

[Supporting information to accompany]
Supramolecular Allosteric Cofacial Porphyrin Complexes

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Figure S1. X-ray crystal structure of **8cCDABCO**.

Figure S2. X-ray crystal structure of **15cCDABCO**.

Table S1. Crystallographic data for structures **8cCDABCO** and **15cCDABCO**.

Figure S3. Uncorrected catalytic data for 2-PC with **14a** and **15a**.

Figure S4. Background reactions for 2-PC with **[Zn(TPP) + 16a]** and **[Zn(TPP) + 16b]**.

Figure S5. Uncorrected catalytic data for 3-PC with **14a** and **15a**.

Figure S6. Background reactions for 3-PC with **[Zn(TPP) + 16a]** and **[Zn(TPP) + 16b]**.

Figure S7. Uncorrected catalytic data for 4-PC with **14a** and **15a**.

Figure S8. Background reactions for 4-PC with **[Zn(TPP) + 16a]** and **[Zn(TPP) + 16b]**.

Crystallographic data in CIF format can be found via the Internet at <http://pubs.acs.org>.

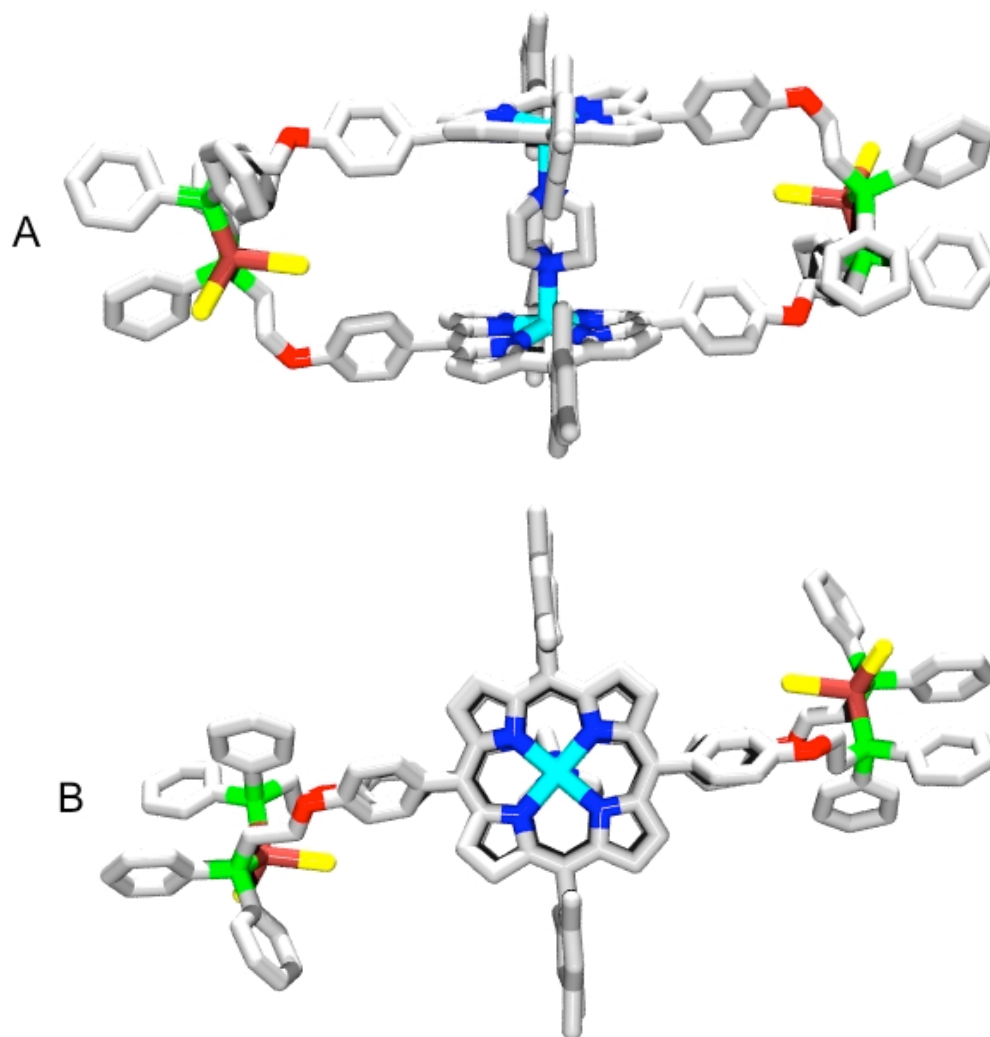


Figure S1. Graphic representations of the X-ray crystal structure of **8cCDABCO** as viewed from the side (A) and from the top (B) containing a molecule of DABCO bridging both Zn atoms. Hydrogen atoms, disordered DABCO carbon atoms, and solvent molecules have been omitted for clarity. Zn-Zn distance: 6.99 Å. Cu-Cu distance: 22.6 Å. Gray = Carbon, Brown = Cu, Red = O, Yellow = Cl, Green = P, Blue = N, Light Blue = Zn.

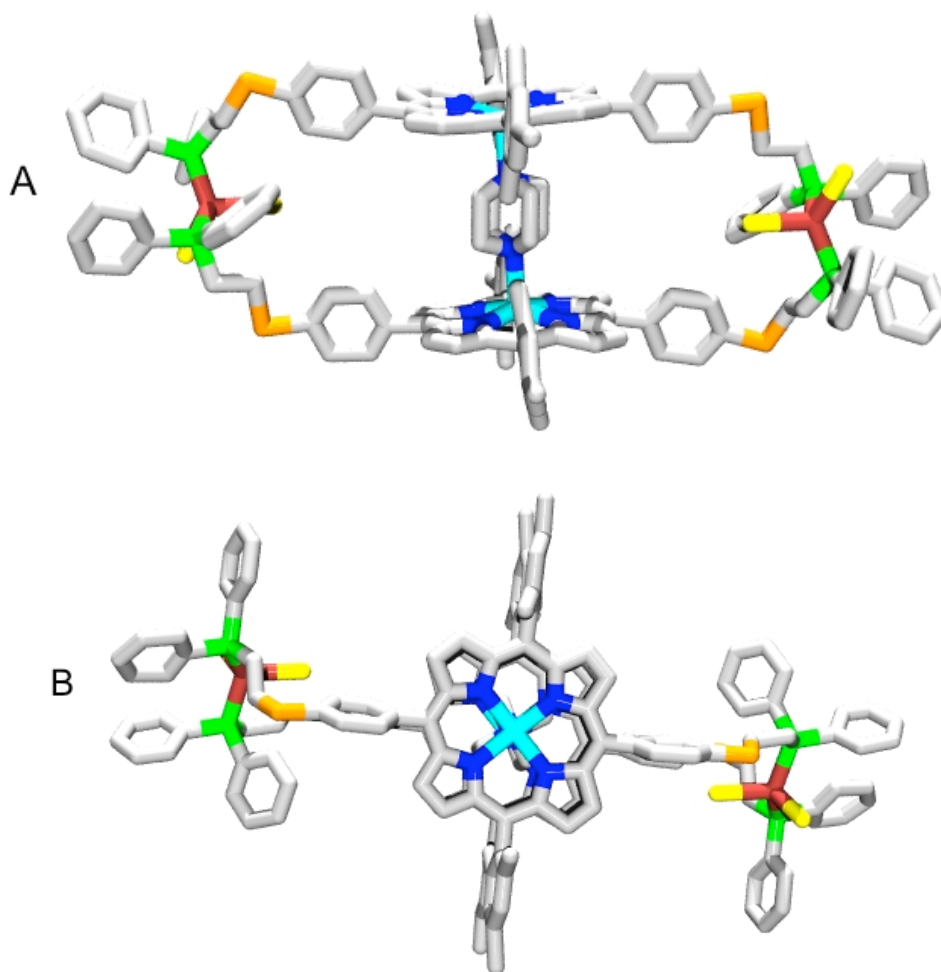


Figure S2. Graphic representations of the X-ray crystal structure of **15cCDABCO** as viewed from the side (A) and from the top (B) containing a molecule of DABCO bridging both Zn atoms. Hydrogen atoms, disordered DABCO carbon atoms, and solvent molecules have been omitted for clarity. Zn-Zn distance: 7.05 Å. Cu-Cu distance: 22.3 Å. Gray = C, Brown = Cu, Red = O, Orange = S, Yellow = Cl, Green = P, Dark Blue = N, Light Blue = Zn.

Table 1. X-ray Crystallographic Data for **8cCDABCO** and **15cCDABCO**.

	8cCDABCO	15cCDABCO
Empirical formula	C ₁₇₄ H ₁₃₆ Cl ₄ Cu ₂ N ₁₄ O ₄ P ₄ Zn ₂	C ₁₇₂ H ₁₅₆ Cl ₄ Cu ₂ N ₁₄ P ₄ S ₄ Zn ₂
Formula weight	3010.46	3094.86
Temperature	153(2) K	153(2) K
Wavelength	0.71000 Å	0.71073 Å
Crystal system, space group	Triclinic, P-1	Triclinic, P-1
Unit cell dimensions	a = 13.127(9) Å α = 97.501(6)° b = 16.349(8) Å β = 96.354(14)° c = 22.544(12) Å γ = 100.273(10)°	a = 13.283(3) Å α = 89.34(3) b = 17.340(4) Å β = 89.58(3)° c = 22.330(5) Å γ = 73.30(3)°
Volume	4675(5) Å ³	4926.0(17) Å ³
Z, Calculated density	1, 1.065 Mg/m ³	1, 1.043 Mg/m ³
Absorption coefficient	0.618 mm ⁻¹	0.628 mm ⁻¹
F(000)	1550	1608
Crystal size	0.080 x 0.040 x 0.010 mm	0.060 x 0.040 x 0.010 mm
Theta range for data collection	2.06 to 21.93 °	1.9 to 21.9 °
Reflections collected / unique	45189 / 10805 [R(int) = 0.0712]	33613 / 8598 [R(int) = 0.083]
Refinement method	Full-matrix least-squares on F ²	Full-matrix least-squares on F ²
Data / restraints / parameters	10756 / 0 / 887	10326 / 0 / 909
Goodness-of-fit on F ²	0.984	0.863
Final R indices [I > 2σ(I)]	R1 = 0.1117, wR2 = 0.2676	R1 = 0.0910, wR2 = 0.2107
R indices (all data)	R1 = 0.1591, wR2 = 0.2846	R1 = 0.1562, wR2 = 0.2311

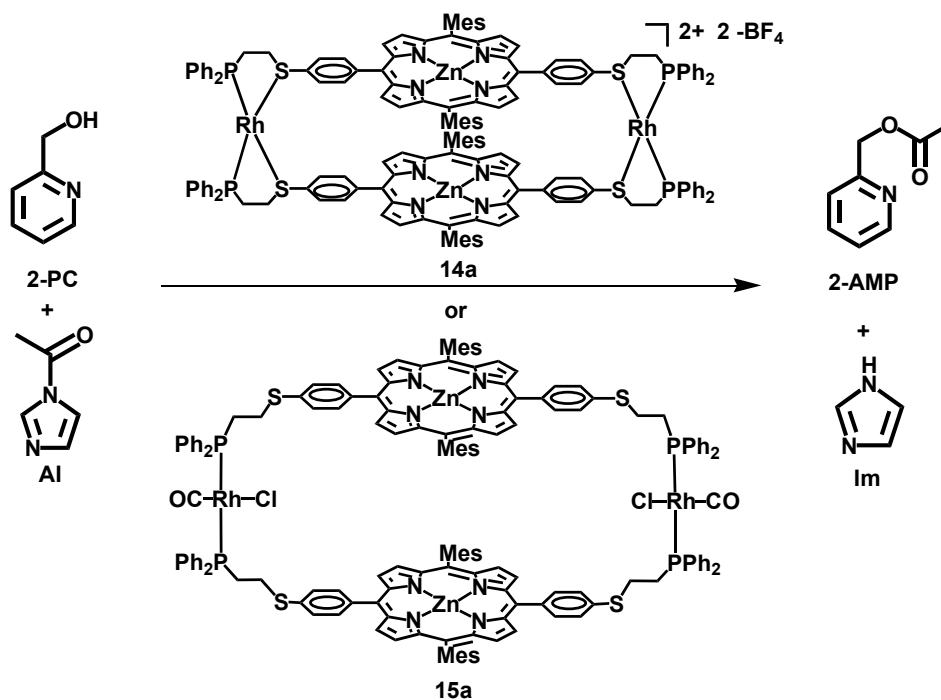
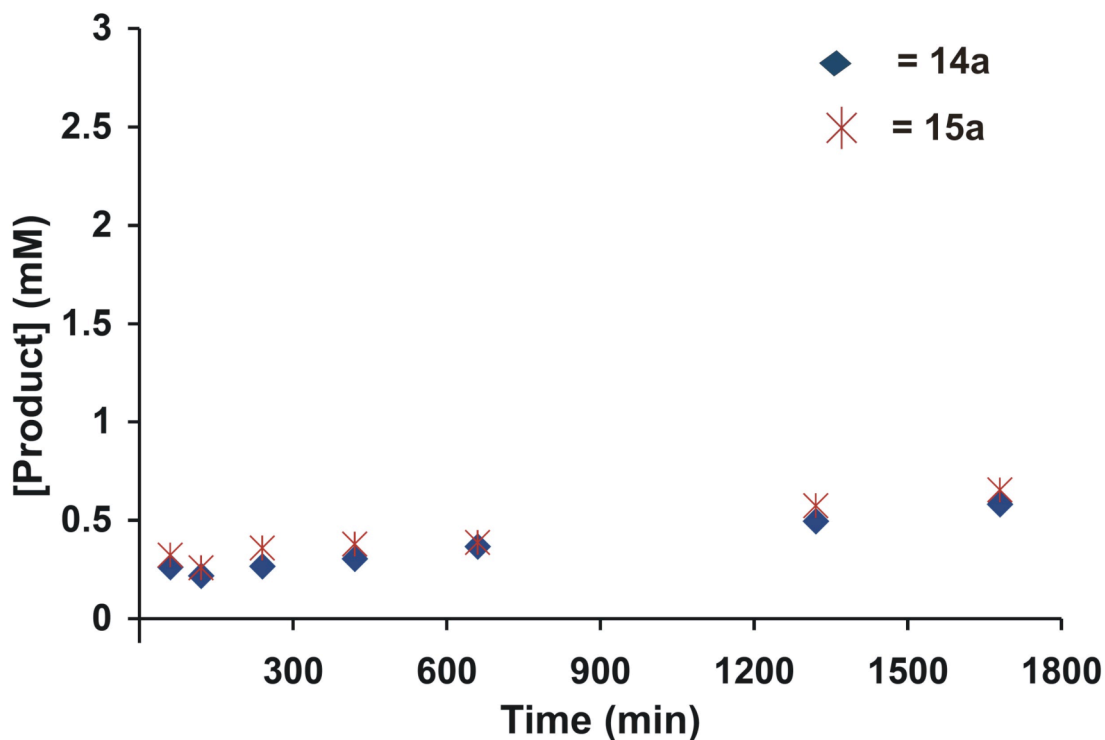


Figure S3. Formation of 2-(acetoxymethyl)pyridine (2-AMP) plotted as concentration vs. time for **14a** and **15a**. All data are uncorrected for background reactions. Conditions: CH_2Cl_2 , rt, 9 mM 2-pyridylcarbinol (2-PC), 6 mM 1-acetylimidazole (AI), 2.5 mM biphenyl (GC reference standard), and 0.3 mM supramolecular catalyst (**14a** or **15a**). CO (1 atm) and appropriate amounts of benzyltriethylammonium chloride when indicated.

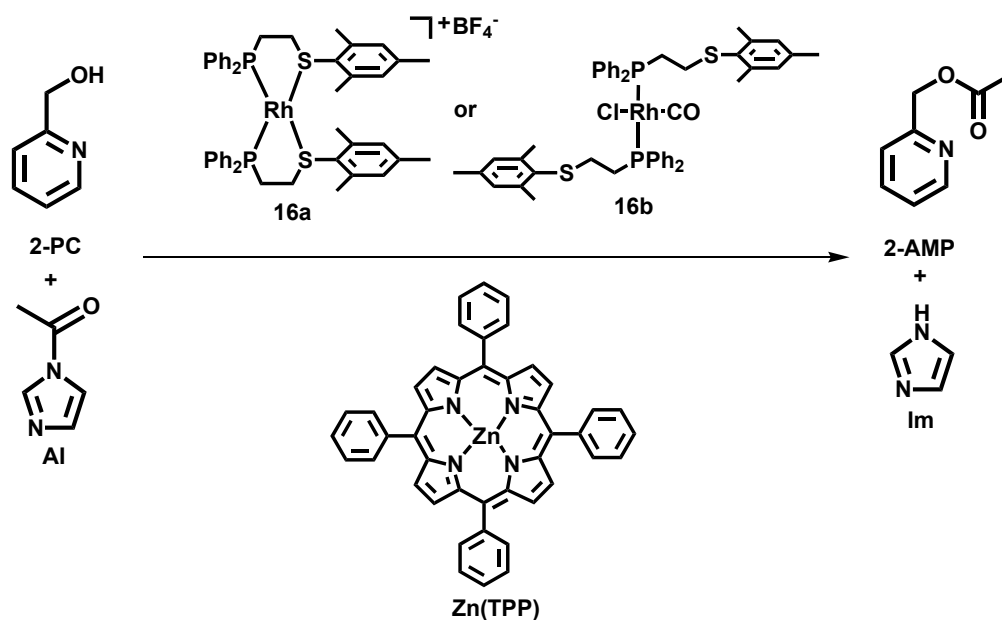
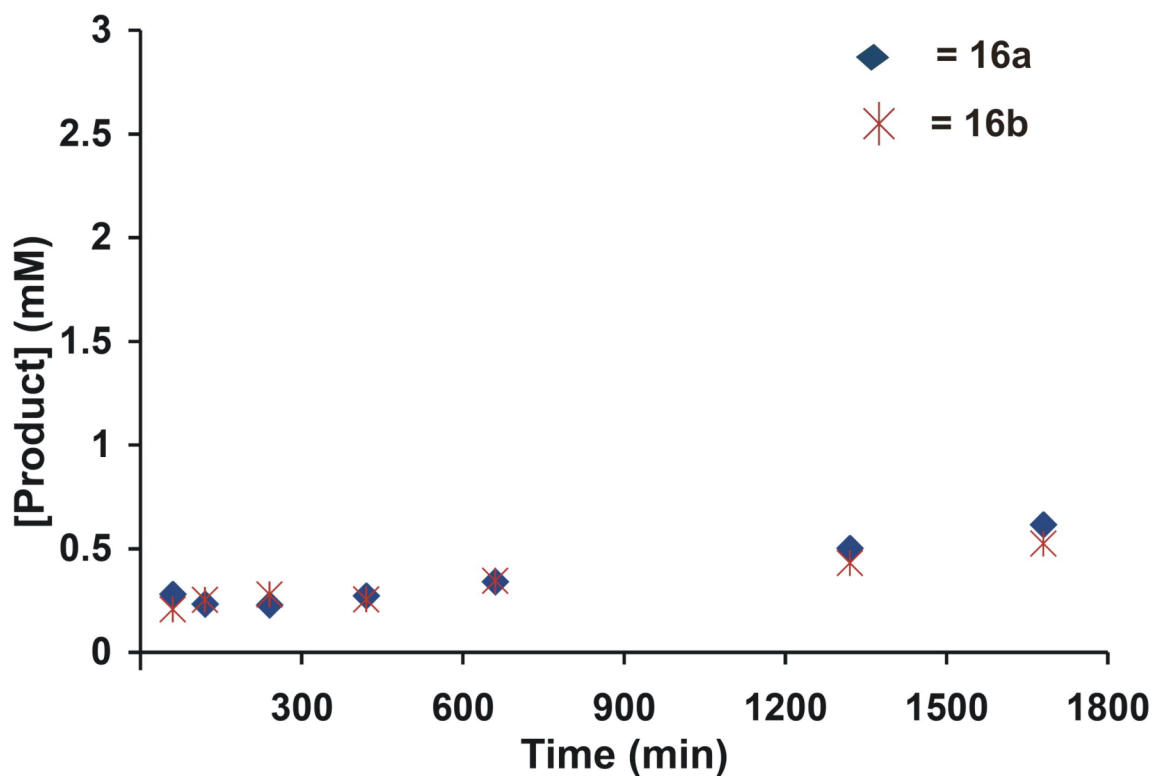


Figure S4. Formation 2-(acetoxymethyl)pyridine (2-AMP) plotted as concentration vs. time for [Zn(TPP) + 16a] and [Zn(TPP) + 16b]. Conditions: CH_2Cl_2 , rt, 9 mM 2-pyridylcarbinol (2-PC), 6 mM 1-acetylimidazole (AI), 2.5 mM biphenyl (GC reference standard), 0.6 mM Rh^{I} -monomer (16a or 16b), and 0.6 mM Zn(TPP). CO (1 atm) and appropriate amounts of benzyltriethylammonium chloride when indicated.

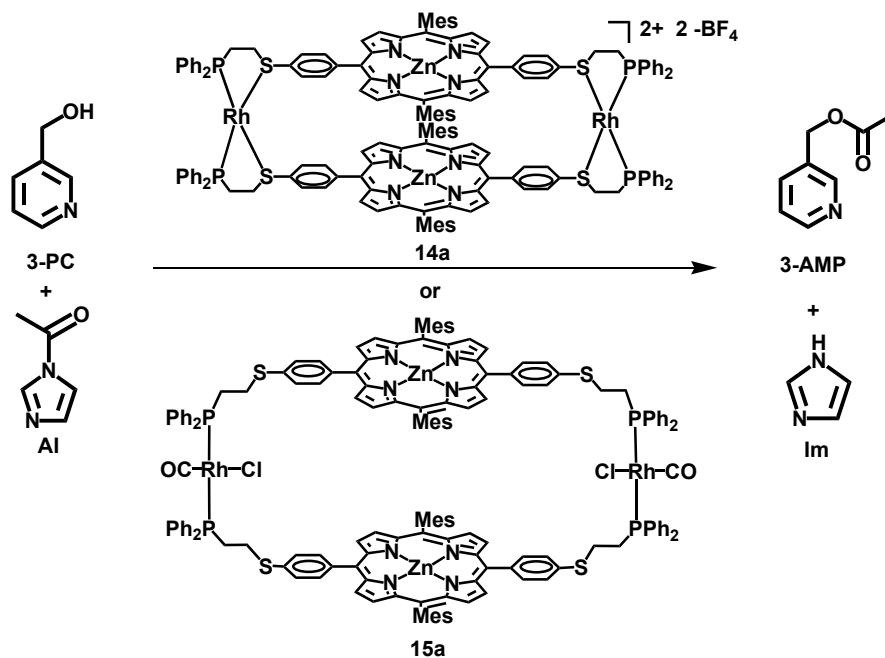
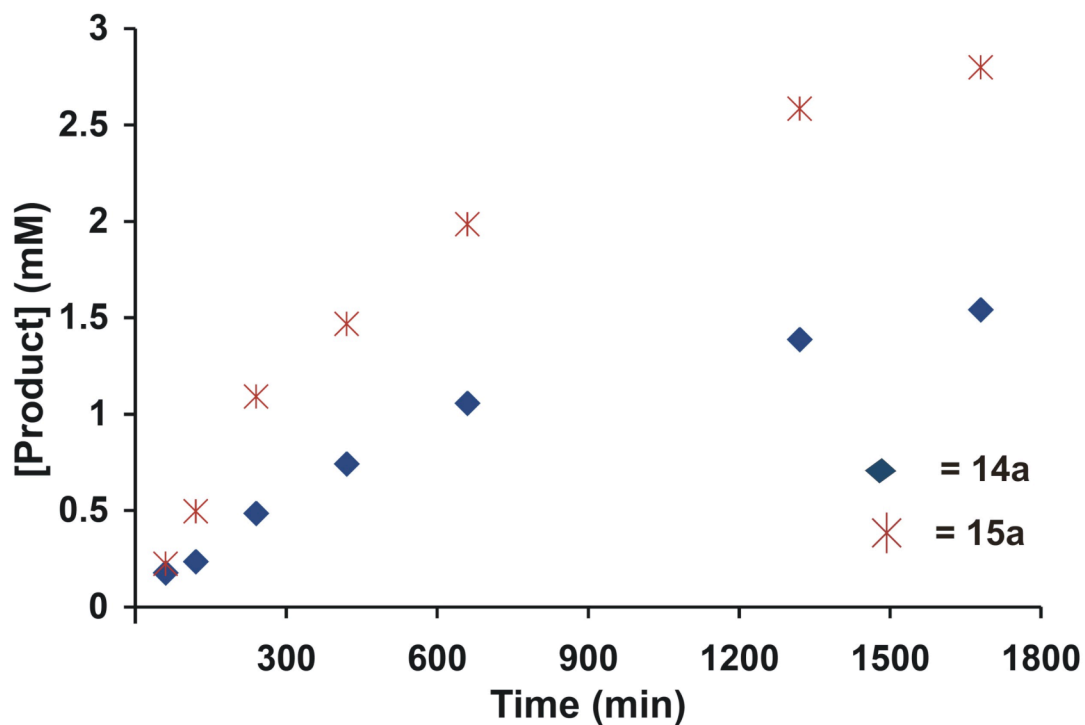


Figure S5. Formation of 3-(acetoxymethyl)pyridine (3-AMP) plotted as concentration vs. time for **14a** and **15a**. All data are uncorrected for background reactions. Conditions: CH₂Cl₂, rt, 9 mM 3-pyridylcarbinol (3-PC), 6 mM 1-acetylimidazole (AI), 2.5 mM biphenyl (GC reference standard), and 0.3 mM supramolecular catalyst (**14a** or **15a**). CO (1 atm) and appropriate amounts of benzyltriethylammonium chloride when indicated

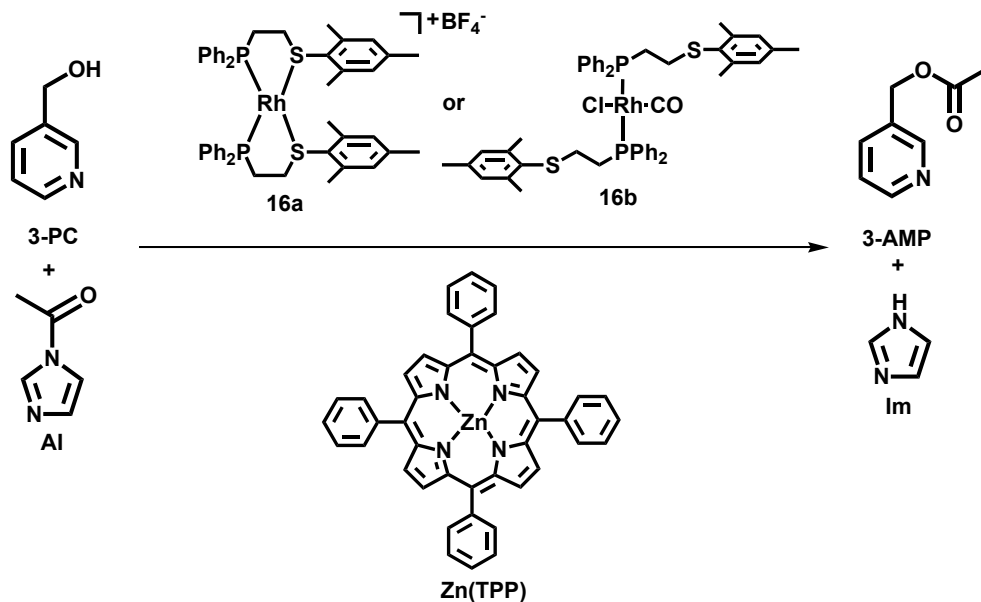
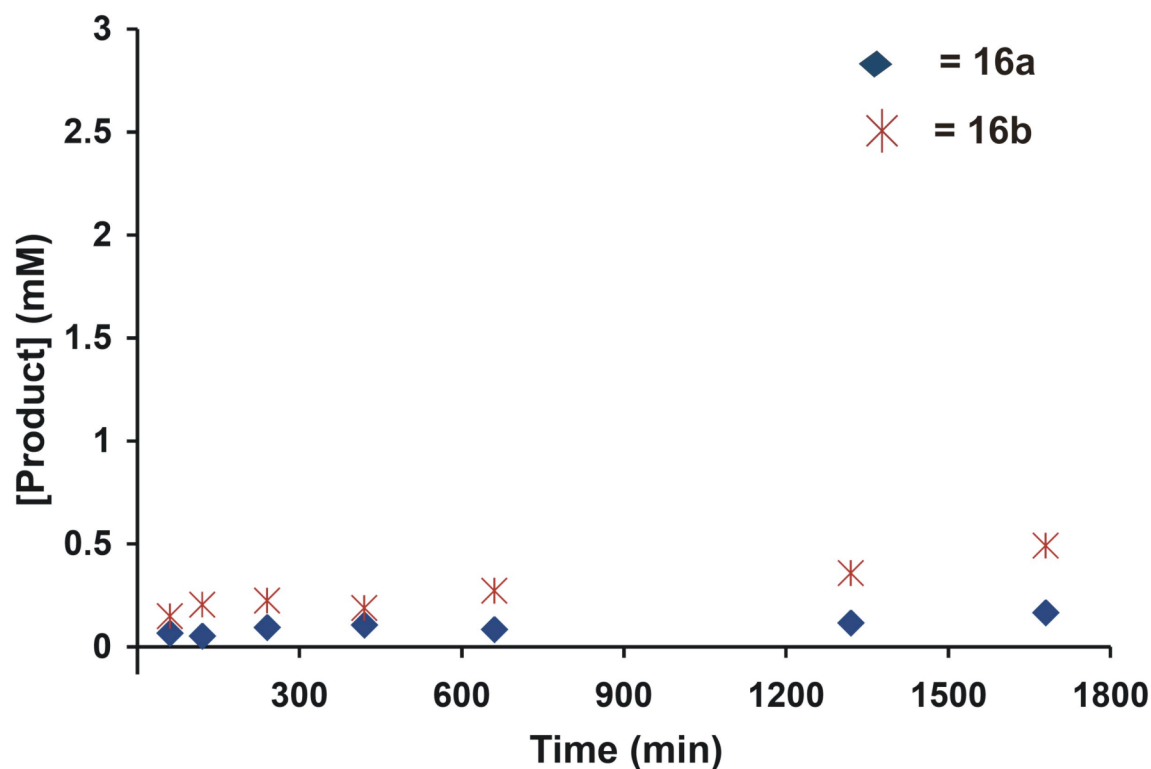


Figure S6. Formation 3-(acetoxymethyl)pyridine (3-AMP) plotted as concentration vs. time for [Zn(TPP) + 16a] and [Zn(TPP) + 16b]. Conditions: CH₂Cl₂, rt, 9 mM 3-pyridylcarbinol (3-PC), 6 mM 1-acetylimidazole (AI), 2.5 mM biphenyl (GC reference standard), 0.6 mM Rh^I-monomer (16a or 16b), and 0.6 mM Zn(TPP). CO (1 atm) and appropriate amounts of benzyltriethylammonium chloride when indicated.

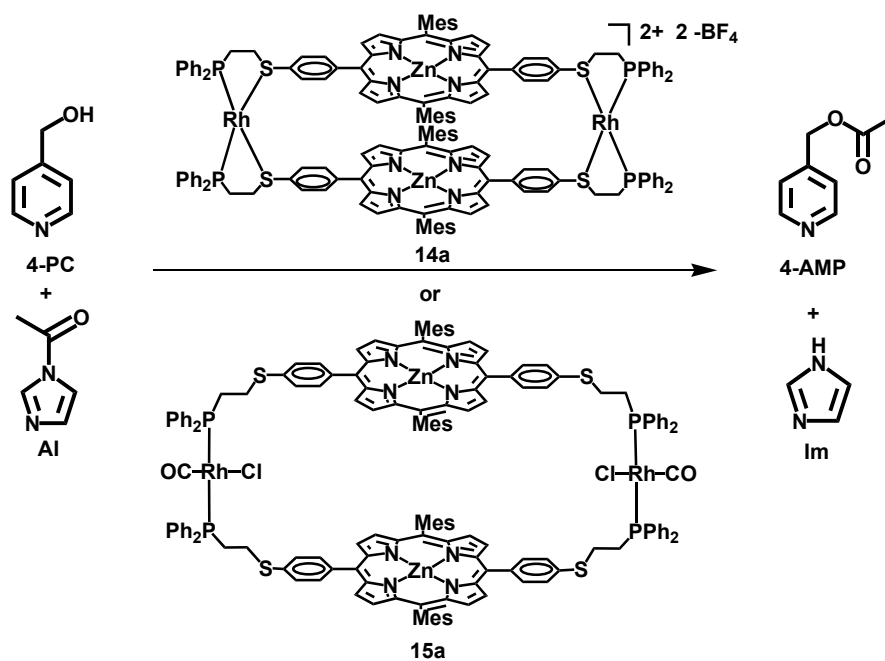
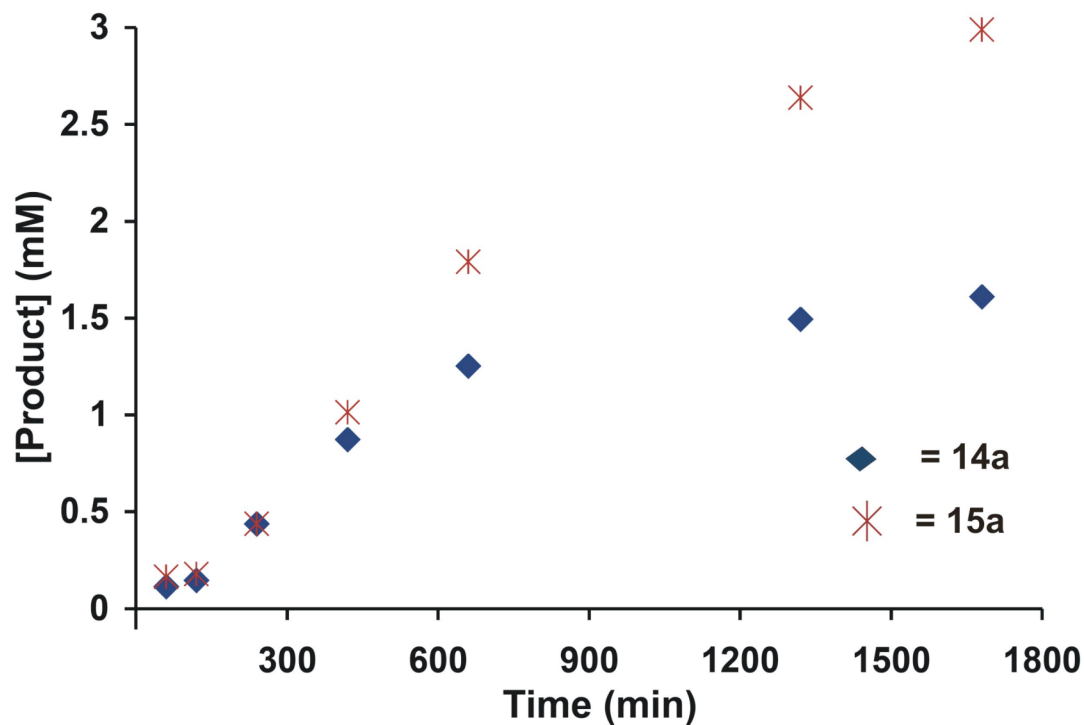


Figure S7. Formation of 4-(acetoxymethyl)pyridine (4-AMP) plotted as concentration vs. time for **14a** and **15a**. All data are uncorrected for background reactions. Conditions: CH₂Cl₂, rt, 9 mM 4-pyridylcarbinol (4-PC), 6 mM 1-acetylimidazole (AI), 2.5 mM biphenyl (GC reference standard), and 0.3 mM supramolecular catalyst (**14a** or **15a**). CO (1 atm) and appropriate amounts of benzyltriethylammonium chloride when indicated.

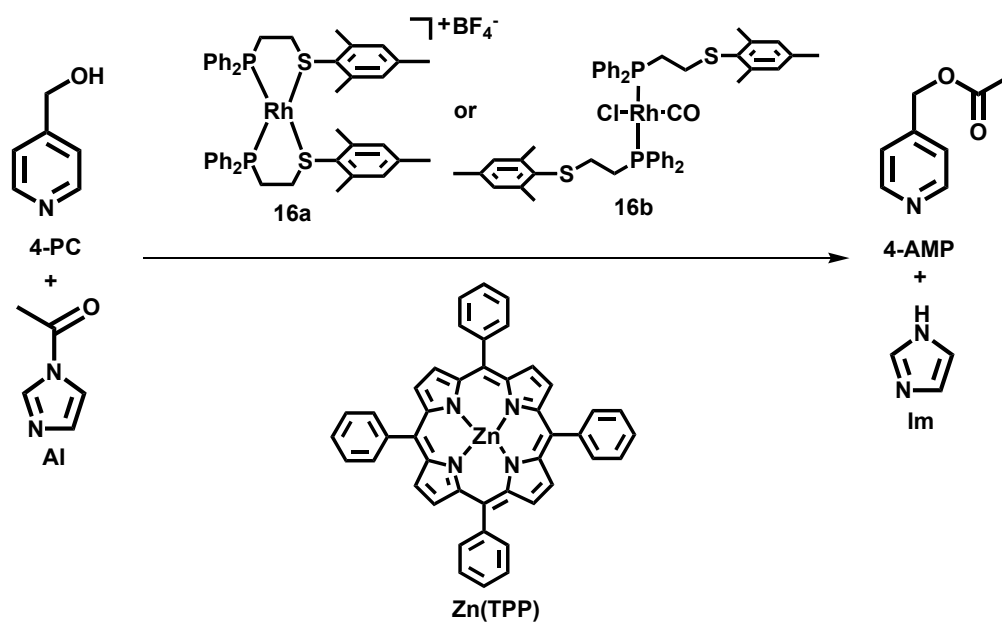
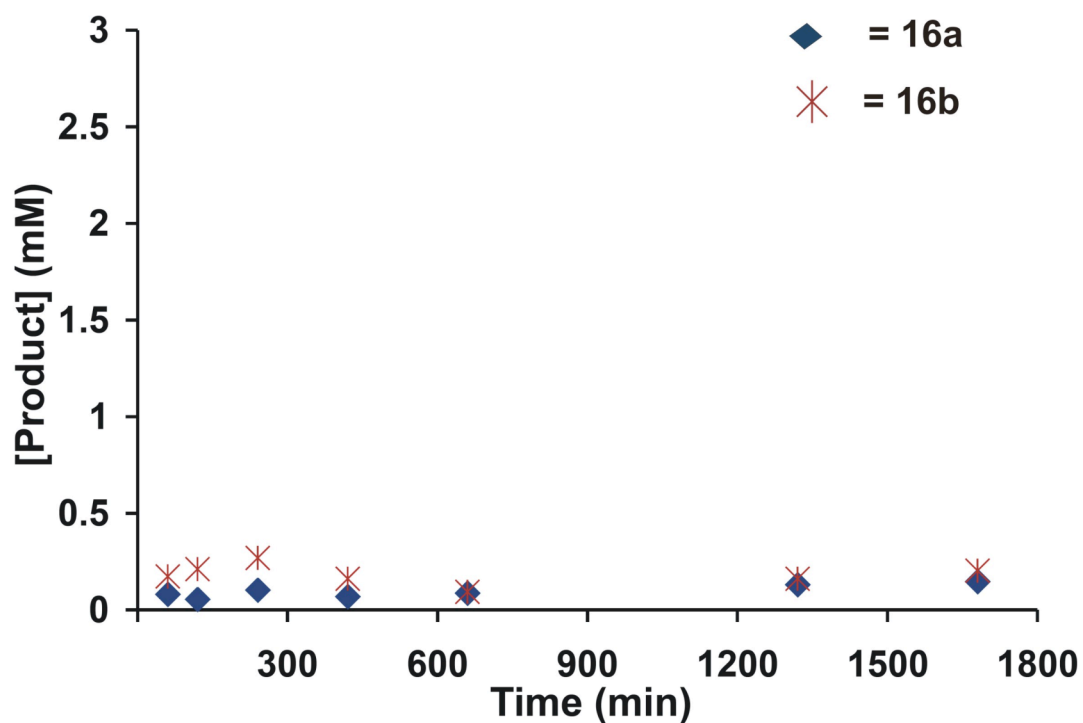


Figure S8. Formation 4-(acetoxymethyl)pyridine (4-AMP) plotted as concentration vs. time for [Zn(TPP) + 16a] and [Zn(TPP) + 16b]. Conditions: CH_2Cl_2 , rt, 9 mM 4-pyridylcarbinol (4-PC), 6 mM 1-acetylimidazole (AI), 2.5 mM biphenyl (GC reference standard), 0.6 mM Rh^{I} -monomer (16a or 16b), and 0.6 mM Zn(TPP). CO (1 atm) and appropriate amounts of benzyltriethylammonium chloride when indicated.