Incorporation of Oxygen-18 into the 25-Position of Cholecalciferol by Hepatic Cholecalciferol 25-Hydroxylase

By THELMA C. MADHOK, HEINRICH K. SCHNOES and HECTOR F. DELUCA

Department of Biochemistry, College of Agricultural and Life Sciences, University of Wisconsin-Madison, Madison, WI 53706, U.S.A.

(Received 13 March 1978)

The oxygen enzymically inserted as a hydroxy function by rat liver post-mitochondrial fraction into the 25-position of cholecalciferol to give 25-hydroxycholecalciferol is derived exclusively from molecular O_2 . Therefore like the other two cholecalciferol hydroxylases, i.e. 25-hydroxycholecalciferol 1α -hydroxylase and 25-hydroxycholecalciferol 24-hydroxylase, the cholecalciferol 25-hydroxylase is also a mono-oxygenase ('mixed-function oxidase').

It is well recognized that cholecalciferol must be transformed into more polar biologically active metabolites before it elicits its biological function (DeLuca, 1974; Kodicek, 1974). Thus it is first converted into $25(OH)D_3$ and subsequently to $1\alpha,25(OH)_{2}D_{3}$ before stimulating the intestine and bone (DeLuca, 1974; Kodicek, 1974).

The 25-hydroxylation occurs in the liver (Ponchon et al., 1969; Olson et al., 1976; Horsting & DeLuca, 1969) and to a smaller extent in the intestine (Tucker et al., 1973; Bhattacharyya & DeLuca, 1974). Studies in vitro have shown the cholecalciferol 25-hydroxylase of rat liver to be microsomal, requiring NADPH and ^a cytoplasmic protein (Bhattacharyya & DeLuca, 1974) for activity, but little else is known concerning the nature of this reaction. On the other hand it has been clearly demonstrated that 25-hydroxycholecalciferol 1α -hydroxylase is a 3-component monooxygenase similar to the steroidogenesis system of the adrenals (Ghazarian et al., 1974a; Pedersen et al., 1976). We have therefore examined the question of whether the cholecalciferol 25-hydroxylase is a mixed function mono-oxygenase.

In the present paper it will be demonstrated that the oxygen enzymically introduced as a hydroxy function into the 25-position of cholecalciferol to yield $25(OH)D₃$ is derived exclusively from molecular $O₂$ providing strong evidence that the 25-hydroxylase is a mono-oxygenase rather than a dehydrogenasehydratase system.

Methods

Animals

Abbreviations used: 25(OH)D₃, 25-hydroxycholecalciferol; 1,25(OH)₂D₃, 1,25 - dihydroxycholecalciferol; $24,25(OH)₂D₃$, 24,25-dihydroxycholecalciferol.

hanging wire cages. They were given water and fed ad libitum a low-calcium $(0.02\% \text{ calcium}, \text{w/w})$, cholecalciferol-deficient diet for 3 weeks.

Preparation of the enzyme

Rats were killed by decapitation and exsanguination. The livers were immediately removed, carefully separated from the adhering connective tissues and weighed. The livers were then rinsed with ice-cold 0.25M-sucrose, minced, transferred to an ice-cold Potter-Elvehjem homogenizer fitted with a Teflon pestle and homogenized in ¹ vol. of the 0.25 M-sucrose. The homogenate was centrifuged at $128g$ at 4° C for ¹⁰ min in ^a Lourdes centrifuge with ^a ⁹ RA rotor. The resulting supernatant was further centrifuged at 10000g for 30 min to remove mitochondrial particles. The resulting post-mitochondrial supernatant was used for incubation.

Incubation conditions and extraction of samples

Some 25 incubations, each with a total volume of lOml, were carried out in 21 ml full-capacity serum bottles. The assay mixture consisted of 5 ml of postmitochondrial supernatant (246 mg of protein), 2.5 ml of phosphate-cofactor solution $(0.1 M-K₂HPO₄)$, 0.4mM-NADP, 160mM-nicotinamide, 20mM-ATP, 22.4mM-glucose 6-phosphate adjusted to pH7.4) and 2.5 ml of salt solution (5mM-MgCl₂ and 0.1M-KCl plus 0.25 unit of glucose 6-phosphate dehydrogenase/ ml; one unit as defined by Sigma Chemical Co., St. Louis, MO, U.S.A.) (Bhattacharyya & DeLuca, 1974). The bottles were sealed with serum-bottle caps and connected through needle outlets to a vacuum train. The bottles were then evacuated repeatedly, followed by flushings with purified N_2 passed over heated copper. Finally, 4ml of 99.3 atom %1802 (Bio-Rad Laboratories, Richmond, CA,

Fig. 1. Chromatographic profile during the isolation and purification of $25(^{18}OH)[3\alpha-3H]D_3$ A Lipidex 5000 column $(2 \text{cm} \times 60 \text{cm})$ was used that

was packed and eluted with a hexane/chloroform $(9:1, v/v)$ solvent system. ----, Radioactivity of sample; ----, radioactivity in each fraction volume. The peak between the arrows is due to $25(OH)D₃$.

U.S.A.) and 7ml of purified N_2 were introduced simultaneously into each of the 25 incubation bottles using gas-tight syringes. The hydroxylation reactions were initiated by introduction (with a Hamilton syringe) of 10μ g of $[3\alpha - 3H]$ cholecalciferol (synthesized chemically by S. Yamada in this laboratory) in 10μ l of 95% ethanol with a specific radioactivity of 58000 c.p.m./ μ g. After 4h of incubation at 37°C at 120 oscillations/min, the reactions were terminated by the immediate transfer of the mixtures into 500ml separating funnels (six incubation mixtures per funnel) each containing 60ml of alcohol and 240ml of hexane. The reaction mixture was vigorously shaken for 7 min and left at 4°C overnight. The lower phase was emptied into a clean beaker and the upper phase was filtered through Shark-skin filter paper into a 500ml round-bottomed flask. The lower phase was put back into the funnel and re-extracted twice with 240 ml portions of hexane. The combined hexane extracts contained 100% of the added radioactivity.

Chromatography

The combined hexane extracts were evaporated to dryness by using a rotary evaporator at 30°C. The residue was dissolved in $500 \mu l$ of hexane/chloroform $(9:1, v/v)$ and applied to a glass column $(2 \text{ cm} \times 60 \text{ cm})$ containing lOOg of Lipidex 5000 (hydroxyalkoxypropyl derivative of Sephadex; Packard Instrument Co., Downers Grove, IL, U.S.A.), packed and eluted in the same solvent system. A total of ¹⁴⁰ fractions (6ml at a flow rate of 1 ml/min) was collected and 10-200 μ l of each fraction was used for counting (Fig. 1) in a Packard model 3255 liquid-scintillation spectrometer (Packard) with the scintillation mixture described previously (Ghazarian et al., 1974b). The efficiency of counting for ${}^{3}H$ content was 45%. The relative elution position of the metabolites has been determined previously by chromatography of the pure metabolites under identical experimental conditions (Holick & DeLuca, 1971). The peak region $25(OH)D₃$ (fractions 97–119) was evaporated to dryness by using a rotary evaporator at 30°C to yield a total radioactivity of 53000 c.p.m. equivalent to 914ng of 25(OH) D_3 (based on 58000c.p.m./ μ g, the specific radioactivity of the substrate cholecalciferol). The residue was dissolved in $20 \mu l$ of water/ methanol $(1:4, v/v)$ and applied to a octadecylsilanebonded microparticulate silica column (4.6mm x 25 cm) (Zorbax-ODS column; DuPont, Wilmington, DE, U.S.A.) in ^a DuPont 830 LCapparatus fitted with a Waters U-6-K injection port. Elution was accomplished with the same solvent system at 14.5 MPa. A total of 55 fractions (1.5ml/min per fraction) was collected. The elution profile is shown in Fig. $2(a)$. Peak fractions 36-44 were combined, dried under N_2 and re-injected into the high-pressure liquid chromatography apparatus. The elution profile is shown in Fig. 2(b). The $25(OH)D_3$ peak region (fractions 34-42) was dried under N₂. The residue was dissolved in 20μ 1 of isopropanol/hexane $(2.5: 97.5, v/v)$ and applied to a microparticulate silica column $(4.6 \text{mm} \times 50 \text{cm})$ (PXS 1050 Partisil-10, Whatman Reeve Angel Inc., Clifton, NJ, U.S.A.) in ^a DuPont 830 LC apparatus. Elution was carried out with the same solvent system at 6.9MPa. A total of ³⁶ fractions (2.6ml/min per fraction) was collected. The elution profile is shown in Fig. $2(c)$. The peak fractions 16-21 were combined and dried under N_2 to yield a total of 582ng of $25(OH)D₃$. This material was analysed in the mass spectrometer (Associated Electrical Industries, Manchester, U.K.; model MS-902) by direct-probe inlet at 130–145°C above ambient temperature and 70eV ionization energy.

Results

Incubation of cholecalciferol with rat liver postmitochondrial fractions has resulted in an overall enzymic conversion of 0.4% into $25(OH)D₃$ (914ng) as calculated from the initial amount of substrate added, which was $250 \mu g$). Previous reports have shown that unlike the other two cholecalciferol hydroxylases, rat liver 25-hydroxylase exhibits lower activity when assayed in vitro. Besides, addition of saturating concentrations of cholecalciferol would be expected to decrease the percentage conversion of the substrate into the product. We have utilized such large amounts of cholecalciferol for isolation of sufficient amounts of the metabolite for analysis.

Fig. 2. High-pressure liquid chromatography of $25-(^{18}OH)[3\alpha^{3}H]D_3$

(a) Zorbax ODS (Dupont) column [4.6 mm ^x ²⁵ cm, eluted with water/methanol $(1:4, v/v)$] profile of the $25(^{18}OH)D_3$ region isolated from the Lipidex 5000 column shown in Fig. 1. The peaks on the left are due to impurities. (b) Zorbax-ODS (Dupont) column $[4.6 \text{mm} \times 25 \text{cm}]$, eluted with water/methanol $(1:4, 1:4)$ v/v] profile of 25(¹⁸OH)D₃ region isolated from (*a*). The peaks on the left are due to impurities. (c) Partisil-10 (Whatman) column [4.6mm×50cm; eluted with isopropanol/hexane (2.5:97.5, v/v)] profile of 25(18 OH)D₃ region isolated from (b). The peak on the left is due to the solvent front. The peaks between the arrows in (a), (b) and (c) are due to $25(OH)D_3$

The purification of the ¹⁸O metabolite depended on liquid/gel partition chromatography using Lipidex 5000 suspended in double-distilled organic solvents. The more refined technique of high-pressure liquid chromatography was used for further purification of the metabolite. An excellent resolution (Fig. 1) of $25(OH)D₃$ was observed from the unaltered undegraded substrate cholecalciferol, which is the major radioactive peak from the Lipidex column. As observed from the same Figure, several other radioactive peaks were obtained. Fractions 30-40 may well be esters of cholecalciferol or precholecalciferol. The other radioactive peaks, fractions 62-75 and appearing as shoulder peak of cholecalciferol (a) fractions 80-96 (b) and fractions $120-140$ (c) are unknown metabolites. The $25(OH)D_3$ obtained after three passes on high-pressure liquid chromatography was sufficiently pure to allow mass-spectral analysis (Fig. 3).

Fig. 3. Mass spectrum of isolated $25(^{18}OH)[3\alpha-^3H]D_3$ The product (582ng) from the Partisil-10 column (Fig. 2c) was analysed by mass spectrometry. The molecular ion at m/e 402 and the absence of one at m/e 400 are especially noteworthy, as are the fragments at m/e271, 253, 136 and 118.

The isolation and identification of the $25(OH)D_3$ have been reported by Blunt et al. (1968). The mass spectrum of the natural hydroxymetabolite was shown to exhibit a molecular ion peak at m/e 400 and characteristic fragment peaks at m/e 367 $(M-CH_3-H_2O)$, 349 (367-H₂O), 341 ($M-59$, loss of C_3H_7O from ring A), 271 (M – side chain), 253 (271 $-H₂O$), 158 (ring A+4 carbon atoms), 136 (ring A+ C-6 and C-7), 118 (136 – H₂O) and 59 [(CH₃)₂C=OH⁺, including C-25, C-26 and C-27 of side chain].

The mass spectrum of the ¹⁸O-labelled 25-hydroxy analogue of cholecalciferol (Fig. 3) was compared with that of the natural metabolite (Blunt et al., 1968). The molecular-ion peak at m/e 402 instead of 400 and the fragment peaks at m/e 369 (M - CH₃ - H₂O), 349 $(369 - H₂¹⁸O), 343 (M – 59), 271, 253, 158, 136, 118)$ and 61 $\text{[(CH₃²C=¹⁸OH) establish the structure of the$ metabolite as $[{}^{18}O]25(OH)D_3$. The incorporation of molecular O_2 as a hydroxy function into the C-25 position is shown by the peak at m/e 61 and the fact that all fragments do not include the side chain (e.g. m/e 271, 158, 136) and also do not contain the ¹⁸O-isotope. The absence of m/e 400, which would arise if water had provided the oxygen for the 25-hydroxy function, is noteworthy.

Discussion

The 25-hydroxycholecalciferol 1α -hydroxylase is the first of the cholecalciferol enzymes that was shown to be a mono-oxygenase ('mixed-function oxidase'). Ghazarian et al. (1973) demonstrated with $^{18}O_2$ that in this hydroxylation all of the oxygen inserted by chick kidney mitochondria into the 1α -position of $25(OH)D₃$ is derived from molecular $O₂$. The chick kidney mitochondrial 24-hydroxylase also incorporates molecular $^{18}O_2$ into 25(OH) D_3 to form 24,25- $(OH)_2[^{18}O]D_3$, which suggests that this system is also a mixed-function oxidase (Madhok et al., 1977). That the 25-hydroxylase also incorporates molecular ${}^{18}O_2$ into cholecalciferol to form $25(OH)[^{18}O]D_3$ demonstrates that this system is also a mixed-function oxidase. Thus investigations can now be directed to the exact enzymic machinery involved in this mixedfunction mono-oxygenase system.

We thank Dr. Robert Burris and Mr. Daniel Arp of this department for their assistance in handling commercial $18O₂$. This work was supported by U.S. Public Health Service Program Project grant no. AM-14881 and the Harry Steenbock Research Fund.

References

- Bhattacharyya, M. H. & DeLuca, H. F. (1974) Arch. Biochem. Biophys. 160, 58-62
- Blunt, J. W., DeLuca, H. F. & Schnoes, H. K. (1968) Biochemistry 7, 3317-3322
- DeLuca, H. F. (1974) Fed. Proc. Fed. Am. Soc. Exp. Biol. 33, 2211-2219
- Ghazarian, J. G., Schnoes, H. K. & DeLuca, H. F. (1973) Biochemistry 12, 2555-2558

 \bar{z}

- Ghazarian, J. G., Jefcoate, C. R., Knutson, J. C., Orme-Johnson, W. H. & DeLuca, H. F. (1974a) J. Biol. Chem. 249, 3026-3033
- Ghazarian, J. G. & DeLuca, H. F. (1974b) Arch. Biochem. Biophys. 160, 63-72
- Holick, M. F. & DeLuca, H. F. (1971) J. Lipid Res. 12, 460-465
- Horsting, M. & DeLuca, H. F. (1969) Biochem. Biophys. Res. Comnmun. 36, 251-256
- Kodicek, E. (1974) Lancet i, 325-329
- Madhok, T. C., Schnoes, H. K. & DeLuca, H. F. (1977) Biochemistry 16, 2142-2144
- Olson, E. B., Knutson, J. C., Bhattacharyya, M. H. & DeLuca, H. F. (1976) J. Clin. Invest. 57, 1213-1220
- Pedersen, J. I., Ghazarian, J. G., Orme-Johnson, N. R. & DeLuca, H. F. (1976) J. Biol. Chem. 251, 3933-3941
- Ponchon, G., Kennan, A. L. & DeLuca, H. F. (1969) J. Clin. Invest. 48, 2032-2037
- Tucker, G., Gagnon, R. E. & Haussler, M. R. (1973) Arch. Biochem. Biophys. 155, 47-57