SUPPORTING INFORMATION

Thirdhand exposures to tobacco-specific nitrosamines through inhalation, dust ingestion, dermal uptake, and epidermal chemistry.

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Table of Contents

Table S1: Tobacco smoke constituents in IARC Group 1 (carcinogenic to humans)

Section S1: TSNA mutagenicity and carcinogenicity

Both NNK and NNN are activated metabolically, with formation of DNA adducts being considered critical for their mutagenicity and carcinogenicity [1, 2]. The carcinogenicity of NNK was tested in mice, rats, hamsters and mink, with administration via drinking water, gavage, subcutaneous or intraperitoneal injection, and skin painting [3-6]. NNK causes lung, liver, pancreas and other types of cancers and is more carcinogenic than NNN for the induction of lung and liver tumors in F344 rats or A/J mice [5, 7]. DNA adduct formation by NNK and NNN, after activation by the cytochrome P450 system, is considered a central mechanism for tumorigenesis [8]. In addition, the binding of NNK and NNN to the nicotinic acetylcholine receptor promotes tumor growth by enhancing and deregulating cell proliferation, migration, and invasion [9], thereby creating a microenvironment for tumor growth. The tumorigenic activity of NNA, together with NNK and NNN, was studied in strain A/J mice, and found to cause tumors, though at a lower incidence than NNK and NNN, with 36% for NNA compared to 87% and 76% for NNK and NNN, respectively [10]. In a study using a HPRT locus mutagenicity assay, NNA exhibited a mutagenic activity comparable to that of NNK in a human B-lymphoblastoid cell line expressing P450 CYP2D6 cDNA [11]. Similarly, the DNA damage caused by NNA and NNK in a comet assay quantifying strand breaks in human HepG2 cells showed a similar damage and dose response, suggesting comparable genotoxicity for both compounds [12]. Several NNA adducts were identified *in vitro* from its reaction with deoxyguanosine [13].

Figure S1: Experimental setup

Table S2: Substrates and conditions used in the evaluation of the effects of skin liquids on the nitrosation of nicotine

Figure S2: Specimens used in the experiments

* adapted from Pavilonis et al, 2014 – *Risk Analysis* [14]

Section S2. Quantification of nicotine and TSNAs on cellulose and cotton substrates

The determination of nicotine and TSNAs is based on the method described by Whitehead et al. [28] **NNN and NNK:** 200 μL of standards, QCs, or sample extracts and 100 μL internal standard solution were pipetted into a 13 × 100 mm glass culture tube. These were mixed with 0.75 mL of 1:1 saturated NaHCO₃/50% K₂CO₃ and 4 mL of 40:40:15:5 pentane/dichloromethane/ethyl acetate/isopropanol, added for the extraction of analytes. The tubes were vortexed, centrifuged, and placed in dry ice acetone bath to freeze the aqueous layers. The organic layers were poured into new tubes containing 200 µL 0.2N HCl in MeOH, and evaporated to dryness in a Thermo Speedvac centrifugal evaporator at (1.5h/45°C). The residues were reconstituted in 250 µL LC mobile phase and 10 µL were injected into the LC-MS/MS system. The LOQs were 0.05 ng/mL.

Nicotine: the same samples were diluted and analyzed by GC/MS to determine nicotine concentrations. The LOQ was 100 ng/mL.

NNA: A different aliquot from each sample was used. To 1 mL standards, QCs, and samples were added NNA-d₃ internal standard, followed by 750 μ L 1:1 saturated NaHCO₃/50% K₂CO₃ and 4 mL 45:45:10 dichloromethane/pentane/ethyl acetate. The samples were vortexed, centrifuged, and placed in a dry ice/acetone bath to freeze the aqueous layers. The organic layers were poured to culture tubes containing 100 µL 0.1N HCl in methanol. The solvent was evaporated to dryness using a centrifugal vacuum evaporator (\approx 1.5 hr at 45 deg C). To the tubes were added 200 µL of derivatizing agent, pentafluorophenylhydrazine (PFPH), 3 mg/mL in acetonitrile. The tubes were heated for 30 min at 60°. After cooling to room temperature, 0.5 mL of the base (1:1 saturated NaHCO₃/50% K₂CO₃) and 4 mL of extraction solvent (45:45:10 dichloromethane/pentane/ethyl acetate) was added. The samples were vortexed, centrifuged, and placed in a dry ice/acetone bath to freeze the aqueous layers. The organic layers were poured to culture tubes containing 0.5 mL of 1 M sulfuric acid, and the samples were vortexed, centrifuged, and placed in a dry ice/acetone bath to freeze the aqueous layers. The organic layers were poured off and discarded. The above base, 0.5 mL, and 4 mL of the above extraction solvent were added. The samples were vortexed, centrifuged, and placed in a dry ice/acetone bath to freeze the aqueous layers. The organic layers were poured to culture tubes and the solvent was evaporated at 55° with a stream of nitrogen. The evaporated samples were reconstituted in 50 µL mobile phase, 10 mM ammonium formate in 85:15 water/methanol. 20 µL were injected into the LC-MS/MS system, with a Phenomenex Phenyl-Hexyl 3 x 150 mm column, mass spectrometer with a heated ESI source (HESI), run in SRM mode, as described in Whitehead et al. The LOQ was 0.05 ng/mL.

S7

7 8 9 10 11 12 13 14

day

Section S3: Human dermal uptake of nicotine and NNK from tobacco-laden clothing

Three volunteers wore tobacco-contaminated clothing inside an environmental chamber with nearly 0.85 air changes per minute. This is approximately 100 times more air exchange than is normally seen in homes in the US [15, 16]. Thus, dermal absorption, rather than inhalation, was the dominant form of exposure in this study. The clothing was long-sleeved shirts and full-length pants that had been exposed to cigarette smoke for 30 days, at concentrations similar to those found in the home of a pack-a-day smoker (3 mgs total particulate material). Each participant also completed a control exposure where they wore similar clothing that had not been exposed to smoke and the order of the exposures was randomized. The participants did not smoke tobacco or cannabis and were not exposed to smoke at home or work. They wore the clothing for three hours and exercised enough to perspire for thirty minutes out of each of hour. Urine specimens were collected prior to exposure, and at 8 hours after the start of exposure. Metabolites of nicotine (cotinine) and NNK (NNAL) were analyzed by published methods [17, 18].

The results show that both nicotine and NNK were absorbed through the human skin. Eight hours after the start of exposure, the urinary NNAL concentration was 86-fold higher than background levels, when the participants wore THS clothing, but remained at background levels when they wore clean clothing. The corresponding cotinine concentrations increased by a factor of 18 during the same period when tobacco-contaminated clothing was worn. These findings suggest that despite having a higher molecular weight than nicotine, NNK passes through the human dermis and into the bloodstream.

Table S4: Biomarkers of nicotine and NNK measured in urine before (background) and eight hours after exposure to tobacco-contaminated clothing.

Table S5: Measured formation rates and predicted rate constants for NNK, NNN and NNA resulting from the nitrosation of skin-bound nicotine

Section S4: Determination of the NNK formation rate constant

The quantitative determination of nicotine, HONO and TSNA concentrations allowed for the estimation of the rate constant k_{NNK} for NNK formation through epidermal chemistry. The formation rate r_{NNK} on experiments carried out on cellulose and cotton was between 24 and 72 nmol m⁻² h⁻¹. For these experiments, considering a nicotine skin surface concentration C_N = 1.5 \times 10³ µmol m⁻², a HONO concentration [HONO] = 700 ppb, and a median NNK formation rate r_{NNK} = 37 nmol m⁻² h⁻¹, a bimolecular reaction rate constant k_{NNK} = 5.5 \times 10⁻⁷ – 1.6 \times 10⁻⁶ was determined as the ratio:

$$
k_{NNK} = \frac{r_{NNK}}{\left[\text{HONO}\right] \times C_N} \tag{S1}
$$

Figure S5. NNA concentration on substrates modified with artificial sweat surrogate mixture at pH = 4 (light blue) and pH = 7 (dark blue)

Table S6: TSNA and nicotine concentrations in indoor and outdoor air

Table S7: TSNA and nicotine concentrations in settled indoor dust

Table S8: TSNA and nicotine concentrations in indoor surfaces and skin

(a) assumes 42 cm^2 per wipe

(b) assumes 100 cm² per wipe

Table S9: TSNA concentrations measured simultaneously in the gas phase and particle phase at different times after smoking ended, and fraction of each compound in the gas phase (results published in Tang et al, 2021 [19], Figure 2)

Figure S6. NNK/nicotine ratio from different studies reported in the literature in A) indoor air, B) indoor settled dust, and C) indoor surfaces. Labels "SHS" and "THS" indicate samples that are more typically associated with secondhand and thirdhand smoke, respectively.

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