Supporting Information

Mapping the FF domain folding pathway via structures of transiently populated folding intermediates

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Supplementary text Figures S1 to S7 Tables S1 to S4 Legend for supplementary file A17GFF_I2_lowest10.pdb. SI References

Chemical Exchange Saturation Transfer

Consider a simple two-state reaction, $F \rightleftharpoons U$, where states F and U are in slow exchange on the NMR chemical shift timescale, with $p_F \gg p_U$ (1). A set of spectra is acquired with a weak radio frequency (B_I) pulse applied at a given frequency for an exchange time, T_{EX} (one spectrum for each position of the B_1 field), and the effect of the B_1 irradiation on the 'visible' major state spectrum is monitored (2-4). A plot of the normalised intensity of a peak derived from the visible (F) state (I/I_0) vs ϖ_{RF} will have two dips, one at the chemical shift (ppm) of the major state, $\overline{\omega}_F$, and, importantly, one at the chemical shift of the minor state ϖ_U . Here ϖ_{RF} is the offset (ppm) at which the B_I field is applied. The exchange parameters ($k_{ex,FU}$, p_U), major/minor state chemical shifts, major state longitudinal relaxation rate, and major and minor state transverse-relaxation rates can all be obtained by analysing CEST profiles recorded with two different B_1 values (5, 6). Exchange occurring at rates spanning the ~ 10 to $\sim 10,000$ s⁻¹ range can be studied using amide ¹⁵N CEST experiments, although as exchange becomes fast only a single dip will be observed in each CEST profile at a frequency of $\sim \overline{\omega}_F$, and its shape could be asymmetric – slightly tilted towards the position of the minor state (6-9). Interestingly, introduction of a third state I that is in rapid exchange with U, $F \rightleftharpoons$ $I \rightleftharpoons U$ such that $p_F \gg p_U \gg p_I$ and $k_{ex,IU} \gg k_{ex,FI}$ results in a CEST intensity profile with two dips, one corresponding to state F and one arising largely from state U, but with the minor U state dip broadened and moved from $\overline{\omega}_{U}$ towards $\overline{\omega}_{I}$ (Fig. S1). Recently we have shown that the broadening of the 'U' state dip can be used to detect other minor states, such as I in this example, that otherwise would have escaped detection by other methods (10, 11).

Materials and Methods

NMR Samples. Isotopically enriched A17G FF and A17G S56P FF domains were overexpressed in *E coli* BL21(DE3) cells transformed with the appropriate plasmid and grown in the appropriate M9 media (12, 13). As all the samples used in this study were uniformly ¹⁵N enriched, the M9 media used to grow the cells consisted of 1g/L ¹⁵NH₄Cl as the sole nitrogen source. To prepare [U-¹⁵N] labelled protein 5 g/L of glucose was added as the sole carbon source. [U-¹⁵N, ¹³C] labelled protein was expressed in cells grown in M9 media consisting of 3 g/L of [¹³C₆] glucose as the sole carbon source. [U-¹⁵N], ¹³C^α labelled protein was expressed in cells grown in M9 media consisting of 3 g/L of [¹³C₆] glucose as the sole carbon source. [U-¹⁵N], ¹³C^α labelled protein was expressed in cells grown in M9 media consisting of 3 g/L of [2-¹³C] glucose as the sole carbon source (12, 14). [U-¹⁵N, ¹³C] 50% ²H labelled protein was expressed in cells grown in 60% D₂O M9 media with 3 g/L of [¹³C₆] glucose as the sole carbon source (15, 16). The NOESY experiments described in the text were measured on a sample that is [U-¹⁵N, ¹³C] enriched everywhere except for the side-chains of Ile, Leu and Val that are [Ile δ 1 - ¹³CH₃, ²H; Leu, Val - ¹³CH₃/¹²CD₃, ²H]. This sample was prepared by growing cells initially in M9 media with 3 g/L of [¹³C₆] glucose as the sole carbon source. One hour prior to induction of protein expression 50 mg/L 2-keto-3-d₂-4-¹³C butyrate and 100 mg/L 2-keto-3-methyl-d₃-3-d₁-4-¹³C-butyrate were added to the media (13). The overexpressed protein with the desired labelling was purified

from the *E coli* cells using a two-step procedure consisting of a cation exchange chromatography step followed by a size exclusion chromatography step, as described previously (16).

NMR Samples. See SI Appendix, Table S1 for a list of all samples used in the present study.

NMR Experiments. All CEST and NOESY experiments were recorded on a 700 MHz Bruker Avance III HD spectrometer equipped with a triple-resonance cryogenically cooled probe with a Z axis gradient. The assignment experiments were performed on a 500 MHz Bruker NEO spectrometer equipped with a room temperature triple resonance Z-gradient probe.

In this study we have used CEST rather than CPMG experiments to obtain ϖ_{I2} values because i) ¹³C CEST experiments can be performed using uniformly ¹³C enriched samples (17, 18), unlike the case for CPMG studies (12, 14), resulting in a more extensive set of I2 state chemical shifts, ii) accurate exchange parameters can be obtained from CEST experiments performed at a single B_0 field (4, 5) and iii) ϖ_{I2} values are directly obtained from the analysis of CEST profiles, while only the absolute value of $\Delta \varpi$ ($|\Delta \varpi|$) is available from the analysis of CPMG data. Thus, additional experiments, sometimes at multiple B_0 field strengths, must be performed to obtain the sign of $\Delta \varpi$ and reconstruct the minor state spectrum from CPMG data (19-21).

All CEST experiments were carried out in a pseudo 3D manner (B_1 offset in the third dimension), with CEST profiles generated by quantifying peak intensities from two-dimensional correlation maps. Amide ¹⁵N ϖ_{12} shifts were obtained from ¹⁵N CEST datasets recorded on a [U-¹⁵N, ¹³C] A17G FF sample (sample 1; SI Appendix, Table S1) using the standard ¹⁵N CEST experiment (4). Amide ¹H^N ϖ_{12} shifts were obtained by recording ¹H^N CEST datasets with suppression of NOE-based dips (22, 23) ([U-¹⁵N, ¹³C] A17G FF sample; sample 1). ¹³C^O $\overline{\omega}_{I2}$ shifts were obtained from ¹³C^O CEST datasets recorded on a [U-¹⁵N, ¹³C] A17G FF sample (sample 1), either using HN(CO)-type or H(N)CO-type sequences in which CEST profiles were obtained by quantifying intensities from ¹⁵N-¹H^N or ¹³C^O(i-1)-¹H^N(i) correlation maps, respectively (24). ¹³C^{α} $\overline{\omega}_{I2}$ shifts were measured using four different CEST-based experiments: i) ¹³C^{α} CEST via an (HACACO)NH scheme recorded on a [U-¹⁵N, ¹³C] A17G FF sample (sample 1), with peak intensities quantified from ¹⁵N-¹H^N correlation maps (25); ii) ¹³C^α CEST based on quantification of CT- $^{13}C^{\alpha}-^{1}H^{\alpha}$ correlation maps (17) and recorded on a 100% D₂O [U- ^{15}N , ^{13}C] A17G FF sample (sample 2); iii) ¹³C^α CEST using a 100% D₂O [U-¹⁵N] ¹³C^α A17G FF sample (sample 3), with peak quantification from ¹³C^α-¹H^α correlation maps (17); iv) Gly-optimized ¹³C^α CEST recorded on a 100% D₂O [U-¹⁵N, ¹³C] A17G FF sample (sample 2) with peak intensities quantified from ${}^{13}C^{\alpha}-{}^{1}H^{\alpha}$ correlation maps (17). ${}^{1}H^{\alpha} \overline{\omega}_{12}$ shifts were obtained from four different CEST experiments all designed to supress NOE dips arising from interactions with remote protons: i) (HACACO)NH-¹H^a CEST using a [U-¹⁵N, ¹³C] A17G FF sample (sample 1), quantifying peak intensities from a series of ${}^{15}N{}^{-1}H^{N}$ correlation maps (26); ii) CT- ${}^{13}C^{\alpha}{}^{-1}H^{\alpha}{}^{-1}H^{\alpha}$ CEST (17, 22) using a 100% D₂O [U-¹⁵N, ¹³C] A17G FF sample (sample 2); iii) ${}^{1}H^{\alpha}$ CEST but using a 100% D₂O [U-¹⁵N] ¹³C^{α} A17G FF sample (sample 3) with peak quantification from ¹³C^{α -1}H^{α} correlation maps (17, 22); iv) Gly-optimized ¹H^{α} CEST using a 100% D₂O [U-¹⁵N, ¹³C] 50% ²H A17G FF sample (sample 4), with peak quantification from a set of ¹³C^{α}-1H^{α} correlation maps (15). The ¹³C^{α} A17G FF sample (sample 3) is crucial to study exchange at ¹H^{α} and ¹³C^{α} sites when the ¹³C^{α} and ¹³C^{β} carbons are strongly coupled, as is the case for S56, for example. ¹³C^{β} ϖ_{I2} shifts were obtained from three different CEST experiments: i) (HBCBCACO)NH-¹³C^{β} CEST using a [U-¹⁵N, ¹³C] A17G FF sample (sample 1) and quantifying peak intensities from ¹⁵N-¹H^N correlation maps (25); ii) Ser-optimized ¹³C^{β} CEST using a 100% D₂O [U-¹⁵N, ¹³C] A17G FF sample (sample 2) and quantifying peak intensities from CT-¹³C^{β -1}H^{β} correlation maps (17); iii) As in ii) but using a Thr-optimized pulse scheme for Thr residues (17); Methyl ¹³C (including Ala ¹³C^{β}) ϖ_{I2} shifts were obtained from methyl ¹³C CEST experiments recorded on a [U-¹⁵N, ¹³C] A17G FF sample (sample 1) (17, 27). Methyl ¹⁴H ϖ_{I2} shifts were obtained using a methyl ¹H CEST experiment that supresses dips in CEST profiles arising from dipolar interactions with remote protons (22) (100% D₂O [U-¹⁵N, ¹³C] A17G FF sample; sample 2). In both methyl ¹³C and ¹H CEST experiments peak intensities were quantified from CT-¹³C-¹H correlation maps. Additional details can be found in *SI Appendix*, Tables S1 and S2.

Amide ¹⁵N CEST (methyl ¹³C D-CEST (9, 28)) experiments were used to obtain two-state (F \rightleftharpoons I2) exchange parameters in A17G S56P FF samples dissolved in 10 (100) % D₂O buffer; these samples were used for measurement of NOEs to confirm the I2 state structure. To estimate p_{II} at 20 °C for the A17G S56P FF sample (sample 13) used to perform the NOESY experiments (Fig. 4), amide ¹⁵N CEST datasets were recorded with four different B_I (T_{EX}) values: 3.2 Hz (400 ms), 53.6 Hz (250 ms), 107.3 Hz (175 ms), and 214.5 Hz (175 ms).

NOEs between G17 H^{α} and L55 H^{δ} were recorded using 3D HSQC-NOESY-HSQC or HSQC-NOESY-HMQC experiments, [Gly ¹³C^{α} (t₁), methyl ¹³C (t₂), methyl ¹H (t₃)], using A17G FF (sample 5) or A17G S56P FF (sample 13,14) samples that are [U-¹⁵N, ¹³C] enriched at all positions except for the side-chains of Ile, Leu and Val that are [U-¹⁵N], [Ile δ 1 - ¹³CH₃, ²H; Leu, Val - ¹³CH₃/¹²CD₃, ²H]. The mixing time was set to 100 ms in all the experiments.

Chemical shift assignments of A17G S56P FF were obtained starting from previous assignments for A17G FF, and completed by analysis of standard HNCA, HNCO, HNCACO, HNCACB, (H)C(CO)NH-TOCSY, and H(C)(CO)NH-TOCSY datasets (1, 29-31) at 15 °C (11.7 T).

NMR Data processing and analysis. NMR data were processed using NMRPipe (32), visualised and assigned using SPARKY (33, 34), with peak intensities quantified using PINT (35). TALOS-N (36) was used to analyse the chemical shifts and obtain residue specific helix propensities, residue specific S^2 values (37), and helix boundaries for the various conformational states of A17G FF. The program *ChemEx* (38) that numerically integrates the Bloch-McConnell equations (39) was used to obtain the best-fit exchange parameters from the CEST data.

Chemical shifts for the I2 folding intermediate were obtained by analysing A17G FF CEST data in a two-state manner with global fitting parameters, $k_{ex,F12}$ and p_{12} , and residue specific fitting parameters $R_{2,F}$, $R_{1,F}$, $R_{2,I2}$, ϖ_F and $\Delta \varpi_{F12}$. In all analyses R_I values were assumed to be the same for all conformers (4, 5). In the analysis of some of the CEST data the exchange parameters were fixed to those obtained from amide ¹⁵N or ¹³C^a CEST data (See Table S2). To extract the best-fit four-state (F, I1, I2 and U) exchange parameters from analysis of the A17G FF amide ¹⁵N CEST profiles, the four-state model in Figure 1B was used, subject to the constraint $R_{2,F} = R_{2,I1} = R_{2,I2} = 2R_{2,U}$ (11). In this case, the global fitting parameters were $k_{ex,F11}$, $k_{ex,F12}$, $k_{ex,F12}$, $k_{ex,FU}$, p_{I1} , p_{I2} and p_U ($k_{ex,I2U} = k_{ex,FU} = 0$ s⁻¹ (11)), with residue specific fitting parameters of: $R_{2,F}$, $R_{1,F}$, ϖ_F , $\Delta \varpi_{F11}$, $\Delta \varpi_{F12}$ and $\Delta \varpi_{FU}$. To obtain urea *m*-values A17G FF ¹⁵N CEST data recorded on samples prepared with five different concentrations of urea were analysed under the assumption that $\Delta \varpi_{F11}$, $\Delta \varpi_{F12}$ and $\Delta \varpi_{FU}$ are independent of urea concentration with $\Delta \varpi_{F11}$, $\Delta \varpi_{F12}$ and $\Delta \varpi_{FU}$ values initialised to those obtained previously from the four-state analysis of the A39G FF data (11).

To extract the best-fit three-state (F, I1, and I2) exchange parameters from the A17G S56P FF amide ¹⁵N CEST data, the three-state triangular model in which all three states interconvert with each other was fit to CEST data subject to the constraint $R_{2,F} = R_{2,I1} = R_{2,I2}$. Global (residue specific) fitting parameters were $k_{ex,FII}$, $k_{ex,FI2}$, $k_{ex,III2}$, p_{II} and p_{I2} ($R_{2,F}$, $R_{I,F}$, ϖ_F , $\Delta \varpi_{FI1}$ and $\Delta \varpi_{FI2}$), and $\Delta \varpi_{FI1}$ and $\Delta \varpi_{FI2}$ were initialised to the values previously determined from the four-state analysis of the A39G FF ¹⁵N CEST data (11). Two-state analysis of the A17G S56P FF CEST data followed as for A17G FF, with $k_{ex,FI2}$ and p_{I2} ($R_{2,F}$, $R_{I,F}$, $R_{2,I2}$, ϖ_F and $\Delta \varpi_{FI2}$) global (residue specific) fitting parameters. Bootstrap or Monte Carlo procedures (100 trials) were used to estimate the uncertainties in all the fitted exchange parameters (40).

Structure calculations. The CS-ROSETTA (41) protocol was used to calculate the structural ensemble of the A17G FF I2 state solely from chemical shifts (41-43). 10,000 structures were calculated and the (rescaled) energy versus C^{α} RMSD to the lowest energy structure clearly shows that calculation has converged, with an RMSD of 0.8 ± 0.2 Å for the lowest energy structure to the 10 lowest energy structures (Fig. S3). As a control the same structure calculation procedure was repeated using F state shifts for the same sites that the I2 state shifts are available. This calculation also converged; the 10 lowest energy structures have a C^{α} RMSD of 0.9 ± 0.2 Å to lowest energy structure and a C^{α} RMSD of 1.4 ± 0.2 Å to the lowest energy F state structure of WT FF obtained by conventional NMR based methods (44) (Fig. S3). Residues W11 to Q68 (S² \geq 0.65) were used for the RMSD calculations. Some of the CS-ROSETTA calculations were performed on NMRbox (45). Structures were visualised and analysed using UCSF Chimera (46).

Extracting urea *m***-values.** Starting from the reaction $F \rightleftharpoons K$, where *F* and *K* are a pair of exchanging states (potentially of a more complex exchange pathway) it follows directly that $\Delta G_{FK} = G_K - G_F = -RT ln(\frac{p_K}{n_F})$, where *R* and *T* are the gas constant and the absolute temperature, respectively. Writing

 $\Delta G_{FK}(urea) = \Delta G_{FK}(0M \ urea) - m_{FK}[urea], \text{ it becomes clear that } \frac{d\Delta G_{FK}}{d[urea]} = -m_{FK} \text{ and, therefore, the slope of the } \Delta G_{FK} = -RTln(\frac{p_K}{p_F}) \text{ vs } [urea] \text{ plot is } -m_{FK}. \text{ Further, it is straightforward to show that for the reaction } F \rightleftharpoons L \rightleftharpoons \dots \rightleftharpoons K \Delta G_{FK} = -RTln(\frac{p_K}{p_F}) \text{ as well, so that the slope of the } \Delta G_{FK} \text{ vs } [urea] \text{ plot is also } -m_{FK}. \text{ In a similar manner the urea } m\text{-value of the transition state connecting states } K \text{ and } L (TS_{KL}), \text{ for example, can be obtained by defining } \Delta G_{FTSKL} = G_{TSKL} - G_F, \text{ where } G_{TSKL} \text{ is the free energy of the transition state. Writing } \Delta G_{FTSKL} = (G_{TSKL} - G_K) - (G_F - G_K) \text{ and noting that } k_{kl} = Cexp(-\frac{G_{TSKL} - G_K}{RT}), \text{ where } k_{KL} \text{ is the forward rate constant for the reaction } K \rightleftharpoons L, \text{ and } C \text{ is a constant, it follows that } \Delta G_{FTSKL} = -RTln(\frac{k_{KL}}{C}) - RTln(\frac{p_K}{p_F}). \text{ Noting that } \Delta G_{FTSKL}(urea) = \Delta G_{FTSKL}(0M \ urea) - m_{TSKL}[urea], \text{ it follows that } \Delta G_{FTSKL} = -RTln(\frac{k_{KL}}{C}) - RTln(\frac{p_K}{p_F}) \text{ Noting that } \Delta G_{FTSKL}(urea) = \Delta G_{FTSKL}(0M \ urea) - m_{TSKL}[urea], \text{ it follows that the slope of the } \Delta G_{FTSKL} = -RTln(\frac{k_{LK}}{C}) - RTln(\frac{p_L}{p_F}) \text{ vs } [urea] \text{ plot is } -m_{TSKL}. \text{ While } C \text{ was set to be 10^7 s^{-1} in this study, the value of } C \text{ does not affect the extracted transition state } m \text{ value}.$



Fig. S1. Large changes in amide ¹⁵N CEST profiles due to exchange between sparsely populated states on the intermediate exchange time-scale. (*A*) ¹⁵N CEST profile calculated for the two-state $F \Leftrightarrow U$ exchange process shown in the figure. There is a clear dip at ϖ_U (7.5 ppm; red line) even though p_U is only 2%. (*B*) ¹⁵N CEST profile calculated for the three-state $F \Leftrightarrow I \Leftrightarrow U$ exchange process in which the F to U interconversion proceeds via a sparsely populated intermediate I. The U state dip has broadened and shifted towards ϖ_I (10 ppm; green line); broadening of minor state dips is thus an indicator of minor exchange. Such CEST profiles can be analysed in a two-state ($F \Leftrightarrow U$) manner with the resulting best-fit parameters reporting on exchange between state F and a second state that is a composite of U and I. The best fit R_2 values of state U will be elevated (beyond what would be expected based on the size of the biomolecule), and consequently elevated minor state R_2 values are a sign of minor exchange (4, 11, 24). The calculations were performed with $\varpi_F = 0$ ppm (black line), $B_0 = 16.4$ T, $B_1 = 20$ Hz, $R_{1,F} = R_{1,U} = R_{1,I} = 1$ s⁻¹, $R_{2,F} = R_{2,U} = R_{2,I} = 10$ s⁻¹ and $T_{EX} = 400$ ms.



Fig. S2. Comparison of $\Delta \varpi_{FI1}$, $\Delta \varpi_{FI2}$, and $\Delta \varpi_{FU}$ values obtained from the four-state analysis of A17G FF amide ¹⁵N CEST data recorded at 10 °C with the corresponding values for A39G FF obtained previously (1 °C) (11). For A17G FF the best-fit exchange parameters for the model shown on top are: $k_{ex,FI1} = 784 \pm 67 \text{ s}^{-1}$, $k_{ex,FI2} = 406 \pm 5 \text{ s}^{-1}$, $k_{ex,I1/2} = 1600 \pm 113 \text{ s}^{-1}$, $k_{ex,I1U} = 11000 \pm 1064 \text{ s}^{-1}$, $p_{I1} = 0.27 \pm 0.01\%$, $p_{I2} = 0.83 \pm 0.01\%$ and $p_U = 0.16 \pm 0.02\%$. The excellent correlation between the $\Delta \varpi$ values from A39G and A17G FF establishes that both variants sample the same 11, I2 and U states. Additional details can be found in Table S4.



Fig. S3. Convergence of CS-ROSETTA structure calculations using CEST derived A17G FF I2 state chemical shifts. (*A*) Convergence plot of the CS-ROSETTA I2 state structure calculations. The rescaled CS-ROSETTA all-atom energy for each of the calculated structures is plotted against the C^a RMSD to the lowest energy structure. (*B*) The ten lowest energy models are tightly clustered (C^a RMSD < 2 Å) with a C^a RMSD to the lowest energy structure of 0.8 ± 0.2 Å, establishing convergence of the structure calculations. Note that the position of the sidechain of L55 is well defined. (*C*-*E*) Control calculations performed using the F state chemical shifts. The chemical shifts for a given atomic site was used in the F state CS-ROSETTA calculations only if the I2 state chemical shift for the same site was available. (*C*) Convergence plot of the CS-ROSETTA F state structure calculations. (*D*) The ten lowest energy models are tightly clustered with a C^a RMSD to the lowest energy structure is 0.9 ± 0.2 Å, showing that the F state structure calculations have also converged. The position of the sidechain of L55 is well defined and different from its position in (*B*). (*E*) Superposition of the ten lowest energy CS-ROSETTA A17G FF F state structures (grey) on the F state structure of WT FF (black; PDB: 1UZC) obtained using conventional NMR experiments (44); C^a RMSD: 1.4 ± 0.2 Å. The sidechain of L55 is in the same position in both the CS-ROSETTA derived A17G FF F state structural ensemble and the NOE-based WT FF structure. Only residues 11 to 68 that have RCI S² ≥ 0.65 were used for the C^a RMSD calculations.



Fig. S4. Validating the CEST derived I2 state structure. (A,B,C) Distances between G17 C^a and A20 C^β, A51 C^β, and L55 $\delta 1/\delta 2$ methyl group(s) in the F (A) and I2 (B) states. Note that the distances between G17 C^a and the L55 methyl δ groups that are large in F (~15 Å) become considerably smaller in I2 (~4.5 Å). The distances between G17 C^α and A20/A51 C^{β} are short in F (~4.5 Å) (A); these are used as references. The distance between G17 C^{α} and A51 C^{β} increases in I2 (~8.1 Å) (B). Distances are summarised in (C) where the mean and the standard deviation based on the ten calculated lowest energy structures is reported while in (A) and (B) distances from the lowest energy structure are shown. The Gly H^{α}-C^{α} (D) and methyl (E) regions of the ¹H-¹³C correlation map of A17G S56P FF (20 °C). In (E) correlations arising from the I2 state are indicated in green. (F-H) Methyl region (ϖ_2, ϖ_3) extracted from a 3D HSQC(ϖ_1)-NOESY-HSQC ($\varpi_2, \, \varpi_3$) spectrum at $\varpi_1 = 47$ ppm, corresponding to the G17 ¹³C^a chemical shift. 3D NOESY datasets were recorded at 15 (F; $k_{ex,Fl2} \sim 18.8 \pm 0.9 \text{ s}^{-1}; p_{l2} \sim 20.4 \pm 0.5 \%$), 20 (G; $k_{ex,Fl2} \sim 35.1 \pm 0.9 \text{ s}^{-1}; p_{l2} \sim 20.4 \pm 0.5 \%$), 20 (G; $k_{ex,Fl2} \sim 35.1 \pm 0.9 \text{ s}^{-1}; p_{l2} \sim 20.4 \pm 0.5 \%$), 20 (G; $k_{ex,Fl2} \sim 35.1 \pm 0.9 \text{ s}^{-1}; p_{l2} \sim 20.4 \pm 0.5 \%$), 20 (G; $k_{ex,Fl2} \sim 35.1 \pm 0.9 \text{ s}^{-1}; p_{l2} \sim 20.4 \pm 0.5 \%$), 20 (G; $k_{ex,Fl2} \sim 35.1 \pm 0.9 \text{ s}^{-1}; p_{l2} \sim 20.4 \pm 0.5 \%$), 20 (G; $k_{ex,Fl2} \sim 35.1 \pm 0.9 \text{ s}^{-1}; p_{l2} \sim 20.4 \pm 0.5 \%$), 20 (G; $k_{ex,Fl2} \sim 35.1 \pm 0.9 \text{ s}^{-1}; p_{l2} \sim 20.4 \pm 0.5 \%$) ~25.1 ± 0.4 %) and 25 (*H*; *k*_{ex,Fl2} ~62.8 ± 2 s⁻¹;*p*_{l2} ~27.6 ± 0.4 %) °C. Correlations are only seen in the NOESY spectra for methyl protons proximal to the G17 H^{α} sites. (*I*) The intensity ratios of the G17H^{α}-A20H^{β} and G17H^{α}-A51H^{β} 'reference' NOEs do not change with temperature as these NOEs largely arises from the F state (short distances in F, long distance for G17H^{α}-A51H^{β} in I2, (A,B,C)). The intensity ratios at each of these three temperatures are scaled by the intensity ratio at 20 °C. (J) The intensity ratios of $G17H^{\alpha}-L55H^{\delta 1}$ and the average of the $G17H^{\alpha}-A20H^{\beta}$ -and G17Ha-A51H^β correlations increase with temperature even as the rotational correlation time decreases, consistent with the G17-L55 NOEs arising from magnetization transfer in the I2 state as p_{I2} increases with temperature. The A17G S56P FF sample used in all these experiments is [U-15N, 13C] enriched everywhere except for the side-chains of Ile, Leu and Val that are [Ileδ1 - ¹³CH₃, ²H; Leu, Val - ¹³CH₃/¹²CD₃, ²H] (sample 13). Hence methyl residues from sites other than Ileδ1, Leuδ1,δ2, Val γ1,γ2 are split in the methyl ¹³C dimension of the ¹H-¹³C HSQC correlation map that was recorded without a constant time period (E). Residual protonation at the carbon adjacent to the methyl group leads to a small shoulder in the ¹³C dimension of the lle δ 1, Leu δ 1, δ 2, Val γ 1, γ 2 correlations.



Fig S5. (*A*-*G*) are similar to Fig S4 (*D*-*J*), except that the protein (sample 14) was dissolved in 100% D₂O, 30% glucose buffer rather than 10% D₂O, 30% glucose buffer and a 3D HSQC-NOESY-HMQC rather than a 3D HSQC-NOESY-HSQC experiment was used to detect NOE correlations between G17 H^a and L55 H^δ. Two-state exchange parameters ($k_{ex,Fl2}$, p_{l2}) obtained from methyl ¹³C D-CEST experiments (17, 27) at 15, 20 and 25 °C are (20 ± 0.7 s⁻¹; 12.5 ± 0.2 %), (44.9 ± 1.5 s⁻¹; 16.4 ± 0.3 %) and (66 ± 3 s⁻¹; 20.9 ± 0.2 %) respectively.



Fig. S6. χ^2_{red} vs p_{l1} obtained from a three-state analysis of A17G S56P FF amide ¹⁵N CEST data recorded at 20 °C ($B_1 = 3.2, 53.6, 107.3$ and 214.5 Hz; 16.4 T; sample 13). A triangular three-state model was fit ($\chi^2_{red} \sim 0.92, k_{ex,Fl1} = 425 \pm 105 \text{ s}^{-1}, k_{ex,Fl2} = 24.4 \pm 4.1 \text{ s}^{-1}, k_{ex,I1l2} = 2348 \pm 210 \text{ s}^{-1}, p_{l1} = 0.61 \pm 0.05$ % and $p_{l2} = 24.6 \pm 1$ %) to amide ¹⁵N CEST data from I43, S50, L52, A53, K59 & V67. ¹⁵N CEST experiments were carried out using the same sample (sample 13) that was used to record the NOESY experiments shown in Fig. 4F,G and *SI Appendix*, Fig. S4. In the calculations of the χ^2_{red} vs p_{l1} profile $\Delta \varpi_{Fl1}$ values were held to within ±2 ppm of those determined previously from the four-state analysis of A39G FF (11).



Fig. S7. (*A*) Ribbon representation of the native state of RhoGAPFF1 (PDB: 2k85) (47) illustrating the elongated H3 helix and that L311 from H3 is close to A273 from H1 in this construct akin to the proximity of L55 and A17 in the WT FF structure. Surface representation of RhoGAPFF1 in its native (*B*) and 'RhoGAPFF1 I1' states (*C*). In the native conformation the oxygen from the OH group of Y308 (coloured red) is inaccessible for phosphorylation (*B*). However when H4 is deleted, 'RhoGAPFF1 I1' state, (the structural equivalent of I1 where H4 is disordered), the oxygen from the OH group of Y308 (red) is clearly visible and accessible for modification (*C*). The molecular orientation in (*A*) differs from that in (*B*) and (*C*). In the HYPA/FBP11 FF domain, the equivalents of RhoGAPFF1 A273, Y308, and L311 are A17, L52, and L55, respectively.

Sample No	Mutant	Protein Labelling	Concentration (mM)	Buffer	Comments				
1	A17G FF	[U- ¹⁵ N, ¹³ C]	~4	50mM Sodium acetate, 100mM NaCl, 2mM EDTA, 2mM NaN₃, 20% [² H]-glucose, 7% D₂O, pH 5.7	Used to obtain the ${}^{15}N$, ${}^{13}C^{\alpha}$, ${}^{13}C^{\alpha}$, ${}^{13}C^{\alpha}$, ${}^{13}C^{\alpha}$, ${}^{13}C^{\beta}$, ${}^{1}H^{N}$, ${}^{1}H^{\alpha}$ and methyl ${}^{13}C$ I2 state shifts.				
2	A17G FF	[U- ¹⁵ N, ¹³ C]	~4	50mM Sodium acetate,100mM NaCl, 2mM EDTA,2mM NaN ₃ , 20% [² H]-glucose,100% D ₂ O, pH 5.7	Used to obtain the methyl ¹ H, Gly ${}^{13}C^{\alpha}$, Ser/Thr ${}^{13}C^{\beta}$ shifts and also the ${}^{13}C^{\alpha/1}H^{\alpha}$ shifts that were missing from sample 1.				
3	A17G FF	[U- ¹⁵ N], ¹³ C ^a	~4	50mM Sodium acetate,100mM NaCl, 2mM EDTA, 2mM NaN₃, 20% [²H]-glucose, 100% D₂O, pH 5.7	Used to obtain I2 state shift ${}^{1}H^{\alpha}$ and ${}^{13}C^{\alpha}$ of S56.				
4	A17G FF	[U- ¹⁵ N, ¹³ C], ~50% ² H	~4	50mM Sodium acetate, 100mM NaCl, 2mM EDTA, 2mM NaN ₃ , 20% [² H]-glucose, 100% D ₂ O, pH 5.7	Used to obtain I2 state shifts of G17 ${}^{1}H^{\alpha 1}$ and ${}^{1}H^{\alpha 2}$.				
5	A17G FF	[U- ¹⁵ N, ¹³ C], Ileδ1-[¹³ CH ₃], Leu, Val- [¹³ CH ₃ , ¹² CD ₃]	~4	50mM Sodium acetate, 100mM NaCl, 2mM EDTA, 2mM NaN ₃ , 30% [² H]-glucose, 10% D ₂ O, pH 5.7	Used to look for G17 $^1\text{H}^{a}\text{-L55}$ $^1\text{H}^{\delta}$ NOEs.				
6	A17G FF	[U- ¹⁵ N]	~2	50mM Sodium acetate,100mM NaCl,2mM EDTA, 2mM NaN₃, 0M Urea, 10% D₂O, pH 5.7	Used for <i>m</i> -value analysis (2.5 °C) and to obtain four-state exchange parameters at 10 °C.				
7	A17G FF	[U- ¹⁵ N]	~2	50mM Sodium acetate, 100mM NaCl, 2mM EDTA, 2mM NaN₃, 0.25M Urea, 10% D₂O, pH 5.7	Used for <i>m</i> -value analysis (2.5 °C).				
8	A17G FF	[U- ¹⁵ N]	~2	50mM Sodium acetate, 100mM NaCl, 2mM EDTA, 2mM NaN₃, 0.5M Urea, 10% D₂O, pH 5.7	Used for <i>m</i> -value analysis (2.5 °C).				
9	A17G FF	[U- ¹⁵ N]	~2	50mM Sodium acetate, 100mM NaCl, 2mM EDTA, 2mM NaN ₃ , 0.75M Urea, 10% D ₂ O, pH 5.7	Used for <i>m</i> -value analysis (2.5 °C).				
10	A17G FF	[U- ¹⁵ N]	~2	50mM Sodium acetate, 100mM NaCl, 2mM EDTA, 2mM NaN ₃ , 1M urea, 10% D ₂ O , pH 5.7	Used for <i>m</i> -value analysis (2.5 °C).				
11	A17G FF	[U- ¹⁵ N]	~2	50mM Sodium acetate, 100mM NaCl, 2mM EDTA, 2mM NaN ₃ , 20% glucose, 10% D2O, pH 5.7	Used to show that the addition of glucose reduces the four-state A17G FF to a two-state system.				
12	A17G S56P FF	[U- ¹⁵ N]	~2	50mM Sodium acetate, 100mM NaCl, 2mM EDTA, 2mM NaN ₃ , 30% glucose, 10% D ₂ O, pH 5.7	Used to test if the S56P mutation has increased <i>p</i> ₁₂ .				
13	A17G S56P FF	[U- ¹⁵ N, ¹³ C], lleõ ₁ -[¹³ CH ₃], Leu, Val- [¹³ CH ₃ , ¹² CD ₃]	~4	50mM Sodium acetate, 100mM NaCl, 2mM EDTA, 2mM NaN ₃ , 30% [² H] glucose, 10% D ₂ O, pH 5.7	Used to look for G17 Ha-L55 H $^{\delta}$ NOEs.				
14	A17G S56P FF	[U- ¹⁵ N, ¹³ C], Ileδ1-[¹³ CH ₃], Leu, Val- [¹³ CH ₃ , ¹² CD ₃]	~4	50mM Sodium acetate, 100mM NaCl, 2mM EDTA, 2mM NaN ₃ , 30% [² H] glucose, 100% D ₂ O, pH 5.7	Used to look for G17 H ^{α} -L55 H ^{δ} NOEs.				

 Table S1.
 Summary of the various A17G FF and A17G S56P FF NMR samples used in this study.

Sample No	Sample Details	CEST Experiment	<i>B</i> ₁ (Hz)	<i>T_{EX}</i> (ms)	<i>k_{ex}</i> (s ⁻¹)	р _{і2} (%)	Comments				
			31.7	350		2.08 ±					
		¹⁵ N	52.9	350	630 ± 6	0.01	¹ H ^N - ¹⁵ N plane				
	[U- ¹⁵ N, ¹³ C] A17G FF	13 C α	40.9	350			¹ H ^{N-15} N plane				
	50mM Sodium	13 C O	46	350			¹ H ^{N_15} N plane ¹ H ^{N_13} C ^O plane				
1	acetate, 100mM NaCl, 2mM	13 C O	46	350	Exc	hange					
	EDTA, 2mM NaN₃, 20% [²H]-	¹³ Cβ	46	300	parame fixed to	ters were the above	¹ H ^N - ¹⁵ N plane				
	glucose, 7% D ₂ O, pH 5.7	¹ H ^N	35.8	350	va	lues.	¹ H ^N - ¹⁵ N plane				
		1 Η α	46	350			¹ H ^N - ¹⁵ N plane				
		Methyl ¹³ C	45.3	350			¹ H- ¹³ C plane				
	[U-¹⁵N, ¹³C] A17G FF	¹³ Cα	35.7	350	474 ± 21	1.71 ±	Used to obtain missing I2 state C ^a shifts				
			46	350	0.02						
	50mM Sodium acetate,100mM NaCl,2mM EDTA,2mM NaN ₃ , 20% [² H]- glucose, 100% D ₂ O, pH 5.7	Gly ¹³ C ^α	42.4	400			¹ H ^a - ¹³ C ^a plane				
2		Ser ¹³ C ^β	40.8	350	Exc	hange	¹ H ^{β-13} C ^β plane				
		Thr ¹³ C ^β	46	350	parame fixed to	ters were the above	${}^{1}H^{\beta}-{}^{13}C^{\beta}$ plane				
		¹ Ηα	46	350	values.		¹ H ^{α-13} C ^α plane				
		Methyl ¹ H	46	400			¹ H- ¹³ C plane				
	[U- ¹⁵ N], ¹³ Ca	13 C a	38.3	350	501 ± 28	1.67 ±	Obtain the S56 ¹³ C ^a shift (¹ H ^a -1 ³ C ^a plane).				
	50mM Sodium acetate,100mM NaCl,2mM EDTA,2mM NaN ₃ , 20% [² H]- glucose, 100% D ₂ O, pH 5.7		49.3	350	0.02						
3			35.5	350	Exc	hange ters were					
		¹ Ηα	45.6	350	fixed to va	the above lues.	Obtain the S56 1 H ^a shift (1 H ^a - 13 C ^a plane).				
	[U- ¹⁵ N, ¹³ C], ~50% ² H		24 7	350							
A	50mM Sodium acetate,100mM	Gly ¹ Ha	<u></u>	0.00	456 + 70	1 85 + 0 1	Gly ¹ Hª CEST experiment (¹ Hª-13Cª				
4	NaCl,2mM EDTA,2mM NaN₃, 20% [²H]- glucose, 100% D₂O, pH 5.7		39.6	350		1.85 ± 0.1	plane).				

Table S2. Summary of the (16.4 T) CEST NMR experiments carried out at 20 °C using various A17G FF samples to obtain the chemical shifts of the I2 state. Sample numbers are from Table S1.

Table S3. ϖ_F and the CEST derived $\Delta \varpi_{FI2}$ values of A17G FF. The I2 state chemical shifts (ϖ_{I2}) can be calculated from the F state chemical shifts (ϖ_F) and $\Delta \varpi_{FI2}$ values as $\varpi_{I2} = \varpi_F + \Delta \varpi_{FI2}$. Experiments were performed at 20 °C using the samples and experiments listed in tables S1 and S2 respectively.

Residue	Nucleus	ϖ _F (ppm)	Δ _{መFl2} (ppm)
W11	С	175.27	0.6 ± 0.1
W11	Cα	56.98	-0.7 ± 0.1
W11	Сβ	29.29	0.0 ± 0.1
W11	Ηα	4.802	0.00 ± 0.02
N12	С	175.35	0.3 ± 0.2
N12	Са	54.65	-0.5 ± 0.1
N12	СВ	40.27	0.0 ± 0.1
N12	HN	9.065	-0.15 ± 0.08
N12	Ηα	4,762	0.13 ± 0.03
N12	N	120.05	-1.7 ± 0.1
T13	C	175.36	0.0 ± 0.1
T13	Са	59.80	-04 + 03
T13	СВ	72.26	0.6 + 0.2
T13	Cv2	21.87	0.0 ± 0.1
T13	HN	7 774	0.08 ± 0.07
T13	На	4 733	-0.15 ± 0.03
T13	Hv2	1.322	0.00 ± 0.00
T13	N	109.97	15 ± 0.01
K14	C C	179.91	-03 + 02
K14	0	59.74	-0.3 ± 0.2
K14	Cu	01 50	0.0 ± 0.1
K14	Ср	31.30	0.0 ± 0.2
K14		8.957	-0.18 ± 0.09
K14	Πū	4.240	0.14 ± 0.03
K14	N C	123.13	-0.8 ± 0.1
EIS	0	179.22	0.4 ± 0.3
E15	Ca	59.61	0.0 ± 0.2
E15	Cβ	28.99	0.0 ± 0.1
E15	HN	8.533	-0.15 ± 0.06
E15	Ηα	4.019	0.19 ± 0.01
E15	N	119.23	0.2 ± 0.1
E16	C	179.99	0.3 ± 0.2
E16	Ca	59.08	0.0 ± 0.1
E16	СВ	30.54	0.0 ± 0.1
E16	HN	8.001	-0.15 ± 0.05
E16	Ηα	4.163	-0.20 ± 0.02
E16	N	119.82	-0.7 ± 0.1
G17	C	174.77	0.0 ± 0.1
G17	Са	47.09	-0.2 ± 0.2
G17	HN	8.369	-0.13 ± 0.02
G17	Ha2	3.349	-0.46 ± 0.01
G17	Ηα1	3.660	-0.90 ± 0.01
G17	N	109.75	-0.3 ± 0.1
K18	С	178.87	0.6 ± 0.3
K18	Cα	60.29	0.0 ± 0.1
K18	Сβ	32.50	0.0 ± 0.1
K18	HN	8.346	-0.13 ± 0.01
K18	Ηα	3.761	0.22 ± 0.01
K18	Ν	121.31	0.0 ± 0.1
Q19	С	178.06	0.0 ± 0.1
Q19	Сα	58.67	0.1 ± 0.1
Q19	Сβ	27.89	0.0 ± 0.1
Q19	HN	7.842	-0.15 ± 0.01
Q19	Ηα	4.043	0.23 ± 0.01

Q19	Ν	118.41	0.2 ± 0.1
A20	С	178.95	0.8 ± 0.2
A20	Сα	54.89	0.0 ± 0.1
A20	Сβ	18.18	0.0 ± 0.1
A20	HN	7.947	-0.19 ± 0.12
A20	Ηα	4.124	-0.16 ± 0.05
A20	Нβ	1.286	0.08 ± 0.02
A20	N	122.53	-0.3 ± 0.1
F21	C	177.08	04 + 01
F21	Ca	61 79	0.7 ± 0.2
F21	Cß	40.36	0.7 ± 0.2
F21	ни	8 106	0.35 ± 0.07
E21	На	3 746	0.00 ± 0.07
F01	N	110.00	0.00 ± 0.02
F21	N C	170.00	0.5 ± 0.1
K22	0	178.90	0.0 ± 0.1
K22	Ca	60.45	0.0 ± 0.2
K22	CR	32.00	0.0 ± 0.1
K22	HN	8.019	-0.16 ± 0.01
K22	Ηα	3.886	0.20 ± 0.01
K22	N	117.43	0.4 ± 0.1
E23	С	178.52	0.2 ± 0.2
E23	Са	59.35	0.0 ± 0.1
E23	Сβ	29.15	0.0 ± 0.1
E23	HN	8.284	-0.18 ± 0.09
E23	Ηα	3.916	0.00 ± 0.03
E23	Ν	120.20	-0.9 ± 0.1
L24	С	176.64	1.2 ± 0.1
L24	Сα	58.24	0.5 ± 0.1
L24	Сβ	40.95	0.0 ± 0.1
L24	Cδ1	22.93	-0.4 ± 0.2
L24	Cδ2	25.85	-0.2 ± 0.3
124	HN	7 746	-0.16 + 0.05
1.24	На	3 753	0.00 + 0.02
1 24	Ηδ1	0.882	0.12 + 0.02
1.24	Нб2	0.002	0.12 ± 0.02
1.24	N	100.710	0.11 ± 0.00
L24		177.64	-0.3 ± 0.1
L25	0	177.04	0.4 ± 0.1
L25	Ca	57.97	0.0 ± 0.1
L25	Cβ	40.80	0.0 ± 0.1
L25	Cδ1	27.25	0.0 ± 0.2
L25	Сδ2	24.24	-0.9 ± 0.1
L25	HN	7.211	-0.15 ± 0.08
L25	Ηα	3.350	0.15 ± 0.02
L25	Hδ1	0.868	-0.41 ± 0.02
L25	Ηδ2	0.675	0.00 ± 0.04
L25	Ν	117.59	0.2 ± 0.1
K26	С	181.22	0.0 ± 0.2
K26	Сα	58.97	-0.3 ± 0.2
K26	Сβ	33.05	0.0 ± 0.1
K26	HN	7.543	-0.04 ± 0.05
K26	Ηα	4.239	0.21 ± 0.01
K26	Ν	115.56	2.2 ± 0.1
E27	С	179.10	0.0 ± 0.2
E27	Са	59.47	0.0 ± 0.1
E27	СВ	30.35	0.3 ± 0.2
E27		9 017	-0.34 + 0.01
1	HIN	0.017	0.07 ± 0.01
F27	HN	3 078	0.33 ± 0.01
E27	ΗΝ Ηα	3.978	0.33 ± 0.01
E27 E27	HN Ha N	3.978 122.66	0.33 ± 0.01 -0.1 ± 0.1
E27 E27 K28	HN Ha N C	3.978 122.66 174.54	$\begin{array}{rrrr} 0.33 & \pm & 0.01 \\ \hline -0.1 & \pm & 0.1 \\ \hline 0.0 & \pm & 0.1 \\ \hline 0.0 & -0.2 \end{array}$
E27 E27 K28 K28	HN Ha N C Ca	3.978 122.66 174.54 54.12	$\begin{array}{rrrr} 0.33 & \pm & 0.01 \\ \hline -0.1 & \pm & 0.1 \\ \hline 0.0 & \pm & 0.1 \\ \hline 0.0 & \pm & 0.2 \\ \hline \end{array}$

K28	HN	7.760	-0.15 ± 0.08
K28	Ηα	4.220	0.21 ± 0.01
K28	Ν	114.05	1.2 ± 0.1
R29	С	176.50	0.4 ± 0.1
R29	Cα	55.87	-0.2 ± 0.1
R29	Сβ	26.66	0.0 ± 0.1
R29	HN	7.914	-0.17 ± 0.04
R29	Ηα	3.867	0.22 ± 0.05
R29	N	115.37	-0.1 ± 0.1
V30	Ca	61.68	-0.3 ± 0.2
V30	Cv1	21.19	-0.5 ± 0.1
V30	Cv2	22,48	0.0 ± 0.1
V30	HN	7 567	-0.10 + 0.06
V30	На	3 897	-0.43 ± 0.05
V30	Hv1	1 220	-0.09 ± 0.02
V30	Hv2	1 107	0.03 ± 0.02
V30	N	120.30	-0.12 ± 0.02
P21	N C	177.52	-0.4 ± 0.1
F31	0	62.29	0.0 ± 0.1
P31	Ca	03.38	-0.3 ± 0.1
P31	CB	32.86	-U.4 ± U.2
P31	Ηα	4.635	0.22 ± 0.01
S32	C	1/3.89	0.0 ± 0.1
S32	Ca	60.86	0.4 ± 0.3
S32	Сβ	62.68	0.8 ± 0.3
S32	HN	8.777	-0.16 ± 0.04
S32	Ηα	3.998	0.24 ± 0.01
S32	Ν	115.90	0.2 ± 0.1
N33	С	175.81	0.0 ± 0.1
N33	Сα	51.61	-0.3 ± 0.1
N33	Сβ	37.78	0.0 ± 0.1
N33	HN	7.832	-0.18 ± 0.09
N33	Ηα	4.819	0.17 ± 0.02
N33	N	115.24	-0.2 ± 0.1
A34	С	177.66	0.0 ± 0.1
A34	Сα	52.42	0.5 ± 0.2
A34	Св	19.59	0.0 ± 0.1
A34	HN	7.534	-0.15 ± 0.01
A34	На	4 464	0.25 ± 0.02
Δ34	НВ	1 610	-0.02 ± 0.02
A34	N	122 75	-0.02 ± 0.01
C25	C C	175 47	-0.2 ± 0.2
000	0	56.60	0.4 ± 0.1
000	Cu	50.09 65 45	0.0 ± 0.1
535	CB	05.45	0.0 ± 0.4
535	HN	ö.322	-0.10 ± 0.07
535	Ηα	4.884	0.00 ± 0.08
535	N	115.79	0.0 ± 0.1
W36	C	175.92	0.6 ± 0.1
W36	Cα	59.63	0.0 ± 0.1
W36	HN	9.379	-0.17 ± 0.03
W36	Ηα	3.980	0.15 ± 0.01
W36	Ν	122.56	0.8 ± 0.1
E37	С	178.96	0.4 ± 0.1
E37	Ca	60.89	-0.7 ± 0.1
E37	Сβ	28.63	0.0 ± 0.1
E37	HN	8.621	-0.22 ± 0.08
E37	Ηα	3.329	0.22 ± 0.01
E37	Ν	115.97	-0.2 ± 0.1
Q38	С	178.32	0.0 ± 0.1
Q38	Ca	58.17	0.0 ± 0.1
Q38	СВ	28.67	0.0 ± 0.1
Q38	HN	7,440	-0.14 + 0.08
200			J 0.00

Q38	Ηα	3.835	0.20 ± 0.03
Q38	N	117.71	1.8 ± 0.1
Δ39	C	179.28	00 + 03
Δ30	Ca	5/ 96	0.0 ± 0.0
A30	CB	10.47	0.0 ± 0.1
A30	Ср	9 117	0.0 ± 0.1
A00	Ha	0.117	-0.14 ± 0.00
A39	Πū	3.052	0.17 ± 0.02
A39	нβ	0.817	0.00 ± 0.02
A39	N	122.17	-0.2 ± 0.1
M40	C	177.28	1.3 ± 0.1
M40	Са	59.15	-2.7 ± 0.1
M40	Сβ	31.43	0.0 ± 0.3
M40	Сε	16.99	-0.6 ± 0.1
M40	HN	8.397	-0.45 ± 0.01
M40	Ηα	3.040	0.74 ± 0.01
M40	Ηε	2.010	-0.50 ± 0.01
M40	Ν	116.35	-1.9 ± 0.1
K41	С	178.26	0.3 ± 0.1
K41	Ca	58.85	-0.3 ± 0.1
K41	Сβ	32.38	0.0 ± 0.1
K41	HN	6.764	0.47 ± 0.01
K41	Ηα	3.702	0.23 ± 0.01
K41	Ν	114.10	3.2 ± 0.1
M42	С	178.00	0.3 ± 0.2
M42	Са	57.95	0.0 ± 0.2
M42	Cß	34.31	-19 ± 04
M/2	Cs	17.34	1.0 ± 0.1
M42	HN	7 322	0.0 ± 0.1
M42	Ha	1.322	0.14 ± 0.03
10142	Πū	4.132	0.00 ± 0.02
M42	HE	2.159	0.00 ± 0.04
M42	N	113.94	0.8 ± 0.1
143	C	175.84	0.9 ± 0.1
l43	Сα	62.11	0.0 ± 0.1
l43	Сβ	39.04	0.0 ± 0.2
l43	Cy2	16.87	0.0 ± 0.1
l43	Сδ	13.86	0.4 ± 0.1
l43	HN	7.101	0.30 ± 0.02
l43	Ηα	4.350	0.04 ± 0.07
l43	Hy2	0.544	-0.07 ± 0.02
l43	Ηδ	0.696	0.00 ± 0.03
l43	Ν	108.94	1.7 ± 0.1
144	С	174.51	0.6 ± 0.1
144	Cα	63.12	-0.4 ± 0.1
144	Сβ	38.02	0.0 ± 0.1
44	Cv2	15.73	0.9 ± 0.1
44	<u>م</u>	14 69	0.3 ± 0.1
44	HN	7 197	-0.06 + 0.01
144	На	2 851	0.36 ± 0.01
144	Huo	2.001	0.00 ± 0.02
144	17ץ2 נוג	0.035	0.00 ± 0.03
144		0.024	0.00 ± 0.11
144	N O	119.29	0.0 ± 0.1
N45	C	1/4.61	0.5 ± 0.1
N45	Са	52.96	-0.1 ± 0.1
N45	Сβ	38.27	0.0 ± 0.1
N45	HN	7.771	-0.23 ± 0.09
N45	Ηα	4.708	0.30 ± 0.01
N45	N	116.29	0.5 ± 0.1
D 40			
D46	Са	51.72	0.6 ± 0.2
D46	Ca HN	51.72 7.677	0.6 ± 0.2 0.31 ± 0.01
D46 D46 D46	Ca HN Ha	51.72 7.677 4.976	$\begin{array}{rrrr} 0.6 & \pm & 0.2 \\ 0.31 & \pm & 0.01 \\ 0.10 & \pm & 0.03 \end{array}$

P47	С	178.80	-0.4 ± 0.1
P47	Сα	65.41	-0.4 ± 0.1
P47	Сβ	32.69	0.0 ± 0.1
P47	Ηα	4.515	0.22 ± 0.04
R48	С	177.22	0.8 ± 0.2
R48	Сα	57.87	-0.4 ± 0.3
R48	Св	30.20	-1.7 ± 0.4
B48	HN	9 778	-0.30 + 0.01
R48	На	3 962	0.42 + 0.02
R48	N	115.95	16 ± 0.02
V/Q	0	176.60	1.0 ± 0.1
V/0	Ca	61.37	-0.4 ± 0.2
149 V40	Cu	01.57	-0.4 ± 0.2
149	Ср	0.450	0.0 ± 0.1
149		0.402	-0.17 ± 0.04
¥49	Ηα	4.028	0.22 ± 0.01
Y49	N	121.70	-3.4 ± 0.1
S50	C	173.92	2.8 ± 0.1
\$50	Ca	58.06	3.7 ± 0.1
S50	Сβ	63.39	-1.3 ± 0.1
S50	HN	7.291	0.50 ± 0.01
S50	Ηα	4.326	-0.36 ± 0.04
S50	Ν	108.01	5.2 ± 0.1
A51	С	176.80	2.7 ± 0.1
A51	Сα	54.39	0.5 ± 0.1
A51	Сβ	19.02	0.0 ± 0.1
A51	HN	7.513	0.28 ± 0.01
A51	Ηα	3.379	0.94 ± 0.01
A51	Ηβ	0.565	1.09 ± 0.01
A51	N	123.39	-1.9 ± 0.1
L52	С	175.39	2.3 ± 0.1
1.52	Са	52.28	30 + 01
1.52	Cß	43.25	0.0 ± 0.1
1.52	Ορ Cδ1	26.13	0.0 ± 0.2
1.50	050	20.10	0.0 ± 0.1
L32		23.11	-0.4 ± 0.1
L52	HIN	5.924	1.77 ± 0.01
L52	Ηα	4.343	0.58 ± 0.02
L52	H01	0.213	0.78 ± 0.01
L52	Ηδ2	0.699	-0.10 ± 0.03
L52	N	111.95	6.6 ± 0.1
A53	С	178.73	0.6 ± 0.1
A53	Сα	54.85	0.6 ± 0.1
A53	Сβ	20.07	-1.7 ± 0.1
A53	HN	8.582	-1.04 ± 0.01
A53	Ηα	3.978	-0.23 ± 0.02
A53	Ηβ	1.497	-0.12 ± 0.01
A53	Ν	125.61	-2.9 ± 0.1
K54	С	177.46	0.8 ± 0.2
K54	Cα	55.46	-0.7 ± 0.2
K54	Cß	33.85	-1.7 ± 0.2
K54	Οp		
K54	HN	7.953	-0.16 ± 0.05
	HN Ha	7.953	-0.16 ± 0.05 -0.08 ± 0.07
K54	HN Ha N	7.953 4.432 115.28	-0.16 ± 0.05 -0.08 ± 0.07 -1.0 ± 0.1
K54	HN Ha N	7.953 4.432 115.28	-0.16 ± 0.05 -0.08 ± 0.07 -1.0 ± 0.1 0.8 ± 0.2
K54 L55	HN Ha N C	7.953 4.432 115.28 178.14	-0.16 ± 0.05 -0.08 ± 0.07 -1.0 ± 0.1 0.8 ± 0.2 -3.0 ± 0.1
K54 L55 L55	HN Ha N C Ca	7.953 4.432 115.28 178.14 58.49	$\begin{array}{rrrr} -0.16 \ \pm \ 0.05 \\ -0.08 \ \pm \ 0.07 \\ -1.0 \ \pm \ 0.1 \\ 0.8 \ \pm \ 0.2 \\ -3.0 \ \pm \ 0.1 \\ \end{array}$
K54 L55 L55 L55	ΗΝ Ηα Ν C Ca Cβ	7.953 4.432 115.28 178.14 58.49 41.03	$\begin{array}{rrrr} -0.16 \pm 0.05 \\ -0.08 \pm 0.07 \\ -1.0 \pm 0.1 \\ 0.8 \pm 0.2 \\ -3.0 \pm 0.1 \\ 1.4 \pm 0.3 \\ 0.5 \end{array}$
K54 L55 L55 L55 L55	Op HN Ha C Ca Cβ Cδ1	7.953 4.432 115.28 178.14 58.49 41.03 23.57	$\begin{array}{rrrr} -0.16 \pm 0.05 \\ -0.08 \pm 0.07 \\ -1.0 \pm 0.1 \\ 0.8 \pm 0.2 \\ -3.0 \pm 0.1 \\ 1.4 \pm 0.3 \\ -0.5 \pm 0.2 \end{array}$
K54 L55 L55 L55 L55 L55	HN Ha C Ca Cβ Cδ1 Cδ2	7.953 4.432 115.28 178.14 58.49 41.03 23.57 25.42	$\begin{array}{rrrr} -0.16 \pm 0.05 \\ -0.08 \pm 0.07 \\ -1.0 \pm 0.1 \\ 0.8 \pm 0.2 \\ -3.0 \pm 0.1 \\ 1.4 \pm 0.3 \\ -0.5 \pm 0.2 \\ 0.0 \pm 0.1 \end{array}$
K54 L55 L55 L55 L55 L55 L55	HN Ha N C Ca Cβ Cδ1 Cδ2 HN	7.953 4.432 115.28 178.14 58.49 41.03 23.57 25.42 8.979	$\begin{array}{rrrr} -0.16 \pm 0.05 \\ -0.08 \pm 0.07 \\ -1.0 \pm 0.1 \\ 0.8 \pm 0.2 \\ -3.0 \pm 0.1 \\ 1.4 \pm 0.3 \\ -0.5 \pm 0.2 \\ 0.0 \pm 0.1 \\ -1.41 \pm 0.01 \end{array}$
K54 L55 L55 L55 L55 L55 L55 L55	HN Ha N C Ca Cβ Cδ1 Cδ2 HN Ha	7.953 4.432 115.28 178.14 58.49 41.03 23.57 25.42 8.979 3.695	$\begin{array}{rrrr} -0.16 \pm 0.05 \\ -0.08 \pm 0.07 \\ -1.0 \pm 0.1 \\ 0.8 \pm 0.2 \\ -3.0 \pm 0.1 \\ 1.4 \pm 0.3 \\ -0.5 \pm 0.2 \\ 0.0 \pm 0.1 \\ -1.41 \pm 0.01 \\ 0.60 \pm 0.01 \end{array}$
K54 L55 L55 L55 L55 L55 L55 L55 L55	6p HN Ha N Ca Cβ Cδ1 Cδ2 HN Ha Ha	7.953 4.432 115.28 178.14 58.49 41.03 23.57 25.42 8.979 3.695 1.042	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$

L55	N	127.78	-5.6 ± 0.1
S56	С	177.34	-2.2 ± 0.1
S56	Сα	61.01	-1.2 ± 0.1
S56	HN	8.677	-0.19 ± 0.09
S56	Ηα	3.801	0.61 ± 0.01
S56	N	112.05	12.0 ± 0.1
E57	С	179.17	-2.6 ± 0.1
E57	Са	58 52	-19 + 01
E57	Cß	30.24	-23 ± 01
E57	HN	6 623	1.46 ± 0.01
E57	На	3 967	0.48 ± 0.02
E57	N	121 50	-36 ± 0.02
LJ7	N C	177.06	-3.0 ± 0.1
K50	0	50.00	0.0 ± 0.2
K58	Ca	59.38	0.9 ± 0.1
K58	СВ	34.01	-1.8 ± 0.6
K58	HN	7.584	0.52 ± 0.02
K58	Ηα	2.968	0.40 ± 0.02
K58	N	121.38	-2.5 ± 0.1
K59	С	179.03	0.5 ± 0.1
K59	Cα	60.09	0.0 ± 0.2
K59	Сβ	31.91	0.0 ± 0.1
K59	HN	7.835	0.48 ± 0.01
K59	Ηα	3.447	0.35 ± 0.04
K59	N	116.45	3.1 ± 0.1
Q60	С	178.92	0.0 ± 0.2
Q60	Cα	59.13	0.0 ± 0.1
Q60	Сβ	28.41	-0.5 ± 0.2
Q60	HN	7.626	0.42 ± 0.01
Q60	На	4.098	0.46 ± 0.02
060	N	119.05	0.6 ± 0.1
A61	C C	180.28	-15 ± 0.1
A61	Ca	5/ 9/	1.5 ± 0.1
A01 A61	CR	18.40	0.5 ± 0.2
461	ци	7 959	0.0 ± 0.1
A01		1.000	-0.19 ± 0.11
A61	на	4.243	0.14 ± 0.01
A61	нβ	1.514	0.12 ± 0.02
A61	N	123.09	-0.1 ± 0.1
F62	С	176.27	0.3 ± 0.2
F62	Cα	60.29	0.0 ± 0.2
F62	Сβ	39.45	0.0 ± 0.1
F62	HN	8.681	-0.17 ± 0.05
F62	Ηα	4.640	0.18 ± 0.01
F62	Ν	120.13	-0.6 ± 0.1
N63	С	177.60	0.4 ± 0.3
N63	Ca	56.36	0.0 ± 0.1
N63	Сβ	37.76	0.0 ± 0.1
N63	HN	8.617	-0.19 ± 0.01
N63	Ηα	4.103	-0.10 ± 0.03
N63	N	118.95	0.4 ± 0.1
A64	С	179.65	0.3 ± 0.2
A64	Ca	54.80	0.0 ± 0.1
A64	CR	18 17	0.0 ± 0.1
Δ64	На	4 1/15	-0.14 + 0.05
A04 A64	ПО	1 564	0.17 ± 0.05
N04		177 44	0.01 ± 0.03
201		1//.44	0.0 ± 0.2
Y65	α	60.34	0.0 ± 0.2
Y65	Сβ	38.91	0.0 ± 0.1
Y65	HN	8.080	-0.29 ± 0.09
Y65	Ηα	4.239	0.16 ± 0.02
Y65	Ν	120.83	0.0 ± 0.2
K66	С	177.84	0.3 ± 0.2

K66	Сα	60.01	0.0 ± 0.1
K66	Сβ	32.04	0.0 ± 0.1
K66	HN	8.000	-0.20 ± 0.01
K66	Ηα	3.492	0.21 ± 0.01
K66	Ν	119.09	-0.1 ± 0.1
V67	С	177.14	0.0 ± 0.1
V67	Сα	63.41	-0.2 ± 0.1
V67	Сβ	31.90	0.0 ± 0.1
V67	Cγ1	21.31	0.0 ± 0.1
V67	Cy2	20.91	0.0 ± 0.1
V67	HN	7.249	-0.14 ± 0.01
V67	Ηα	4.004	0.22 ± 0.03
V67	Hy1	0.971	-0.07 ± 0.01
V67	Hy2	1.043	0.00 ± 0.07
V67	Ν	114.19	1.3 ± 0.1
Q68	С	176.53	-0.4 ± 0.3
Q68	Сα	56.59	0.4 ± 0.1
Q68	Сβ	28.87	0.0 ± 0.1
Q68	HN	7.682	-0.14 ± 0.01
Q68	Ηα	4.256	-0.13 ± 0.03
Q68	Ν	120.45	-0.5 ± 0.2
T69	С	174.62	0.2 ± 0.1
T69	Сα	62.53	0.3 ± 0.1
T69	Сβ	69.53	0.0 ± 0.1
T69	Cy2	21.55	0.0 ± 0.1
T69	HN	7.847	-0.16 ± 0.08
T69	Ηα	4.179	0.00 ± 0.04
T69	Hy2	1.110	0.02 ± 0.02
T69	Ν	113.91	0.3 ± 0.1
E70	С	175.47	0.0 ± 0.1
E70	Сα	56.75	-0.2 ± 0.1
E70	Сβ	30.09	0.0 ± 0.1
E70	HN	8.075	-0.19 ± 0.06
E70	Ηα	4.262	0.08 ± 0.03
E70	Ν	123.26	-0.4 ± 0.1
K71	Сα	57.58	0.2 ± 0.1
K71	HN	7.856	-0.14 ± 0.08
K71	Ηα	4.128	0.11 ± 0.05
K71	Ν	127.07	0.1 ± 0.1

Sample	Protein		Temperature (°C)	Experimental Parameters		Fitted Parameters							
No	Labelling	Buffer		<i>B</i> ₁ (Hz)	<i>T_{EX}</i> (ms)	<i>k_{ex,Fl1}</i> (s ⁻¹)	<i>k_{ex,Fl2}</i> (s ⁻¹)	<i>k_{ex,I1I2}</i> (s ⁻¹)	<i>k_{ex,I1U}</i> (s ⁻¹)	<i>p</i> 11 (%)	р _{і2} (%)	ри (%)	Comments
		50mM Sodium		26	450								Data from
6	[U- ¹⁵ N]	acetate,100mM NaCl,2mM	10	52.1	400	784 ±	406 ±	1600 ±	11000 ±	0.27 ±	0.83 ±	0.16 ±	K26, K28, R29, N33, E37, K41,
		EDTA, 2mm NaN ₃ , 10% D ₂ O, pH 5,7		104.1	350	67	5	113	1064	0.01	0.01	0.02	M42, I43, S50, L52, L55, and
				208.3	350								S56 was used.
		50mM Sodium		26	450								
6	[U- ¹⁵ N]	acetate,100mM NaCl,2mM		51.7	400	372 +	216	886 8	8370 +	0.13	0.55 +	0.11	
	[0]	EDTA, 2mM NaN₃, 10%		103.5	350	21	3	39	€ € 10	0.01	0.01	0.01	
		D ₂ O, ph 5.7		207	350								
	[U- ¹⁵ N]	50mM Sodium acetate,100mM NaCl,2mM EDTA, 2mM NaN₃, 0.25 M urea, 10% D ₂ O, pH 5.7		26	450				836 8890 ± ± 30 462	0.19 ± 0.01	0.52	0.25	For <i>m</i> -value analysis Data from residues T13, K26, K28, R29, N33, E37, K41,
7				51.9	400	385 ± 16	225 ± 2	836					
				103.7	350			30			± 0.01	± 0.01	
				207.4	350								
	[U- ¹⁵ N]	50mM Sodium acetate,100mM NaCl,2mM EDTA, 2mM NaN ₃ , 0.50 M urea, 10% D ₂ O, pH 5.7	2.5	26	450	371 ± 15	236 ± 3						
				52	400			828	828 9490 [±] [±] 26 351	0.25 0. ± 0.01 0.	0.52	0.52	
8				104.1	350			± 26 \$			0.01	± 0.01	
				208.2	350								M42, I43, S50, L52, L55, and
		50mM Sodium acetate,100mM NaCl,2mM		26	450	345	249	853 9510					- S56 was used.
				51.9	400				9510	0.34	0.51	1.11	
9	[U-15N]	EDTA, 2mM NaN ₃ , 0.75 M		103.8	350	± 16	± 3	± 27	± 346	± 0.01	± 0.01	± 0.02	
		D ₂ O, pH 5.7		207.5	350								
		50mM Sodium		26	450								
		acetate,100mM NaCl,2mM		52.2	400	322	256	885 ± 30	9290 ± 370	0.43 ± 0.02	0.51	2.06	
10	[U- ¹⁵ N]	EDTA, 2mM NaN₃, 1.0 M		104.3	350	± 18	± 4				± 0.01	± 0.03	
		Urea, 10% D₂O, pH 5.7		208.6	350								

Table S4. Four-state exchange parameters obtained by analysing A17G FF ¹⁵N CEST data under different conditions. Sample numbers are from Table S1.

Legend for supplementary file A17GFF_I2_lowest10.pdb. This file contains the coordinates of the ten lowest energy A17G FF I2 state structures obtained using the CS-ROSETTA program (41) as described in the Materials and Methods section of the *SI Appendix*.

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