

# Reversibility Effects on the Stereoselectivity of Pt(II)-Mediated Cascade Poly-ene Cyclizations

## Supporting Information

*Jeremiah A. Feducia, Michel R. Gagné*

*Department of Chemistry, University of North Carolina at Chapel Hill,  
Chapel Hill, North Carolina 27599-3290*

### Table of Contents

1. Experimental Procedures
2. Figure S1.  $^{31}\text{P}$  NMR of cyclization of (*E:Z*)-**1** with  $(\text{EtPPPEt})\text{Pt}^{2+}$  in  $\text{CH}_2\text{Cl}_2$ .
3. Figure S2.  $^{31}\text{P}$  NMR of  $[(\text{EtPPPEt})\text{Pt}(1\text{-hexene})][\text{BF}_4]_2$ .
4. Figure S3.  $^{31}\text{P}$  NMR of cyclization of (*E:Z*)-**1** with  $(\text{EtPPP})\text{Pt}^{2+}$  in  $\text{MeNO}_2$ .
5. Table S1.  $^{31}\text{P}$  NMR Data Table.
6. Table S2. Acid or base additive effects on the cyclization of (*E:Z*)-**1**.
7. Crystallographic Data for **4** (Table S3-S7, Figure S3)
8. Crystallographic Data for **8** (Table S8-S13, Figure S4)
9. Crystallographic Data for **9** (Table S14-S19, Figure S5)
10. NMR Spectra for Isolated Compounds

### Experimental Procedures

**General Methods.** All reactions were performed under an inert atmosphere of  $\text{N}_2$  using standard Schlenk techniques or using an MBraun Lab-Master 100 glove box. Solvents for equilibrium studies ( $\text{CH}_2\text{Cl}_2$ ,  $\text{ClCH}_2\text{CH}_2\text{Cl}$ ,  $\text{MeNO}_2$ ,  $\text{EtNO}_2$ ) were dried over  $\text{CaH}_2$ , distilled, degassed by several successive freeze-pump-thaw cycles and stored in a

glove box. All starting [(RPPPR')PtMe][BF<sub>4</sub>] and [(RPPPR')Pt-I][I] compounds,<sup>1</sup> [Ph<sub>2</sub>NH<sub>2</sub>][BF<sub>4</sub>]<sup>2</sup>, [Ph<sub>2</sub>NMeH][BF<sub>4</sub>]<sup>2</sup> and **1**<sup>3</sup> were prepared by published procedures. Diphenylamine was purchased from Aldrich, distilled over CaH<sub>2</sub>, and stored over 4Å molecular sieves in a glove box. 2,3-dihydropyran was purchased from Aldrich and distilled prior to use. NMR spectra were recorded on a Bruker Avance 400 MHz or Bruker Avance 500 MHz spectrometer; chemical shifts are given in ppm and are referenced to residual solvent resonances (<sup>1</sup>H and <sup>13</sup>C). NMR spectra to determine the structure of the products from the cyclization of **10** were recorded on an 800 MHz Varian Inova NMR Spectrometer at Duke University (Durham, NC). <sup>31</sup>P NMR chemical shifts are referenced to an external PPh<sub>3</sub> standard in C<sub>6</sub>D<sub>6</sub> sealed in a capillary tube.

**E-2:** Prepared by a similar procedure to **1**<sup>3</sup> using 2,3-dihydropyran in place of dihydrofuran. The crude material was separated by silica gel chromatography using 4:1 hexanes/EtOAc to yield 647.6 mg (54%) of a pale yellow oil; <sup>1</sup>H NMR: (400 MHz, CDCl<sub>3</sub>) δ 7.72 (d, 2 H, *J* = 7.6 Hz), 7.28 (d, 2 H, *J* = 7.6 Hz), 5.73 (m, 1 H), 4.94 (m, 3 H), 4.37 (br, 1 H), 2.91 (q, 2 H, *J* = 6.8 Hz), 2.41 (s, 3 H), 2.08 (m, 2 H), 1.96 (m, 4 H), 1.58 (s, 3 H), 1.48 (m, 2 H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 143.2, 138.5, 136.9, 135.7, 129.6, 127.0, 123.1, 114.3, 42.8, 38.9, 32.2, 29.4, 24.8, 21.5, 15.9. HRMS (ESI) [M+K]<sup>+</sup>/*z* calc. 346.124, found 346.120.

**E-7:** Prepared by a published procedure<sup>3</sup> using 3-methyl-2,3-dihydrofuran.<sup>4</sup> The crude material was purified by silica gel chromatography using 9:1 hexanes/EtOAc to yield 797.1 mg (65%) of a clear oil; <sup>1</sup>H NMR: (400 MHz, CDCl<sub>3</sub>) δ 5.73 (m, 1 H), 4.96

<sup>1</sup> Feducia, J. A.; Campbell, A. N.; Anthis, J. W.; Gagné, M. R. *Organometallics* **2006**, *25*, 3114-3117.

<sup>2</sup> Forschner, T. C.; Cutler, A. R. *Organometallics* **1985**, *4*, 1247-1257.

<sup>3</sup> Koh, J. H.; Mascarenhas, C.; Gagné, M. R. *Tetrahedron* **2004**, *60*, 7405-7410.

<sup>4</sup> Chan, J.; Jamison, T. F. *J. Am. Chem. Soc.* **2004**, *126*, 10682-10691.

(m, 2 H), 4.85 (d, 1 H,  $J = 9.6$  Hz), 3.45 (m, 1 H), 3.25 (t, 1 H,  $J = 9.2$  Hz), 2.60 (m, 1 H), 2.12 (m, 4 H), 1.63 (s, 3 H), 1.41 (m, 1 H), 0.92 (d, 3 H,  $J = 10.8$  Hz);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  158.0, 138.2, 115.2, 101.4, 77.4, 37.7, 31.2, 27.8, 21.2. HRMS (ESI)  $[\text{M}+\text{Na}]/z$  calc. 177.126, found 177.127.

***E,E*-10**: Prepared by a published procedure<sup>3</sup> using 3-methyl-2,3-dihydrofuran.<sup>4</sup> The crude material was purified by silica gel chromatography using 9:1 hexanes/EtOAc to yield 370.4 mg (53%) of a clear oil;  $^1\text{H}$  NMR: (400 MHz,  $\text{CDCl}_3$ )  $\delta$  5.77 (m, 1 H), 5.07 (m, 1 H), 4.98 (d, 1 H,  $J = 17.2$  Hz), 4.90 (d, 1 H,  $J = 10.4$  Hz), 4.84 (d, 1 H,  $J = 9.2$  Hz), 3.44 (m, 1 H), 3.26 (t, 1 H,  $J = 9.2$  Hz), 2.59 (m, 1 H), 2.07 (m, 8 H), 1.64 (s, 3 H), 1.58 (s, 3 H), 1.38 (m, 1 H), 0.89 (d, 3 H,  $J = 6.8$  Hz);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  139.1, 138.3, 135.3, 127.6, 124.6, 114.7, 68.3, 40.2, 39.4, 35.8, 32.8, 26.8, 17.4, 16.9, 16.4. HRMS (ESI)  $[\text{M}+\text{Na}]/z$  calc. 245.1881, found 245.1877.

**4**: In a glovebox, (PPP) $\text{PtI}_2$  (0.10 mmol) and  $\text{AgBF}_4$  (0.23 mmol) were weighed out into a glass vial. Dichloromethane was added to this and the solution was stirred for 10 minutes upon which (***E:Z***-1) (0.11 mmol) was added to the solution dropwise. After an additional 10 minutes of stirring  $\text{AgI}$  began to precipitate and diphenylmethanamine (0.11 mmol) was added to the suspension. This mixture was shielded from light and stirred for 2 hours. The suspension was filtered through celite to remove  $\text{AgI}$ , washed three times with a saturated  $\text{NaHCO}_3$  solution, dried over  $\text{MgSO}_4$  and concentrated *in vacuo*. The product was precipitated three times from dichloromethane and  $\text{Et}_2\text{O}$  and dried to give 75.4 mg (68%) of a white solid;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.78 (m, 2 H); 7.51 (m, 25 H); 7.20 (d, 2 H,  $J = 8.0$  Hz); 3.35 (m, 3 H); 2.87 (m, 3 H); 2.43 (m, 7 H);

1.79 (m, 1 H); 1.43 (m, 1 H); 1.29 (m, 4 H); 0.97 (m, 4 H); 0.80 (s, 3 H).  $^{31}\text{P}\{^1\text{H}\}$  NMR (162 MHz,  $\text{CDCl}_3$ ):  $\delta$  90.0 (s,  $^3J_{\text{Pt-P}} = 1319$  Hz, 1P); 44.6 (s,  $^3J_{\text{Pt-P}} = 2991$  Hz, 2P).

**8:** In a glovebox, (PPP)PtI<sub>2</sub> (0.53 mmol) and AgBF<sub>4</sub> (0.133 mmol) were weighed out into a glass vial. Dichloromethane (15 mL) and acetone (1.06 mmol) was added to this and the solution was stirred for 1 hour upon which a 3 mL solution of **E,7** (0.58 mmol) and Ph<sub>2</sub>NMe (1.06 mmol) in dichloromethane was added to the solution dropwise. This mixture was stirred for 1 hour. The suspension was filtered through celite to remove AgI, washed three times with a saturated NaHCO<sub>3</sub> solution, washed once with brine and dried over MgSO<sub>4</sub>. The filtrate from drying was concentrated *in vacuo*. The product was precipitated three times from dichloromethane and Et<sub>2</sub>O and dried to give 380.3 mg (74%) of a white solid. Crystals suitable for X-ray crystallography were grown by vapor diffusion with dichloromethane and pentane;  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.77 (m, 2 H); 7.48 (m, 23 H); 3.30 (m, 2 H); 2.98 (t, 1 H,  $J = 8.0$  Hz); 2.79 (m, 2 H); 2.29 (m, 5 H); 1.49 (m, 3 H); 1.22 (m, 2 H); 0.87 (m, 3 H); 0.56 (d, 3 H,  $J = 10.0$  Hz); 0.44 (t, 1 H,  $J = 11.6$  Hz); 0.06 (s, 3 H).  $^{31}\text{P}\{^1\text{H}\}$  NMR (162 MHz,  $\text{CDCl}_3$ ):  $\delta$  89.6 (s, 1 P,  $J_{\text{Pt-P}} = 1316$  Hz); 41.1 (d, 2 P,  $J_{\text{P-P}} = 22$  Hz,  $J_{\text{Pt-P}} = 3001$  Hz).

**9:** In a glovebox, (EtPPP)PtI<sub>2</sub> (0.11 mmol) and AgBF<sub>4</sub> (0.29 mmol) were weighed out into a glass vial. Dichloromethane (8 mL) was added to this and the solution was stirred for 10 minutes upon which **E-7** (0.13 mmol) was added to the solution dropwise. After an additional 10 minutes of stirring AgI began to precipitate and diphenylmethylamine (0.13 mmol) was added to the suspension. This mixture was stirred for 48 hours. The suspension was filtered through celite to remove AgI, washed three times with a saturated NaHCO<sub>3</sub> solution, dried over MgSO<sub>4</sub> and concentrated *in*

*vacuo*. The product was precipitated three times from dichloromethane and Et<sub>2</sub>O and dried to give 53.6 mg (51%) of a white solid. Crystals suitable for X-ray crystallography were grown by vapor diffusion with dichloromethane and diethyl ether; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.68 (m, 4 H); 7.59 (m, 6 H); 7.43 (m, 10 H); 3.70 (t, 1 H, *J* = 8.0 Hz); 3.45 (m, 2 H); 3.09 (t, 1 H, *J* = 8.0 Hz); 2.65 (m, 2 H); 2.34 (m, 5 H); 1.77 (m, 1 H); 1.67 (m, 2 H); 1.33 (m, 2 H); 1.03 (m, 1 H); 0.85 (m, 4 H); 0.78 (m, 2 H); 0.71 (m, 1 H); 0.57 (d, 3 H, *J* = 6.5 Hz); 0.48 (s, 3 H). <sup>31</sup>P{<sup>1</sup>H} NMR (162 MHz, CDCl<sub>3</sub>): δ 90.9 (s, <sup>3</sup>*J*<sub>Pt-P</sub> = 1241 Hz, 1P); 43.8 (s, <sup>3</sup>*J*<sub>Pt-P</sub> = 3012 Hz, 2P).

**Typical Procedure for Equilibrium Measurements.** A solution of [(RPPPR)PtMe][BF<sub>4</sub>] (0.008 mmol), [Ph<sub>2</sub>NH<sub>2</sub>][BF<sub>4</sub>] (0.08 mmol), and (*E:Z*)-**1** or *E-2* (0.08 mmol) in 300 μL of CH<sub>3</sub>NO<sub>2</sub><sup>5</sup> (or other solvent for Table 1) was prepared in a glove box and sealed in a J-Young NMR tube at 25(1) °C. The reaction was monitored by <sup>31</sup>P NMR until no change in peak intensity was observed.

**Equilibrium Calculations.** Equilibrium constants were calculated from the average molar ratios of the Pt(η<sup>2</sup>-alkenyl) and Pt(alkyl) complexes using the equation below. [Pt] was obtained by determining the ratio of Pt-alkyl:Pt(η<sup>2</sup>-alkene) by <sup>31</sup>P NMR. The concentration of acid and base were calculated from the ratio of Pt-alkyl:Pt(η<sup>2</sup>-alkene). 100% cyclization would result in 10 equiv. of acid present in solution while 0% cyclization would result in 9 equiv. of acid and 1 equiv. of base.

---

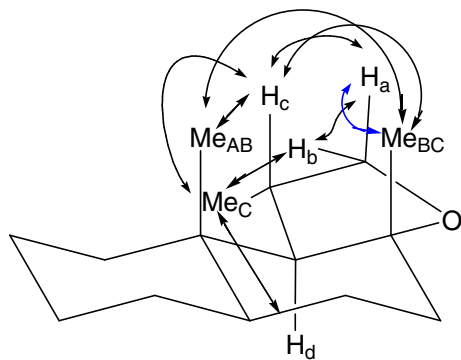
<sup>5</sup> Commercial MeNO<sub>2</sub> contains traces of propionitrile that poison the Pt<sup>2+</sup> catalyst, for purification see: Parrett, F. W.; Sun, M. S. *J. Chem. Educ.* **1977**, *54*, 448-449.

$$K_{\text{eq}} = \frac{[\text{Pt-alkyl}][\text{Ph}_2\text{NH}_2^+]}{[\text{Pt}(\eta^2\text{-alkene})][\text{Ph}_2\text{NH}]}$$

**Structural Assignment for Products of Cyclization.** Stereochemical assignments for the products of the cyclization of *E:Z-1* and *E-7* were determined by the organic fragment of the X-ray structures of **4**, **8**, and **9** respectively. Reductive cleavage of the organic fragment with NaBH<sub>4</sub> followed by GC analysis resulted in a more expedient process to determine the products of cyclization with various Pt<sup>2+</sup> sources.<sup>6</sup> Products from the cyclization of **10** were isolated following the standard protocol for Pt-alkyl reductive cleavage with NaBH<sub>4</sub>.<sup>6</sup> The stereochemistry was determined by NOESY experiments on a 600 and 800 MHz spectrometer. *J*<sub>HH</sub> for the protons shown below was deduced from selective decoupling experiments performed on an 800 MHz spectrometer. Blue arrows indicate a weak NOE interaction.

---

<sup>6</sup> Koh, J. H.; Gagné, M. R. *Angew. Chem. Int. Ed.* **2004**, *43*, 3459-3461.

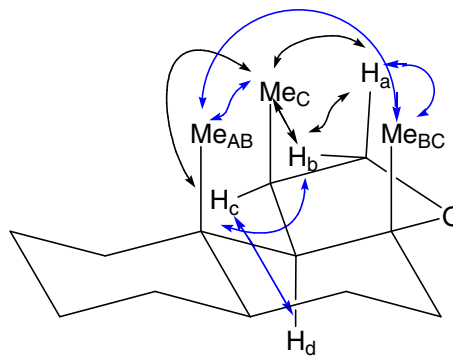
**Major Diastereomer**

$\delta$  (ppm) (NOEs)

H<sub>a</sub>: 3.99 H<sub>b</sub>, H<sub>c</sub>, Me<sub>BC</sub>(w)  
 H<sub>b</sub>: 3.22 H<sub>a</sub>, Me<sub>C</sub>  
 H<sub>c</sub>: 2.25 H<sub>a</sub>, Me<sub>BC</sub>, Me<sub>C</sub>, Me<sub>AB</sub>  
 H<sub>d</sub>: 1.60 Me<sub>C</sub>

d,  $J_{HH} = 11.5$  Hz

Me<sub>AB</sub>: 0.74 H<sub>c</sub>, Me<sub>BC</sub>  
 Me<sub>BC</sub>: 1.08 H<sub>a</sub>(w), H<sub>c</sub>, Me<sub>AB</sub>  
 Me<sub>C</sub>: 1.02 H<sub>b</sub>, H<sub>c</sub>, H<sub>d</sub>  
 d,  $J_{HH} = 6.6$  Hz

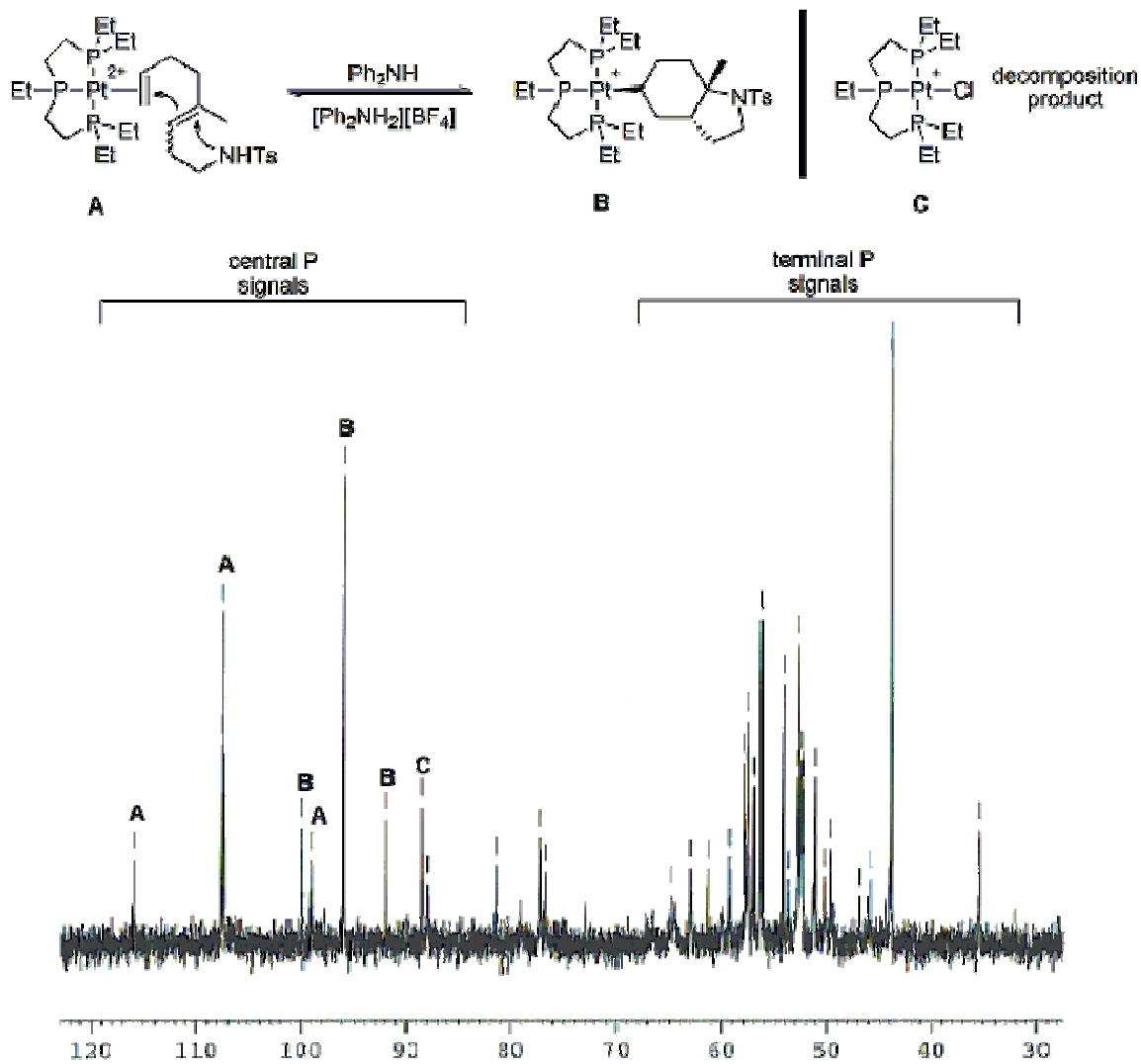
**Minor Diastereomer**

$\delta$  (ppm) (NOEs)

3.43 H<sub>b</sub>, Me<sub>BC</sub>(w), Me<sub>C</sub>(w)  
 4.11 H<sub>a</sub>, H<sub>c</sub>(w), Me<sub>C</sub>(w)  
 2.43 H<sub>b</sub>(w), H<sub>D</sub>(w), Me<sub>C</sub>(w)  
 1.38 H<sub>C</sub>

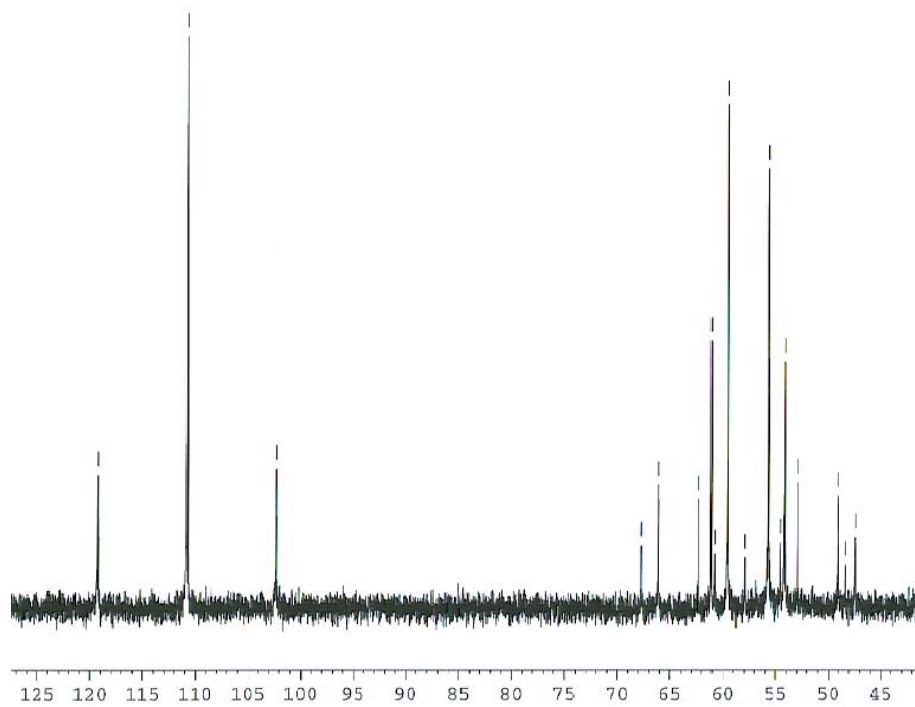
d,  $J_{HH} = 9.0$  Hz

0.87 Me<sub>BC</sub>(w), Me<sub>C</sub>(w)  
 1.24 H<sub>a</sub>(w), Me<sub>AB</sub>(w)  
 1.11 H<sub>a</sub>, H<sub>b</sub>, H<sub>c</sub>, Me<sub>AB</sub>(w)  
 d,  $J_{HH} = 7.2$  Hz

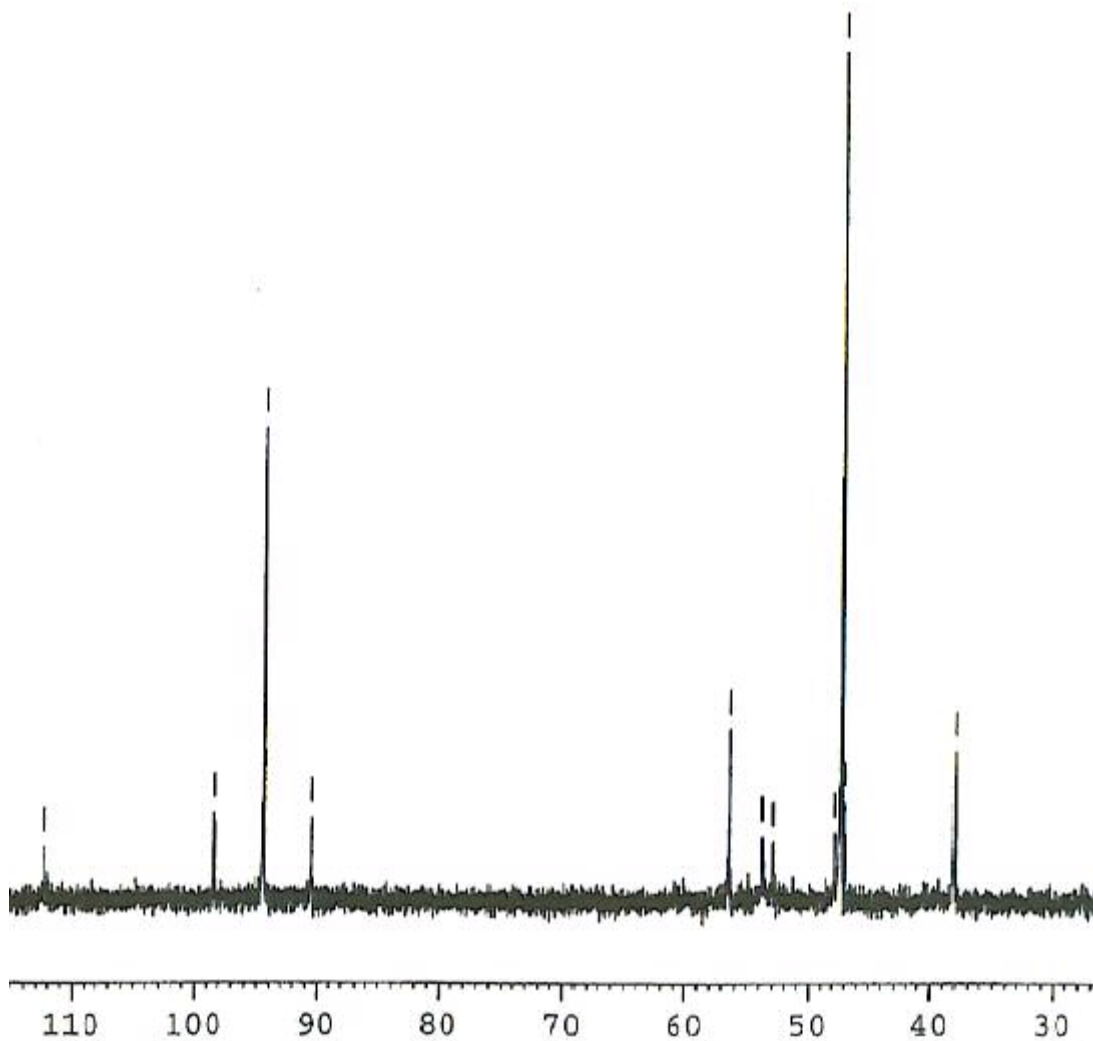


**Figure S1.**  $^{31}\text{P}$  NMR of cyclization of  $(E:Z)$ -1 with  $(\text{EtPPPEt})\text{Pt}^{2+}$  in  $\text{CH}_2\text{Cl}_2$ .





**Figure S2.**  $^{31}\text{P}$  NMR of  $[(\text{EtPPPEt})\text{Pt}(1\text{-hexene})][\text{BF}_4]_2$ .



**Figure S3.**  $^{31}\text{P}$  NMR of cyclization of (*E:Z*)-**1** with  $(\text{EtPPP})\text{Pt}^{2+}$  in  $\text{MeNO}_2$ .

**Table S1.**  $^{31}\text{P}$  NMR chemical shifts<sup>7</sup> for  $\text{Pt}(\eta^2\text{-alkenyl})$  and  $\text{Pt}(\text{alkyl})$  complexes.

Ligand	$\text{Pt}(\eta^2\text{-alkenyl})$	$\text{Pt}(\text{alkyl})$
EtPPPEt	110.2 ppm	98.3 ppm
tBuPPPEt	136.4 ppm	120.6 ppm
PPPEt	108.1 ppm	95.3 ppm
EtPPP	110.7 ppm	92.9 ppm
PPP	-	91.4 ppm

<sup>7</sup> Chemical shifts are given for the central P of the triphosphine ligand. This signal is the most diagnostic as the chemical shifts for the terminal phosphines in the  $\text{Pt}(\eta^2\text{-alkenyl})$  and the  $\text{Pt}(\text{alkyl})$  overlap.

**Table S2.** Acid or base additive effects on the cyclization of (*E:Z*)-**1**.<sup>a</sup>

Entry	Additive	A : B
1	None	14 : 1
1*	Ph <sub>2</sub> NH	1.6 : 1
4	None	1 : 10
4*	Ph <sub>2</sub> NH <sub>2</sub> <sup>+</sup>	1 : 4

<sup>a</sup> Reaction conditions: [Pt] = 0.027 M, [Ph<sub>2</sub>NH<sub>2</sub>][BF<sub>4</sub>] = 0.27 M, [1] = 0.27 M, 25(1) °C. <sup>b</sup> Relative concentrations determined by <sup>31</sup>P NMR. Average of three measurements.

**Reductive Cleavage of Pt-alkyls.** To a 0.02 M solution of Pt-alkyl in dry Et<sub>2</sub>O (for *E-5* or *E-7*) or THF (for *E,E-10*) at 0°C was added 10 equivalents of NaBH<sub>4</sub> under N<sub>2</sub>. The reaction mixture was warmed to room temperature and stirred for 3-5 hours. At this point the reaction was quenched with H<sub>2</sub>O, diluted with brine and the cleaved products were extracted with Et<sub>2</sub>O and washed three times with brine. The organic layer was separated, dried over MgSO<sub>4</sub>, filtered and concentrated. For *E-5*, the product was concentrated on a rotary evaporator at 0 °C. The crude material was purified by chromatography on silica with 300:1 pentane/Et<sub>2</sub>O to yield 13.0 mg (26%) of clear oil whose purity was compared to known spectral data. For *E-7*, the products were concentrated on a rotary evaporator at 0 °C and injected onto the GC (HP 6890 with a DB-1 column; 80 °C for 8 min, 20 °C/min to 250 °C, hold 15 min): t<sub>R</sub> 8.87 min (kinetic-PPP) and 9.34 min (thermodynamic-EtPPP). For *E,E-10*, the products were concentrated under vacuum and the 2:1 mixture of diastereomers was obtained by chromatography on silica gel using 300:1 hexanes/EtOAc.

### X-Ray Structure Determinations of **4**, **8** and **9**.

**General Procedure.** Crystals of **4** suitable for X-ray crystallography were grown from slow evaporation of a dichloromethane solution. Crystals of **8** and **9** suitable for X-

ray crystallography were grown from vapor diffusion using dichloromethane and pentane at room temperature (48 h for **9**, 7 days for **8**). Single crystals were mounted in oil on the end of a fiber. Intensity data were collected on a Siemens SMART diffractometer with CCD detection using Mo K $\alpha$  radiation of wavelength 0.71073 Å (to scan mode). The structure was solved by direct methods and refined by least-squares techniques on  $F$  using structure solution programs from the NARCVAX system.<sup>8</sup> All non-hydrogen atoms were refined anisotropically. Hydrogen atoms were placed in calculated positions (C-H = 0.96 Å) and allowed to ride on the atoms to which they were bonded. Crystal data, data collection, and refinement parameters for **4**, **8** and **9** are listed in Table S3, S8 and S14 respectively. Absorption corrections were made using SADABS. The X-ray structure for compound **4** was refined as an enantiomeric twin which resulted in the expected 50:50 mix. For compound **8**, the structure was solved in P2(1)/c (with  $Z = 8$ ,  $Z' = 2$ ). One molecule was disordered but eventually refined with the disordered portion restrained to have a consistent geometry. This still led to an unstable refinement and weak restraints on thermal parameters were put in place. In the structure shown for **9**, PLATON/Squeeze was used to model the solvent as it was too disordered to model with atomic positions (results are found at the bottom of the .cif file).

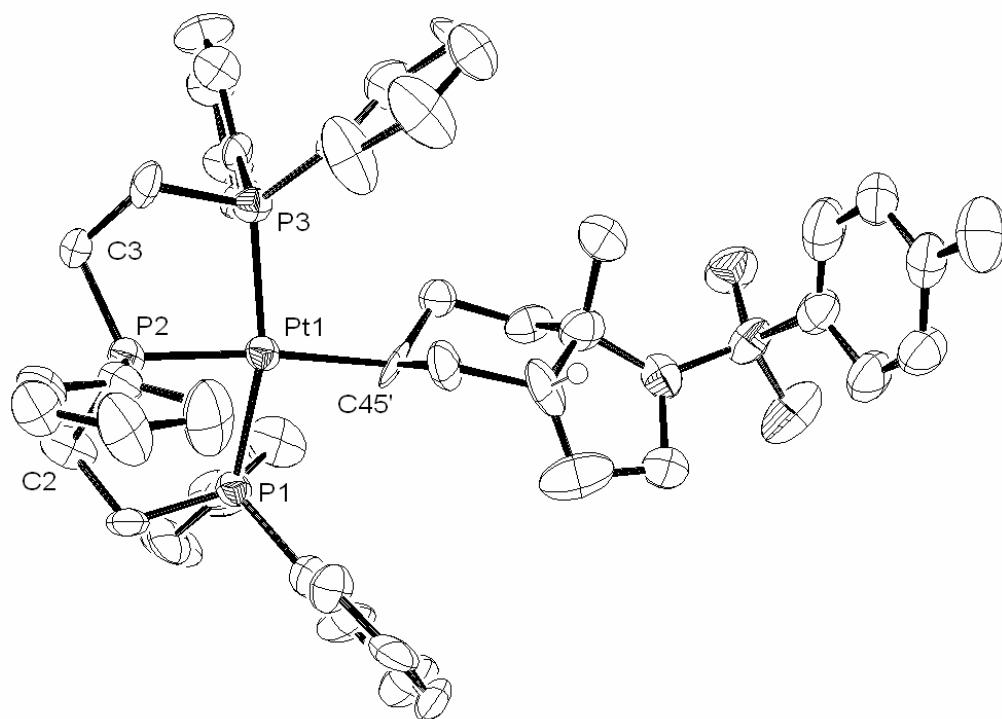
---

<sup>8</sup> Gabe, E. J.; Le Page, Y.; Charland, J. P.; Lee, F. L.; White, P. S. *J. Appl. Crystallogr.* **1989**, 22, 384-392.

**Table S3.** Crystallographic data and collection parameters for **4**.

	<b>4</b>
Empirical Formula	C <sub>51</sub> H <sub>57</sub> BCl <sub>2</sub> F <sub>4</sub> NO <sub>2</sub> P <sub>3</sub> PtS
FW	1193.75
Space Group	Orthorhombic, P2 <sub>1</sub> 2 <sub>1</sub> 2
<i>A</i>	18.7099(12)
<i>B</i>	25.8448(13)
<i>C</i>	10.7596(4)
<i>V</i> , Å <sup>3</sup>	5202.8(5)
<i>Z</i>	4
<i>T</i> , K	173(2)
<i>D</i> <sub>c</sub> , Mg/m <sup>3</sup>	1.524
<i>λ</i> , Å	Mo Kα (0.71073)
<i>μ</i> , mm <sup>-1</sup>	2.986
<i>R</i> <sub>f</sub> <sup><i>a</i></sup>	0.0706
<i>R</i> <sub>w</sub> <sup><i>b</i></sup>	0.1536
GOF <sup><i>c</i></sup>	1.093

<sup>*a*</sup> $R_f = \Sigma(F_o - F_c)/\Sigma F_o$ . <sup>*b*</sup> $R_w = [\Sigma\omega(F_o - F_c)^2/\Sigma\omega F_o^2]^{1/2}$ . <sup>*c*</sup>GOF =  $[\Sigma\omega(F_o - F_c)^2/(n - p)]^{1/2}$ , where *n* = number of reflections and *p* = number of parameters.



**Figure S3.** ORTEP drawing of **4**.  $\text{BF}_4^-$  counter ion and solvent molecule not shown.

**Table S4.** Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **4**.  $U(\text{eq})$  is defined as one third of the trace of the orthogonalized  $U^{ij}$  tensor.

	x	y	z	$U(\text{eq})$
Pt (1)	2931 (1)	8829 (1)	-457 (1)	31 (1)
P (1)	2969 (3)	9696 (1)	-781 (3)	35 (1)
P (2)	4067 (2)	8843 (2)	-1136 (3)	33 (1)
P (3)	3016 (2)	7946 (1)	-753 (3)	33 (1)
S (1)	-461 (2)	8601 (2)	3652 (4)	50 (1)
C (1)	3896 (9)	9867 (5)	-1201 (13)	42 (4)
C (2)	4207 (10)	9456 (6)	-1973 (13)	50 (5)
C (3)	4265 (8)	8247 (6)	-1980 (13)	36 (4)
C (4)	3953 (7)	7824 (6)	-1208 (14)	41 (4)
C (5)	2720 (8)	10133 (6)	513 (17)	49 (4)
C (6)	3087 (11)	10050 (7)	1627 (12)	58 (5)
C (7)	2902 (16)	10347 (7)	2650 (18)	83 (7)
C (8)	2346 (11)	10704 (9)	2560 (20)	81 (8)
C (9)	2006 (12)	10755 (8)	1470 (19)	79 (6)
C (10)	2200 (9)	10483 (6)	414 (19)	64 (5)
C (11)	2390 (9)	9883 (6)	-2055 (14)	42 (4)
C (12)	2567 (11)	10280 (7)	-2905 (16)	64 (6)
C (13)	2088 (13)	10429 (7)	-3802 (16)	74 (6)
C (14)	1448 (12)	10164 (7)	-3980 (18)	75 (6)
C (15)	1252 (12)	9767 (8)	-3151 (18)	73 (6)
C (16)	1728 (11)	9629 (7)	-2236 (15)	65 (6)
C (17)	4705 (7)	8838 (6)	102 (12)	42 (3)
C (18)	5426 (8)	8849 (7)	-51 (12)	47 (4)
C (19)	5914 (9)	8837 (8)	931 (13)	57 (4)
C (20)	5644 (10)	8858 (9)	2160 (13)	71 (6)
C (21)	4926 (10)	8827 (10)	2353 (14)	81 (6)
C (22)	4451 (8)	8875 (7)	1381 (11)	49 (4)
C (24)	2097 (9)	8086 (5)	-2777 (12)	42 (3)
C (25)	1747 (9)	7943 (6)	-3798 (15)	50 (4)
C (26)	1794 (9)	7451 (7)	-4230 (13)	54 (5)
C (27)	2220 (10)	7093 (7)	-3646 (16)	61 (5)
C (28)	2583 (9)	7246 (7)	-2574 (14)	47 (4)
C (29)	2519 (8)	7733 (6)	-2117 (11)	31 (3)
C (30)	2821 (8)	7506 (5)	492 (15)	40 (3)
C (31)	2244 (10)	7164 (7)	479 (19)	65 (5)
C (32)	2103 (14)	6837 (6)	1519 (15)	73 (6)
C (33)	2531 (11)	6850 (7)	2517 (16)	60 (5)
C (34)	3073 (13)	7179 (9)	2576 (16)	89 (8)
C (35)	3250 (10)	7486 (9)	1571 (16)	79 (7)
C (37)	1124 (8)	9020 (20)	240 (70)	37 (4)
C (38)	661 (7)	8651 (6)	1020 (12)	44 (4)
C (45)	1913 (6)	8879 (5)	461 (11)	49 (4)
C (37')	1303 (7)	8521 (6)	180 (14)	37 (4)
C (38')	661 (7)	8651 (6)	1020 (12)	44 (4)
C (45')	1913 (6)	8879 (5)	461 (11)	49 (4)
C (39)	859 (9)	8561 (7)	2391 (14)	49 (4)

N(40)	282(6)	8820(6)	3125(11)	49(3)
C(41)	558(12)	9310(8)	3780(20)	86(7)
C(42)	1247(13)	9369(8)	3160(20)	88(7)
C(43)	1517(8)	8840(8)	2826(12)	58(5)
C(44)	2159(8)	8747(6)	1885(11)	49(4)
O(46)	-722(7)	8236(5)	2796(11)	68(4)
O(47)	-903(7)	9052(6)	3967(11)	80(4)
C(48)	-303(9)	8292(7)	5058(15)	52(4)
C(49)	-216(11)	7788(9)	5113(17)	77(6)
C(50)	-54(12)	7523(8)	6192(18)	78(6)
C(51)	47(9)	7795(8)	7273(14)	55(4)
C(52)	-64(10)	8333(7)	7235(16)	58(5)
C(53)	-246(10)	8589(8)	6160(14)	61(5)
C(54)	311(12)	7549(10)	8450(17)	91(7)
C(55)	912(9)	7976(6)	2697(15)	50(4)
B(61)	5807(10)	8877(10)	5878(15)	55(5)
F(62)	5229(12)	9027(13)	5378(15)	274(17)
F(63)	5618(15)	8352(12)	6010(30)	320(20)
F(64)	5895(6)	9005(5)	7070(8)	81(4)
F(65)	6416(5)	8860(5)	5203(8)	74(3)
C(71)	3568(15)	8934(6)	4550(30)	170(40)
Cl(1)	3407(7)	8268(5)	4416(13)	92(3)
Cl(2)	2892(8)	9217(6)	5431(13)	124(5)

**Table S5.** Bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ] for **4**.

Pt(1) - C(45)	2.150(11)
Pt(1) - P(2)	2.247(3)
Pt(1) - P(1)	2.269(4)
Pt(1) - P(3)	2.308(4)
P(1) - C(11)	1.814(15)
P(1) - C(1)	1.845(16)
P(1) - C(5)	1.854(17)
P(2) - C(17)	1.789(13)
P(2) - C(3)	1.826(14)
P(2) - C(2)	1.842(15)
P(3) - C(30)	1.795(15)
P(3) - C(29)	1.824(13)
P(3) - C(4)	1.847(14)
S(1) - O(46)	1.406(12)
S(1) - O(47)	1.469(13)
S(1) - N(40)	1.605(13)
S(1) - C(48)	1.736(17)
C(1) - C(2)	1.47(2)
C(3) - C(4)	1.49(2)
C(5) - C(10)	1.33(2)
C(5) - C(6)	1.40(2)
C(6) - C(7)	1.38(2)
C(7) - C(8)	1.40(3)
C(8) - C(9)	1.34(3)
C(9) - C(10)	1.38(2)
C(11) - C(12)	1.41(2)
C(11) - C(16)	1.42(2)
C(12) - C(13)	1.37(3)
C(13) - C(14)	1.39(3)



C(14) - C(15)	1.41 (3)
C(15) - C(16)	1.37 (2)
C(17) - C(18)	1.361 (19)
C(17) - C(22)	1.458 (19)
C(18) - C(19)	1.397 (19)
C(19) - C(20)	1.42 (2)
C(20) - C(21)	1.36 (2)
C(21) - C(22)	1.38 (2)
C(24) - C(25)	1.33 (2)
C(24) - C(29)	1.400 (19)
C(25) - C(26)	1.35 (2)
C(26) - C(27)	1.37 (2)
C(27) - C(28)	1.40 (2)
C(28) - C(29)	1.36 (2)
C(30) - C(31)	1.40 (2)
C(30) - C(35)	1.41 (2)
C(31) - C(32)	1.43 (2)
C(32) - C(33)	1.34 (3)
C(33) - C(34)	1.33 (3)
C(34) - C(35)	1.38 (2)
C(37) - C(45)	1.544 (10)
C(37) - C(38)	1.547 (10)
C(38) - C(39)	1.54 (2)
C(45) - C(44)	1.636 (18)
C(39) - C(43)	1.50 (2)
C(39) - N(40)	1.50 (2)
C(39) - C(55)	1.55 (2)
N(40) - C(41)	1.54 (2)
C(41) - C(42)	1.46 (3)
C(42) - C(43)	1.50 (3)
C(43) - C(44)	1.59 (2)
C(48) - C(49)	1.31 (3)
C(48) - C(53)	1.42 (2)
C(49) - C(50)	1.38 (3)
C(50) - C(51)	1.37 (2)
C(51) - C(52)	1.40 (2)
C(51) - C(54)	1.50 (2)
C(52) - C(53)	1.38 (2)
B(61) - F(62)	1.27 (2)
B(61) - F(64)	1.335 (19)
B(61) - F(65)	1.35 (2)
B(61) - F(63)	1.41 (4)
C(71) - Cl(2)	1.742 (10)
C(71) - Cl(1)	1.753 (10)
C(45) - Pt(1) - P(2)	170.5 (4)
C(45) - Pt(1) - P(1)	92.2 (4)
P(2) - Pt(1) - P(1)	84.51 (16)
C(45) - Pt(1) - P(3)	100.6 (4)
P(2) - Pt(1) - P(3)	84.63 (16)
P(1) - Pt(1) - P(3)	162.28 (10)
C(11) - P(1) - C(1)	108.2 (8)
C(11) - P(1) - C(5)	104.7 (7)
C(1) - P(1) - C(5)	105.9 (6)
C(11) - P(1) - Pt(1)	111.2 (5)
C(1) - P(1) - Pt(1)	107.7 (5)
C(5) - P(1) - Pt(1)	118.6 (5)

C(17) - P(2) - C(3)	103.3 (7)
C(17) - P(2) - C(2)	105.9 (7)
C(3) - P(2) - C(2)	117.0 (6)
C(17) - P(2) - Pt(1)	112.9 (4)
C(3) - P(2) - Pt(1)	109.9 (5)
C(2) - P(2) - Pt(1)	107.9 (6)
C(30) - P(3) - C(29)	107.8 (7)
C(30) - P(3) - C(4)	106.4 (7)
C(29) - P(3) - C(4)	102.6 (7)
C(30) - P(3) - Pt(1)	120.7 (5)
C(29) - P(3) - Pt(1)	112.0 (5)
C(4) - P(3) - Pt(1)	105.7 (5)
O(46) - S(1) - O(47)	119.3 (8)
O(46) - S(1) - N(40)	107.8 (8)
O(47) - S(1) - N(40)	106.8 (9)
O(46) - S(1) - C(48)	108.7 (9)
O(47) - S(1) - C(48)	105.1 (8)
N(40) - S(1) - C(48)	108.9 (8)
C(2) - C(1) - P(1)	109.7 (11)
C(1) - C(2) - P(2)	106.8 (11)
C(4) - C(3) - P(2)	105.1 (10)
C(3) - C(4) - P(3)	113.2 (11)
C(10) - C(5) - C(6)	122.1 (17)
C(10) - C(5) - P(1)	122.6 (14)
C(6) - C(5) - P(1)	115.2 (13)
C(7) - C(6) - C(5)	118 (2)
C(6) - C(7) - C(8)	120 (2)
C(9) - C(8) - C(7)	118.4 (18)
C(8) - C(9) - C(10)	123 (2)
C(5) - C(10) - C(9)	118.1 (19)
C(12) - C(11) - C(16)	116.9 (15)
C(12) - C(11) - P(1)	122.9 (14)
C(16) - C(11) - P(1)	120.2 (13)
C(13) - C(12) - C(11)	120.5 (18)
C(12) - C(13) - C(14)	121.3 (16)
C(13) - C(14) - C(15)	119.7 (17)
C(16) - C(15) - C(14)	118 (2)
C(15) - C(16) - C(11)	123.0 (17)
C(18) - C(17) - C(22)	115.8 (12)
C(18) - C(17) - P(2)	124.9 (11)
C(22) - C(17) - P(2)	119.1 (10)
C(17) - C(18) - C(19)	123.9 (13)
C(18) - C(19) - C(20)	118.1 (15)
C(21) - C(20) - C(19)	119.5 (15)
C(20) - C(21) - C(22)	121.0 (15)
C(21) - C(22) - C(17)	120.1 (14)
C(25) - C(24) - C(29)	121.0 (14)
C(24) - C(25) - C(26)	120.7 (15)
C(25) - C(26) - C(27)	120.9 (15)
C(26) - C(27) - C(28)	117.9 (16)
C(29) - C(28) - C(27)	121.3 (15)
C(28) - C(29) - C(24)	118.1 (13)
C(28) - C(29) - P(3)	121.8 (11)
C(24) - C(29) - P(3)	119.9 (11)
C(31) - C(30) - C(35)	115.1 (16)
C(31) - C(30) - P(3)	123.5 (14)
C(35) - C(30) - P(3)	121.4 (12)

C(30) -C(31) -C(32)	120.7 (18)
C(33) -C(32) -C(31)	120.2 (18)
C(34) -C(33) -C(32)	120.8 (16)
C(33) -C(34) -C(35)	120.9 (17)
C(34) -C(35) -C(30)	122.0 (17)
C(45) -C(37) -C(38)	107.3 (13)
C(39) -C(38) -C(37)	119 (3)
C(37) -C(45) -C(44)	118 (3)
C(37) -C(45) -Pt(1)	142 (2)
C(44) -C(45) -Pt(1)	99.7 (8)
C(43) -C(39) -N(40)	102.2 (13)
C(43) -C(39) -C(38)	115.1 (13)
N(40) -C(39) -C(38)	105.3 (12)
C(43) -C(39) -C(55)	110.5 (14)
N(40) -C(39) -C(55)	111.7 (13)
C(38) -C(39) -C(55)	111.5 (13)
C(39) -N(40) -C(41)	111.6 (13)
C(39) -N(40) -S(1)	130.8 (13)
C(41) -N(40) -S(1)	114.8 (11)
C(42) -C(41) -N(40)	99.9 (15)
C(41) -C(42) -C(43)	108.1 (18)
C(39) -C(43) -C(42)	103.7 (15)
C(39) -C(43) -C(44)	110.4 (13)
C(42) -C(43) -C(44)	122.9 (16)
C(43) -C(44) -C(45)	110.7 (11)
C(49) -C(48) -C(53)	119.3 (16)
C(49) -C(48) -S(1)	121.1 (14)
C(53) -C(48) -S(1)	119.5 (14)
C(48) -C(49) -C(50)	123.8 (18)
C(51) -C(50) -C(49)	119.2 (19)
C(50) -C(51) -C(52)	117.5 (17)
C(50) -C(51) -C(54)	123 (2)
C(52) -C(51) -C(54)	119.5 (17)
C(53) -C(52) -C(51)	122.5 (17)
C(52) -C(53) -C(48)	117.5 (17)
F(62) -B(61) -F(64)	115.9 (18)
F(62) -B(61) -F(65)	120.1 (18)
F(64) -B(61) -F(65)	114.9 (16)
F(62) -B(61) -F(63)	97 (3)
F(64) -B(61) -F(63)	100 (2)
F(65) -B(61) -F(63)	103.6 (18)
Cl(2) -C(71) -Cl(1)	109.4 (12)

**Table S6.** Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **4**. The anisotropic displacement factor exponent takes the form:  $-2\pi^2[h^2a^*2U^{11} + \dots + 2hk a^* b^* U^{12}]$ .

	$U_{11}$	$U_{22}$	$U_{33}$	$U_{23}$	$U_{13}$	$U_{12}$
Pt(1)	40 (1)	31 (1)	22 (1)	3 (1)	1 (1)	-4 (1)

P (1)	45 (2)	31 (2)	28 (2)	2 (1)	4 (2)	-6 (2)
P (2)	42 (2)	36 (2)	22 (2)	4 (2)	0 (1)	-3 (2)
P (3)	40 (2)	32 (2)	26 (2)	-2 (1)	-5 (2)	-1 (2)
S (1)	53 (3)	61 (3)	38 (2)	-13 (2)	-2 (2)	8 (2)
C (1)	71 (12)	22 (8)	35 (8)	-6 (6)	10 (8)	-9 (7)
C (2)	83 (13)	49 (11)	19 (7)	11 (7)	-24 (8)	-19 (9)
C (3)	37 (9)	38 (9)	34 (8)	-3 (7)	12 (6)	2 (7)
C (4)	22 (8)	57 (11)	43 (9)	-13 (8)	8 (7)	5 (7)
C (5)	55 (11)	51 (9)	43 (8)	-6 (9)	7 (9)	-12 (7)
C (6)	89 (15)	62 (11)	22 (7)	-3 (7)	12 (8)	-9 (10)
C (7)	140 (20)	39 (10)	66 (12)	-3 (8)	40 (15)	-45 (15)
C (8)	66 (13)	75 (15)	102 (17)	-57 (13)	55 (13)	-27 (11)
C (9)	54 (12)	99 (16)	84 (13)	-53 (12)	-16 (12)	-1 (12)
C (10)	56 (11)	60 (11)	77 (11)	-40 (10)	-21 (12)	-2 (9)
C (11)	49 (10)	42 (10)	34 (8)	-15 (7)	-12 (7)	10 (8)
C (12)	97 (15)	53 (12)	42 (10)	20 (9)	19 (10)	-19 (10)
C (13)	107 (16)	56 (11)	60 (11)	38 (9)	-36 (13)	-22 (13)
C (14)	95 (16)	56 (13)	75 (13)	13 (10)	-37 (12)	22 (12)
C (15)	91 (16)	59 (13)	70 (13)	-6 (10)	-13 (12)	9 (11)
C (16)	94 (15)	65 (13)	35 (9)	30 (9)	-29 (9)	-2 (11)
C (17)	39 (8)	33 (7)	54 (8)	4 (8)	-3 (6)	-7 (8)
C (18)	57 (9)	45 (9)	38 (7)	12 (7)	-7 (6)	-10 (9)
C (19)	55 (10)	66 (11)	51 (8)	4 (9)	-1 (7)	-11 (11)
C (20)	76 (12)	113 (16)	25 (7)	16 (11)	-12 (7)	-34 (14)
C (21)	75 (13)	136 (19)	33 (8)	-9 (13)	3 (8)	-23 (16)
C (22)	58 (9)	67 (11)	22 (6)	-5 (8)	2 (6)	-20 (10)
C (24)	52 (9)	38 (8)	35 (7)	6 (6)	-1 (8)	22 (9)
C (25)	60 (10)	44 (10)	47 (9)	4 (8)	-19 (8)	-4 (8)
C (26)	69 (12)	58 (11)	36 (9)	3 (7)	-15 (7)	6 (9)
C (27)	72 (14)	48 (11)	62 (10)	-19 (8)	-29 (10)	15 (9)
C (28)	50 (10)	53 (11)	39 (8)	4 (8)	-13 (7)	-3 (8)
C (29)	39 (8)	40 (9)	15 (6)	-3 (6)	4 (6)	-1 (7)
C (30)	55 (9)	25 (7)	40 (7)	-4 (7)	12 (9)	2 (7)
C (31)	86 (14)	61 (11)	47 (9)	-6 (9)	-2 (11)	-4 (10)
C (32)	126 (17)	40 (10)	54 (10)	17 (8)	2 (13)	-43 (13)
C (33)	75 (13)	63 (13)	41 (10)	31 (9)	-2 (9)	-4 (11)
C (34)	110 (20)	111 (18)	43 (10)	33 (11)	-35 (11)	-48 (16)
C (35)	66 (13)	130 (20)	45 (11)	40 (12)	-24 (9)	-34 (12)
C (38)	54 (10)	42 (10)	37 (7)	-8 (6)	-15 (7)	5 (7)
C (45)	82 (11)	47 (9)	17 (5)	6 (7)	30 (7)	35 (9)
C (38')	54 (10)	42 (10)	37 (7)	-8 (6)	-15 (7)	5 (7)
C (45')	82 (11)	47 (9)	17 (5)	6 (7)	30 (7)	35 (9)
C (39)	61 (11)	49 (10)	39 (8)	-6 (8)	-4 (7)	-1 (8)
N (40)	45 (7)	50 (8)	52 (7)	7 (8)	4 (6)	-5 (8)
C (41)	96 (17)	59 (14)	104 (16)	-31 (12)	46 (14)	-33 (12)
C (42)	114 (19)	75 (15)	76 (14)	-41 (12)	-43 (14)	1 (14)
C (43)	68 (10)	86 (13)	19 (6)	18 (9)	-4 (6)	-38 (12)
C (44)	56 (10)	65 (11)	27 (6)	3 (7)	10 (7)	-20 (10)
O (46)	69 (8)	71 (9)	64 (8)	-27 (7)	-6 (6)	-9 (7)
O (47)	83 (9)	107 (11)	49 (7)	-21 (7)	-25 (6)	48 (8)
C (48)	47 (10)	52 (11)	55 (10)	-6 (8)	9 (7)	11 (8)
C (49)	77 (14)	107 (19)	46 (11)	-12 (11)	14 (9)	-12 (13)
C (50)	94 (16)	72 (14)	68 (13)	-16 (11)	39 (12)	-24 (12)
C (51)	47 (10)	77 (13)	41 (9)	-6 (9)	8 (7)	-7 (9)
C (52)	77 (13)	52 (12)	44 (10)	-1 (8)	3 (9)	13 (10)
C (53)	67 (12)	74 (13)	43 (9)	-18 (9)	6 (9)	-6 (10)
C (54)	99 (17)	120 (20)	49 (11)	18 (12)	8 (11)	3 (15)

C (55)	57 (11)	49 (10)	46 (9)	15 (8)	-11 (8)	1 (8)
B (61)	54 (11)	76 (15)	37 (9)	-9 (10)	0 (8)	6 (13)
F (62)	230 (20)	530 (50)	60 (9)	-8 (18)	-52 (13)	230 (30)
F (63)	270 (30)	350 (30)	350 (40)	-280 (30)	250 (30)	-250 (30)
F (64)	93 (8)	120 (11)	29 (5)	-27 (5)	-5 (5)	-1 (7)
F (65)	74 (6)	101 (8)	48 (5)	-13 (6)	23 (5)	-13 (7)
C (71)	60 (30)	430 (120)	27 (16)	50 (50)	10 (20)	-50 (40)
Cl (1)	96 (8)	85 (8)	95 (8)	-2 (7)	4 (7)	-1 (6)
Cl (2)	138 (11)	136 (11)	98 (8)	-29 (9)	-45 (11)	13 (10)

**Table S7.** Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **4**.

	x	y	z	U (eq)
H (1A)	3899	10198	-1665	51
H (1B)	4184	9912	-438	51
H (2A)	4724	9519	-2101	60
H (2B)	3969	9445	-2795	60
H (3A)	4044	8253	-2816	43
H (3B)	4788	8200	-2073	43
H (4A)	4245	7782	-448	49
H (4B)	3976	7496	-1680	49
H (6)	3453	9797	1681	69
H (7)	3154	10307	3410	99
H (8)	2210	10907	3260	97
H (9)	1615	10987	1419	95
H (10)	1970	10545	-358	77
H (12)	3020	10444	-2854	77
H (13)	2196	10719	-4313	89
H (14)	1146	10251	-4658	90
H (15)	802	9599	-3222	88
H (16)	1607	9350	-1700	78
H (18)	5609	8866	-873	56
H (19)	6414	8815	780	69
H (20)	5961	8894	2844	85
H (21)	4750	8770	3169	97
H (22)	3959	8933	1544	59
H (24)	2060	8433	-2492	50
H (25)	1461	8188	-4229	61
H (26)	1529	7354	-4945	65
H (27)	2266	6751	-3962	73
H (28)	2880	7004	-2156	57
H (31)	1942	7149	-229	78
H (32)	1705	6609	1503	88
H (33)	2444	6620	3190	71
H (34)	3344	7205	3321	107
H (35)	3672	7690	1607	95
H (37A)	1004	8991	-656	45
H (37B)	1036	9387	494	45
H (38A)	163	8779	996	53

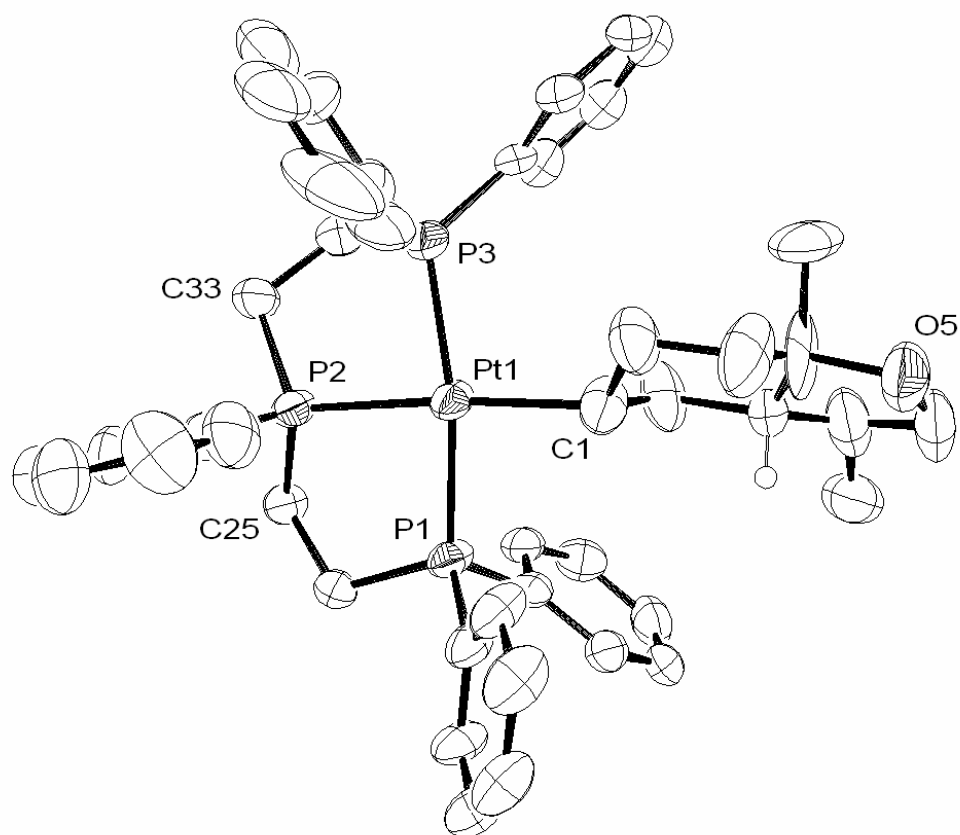
H(38B)	665	8310	599	53
H(37C)	1454	8159	322	45
H(37D)	1164	8557	-704	45
H(38C)	520	9016	897	53
H(38D)	249	8429	793	53
H(41A)	242	9611	3630	104
H(41B)	615	9256	4685	104
H(42A)	1591	9544	3716	106
H(42B)	1192	9581	2398	106
H(43)	1660	8673	3625	69
H(44A)	2317	8382	1936	59
H(44B)	2568	8970	2119	59
H(49)	-268	7594	4368	92
H(50)	-14	7157	6187	93
H(52)	-10	8527	7979	69
H(53)	-331	8951	6157	73
H(54A)	834	7566	8475	136
H(54B)	158	7187	8477	136
H(54C)	114	7734	9167	136
H(55A)	1382	7846	2443	76
H(55B)	537	7788	2249	76
H(55C)	851	7925	3593	76
H(71A)	4036	8993	4954	208
H(71B)	3581	9093	3713	208

---

**Table S8.** Crystallographic data and collection parameters for **8**.

<b>8</b>	
Empirical Formula	C <sub>44</sub> H <sub>50</sub> BF <sub>4</sub> NOP <sub>3</sub> Pt
FW	969.65
Space Group	Orthorhombic, P2 <sub>1</sub> /n
<i>A</i>	10.5334(14)
<i>B</i>	44.022(6)
<i>C</i>	19.544(3)
<i>V</i> , Å <sup>3</sup>	8885(2)
<i>Z</i>	8
<i>T</i> , K	100(2)
<i>D</i> <sub>c</sub> , Mg/m <sup>3</sup>	1.450
$\lambda$ , Å	Mo K $\alpha$ (0.71073)
$\mu$ , mm <sup>-1</sup>	3.314
<i>R</i> <sub>f</sub> <sup>a</sup>	0.0771
<i>R</i> <sub>w</sub> <sup>b</sup>	0.1770
GOF <sup>c</sup>	1.031

<sup>a</sup> $R_f = \Sigma(F_o - F_c)/\Sigma F_o$ . <sup>b</sup> $R_w = [\Sigma\omega(F_o - F_c)^2/\Sigma\omega F_o^2]^{1/2}$ . <sup>c</sup>GOF =  $[\Sigma\omega(F_o - F_c)^2/(n - p)]^{1/2}$ , where *n* = number of reflections and *p* = number of parameters.



**Figure S4.** ORTEP representation of **8**.  $\text{BF}_4^-$  counter ion and solvent molecule not shown.



**Table S9.** Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **8**.  $U(\text{eq})$  is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor.

	x	y	z	U(eq)
Pt(1)	2181(1)	604(1)	3168(1)	43(1)
P(1)	3969(3)	498(1)	3983(2)	41(1)
P(2)	1144(3)	616(1)	4094(2)	42(1)
C(1)	3309(13)	597(3)	2359(7)	61(4)
C(2)	2948(15)	855(3)	1839(7)	72(4)
C(3)	3885(18)	871(4)	1318(10)	111(7)
C(4)	4082(18)	576(4)	1054(8)	100(7)
O(5)	5043(11)	560(3)	609(5)	88(3)
P(5)	7098(4)	1679(1)	10096(2)	62(1)
C(6)	5692(18)	287(5)	697(9)	112(7)
C(7)	4967(18)	92(4)	1166(8)	94(5)
C(8)	4400(12)	333(3)	1538(6)	58(3)
C(9)	3442(15)	300(3)	2040(7)	68(4)
C(10)	2772(15)	520(5)	484(9)	131(9)
C(11)	5814(15)	-151(4)	1552(8)	90(5)
C(12)	4744(10)	146(2)	3789(5)	40(2)
C(13)	3999(10)	-111(3)	3696(5)	45(3)
C(14)	4483(12)	-386(3)	3507(6)	53(3)
C(15)	5784(13)	-388(3)	3406(6)	58(3)
C(16)	6504(11)	-135(3)	3482(5)	45(3)
C(17)	6022(11)	137(3)	3668(6)	48(3)
C(18)	5251(10)	780(2)	4186(6)	45(3)
C(19)	5179(11)	1045(3)	3774(7)	54(3)
C(20)	6153(13)	1257(3)	3918(8)	68(4)
C(21)	7182(13)	1219(3)	4463(8)	67(4)
C(22)	7227(11)	960(3)	4861(7)	57(3)
C(23)	6288(10)	738(3)	4721(6)	50(3)
C(24)	3470(10)	425(3)	4817(5)	45(3)
C(25)	2023(9)	344(2)	4720(5)	41(3)
C(26)	1150(11)	963(3)	4571(6)	53(3)
C(27)	1542(15)	1235(3)	4322(8)	75(4)
C(28)	1478(19)	1505(3)	4665(9)	94(5)
C(29)	1093(17)	1518(4)	5289(9)	85(5)
C(30)	677(14)	1251(4)	5548(8)	73(4)
C(32)	697(12)	977(3)	5194(6)	60(3)
C(33)	-553(11)	512(3)	3806(6)	49(3)
C(34)	-813(12)	391(4)	3045(6)	69(4)
C(35)	-580(16)	990(4)	2470(7)	77(5)
C(36)	203(19)	1240(4)	2561(8)	90(6)
C(37)	-260(30)	1539(5)	2553(10)	158(12)
C(38)	-1670(30)	1553(6)	2442(9)	114(8)
C(39)	-2460(30)	1305(6)	2342(9)	126(9)
C(40)	-1930(16)	1033(5)	2357(8)	104(7)
C(41)	-366(11)	445(4)	1665(6)	64(4)
C(42)	-623(12)	649(4)	1076(7)	83(5)
C(43)	-980(13)	508(5)	415(7)	88(6)
C(44)	-1056(14)	198(5)	352(8)	88(5)
C(45)	-796(13)	19(4)	934(8)	81(5)
C(46)	-437(13)	130(4)	1579(8)	75(4)

Pt(2)	5688(1)	1799(1)	9090(1)	54(1)
P(3)	111(3)	616(1)	2528(2)	61(1)
P(4)	5380(4)	1290(1)	9032(2)	78(1)
C(51)	4302(16)	1880(4)	8147(9)	96(3)
C(52)	4635(17)	2086(4)	7668(9)	103(3)
C(53)	3596(16)	2097(5)	6938(9)	103(3)
C(54)	2255(17)	2161(5)	7074(10)	100(2)
O(55)	1299(11)	2135(3)	6465(6)	97(2)
C(56)	134(16)	2095(4)	6709(9)	96(3)
C(57)	507(16)	1987(4)	7493(9)	98(3)
C(58)	1964(16)	1927(4)	7527(9)	98(2)
C(59)	2867(15)	1908(4)	8263(9)	94(3)
C(60)	2150(30)	2489(5)	7425(13)	168(11)
C(61)	-290(20)	1687(6)	7561(12)	149(9)
C(62)	3761(15)	1128(3)	8695(10)	86(5)
C(63)	3540(16)	1022(4)	7980(11)	112(7)
C(64)	2312(18)	909(4)	7689(13)	117(7)
C(65)	1440(20)	889(4)	8095(14)	115(7)
C(66)	1620(30)	993(5)	8742(15)	129(8)
C(67)	2840(30)	1131(6)	9080(12)	140(9)
C(68)	6383(16)	1051(4)	8704(10)	90(5)
C(69)	6506(18)	731(3)	8880(9)	86(5)
C(70)	7225(18)	550(5)	8570(8)	99(6)
C(71)	7890(14)	668(3)	8022(10)	85(5)
C(72)	7764(18)	962(5)	7829(13)	129(8)
C(73)	6934(14)	1180(3)	8217(7)	72(4)
C(74)	5830(40)	1167(5)	9936(13)	250(20)
C(75)	6690(19)	1302(3)	10382(7)	82(5)
C(76)	8800(16)	1668(3)	10108(8)	70(4)
C(77)	9683(19)	1556(4)	10672(9)	87(5)
C(78)	11000(20)	1559(4)	10708(11)	102(6)
C(79)	11460(20)	1678(5)	10158(13)	117(7)
C(80)	10633(18)	1815(5)	9607(12)	115(7)
C(81)	9313(16)	1801(4)	9589(10)	95(5)
P61	6301(8)	2282(2)	9473(4)	63(1)
C821	7020(40)	1963(9)	10780(20)	63(2)
C831	7120(30)	2258(6)	10427(14)	64(2)
C841	7374(19)	2506(4)	9066(11)	79(2)
C851	7621(19)	2427(4)	8416(11)	78(3)
C861	8389(19)	2613(4)	8091(9)	79(3)
C871	8910(18)	2879(4)	8416(10)	80(3)
C881	8663(19)	2958(4)	9066(10)	81(3)
C891	7890(20)	2772(4)	9390(9)	80(2)
C901	4880(16)	2522(4)	9497(11)	72(2)
C911	4580(20)	2761(5)	9028(11)	75(2)
C921	3490(20)	2938(4)	9036(11)	77(2)
C931	2701(17)	2876(4)	9512(11)	77(2)
C941	3003(18)	2637(4)	9981(10)	76(3)
C951	4093(19)	2460(4)	9973(10)	74(2)
P62	5585(8)	2280(2)	9552(4)	63(1)
C822	6770(30)	1964(10)	10730(20)	63(2)
C832	5660(30)	2193(6)	10524(14)	64(2)
C842	6949(16)	2512(4)	9491(11)	77(2)
C852	7251(18)	2553(4)	8836(9)	79(2)
C862	8280(19)	2739(4)	8759(9)	81(2)
C872	9007(16)	2884(4)	9337(11)	82(3)

C882	8705(18)	2844(4)	9992(9)	83(3)
C892	7676(19)	2658(5)	10069(9)	81(3)
C902	4243(16)	2535(4)	9370(12)	74(2)
C912	4376(17)	2824(4)	9103(13)	75(2)
C922	3340(20)	3025(3)	9008(12)	77(3)
C932	2174(17)	2939(4)	9179(11)	77(3)
C942	2040(15)	2650(4)	9445(11)	77(3)
C952	3075(19)	2448(3)	9541(11)	76(3)
B(1)	4270(20)	1772(4)	1818(15)	93(7)
F(1)	5377(11)	1787(3)	2322(8)	166(6)
F(2)	4017(12)	2068(2)	1651(7)	142(4)
F(3)	4446(11)	1618(3)	1285(8)	134(4)
F(4)	3266(12)	1663(3)	2100(7)	145(5)
B(2)	9505(16)	440(4)	6402(8)	60(4)
F(5)	10681(10)	450(4)	6425(9)	218(9)
F(6)	9094(15)	740(3)	6506(7)	158(5)
F(7)	9176(8)	247(2)	6899(4)	80(2)
F(8)	8704(11)	404(3)	5790(6)	138(5)

**Table S10.** Bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ] for **8**.

Pt(1)-C(1)	2.158(11)
Pt(1)-P(1)	2.262(3)
Pt(1)-P(2)	2.290(3)
Pt(1)-P(3)	2.290(3)
P(1)-C(18)	1.819(12)
P(1)-C(12)	1.827(11)
P(1)-C(24)	1.838(10)
P(2)-C(26)	1.787(13)
P(2)-C(33)	1.825(11)
P(2)-C(25)	1.827(10)
C(1)-C(9)	1.467(17)
C(1)-C(2)	1.522(16)
C(2)-C(3)	1.553(18)
C(3)-C(4)	1.43(2)
C(4)-C(8)	1.423(17)
C(4)-O(5)	1.460(15)
C(4)-C(10)	1.61(3)
O(5)-C(6)	1.374(19)
P(5)-C(76)	1.788(17)
P(5)-C(75)	1.831(13)
P(5)-C821	1.850(5)
P(5)-C822	1.850(5)
P(5)-Pt(2)	2.281(4)
C(6)-C(7)	1.56(2)
C(7)-C(8)	1.479(19)
C(7)-C(11)	1.50(2)
C(8)-C(9)	1.546(16)
C(12)-C(13)	1.367(15)
C(12)-C(17)	1.412(14)
C(13)-C(14)	1.390(16)
C(14)-C(15)	1.423(17)
C(15)-C(16)	1.338(17)
C(16)-C(17)	1.378(16)

C(18)-C(23)	1.368(15)
C(18)-C(19)	1.411(14)
C(19)-C(20)	1.375(17)
C(20)-C(21)	1.372(19)
C(21)-C(22)	1.379(17)
C(22)-C(23)	1.377(16)
C(24)-C(25)	1.540(14)
C(26)-C(27)	1.388(17)
C(26)-C(32)	1.393(16)
C(27)-C(28)	1.37(2)
C(28)-C(29)	1.36(2)
C(29)-C(30)	1.39(2)
C(30)-C(32)	1.395(18)
C(33)-C(34)	1.551(16)
C(34)-P(3)	1.828(14)
C(35)-C(36)	1.36(2)
C(35)-C(40)	1.41(2)
C(35)-P(3)	1.794(16)
C(36)-C(37)	1.40(2)
C(37)-C(38)	1.46(3)
C(38)-C(39)	1.36(3)
C(39)-C(40)	1.32(2)
C(41)-C(46)	1.40(2)
C(41)-C(42)	1.440(19)
C(41)-P(3)	1.825(13)
C(42)-C(43)	1.41(2)
C(43)-C(44)	1.37(2)
C(44)-C(45)	1.37(2)
C(45)-C(46)	1.34(2)
Pt(2)-C(51)	2.144(17)
Pt(2)-P(4)	2.267(4)
Pt(2)-P61	2.301(8)
Pt(2)-P62	2.311(8)
P(4)-C(68)	1.701(17)
P(4)-C(74)	1.82(2)
P(4)-C(62)	1.845(16)
C(51)-C(52)	1.40(2)
C(51)-C(59)	1.58(2)
C(52)-C(53)	1.62(2)
C(53)-C(54)	1.51(2)
C(54)-O(55)	1.40(2)
C(54)-C(58)	1.43(2)
C(54)-C(60)	1.61(3)
O(55)-C(56)	1.412(18)
C(56)-C(57)	1.58(2)
C(57)-C(58)	1.55(2)
C(57)-C(61)	1.59(3)
C(58)-C(59)	1.56(2)
C(62)-C(67)	1.34(3)
C(62)-C(63)	1.45(2)
C(63)-C(64)	1.40(2)
C(64)-C(65)	1.33(3)
C(65)-C(66)	1.32(3)
C(66)-C(67)	1.46(3)
C(68)-C(73)	1.33(2)
C(68)-C(69)	1.45(2)

C(69)-C(70)	1.33(2)
C(70)-C(71)	1.48(2)
C(71)-C(72)	1.35(2)
C(72)-C(73)	1.59(2)
C(74)-C(75)	1.28(2)
C(76)-C(81)	1.37(2)
C(76)-C(77)	1.39(2)
C(77)-C(78)	1.37(3)
C(78)-C(79)	1.37(3)
C(79)-C(80)	1.38(3)
C(80)-C(81)	1.38(2)
P61-C841	1.802(15)
P61-C901	1.841(15)
P61-C831	1.90(3)
C821-C831	1.49(6)
C841-C851	1.3900
C841-C891	1.3900
C851-C861	1.3900
C861-C871	1.3900
C871-C881	1.3900
C881-C891	1.3900
C901-C911	1.3900
C901-C951	1.3900
C911-C921	1.3900
C921-C931	1.3900
C931-C941	1.3900
C941-C951	1.3900
P62-C842	1.786(15)
P62-C902	1.786(15)
P62-C832	1.93(3)
C822-C832	1.53(5)
C842-C852	1.3900
C842-C892	1.3900
C852-C862	1.3900
C862-C872	1.3900
C872-C882	1.3900
C882-C892	1.3900
C902-C912	1.3900
C902-C952	1.3900
C912-C922	1.3900
C922-C932	1.3900
C932-C942	1.3900
C942-C952	1.3900
B(1)-F(3)	1.29(2)
B(1)-F(2)	1.36(2)
B(1)-F(1)	1.37(3)
B(1)-F(4)	1.37(2)
B(2)-F(5)	1.232(18)
B(2)-F(8)	1.331(18)
B(2)-F(7)	1.385(16)
B(2)-F(6)	1.417(19)
C(1)-Pt(1)-P(1)	90.6(4)
C(1)-Pt(1)-P(2)	175.2(4)
P(1)-Pt(1)-P(2)	84.80(10)

C(1)-Pt(1)-P(3)	101.7(4)
P(1)-Pt(1)-P(3)	163.75(11)
P(2)-Pt(1)-P(3)	83.15(10)
C(18)-P(1)-C(12)	106.3(5)
C(18)-P(1)-C(24)	104.8(5)
C(12)-P(1)-C(24)	104.6(5)
C(18)-P(1)-Pt(1)	119.5(4)
C(12)-P(1)-Pt(1)	112.0(4)
C(24)-P(1)-Pt(1)	108.5(4)
C(26)-P(2)-C(33)	106.2(6)
C(26)-P(2)-C(25)	105.1(5)
C(33)-P(2)-C(25)	112.2(5)
C(26)-P(2)-Pt(1)	118.6(4)
C(33)-P(2)-Pt(1)	109.9(4)
C(25)-P(2)-Pt(1)	104.8(3)
C(9)-C(1)-C(2)	114.4(11)
C(9)-C(1)-Pt(1)	116.0(8)
C(2)-C(1)-Pt(1)	112.2(8)
C(1)-C(2)-C(3)	111.3(11)
C(4)-C(3)-C(2)	110.7(13)
C(8)-C(4)-C(3)	118.3(14)
C(8)-C(4)-O(5)	105.0(12)
C(3)-C(4)-O(5)	115.3(14)
C(8)-C(4)-C(10)	113.6(16)
C(3)-C(4)-C(10)	103.0(15)
O(5)-C(4)-C(10)	100.4(13)
C(6)-O(5)-C(4)	110.8(11)
C(76)-P(5)-C(75)	105.5(8)
C(76)-P(5)-C821	101.4(13)
C(75)-P(5)-C821	110.2(17)
C(76)-P(5)-C822	109.4(12)
C(75)-P(5)-C822	108.8(18)
C821-P(5)-C822	8(2)
C(76)-P(5)-Pt(2)	119.9(5)
C(75)-P(5)-Pt(2)	108.8(6)
C821-P(5)-Pt(2)	110.7(16)
C822-P(5)-Pt(2)	104.2(15)
O(5)-C(6)-C(7)	105.3(12)
C(8)-C(7)-C(11)	121.6(13)
C(8)-C(7)-C(6)	100.6(14)
C(11)-C(7)-C(6)	112.1(15)
C(4)-C(8)-C(7)	106.1(11)
C(4)-C(8)-C(9)	113.3(11)
C(7)-C(8)-C(9)	128.3(12)
C(1)-C(9)-C(8)	108.3(10)
C(13)-C(12)-C(17)	119.8(10)
C(13)-C(12)-P(1)	117.5(8)
C(17)-C(12)-P(1)	122.5(9)
C(12)-C(13)-C(14)	121.6(10)
C(13)-C(14)-C(15)	117.0(11)
C(16)-C(15)-C(14)	121.4(11)
C(15)-C(16)-C(17)	121.4(11)
C(16)-C(17)-C(12)	118.7(11)
C(23)-C(18)-C(19)	119.5(11)
C(23)-C(18)-P(1)	121.5(8)
C(19)-C(18)-P(1)	119.0(9)

C(20)-C(19)-C(18)	119.4(12)
C(21)-C(20)-C(19)	121.4(12)
C(20)-C(21)-C(22)	118.2(12)
C(23)-C(22)-C(21)	122.1(13)
C(18)-C(23)-C(22)	119.5(11)
C(25)-C(24)-P(1)	112.3(7)
C(24)-C(25)-P(2)	106.9(7)
C(27)-C(26)-C(32)	116.4(12)
C(27)-C(26)-P(2)	121.6(10)
C(32)-C(26)-P(2)	121.9(9)
C(28)-C(27)-C(26)	122.1(14)
C(29)-C(28)-C(27)	121.7(15)
C(28)-C(29)-C(30)	117.6(14)
C(29)-C(30)-C(32)	121.1(13)
C(26)-C(32)-C(30)	120.9(12)
C(34)-C(33)-P(2)	111.0(8)
C(33)-C(34)-P(3)	109.4(9)
C(36)-C(35)-C(40)	118.6(17)
C(36)-C(35)-P(3)	120.2(12)
C(40)-C(35)-P(3)	121.2(17)
C(35)-C(36)-C(37)	124(2)
C(36)-C(37)-C(38)	113(2)
C(39)-C(38)-C(37)	124(2)
C(40)-C(39)-C(38)	119(2)
C(39)-C(40)-C(35)	122(2)
C(46)-C(41)-C(42)	121.6(13)
C(46)-C(41)-P(3)	121.2(11)
C(42)-C(41)-P(3)	117.1(13)
C(43)-C(42)-C(41)	115.6(17)
C(44)-C(43)-C(42)	121.2(15)
C(45)-C(44)-C(43)	120.1(16)
C(46)-C(45)-C(44)	123.1(18)
C(45)-C(46)-C(41)	118.3(15)
C(51)-Pt(2)-P(4)	93.1(5)
C(51)-Pt(2)-P(5)	175.8(5)
P(4)-Pt(2)-P(5)	82.86(13)
C(51)-Pt(2)-P61	103.1(6)
P(4)-Pt(2)-P61	163.3(2)
P(5)-Pt(2)-P61	80.8(2)
C(51)-Pt(2)-P62	96.2(6)
P(4)-Pt(2)-P62	155.3(2)
P(5)-Pt(2)-P62	87.0(2)
P61-Pt(2)-P62	19.9(2)
C(35)-P(3)-C(41)	106.6(7)
C(35)-P(3)-C(34)	106.2(7)
C(41)-P(3)-C(34)	102.0(7)
C(35)-P(3)-Pt(1)	112.7(6)
C(41)-P(3)-Pt(1)	123.7(4)
C(34)-P(3)-Pt(1)	103.8(4)
C(68)-P(4)-C(74)	97.0(15)
C(68)-P(4)-C(62)	103.1(8)
C(74)-P(4)-C(62)	106.0(12)
C(68)-P(4)-Pt(2)	122.4(6)
C(74)-P(4)-Pt(2)	103.8(8)
C(62)-P(4)-Pt(2)	120.9(4)
C(52)-C(51)-C(59)	114.9(14)

C(52)-C(51)-Pt(2)	117.8(13)
C(59)-C(51)-Pt(2)	113.5(11)
C(51)-C(52)-C(53)	113.3(16)
C(54)-C(53)-C(52)	110.1(15)
O(55)-C(54)-C(58)	105.2(16)
O(55)-C(54)-C(53)	112.0(15)
C(58)-C(54)-C(53)	107.3(15)
O(55)-C(54)-C(60)	109.6(15)
C(58)-C(54)-C(60)	109.9(18)
C(53)-C(54)-C(60)	112.6(18)
C(54)-O(55)-C(56)	104.5(13)
O(55)-C(56)-C(57)	107.4(13)
C(58)-C(57)-C(56)	98.4(13)
C(58)-C(57)-C(61)	113.3(17)
C(56)-C(57)-C(61)	107.2(16)
C(54)-C(58)-C(57)	100.6(14)
C(54)-C(58)-C(59)	116.3(16)
C(57)-C(58)-C(59)	117.8(14)
C(58)-C(59)-C(51)	107.3(13)
C(67)-C(62)-C(63)	123.4(18)
C(67)-C(62)-P(4)	120.9(17)
C(63)-C(62)-P(4)	115.5(15)
C(64)-C(63)-C(62)	117.8(19)
C(65)-C(64)-C(63)	118(2)
C(66)-C(65)-C(64)	124(2)
C(65)-C(66)-C(67)	122(2)
C(62)-C(67)-C(66)	114(2)
C(73)-C(68)-C(69)	123.5(16)
C(73)-C(68)-P(4)	113.3(13)
C(69)-C(68)-P(4)	122.8(14)
C(70)-C(69)-C(68)	120.6(19)
C(69)-C(70)-C(71)	120.8(18)
C(72)-C(71)-C(70)	120.3(17)
C(71)-C(72)-C(73)	119(2)
C(68)-C(73)-C(72)	116.0(16)
C(75)-C(74)-P(4)	122.6(17)
C(74)-C(75)-P(5)	113.1(13)
C(81)-C(76)-C(77)	116.1(16)
C(81)-C(76)-P(5)	121.3(12)
C(77)-C(76)-P(5)	122.3(13)
C(78)-C(77)-C(76)	123.3(18)
C(79)-C(78)-C(77)	118.5(18)
C(78)-C(79)-C(80)	120(2)
C(81)-C(80)-C(79)	119(2)
C(76)-C(81)-C(80)	122.7(18)
C841-P61-C901	106.2(10)
C841-P61-C831	105.1(12)
C901-P61-C831	102.7(12)
C841-P61-Pt(2)	121.6(8)
C901-P61-Pt(2)	111.2(8)
C831-P61-Pt(2)	108.3(8)
C831-C821-P(5)	103(2)
C821-C831-P61	116.5(19)
C851-C841-C891	120.0
C851-C841-P61	120.8(12)
C891-C841-P61	119.1(11)



C861-C851-C841	120.0
C851-C861-C871	120.0
C881-C871-C861	120.0
C891-C881-C871	120.0
C881-C891-C841	120.0
C911-C901-C951	120.0
C911-C901-P61	120.0(11)
C951-C901-P61	120.0(11)
C901-C911-C921	120.0
C911-C921-C931	120.0
C941-C931-C921	120.0
C931-C941-C951	120.0
C941-C951-C901	120.0
C842-P62-C902	104.0(10)
C842-P62-C832	107.7(12)
C902-P62-C832	101.4(11)
C842-P62-Pt(2)	113.4(8)
C902-P62-Pt(2)	126.4(8)
C832-P62-Pt(2)	102.0(8)
C832-C822-P(5)	120(3)
C822-C832-P62	106(2)
C852-C842-C892	120.0
C852-C842-P62	117.9(12)
C892-C842-P62	122.1(12)
C862-C852-C842	120.0
C852-C862-C872	120.0
C882-C872-C862	120.0
C872-C882-C892	120.0
C882-C892-C842	120.0
C912-C902-C952	120.0
C912-C902-P62	121.0(11)
C952-C902-P62	118.9(11)
C922-C912-C902	120.0
C932-C922-C912	120.0
C922-C932-C942	120.0
C952-C942-C932	120.0
C942-C952-C902	120.0
F(3)-B(1)-F(2)	111(2)
F(3)-B(1)-F(1)	111.7(16)
F(2)-B(1)-F(1)	103.1(18)
F(3)-B(1)-F(4)	112.7(18)
F(2)-B(1)-F(4)	107.5(15)
F(1)-B(1)-F(4)	110(2)
F(5)-B(2)-F(8)	119.5(16)
F(5)-B(2)-F(7)	112.7(13)
F(8)-B(2)-F(7)	110.8(14)
F(5)-B(2)-F(6)	107.1(17)
F(8)-B(2)-F(6)	94.7(12)
F(7)-B(2)-F(6)	110.3(12)

---

**Table S11.** Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **8**. The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
Pt(1)	32(1)	61(1)	39(1)	21(1)	9(1)	12(1)
P(1)	27(1)	52(2)	43(2)	18(1)	6(1)	3(1)
P(2)	28(1)	60(2)	38(2)	16(1)	6(1)	2(1)
C(1)	65(8)	64(8)	60(8)	29(6)	27(7)	26(7)
C(2)	89(11)	77(10)	58(8)	14(7)	33(8)	8(8)
C(3)	109(14)	136(16)	110(14)	78(13)	72(12)	46(12)
C(4)	138(15)	106(12)	85(11)	68(10)	95(12)	56(11)
O(5)	81(7)	121(9)	75(7)	21(6)	43(6)	16(7)
P(5)	89(3)	50(2)	49(2)	1(2)	14(2)	-9(2)
C(6)	101(13)	190(20)	63(10)	15(11)	50(10)	44(14)
C(7)	106(13)	130(15)	57(9)	4(9)	41(10)	20(11)
C(8)	47(7)	92(10)	39(7)	4(6)	15(6)	9(7)
C(9)	107(11)	56(8)	54(8)	20(6)	42(8)	4(8)
C(10)	45(9)	250(30)	89(13)	98(15)	-4(9)	10(12)
C(11)	80(11)	116(13)	64(10)	-24(9)	-7(8)	38(10)
C(12)	34(6)	53(7)	30(6)	15(5)	2(5)	10(5)
C(13)	31(6)	67(8)	36(6)	11(5)	2(5)	1(5)
C(14)	55(7)	51(7)	49(7)	10(6)	-1(6)	4(6)
C(15)	69(9)	65(8)	41(7)	-3(6)	15(6)	21(7)
C(16)	45(6)	61(8)	29(6)	4(5)	8(5)	4(6)
C(17)	39(6)	70(8)	36(6)	12(5)	8(5)	-3(6)
C(18)	35(6)	49(6)	52(7)	15(5)	12(5)	10(5)
C(19)	39(6)	54(7)	73(9)	31(6)	20(6)	9(6)
C(20)	61(9)	48(8)	100(11)	23(7)	30(9)	-4(7)
C(21)	47(8)	56(8)	101(11)	3(8)	26(8)	-1(6)
C(22)	40(7)	57(8)	73(9)	4(6)	8(6)	13(6)
C(23)	33(6)	55(7)	59(8)	24(6)	3(6)	10(5)
C(24)	39(6)	62(7)	31(6)	7(5)	3(5)	8(5)
C(25)	35(6)	54(7)	35(6)	20(5)	10(5)	-4(5)
C(26)	47(7)	60(8)	51(7)	11(6)	9(6)	-7(6)
C(27)	84(10)	88(11)	60(9)	10(8)	27(8)	-6(8)
C(28)	138(16)	44(8)	98(13)	-7(8)	22(12)	-4(9)
C(29)	99(12)	75(11)	88(12)	-20(9)	31(10)	13(9)
C(30)	66(9)	93(11)	62(9)	-12(8)	18(8)	7(8)
C(32)	61(8)	71(9)	54(8)	5(6)	26(7)	2(7)
C(33)	38(6)	67(8)	40(6)	2(5)	3(5)	-6(5)
C(34)	36(7)	118(12)	50(8)	-13(8)	1(6)	2(7)
C(35)	77(10)	118(13)	33(7)	5(8)	7(7)	52(10)
C(36)	107(14)	93(12)	56(9)	1(8)	-17(9)	66(11)
C(37)	240(30)	123(17)	75(13)	-9(11)	-56(16)	99(19)
C(38)	150(20)	128(18)	53(11)	0(11)	-18(12)	77(16)
C(39)	160(20)	170(20)	46(10)	-2(13)	3(12)	100(20)
C(40)	83(11)	173(18)	57(9)	18(10)	16(9)	85(12)
C(41)	27(6)	124(13)	38(7)	-5(7)	-3(5)	26(7)
C(42)	46(7)	154(15)	47(8)	22(9)	6(6)	47(9)
C(43)	52(8)	173(18)	36(8)	8(9)	0(7)	57(10)
C(44)	59(9)	145(16)	60(10)	-15(11)	11(8)	34(11)
C(45)	53(8)	130(14)	63(10)	4(10)	16(8)	29(9)
C(46)	54(8)	114(13)	57(9)	2(8)	15(7)	28(8)
Pt(2)	68(1)	50(1)	49(1)	7(1)	23(1)	5(1)

P(3)	40(2)	105(3)	39(2)	16(2)	7(1)	29(2)
P(4)	72(3)	56(2)	92(3)	-10(2)	-19(2)	10(2)
C(51)	72(5)	124(6)	94(6)	37(5)	20(5)	18(5)
C(52)	78(5)	134(6)	98(6)	33(5)	18(5)	10(5)
C(53)	78(5)	134(6)	96(6)	32(5)	17(5)	9(5)
C(54)	76(4)	128(5)	94(5)	34(4)	15(4)	12(4)
O(55)	75(4)	124(5)	91(5)	28(4)	15(4)	9(4)
C(56)	73(5)	122(6)	94(6)	34(5)	18(5)	17(5)
C(57)	72(5)	126(6)	97(6)	39(5)	20(5)	18(5)
C(58)	73(4)	125(5)	95(5)	41(4)	17(4)	17(4)
C(59)	70(5)	121(6)	92(6)	40(5)	20(5)	20(5)
C(60)	220(30)	90(15)	170(20)	-40(15)	-40(20)	6(16)
C(61)	119(18)	180(20)	130(20)	42(17)	-2(15)	-32(17)
C(62)	72(10)	37(7)	134(16)	-16(8)	-20(10)	21(7)
C(63)	62(10)	129(16)	139(17)	-75(14)	5(11)	-5(10)
C(64)	85(13)	94(13)	170(20)	-61(14)	25(14)	-10(11)
C(65)	150(20)	70(12)	140(20)	-14(12)	57(18)	-31(12)
C(66)	140(20)	99(16)	160(20)	54(15)	63(19)	-19(14)
C(67)	150(20)	180(20)	112(17)	39(16)	58(17)	-30(19)
C(68)	70(10)	81(11)	117(14)	-13(10)	11(10)	-8(9)
C(69)	114(13)	58(9)	75(10)	-17(8)	-10(10)	14(9)
C(70)	104(13)	146(16)	57(10)	1(10)	41(10)	21(12)
C(71)	60(9)	64(10)	127(15)	-34(9)	9(9)	5(7)
C(72)	79(13)	97(15)	200(20)	17(15)	-3(14)	30(11)
C(73)	82(10)	77(10)	63(9)	-11(7)	29(8)	-19(8)
C(74)	410(50)	127(19)	130(20)	76(16)	-170(30)	-130(30)
C(75)	159(16)	47(8)	45(8)	0(6)	35(10)	-12(9)
C(76)	91(11)	49(7)	65(9)	-1(7)	1(9)	5(7)
C(77)	106(14)	79(11)	71(11)	1(8)	2(10)	10(10)
C(78)	99(15)	103(14)	84(13)	4(11)	-37(12)	8(11)
C(79)	81(13)	139(18)	126(19)	-15(15)	3(14)	9(13)
C(80)	69(11)	156(19)	118(17)	37(14)	14(11)	19(12)
C(81)	65(10)	124(15)	96(13)	10(11)	18(10)	6(10)
P61	62(3)	49(2)	80(3)	7(2)	20(3)	5(3)
C821	62(4)	66(3)	60(3)	-6(3)	9(3)	-6(3)
C831	63(4)	66(3)	61(3)	-6(3)	8(3)	-6(3)
C841	70(5)	70(4)	97(5)	6(4)	19(4)	-8(4)
C851	69(5)	69(5)	97(6)	6(5)	19(5)	-8(4)
C861	70(5)	69(5)	98(6)	7(5)	19(5)	-8(5)
C871	71(5)	70(5)	99(6)	7(5)	19(5)	-9(5)
C881	72(5)	71(5)	100(6)	6(5)	17(5)	-9(4)
C891	72(4)	70(4)	98(5)	5(4)	18(4)	-8(4)
C901	75(4)	57(4)	86(4)	4(4)	23(4)	13(4)
C911	79(5)	60(5)	88(5)	4(4)	23(4)	14(4)
C921	81(5)	62(5)	89(5)	5(4)	24(4)	16(5)
C931	81(5)	63(5)	89(5)	6(4)	24(4)	15(5)
C941	80(5)	62(5)	89(5)	5(4)	25(5)	15(5)
C951	78(5)	60(5)	88(5)	5(4)	25(4)	13(5)
P62	62(3)	49(2)	80(3)	7(2)	20(3)	5(3)
C822	62(4)	66(3)	60(3)	-6(3)	9(3)	-6(3)
C832	63(4)	66(3)	61(3)	-6(3)	8(3)	-6(3)
C842	70(4)	67(4)	95(5)	5(4)	17(4)	-6(4)
C852	70(5)	71(5)	98(5)	6(4)	18(4)	-8(4)
C862	72(5)	72(4)	99(5)	6(4)	17(4)	-9(4)
C872	73(5)	72(5)	101(6)	5(5)	16(5)	-9(4)
C882	74(5)	73(5)	101(6)	5(5)	15(5)	-8(5)

C892	73(5)	71(5)	99(6)	5(5)	16(5)	-7(4)
C902	79(5)	59(4)	87(5)	4(4)	23(4)	14(4)
C912	79(5)	61(4)	88(5)	5(4)	23(4)	14(4)
C922	82(5)	63(5)	89(5)	6(4)	24(5)	15(5)
C932	82(5)	63(5)	90(5)	6(5)	24(5)	15(5)
C942	81(5)	63(5)	90(5)	5(5)	24(5)	15(5)
C952	80(5)	61(5)	89(5)	6(4)	23(5)	14(5)
B(1)	79(14)	54(11)	160(20)	2(12)	53(15)	19(10)
F(1)	74(7)	220(15)	186(13)	-32(11)	-19(8)	68(8)
F(2)	142(10)	79(7)	203(13)	18(7)	35(9)	30(7)
F(3)	109(8)	113(8)	194(13)	-28(8)	60(9)	1(7)
F(4)	126(10)	133(9)	184(12)	6(9)	53(9)	-43(8)
B(2)	57(10)	75(11)	47(9)	13(8)	12(8)	23(8)
F(5)	57(6)	350(20)	257(17)	226(16)	56(9)	51(9)
F(6)	180(13)	184(13)	125(10)	26(9)	71(10)	50(11)
F(7)	66(5)	94(6)	86(6)	26(5)	29(5)	11(4)
F(8)	130(9)	135(9)	122(9)	60(7)	-44(8)	-60(8)

**Table S12.** Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **8**.

	x	y	z	U(eq)
H(1)	4207	647	2608	73
H(2A)	2976	1050	2094	87
H(2B)	2052	824	1577	87
H(3A)	3520	1007	926	134
H(3B)	4727	956	1557	134
H(6A)	6607	318	928	135
H(6B)	5667	187	241	135
H(7)	4234	-12	849	113
H(8)	5185	415	1857	70
H(9A)	2589	230	1779	82
H(9B)	3771	148	2406	82
H(10A)	2808	319	270	196
H(10B)	2687	677	123	196
H(10C)	2024	528	713	196
H(11A)	6588	-58	1835	134
H(11B)	6072	-293	1217	134
H(11C)	5336	-261	1856	134
H(13)	3130	-102	3761	54
H(14)	3967	-565	3448	64
H(15)	6149	-572	3282	69
H(16)	7367	-144	3407	54
H(17)	6539	316	3714	57
H(19)	4464	1076	3400	65
H(20)	6112	1433	3634	82
H(21)	7845	1368	4563	80
H(22)	7928	933	5245	68
H(23)	6360	558	4995	60
H(24A)	3991	255	5060	54
H(24B)	3648	607	5117	54
H(25A)	1731	359	5170	49

H(25B)	1869	134	4540	49
H(27)	1864	1235	3901	90
H(28)	1708	1688	4460	112
H(29)	1109	1704	5539	103
H(30)	373	1255	5975	88
H(32)	397	797	5380	72
H(33A)	-784	353	4119	59
H(33B)	-1106	691	3835	59
H(34A)	-1749	405	2841	83
H(34B)	-553	175	3041	83
H(36)	1113	1209	2635	108
H(37)	282	1713	2613	189
H(38)	-2064	1748	2438	137
H(39)	-3375	1328	2264	152
H(40)	-2478	860	2289	125
H(42)	-557	863	1128	99
H(43)	-1171	631	8	106
H(44)	-1290	108	-97	106
H(45)	-874	-195	876	98
H(46)	-236	-2	1971	90
H(51)	4288	1682	7896	115
H(52A)	5492	2029	7571	124
H(52B)	4713	2291	7878	124
H(53A)	3843	2258	6636	123
H(53B)	3596	1901	6692	123
H(56A)	-413	1940	6424	116
H(56B)	-354	2288	6677	116
H(57)	354	2149	7828	117
H(58)	2036	1729	7288	117
H(59A)	2639	1729	8521	113
H(59B)	2771	2093	8538	113
H(60A)	2091	2647	7067	252
H(60B)	2922	2523	7788	252
H(60C)	1377	2495	7632	252
H(61A)	-377	1568	7131	223
H(61B)	-1159	1742	7636	223
H(61C)	155	1567	7957	223
H(63)	4208	1029	7716	134
H(64)	2106	848	7214	141
H(65)	636	794	7909	138
H(66)	941	976	8994	155
H(67)	2976	1217	9535	168
H(69)	6067	651	9221	104
H(70)	7316	342	8701	119
H(71)	8407	536	7807	102
H(72)	8164	1037	7467	155
H(73)	6835	1391	8109	86
H(74A)	5022	1168	10127	303
H(74B)	6098	952	9927	303
H(75A)	7483	1175	10474	98
H(75B)	6379	1322	10826	98
H(77)	9363	1473	11053	105
H(78)	11573	1481	11105	123
H(79)	12362	1666	10154	141
H(80)	10961	1917	9250	138
H(81)	8740	1887	9200	114

H82A1	6189	1948	10947	76
H82B1	7743	1937	11186	76
H83A1	8050	2306	10463	77
H83B1	6748	2418	10684	77
H851	7265	2246	8194	94
H861	8558	2559	7647	94
H871	9435	3006	8194	96
H881	9019	3139	9288	97
H891	7726	2826	9835	96
H911	5116	2803	8703	90
H921	3282	3101	8716	92
H931	1957	2997	9517	92
H941	2465	2595	10306	91
H951	4299	2297	10293	89
H82A2	6600	1852	11146	76
H82B2	7570	2082	10888	76
H83A2	5831	2380	10807	77
H83B2	4828	2105	10595	77
H852	6755	2453	8441	95
H862	8487	2766	8311	97
H872	9710	3011	9285	98
H882	9201	2943	10387	100
H892	7469	2630	10516	97
H912	5174	2883	8987	90
H922	3432	3223	8826	92
H932	1466	3077	9114	93
H942	1242	2591	9562	92
H952	2984	2251	9723	91

**Table S13.** Torsion angles [ $^{\circ}$ ] for **8**.

C(1)-Pt(1)-P(1)-C(18)	67.9(5)
P(2)-Pt(1)-P(1)-C(18)	-110.6(4)
P(3)-Pt(1)-P(1)-C(18)	-152.9(5)
C(1)-Pt(1)-P(1)-C(12)	-57.2(5)
P(2)-Pt(1)-P(1)-C(12)	124.2(4)
P(3)-Pt(1)-P(1)-C(12)	81.9(6)
C(1)-Pt(1)-P(1)-C(24)	-172.2(5)
P(2)-Pt(1)-P(1)-C(24)	9.3(4)
P(3)-Pt(1)-P(1)-C(24)	-33.0(6)
C(1)-Pt(1)-P(2)-C(26)	68(4)
P(1)-Pt(1)-P(2)-C(26)	85.8(5)
P(3)-Pt(1)-P(2)-C(26)	-105.1(5)
C(1)-Pt(1)-P(2)-C(33)	-169(4)
P(1)-Pt(1)-P(2)-C(33)	-151.8(4)
P(3)-Pt(1)-P(2)-C(33)	17.3(4)
C(1)-Pt(1)-P(2)-C(25)	-49(4)
P(1)-Pt(1)-P(2)-C(25)	-31.1(4)
P(3)-Pt(1)-P(2)-C(25)	138.0(4)
P(1)-Pt(1)-C(1)-C(9)	88.4(10)
P(2)-Pt(1)-C(1)-C(9)	106(4)
P(3)-Pt(1)-C(1)-C(9)	-80.9(10)
P(1)-Pt(1)-C(1)-C(2)	-137.6(11)
P(2)-Pt(1)-C(1)-C(2)	-120(4)

P(3)-Pt(1)-C(1)-C(2)	53.1(11)
C(9)-C(1)-C(2)-C(3)	-52.6(19)
Pt(1)-C(1)-C(2)-C(3)	172.7(12)
C(1)-C(2)-C(3)-C(4)	46(2)
C(2)-C(3)-C(4)-C(8)	-49(2)
C(2)-C(3)-C(4)-O(5)	-174.5(14)
C(2)-C(3)-C(4)-C(10)	77.2(15)
C(8)-C(4)-O(5)-C(6)	14(2)
C(3)-C(4)-O(5)-C(6)	146.0(17)
C(10)-C(4)-O(5)-C(6)	-104.1(16)
C(4)-O(5)-C(6)-C(7)	8(2)
O(5)-C(6)-C(7)-C(8)	-25.7(18)
O(5)-C(6)-C(7)-C(11)	-156.4(14)
C(3)-C(4)-C(8)-C(7)	-161.4(16)
O(5)-C(4)-C(8)-C(7)	-31.1(19)
C(10)-C(4)-C(8)-C(7)	77.6(16)
C(3)-C(4)-C(8)-C(9)	53(2)
O(5)-C(4)-C(8)-C(9)	-176.9(13)
C(10)-C(4)-C(8)-C(9)	-68.2(17)
C(11)-C(7)-C(8)-C(4)	159.1(17)
C(6)-C(7)-C(8)-C(4)	34.7(18)
C(11)-C(7)-C(8)-C(9)	-62(2)
C(6)-C(7)-C(8)-C(9)	173.6(14)
C(2)-C(1)-C(9)-C(8)	53.4(16)
Pt(1)-C(1)-C(9)-C(8)	-173.6(9)
C(4)-C(8)-C(9)-C(1)	-51.7(18)
C(7)-C(8)-C(9)-C(1)	171.7(14)
C(18)-P(1)-C(12)-C(13)	172.8(8)
C(24)-P(1)-C(12)-C(13)	62.3(9)
Pt(1)-P(1)-C(12)-C(13)	-55.0(9)
C(18)-P(1)-C(12)-C(17)	-13.5(10)
C(24)-P(1)-C(12)-C(17)	-124.0(9)
Pt(1)-P(1)-C(12)-C(17)	118.7(8)
C(17)-C(12)-C(13)-C(14)	2.4(16)
P(1)-C(12)-C(13)-C(14)	176.2(8)
C(12)-C(13)-C(14)-C(15)	-0.8(16)
C(13)-C(14)-C(15)-C(16)	-0.7(17)
C(14)-C(15)-C(16)-C(17)	0.6(18)
C(15)-C(16)-C(17)-C(12)	0.9(16)
C(13)-C(12)-C(17)-C(16)	-2.4(15)
P(1)-C(12)-C(17)-C(16)	-176.0(8)
C(12)-P(1)-C(18)-C(23)	-58.8(10)
C(24)-P(1)-C(18)-C(23)	51.6(11)
Pt(1)-P(1)-C(18)-C(23)	173.3(8)
C(12)-P(1)-C(18)-C(19)	120.3(9)
C(24)-P(1)-C(18)-C(19)	-129.2(9)
Pt(1)-P(1)-C(18)-C(19)	-7.5(11)
C(23)-C(18)-C(19)-C(20)	0.0(18)
P(1)-C(18)-C(19)-C(20)	-179.2(10)
C(18)-C(19)-C(20)-C(21)	-1(2)
C(19)-C(20)-C(21)-C(22)	1(2)
C(20)-C(21)-C(22)-C(23)	0.9(19)
C(19)-C(18)-C(23)-C(22)	1.7(17)
P(1)-C(18)-C(23)-C(22)	-179.1(9)
C(21)-C(22)-C(23)-C(18)	-2.2(18)
C(18)-P(1)-C(24)-C(25)	148.3(8)

C(12)-P(1)-C(24)-C(25)	-100.0(8)
Pt(1)-P(1)-C(24)-C(25)	19.7(9)
P(1)-C(24)-C(25)-P(2)	-45.2(10)
C(26)-P(2)-C(25)-C(24)	-75.0(8)
C(33)-P(2)-C(25)-C(24)	170.0(7)
Pt(1)-P(2)-C(25)-C(24)	50.7(8)
C(33)-P(2)-C(26)-C(27)	-111.9(12)
C(25)-P(2)-C(26)-C(27)	129.1(11)
Pt(1)-P(2)-C(26)-C(27)	12.4(13)
C(33)-P(2)-C(26)-C(32)	64.6(12)
C(25)-P(2)-C(26)-C(32)	-54.4(12)
Pt(1)-P(2)-C(26)-C(32)	-171.1(9)
C(32)-C(26)-C(27)-C(28)	-1(2)
P(2)-C(26)-C(27)-C(28)	176.1(13)
C(26)-C(27)-C(28)-C(29)	4(3)
C(27)-C(28)-C(29)-C(30)	-4(3)
C(28)-C(29)-C(30)-C(32)	2(2)
C(27)-C(26)-C(32)-C(30)	-2(2)
P(2)-C(26)-C(32)-C(30)	-178.2(10)
C(29)-C(30)-C(32)-C(26)	1(2)
C(26)-P(2)-C(33)-C(34)	138.8(9)
C(25)-P(2)-C(33)-C(34)	-107.0(10)
Pt(1)-P(2)-C(33)-C(34)	9.3(10)
P(2)-C(33)-C(34)-P(3)	-39.2(11)
C(40)-C(35)-C(36)-C(37)	0(2)
P(3)-C(35)-C(36)-C(37)	-177.2(13)
C(35)-C(36)-C(37)-C(38)	1(3)
C(36)-C(37)-C(38)-C(39)	-1(3)
C(37)-C(38)-C(39)-C(40)	1(3)
C(38)-C(39)-C(40)-C(35)	0(3)
C(36)-C(35)-C(40)-C(39)	-1(2)
P(3)-C(35)-C(40)-C(39)	177.0(13)
C(46)-C(41)-C(42)-C(43)	1.5(17)
P(3)-C(41)-C(42)-C(43)	179.5(9)
C(41)-C(42)-C(43)-C(44)	-0.7(19)
C(42)-C(43)-C(44)-C(45)	1(2)
C(43)-C(44)-C(45)-C(46)	-1(2)
C(44)-C(45)-C(46)-C(41)	2(2)
C(42)-C(41)-C(46)-C(45)	-2.2(19)
P(3)-C(41)-C(46)-C(45)	179.8(10)
C(76)-P(5)-Pt(2)-C(51)	118(7)
C(75)-P(5)-Pt(2)-C(51)	-3(7)
C821-P(5)-Pt(2)-C(51)	-124(7)
C822-P(5)-Pt(2)-C(51)	-119(7)
C(76)-P(5)-Pt(2)-P(4)	101.4(5)
C(75)-P(5)-Pt(2)-P(4)	-20.0(6)
C821-P(5)-Pt(2)-P(4)	-141.2(15)
C822-P(5)-Pt(2)-P(4)	-135.9(16)
C(76)-P(5)-Pt(2)-P61	-82.2(5)
C(75)-P(5)-Pt(2)-P61	156.5(6)
C821-P(5)-Pt(2)-P61	35.2(15)
C822-P(5)-Pt(2)-P61	40.5(16)
C(76)-P(5)-Pt(2)-P62	-101.3(5)
C(75)-P(5)-Pt(2)-P62	137.4(6)
C821-P(5)-Pt(2)-P62	16.2(15)
C822-P(5)-Pt(2)-P62	21.4(16)



C(36)-C(35)-P(3)-C(41)	-115.7(12)
C(40)-C(35)-P(3)-C(41)	66.8(12)
C(36)-C(35)-P(3)-C(34)	136.1(12)
C(40)-C(35)-P(3)-C(34)	-41.4(13)
C(36)-C(35)-P(3)-Pt(1)	23.1(13)
C(40)-C(35)-P(3)-Pt(1)	-154.4(10)
C(46)-C(41)-P(3)-C(35)	-155.1(11)
C(42)-C(41)-P(3)-C(35)	26.9(11)
C(46)-C(41)-P(3)-C(34)	-43.9(11)
C(42)-C(41)-P(3)-C(34)	138.0(10)
C(46)-C(41)-P(3)-Pt(1)	71.8(12)
C(42)-C(41)-P(3)-Pt(1)	-106.2(9)
C(33)-C(34)-P(3)-C(35)	-67.1(11)
C(33)-C(34)-P(3)-C(41)	-178.5(9)
C(33)-C(34)-P(3)-Pt(1)	52.0(9)
C(1)-Pt(1)-P(3)-C(35)	-100.5(6)
P(1)-Pt(1)-P(3)-C(35)	121.4(6)
P(2)-Pt(1)-P(3)-C(35)	78.9(5)
C(1)-Pt(1)-P(3)-C(41)	30.1(7)
P(1)-Pt(1)-P(3)-C(41)	-108.0(7)
P(2)-Pt(1)-P(3)-C(41)	-150.4(7)
C(1)-Pt(1)-P(3)-C(34)	145.0(6)
P(1)-Pt(1)-P(3)-C(34)	6.9(7)
P(2)-Pt(1)-P(3)-C(34)	-35.5(5)
C(51)-Pt(2)-P(4)-C(68)	98.0(9)
P(5)-Pt(2)-P(4)-C(68)	-83.2(8)
P61-Pt(2)-P(4)-C(68)	-95.6(11)
P62-Pt(2)-P(4)-C(68)	-149.9(9)
C(51)-Pt(2)-P(4)-C(74)	-154.3(15)
P(5)-Pt(2)-P(4)-C(74)	24.5(14)
P61-Pt(2)-P(4)-C(74)	12.2(17)
P62-Pt(2)-P(4)-C(74)	-42.1(15)
C(51)-Pt(2)-P(4)-C(62)	-35.7(9)
P(5)-Pt(2)-P(4)-C(62)	143.1(8)
P61-Pt(2)-P(4)-C(62)	130.7(11)
P62-Pt(2)-P(4)-C(62)	76.4(10)
P(4)-Pt(2)-C(51)-C(52)	-136.0(15)
P(5)-Pt(2)-C(51)-C(52)	-153(6)
P61-Pt(2)-C(51)-C(52)	48.0(16)
P62-Pt(2)-C(51)-C(52)	67.0(16)
P(4)-Pt(2)-C(51)-C(59)	85.7(13)
P(5)-Pt(2)-C(51)-C(59)	69(7)
P61-Pt(2)-C(51)-C(59)	-90.3(13)
P62-Pt(2)-C(51)-C(59)	-71.4(13)
C(59)-C(51)-C(52)-C(53)	-49(2)
Pt(2)-C(51)-C(52)-C(53)	172.9(12)
C(51)-C(52)-C(53)-C(54)	55(2)
C(52)-C(53)-C(54)-O(55)	-172.5(14)
C(52)-C(53)-C(54)-C(58)	-58(2)
C(52)-C(53)-C(54)-C(60)	63(2)
C(58)-C(54)-O(55)-C(56)	44.1(18)
C(53)-C(54)-O(55)-C(56)	160.3(15)
C(60)-C(54)-O(55)-C(56)	-74.0(19)
C(54)-O(55)-C(56)-C(57)	-19.0(19)
O(55)-C(56)-C(57)-C(58)	-10(2)
O(55)-C(56)-C(57)-C(61)	-127.6(17)

O(55)-C(54)-C(58)-C(57)	-50.0(19)
C(53)-C(54)-C(58)-C(57)	-169.4(17)
C(60)-C(54)-C(58)-C(57)	68(2)
O(55)-C(54)-C(58)-C(59)	-178.4(14)
C(53)-C(54)-C(58)-C(59)	62(2)
C(60)-C(54)-C(58)-C(59)	-61(2)
C(56)-C(57)-C(58)-C(54)	34(2)
C(61)-C(57)-C(58)-C(54)	147.4(18)
C(56)-C(57)-C(58)-C(59)	161.9(16)
C(61)-C(57)-C(58)-C(59)	-85(2)
C(54)-C(58)-C(59)-C(51)	-55(2)
C(57)-C(58)-C(59)-C(51)	-174.3(16)
C(52)-C(51)-C(59)-C(58)	47(2)
Pt(2)-C(51)-C(59)-C(58)	-173.4(12)
C(68)-P(4)-C(62)-C(67)	143.9(17)
C(74)-P(4)-C(62)-C(67)	43(2)
Pt(2)-P(4)-C(62)-C(67)	-75.0(18)
C(68)-P(4)-C(62)-C(63)	-40.8(15)
C(74)-P(4)-C(62)-C(63)	-142.1(17)
Pt(2)-P(4)-C(62)-C(63)	100.4(12)
C(67)-C(62)-C(63)-C(64)	-2(3)
P(4)-C(62)-C(63)-C(64)	-177.4(15)
C(62)-C(63)-C(64)-C(65)	-4(3)
C(63)-C(64)-C(65)-C(66)	6(4)
C(64)-C(65)-C(66)-C(67)	-2(4)
C(63)-C(62)-C(67)-C(66)	5(3)
P(4)-C(62)-C(67)-C(66)	-179.5(14)
C(65)-C(66)-C(67)-C(62)	-4(3)
C(74)-P(4)-C(68)-C(73)	-140.6(15)
C(62)-P(4)-C(68)-C(73)	111.1(14)
Pt(2)-P(4)-C(68)-C(73)	-29.4(16)
C(74)-P(4)-C(68)-C(69)	45.9(17)
C(62)-P(4)-C(68)-C(69)	-62.4(16)
Pt(2)-P(4)-C(68)-C(69)	157.2(11)
C(73)-C(68)-C(69)-C(70)	2(3)
P(4)-C(68)-C(69)-C(70)	174.8(14)
C(68)-C(69)-C(70)-C(71)	-2(3)
C(69)-C(70)-C(71)-C(72)	0(3)
C(70)-C(71)-C(72)-C(73)	2(3)
C(69)-C(68)-C(73)-C(72)	0(2)
P(4)-C(68)-C(73)-C(72)	-173.7(12)
C(71)-C(72)-C(73)-C(68)	-2(2)
C(68)-P(4)-C(74)-C(75)	95(3)
C(62)-P(4)-C(74)-C(75)	-160(3)
Pt(2)-P(4)-C(74)-C(75)	-31(4)
P(4)-C(74)-C(75)-P(5)	16(4)
C(76)-P(5)-C(75)-C(74)	-122(2)
C821-P(5)-C(75)-C(74)	129(3)
C822-P(5)-C(75)-C(74)	121(3)
Pt(2)-P(5)-C(75)-C(74)	8(3)
C(75)-P(5)-C(76)-C(81)	139.4(13)
C821-P(5)-C(76)-C(81)	-106(2)
C822-P(5)-C(76)-C(81)	-104(2)
Pt(2)-P(5)-C(76)-C(81)	16.4(15)
C(75)-P(5)-C(76)-C(77)	-47.8(14)
C821-P(5)-C(76)-C(77)	67(2)

C822-P(5)-C(76)-C(77)	69(2)
Pt(2)-P(5)-C(76)-C(77)	-170.8(10)
C(81)-C(76)-C(77)-C(78)	-3(2)
P(5)-C(76)-C(77)-C(78)	-176.7(14)
C(76)-C(77)-C(78)-C(79)	0(3)
C(77)-C(78)-C(79)-C(80)	6(3)
C(78)-C(79)-C(80)-C(81)	-7(3)
C(77)-C(76)-C(81)-C(80)	2(3)
P(5)-C(76)-C(81)-C(80)	175.3(16)
C(79)-C(80)-C(81)-C(76)	3(3)
C(51)-Pt(2)-P61-C841	-75.4(10)
P(4)-Pt(2)-P61-C841	118.5(11)
P(5)-Pt(2)-P61-C841	106.1(9)
P62-Pt(2)-P61-C841	-146.9(15)
C(51)-Pt(2)-P61-C901	50.7(9)
P(4)-Pt(2)-P61-C901	-115.3(10)
P(5)-Pt(2)-P61-C901	-127.8(8)
P62-Pt(2)-P61-C901	-20.8(10)
C(51)-Pt(2)-P61-C831	162.9(10)
P(4)-Pt(2)-P61-C831	-3.2(14)
P(5)-Pt(2)-P61-C831	-15.6(9)
P62-Pt(2)-P61-C831	91.4(13)
C(76)-P(5)-C821-C831	79(2)
C(75)-P(5)-C821-C831	-169.6(19)
C822-P(5)-C821-C831	-88(24)
Pt(2)-P(5)-C821-C831	-49(2)
P(5)-C821-C831-P61	36(3)
C841-P61-C831-C821	-142(2)
C901-P61-C831-C821	107(2)
Pt(2)-P61-C831-C821	-11(2)
C901-P61-C841-C851	-113.1(13)
C831-P61-C841-C851	138.5(13)
Pt(2)-P61-C841-C851	15.2(15)
C901-P61-C841-C891	63.0(14)
C831-P61-C841-C891	-45.4(15)
Pt(2)-P61-C841-C891	-168.6(9)
C891-C841-C851-C861	0.0
P61-C841-C851-C861	176.1(16)
C841-C851-C861-C871	0.0
C851-C861-C871-C881	0.0
C861-C871-C881-C891	0.0
C871-C881-C891-C841	0.0
C851-C841-C891-C881	0.0
P61-C841-C891-C881	-176.2(16)
C841-P61-C901-C911	23.2(15)
C831-P61-C901-C911	133.3(13)
Pt(2)-P61-C901-C911	-111.0(11)
C841-P61-C901-C951	-158.3(13)
C831-P61-C901-C951	-48.2(15)
Pt(2)-P61-C901-C951	67.5(13)
C951-C901-C911-C921	0.0
P61-C901-C911-C921	178.5(16)
C901-C911-C921-C931	0.0
C911-C921-C931-C941	0.0
C921-C931-C941-C951	0.0
C931-C941-C951-C901	0.0

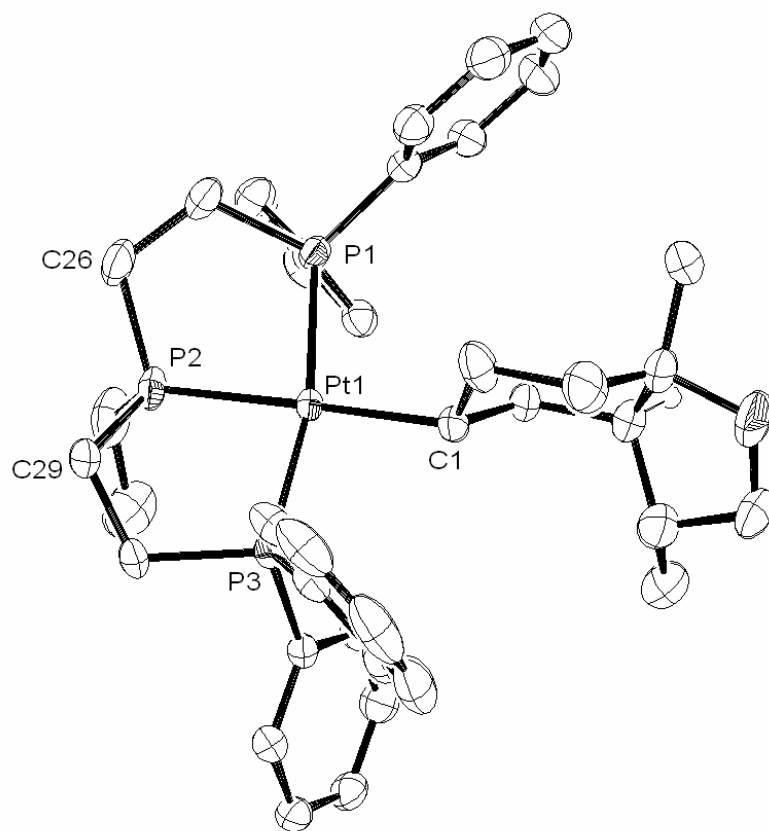
C911-C901-C951-C941	0.0
P61-C901-C951-C941	-178.5(16)
C(51)-Pt(2)-P62-C842	-104.3(9)
P(4)-Pt(2)-P62-C842	144.2(8)
P(5)-Pt(2)-P62-C842	78.4(8)
P61-Pt(2)-P62-C842	7.4(10)
C(51)-Pt(2)-P62-C902	26.1(10)
P(4)-Pt(2)-P62-C902	-85.4(11)
P(5)-Pt(2)-P62-C902	-151.2(10)
P61-Pt(2)-P62-C902	137.8(15)
C(51)-Pt(2)-P62-C832	140.2(10)
P(4)-Pt(2)-P62-C832	28.7(12)
P(5)-Pt(2)-P62-C832	-37.1(9)
P61-Pt(2)-P62-C832	-108.1(14)
C(76)-P(5)-C822-C832	134(3)
C(75)-P(5)-C822-C832	-112(3)
C821-P(5)-C822-C832	147(27)
Pt(2)-P(5)-C822-C832	4(4)
P(5)-C822-C832-P62	-34(4)
C842-P62-C832-C822	-73(2)
C902-P62-C832-C822	178(2)
Pt(2)-P62-C832-C822	47(2)
C902-P62-C842-C852	-85.9(13)
C832-P62-C842-C852	167.1(12)
Pt(2)-P62-C842-C852	55.0(12)
C902-P62-C842-C892	92.2(14)
C832-P62-C842-C892	-14.9(16)
Pt(2)-P62-C842-C892	-127.0(10)
C892-C842-C852-C862	0.0
P62-C842-C852-C862	178.1(15)
C842-C852-C862-C872	0.0
C852-C862-C872-C882	0.0
C862-C872-C882-C892	0.0
C872-C882-C892-C842	0.0
C852-C842-C892-C882	0.0
P62-C842-C892-C882	-178.0(16)
C842-P62-C902-C912	11.7(15)
C832-P62-C902-C912	123.4(14)
Pt(2)-P62-C902-C912	-122.3(11)
C842-P62-C902-C952	-164.2(13)
C832-P62-C902-C952	-52.5(16)
Pt(2)-P62-C902-C952	61.8(16)
C952-C902-C912-C922	0.0
P62-C902-C912-C922	-175.8(17)
C902-C912-C922-C932	0.0
C912-C922-C932-C942	0.0
C922-C932-C942-C952	0.0
C932-C942-C952-C902	0.0
C912-C902-C952-C942	0.0
P62-C902-C952-C942	175.9(17)

---

**Table S14.** Crystallographic data and collection parameters for **9**.

<b>9</b>	
Empirical Formula	C <sub>40</sub> H <sub>50</sub> BF <sub>4</sub> OP <sub>3</sub> Pt
FW	921.61
Space Group	Triclinic, P-1
<i>A</i>	10.7092(9)
<i>B</i>	11.3859(9)
<i>C</i>	18.7092(15)
<i>V</i> , Å <sup>3</sup>	2093.4(3)
<i>Z</i>	2
<i>T</i> , K	100(2)
<i>D</i> <sub>c</sub> , Mg/m <sup>3</sup>	1.462
<i>λ</i> , Å	Mo Kα (0.71073)
<i>μ</i> , mm <sup>-1</sup>	3.513
<i>R</i> <sub>f</sub> <sup>a</sup>	0.0328
<i>R</i> <sub>w</sub> <sup>b</sup>	0.0836
GOF <sup>c</sup>	1.086

<sup>a</sup> $R_f = \Sigma(F_o - F_c)/\Sigma F_o$ . <sup>b</sup> $R_w = [\Sigma\omega(F_o - F_c)^2/\Sigma\omega F_o^2]^{1/2}$ . <sup>c</sup>GOF =  $[\Sigma\omega(F_o - F_c)^2/(n - p)]^{1/2}$ , where *n* = number of reflections and *p* = number of parameters.



**Figure S5.** ORTEP representation of **9**.  $\text{BF}_4^-$  counter ion and solvent molecule not shown.

**Table S15.** Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **9**.  $U(\text{eq})$  is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor.

	x	y	z	U(eq)
Pt(1)	-706(1)	7493(1)	7095(1)	21(1)
P(1)	-2117(1)	7994(1)	7744(1)	25(1)
P(2)	-2413(1)	7498(1)	6081(1)	30(1)
P(3)	237(1)	6790(1)	6244(1)	24(1)
C(1)	829(3)	7256(4)	7980(2)	30(1)
C(2)	1483(4)	8362(4)	8610(2)	35(1)
C(3)	2625(4)	8144(4)	9220(2)	41(1)
C(4)	2200(4)	6942(5)	9522(2)	48(1)
C(6)	4519(5)	7400(5)	9622(3)	51(1)
C(7)	3903(4)	8023(5)	8995(2)	45(1)
C(8)	1453(5)	5827(5)	8884(3)	49(1)
C(9)	325(5)	6065(4)	8294(2)	39(1)
C(10)	1412(5)	7035(6)	10096(2)	56(1)
C(11)	4827(5)	9253(5)	8901(3)	57(1)
C(12)	-2128(3)	7717(3)	8684(2)	28(1)
C(13)	-2544(4)	6488(4)	8816(2)	34(1)
C(14)	-2500(4)	6258(4)	9539(2)	41(1)
C(15)	-2052(4)	7241(4)	10136(2)	43(1)
C(16)	-1637(5)	8467(4)	10016(2)	42(1)
C(17)	-1684(4)	8703(4)	9293(2)	33(1)
O(18)	3472(4)	6782(4)	9919(2)	65(1)
C(19)	-2003(3)	9625(3)	7767(2)	28(1)
C(20)	-724(4)	10503(3)	7912(2)	32(1)
C(21)	-581(4)	11769(4)	7941(2)	39(1)
C(22)	-1716(5)	12157(4)	7807(2)	45(1)
C(23)	-2997(5)	11286(4)	7652(3)	49(1)
C(24)	-3141(4)	10030(4)	7632(2)	39(1)
C(25)	-3830(3)	7065(3)	7161(2)	34(1)
C(26)	-4030(4)	7180(4)	6331(2)	43(1)
C(27)	-2413(5)	8911(5)	5679(3)	57(1)
C(28)	-1204(6)	9497(6)	5439(4)	67(2)
C(29)	-2486(4)	6240(4)	5358(2)	38(1)
C(30)	-1038(4)	6354(4)	5312(2)	34(1)
C(31)	1714(3)	7905(3)	6109(2)	25(1)
C(32)	2266(3)	7648(3)	5526(2)	29(1)
C(33)	3368(4)	8534(3)	5418(2)	32(1)
C(34)	3925(4)	9690(4)	5891(2)	35(1)
C(35)	3376(4)	9953(4)	6468(2)	39(1)
C(36)	2277(4)	9066(3)	6580(2)	32(1)
C(37)	709(4)	5397(3)	6393(2)	29(1)
C(38)	2058(4)	5403(4)	6562(2)	36(1)
C(39)	2348(6)	4294(4)	6663(2)	48(1)
C(40)	1305(7)	3203(5)	6597(2)	57(1)
C(41)	-24(6)	3198(4)	6440(2)	50(1)
C(42)	-333(5)	4295(3)	6342(2)	38(1)
B(1)	4273(4)	6514(4)	3585(3)	38(1)
F(1)	5334(6)	7170(4)	3383(4)	132(2)
F(2)	4658(5)	5801(6)	4047(4)	157(3)

F(4)	3346(6)	5739(5)	2958(4)	162(3)
F(3)	3629(4)	7364(4)	3803(2)	80(1)

---

**Table S16.** Bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ] for **9**.

---

Pt(1)-C(1)	2.134(3)
Pt(1)-P(3)	2.2722(8)
Pt(1)-P(2)	2.3016(8)
Pt(1)-P(1)	2.3032(8)
P(1)-C(12)	1.819(3)
P(1)-C(19)	1.822(3)
P(1)-C(25)	1.846(4)
P(2)-C(29)	1.842(4)
P(2)-C(27)	1.843(5)
P(2)-C(26)	1.850(4)
P(3)-C(31)	1.818(3)
P(3)-C(37)	1.822(4)
P(3)-C(30)	1.861(3)
C(1)-C(2)	1.512(5)
C(1)-C(9)	1.541(6)
C(2)-C(3)	1.556(5)
C(3)-C(4)	1.525(7)
C(3)-C(7)	1.562(6)
C(4)-O(18)	1.457(5)
C(4)-C(10)	1.528(6)
C(4)-C(8)	1.534(7)
C(6)-O(18)	1.412(6)
C(6)-C(7)	1.540(6)
C(7)-C(11)	1.526(7)
C(8)-C(9)	1.534(6)
C(12)-C(17)	1.405(5)
C(12)-C(13)	1.410(5)
C(13)-C(14)	1.397(6)
C(14)-C(15)	1.388(6)
C(15)-C(16)	1.399(6)
C(16)-C(17)	1.397(5)
C(19)-C(24)	1.397(5)
C(19)-C(20)	1.397(5)
C(20)-C(21)	1.400(5)
C(21)-C(22)	1.382(6)
C(22)-C(23)	1.394(7)
C(23)-C(24)	1.390(6)
C(25)-C(26)	1.541(6)
C(27)-C(28)	1.475(8)
C(29)-C(30)	1.542(5)
C(31)-C(36)	1.400(5)
C(31)-C(32)	1.403(4)
C(32)-C(33)	1.392(5)
C(33)-C(34)	1.397(5)
C(34)-C(35)	1.392(6)
C(35)-C(36)	1.394(5)
C(37)-C(38)	1.401(5)
C(37)-C(42)	1.403(5)



C(38)-C(39)	1.402(6)
C(39)-C(40)	1.391(8)
C(40)-C(41)	1.381(8)
C(41)-C(42)	1.397(6)
B(1)-F(2)	1.315(6)
B(1)-F(1)	1.333(6)
B(1)-F(4)	1.375(7)
B(1)-F(3)	1.418(6)

C(1)-Pt(1)-P(3)	90.04(9)
C(1)-Pt(1)-P(2)	173.01(10)
P(3)-Pt(1)-P(2)	85.87(3)
C(1)-Pt(1)-P(1)	101.07(9)
P(3)-Pt(1)-P(1)	166.86(3)
P(2)-Pt(1)-P(1)	82.35(3)
C(12)-P(1)-C(19)	105.17(15)
C(12)-P(1)-C(25)	104.01(17)
C(19)-P(1)-C(25)	107.12(17)
C(12)-P(1)-Pt(1)	125.26(11)
C(19)-P(1)-Pt(1)	110.17(11)
C(25)-P(1)-Pt(1)	103.82(12)
C(29)-P(2)-C(27)	109.9(2)
C(29)-P(2)-C(26)	110.85(18)
C(27)-P(2)-C(26)	99.6(2)
C(29)-P(2)-Pt(1)	104.73(13)
C(27)-P(2)-Pt(1)	121.81(16)
C(26)-P(2)-Pt(1)	109.86(13)
C(31)-P(3)-C(37)	105.96(16)
C(31)-P(3)-C(30)	104.83(15)
C(37)-P(3)-C(30)	105.07(17)
C(31)-P(3)-Pt(1)	116.13(11)
C(37)-P(3)-Pt(1)	115.68(10)
C(30)-P(3)-Pt(1)	108.12(12)
C(2)-C(1)-C(9)	110.1(3)
C(2)-C(1)-Pt(1)	114.2(2)
C(9)-C(1)-Pt(1)	112.4(2)
C(1)-C(2)-C(3)	111.4(3)
C(4)-C(3)-C(2)	114.7(3)
C(4)-C(3)-C(7)	100.8(4)
C(2)-C(3)-C(7)	117.5(3)
O(18)-C(4)-C(3)	104.5(4)
O(18)-C(4)-C(10)	106.8(3)
C(3)-C(4)-C(10)	112.8(4)
O(18)-C(4)-C(8)	108.3(4)
C(3)-C(4)-C(8)	111.3(3)
C(10)-C(4)-C(8)	112.7(4)
O(18)-C(6)-C(7)	107.8(3)
C(11)-C(7)-C(6)	116.1(4)
C(11)-C(7)-C(3)	113.5(4)
C(6)-C(7)-C(3)	101.4(4)
C(4)-C(8)-C(9)	112.8(4)
C(8)-C(9)-C(1)	112.8(4)
C(17)-C(12)-C(13)	118.8(3)
C(17)-C(12)-P(1)	121.4(3)
C(13)-C(12)-P(1)	119.8(3)

C(14)-C(13)-C(12)	120.5(4)
C(15)-C(14)-C(13)	120.0(4)
C(14)-C(15)-C(16)	120.3(4)
C(17)-C(16)-C(15)	119.9(4)
C(16)-C(17)-C(12)	120.5(4)
C(6)-O(18)-C(4)	109.8(3)
C(24)-C(19)-C(20)	118.7(3)
C(24)-C(19)-P(1)	122.8(3)
C(20)-C(19)-P(1)	118.5(3)
C(19)-C(20)-C(21)	120.7(4)
C(22)-C(21)-C(20)	119.9(4)
C(21)-C(22)-C(23)	119.8(4)
C(24)-C(23)-C(22)	120.4(4)
C(23)-C(24)-C(19)	120.4(4)
C(26)-C(25)-P(1)	112.1(3)
C(25)-C(26)-P(2)	111.4(2)
C(28)-C(27)-P(2)	115.7(4)
C(30)-C(29)-P(2)	109.2(2)
C(29)-C(30)-P(3)	113.2(2)
C(36)-C(31)-C(32)	119.3(3)
C(36)-C(31)-P(3)	119.1(2)
C(32)-C(31)-P(3)	121.6(2)
C(33)-C(32)-C(31)	120.5(3)
C(32)-C(33)-C(34)	119.9(3)
C(35)-C(34)-C(33)	119.9(3)
C(34)-C(35)-C(36)	120.4(3)
C(35)-C(36)-C(31)	120.1(3)
C(38)-C(37)-C(42)	120.1(4)
C(38)-C(37)-P(3)	122.3(3)
C(42)-C(37)-P(3)	117.7(3)
C(37)-C(38)-C(39)	119.3(4)
C(40)-C(39)-C(38)	120.1(5)
C(41)-C(40)-C(39)	120.7(4)
C(40)-C(41)-C(42)	120.0(4)
C(41)-C(42)-C(37)	119.8(4)
F(2)-B(1)-F(1)	109.5(5)
F(2)-B(1)-F(4)	107.0(5)
F(1)-B(1)-F(4)	108.3(6)
F(2)-B(1)-F(3)	120.0(6)
F(1)-B(1)-F(3)	106.8(4)
F(4)-B(1)-F(3)	104.8(4)

**Table S17.** Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **9**. The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [ h^2 a^* 2 U^{11} + \dots + 2 h k a^* b^* U^{12} ]$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
Pt(1)	18(1)	26(1)	17(1)	2(1)	1(1)	7(1)
P(1)	20(1)	28(1)	25(1)	2(1)	4(1)	9(1)
P(2)	26(1)	33(1)	25(1)	7(1)	-4(1)	7(1)
P(3)	22(1)	29(1)	17(1)	2(1)	4(1)	5(1)
C(1)	27(2)	46(2)	21(1)	7(1)	5(1)	19(1)
C(2)	24(2)	48(2)	24(2)	6(1)	2(1)	5(1)

C(3)	26(2)	60(2)	32(2)	7(2)	1(1)	10(2)
C(4)	29(2)	81(3)	37(2)	25(2)	4(2)	22(2)
C(6)	33(2)	73(3)	47(2)	11(2)	1(2)	28(2)
C(7)	37(2)	55(2)	39(2)	1(2)	6(2)	15(2)
C(8)	55(3)	64(3)	43(2)	21(2)	13(2)	38(2)
C(9)	54(2)	41(2)	27(2)	7(1)	6(2)	28(2)
C(10)	39(2)	95(4)	31(2)	22(2)	8(2)	17(2)
C(11)	36(2)	73(3)	57(3)	15(2)	11(2)	10(2)
C(12)	22(1)	35(2)	29(2)	7(1)	8(1)	11(1)
C(13)	30(2)	37(2)	38(2)	9(1)	8(1)	12(1)
C(14)	35(2)	46(2)	46(2)	20(2)	13(2)	16(2)
C(15)	38(2)	60(3)	35(2)	17(2)	13(2)	20(2)
C(16)	46(2)	52(2)	29(2)	5(2)	12(2)	17(2)
C(17)	34(2)	38(2)	31(2)	7(1)	11(1)	12(1)
O(18)	38(2)	105(3)	55(2)	38(2)	0(2)	30(2)
C(19)	30(2)	31(2)	24(1)	5(1)	6(1)	14(1)
C(20)	34(2)	33(2)	29(2)	8(1)	8(1)	11(1)
C(21)	47(2)	34(2)	31(2)	7(1)	9(2)	7(2)
C(22)	66(3)	33(2)	38(2)	10(2)	13(2)	21(2)
C(23)	52(3)	47(2)	58(3)	13(2)	15(2)	31(2)
C(24)	34(2)	41(2)	46(2)	7(2)	9(2)	19(2)
C(25)	21(1)	36(2)	41(2)	-3(1)	2(1)	9(1)
C(26)	26(2)	59(2)	38(2)	-3(2)	-4(1)	19(2)
C(27)	51(3)	54(3)	53(3)	24(2)	-7(2)	12(2)
C(28)	54(3)	65(3)	84(4)	37(3)	13(3)	18(3)
C(29)	24(2)	54(2)	22(1)	-2(1)	1(1)	1(2)
C(30)	29(2)	47(2)	17(1)	-2(1)	1(1)	4(1)
C(31)	22(1)	29(1)	21(1)	4(1)	4(1)	6(1)
C(32)	30(2)	31(2)	24(1)	4(1)	7(1)	9(1)
C(33)	29(2)	41(2)	29(2)	12(1)	11(1)	13(1)
C(34)	25(2)	40(2)	37(2)	12(1)	8(1)	5(1)
C(35)	33(2)	36(2)	37(2)	-1(1)	7(2)	1(1)
C(36)	29(2)	35(2)	27(2)	-2(1)	6(1)	6(1)
C(37)	38(2)	32(2)	18(1)	2(1)	9(1)	10(1)
C(38)	43(2)	45(2)	22(1)	5(1)	5(1)	20(2)
C(39)	68(3)	55(2)	27(2)	8(2)	6(2)	36(2)
C(40)	106(4)	47(2)	26(2)	11(2)	15(2)	40(3)
C(41)	85(4)	35(2)	29(2)	10(2)	23(2)	14(2)
C(42)	54(2)	33(2)	25(2)	3(1)	17(2)	6(2)
B(1)	30(2)	29(2)	57(3)	16(2)	12(2)	7(2)
F(1)	128(4)	80(3)	269(7)	89(4)	148(5)	59(3)
F(2)	78(3)	192(6)	183(6)	154(5)	0(3)	5(3)
F(4)	136(5)	98(4)	201(6)	-75(4)	-51(4)	73(4)
F(3)	65(2)	102(3)	66(2)	-23(2)	23(2)	26(2)

---

**Table S18.** Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **9**.

	x	y	z	U(eq)
H(1)	1560	7151	7757	36
H(2A)	787	8517	8838	42
H(2B)	1868	9109	8406	42
H(3)	2934	8852	9650	50
H(6A)	4915	6799	9420	61
H(6B)	5245	8034	10018	61
H(7)	3598	7428	8511	54
H(8A)	1054	5095	9095	59
H(8B)	2112	5631	8640	59
H(9A)	-404	6132	8520	47
H(9B)	-64	5345	7877	47
H(10A)	1336	6315	10349	83
H(10B)	505	7049	9841	83
H(10C)	1891	7800	10464	83
H(11A)	5058	9873	9353	86
H(11B)	4359	9542	8470	86
H(11C)	5657	9134	8819	86
H(13)	-2858	5812	8409	41
H(14)	-2776	5427	9624	49
H(15)	-2027	7082	10627	51
H(16)	-1323	9138	10426	50
H(17)	-1415	9537	9212	40
H(20)	57	10239	7993	39
H(21)	294	12361	8051	46
H(22)	-1623	13014	7821	54
H(23)	-3777	11551	7558	58
H(24)	-4018	9443	7526	47
H(25A)	-3984	6181	7209	41
H(25B)	-4509	7346	7346	41
H(26A)	-4417	7861	6229	51
H(26B)	-4681	6398	6015	51
H(27A)	-3219	8700	5241	68
H(27B)	-2510	9530	6054	68
H(28A)	-387	9616	5847	101
H(28B)	-1232	10304	5309	101
H(28C)	-1188	8959	5001	101
H(29A)	-2902	5424	5484	45
H(29B)	-3049	6299	4868	45
H(30A)	-771	6987	5004	41
H(30B)	-1037	5549	5058	41
H(32)	1885	6864	5201	34
H(33)	3742	8352	5024	38
H(34)	4677	10296	5819	42
H(35)	3753	10742	6787	46
H(36)	1910	9251	6976	38
H(38)	2768	6150	6607	43
H(39)	3260	4288	6777	57
H(40)	1508	2453	6661	68
H(41)	-729	2448	6398	59

H(42)                                      -1245                                      4296                                      6241                                      46

---

**Table S19.** Torsion angles [°] for **9**.

C(1)-Pt(1)-P(1)-C(12)	-17.64(18)
P(3)-Pt(1)-P(1)-C(12)	129.65(18)
P(2)-Pt(1)-P(1)-C(12)	156.18(15)
C(1)-Pt(1)-P(1)-C(19)	109.30(16)
P(3)-Pt(1)-P(1)-C(19)	-103.41(17)
P(2)-Pt(1)-P(1)-C(19)	-76.87(12)
C(1)-Pt(1)-P(1)-C(25)	-136.28(17)
P(3)-Pt(1)-P(1)-C(25)	11.0(2)
P(2)-Pt(1)-P(1)-C(25)	37.54(14)
C(1)-Pt(1)-P(2)-C(29)	-24.4(8)
P(3)-Pt(1)-P(2)-C(29)	30.00(13)
P(1)-Pt(1)-P(2)-C(29)	-144.15(13)
C(1)-Pt(1)-P(2)-C(27)	-149.7(8)
P(3)-Pt(1)-P(2)-C(27)	-95.3(2)
P(1)-Pt(1)-P(2)-C(27)	90.5(2)
C(1)-Pt(1)-P(2)-C(26)	94.7(8)
P(3)-Pt(1)-P(2)-C(26)	149.10(16)
P(1)-Pt(1)-P(2)-C(26)	-25.06(16)
C(1)-Pt(1)-P(3)-C(31)	-79.44(15)
P(2)-Pt(1)-P(3)-C(31)	106.24(12)
P(1)-Pt(1)-P(3)-C(31)	132.59(16)
C(1)-Pt(1)-P(3)-C(37)	45.70(17)
P(2)-Pt(1)-P(3)-C(37)	-128.62(13)
P(1)-Pt(1)-P(3)-C(37)	-102.27(18)
C(1)-Pt(1)-P(3)-C(30)	163.15(17)
P(2)-Pt(1)-P(3)-C(30)	-11.17(14)
P(1)-Pt(1)-P(3)-C(30)	15.2(2)
P(3)-Pt(1)-C(1)-C(2)	132.1(2)
P(2)-Pt(1)-C(1)-C(2)	-173.7(6)
P(1)-Pt(1)-C(1)-C(2)	-54.9(3)
P(3)-Pt(1)-C(1)-C(9)	-101.5(2)
P(2)-Pt(1)-C(1)-C(9)	-47.3(9)
P(1)-Pt(1)-C(1)-C(9)	71.5(2)
C(9)-C(1)-C(2)-C(3)	54.4(4)
Pt(1)-C(1)-C(2)-C(3)	-178.0(2)
C(1)-C(2)-C(3)-C(4)	-52.7(5)
C(1)-C(2)-C(3)-C(7)	65.4(5)
C(2)-C(3)-C(4)-O(18)	165.4(4)
C(7)-C(3)-C(4)-O(18)	38.2(4)
C(2)-C(3)-C(4)-C(10)	-79.0(5)
C(7)-C(3)-C(4)-C(10)	153.7(3)
C(2)-C(3)-C(4)-C(8)	48.8(5)
C(7)-C(3)-C(4)-C(8)	-78.4(4)
O(18)-C(6)-C(7)-C(11)	146.5(5)
O(18)-C(6)-C(7)-C(3)	23.1(5)
C(4)-C(3)-C(7)-C(11)	-161.7(4)
C(2)-C(3)-C(7)-C(11)	72.9(5)
C(4)-C(3)-C(7)-C(6)	-36.5(4)
C(2)-C(3)-C(7)-C(6)	-161.9(4)
O(18)-C(4)-C(8)-C(9)	-163.2(4)

C(3)-C(4)-C(8)-C(9)	-48.9(5)
C(10)-C(4)-C(8)-C(9)	79.0(5)
C(4)-C(8)-C(9)-C(1)	53.9(5)
C(2)-C(1)-C(9)-C(8)	-56.3(4)
Pt(1)-C(1)-C(9)-C(8)	175.1(3)
C(19)-P(1)-C(12)-C(17)	-18.1(3)
C(25)-P(1)-C(12)-C(17)	-130.5(3)
Pt(1)-P(1)-C(12)-C(17)	110.9(3)
C(19)-P(1)-C(12)-C(13)	164.1(3)
C(25)-P(1)-C(12)-C(13)	51.7(3)
Pt(1)-P(1)-C(12)-C(13)	-66.9(3)
C(17)-C(12)-C(13)-C(14)	-0.8(5)
P(1)-C(12)-C(13)-C(14)	177.0(3)
C(12)-C(13)-C(14)-C(15)	0.4(6)
C(13)-C(14)-C(15)-C(16)	-0.3(6)
C(14)-C(15)-C(16)-C(17)	0.6(6)
C(15)-C(16)-C(17)-C(12)	-1.0(6)
C(13)-C(12)-C(17)-C(16)	1.1(5)
P(1)-C(12)-C(17)-C(16)	-176.7(3)
C(7)-C(6)-O(18)-C(4)	0.7(6)
C(3)-C(4)-O(18)-C(6)	-25.2(5)
C(10)-C(4)-O(18)-C(6)	-145.0(5)
C(8)-C(4)-O(18)-C(6)	93.5(5)
C(12)-P(1)-C(19)-C(24)	-87.2(3)
C(25)-P(1)-C(19)-C(24)	23.0(4)
Pt(1)-P(1)-C(19)-C(24)	135.3(3)
C(12)-P(1)-C(19)-C(20)	94.0(3)
C(25)-P(1)-C(19)-C(20)	-155.7(3)
Pt(1)-P(1)-C(19)-C(20)	-43.4(3)
C(24)-C(19)-C(20)-C(21)	1.6(5)
P(1)-C(19)-C(20)-C(21)	-179.6(3)
C(19)-C(20)-C(21)-C(22)	-1.5(6)
C(20)-C(21)-C(22)-C(23)	0.6(6)
C(21)-C(22)-C(23)-C(24)	0.2(7)
C(22)-C(23)-C(24)-C(19)	-0.1(7)
C(20)-C(19)-C(24)-C(23)	-0.8(6)
P(1)-C(19)-C(24)-C(23)	-179.6(3)
C(12)-P(1)-C(25)-C(26)	179.8(3)
C(19)-P(1)-C(25)-C(26)	68.8(3)
Pt(1)-P(1)-C(25)-C(26)	-47.8(3)
P(1)-C(25)-C(26)-P(2)	28.5(4)
C(29)-P(2)-C(26)-C(25)	118.6(3)
C(27)-P(2)-C(26)-C(25)	-125.7(3)
Pt(1)-P(2)-C(26)-C(25)	3.3(3)
C(29)-P(2)-C(27)-C(28)	-66.4(5)
C(26)-P(2)-C(27)-C(28)	177.2(5)
Pt(1)-P(2)-C(27)-C(28)	56.5(5)
C(27)-P(2)-C(29)-C(30)	85.6(3)
C(26)-P(2)-C(29)-C(30)	-165.3(3)
Pt(1)-P(2)-C(29)-C(30)	-46.9(3)
P(2)-C(29)-C(30)-P(3)	40.0(4)
C(31)-P(3)-C(30)-C(29)	-139.2(3)
C(37)-P(3)-C(30)-C(29)	109.4(3)
Pt(1)-P(3)-C(30)-C(29)	-14.7(3)
C(37)-P(3)-C(31)-C(36)	-123.1(3)
C(30)-P(3)-C(31)-C(36)	126.0(3)

Pt(1)-P(3)-C(31)-C(36)	6.8(3)
C(37)-P(3)-C(31)-C(32)	59.6(3)
C(30)-P(3)-C(31)-C(32)	-51.2(3)
Pt(1)-P(3)-C(31)-C(32)	-170.4(2)
C(36)-C(31)-C(32)-C(33)	0.5(5)
P(3)-C(31)-C(32)-C(33)	177.7(3)
C(31)-C(32)-C(33)-C(34)	-0.4(5)
C(32)-C(33)-C(34)-C(35)	0.0(6)
C(33)-C(34)-C(35)-C(36)	0.4(6)
C(34)-C(35)-C(36)-C(31)	-0.3(6)
C(32)-C(31)-C(36)-C(35)	-0.1(5)
P(3)-C(31)-C(36)-C(35)	-177.4(3)
C(31)-P(3)-C(37)-C(38)	15.0(3)
C(30)-P(3)-C(37)-C(38)	125.7(3)
Pt(1)-P(3)-C(37)-C(38)	-115.2(3)
C(31)-P(3)-C(37)-C(42)	-165.7(2)
C(30)-P(3)-C(37)-C(42)	-55.0(3)
Pt(1)-P(3)-C(37)-C(42)	64.1(3)
C(42)-C(37)-C(38)-C(39)	1.3(5)
P(3)-C(37)-C(38)-C(39)	-179.4(3)
C(37)-C(38)-C(39)-C(40)	-0.1(6)
C(38)-C(39)-C(40)-C(41)	-0.7(6)
C(39)-C(40)-C(41)-C(42)	0.3(6)
C(40)-C(41)-C(42)-C(37)	0.9(6)
C(38)-C(37)-C(42)-C(41)	-1.7(5)
P(3)-C(37)-C(42)-C(41)	179.0(3)

---