# The Disulphide Bonds of Erabutoxin a, a Neurotoxic Protein of a Sea-Snake (Laticauda semifasciata) Venom

BY Y. ENDO, S. SATO, S. ISHII\* AND N. TAMIYA

Department of Chemistry, Tohoku University, Sendai, Japan, and \*Department of Pharmacy, Hokkaido University, Sapporo, Japan

(Received 9 November 1970)

Erabutoxin a was partially hydrolysed with enzymes and sulphuric acid and the resulting peptides were separated from each other by column chromatography and paper electrophoresis. From the results of amino acid analyses of the sulphurcontaining peptides and their oxidized components, all four disulphide bridges in the toxin molecule were located. The disulphide bonds were found between halfcystine residues at positions 3 and 24, 17 and 41, 43 and 54, and 55 and 60 from the N-terminus.

Erabutoxin a is a crystalline neurotoxic protein obtained from a sea-snake (Laticauda semifasciata) There are some similar neurotoxins, venom. namely erabutoxin b and laticotoxin a of sea snakes (Tamiya & Arai, 1966; Sato, Yoshida, Abe & Tamiya, 1969), cobrotoxin (Yang, 1965) and toxins α (Boquet, Izard, Jouannet & Meaume, 1966; Karlsson, Eaker & Porath, 1966; Botes & Strydom, 1969) of cobras. The amino acid sequences of toxin  $\alpha$ of Naja nigricollis (Eaker & Porath, 1967), toxin a of Naja haje haje (Botes & Strydom, 1969), cobrotoxin of Naja naja atra (Yang, Yang & Huang, 1969) and erabutoxins a and b of Laticauda semifasciata (Sato & Tamiya, 1971) have been elucidated. All five toxins consist of 61 or 62 amino acid residues with eight half-cystine residues among them. All the half-cystine residues are found at corresponding positions of these toxin molecules.

The present study was undertaken to locate the disulphide linkages in the erabutoxin a molecule.

#### MATERIALS AND METHODS

Erabutoxin a. Erabutoxin a was isolated by the method of Tamiya & Arai (1966), desalted by gel filtration (Sato et al. 1969) and freeze-dried.

Proctase A. An acid protease produced by Aspergillus niger var. macrosporus, named Proctase A, was a kind gift from Dr Yamasaki (Yamasaki, Shiraki, Horiuchi, Yamada & Goi, 1967).

Paper electrophoresis. Paper electrophoresis was carried out on Toyo no. 50 filter paper (Toyo Scientific Instruments, Tokyo, Japan), cooled in chilled hexane at  $0-5^{\circ}$ C, with solvent mixtures pyridine-acetic acid-water (1:10:89, by vol., for pH3.6 and 10:7:233, by vol., for pH5.0) at 40 V/cm. The paper was dried in the air after the electrophoresis and the peptides were located on the paper, if necessary, by spraying with ninhydrin (0.24%, w/v, in acetone), platinochloric acid (Toennies & Kolb, 1951),  $\alpha$ -nitrosonaphthol (Acher & Crocker, 1952), Ehrlich reagent (Smith, 1953) or Sakaguchi reagent (Jepson & Smith, 1953). Successive applications of the above sprays were also used (Smith, 1958).

Amino acid analysis. About  $0.1 \mu mol$  of a peptide was hydrolysed with 6 M-HCl (0.3 ml) in an evacuated and sealed glass tube at 110°C for 24h. The hydrolysate was dried in an evacuated desiccator and analysed for the amino acid contents with a JLC 5AH automatic amino acid analyser (Japan Electron Optics Laboratory Co., Tokyo, Japan).

## EXPERIMENTAL AND RESULTS

Digestion of erabutoxin a with Proctase A and the separation of resulting peptides. Erabutoxin a (42mg) was dissolved in 4.5ml of 1M-formic acid-1M acetic acid (2:1, v/v) mixture and Proctase A (1.2mg) was added. The mixture was incubated at 37°C for 12h and freeze-dried. The number of amino groups, as measured by trinitrobenzenesulphonic acid method (Satake, Okuyama, Ohashi & Shinoda, 1960) was 2.37 times the original value. About 7 new amino groups appeared/molecule of the toxin.

The freeze-dried material was dissolved in  $0.1 \,\mathrm{M}$ -formic acid (2.0ml) and passed with the formic acid through a Sephadex G-25 column ( $2 \,\mathrm{cm} \times 270 \,\mathrm{cm}$ ) that had been equilibrated with the formic acid. The elution pattern as followed by extinction at 230 and 280 nm is shown in Fig. 1.

The u.v.-absorbing fractions were collected and freeze-dried. Each one-hundredth part of the dried material was subjected to paper electrophoresis at pH 5.0 for 100 min. The results are shown in Fig. 2. The sulphur-containing (platinochloric acidpositive) peptides (shadowed in Fig. 2) were found in peaks II, III and V of Fig. 1.



Fig. 1. Sephadex column chromatography of Proctase A digests of erabutoxin a. The Proctase A digests of erabutoxin a were applied on a Sephadex G-25 column  $(2 \text{ cm} \times 270 \text{ cm})$  and eluted with 0.1 M-formic acid. Fractions (each 5 ml) were assayed for the extinction at 230 nm ( $\odot$ ) and 280 nm ( $\bigcirc$ ) in 1 cm cells.



Fig. 2. Paper electrophoresis of Proctase A peptides of erabutoxin a. The Sephadex G-25 fractions of Proctase A digests were subjected to paper electrophoresis at pH5.0for 100 min. The shadowed spots are platinochloric acidpositive.

The remaining main parts of the dried material were subjected to paper electrophoresis under the same conditions, and peptides were oxidized with performic acid vapour on the paper (Brown & Hartley, 1966), extracted with 2.5% (v/v) acetic acid (each 3ml) from the spots and freeze-dried. A part of the sulphur-containing and oxidized peptides were again subjected to paper electrophoresis at pH 3.6 and separated into their components. The results of amino acid analyses of the oxidized peptides and their components are shown in Table 1. The results of amino acid analyses of sulphur-free peptides are also shown. Peptide III-2 was separated into three components on the second electrophoresis after the oxidation, whereas peptides III-1 and V-1 were each separated into two components. Peptide III-3 again gave a single component after the oxidation. The amino acids of peptide III-2 can be explained as the sum of those of peptides III-1 and V-1.



Fig. 3. Paper electrophoresis of sulphuric acid hydrolysate of a Proctase A peptide from erabutoxin a. Peptide III-3, a Proctase A peptide from erabutoxin a, was partially hydrolysed with  $H_2SO_4$  and the hydrolysate subjected to paper electrophoresis at pH5.0 for 4h. The shadowed spots are platinochloric acid-positive.

Sulphuric acid hydrolysis of a larger fragment. Peptide III-3  $(0.8 \mu \text{mol})$ , which contained two disulphide bonds, was partially hydrolysed with 10<sub>M</sub>-sulphuric acid-acetic acid-water-thioglycollic acid mixture (15:9:6:0.01, by vol.) (0.5ml) in an evacuated and sealed glass tube, for 45min at 100°C (Ryle, Sanger, Smith & Kitai, 1955) and the hydrolysate was applied to a column  $(1.4 \text{ cm} \times 5 \text{ cm})$ of weakly basic resin (Amberlite CG-4B, OH<sup>-</sup> form) to remove the sulphuric acid. The peptides were eluted from the column with 5% (v/v) acetic acid (25ml). The residue obtained by evaporating the eluate at below 32°C in a flash evaporator was dissolved in 2.5% (v/v) acetic acid (0.4 ml) and separated into its components by paper electrophoresis at pH 5.0 for 4h (Fig. 3). Six spots, out of seven detected by ninhydrin spray, were chloroplatinic acid-positive. Spots 1-5 of the main run of the electrophoresis were oxidized on the paper with performic acid vapour and extracted with 2.5% (v/v) acetic acid (each 3 ml). The acetic acid extracts were evaporated to dryness under reduced pressure at  $32^{\circ}$ C, dissolved in 2.5% (v/v) acetic acid (each 0.2ml) and subjected again to paper electrophoresis at pH3.6. Free cysteic acid was detected as one of the oxidation products from the peptides of

	Peptide VI-2	1.00 (1)*			0.91(1)*		1.05 (1)*	1.03 (1)*	4
	Peptide VI-1		(1) 16.0	1.09 (1)*			*(1)06.0		1 4
Peptide V-1	Peptide V-2		0.94 (1)	1.00 (1)*		1.53 (2)			4
	Peptide IV-1	1.06 (1)*	1.62 (2)	1.03 (1)*				0.91 (1)*	5
	Com- ponent 2	$1.21(1)^{\circ}$ 0.71(1)^{\circ}	1.54 (2)	0.96 (1)*	1.11 (1)*		1 (1)		7
	Com- ponent 1	1.01 (1)*				1.01 (1)*	0.89 (1)*	1.09 (1)*	4
	Total	1.82 (2)* 0.90 (1)*	1.90 (2)	1.25 (1)*	1.37 (1)	1.04 (1)*	$\begin{array}{ccc} 1 & (1) \\ 0.91 & (1)^{*} \end{array}$	1.08 (1)*	11
	Peptide III-3	3.94 (4)* 2.07 (2)*	0.95(1) 1.75(2)	2.10 (2)*	2.06(2) 1.14(1)*	1.83(2) 0.96(1)	1.89 (2)*	Ì	20
Peptide III-2	Com-	2.02 (2)* 0.87 (1)*	1.23(1) 3.13(3)	1.12 (1)*	1.12 (1) 2.12 (1)		1 (1) 0.98 (1)*		12
	Com- ponent 2	1.16 (1)*				1.16 (1)*	0.76 (1)*	<b>*(1) 16</b> .0	4
	Com- ponent 1	1.04 (1)*			2.22 (2)*			0.74 (1)*	4
	Total	4.15 (4)* 0.96 (1)*	1.23(1) 2.81(3)	1.44 (1)*	1.17(1) 3.92(3)	0.93 (1)*	$\begin{array}{ccc} 1 & (1) \\ 0.97 & (1)^{*} \\ 1.02 & (1)^{*} \end{array}$	1.56 (2)*	20
Peptide III-1	Com-	1.34 (1)*	$1.06(1)^{\circ}$ $0.89(1)^{\circ}$		0.96 (1)*		0 74 (1)*		5
	Com- ponent 1	1.27 (1)*			1.91 (2)*			0.82 (1)*	4
	Total	1.82 (2)*	0.86 (1)* 0.94 (1)*		0.52 (1) 2.37 (2)*		(1) 00 0	1.08 (1)	6

Abbreviation: Cya, cysteic acid.

DISULPHIDE BONDS OF ERABUTOXIN a

	Amino acid com mol of j	mposition (mol/ peptide)		
	Peptide 5a	Peptide 5b		
Суа	1.20 (1)*	0.82 (1)*		
Asp	0.80 (1)*			
Glu	1.02 (1)*	1.18 (1)*		
Val	1.00 (1)*			

\* The total numbers of these amino acids were taken as standard round numbers.

spots 1-4. The spot 5 peptide was separated into two components, 5a and 5b, after the oxidation. The results of amino acid analysis of the components are shown in Table 2.

### DISCUSSION

The amino acid compositions of the peptides presented in Tables 1 and 2 can be explained on the basis of the amino acid sequence of erabutoxin a (Sato & Tamiya, 1971), assuming disulphide linkages between half-cystine residues at positions 3 and 24, 17 and 41, 43 and 54 and 55 and 60 from the N-terminus. The structures of the peptides are shown in Fig. 4.

The whole structure of erabutoxin a is shown in Fig. 5. The only difference in the amino acid sequences of erabutoxins a and b is that the latter has histidine at position 26 instead of asparagine in the former. It is therefore reasonable to conclude that erabutoxin b has the same arrangement of disulphide bridges as erabutoxin a.

In addition to the above peptides, the following two peptides were isolated from the Proctases A and B digests of the 'tryptic core' of erabutoxin a:

16	17	18	19
Thr -	Cys-	Pro-	Ser
40	1 <sup>°</sup>	42	
Gly-	Ċys-	Gly	
	2	з	4
Il	e-Cy	s-Ph	ie
2	3 I Š	2	25
Se	r-Ċv	s-Tv	r

The results confirm the above conclusion.

Essentially the same overall arrangement of disulphide bridges was presented at the Second International Symposium on Animal and Plant Toxins held in Tel-Aviv, Israel in February 1970, for cobrotoxin by Dr C. C. Yang and for toxin  $\alpha$  by

Table 1. Amino acid compositions of Proctase A peptides and their oxidized components from erabutoxin a

Abbreviation: Cya, cysteic acid



Fig. 4. Structures of the peptides obtained from erabutoxin a.



Fig. 5. Structure of erabutoxins a and b. The amino acid at position 26 is asparagine for erabutoxin a and histidine for erabutoxin b.

Dr D. P. Botes. This overall structure therefore seems to be common to neurotoxins of cobras and sea snakes.

The research fund was provided by the Ministry of Education, the Japanese Government.

#### REFERENCES

- Acher, R. & Crocker, C. (1952). Biochim. biophys. Acta, 9, 704.
- Boquet, P., Izard, Y., Jouannet, M. & Meaume, J. (1966). C. r. hebd. Séanc. Acad. Sci., Paris, 262, 1134.
- Botes, D. P. & Strydom, D. J. (1969). J. biol. Chem. 244, 4147.
- Brown, J. R. & Hartley, B. S. (1966). Biochem. J. 101, 214.
- Eaker, D. & Porath, J. (1967). Abstr. 7th int. Cong. Biochem., Tokyo, p. 1087.
- Jepson, J. B. & Smith, I. (1953). Nature, Lond., 172, 1100.

- Karlsson, E., Eaker, D. & Porath, J. (1966). Biochim. biophys. Acta, 127, 502.
- Ryle, A. P., Sanger, F., Smith, L. F. & Kitai, R. (1955). Biochem. J. 60, 541.
- Satake, K., Okuyama, T., Ohashi, M. & Shinoda, T. (1960). J. Biochem., Tokyo, 47, 654.
- Sato, S. & Tamiya, N. (1971). Biochem. J. 122, 453.
- Sato, S., Yoshida, H., Abe, H. & Tamiya, N. (1969). Biochem. J. 115, 85.
- Smith, I. (1953). Nature, Lond., 171, 43.
- Smith, I. (1958). Chromatographic Techniques, p. 60. Ed. by Smith, I. London: William Heinemann (Medical Books) Ltd.
- Tamiya, N. & Arai, H. (1966). Biochem. J. 99, 624.
- Toennies, G. & Kolb, J. J. (1951). Analyt. Chem. 23, 823.
- Yamasaki, M., Shiraki, M., Horiuchi, S., Yamada, Y. & Goi, H. (1967). Abstr. 7th int. Cong. Biochem., Tokyo, p. 762.
- Yang, C. C. (1965). J. biol. Chem. 240, 1616.
- Yang, C. C., Yang, H. J. & Huang, J. S. (1969). Biochim. biophys. Acta, 188, 65.