

Supporting Information

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An Advanced Mechanically Active Osteoarthritis-on-Chip Model to Test Injectable Therapeutic Formulations: The SYN321 Case Study

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SUPPORTING INFORMATION

An advanced mechanically active osteoarthritis-on-chip model to test injectable therapeutic formulations: the SYN321 case study

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Hyaluronic acid (HA) description and molecular structure

Hyaluronic acid (HA) is an anionic, non-sulphated glycosaminoglycan distributed throughout connective, epithelial, and neural tissues in humans and in other vertebrates. It is a polysaccharide built of disaccharide repeating residues of β -D-glucuronic acid and *N*-acetyl- β -D-glucosamine, where the linkage is (1 \rightarrow 3) from the glucuronic acid to the glucosamine, and (1 \rightarrow 4) from the glucosamine to the glucuronic acid. Hyaluronan refers to all physiological forms of hyaluronic acid, the most common being the sodium salt (sodium hyaluronate; NaHA). However, the term hyaluronic acid is commonly used in the literature for referring to any of its forms.



Hyaluronic acid (X = H)Sodium hyaluronate (X = Na)

Figure S1. Structure of hyaluronic acid (HA) and sodium hyaluronate (NaHA)

SYN321 – Synthesis

The synthesis of SYN321 followed the protocol outlined in EP 3 226 905 81 (entitled *'Hyaluronan conjugates with pharmaceutically active substances, methods and compositions'*). In details, the procedure involved the subsequent steps:

- i. Synthesis of [2-(2,6-dichloro-phenylamino)-phenyl]-acetic acid 2-(2-tert-butoxycarbonylamino-ethoxy)-ethyl ester (E):
 Diclofenac (1.2 g), [2-(2-hydroxy-ethoxy)-ethyl]-carbamic acid tert-butyl ester (1,4 g) and 4-dimethylamino- pyridine (DMAP) (76 mg) were dissolved in dichloromethane (DCM) (6 mL). The reaction mixture was cooled on an ice- water bath, 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide (EDC) (1.31 g) was added and was stirred for 4 h while the ice was melting. LCMS of the reaction mixture showed that the expected product was formed and no starting material was left. The reaction mixture was transferred to a separation funnel and about 7 mL of DCM was added. The DCM phase was washed with water (~3*10 mL) and the DCM phase was evaporated. Obtained 2 g crude material. Flash chromatography in EtOAc/Heptane 1/1 gave 1.166 g.
- ii. Synthesis of [2-(2,6-dichloro-phenylamino)-phenyl]-acetic acid 2-(2-amino-ethoxy)-

ethyl ester (F):

[2-(2,6-Dichloro-phenylamino)-phenyl]-acetic acid 2-(2-tert-butoxycarbonylaminoethoxy)-ethyl ester (1166 mg) was dissolved in DCM (9 mL). Trifluoroacetic acid (TFA) (1 mL) was added. The reaction mixture was heated at 40 °C for 15 min. [2-(2,6-Dichloro-phenylamino)-phenyl]-acetic acid 2-(2-amino-ethoxy)-ethyl ester was obtained after evaporation as the ditrifluoroacetate salt, 1.47 g.

iii. <u>Synthesis of hyaluronan succinyl ester (C):</u>

Sodium hyaluronate (1000 mg) was dissolved in formamide (100 mL). Pyridine (Py) (2014 mL), DMAP (30 mg) and succinic anhydride (2494 mg) were added. The reaction mixture was stirred at Room temperature. The reaction mixture was dialyzed in water for 24 hours. The reaction mixture was dialyzed in 1% NaCl for 24 h. The product was precipitated in ethanol (1 L), collected and dried in vacuum overnight. Hyaluronan succinyl ester was obtained 1.043 g.

iv. Synthesis of SYN321:

Hyaluronan succinyl ester (200 mg) was dissolved in water (5 mL). DMF (15 mL) was added to obtain a 30 solution of succinyl hyaluronan in water/DMF, 1/3 (20 mL). *N*-methylmorpholine (33 mL), HOBT (0,5 mg) and a DMF solution of [2-(2,6-dichlorophenylamino)-phenyl]-acetic acid 2-(2-amino-ethoxy)-ethyl ester 397 mg/mL (46,4 mL, 18,4 mg) were added to the stirred succinyl-hyaluronan solution. EDC (5.8 mg) was added. The reaction mixture was mixed thoroughly and left over night at room temperature. Sodium chloride (200 mg) was added as a sat solution (359 mg/mL) to the DMF-water solution. The product was precipitated in ethanol (100 ml) and stirred for 2.5 h. The precipitate was 35 collected and dissolved in 1% NaCl (20 mL) and precipitated in ethanol (80 mL). The solid material was collected, and once more dissolved in 1% NaCl and precipitated. The precipitate was dissolved in water and lyophilized. Obtained 177 mg. According to proton ¹H-NMR spectroscopy the degree of substitution was 0.22.



SYN321

Figure S2: Reaction scheme for the synthesis of SYN321 (a = 1, *b* = 5)

Evaporation within the uBeat® MultiCompress platform

To test the evaporation rate within the uBeat® MultiCompress platform, three devices were injected with fibrin gel only (i.e., final concentration 2 U/ml Thrombin, 10mg/ml fibrinogen) and cultured at 37°C 5%CO2 over weekend with PBS in channels 1-3-5. Then, using Bovine Serum Albumin (BSA) from Pierce BCA Protein Assay Kit, ThermoFisher, a solution of 0.5 mg/ml of BSA in PBS was prepared and 200 µl were placed in the channels 1-3-5 and the corresponding reservoirs of each chamber of the devices. After three hours, BCA solution was collected from each chamber of one device and stored at +4°C. The other two devices were kept in culture for a total of three days, one in static and one in dynamic conditions (2h of stimulation, 4h of rest, 2h of stimulation and 16h of rest) and then the BSA solution was collected from them. A BSA calibration scale was prepared (2mg/ml, 1mg/ml, 0.5mg/ml, 0.25mg/ml, 0.125mg/ml, 0.0625mg/ml, 0.03125mg/ml, 0mg/ml) and 25µl of scale's duplicates were transferred in a 96 multiwell plate together with 25µl of sample's duplicates. A solution of working reagent (WR) was prepared by mixing 50 parts of BCA reagent A with 1 part of BCA reagent B, then 200µl of WR were transferred in each well and the plate was incubated for 30 min at 37°C. Then, Absorbance at 550 nm was measured with Tecan spectrophotometer. Results were analysed in Excel. The blank value was subtracted from all the wells, a calibration curve was created by linear regression and samples' concentrations were obtained. Values were plotted using GraphPad Prism. Statistical analysis was performed using nonparametric Kruskal-Wallis test. As illustrated in the figure below, the BSA concentration in PBS increased after three days of static cultures (T2) within the uBeat® MultiCompress platform, compared to samples maintained for only three hours (T0). This suggests that some evaporation occurs within the device. Additionally, a higher BSA concentration was observed at T2 in dynamic samples compared to static samples, indicating that mechanical actuation further promotes evaporation.



Figure S3: BSA concentration in PBS over time within the uBeat \mathbb{B} MultiCompress platform comparing T0 (i.e. 3 hours in static conditions), T2 static (i.e. 3 days in static conditions) and T2 dynamic (i.e. 3 days in dynamic conditions). N=3.

SYN321 hydrolysis in synovial fluid and human plasma

Instrumentation		Waters Acquity UPLC + Thermo Q-Exactive Focus Orbitran MS						
		Watara A aquity	Waters Δ caulty HSS T3 (2.1 × 50 mm 1.8 µm) column with					
Column		waters Acquiry HSS 15 (2.1 \times 50 mm, 1.8 µm) column with						
Invitation mode			pre-column mer					
Sheeth as		ESI-		t a)				
A suriliarra and		nitrogen 30 (A	roitrary uni	ts)				
Auxiliary gas		nitrogen 10 (A		ls)				
Sweep gas		nitrogen 3 (Art	offrary units	5)				
Capillary voltag	je	3200 V						
Capillary tempe	rature	320 °C						
Auxiliary gas te	mperature	500 °C						
Mass range		$m/z \ 100 - 600$						
RF lens		80						
Resolution		70 000 (FWHN	А @ m/z 20	00) for full scan				
Normalized coll	lision energy	Off for full sca	n					
Calibration		External						
Software		Thermo Xcalibur 4.1.31.9						
			First 0.5 min of the run was directed into waste by using a					
Other information	on	divert valve to decrease the ion source contamination by						
		early eluting matrix constituents.						
Gradient Elution	h; A = 0.1% acet	ic acid, $B = 85:15$ acetonitrile / isopropanol (v/v)						
Time (min)	Flow (ml/min)	A%	В%	curve				
0.0	0.600	100	0	-				
0.5	0.600	100	0	6				
2.0	0.600	80	20	6				
3.0	0.600	10	90	6				
4.0	0.600	2	98	6				
5.0	0.600	100	0	1				
Column oven te	mperature 60 °C							
Injection volume 4 µl								
Ion chromatogi	rams were extra	acted from the	Q-Exactive	e Focus Orbitrap MS total ion				
chromatograms	using calculated	l monoisotopic	accurate m	asses for deprotonated molecular				
ions with 10 pp	m window.	_		-				
Compound	m/z	Retention t	ime (min)	Polarity				
Diclofenac	294.0094	3.05	. ,	ESI-				
Warfarin (IS)	307.0976	2.96		ESI-				

Table S1. LC-MS method for analysis and quantitation of diclofenac (1).

Table S2. LC-MS method for analysis and quantitation of diclofenac lactam (2) an	d linker
(3).	

Instrumentation	Waters Acquity UPLC + Thermo O-Exploris Orbitrap MS							
Column		Waters Atlan	Waters Atlantis premier BEH C18 AX (2.1 × 100 mm,					
		1.7 μm) colu	mn with p	re-column filter				
Ionization mode	;	ESI+						
Sheath gas		nitrogen 50 (Arbitrary	units)				
Auxiliary gas		nitrogen 10 (Arbitrary	units)				
Sweep gas		nitrogen 3 (A	Arbitrary u	nits)				
Capillary voltag	e	3000 V						
Capillary tempe	rature	320 °C						
Auxiliary gas te	mperature	450 °C						
Mass range		m/z 100 - 60)0					
RF lens		80						
Resolution		120 000 (FW	/HM @ m/	z 200) for full scan				
Calibration		External	_					
Software		Thermo Xca	libur 4.1.3	1.9				
		First 0.5 min of the run was directed into waste by using						
Other information	on	a divert valv	a divert valve to decrease the ion source contamination					
		by early eluting matrix constituents.						
Gradient Elution	n; A = 2 mM ammo	nium formate, $B = 85:15$ acetonitrile / isopropanol (v/v)						
Time (min)	Flow (ml/min)	A%	В%	curve				
0.0	0.500	98	2	-				
0.5	0.500	98	2	6				
3.0	0.500	10	90	6				
5.0	0.500	10	90	6				
5.5	0.500	2	98	6				
8.0	0.500	98	2	1				
Column oven te	mperature 60 °C							
Injection volum	e 4 µl							
Ion chromatogra	ams were extracted	from the O-Ex	ploris Orb	itrap MS total ion chromatograms				
using calculated	monoisotopic accu	rate masses fo	or deprotor	nated molecular ions with 10 ppm				
window.								
Compound	m/z	Retention ti	me (min)	Polarity				
Lactam	278.0134	3.00		ESI+				
Linker	206.1023	1.54		ESI+				
Warfarin (IS)	309.1121	3.25		ESI+				

Matrix	Human plasma							
Compound	Diclofenac (1)	Lactam (2)	Linker (3)					
Detection limit (nM)	10	10	10					
Quantitation limit (nM)	10	10	10					
Range (nM)	10 - 50 000	10 - 1 000	10 - 1 000					
R ²	>0.999	0.996	>0.999					
Concentration (nM)	Accuracy % (n=2)	Accuracy % (n=2)	Accuracy % (n=2)					
10	121.8	100.0	100.0					
100	86.4	100.0	100.0					
1000	90.5	100.0	100.0					
10000	101.3	-	-					
50000	100.0	-	-					
Snedecor precision (%)*	17.4	14.7	9.5					
Calibration curve fitting	Ouadratic fitting, weighting 1/X							
Internal standard	Warfarin							

Table S3. Method performance of compounds quantitation in human plasma.

Table S4. Method performance of compounds quantitation in human synovial fluid.

Matrix	Human synovial fluid							
Compound	Diclofena	c (1)	Lactam (2	2)	Linker (3))		
Detection limit (ng/ml)	10		10		100			
Quantitation limit (ng/ml)	10		10		100			
Range (ng/ml)	10 - 50 00	00	10 - 50 00)0	100 - 50 (000		
R ²	> 0.999		0.993		0.986	0.986		
Cona (ng/ml)	Acc. %	Prec. %	Acc. %	Prec. %	Acc. %	Prec. %		
	(n=3)	(n=3)	(n=3)	(n=3)	(n=3)	(n=3)		
10	105.0	14.5	117.7	15.3	-	-		
100	97.3	8.6	86.0	18.8	75.6	14.9		
1 000	97.4	0.9	95.5	19.4	127.8	20.2		
10 000	100.3	2.0	100.8	16.6	96.6	16.4		
50 000	100.0	1.8	100.1	9.1	96.4	15.4		
Calibration curve fitting	Quadratic, weighting 1/X							
Internal standard	Warfarin							

<u>Effect of SYN321 in a rat MIA model of osteoarthritis: tables reporting</u> <u>results of body weight, Weight Bearing tests and distance</u>

	Group 1		Group 2		Group 3		Group 4		Group 5	
Days	Mean	SEM	Mean	SEM	Mean	SEM	Mean	SEM	Mean	SEM
Baseline	241.00	3.13	239.40	3.13	243.00	4.11	235.10	2.17	237.20	2.48
11	300.40	4.20	292.10	3.80	305.10	5.05	295.90	2.75	294.40	4.50
15	318.78	5.97	295.20	8.84	320.00	6.21	311.89	3.19	313.30	5.65
19	341.11	6.62	324.30	4.76	338.78	8.28	334.11	3.47	336.00	6.18
23	369.00	8.02	346.00	5.98	369.22	8.39	357.44	4.65	357.80	7.42
28	382.33	8.92	357.80	6.22	371.00	8.87	369.44	5.11	367.30	8.79
32	395.11	9.42	369.70	6.79	384.44	9.25	382.00	5.19	379.10	9.45
36	403.56	11.86	375.44	7.79	390.78	9.34	394.24	4.84	384.40	10.71
40	414.22	12.30	384.22	8.12	399.56	9.15	402.78	4.90	392.20	11.99
44	432.33	12.62	400.63	10.24	415.33	9.48	414.89	5.65	406.40	12.29

Table S5: Mean group body weight (g).

 Table S6: Mean group body weight (% from baseline).

	Group 1		Group 2		Group 3		Group 4		Group 5	
Days	Mean	SEM								
Baseline	100.00	0.00	100.00	0.00	100.00	0.00	100.00	0.00	100.00	0.00
11	124.65	0.77	122.13	1.78	125.58	0.75	125.88	0.74	124.10	1.23
15	132.77	1.04	123.51	3.95	131.67	1.05	133.18	0.97	132.01	1.22
19	142.06	1.24	135.50	1.44	140.10	1.54	142.67	1.11	141.57	1.38
23	153.65	1.67	144.53	1.76	152.80	2.34	152.60	1.29	150.72	1.78
28	159.17	1.97	149.47	1.91	153.52	2.43	157.74	1.67	154.74	2.65
32	164.50	2.21	154.43	2.05	159.07	2.44	163.08	1.45	159.71	2.98
36	167.90	2.97	156.63	2.16	161.72	2.66	168.34	1.75	161.92	3.53
40	172.34	3.17	160.30	2.42	165.34	2.40	172.00	1.82	165.19	4.09
44	179.88	3.18	167.25	3.10	171.94	3.08	177.22	2.63	171.16	4.10

Table S7: Mean group Weight Bearing (difference between legs) (L%-R%).

	Group 1		Group 2		Group 3		Group 4		Group 5	
Days	Mean	SEM	Mean	SEM	Mean	SEM	Mean	SEM	Mean	SEM
Baseline	-0.29	0.58	-0.45	0.49	-0.44	0.63	-0.60	0.49	-0.15	0.57
10	17.83	0.85	17.00	1.30	19.24	1.12	13.87	0.71	17.24	0.80
12	22.39	2.48	18.12	2.28	11.60 ^a	2.34	13.56	4.31	15.96	2.75
14	25.59	1.49	18.56	1.43	19.62	2.09	18.68	2.47	18.14	2.73
17	27.23	1.49	23.38	2.55	20.77	1.54	22.02	3.15	20.15	1.51
24	24.63	2.62	24.06	3.01	21.54	2.21	20.97	2.37	18.43	2.12
31	21.24	1.46	25.29	3.05	22.54	2.84	19.61	3.16	18.71	1.68
38	20.37	2.34	23.14	2.03	19.12	2.91	18.33	2.15	21.85	2.22
45	18.72	1.80	19.89	3.15	19.51	1.82	17.06	0.78	21.85	2.00

^ap<0.05 vs. Vehicle (Group 1) using one way ANOVA followed by Dunnett's test.

	Group 1		Group 2		Group 3		Group 4		Group 5	
Days	Mean	SEM								
10	29.97	1.33	31.30	1.71	28.75	1.30	20.65	3.85	26.54	2.67
24	35.96	1.93	25.95	2.85	25.97	4.15	30.76	3.12	27.83	4.25
38	24.19	2.16	13.07	2.66	17.20	2.49	19.95	2.02	12.81	3.47