Calcium stimulation of gastrin and gastric acid secretion: effect of small doses of calcium carbonate

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SUMMARY Oral calcium carbonate (0.5 g, pH 9.4) increased serum gastrin and gastric acid output with slight but insignificant change in serum calcium. A similar rise in serum calcium during an intravenous infusion of calcium gluconate failed to increase serum gastrin and gastric acid output. Both intragastric calcium actions were abolished by acidification of the calcium carbonate solution (pH 1.0). The increase in serum gastrin and gastric acid output after intragastric calcium carbonate was not affected, however, by a simultaneous intraduodenal acid load. Equivalent neutralising doses of magnesium hydroxide (pH 9.4) did not increase serum gastrin and gastric acid output above basal levels, whereas antral acidification with 20 ml 0.1 N HCl resulted in a slight decrease in serum gastrin. Intraduodenal calcium carbonate (pH 3.0) also increased serum gastrin and gastric acid output, whereas an equivalent volume of intraduodenal saline (pH 3.0) had no effect. These findings indicate that calcium increases serum gastrin by local stimulation of antral and duodenal mucosa. They also suggest that the action of calcium on gastric secretion is partly mediated by gastrin.

Calcium increases the serum levels of immunoreactive gastrin and stimulates gastric acid secretion (Barreras, 1973). Intravenous administration of calcium gluconate causes a progressive rise in serum calcium, stimulates gastric acid secretion (Murphy *et al.*, 1966; Barreras and Donaldson, 1967; Smallwood, 1967), and increases the levels of serum gastrin (Reeder *et al.*, 1970). Oral ingestion of calcium stimulates acid secretion in duodenal ulcer patients in both the fasting and post-cibal states (Barreras, 1970; Fordtran, 1968), with a concomitant increase in serum gastrin (Reeder *et al.*, 1971).

However, the mechanisms whereby calcium increases the serum levels of immunoreactive gastrin and stimulates gastric acid secretion have not been conclusively defined. There is conflicting evidence as to whether ingestion of calcium carbonate selectively increases gastrin by virtue of calcium (Levant *et al.*, 1973) or as the result of antral alkalinisation (McGuigan and Trudeau, 1970; Hansky *et al.*, 1971). Furthermore, it has been proposed that calcium-

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stimulated gastric acid secretion is partly mediated by gastrin. The possibility, however, that the increase in serum gastrin and gastric secretion in response to calcium are parallel but independent phenomena has not been excluded (Levant *et al.*, 1973; Rosato *et al.*, 1973).

The present studies were designed to investigate the interaction of calcium, serum levels of immunoreactive gastrin, and gastric acid secretion with respect to: (1) the site of calcium stimulation; and (2) whether the calcium effect on gastric acid secretion is mediated by gastrin.

Methods

Twenty-five male patients with uncomplicated duodenal ulcer (DU) were studied. Their mean average was 54 years. DU was documented by upper gastrointestinal radiographs and/or duodenoscopy. Patients with duodenal ulcer participated at least in two and in as many as five paired studies. The means \pm SD of basal acid secretion (BAO) was $6\cdot8 \pm 4\cdot5$ and histamine-stimulated gastric acid secretion (PAO) was $40\cdot2 \pm 10\cdot0$ mEq per hour. Informed consent was obtained from all patients.

TECHNIQUES

Paired, randomised gastric secretory studies were carried out after a 12 hour overnight fast. A radioopaque Levin Tube (F no. 14) was placed in the most dependent portion of the stomach under fluoroscopic control. After the stomach was completely emptied, basal and stimulated gastric secretions were collected by continuous pump and hand suction. Gastric juice hydrogen ion concentration was determined by the pH electrode (Moore and Scarlata, 1965). The neutralising equivalence of CaCO₃ and MgOH was determined *in vitro* (US *Pharmacopeia*, 1970). Each test was performed on a different day.

Total serum and gastric juice calcium were determined by atomic absorption spectroscopy. Serum gastrin was determined by radioimmunoassay by a modification of previously described methods (McGuigan and Trudeau, 1970; Yalow and Berson, 1970; Stadil and Rehfeld, 1973). All serum samples were stored at -25° C within one hour of collection and remained at this temperature until immediately before assay. Synthetic human gastrin (Imperial Chemical Industries, England) (SHG-I) and rabbit antiserum cospecific for SHG-I and II (Calbiochem Corp., California) were used. SHG-I was iodinated by a modification of the chloramine-T procedure (Hunter and Greenwood, 1962) with Na¹²⁵I (Amersham Searle Corp., Chicago, Ill.) (less than 1%¹²⁶I). All assays were performed in triplicate on two separate occasions. Gastrin values from individual assays were determined as a mean gastric value (n = 6), and its concentration expressed in picomole of immunoreactive gastrin/l serum (Stadil and Rehfeld, 1973). The coefficient of variation was less than 6% over a range of 12 to 210 pmol gastrin/l serum.

In vitro binding of calcium to gastric juice proteins was studied using pooled gastric juice and Amicon Centriflo ultrafiltration cones (Amicon Corp., Lexington, Mass.) for separation of free and protein bound calcium. Gastric juice was obtained during histamine stimulation from six DU patients and collected at 4°C; 0.555 g (5 mmol) calcium chloride was added to 20 ml aliquots of gastric juice and the final pH was adjusted with HCl or NaOH to pH 1.0, 2.0, 3.0, 4.0, and 7.0. Each aliquot was allowed to equilibrate in a water bath at 37° with continuous and gentle shaking for 30 minutes. Then 5 ml of the test sample was pipetted into the ultrafiltration cone. The assembly was centrifuged at 800 rpm for 10 minutes to collect a volume of ultrafiltrate (U) equivalent to approximately 10% of the original (O) volume. Calcium concentration was determined in the U and O samples. The percent calcium bound to gastric juice proteins is the difference between the two concentrations (C) and calculated as follows:

%CA bound =
$$\frac{CO - CU}{CO} \times 100$$

EXPERIMENTAL DESIGN

Studies with calcium carbonate or magnesium hydroxide

Experiments were performed with single doses of calcium carbonate (0.5 g) suspended in water (pH 9.4) or adjusted to pH1.0 or 3.0 with HCl. Magnesium hydroxide gel was used at an equivalent neutralising dose of 0.4 g magnesium hydroxide in water (pH 9.4). Basal gastric secretion was collected for 30 minutes immediately before the antacid administration. Then the antacid, in a final volume of 20 ml, was administered through the nasogastric tube. In certain studies $CaCO_3$ (0.5 g), saline or HCl solutions were instilled in the third portion of the duodenum through a radio-opaque tube F no. 12 positioned under fluoroscopic control. The gastric tube was clamped for 25 minutes. The stomach was emptied for five minutes and gastric collections were obtained in subsequent 30-minute intervals (30-60, 60-90, and 90-120 minutes). Blood samples were drawn 30 and 15 minutes before, and 30, 60, 90, and 120 minutes after the antacid administration for analysis of serum calcium and gastrin.

Studies with intravenous calcium gluconate

Intravenous infusions of calcium gluconate were administered as previously described (Barreras and Donaldson, 1967). Patients received 2 mg elemental calcium per kg of body weight per hour. Gastric secretion was collected in 30 minute periods during the saline (one hour) and calcium (two hours) infusions. Blood samples were obtained every 30 minutes for serum calcium and gastric determination.

Statistical analysis

The Willcoxon-sign test was applied to all experiments with six or more observations and the binomial distribution was applied to experiments with less than six observations.

Results

STUDIES WITH CALCIUM CARBONATE

Small doses of intragastric calcium carbonate in water (pH 9·4) increased serum gastrin (P < 0.05) and gastric acid secretion (P < 0.001) above basal levels, without a significant increase in serum calcium concentration (P > 0.05) in six patients with DU (Fig. 1). The same dose of calcium carbonate adjusted to pH 1.0, however, did not cause an

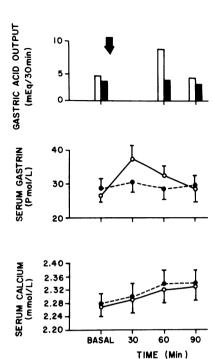


Fig. 1 Effect of intragastric dose of 0.5 CaCO_3 in 20 ml water (pH 9·4) or in 20 ml HCl (pH 1·0) on serum calcium, serum gastrin, and gastric acid output in six DU patients. Arrow indicates the timing of administration of test solution. Values of gastric acid output are means and of serum gastrin and calcium are means \pm SE. White symbols: $0.5 \text{ g CaCO}_3(\text{pH 9·4})$. Black symbols: 0.5 g CaCO₃ (pH 1·0)

increase in serum gastrin and gastric acid output (P > 0.05). The effect of the final pH of the calcium solutions on antral pH was determined in separate studies in four of these patients. The antral pH was measured in 10 ml aliquots of gastric juice collected before and at five minute intervals for 30 minutes. Basal pH of 1.3 increased after the administration of calcium carbonate (pH 9.4) and remained above 5.5 throughout the 30 minute period. A decrease in antral pH was observed after calcium carbonate (pH 1.0) remaining below 1.3 for the 30 minute period.

To investigate the effect of acid on the calcium action, 0.5 g CaCO₃ (pH 9.4) was instilled in the antrum through a gastric tube and 20 ml HCl (pH 1.0) was instilled in the third portion of the duodenum through a duodenal tube, simultaneously. Under these experimental conditions, calcium caused an increase in serum gastrin (P < 0.05) and gastric acid output (P < 0.05) above basal levels in six patients with DU (Fig. 2). The increase in serum gastrin and gastric acid output after the simultaneous

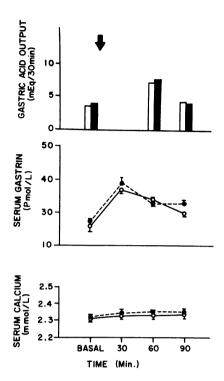


Fig. 2 Serum calcium, serum gastrin, and gastric acid output in response to either simultaneous intragastric dose of 0.5 g CaCO₃ (pH 9·4) and 20 ml intraduodenal HCl (pH 1·0) or to intragastric dose of 0·5 g CaCO₃ (pH 9·4) alone in six DU patients. Arrow indicates the timing of administration of test solutions. Values of gastric acid output are means and of serum gastrin and calcium are means \pm SE. Antrum: \bigcirc 0·5 g CaCO₃ (pH 9·4) in antrum; \bigcirc 0·5 g CaCO₃ (pH 9·4) in the antrum and 20 ml HCl (pH 1·0) in the duodenum

administration of intragastric calcium carbonate and intraduodenal HCl was not different from that observed after intragastric calcium carbonate alone (P > 0.05). Equivalent neutralising doses of magnesium hydroxide (0.4 g in water, ph 9.4) did not increase serum gastrin above basal levels (Fig. 3, P > 0.05) in six patients with DU. Administration of 20 ml 0.1 N HCl (pH 1.0) in the antrum caused a slight decrease in serum gastrin below basal levels at 60 minutes (P < 0.05).

Studies were conducted in seven patients to determine the site of the action by performing the following experiments: (1) CaCO₃ (pH 9·4) instilled in the antrum through the gastric tube or adjusted to pH $3\cdot0$ instilled in the third portion of the duodenum through a duodenal tube. The concentration and output of calcium was measured in the gastric contents before and every 30 minutes after the administration of calcium carbonate in the duodenum. Calcium stimulation of gastrin and gastric acid secretion

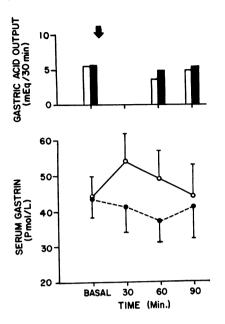


Fig. 3 Effect of intragastric dose of magnesium hydroxide (0.4 g) in 20 ml water (pH 9.4) or of 20 ml HCl(pH 1.0) on serum gastrin and gastric acid output in six DU patients. Arrow indicates the timing of administration of test solutions. Values of gastric acid output are means and of serum gastrin are means \pm SE. $\bigcirc \square$ Mg OH. $\bigcirc \blacksquare$ HCl.

Any increase in total calcium in the gastric contents above basal levels was considered calcium reflux in the antrum. Two patients with calcium reflux in the antrum were excluded; and (2) CaCO₃ or a similar volume of isotonic saline (pH 3·0) was instilled in the third portion of the duodenum through a duodenal tube in the remaining five patients. Serum gastrin and gastric acid output increased above basal levels in response to calcium administration at pH 9·4 in the antrum (P < 0·05) or pH 3·0 in the duodenum (P < 0·05). Both secretory responses were equivalent (P > 0·05, Fig. 4). Serum gastrin and gastric acid output, however, did not increase in response to intraduodenal saline at pH 3·0 (Fig. 5, P > 0·05).

STUDIES WITH INTRAVENOUS CALCIUM GLUCONATE (Fig. 6)

During the constant intravenous infusion of calcium gluconate, serum calcium steadily rose; however, the immunoreactive levels of serum gastrin and gastric acid output began to rise 60 minutes after the onset of the calcium infusion. An increase in serum calcium from 2.31 ± 0.01 to 2.38 ± 0.2 mmol/l (d = 0.07 mmol/l) had no effect on serum gastrin or gastric acid output.

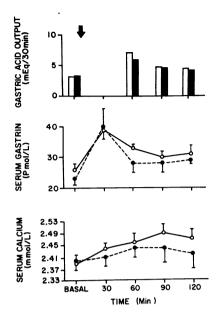


Fig. 4 Effect of intragastric dose (0.5 g) of calcium carbonate (pH 9.4) or of intraduodenal dose (0.5 g) of calcium carbonate (pH 3.0) on serum calcium, serum gastrin, and gastric acid output in five DU patients. Arrow indicates the timing of administration of test solutions. Values of gastric acid output are means and of serum gastrin and calcium are means $\pm SE$. $\bigcirc \square$ Gastric.

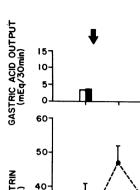
In vitro CALCIUM BINDING STUDIES

These experiments were designed to determine the influence of pH of gastric juice on calcium binding to protein. Deacidification of gastric juice caused a gradual increase in calcium binding in six *in vitro* studies. When compared with the calcium binding observed at pH of 1.0 and 2.0, alkalinisation of the aliquots of gastric juice increased the mean \pm SE binding of calcium by 4.2 \pm 0.4% at pH 3.0 by 5.9 \pm 0.6% at pH 4.0, and by 10.4 \pm 0.5% at pH 7.0 (p < 0.05).

Discussion

Small doses of intragastric calcium carbonate (0.5 g)increased gastric acid secretion and serum levels of immunoreactive gastrin. These small doses of calcium carbonate caused a slight but statistically insignificant increase in serum calcium. Similar increases in serum calcium during intravenous infusion of calcium gluconate, however, failed to stimulate gastrin secretion and gastric acid output.

The specificity of the calcium effect was evidenced by the observation that equivalent neutralising doses



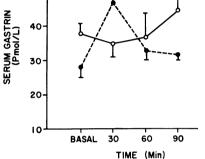


Fig. 5 Serum gastrin and gastric acid output in response to intraduodenal dose of 0.5 g calcium carbonate (pH 3.0) or 20 ml saline (pH 3.0) in five patients with DU. Arrow ndicates the timing of administration of test solutions. Values of gastric acid output are means and of serum gastrin are means \pm SE. \bigcirc Saline. \bigcirc \bigcirc 0.5 g CaCO₃

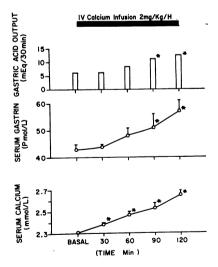


Fig. 6 Effect of intravenous saline (basal) and calcium gluconate (2 mg/kg body wt/h) on serum calcium, immunoreactive levels of serum gastrin, and gastric acid output. Values of gastric acid output are means and of serum gastrin and calcium are means \pm SE of six experiments. *Means significantly greater than basal values

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of magnesium hydroxide did not increase gastrin or stimulate acid secretion. These calcium effects, however, were completely blocked by acidification of the calcium carbonate solution to pH 1.0. Acidification of the calcium solution appears to inhibit the effect of calcium on the antral mucosa and perhaps on the fundic mucosa as well (Debas and Grossman, 1975), as these observations cannot be entirely explained by simultaneous inhibition of spontaneous gastrin secretion, by secretin stimulation, or by the gastric inhibitory response to intraduodenal acid. Evidence has been presented previously that exogenous secretin decreased basal serum levels of immunoreactive gastrin (Hansky et al., 1971) and that duodenal acidification inhibited the pentagastrin stimulated gastric acid secretion (Ward and Bloom, 1974) in duodenal ulcer patients. It is conceivable that the small intragastric and intraduodenal acid loads used in this study were not sufficient to stimulate the hormonal and neural mechanisms required to influence the effect of calcium carbonate on serum gastrin and gastric acid output. These findings are consistent with the hypothesis that small doses of calcium carbonate increase serum gastrin by local stimulation of the antral and duodenal mucosa which is favoured in an alkaline environment (Feurle, 1965). Thus, the action of calcium appears to be similar to the action of all other mucosal stimuli of antral gastrin which are inhibited by antral acidification (Redford and Schofield, 1965: Grossman, 1967).

The parallel increase in serum gastrin and gastric acid output in response to small doses of calcium carbonate, coupled with simultaneous inhibition by acidification of the calcium solution, suggest that calcium-stimulated gastric acid secretion could be partly mediated by gastrin. This assumption is further supported by the finding that calcium stimulates gastric acid secretion without coming in direct contact with parietal cells. In contrast with intraduodenal saline, small doses of calcium carbonate in the duodenum increased serum gastrin and gastric acid secretion without changing serum calcium significantly. Gastrin has been detected in the human duodenum by bioassay and immunoassay methods (Emås et al., 1971) and duodenal gastrin has been stimulated by amino-acids (Haves et al., 1974). These findings, however, do not preclude the possibility that small doses of calcium carbonate stimulated parietal cell secretion by direct action which could be inhibited by acidification (Debas and Grossman, 1975; Holtermuller et al., 1974), or by additional humoral (Holtermuller et al., 1976) or neural mechanisms triggered from the stomach or duodenum (Harper et al., 1959; Sircus, 1953).

The physicochemical interactions of calcium

carbonate and the components of gastric secretion were not adversely affected by acidification. Calcium must be solubilised to exert an effect and partial ionisation of calcium carbonate by adjustment to pH 3 is required to increase the serum levels of immunoreactive gastrin in patients with pernicious anaemia (Behar and Smyth, unpublished observations). Calcium binding to gastric proteins *in vitro* was also reduced by acidification, in accord with previous observations (Carr, 1955) of maximum binding of calcium to crystalline pepsinogen at pH $7\cdot4$.

In agreement with the concept advanced by Grossman that antral deacidification enhances the release of gastrin only in the presence of positive stimuli (Grossman, 1967), the antral action of small doses of calcium appears to require a relatively alkaline environment (Feurle, 1975). However, antral alkalinisation *per se* does not explain the calcium effect. Equivalent neutralising solutions of small doses of non-calcium antacids failed to increase serum gastrin and gastric acid output above basal levels (Barreras, 1970). Furthermore, partially solubilised (ionised) calcium carbonate is able to increase the levels of serum gastrin in patients with pernicious anaemia and histamine-achlorhydria.

In summary, these studies indicate that small doses of calcium carbonate increase the serum levels of immunoreactive gastrin by local stimulation of the antral and duodenal mucosa. They also suggest that calcium stimulation of gastric acid secretion is partly mediated by increasing the serum levels of immunoreactive gastrin. Additional direct calcium action on parietal cells or through neural and other humoral mechanisms could not be excluded.

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