

## Supplementary Information

### Design, synthesis, and biological evaluation of novel halogenated chlorido[*N,N'*-bis(salicylidene)-1,2-bis(3-methoxyphenyl)ethylenediamine]iron(III) complexes as anticancer agents

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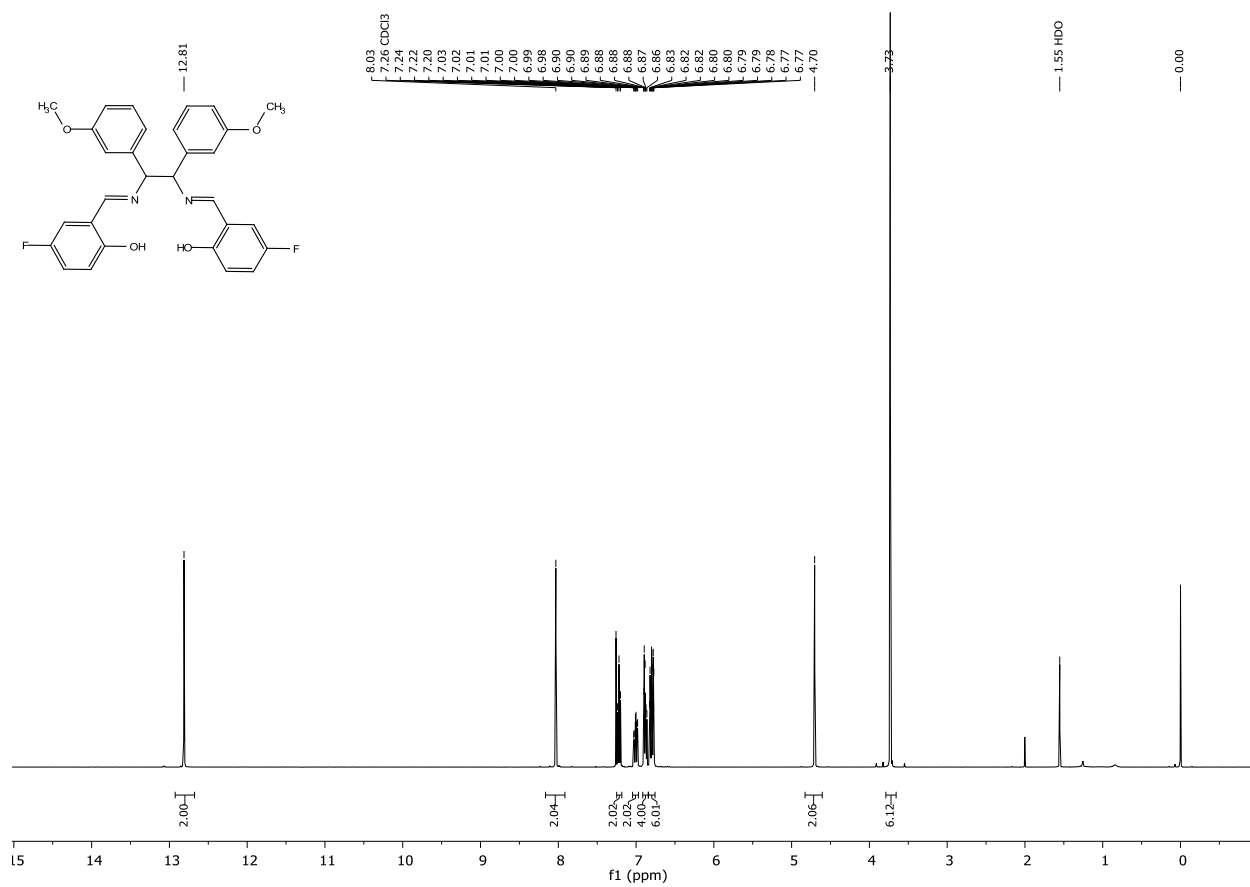
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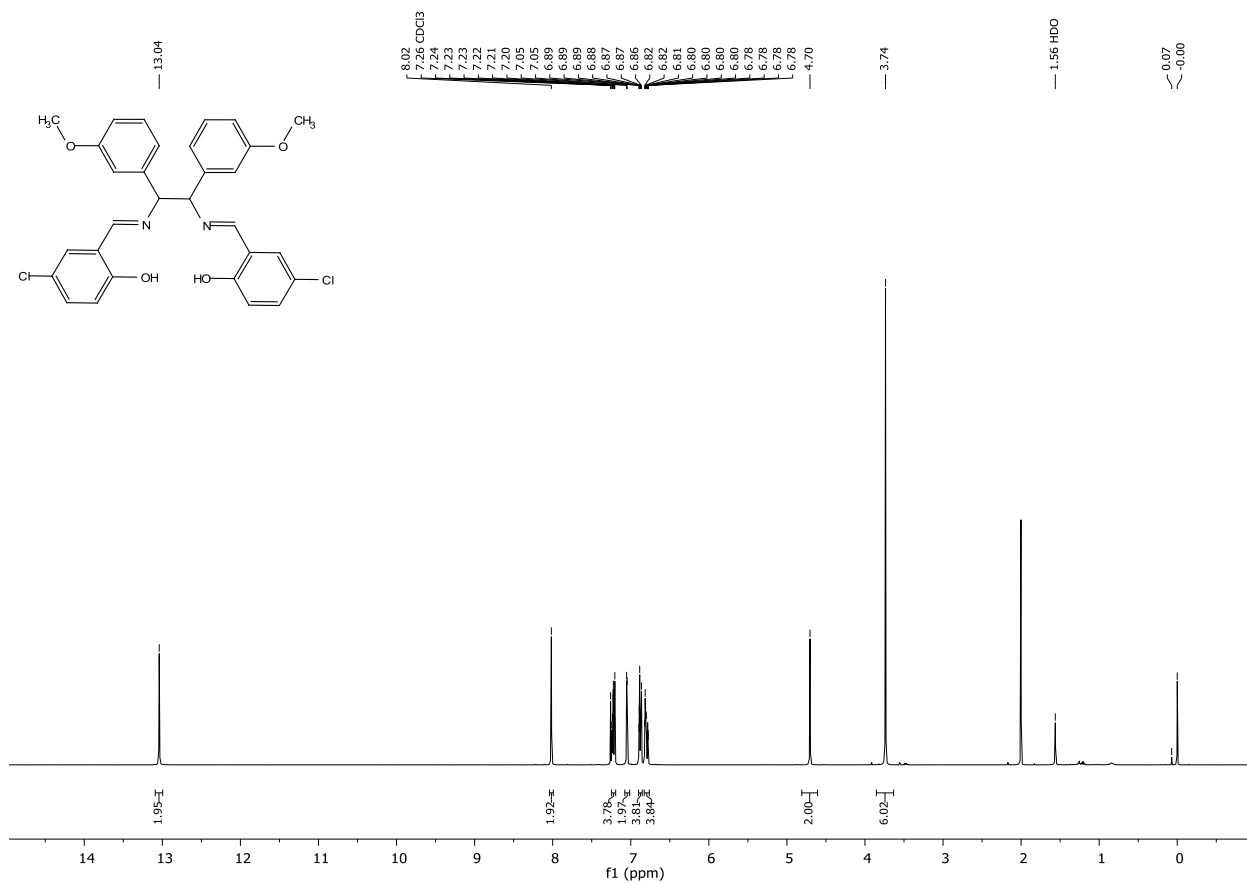
## 1. Characterisation of the ligands L1 – L3

### 1.1. $^1\text{H}$ NMR spectra of the ligands



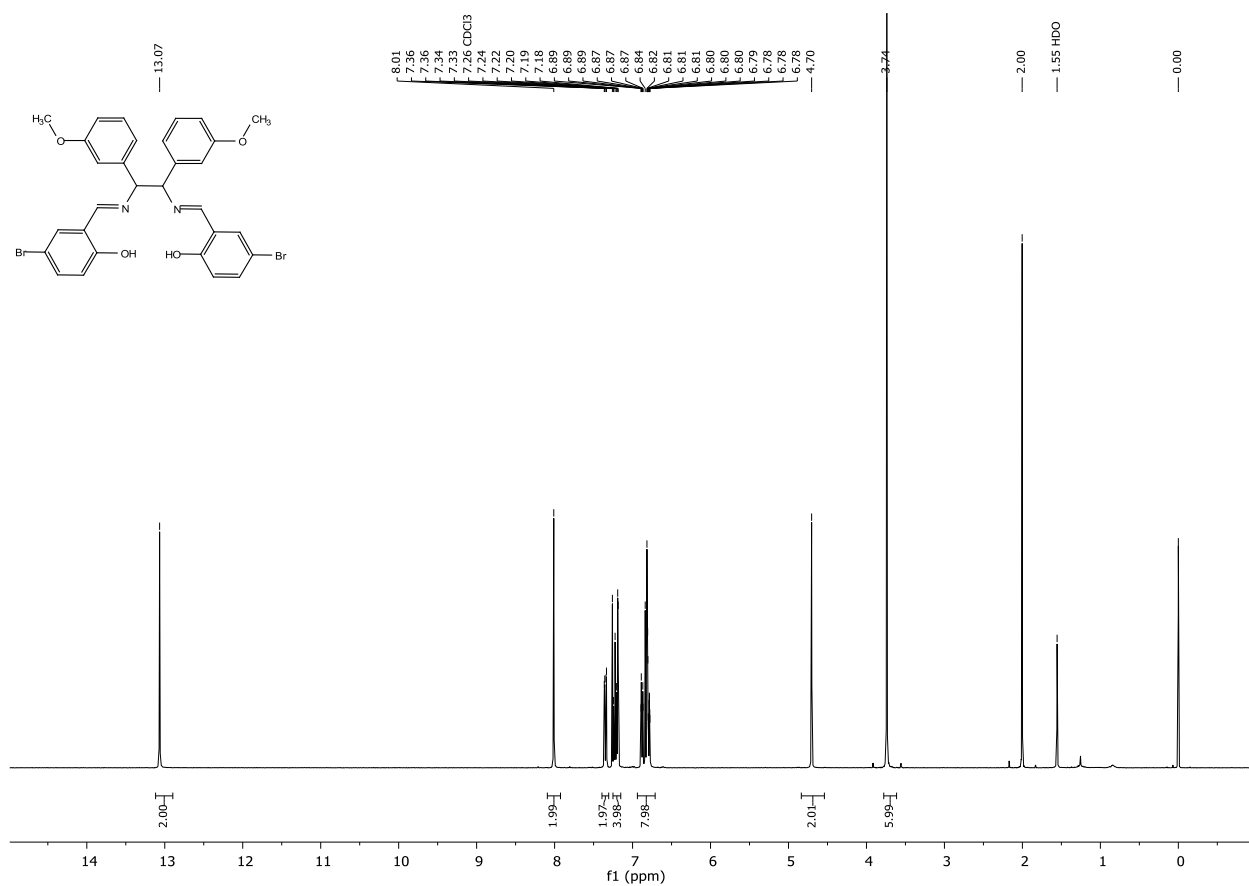
**Figure S1:**  $^1\text{H}$  NMR of L1 in  $\text{CDCl}_3$

$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ -*d*)  $\delta$  12.81 (s, 2H), 8.03 (s, 2H), 7.22 (t,  $J = 7.9$  Hz, 2H), 7.01 (ddd,  $J = 9.1, 8.1, 3.1$  Hz, 2H), 6.92 – 6.84 (m, 4H), 6.84 – 6.75 (m, 6H), 4.70 (s, 2H), 3.73 (s, 6H)



**Figure S2:** <sup>1</sup>H NMR of **L2** in CDCl<sub>3</sub>

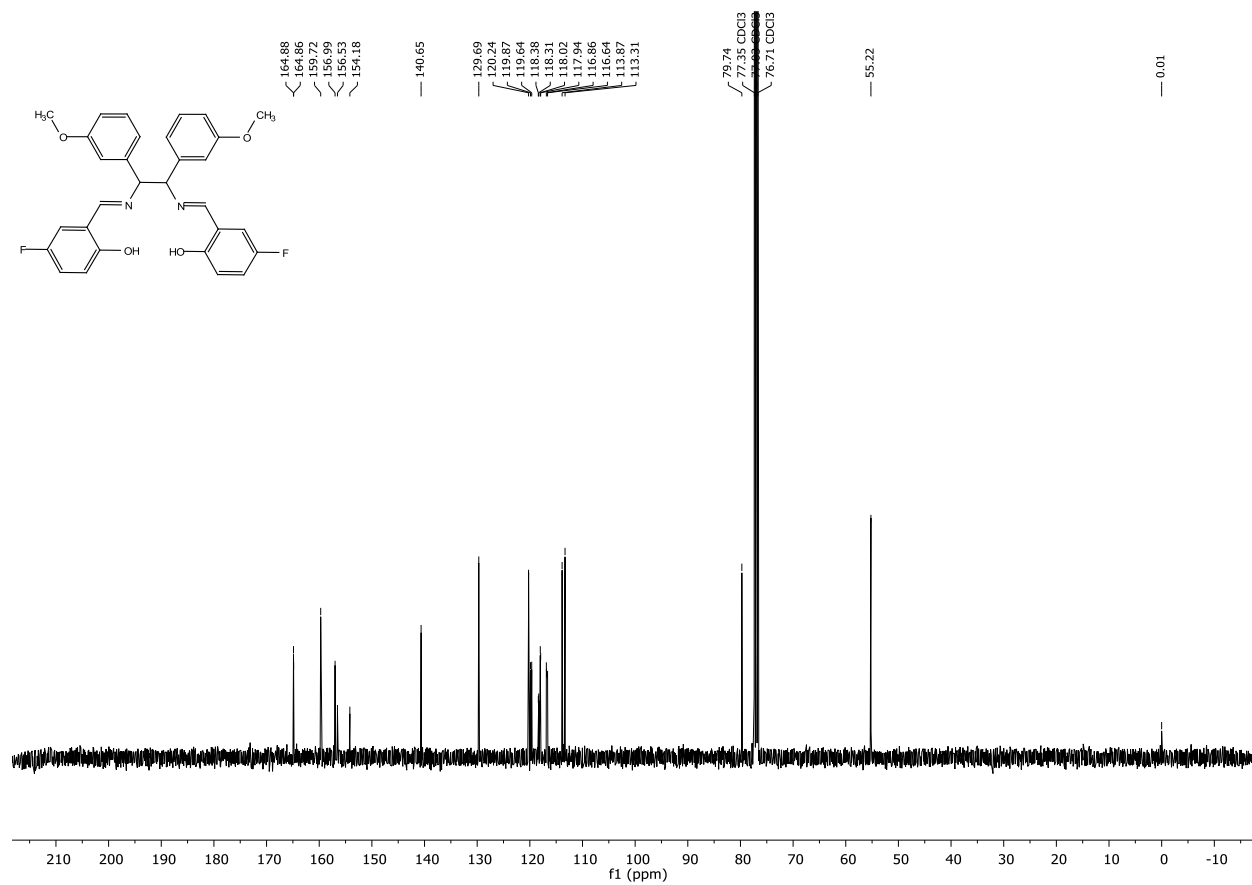
<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>-d) δ 13.04 (s, 2H), 8.02 (s, 2H), 7.27 – 7.18 (m, 4H), 7.05 (d, *J* = 2.5 Hz, 2H), 6.92 – 6.84 (m, 4H), 6.84 – 6.75 (m, 4H), 4.70 (s, 2H), 3.74 (s, 6H)



**Figure S3:**  $^1\text{H}$  NMR of **L3** in  $\text{CDCl}_3$

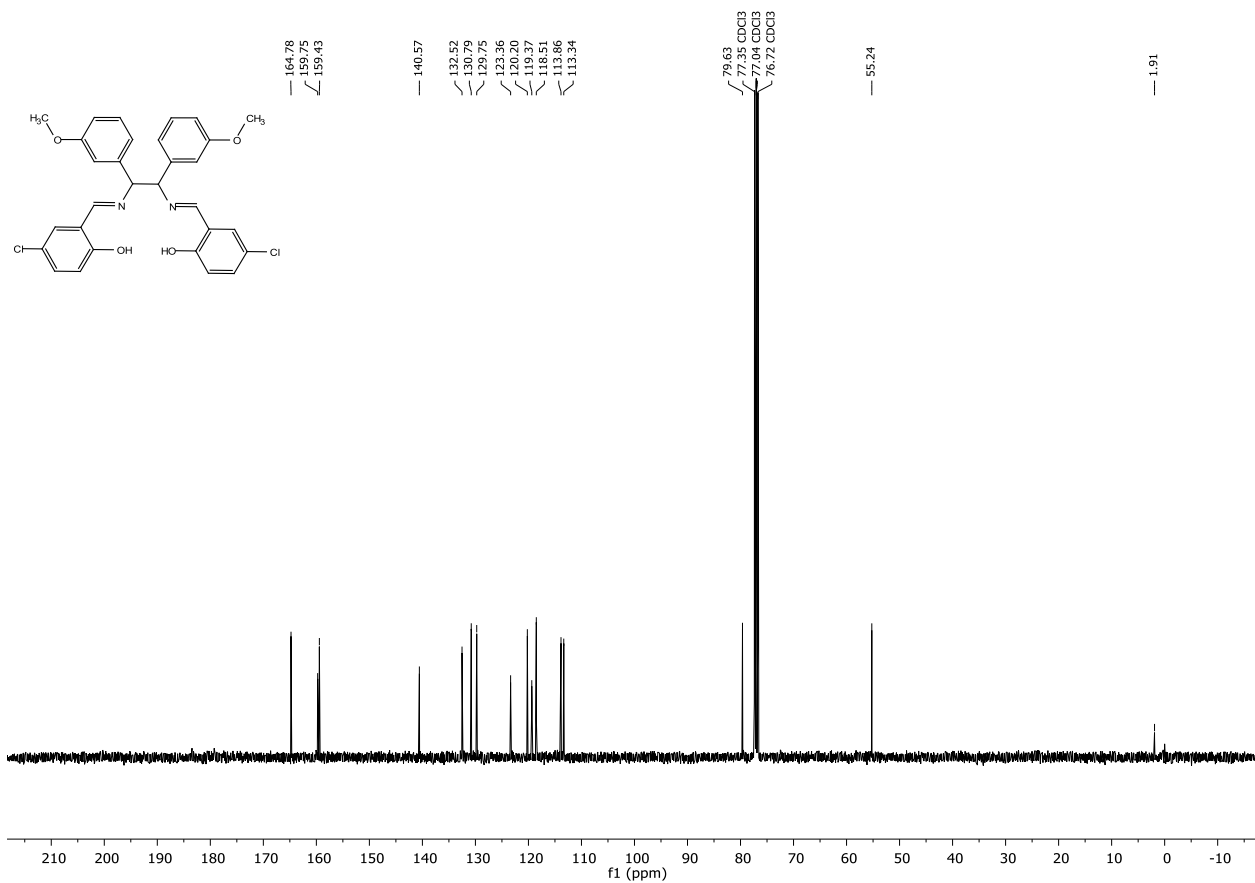
$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ -*d*)  $\delta$  13.07 (s, 2H), 8.01 (s, 2H), 7.35 (dd,  $J = 8.8, 2.5$  Hz, 2H), 7.27 – 7.16 (m, 4H), 6.91 – 6.75 (m, 8H), 4.70 (s, 2H), 3.74 (s, 6H)

## 1.2. $^{13}\text{C}$ NMR spectra of the ligands



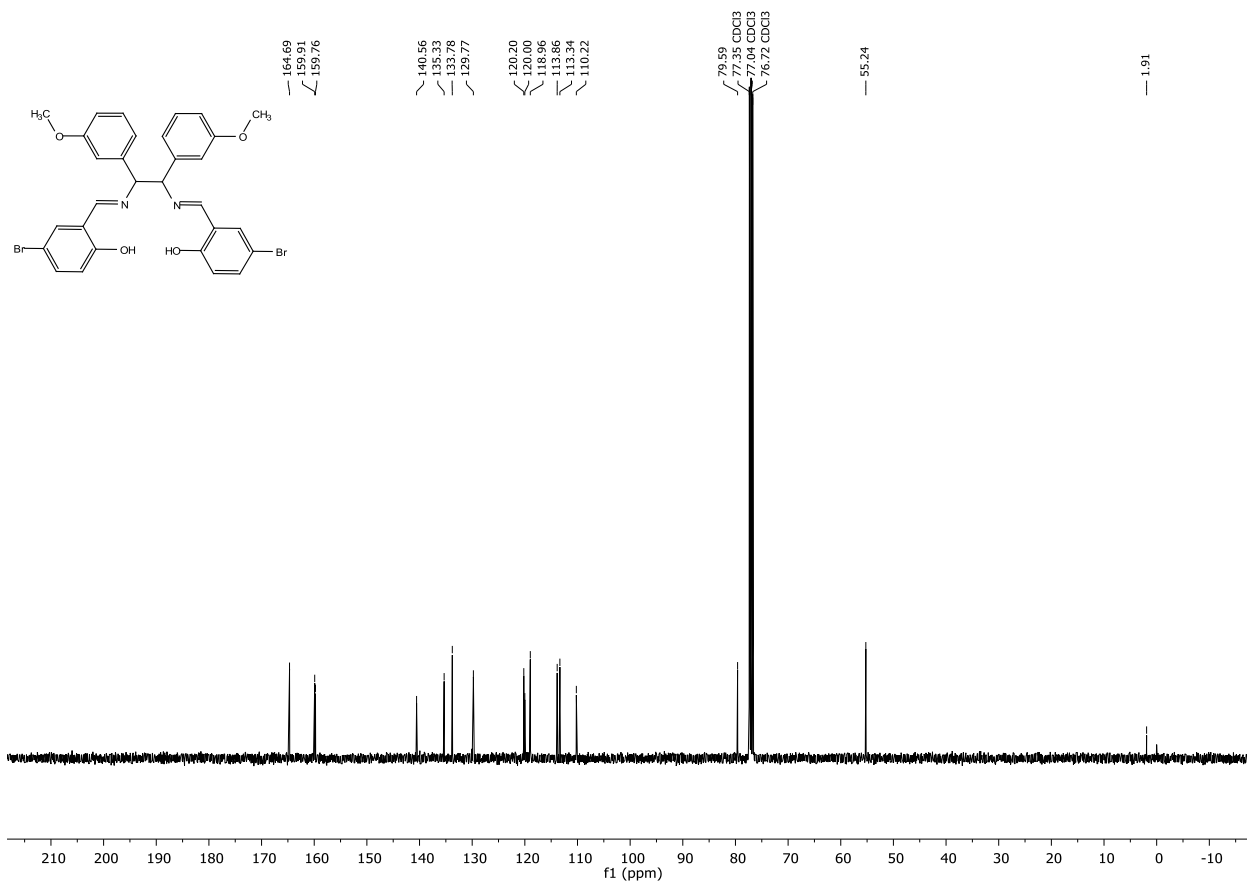
**Figure S4:**  $^{13}\text{C}$  NMR of **L1** in  $\text{CDCl}_3$

$^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ -*d*)  $\delta$  164.87 (d,  $J = 2.7$  Hz), 161.75 – 152.15 (m), 140.65, 129.69, 120.24, 119.76 (d,  $J = 23.2$  Hz), 118.16 (dd,  $J = 37.0, 7.3$  Hz), 116.75 (d,  $J = 22.8$  Hz), 113.87, 113.31, 79.74, 55.22



**Figure S5:** <sup>13</sup>C NMR of L2 in CDCl<sub>3</sub>

<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>-d) δ 164.78, 159.75, 159.43, 140.57, 132.52, 130.79, 129.75, 123.36, 120.20, 119.37, 118.51, 113.86, 113.34, 79.63, 55.24

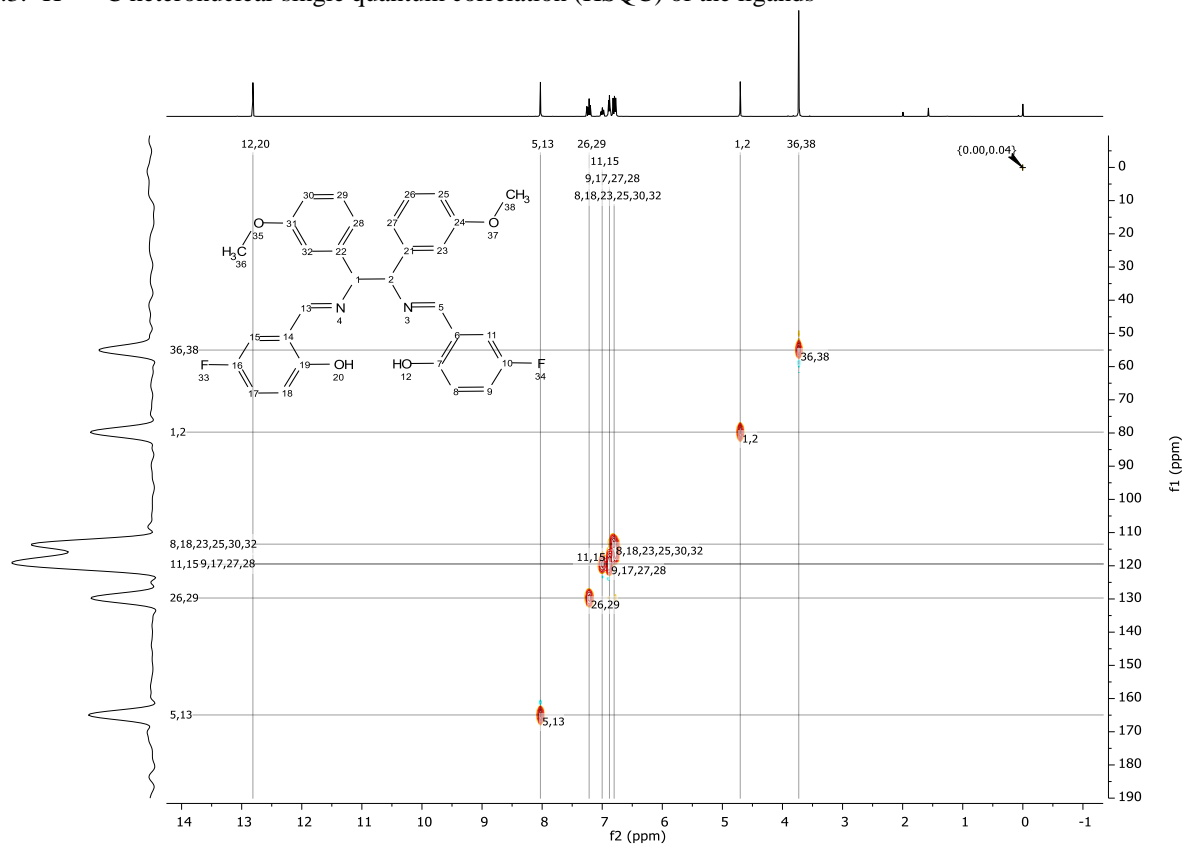


**Figure S6:** <sup>13</sup>C NMR of L3 in CDCl<sub>3</sub>

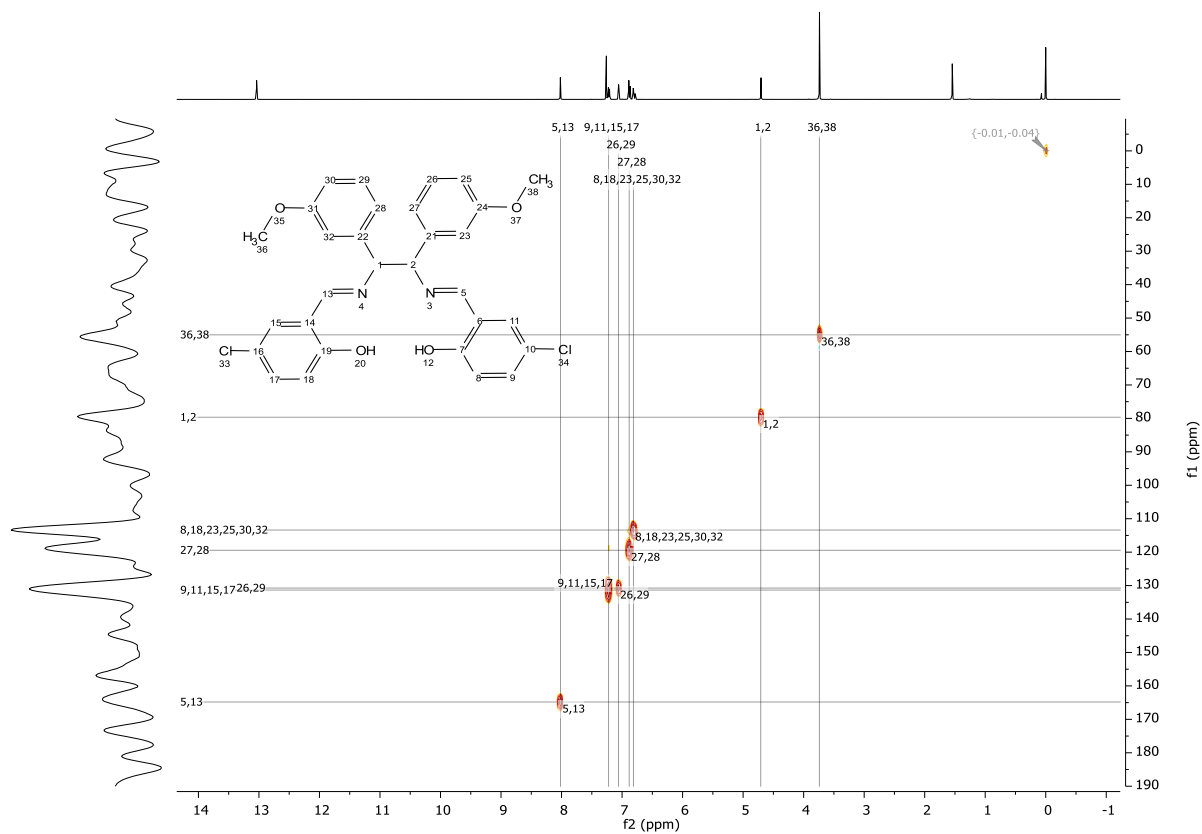
<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>-d) δ 164.69, 159.91, 159.76, 140.56, 135.33, 133.78, 129.77, 120.20, 120.00, 118.96, 113.86, 113.34, 110.22, 79.59, 55.24



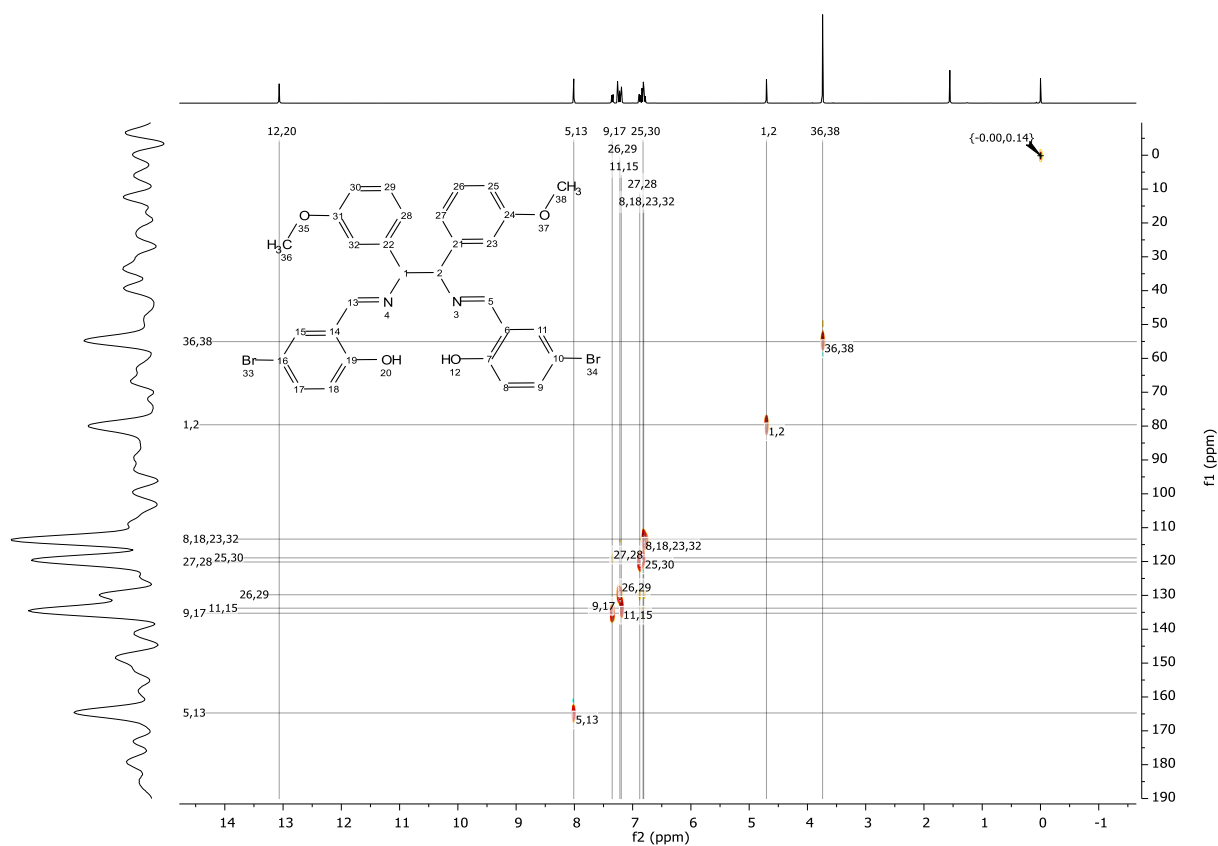
### 1.3. $^1\text{H} - ^{13}\text{C}$ heteronuclear single quantum correlation (HSQC) of the ligands



**Figure S7:**  $^1\text{H} - ^{13}\text{C}$  HSQC spectrum of L1 in  $\text{CDCl}_3$

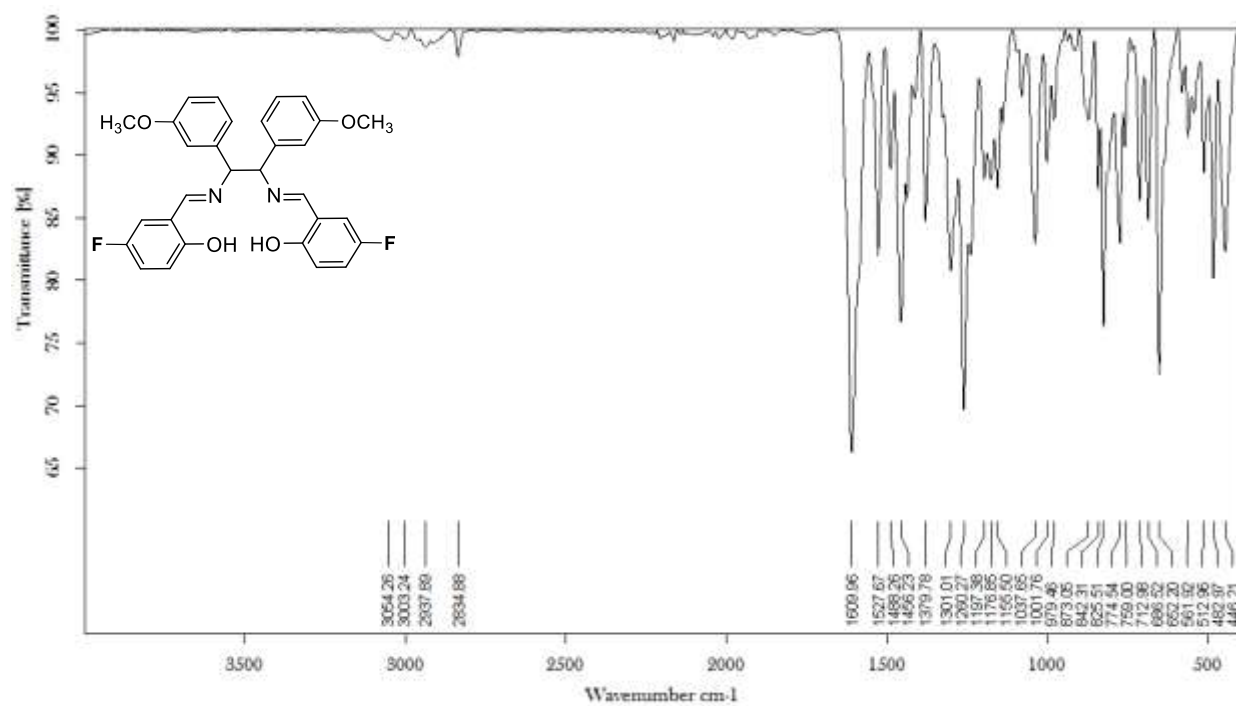


**Figure S8:**  $^1\text{H} - ^{13}\text{C}$  HSQC spectrum of L2 in  $\text{CDCl}_3$



**Figure S9:**  $^1\text{H}$  –  $^{13}\text{C}$  HSQC spectrum of L3 in  $\text{CDCl}_3$

#### 1.4. FT-IR spectra of the ligands



**Figure S10:** FT-IR spectrum of the neat L1

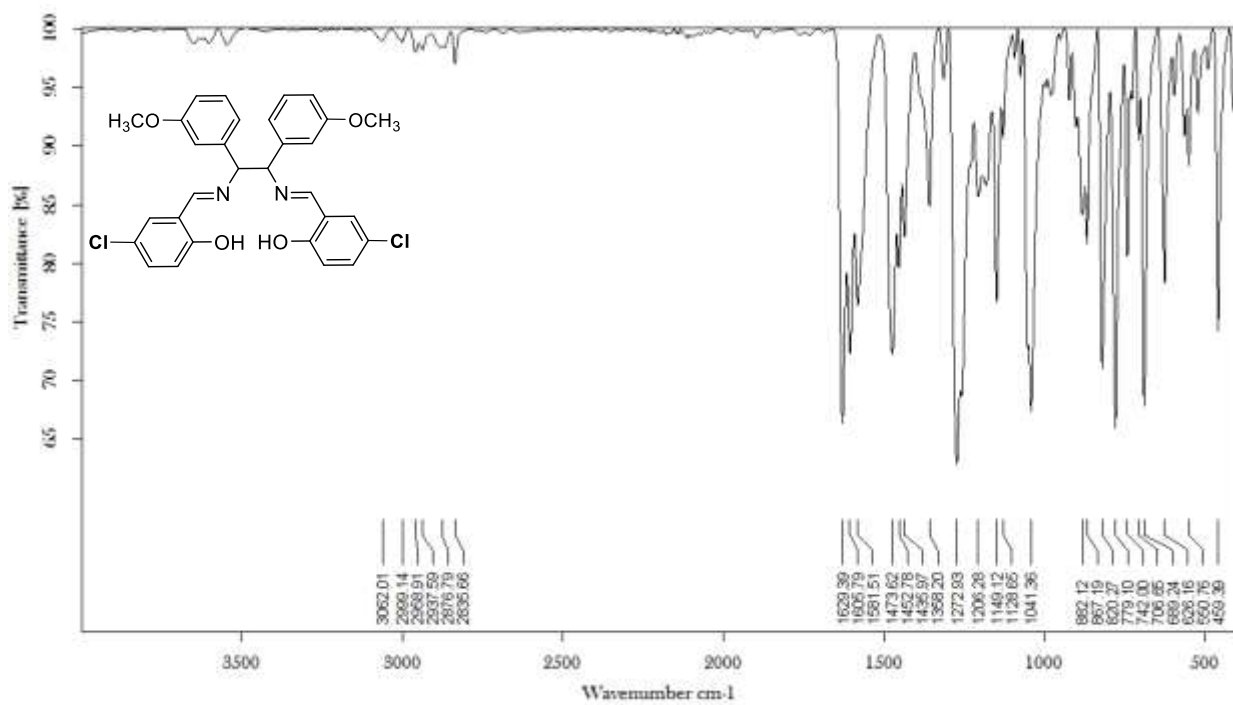


Figure S11: FT-IR spectrum of the neat L2

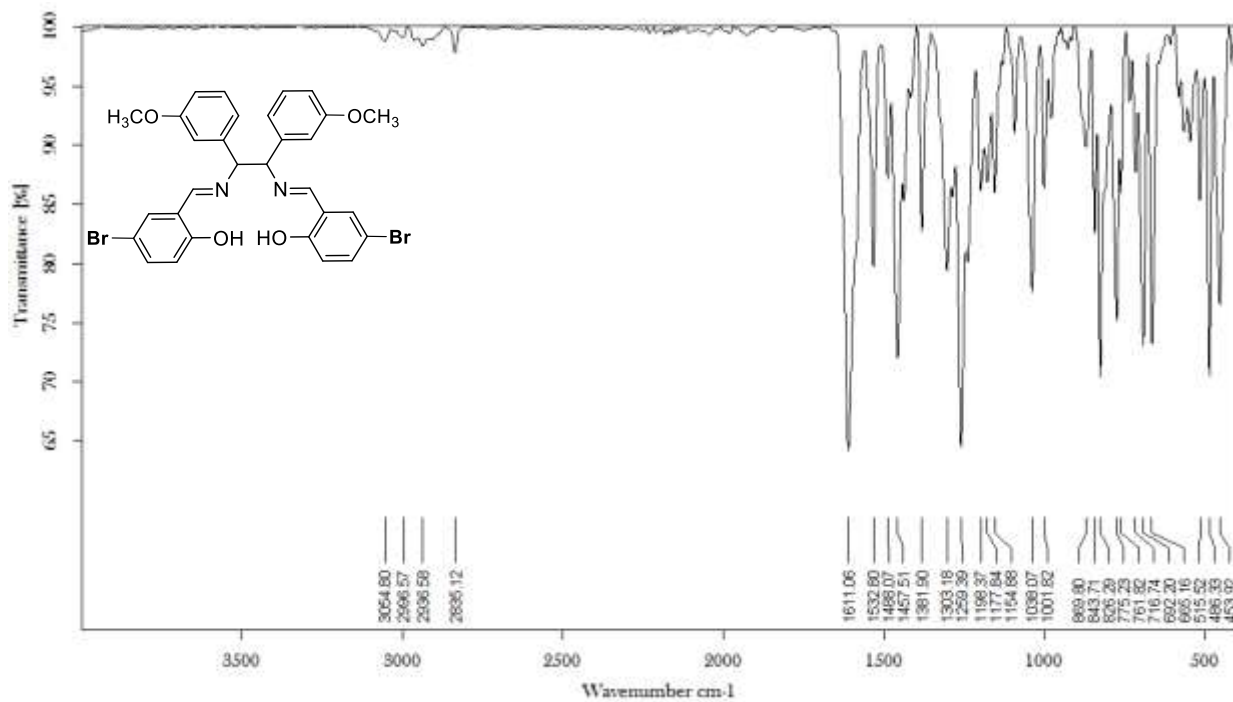
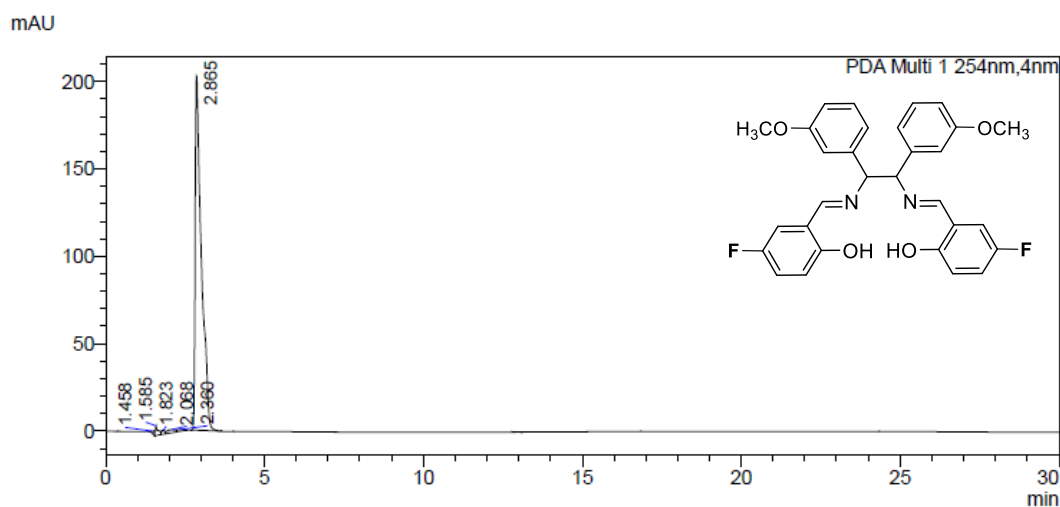


Figure S12: FT-IR spectrum of the neat L3

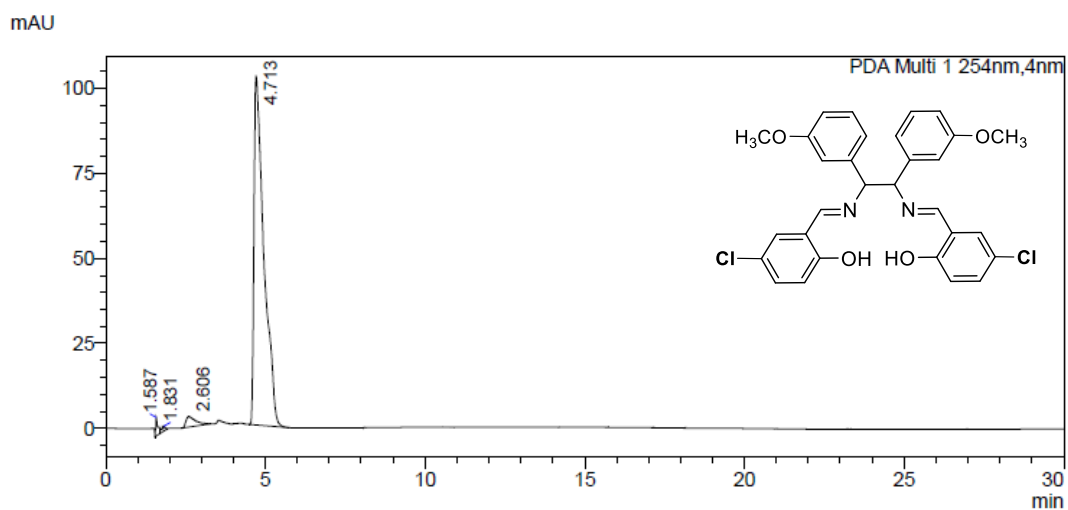
## 1.5. HPLC chromatograms of the ligands

**Table S1: HPLC data L1 – L3**

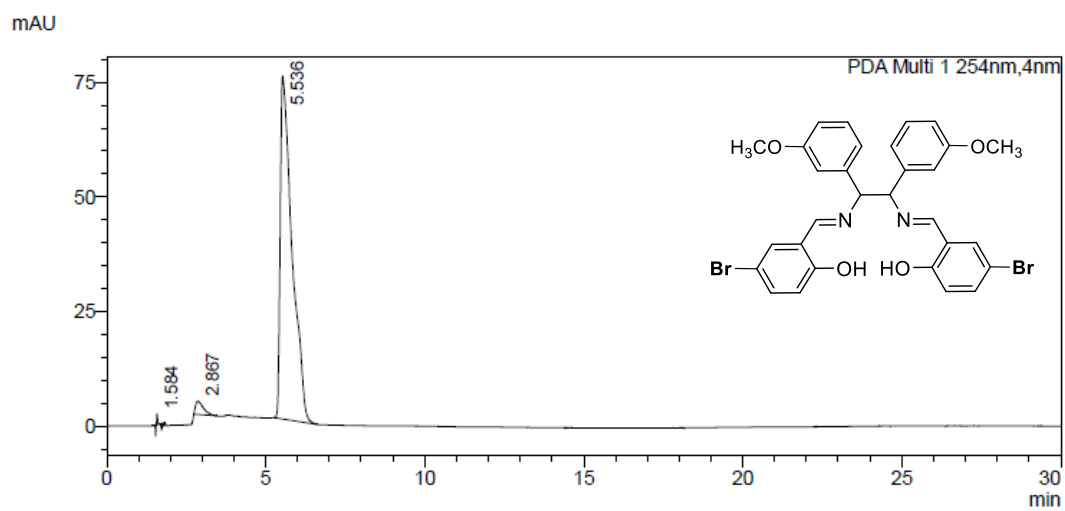
Nr.	Ret. Time in Min.	Area %
L1	2.865	96.378
L2	4.713	95.724
L3	5.536	95.852



**Figure S13: HPLC chromatogram of L1**



**Figure S14: HPLC chromatogram of L2**



**Figure S15:** HPLC chromatogram of L3

## 2. Characterisation of the complexes X1 – X3

### 2.1. FT-IR spectra of the complexes

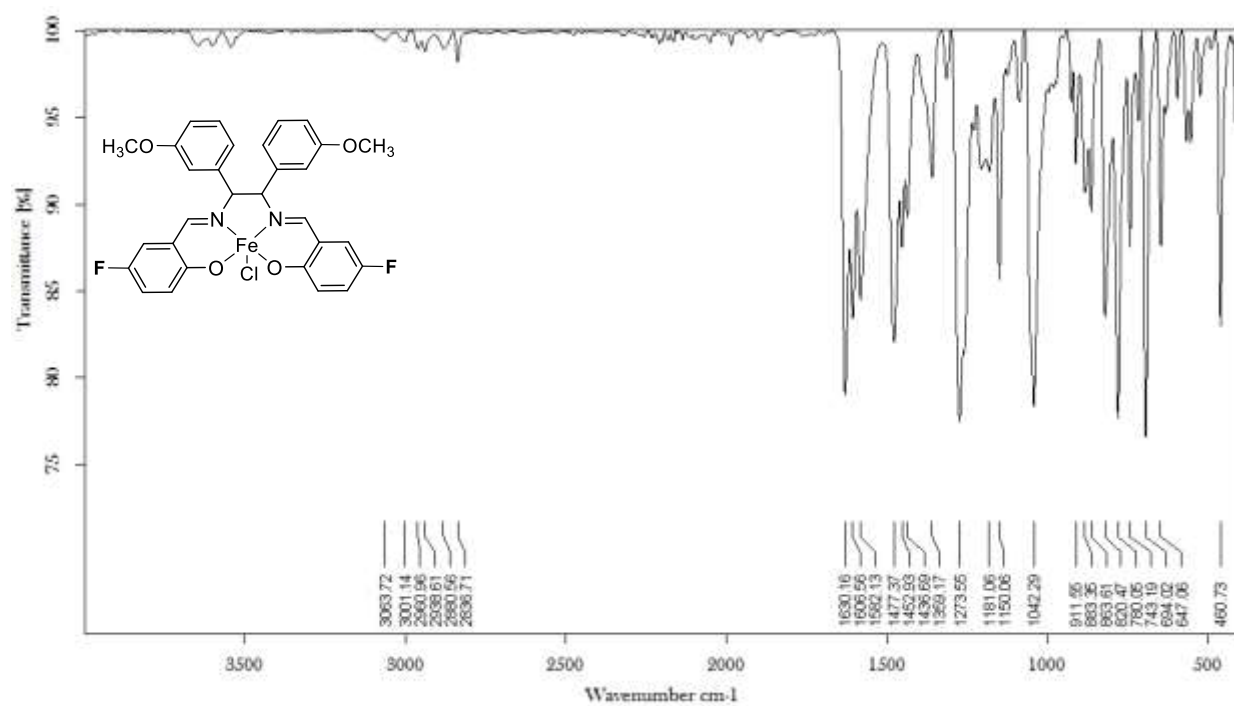


Figure S16: FT-IR spectrum of the neat X1

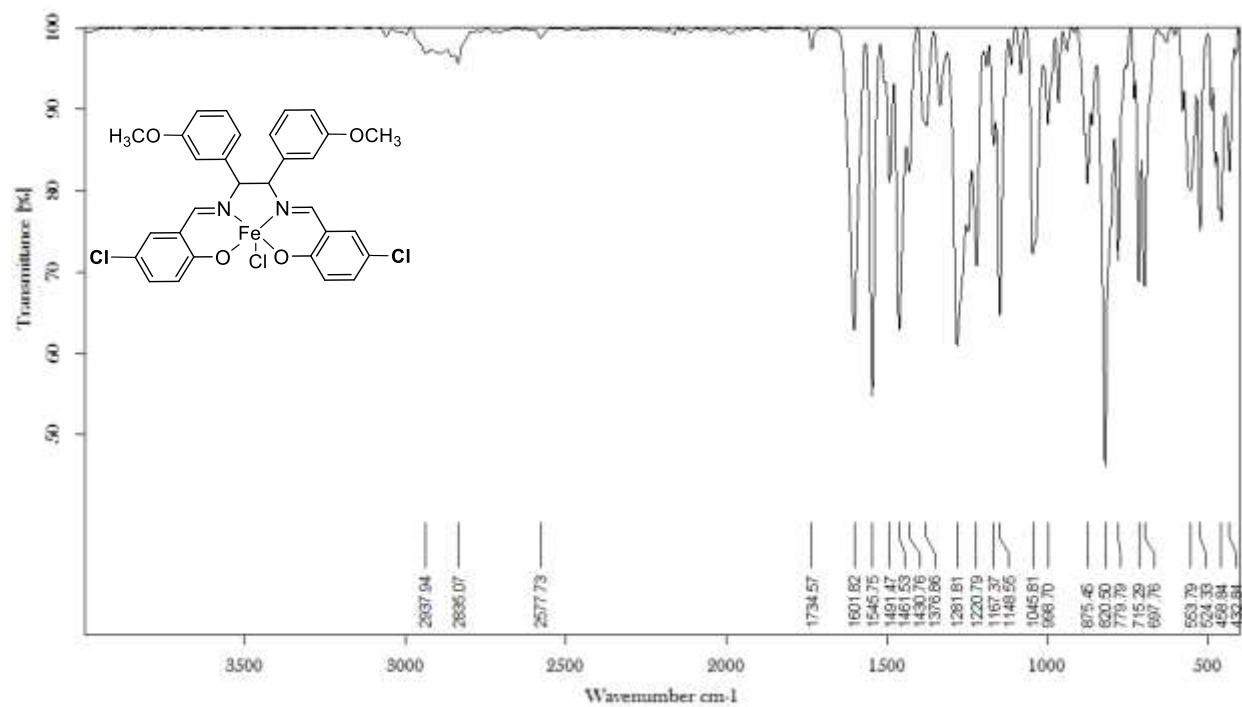


Figure S17: FT-IR spectrum of the neat X2

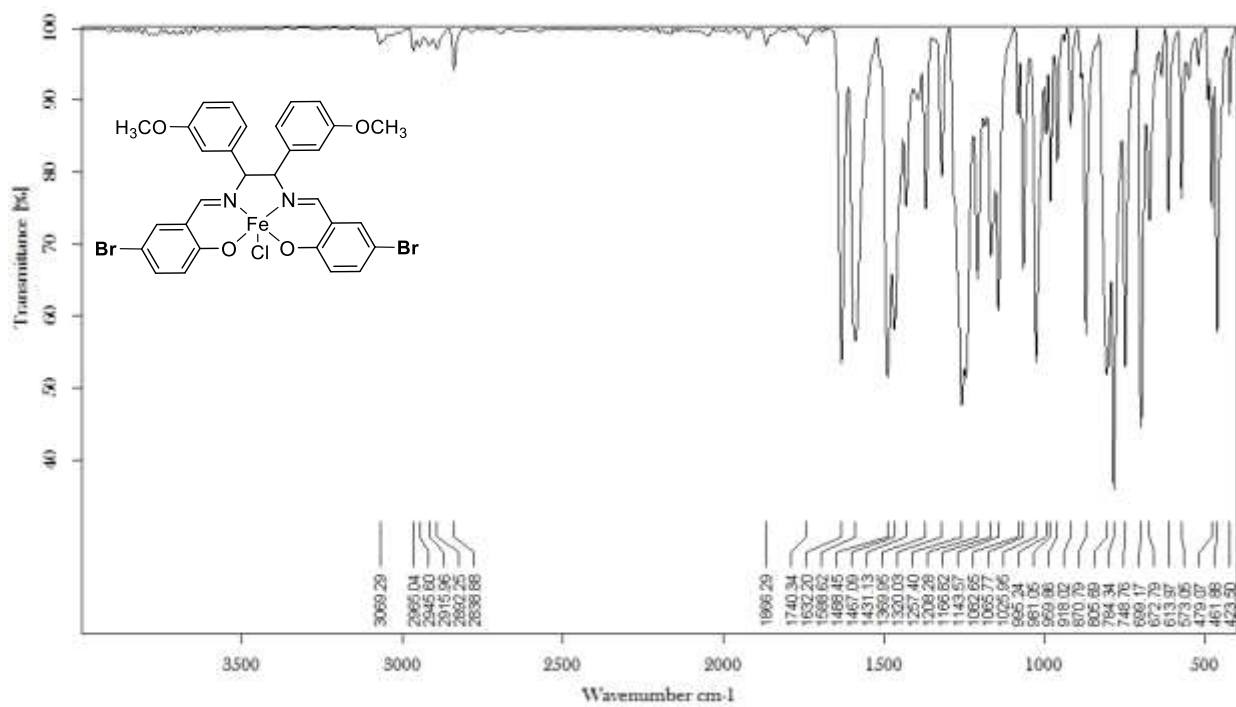
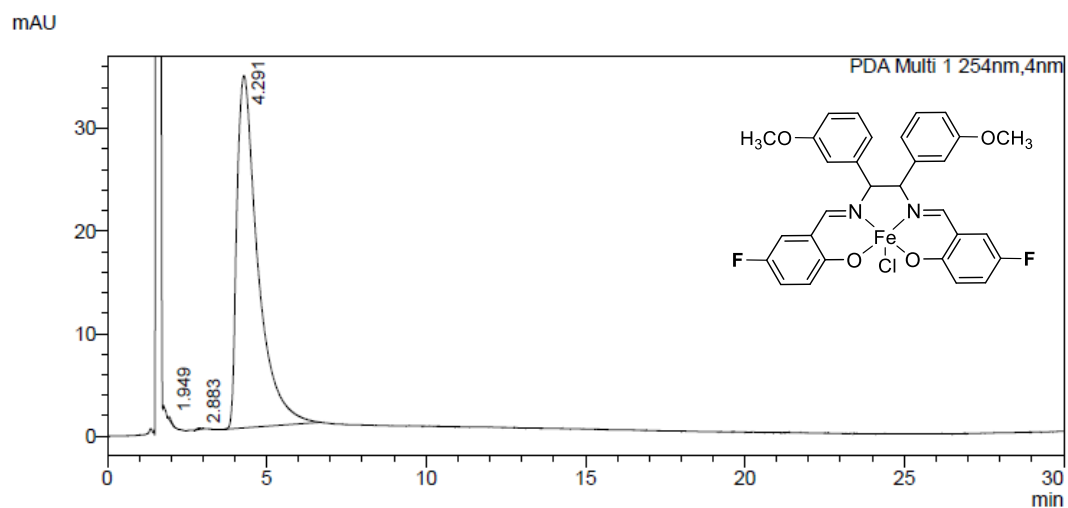


Figure S18: FT-IR spectrum of the neat X3

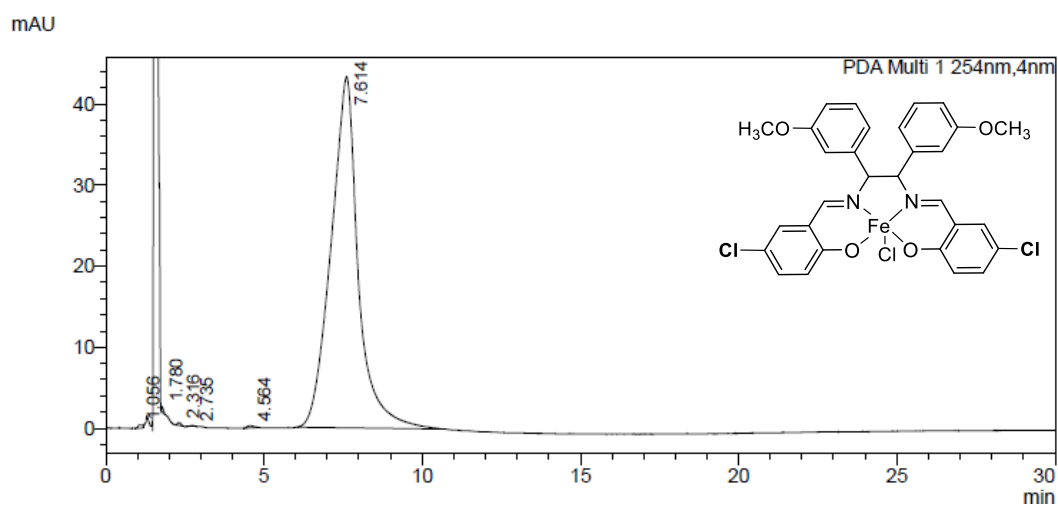
## 2.2. HPLC chromatograms of the complexes

**Table S2: HPLC data X1 – X3**

Nr.	Ret. Time in Min.	Area %
<b>X1</b>	4.291	99.933
<b>X2</b>	7.614	99.794
<b>X3</b>	9.634	99.425

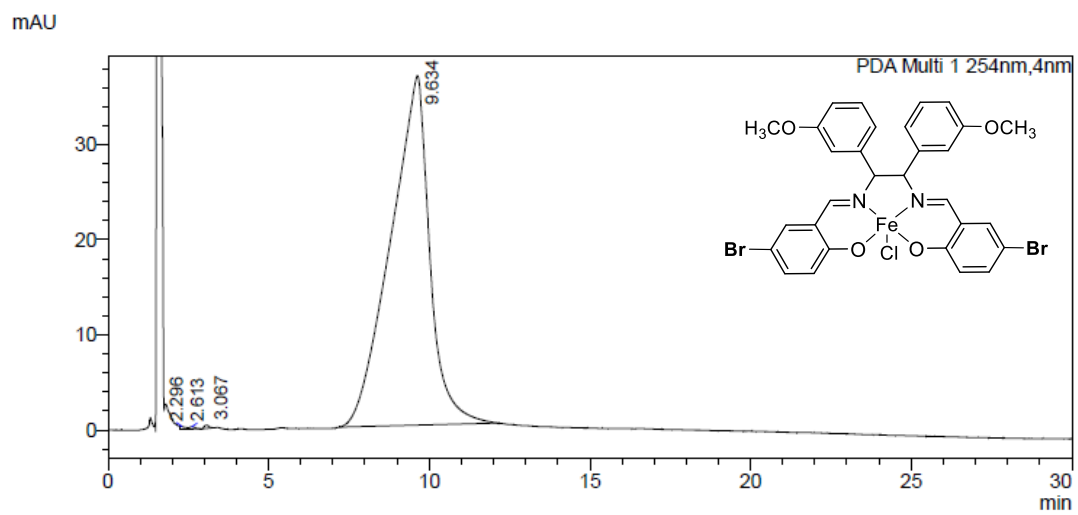


**Figure S19: HPLC chromatogram of X1**



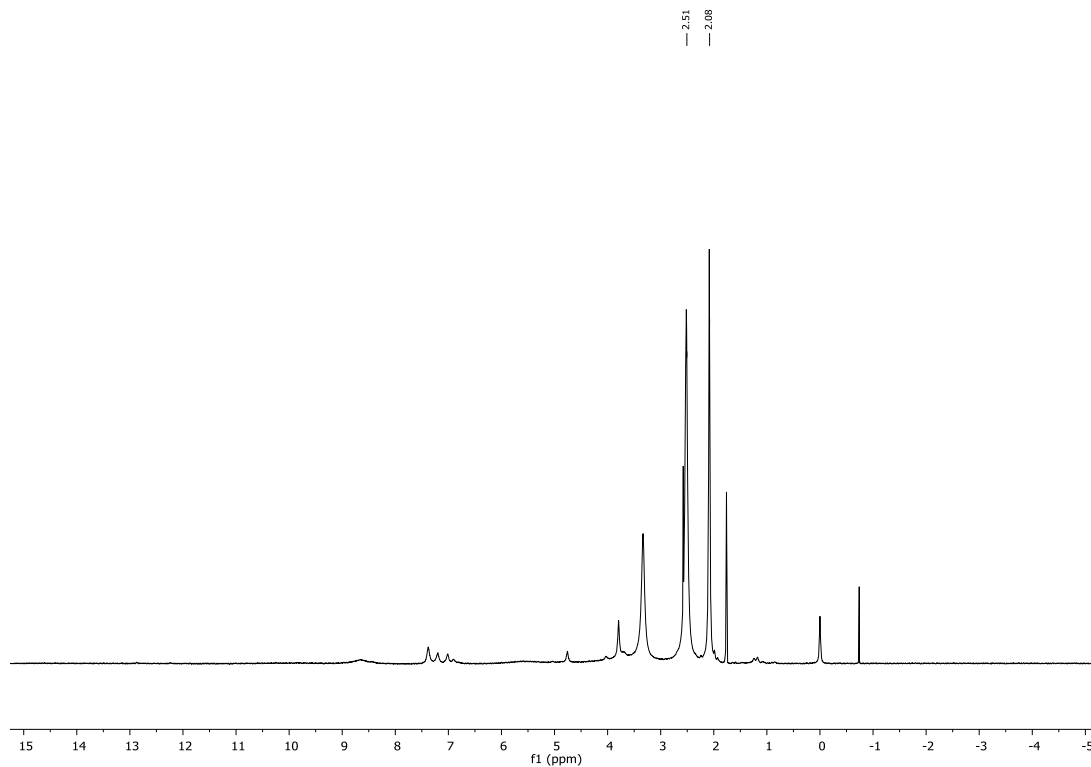
**Figure S20: HPLC chromatogram of X2**



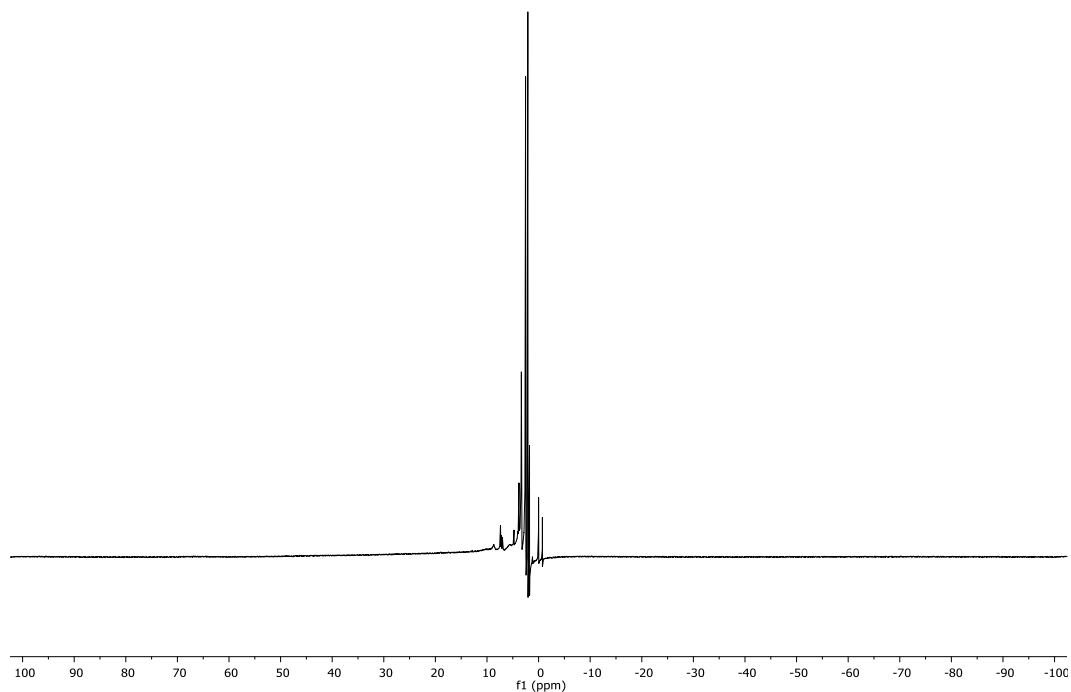


**Figure S21:** HPLC chromatogram of **X3**

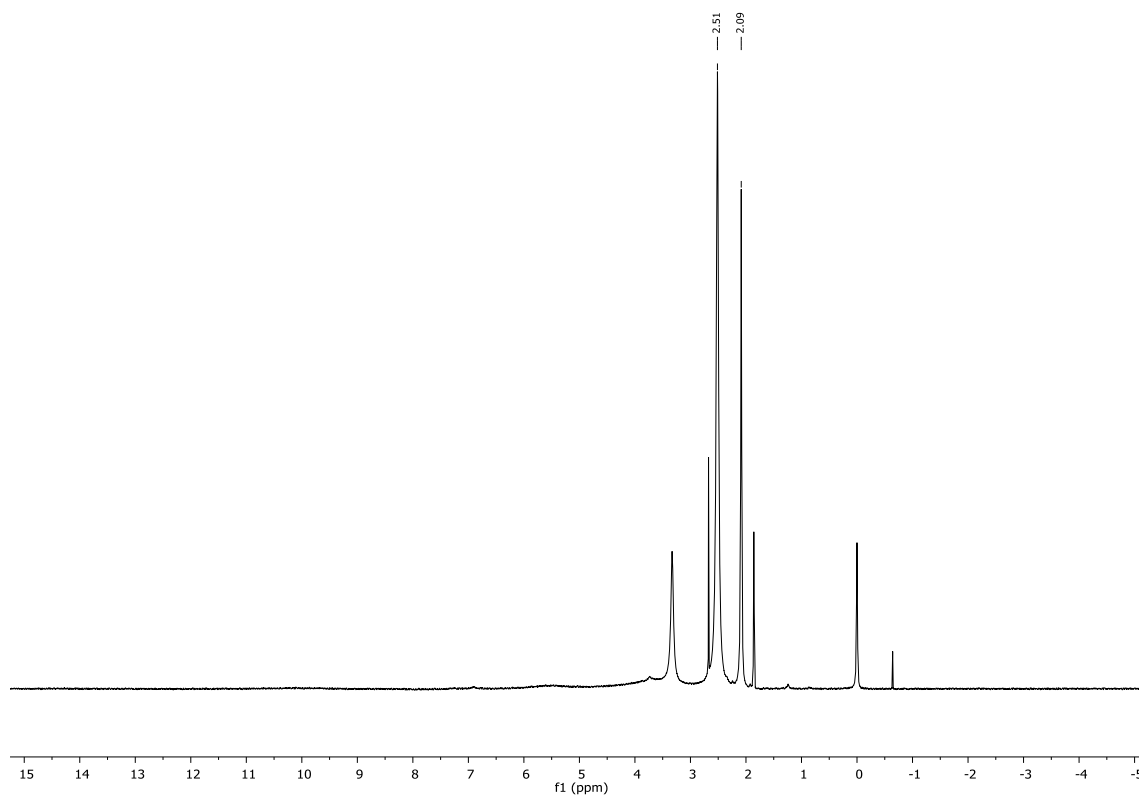
### 2.3. NMR spectra with Evans method of the complexes



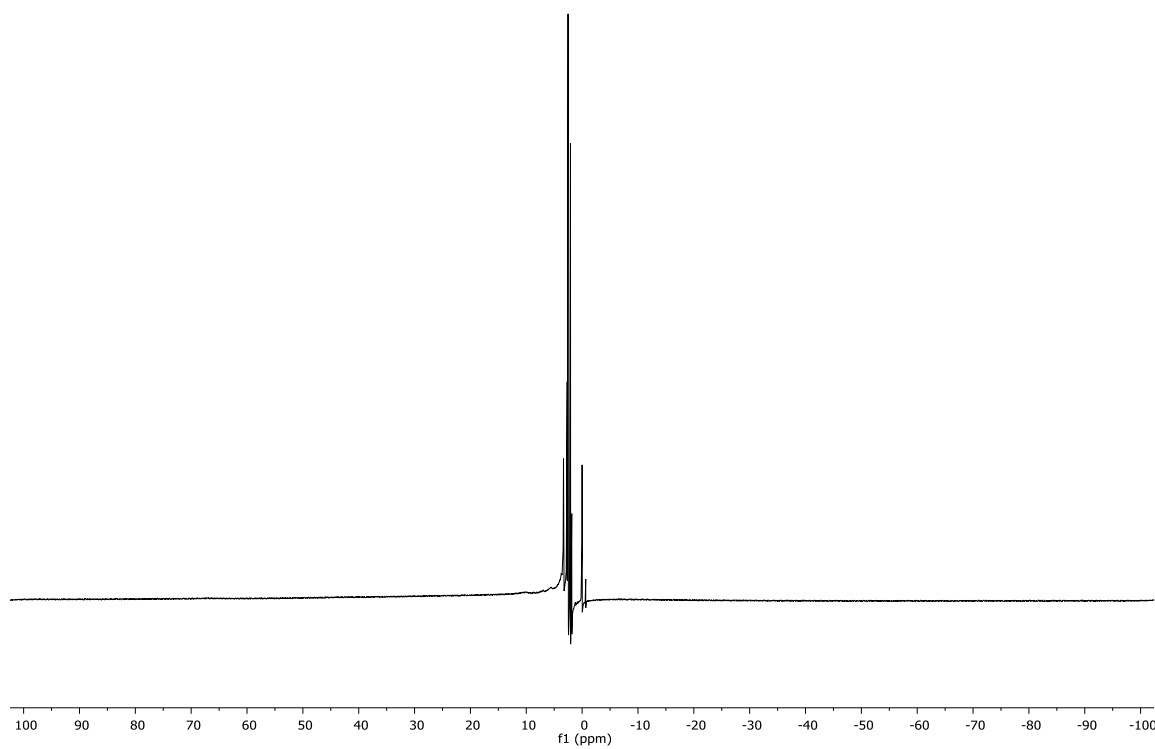
**Figure S22:** <sup>1</sup>H NMR spectrum of 5 mg of paramagnetic **X1** in DMSO-*d*<sub>6</sub> at 298 K with a DMSO-*d*<sub>6</sub> capillary. The shifted and unshifted solvent peaks are highlighted.



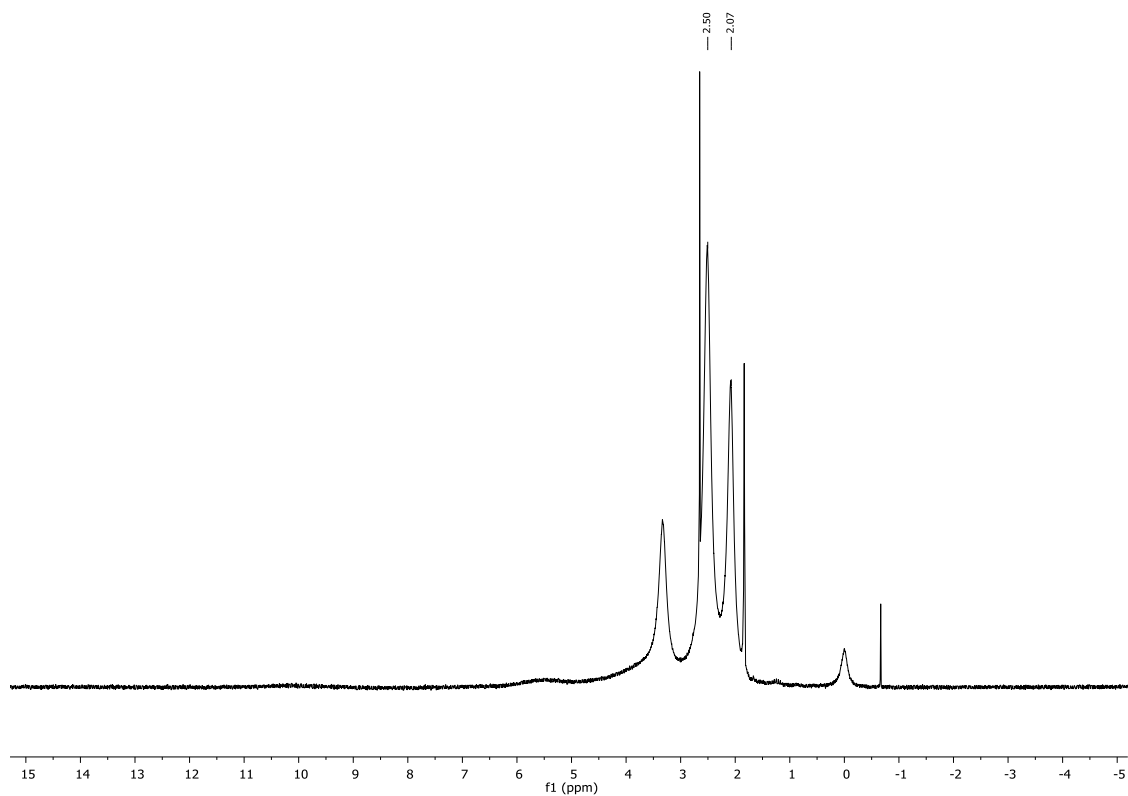
**Figure S23:** <sup>1</sup>H NMR spectrum of 5 mg of paramagnetic **X1** in DMSO-*d*<sub>6</sub> at 298 K



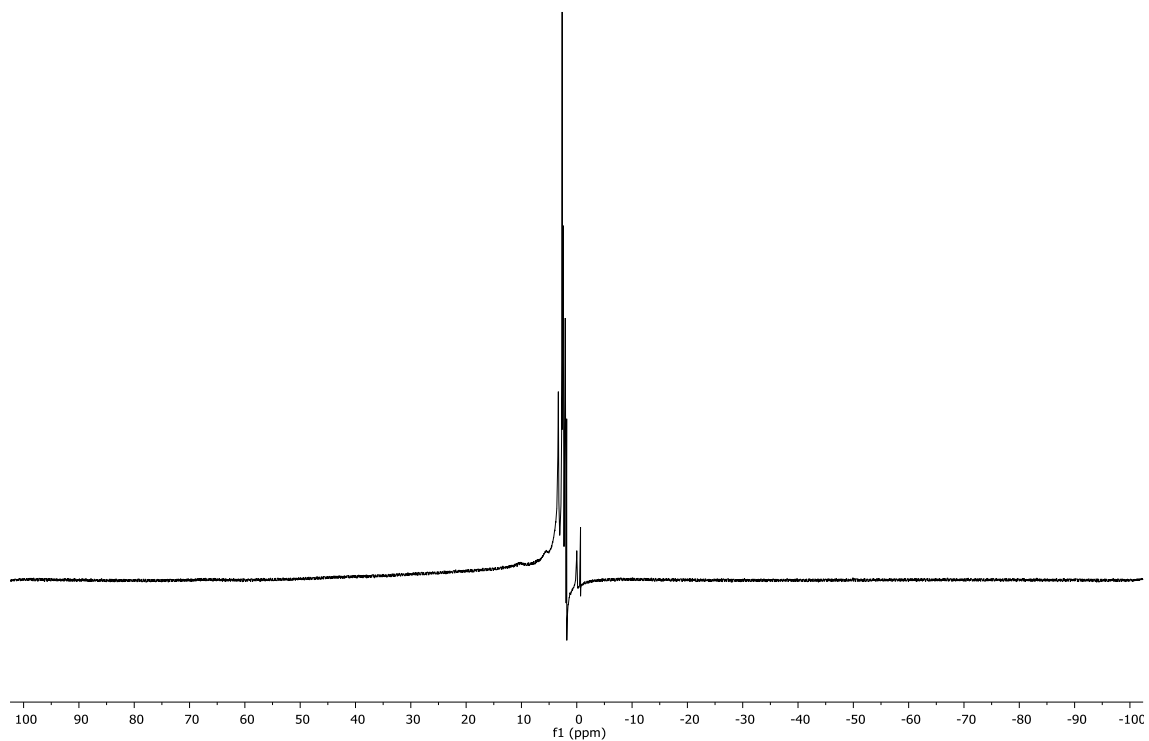
**Figure S24:**  $^1\text{H}$  NMR spectrum of 5 mg of paramagnetic **X2** in  $\text{DMSO-}d_6$  at 298 K with a  $\text{DMSO-}d_6$  capillary. The shifted and unshifted solvent peaks are highlighted.



**Figure S25:**  $^1\text{H}$  NMR spectrum of 5 mg of paramagnetic **X2** in  $\text{DMSO-}d_6$  at 298 K

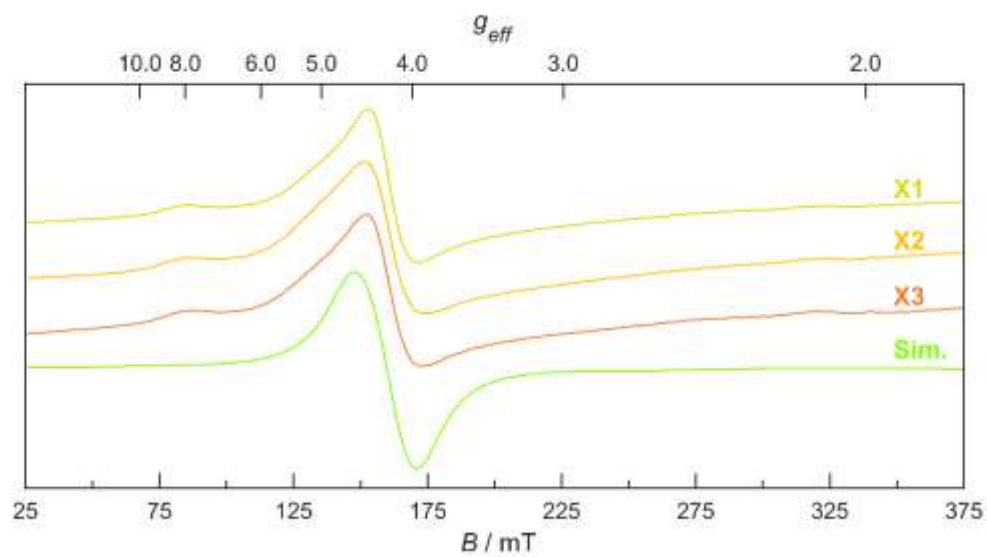


**Figure S26:**  $^1\text{H}$  NMR spectrum of 5 mg of paramagnetic **X3** in  $\text{DMSO-}d_6$  at 298 K with a  $\text{DMSO-}d_6$  capillary. The shifted and unshifted solvent peaks are highlighted.



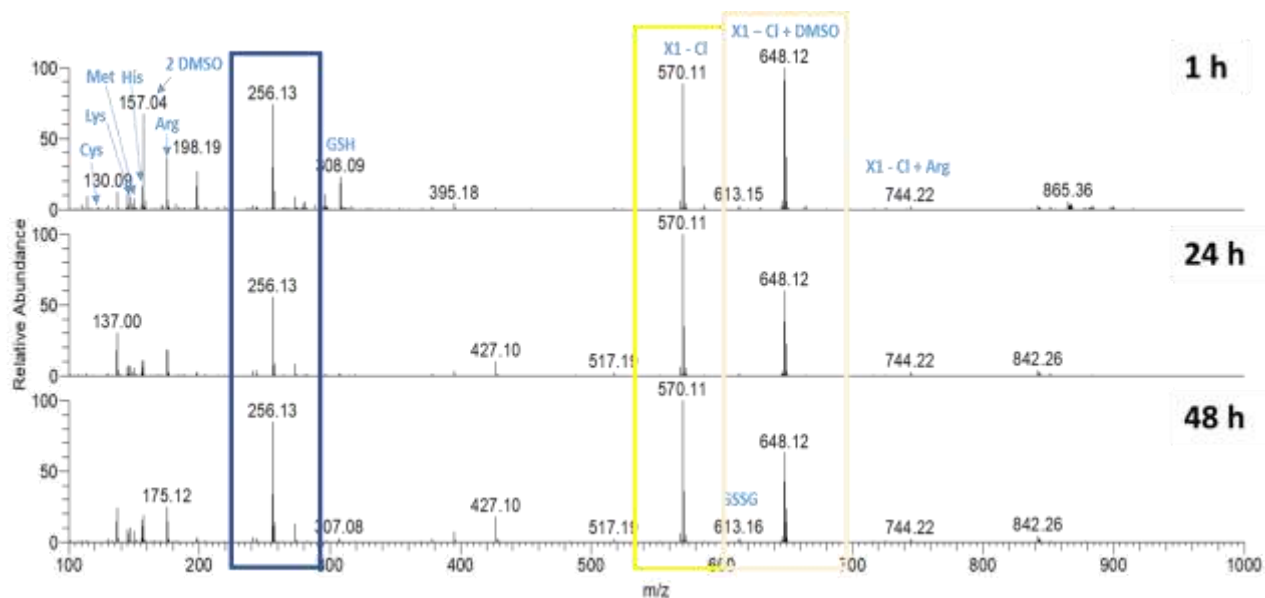
**Figure S27:**  $^1\text{H}$  NMR spectrum of 5 mg of paramagnetic **X3** in  $\text{DMSO-}d_6$  at 298 K

## 2.4. EPR spectra of the complexes

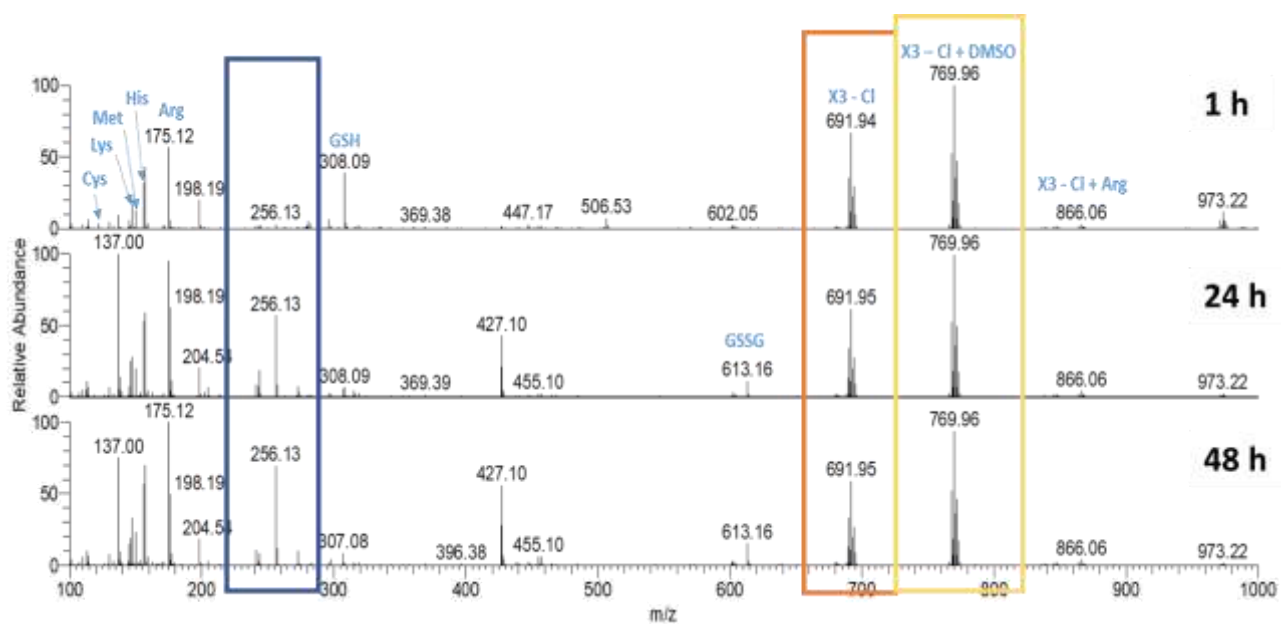


**Figure S28:** EPR spectra of **X1**, **X2** and **X3** in DMSO at 98 K. Simulation of a rhombic spin 5/2  $\text{Fe}^{3+}$  system showing the observed signature. Simulation parameters:  $g_x=g_y=g_z=2.0023$ ,  $D = 10\,000$  MHz,  $E/D = 0.28$ . Note: The intensity of the signal at  $g\sim 7.9$  is not accurately reproduced in the simulation due to line broadening effects.

### 3. Reactivity toward biological nucleophiles



**Figure S29:** HR-MS spectra of X1 in an aqueous solution containing equimolar amounts of Arg, Cys, His, Lys, Met and GSH.

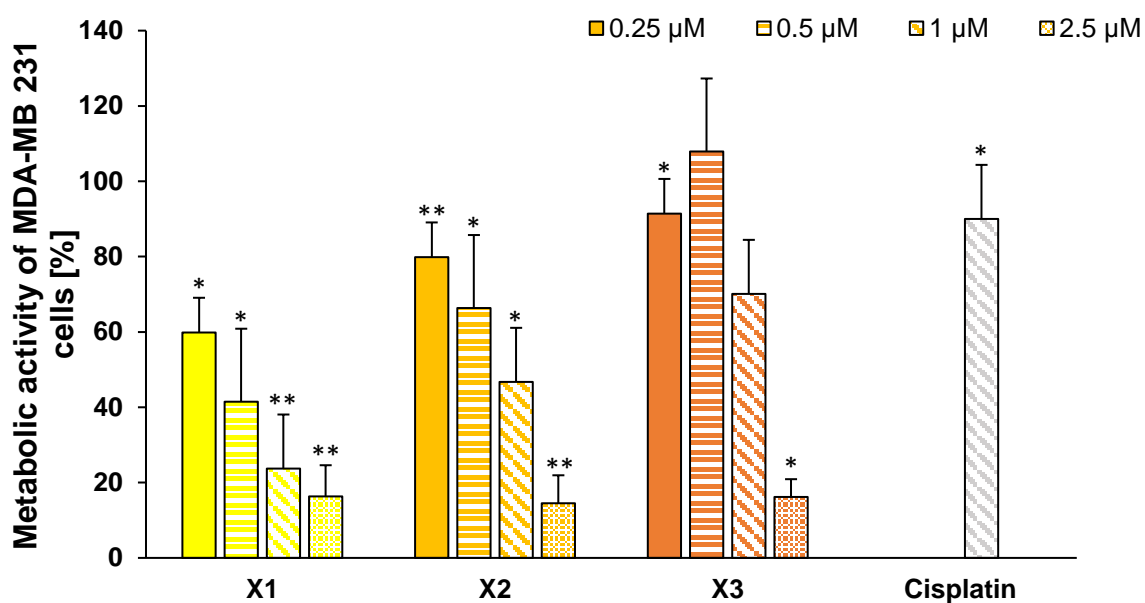


**Figure S30:** HR-MS spectra of X3 in an aqueous solution containing equimolar amounts of Arg, Cys, His, Lys, Met and GSH.

**Table S3:** List of ions observed in the HR-MS analysis of **X1** – **X3** in aqueous solution. Oleamide is a common contaminant.

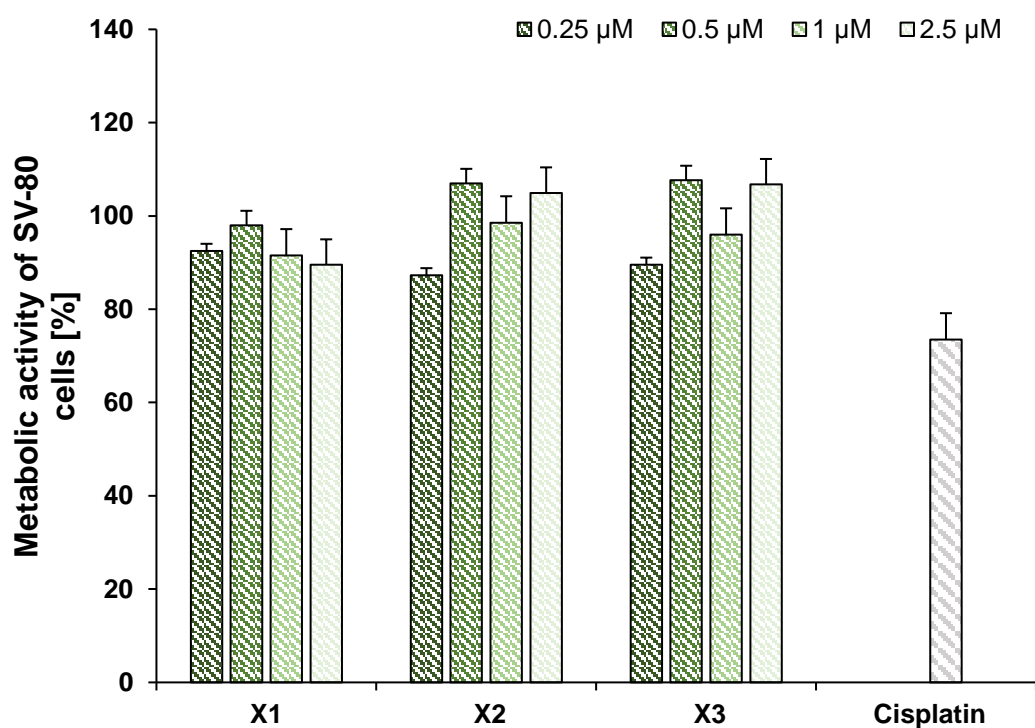
	species	m <sub>exp.</sub>	m <sub>calc</sub>	error [ppm]
<b>biomolecules</b>	[Cys + H] <sup>+</sup>	122.0272	122.0270	1.64
	[Lys + H] <sup>+</sup>	147.1130	147.1128	1.36
	[Met + H] <sup>+</sup>	150.0585	150.0583	1.33
	[His + H] <sup>+</sup>	156.0769	156.0768	0.64
	[Arg + H] <sup>+</sup>	175.1191	175.1190	0.57
	[GSH + H] <sup>+</sup>	308.0914	308.0911	0.97
	[GSSG + H] <sup>+</sup>	613.1604	613.1592	1.96
<b>hydrolysis products</b>	[C <sub>16</sub> H <sub>17</sub> O <sub>2</sub> N + H] <sup>+</sup>	256.1336	256.1332	1.56
	[C <sub>16</sub> H <sub>20</sub> O <sub>2</sub> N <sub>2</sub> + H] <sup>+</sup>	273.1602	273.1598	1.46
	[C <sub>23</sub> H <sub>20</sub> O <sub>3</sub> N <sub>1</sub> Cl + H] <sup>+</sup>	394.1212	394.1204	2.03
	[C <sub>23</sub> H <sub>23</sub> O <sub>3</sub> N <sub>2</sub> Cl + H] <sup>+</sup>	411.1478	411.1470	1.95
<b>X1</b>	[X1 - Fe - Cl + 3H] <sup>+</sup>	517.1943	517.1933	1.93
	[X1 - Cl] <sup>+</sup>	570.1057	570.1049	1.40
	[X1 - Cl + DMSO] <sup>+</sup>	648.1198	648.1190	1.23
	[X1 - Cl + Lys] <sup>+</sup>	716.2118	716.2104	1.95
	[X1 - Cl + His] <sup>+</sup>	725.1758	725.1741	2.34
	[X1 - Cl + Arg] <sup>+</sup>	744.2180	744.2167	1.75
	[X1 - Cl + C <sub>16</sub> H <sub>20</sub> O <sub>2</sub> N <sub>2</sub> ] <sup>+</sup>	842.2592	842.2573	2.26
	[X1 - Cl + GSH] <sup>+</sup>	877.1903	877.1888	1.71
<b>X2</b>	[X2 - Cl] <sup>+</sup>	602.0470	602.0457	2.16
	[X2 - Cl + DMSO] <sup>+</sup>	680.0611	680.0596	2.21
	[X2 - Cl + Lys] <sup>+</sup>	748.1531	748.1512	2.54
	[X2 - Cl + His] <sup>+</sup>	757.1170	757.1152	2.38
	[X2 - Cl + Arg] <sup>+</sup>	776.1594	776.1574	2.58
	[X2 - Cl + oleamide] <sup>+</sup>	883.3197	883.3174	2.60
	[X2 - Cl + GSH] <sup>+</sup>	909.1316	909.1295	2.31
<b>X3</b>	[X3 - Cl] <sup>+</sup>	691.9440	691.9426	2.02
	[X3 - Cl + DMSO] <sup>+</sup>	769.9585	769.9566	2.47
	[X3 - Cl + Lys] <sup>+</sup>	838.0500	838.0482	2.15
	[X3 - Cl + His] <sup>+</sup>	847.0140	847.0121	2.24
	[X3 - Cl + Arg] <sup>+</sup>	866.0563	866.0543	2.31
	[X3 - Cl + oleamide] <sup>+</sup>	973.2168	973.2145	2.36
	[X3 - Cl + GSH] <sup>+</sup>	999.0286	999.0264	2.20

#### 4. Biological activity

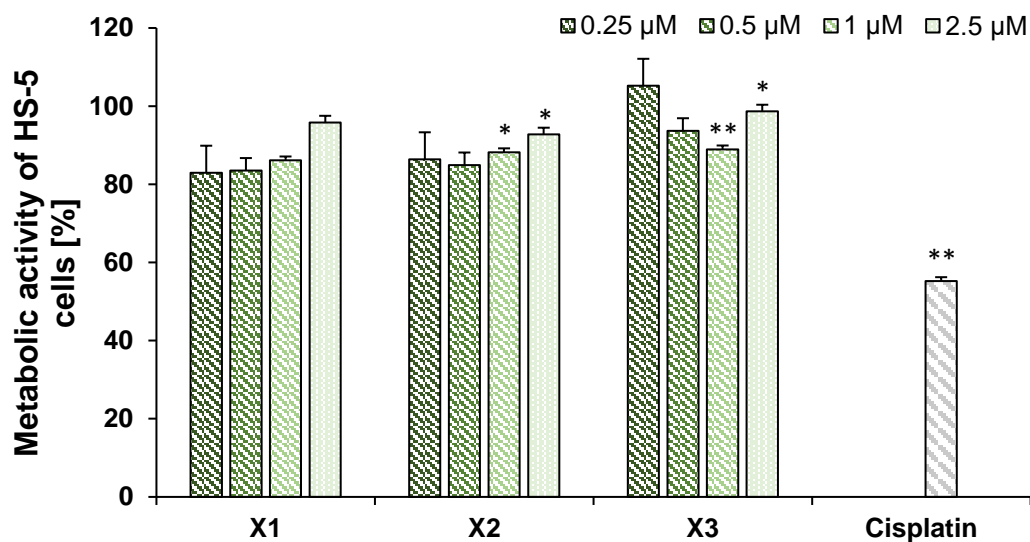


**Figure S31:** Effect of the iron(III) salen complexes **X1** – **X3** on the metabolic activity of MDA-MB 231. The mean proliferation + SE of eight independent experiments is plotted. The metabolic activity of cells only (without compound) was set at 100% (data not shown). The asterisks (\*  $p < 0.05$ , \*\*  $p < 0.005$  against no compound) represents statistical significance.

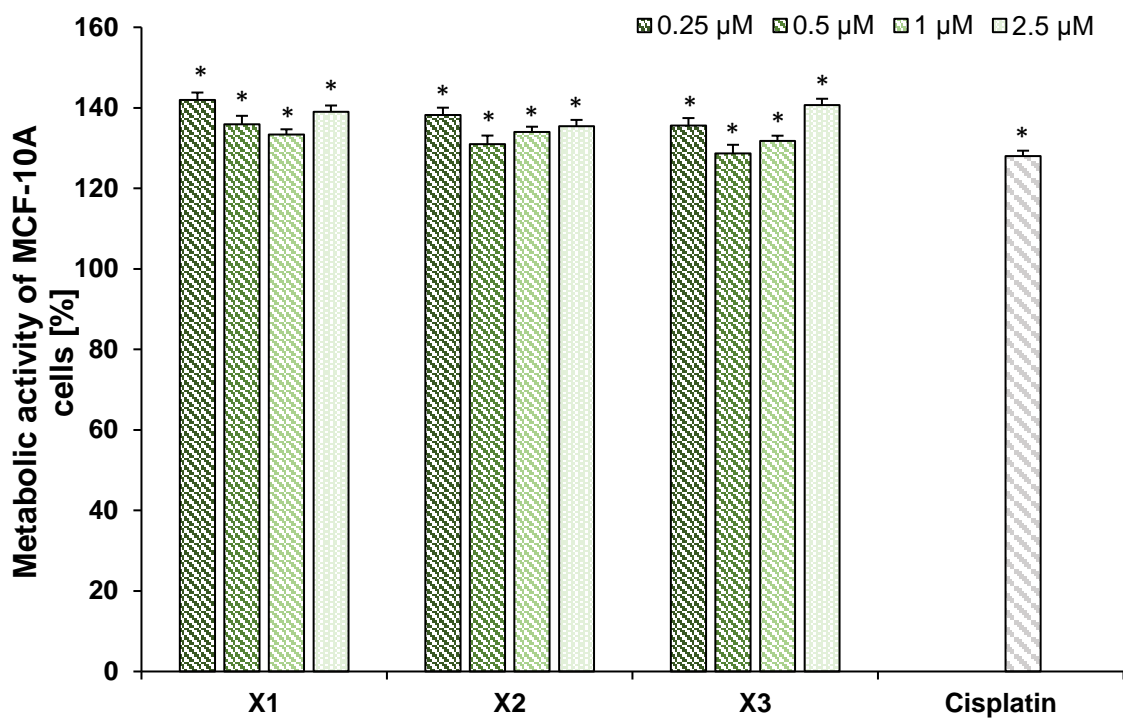




**Figure S32:** Effect of the iron(III) salene complexes **X1** – **X3** on the metabolic activity of SV-80 transformed fibroblasts. The mean metabolic activity + SE of five independent experiments is plotted. The metabolic activity of cells only (without compound) was set at 100%.



**Figure S33:** Effect of the iron(III) salene complexes **X1** – **X3** on the metabolic activity of HS-5 cells. The mean metabolic activity + SE of six independent experiments is plotted. The metabolic activity of cells only (without compound) was set at 100%. The asterisks (\*  $p < 0.05$ , \*\*  $p < 0.005$  against no compound) represents statistical significance.



**Figure S34:** Effect of the iron(III) salene complexes **X1** – **X3** on the metabolic activity of MCF-10A cells. The mean metabolic activity + SE of six independent experiments is plotted. The metabolic activity of cells only (without compound) was set at 100%. The asterisks (\*  $p < 0.005$  against no compound) represent statistical significance.