

$^1\text{H}$  NMR study of the magnetic properties and  
electronic structure of the hydroxide complex of  
substrate-bound heme oxygenase from *Nesseria*  
*meningitidis*; Influence of the axial water deprotonation  
on the distal H-bond network

Li-Hua Ma<sup>†</sup>, Yangzhong Liu<sup>†</sup>, Xuhang Zhang<sup>‡</sup>, Tadashi Yoshida<sup>‡</sup> and Gerd N. La Mar<sup>†\*</sup>

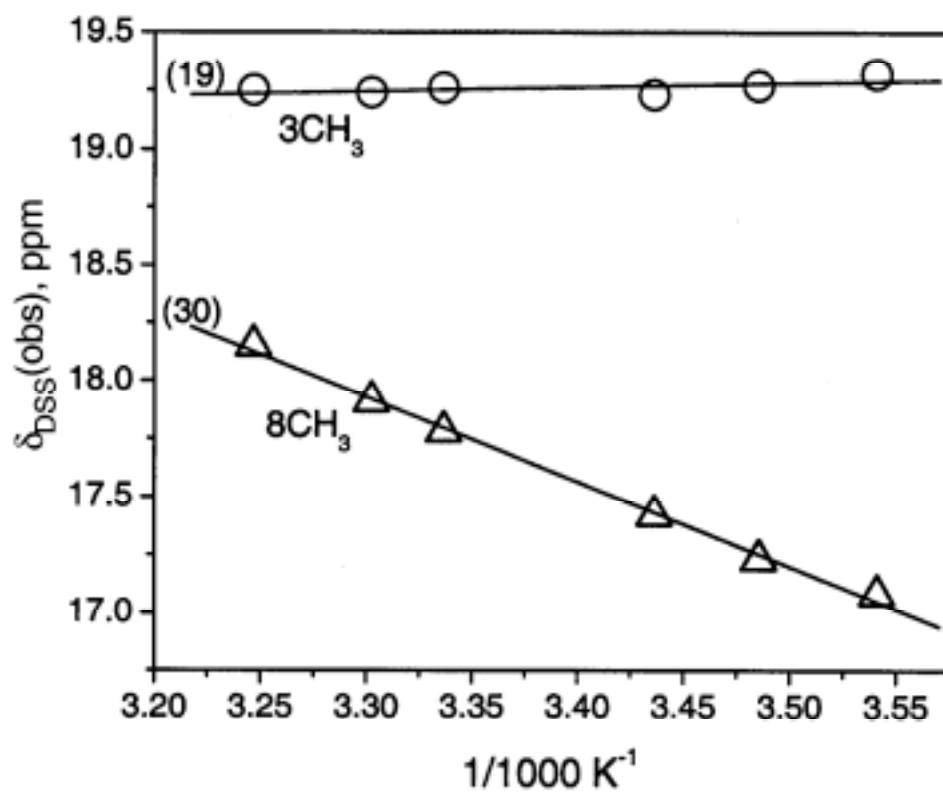
Supporting Information

Four figures (partially relaxed spectra for *NmHO*-PH-OH, Curie plot for heme methyls and comparison of magnetic axes determination for *NmHO*-PH-CN with and without considerations of rhombic anisotropies, and labile proton saturation factors) and one Table (chemical shifts for assigned *NmHO*-PH-OH residues), total 5 pages.

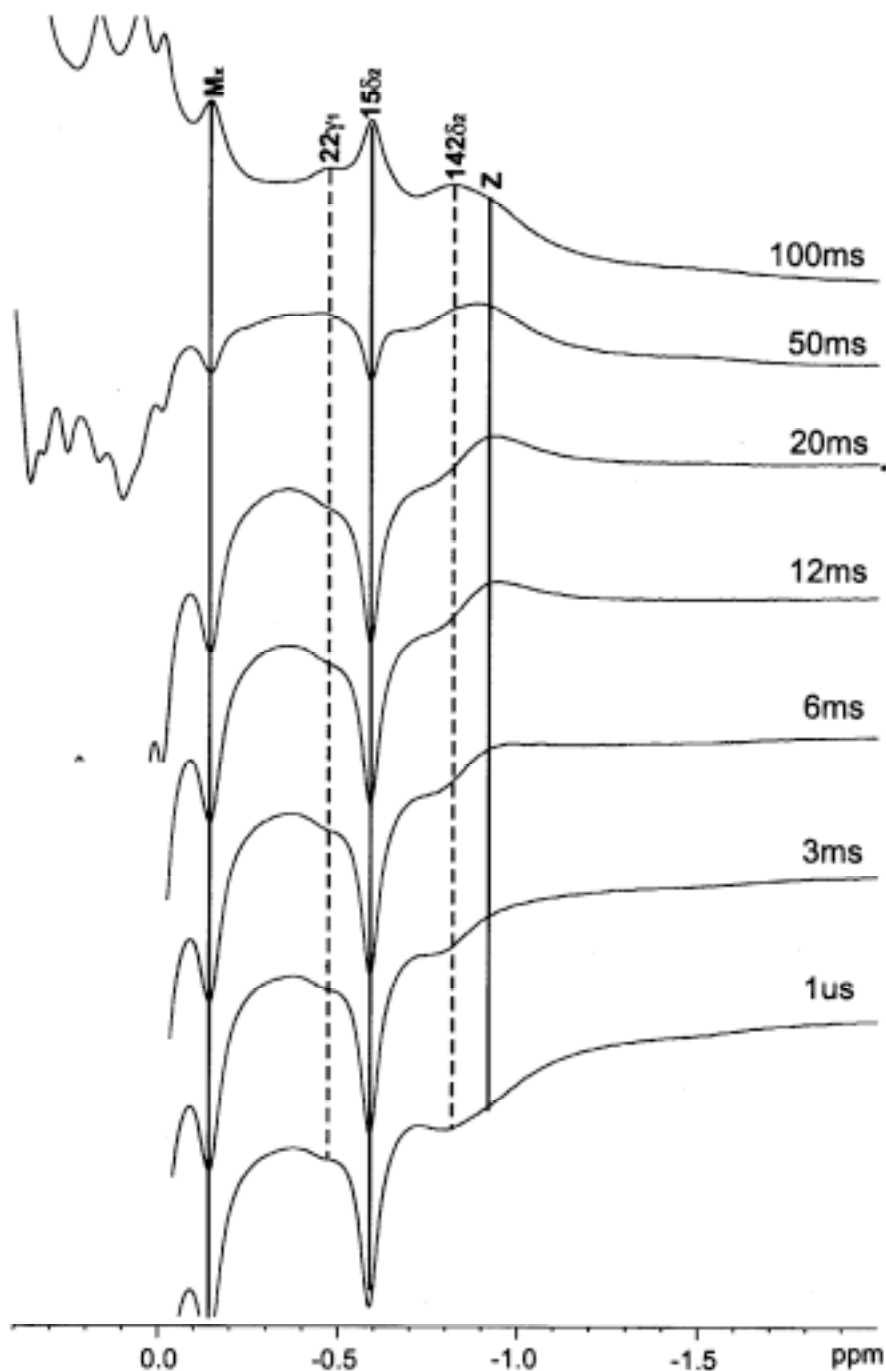
**Table S1.** Observed chemical shifts for assigned residues in *NmHO-PH-OH*<sup>a</sup>

		$\delta_{\text{DSS}}(\text{obs})$				
Ala12	C $_{\alpha}$ H	3.39		His141	C $_{\delta 2}$ H	6.30
	C $_{\beta}$ H <sub>3</sub>	1.24			C $_{\epsilon 1}$ H	7.60
Leu15	C $_{\delta 1}$ H <sub>3</sub>	-0.60		Leu142	C $_{\delta 2}$ H <sub>3</sub>	0.80
Thr20	C $_{\gamma}$ H <sub>3</sub>	1.30			C $_{\gamma}$ H	0.05
Ala21	C $_{\beta}$ H <sub>3</sub>	1.47		Asn147	C $_{\alpha}$ H	4.15
Val22	C $_{\gamma 1}$ H <sub>3</sub>	0.70			C $_{\beta 1}$ H	1.42
	C $_{\gamma 2}$ H <sub>3</sub>	0.98			C $_{\beta 2}$ H	1.96
Asp24	C $_{\beta 1}$ H	2.82		Trp153	C $_{\delta 1}$ H	6.95
	C $_{\beta 2}$ H	3.08		Val157	C $_{\alpha}$ H	2.85
Phe52	C $_{\alpha}$ H	4.40			C $_{\beta}$ H	1.55
	C $_{\beta 1}$ H	3.03			C $_{\gamma 1}$ H <sub>3</sub>	0.37
	C $_{\beta 2}$ H	3.13			C $_{\gamma 2}$ H <sub>3</sub>	-0.02
	C $_{\epsilon}$ Hs	7.43		Ala180	C $_{\alpha}$ H	2.58
	C $_{\zeta}$ H	7.31			C $_{\beta}$ H <sub>3</sub>	0.58
His53	C $_{\delta 1}$ H	7.45		Phe181	C $_{\beta 1}$ H	3.57
	C $_{\epsilon 2}$ H	7.95			C $_{\beta 2}$ H	3.03
Tyr112	C $_{\beta 1}$ H	3.02		Ala182	C $_{\alpha}$ H	4.10
	C $_{\beta 2}$ H	3.30			C $_{\beta}$ H <sub>3</sub>	1.66
	C $_{\delta}$ Hs	6.90		Phe183	C $_{\alpha}$ H	4.53
Cys113	C $_{\beta 1}$ H	2.72			C $_{\beta 1}$ H	3.68
	C $_{\beta 2}$ H	2.76			C $_{\beta 2}$ H	3.42
Ala114	C $_{\alpha}$ H	4.37			C $_{\delta}$ Hs	7.64
	C $_{\beta}$ H <sub>3</sub>	1.90			C $_{\epsilon}$ Hs	7.37
Asn118	C $_{\beta 1}$ H	3.12		Tyr184	C $_{\delta}$ Hs	6.18
	C $_{\beta 2}$ H	3.02			C $_{\epsilon}$ Hs	6.48
Ala121	C $_{\beta}$ H <sub>3</sub>	2.30		Val186	C $_{\alpha}$ H	3.64
Phe125	C $_{\delta}$ Hs	7.04			C $_{\beta}$ H	2.40
	C $_{\epsilon}$ Hs	7.12			C $_{\gamma 2}$ H <sub>3</sub>	1.25
	C $_{\zeta}$ H	7.20		Val187	C $_{\alpha}$ H	3.42
Ala139	C $_{\alpha}$ H	4.11			C $_{\beta}$ H	1.50
					C $_{\gamma 1}$ H <sub>3</sub>	0.69
					C $_{\gamma 2}$ H <sub>3</sub>	0.08

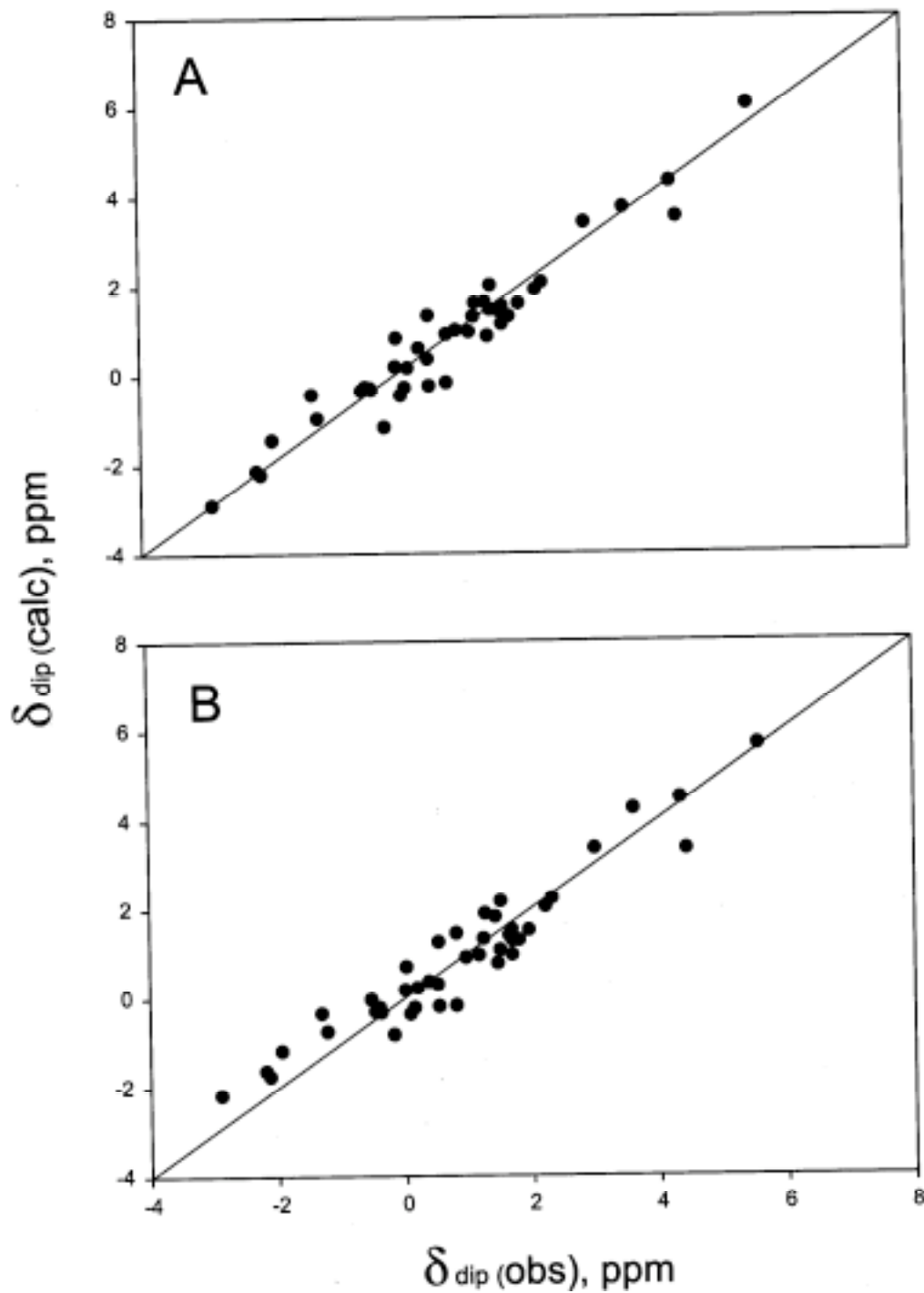
a)  $\delta_{\text{DSS}}(\text{obs})$  in ppm referenced to DSS via the solvent resonance, in <sup>2</sup>H<sub>2</sub>O at pH 9.1, 100 mM bicarbonate at 25°C.



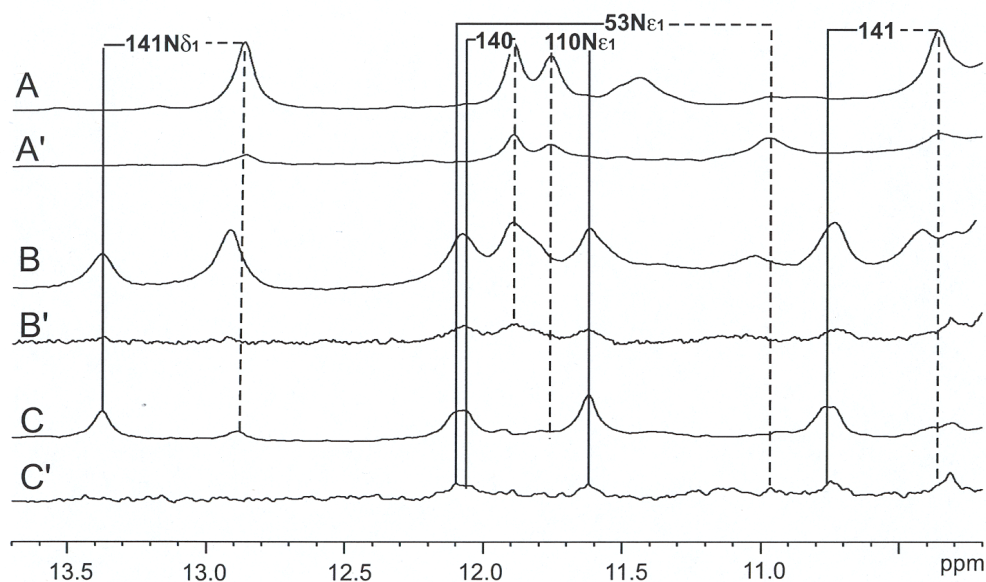
**Figure S1.** Curie plot ( $^1H$  NMR  $\delta_{DSS}(obs)$  versus reciprocal absolute temperature ( $T^{-1}$ )) for the resolved heme methyl peak for *NmHO-PH-OH* at pH 10.5, in  $^2H_2O$ , 100 mM bicarbonate at 25°C. The apparent intercepts, in ppm, at  $T^{-1} = 0$ , are given in parenthesis.



**Figure S2.** Partially relaxed spectra in an inversion-recovery pulse sequence for the resolved upfield 600 MHz  $^1\text{H}$  NMR spectrum of *NmHO*-PH- $^2\text{H}_2\text{O}$  in  $^2\text{H}_2\text{O}$  100 mM bicarbonate, pH 10.5 at 25°C. The relaxation delay times are given on the right. Peaks are labeled by amino acid number and proton position, with solid and dashed lines representing signals in *NmHO*-PH-OH and *NmHO*-PH- $\text{H}_2\text{O}$ , respectively. The single proton peak z exhibits a null at  $\sim 5$  ms, or a  $T_1 \sim 8$  ms, and exhibits the shift<sup>38, 49</sup> and  $T_1$  are consistent with it arising from a vinyl  $\text{H}_\beta$ .



**Figure S3.** Plots of  $\delta_{\text{dip}}(\text{obs})$  versus  $\delta_{\text{dip}}(\text{calc})$  obtained from the magnetic axes determination of *NmHO-PH-CN* based on: (A) five-parameter search to determine  $\Delta\chi_{\text{ax}} (= 2.48 \pm 0.1 \times 10^{-8} \text{ m}^3/\text{mol})$ ,  $\Delta\chi_{\text{rh}} (= -0.52 \pm 0.09 \times 10^{-8} \text{ m}^3/\text{mol})$ , the tilt of magnitude ( $\beta = 8 \pm 1^\circ$ ), tilt direction ( $\alpha = 280 \pm 10^\circ$ ) and rhombic axes ( $\kappa = 41 \pm 10^\circ$ ); and a three-parameter search keeping  $\Delta\chi_{\text{rh}} = 0$  and  $\gamma = 0$ , to determine  $\Delta\chi_{\text{ax}} (= 2.4 \pm 0.1 \times 10^{-8} \text{ m}^3/\text{mol})$ , tilt magnitude ( $\beta = 4 \pm 1^\circ$ ) and tilt direction ( $\alpha = 260 \pm 10^\circ$ ). The residues error function,  $F/n$ , is smaller for the five-parameter ( $F/n = 0.05$ ) than three-parameter ( $F/n = 0.06$ ) search.



**Figure 4S.** 3:9:19 reference spectra<sup>52</sup> for the labile proton involved in strong H-bonds in *NmHO-PH-H<sub>2</sub>O/NmHO-PH-OH* mixture in <sup>1</sup>H<sub>2</sub>O, 50 mM in phosphate at 25°C, as a function of pH. Assigned peak in *NmHO-PH-H<sub>2</sub>O*<sup>50</sup> and *NmHO-PH-OH* are connected by bars over the reference spectra, with dashed and solid vertical lines representing the *NmHO-PH-H<sub>2</sub>O* and *NmHO-PH-OH* complexes, respectively. (A) pH 7.2 (~98% *NmHO-PH-H<sub>2</sub>O*/2% *NmHO-PH-OH*) (B) pH 9.4 (~60% *NmHO-PH-H<sub>2</sub>O*/~40% *NmHO-PH-OH*) and (C) pH 10.2 (20% *NmHO-H<sub>2</sub>O*/80% *NmHO-OH*). Magnetization difference 3:9:19 traces upon subtracting the off-resonance from the on-resonance saturation of the water signals as a function of pH as shown in A', B' and C', which demonstrate the saturation factor due to exchange and not increased (for His141 N<sub>δ</sub>H actually decreased) upon increasing pH from 7.2 to 10.2.