AGONIST-EVOKED CHANGES IN CYTOSOLIC pH AND CALCIUM CONCENTRATION IN HUMAN PLATELETS: STUDIES IN PHYSIOLOGICAL BICARBONATE

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SUMMARY

1. Cytosolic pH (pH_i) and calcium concentration ($[Ca^{2+}]_1$) have been investigated in the presence and absence of physiological HCO_3^- in human platelets co-loaded with the fluorescent indicators BCECF and Fura-2. Basal pH_i and changes evoked by butyrate, thrombin, platelet activating factor (PAF), ADP and phorbol ester were investigated, as were the effects of removing external Na⁺.

2. In the presence of physiological HCO_3^- and CO_2 , basal pH_i was 7.02 ± 0.04 compared with 7.15 ± 0.05 in the absence of $HCO₃^-$. Estimated cytosolic buffering power was reduced from 35.6 ± 3.0 to 14.5 ± 0.4 mm/pH unit by the omission of $HCO₃$.

3. Thrombin evoked an immediate acidification of 0.03 ± 0.01 pH units in the presence of HCO_3^- and 0.07 ± 0.01 pH units in its absence. The acidifications were followed by a slow alkalinization. The final pH_i was 0.10 ± 0.01 units above basal in the presence of HCO_3^- and 0.08 ± 0.02 units above basal in the absence of HCO_3^- . The initial acidification was significantly greater in the absence of $HCO₃$. The subsequent increase in pH_i was similar in the presence and absence of this ion, but the calculated loss of proton equivalents was greater in the presence of $HCO₃^-$.

4. Replacement of extracellular Na^+ with N-methyl-D-glucamine resulted in a fall in basal pH_i and abolished recovery from thrombin-evoked acidification in both the presence and absence of $HCO₃⁻$.

5. In the presence of $HCO₃$, PAF and ADP evoked an intracellular acidification similar to that caused by thrombin. However, with PAF and ADP, the subsequent recovery in pH_i was slow and did not rise above basal levels. Phorbol dibutyrate, an activator of protein kinase C, evoked a similar elevation in pH_i of 0.04 ± 0.01 units over 3 min in the presence and absence of $HCO₃⁻$.

6. Stopped-flow fluorimetric measurements were made of both BCECF and Fura-2 fluorescence in the presence of $HCO₃⁻$. In the presence and absence of external $Ca²⁺$, thrombin-evoked rises in $[\text{Ca}^{2+}]_i$ peaked before any cytoplasmic alkalinization occurred. ADP evoked rapid elevations in $[Ca^{2+}]_i$, but caused no alkalinization.

7. These results (i) indicate that the absence of physiological $HCO₃⁻$ does not

qualitatively alter agonist-evoked changes in pH_i , although the presence of this ion affects basal pH_i and increases cytosolic buffering power, (ii) show that, even in the presence of HCO₃⁻, agonist-evoked alkalinization is dependent on external Na⁺, and (iii) provide further evidence that activation of $Na⁺-H⁺$ exchange and an elevation in pH_i are not a prerequisite for elevation of $[Ca^{2+}]$ _i in platelets.

INTRODUCTION

Stimulus-evoked increases in cytosolic pH (pH_i) or H^+ efflux have been widely reported in platelets (Horne, Norman, Schwartz & Simons, 1981; Zavoico, Cragoe & Feinstein, 1986; Siffert & Akkerman, 1987a; Simpson & Rink, 1987; Siffert, Siffert & Scheid, 1987; Zavoico & Cragoe, 1988). Similarly, thrombin or activators of protein kinase C have been demonstrated to promote $Na⁺-H⁺$ exchange across the platelet plasma membrane (Siffert et al. 1987). Yet, although stimulus-response coupling in platelets is comparatively well understood, with Ca²⁺, inositol phosphates, diacylglycerol and cyclic nucleotides recognized as having important signalling roles (e.g. Rink $\&$ Hallam, 1984), there is disagreement as to the part stimulus-evoked changes in pH_i may play in activation. Siffert & Akkerman (1987*a*) have proposed that cytosolic alkalinization is a prerequisite of Ca^{2+} mobilization in platelets, whilst others (e.g. Simpson & Rink, 1987; Zavoico & Cragoe, 1988; Alonso, Collazos & Sanchez, 1988) have concluded that changes in pH_i do not form a necessary component of the pathways leading to Ca^{2+} mobilization or the stimulation of shape change or aggregation. Moreover, these earlier reports were of experiments conducted in HEPES-buffered salines which lacked $HCO₃^-$. In many cell types the absence of HCO_3^- profoundly affects pH_i regulation (Thomas, 1984; Ganz, Boyarsky, Boron & Sterzel, 1987; Gillespie & Greenwell, 1988; Kettermann & Schlue, 1988; Ganz, Boyarsky, Sterzel & Boron, 1989; Thomas, 1989; Grinstein, Rotin & Mason, 1989), and at least three types of HCO_3^- transport across cell membranes have been identified (e.g. Boron, 1986). A very recent preliminary report shows data suggesting that HCO_3^- potentiates thrombin-evoked elevation of pH_i in human platelets (Ozaki, Yatomi, Kariya & Kume, 1989). We have therefore studied basal pH_i and the changes following addition of butyrate (to acidify the cytosol) and various agonists in platelets in the presence and absence of a physiological concentration of $HCO₃$. To assess the possible role of changes in pH_{i} in generating changes in cytosolic calcium concentration $([Ca^{2+}]_1)$ we have investigated both variables by stopped-flow fluorimetry (Rink & Sage, 1985; Sage & Rink, 1987) using cells co-loaded with the fluorescent indicators BCECF and Fura-2. In addition, we have examined changes in pH_i evoked by the Ca²⁺ mobilizing agonists ADP and platelet activating factor (PAF), and compared them with changes evoked by the more powerful activator, thrombin, on which earlier studies have concentrated.

METHODS

Preparation of cells

Blood was drawn from healthy volunteers and 8-5 ml aliquots were added to 1-5 ml acid citrate dextrose anticoagulant in plastic test tubes. The anticoagulant contained: trisodium citrate (2-5 g/

100 ml), citric acid (1.5 g/100 ml) and D-glucose (2 g/100 ml). This procedure gave a final whole anticoagulated blood citrate concentration of ²² mm and ^a pH of ⁶'5. Platelet-rich plasma was prepared by centrifugation for 5 min at 700 g at room temperature and 100 μ M-aspirin added. Platelets were then co-loaded with Fura-2 and BCECF by incubation for 45 min at 37 °C with 3μ M each of the respective acetoxymethylesters added from 5 mm stocks in dimethylsulphoxide (DMSO).The cells were then pelleted by centrifugation at 350 ^g for 20 min and as much of the supernatant as possible removed prior to resuspension.

In the present work, as in previous studies with platelets co-loaded with Fura-2 and BCECF, there was no evidence of significant cross-talk between the signals from the two dyes. The excitation wavelength for BCECF gives no Fura-2 emission (see e.g. Zavoico & Cragoe, 1988).The Fura-2 signals seen in the present work were closely similar to those we have previously seen in studies of singly loaded cells, as also found by Simpson & Rink (1987).

Solutions

The HEPES-buffered saline (HEPES) had the composition (mm) : NaCl, 145; KCl, 5; MgSO₄, 1; HEPES, 10 ; p-glucose, 10 . The HEPES was half-neutralized by the addition of 5 mm-NaOH , giving a final pH of 7.4 at 37 °C. The HCO_3^- -containing buffer (HCO_3^-) had the composition (mM): NaCl, 122 ; NaHCO₃, 23 ; KCl, 5; MgSO₄, 1; HEPES, 10; p-glucose, 10. The HEPES was halfneutralized by the addition of 5 mm-NaOH prior to the addition of the NaHCO₃ and the solution was gassed with 5% CO_2 -95% O_2 to give a final pH of 7.4 at 37 °C. Nominally Na⁺-free HCO₃⁻containing buffer (0 $\text{Na}^+\text{-HCO}_3^-$) had the following composition (mM): N-methyl-D-glucamine $(NMDG)$, 127, HCl, 122; $MgSO₄$, 1; HEPES, 10; KHCO₃, 23. The HEPES was thus halfneutralized by NMDG and the final pH was 7.4 at 37 \degree C when the solution was gassed with 5% CO₂. Nominally Na+-free HEPES-buffered saline (0 Na-HEPES) had the composition (mM): NMDG, 127; HCl, 122; $MgSO₄$, 1; HEPES, 10; KCl, 23; pH 7.4 at 37 °C. Potassium was elevated in this buffer so as to be comparable with the concentration necessitated in 0 Na- $HCO₃$ by the use of $KHCO₃$. To control for effects of elevated K⁺, modified Na⁺-containing media were prepared in which NaHCO₃ was replaced by KHCO₃ (23 mm-K⁺-HCO₃⁻) or NaCl was partially replaced by 23 mm-KCl (23 mm-K⁺-HEPES). The additional 5 mm-KCl present in control buffers was omitted from all four buffers used in the low-Na⁺ experiments (i.e. $[K^+]$ was 23 mm in each). Apyrase $(20 \ \mu g/ml)$ was added to all suspensions to prevent activation by residual traces of ADP. Solutions containing $HCO₃^-$ were bubbled with 5% $CO₂-95% O₂$ for at least 30 min prior to use. Platelet suspensions containing HCO_3^- were kept in equilibrium with a 5% CO_2 -95% CO_2 mixture at all times, the gas being delivered by polythene tubing to the plastic vessels containing stock suspensions, the reaction cuvettes and the holding reservoirs of the stopped-flow apparatus. Suspensions were maintained at 37 °C at all times. In some experiments, 1 mm-CaCl₂ was added to the cell suspension prior to use.

Measurement of pH,

The cell suspension was placed in cylindrical glass cuvettes in a specially constructed thermostatted holder in ^a Perkin-Elmer MPF 44A spectrophotometer. The suspension could be stirred with a magnetic stir-bar. If agonists were added, stirring was usually stopped 5 s after the addition to prevent aggregation. The BCECF fluorescence, with excitation at 490 nm, emission at 530 nm and a slit width of ⁵ nm, was recorded on a chart recorder. Dye leakage was assessed by measuring at intervals the supernatant fluorescence after centifugation to remove the cells. The leak signal, which was found to increase linearly at $0.44 \pm 0.02\%$ /min (s.E.M., $n = 50$), was corrected for pH if necessary and subtracted from all experimental values. The remaining signal was then scaled to that which would have been obtained if all the dye had been intracellular. The corrected signals were then calibrated by comparison with a calibration curve obtained from an aliquot of the same cell suspension which had been lysed to release trapped dye by the addition of 50 μ M-digitonin. The lysate pH was stepped from pH 60 to pH 80 by the addition of 1 M-HCl or ¹ M-NaOH as required. The pH was monitored in the cuvette by ^a WPA EL26 pH electrode (WPA Ltd, Linton, Cambridgeshire) connected to ^a Universal PTI-6 digital pH meter. The pH value obtained was then corrected for the red shift which occurs in the intracellular environment by the method of Rink, Tsien & Pozzan (1982). Briefly, dye-loaded cells were suspended in a modified HEPES-buffered saline containing ¹³⁵ mM-KCl and ¹⁵ mM-NaCl. The pH of the medium was then stepped from 8.0 to 6.0 by the addition of 1 M-NaOH or 1 M-HCl as required, after the addition of

33

nigericin (0-2 μ g/ml; from 0-2 mg/ml stock in DMSO). Nigericin is a H⁺-K⁺ antiporter which sets $[H^+]_i/[H^+]_o = [K^+]_i/[K^+]_o$ (Thomas, Buchsbaum, Zimniak & Racker, 1979). The cells were then lysed by the addition of 50 μ M-digitonin and the pH returned stepwise to 8.0 by the addition of ¹ M-NaOH, and correction curves comparing lysate and intracellular fluorescence plotted after adjustment for dye leakage as detailed above.

The results are expressed as mean \pm s.E. of mean (S.E.M.).

Measurement of $[Ca^{2+}]_1$, shape change and aggregation

Fura-2 fluorescence, with excitation at 339 nm and emission at 500 nm, was monitored as for BCECF above. The signals were corrected for dye leakage according to the method of Rink & Pozzan (1985) and calibrated in terms of $\lceil Ca^{2+} \rceil$ as previously described (Pollock, Rink & Irvine, 1986).

The optical density of the stirred platelet suspension could be measured simultaneously with fluorescence. Shape change and aggregation could thus be monitored by the turbidimetric method of Born (1970). Stirring was maintained in experiments assessing these parameters.

The results are expressed as mean \pm s.E.M..

Stopped-flow fluorimetry

Stopped-flow kinetic measurements of Fura-2 fluorescence were made as previously described (Sage & Rink, 1987). For BCECF fluorescence, the procedure was modified since an artificial fluorescence increase occurred on mixing cell and agonist solutions. The artifact was presumably due to slight pH imbalance between the cell suspension and agonist solution. Traces were corrected for this artifact by subtracting the fluorescence change determined on mixing cell suspension with agonist-free buffer.

The results are expressed as mean \pm s.E.M.

Materials

The acetoxymethyl esters of Fura-2 and BCECF were from Molecular Probes, Eugene, OR, USA. The Na2-ADP, apyrase, aspirin, butyric acid, 4,4'-diisothiocyanatostilbene-2,2'-disulphonic acid (DIDS), N-methyl-D-glucamine, nigericin, phorbol dibutyrate, sodium butyrate and bovine thrombin were from Sigma, Poole, Dorset. Platelet activating factor and HEPES were from Calbiochem, San Diego, CA, USA. Digitonin was from BDH Ltd, Poole, Dorset.

RESULTS

Basal pH_i and butyrate-induced acidification in the presence and absence of bicarbonate

In $HCO₃$ -containing medium, the basal pH_i determined from BCECF fluorescence was 7.02 ± 0.04 (s.e.m., $n = 10$), compared with 7.15 ± 0.05 ($n = 10$) in the absence of $HCO₃$. The difference was significant (Student t test, difference means, $0.05 > P >$ 0.02). Basal pH_i was similarly elevated in the absence of $HCO₃⁻$ in Na⁺-containing and $Na⁺$ -free media containing 23 mm- $K⁺$ (Table 1).

The addition of 50 mm-NMDG butyrate to platelets suspended in $Na⁺$ -free medium induced an initial fall in pH_i of 0.48 ± 0.04 units ($n = 6$) in the presence of $HCO₃⁻$ and 0.92 ± 0.06 units (n = 6) in the absence of $HCO₃⁻$ (not shown). Cytosolic buffering power, calculated as described by Szatkowski & Thomas (1986), was 35.6 ± 3.0 and 14.5 ± 0.04 mm/pH unit in the presence and absence of $HCO_3^$ respectively. The difference was significant (paired Student t test, $P < 0.001$).

Thrombin-evoked changes in pH_i in the presence and absence of bicarbonate

Figure ¹ shows responses evoked by thrombin (0 5 unit/ml) in platelets co-loaded with BCECF and Fura-2 and suspended in HEPES or HCO_3^- media. Figure 1A

Fig. 1. Responses evoked by thrombin (0.5 units/ml) in the presence and absence of $HCO₃$. $A,$ BCECF fluorescence calibrated in terms of pH_i. B, Fura-2 fluorescence calibrated in terms of $[Ca²⁺]$ _i (upper traces) and optical density (lower traces). Traces on the left are from experiments conducted in 23 mm-HCO₃⁻ and those on the right in the absence of HCO₃⁻.

shows BCECF fluorescence. Stimulation evoked an initial acidification followed by a later alkalinization. The fall below basal pH_i was 0.03 ± 0.01 units (s.e.m., $n = 10$) in cells in the presence of HCO_3^- and 0.07 ± 0.01 units (n = 10) in the absence of HCO_3^- . The difference was significant (Student t test, difference means, $0.01 > P > 0.001$). Three minutes after addition of thrombin, the pH_i had risen 0.10 ± 0.01 units (n = 10) above the basal level in HCO_3^- and 0.08 ± 0.02 units (n = 10) above basal in HEPES. The difference in the increment in pH_i was not significant ($P > 0.05$). The

35

initial rate of loss of proton equivalents, calculated using the cytosolic buffering powers determined above were $11 \cdot 1 \pm 1 \cdot 1$ and $5 \cdot 8 \pm 0 \cdot 7$ mm/min (s.g.m., $n = 10$ for each) in the presence and absence of $HCO₃⁻$ respectively.

Thrombin-evoked elevation in $\lceil Ca^{2+} \rceil$

The upper traces of Fig. $1B$ show Fura-2 fluorescence. Thrombin evoked similar rises in $[\text{Ca}^{2+}]_i$ in the presence and absence of HCO_3^- . In the presence of external

Fig. 2. Effect of Na⁺ removal on $\rm pH_{i}$ responses evoked by thrombin (0.5 units/ml) in the presence and absence of $HCO₃$. The upper traces show responses in control medium and lower traces responses in medium in which Na⁺ was completely replaced with NMDG. Traces on the left are from experiments conducted in 23 mm-HCO_3^- , those on the right in the absence of $HCO₃⁻$. [K⁺] was 23 mm in all media (see Methods).

Ca²⁺, $[Ca^{2+}]$ _i was elevated by 628 ± 30 nm (s.e.m., $n = 5$) in HCO_3^- and 611 ± 28 nm $(n = 5)$ in HEPES. The lower traces of Fig. 1B show the optical density of the platelet suspension. There was no discernable difference in the shape change or aggregation evoked by thrombin in the presence or absence of $HCO₃⁻$.

Effects of Na⁺ substitution on thrombin-evoked changes in pH_i

Figure 2 shows the effects of $Na⁺$ substitution in the presence and absence of $HCO₃$. On complete replacement of Na⁺ with NMDG, basal pH_i fell significantly in both the presence and absence of $HCO₃⁻$ (Table 1). Thrombin-evoked acidification was slightly, but not significantly, increased in the absence of external $Na⁺$, whilst the subsequent alkalinization was completely abolished in both the presence and absence of HCO_3^- (Table 1). To allow replacement of NaHCO₃ with KHCO₃, [K⁺] was elevated to 23 mm in control and Na⁺-free media (see Methods). This manoeuvre did not significantly alter basal or thrombin-evoked changes in pH_i in the presence or absence of $HCO₃⁻$ (Table 1).

The effects on thrombin-evoked changes in pH_i of DIDS, which blocks the three known $HCO₃$ ⁻ transport systems in other cell types, were also tested. In HCO_3^- -containing medium, 0.5 mM-DIDS slowed the rate of alkalinization following initial acidification evoked by thrombin (05 units/ml) from 0.22 ± 0.03 to 0.12 ± 0.03 units/min (S.E.M., both $n = 9$) (data not shown). The difference was

significant (paired Student t test, $0.05 > P > 0.02$). However, thrombin-evoked alkalinization was also significantly (001 > P > 0001) slowed by 0.5 mm-DIDS in HEPES-buffered, $HCO₃$ -free medium from 0.46 ± 0.09 to 0.16 ± 0.03 pH units/min. Since similar effects of DIDS were seen in the presence and absence of $HCO₃⁻$, these results suggest that the DIDS-induced slowing of

*** = $0.01 > P > 0.001$; **** = $P < 0.001$.

alkalinization is due to an action or actions other than on HCO_3^- transport. Somewhat surprisingly, 0 ¹ mM-DIDS increased the rate of alkalinization evoked by thrombin (05 units/ml) from 0.20 ± 0.03 to 0.34 ± 0.03 (both $n = 6$) pH units/min in the presence of HCO_3^- and from 0.29 ± 0.05 to 0.44 \pm 0.07 (both $n = 6$) pH units/min in the absence of HCO₃⁻. The rate of recovery from an intracellular acidification evoked by the addition of 50 mM-sodium butyrate in the presence of $HCO₃^-$ was also increased by the addition of DIDS (0.5 mm) from 0.02 \pm 0.00 (n = 9) to 0.03 \pm 0.00 $(n = 8)$ pH units/min. At present we can offer no explanation for the acceleration of alkalinization by the lower concentration of DIDS.

Changes in pH_i evoked by platelet activating factor, ADP and phorbol ester

Figure 3 shows changes in pH_i evoked by ADP, platelet activating factor (PAF) and phorbol dibutyrate in $HCO₃$ -containing medium in the presence or absence of external Ca²⁺. Figure 3A shows, in this batch of cells, the usual biphasic pH_i response to thrombin. Figure 3B and 3C show that in response to PAF (20 ng/ml), an initial acidification was followed by a slow recovery in pH_i to around the basal level; following ADP (20 μ M), pH_i was still below basal levels 3 min after stimulation. Figure 3D shows that phorbol dibutyrate evoked a slow rise in $\rm pH$, which was not preceded by a measurable acidification. Changes in pH_i evoked by all four agonists were not significantly different in the presence or absence of external Ca^{2+} (Table 2).

Stopped-flow kinetic measurements in the presence of bicarbonate

Figure 4 shows changes in BCECF and Fura-2 fluorescence evoked by thrombin (0.5 units/ml) or ADP (40 μ M) in the presence of HCO₃⁻ and recorded by stoppedflow fluorimetry. Cells were stimulated in both the presence and absence of external $Ca²⁺$ and responses recorded with a sampling interval of 100 or 10 ms. The agonistevoked falls in pH_i were variable in size. Acidification, when detectable, proceeded on a similar time course to agonist-evoked rises in $[Ca^{2+}]_i$. Thrombin-evoked alkalinization was markedly slower in onset.

Figure 4A shows that in the presence of external Ca^{2+} , the thrombin-evoked rise in $[Ca^{2+}]_i$, indicated by increased Fura-2 fluorescence, was delayed in onset by

Fig. 3. Cytosolic pH responses evoked in the presence of $HCO₃$ by: A, thrombin (0.5) units/ml), B, PAF, 20 μ l/ml) C, ADP (20 μ m) and D, phorbol dibutyrate (PDB; 200 nm). Traces on the left are from experiments conducted in the presence of 1 mm-external Ca^{2+} and those on the left in the absence of added $Ca²⁺$.

TABLE 2. Changes in pH_i evoked by thrombin, PAF, ADP and phorbol dibutyrate (PDB) in the presence of $HCO₃$

Agonist	$Ca02+$	Basal pH.	Initial fall in pH,	Final change in pH,	\boldsymbol{n}
Thrombin	\div	$7.04 + 0.03$	$-0.06 + 0.01$	$+0.14 + 0.01$	16
(0.5 units/ml)		$7.04 + 0.03$	-0.05 ± 0.01	$+0.11 \pm 0.01$	16
PAF	$\ddot{}$	$7.07 + 0.02$	$-0.04 + 0.01$	$+0.01 + 0.01$	17
(20 ng/ml)		$7.08 + 0.02$	-0.05 ± 0.01	$0.00 + 0.01$	17
ADP	\div	$7.03 + 0.03$	$-0.07 + 0.01$	$-0.04 + 0.01$	16
$(20 \mu \text{m})$		7.03 ± 0.02	$-0.07 + 0.01$	$-0.05 + 0.01$	16
PDB (200 nM)	\div	$7.12 + 0.04$ $7.11 + 0.05$		$+0.04 \pm 0.01$ $+0.04 + 0.01$	13 14

 0.57 ± 0.3 s (S.E.M., $n = 8$) and peaked at 4.4 ± 0.4 s ($n = 8$). The pH_i, indicated by BCECF fluorescence, did not rise back to the basal level until 6.0 ± 0.3 s ($n = 25$) after stimulation, by which time $[Ca^{2+}]_i$ had peaked. In the absence of external Ca^{2+} (Fig. 4B), the onset of rise in $[\text{Ca}^{2+}]_i$ was delayed by 0.67 \pm 0.3 s (n = 8) and peaked at 3.9 ± 0.5 s (n = 8), whilst pH_i did not rise back to the basal level until 5.7 ± 0.6 s $(n = 8)$ after stimulation.

39

Figure 4C shows that in the presence of external Ca^{2+} , ADP evoked a rise in $[Ca^{2+}]$ without measurable delay ($n = 5$). In the absence of external Ca²⁺ (Fig. 4D), the delay in onset of ADP-evoked rise in $[\text{Ca}^{2+}]_1$ was 210 ± 20 ms ($n = 5$). These Fura-2 signals are closely similar to those seen with these concentrations of thrombin and ADP in singly loaded cells in the absence of $HCO₃⁻$ (Sage & Rink, 1987). The small ADP-evoked acidifications, which were not followed by alkalinization, were not resolvable by the stopped-flow system and are not shown.

Fig. 4. Stopped-flow recordings. A and B show Fura-2 and BCECF fluorescence responses evoked by thrombin (0.5 units/ml) in the presence of 1 mm-external Ca^{2+} and HCO_3^- (A) or the absence of added Ca²⁺ (B). C and D show Fura-2 fluorescence responses evoked by 40 μ M-ADP in the presence and absence of external Ca²⁺. Note different time bases of thrombin and ADP traces.

DISCUSSION

These results provide the first reported measurements of ${[Ca^{2+}]}_i$ and pH_i in human platelets resuspended in media containing physiological concentrations of $CO₂$ and $HCO₃$. Fortunately, for us and the many other groups who have studied these variables in resting and stimulated platelets, we found no significant differences in the extent or kinetics of $[\text{Ca}^{2+}]_i$ rises evoked by thrombin, PAF or ADP in the presence and absence of $HCO₃⁻$. Nor were the changes in pH_i caused by these agonists markedly affected, being modified mainly in the way expected by the additional cytosolic buffering capacity in $HCO₃$ -containing medium. We find no evidence for an important role for $HCO₃$ -dependent $H⁺$ transport in human platelets under the conditions and within the time frame of our experiments.

Basal pH_i and cytosolic buffering power in the presence and absence of bicarbonate

Basal pH_i was significantly higher in the absence than in the presence of $HCO₃$. A similar elevation in basal pH_i in the absence of $HCO₃⁻$ has recently been reported in embryonic chick somitic cells (Gillespie & Greenwell, 1988). This difference may partly be due to a rise in pH_i caused by the loss of CO_2 from cells when resuspended in $HCO₃$ -free, HEPES-buffered medium from platelet-rich plasma. In addition, the platelet plasma membrane may have an appreciable HCO_3^- conductance which would result in the expulsion of this ion from the cytosol under the influence of a negative membrane potential, equivalent to an entry of H^+ . A small, progressive decline in basal pH_i was observed in all media, which was more pronounced in $HCO₃$ -free solutions. This may reflect a recovery following an elevation in pH_i, presumably a result of preparation of cells in the absence of $CO₂$. These results indicate that caution is required to allow recovery from disturbances in $\rm pH_i$ when changing solutions and that basal pH_i may be elevated under the unphysiological conditions of absence of $HCO₃⁻$. Our results suggest that $HCO₃⁻$ -dependent transport processes are not required for maintaining basal pH_i at levels far higher than electrochemical equilibrium in platelets; this contrasts with some other cells, in which pH_i is higher in the presence of $HCO₃⁻$ than in its absence (Ganz *et al.* 1987; Gillespie & Greenwell, 1988).

The absence of $HCO₃$ significantly reduced cytosolic buffering power. Hence many earlier reports may have overestimated changes in $\rm pH_{i}$ resulting from the activation of cells under unphysiological conditions.

Thrombin-evoked changes in pH_i in the presence and absence of bicarbonate

Thrombin evoked a small initial fall in pH_i followed by a sustained rise above basal levels in the presence and absence of $HCO₃⁻$. Similar responses, in the absence of $HCO₃$, have been reported previously (e.g. Zavoico *et al.* 1986; Simpson & Rink, 1987; Siffert & Akkerman, 1987 a; Zavoico & Cragoe, 1988). The thrombin-evoked acidification was twice as large in the absence of $HCO₃⁻$ as in its presence, reflecting the greater cytosolic buffering power in cells suspended in $HCO₃⁻$ -containing medium. Siffert, Siffert, Scheid & Akkerman (1989) have recently claimed that agonist-evoked intracellular acidifications are an artifact of platelet shape change. The present data do not support this hypothesis, since the absence of $HCO_3^$ increased the thrombin-evoked intracellular acidification but had no measurable effect on shape change. The reported absence of thrombin-induced acidification in platelets which had already changed shape (Siffert et al. 1989) suggests that the cytoskeletal changes or an associated increased metabolism are the main cause of the acidification. The alternative idea that the elevation in ${[Ca^{2+}]}_i$ causes the fall in pH_i seems unlikely in view of the observation that ionophore-induced elevation of $[Ca^{2+}]_i$ in platelets had no detectable effect on pH_i (Simpson & Rink, 1987).

Thrombin-evoked alkalinization is believed to be mediated by the activation of Na+-H+ exchange via the production of diacylglycerol and the activation of protein kinase C (e.g. Siffert et al. 1987; Siffert, Fox, Muckenoff & Scheid, 1984; Zavoico

et al. 1986). In the present experiments, similar thrombin-evoked increases in pH_i were observed in the presence and absence of $HCO₃$. However, the initial rate of loss of proton equivalents was greater in $HCO₃⁻$ containing medium. This might partly reflect the lower pH_i in the presence of HCO_3^- , since the Na⁺-H⁺ exchanger is enhanced by lowered pH_i; however, another role for HCO_{3}^{-} in speeding proton efflux cannot be ruled out.

Replacement of external Na^+ with NMDG resulted in a fall in basal pH_i in both the presence and absence of $HCO₃⁻$ (Table 1). Replacement of Na⁺ slightly, but not significantly, increased the thrombin-evoked acidification in both the presence and absence of HCO_3^- . There was no recovery following acidification in Na^+ -free medium with or without $HCO₃⁻$, confirming the Na⁺ dependence of the alkalinization. Our results contrast with those of Ozaki *et al.* (1989), who reported (i) that the presence of 10 mM-HCO₃⁻ enhanced thrombin-induced alkalinization and more strikingly, (ii) that the addition of 10 mm- HCO_3^- to Na⁺-free medium could restore a thrombinevoked elevation of pH_i . We do not understand the basis of the apparent contradiction between those results and those of the present study. However, we note that Ozaki et al. (1989) used Tris rather than HEPES buffer and that they give no details of their control of P_{CO_2} and external pH in HCO_3^- -containing media. These authors could not explain their data in terms of Na^+ -coupled HCO_3^- transport since the effects occurred in $Na⁺$ -free media; thus they appear to favour a role for 'anion channels'.

Thrombin-evoked rises in $[Ca^{2+}]$, were the same in the presence and absence of $HCO₃$, indicating that the observed differences in basal pH_i were without effect on the mechanisms gating Ca^{2+} fluxes.

Comparison of changes in pH_i evoked by thrombin, platelet activating factor, ADP and phorbol ester

Thrombin, PAF and ADP all evoked an early fall in pH_i in HCO_3^- -containing medium, in the presence or absence of external Ca^{2+} . Subsequent recovery of pH₁ varied between agonists. Only thrombin evoked a significant elevation in pH_i above basal levels. After PAF-evoked acidification, pH_i returned slowly to the basal level whilst after stimulation by ADP, pH_i was still below basal levels 3 min after stimulation. Recovery was not dependent on external $Ca²⁺$.

Phorbol dibutyrate evoked a slow alkalinization in the presence and absence of external Ca²⁺, confirming previous reports that $Na⁺-H⁺$ exchange is activated by protein kinase C (Zavoico et al. 1986; Siffert et al. 1987). The lack of acidification following stimulation by phorbol esters may be correlated with the lack of shape change evoked by these agents (e.g. Rink, Sanchez & Hallam, 1983). Physiologically, protein kinase C is activated by diacylglycerol, a product of the agonist-evoked hydrolysis of phosphatidylinositol bisphosphate. Thrombin is a more powerful stimulus for phosphatidylinositol bisphosphate hydrolysis than is PAF, whilst ADP is reported to evoke much less, or no, hydrolysis of it (Fisher, Bakshian & Baldassare, 1985; Daniel, Dangelmaier, Selak & Smith, 1986). The ability of thrombin alone to stimulate a sustained cytoplasmic alkalinization in platelets most probably reflects the greater effectiveness of this agonist in promoting a sustained production of diacylglycerol. The weak effect of PAF in promoting $Na⁺-H⁺$ exchange mirrors the

weak effect of this agonist in promoting Ca^{2+} efflux under conditions where thrombin or phorbol ester markedly speed Ca^{2+} extrusion (Rink & Sage, 1987). The much slower rises in pH_i evoked by PAF and ADP, which do not exceed basal levels, may reflect activation of H^+ extrusion by the fall in pH.

Relationship between changes in pH_i and $[Ca^{2+}]$ _i

Siffert & Akkerman (1987a) have suggested that a rise in $\rm pH_{i}$ is a prerequisite for $Ca²⁺$ mobilization in platelets. Their argument is based on their finding that replacement of external Na+ can suppress thrombin-evoked discharge of internal $Ca²⁺$ stores and slows and reduces $Ca²⁺$ influx. We have reported previously, however, that Na^+ replacement with choline has no significant effect on Ca^{2+} influx or release from internal stores evoked by thrombin or PAF (Sage & Rink, 1986). These conflicting results are not easily explained, although there were some differences in the methods of platelet preparation (Rink, 1987). Siffert & Akkerman (1987b) have also stated that thrombin-evoked Ca^{2+} mobilization is pH sensitive only at low doses of thrombin. A similar sensitivity of thrombin-evoked responses to $Na⁺$ removal has been reported by Connolly & Limbird (1983), who attributed the effect to a requirement for Na^+ in the generation of thromboxane A_2 , itself a stimulus to hydrolysis of phosphatidylinositol bisphosphate and the elevation of $[Ca^{2+}]$ in platelets (Hallam, Rink & Sanchez, 1983; Seiss, Siegel & Lapetina, 1983; Watson, Reep, McConnell & Lapetina, 1985). It has since been suggested that low doses of thrombin, as well as the weaker platelet agonists ADP and adrenaline, stimulate arachidonic acid release and so thromboxane A_2 production by evoking $Na^+ - H^+$ exchange which in turn activates a phospholipase A_2 (Sweatt, Blair, Cragoe & Limbird, 1986; Sweatt, Connolly, Cragoe & Limbird, 1986). Our results do not indicate any essential role for such a mechanism, since all the data were obtained using platelets in which cyclooxygenase, and so production of thromboxane A_2 , was blocked by prior treatment with aspirin.

Strong evidence against an essential role for an elevation in pH_i in platelet activation also comes from the present results. Only thrombin was found to evoke an eventual rise in pH_{i} , whilst after stimulation with PAF and ADP, pH_{i} remained below basal levels for several minutes. All three agonists are known to elevate $[Ca^{2+}]$ by stimulating both Ca^{2+} influx and the release of Ca^{2+} from intracellular stores (Rink, Smith & Tsien, 1982; Hallam, Sanchez & Rink, 1984; Hallam & Rink, 1985). Further, we believe compelling, evidence against a causal role for elevation of pH_i in agonist-evoked rises in $[Ca^{2+}]_i$ comes from the stopped-flow studies. In both the presence and absence of external Ca^{2+} , the thrombin-evoked rise in $[Ca^{2+}]_i$, as indicated by Fura-2 fluorescence, was under way at least $5s$ before pH_i , as indicated by BCECF fluorescence, exceeded basal values. In fact, in both the presence and absence of external Ca²⁺, thrombin-evoked rises in $[Ca^{2+}]_i$ had peaked before pH_i was above basal levels. We have reported previously that ADP, in the presence of external Ca²⁺, evokes rises in $[Ca^{2+}]$ _i without measurable delay, whilst in the absence of external Ca²⁺ ADP-evoked rises in $[Ca^{2+}]_i$ are delayed by only 200 ms (Sage & Rink, 1987). Here we have demonstrated similar ADP-evoked rises in $[\text{Ca}^{2+}]_i$ in the presence of $HCO₃$. These rapid $Ca²⁺$ signals are generated during the period of agonist-evoked cytoplasmic acidification. It has previously been reported that

changes in pH_i imposed by addition of $NH₄Cl$, CO₂ or nigericin do not significantly alter $[Ca^{2+}]$, (Simpson & Rink, 1987).

We therefore conclude that an elevation in cytoplasmic pH_i is not a *requirement* for the mobilization of Ca^{2+} in platelets. Agonist-evoked alkalinization, when present, may, however, serve to protect the cell from severe cytosolic acidification following a rapid increase in metabolic activity upon activation, as previously suggested (Simpson & Rink, 1987). In our hands, thrombin, but not PAF or ADP, evokes significant elevation in pH_i by activating $Na^+ - H^+$ exchange. All three agonists, however, mobilize internal Ca^{2+} (e.g. Hallam et al. 1984; Hallam & Rink, 1985; Rink & Sage, 1987). The role of agonist-evoked alkalinization, if any, does not thus appear to be regulation of $[Ca^{2+}]_1$. Even a role for changes in pH_i in modulating Ca^{2+} fluxes, for example by enhancing inositol trisphosphate-evoked discharge of the internal $Ca²⁺ stores (Brass & Joseph, 1985), seems unlikely in view of the relative time courses$ revealed by stopped-flow fluorimetry.

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