

## Supplemental Materials

# Electrostatic influence of PsaC protein binding to the PsaA/PsaB heterodimer in Photosystem I

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## Supplemental Discussion

**Difference between the Henderson-Hasselbalch and the effective  $pK_a$ .** The difference between the Henderson-Hasselbalch  $pK_a$  and the effective  $pK_a$  is pronounced especially for strong electrostatic coupling between different titratable residues. The latter  $pK_a$  relates directly to measurable protonation probability of the considered residue. For instance, we obtained a protonation probability of 0.27  $H^+$  for Asp-B575 in the  $A_{1B}^-$  state of the P700- $F_X$  core at pH 7 (see main text). This protonation probability yields the effective  $pK_a$  value of 6.6 for this residue according to the following equation (1)

$$pK_a = pH + \frac{1}{\ln 10} \ln \frac{\langle x \rangle}{1 - \langle x \rangle} \quad (\text{eq. S1})$$

where  $\langle x \rangle$  is the protonation probability of the considered titratable residue. In cases where  $pK_a$  and solvent pH differ too much, the protonation probability  $\langle x \rangle$  is close to zero or one. Under such circumstances the evaluated  $pK_a$  is error prone, since the numerical accuracies of the computed probabilities  $\langle x \rangle$  are too low. In this case, the Henderson-Hasselbalch  $pK_a$  definition becomes more useful. Although, it is an invasive method, since it changes the protonation state of the considered residue by using a bias potential.

For the same charge state  $A_{1B}^-$ , the Henderson-Hasselbalch  $pK_a$  of Asp-B575 in the P700- $F_X$  core evaluated with application of a bias potential is 6.9 (Table 1 in the main text), which differs by 0.3  $pK_a$  units (corresponding to an energy of 18 meV) from that obtained with eq. S1. Asp-B575 is located in the inner water network between  $A_{1A/B}$  and  $F_X$  (2). When we constrain the protonation state of Asp-B575 to be 0.5  $H^+$  as is done with a bias potential (i.e. using the definition of the Henderson-Hasselbalch  $pK_a$ ), we simultaneously obtain tiny changes of the protonation pattern of nearby titratable residues, namely deprotonation at Asp-B555, Asp-B558, Glu-B682 and Glu-F98 (see Fig. 1A in the main text). These local deprotonations induced by constraining Asp-B575 to be half protonated are the main source that yields different values for the Henderson-Hasselbalch and the effective  $pK_a$ .

## Supplemental References

1. Ullmann, G.M., and E.W. Knapp. 1999. Electrostatic models for computing protonation and redox equilibria in proteins. *Eur. Biophys. J.* 28:533-551.
2. Jordan, P., P. Fromme, H.T. Witt, O. Klukas, W. Saenger, and N. Krauß. 2001. Three-dimensional structure of cyanobacterial photosystem I at 2.5 Å resolution. *Nature* 411:909-917.

## Supplemental Table

**Table S1:** Calculated  $pK_a$  values of residues for the native PSI complex and the P700-F<sub>X</sub> core in the P<sub>B</sub><sup>0</sup> state.

residues of PsaA	redox state <sup>a</sup>	native PSI	P700-F <sub>X</sub> core	residues of PsaB	redox state <sup>a</sup>	native PSI	P700-F <sub>X</sub> core
Asp-A568		3.9	4.7	Asp-B555		4.1	5.6
(Asn-A571) <sup>b</sup>				Asp-B558		8.5	5.7
Asp-A579		-0.2	2.0	Asp-B566		-4.8	3.2
Arg-A583		22.3	13.6	Arg-B570		21.4	15.5
(Gln-A588) <sup>b</sup>				Asp-B575		5.4	5.1
					A <sub>1A</sub> <sup>-</sup>	8.9	8.3
					A <sub>1B</sub> <sup>-</sup>	7.4	7.0
					F <sub>X</sub> <sup>-</sup>	9.4	8.6
Glu-A699		-2.6	-2.4	Glu-B679		-4.2	-4.3
	A <sub>1A</sub> <sup>-</sup>	-2.7	-2.5		A <sub>1B</sub> <sup>-</sup>	-3.9	-3.9
Glu-A702		-1.3	2.6	Glu-B682		4.8	5.2
	A <sub>1A</sub> <sup>-</sup>	-1.3	2.8		A <sub>1B</sub> <sup>-</sup>	6.3	5.9
(Gln-A718) <sup>b</sup>				Lys-B702		16.4	10.0

<sup>a</sup> If no redox state is indicated all redox active cofactors are in the neutral charge state.

<sup>b</sup> Non-titratable residue in PsaA, which is symmetry related to a titratable residue in PsaB.