## Hydrogen-Mediated Mannose Uptake in Azotobacter vinelandii<sup>†</sup>

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Azotobacter vinelandii can grow mixotrophically with  $H_2$  plus mannose under  $N_2$ -fixing conditions (T. Y. Wong and R. J. Maier, J. Bacteriol. 163:528–533, 1985). Mixotrophically grown cultures incubated in  $H_2$  transported mannose with a  $V_{max}$  fourfold greater than that observed for cultures incubated in argon, but  $H_2$  did not change the apparent  $K_m$  for mannose. Respiratory inhibitors, such as potassium cyanide, hydroxylamine, and *p*-chloromercuribenzoic acid, as well as the proton conductor carbonyl cyanide *m*-chlorophenyl-hydrazone inhibited mannose uptake. We suggest that one of the roles of  $H_2$  in mixotrophic metabolism is to supply energy that facilitates mannose transport.

Azotobacter vinelandii can grow mixotrophically, with H<sub>2</sub> as the inorganic energy source and mannose as the carbon source (16). Both the growth rate and the cell yield are dependent on the amounts of both substrates, H<sub>2</sub> and mannose. Hydrogen does not significantly stimulate the growth of A. vinelandii with other carbon sources, and the mannose-dependent mixotrophic growth occurs only when the cultures are fixing nitrogen (16). In ammonium-containing medium, mannose-mediated growth of A. vinelandii is not dependent on added  $H_2$  (16). In Azotobacter chroococcum, H<sub>2</sub> oxidation via hydrogenase is also beneficial for the growth of the organism under N<sub>2</sub>-fixing but not NH<sub>4</sub><sup>+</sup>utilizing conditions (1). It has been proposed that under C-limiting conditions, hydrogenase is crucial for the initiation of diazotrophic growth (1). Hydrogen uptake is known to be able to support oxidative phosphorylation in A. vinelandii (8), and H<sub>2</sub> oxidation results in electron flow ultimately to  $O_2$  (8, 10) through a cytochrome-dependent respiratory chain (8, 15).

<sup>14</sup>C-sugar uptake experiments on mixotrophically grown cultures (16) suggested that  $H_2$  plays a role in mannose transport; the mannose uptake rate when cells were incubated with  $H_2$  was two- to threefold greater than that when cells were incubated with argon. In this report we demonstrate that  $H_2$  has the ability to provide energy for mannose transport and that  $H_2$  provided to cultures causes a significant increase in the  $V_{\text{max}}$  for mannose transport in A. vinelandii.

Mannose uptake experiments were performed similarly to those previously described (16). All experiments used A. vinelandii CA (provided by Paul Bishop, North Carolina State University, Raleigh), which is the same as strain OP (4). Strain OP was utilized previously for H<sub>2</sub>-dependent mixotrophic-growth experiments (16). Cells were grown in Burk medium (6) with H<sub>2</sub> and 2% mannose as described previously (16), harvested at a concentration of  $4 \times 10^8$  to  $6 \times 10^8$  cells per ml, and then washed three times with 50 mM potassium phosphate buffer (pH 7.5). The cells were suspended in carbon-free Burk medium to about  $1.5 \times 10^8$  cells per ml, and the culture was then incubated at 30°C for approximately 2 h with gentle stirring to deplete the endogenous substrate (16). Samples (2.5 ml) were removed into 70-ml serum vials. The vials were stoppered tightly, and the appropriate gas (H<sub>2</sub> or Ar) was added. At 1 min after addition of 5 cm<sup>3</sup> of H<sub>2</sub> or argon, [<sup>14</sup>C]mannose was injected to initiate the 1-min assay. The [<sup>14</sup>C]mannose (D-[U-<sup>14</sup>C]man nose) (Amersham Corp., Arlington Heights, Ill.) had a specific activity of 280 mCi/mmol. Unless otherwise indicated, each reaction vial received 0.5 mM mannose at a specific activity of 0.5 µCi/µmol. The vials were placed on a rotary shaker at 300 cycles per min during the 1-min assay. For the kinetic experiments in which the substrate concentration was varied, the lower concentrations tested (0.05,0.10, 0.20, 0.30, and 0.40 mM) contained 2.0 µCi/µmol, and the other concentrations (0.6, 0.8, 1.0, 1.5, and 2.0 mM) had  $0.5 \ \mu Ci/\mu mol$  in the reaction vial. After 1 min at 30°C with shaking, a 2.0-ml sample was removed and four 0.5-ml portions from this sample were rapidly placed on a filter unit (12-sample filtering manifold; Millipore Corp., Bedford, Mass.) under vacuum. The filters (pore size, 0.22 µm; GSWP 02400; Millipore) were immediately washed with two 10-ml volumes of ice-cold 50 mM potassium phosphate (pH 7.5). Filters were allowed to air dry and placed into 6-ml scintillation vials, and then 5 ml of Aquasol-2 (New England Nuclear Corp., Boston, Mass.) was added and radioactivity was counted in an LKB model 1218 scintillation spectrometer. Background counts from filters receiving the  $[^{14}C]$ mannose solution but without cells were subtracted from all experimental values, and control experiments with boiled cells demonstrated negligible incorporated radioactivity. The viable cell number was determined by plate counts of appropriate dilutions, and protein amounts were determined from a standard curve of A. vinelandii viable cell number versus protein concentration (15). For the inhibitor studies the amount of mannose taken up in 1 min by cells with Ar plus the inhibitor was subtracted from the rate with H<sub>2</sub> plus the inhibitor. The difference is called the  $H_2$ -dependent rate (Table 1). The [<sup>14</sup>C]mannose was added 1 min after the injection of H<sub>2</sub> or Ar, and the samples were removed 1 min after the addition of mannose. For the study of mannose uptake with time (Fig. 1), 10 ml of cells at  $2 \times 10^8$  viable cells per ml was added to 250-ml prescription bottles. The bottles were sealed with a serum stopper, the appropriate gas (20  $cm^3$  of H<sub>2</sub> or Ar) was added, and then 0.5 mM mannose at 1.0  $\mu$ Ci/ $\mu$ mol was added to initiate the assay.

Previous results indicated that cultures incubated with H<sub>2</sub>

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Inhibitor	Concn (mM)	Uptake rate (% of uninhibited)	
		Oxygen <sup>b</sup>	Mannose <sup>c</sup>
KCN	0.02	81 ± 5	88 ± 18
	0.10	$72 \pm 5$	46 ± 7
	0.70	$5 \pm 0.3$	$1.5 \pm 0.3$
Hydroxylamine	0.10	$133 \pm 7$	$104 \pm 4$
	1.0	$105 \pm 7$	$104 \pm 4$
	5.0	$55 \pm 13$	84 ± 6
	20.0	$42 \pm 6$	$43 \pm 9$
Antimycin A	0.05	$104 \pm 6$	$107 \pm 3$
	0.5	95 ± 8	99 ± 4
	10.0	$51 \pm 5$	$50 \pm 10$
pCMPS	0.05	84 ± 9	$110 \pm 13$
	0.5	$70 \pm 13$	81 ± 4
	2.0	$38 \pm 5$	$67 \pm 5$
СССР	0.01	$89 \pm 10$	75 ± 2
	0.1	$113 \pm 17$	$41 \pm 3$
HQNO	0.5	97 ± 6	$93 \pm 6$
	5.0	$79 \pm 5$	$61 \pm 9$
рСМВ	0.01	45 ± 9	$80 \pm 5$
	0.05	$28 \pm 5$	$33 \pm 3$
	0.5	$6 \pm 3$	$16 \pm 4$
Chlorpromazine	0.1	84 ± 7	ND
	1.0	89 ± 5	ND
	5.0	79 ± 10	ND

TABLE 1. Effect of respiratory inhibitors on  $H_2$ -mediated  $O_2$  uptake and mannose uptake rates<sup>*a*</sup>

<sup>a</sup> Cultures were grown with H<sub>2</sub> and mannose, the cells were harvested by centrifugation, and the pellet was washed three times and suspended in carbon-free Burk medium (see text). After depletion of endogenous substrate, samples were incubated with the indicated inhibitor for 1 h, and then O<sub>2</sub> or mannose uptake in the presence of H<sub>2</sub> was determined (see text). [<sup>14</sup>C]mannose (0.5 mM at 0.5  $\mu$ Ci/ $\mu$ mol) was added 1 min after the addition of Ar or H<sub>2</sub> to initiate the 1-min mannose uptake assay. Abbreviations: CCCP, carbonyl cyanide *m*-chlorophenylhydrazone; HQNO, 2-*n*-heptyl-4-hydroxy-quinoline-*N*-oxide; pCMB, *p*-chloromercuribenzoic acid; pCMPS, *p*-chloromerc

<sup>b</sup> Values are means  $\pm$  standard deviations for five determinations; 100% activity (uninhibited rate) was 26.5  $\pm$  2.6 nmol of O<sub>2</sub> per min per 10<sup>8</sup> cells.

<sup>c</sup> Values are means  $\pm$  standard deviations for four samples; 100% activity (uninhibited rate) was  $0.15 \pm 0.02$  nmol of mannose per min per  $10^8$  cells. The 100% activity value is the mean  $\pm$  standard deviation for eight filtered samples.

transport mannose at a rate about twofold greater than that of cultures incubated with argon (16). In this study, we repeated those experiments and also performed similar ones with a larger gas phase during the 1-min mannose uptake assay (see above). A typical mannose uptake experiment over a 2-min period is shown in Fig. 1. It is clear that the culture with H<sub>2</sub> was better able to transport mannose than the argon-supplied culture was. This difference is clear even in the first 30 s of assay (Fig. 1). Exchange experiments with cultures that were provided with cold (nonradioactive) mannose after they took up radioactive mannose indicated that most of the mannose taken up in 1 min was not exchangeable with cold mannose. Therefore, it was evident that the mannose taken up from the medium was either tightly associated with cellular components or metabolized, rather than remaining loosely bound to the cell surface. By use of large vials (70 ml) and small culture volumes (2.5 ml) with rapid shaking during the assay (conditions used for all the



FIG. 1. Mannose uptake in the presence of  $H_2$  (O) or Ar ( $\bullet$ ). Cells grown in Burk medium with mannose plus  $H_2$  were harvested and treated as described in the text, and 10 ml of cells was added to a 250-ml prescription bottle. The bottle was sealed with a serum stopper, the appropriate gas (20 cm<sup>3</sup> of  $H_2$  or Ar) was added, and then a [<sup>14</sup>C]mannose solution was injected to initiate the assay (see text). Samples were removed from the sealed bottle by use of an N<sub>2</sub>-sparged syringe. Three 0.5-ml samples were filtered for each indicated time point.

1-min assays), we observed up to a fivefold greater amount of mannose uptake in 1 min by cultures incubated in  $H_2$ rather than in Ar (data not shown). The above-described conditions were designed to maximize the respiratory ability of the cells by ensuring the availability of ample dissolved  $O_2$ to the cells in the sealed vial.

We determined the mannose uptake rate in 1 min at various substrate concentrations to learn more about the nature of H<sub>2</sub>-supported mannose uptake. A plot of uptake versus substrate concentration (Fig. 2) revealed that the system showed Michaelis-Menten kinetics. A double-reciprocal plot (Fig. 2) of the same data indicated that the apparent  $K_m$  for mannose of the cells incubated in H<sub>2</sub> was similar to that of Ar-incubated cells. In that experiment both cultures had a  $K_m$  of 0.30 to 0.36 mM. However, the  $V_{max}$  was significantly different, i.e., 0.6 nmol/min per mg of protein for the cultures in H<sub>2</sub>. Other experiments gave similar results, with the  $V_{max}$  of the cultures in H<sub>2</sub> being three to five times greater than that of the cultures in Ar.

Inhibitor studies were done to examine the possibility that  $H_2$  supplied energy for mannose uptake. Inhibitors that have previously been shown to be effective at inhibiting *A. vinelandii* respiratory activity (3, 9) were used. Initially, inhibitor concentrations that significantly inhibited  $O_2$  uptake in the presence of  $H_2$  were found (Table 1). However, some of the inhibitors (such as chlorpromazine and 2-*n*-heptyl-4-hydroxyquinoline-*N*-oxide) did not significantly inhibit respiratory activity and probably did not penetrate the cell wall very effectively. The inhibitor concentrations that were effective at inhibiting  $H_2$ -dependent  $O_2$  uptake were then tested for their effect on  $H_2$ -mediated mannose uptake. At the concentrations that were effective at inhibiting respiratory inhibitors also significantly inhibited mannose uptake (Table 1). The best inhibitors of



I/S (mM Mannose)

FIG. 2. Mannose uptake rate with various substrate concentrations in the presence of  $H_2(\bigcirc)$  or Ar ( $\bullet$ ). The mannose uptake in 1 min was determined at 10 different mannose concentrations as described in the text. (A) Plot of uptake rate versus substrate concentration. (B) Double-reciprocal plot with 8 of the 10 substrate concentrations used; the two lowest substrate concentrations (0.05 and 0.1 mM) were not plotted.

respiratory activity, KCN and *p*-chloromercuribenzoic acid, were also the most effective at inhibiting H<sub>2</sub>-mediated mannose uptake. The uncoupler carbonyl cyanide *m*-chlorophenylhydrazone, which may have stimulated O<sub>2</sub> uptake slightly, also inhibited H<sub>2</sub>-dependent mannose uptake. The results indicate that energy provided by H<sub>2</sub> oxidation can support mannose uptake. A glucose transport system in *A. vinelandii* membranes that depends primarily on malate oxidation for energy has been described (2, 3). The glucose uptake was also inhibited by respiratory inhibitors (2, 3).

The roles of  $H_2$ -oxidizing hydrogenases in bacterial physiology are not well understood at the molecular level (5). The oxidation of  $H_2$  certainly has the potential to produce a proton gradient or a membrane potential that allows energy transduction (11). Useful work, such as transport, is thought to be coupled to the energetics of the bacterial membrane by several different mechanisms (7); the molecular nature of  $H_2$ - mediated mannose transport in A. vinelandii cannot yet be predicted.

H<sub>2</sub> oxidation in A. vinelandii can be coupled to oxidative phosphorylation (8), and  $H_2$  oxidation is also believed to be involved in enhancing respiratory protection of nitrogenase (13). Our results indicate an additional role for  $H_2$  oxidation in sugar (mannose) transport, and this must be one role for  $H_2$  in mixotrophic metabolism. It is possible that  $H_2$  oxidation plays such a role in other  $H_2$ -metabolizing bacteria. For example, Pankhania et al. (12) observed slow growth of lactate-limited Desulfovibrio vulgaris in a gas phase of  $N_2$ -CO<sub>2</sub> but rapid growth when the gas phase was changed to  $H_2$ -CO<sub>2</sub>. However, on the basis of other experiments they suggested that the effect of  $H_2$  was not to serve as an energy source for growth but perhaps had some other role in facilitating lactate-dependent growth. In another bacterium, H<sub>2</sub> oxidation confers increased nickel transport ability: constitutive strains of Bradyrhizobium japonicum exhibiting hydrogenase activity had a  $V_{max}$  for nickel that was more than fourfold greater than that of the wild type not expressing hydrogenase (14). Other roles for  $H_2$  in facilitating transport processes in microorganisms seem possible. The use of membrane vesicles of A. vinelandii that have the capacity to transport mannose in a H<sub>2</sub>-dependent manner would be most useful for further studies of the mechanisms involved.

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