Probing Ligand Binding to Duplex DNA using KMnO₄ Reactions and Electrospray

Ionization Tandem Mass spectrometry

Carolyn L. Mazzitelli and Jennifer S. Brodbelt

Department of Chemistry and Biochemistry, 1 University Station A5300, University of Texas at

Austin, Austin, TX 78751

Supporting Information Section Table of Contents:

Figure S-1. ESI-mass spectra of d(GCAGTGA/TCACTGC) before the KMnO₄ reaction and with echinomycin after the KMnO₄ reaction: Page S-2.

FigureS-2.ESI-massspectraofd(GGACAGTGAGGGCAGTGAGGG/CCTGTCACTCCCGTCACTCCC)beforetheKMnO₄reaction and with echinomycin after the KMnO₄ reaction: Page S-3.

Figure S-3. ESI-mass spectra of d(GCGGGGGATGGGGGCG/CGCCCCATCCCCGC) before the $KMnO_4$ reaction and with actinomycin-D after the $KMnO_4$ reaction: Page S-4

Figure S-4. ESI-mass spectra of d(GTAGAGTCGACCTG/CAGGTCGACTCTAC) with ethidium bromide before and after the KMnO4 reaction: Page S-5.

Figure S-5. ESI-MS³ spectra for $[ds + E + O]^{6-}$ containing d(GCGGATATATGGCG/CGCCATATATCCGC) and echinomycin: Page S-6.



Figure S-1. ESI-mass spectra showing solutions containing the duplex d(GCAGTGA/TCACTGC) (a) after 20 min, reaction with KMnO₄, and (b) with echinomycin (E), after 20 min. reaction with KMnO₄. Ions containing oxidized thymines are labeled with $\mathbf{\nabla}$, with the number in parenthesis indicating the number of oxidation adducts.



Figure S-2. ESI-mass spectra showing solutions containing the duplex d(GGACAGTGAGGGCAGTGAGGG/CCTGTCACTCCCGTCACTCCC) (a) after 20 min, reaction with KMnO₄, and (b) with echinomycin (E), after 20 min. reaction with KMnO₄. Ions containing oxidized thymines are labeled with $\mathbf{\nabla}$, with the number in parenthesis indicating the number of oxidation adducts.



Figure **S-3. ESI-mass** spectra showing solutions containing the duplex d(GCGGGGGATGGGGCG/CGCCCCATCCCCGC) (a) after 20 min, reaction with KMnO₄, and (b) with actinomycin-D (ACD), after 20 min. reaction with KMnO₄. Ions containing oxidized thymines are labeled with $\mathbf{\nabla}$, with the number in parenthesis indicating the number of oxidation adducts. The single strand d(GCGGGGGATGGGGCG) is abbreviated as ss1 and d(CGCCCCATCCCCGC) is ss2.



Figure S-4. ESI-mass spectra showing solutions containing the duplex d(GTAGAGTCGACCTG/CAGGTCGACTCTAC) (a) with ethidium bromide (EB) prior to the KMnO₄ reaction, and (b) with ethidium bromide after 20 min. reaction with KMnO₄. Ions containing oxidized thymines are labeled with $\mathbf{\nabla}$, with the number in parenthesis indicating the number of oxidation adducts. The single strand d(GTAGAGTCGACCTG) is abbreviated as ss1 and d(CAGGTCGACTCTAC) is ss2.



ESI-MS³ O]⁶⁻ experiments Figure S-5. for [ds containing Е ++ d(GCGGATATATGGCG/CGCCATATATCCGC) and echinomycin (E): (a) CAD spectrum of the initial $[ds + E + O]^{6-}$ complex and (b) MS³ spectrum of the $[ss2 + O]^{3-}$ product ion. "O" is indicative of an oxidation adduct. The sequence structure in Figure 5b summarizes the sequence coverage. The fragments containing an oxidation adduct are labeled with a "▼" symbol. The thymine that was determined to be oxidized is underlined in the sequence shown in Figure 2b.