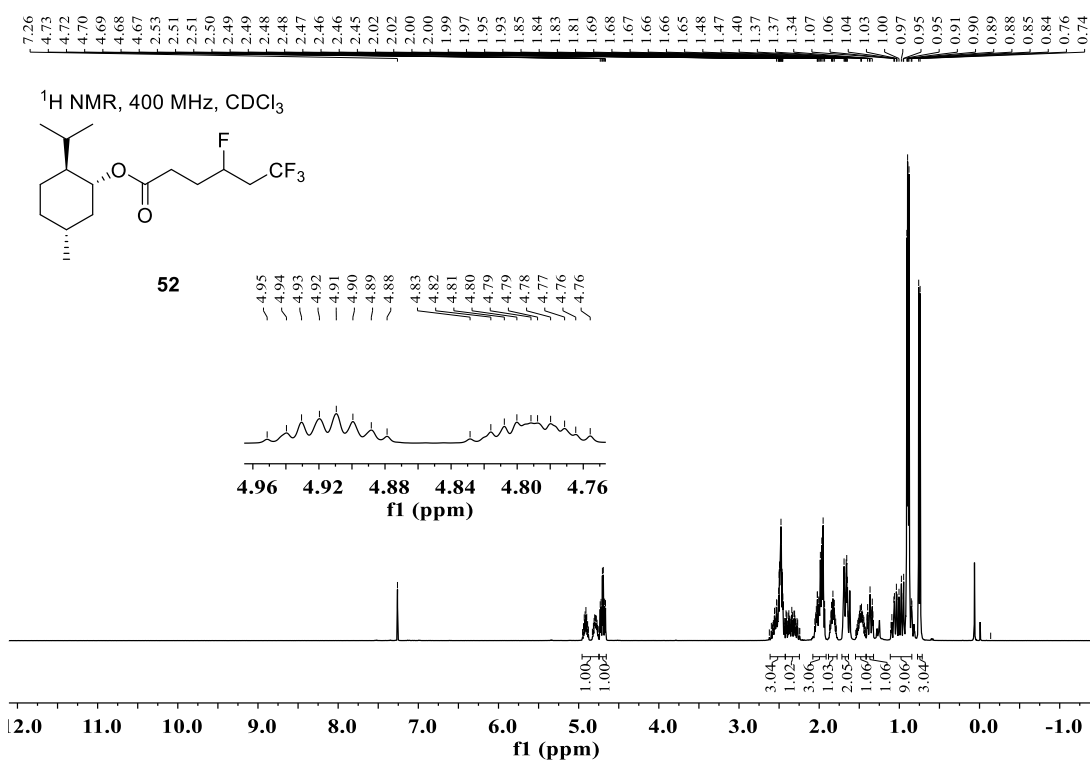
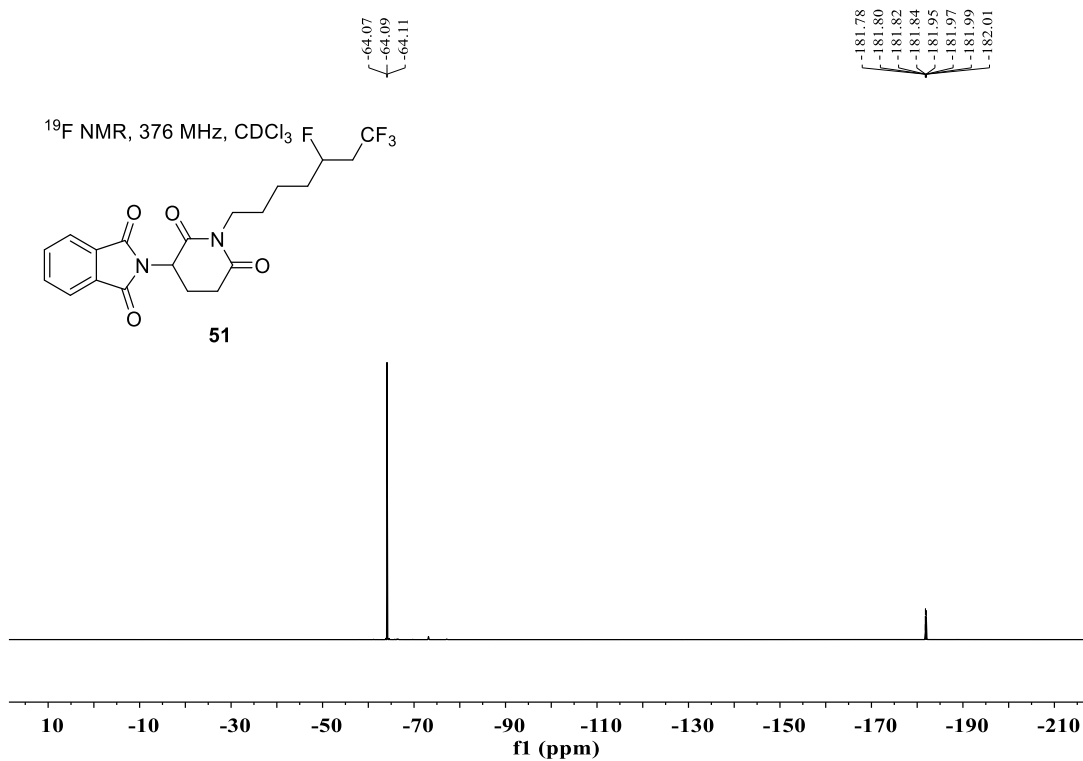


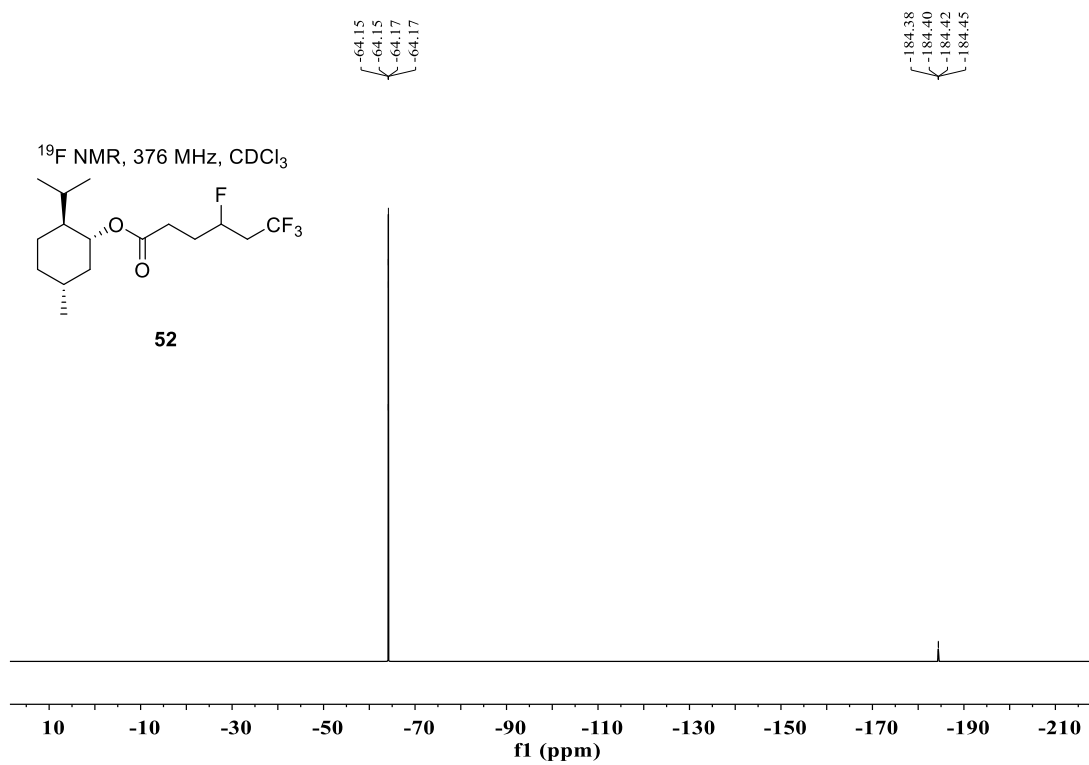
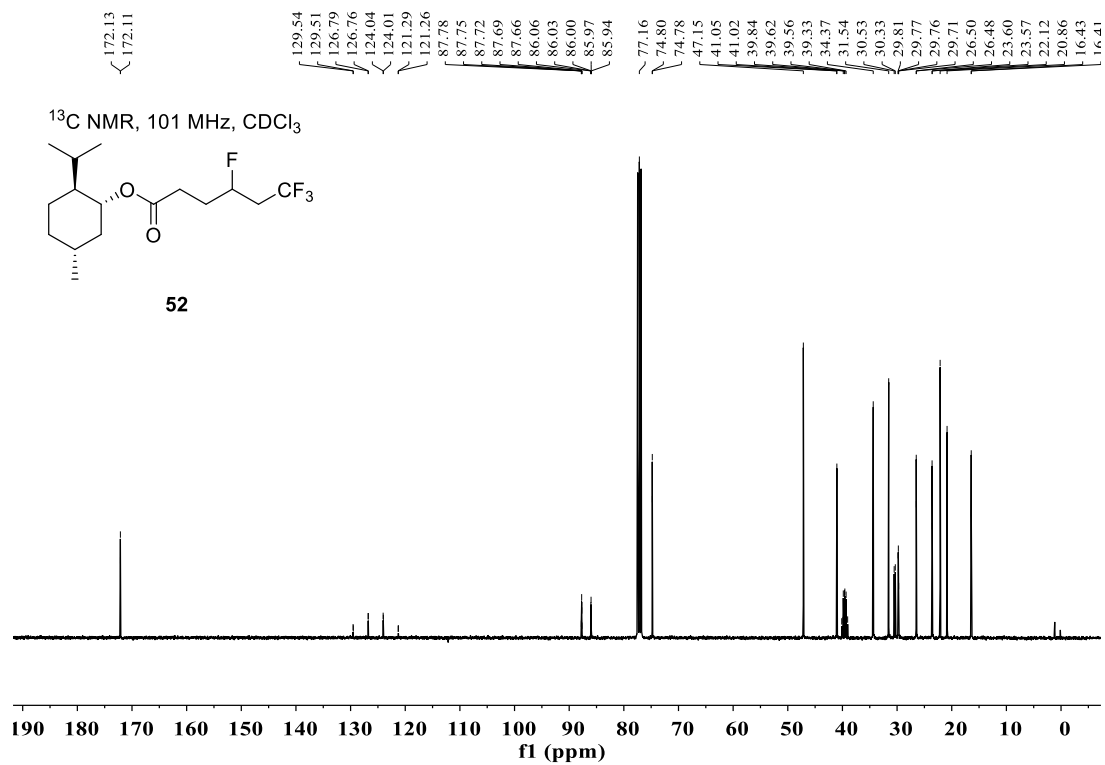


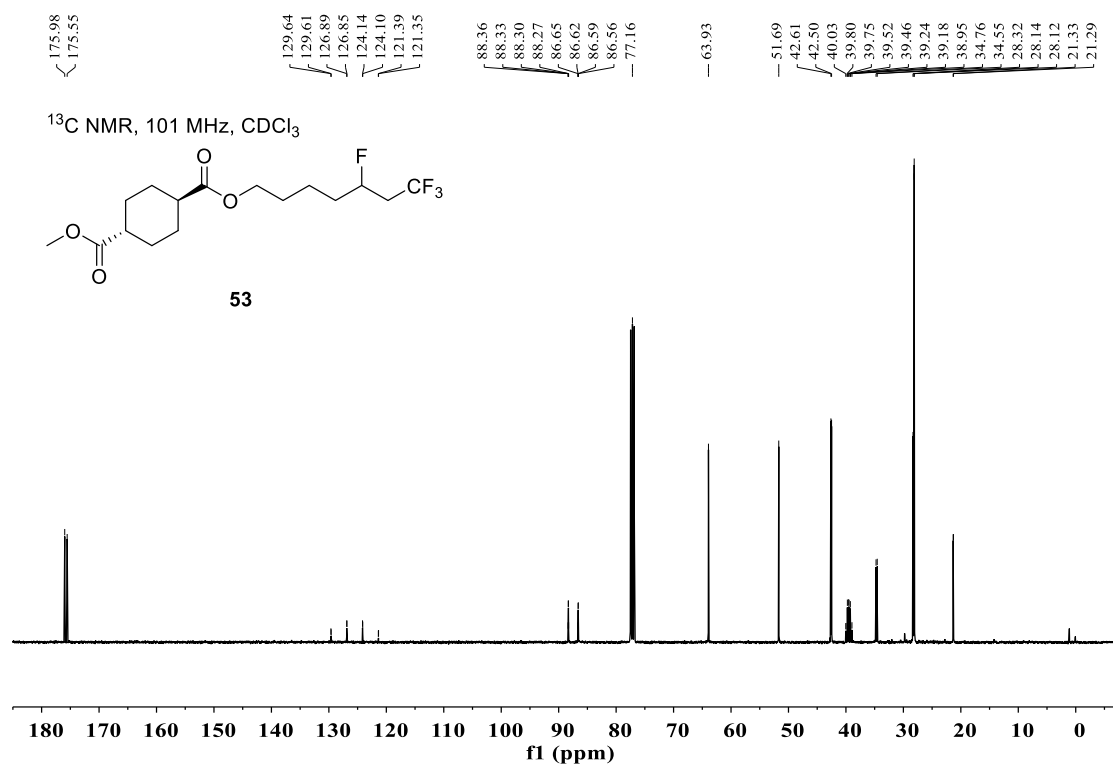
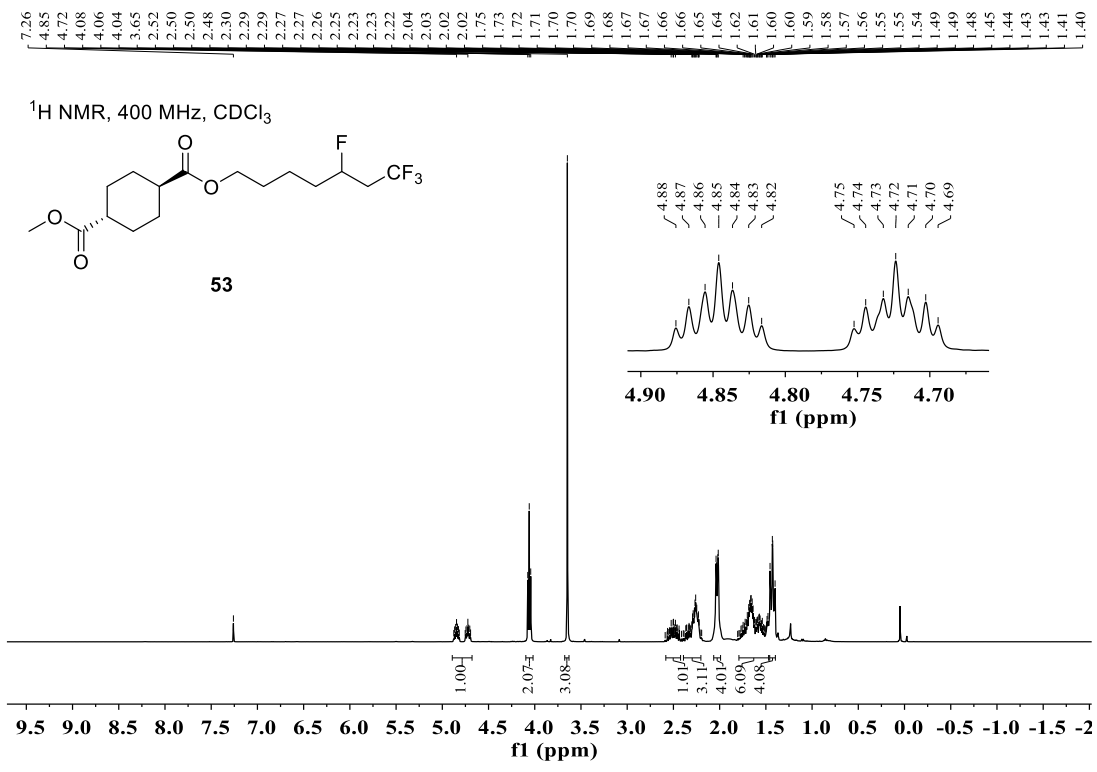


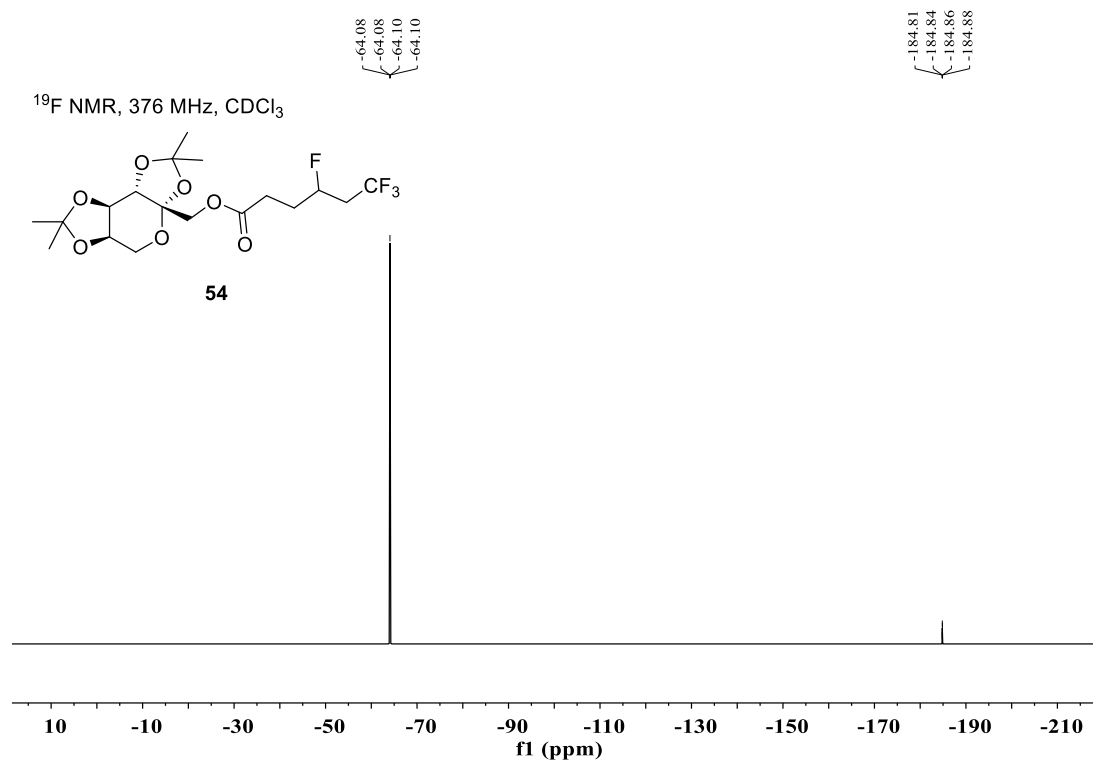
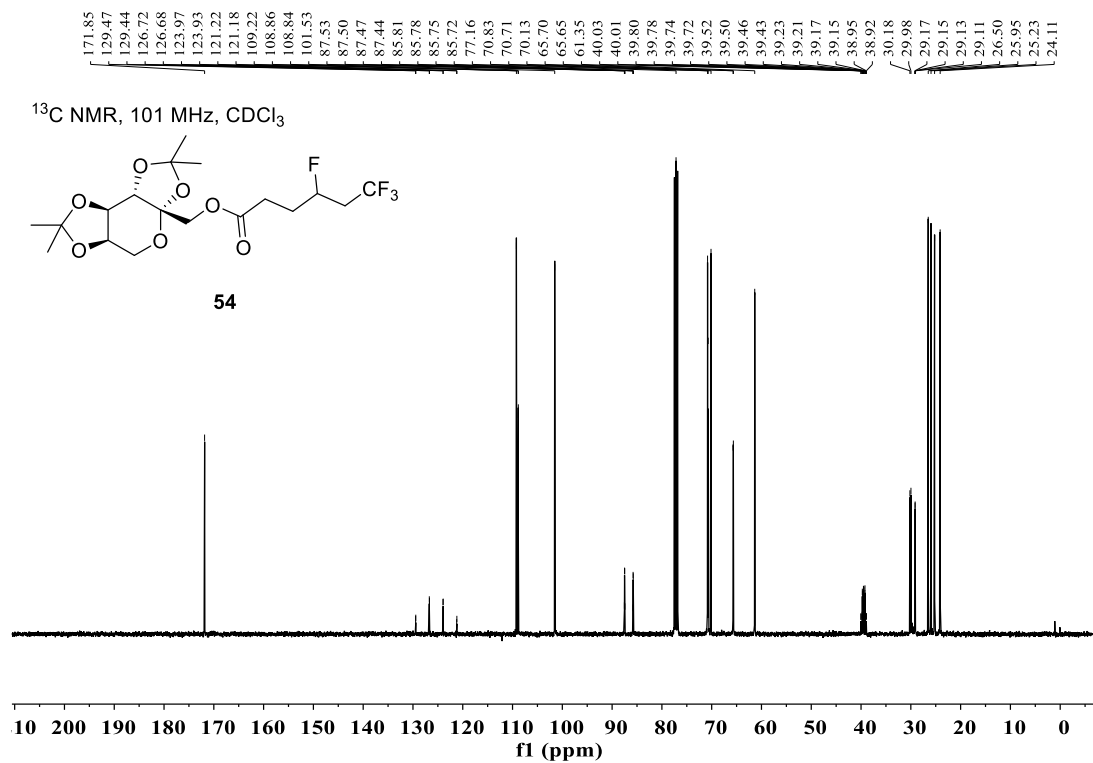


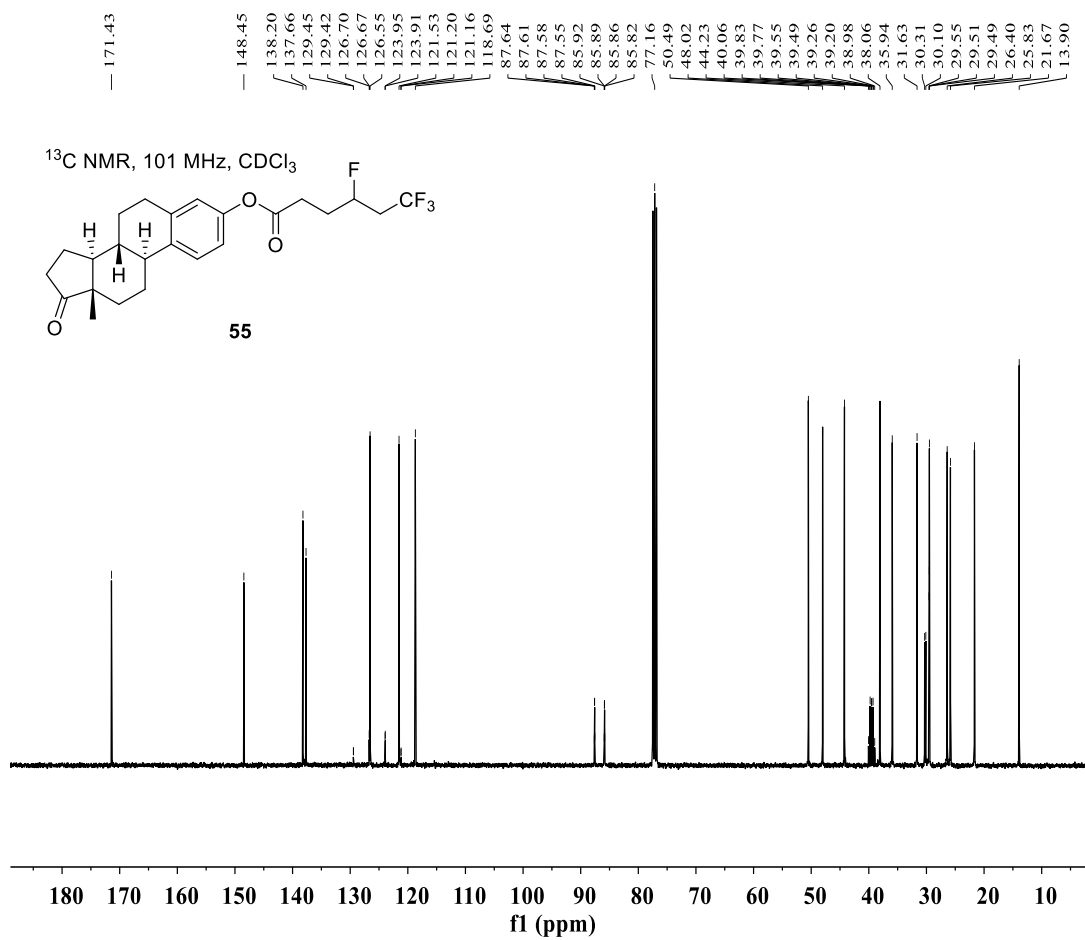
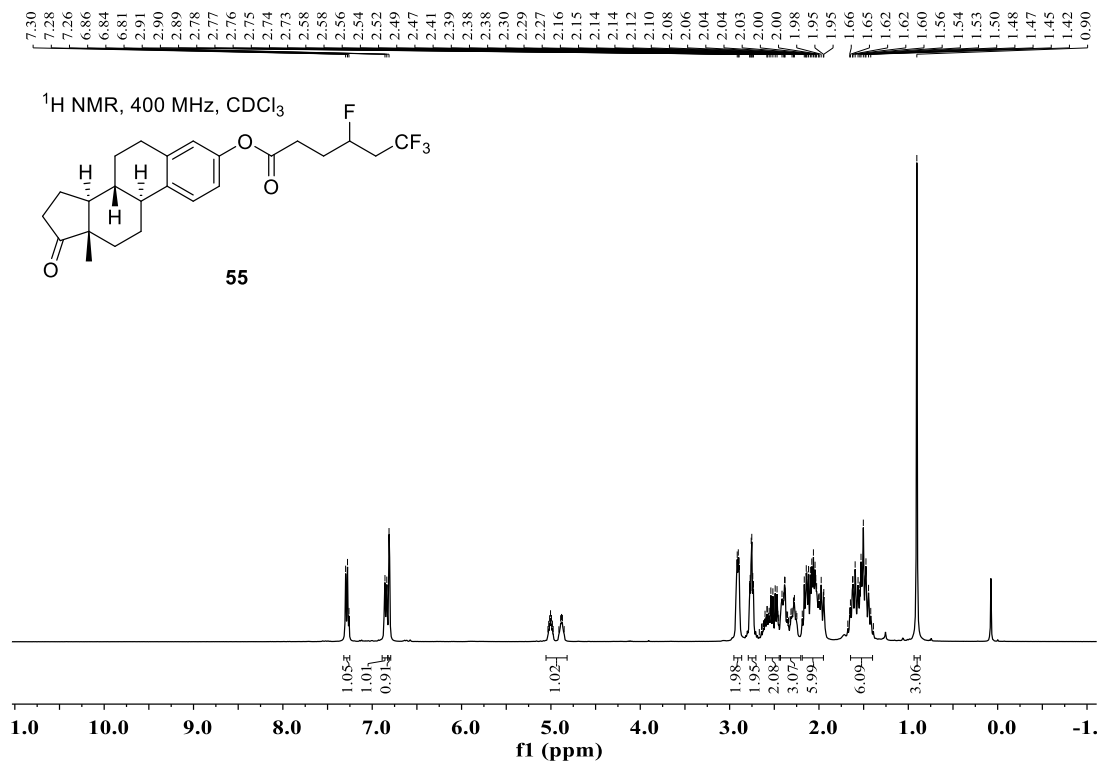


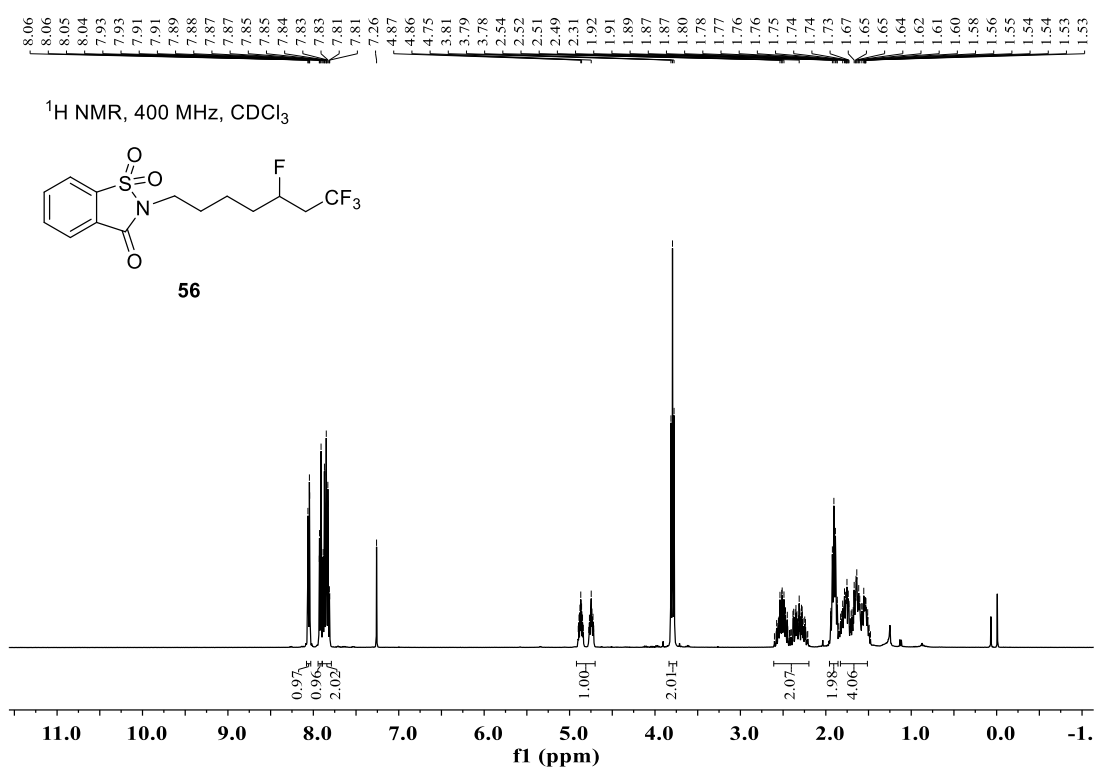
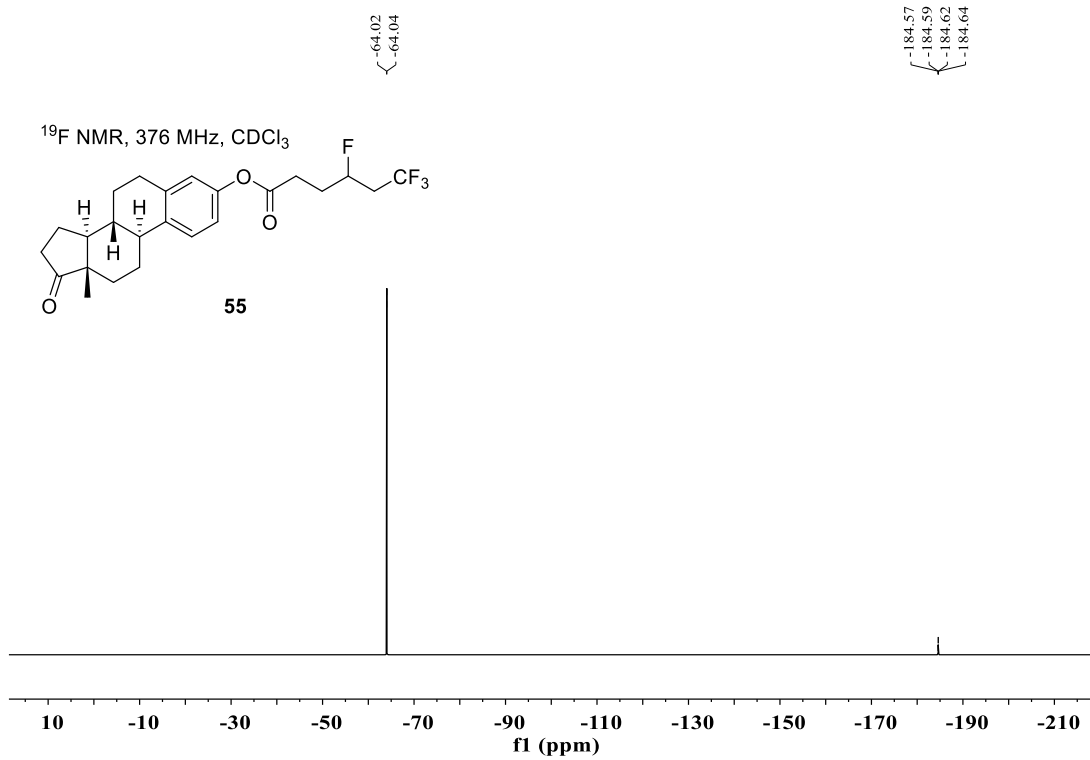


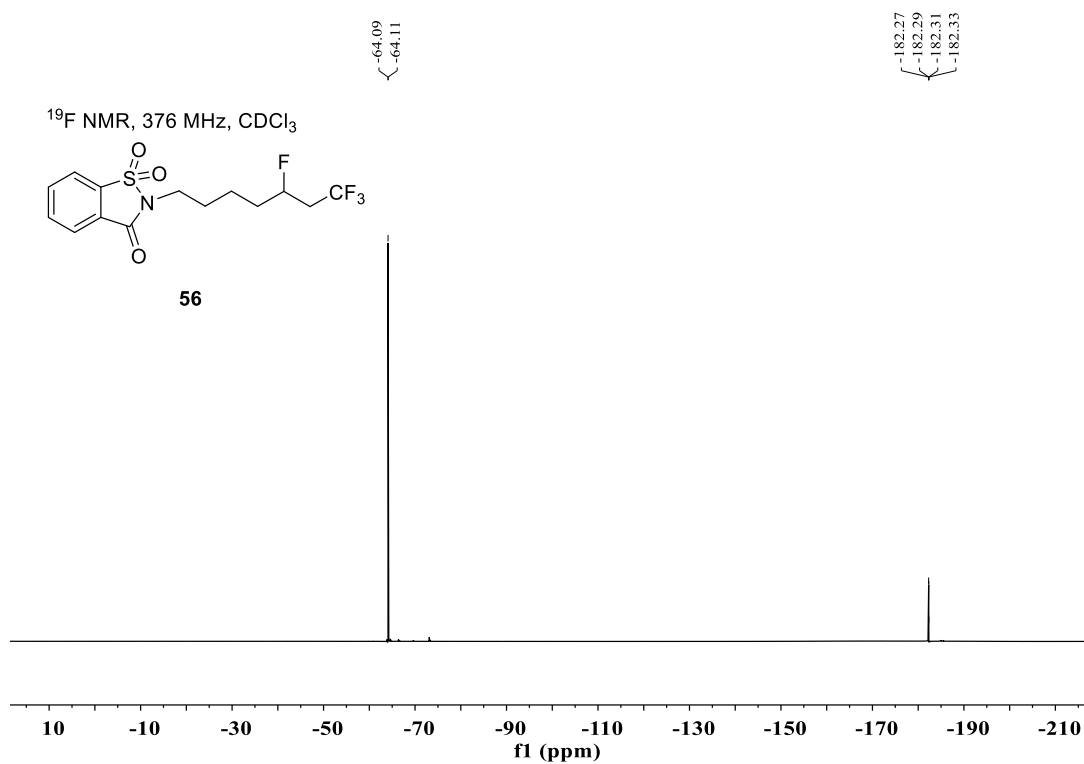
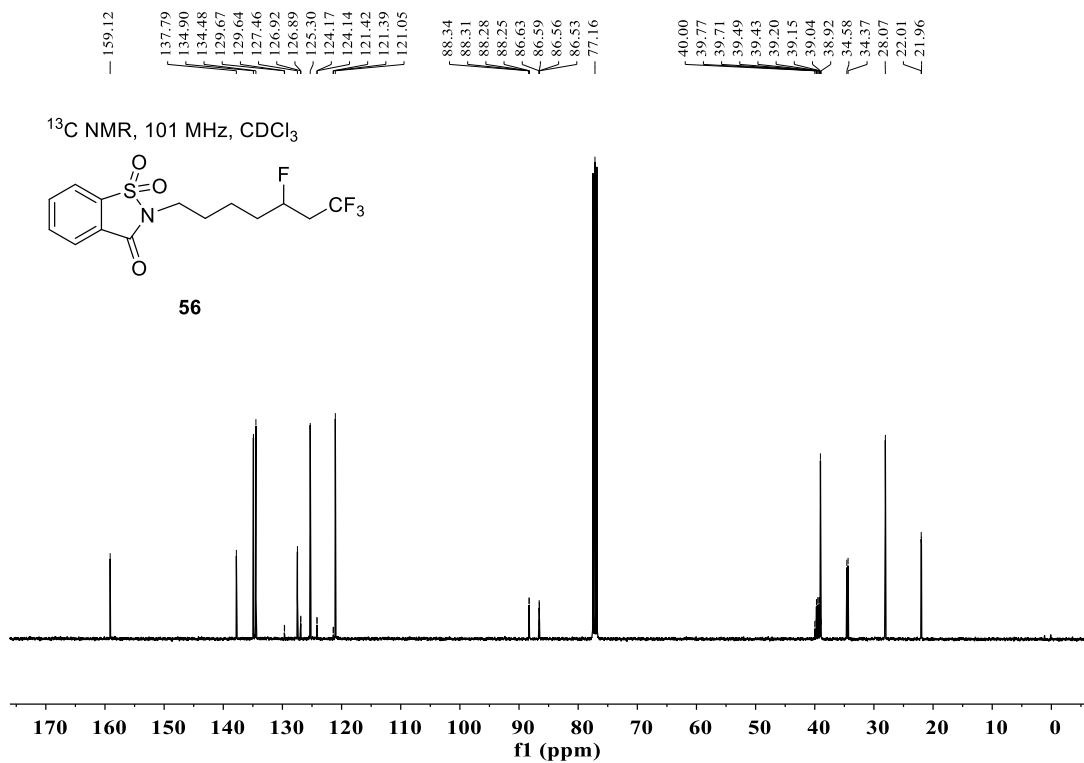


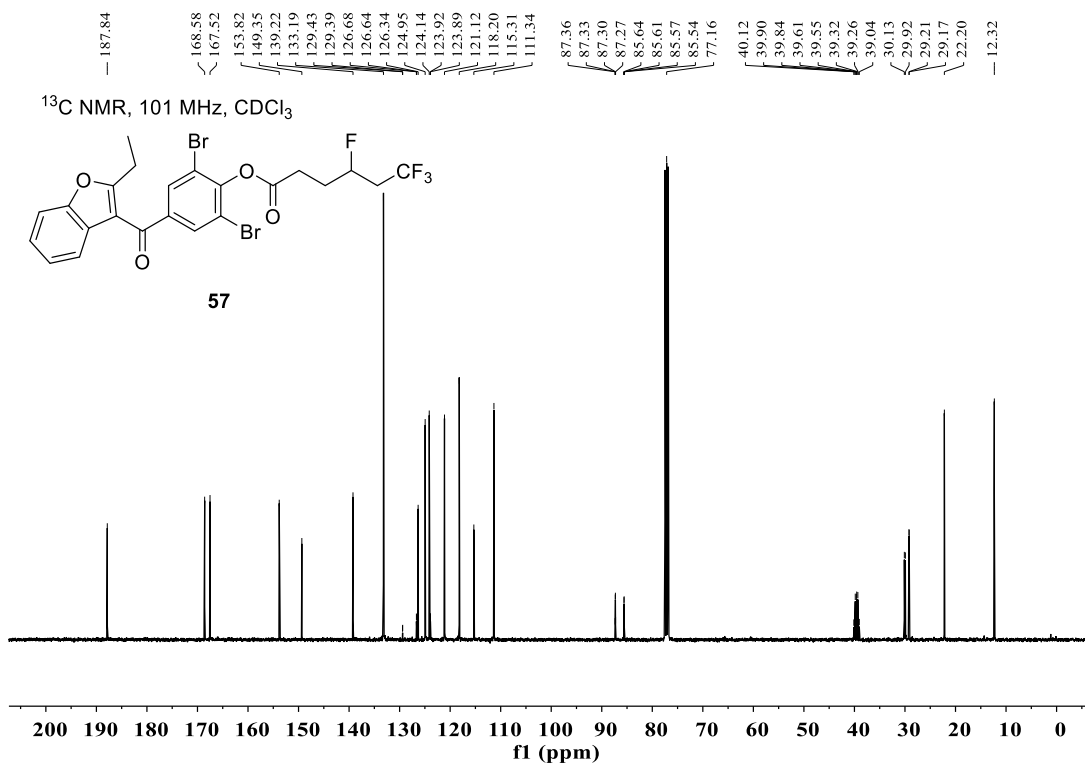
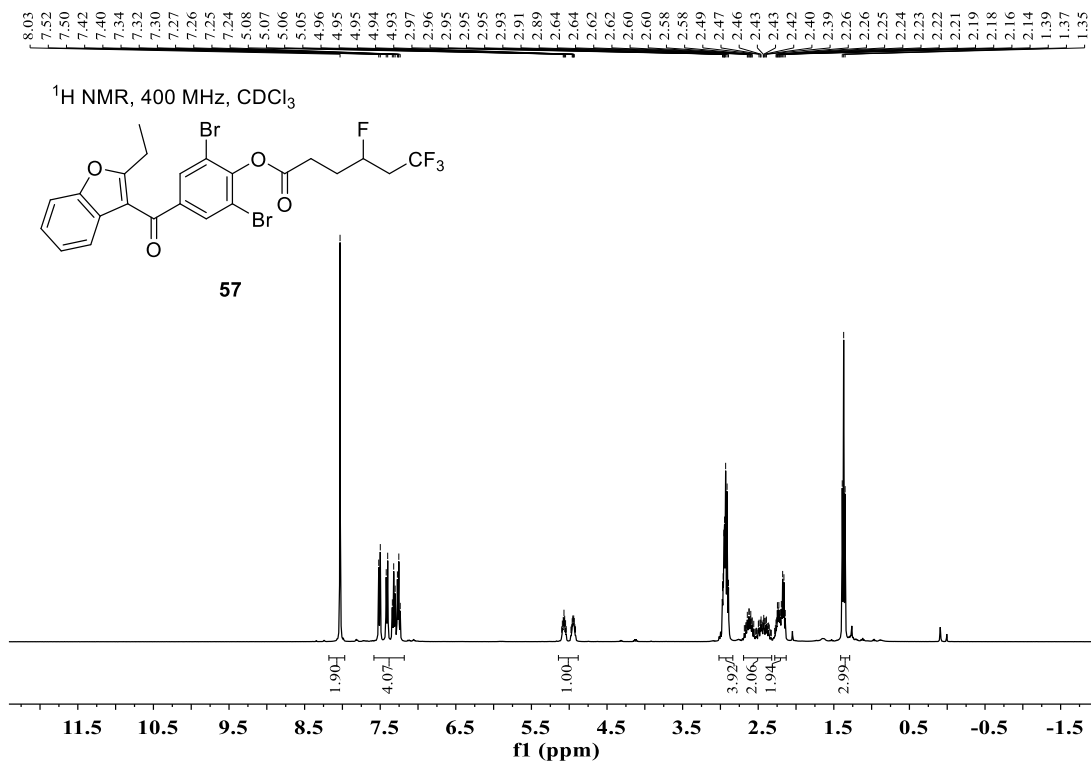


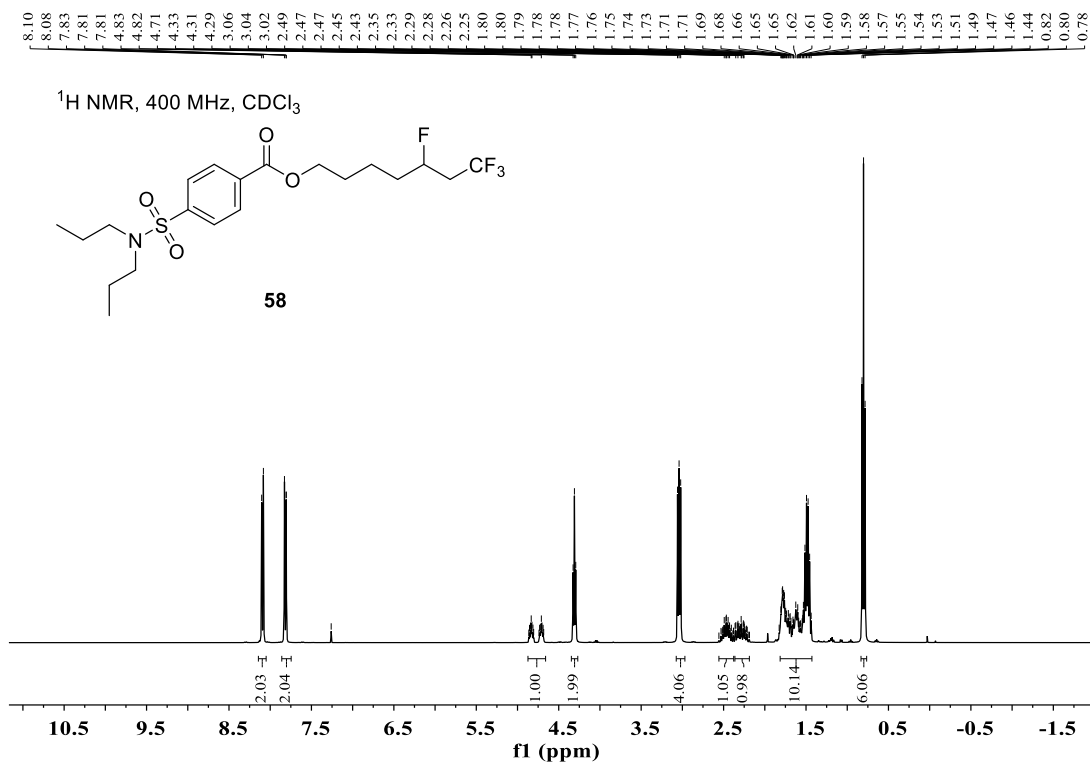
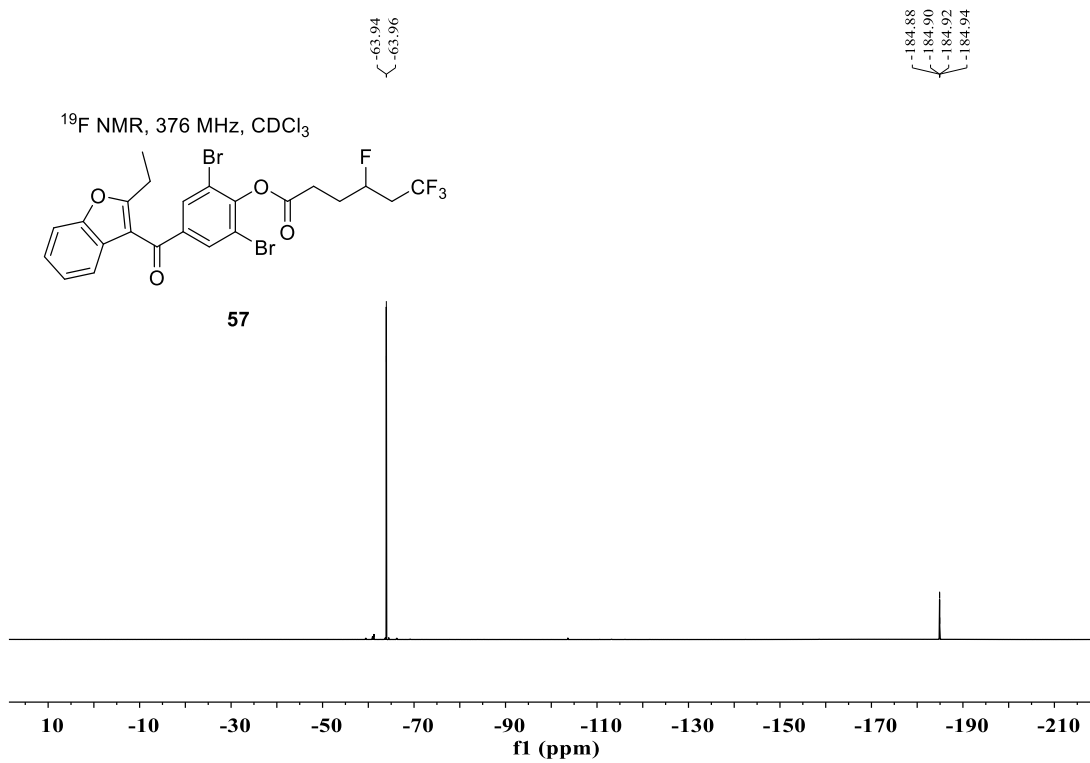


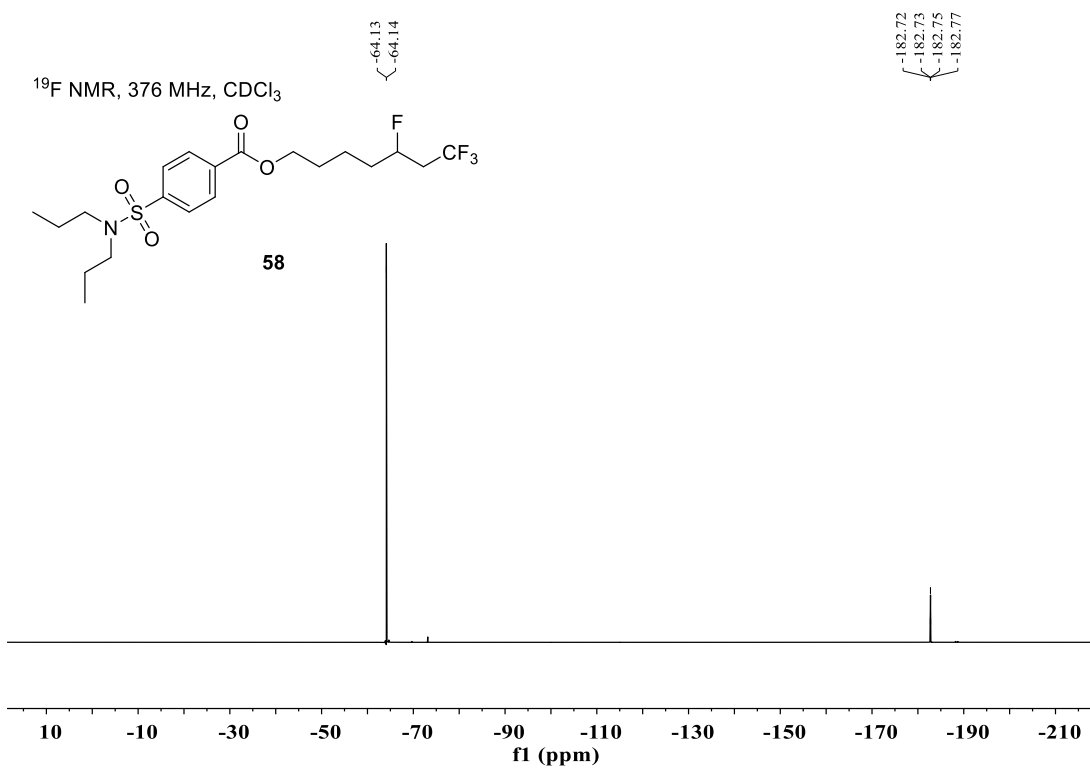
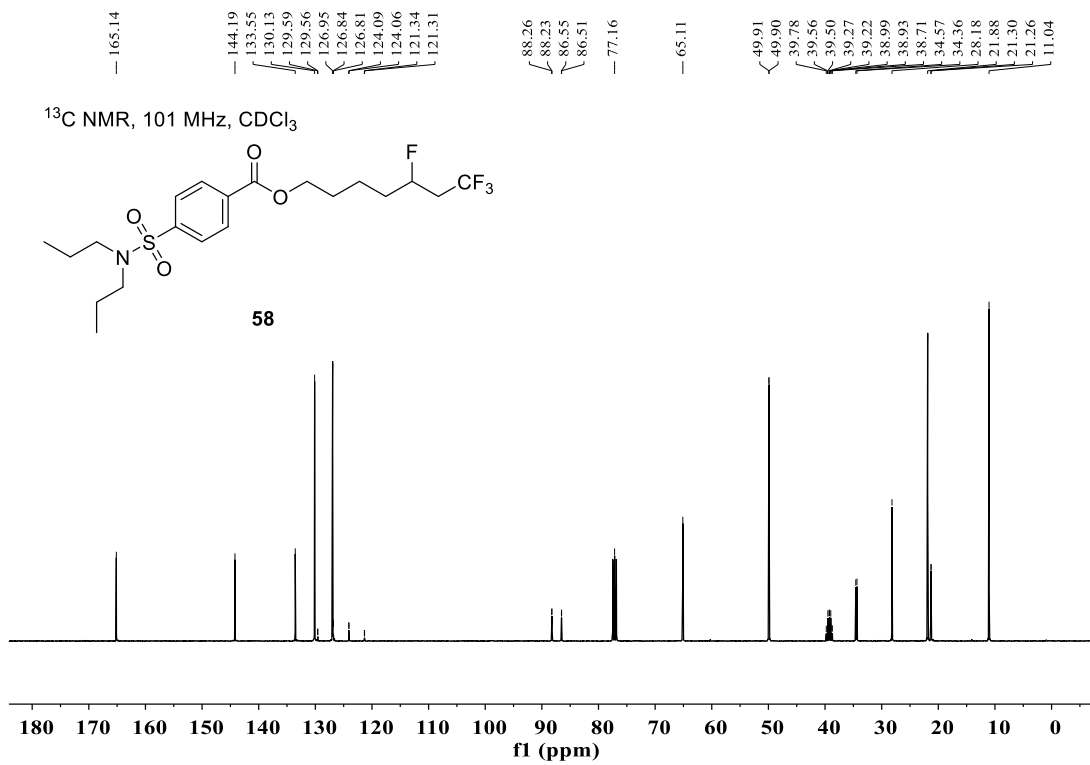


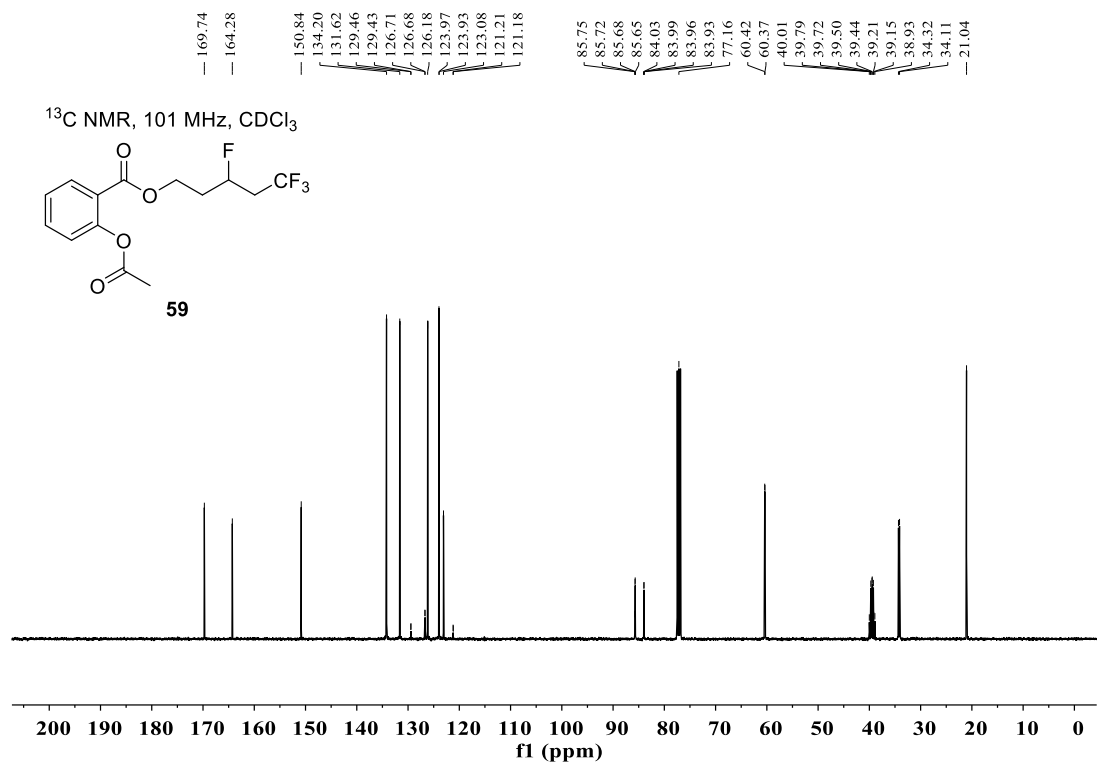
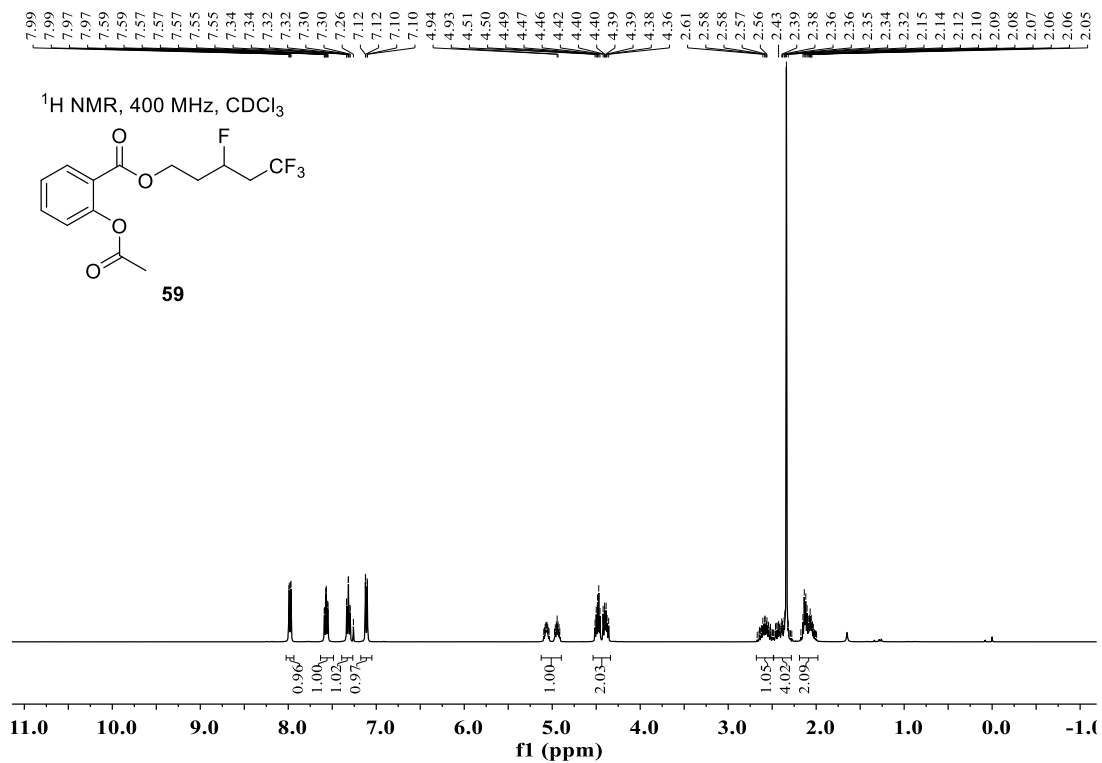


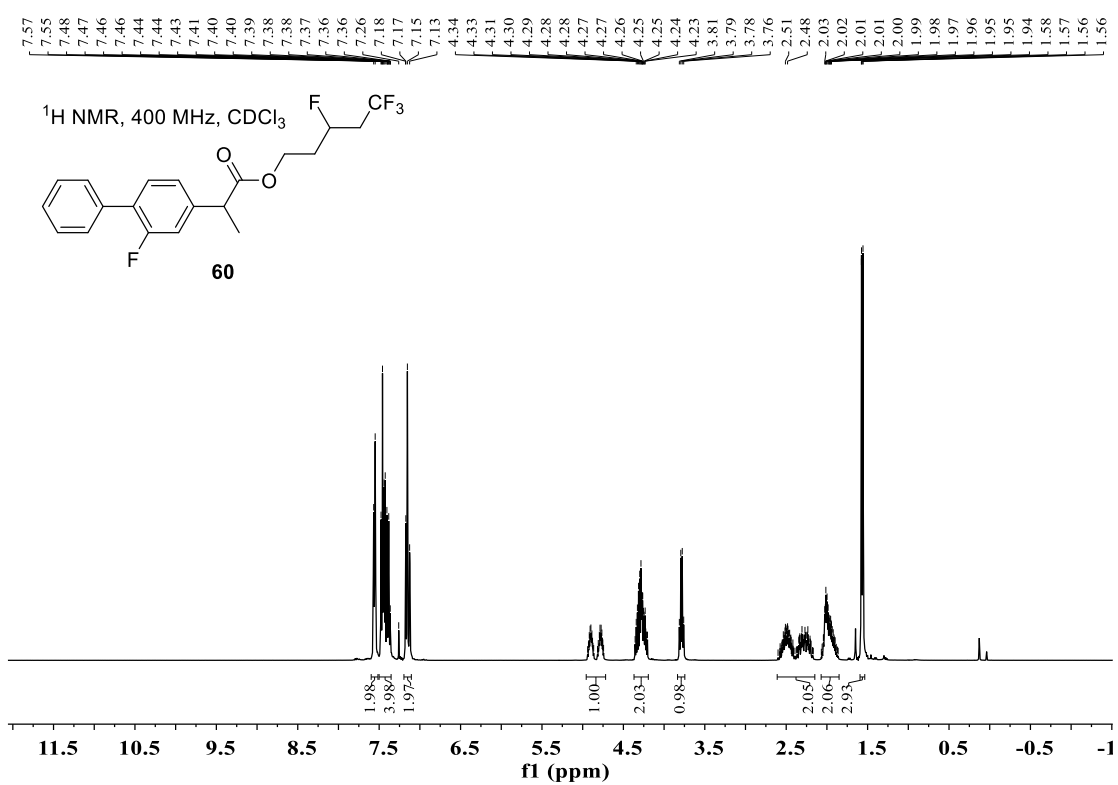
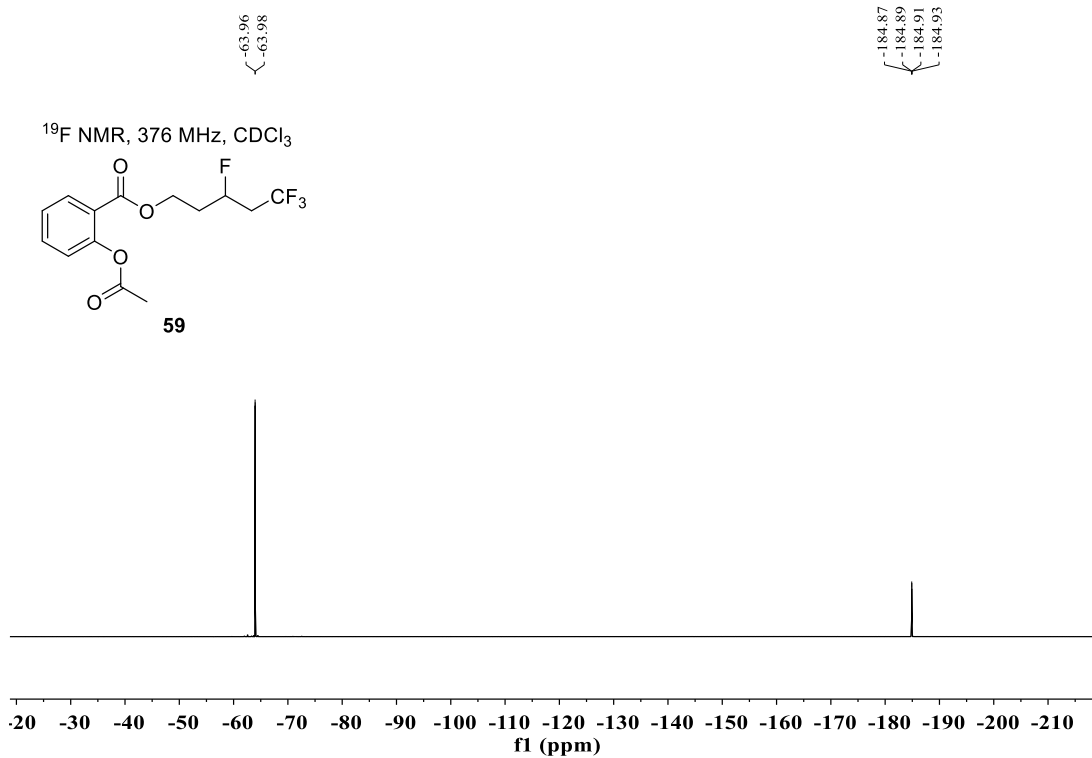


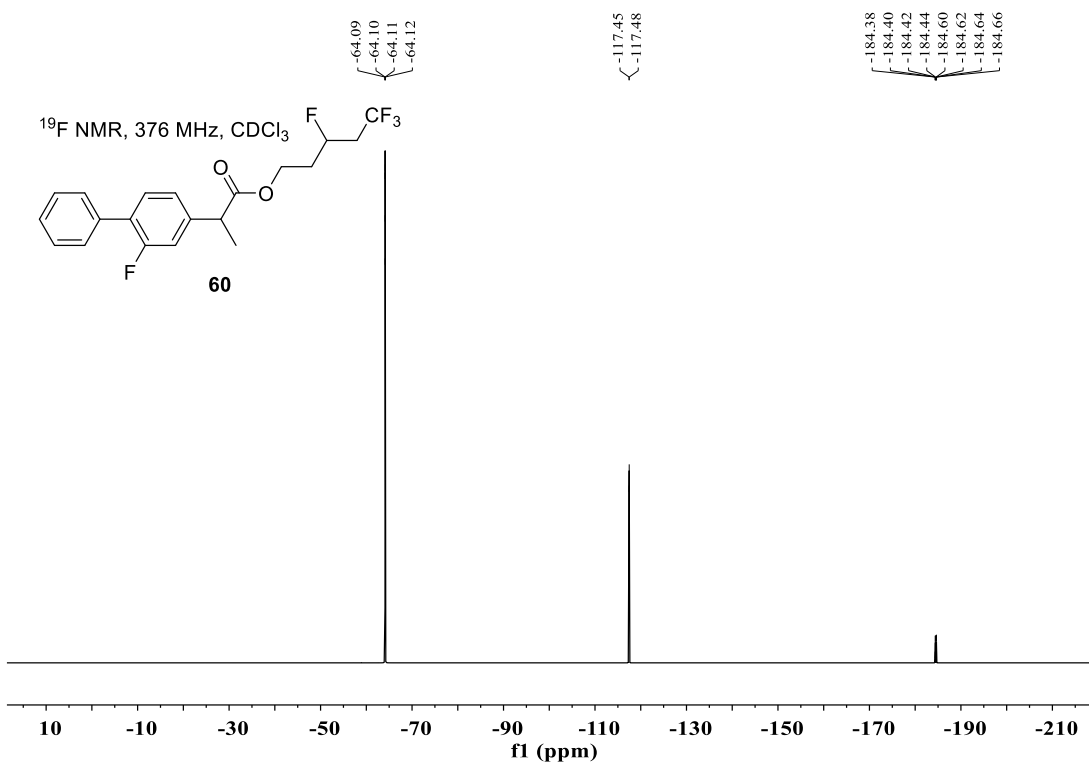
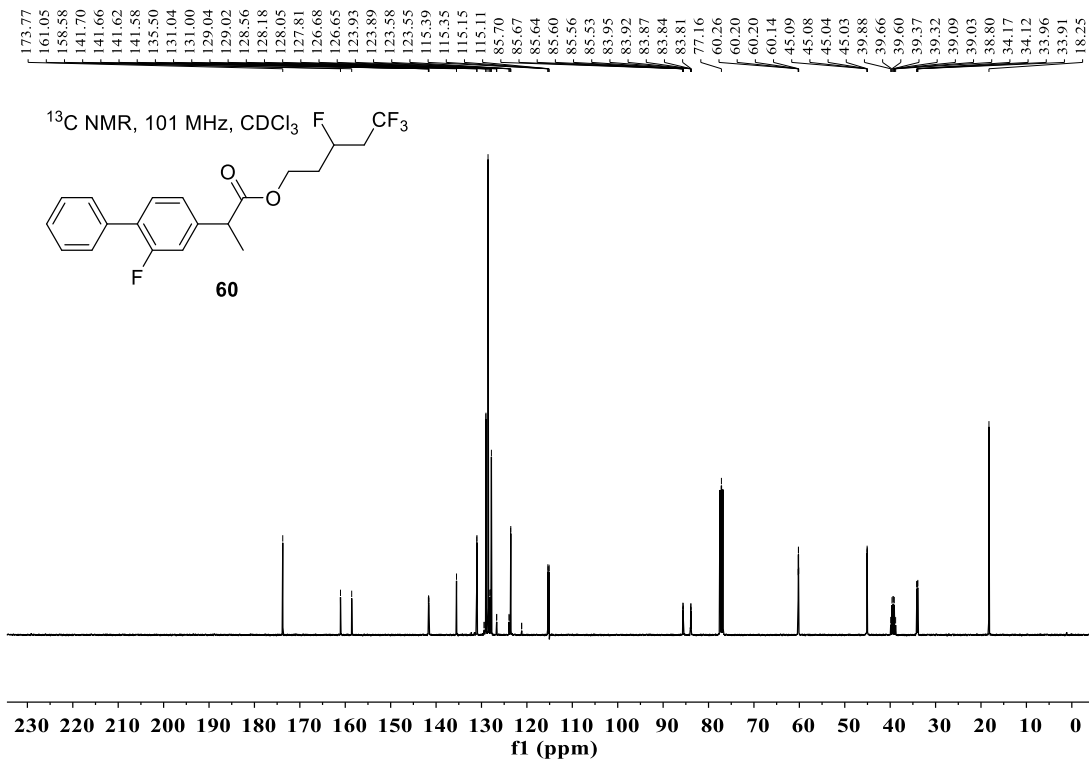


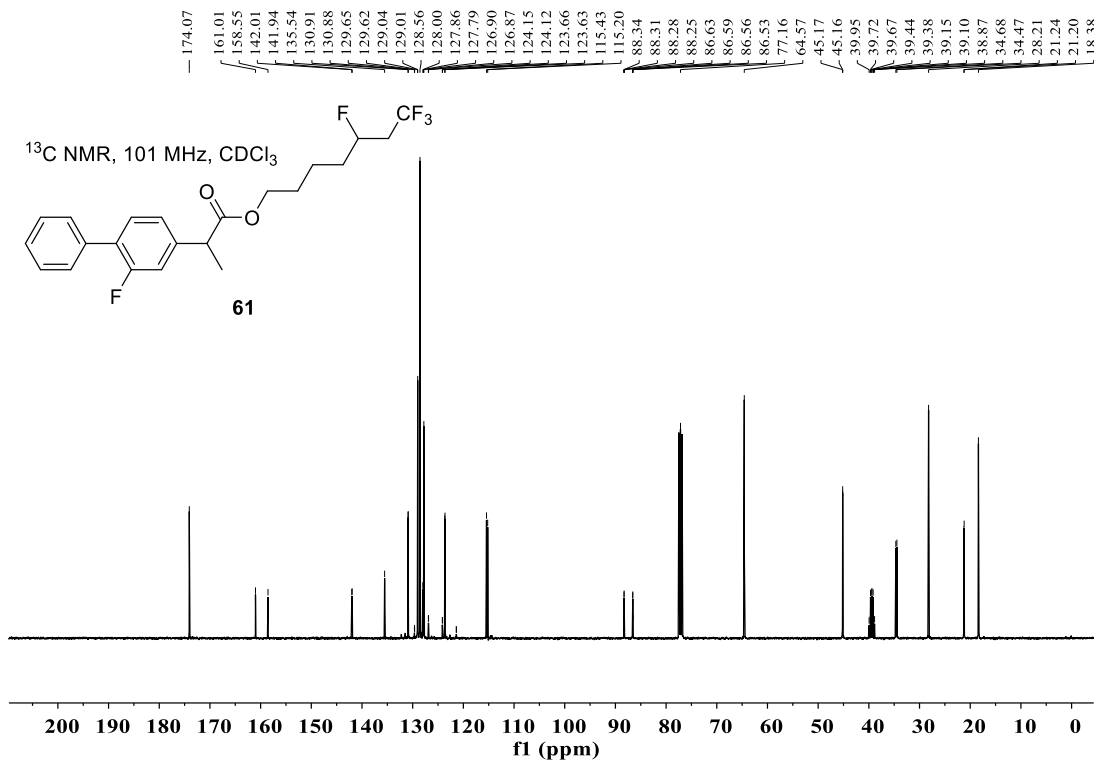
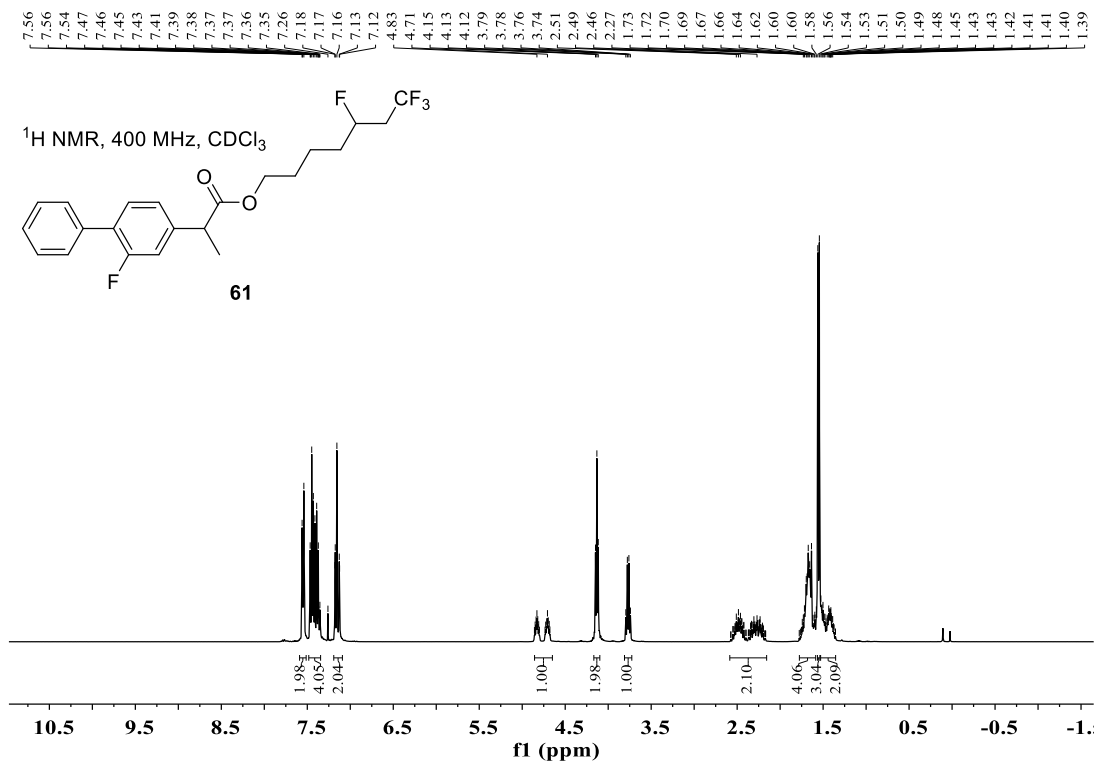


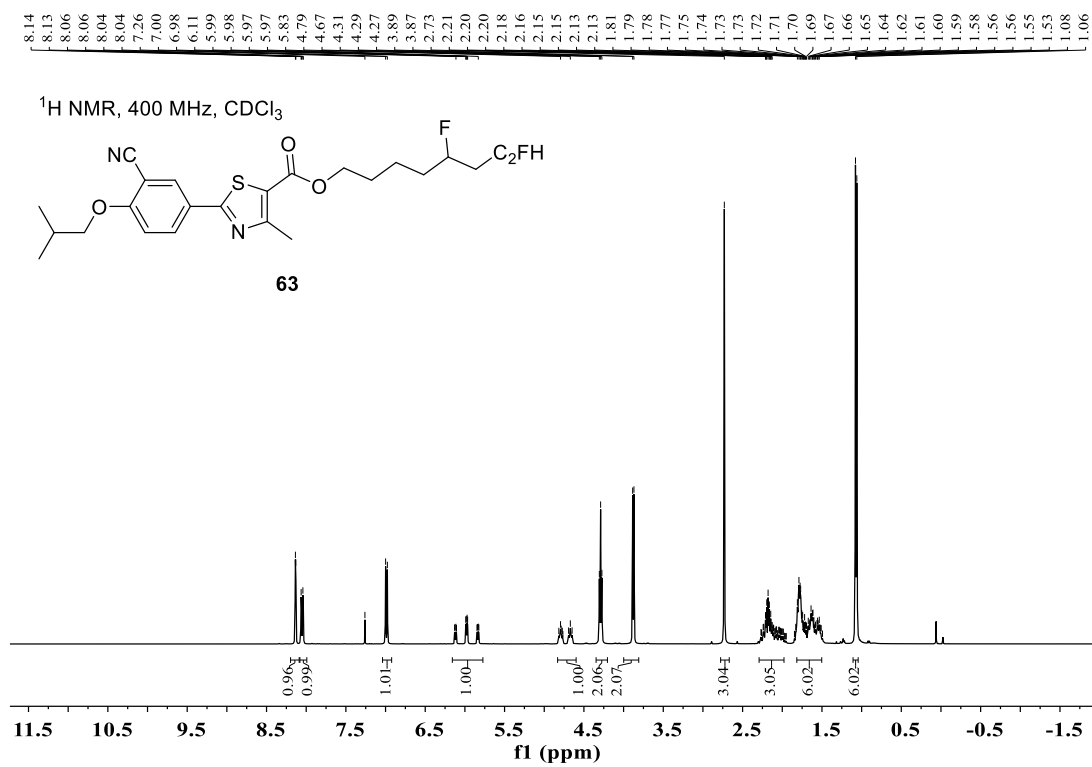
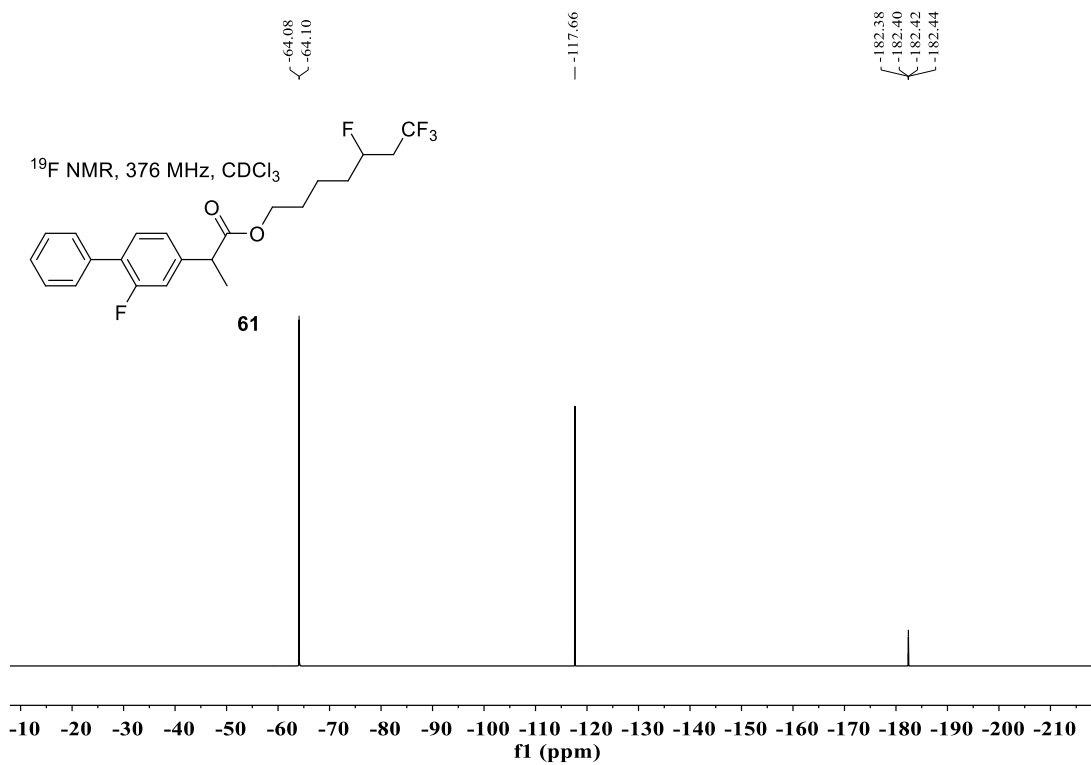


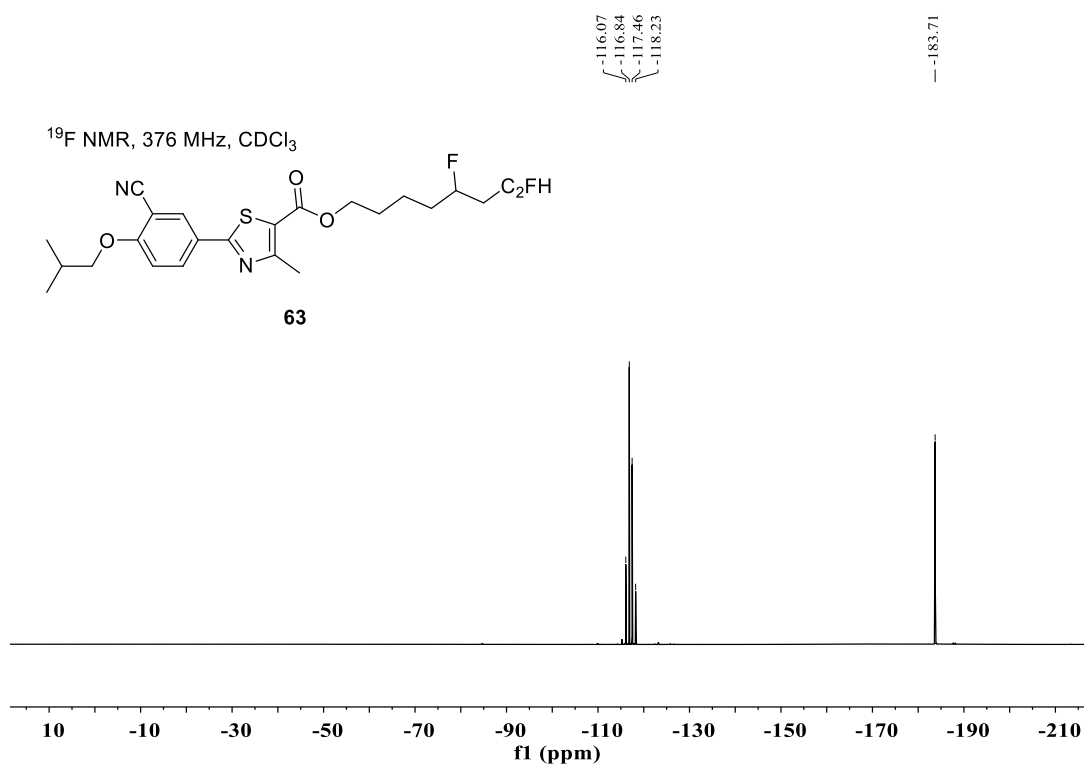
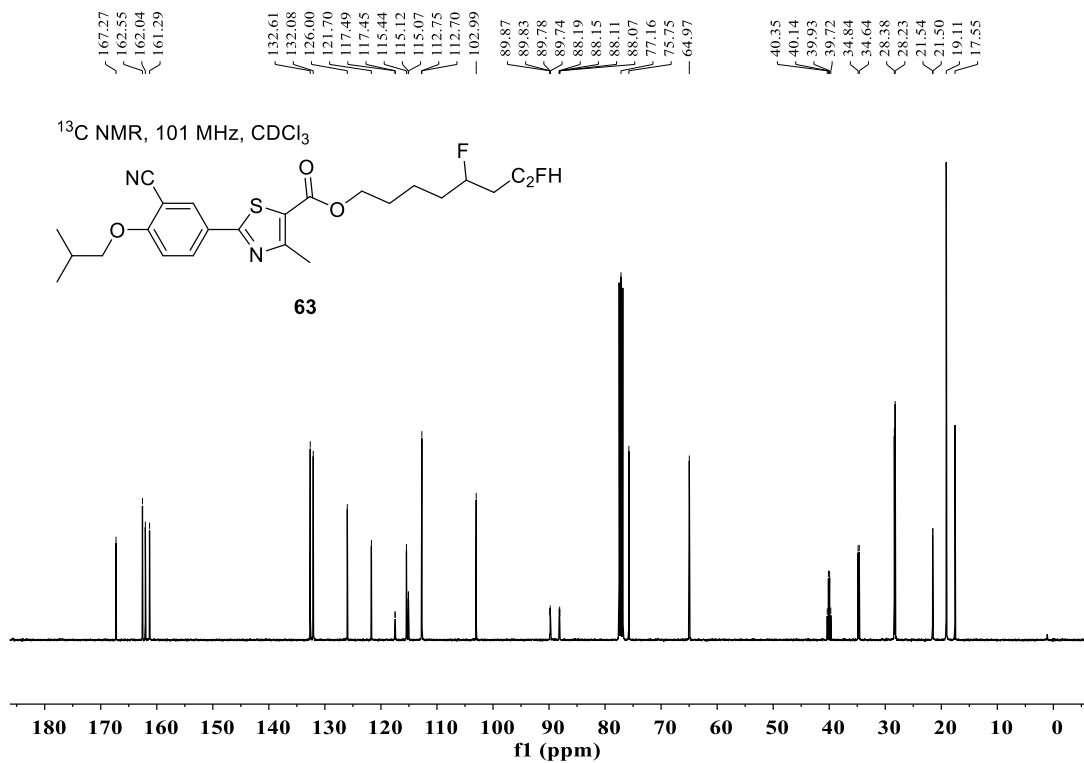


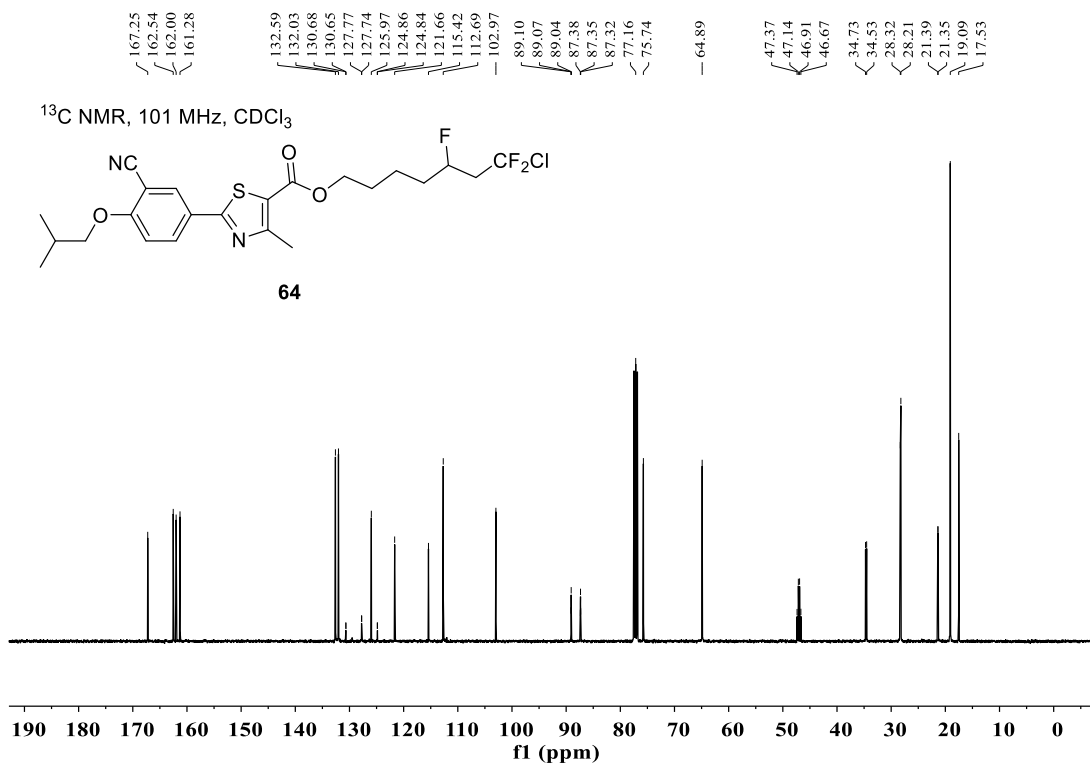
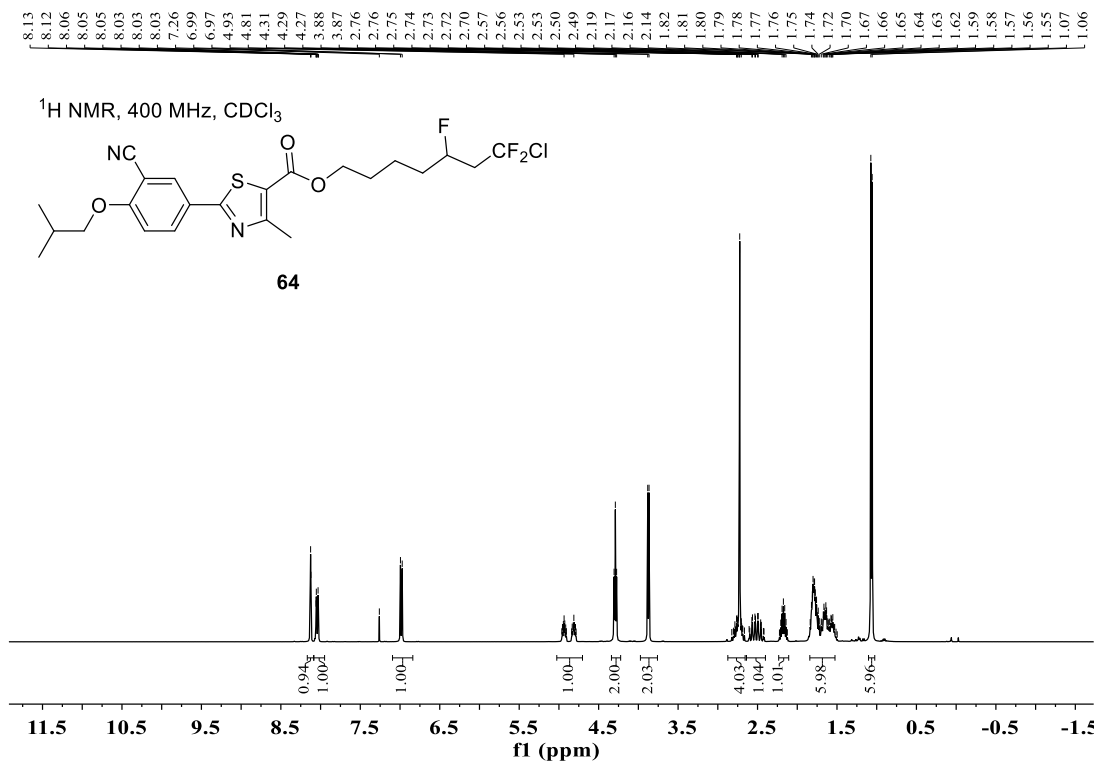


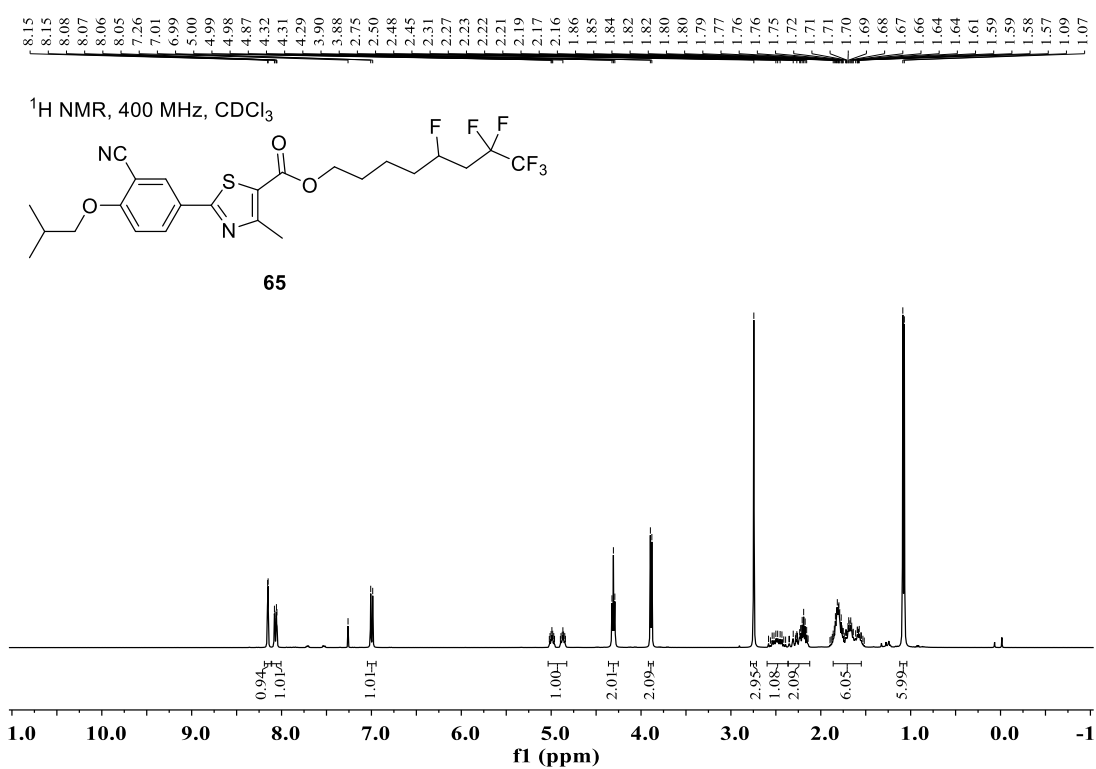
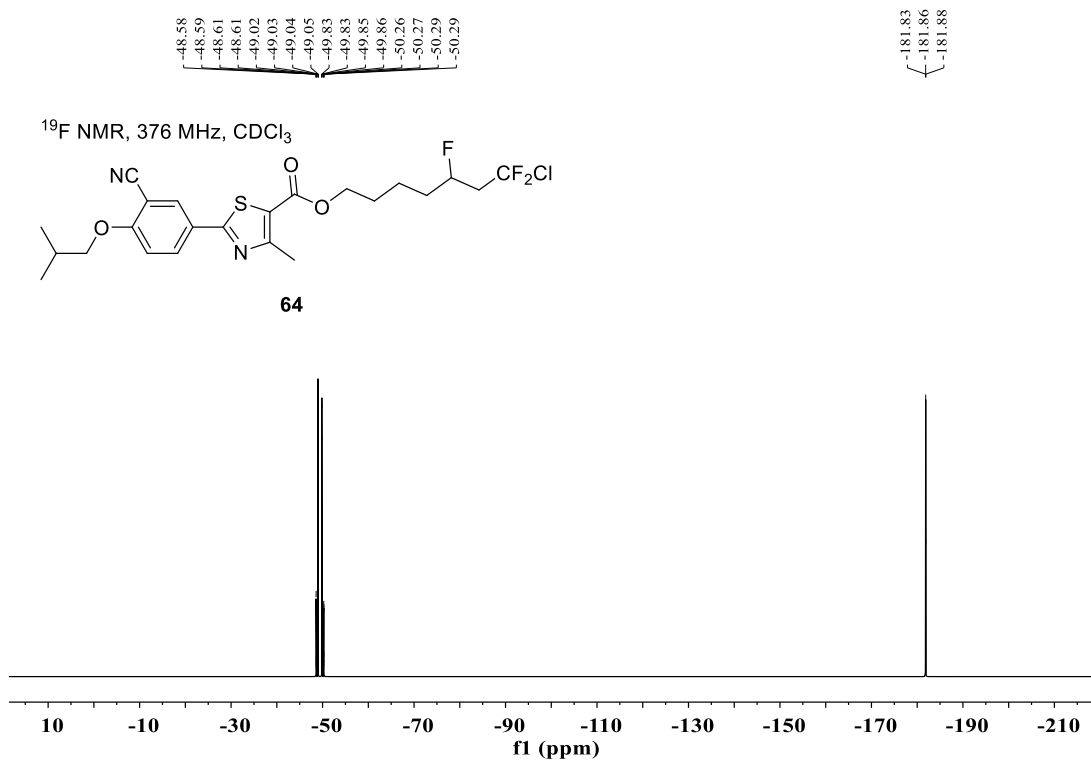


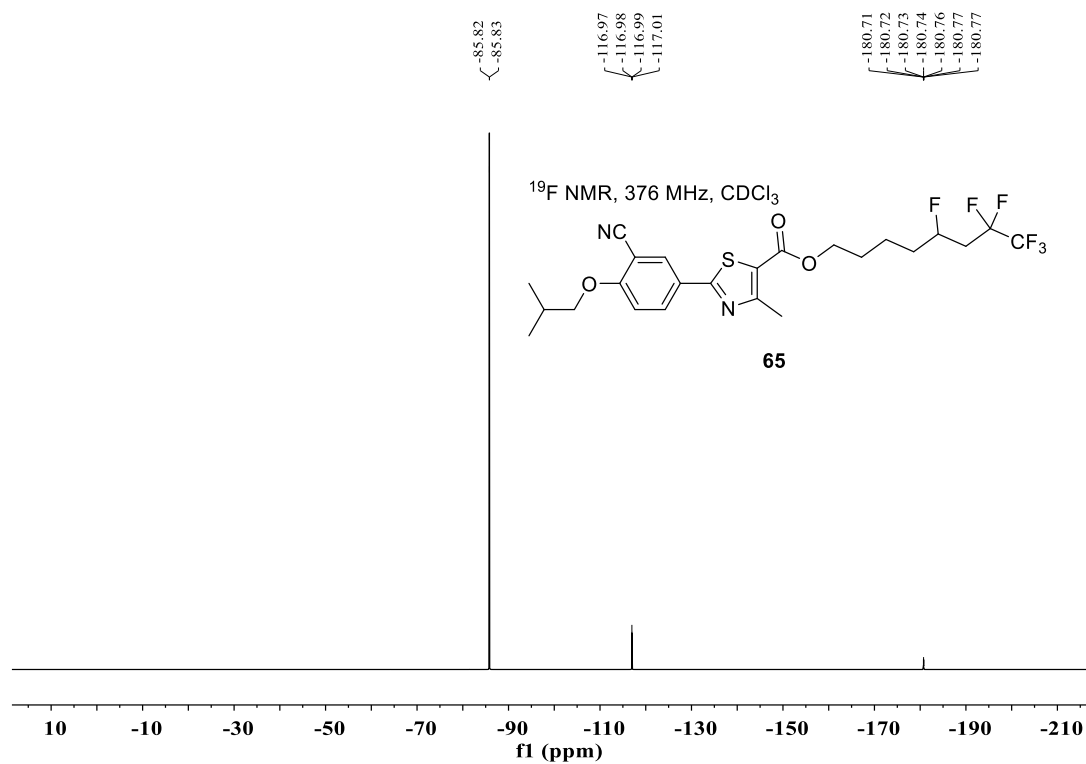
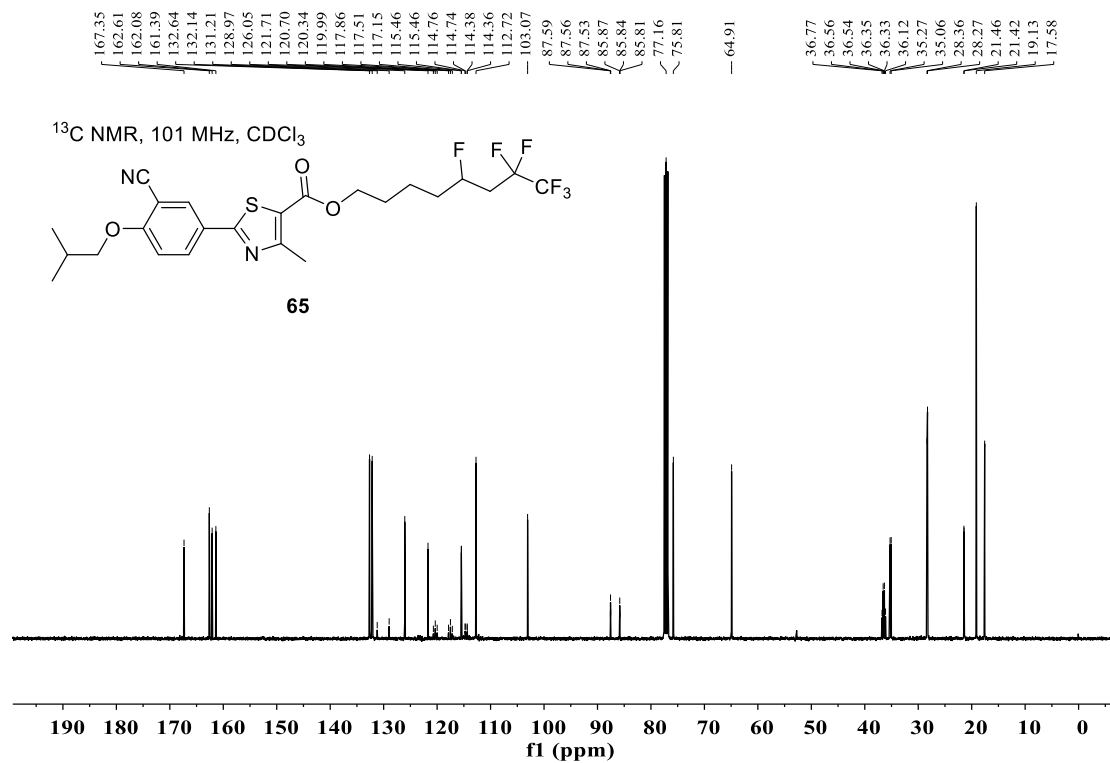


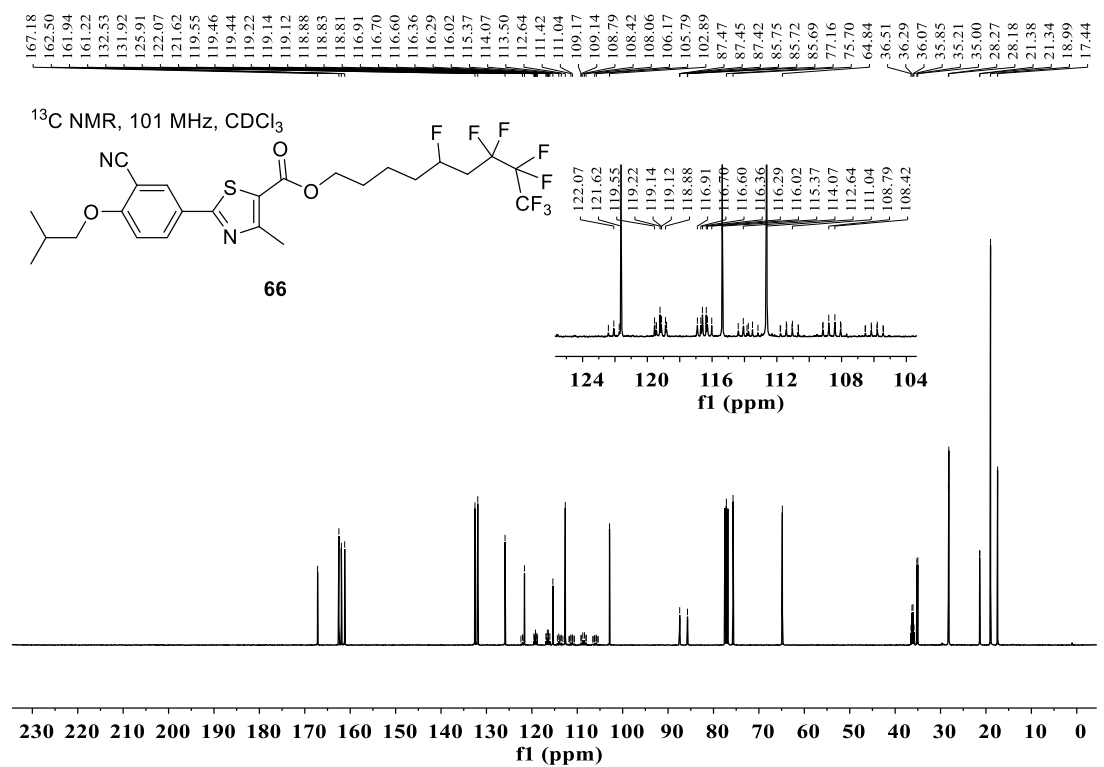
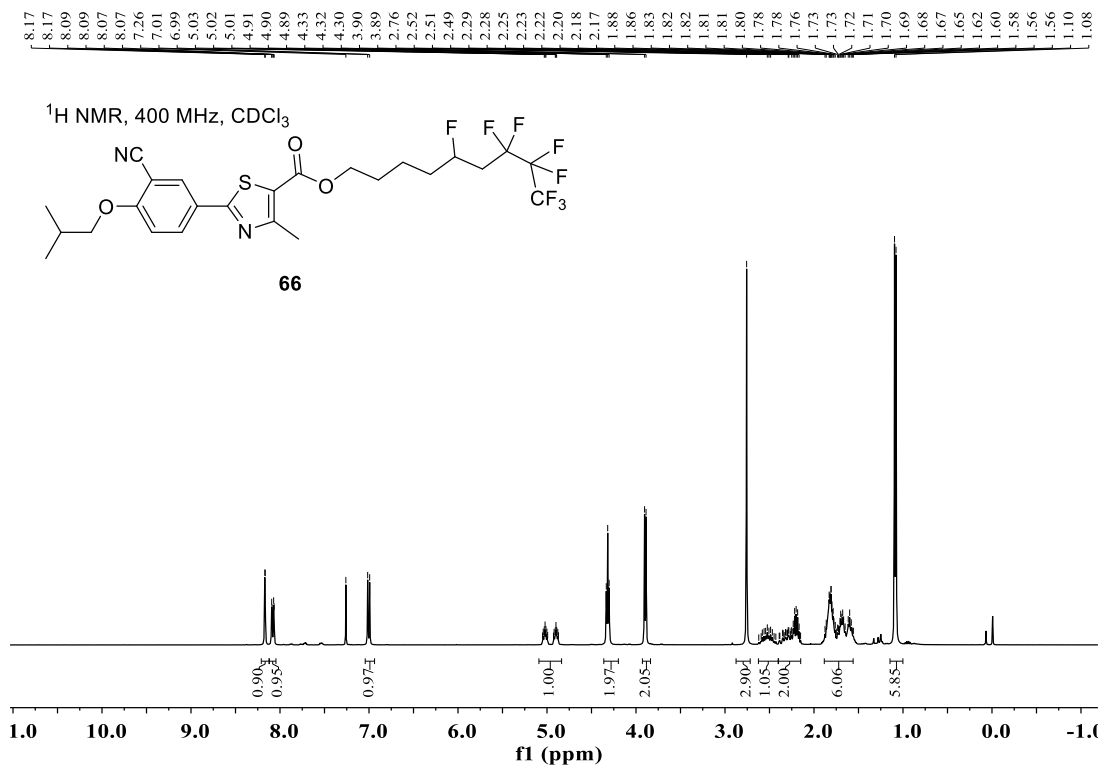


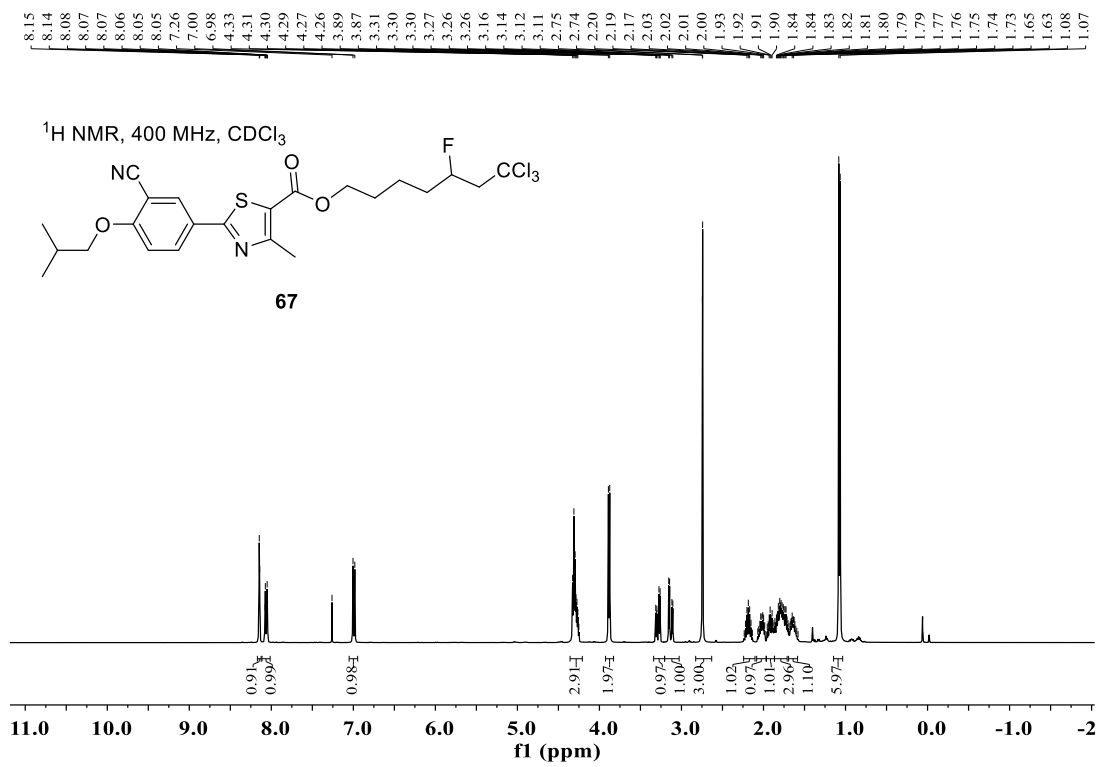
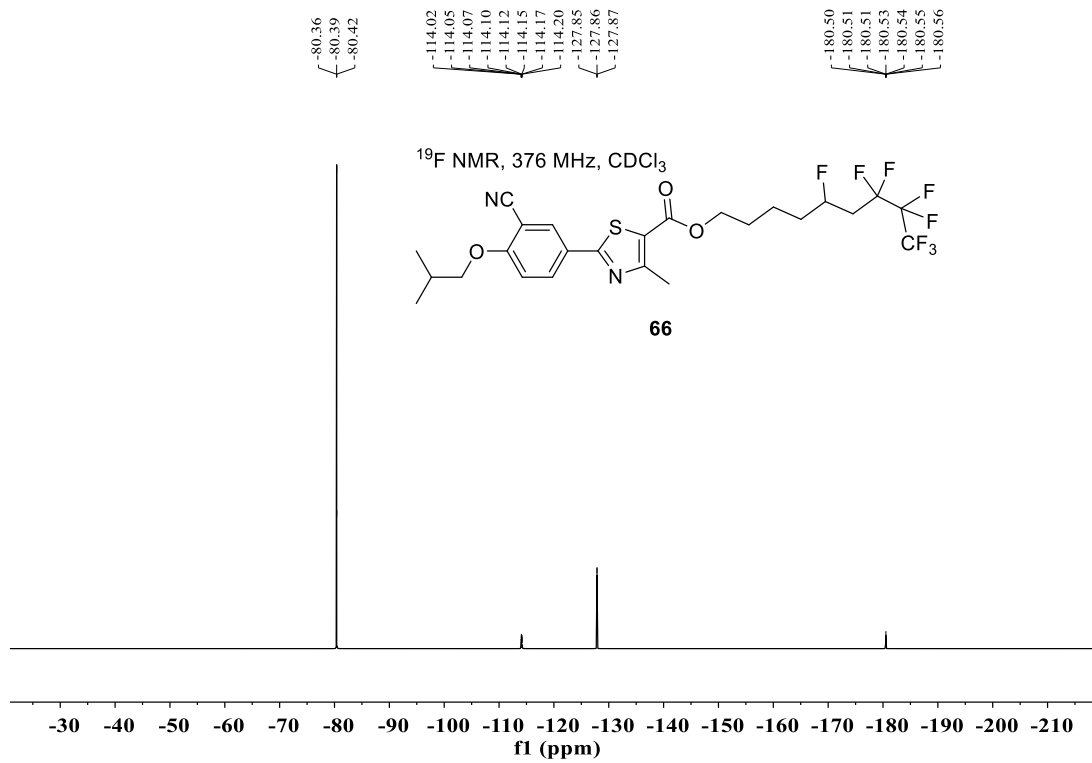


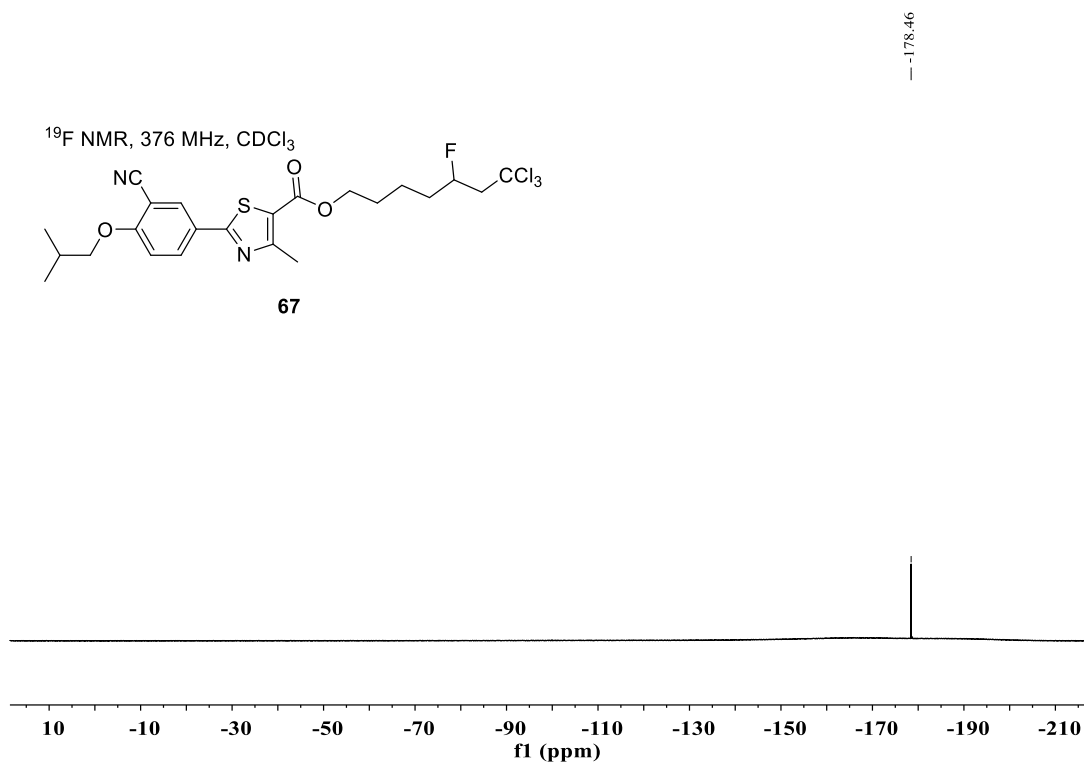
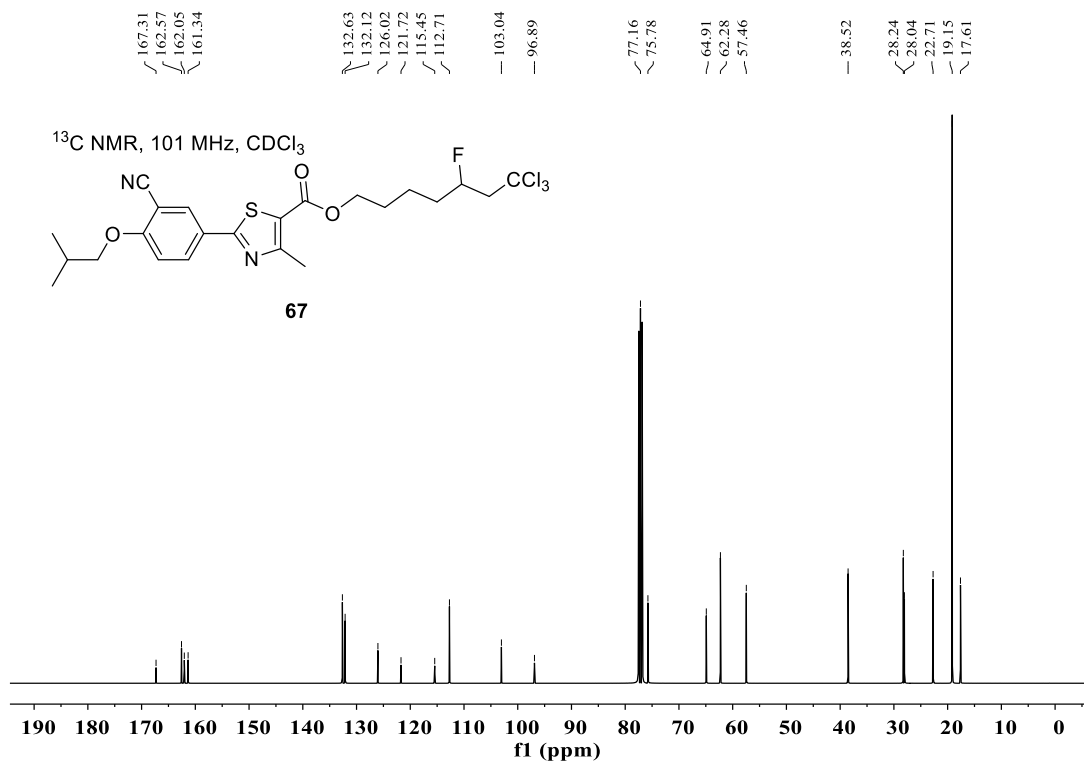


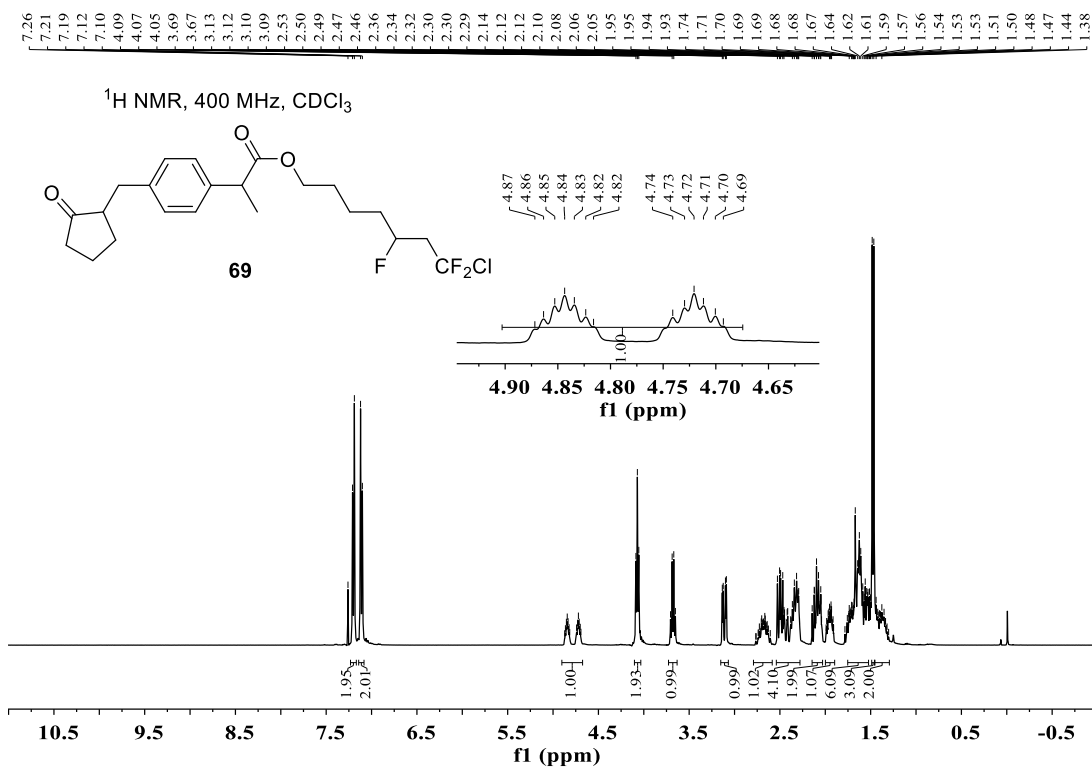
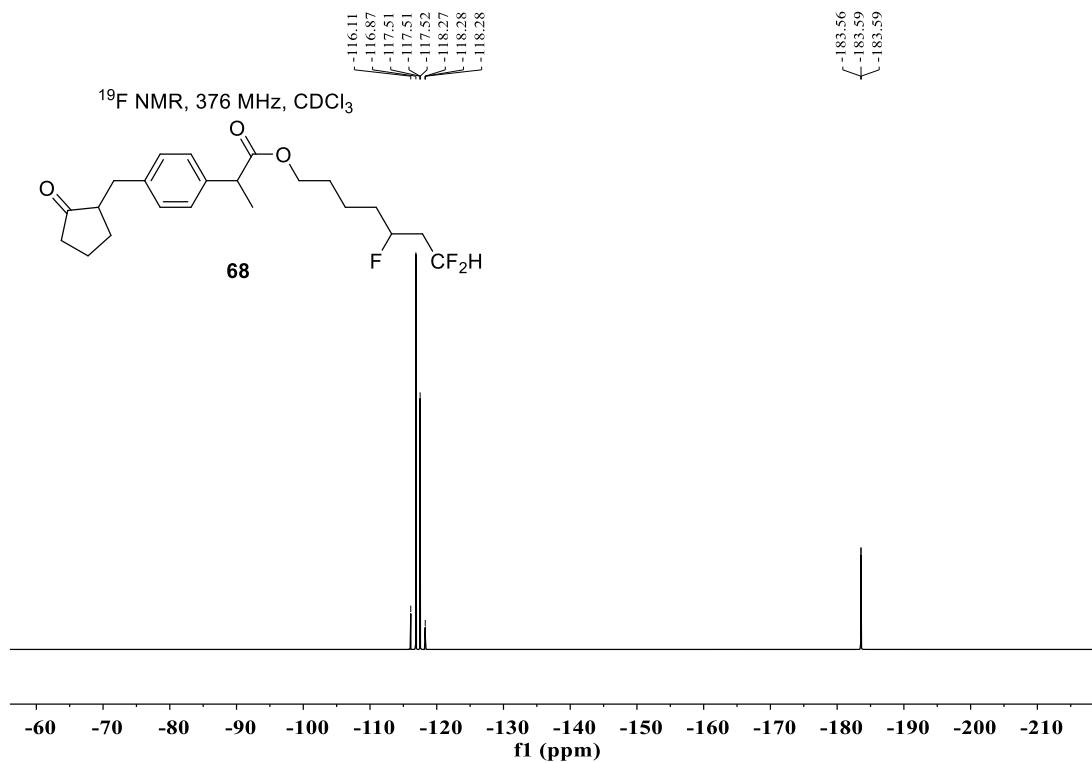


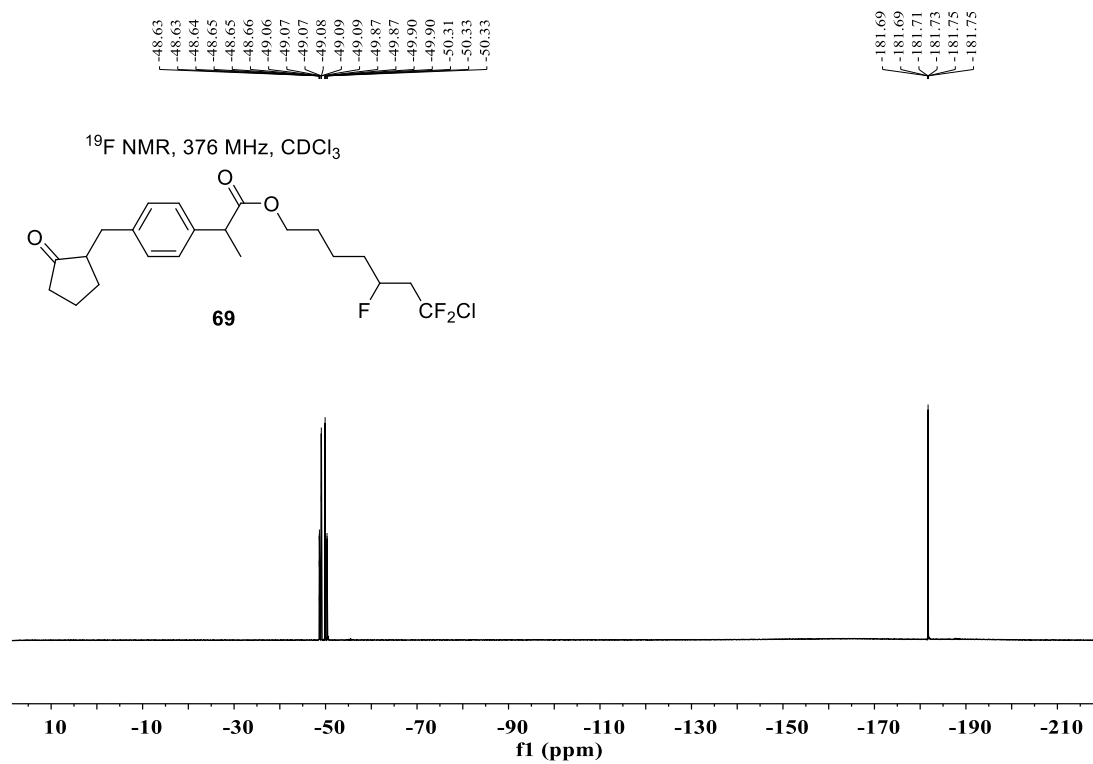
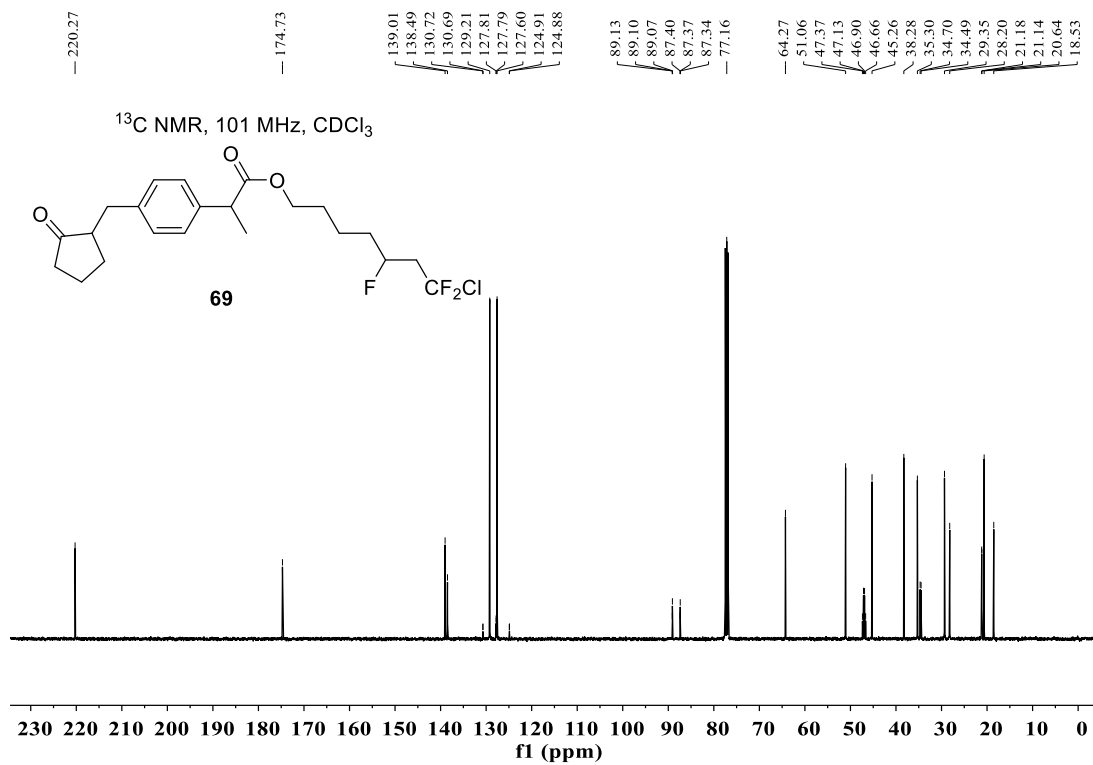


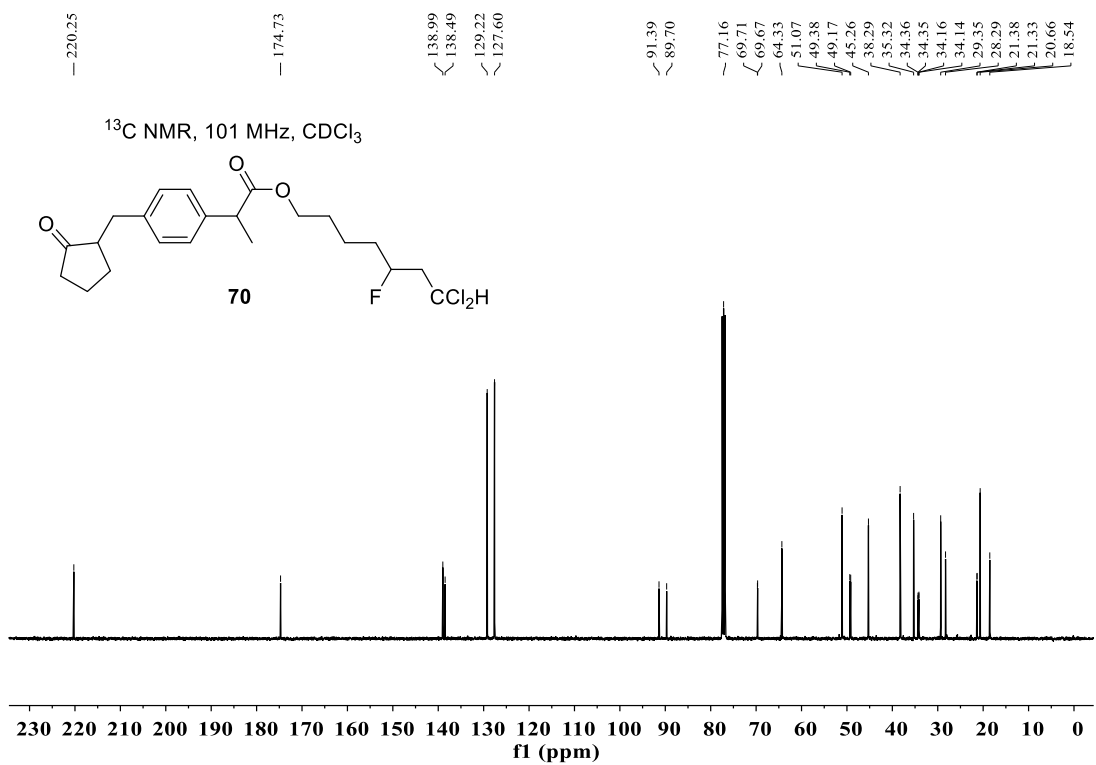
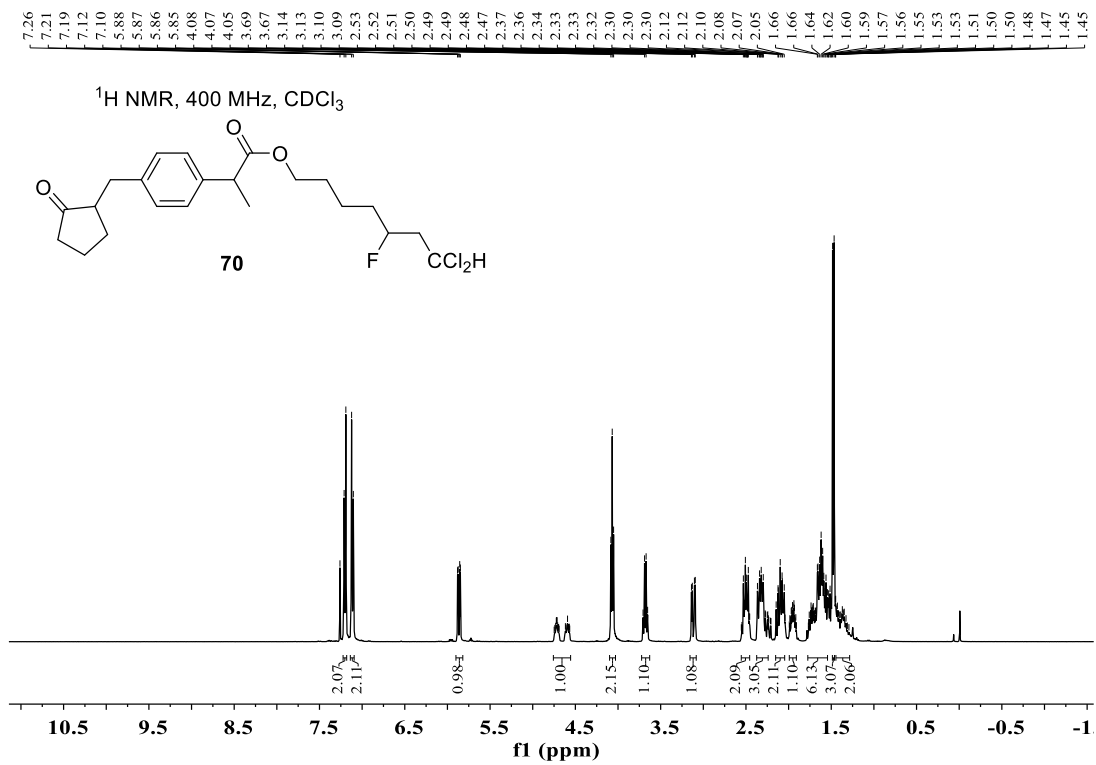


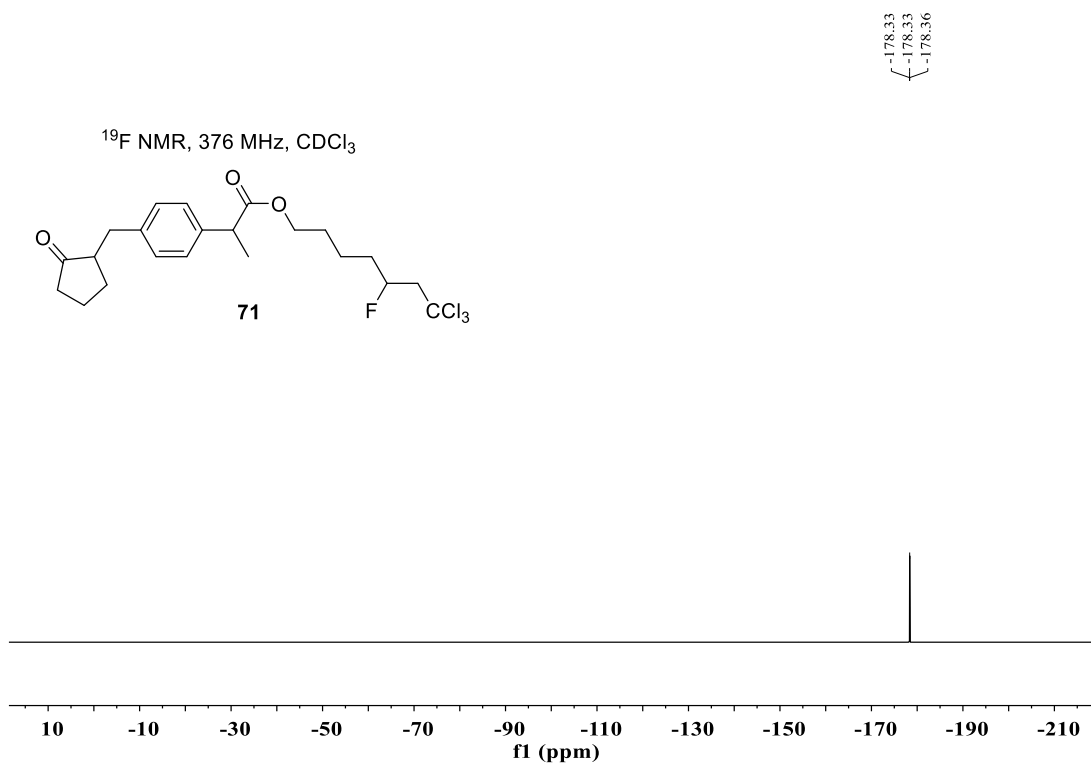
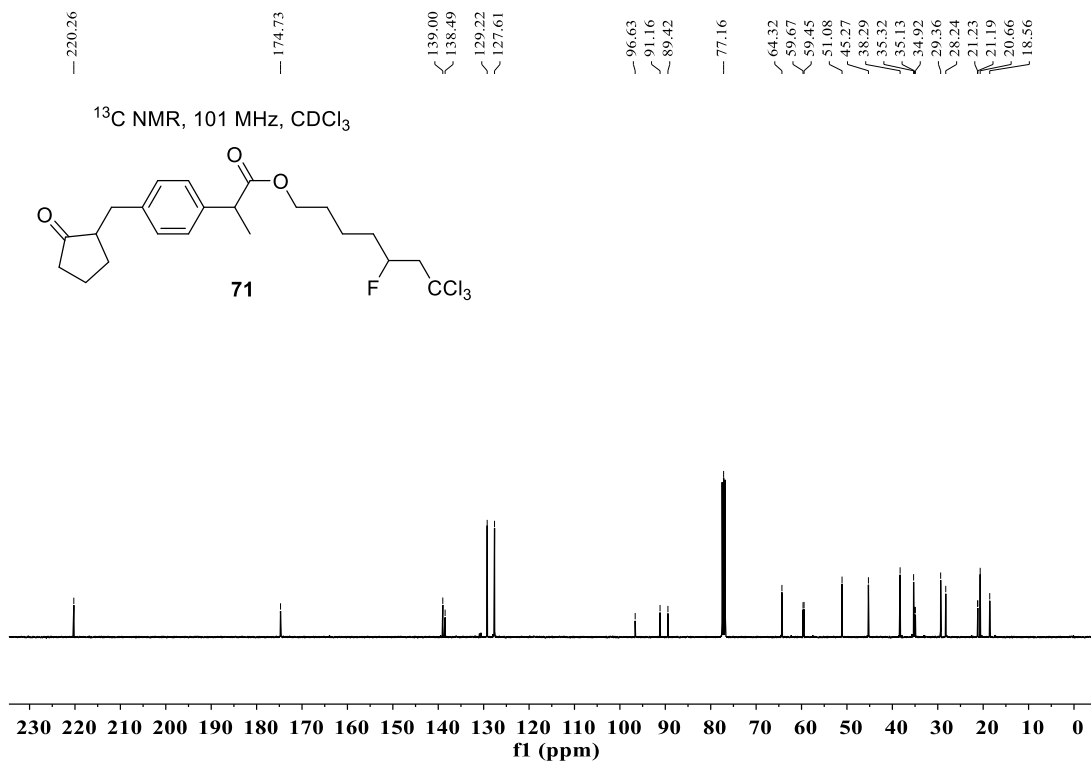


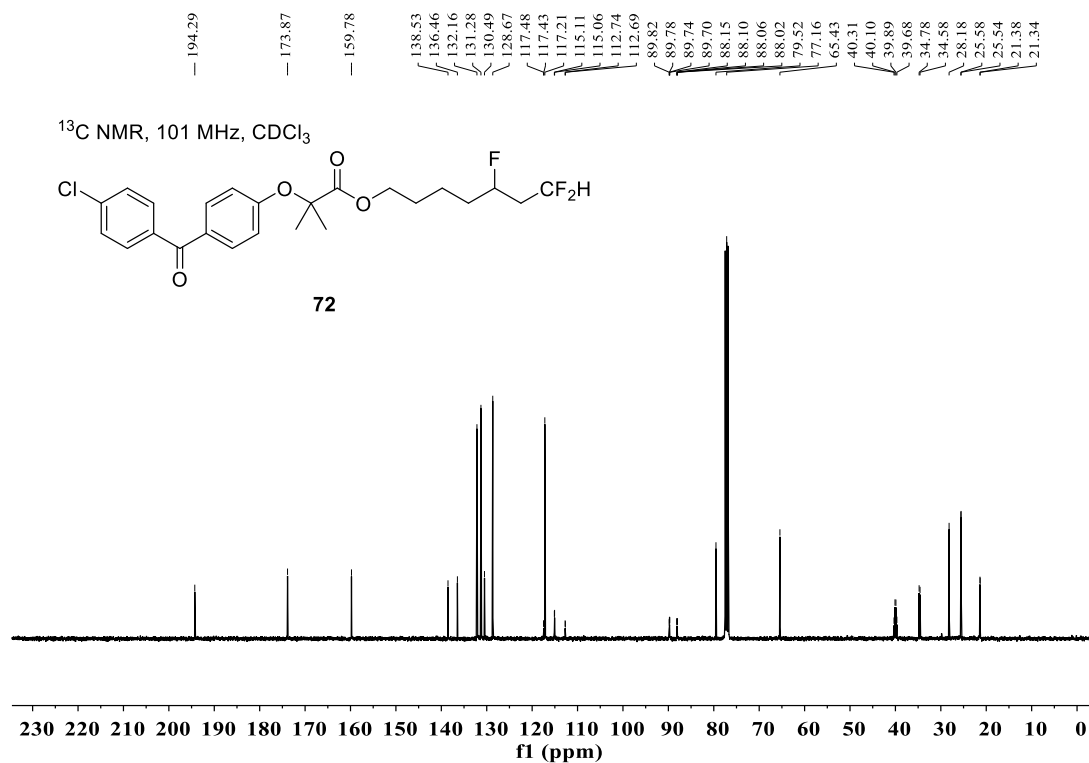
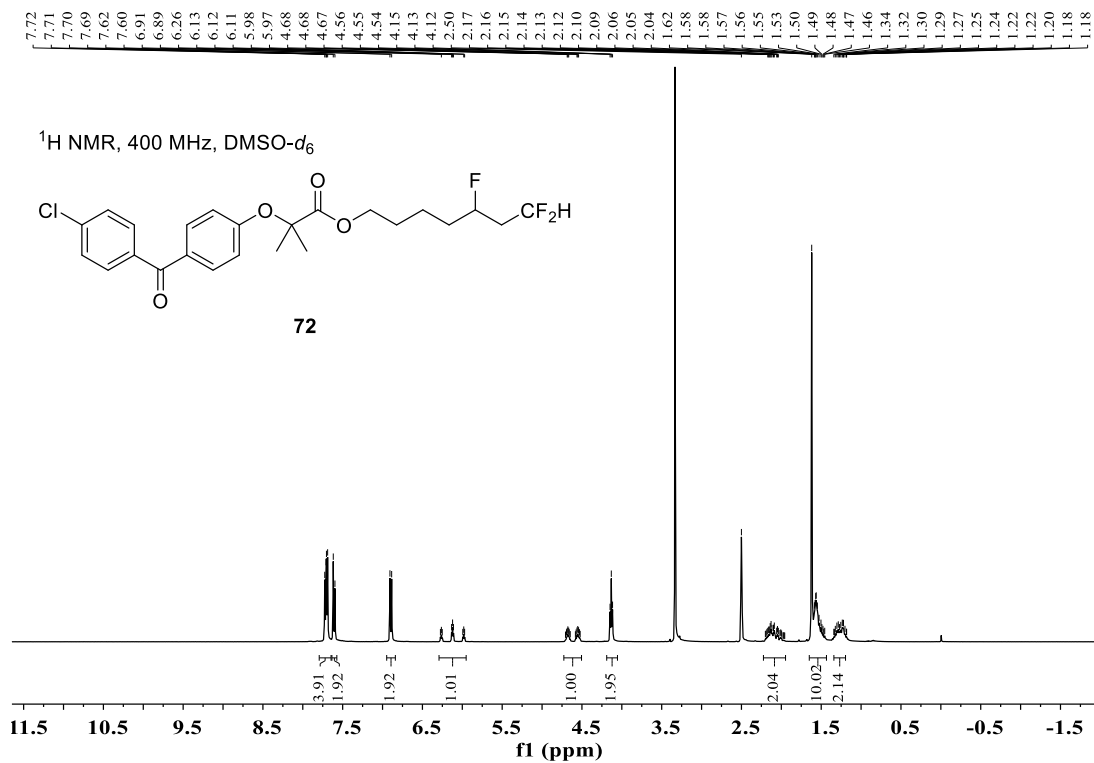


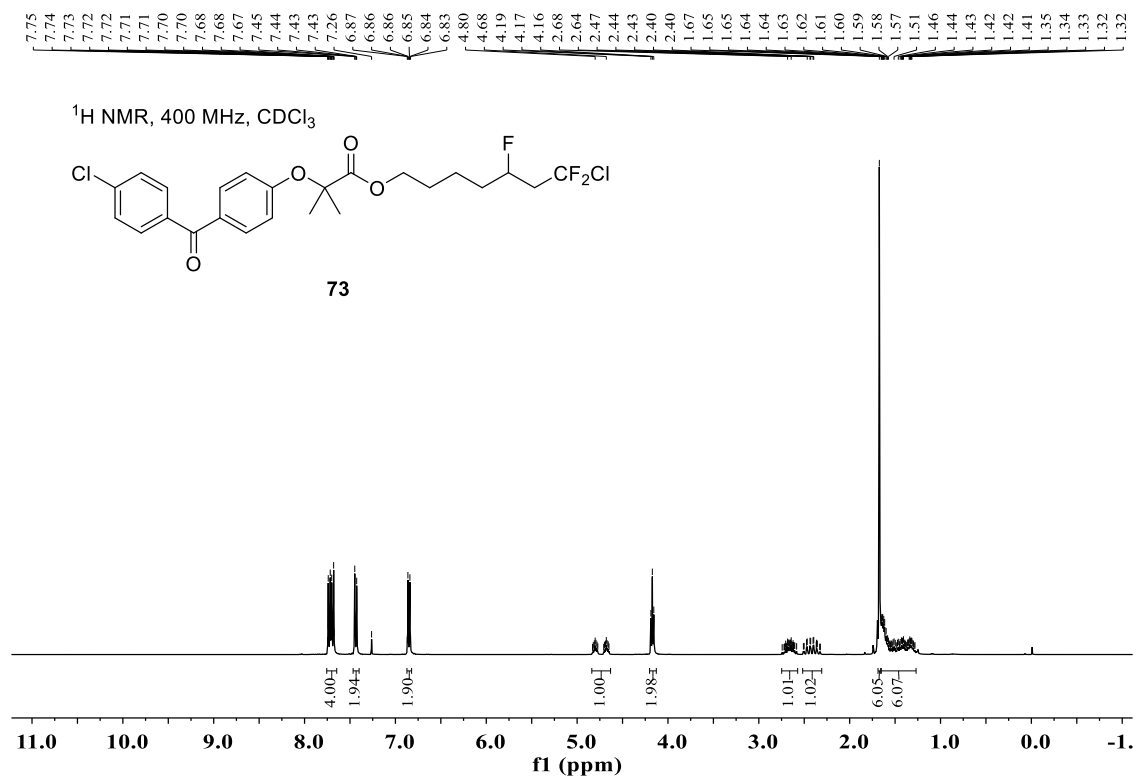
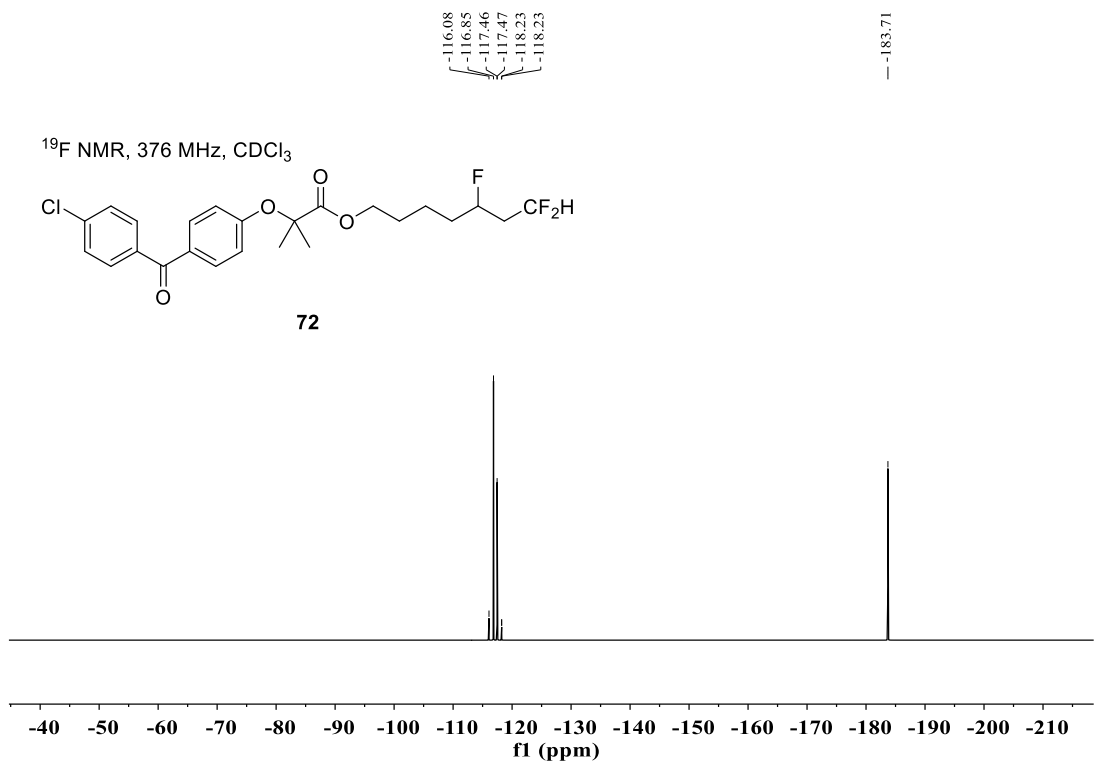


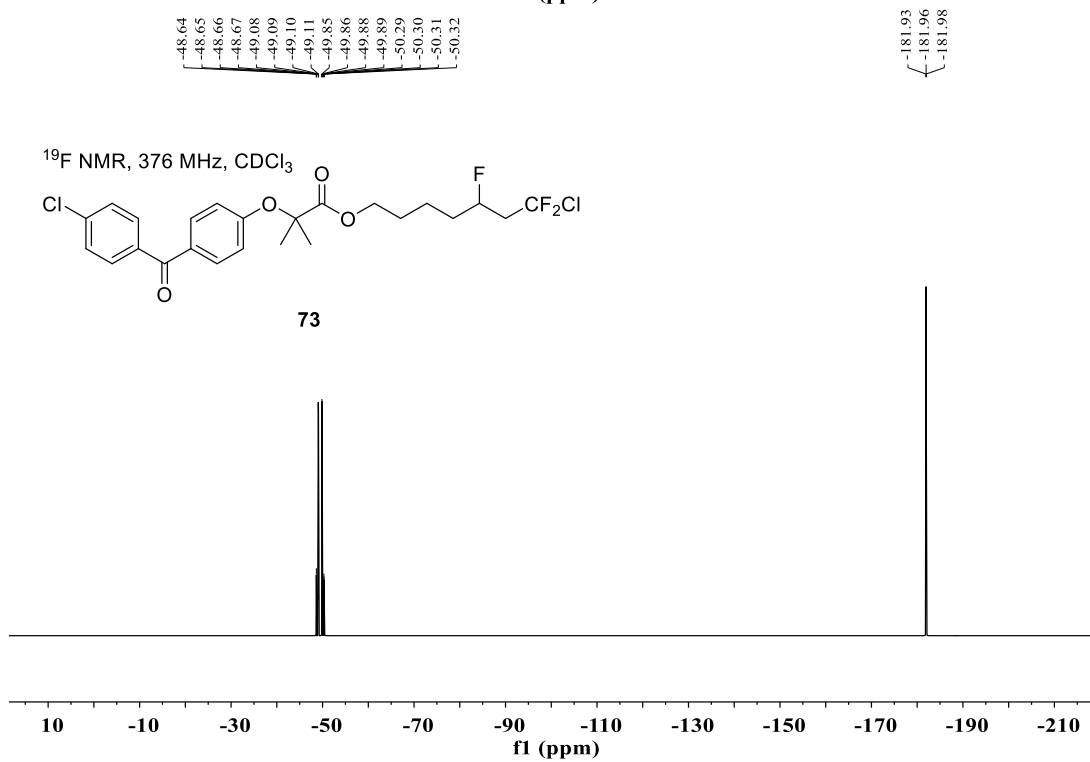
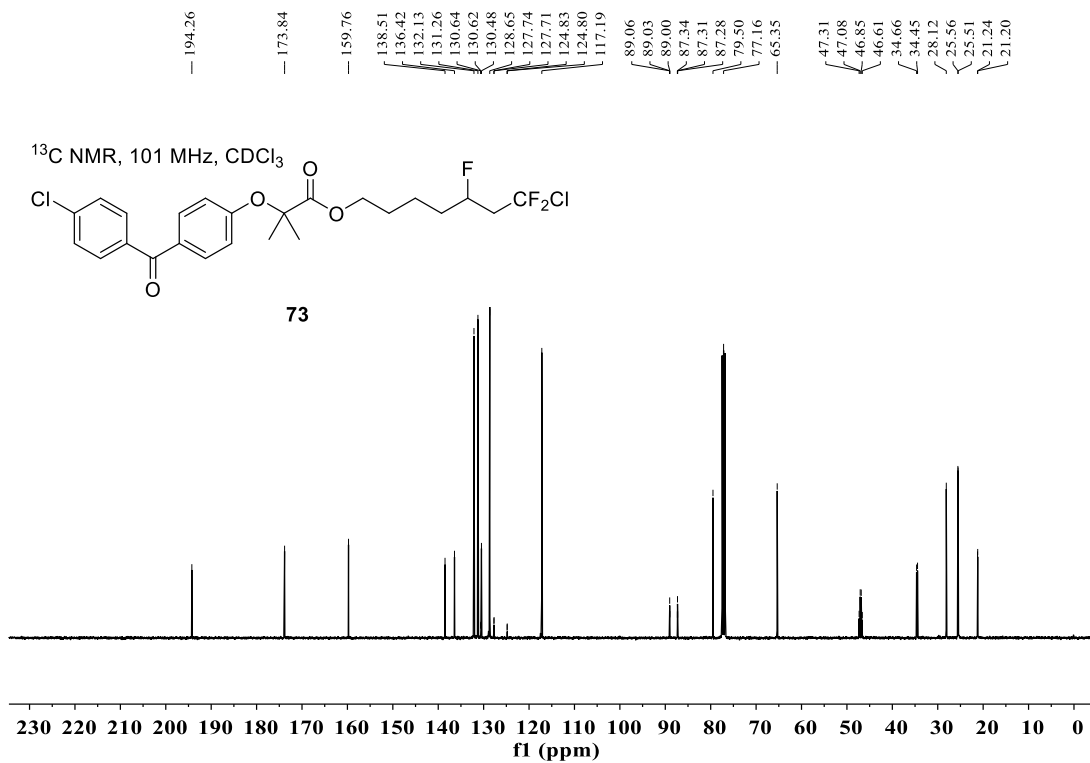


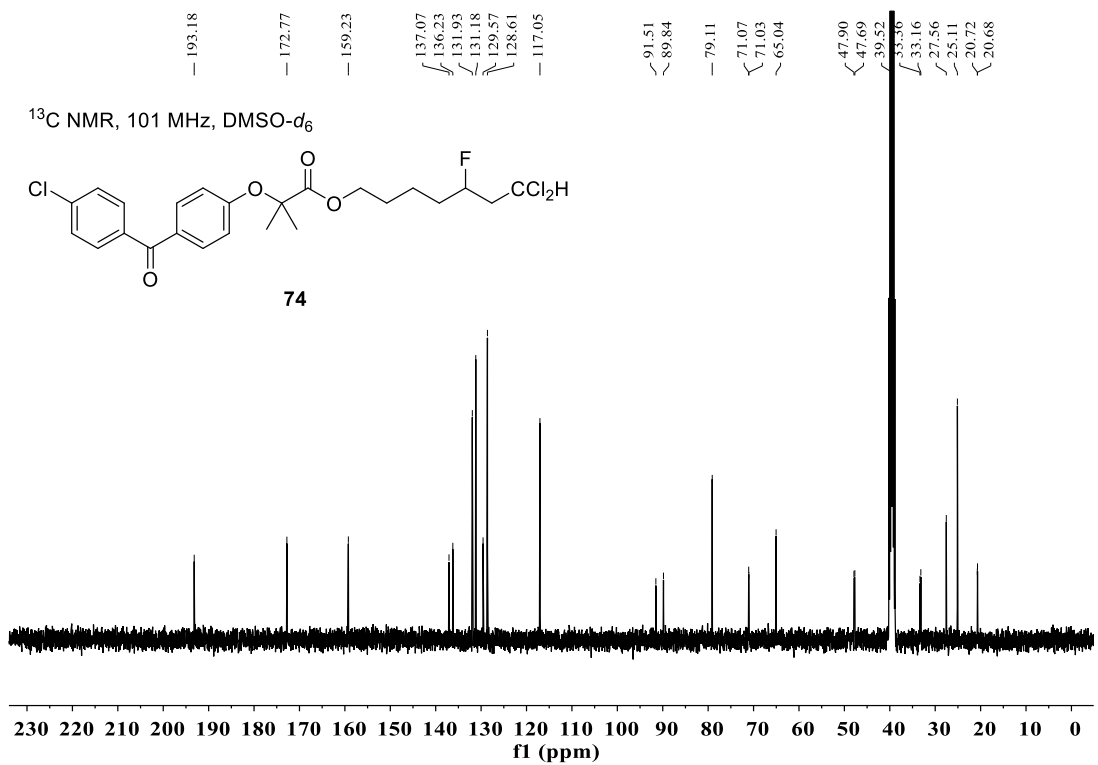
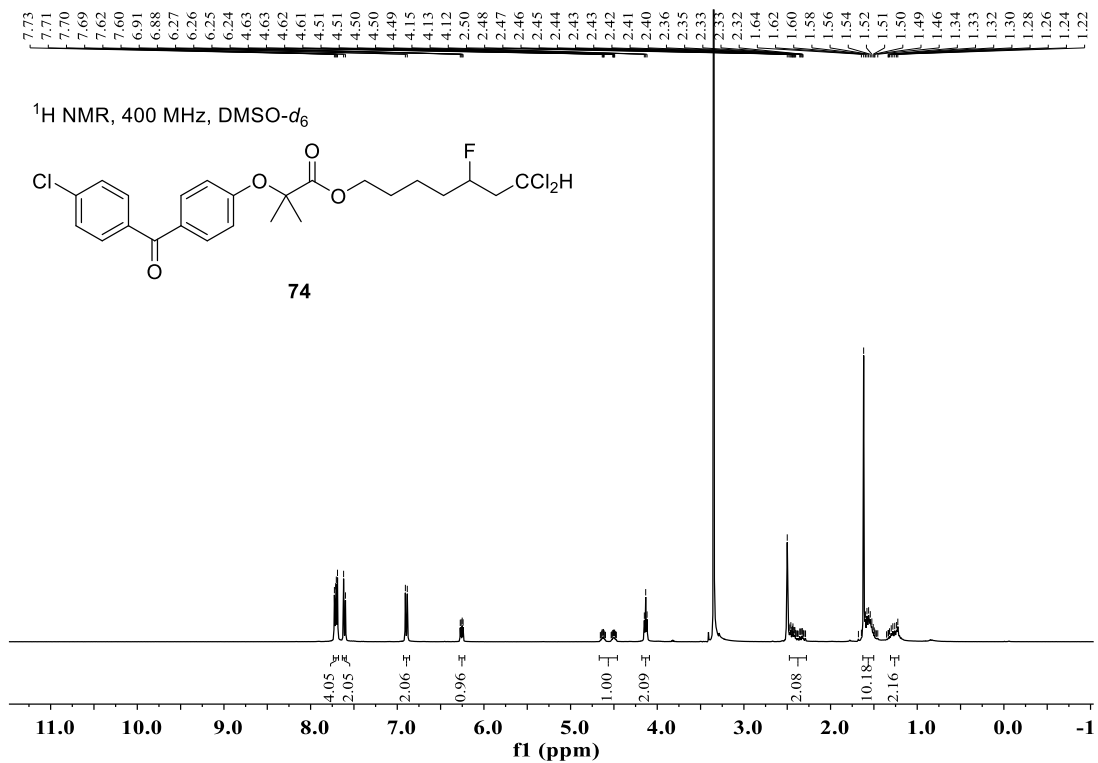




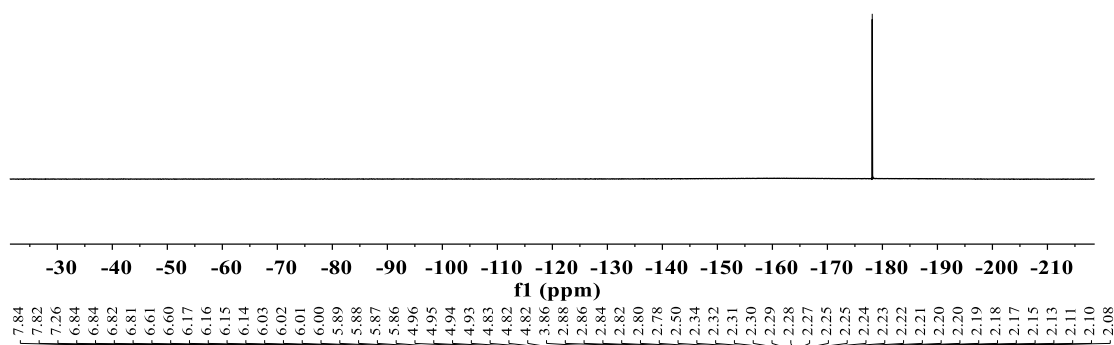
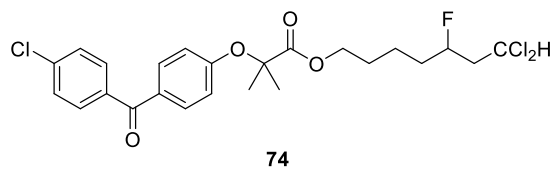




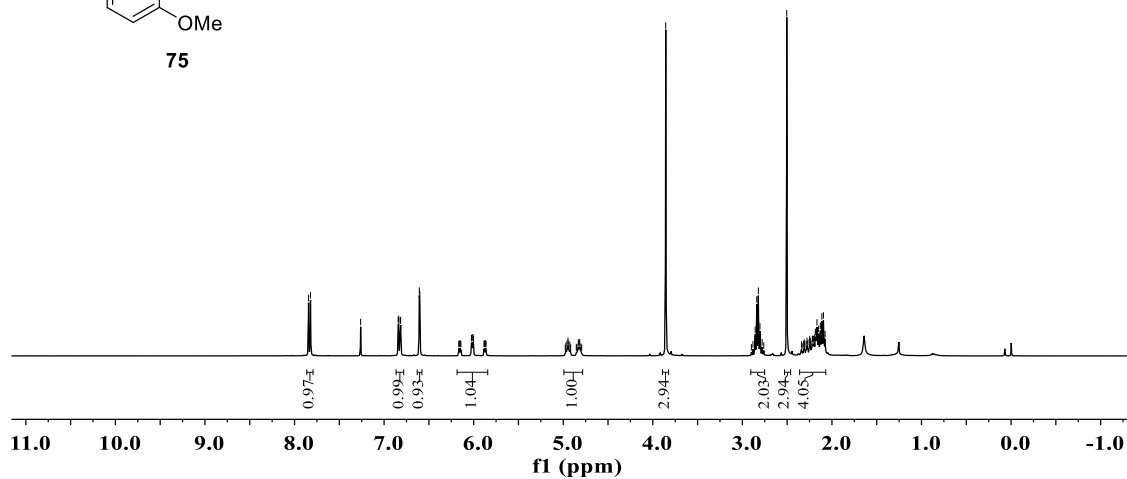
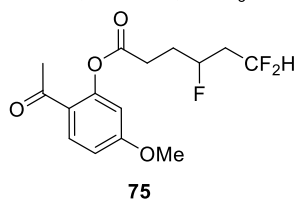


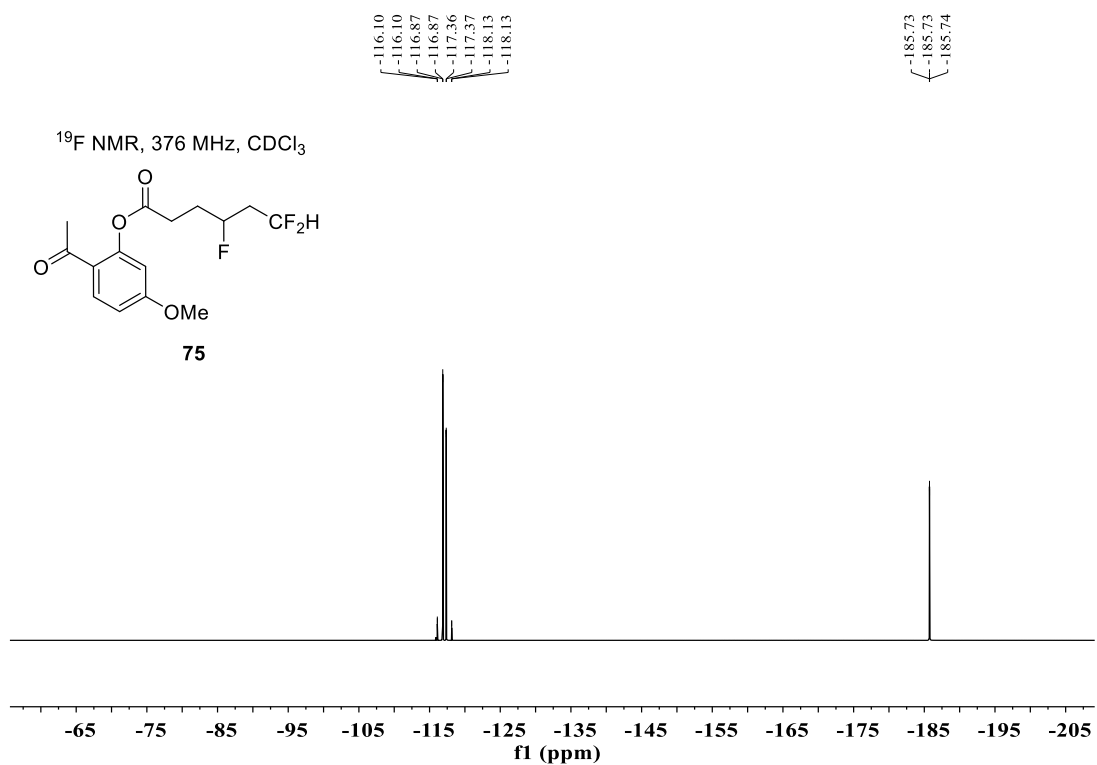
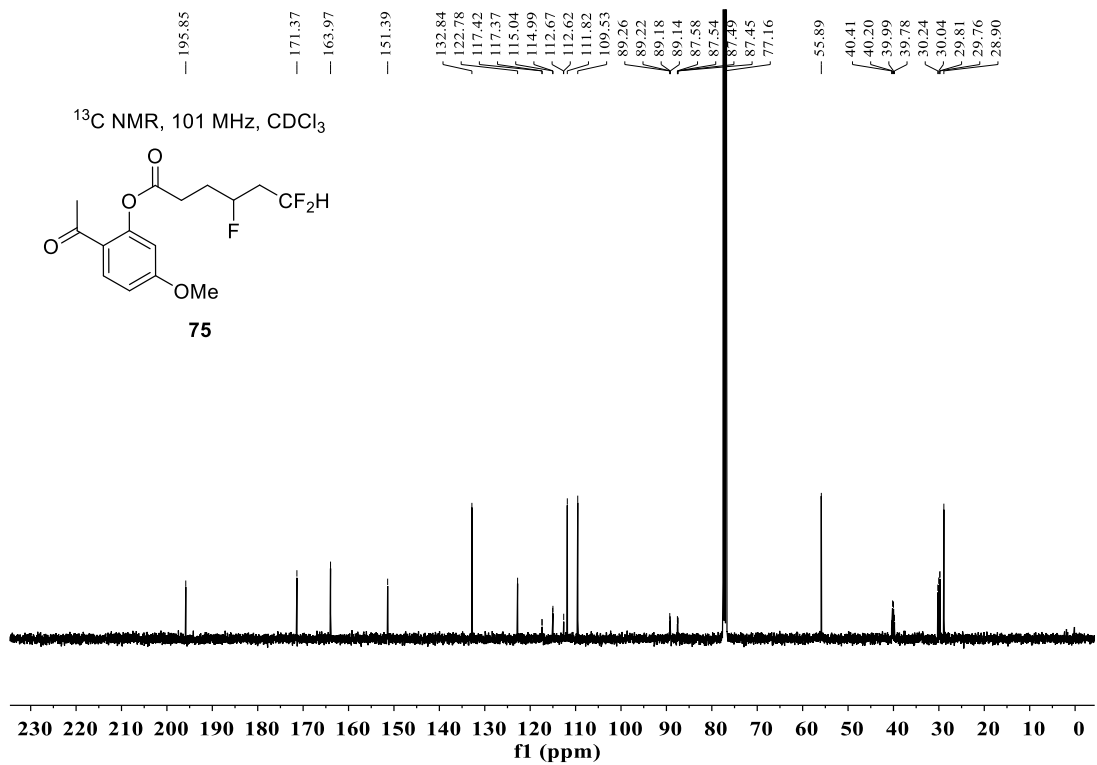


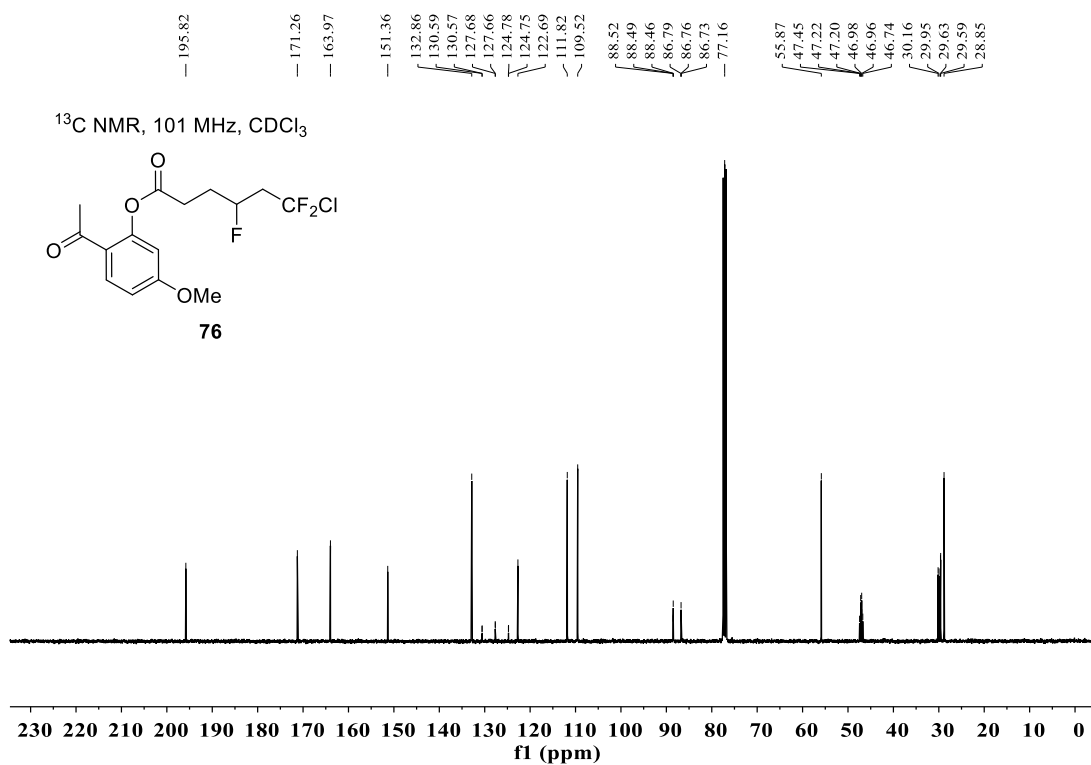
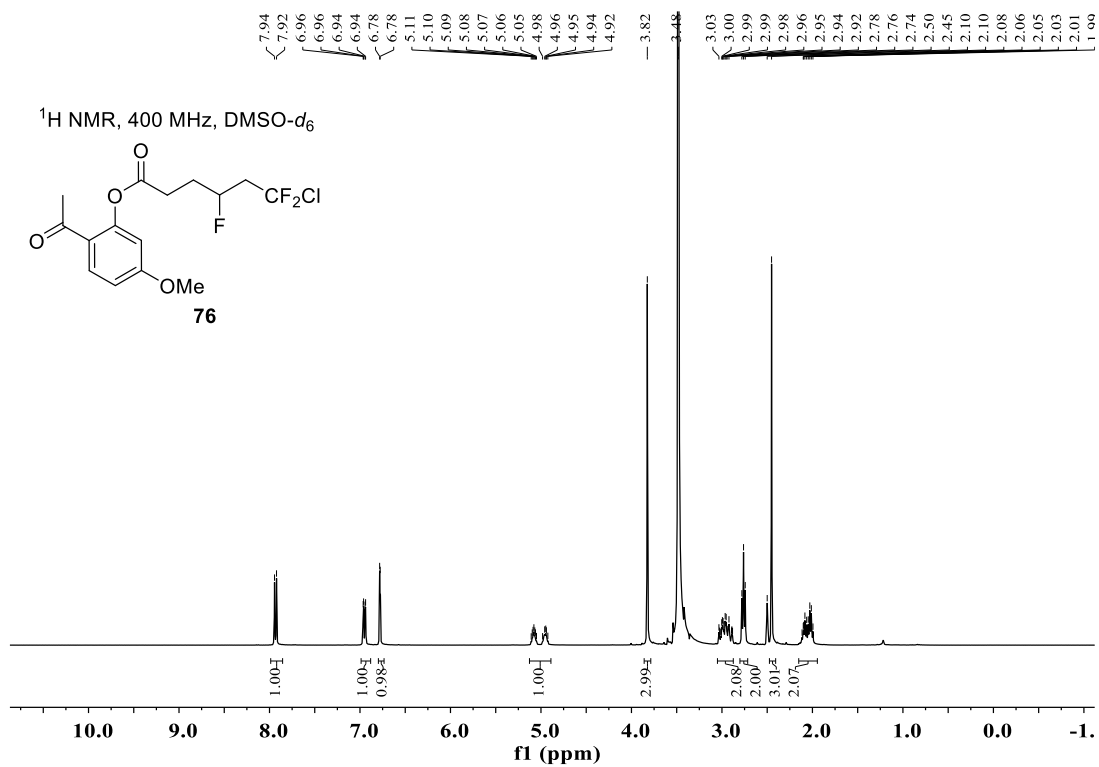
¹⁹F NMR, 376 MHz, DMSO-*d*₆

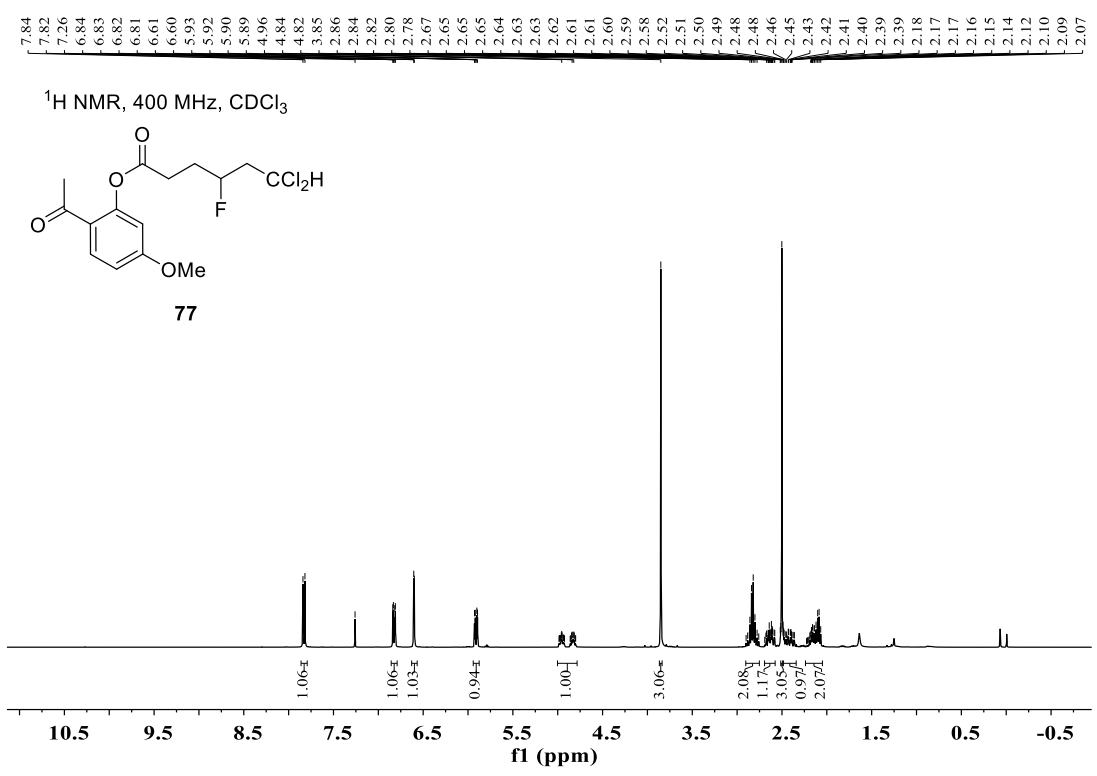
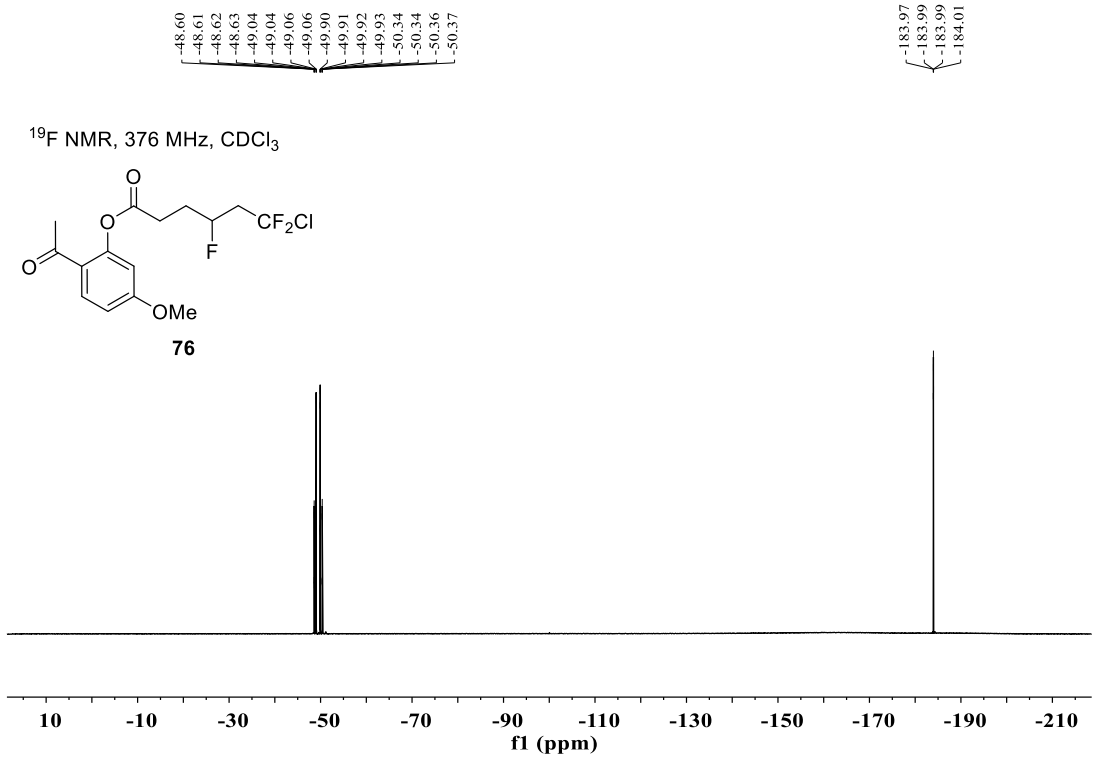


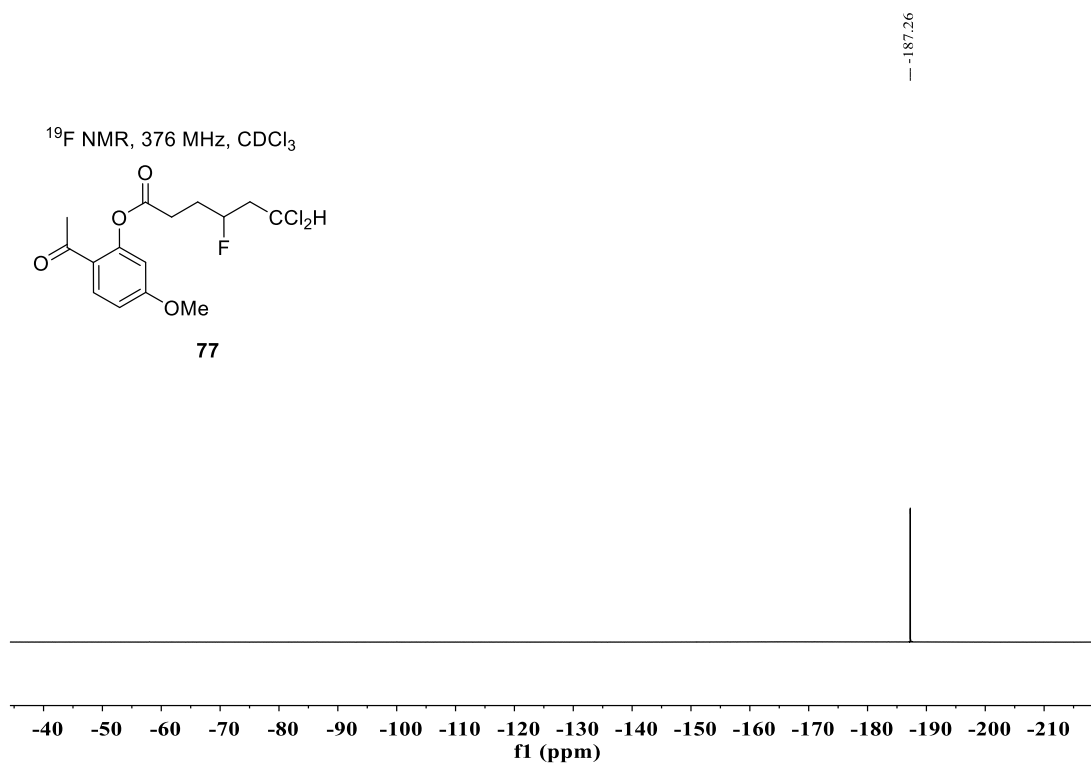
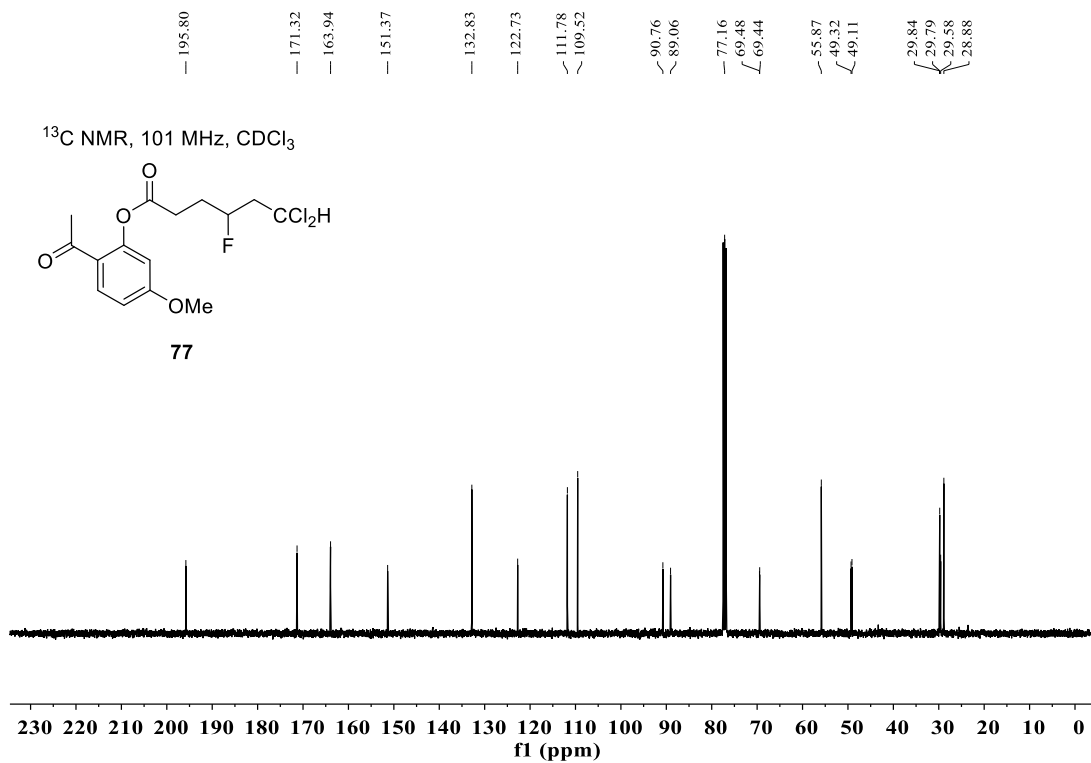
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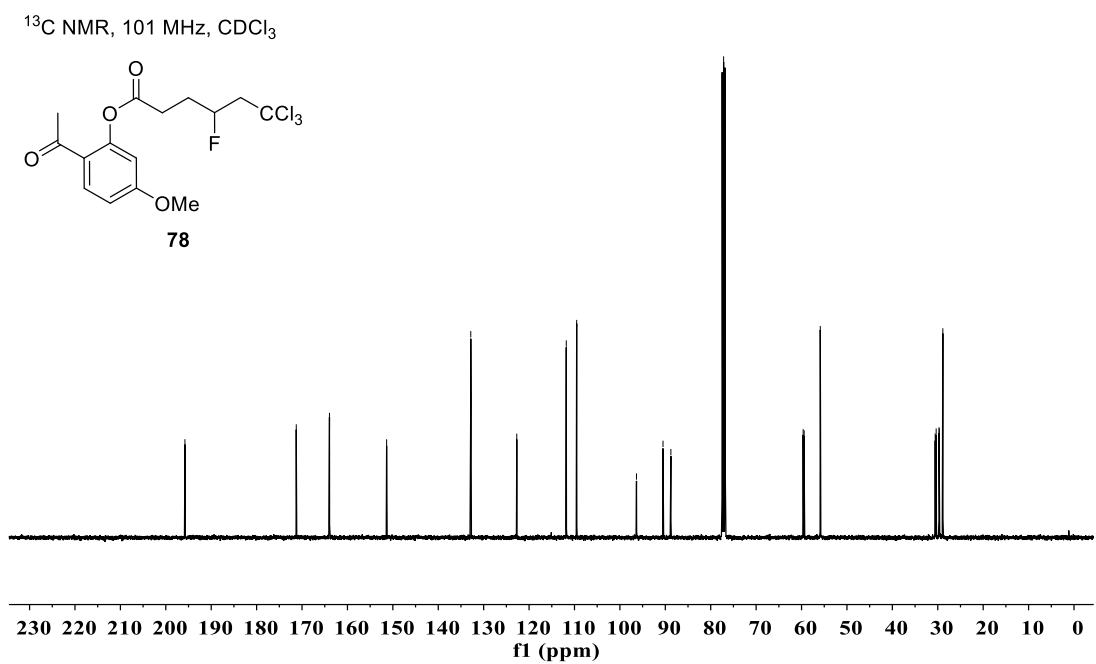
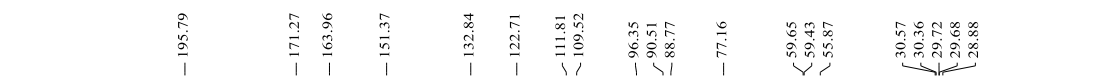
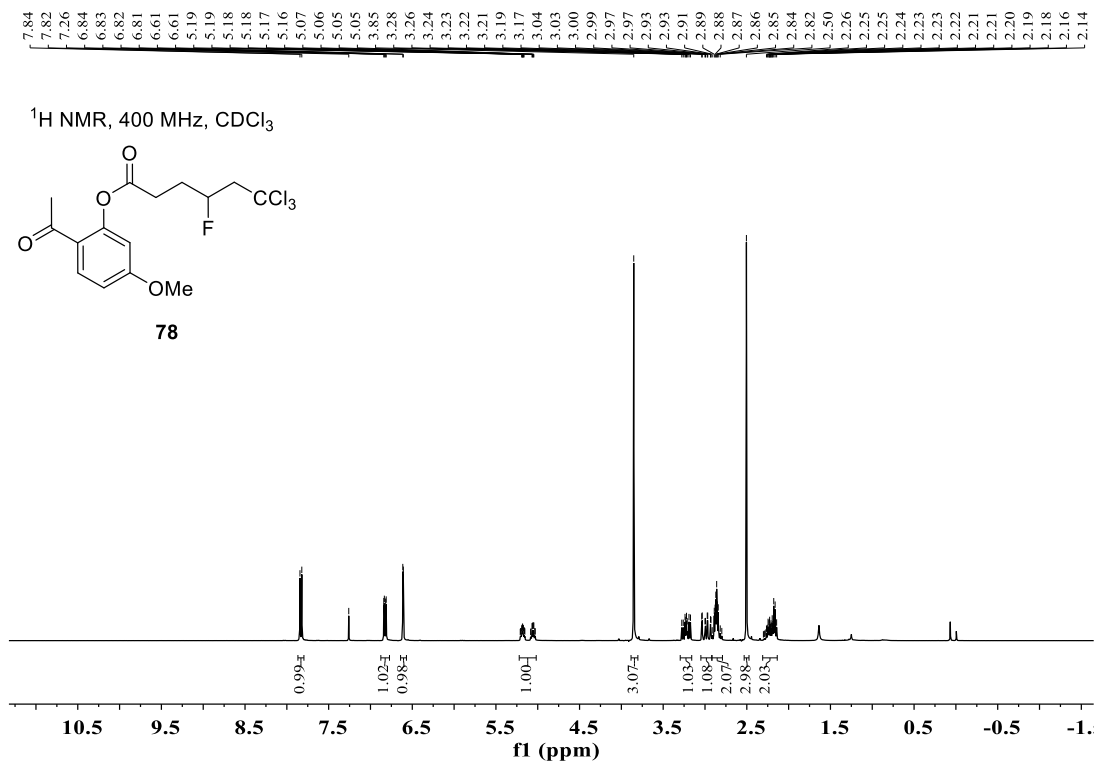




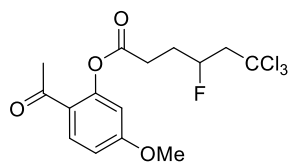




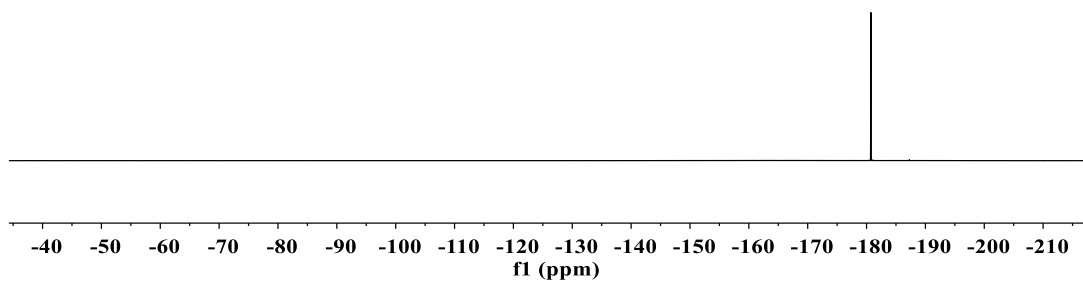




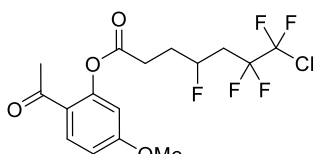
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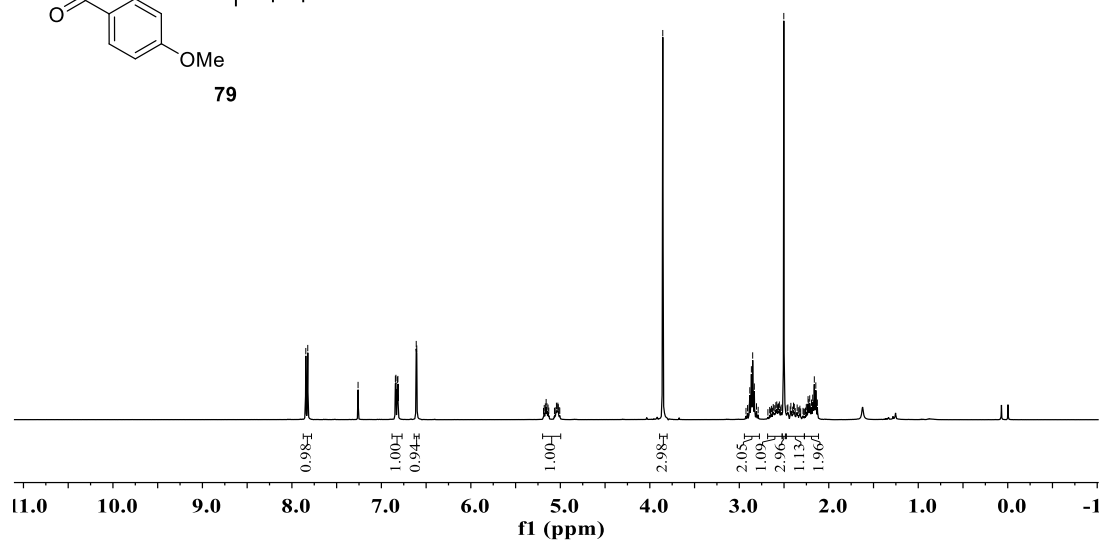
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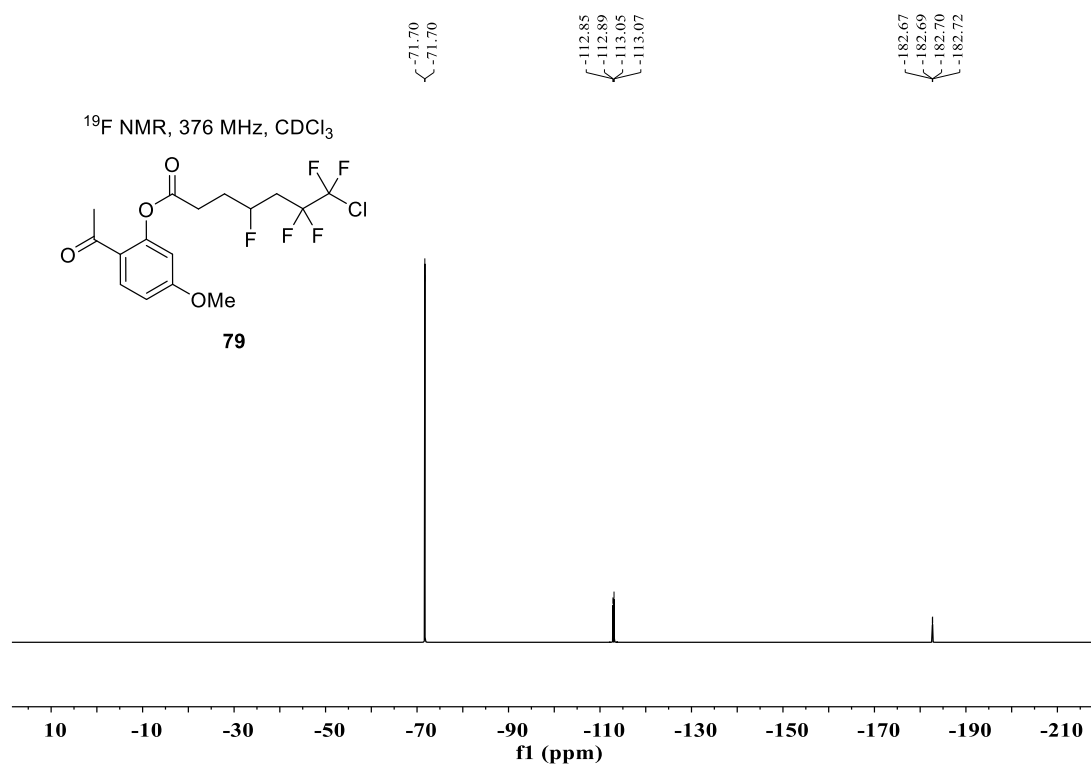
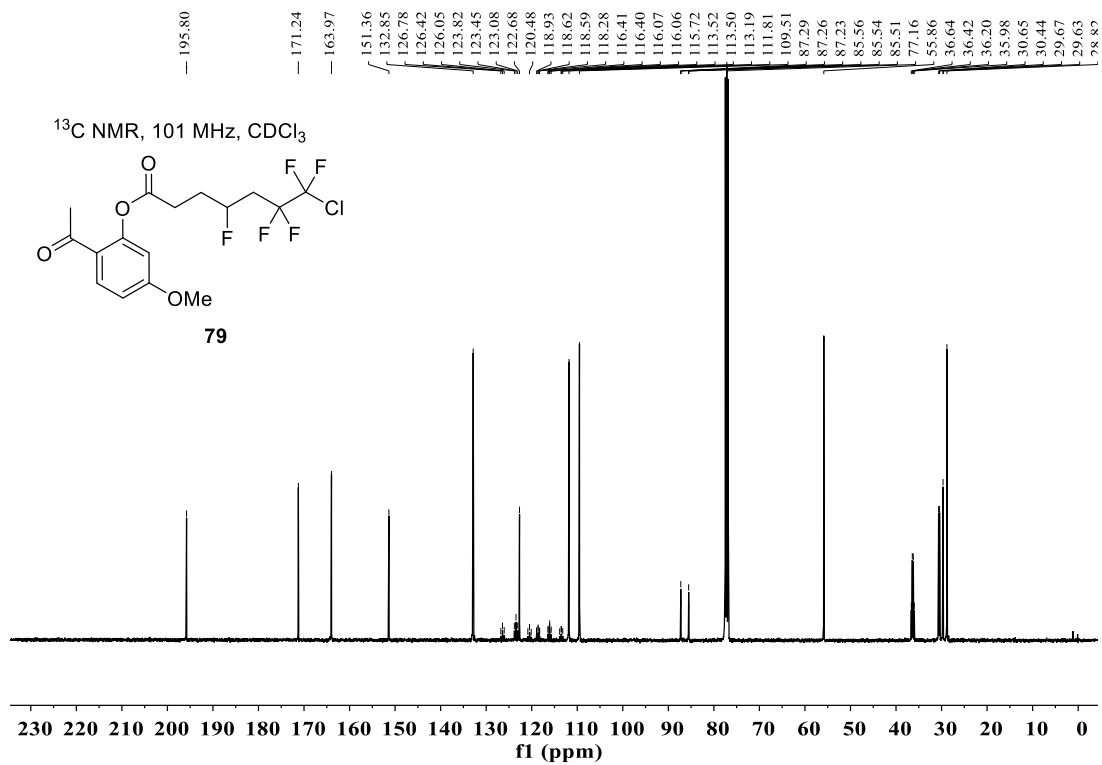


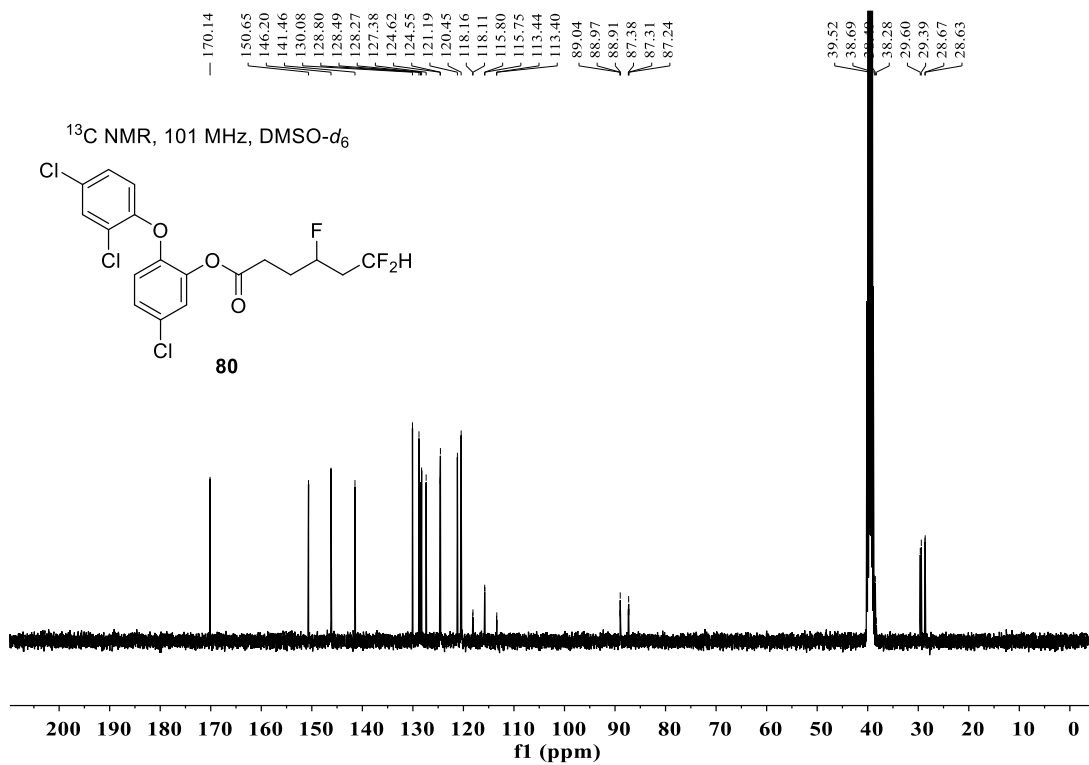
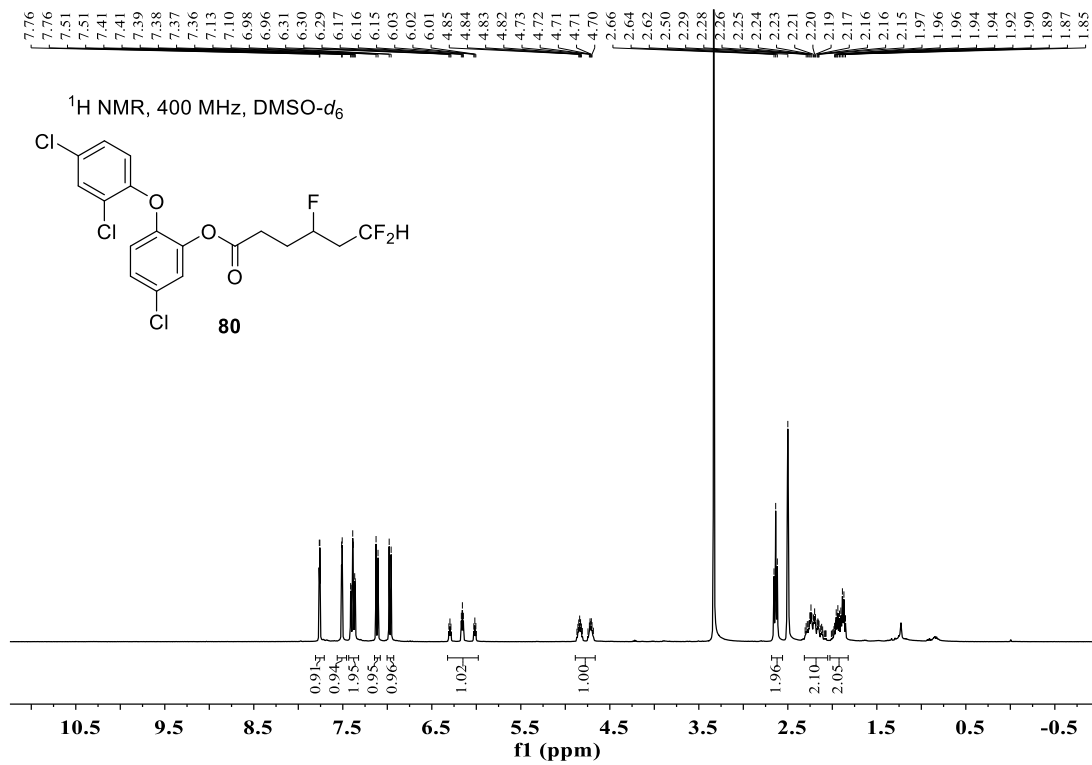
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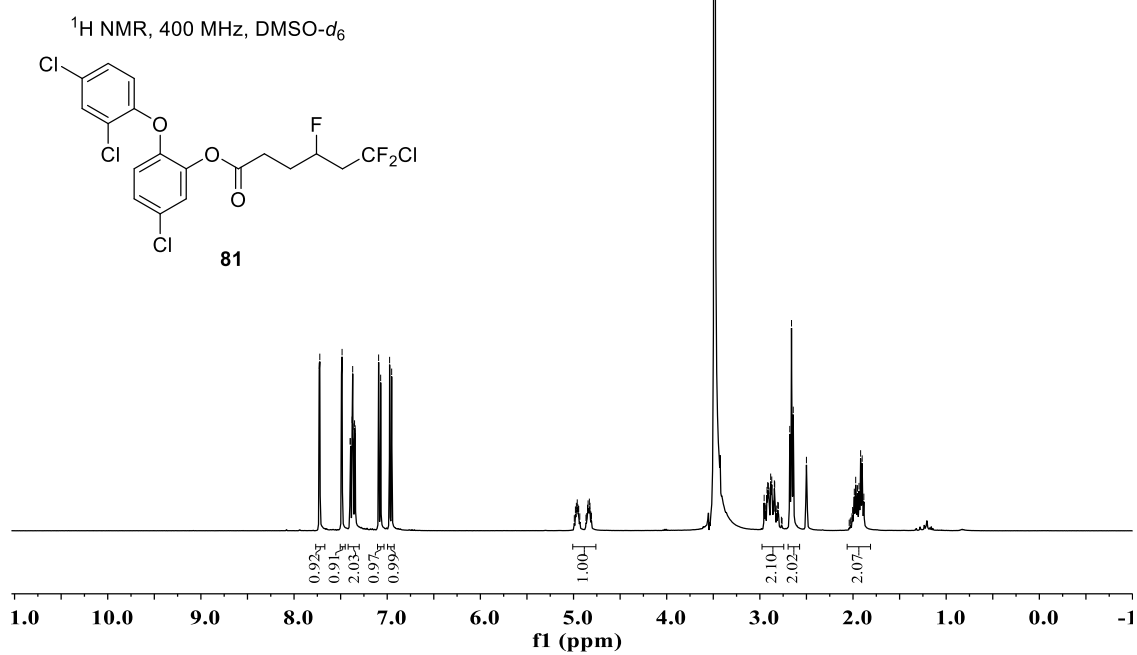
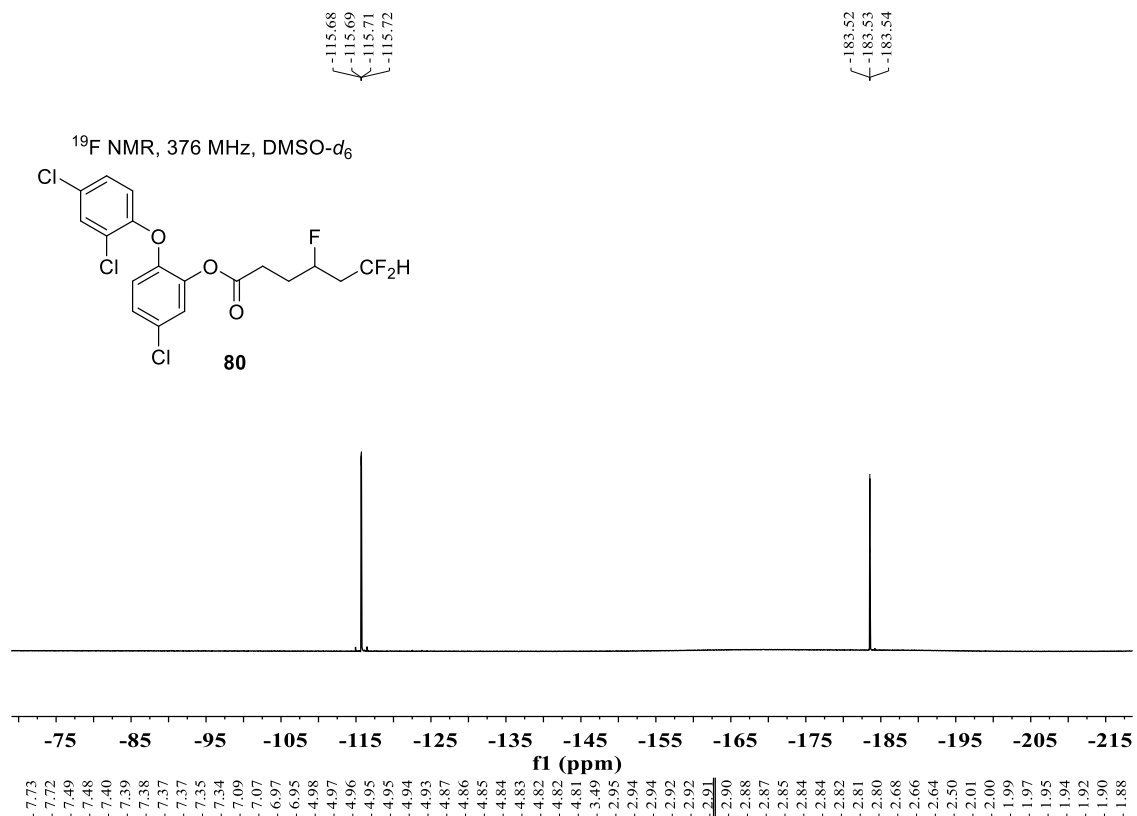


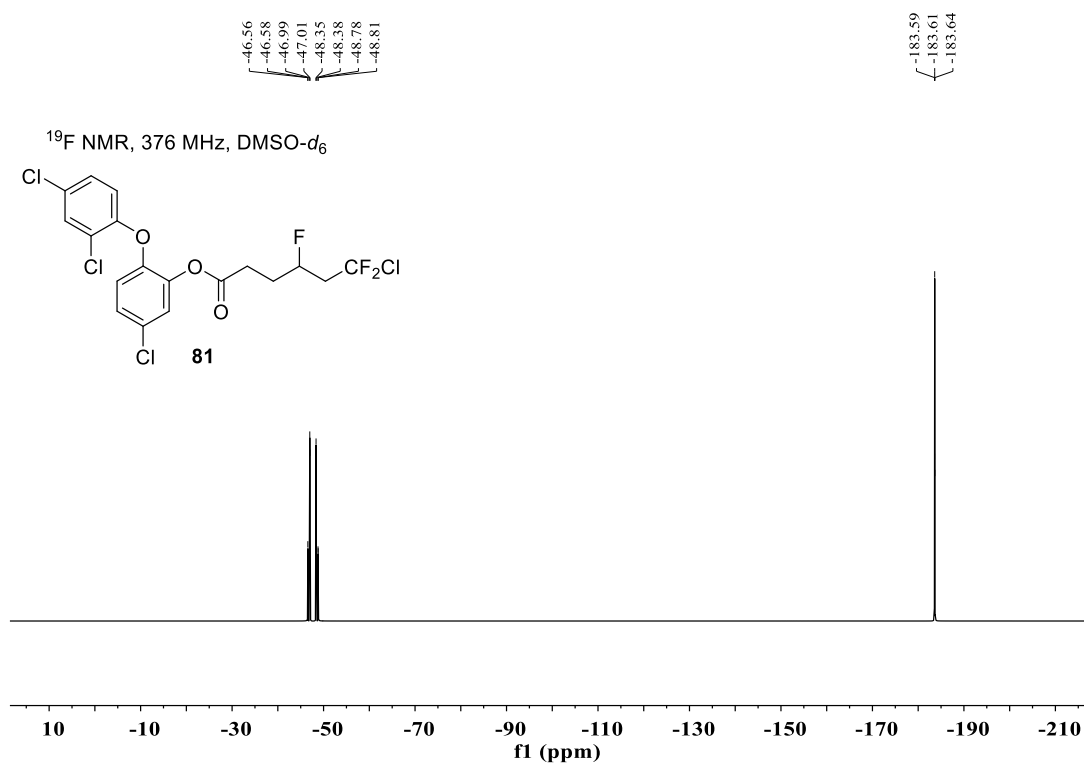
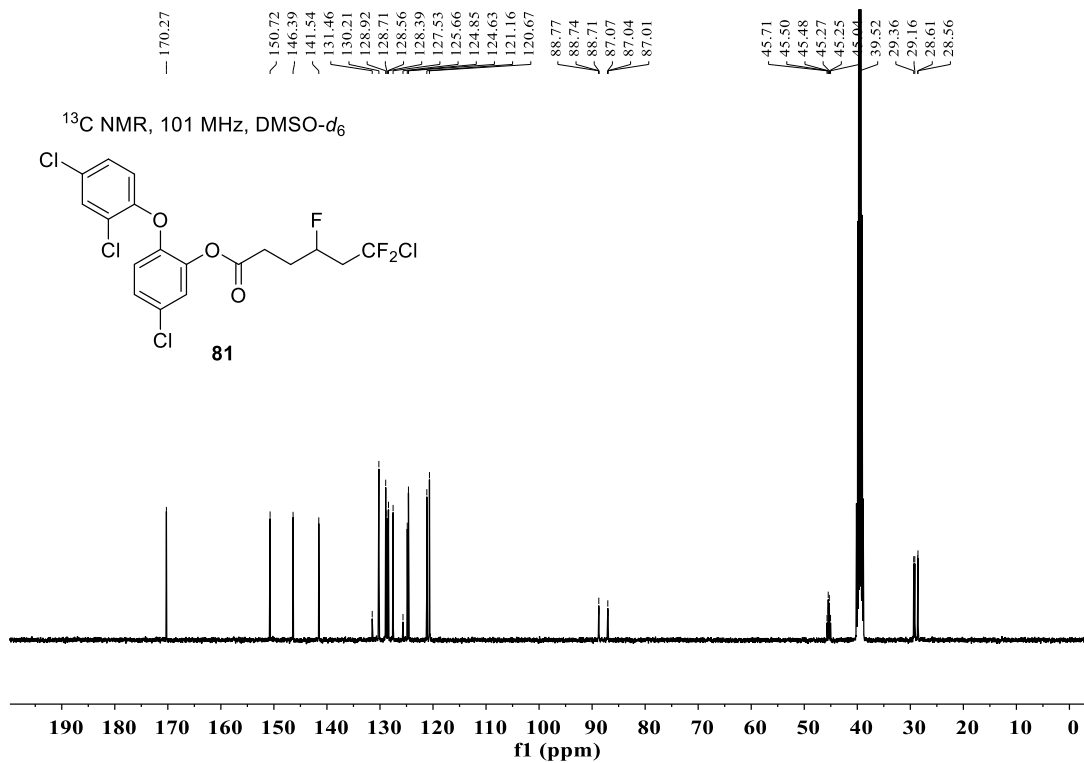
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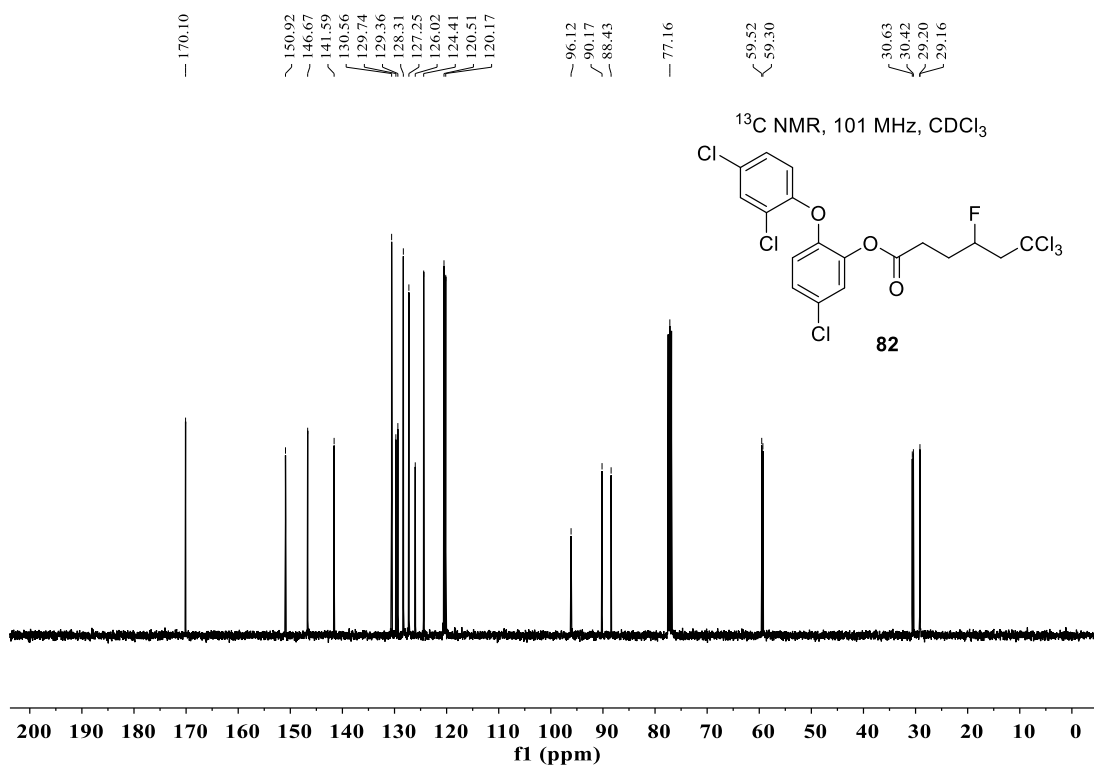
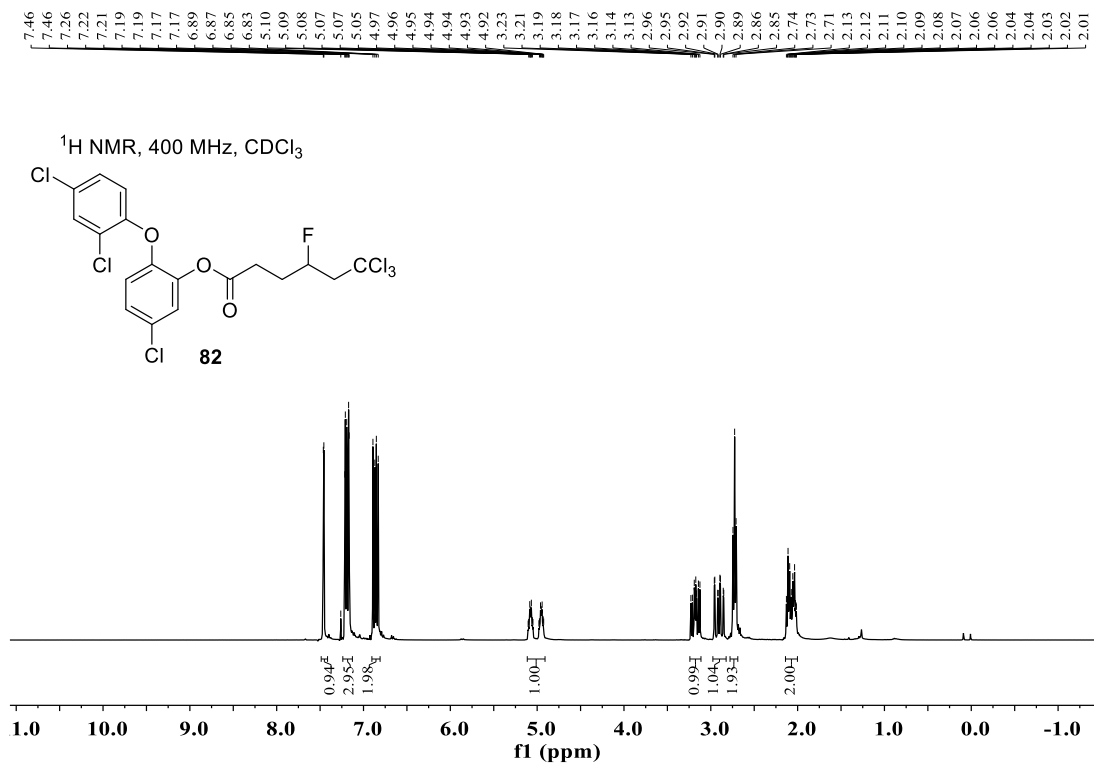


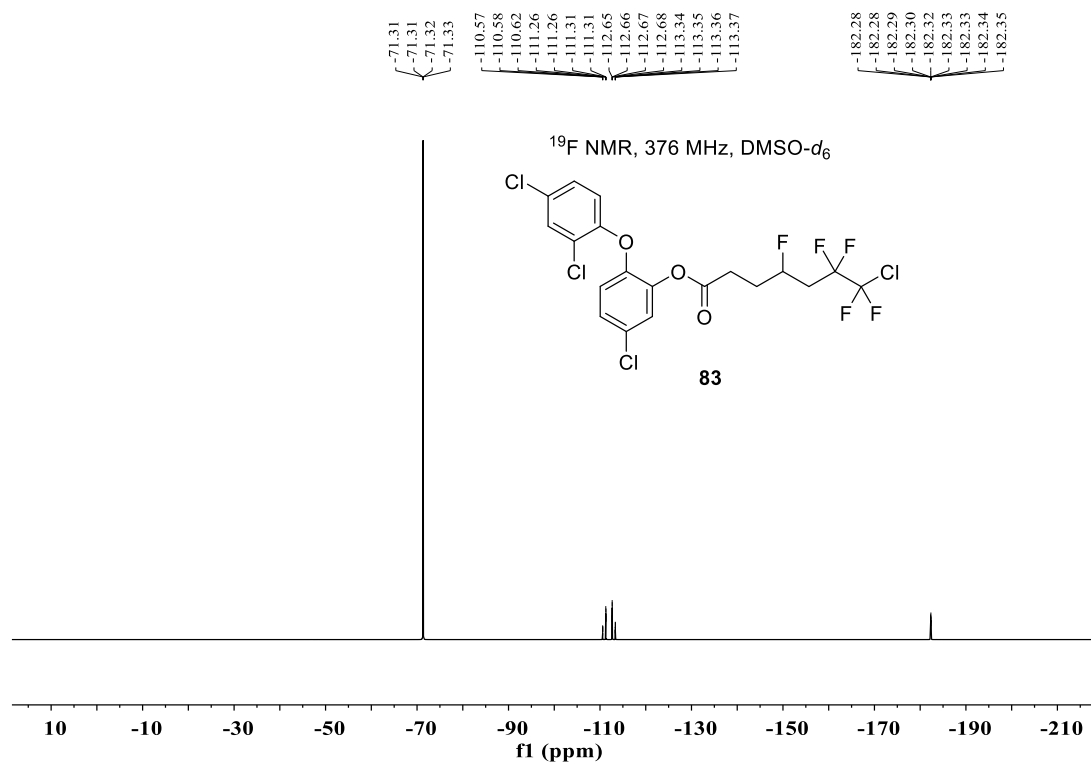
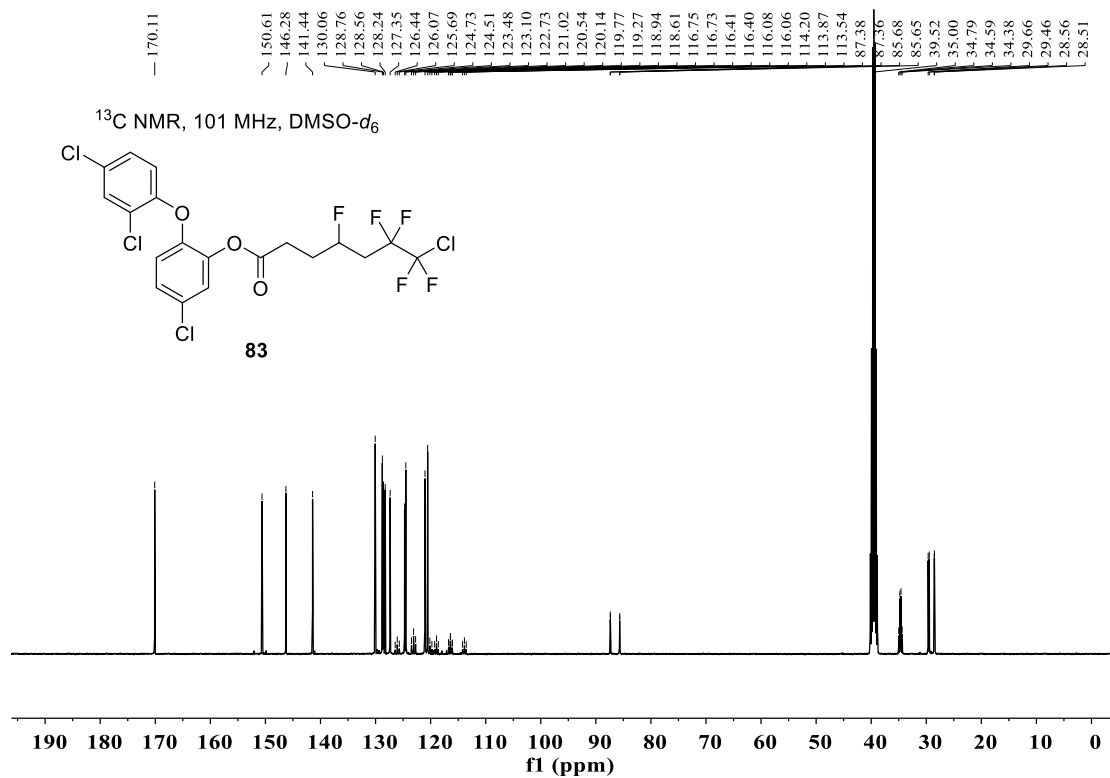


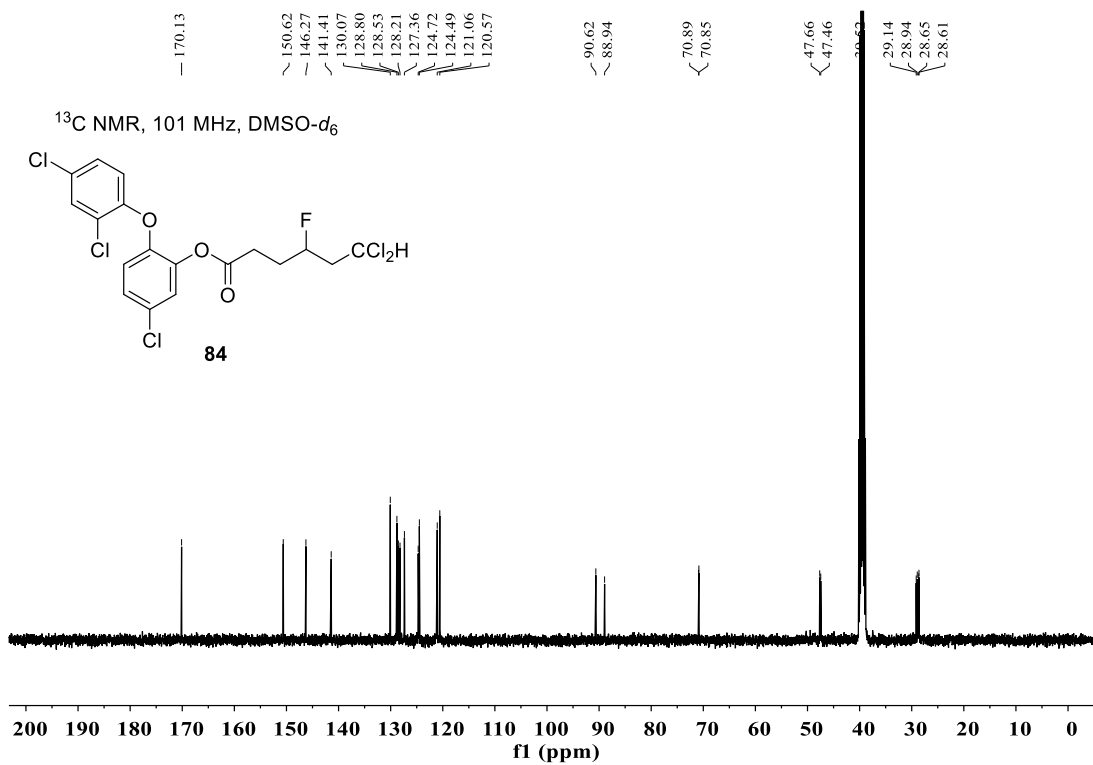
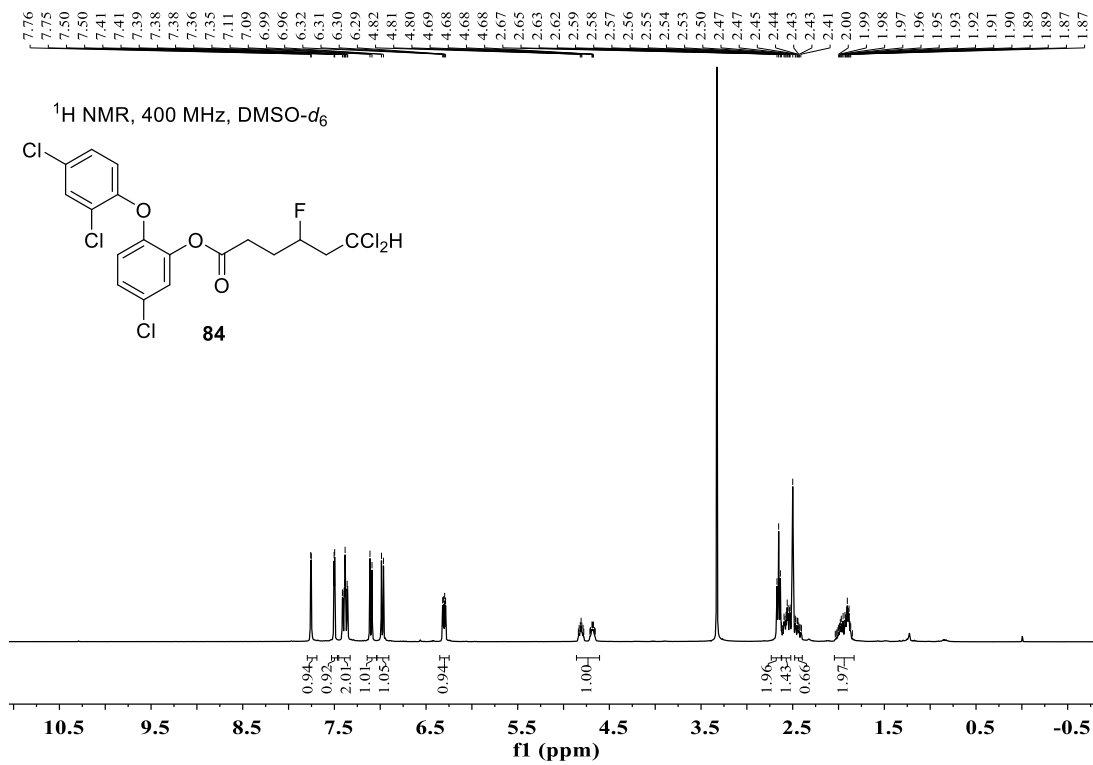


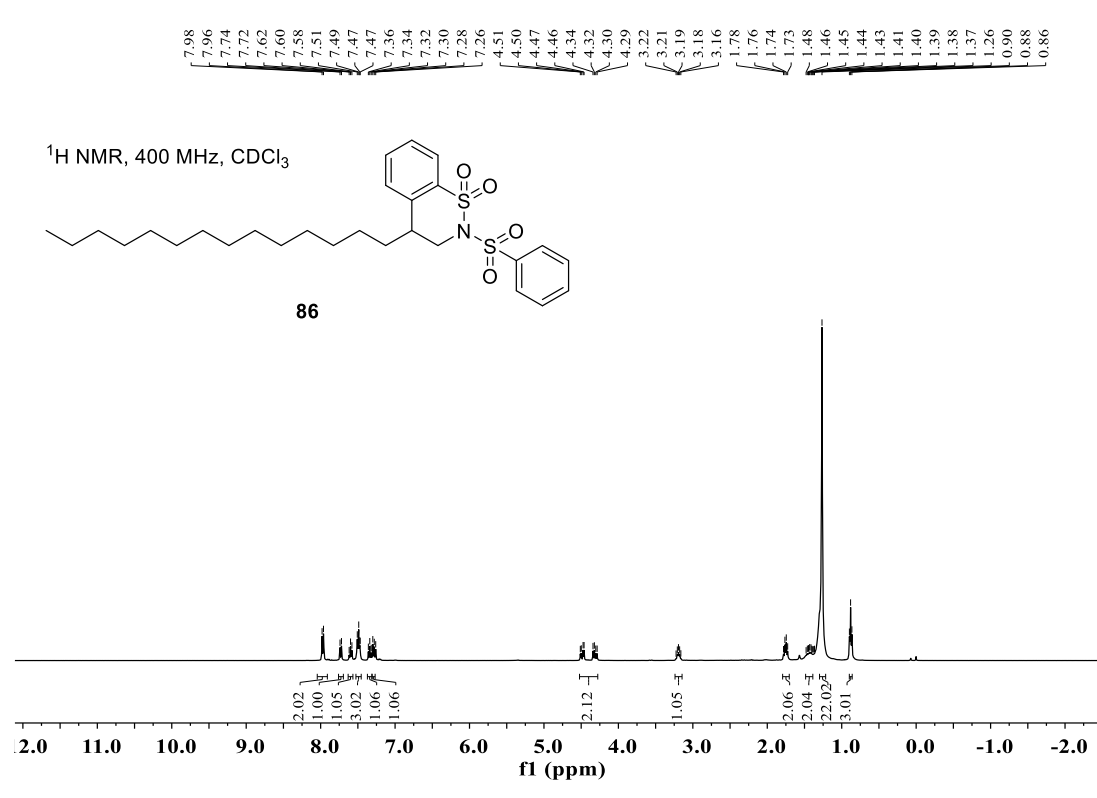
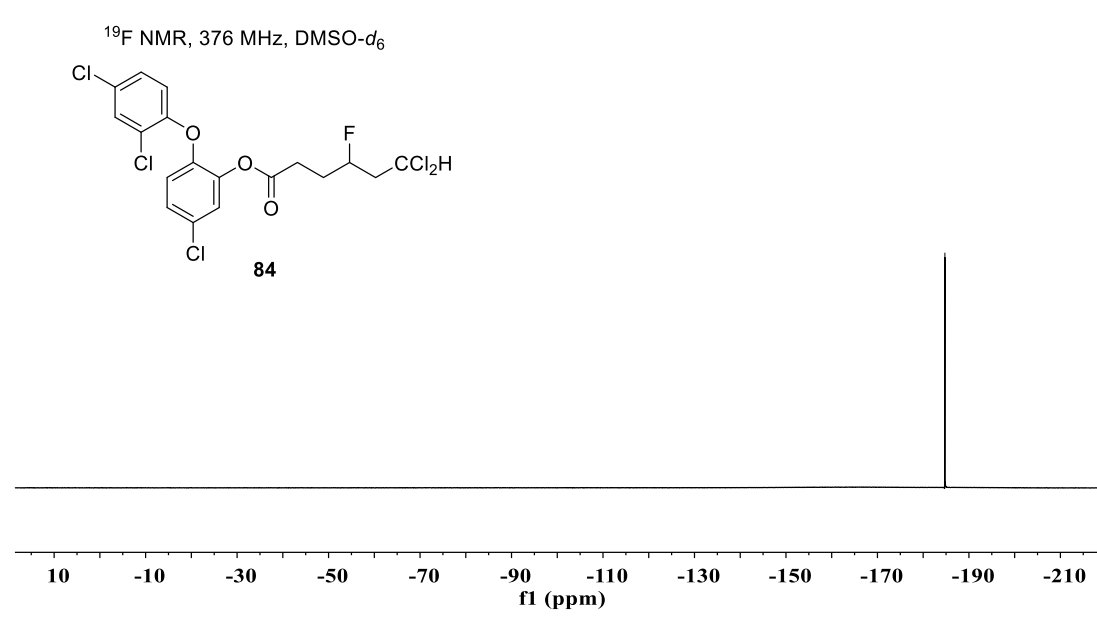


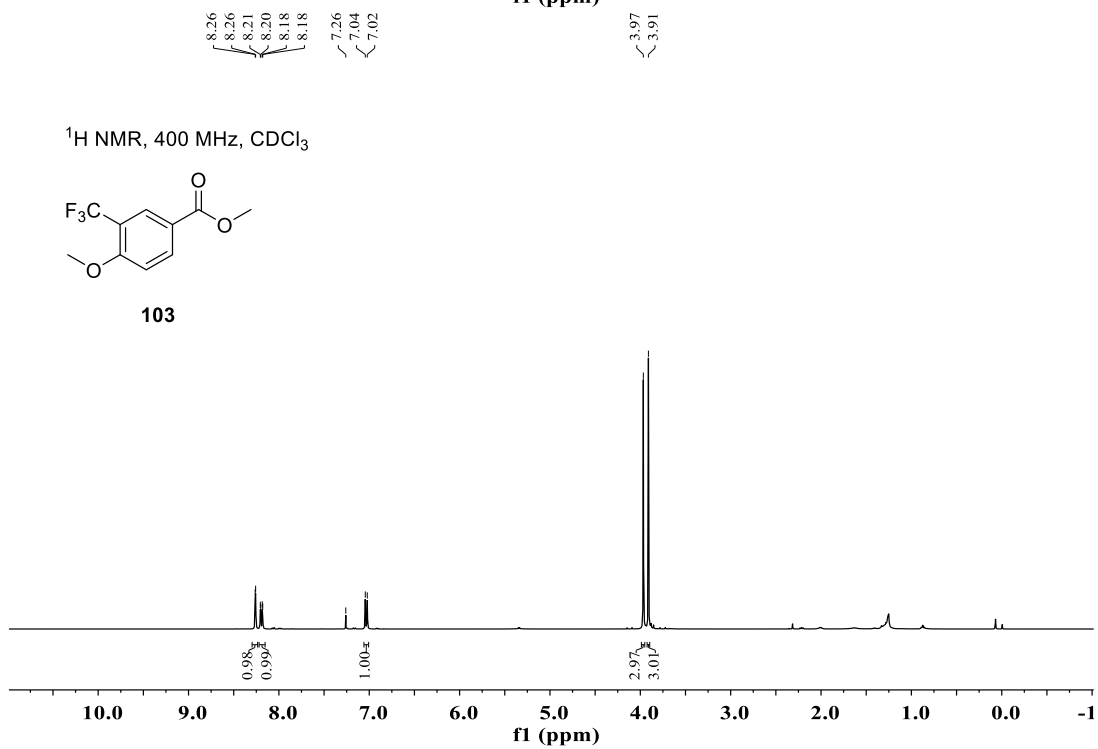
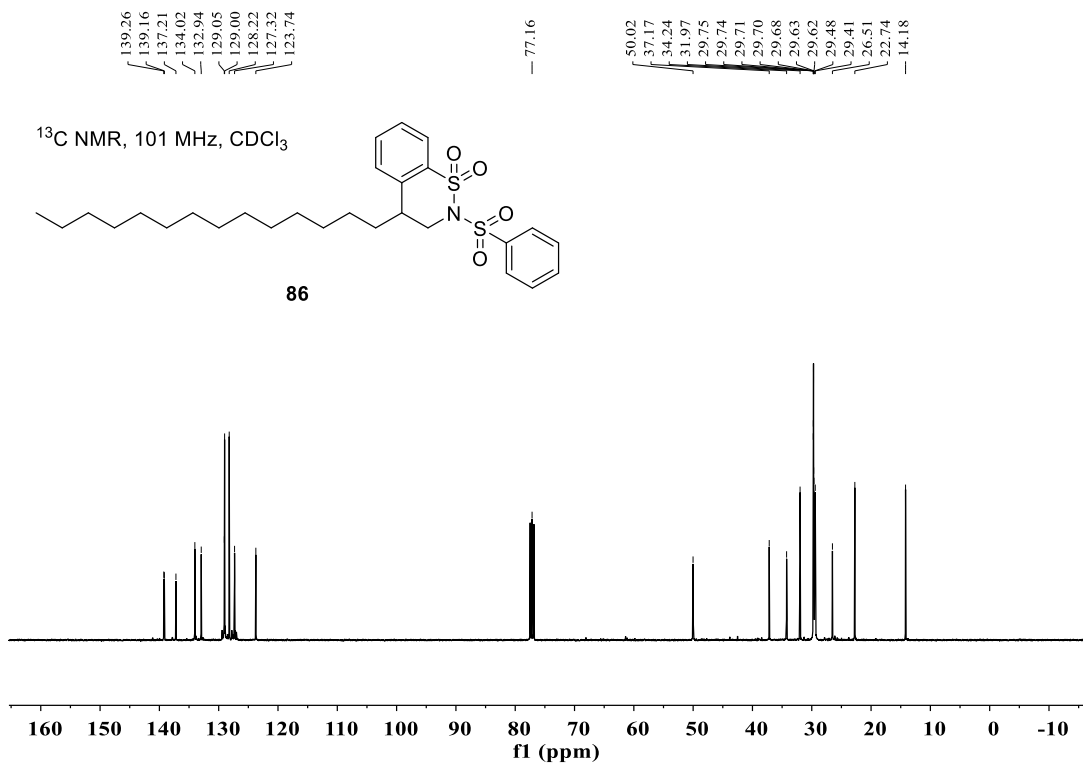


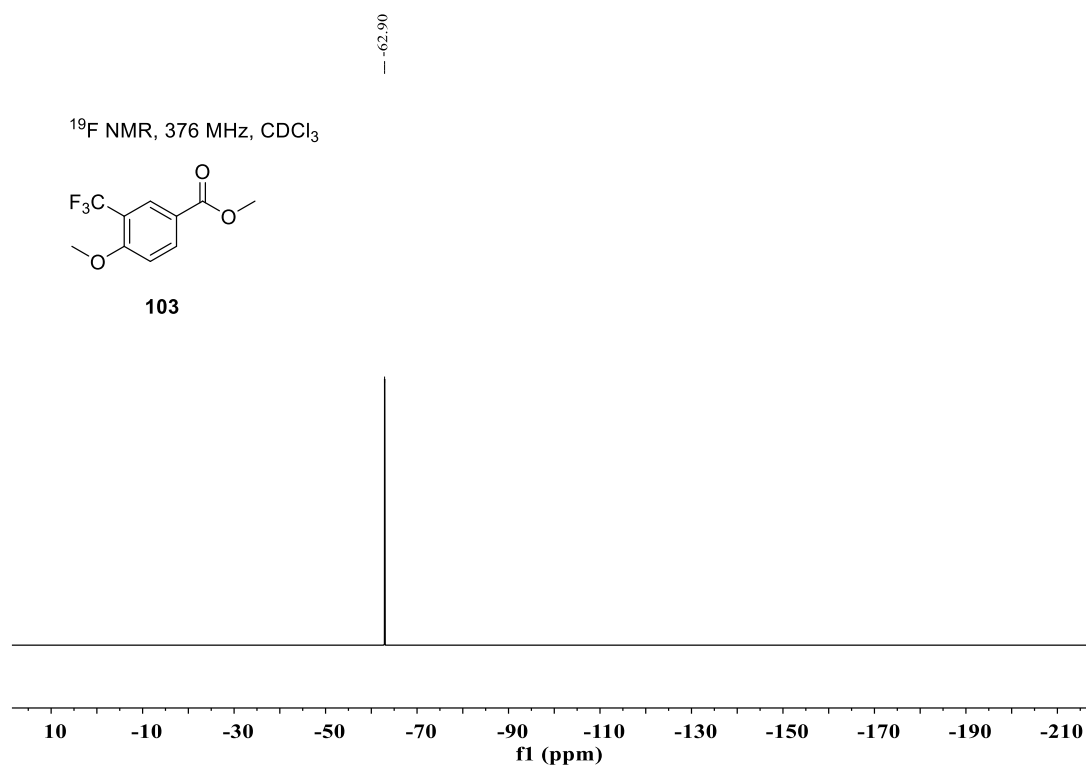
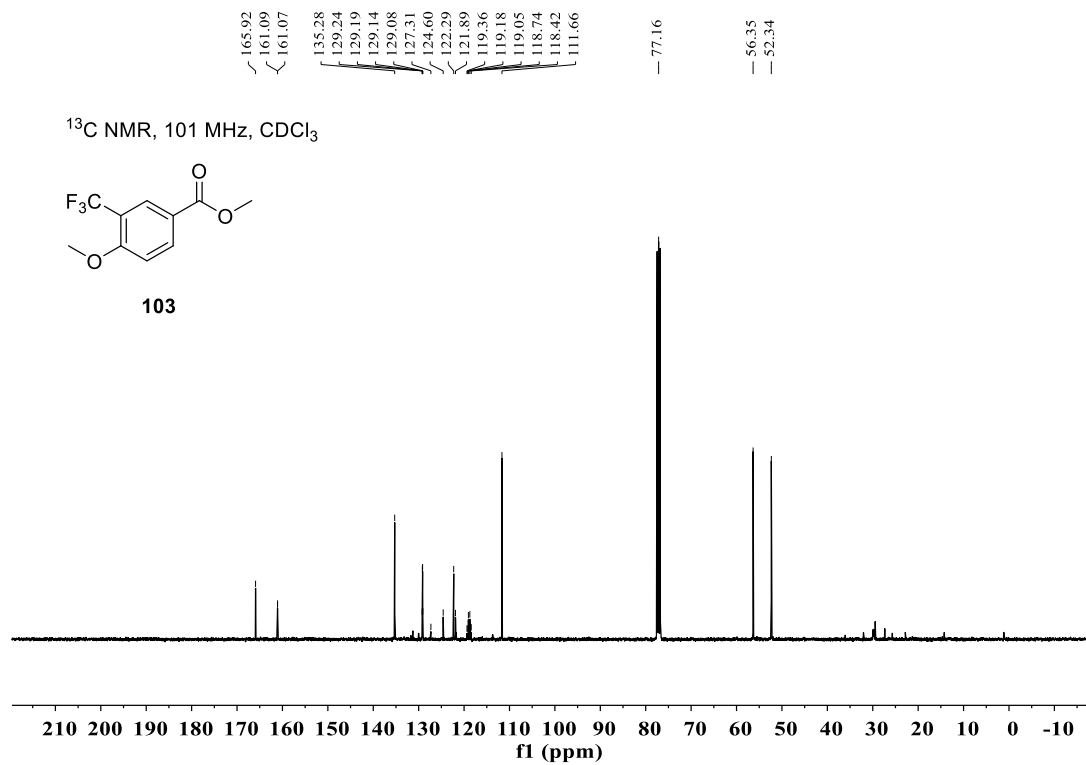












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