# **Supplemental Data**

# **Architecture of the 99 bp DNA-Six-Protein Regulatory Complex of the** λ *att* **Site**

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### **FRET Calculations**

The determinations of the efficiency of energy transfer  $E$  and  $R_0$  and FRET calculations were performed as described (Radman-Livaja et al., 2005). The efficiency of energy transfer *E* was determined from the extent of donor fluorescence quenching in doubly labeled (donor and acceptor) P-arm bound with six proteins compared with donor-onlylabeled P-arm bound with six proteins. Donor fluorescence quenching due to FRET is described by the following equation:

$$
\frac{I_j^D - I_i^{DA}}{I_j^D} = 1 - \frac{[complex]_i^{DA}}{[complex]_j^D} (1 - rE),
$$
 [1]

where  $I_i^{DA}$  is the fluorescence intensity of a doubly labeled P-arm-protein complex, *i.*  $I_j^D$  is the florescence intensity of a donor-only labeled P-arm- protein complex.  $[complex]$ <sup> $D_A$ </sup> and  $[complex]$ <sup> $D$ </sup> are the total amounts of doubly and singly labeled P-arm-protein complexes in bands *i* and *j*, respectively, and *r* is the efficiency of DNA labeling with the acceptor.

Eq.1 can be written as follows:

$$
I_i^{DA} \frac{[complex]_j^D}{[complex]_i^{DA}} = I_j^D (1 - rE)
$$
 [2]

[complex] values were obtained from a Phosphorimager scan. *E* was obtained from the slope of the line formed by plotting  $I_i^{DA}$  [complex] $\int_i^D$  /[complex] $\int_i^{DA}$ *i D j*  $I_i^{DA}$ [*complex*] $_i^D$  /[*complex*] $_i^{DA}$  against  $I_j^D$ , the slope being equal to (1-*rE*). The value *r* was calculated from the absorbance spectrum of each acceptor-labeled oligonucleotide by using  $(A^R(560)\varepsilon^{ssDNA}(260)/A^{ssDNA}$  $(260)\varepsilon^R(560)$ , with  $\varepsilon^R(560 \text{ nm})=91.000 \text{ M}^1 \text{ cm}^1$ .

The distance R between donor and acceptor is calculated from the measured *E* value by using the Förster equation for FRET:

$$
R = R_0 \sqrt[6]{\left(\frac{1}{E} - 1\right)},
$$

where  $R_0$  is the Förster radius determined for each donor-acceptor pair.

#### *R0* **Determination**

*R0* was measured by the method of Wu and Brand (Wu and Brand, 1994) for each donoracceptor pair by using the following equations:  $R_0 = 0.211 \sqrt[6]{\kappa^2 n^{-4} \phi^D J(\lambda)}$  in Å.  $\kappa^2$  is the transition dipoles orientation factor estimated to be 2/3 for fluorophores that have a fast rotational diffusion rate in solution (Dale et al., 1979; Dale et al., 1980). *n* is the refractive index of the medium (1.4 for DNA aqueous solutions).  $\phi^D$  is the emission quantum yield of the donor.  $J(\lambda)$  [in M<sup>-1</sup>.cm<sup>-1</sup>.(nm)<sup>4</sup>] is the overlap integral between the donor emission spectrum and the acceptor absorbance spectrum defined by the equation

$$
J(\lambda) = \frac{\sum_{\Delta\lambda} [F_D(\lambda)\varepsilon_A(\lambda)\lambda^4\Delta\lambda]}{\sum_{\Delta\lambda} [F_D(\lambda)\Delta\lambda]}
$$
, where  $F_D(\lambda)$  is the donor emission spectrum and  $\varepsilon_A(\lambda)$  is

the spectrum for acceptor extinction coefficients.

 $\phi^D$  for for each donor-labeled P-arm substrate was calculated as described in (Lakpwicz,

1999) by using the equation  $\phi^D = \phi^R \frac{I}{I^R} \frac{\partial D}{\partial D^D}$ *R R*  $D = \mathcal{A}^R$ <sup>D</sup> *OD OD I*  $\phi^D = \phi^R \frac{I^D}{I^R} \frac{OD^R}{OD^R}$ , where  $\phi^R$  is the fluorescence quantum yield of a fluorescence standard (Fluka, Sigma-Aldrich);  $I^D$  and  $I^R$  are the fluorescence emission intensities of Bodipy-Fl and the standard, respectively; and *OD*<sup>*D*</sup> and *OD*<sup>*R*</sup> are absorbance values of Bodipy-Fl at 488 nm and the standard at 496 nm, respectively.  $\phi^R =$  $0.95\pm 0.03$  at  $\lambda$ ex = 496 nm, when fluorescein is in 0.1 M NaOH (Brannon and Magde, 1978). The Bodipy-Fl-labeled P-arm substrates  $(0.8 \mu M)$  or the standard  $(0.4 \mu M)$  were dissolved either in buffer A (see *Methods*) or 0.1 M NaOH, respectively. A Beckman DU 520 UV/VIS spectrometer was used to collect the ODs. Fluorescence spectra of the same solutions were recorded on a Fluoromax 2 instrument (Instruments S.A.) by using a slit width of 5 nm. Excitation with Xenon lamp was at 488 or 496 nm for Bodipy-Fl or the standard, respectively. Emission spectra were collected over a wavelength range of 500- 600 nm and corrected for buffer signals and lamp fluctuations. The emission maximum for Bodipy-Fl and the fluorescein standard was at 520 nm when excited at 488 and 496 nm, respectively. Fluorescence intensities were determined by integrating the curve  $\pm 20$ nm around the emission peak.

The emission spectra  $F_D(\lambda)$  of Bodipy-Fl-labeled P-arm substrates for  $J(\lambda)$  calculations were recorded as described in the previous paragraph.  $\varepsilon_A(\lambda)$  for acceptor (tetramethylrhodamine, TAMRA) labeled P-arm substrates was obtained from absorbance spectra of 0.50 µM substrate solutions (10 mM Tris-HCl pH 7.5, 50 nM NaCl, 1 mM EDTA) collected in the wavelength range of 190-800 nm on a HP8452A UV/VIS spectrophotometer. Absorbance spectra  $A(\lambda)$  were then converted to  $\varepsilon_A(\lambda)$ 

spectra by using the "50 µg/ml rule" (i.e.,  $\varepsilon_A(\lambda) = \frac{A(\lambda)}{A(260nm)} \varepsilon_{DNA}(260nm)$ )  $\varepsilon_A(\lambda) = \frac{A(\lambda)}{A(260nm)} \varepsilon_{DNA}(260nm)$ , for  $\lambda = 500$ 

to 600 nm and 
$$
\varepsilon_{DNA}(260nm) = \frac{50*10^{-3} g/l}{Mr(DNA)*1cm}
$$
, where *Mr(DNA)* is the molecular weight

of the DNA substrate in  $g/mol$ .  $J(\lambda)$  calculations and spectral data manipulations were done by using Microsoft Excel spreadsheets.

#### **Anisotropy Measurements**

Steady-state anisotropies (*r*) were measured for all Bodipy-Fl-labeled or acceptor (TAMRA)-labeled P-arms both unbound and in the context of P-arm-six protein complex. Samples contained the labeled substrates (50 nM) in binding buffer (see above) with or without six proteins. *r* was calculated from  $_{vv}$  + 201 $_{vH}$  $V$ *V*  $V$  *VH*  $I_{yy}$  + 2*GI*  $r = \frac{I_{VV} - GI}{I}$  $+2$  $=\frac{I_{VV}-GI_{VH}}{I}$ . *I* is the fluorescence

intensity of the sample. The first and second letters in the subscript describe the positions of the excitation and emission polarizers, respectively (*V*, vertical; *H*, horizontal). *G* is the correction factor for differences in sensitivities of the detection system for vertically and horizontally polarized light and is equal to  $I_{HV}/I_{HH}$  (Lakowicz, 1999). Fluorescence intensities were measured on a Fluoromax 2 fluoimeter with autopolarizers (Instrument S.A.) using excitation wavelengths of 488 and 560 nm for Bodipy-Fl and TAMRA, respectively. Emission was recorded over 500-540 nm and 570-610 nm ranges for Bodipy-Fl and TAMRA, respectively. *r* values ranged 0.16-0.20 for unbound Bodipy-Fl P-arms. Proteins addition increased *r* only slightly, in the range of 0.20-0.26. *r* values of acceptor-labeled P-arms were 0.18-0.24 and 0.22-0.25 with and without proteins, respectively. While the measured anisotropies suggest somewhat constrained rotational

motion of the dyes, we assumed they were low enough to justify the use of 2/3 for  $\kappa^2$ , as

this value has been widely accepted for fluorophores with our anisotropies ranges

(Lorenz et al., 1999; Mekler et al., 2002; Rasnik et al., 2004) for review see(Lakowicz,

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 $*$  X1 and X2 are underlined.

			$J(\lambda)^\# (\text{M}^{-1}$ cm-	
Donor	$\Phi$ <sup>D.</sup>	Acceptor	$1.nm4$ /10 <sup>15</sup>	$R_0^{\%}(\AA)$
fP2	0.81	fX1	2.20	54.5
fP2	0.81	fX1.5	2.41	55.9
fP2	0.81	fH2A	2.97	57.9
fP2	0.81	fH <sub>2</sub> B	2.79	57.3
fP2	0.81	fH <sub>2</sub> C	3.08	58.3
fP2	0.81	fH <sub>2</sub> D	2.65	56.8
fX1	0.77	fX1.5	2.49	55.6
fX1	0.77	fH <sub>2</sub> A	3.48	58.9
fX1	0.77	fH <sub>2</sub> B	2.54	55.9
fX1	0.77	fH <sub>2</sub> C	2.34	55.2
fX1	0.77	fH <sub>2</sub> D	2.74	56.7
fF	0.82	fX1	2.20	55.2
fF	0.82	fX1.5	1.69	52.8
$\operatorname{f\!}$	0.82	fH <sub>2</sub> A	1.95	54.1
$\operatorname{f\!}$	0.82	fH2B	1.63	52.5
$\operatorname{f\!}$	0.82	fH <sub>2C</sub>	2.40	56.0
$\operatorname{f\!}$	0.82	fH <sub>2</sub> D	2.66	57.0
fH <sub>2</sub> A	0.86	fX1.5	2.05	55.0
fH <sub>2</sub> A	0.86	fH2B	2.28	56.0
fH <sub>2</sub> A	0.86	fH <sub>2</sub> C	1.68	53.2
fH <sub>2</sub> A	0.86	fH <sub>2</sub> D	1.82	53.9
fH <sub>2</sub> B	0.74	fX1.5	1.82	52.5
fH <sub>2</sub> B	0.74	fH <sub>2</sub> C	2.62	55.8
fH <sub>2</sub> B	0.74	fH <sub>2</sub> D	2.28	54.6
fH <sub>2C</sub>	0.70	fX1.5	1.76	51.8
fH <sub>2</sub> C	0.70	fH2D	2.13	53.5
fH <sub>2</sub> D	0.88	fX1.5	2.28	56.2

Table S2.  $R_0$  Measurements

 $*$   $\Phi$ <sup>D</sup> is the fluorescence quantum yield of the donor.

$$
{}^*J(\lambda) = \frac{\sum_{\Delta\lambda} [F_{D}(\lambda)\varepsilon_{A}(\lambda)\lambda^4 \Delta \lambda]}{\sum_{\Delta\lambda} [F_{D}(\lambda)\Delta \lambda]}
$$
 Where  $F_{D}(\lambda)$  is the fluorescence intensity of the donor and

 $\varepsilon_A(\lambda)$  is the extinction coefficient of the acceptor at wavelength  $\lambda$ .

 ${}^{96}R_0 = 0.211 \sqrt[6]{\kappa^2 n^{-4} \phi^D J(\lambda)}$ ,  $R_0$  is the Förster distance between donor and acceptor.  $\kappa^2$ (transition dipoles orientation factor) =  $2/3$  and *n* (refractive index of the medium) = 1.4.