

## Synthesis of (+)-Coronafacic Acid

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### Supporting Information

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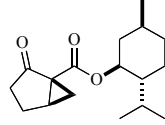
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## Experimental Section

**General.**  $^1\text{H}$  NMR (at 400 MHz) and  $^{13}\text{C}$  NMR (at 400 MHz) spectra were obtained as solutions in deuteriochloroform ( $\text{CDCl}_3$ ). The infrared (IR) spectra were determined as neat oils.  $R_f$  values indicated refer to thin layer chromatography (TLC) on 5.0 x 10 cm, 250  $\mu\text{m}$  analytical plates coated with silica gel 60 F<sub>254</sub>, developed in the solvent system indicated. Column chromatography was carried out on using silica gel 60 particle size 0.015 – 0.040 mm. The solvent mixtures reported are volume/volume mixtures. EtOAc is ethyl acetate. All glassware was oven dried and reactions were carried out under a flow of nitrogen. All Wittig reactions were stirred magnetically, under dry  $\text{N}_2$  unless otherwise noted.  $\text{Fe}(\text{CO})_5$  mediated carbonylation was performed in a Rayonet reactor using a tube inside a tube setup, such that the reaction mixture is a thin layer between the outer tube and the inner tube.

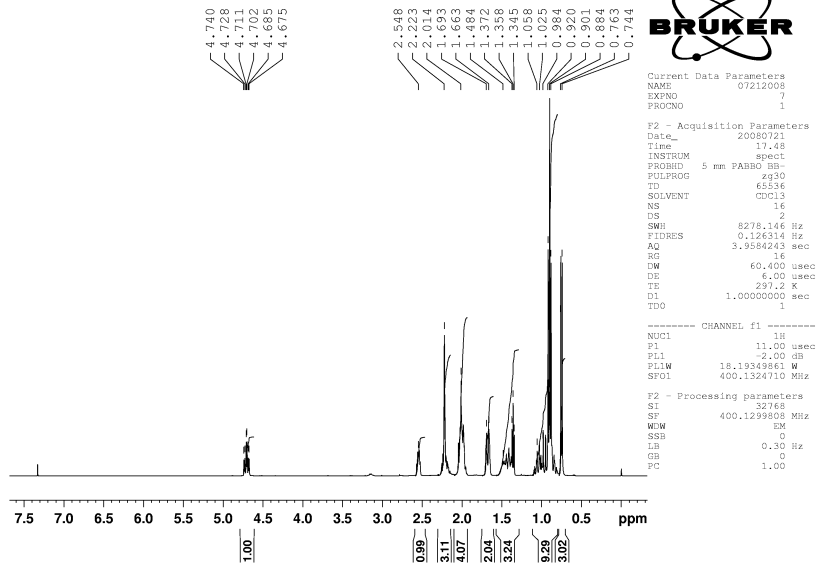


**Diazo ester 1:** 5-hexen-2-one (20 g, 200 mmol) was mixed with NaH (20 g) in toluene (200 mL) at rt. The mixture was heated to reflux for 10 min before dimethyl carbonate (56 g) was added dropwise over 30 min. Caution: the reaction is highly exothermic! After an additional 3 h at reflux, the mixture was partitioned between MTBE and 1N aqueous HCl and brine. The combined organic extract was dried ( $\text{Na}_2\text{SO}_4$ ) and concentrated. The residue was bulb to bulb distilled to afford the  $\beta$ -ketoester as a colorless oil (27.24 g, 86% yield). TLC  $R_f(\text{PE}/\text{MTBE} = 8/2) = 0.45$ . The  $\beta$ -ketoester (8.80 g, 56 mmol) was mixed with (+)-menthol (7.4 g), DMAP (1.15 g) and molecular sieve (4 Å, 12.5 g) in toluene (150 mL). The reaction mixture was heated to reflux overnight, then passed through a pad of silica gel. The eluant was concentrated and the residue was chromatographed to afford the menthol ester as a colorless oil (9.27 g, 60% yield). TLC  $R_f(\text{PE}/\text{MTBE} = 8/2) = 0.80$ . The  $\beta$ -ketoester (4.13 g, 14.7 mmol) was mixed with TEA (3.1 g, 30 mmol) and mesyl azide (2.7 g, 22 mmol) in MeCN (50 mL) at rt. After 8 h the solvent was removed *in vacuo*. The residue was partitioned between MTBE and 1N aqueous NaOH, saturated aqueous  $\text{NaHCO}_3$  and brine. The combined organic extract was dried ( $\text{Na}_2\text{SO}_4$ ) and concentrated. The residue was chromatographed to afford the diazo ester **1** as a colorless oil (4.36 g, 72% yield). TLC  $R_f(\text{PE}/\text{MTBE} = 8/2) = 0.50$ . The spectroscopic data was consistent with those that were earlier reported.<sup>11a</sup>

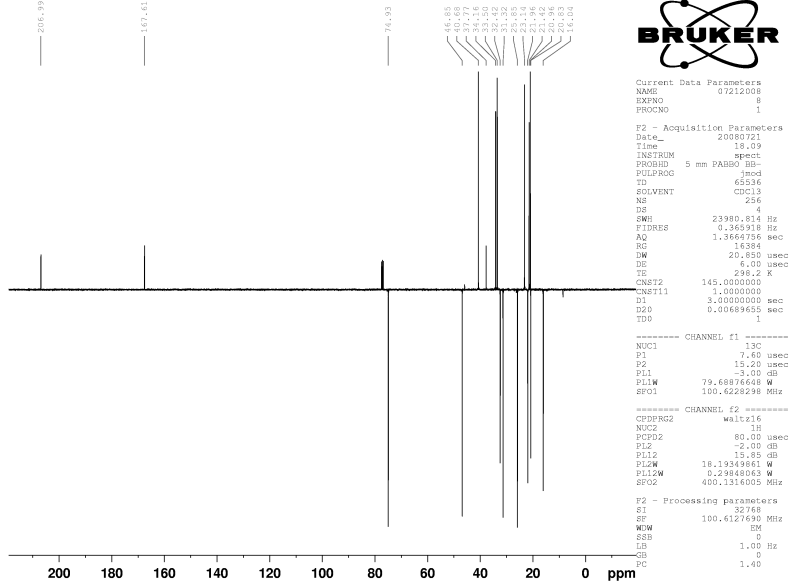


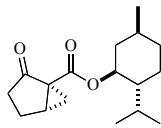
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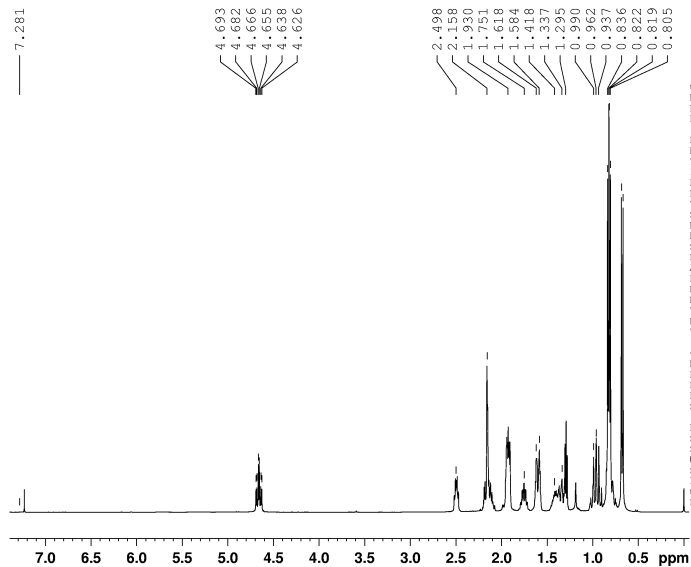
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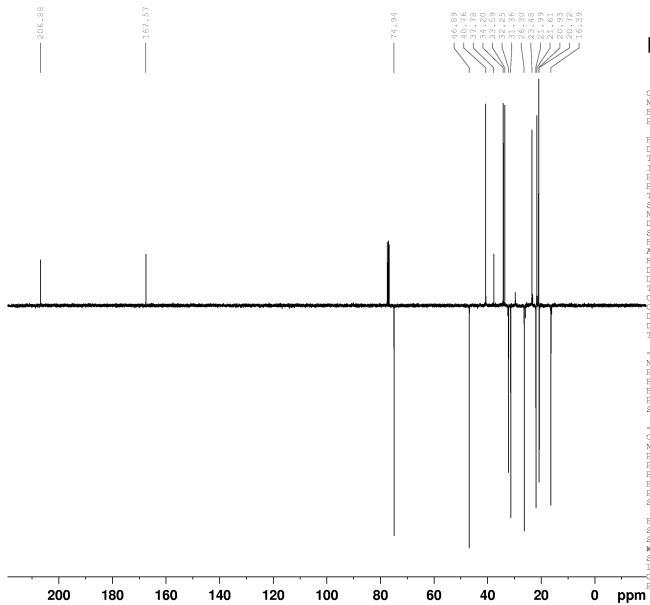
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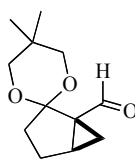
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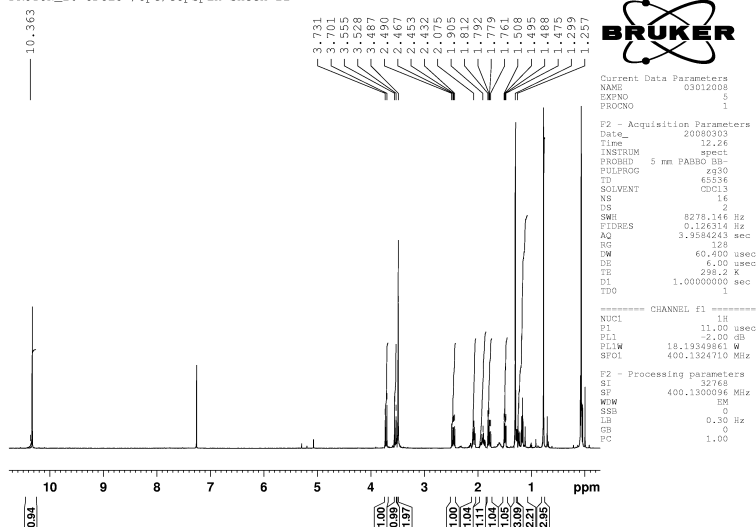
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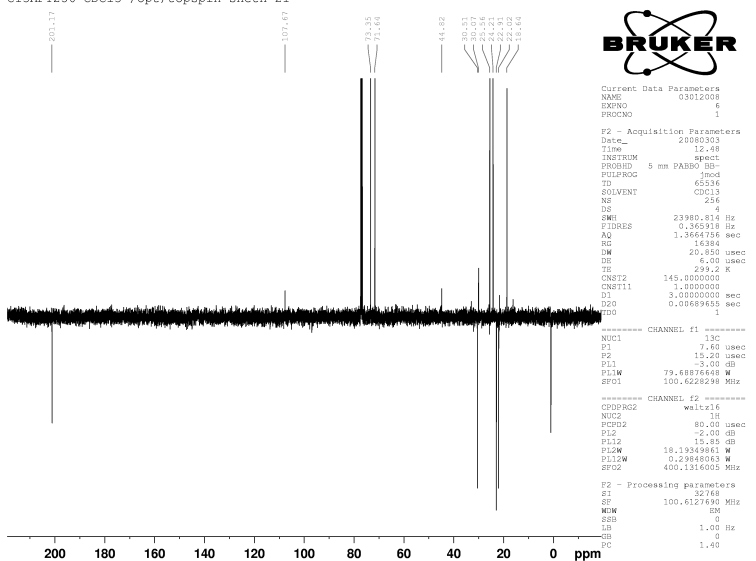


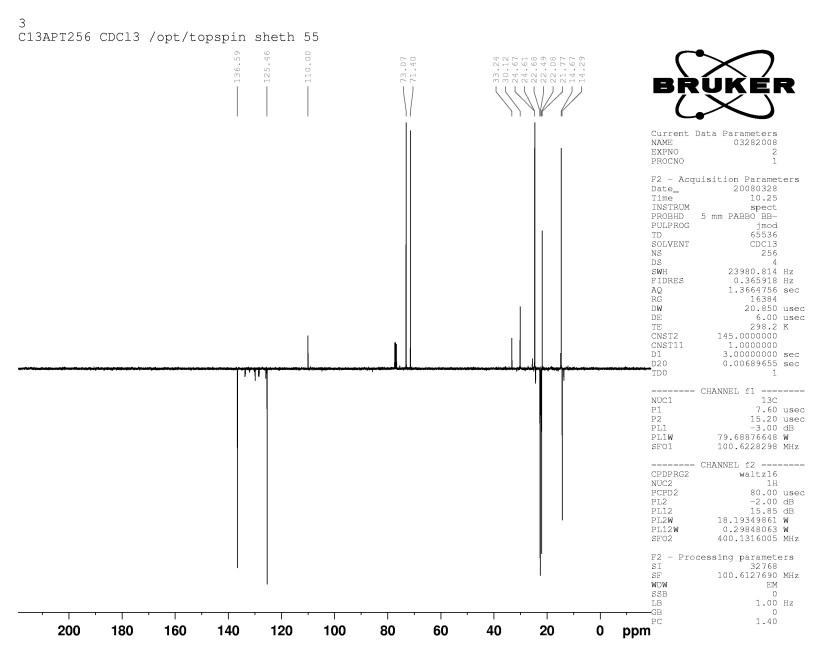
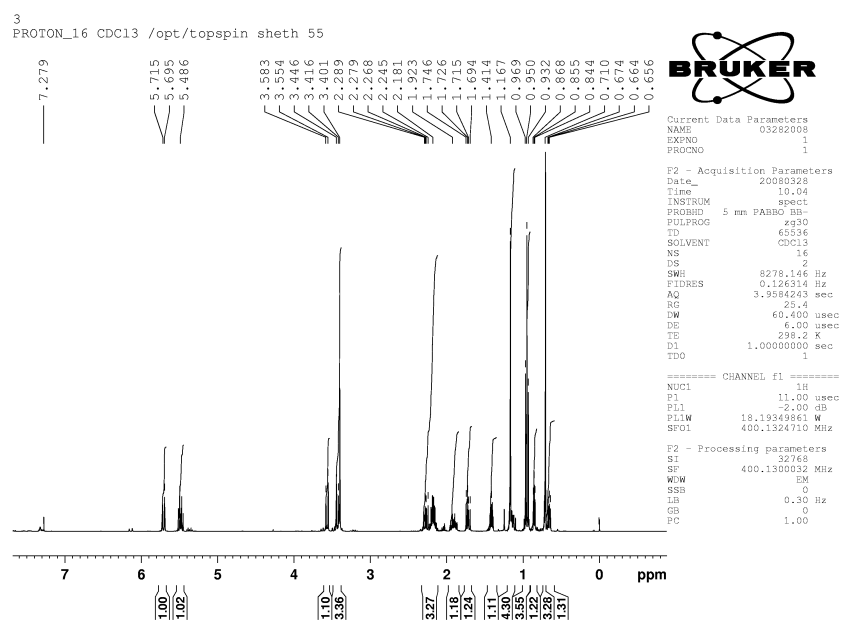
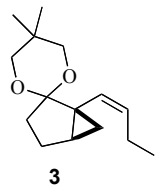
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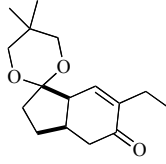


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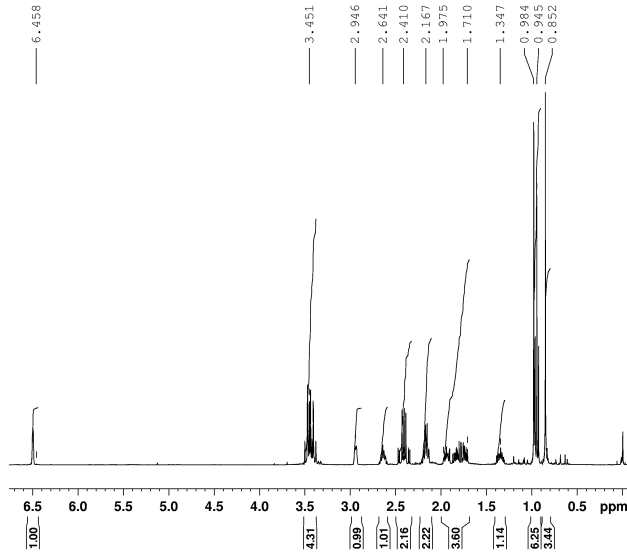






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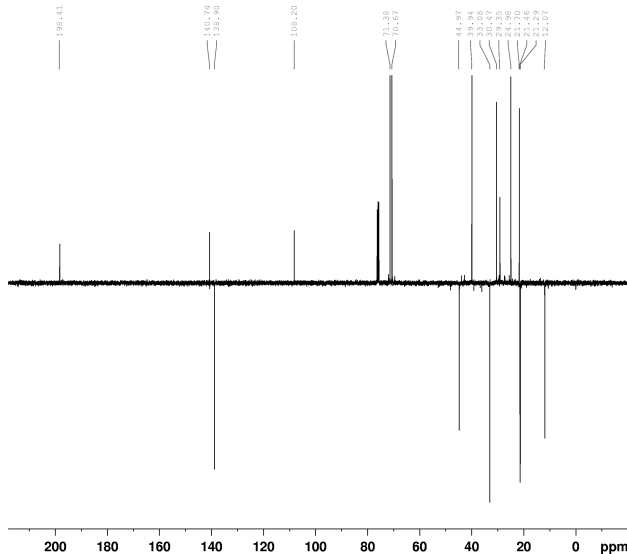
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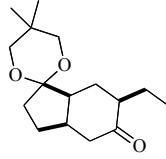
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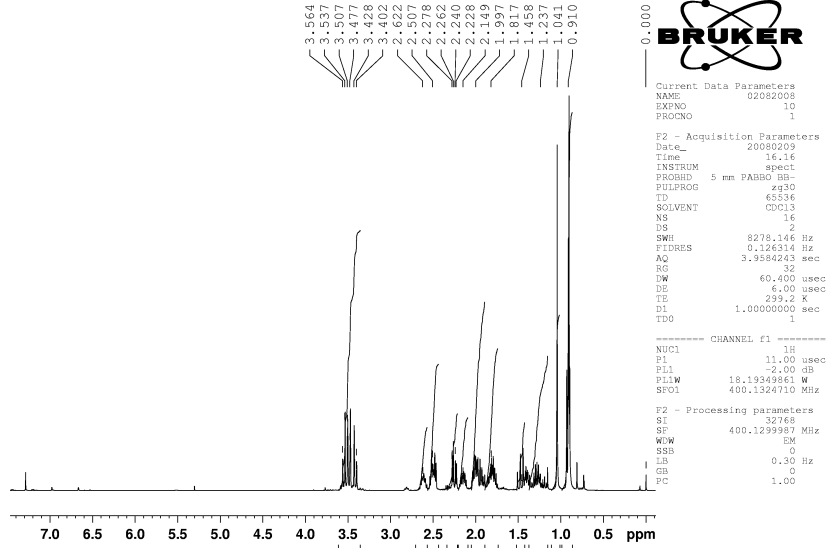
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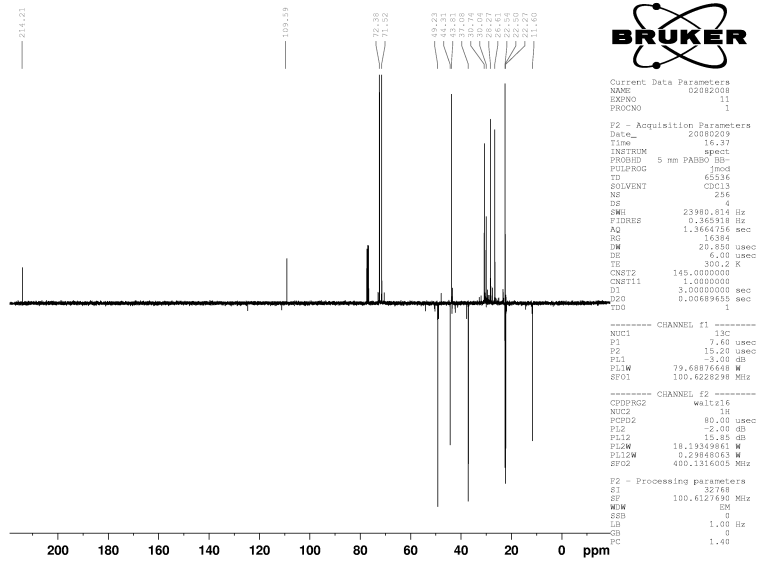


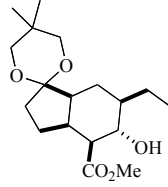
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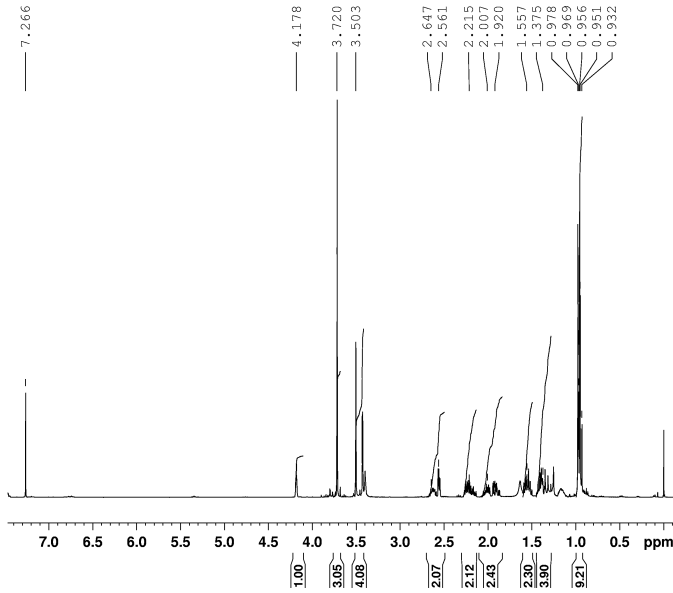
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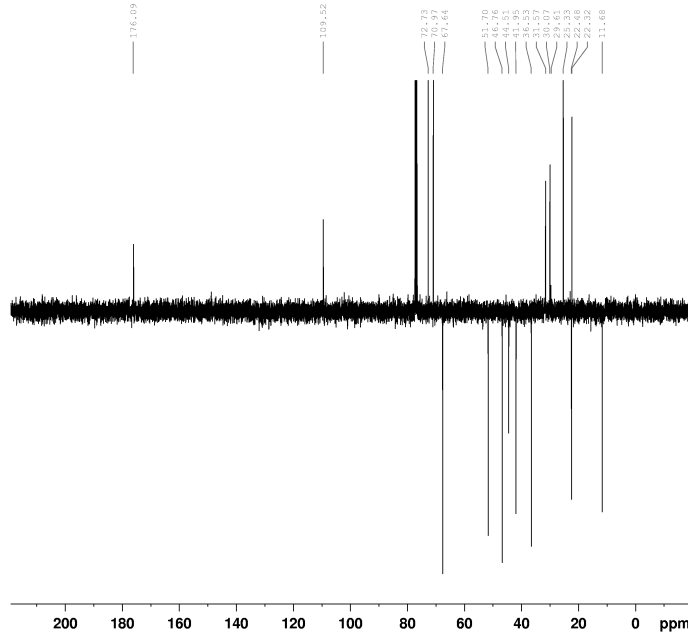
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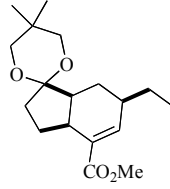
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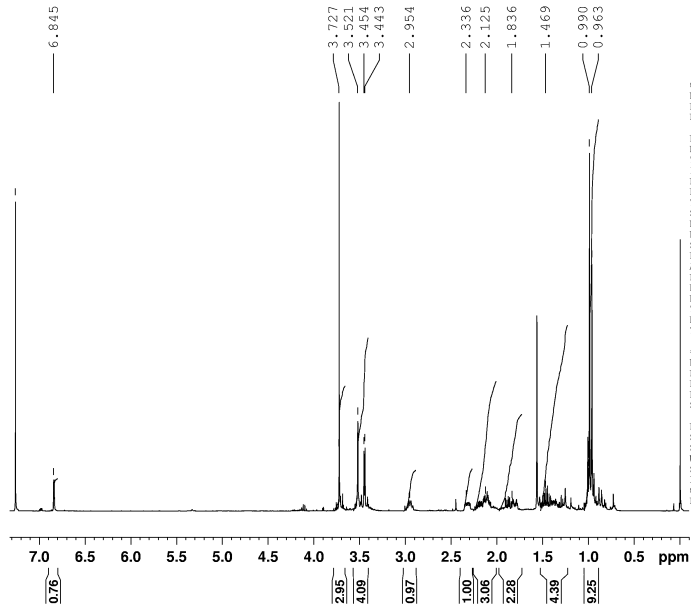
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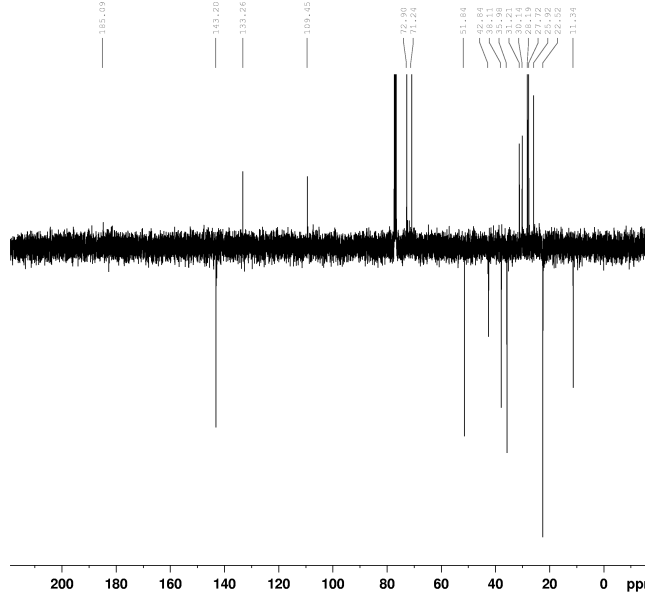
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