

Table 3. Volatile headspace constituents from *Agave palmeri* and *Datura wrightii* identified by GC-MS

Peak no. Fig. 2	Ret. t. (min)	Compound	<i>Agave palmeri</i> (%)	<i>Datura wrightii</i> (%)	MW	Molecular formula	notes	Structure class
	2.77	Dimethyl disulfide	<1		94	C ₂ H ₆ S ₂	b, d, e	Disulfide
	3.52	2-hexanone		<1	100	C ₆ H ₁₂ O	d, e	Ketone
7	5.00	Ethyl isovalerate	3.4		130	C ₇ H ₁₄ O ₂	b, d, e	Ester
8	6.48	Xylene	9.7		106	C ₈ H ₁₀	d, e, i	Aromatic
	7.70	Unk. monoterpene	<1		136	C ₁₀ H ₁₆		Terpene
	7.88	A-thujene	<1		136	C ₁₀ H ₁₆	d, j	Terpene
9	8.23	A-pinene	16.3		136	C ₁₀ H ₁₆	b, d, e	Terpene
	8.45	Ethyl tiglate	<1		128	C ₇ H ₁₂ O ₂	b, d, e	Ester
10	8.82	Camphene	11.0		136	C ₁₀ H ₁₆	b, d, e	Terpene
	9.22	Benzaldehyde	<1	<1	106	C ₇ H ₆ O	c, d, e, f	Aromatic
	9.38	Ethyl hexanoate	<1		144	C ₈ H ₁₆ O ₂	d	Ester
	9.57	Sabinene	<1	<1	136	C ₁₀ H ₁₆	b, c, d, e, j	Terpene
11	9.73	β-pinene	14.8		136	C ₁₀ H ₁₆	b, d, e	Terpene
1	10.07	Beta-myrcene	<1	2.5	136	C ₁₀ H ₁₆	b, c, d, e	Terpene
12	10.60	Ethyl hexenoate	4.7		142	C ₈ H ₁₄ O ₂	d	Ester
	10.82	Methyl sorbate	<1		126	C ₇ H ₁₀ O ₂	b, d, f	Ester
	11.03	Cymene	<1		134	C ₁₀ H ₁₄	b, d, e	Terpene
13	11.18	Limonene	12.7	<1	136	C ₁₀ H ₁₆	b, d, e	Terpene
	11.21	Cineol		<1	154	C ₁₀ H ₁₈ O	c, d	Terpene
2	11.29	Benzyl alcohol		9.5	109	C ₇ H ₈ O	c, d, e	Aromatic
	11.57	Ethyl isohexenoate	<1		142	C ₈ H ₁₄ O ₂	d	Ester
3	11.82	Beta-ocimene	<1	32.4	136	C ₁₀ H ₁₆	c, e, d, j	Terpene
14	12.02	Ethyl sorbate ^k	2.0		140	C ₈ H ₁₂ O ₂	b, d, f, h	Ester
	12.04	Acetophenone		<1	120	C ₈ H ₈ O	d, e	Aromatic
	12.42	Unk. monoterpene	<1		136	C ₁₀ H ₁₆		Terpene
		Methyl benzoate		2.6	136	C ₈ H ₈ O ₂	c, d, e, f, g	Ester
15	12.67	Ethyl sorbate	16.9		140	C ₈ H ₁₂ O ₂	b, d, e	Ester
	12.74	Linalool		<1	154	C ₁₀ H ₁₈ O	c, d, e, f	Terpene
	12.78	Ethyl sorbate ^k	<1		140	C ₈ H ₁₂ O ₂	b, d	Ester
	12.88	Nonanal		<1	142	C ₉ H ₁₈ O	c, d, e, f	Aldehyde
	13.25	Phenethanol	<1		122	C ₈ H ₁₀ O	c, b, d, e, f	Aromatic
	14.03	Propyl sorbate ^k	<1		154	C ₉ H ₁₄ O ₂	b, d, f	Ester
	14.20	Ethyl benzoate	<1		150	C ₉ H ₁₀ O ₂	d, e, f	Ester
	14.55	Propyl sorbate ^k	<1		154	C ₉ H ₁₄ O ₂	b, d, f	Ester
4	14.63	Methylsalicyclate		3.9	152	C ₈ H ₈ O ₃	c, d, e,	Aromatic
	14.75	Propyl sorbate	<1		154	C ₉ H ₁₄ O ₂	b, d, e, f, g, h	Ester
	14.92	Decanal	<1		156	C ₁₀ H ₂₀ O	c, d, e	Aldehyde
	15.38	Propyl sorbate ^k	<1		154	C ₉ H ₁₄ O ₂	b, d, f	Ester
	15.36	Nerol		<1	154	C ₁₀ H ₁₈ O	c, d, e, f	Terpene

Peak no. Fig. 2	Ret. t. (min)	Compound	<i>Agave palmeri</i> (%)	<i>Datura wrightii</i> (%)	MW	Molecular formula	notes	Structure class
	15.87	Butyl sorbate	<1		154	C ₁₀ H ₁₆ O ₂	d	Ester
5	15.94	Geraniol		35.5	154	C ₁₀ H ₁₈ O	c, d, e, f	Terpene
	16.36	Geranial		1.6	152	C ₁₀ H ₁₆ O	c, d, e, f	Terpene
	16.61	Neral		1.1	152	C ₁₀ H ₁₆ O	c, d, e	Terpene
	17.85	Cycklosativene	<1		204	C ₁₅ H ₃₂	d	Terpene
	17.88	Geranyl acetate		<1	196	C ₁₂ H ₂₀ O ₂	c, d, e	Terpenoid
	18.13	Benzyl pentanoate	<1		192	C ₁₂ H ₁₆ O ₂	c, d	Ester
	18.13	Tetradecane		<1	198	C ₁₄ H ₃₀	d, e	Alkane
	18.83	Alloaromadendrene	<1		204	C ₁₅ H ₃₂	d	Sesquiterpene
	19.10	Pentadecane	<1	<1	212	C ₁₅ H ₃₂	d, e, f	Alkane
6	19.13	A-farnesene		2.3	204	C ₁₅ H ₂₄	c, d, e, f	Terpene
	19.88	Hexadecane	<1		226	C ₁₆ H ₃₄	d, e	Alkane
	20.55	Heptadecane	<1		240	C ₁₇ H ₃₆	d, f	Alkane
	21.14	Benzyl benzoate		<1	212	C ₁₄ H ₁₂ O ₂	c, d, e, f	Ester
	21.24	Octadecane	<1		254	C ₁₈ H ₃₈	d, e, f	Alkane
	22.00	Nonadecane	<1		268	C ₁₉ H ₄₀	d, e, f	Alkane

^aCompounds are listed in order of GC retention time. Compound numbers shown in the first column correspond with Fig. 2.

^bAlso observed by R. A. Raguso (1).

^cAlso observed by Raguso *et al.* (2).

^dDetermined by comparison to spectra in the National Institute of Standards and Technology (NIST) 1998 spectral library.

^eDetermined by comparison of spectra and retention times with standard samples.

^fConfirmed by accurate mass measurement (observed mass within 5 mDa of calculated mass).

^gStandard samples prepared by transesterification reactions under acidic conditions.

^hDouble bond stereochemistry not determined.

ⁱRegiochemistry undetermined.

^jDetermined by comparison to retention indices published by Helmig *et al.* (3).

^kIsomer.

1. Raguso, R. A. (2004) *Ecology* **85**, 1486-1494.
2. Raguso RA, Henzel C, Buchman SL, Nabhan GP (2003) *Int J Plant Sci* 164:877-892.
3. Helmig D, Klinger LF, Guenther A, Vierling L, Geron C, Zimmerman P (1999) *Chemosphere* 38:2163-2187.