

## Supplemental Materials (Molecular Mechanics Parameter Development)

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## Parameters for Mn2+ and Fosfomycin taken from Amber Parameter Database ##  
## Website - http://pharmacy.man.ac.uk/amber/ ##  
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Mn<sup>2+</sup> vdw parameters (from Concanavalin A study by G.M. Bradbrook et al.)

MASS

MN 55.00 0.00

NONB

MN 1.6900 0.0140

Manganese parameters (cont.)

mn.prp

MNG INT 0

CHANGE OMIT DU BEG

0.0

1	DUMM	DU	M	0	-1	-2	0.000	0.000	0.000	0.000
2	DUMM	DU	M	1	0	-1	1.000	0.000	0.000	0.000
3	DUMM	DU	M	2	1	0	1.000	90.000	0.000	0.000
4	MN	MN	M				29.683	127.561	-18.450	2.000

DONE

STOP

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Fosfomycin parameters (calix[4]arenes: PhD Thesis of Marc Baaden - 250800)

MASS

BOND

CT-P 331. 1.82

ANGLE

CT-P -O2	60.	104.6	angles from gopt / force const. (Baaden)
CT-CT-P	35.	113.1	" "
OS-CT-P	50.	124.3	" "
P -CT-H2	60.	127.3	" "

DIHEDRAL

X -CT-P -X 9 3.51 0. 3. tpl - JPC 94

NONBOND

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#####
## Parameters for Fosfomycin and Glutathione (ILG) developed in collaboration ##
## with the research group of Professor Terry Lybrand ##
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Fosfomycin parameters (cont.)

fos.prp

FCN INT 0

CORRECT OMIT DU BEG

0.0

1	DU1	DU	M	0	0	0	0.000	0.000	0.000	
2	DU2	DU	M	1	0	0	1.000	0.000	0.000	
3	DU3	DU	M	2	1	0	1.000	90.000	0.000	
4	C1	CT	M	3	2	1	0.559	119.220	156.267	-0.0043
5	H11	HC	E	4	3	2	1.090	126.907	305.732	-0.0119
6	H12	HC	E	4	3	2	1.092	119.722	97.641	-0.0119
7	H13	HC	E	4	3	2	1.091	73.080	203.685	-0.0119
8	C2	CT	M	4	3	2	1.511	33.238	14.929	0.0434
9	H2	H1	E	8	4	3	1.083	116.259	184.228	-0.0280
10	O3	OS	M	8	4	3	1.469	114.639	49.091	-0.4633
11	C4	CT	M	10	8	4	1.518	59.584	247.771	0.0872
12	H4	H2	E	11	10	8	1.082	108.886	253.746	-0.0113
13	P	P	M	11	10	8	1.877	124.338	116.285	1.0550
14	O2P	O2	E	13	11	10	1.504	103.349	74.682	-0.8810
15	O3P	O2	E	13	11	10	1.519	104.574	313.185	-0.8810
16	O1P	O2	E	13	11	10	1.508	98.477	194.010	-0.8810

LOOP

C2 C4

DONE

STOP

ILG - amino-terminal  $\gamma$ -glutamyl residue from glutathione (GSH)

ILG.prp

ILG INT 0

CORRECT OMIT DU BEG

0.0

1	DU1	DU	M	0	0	0	0.000	0.000	0.000	
2	DU2	DU	M	1	0	0	1.000	0.000	0.000	
3	DU3	DU	M	2	1	0	1.000	90.000	0.000	
4	N	N3	M	3	2	1	1.262	149.168	315.313	-0.8689
5	H1	H	E	4	3	2	1.005	133.615	120.590	0.3395
6	H2	H	E	4	3	2	1.110	83.034	220.761	0.3395
7	H3	H	E	4	3	2	1.007	111.441	327.265	0.3395
8	CA	CT	M	4	3	2	1.480	29.196	65.498	0.2825
9	HA	HP	E	8	4	3	1.085	107.123	107.881	0.0238
10	C	C	B	8	4	3	1.536	107.353	354.046	0.6306
11	O	O2	E	10	8	4	1.339	113.413	20.621	-0.5382
12	OXT	O2	E	10	8	4	1.203	123.852	198.954	-0.5382
13	CB	CT	M	8	4	3	1.533	113.868	229.142	0.0122
14	HB2	HC	E	13	8	4	1.081	108.878	311.188	0.0095
15	HB3	HC	E	13	8	4	1.085	109.375	68.078	0.0095
16	CG	CT	M	13	8	4	1.533	113.389	190.403	-0.0498
17	HG2	HC	E	16	13	8	1.078	110.008	294.996	0.0086
18	HG3	HC	E	16	13	8	1.088	109.915	53.520	0.0086
19	CD	C	M	16	13	8	1.519	110.542	171.884	0.5449
20	OD	O	E	19	16	13	1.222	122.434	343.497	-0.5536

DONE

STOP

## References from Supplemental Materials:

M. Baaden\*, G. Wipff, F. Berny; \*Laboratory of Molecular Biophysics, University of Oxford  
M. Baaden, G. Wipff, M. R. Yaftian, M. Burgard and D. Matt (2000) Cation coordination by calix(*I*)arenes bearing amide and/or phosphine oxide pendant groups: how many arms are needed to bind Li<sup>+</sup> vs Na<sup>+</sup> ? A combined NMR and molecular dynamics study. *J. chem. Soc., Perkin Trans. 2.*, 1315.

Bradbrook GM, Gleichmann T, Harrop SJ, Habash J, Raftery J, Kalb J, Yariv J, Hillier IH, Helliwell JR (1998) X-ray and molecular dynamics studies of concanavalin-A glucoside and mannoside complexes - Relating structure to thermodynamics of binding. *Journal of the Chemical Society-Faraday Transactions.* 94,1603.

Ewing, T.J.A. and Lybrand, T.P. (1994) A Comparison of Perturbation Methods and Poisson-Boltzmann Electrostatics Calculations for Estimation of Relative Solvation Free Energies. *J Phys Chem* **98**:1748-1752.

D.A. Pearlman, D.A. Case, J.W. Caldwell, W.S. Ross, T.E. Cheatham, III, S. DeBolt, D. Ferguson, G. Seibel, and P. Kollman. AMBER, a package of computer programs for applying molecular mechanics, normal mode analysis, molecular dynamics and free energy calculations to simulate the structural and energetic properties of molecules. *Comp. Phys. Commun.*91,1-41 (1995).