# The Immunochemistry of Shigella flexneri Lipopolysaccharides

A QUANTITATIVE ANALYSIS OF THEIR MONOSACCHARIDE CONSTITUENTS

By D. A. R. SIMMONS

Department of Bacteriology, University of Glasgow

(Received 22 October 1965)

1. The lipopolysaccharides of a representative selection of Shigella flexneri serotypes all contain the same constituents as Salmonella chemotype VII, namely, aldoheptose phosphate, <sup>3</sup> - deoxy - <sup>2</sup> - oxo - octonate, 0 - phosphorylethanolamine, D-galactose, D-glucose, N-acetyl-D-glucosamine and L-rhamnose. 2. The presence of all the Salmonella basal sugars in Sh. flexneri lipopolysaccharides is consistent with the view that the latter contain a basal structure or core which is similar to the common basal structure of Salmonella lipopolysaccharides. 3. Although the Sh. flexneri lipopolysaccharides belong to one chemotype, there appear to be quantitative differences in the composition of their 0-specific side chains. The repeating units of Sh. flexneri serotypes la, 2a, 3a, 4a, and variant X contain  $D$ -glucose, N-acetyl-D-glucosamine and L-rhamnose in the proportions  $1:1:2$ respectively. The analogous repeating units of serotypes 5a and 6 contain an additional mole of D-glucose and D-galactose respectively and that of variant Y <sup>1</sup> mole of D-glucose less.

There is now much evidence, recently reviewed by Liideritz, Staub & Westphal (1966), to support the view that the Salmonella lipopolysaccharides share a common basal structure or core to which side chains carrying the different 0-specific factors are attached. The basal sugars of the core, aldoheptose phosphate, 3-deoxy-2-oxo-octonate, 0-phosphorylethanolamine, D-galactose, D-glucose and N-acetyl-D-glucosamine, are therefore found in all Salmonella lipopolysaccharides whereas the 'special' (side chain) sugars associated with 0-specificity differ from chemotype to chemotype (Kauffmann, Luderitz, Stierlin & Westphal, 1960). The more recent demonstration of the same structural sequence in the core of Escherichia coli and Salmonella lipopolysaccharides (Edstrom & Heath, 1964) and the finding of the Salmonella basal sugars in Arizona and E. coli serotypes (Westphal, Kauffmann, Luderitz & Stierlin, 1960) indicate the probable presence of a similar basal structure in all Enterobacteriaceae lipopolysaccharides. In the Sh. flexneri that form the subject of this paper, the presence of all the basal sugars in a single lipopolysaccharide has yet to be demonstrated but reports of aldoheptose phosphate in Sh. flexneri serotype <sup>3</sup> (Slein & Schnell, 1953) and of 0 phosphorylethanolamine in an unspecified Sh. flexneri serotype (Grollman & Osborn, 1964) suggest that the lipopolysaccharides of this genus may also

possess a basal structure similar to that of the Salmonella.

The present study of eight representative Sh. flexneri lipopolysaccharides was undertaken to determine their basal sugar composition and to analyse quantitatively all their sugar constituents including those in the 0-specific side chains. Earlier studies (Goebel, Binkley & Perlman, 1945; Slein & Schnell, 1953; Simmons, 1957, 1962) have demonstrated glucose, N-acetylglucosamine and rhamnose as the major components of this group of polysaccharides.

## MATERIALS AND METHODS

Lipopoly8accharides. The lipopolysaccharides were extracted from acetone-dried cultures of Shigella flexneri by the phenol-water method and purified in the ultracentrifuge as described by Kauffmann et al. (1960). The N.C.T.C. strain numbers and serotypes of the eight smooth Sh. flexneri cultures used are given in Table 1 together with the weights extracted and the yields of lipopolysaccharide obtained.

Chromatography. This was carried out by the descending method on Whatman no. <sup>1</sup> paper with the following solvents: I, butan-l-ol-acetic acid-water (4:1:5, by vol.) (Partridge, 1948); II, butan-l-ol-pyridine-water (6:4: 3, by vol.) (Jeanes, Wise & Dimler, 1951); III, ethyl acetateacetic acid-water (3: 1: 3, by vol.) (Jermyn & Isherwood, 1949). Alkaline AgNO3 (Trevelyan, Procter & Harrison,



0-24 0-81 0-26 0-77 0-40 0-28

Table 1. Yield of lipopolysaccharides extracted from some representative Shigella flexneri



High-voltage electrophoresis. This was carried out on 2043 bmgl paper (Schleicher & Schiill, Dassel, Germany) according to the method of Kickhofen & Westphal (1952) with buffer A: pyridine-acetic acid-water (10:4: 86, by vol.), pH5-3. Electrophoresis was run at 3000v and 120mA for 1 hr.

voltage electrophoresis.

Shigella flexneri serotype la  $9a$ 3a 4a 5a 6 Variant X Variant Y

Hydrolysis of lipopolysaccharides. Hydrolysis was carried out in sealed glass ampoules, each containing approx. 3.0mg. of the lipopolysaccharide under study at  $1\%$  (w/v) concentration. Samples for amino sugar analysis were hydrolysed for  $12\,\text{hr}$ , at  $100^\circ$  with  $4\,\text{N-HCl}$ . Portions with an expected content of  $10-50 \,\mathrm{m} \mu \mathrm{m}$ oles of hexosamine were rapidly dried in vacuo to remove the acid. Samples for the analysis of neutral sugars were hydrolysed for  $4hr.$  at  $100^\circ$  with  $1 \text{ N-H}_2\text{SO}_4$ . The hydrolysate was diluted with 2 vol. of water and neutralized with Amberlite IR-410  $(HCO<sub>3</sub>– form)$  before drying portions in vacuo for microanalysis. Each portion contained  $10-100 \,\mathrm{m}_{\mu}$ moles of the sugar under study. For the determination of 3-deoxy-2 oxo-octonate, hydrolysis was for lOmin. at 100° with 0-25N-H2SO4. Negatively charged oligosaccharides, obtained by the elution of high-voltage electrophoretograms, were hydrolysed for  $12$  hr. at  $100^{\circ}$  with  $2N-\text{H}_2\text{SO}_4$ .

Quantitative microanalysis. Analyses were performed in 7-5 cm. x 1-0cm. tubes with Carlsberg constriction micropipettes (H. E. Pedersen, Copenhagen, Denmark) as follows: heptose by a modified  $H_2SO_4$ -cysteine method (Osborn, 1963); 3-deoxy-2-oxo-octonate by the thiobarbituric acid method of Waravdekar & Saslaw (1959); O-phosphorylethanolamine by the dinitrofluorobenzene method (Ghuysen & Strominger, 1963); D-galactose both by the galactose dehydrogenase method (Wallenfels & Kurz, 1962) and the galactose oxidase method (Fischer & Zapf, 1964); D-glucose by the glucose oxidase method (Kruger, Luderitz, Strominger & Westphal, 1962); L-rhamnose both by the  $H_2SO_4/c$ ysteine reaction (Dische & Shettles, 1951) and the L-rhamnose isomerase method (Krüger et al. 1962); free hexosamine by a modified Morgan-Elson reaction after N-acetylation (Strominger, Park & Thompson, 1959); D-glucosamine by the D-glucosamine-6 phosphate N-acetylase method (Luideritz, Simmons, West-

 $d$  & Strominger, 1964); N-acetylhexosamine by a dified Morgan-Elson reaction (Reissig, Strominger  $\&$ oir, 1955), and phosphate by the method of Lowry, perts, Leiner, Wu & Farr (1954). All measurements were made in the Unicam SP.500 spectrophotometer with standard microanalytical accessories. The results of the analyses are given in Table 2.

0-80 1-93  $0.52$ 2-56 1-11 2-15

Materials. The enzymes were obtained or prepared as follows: glucose oxidase (C. F. Boehringer und Soehne G.m.b.H., Mannheim, Germany); galactose oxidase and horse-radish peroxidase (Seravac Laboratories Ltd., Maidenhead, Berks.); galactose dehydrogenase was prepared by the method of Wallenfels & Kurz (1962) from a culture of Pseudomonas saccharophila (a product of Ciba, Basle, Switzerland) kindly given by Dr 0. Luderitz, Freiburg, Germany; D-glucosamine 6-phosphate N-acetylase was prepared according to Brown (1962) and L-rhamnose isomerase (also kindly given by Dr 0. Luderitz) was prepared by Krüger et al. (1962).

The chemicals were obtained or prepared as follows: D-glycero-L-mannoheptose was a gift from Dr N. K. Richtmyer (National Institutes of Health, Bethesda, Md., U.S.A.) and penta-acetyl methyl ester of 3-deoxy-2-oxo-octonate was a gift from Professor E. C. Heath (Johns Hopkins University, Baltimore, Md., U.S.A.). Free 3-deoxy-2-oxooctonate was prepared from this derivative by the method of E. C. Heath (personal communication) as follows: 10mg. of the derivative was dissolved in 1-Oml. of methanol; 1-Oml. of 0-2N-NaOH was added; after mixing and incubating at 37° for 15min. the mixture was treated with just sufficient Dowex 50 ( $H^+$  form) resin to adjust it to p $H5-6$ ; the resin was removed by filtration and the solution concentrated to dryness under reduced pressure. The sample was dissolved in water to give a molar concentration of 0-01. The pH should be adjusted to 6-5 if necessary.

All other chemicals, reagents and reference sugars were obtained from British Drug Houses Ltd., Poole, Dorset, and were either AnalaR grade or the highest grade obtainable.

### RESULTS

The quantitative results given in Table 2 show that the major constituents of the Sh. flexneri lipopolysaccharide hydrolysates are glucose, glucosamine and rhamnose. Galactose, heptose

phosphate, 3 - deoxy - 2 - oxo - octonate and 0 phosphorylethanolamine are present in smaller amounts. From their reactions with the specific enzymes used in the analyses the glucose, galactose and glucosamine are of the D-configuration and the rhamnose is of the L-configuration. The value obtained for the total free hexosamine in each hydrolysate as measured by the method of Strominger et al. (1959) was the same as that obtained for glucosamine by the specific N-acetylase technique (Liuderitz et al. 1964). The total amino sugar content ofeach lipopolysaccharide is thus accounted for as glucosamine. This conclusion was confirmed by the chromatographic absence of other hexosamines and by the finding of arabinose only when the hydrolysates were treated with ninhydrinpyridine buffer (Stoffyn & Jeanloz, 1954), which converts amino sugars into their parent pentoses. There is evidence that the free glucosamine in the Sh. flexneri lipopolysaccharide hydrolysates is N-acetylated in the native polysaccharides (Simmons, 1962). The values given in Table 2 are therefore expressed in terms of N-acetyl-Dglucosamine. The presence of galactose, glucose, rhamnose and O-phosphorylethanolamine was also confirmed chromatographically. Chromatographic evidence for the presence of heptose and 3-deoxy-2 oxo-octonate in these hydrolysates is unsatisfactory because the former could not be separated from a galactose/glucose mixture and the latter may be confused with oligosaccharides produced during hydrolysis. As heptose phosphate and 3-deoxy-2 oxo-octonate are negatively charged, their presence was convincingly demonstrated by high-voltage electrophoresis as follows: the electrophoretograms of the hydrolysates showed that the bulk of the material was in the neutral fraction but cathodic and anodic fractions were also present, the former being oligosaccharides containing basic amino sugars. Four anodic fractions (electrophoretic mobilities  $0.5, 0.6, 0.8$  and  $1.1$  times that of glucose 6-phosphate in buffer A) could be detected in each lipopolysaccharide hydrolysate and these were designated Al to A4 in order of increasing electrophoretic mobility. A4 was electrophoretically indistinguishable from authentic 3-deoxy-2-oxooctonate. Onelution, it reacted in thethiobarbituric acid reaction (Waravdekar & Saslaw, 1959) to give a chromogen with a spectrum indistinguishable from authentic 3-deoxy-2-oxo-octonate. Al, A2 and A3 all contained phosphorus on spraying with the reagent of Hanes & Isherwood (1949) and with naphtharesorcinol reagent all gave a purple colour indistinguishable from that of glucose 6-phosphate but readily distinguishable from the brightblue colour of glucuronic acid. A3, when eluted, hydrolysed and chromatographed, gave a single spot indistinguishable from authentic D-glycero-L-

mannoheptose in solvents I and II. This experiment does not differentiate between D-glycero-L-mannoheptose and L-glycero-D-mannoheptose but in all Enterobacteriaceae lipopolysaccharides where the configuration has been determined, it has proved to be the latter. Quantitative analysis of the free phosphate in this fraction gave the aldoheptose: phosphate ratio as  $1.00:0.71$  under the conditions of the experiment. Thus A3 appears to be Lglycero-D-mannoheptose phosphate or its optical enantiomorph. Al and A2 also contained heptose as evidence by absorption peaks at  $510 \text{m} \mu$  in the sulphuric acid/cysteine reaction (Dische & Shettles, 1951). Absorption peaks at  $410 \text{m} \mu$  in the same reaction indicated the presence of hexose, which on subsequent chromatography proved to be galactose. Al and A2 therefore appear to be oligosaccharides containing aldoheptose phosphate and galactose. The ratio of the sugars in these anodic oligosaccharides has not yet been established by quantitative analysis. Pentoses, 3,6-dideoxyhexoses and hexuronic acids have not been found by chromatography among the hydrolysis products of these lipopolysaccharides.

The finding of comparable amounts of all the Salmonella basal sugars in Sh. flexneri lipopolysaccharides is consistent with the view that both genera have a similar type of basal structure or core. If this basal structure contains approximately equivalent amounts of galactose, glucose and N-acetylglucosamine as in the Salmonella basal structure (Sutherland, Lüderitz & Westphal, 1965). then the bulk of the glucose and N-acetylglucosamine of the Sh. flexneri lipopolysaccharides lies in the 0-specific side chains and the ratio of the sugars in these side chains may be determined as in Table 3.

In the Sh. flexneri serotype 6 lipopolysaccharide the galactose ratio is so high that it appears probable that this sugar is also present in the 0-specific side chains. Here it has been assumed that the basal galactose accounts for  $1\%$  of the total lipopolysaccharide and that equimolar amounts of glucose and N-acetylglucosamine are also of basal origin. The analytical results (Table 2) show that the carbohydrate accounted for in serotype 6 is lower than with other serotypes. This lipopolysaccharido may be more resistant to hydrolysis as larger amounts of oligosaccharide were noted when equivalent weights of all lipopolysaccharide hydrolysates were chromatographed.

### DISCUSSION

The eight Shigella flexneri lipopolysaccharides used in this study all contain the same constituents, namely, aldoheptose phosphate, 3 deoxy-2-oxooctonate, 0-phosphorylethanolamine, D-galactose,

Table 2. Quantitative analysis of the sugar components of Shigella flexneri lipopolysaccharides

Each figure is a mean of eight or twelve determinations on each lipopolysaccharide hydrolysate. The values given are believed to be accurate within the following limits: heptose $\pm 0.9$ ; 3-deoxy-2-oxo-octonate $\pm 1.0$ ; O-phosphorylethanolamine $\pm 0.3$ ; galactose $\pm 1.2$ ; glucose $\pm 0.7$ ; N-acetylglucosamine $\pm 1.5$ ;  $channel: 0.$ 



The precise amount of the lipid A component in each lipopolysaccharide remains undetermined. From unpublished observations made during the ether extraction of acetic acid-degraded antigens (Simmons, 1957) it constitutes approx.  $30\%$  of the lipopolysaccharide complex.

# Table 3. Molar ratios of sugars in the specific side chains of Shigella flexneri lipopolysaccharides

The 'percentage of sugar in the O-specific side chains' was derived from the data in Table 2, with the following assumptions. All the galactose and equivalent amounts of glucose and  $N$ -acetylglucosamine are basal in origin. All the rhamnose lies in the O-specific side chains. In serotype 6 lipopolysaccharide where galactose appears to be a side-chain constituent, basal galactose amounts to 1% of the total lipopolysaccharide. In calculating the molar ratios, rhamnose was taken as 2.00.



D-glucose, N-acetyl-D -glucosamine and L-rhamnose. They therefore belong to a single chemotype indistinguishable from Salmonella chemotype VII (Kauffmann et al. 1960). The occurrence of glucose, N-acetylglucosamine and rhamnose as the major constituents of these lipopolysaccharides confirms earlier reports by Slein & Schnell (1953) in respect ofSh. flexneri type 3 polysaccharide and of Simmons (1957, 1962) in respect of the degraded haptens of all eight serotypes. Reports of small amounts of galactose in the polysaccharides of serotypes 6 (Efimova, 1959), 2a (Sasaki, 1956) and variant Y (Sasaki, 1957), of aldoheptose phosphate in serotype <sup>3</sup> polysaccharide (Slein & Schnell, 1953) and of O-phosphorylethanolamine in a Sh. flexneri species (Grollman & Osborn, 1964) are also confirmed and extended to the other types used in this study. 3-Deoxy-2-oxo-octonate, which was first isolated from  $E.$  coli 0111 (Heath & Ghalambor, 1963) and subsequently shown to be a common constituent of Salmonella lipopolysaccharides (see review by Lüderitz et al. 1966), has now been demonstrated for the first time in the analogous Sh. flexneri preparations. The presence of xylose (Sasaki, 1957), ofmannose and xylose (Sasaki, 1957) and of uronic acids (Efimova, 1959) in these lipopolysaccharides has not been confirmed with the techniques described above.

The finding of all the Salmonella basal sugars in many E. coli (see review by Lüderitz et al. 1966) and now in Sh. flexneri lipopolysaccharides suggests that many Enterobacteriaceae species have a similar lipopolysaccharide core. However, it should be stressed that the presence of a similar core in all Enterobacteriaceae does not necessarily imply that it is structurally identical from genus to genus. Indeed, comparative studies of the lipopolysaccharides from mutants of Salmonella typhimurium and E. coli 0111 have already demonstrated that, whereas the sugar sequence is the same in both (Edstrom & Heath, 1964), the nature of the linkage in a glucosylgalactose from the  $E$ . coli lipopolysaccharide is definitely different from that in the analogous glucosylgalactose from S. minneaota RII lipopolysaccharide (Sutherland et al. 1965).

The O-specific side chains of the Sh. flexneri lipopolysaccharides comprise the greater bulk of the total molecule and are probably composed of a number of repeating units analogous to those proposed for Salmonella groups B,D,E,G,N and U (Staub & Raynaud, 1964; Robbins & Uchida, 1962; Simmons, Lüderitz & Westphal, 1965a,b; Lüderitz, Simmons & Westphal, 1965). Although the Sh. fiexneri lipopolysaccharides all belong to one chemotype, there appear to be quantitative differences in the composition of their 0-specific side chains. The results in Table 3 are consistent with the view that the repeating units of serotypes

la, 2a, 3a, 4a and variant X contain glucose, N-acetylglucosamine and rhamnose in the proportions 1: 1: 2 respectively. The analogous repeating units of serotypes 5a and 6 appear to contain an additional mole ofglucose and galactose respectively while that of variant Y has <sup>1</sup> mole of glucose less. It remains to be seen if these quantitative differences are of significance in the immunochemistry of the determinant groups of the different Sh. flexneri serotypes.

The author acknowledges with thanks his indebtedness to: Professor Dr O. Westphal and Dr O. Lüderitz of the Max-Planck-Institut fur Immunbiologie, Freiburg, Germany, where the high-voltage-electrophoresis experiment was performed during the author's tenure of a Medical Research Council (Sir Henry Wellcome) Research Fellowship; the donors of enzymes and chemicals as indicated in the text and the Departments of Biochemistry at Queen's College, Dundee, and Glasgow University for the use of the ultracentrifuge facilities required in the preparation of the lipopolysaccharides. Part of this work was carried out while the author was in receipt of a Medical Research Council grant.

### REFERENCES

- Brown, D. H. (1962). In Methoden der enzymatischen Analyse, p. 151. Ed. by Bergmeyer, H.-U. Weinheim: Verlag Chemie.
- Consden, R., Gordon, A. H. & Martin, A. J. P. (1944). Biochem. J. 38, 224.
- Dische, Z. & Shettles, L. B. (1951). J. biol. Chem. 192, 279.
- Edstrom, R. D. & Heath, E. C. (1964). Biochem. biophys. Res. Commun. 16, 576.
- Efimova, A. A. (1959). Zh. Mikrobiol. 30, 100.
- Fischer, W. & Zapf, J. (1964). Hoppe-Seyl. Z. 337, 186.
- Ghuysen, J. M. & Strominger, J. L. (1963). Biochemistry, 2, 1110.
- Goebel, W. F., Binkley, F. & Perlman, E. (1945). J. exp. Med. 81, 315.
- Groilman, A. P. & Osborn, M. J. (1964). Biochemistry, 3, 1571.
- Hanes, C. S. & Isherwood, F. A. (1949). Nature, Lond., 164, 1107.
- Heath, E. C. & Ghalambor, M. A. (1963). Bibchem. biophys. Res. Commun. 10, 340.
- Jeanes, A., Wise, C. A. & Dimler, R. J. (1951). Analyt. Chem. 23, 415.
- Jermyn, M. A. & Isherwood, F. A. (1949). Biochem. J. 44, 402.
- Kauffmann, F., Liuderitz, O., Stierlin, H. & Westphal, 0. (1960). Zbl. Bakt., I. Abt. Orig., 178, 442.
- Kickh6fen, B. & Westphal, 0. (1952). Z. Naturf. 7b, 655.
- Krüger, L., Lüderitz, O., Strominger, J. L. & Westphal, O. (1962). Biochem. Z. 335, 548.
- Lowry, 0. H., Roberts, N. R., Leiner, K. Y., Wu, M. L. & Farr, A. L. (1954). J. biol. Chem. 207, 1.
- Luderitz, O., Simmons, D. A. R. & Westphal, 0. (1965). Biochem. J. 97, 820.
- Liuderitz, O., Simmons, D. A. R.,Westphal, 0. & Strominger, J. L. (1964). Analyt. Biochem. 9, 263.
- Lüderitz, O., Staub, A. M. & Westphal, O. (1966). Bact. Rev. (in the Press).
- Osborn, M. J. (1963). Proc. nat. Acad. Sci., Wash., 50, 499.
- Partridge, S. M. (1948). Biochem. J. 42, 238.
- Reissig, J. L., Strominger, J. L. & Leloir, L. F. (1955). J. biol. chem. 217, 959.
- Robbins, P. W. & Uchida, T. (1962). Biochemistry, 1, 323.
- Sasaki, T. (1956). Jap. J. Bact. 11, 1093.
- Sasaki, T. (1957). Jap. J. Bact. 12, 19.
- Simmons, D. A. R. (1957). J. gen. Microbiol. 17, 650.
- Simmons, D. A. R. (1962). Biochem. J. 84, 353.
- Simmons, D. A. R., Liideritz, 0. & Westphal, 0. (1965a). Biochem. J. 97, 807.
- Simmons, D. A. R., Lüderitz, O. & Westphal, O. (1965b). Biochem. J. 97, 815.
- Slem, M. W. & Schnell, G. W. (1953). Proc. Soc. exp. Biol., N.Y., 82, 734.
- Staub, A. M. & Raynaud, M. (1964). In The World Problem of Salmonellosi8, p. 8. Ed. by E. Van Oye. The Hague: Junk.
- Stoffyn, P. J. & Jeanloz, R. W. (1954). Arch. Biochem. Biophy8. 52, 373.
- Strominger, J. L., Park, J. T. & Thompson, R. E. (1959). J. biol. Chem. 234, 3263.
- Sutherland, I. W., Lüderitz, O. & Westphal, O. (1965). Biochem. J. 96, 439.
- Trevelyan, W. E., Procter, D. P. & Harrison, J. S. (1950). Nature, Lond., 166, 444.
- Wallenfels, K. & Kurz, G. (1962). Biochem. Z. 335, 559.
- Waravdekar, V. S. & Saslaw, L. D. (1959). J. biol. Chem. 234, 705.
- Westphal, O., Kauffmann, F., Liuderitz, 0. & Stierlin, H. (1960). Zbl. Bakt., 1. Abt. Orig., 179, 336.