# HUMAN PLATELET PHENOLSULPHOTRANSFERASE M AND P: SUBSTRATE SPECIFICITIES AND CORRELATION WITH IN VIVO SULPHOCONJUGATION OF PARACETAMOL AND SALICYLAMIDE

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- 1 Human platelet phenolsulphotransferase exists in two functional forms, M and P. In this study the substrate specificity of the two forms has been further delineated by correlating activities in different individuals with various substrates. *m*-Tyramine, noradrenaline, adrenaline, 5-hydroxytryptamine, *p*-hydroxyamphetamine, isoprenaline, salbutamol and l-naphthol were all specific substrates for the M form of the enzyme.
- 2 Paracetamol, a mixed substrate, was predominantly metabolized by the M form. Salicylamide at 5  $\mu$ M was a substrate for the P form but became an M substrate at higher concentration. Phenol itself, a specific substrate for phenolsulphotransferase P at 10  $\mu$ M, also became an M substrate at 1 mM concentration. These substrate specificities were confirmed with the selective inhibitor, dichloronitrophenol.
- 3 In this study, we measured phenolsulphotransferase activity in platelets from 13 individuals selected on the basis of their wide variation in ability to sulphoconjugate paracetamol and salicylamide *in vivo*. There was no significant relationship between the *in vivo* pattern with either drug and the activity of platelet phenolsulphotransferase assayed with paracetamol or salicylamide respectively.

#### Introduction

Sulphate conjugation, catalyzed by phenolsulphotransferase (PST) (EC.2.8.2.1), is an important metabolic pathway, not only for phenolic monoamines and some of their metabolites but also for phenol and some phenolic drugs, including paracetamol and salicylamide (Williams, 1959). Although there is a great deal of inter-individual variation in ability to form both the glucuronide and sulphate conjugates of these drugs, some individuals are relatively unable to form the sulphate conjugate of one or other (Caldwell et al., 1982). One possible reason for this finding is a decrease in PST activity. To investigate the possibility, we have measured the enzyme, using paracetamol and salicylamide as substrates, in platelets from a group of normal

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Human platelet PST (Hart et al., 1979; Anderson & Weinshilboum, 1980; Rein et al., 1981) exists in two functional forms, PST P (specific for 30 µm phenol) and PST M (specific for the monoamines, dopamine and tyramine) (Rein et al., 1982). These two forms differ in substrate specificity, tissue distribution and inhibitor sensitivity, the P form being selectively inhibited 10<sup>-6</sup>M 2,6-dichloro-4-nitrophenol (DCNP) (Rein et al., 1982); they appear to be under separate control (Bonham Carter et al., 1981). To determine which forms of the enzyme metabolize paracetamol and salicylamide, we have examined the effect of DCNP on platelet PST activity using these drugs as substrates. We have confirmed our findings by correlating individual platelet PST activity values from the group of normal volunteers, using paracetamol and salicylamide as substrates, with those

obtained using p-tyramine (for PST M) and phenol (for PST P) as substrates. There is a high degree of correlation between activities in different individuals with different substrates acting at the same active site (Bonham Carter et al., 1981). We have also tested eight other phenolic compounds in this way in order to characterize the substrate specificities of PST M and P further.

The efficiency of the assay procedure was assessed for each of the different substrates used as there is evidence that it can vary considerably with different phenolic acceptors (Foldes & Meek, 1973; Baranczyk-Kuzma et al., 1981).

#### Methods

#### Materials

Materials for PST assay were essentially the same as previously described (Bonham Carter et al., 1981) except for the following substrates: paracetamol (N-acetyl-p-aminophenol),salicylamide, hydroxytryptamine creatinine sulphate complex (5-HT), adrenaline (free base) and noradrenaline (free base), which were all purchased from Sigma Chemical Co., Poole, Dorset, U.K. I-Naphthol was purchased from Hopkin & Williams, Chadwell Heath, Essex, U.K. p-Hydroxyamphetamine hydrobromide was a gift from Smith Kline & French Laboratories Ltd, Welwyn Garden City, Herts, U.K., isoprenaline hydrochloride was a gift from Pharmax Ltd, Bexley, Kent, U.K., salbutamol sulphate was a gift from Allen & Hanburys Ltd. Bethnal Green, London, E.2, U.K. and m-tyramine was a gift from Roche Products Ltd, Welwyn Garden City, Herts, U.K. (Ro 4-4657). DCNP was obtained from Koch-Light Laboratories Ltd, Colnbrook, Bucks, U.K.

# Sample collection and platelet preparation

Venous blood samples (10 ml) were collected from 13 normal volunteers and platelets prepared on the same day, as previously described (Bonham Carter et al., 1981); the platelet pellet was washed with isotonic sucrose (1 ml/10 ml blood), resuspended in phosphate buffer (10  $\mu$ M, pH 7.4; 1 ml/10 ml blood), divided into aliquots as required and stored deep frozen  $(-20^{\circ}\text{C})$ . Care was taken not to allow plasma or platelet suspensions to heat up during centrifugation as PST M is thermolabile above 30°C (Bonham Carter et al., 1981). The amount of glucuronide and sulphate conjugates of paracetamol or salicylamide excreted after administration of 1 g of either of these drugs had previously been estimated in 11 of the subjects for paracetamol and 9 of the subjects for salicylamide (Caldwell et al., 1982).

### PST assay

Platelet PST activities were measured by the same method as previously described (Bonham Carter et al., 1981) based on that of Foldes & Meek (1973). A solution of  $^{35}$ S-3'-phosphoadenosine-5'-phosphosulphate (PAPS) and non-radioactive PAPS in water was employed as sulphate donor (final concentration in incubation mixture,  $0.6~\mu$ M).  $10~\mu$ l (containing 18–35  $\mu$ g protein) of platelet suspension was used with the following substrates: p-tyramine, m-tyramine, noradrenaline, adrenaline, 5-HT, p-hydroxyamphetamine, isoprenaline, salbutamol and salicylamide, and  $20~\mu$ l (containing 35–70  $\mu$ g protein) with the remaining substrates: phenol, paracetamol and salicylamide, at low concentration. Less than 20% of the PAPS was consumed during the assay.

# Efficiency studies

In this method, excess (unreacted) PAPS is removed by 2 successive precipitations with barium hydroxide (200  $\mu$ l, 0.1 M) and zinc sulphate (200  $\mu$ l, 0.1 M). Foldes & Meek (1973) have demonstrated that this procedure removes at least 99.5% of unreacted PAPS in control experiments in the absence of enzyme and we have confirmed this. Any counts above a water blank level that remain in the supernatant after 2 precipitations in an assay are likely to be due to products of the PST enzyme reaction, i.e. sulphate esters. In order to assess the efficiency for different substrates, we carried out assays using the same batch of platelets and a range of phenolic acceptors with 6 tubes for each substrate. In each case, both supernatant and precipitate were counted, in duplicate for each substrate, after 1 precipitation, after 2 precipitations and after 3 precipitations. Blanks containing platelets, but water instead of acceptor substrate, were treated in the same way, and these counts were subtracted from those from the corresponding tubes containing acceptor substrate. In this way, the percentage radioactivity remaining in the third supernatant and that precipitated were calculated giving an indication of the solubility of the sulphate conjugates for each substrate.

# K<sub>m</sub> determinations

Human platelet PST activity was measured using different concentrations of the substrates for which apparent  $K_m$  values were not already known. The range of substrate concentrations used were as follows: for salicylamide 1–180  $\mu$ M; for paracetamol 100–800  $\mu$ M, for m-tyramine, p-hydroxyamphetamine and isoprenaline 1  $\mu$ M – 1 mM. Apparent  $K_m$  values were calculated by the direct linear plot method (Eisenthal & Cornish-Bowden, 1974). Apparent  $K_m$  values for most of the remaining sub-

strates used have previously been reported (Rein et al., 1981; Bonham Carter et al., 1981; R. Sodha, personal communication).

## Inhibitor studies

The effect of the selective inhibitor, DCNP, on platelet PST activities using several acceptor substrates was examined using different concentrations of DCNP as described previously (Rein et al., 1982). PST assay conditions were as described above with the following acceptor substrate concentrations: p-tyramine, 30  $\mu$ m; phenol, 30  $\mu$ m and 1 mm; paracetamol, 850  $\mu$ m and salicylamide, 5  $\mu$ m.

#### Protein estimation

Protein concentrations of platelet suspensions were measured by the method of Lowry *et al.* (1951), using bovine serum albumin as standard.

#### Results

Efficiency of the assay using different substrates

This assay procedure relies on removing excess PAPS by precipitating twice with barium hydroxide and zinc sulphate without precipitating the sulphate esters of the phenolic acceptors, i.e. the products of the reaction. In order to assess the efficiency of the assay with the substrates used, a third precipitation with barium hydroxide and zinc sulphate was carried out as described in the **Methods** section. The recovery in the third supernatant was 90% or more for p-tyramine, m-tyramine, 5-HT, p-hydroxyamphetamine, salbutamol, paracetamol, l-naphthol and phenol, and 80% or more for salicylamide and isoprenaline. However, for the catecholamines, adrenaline and noradrenaline, it was much less, 68% for adrenaline and only 48% for noradrenaline.

# Apparent K<sub>m</sub> values

The apparent  $K_m$  values found were as follows: m-tyramine, 5  $\mu$ M; p-hydroxyamphetamine, 30  $\mu$ M; isoprenaline, 2.5  $\mu$ M; paracetamol, 850  $\mu$ M and salicylamide, 5  $\mu$ M. An apparent  $K_m$  of 7  $\mu$ M has previously been reported for phenol (Rein et~al., 1981). In this study, by using a higher phenol concentration range a second, higher, apparent  $K_m$  of 560  $\mu$ M was found.

# Activity of platelet PST with different substrates

The mean specific activity of platelet PST from the 13 subjects, using the range of acceptor substrates described, is given in Table 1. The concentrations of

acceptor substrates at which the specific activities were measured are also shown in Table 1. They were, in general, used at approximately their apparent  $K_m$ and in some cases at one-tenth, 10 times or one hundred times  $K_m$ , except for 1-naphthol where the  $K_m$  was unknown, and noradrenaline where the activity was not sufficiently greater than blank at 5 μM, presumably because of the precipitation problem described in the efficiency section. Each substrate was used at its determined  $K_m$  value in case it was also a substrate for the other form of the enzyme with a different  $K_m$ , as it is for phenol. The specific activities quoted in Table 1 are those that were actually obtained and have not been corrected for losses occurring during the precipitation procedure which vary from substance to substance. Table 1 also gives the mean activity for each substrate relative to that with ptyramine at 13  $\mu$ M. There was no measurable PST activity using paracetamol at 3 concentrations lower than the  $K_m$  (30, 60 and 90  $\mu$ M).

Correlation of platelet PST activities with different substrates

There was a high degree of correlation between platelet PST activities in the 13 subjects obtained with p-tyramine as substrate (for PST M) and those using m-tyramine, noradrenaline, adrenaline, 5-HT, phydroxyamphetamine, isoprenaline, salbutamol and l-naphthol (noradrenaline, P < 0.01; all remaining substrates, P < 0.001) (Table 1) but no significant correlation of the activities using any of these substrates and those with phenol at 10  $\mu$ M (for PST P) (Table 1). However, although the individual platelet PST activity values using paracetamol, at 850  $\mu$ M and 8.5 mm correlated highly with those using p-tyramine, the correlation coefficients obtained between paracetamol values and those for phenol (at 10  $\mu$ M) were higher than for the other substrates, just reaching significance (P < 0.05) at the lower paracetamol concentration (850  $\mu$ M) (Table 1). Platelet activity values using salicylamide as substrate at 5 and 50  $\mu$ M (but not 500  $\mu$ M) concentration showed a highly significant degree of correlation with 10 μM phenol values (Table 1). However, the reverse was true when the salicylamide data were correlated with activities using 13  $\mu$ M p-tyramine. There was a highly significant correlation with salicylamide at 500  $\mu$ M, one which was just significant at 50  $\mu$ M but no significant correlation at 5  $\mu$ M (Table 1).

With 133  $\mu M$  p-tyramine (approximately 10 times  $K_m$ ), similar correlation coefficients were obtained with the other substrates as with 13  $\mu M$  p-tyramine; m-tyramine at 500  $\mu M$  (approximately one hundred times  $K_m$ ) gave similar correlation coefficients as it did at its lower concentration. However, although the activities with p-tyramine at 1.3  $\mu M$  correlated highly both with p-tyramine at 13  $\mu M$  (Table 1) and all the

**Table 1** Specific activities of human platelet (PST) using different substrates, their activities relative to that using p-tyramine at 13  $\mu$ M and the correlation coefficients of PST activities in 13 individuals with those using p-tyramine or phenol

Substrate	Concentration (μΜ)	Specific activity (nmoles product/ mg protein/10 min) Mean ± s.e. mean	Relative activity (p-tyramine 13 µм = 100)	Pearson correlation coefficient with p-tyramine (13 µM)	Pearson correlation coefficient with phenol (10 µM)
p-Tyramine	13†	$0.20 \pm 0.01$	100		0.14
p-Tyramine	133	$0.49 \pm 0.04$	245	0.96***	0.23
p-Tyramine	1.3	$0.03 \pm 0.002$	15	0.85***	0.50*
m-Tyramine	5†	$0.19 \pm 0.02$	95	0.89***	0.30
Noradrenaline	20	$0.05 \pm 0.01$	25	0.65**	0.20
Adrenaline	3†	$0.09 \pm 0.01$	45	0.85***	0.16
5-HT	160+	$0.09 \pm 0.01$	45	0.88***	0.16
p-Hydroxy-					
amphetamine	30 <sup>+</sup>	$0.14 \pm 0.01$	70	0.87***	0.17
Isoprenaline	2.5†	$0.11 \pm 0.01$	55	0.92***	0.13
Salbutamol	150†	$0.16 \pm 0.01$	80	0.82***	0.31
l-Naphthol	20	$0.19 \pm 0.02$	95	0.94***	-0.01
Paracetamol	850 <sup>†</sup>	$0.07 \pm 0.01$	35	0.85***	0.48*
Paracetamol	8,500	$0.19 \pm 0.01$	95	0.90***	0.46
Salicylamide	5†	$0.04 \pm 0.01$	20	0.33	0.95***
Salicylamide	50	$0.06 \pm 0.01$	30	0.51*	0.82***
Salicylamide	500	$0.15 \pm 0.01$	75	0.88***	0.002
Phenol	10†	$0.05 \pm 0.01$	25	0.14	_
Phenol	1,000	$0.21 \pm 0.01$	105	0.81***	0.05

Significance: \*P < 0.05; \*\*P < 0.01; \*\*\*P < 0.001.

other PST M-type substrates, they also correlated significantly with PST P activities — phenol at  $10~\mu M$  (Table 1) and salicylamide at  $5~\mu M$  (correlation coefficient r=0.68, P<0.01).

#### Inhibition by DCNP

As shown in Figure 1a, PST activity with phenol at 30  $\mu$ M was 98% inhibited in the presence of 10<sup>-6</sup>M DCNP, but with phenol at 1 mM, it was only inhibited by 17%. Activity with p-tyramine was only inhibited by 7% in the presence of 10<sup>-6</sup>M DCNP. PST activity with 5  $\mu$ M salicylamide was 98% inhibited by 10<sup>-6</sup>M DCNP, showing a similar pattern to 30  $\mu$ M phenol; paracetamol sulphation on the other hand, whilst showing greater sensitivity to 10<sup>-6</sup>M DCNP than that of p-tyramine, was still only 43% inhibited (Figure 1b).

Correlation of PST activities with the in vivo sulphate conjugation of paracetamol and salicylamide

The ratios of glucuronide to sulphate conjugates of paracetamol and salicylamide excreted by the group of normal volunteers after ingestion of 1 g of these drugs (Caldwell et al., 1982) are shown in Table 2. There is a substantial degree of interindividual variation in relative conjugating ability, with some evidence, in the case of paracetamol, that the effect derives from the inability of certain individuals to form sulphate conjugates (Caldwell et al., 1980, 1982). These ratios were ranked and correlated with platelet PST activities in the same individuals using these 2 drugs, and p-tyramine and phenol, as substrates. Spearman rank correlation coefficients are shown in Table 3. There was no significant positive correlation between paracetamol excretion data (11 individuals) and PST activities with paracetamol or p-tyramine (for PST M) as substrates. However, with phenol (for PST P) as substrate, there was a positive correlation which just reached statistical significance (P < 0.05, Table 3). There was no significant positive correlation between salicylamide excretion data (9 individuals) and PST activities with salicylamide, p-tyramine or phenol as substrates; in fact, all these correlation coefficients were negative, for reasons which are not immediately apparent.

<sup>†</sup>approximate apparent  $K_m$  value

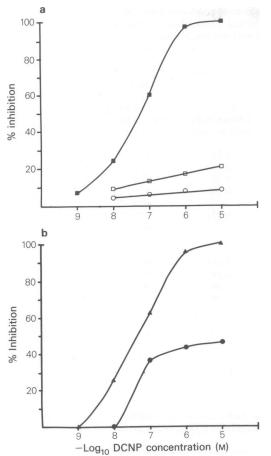


Figure 1 Inhibition of human platelet phenolsulphotransferase by various concentrations of dichloronitrophenol (DCNP) with (a) p-tyramine at 30  $\mu$ M  $\odot$ ; phenol at 30  $\mu$ M  $\odot$  and 1 mM  $\odot$ ; (b) paracetamol at 850  $\mu$ M  $\odot$ ; salicylamide at 5  $\mu$ M  $\Delta$ .

Results are expressed relative to values obtained in the absence of inhibitor.

## Discussion

The results obtained in the recovery studies demonstrate that it is possible to use a wide range of neutral and basic phenols for this PST assay and obtain reasonably accurate specific activity values. Recovery values for the catecholamines, adrenaline and noradrenaline, were much poorer; the specific activities quoted have not been corrected and are therefore low. In their original paper, Foldes & Meek (1973) examined the recovery for a range of substrates by preparing the sulphate esters, dissolving them in phosphate buffer and then precipitating with barium hydroxide and zinc sulphate as in the assay procedure. They claimed an 82–100% range of

recoveries for neutral and basic phenols, including noradrenaline. These values are somewhat higher than we have observed; however, we carried out our recovery experiments in the presence of barium acetate, which is used to stop the PST reaction in the assay. It is conceivable that the poor recoveries are due to low solubility of the barium salts of corresponding sulphate esters, and if this is so, then the barium acetate will provide an additional source of barium to form such salts. The fact that recoveries for acidic substrates in this assay are very poor (Foldes & Meek, 1973; Baranczyk-Kuzma et al., 1981) may also be for this reason.

When different substrates are employed, variable recoveries in the assay will affect absolute specific activities but should not affect the correlation of individual values with those using either the PST M substrate (p-tyramine at 13 µm) or the PST P substrate (phenol at 10 µm). Rein et al. (1982) have shown that dopamine and p-tyramine are substrates for PST M, whilst 30 µm phenol is a substrate for PST P. using the selective inhibitor DCNP. We have previously confirmed that both dopamine and 4hydroxy-3-methoxyphenylglycol (HMPG) are substrates for PST M by correlation of individual PST activities. That study also demonstrated that the 2 enzymes are under separate control (Bonham Carter et al., 1981). Anderson et al. (1981) have also shown that individual PST activities with HMPG as substrate correlate significantly with values obtained using dopamine and p-tyramine. Using the same correlation procedure for the data obtained in this study, we have shown that m-tyramine, noradrenaline, p-hydroxyamphetamine, adrenaline. 5-HT. isoprenaline, salbutamol and l-naphthol are all substrates, for the M form of the enzyme only, at the concentrations we have used. However, paracetamol appears to be a substrate for both forms of the enzyme, although it shows more M than P type activity at the concentrations used, a conclusion confirmed by experiments with the selective inhibitor DCNP. Anderson et al. (1981) have shown similarly that individual platelet PST activity values, using 5-HT and paracetamol, correlate significantly with those for HMPG, indicating that both are substrates for PST M; they have not examined the relationship with anv PST P substrate.

We have shown here that salicylamide is a PST P substrate at low concentration, becoming an M substrate at high concentration. Presumably the relatively low PST P activity is swamped at high concentration, because the PST M activity is so much greater. It is interesting that phenol itself becomes a substrate, again apparently exclusively for PST M at high concentration; this was confirmed by using DCNP and also by the second much higher  $K_m$  obtained for phenol sulphation. This finding may explain why Mulder & Scholtens (1977) found that

**Table 2** In vivo excretion of conjugated paracetamol and salicylamide after administration of 1 g of either drug to normal volunteers. The results are expressed as the ratio of glucuronide to sulphate conjugate excreted 0–8 h after taking the drug

	Para	Paracetamol		Salicylamide	
Subject	Glucuronide/ sulphate ratio	Ranked data (1 = highest sulphate con- jugate excretor)	Glucuronide/ sulphate ratio	Ranked data (1 = highest sulphate con- jugate excretor)	
1	2.54	10	0.55	2	
2	_		1.20	6	
2 3	_		0.72	3.5	
4	0.74	5	0.50	1	
5	2.29	9			
6	0.71	4	0.72	3.5	
7	0.07	1	_	_	
8	0.14	2		_	
9	1.4	8	_		
10	0.56	3	2.07	9	
11	4.28	11	0.93	5	
12	1.1	7	1.89	8	
13	0.82	6	1.32	7	

**Table 3** Correlation coefficients (Spearman rank) for *in vivo* sulphate conjugation of paracetamol or salicylamide with platelet PST activities in the same individuals using different substrates

PST substrate (concentration)	Paracetamol (in vivo sulphoconjugation) (n = 11)	Salicylamide (in vivo sulphoconjugation) (n = 9)
Paracetamol (850 μM)	$r_{s} = 0.17$	
Paracetamol (8,500 μM)	$r_{s} = 0.12$	_
Salicylamide (5 μM)	_	$r_{s} = -0.27$
Salicylamide (50 µM)		$r_{\rm s} = -0.27$
Salicylamide (500 μM)	_	$r_{\rm s}=-0.61$
p-Tyramine (13 μM)	$r_{s} = 0.10$	$r_s = -0.61$
Phenol (10 μM)	$r_s = 0.54*$	$r_{s} = -0.20$

Significance: \*P < 0.05

sulphation of this substrate was not substantially inhibited by  $10^{-6}$ M DCNP; they used phenol at 1 mm concentration. It is also noteworthy that p-tyramine sulphation at very low concentration  $(1.3 \, \mu\text{M})$  showed a significant correlation with PST P activity, both with phenol and salicylamide (at  $5 \, \mu\text{M}$ ) as substrates, although the degree of correlation with M substrates is considerably greater. It is possible that other substrates of human platelet PST are substrates for both forms but that M activity has dominated at the substrate concentrations used for investigation.

It appears from this study that the consistently low excretion of the sulphate conjugates of paracetamol or salicylamide by some individuals in this group of volunteers (Caldwell et al., 1980, 1982) cannot be explained by a generalized deficiency of PST. Most sulphate conjugation of orally administered substances occurs in the gut and since there are similarities between gut and platelet enzymes (Anderson & Weinshilboum, 1980; Rein et al., 1981; Anderson et al., 1981), the platelet seems to be a reasonable model to use for a study of this type where gut biopsies are not readily available. Whether the significant correlation between paracetamol sulphate excretion and PST P activity is a chance finding needs further investigation. Paracetamol is, to some extent, a

substrate for PST P; however, at 850  $\mu$ M and 8.5 mM it is predominantly a substrate for PST M and at the lower concentration of 90  $\mu$ M there was no measurable PST activity at all.

There are other possible explanations for some individuals' relative inability to form sulphate conjugates of orally administered compounds, e.g. reduced drug availability at the conjugation site or modification of in vivo enzyme activity by the presence of endogenous inhibitors (Anderson & Weinshilboum, 1979). These may also be the reasons why patients suffering from depressive illness who excrete significantly low levels of tyramine-Osulphate after oral tyramine administration (Sandler et al., 1975; Bonham Carter et al., 1978), have normal platelet PST activity (Bonham Carter et al., 1981). It interesting. however, that poor sulphate conjugators of paracetamol are not necessarily poor sulphate conjugators of salicyclamide and vice versa (Caldwell et al., 1982). Whether either of these groups are also poor sulphate conjugators of tyramine or another phenolic compound must be determined by future investigation.

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