Supporting Information

Electrochemical oxidation of low concentration methane on Pt/Pt and Pt/CP under ambient conditions

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Figure S1. SEM images of (a) unmodified Pt mesh and (b) carbon paper (scale bars 1 µm).



Figure S2. XRD pattern of Pt/Pt mesh and Pt/CP, inlet is the zoom-in of Pt peaks on CP.



Figure S3. A plot of charge density (q) vs. activation time at $E_{app} = 0.15$ V for Pt/Pt in CH₄ saturated 0.5 M HClO₄.



Figure S4. (a) Chronoamperometry during CH₄ electrochemical activation (with N₂ background subtracted) recorded at $E_{app} = 0.15$ V for 1800 s in 0.5 M HClO₄ on Pt/Pt. (b) Chronoamperometry during CH₄ activation (N₂ background subtracted) recorded at $E_{app} = 0.3$ V for 1800 s in 0.5 M HClO₄ on Pt/CP.



Figure S5. HPLC result obtained from Pt/CP at $E_{app} = 0.3$ V for 15 h reaction in CH₄ saturated 0.5 M HClO₄.



Figure S6. (a) GC results of 1% CH₄ in N₂ electro-oxidation on Pt/Pt before (black line) and after (red line) one cycle of electrochemical oxidation including CA for 30 min at 0.15 V and two CV scans, (b) GC results of 1% CH₄ in N₂ for three cycles of electrochemical oxidation on Pt/Pt.

Figure S7. GC results of N_2 electro-oxidation on Pt/Pt before (black line) and after (red line) one cycle of CA and two CV scans for 30 min at 0.15 V in the absence of CH₄.

Figure S8. GC results obtained from Pt/CP before (black line) and after (red line) CA at 0.3 V for 30 min and two CV scans for 0.5% CH₄ in N₂.

Figure S9. GC results before CH₄ oxidation (black line) and after (red line) 30 cycles of CH₄ oxidation using 0.5% CH₄ in N₂ with $E_{app} = 0.3$ V for 30 min CA and two CV scans during each cycle on Pt/CP.

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