

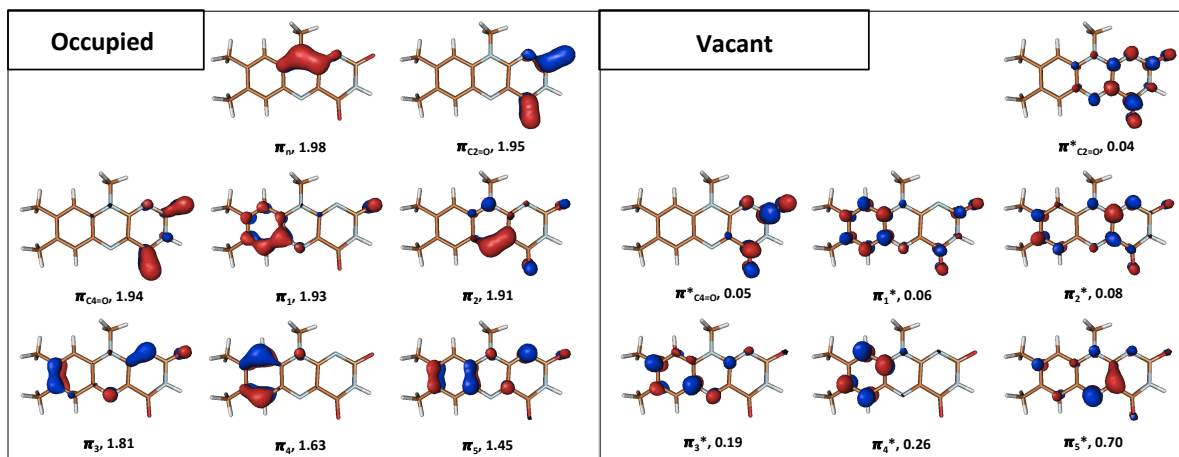
**Supporting Information for:**

**Electronic Structure Methods for Simulating Flavin's Spectroscopy and Photophysics:  
Comparison of Multi-Reference, TD-DFT, and Single-Reference Wave Function Methods**

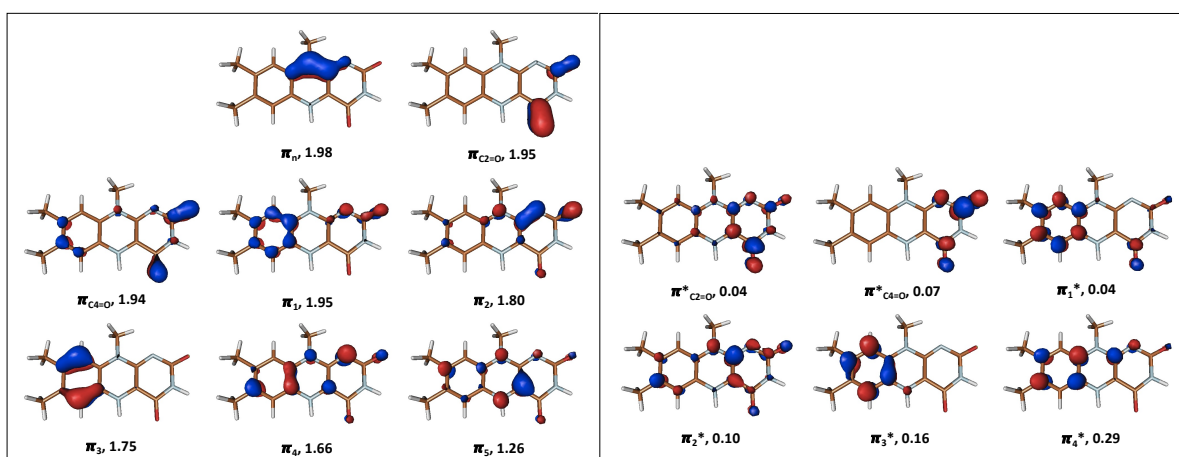
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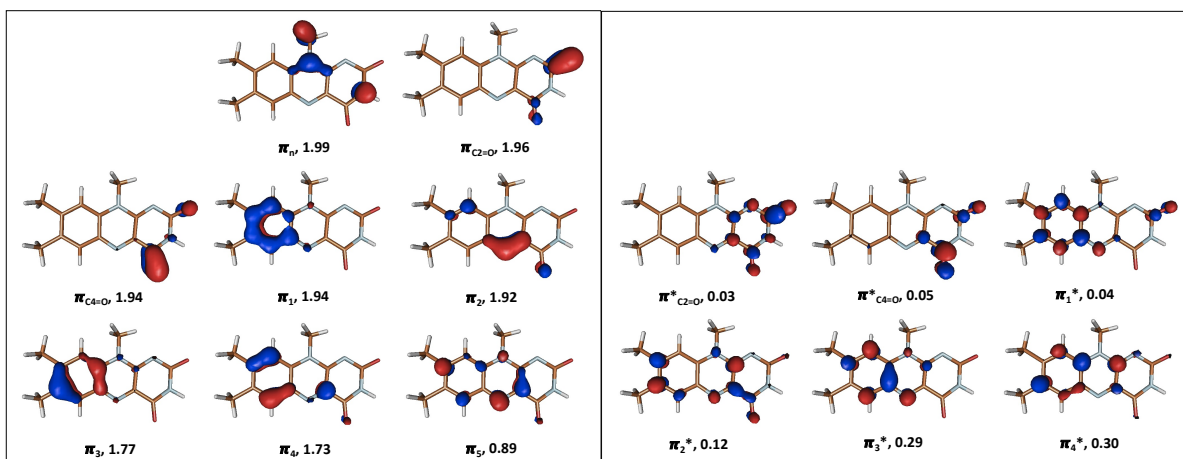
a. FI



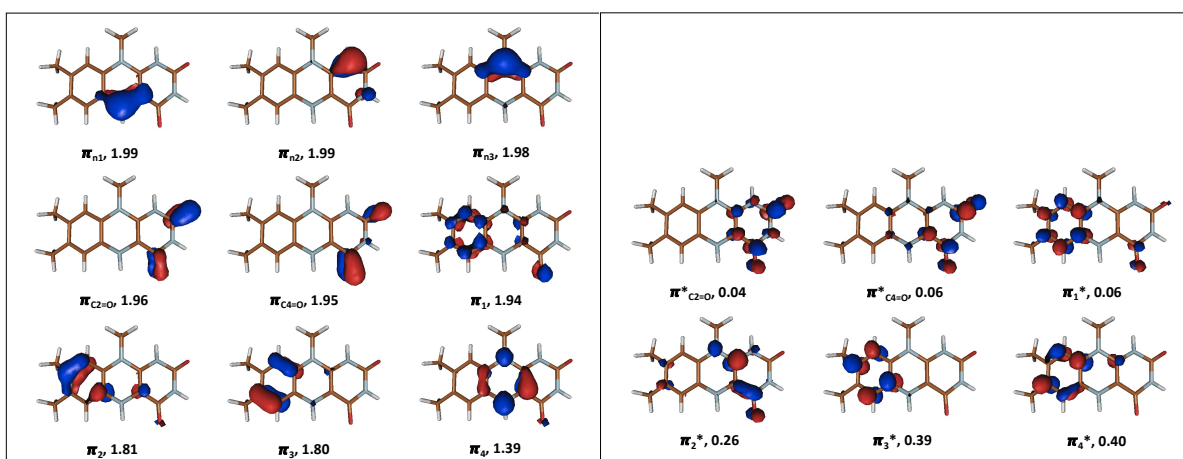
b. FIH<sup>-</sup>



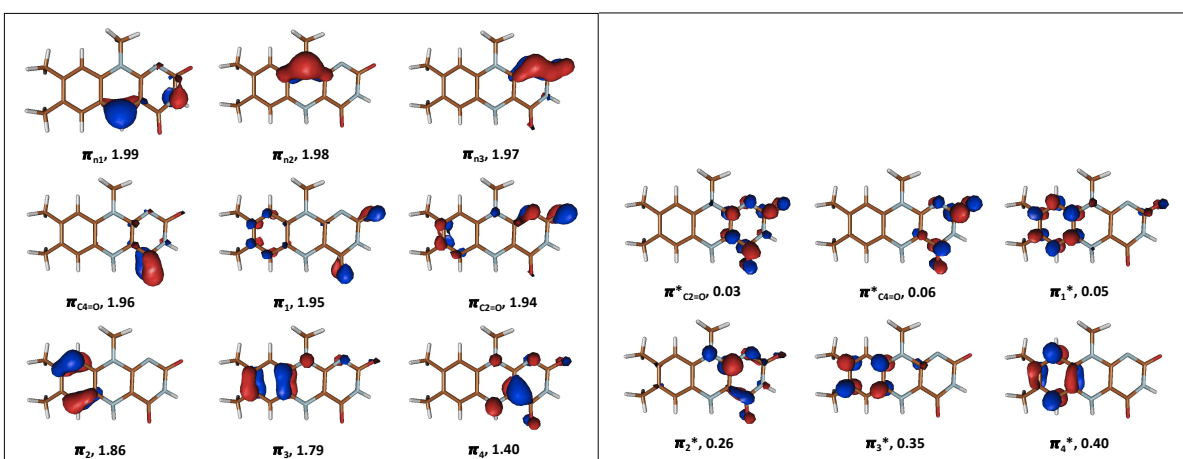
c. FI<sup>-</sup>



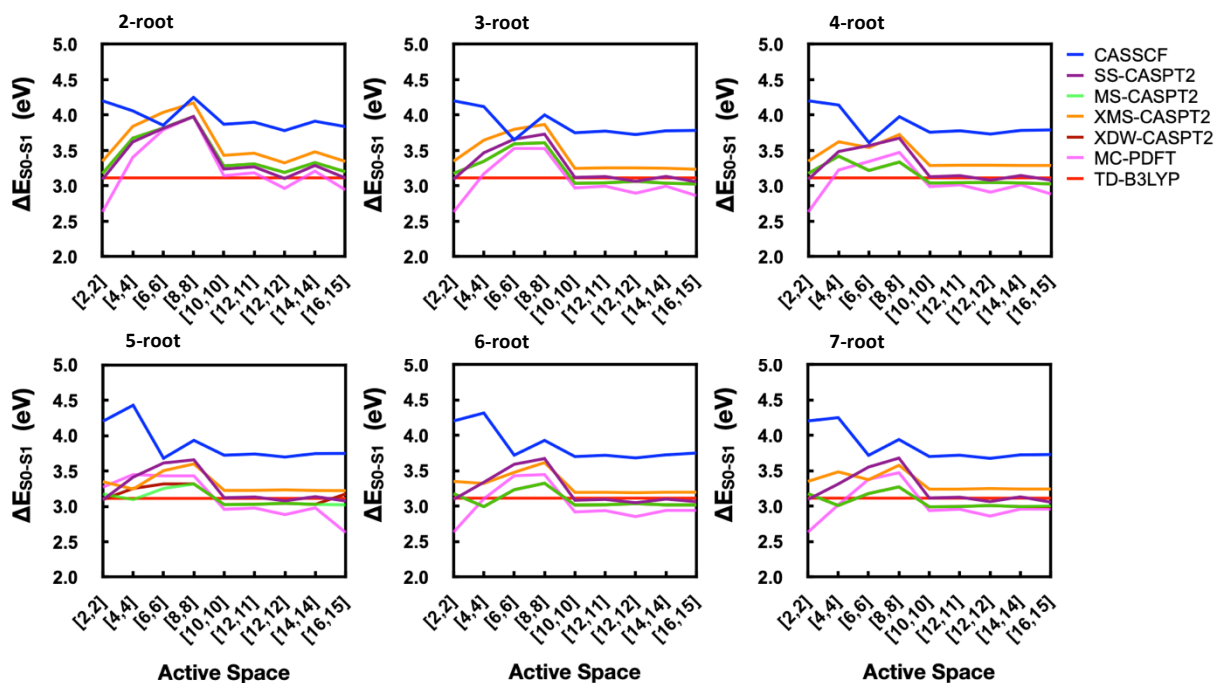
d. FIH<sub>2</sub>



e. FIH<sup>-</sup>



**Figure S1.** Occupied (left) and vacant (right) orbitals used as a starting point for the active space of flavin in each of five redox and protonation states. **a.** oxidized quinone (FI), **b.** neutral semiquinone radical (FIH<sup>•</sup>), **c.** anionic semiquinone radical (FI<sup>-</sup>), **d.** neutral hydroquinone (FIH<sub>2</sub>), and **e.** anionic hydroquinone (FIH<sup>-</sup>).



**Figure S2.** MR-PT2 and CASSCF  $S_1$  excitation energies as a function of active space [number of electrons, number of orbitals] with 2-root averaging, 3-root state averaging, 4-root state averaging, 5-root state averaging, 6-root state averaging, and 7-root state averaging for LF in the oxidized form. The TD-B3LYP excitation energy is shown as a red line for reference.