The Properties of a Nuclear Acidic Protein Fraction that Binds $[6,7-3H]$ Oestradiol-17 β

By R. J. B. KING, J. GORDON AND A. W. STEGGLES

Department of Hormone Biochemistry, Imperial Cancer Research Fund, Lincoln's Inn Fields, London W.C.2

(Received 11 March 1969)

1. Additional evidence was obtained that the nuclear oestradiol-17 β receptor is an acidic protein. Partial purification of the receptor protein was obtained by chromatography on hydroxyapatite and it contains protein-bound phosphate. 2. The nuclear '5s' and cytoplasmic '9 5s' and '5s' receptors from uterus, dimethylbenzanthracene-induced mammary adenocarcinoma and kidney are precipitated together with bound oestradiol-17 β by protamine sulphate. This common property suggests that the nuclear and cytoplasmic receptors are related to each other. 3. The properties of two acidic protein fractions from both liver and dimethylbenzanthracene-induced mammary adenocarcinoma are described. Fraction ¹ contains two major components and fraction 2 contains one component, as judged from polyacrylamide-gel electrophoresis. Fraction ² contains RNA and both fractions contain protein-bound phosphate. 4. These fractions form insoluble complexes with calf thymus histone, protamine sulphate and poly-L-lysine. The formation of these complexes is markedly affected by ionic strength and pH. Ionization of both the ϵ -amino group of lysine and carboxyl group are involved. RNA and DNA do not appear to be involved. The interaction is not affected by EDTA or 1mm-Na^+ , $-K^+$, Ca^{2+} , Mg^{2+} or Mn^{2+} . Per unit weight, whole histone has 4-5 times as many binding sites for the acidic proteins as the latter have for the former. 5. No convincing evidence was obtained for DNA-acidic protein interaction, but, as juidged from precipitation experiments, there was competition between DNA and acidic protein for histone. 6. Relatively large amounts of acidic protein partly relieved the histone inhibition of the template activity of DNA for Escherichia coli RNA polymerase (EC 2.7.7.6).

Previous studies from this laboratory have shown that, in oestrogen-sensitive tissues, [6,7-3H]- α oestradiol-17 β is non-covalently bound to an acidic protein component of both euchromatin and heterochromatin and that this component can be extracted with sodium chloride (King, Gordon, Cowan & Inman, 1966; King & Gordon, 1968). The present paper describes the further characterization of this material.

Certain nuclear acidic proteins form insoluble complexes with histones and alter the template activity of DNA-histone for RNA polymerase (Langan, 1967; Wang & Johns, 1968; Wang, 1968). These properties of the protein fractions described below were also investigated.

MATERIALS AND METHODS

Tissue fractions. Mammary tumours were induced by intragastric administration of 50mg. of dimethylbenzanthracene in corn oil (Huggins, Briziarelli & Sutton, 1959;

Young, Cowan & Sutherland, 1963). 3H-labelled mammary tumours for the fractionation studies were obtained from intact female rats ¹ hr. after the suboutaneous injection of [6,7-3H]oestradiol-17 β (0.1 μ g./100g. body wt.). Homogenates (10%, w/v) were prepared in 0.25M-sucrose containing 3mM-CaCI2, and nuclei were isolated either by centrifugation at 700g for 10min. (method A) or by densitygradient centrifugation (Allfrey, Littau & Mirsky, 1964) (method B). In both cases the nuclear fraction was washed twice with lOmx-tris-HCI buffer, pH 7-4, containing 3mM-CaCl2 (tris-Ca2+). This removed negligible amounts of radioactivity.

When non-radioactive tissues were used, nuclei were prepared from 40-80g. of tissue by method A, washed twice with tris-Ca²⁺ and resuspended in 2.2 M-sucrose containing 3mm-CaCl2. This was centrifuged at 28000rev./ min for ¹ hr. in a Spinco no. 30 rotor and the nuclear pellet was washed twice with tris-Ca2+ (method C).

For the experiments with protamine sulphate, kidneys and mammary tumours were obtained 1hr. after the subcutaneous injection of 0.2μ g. of [6,7-3H]oestradiol-17 β /100g. body wt. The rats had been ovariectomized 8 days before use and all the tumours decreased in size during that

period. Homogenates (approx. 30% , w/v) in 10mm-tris- 1.5 mM-EDTA-2mM- β -mercaptoethanol, pH7.0, were prepared by using a Silverson homogenizer (Silverson Machines Ltd., London S.E.1) at maximum speed for 1min. The homogenate was divided into two equal parts. The first was centrifuged at 10^5 g for 1 hr. to give a clear supernatant fraction. The second part was diluted with 0.36 M-sucrose to a final concentration of 0-25M with respect to sucrose. This was centrifuged at 500g for 10min. and the crude nuclear pellet washed once with 10ml. of tris-Ca2+. The pellet was extracted with 0 5M-KCl-10mM-tris-1mm-EDTA, pH 7-0, for 15min. (0-5ml. for the kidney, 1-Oml. for the tumour) and centrifuged at 3000g for 30min. The supernatant was called the nuclear extract. Samples (0-2ml.) of both the nuclear extract and supernatant fractions were applied directly to linear $5-20\%$ (w/v) sucrose gradients. For the supernatant fractions the gradient was made up in 10mM-tris-HCl buffer, pH7-4, but the nuclear gradients also contained 0-3M-KCI. The sedimentation coefficients quoted in this paper are those reported by Jensen et al. (1968). Another sample (0-4ml.) of each fraction was added to 0-2ml. of protamine sulphate $(7.5 \text{mg./ml. water})$ and kept at 0° for 10min. The bulky precipitate was sedimented at 103g for 10min. and 0-2ml. portions of the supernatant were applied to the sucrose gradients. Portions were also taken for measurement of radioactivity. The pellet was dissolved in $1 M-NaOH$ and its radioactivity measured.

Uteri from eight to 12 immature Sprague-Dawley rats were incubated with 5ml. of $1nM - [6,7-3H]$ oestradiol-17 β in Krebs-Ringer phosphate medium II (Dawson & Elliott, 1959) at 4° for a total time of $2\,\text{hr}$. The medium was replaced twice during this period. The uteri were washed twice with 5ml. of the homogenizing medium and a homogenate (approx. $30\%, w/v$) was prepared as detailed above that is essentially the same as that described by other workers (Toft & Gorski, 1966; Jensen et al. 1968; Rochefort & Baulieu, 1968).

Columns (3cm. x 4.5cm.) of hydroxyapatite (Bio-Gel HTP; BioRad Laboratories, Richmond, Calif., U.S.A.) were prepared in 1 mM-potassium phosphate buffer, pH7, as described by Velle & Engel (1964). For routine experiments the column was eluted with 140ml. of ¹ mM-potassium phosphate buffer, pH7, followed by 0.1M-potassium phosphate buffer, pH7. Fractions (20ml.) were collected until no further material absorbing at 280nm. was eluted (usually about 100ml.). The concentration of potassium phosphate buffer, pH7, was then changed to 0-2M and the process repeated, and the column was eluted with 0-4Mpotassium phosphate buffer, pH7, in the same way. The column was washed with 1M-potassium phosphate buffer, pH7, exhaustively washed with ImM-potassium phos. phate buffer, $pH7$, and stored in this solution at 4° until re-used.

The material eluted with 0.1M-phosphate (fraction 1) was 50% saturated with $(NH_4)_2SO_4$ at pH 7.0, the precipitate dissolved in 10mx-tris-HCl buffer, pH8-5, and the solution dialysed overnight against the same buffer. The material eluted with 0.2 M-phosphate (fraction 2) was treated in the same way. In a typical preparation from 80g. of liver, fractions ¹ and 2 contained 4mg. and 5mg. of protein respectively. From 70g. of tumour, fractions ¹ and 2 contained 3mg. and 4mg. of protein respectively.

[3H]Tryoptphan-labelled fractions ¹ and 2 were obtained from livers taken 30min. after the intraperitoneal injection of $100\,\mu\text{C}$ of L-[3H]tryptophan (specific radioactivity $2\,\text{C}/$ m-mole; The Radiochemical Centre, Amersham, Bucks). Only 10g. of liver was used for each of these experiments. After elution from the hydroxyapatite fractions ¹ and 2 were exhaustively dialysed against water at 4°, freeze-dried and redissolved in 10mM-tris-HCl buffer, pH8-5.

Histone-acidic protein interaction. Unless stated otherwise, this was carried out in 10mM-tris-HCl buffer pH8-5, in a final volume of 0.25 ml. at 0° for 10min. The acidic protein was always added last. The complex was sedimented at 10OOg for 10min. and washed twice with 0-5ml. of water, and protein in the pellet was determined.

Protein was determined by the method of Lowry, Rosebrough, Farr & Randall (1951), with crystallized bovine serum albumin (Armour Pharmaceuticals Ltd., Eastbourne, Sussex) as standard. This standard was used throughout these experiments, so that quantitative comparisons cannot be made between the amounts of protein precipitated with protamine and poly-L-lysine. DNA was measured by the diphenylamine reaction (Burton, 1956) and phosphate by a micro modification of the method described by King & Wooton (1956). Protein-bound phosphate was released by treatment with 1M-NaOH for 10min. in a boiling-water bath (Langan, 1967). Control determinations were carried out both on protein samples that had not been hydrolysed and on solutions that contained no protein but had been hydrolysed. The orcinol reaction for RNA was carried out by the method of Greenbaum & Slater (1957).

Gel electrophoresis was carried out (1) towards the anode at pH8-5 in gels containing 7.5% (w/v) polyacrylamide (Cyanogum 41; British Drug Houses Ltd., Poole, Dorset) and (2) towards the cathode at pH4 in 15% (w/v) polyacrylamide gels as described by McAllister, Wan & Irvin (1963) for histones. They were stained with Coomassie Blue for protein (Crambach, Reisfeld, Wyckoff & Zaccari, 1967) or Acridine Orange for RNA (Richards, Coll & Gratzer, 1965).

RNA polymerase (EC 2.7.7.6) (ribonuclease-free) was prepared from E8cherichia coli strain M.R.E. 620 up to stage 3 of the method described by Chamberlin & Berg (1962). The template activity of the DNA (salmon sperm) was measured by the same method, with a 20min. incubation period and 10μ g. of DNA/tube.

'Melting' profiles of DNA were measured in ^a Gilford model 2002 spectrophotometer fitted with an ethylene glycol-heated cell housing. The measurements were made in 10mM-tris-HCI buffer, pH8-5.

The effect of DNA on the sedimentation of [3H]tryptophan-labelled acidic proteins was carried out by equilibrating 50μ g. of calf thymus DNA with either fraction 1 or fraction 2 in 0-2ml. of 10mM-tris-HCl buffer, pH7-4, for 10min. at 0° and then layering it on top of a linear 5-20% (w/v) sucrose gradient prepared in 10mM-tris-HCl buffer, pH7-4. This was spun at 39000rev./min. for 5hr. in a Spinco SW39 rotor.

Materials. Protamine sulphate (salmine sulphate) was purchased from British Drug Houses Ltd. Calf thymus histones, either unfractionated or the Fl and F3 fractions, were kindly given by Dr A. J. MacGillivray and had been prepared by method 2 of Johns (1964). The Fl fraction had a lysine/arginine molar ratio of $21.6:3.6\%$ and the F3 fraction had a lysine/arginine molar ratio of 11.9:9.9%. Phosvitin was obtained from Mann Research Laboratories, New York, N.Y., U.S.A. Yeast 'soluble' RNA was obtained from C. F. Boehringer und Soehne G.m.b.H., Mannheim, Germany. Poly-L-lysine (mol.wt. 2500), bovine pancreatic ribonuclease (type IA), pancreatic non-crystalline deoxyribonuclease ¹ and calfthymus DNA (type 1) were purchased from Sigma Chemical Co., St Louis, Mo., U.S.A. Salmonsperm DNA was obtained from Calbiochem Ltd., London W.1.

The [6,7-3H]oestradiol-17 β (specific radioactivity 235μ c/ μ g.) was prepared and purified as described by King & Gordon (1966).

Measurement of radioactivity. This was carried out in a Packard Tri-Carb model 3003 liquid-scintillation spectrometer. The phosphor had the following composition: 100g. of naphthalene, 3g. of 2,5-diphenyloxazole, 230ml. of ethanol, 385ml. of xylene and 385ml. of dioxan. With the exception of the sucrose gradients (expressed as c.p.m.) the radioactivity of all samples was corrected for quenching by a channels-ratio method (Baille, 1960) and values are expressed as d.p.m.

RESULTS

Fractionation of the bound $[6,7.3H]$ oestradiol-17 β from mammary-tumour nuclei. Optimum extraction of bound $[6,7.3H]$ oestradiol-17 β from nuclei (isolated by method A) occurred with 2ml. of 1 M-sodium chloride/g. original wt. of tissue. When nuclei were prepared by method B, ¹ ml. of ¹ M-sodium chloride/ g. original wt. of tissue was used. This extracted 80% of the nuclear [6,7-3H]oestradiol-17 β . Dilution of this extract to 0.15 M-sodium chloride by addition of ¹ mM-potassium phosphate buffer, pH 7, precipitated over 90% of the DNA together with 11% of the protein without precipitation of radioactivity. In later experiments it was found that dilution to

0 4M-sodium chloride gave similar results, and this was then used as a routine.

Fractionalprecipitationwithammoniumsulphate did not give further purification and also partly dissociated the oestradiol-protein complex. It was not used further with $[6,7.3H]$ oestradiol-17 β labelled fractions. The extract in 0.15M-sodium chloride solution was applied directly to the hydroxyapatite column and eluted successively with 1mm -, 50mm -, 0.1m -, 0.2m - and 0.4m -potassium phosphate buffer, pH7 (Fig. 1). The radioactivity eluted in the ¹ mM-phosphate fraction was due to free [6,7-3H]oestradiol-17 β . Radioactivity, together with material absorbing at 280nm., was eluted with each of the 0.1M-, 0.2M- and 0.4Mphosphate fractions. For the large-scale isolation of fractions ¹ and 2 from both tumour and liver, the diluted sodium chloride extract was 80% saturated with ammonium sulphate, the precipitate dissolved in ¹ mm-potassium phosphate buffer, pH 7, and applied to the column.

Composition of fractions 1 and 2. Only the 0.4 m phosphate fraction contained small amounts of DNA, and this fraction was not investigated further. Similar results were obtained when the nuclei were isolated by method B except that all the fractions contained less protein and RNA.

Analysis showed that fractions ¹ and 2 from liver and tumour both contained about 10μ g. of proteinbound P/mg. of protein. In six separate estimations on liver this varied from 7 to 44μ g. of P/mg. of protein. Fraction 2 gave a positive orcinol test for RNA but fraction ¹ did not.

In view of the possible presence of contaminating basic proteins in the 0 4M-phosphate extract, the

Fig. 1. Fractionation of NaCl extract of nuclei from mammary tumour labelled in vivo with $[6,7.3H]$ oestradiol-17 β on hydroxyapatite. The column $(3cm. \times 4.5cm.)$ was eluted with a discontinuous gradient of potassium phosphate buffer, pH7, the concentration of which is indicated by the arrows. $---, E_{280}$; $---, Radioactivity.$

affect of adding 1g. of the cation-exchange resin Amberlite CG-50 (Na+ form) to the 0.4 m-phos phate extract was tested. This removed about 30% of the protein present at that stage but influenced neither the final yield of fractions ¹ and 2 nor their phosphate content or ability to complex with histone. This stage was therefore omitted from routine preparations. Polyacrylamide-gel electrophoresis patterns of the liver preparations are shown in Fig. 2. At pH8-5 fraction ¹ contained two main components and several fast-moving minor ones, whereas fraction 2 contained one main component with a mobility similar to that of component 2 of fraction 1. Addition of 8M-urea to the gels did not alter the properties of the bands. Neither fraction contained detectable amounts of protein that migrated at pH4. As judged from Acridine Orange staining, no RNA-containing fraction migrated at pH8-5. Similar results were obtained with mammary tumour. B. J. B. KING, J. GORDON AND A. W. STEGGLES

test of adding 1g. of the station-scaling sears parties. With the same model test of a continue of the station of th It is 1. In S. IN IN . IN THE COLENARY AND A. W. STEGGOLES all the sine of the strip of the formula in the strip of the formula in the strip of the formula B, J. B. K. K. NGRO A. G. ORDENOVAND A. W. STEOGLES and the set of the set of the form of the form of the set of the set of the form of the set of the form of the set of the form of the set of the set of the form of the s CING, J. GO

on-exchange is the 0.4 M-F

on-exchange is the 0.4 M-F

is removed a that stage

of fractions 1

ability to committed

fore omitted

and e-gel elections are sharalized worst

proposed in the pelsidic . Neithe ING, J.

n-exchan

the 0-4

removeed that states of fraction

that states of fraction

bility to e omit and speak of fraction

mide-gel

nained the protection might be described in the speak of

speak of protection

ind p

A88ociation with polycation& Both fractions formed insoluble complexes with calf thymus histone, protamine sulphate and poly-L-lysine (Table 1), and this was influenced by ionic strength (Fig. 3) and pH (Fig. 4). The decrease at $pH11$ indicated that the lysine ϵ -amino group might be involved, and the effect at acid pH could be due to carboxyl-group ionization. pH6 and 7-5 also affected the interaction. The interaction was unaffected by 1mm-EDTA, 1mmpotassium chloride or lmM-sodium chloride, all at

Fig. 2. Polyacrylamide-gel electrophoresis of acidic protein fractions 1 and 2 from liver nuclei. A $100\,\mu$ g. portion of each fraction was applied to the gels (7-5% polyacrylamide, pH8-5), which were run for 45min. 7mi/gel. 'a' indicates the protein band common to both fractions 1 and 2.

pH 8-5. With the same molarity of calcium chloride, magnesium chloride or manganous chloride no consistent results were obtained: they either had no effect or gave a small inhibition. As fractions ¹ and ² contained both RNA and protein-bound phosphate it was possible that phosphate ionization might explain this effect at neutral pH, and this was investigated by using the model compounds phosvitin and 'soluble' RNA. Phosvitin did form complexes with poly-L-lysine at pH5-5, but not at 7-5, and these were not affected by EDTA. At pH ⁸ ¹ mM-magnesium chloride enhanced RNApoly-L-lysine interaction, which indicated that neither of these two types of interaction could KING, J. GORĐON AND A. W. STEGGLES

1969

ion-scelange resin pHS 6. With the same molecular container of the b4 skeps

io the b4 skephole compares and this same contained they either had

is temperature and the difference NG, J. GORDON AND A. W. STEGGLES

cardsage resin prits 5. With the same moladies of the other hand the
same points of the same model and the same model and the same model and
the other hand that stages but no effect or ga

Table 1. Precipitation of acidic protein by polycation8

Results are expressed as μ g. of protein precipitated and represent the means \pm s.E.M. of at least four determinations. A 50μ g. portion of each component was present in a final volume of 0.25ml. of 10mm-tris-HCl buffer, pH8-5.
Acidie protein Histone Proteining Poly-t-lysine Acidic protein Histone Protamine Poly-L-lysine

 \bullet , Effect on complex-formation. Equal amounts (50 μ g.) of common to both fractions 1 and 2.

common to NaCl (x)

common fraction of NaCl (x)

Fig. 3. Effect of NaCl on histone-acidic protein interaction.

Fig. 3. Effect on complex formation. Equal anounuts $(50\mu g)$, of
 \bullet fraction 2 and calf thymus histone were mixed in a final at................... :: $\sum_{i=1}^{n}$ $\sum_{i=1}^{n}$ volume of O-5ml. of NaCl. The amount of protein precipitated relative to that in lOmM-tris-HCl buffer, pH8-5, was measured. \blacksquare , Solubility of the complex. The histonefraction 2 complex was formed in lOmm-tris-HCl buffer, pH8-5, and sedimented at 8OOg for 10min. Then 0-5ml. of NaCl was added and the amount of protein solubilized relative to that in the control $(0.5$ ml. of water) was measured.

explain the EDTA effect on poly-L-lysine-acidic protein interaction.

Incubation of fraction 2 with either ribonuclease or deoxyribonuclease for 15min. at room temperature before addition of the histone did not diminish the amount of complex formed or its E_{280}/E_{260} ratio of 1-3.

The effect of titrating 50μ g. of the acidic proteins with histone is shown in Figs. $5(a)$ and $5(b)$. The arginine-rich F3 histone was more efficient at formation of complexes with the acidic protein than was the lysine-rich Fl fraction. At optimum histone concentrations more protein was precipitated with

Fig. 4. Effect of pH on protein-polyeation interaction. Equal amounts (50μ g.) of the two reactants were mixed in a final volume of 05ml. of buffer and the precipitate was sedimented at 800g for 10min. The following buffers were used: pH 3*5-6 5, lOmx-sodium citrate-HCI; pH 7-0- ⁹ 0, lOmx-tris-HCl; pH 9 5-11-5, lOmM-glycine-NaOH. \bullet , Fraction 2+ calf thymus histone; \wedge , fraction 2+poly-Llysine; o, phosvitin+poly-L-lysine.

fraction ² than with fraction 1. Maximum precipitation occurred with a whole histone/acidic protein weight ratio of about 1 :1.

When 50μ g. of histone was titrated with increasing amounts of acidic protein (Fig. 6), there was a linear increase in the amount of protein precipitated up to a ratio of 1μ g. of histone to $4-5 \mu$ g. of acidic protein, indicating that, per unit weight, the histone contained 4-5 as many binding sites for the acidic protein as the latter had for the former.

The complex was soluble in 10mm - and 0.2m hydrochloric acid and 1M-sodium hydroxide and in high concentrations of salt (Fig. 3). When poly-Llysine was used, the pH-solubility profile was characterized by a marked decrease at pH7 that was eliminated if ¹ mM-EDTA was present.

A88ociation with DNA. Table ² shows the interaction with DNA either alone or in the presence of histone. This experiment was set up such that there was an excess of DNA relative to histone. Some competition existed between the DNA and acidic protein for histone, as the amount of protein precipitated with the DNA was less than in its absence. Similarly the amount of DNA precipitated in the presence of histone and acidic protein was less than with histone alone. The amount of protein precipitated/ μ g. of DNA increased in the presence of the acidic protein, indicating the formation of a DNA-histone-acidic protein complex. This experiment was repeated a further three times with similar results, except that the actual amount of protein precipitated varied from batch to batch of the acidic protein.

Owing to the crudeness of the estimation methods, this did not provide conclusive evidence as to whether the acidic fraction did or did not form complexes with DNA. This was checked by using

Fig. 5. Effect of increasing amounts of calf thymus histone on its interaction with acidic protein fraction 1(a) or fraction 2(b). Calf thymus histone was mixed with $50\,\mu$ g. of acidic protein in a final volume of 0.5ml. of 10mmtris-HCl buffer, pH8-5. \blacksquare , Whole histone; \bullet , histone F3; \circ , histone Fl.

Table 2. Interaction of DNA, histone and acidic protein

fractions that had been labelled in vivo with [3H] tryptophan. The presence of DNA did not influence the sedimentation pattern of either fraction 1 or fraction 2 in a 5-20% sucrose gradient, which supports the conclusion that there was no DNAacidic protein interaction under these conditions. Neither fraction showed any definite sedimentation peaks, most of the radioactivity being in the upper third of the gradient. Under these conditions the DNA migrated half way down the gradient.

Further evidence for the lack of DNA-acidic protein interaction was the observation that fraction 2 did not alter the T_m of DNA (75 μ g. of protein + 75μ g. of DNA).

Influence of fraction 2 on the template activity of DNA for RNA polymerase. These experiments were set up to give maximal histone inhibition $(20 \,\mu\text{g. of})$ histone + 10μ g. of DNA). All the reactants except the DNA, [³H]UTP (obtained from The Radiochemical Centre) and RNA polymerase were mixed and kept at 0° for 10min. to allow histone-acidic protein interaction to occur. The other reactants were then added and the tubes incubated for 20min. at 37°. In separate experiments it was shown that this incubation medium (minus DNA) allowed complete interaction between the two types of protein, whereas the medium of higher ionic strength described by Fuchs, Millette, Zillig & Walter (1967) did not. Histone/acidic protein ratios of 1:1 and 1: 5 were tested. Low concentrations of fraction ² had very little effect on the template activity of DNA in either the presence or the absence of histone (Table 3). The higher concentration of fraction 2 occasionally had a small inhibitory effect on its own but gave some release of the histone inhibitiqp. This experiment was repeated a further three times with each tissue with essentially similar results, except that the effect of the 100μ g. of acidic protein on the DNA--histone varied from 28% to 76% for liver and from 28% to ⁴⁴% for the tumour. Different batches of acidic fraction were prepared for each experiment.

Fig. 6. Effect of increasing amounts of acidic protein on their interaction with calf thymus histone. Details are as given in Fig. 5. \circ . Fraction 1; \bullet , fraction 2.

Table 3. Effect of acidic and basic proteins on the template activity of DNA for RNA polymerase

All tubes contained 0.1μ mole each of ATP, GTP, CTP and [3H]UTP, 1.25μ moles of MnSO₄, 5μ moles of MgCl₂, 10.5 μ moles of tris buffer, 5 μ moles of β -mercaptoethanol and $10\,\mu$ g. of DNA in a final volume of 0.25ml. In the absence of the additions shown in the table this incorporated 340pmoles of UMP/20min.

Precipitation of bound $[6,7.3H]$ oestradiol-17 β with protamine sulphate. As the nuclear oestradiol-binding protein fraction from mammary tumour formed an insoluble complex with protamine sulphate it was decided to test whether the oestradiol-binding proteins from other tissues reacted in the same way.

Table 4 shows that protamine sulphate precipitated an appreciable amount of the radioactivity from nuclear extracts of uterus, kidney and mammary tumour. The precipitation step was carried

Table 4. Precipitation of radioactivity from tissue extracts by protamine sulphate

A 0-4ml. portion of each fraction was mixed with 0-2ml. of protamine sulphate $(7.5 \text{mg.}/\text{ml.})$ at 0° for 10min. Results represent the means \pm s.g.m. of at least four determinations.

out in a final concentration of 0.32 M-potassium chloride, which appreciably inhibits histone-acidic protein interaction (Fig. 3). It is likely therefore that the amounts of protein and radioactivity precipitated in these experiments represent submaximal values.

Radioactivity was also precipitated from the supernatant fractions, but as these were carried out at lower ionic strength than the nuclear extracts these values should represent maximal precipitation. The absence of precipitation of radioactivity from solutions of bovine serum albumin and rat plasma containing comparable amounts of protein and $[6,7.3H]$ oestradiol-17 β indicates that this is not a non-specific interaction. To check this in a more definitive manner the effect of protamine sulphate precipitation on the sedimentation behaviour of the radioactive material on sucrose gradients was investigated (Fig. 7). As far as the supernatant fractions were concerned, protamine sulphate precipitated all of the 9-5s component from both tumour and uterus and a large part of the radioactivity in the 3-6s region. This was virtually complete with uterus but less so with tumour and kidney, which agrees with results given in Table 4. This difference may be due to albumin binding. Serum albumin does bind [6,7-3H] oestradiol-17 β , it sedimented in this region of the gradient and was not affected by protamine

Fig. 7. Sedimentation profiles of tissue extracts labelled with $[6,7.3H]$ oestradiol-17 β . The gradients $[5-20\%$ (w/v) sucrose in 1OmM-tris-HCl buffer, pH7.4] were centrifuged for 16-18hr. at 39000rev./min. in the Spinco SW39 rotor. The gradients for the nuclear extracts in addition contained 0-3M-KCI. (a) Uterus labelled in vitro with $1 \text{nm-}\{6,7\cdot{}^{3}H\}$ oestradiol-17 β . \blacktriangle , $10^{5}g$ supernatant; \blacksquare , $10^{5}g$ supernatant after protamine sulphate treatment; O, 0-5M-KCl extract of nuclei; \Box , 0-5M-KCl extract of nuclei after protamine sulphate treatment. (b) Mammary tumour labelled in vivo with [6,7-3H]oestradiol-17 β . \blacktriangle , 10⁵g supernatant; \blacksquare , 10⁵g supernatant after protamine sulphate treatment; \circ , 0.5m-KCl extract of nuclei; \Box , 0.5m-KCl extract of nuclei after protamine sulphate treatment.

sulphate. As judged from the haemoglobin colour, the tumour and kidney supernatant fractions contained some blood whereas the uterus contained very little.

Under the conditions used for extraction of the nuclei, only a 5s component was obtained and the amount ofit was markedly diminished by protamine sulphate treatment.

DISCUSSION

The evidence presented here provides further proof that the nuclear oestradiol-receptor protein in uterus, kidney and dimethylbenzanthraceneinduced rat mammary adenocarcinoma is acidic (King, 1967). The ability of both it and the cytoplasmic receptor to form complexes with protamine sulphate indicates some similarity in the properties of the nuclear and cytoplasmic receptor(s). It is noteworthy that the androgen-receptor protein in the cytoplasm of rat prostate gland is also precipitated by protamine sulphate (W. I. P. Mainwaring, personal communication).

The presence of protein-bound phosphate in the nuclear oestradiol receptor fraction is of interest in view of the possible role of these proteins in metabolic control processes (Kleinsmith, Allfrey & Mirsky, 1966; Langan, 1967), but it remains to be seen whether the phosphate is actually present in the binding protein or merely a contaminant of this, as yet, impure fraction. The enzyme phosphoprotein phosphatase does not alter the binding characteristics of either the nuclear or the cytoplasmic receptors, which indicates that phosphate is not involved in oestradiol-17 β binding (R. J. B. King, unpublished work).

The acidic protein fractions described here represent a very small proportion of the nuclear acidic proteins present in the nucleus. The total nuclear acidic protein in liver is about 2mg./g. wet wt. (Steele & Busch, 1963) and in mammary tumours about 8mg./g. wet wt. (Smith, King, Meggitt & Allen, 1966). Allowing for a 50% yield of nuclei in our experiments, the amount of protein recovered in fractions ¹ and 2 combined would represent about 10% of the total nuclear acidic protein. Even so, the two fractions are still heterogeneous. Fraction ¹ contained a number of minor protein bands on polyacrylamide-gel electrophoresis in addition to the two main components. Although fraction 2 ran as a single component on polyacrylamide gel, it too is heterogeneous, as judged from its sedimentation behaviour in a sucrose gradient. The ready solubility of both fractions in dilute tris and the absence of detectable protein migrating at pH⁴ indicated that very little, if any, basic protein was present.

The interrelationship of fractions ¹ and 2 is not

clear. In mammary tumour they both contained bound oestradiol-17 β and, as judged from their electrophoretic mobility, fraction ¹ contains one component that is the same as fraction 2. However, this cannot be explained by poor fractionation on the hydroxyapatite because, if fraction 1 is rechromatographed on a fresh column, it is still eluted with 01M-phosphate. Both fractions react with polycations, although fraction ¹ does this less efficiently than fraction 2. This may be due to the greater heterogeneity of this fraction rather than to an intrinsic property of the acidic protein present in it.

The ability to form insoluble complexes with polycations is similar to that reported by other workers (Langan, 1967; Wang & Johns, 1968), but differs in the greater efficiency of precipitation in our experiments. This may be due to the study of different proteins or to more homogeneous preparation. The formation of this complex is markedly affected by ionic strength and pH. The interaction is completely inhibited by 0.6 M-sodium chloride, and 85% inhibition occurs with 0.3 M-sodium chloride. Once the complex has been formed a higher salt concentration is required to dissociate it. The solubility of the complex in salt solutions would explain the previous observation that a high salt concentration was required to solubilize the nuclear oestradiol-17 β -receptor protein even after removal of DNA with deoxyribonuclease (King & Gordon, 1967). The effect of ionic strength on the association suggests that ionic bonds are primarily involved, and this is supported by the pH effect. Carboxyl-group ionization could account for the dissociation in acid. Changes occurring at about $pH6$ and 7.5 are also involved. The differences in the pH curves for the interaction of acidic protein with whole histone and with poly-L-lysine suggest that some of the interaction at intermediate pH values is due to non-lysine residues of the histone. As both fractions contain RNA and protein-bound phosphate it is possible that these may also play a part in the interaction. Bonner (1967) has described an acidic protein present in chromatin that contains RNA, and the protein described here has some of the properties of this 'histone-binding protein'. However, the lack of effect of ribonuclease and the enhancement of RNA-histone interaction by Mg^{2+} suggests that RNA is not involved in the present experiments. As no RNA was detected in the protein band on gel electrophoresis it is possible that the RNAin these fractions is an impurity rather than being associated with the protein. The involvement of protein-bound phosphate is less clear. The phosvitin-poly-L-lysine interaction has some of the characteristics of the poly-L-lysine-acidic protein interaction, but differences do exist. The decrease in binding of phosvitin at neutral pH cannot be overcome by EDTA whereas that of the acidic protein can. The ability of the acidic proteins to form complexes with at least three polycations indicates that no great specificity is involved as far as the cation is concerned, but some specificity must exist for the acidic protein because acidic proteins such as serum albumin do not form insoluble complexes with the polycations used in these experiments.

The stoicheiometry of the histone-acidic protein interaction is also of interest as, on a weight-forweight basis, the whole histone has about 4 times as many binding sites for the acidic protein as the latter has for the former.

The evidence presented here suggests that these particular acidic proteins do not associate with DNA, as do some of the bacteriophage (Ptashne & Hopkins, 1968) and liver nuclear acidic proteins (Wang, 1967), but that triplex DNA-histone-acidic protein molecules can be formed with some competition between the DNAand acidic protein for the histone binding sites. The precipitation experiments indicate that DNA-histone interaction takes preference over histone-acidic protein interaction. This is supported by the experiments on the template activity of the DNA. Even though histoneacidic protein interaction was allowed to take place before addition of the DNA template, ^a fivefold weight excess of acidic protein to histone was required to produce any marked relief of the histone blocking of the DNA template.

As far as can be judged, the acidic protein fractions from dimethylbenzanthracene-induced mammary tumour possessed similar properties to those from liver. As judged by the precipitation of o estradiol-17 β -receptor protein from uterus and kidney by protamine sulphate the same is true of these tissues, but the relevance of the properties reported in this paper to the mechanism of oestradiol action remains to be investigated. It is compatible with an effect of the hormone on the chromatin (King, 1967; Warren & Barker, 1967; Hamilton, 1968), and it is tempting to speculate that oestradiol-17 β may alter the structure of the chromatin by affecting the interaction of DNA, histone and the acidic proteins.

The authors are grateful to Profesor E. E. Baulieu for his advice on the labelling of uteri in vitro with [6,7-3H] oestradiol-17 β .

REFERENCES

- Allfrey, V. G., Littau, V. C. & Mirsky, G. E. (1964). J. Cell Biol. 21, 213.
- Baille, L. A. (1960). Int. J. appl. Radiat. 8, 1.
- Bonner, J. (1967). In Biochim. biophy8. Acta Library vol.

20: Symposium on Regulatory Mechanisms in Nucleic Acid and Protein Synthesis, p. 211. Ed. by Koningsberger, V. V. & Bosch, L. Amsterdam: Elsevier Publishing Co. Burton, K. (1956). Biochem. J. 62, 315.

- Chamberlin, M. & Berg, P. (1962). Proc. nat. Acad. Sci., Wash., 48, 81.
- Crambach, A., Reisfeld, R. A., Wyckoff, M. & Zaccari, J. (1967). Analyt. Biochem. 20,150.
- Dawson, R. M. C. & Elliott, W. H. (1959). In Data for Biochemical Research, p. 208. Ed. by Dawson, R. M. C., Elliott, D. C., Elliott, W. H. & Jones, K. M. Oxford: Clarendon Press.
- Fuchs, E., Millette, R. L., Zillig, W. & Walter, G. (1967). Europ. J. Biochem. 3, 183.
- Greenbaum, A. L. & Slater, T. F. (1957). Biochem. J. 66,155. Hamilton, T. H. (1968). Science, 161, 649.
- Huggins, C., Briziarelli, G. & Sutton, H. (1959). J. exp. Med. 109, 25.
- Jensen, E. V., Suzuki, T., Kawashima, T., Stumpf, W. E., Jungblut, P. W. & DeSombre, E. R. (1968). Proc. nat. Acad. Sci., Wash., 59, 632.
- Johns, E. W. (1964). Biochem. J. 92, 55.
- King. E. J. & Wooton, I. D. P. (1956). Micro-analysis in Medical Chemistry, pp. 75-81. 3rd ed., London: J. and A. Churchill Ltd.
- King, R. J. B. (1967). Arch. Anat. microsc. Morph. exp. 56, 570.
- King, R. J. B. & Gordon, J. (1966). J. Endocrin. 34,431.
- King, R. J. B. & Gordon, J. (1967). J. Endocrin. 89, 533.
- King, R. J. B. & Gordon, J. (1968). J. Endocrin. 40, 195.
- King, R. J. B., Gordon, J., Cowan, D. M. & Inman, D. R. (1966). J. Endocrin. 86, 139.
- Kleinsmith, L. G., Allfrey, V. E. & Mirsky, A. E. (1966). Proc. nat. Acad. Sci., Wash., 55,1182.
- Langan, T. A. (1967). In Biochim. biophys. Acta Library vol. 20: Symposium on Regulatory Mechanisms in Nucleic Acid and Protein Synthesis, p. 232. Ed. by Koningsberger, V. V. & Bosch, L. Amsterdam: Elsevier Publishing Co.
- Lowry, 0. H., Rosebrough, N. J., Farr, A. L. & Randall, R. J. (1951). J. biol. Chem. 198, 265.
- McAllister, H. C., Wan, Y. C. & Irvin, J. J. (1963). Analyt. Biochem. 5,321.
- Ptashne, M. & Hopkins, N. (1968). Proc. nat. Acad. Sci., Wash., 60, 1282.
- Richards, E. G., Coll, J. A. & Gratzer, W. B. (1965). Analyt. Biochem. 12, 452.
- Rochefort, H. & Baulieu, E. E. (1968). C. B. Acad. Sci., Paris, 267, 662.
- Smith, J. A., King, R. J. B., Meggitt, B. F. & Allen, L. N. (1966). Brit. J. Cancer, 20, 335.
- Steele, W. J. & Busch, H. (1963). Cancer Res. 28, 1153.
- Toft, D. & Gorski, J. (1966). Proc. nat. Acad. Sci., Wash., 55, 1574.
- Velle, W. & Engel, L. L. (1964). Endocrinology, 74, 429.
- Wang, T. Y. (1967). Arch. Biochem. Biophys. 122, 629.
- Wang, T. Y. (1968). Exp. Cell Res. 58, 288.
- Wang, T. Y. & Johns, E. W. (1968). Arch. Biochem. Biophys. 124, 176.
- Warren, J. C. & Barker, K. L. (1967). Biochim. biophys. Acta, 188, 421.
- Young, S., Cowan, D. M. & Sutherland, L. E. (1963). J. Path. Bact. 85, 331.