Effects of Proteolytic Digestion by Chymotrypsin on the Structure and Catalytic Properties of Reduced Nicotinamide-Adenine Dinucleotide-Ubiquinone Oxidoreductase from Bovine Heart Mitochondria

By SUSAN E. CROWDER and C. IAN RAGAN* Department of Physiology and Biochemistry, University of Southampton, Southampton 509 3TU, U.K.

(Received 13 January 1977)

1. Incubation of NADH-ubiquinone oxidoreductase (Complex I) with chymotrypsin caused loss of rotenone-sensitive ubiquinone-1 reduction and an increase in rotenoneinsensitive ubiquinone reduction. 2. Within the same time-course, NADH- $K_3Fe(CN)_6$ oxidoreductase activity was unaffected. 3. Mixing of chymotrypsin-treated Complex ^I with Complex III did not give rise to NADH-cytochrome c oxidoreductase activity. 4. Gel electrophoresis in the presence of sodium dodecyl sulphate revealed selective degradation of several constituent polypeptides by chymotrypsin. 5. With higher chymotrypsin concentrations and longer incubation times, a decrease in NADH- $K_3Fe(CN)_6$ oxidoreductase was observed. The kinetics of this decrease correlated with solubilization of the low-molecular-weight type-II NADH dehydrogenase (subunit mol.wts. ⁵³⁰⁰⁰ and 27000) and with degradation of a polypeptide of mol.wt. 30000. 6. Phospholipiddepleted Complex ^I was more rapidly degraded by chymotrypsin. Specifically, a subunit of mol.wt. 75000, resistant to chymotrypsin in untreated Complex I, was degraded in phospholipid-depleted Complex I. In addition, the 30000-mol.wt. polypeptide was also more rapidly digested, correlating with an increased rate of transformation to type II NADH dehydrogenase.

Selective proteolytic digestion of enzymes has proven to be a useful technique for the study of the structure and function of enzymes. As part of a general survey of the properties of mitochondrial NADHubiquinone oxidoreductase (Complex ^I of Hatefi et al., 1962) we have been investigating the effects of proteolytic degradation on the activity of this enzyme and on its polypeptide subunits analysed by polyacrylamide-gel electrophoresis in the presence of sodium dodecyl sulphate (Weber & Osbom, 1969). It has been known for some time that both trypsin and chymotrypsin could transform the soluble high-molecular-weight (type I) NADH dehydrogenase (Ringler et al., 1963) into the low-molecularweight (type II) NADH dehydrogenase (Cremona et al., 1963). In view of the similarity between Complex ^I and the soluble type-I NADH dehydrogenase (for ^a general review of NADH dehydrogenation see Ragan, 1976c), we have examined the effects of chymotrypsin on Complex ^I with a view to determining the underlying structural alterations resulting in modification of enzyme activities and transformation to the type II NADH dehydrogenase. The effects of trypsin have been described (Ragan,

1976b). In particular, it was found that nicotinamideadenine dinucleotide transhydrogenase was very rapidly inactivated by trypsin and that this loss of activity appeared to be correlated with degradation of one of the Complex-I subunits.

Materials and Methods

Chemicals

a-Chymotrypsin (EC 3.4.21.1) (type II) and phenylmethanesulphonyl fluoride were obtained from Sigma (London) Chemical Co., Kingstonupon-Thames, Surrey, U.K. Ubiquinone-1 was generously given by Hoffman La Roche, Basle, Switzerland.

Biological materials

Complex ^I (NADH dehydrogenase, EC 1.6.99.3) was prepared as described by Hatefi et al. (1962). Phospholipid depletion of Complex ^I with cholate and $(NH_4)_2SO_4$ was carried out in the absence of reducing agents as described by Ragan & Racker (1973). Complex III (ubiquinone-cytochrome c oxidoreductase) was prepared bythe method ofRieske (1967).

^{*} To whom reprint requests should be addressed.

Chymotrypsin treatment

Complex ^I (1 mg of protein/mi) was incubated with chymotrypsin (20 or 100μ g of protein/ml) in 50 mmsodium phosphate, pH7.6, at 21°C. At intervals, portions were removed for assay or for preparation of samples for electrophoresis.

Gel electrophoresis

After incubation with chymotrypsin, portions (200 μ l) were incubated at 21°C for 30min with 10 μ l of 5 mM-phenylmethanesulphonyl fluoride in ethanol. The samples were then dissociated by adding 20μ of 10% (w/v) SDS,* 20 μ l of 10% (v/v) 2-mercaptoethanol, 2μ l of 0.4% Bromophenol Blue, 20μ l of ¹ M-sucrose and boiling for 2min. Electrophoresis of up to 100μ g of protein was performed on $12 \text{cm} \times$ 6mm (internal diam.) gels containing 12.5% (w/v) acrylamide and 0.34% bisacrylamide at 5.5mA/gel for 16h by the procedure of Weber & Osborn (1969). Alternatively, electrophoresis was performed as described by Neville & Glossman (1974) by using ^a 3% (w/v) acrylamide stacking gel, a running gel as above, and the discontinuous buffer system J4179 of Jovin et al. (1971). It was unnecessary to change the ionic composition of the sample buffer from sodium phosphate to the Tris/borate buffer used in the cathodic compartment of the electrophoresis. Electrophoresis was performed at 2mA/gel for 4h. Staining and destaining were as described by Weber & Osborn (1969). Destained gels were scanned with a Joyce-Loebl densitometer (Joyce, Loebl and Co., Gateshead, Co. Durham, U.K.). The scanning speed was 2.5 cm/min, and the optical slit-width was approx. 0.3 mm. For duplicate samples electrophoresed in parallel, the peak heights from the densitometer traces agreed within $10-15\%$ of each other. When untreated and chymotrypsin-treated samples were compared, changes in peak height of up to 20% were therefore not considered significant.

Polypeptide molecular weights were determined separately for each electrophoresis system as described by Ragan (1976a). For reference, these are listed in Table 1.

Assays

NADH-ubiquinone-1 oxidoreductase and $NADH-K₃Fe(CN)₆ oxidoreductase were measured$ at 21°C as described by Ragan (1976a). NADHcytochrome c oxidoreductase was assayed at 30 $\rm ^{\circ}C$ as described by Hatefi & Rieske (1967). Protein was measured by the method of Lowry et al. (1951), with bovine serum albumin (Sigma, fraction V) as a standard.

* Abbreviation: SDS, sodium dodecyl sulphate.

Table 1. Subunit composition of Complex I The molecular weights of the Complex ^I subunits were determined separately for both electrophoretic systems as described by Ragan (1976a). Subunits that were quantitatively purified into the iron-protein fraction or the type-II NADH dehydrogenase obtained from treatment with chaotropic agents (Hatefi & Stempel, 1967) are also indicated.

Subunit molecular weights determined in the system of:

Weber & Osborn (1969)	Jovin et al. (1971)
75000 (iron-protein)	75000 (iron-protein)
53000	53000* (type II dehydrogenase)
	49000* (iron-protein)
42000	42000
39000	38000
33000+	30000
30000† (iron-protein)	
27000 (type II de-	27000 (type II dehydrogenase)
hydrogenase)	
25000	26000
	25000
23500	23500
22000	22000
20500	20500
18000	18000
15500	16500
	15500
8000	8000
5000	5000

* The subunits of mol.wt. 53000 and 49000 co-migrate in the Weber & Osbom (1969) system.

t The subunits of mol.wt. 33000 and 30000 co-migrate in the Jovin et al. (1971) system.

Results

Effect of limited digestion by chymotrypsin on Complex-I activity

Incubation of Complex ^I (1 mg of protein/ml) with chymotrypsin (20 μ g of protein/ml) at 21°C caused progressive loss of rotenone sensitivity of NADHubiquinone-1 oxidoreductase, although the overall rate of ubiquinone-1 reduction was only slightly decreased (Fig. 1). Closer examination of this effect revealed that the V_{max} for ubiquinone-1 was unaffected by chymotrypsin digestion $(6.0 \,\mu\text{mol/min})$ per mg of protein at 30°C), but the K_m increased from 54 μ M to 85 μ M after 4h, by which time the percentage inhibition by rotenone had fallen from 90 to 10%. Within the same time-course, NADH- $K_3Fe(CN)_{6}$ oxidoreductase was unaffected.

Mixing of Complex ^I and Complex III leads to reconstitution of NADH-cytochrome c oxidoreductase activity (Hatefi, 1963). Table 2 shows that mixing of chymotrypsin-treated Complex ^I with Complex III gave rise to low rates of NADH-cytochrome c oxidoreductase, which were only partially inhibited by rotenone. Therefore the rapid rotenone-insensitive pathway of ubiquinone-1 reduction in the modified Complex ^I is not related to the physiological pathway of ubiquinone-10 reduction.

Effect of limited digestion by chymotrypsin on the polypeptide composition of Complex I

The polypeptides of Complex ^I were resolved by polyacrylamide-gel electrophoresis in the presence of SDS (Weber & Osborn, 1969). Fig. ² shows the progressive changes in polypeptide profile with time when Complex ^I was treated with chymotrypsin as in Fig. 1. Control experiments established that no proteolytic

Fig. 1. Effect of limited digestion by chymotrypsin on Complex-I activities

Complex I was incubated with chymotrypsin $(20 \mu g)$ ml) and assayed as described in the Materials and Methods section. (.), NADH-ubiquinone-1 oxidoreductase; (A), NADH-ubiquinone-1 oxidoreductase assayed in the presence of 1μ g of rotenone/ml; (\blacksquare), NADH-K₃Fe(CN)₆ oxidoreductase.

digestion occurred after treatment with SDS. Polypeptides of mol.wts. 75000, 53000 and 30000 are major components of the iron-protein fragment obtained from Complex ^I by chaotropic resolution (Hatefi & Stempel, 1967; Ragan, 1976a). The lowmolecular-weight type II NADH dehydrogenase has subunits ofmol.wts. 53 000 and 27000 (Ragan, 1976a). In the experiment of Fig. 2, the 75000- and 30000-

Direction of migration towards anode \rightarrow

Fig. 2. Analysis of the effect of limited digestion by chymotrypsin on the polypeptides of Complex I by gel electrophoresis in the Weber & Osborn (1969) system

Complex ^I was treated with chymotrypsin as in Fig. ¹ and, after the indicated times, samples were electrophoresed according to Weber & Osborn (1969) and the gels were scanned as described in the Materials and Methods section. Molecular weights are indicated in thousands by the postscript K.

Table 2. Reconstitution of NADH-cytochrome c oxidoreductase

Complex I (1 mg of protein in 1 ml) was incubated with or without 20μ g of chymotrypsin at 21° C for 5 h as described in the Materials and Methods section. Samples were diluted to approx. 10ml with 50mm-sodium phosphate, pH7.6, and centrifuged for 30 min at 150000g and 4°C inthe 10 × 10ml rotor of an MSE-65 centrifuge. The pellets were suspended by homogenization in the same buffer and re-centrifuged as above. The pellets were finally homogenized in approx. 0.3ml of 0.67M-sucrose/50mM-Tris/HCl, pH8.0, and diluted to a final protein concentration of 2.5mg/mi. Portions $(20\,\mu$) were incubated at 0°C for 45 min with or without 20 μ l of Complex III (35 mg of protein/ml) and 0.67M-sucrose/ 50mM-Tris/HCl, pH8.0, to a final volume of 50 μ l. Complex III alone did not catalyse NADH-cytochrome c oxidoreduction. Abbreviation: N.D., not determined.

mol.wt. polypeptides were clearly unaffected, but one of the 53000-mol.wt. polypeptides and others of mol.wts. 42000, 39000 and 33000 were degraded. To distinguish the two 53000-mol.wt. polypeptides and to provide better resolution of the low-molecularweight polypeptides, electrophoresis was also performed at pH9.5 in the discontinuous buffer system J4179 of Jovin et al. (1971) . In this system (Fig. 3) the 53000-mol.wt. polypeptide of the iron-protein fraction migrates with an apparent mol.wt. of 49000, and it is clear that it is this polypeptide which is susceptible to proteolytic digestion. The subunits of the type II NADH dehydrogenase (mol.wts. 53000 and 27000) were clearly unaffected. The 33000-mol.wt. polypeptide co-migrates with the 30000-mol.wt. polypeptide in the Jovin et al. (1971) system, whereas the low-molecular-weight components of Complex ^I are more clearly resolved than in the Weber & Osborn (1969) system. Of these, the 26000-, 23500-, 22000-, 18000- and 15500-mol.wt. polypeptides were to a greater or lesser extent degraded, but the 25000-, 20500- and 16500-mol.wt. polypeptides were relatively resistant.

The time-course of degradation of the Complex-I subunits was approximately parallel to the loss of rotenone sensitivity, except for the 53 000-mol.wt. polypeptide, which was degraded far more rapidly than the others and more rapidly than the changes in enzyme activity. The effect of chymotrypsin on the 53 000-mol.wt. polypeptide of the iron-protein fragment were confirmed by isolating the latter from chymotrypsin-treated Complex I. In Fig. 4, ironprotein isolated from intact and chymotrypsintreated Complex ^I was compared with isolated type II dehydrogenase; clearly the iron-protein shows specific degradation of the 53000-mol.wt. polypeptide.

Effect of extensive digestion by chymotrypsin on Complex Iactivity

Fig. 1 shows that NADH- $K_3Fe(CN)_6$ oxidoreductase activity showed considerable resistance to chymotrypsin. Loss of this activity was only encountered under conditions of more extensive degradation by chymotrypsin, e.g. 100μ g of protein/ ml for longer times. Little effect on this activity was detected for approx. 120min, after which the rate declined to a minimum of about 30 $\%$ of the original after 300min.

Effect of extensive digestion by chymotrypsin on the polypeptide composition of Complex I

The polypeptide profile after 120 min digestion was similar to that of Fig. 2 and not greatly changed after 300min. However, the decrease in NADH- $K_3Fe(CN)_6$ oxidoreductase activity was paralleled by solubilization of type-II NADH dehydrogenase,

Direction of migration towards anode \rightarrow

Fig. 3. Analysis of the effect of limited digestion by chymotrypsin on the polypeptides of Complex I by gel electrophoresis in the Neville & Glossmann (1974) system Samples from an experiment similar to that of Fig. 2 were analysed by electrophoresis by the method of Neville & Glossmann (1974).

Direction of migration towards anode \rightarrow

Fig. 4. Subunit composition of isolated iron-protein and type II dehydrogenase

Iron-protein (a) and type II dehydrogenase (c) were purified from Complex ^I as described by Hatefi & Stempel (1967). Iron-protein from chymotrypsintreated Complex ^I (b) was isolated as follows. Complex ^I (10mg of protein in lOml of 50mMsodium phosphate, pH7.6) was incubated with chymotrypsin (200 μ g of protein) for 100 min at 21°C. After cooling on ice, the sample was centrifuged at 100000g for 20 min at 4° C in the 10×10 ml rotor of a MSE ⁶⁵ centrifuge. The pellet was resuspended by homogenization in 1 ml of 0.67 M-sucrose/50 mm-Tris/HCl, pH8.0 at 0° C, and left for 1 h at 0° C after the addition of a 25mm solution of phenylmethanesulphonyl fluoride in ethanol. Then 4M-NaClO₄ was added to a final concentration of0.9 M and the mixture incubated at 21°C for lOmin. Subsequent steps were exactly as described by Hatefi & Stempel (1967). For electrophoresis, portions containing approx. $30\,\mu$ g of protein were dissociated, and run by the method of Neville & Glossmann (1974).

shown by the presence of the subunits of mol.wts. 53000 and 27000 and of NADH-K₃Fe(CN)₆ oxidoreductase activity in the supernatant after centrifugation at 1000OOg for 30min (results not shown). Small amounts of the 75000-mol.wt. polypeptide and the subunits of chymotrypsin itself were also present in the soluble fraction. The unusual kinetics of release of the type II NADH dehydrogenase suggest that the structure of Complex ^I is maintained up to a certain extent of degradation, beyond which progressive instability and relatively rapid fragmentation occur. Presumably it is degradation at one particular site which is involved in release of the type II dehydrogenase and of those polypeptides that are clearly resolved on the gels; the 30000-mol.wt. polypeptide of the iron-protein fraction was degraded with time in a manner that quite closely followed the release of the type II dehydrogenase. Thus there was little change in this polypeptide for up to 120min after chymotrypsin addition, but it was subsequently degraded within the next 240min. These results were confirmed by using the discontinuous-electrophoresis system, which also showed little change in the low-molecularweight polypeptides between 120 and 360min incubation with chymotrypsin. Substantial degradation of the 30000-mol.wt. polypeptide within this time range was also observed, except that there was a decrease in peak height within the first 30min of the incubation owing to degradation of the 33000-mol.wt. polypeptide, which co-migrates with the 30000 mol.wt. polypeptide in this electrophoresis system.

Effect of digestion by chymotrypsin on phospholipiddepleted Complex I

The transformation by trypsin (EC 3.4.21.4) of Complex ^I into ^a type II NADH dehydrogenase was facilitated by depletion of the endogenous phospholipid of Complex I by cholate and $(NH₄)₂SO₄$ treatment (Ragan, 1976a). To investigate further the factors involved in solubilization of the type II dehydrogenase by chymotrypsin, the effects of phospholipid depletion were investigated.

Several differences were noted. First, transformation to type II dehydrogenase occurred more readily, e.g. NADH- $K_3Fe(CN)_6$ oxidoreductase began to decrease after treatment with 20μ g of chymotrypsin/ml for 60min only, conditions that would not have affected undepleted Complex I. Secondly, there were marked differences in the polypeptide profile, especially in the high-molecularweight region. As shown in Fig. $5(a)$, after 100 min with chymotrypsin there had been more extensive degradation than for the same time in Fig. 3. The major differences were that after 300min the 75000 mol.wt. polypeptide had been considerably digested, whereas the 30000-mol.wt. polypeptide had almost completely disappeared. The 30000-mol.wt. poly-

Direction of migration towards anode \rightarrow

Fig. 5. Effect of chymotrypsin digestion on the polypeptides of phospholipid-depleted Complex I

Complex ^I was depleted of phospholipids as described by Ragan & Racker (1973). Chymotrypsin treatment and electrophoresis (a) were exactly as in Fig. 3. In (b), the depleted Complex ^I was treated with 100μ g of chymotrypsin/ml for 6h, and electrophoresed by the method of Neville & Glossmann (1974).

peptide was largely degraded between 100 and 240min after the addition of chymotrypsin, following the kinetics of transformation to a type II dehydrogenase. Fig. $5(b)$ shows the effect of 360 min incubation with $100 \mu g$ of chymotrypsin/ml. Only the 53000- and 27000-mol.wt. polypeptides of the type II dehydrogenase remained in the high-molecularweight region, and a few others together with chymotrypsin subunits, in the low-molecular-weight region.

Discussion

The conversion by chymotrypsin of rotenonesensitive into rotenone-insensitive NADH-ubiquinone-1 oxidoreductase is similar to the change induced by trypsin treatment (Ragan, 1976b), except that the total ubiquinone-1 reductase activity declined rather more in the latter case. The changes in polypeptide composition accompanying this modification of ubiquinone-1 reduction are also very similar to those caused by trypsin, e.g. the degradation of the 53000 mol.wt. polypeptide of the iron protein, and of the 42000-, 39000-, 33000- and 26000-mol.wt. polypeptides, and the lack of digestion of the 75000-, 53000- (type II dehydrogenase) 30000- and 27000 mol.wt. polypeptides. In the low-molecular-weight region, however, only the 20500-mol.wt. polypeptide was rapidly digested by trypsin, whereas several components were degraded by chymotrypsin under similar conditions. The rapid degradation of the 20500-mol.wt. polypeptide by trypsin followed the same time-course as loss of nicotinamideadenine dinucleotide transhydrogenase, and the suggestion was therefore made that this polypeptide was a subunit of this enzyme (Ragan, 1976b). Chymotrypsin, however, induced loss of transhydrogenase activity at only ⁷ % of the rate induced by trypsin (results not shown). The resistance of the 20 500-mol.wt. polypeptide to degradation by chymotrypsin therefore supports the original proposal that this polypeptide is a transhydrogenase subunit.

In using any probe to study three-dimensional organization, failure of the probe to react with a particular polypeptide can be due either to inaccessibility of the polypeptide or to the absence of susceptible residues from that polypeptide. In this context, the failure of trypsin and chymotrypsin to degrade the subunits of the type II dehydrogenase and the 75000- and 30000-mol.wt. subunits of the iron-protein cannot be interpreted unambiguously, particularly as the isolated type II dehydrogenase is also resistant to degradation. However, by using lactoperoxidase-catalysed iodination as a structural probe (Ragan, 1975a) it was found that the type II dehydrogenase subunits were not labelled in intact Complex I, but were labelled when isolated type II dehydrogenase was used. Moreover, the 75000-, 53 000- and 30000-mol.wt. polypeptides of the ironprotein were labelled to only a small extent or not at all in intact Complex I, and heavily labelled in isolation. These result taken together would therefore suggest that in Complex ^I the type II dehydrogenase and iron-protein subunits are buried in the structure with limited access to the 75000- and 53000-mol.wt. polypeptides of the iron-protein. The remaining clearly resolved polypeptides of mol.wts. 42000, 39000 and 33000 were labelled by lactoperoxidase (Ragan, 1976a) and degraded by both trypsin and chymotrypsin, suggesting that they have an exposed configuration in Complex I. The poor resolution of the low-molecular-weight polypeptides and the absence of information as to their possible function makes interpretation of their structural organization unreliable.

The solubilization of type II dehydrogenase by chymotrypsin occurred with a marked lag, during which time most of the proteolytic degradation of the enzyme had already occurred. However, up to the time at which the type II dehydrogenase began to be released, the 30000-mol.wt. polypeptide did not show evidence of degradation, but was almost completely degraded during the release of the type II enzyme. Tentatively, we conclude that the 30000-mol.wt. polypeptide is involved in binding the type II dehydrogenase subunits into the overall structure. In this context, it is noteworthy that papain (EC 3.4.22.2) is capable of degrading the 30000-mol.wt. polypeptide much more readily than chymotrypsin or trypsin (Ragan, 1976c) and that this occurs without transformation to type ¹¹ NADH dehydrogenase, i.e. without loss of NADH- $K_3Fe(CN)_6$ oxidoreductase activity. Obviously, factors other than the 30000-mol.wt. polypeptide are probably involved in the binding of the type II dehydrogenase subunits into the overall structure, but it is also possible that the hydrolysis of the polypeptide by papain does not affect the type II binding site. Preliminary observations suggest that initially this polypeptide is only degraded to a 28000-mol.wt. fragment, and that subsequent cleavage of this on longer treatment with papain does occur with loss of $NADH-K₃Fe(CN)₆ oxidoreductase.$

After incubation with $100 \mu g$ of chymotrypsin per ml for 6h, solubilization of the type II dehydrogenase subunits was nearly quantitative, judged from the staining intensity of the 53000-mol.wt. subunit on gels of the soluble fraction obtained after centrifugation of the digested Complex I. By this time, no further losses of NADH- $K_3Fe(CN)_6$ oxidoreductase activity were observed. Apart from some of the 75000-mol.wt. polypeptide, no other subunits of Complex ^I were solubilized by chymotrypsin and the released enzyme was estimated to be about 75 $\%$ pure from the Coomassie Blue staining.

The effects of phospholipid removal on the susceptibility of Complex ^I to chymotrypsin were broadly similar to those seen with trypsin (Ragan, 1976b). In general, all susceptible polypeptides were rather more rapidly degraded, but in particular, the 75000- and 30000-mol.wt. polypeptides were now degraded by low concentrations of chymotrypsin. These effects were also observed with trypsin (Ragan, 1976b), suggesting increased accessibility of the 75000-mol.wt. polypeptideby phospholipid removal. The more rapid degradation of the 30000-mol.wt. polypeptide by chymotrypsin was probably due to a shortening of the lag period by faster digestion of other polypeptides and was correlated with a more rapid conversion of Complex ^I into type II NADH dehydrogenase.

This work was supported by a Science Research Council grant to C. I. R.

References

- Cremona, T., Kearney, E. B., Villavicencio, M. & Singer, T. P. (1963) Biochem. Z. 338, 407-442
- Hatefi, Y. (1963) Enzymes 2nd Ed. 1, 495-515
- Hatefi, Y. & Rieske, J. (1967) Methods Enzymol. 10, 225-231
- Hatefi, Y. & Stempel, K. E. (1967) Biochem. Biophys. Res. Commun. 26,301-308
- Hatefi, Y., Haavik, A. G. & Griffiths, D. E. (1962) J. Biol. Chem. 237, 1676-1680
- Jovin, T. K., Dante, M. L. & Chramback, A. (1971) Multiphasic Buffer Systems Output, Federal Scientific and Technical Information, U.S. Department of Commerce, PB196085-196091, Springfield, VA
- Lowry, 0. H., Rosebrough, N. J., Farr, A. L. & Randall, R. J. (1951)J. Biol. Chem. 193, 265-275
- Neville, D. M. & Glossmann, H. (1974) Methods Enzymol. 32,92-102
- Ragan, C. I. (1976a) Biochem. J. 154,295-305
- Ragan, C. I. (1976b) Biochem. J. 156, 367-374
- Ragan, C. I. (1976c) Biochim. Biophys. Acta 456, 249-290

 \sim

- Ragan, C. I. & Racker, E. (1973) J. Biol. Chem. 248, 6876-6884
- Rieske, J. S. (1967) Methods Enzymol. 10, 239-245
- Ringler, R. L., Minakami, S. & Singer, T. P. (1963) J. Biol. Chem. 238, 801-810
- Weber, K. & Osborn, M. (1969) J. Biol. Chem. 244, 4406-4412