

## **Supporting Information**

# **Structural and Synthetic Investigations of Tanikolide Dimer, a SIRT2 Selective Inhibitor, and Tanikolide Seco Acid from the Madagascar Marine Cyanobacterium *Lyngbya majuscula***

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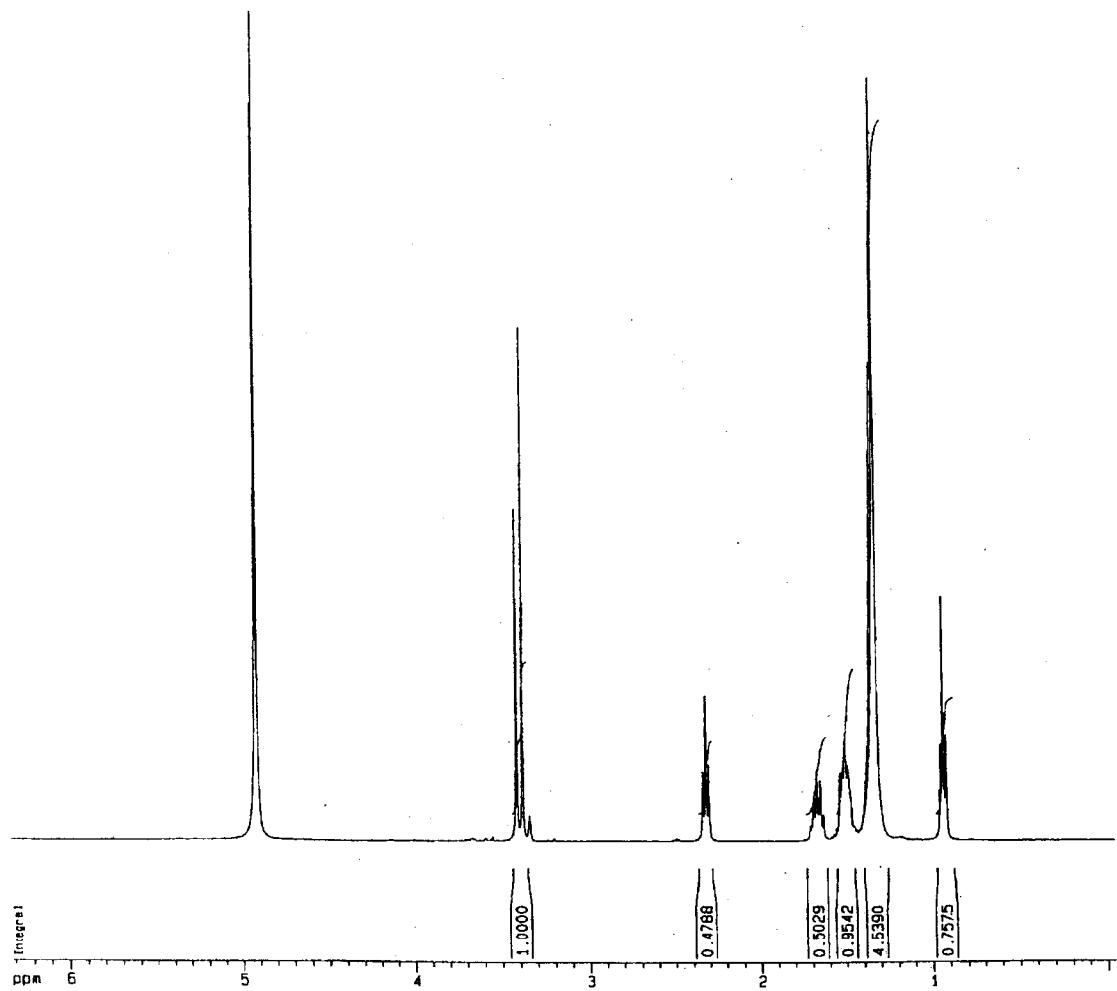
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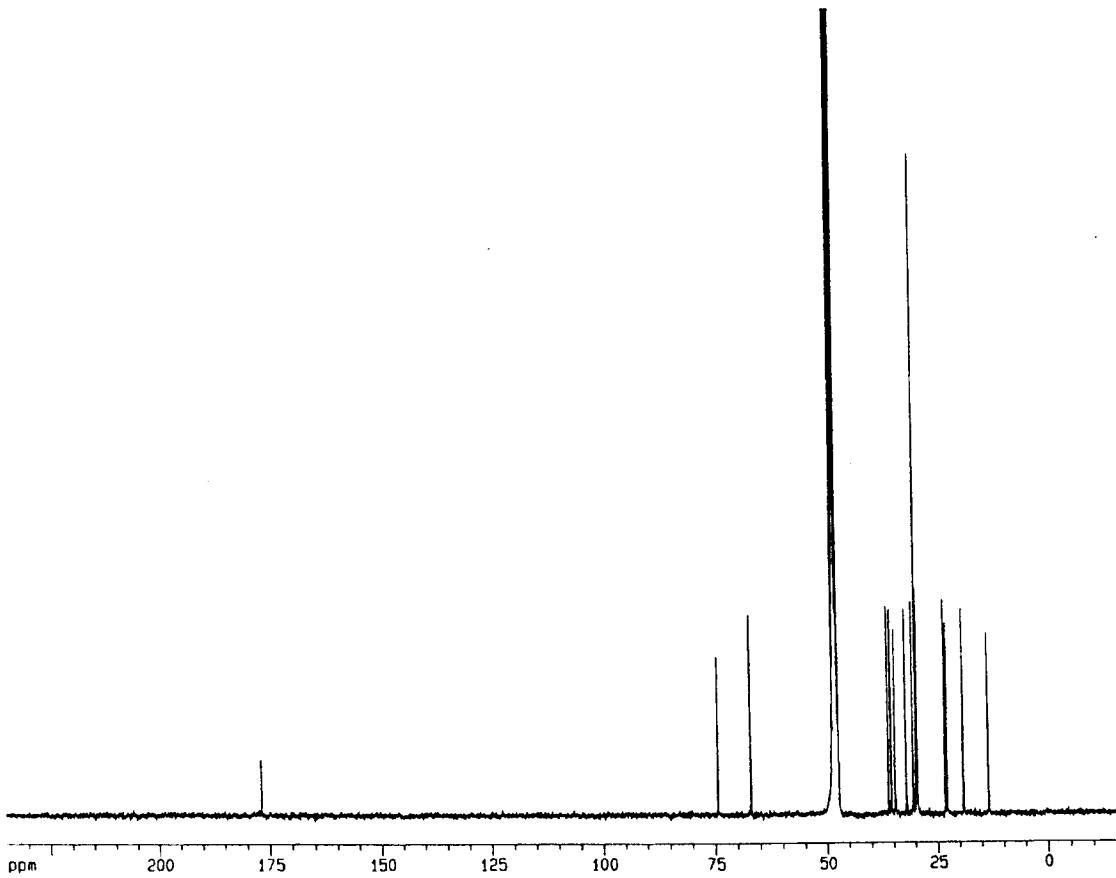
**General Experimental Methods – Natural Products:** NMR spectra were referenced to residual solvent signal with resonances at  $\delta_{\text{H/C}}$  7.26 / 77.1 ( $\text{CDCl}_3$ ) and  $\delta_{\text{H/C}}$  3.31/49.15 ( $\text{CD}_3\text{OD}$ ). X-ray diffraction data were collected on a diffractometer with a HiStar Cu K $\alpha$  radiation area detector. Structure solution and refinement was carried out using SHELXS and SHELXL, respectively. Chiral GC-MS analyses was performed on a Cyclosil-B chiral column (30m x 0.250 mm).

**General Experimental Methods – Synthesis:** NMR chemical shift values were recorded as parts per million relative to tetramethylsilane as an internal standard unless otherwise indicated, and coupling constants in Hertz. The progress of chemical reactions were checked on Merck TLC plates (Merck 5554 Kiesel gel 60 F254), and the spots were visualized under 254 nm UV light and/or by charring after dipping the TLC plate into a vanillin solution (9.0 g of vanillin and 1.5 mL of concentrated  $\text{H}_2\text{SO}_4$  in 300 mL of MeOH), a  $\text{KMnO}_4$  solution (3 g of  $\text{KMnO}_4$ , 20 g of  $\text{K}_2\text{CO}_3$ , and 5 mL of 5% NaOH solution in 300 mL of H<sub>2</sub>O), or a phosphomolybdic acid solution (250 mg phosphomolybdic acid in 50 mL EtOH). Column chromatography was performed on Merck silica gel (9385 Kiesel gel 60) using hexanes-EtOAc (v/v). The solvents were simple distilled unless otherwise noted. Unless otherwise specified, all reactions were conducted under a slight positive pressure of dry N<sub>2</sub>. The usual work-up refers to washing the quenched reaction mixture with brine, drying the organic extracts over anhyd MgSO<sub>4</sub> and evaporating under reduced pressure using a rotary evaporator. Solvents used in the reactions were dried under a N<sub>2</sub> atmosphere. THF was distilled from Nabenzophenone, and CH<sub>2</sub>Cl<sub>2</sub> was distilled from P<sub>2</sub>O<sub>5</sub>. Benzene was washed with conc. H<sub>2</sub>SO<sub>4</sub>, distilled from Na-benzophenone, and stored over 4 Å molecular sieves. Et<sub>2</sub>O was distilled from LiAlH<sub>4</sub>. CH<sub>3</sub>CN was distilled from CaH<sub>2</sub> and stored over 4 Å molecular sieves. Pyridine and triethylamine were distilled over KOH and stored over 4 Å molecular sieves.

400 MHz  $^1\text{H}$  NMR of compound **2** in  $\text{CD}_3\text{OD}$



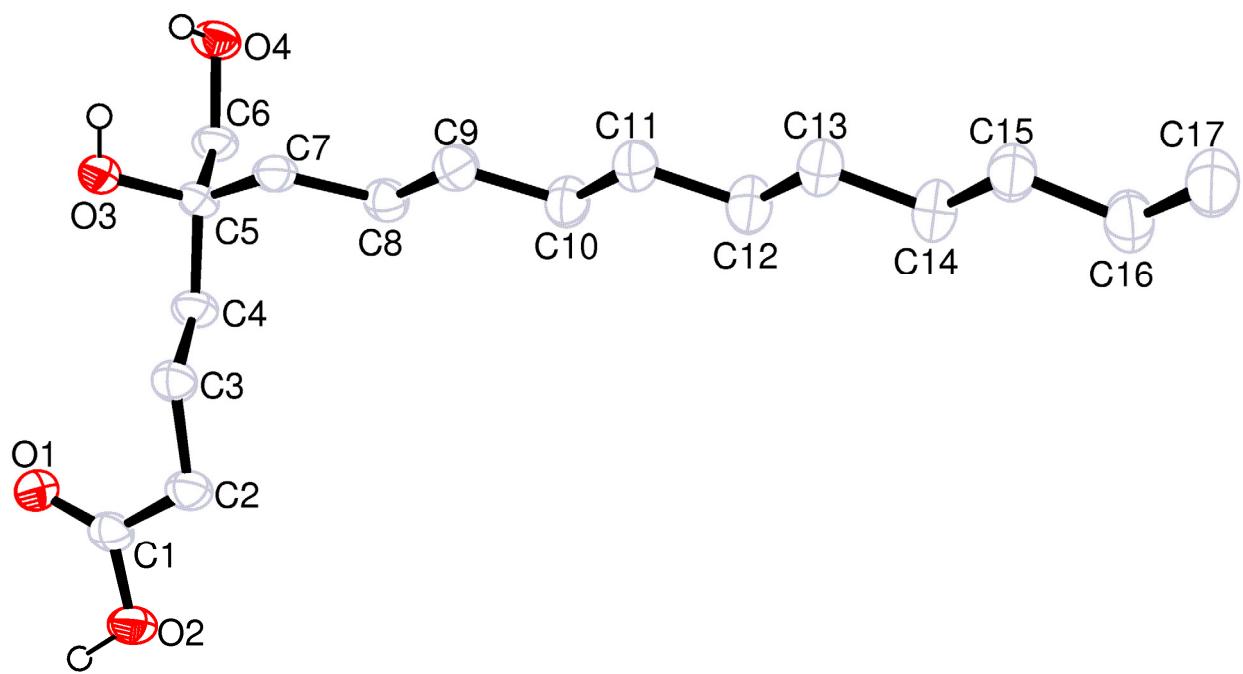
100 MHz  $^{13}\text{C}$  NMR of compound **2** in  $\text{CD}_3\text{OD}$



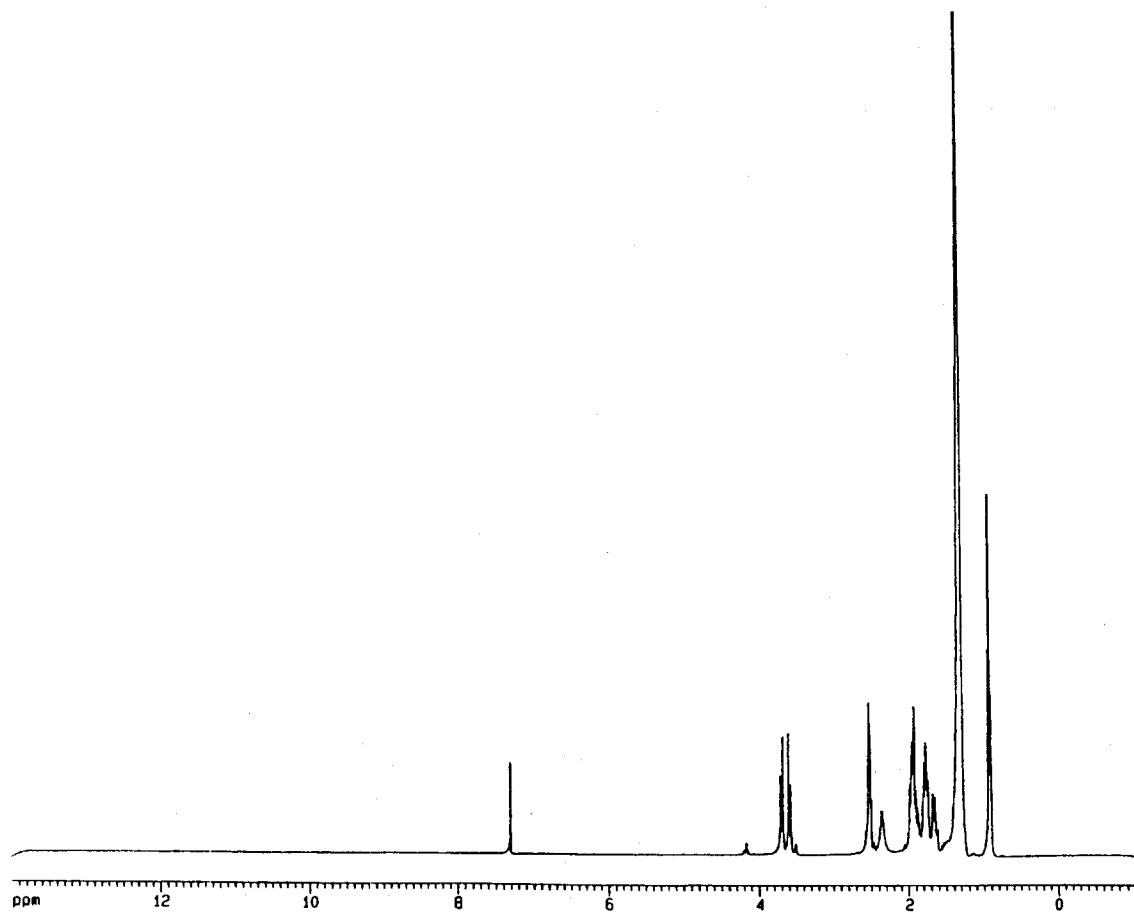
X-ray crystallographic data and structure refinement for compound **2**

Identification code	WG112003		
Empirical formula	$C_{17}H_{34}O_4$		
Formula weight	302.44		
Temperature	100(2) K		
Wavelength	1.54180 Å		
Crystal system	Monoclinic		
Space group	P2 <sub>1</sub>		
Unit cell dimensions	$a = 5.524(4)$ Å	$\alpha = 90^\circ$	
	$b = 8.048(5)$ Å	$\beta = 97.180(14)^\circ$	
	$c = 20.295(12)$ Å	$\gamma = 90^\circ$	
Volume	895.2(10) Å <sup>3</sup>		
Z	2		
Density (calculated)	1.122 Mg/m <sup>3</sup>		
Absorption coefficient	0.618 mm <sup>-1</sup>		
F(000)	336		
Crystal size	0.30 x 0.10 x 0.02 mm <sup>3</sup>		
Theta range for data collection	4.39 to 71.19°.		
Index ranges	0≤h≤6, -9≤k≤9, -24≤l≤24		
Reflections collected	8951		
Independent reflections	2294 [R(int) = 0.1640]		
Completeness to theta = 71.19°	85.5 %		
Absorption correction	Semi-empirical from equivalents		
Max. and min. transmission	1.0000 and 0.1337		
Refinement method	Full-matrix least-squares on F <sup>2</sup>		
Data / restraints / parameters	2294 / 1 / 196		
Goodness-of-fit on F <sup>2</sup>	1.071		
Final R indices [I>2sigma(I)]	R1 = 0.0882, wR2 = 0.2253		
R indices (all data)	R1 = 0.0975, wR2 = 0.2358		
Absolute structure parameter	0.5(7)		
Largest diff. peak and hole	0.274 and -0.339 e.Å <sup>-3</sup>		

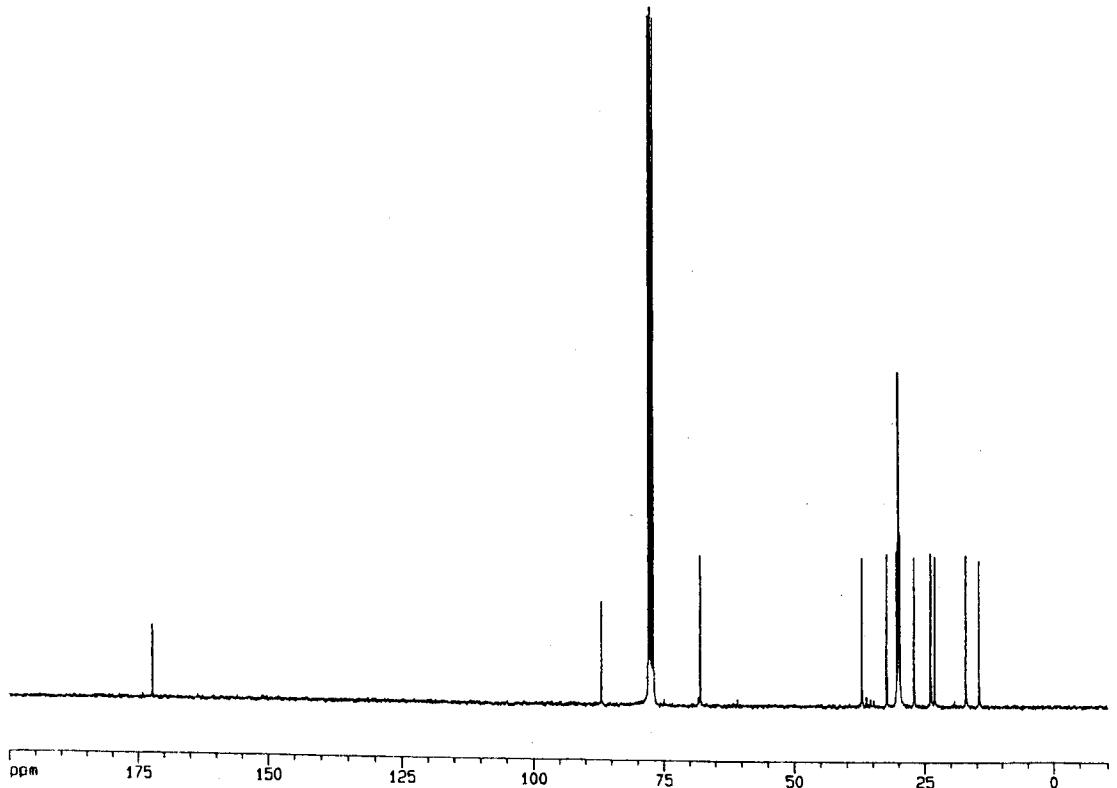
ORTEP of tanikolide seco acid **2** with 50% displacement ellipsoids



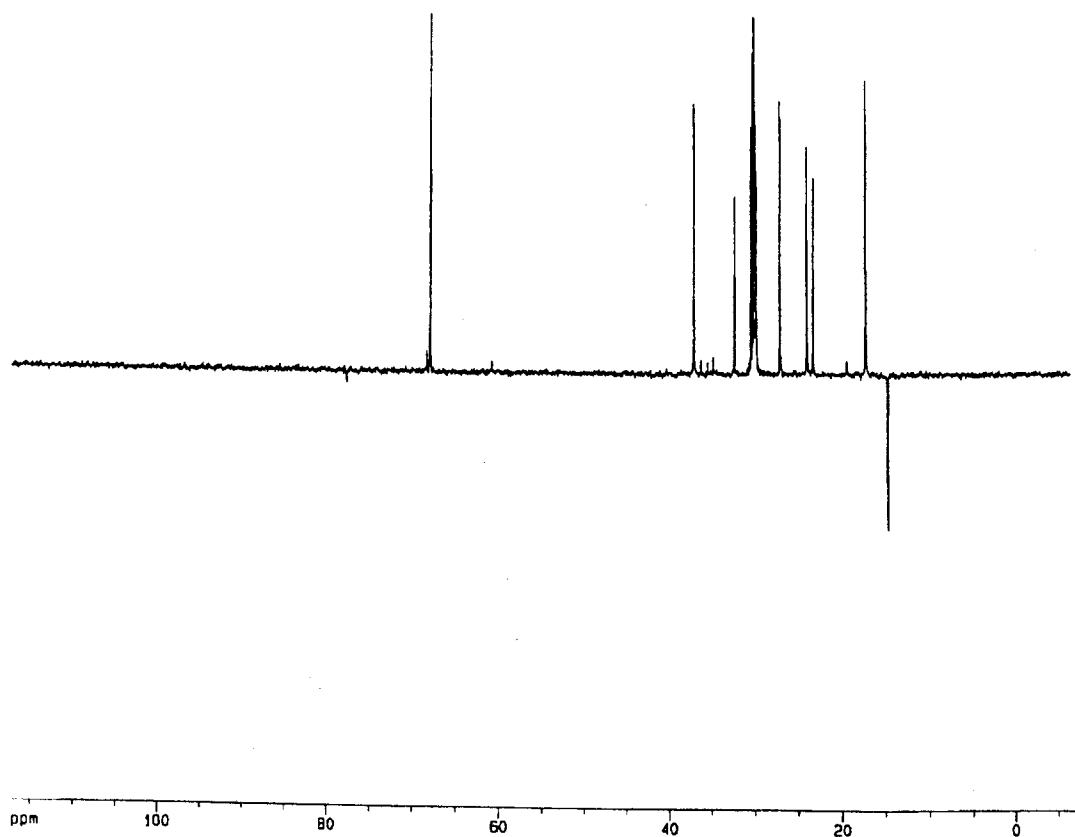
400 MHz  $^1\text{H}$  NMR of compound **3** in  $\text{CDCl}_3$



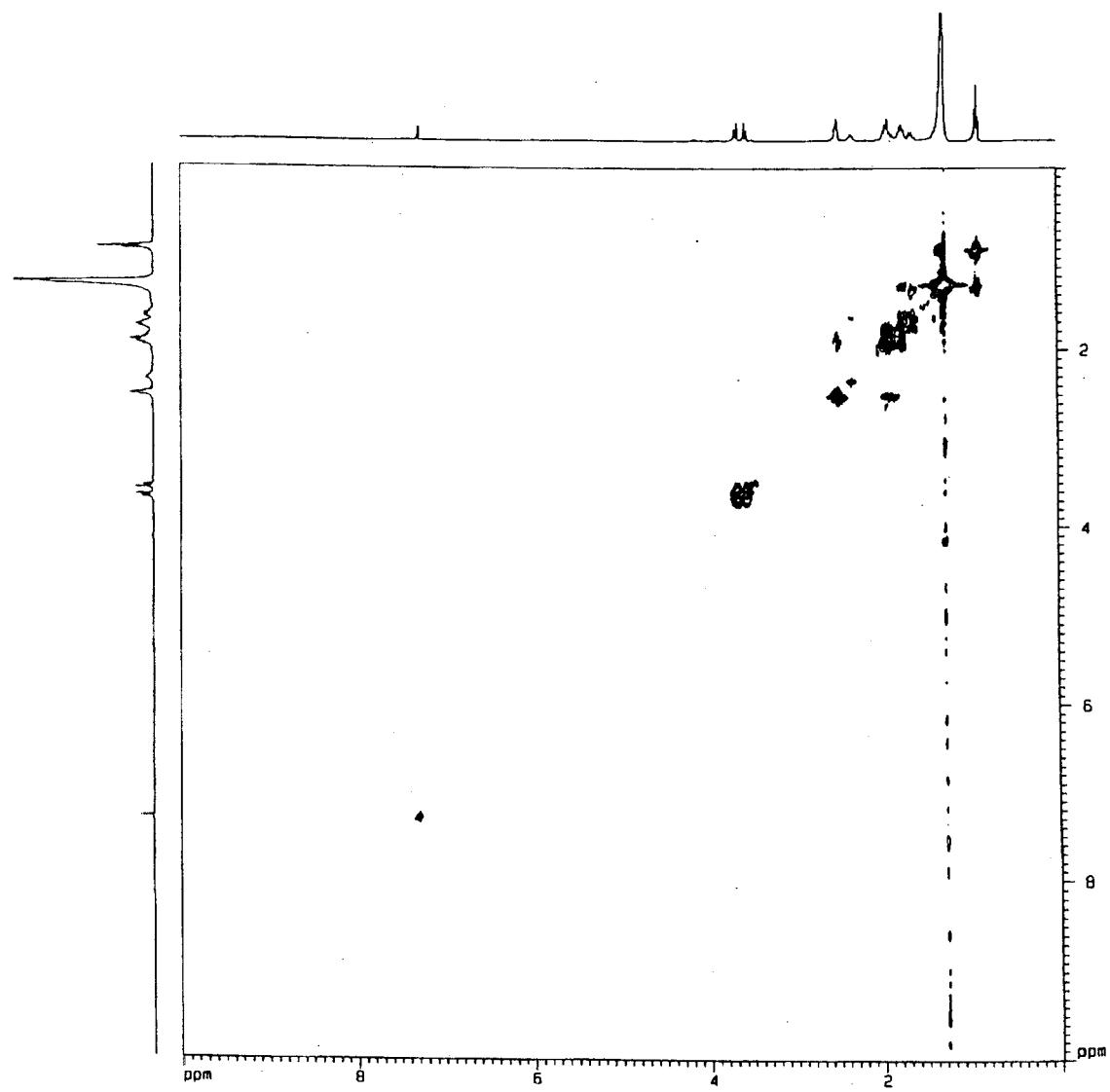
100 MHz  $^{13}\text{C}$  NMR of compound **3** in  $\text{CDCl}_3$



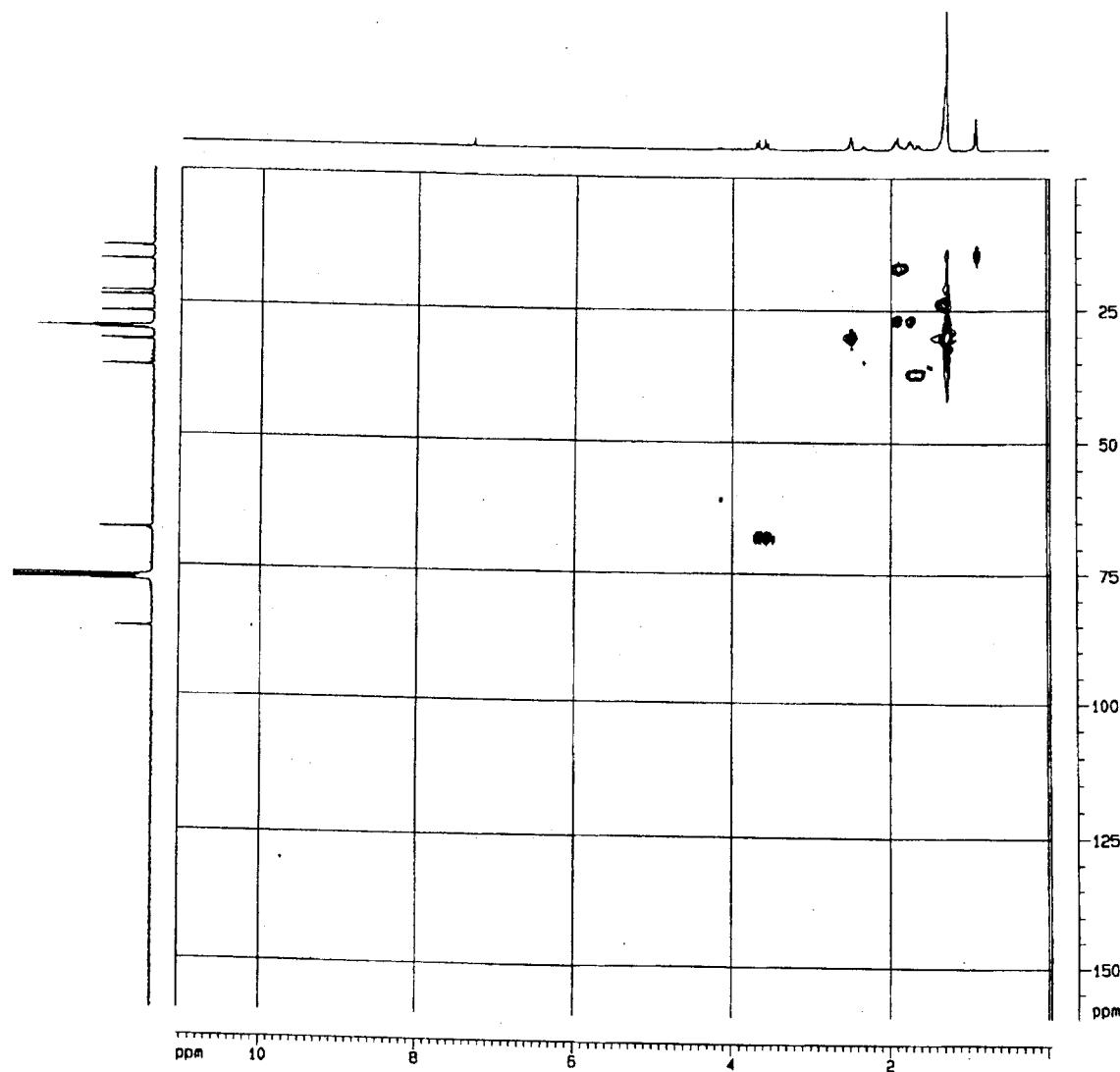
100 MHz DEPT 135 spectrum of compound **3** in  $\text{CDCl}_3$



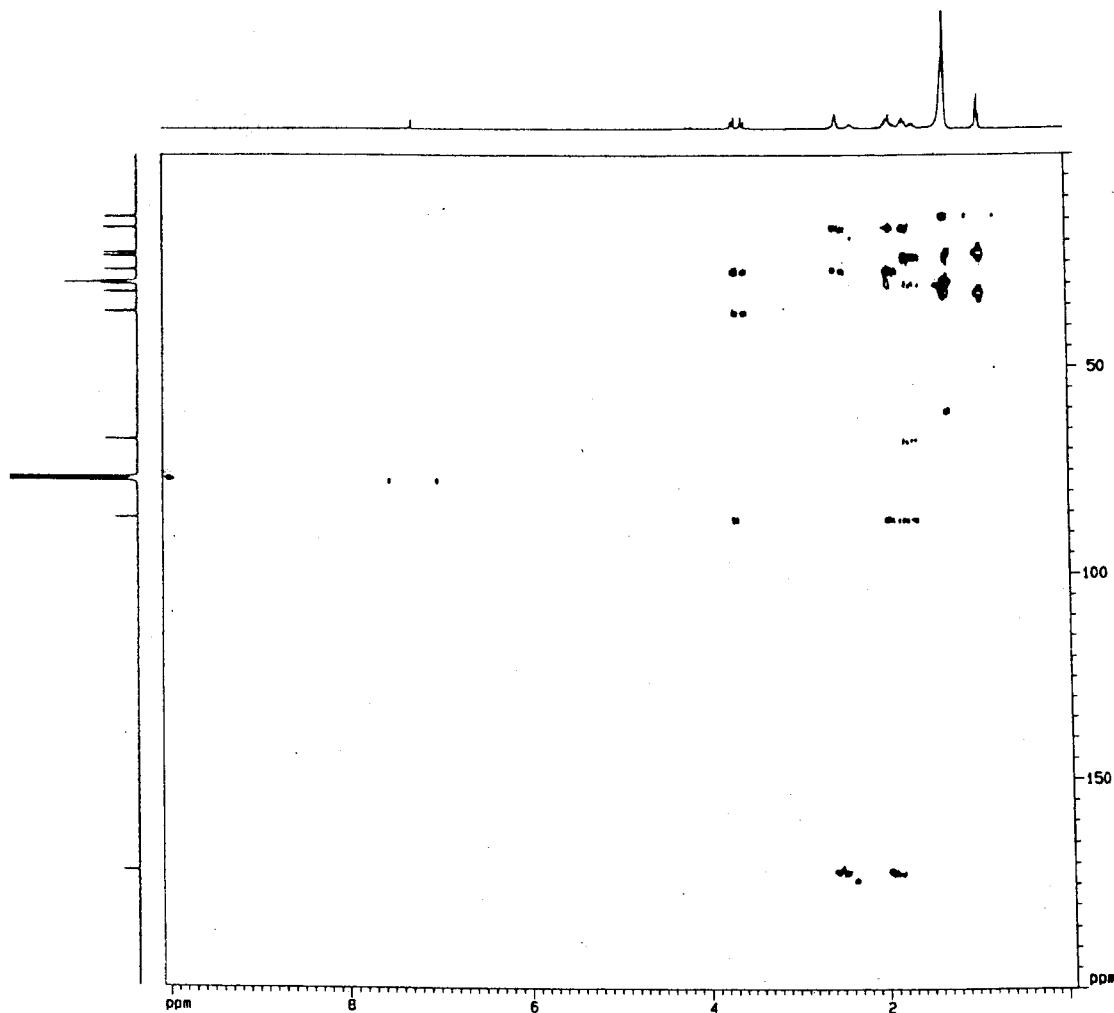
400 MHz  $^1\text{H}$ - $^1\text{H}$  COSY spectrum of compound **3** in  $\text{CDCl}_3$



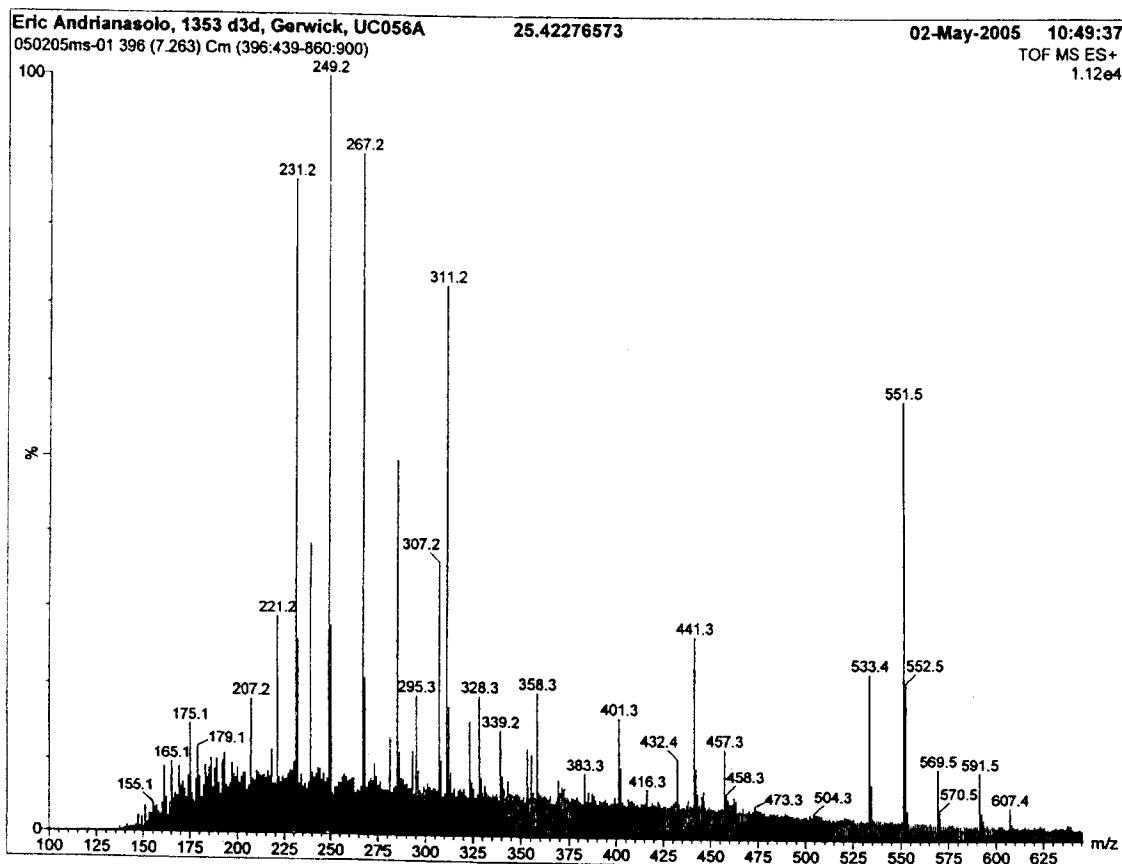
400 MHz Multi-edited HSQC spectrum of compound 3 in  $\text{CDCl}_3$



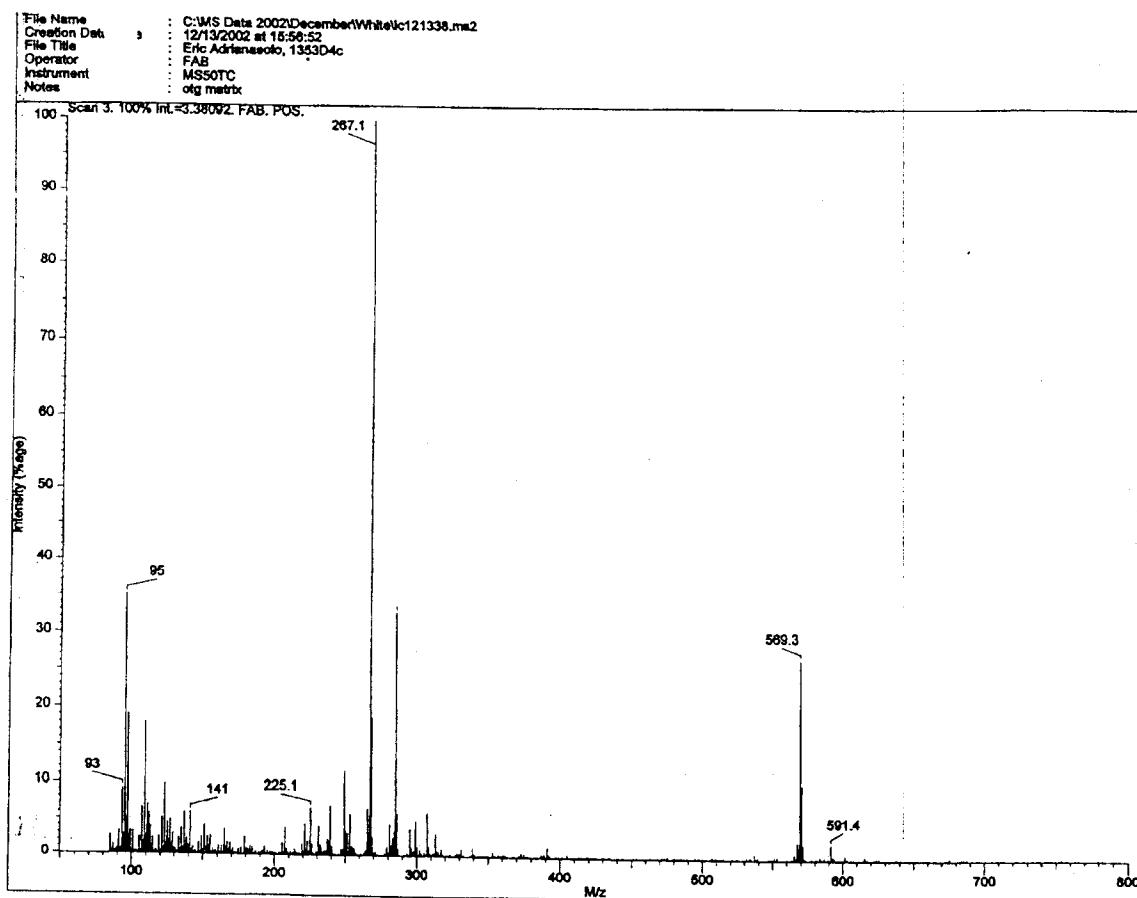
400 MHz HMBC spectrum of compound **3** in  $\text{CDCl}_3$  optimized for  $J = 8 \text{ Hz}$



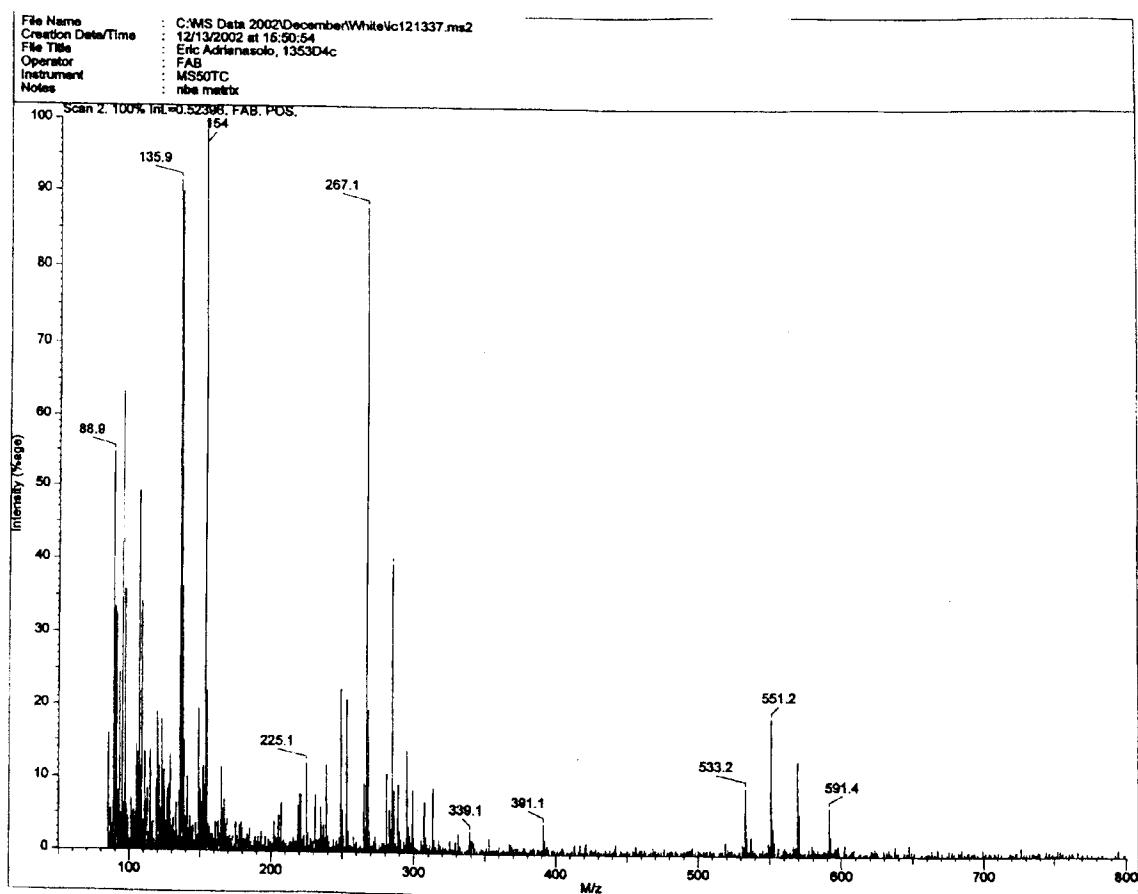
LR TOF MS ES+ spectrum of compound 3

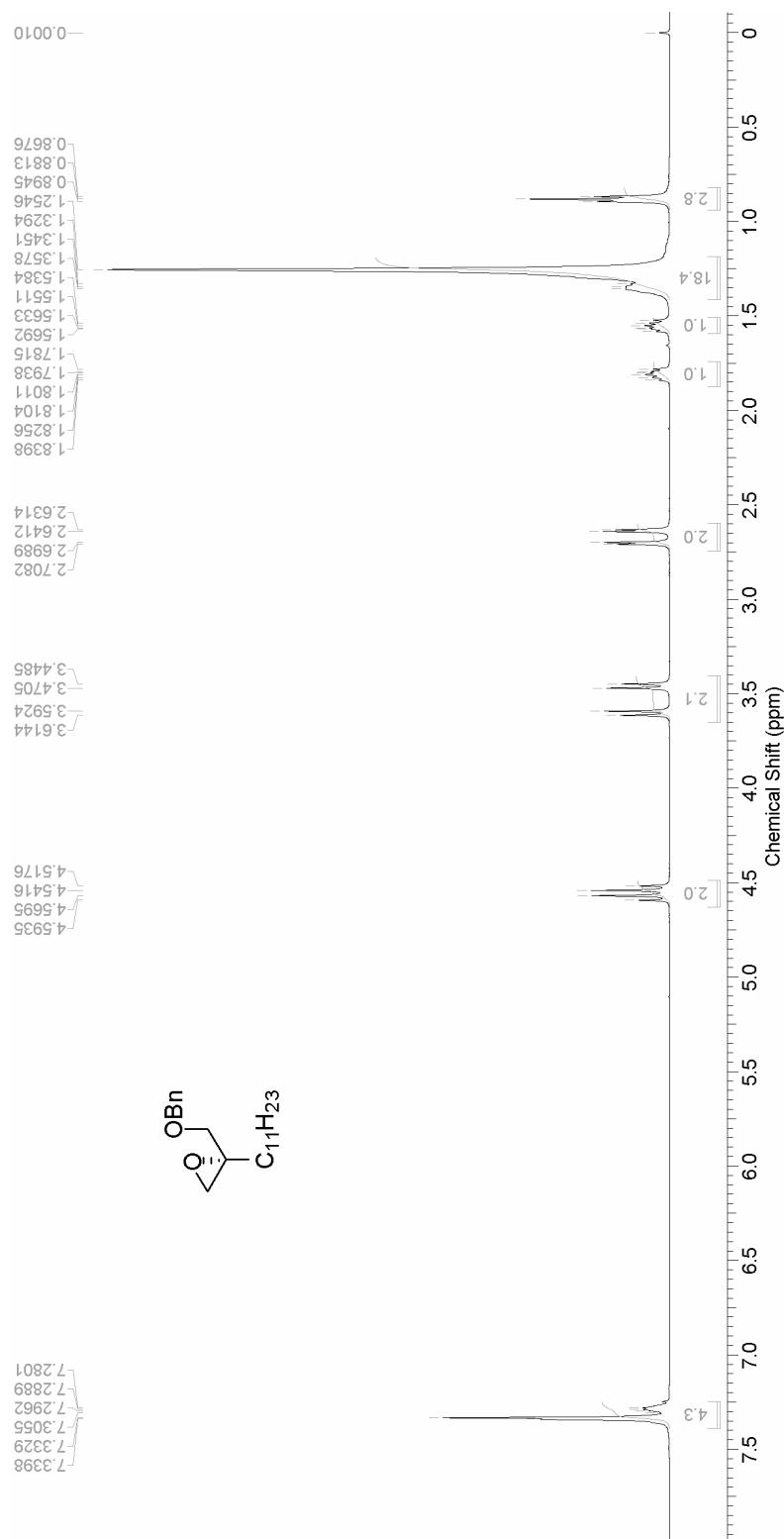


### LR FAB (otg matrix) spectrum of compound 3

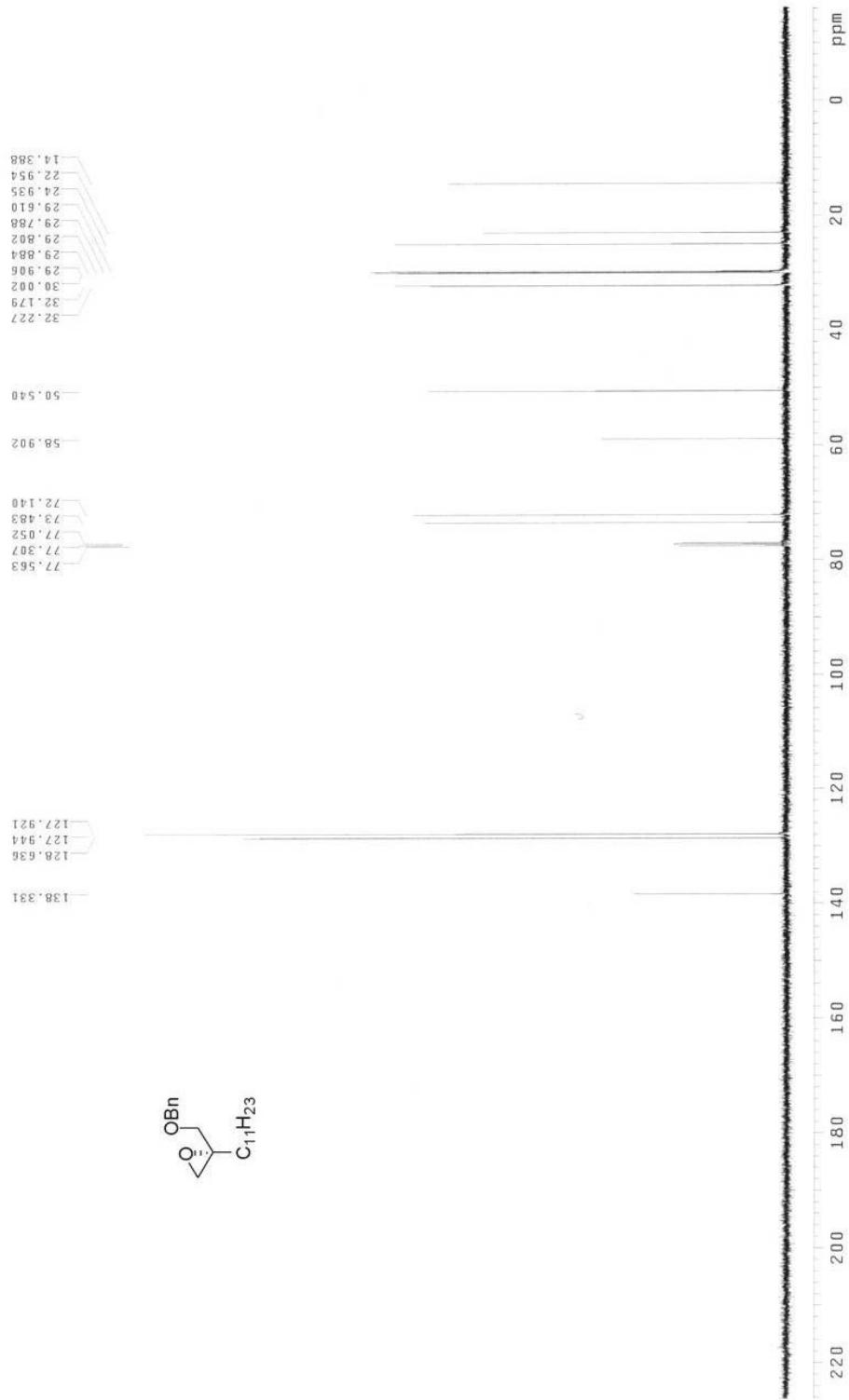


LR FAB (nba matrix) spectrum of compound 3

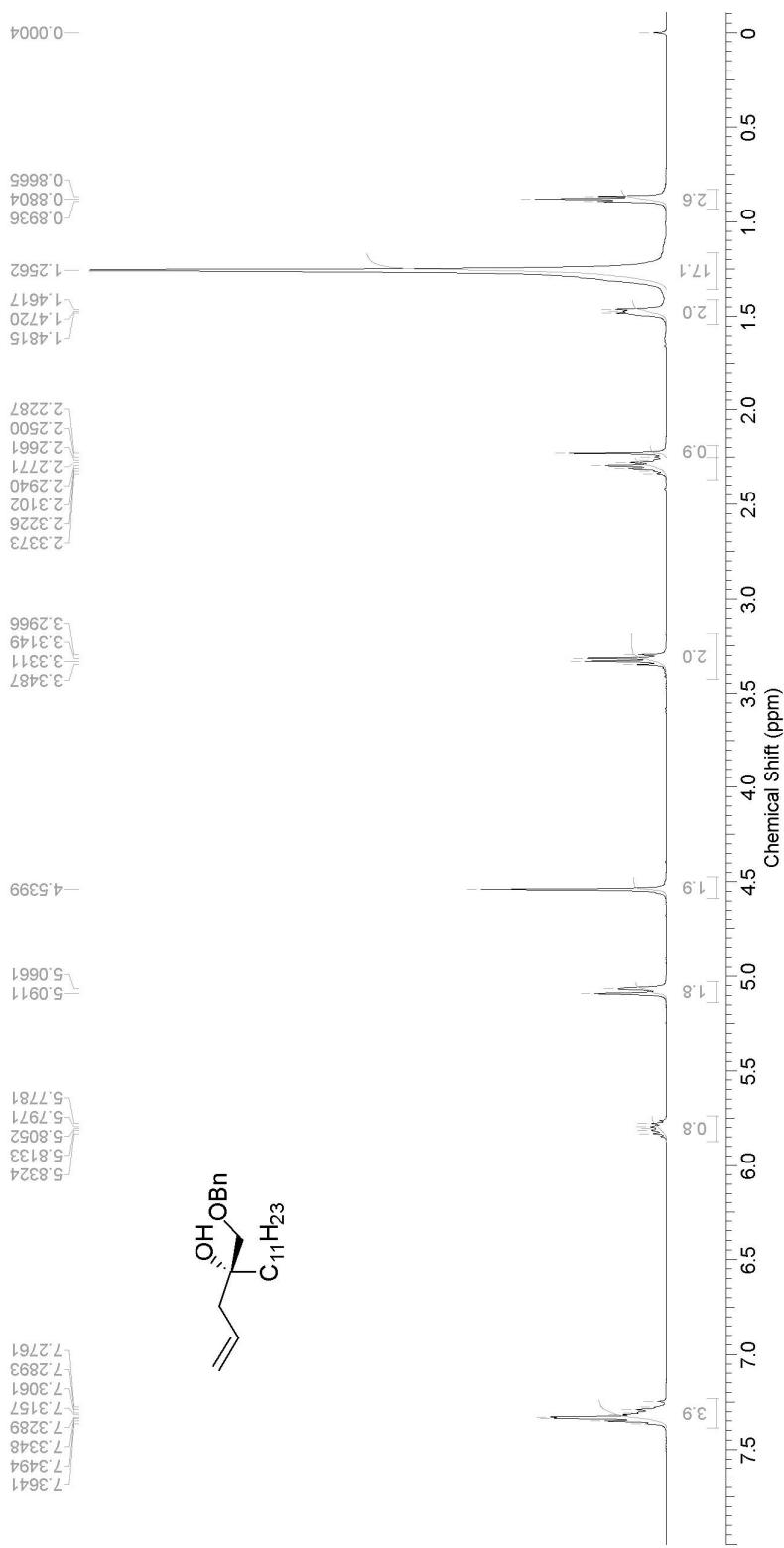


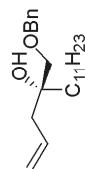
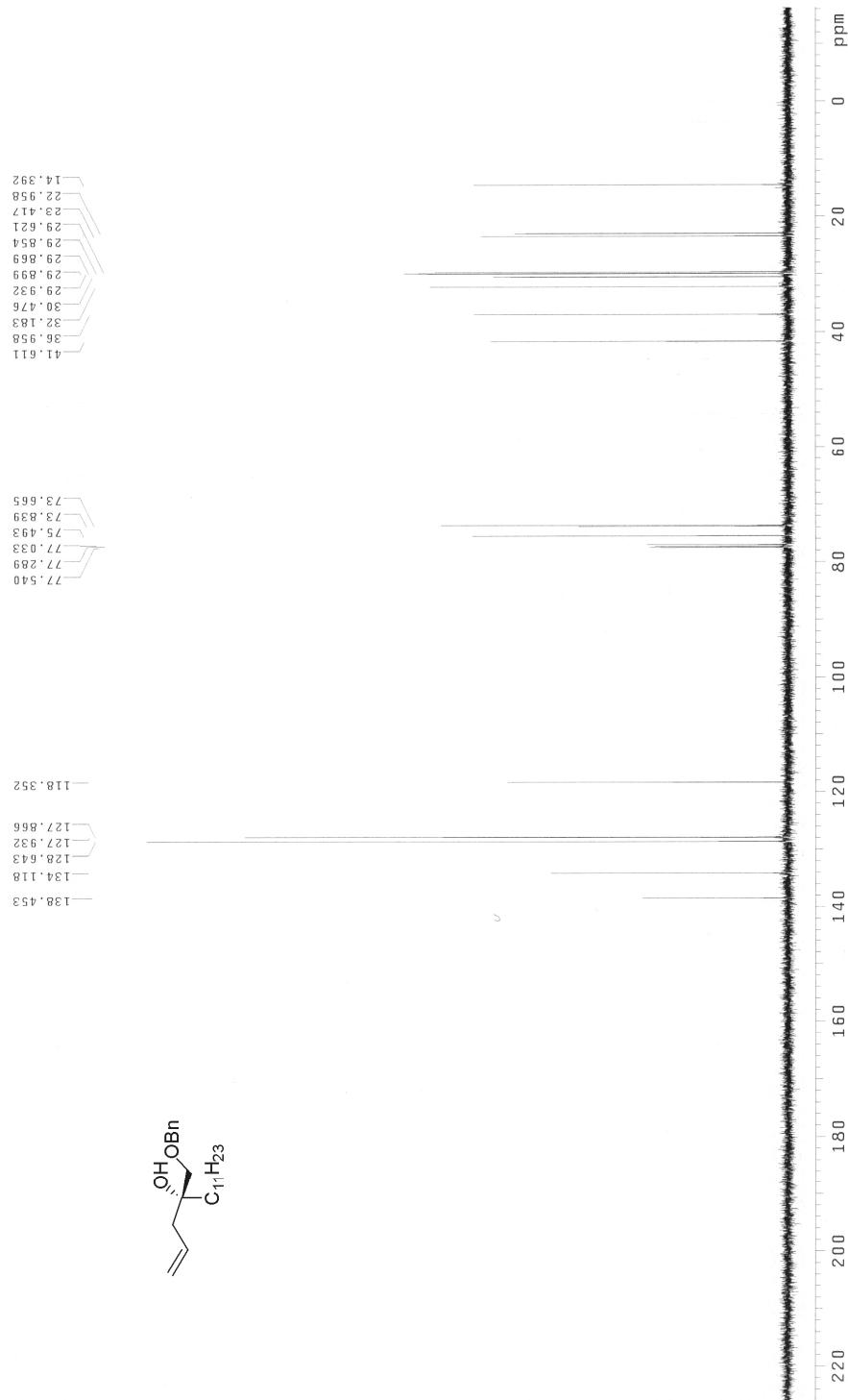


<sup>1</sup>H-NMR (500MHz, CDCl<sub>3</sub>) of the benzyl ether of epoxide 6

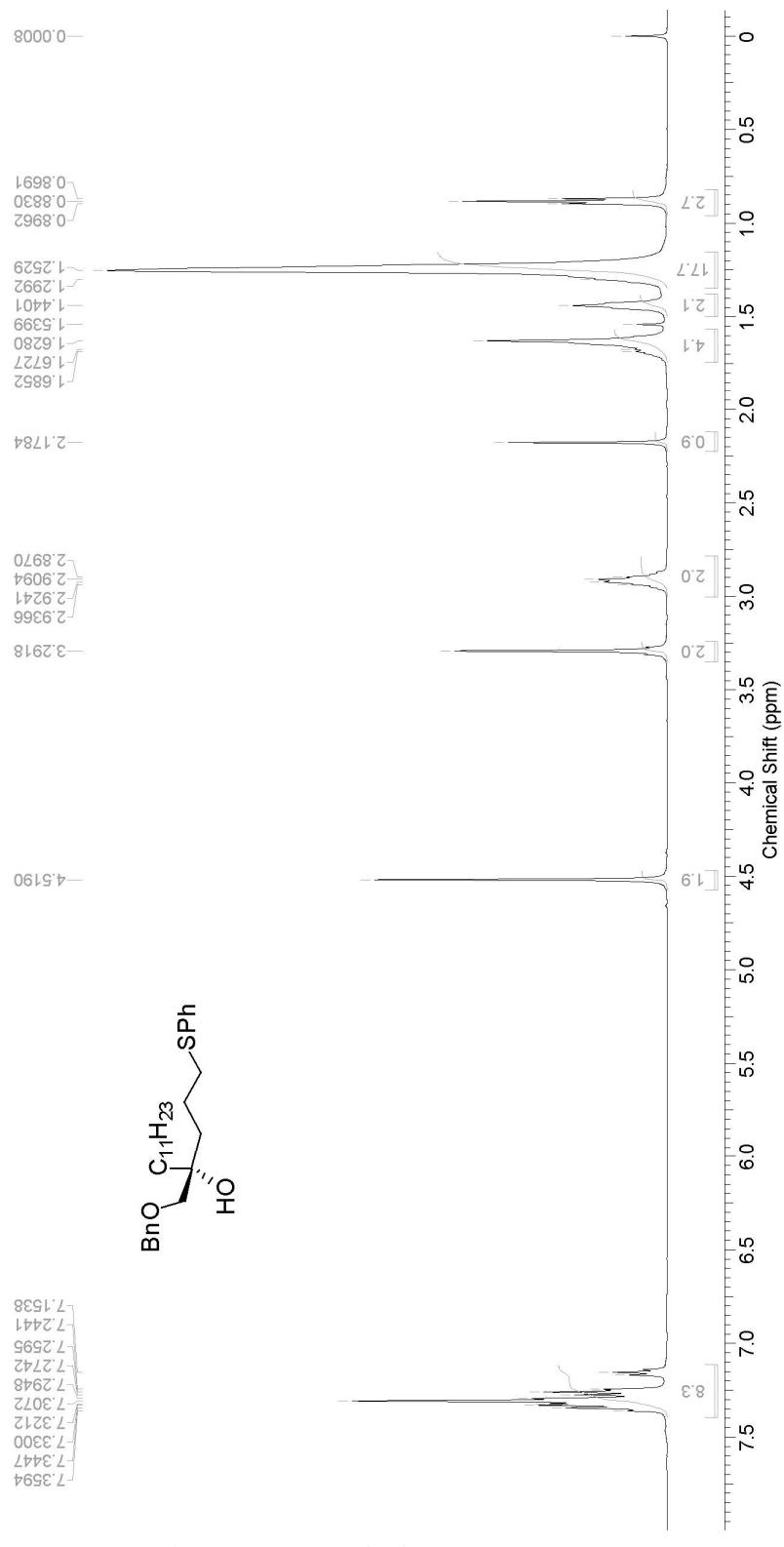


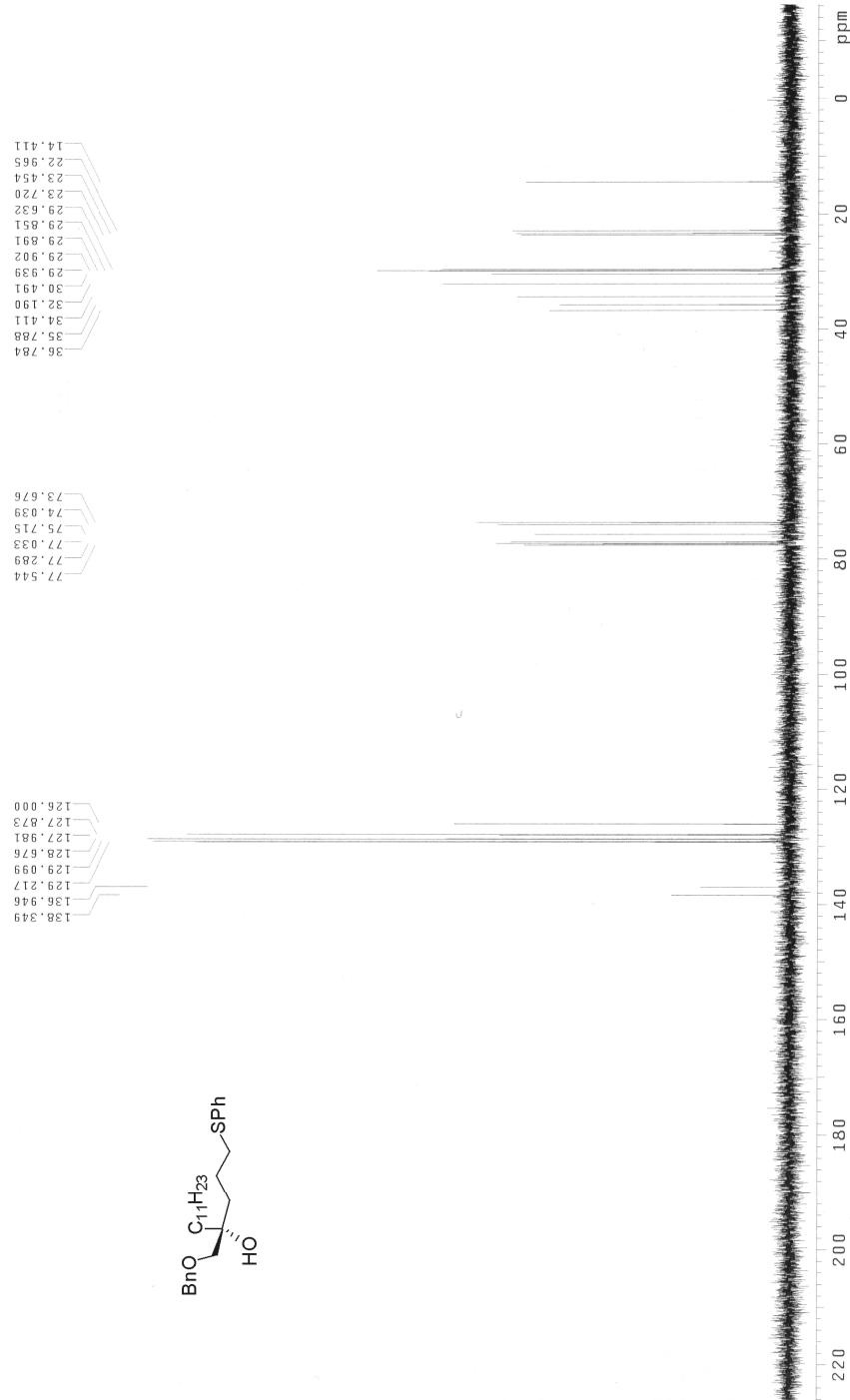
$^{13}\text{C}$ -NMR (125 MHz,  $\text{CDCl}_3$ ) of the benzyl ether of epoxide **6**



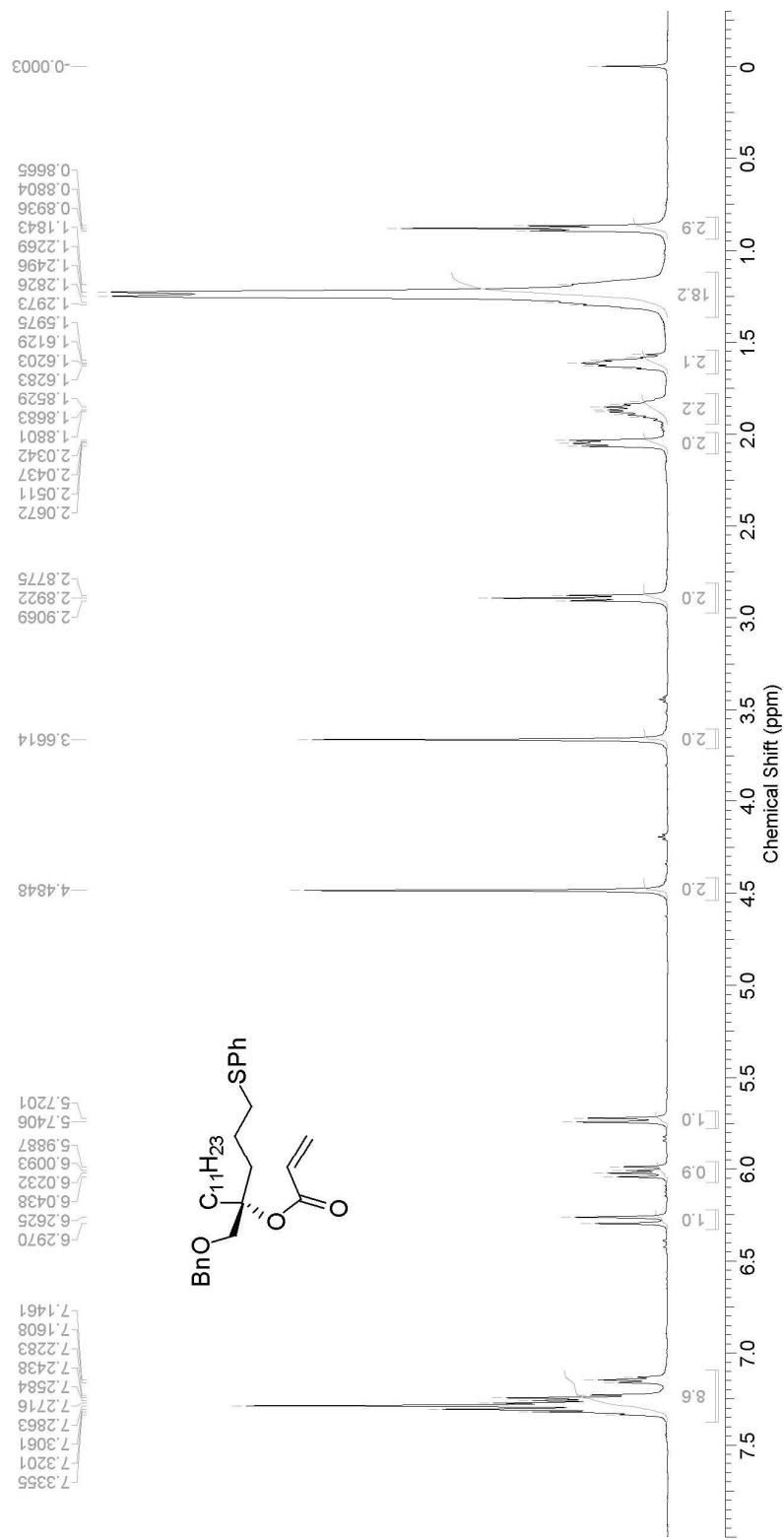


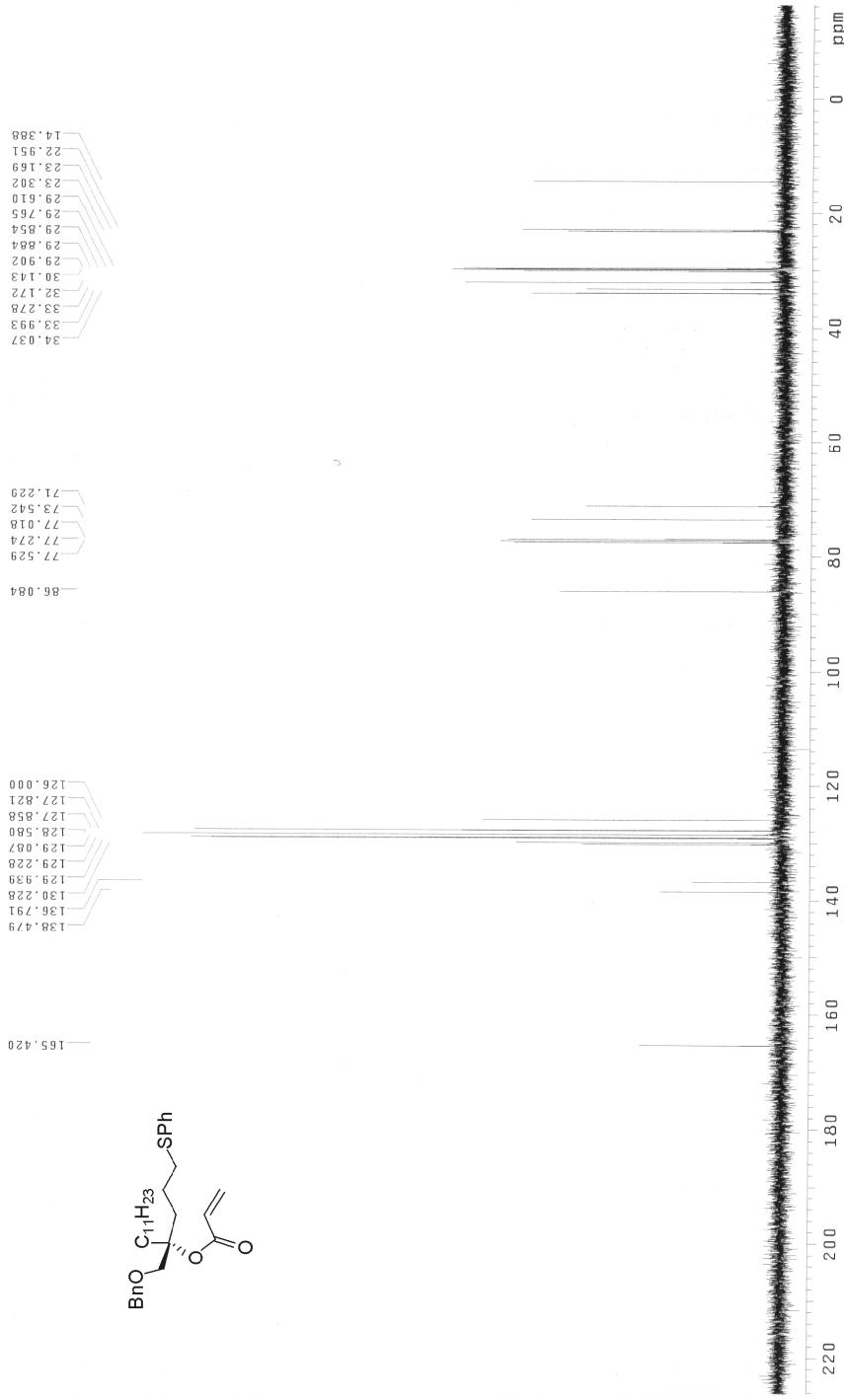
$^{13}\text{C}$ -NMR (125 MHz,  $\text{CDCl}_3$ ) of 7



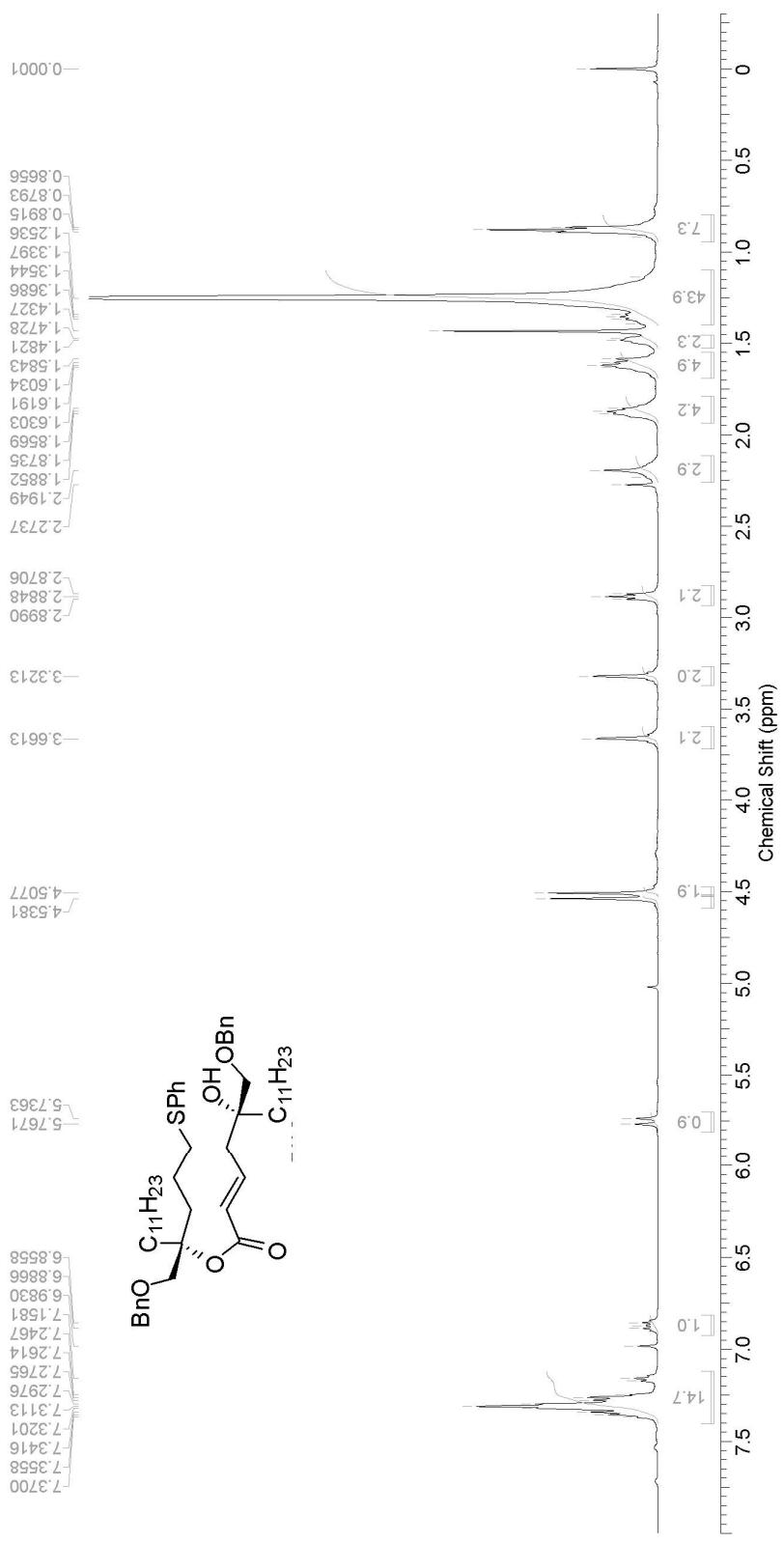


<sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>) of sulfide derivative of **7**

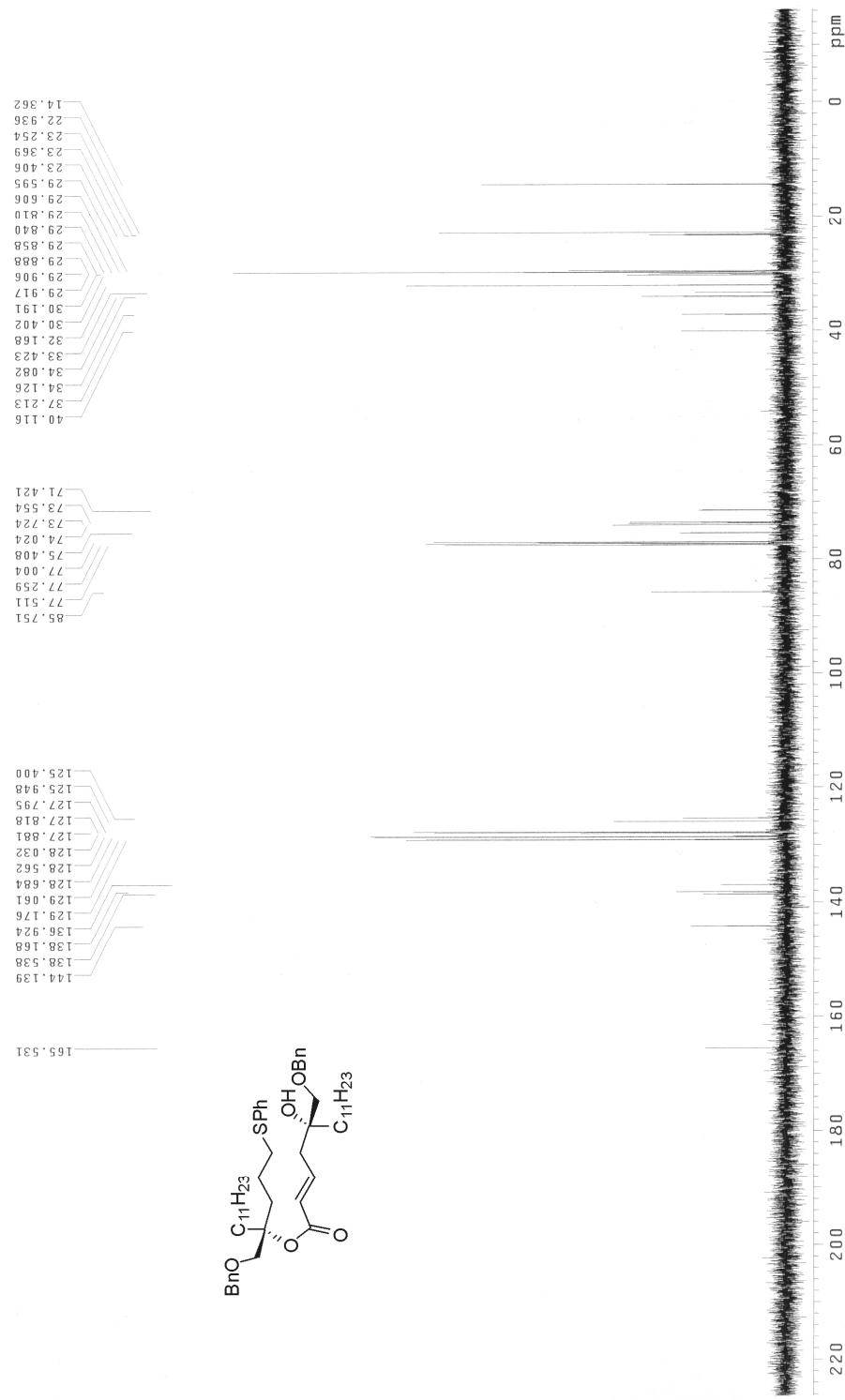


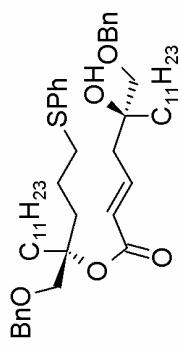
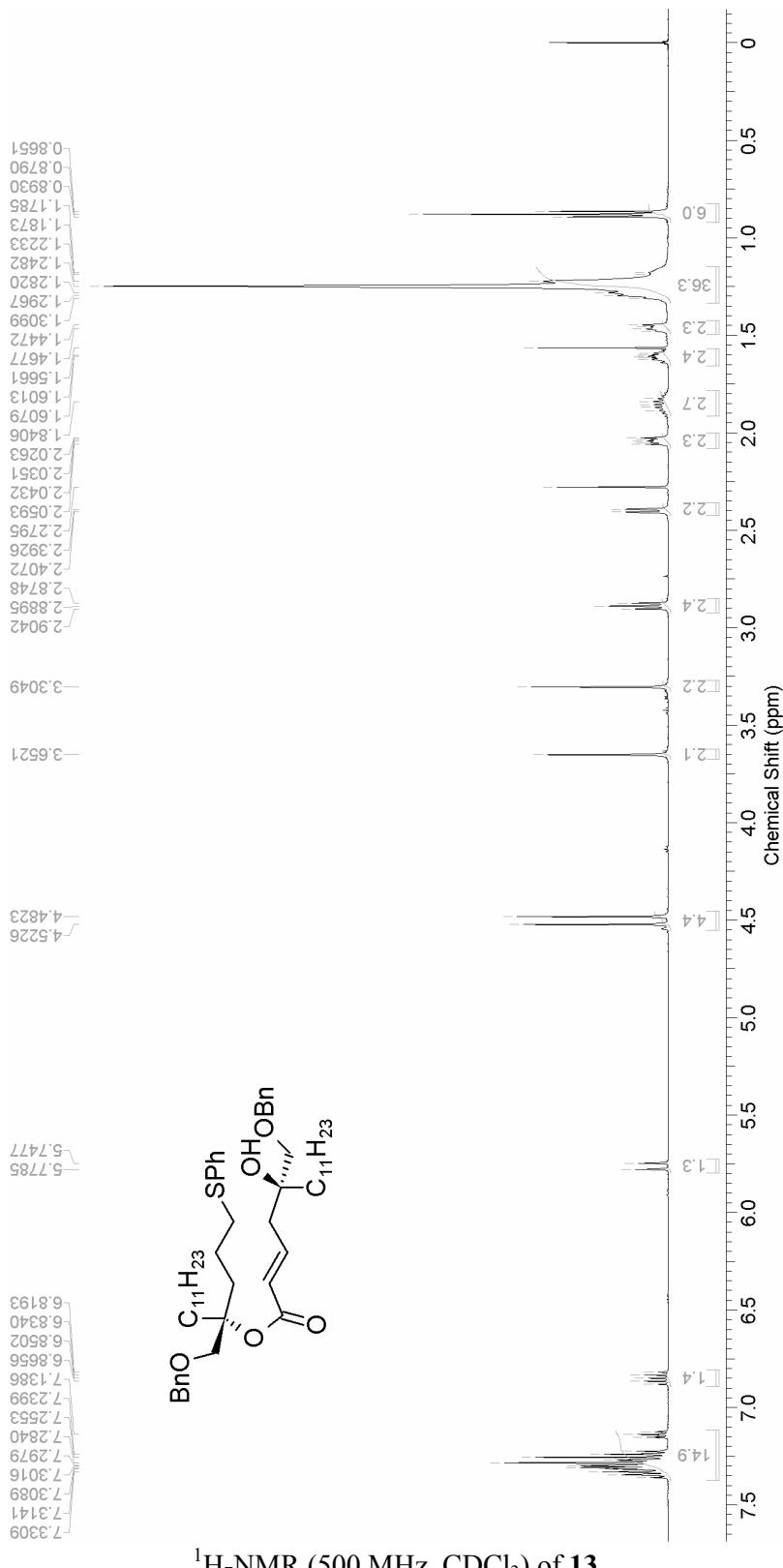


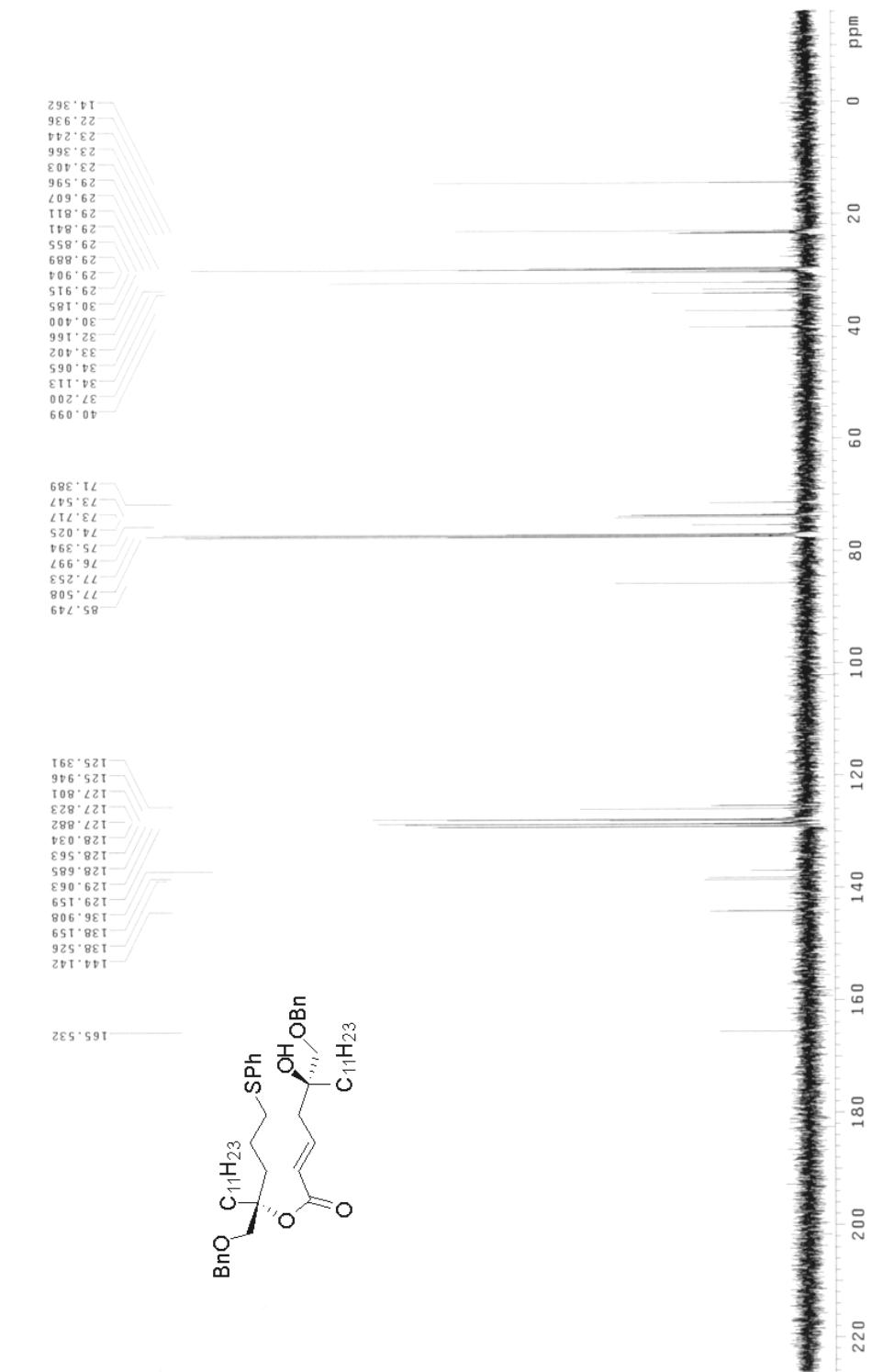
<sup>13</sup>C-NMR ( $125\text{ MHz}$ ,  $\text{CDCl}_3$ ) of **8**

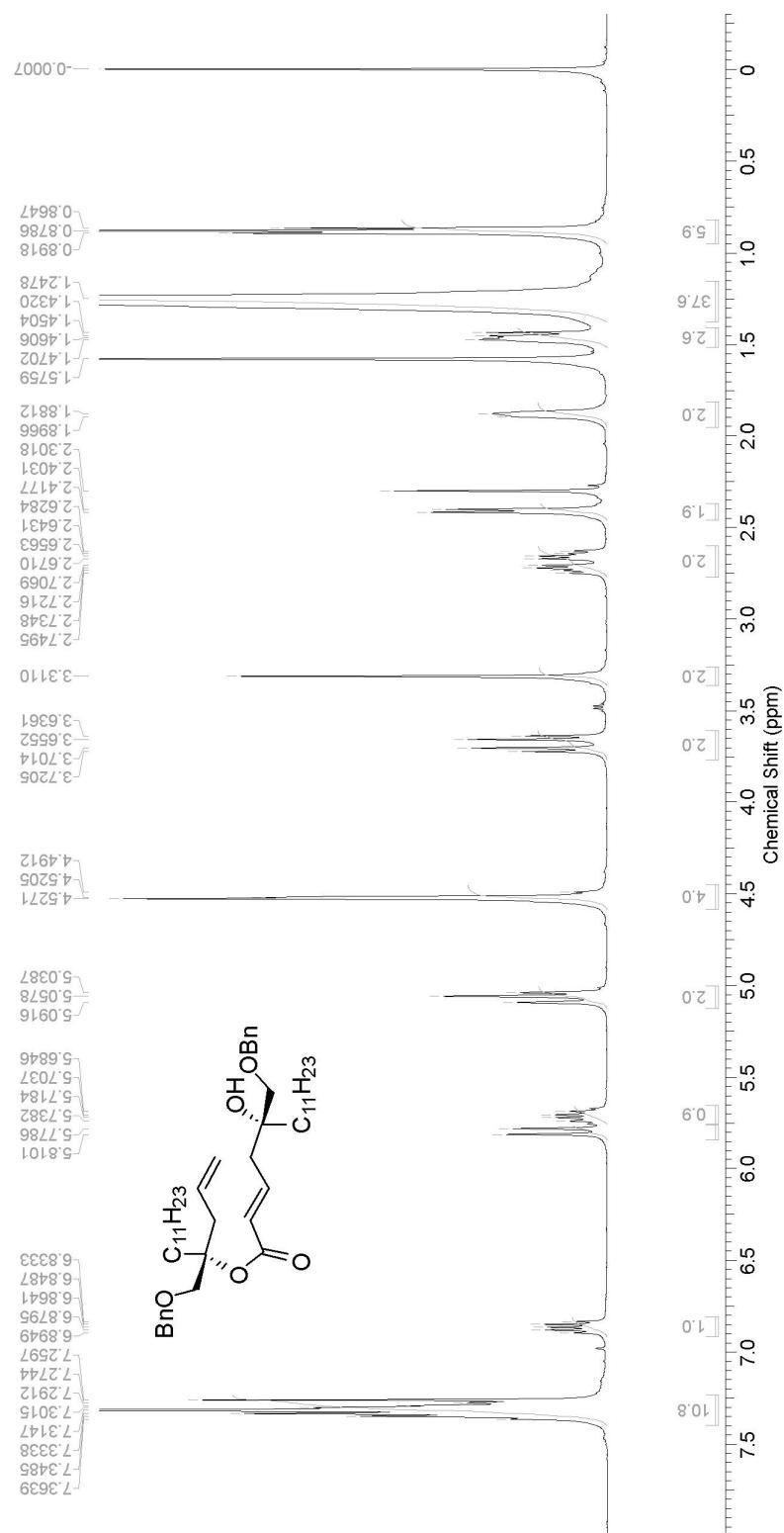


<sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>) of **10**

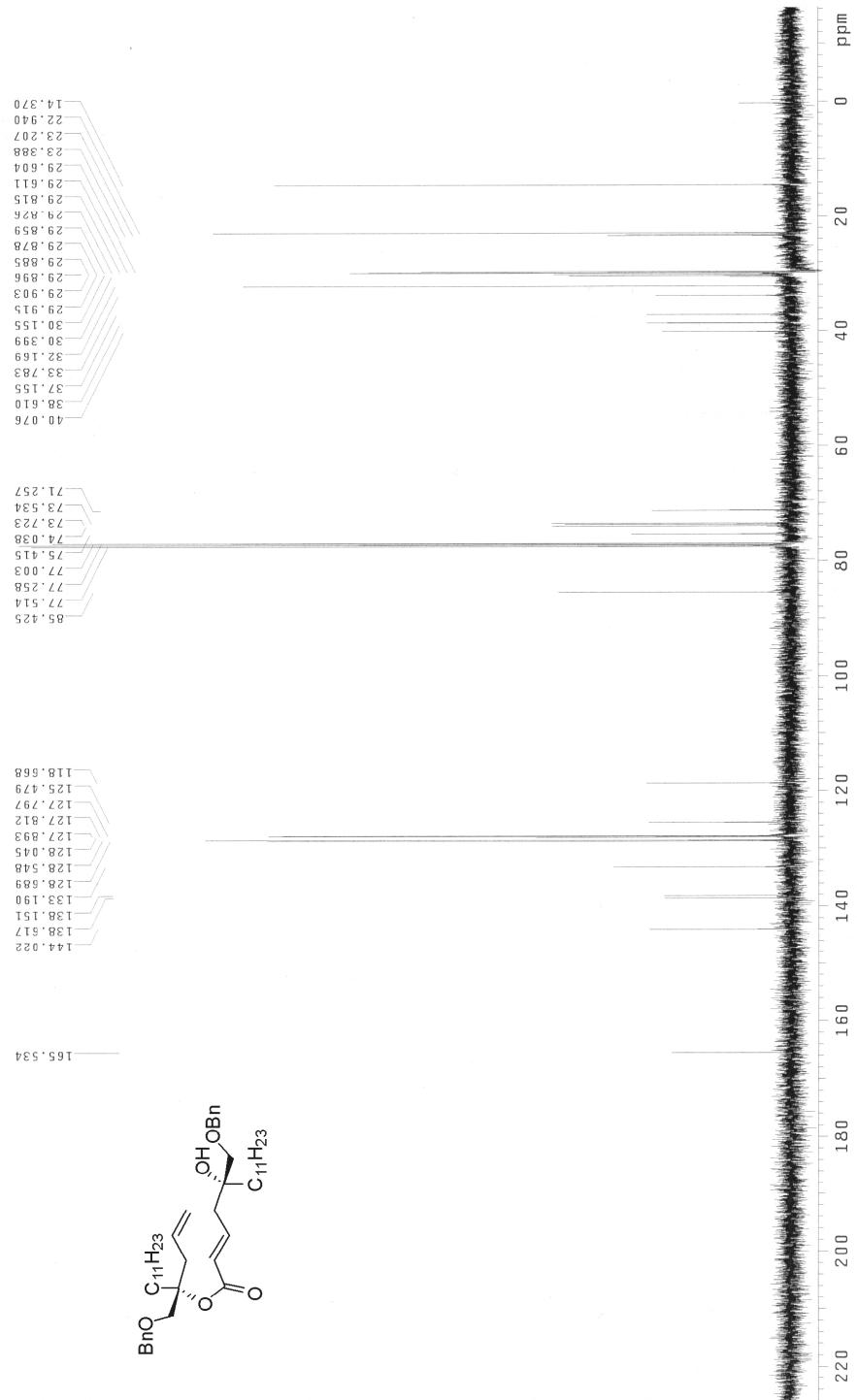




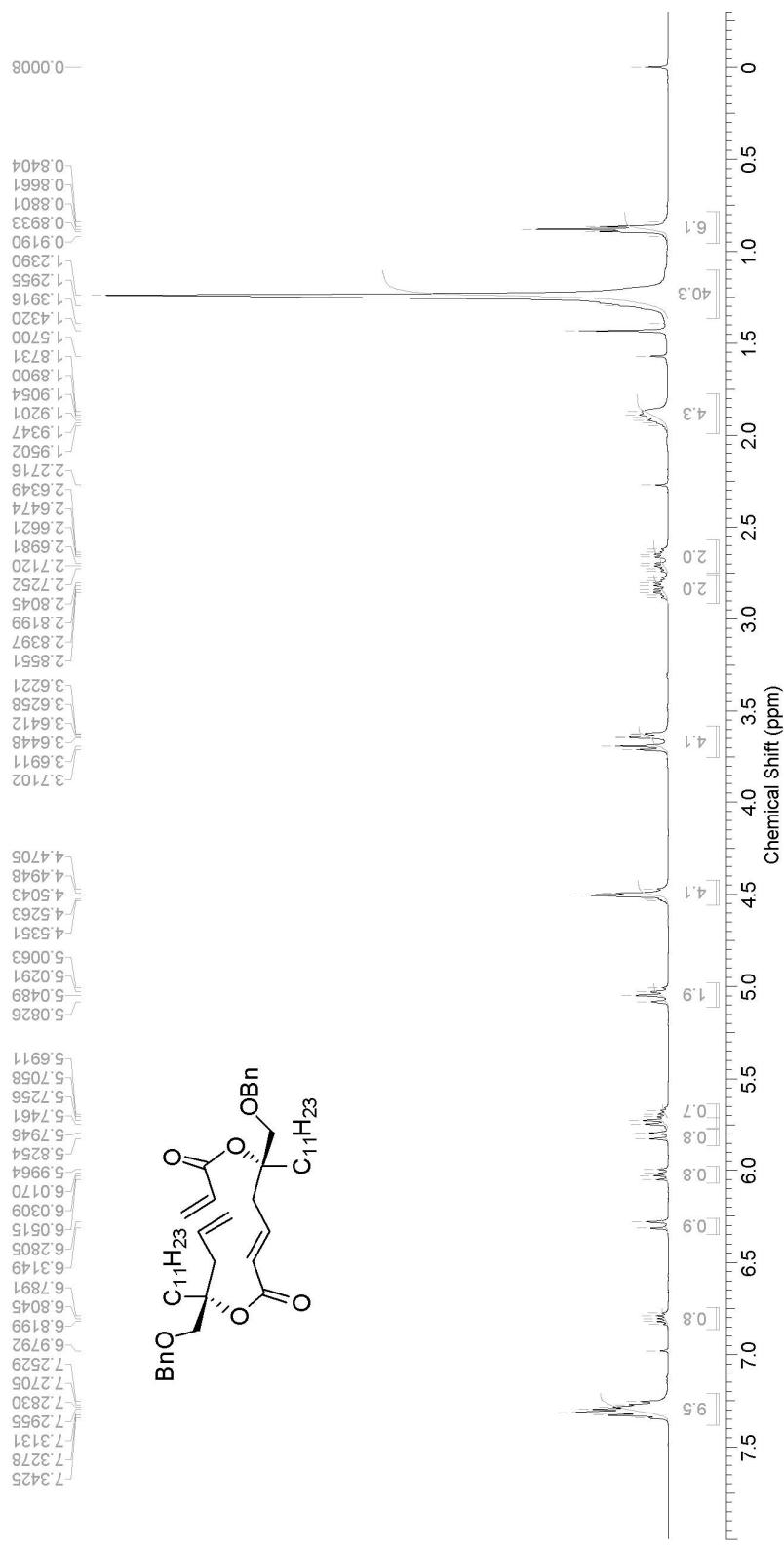




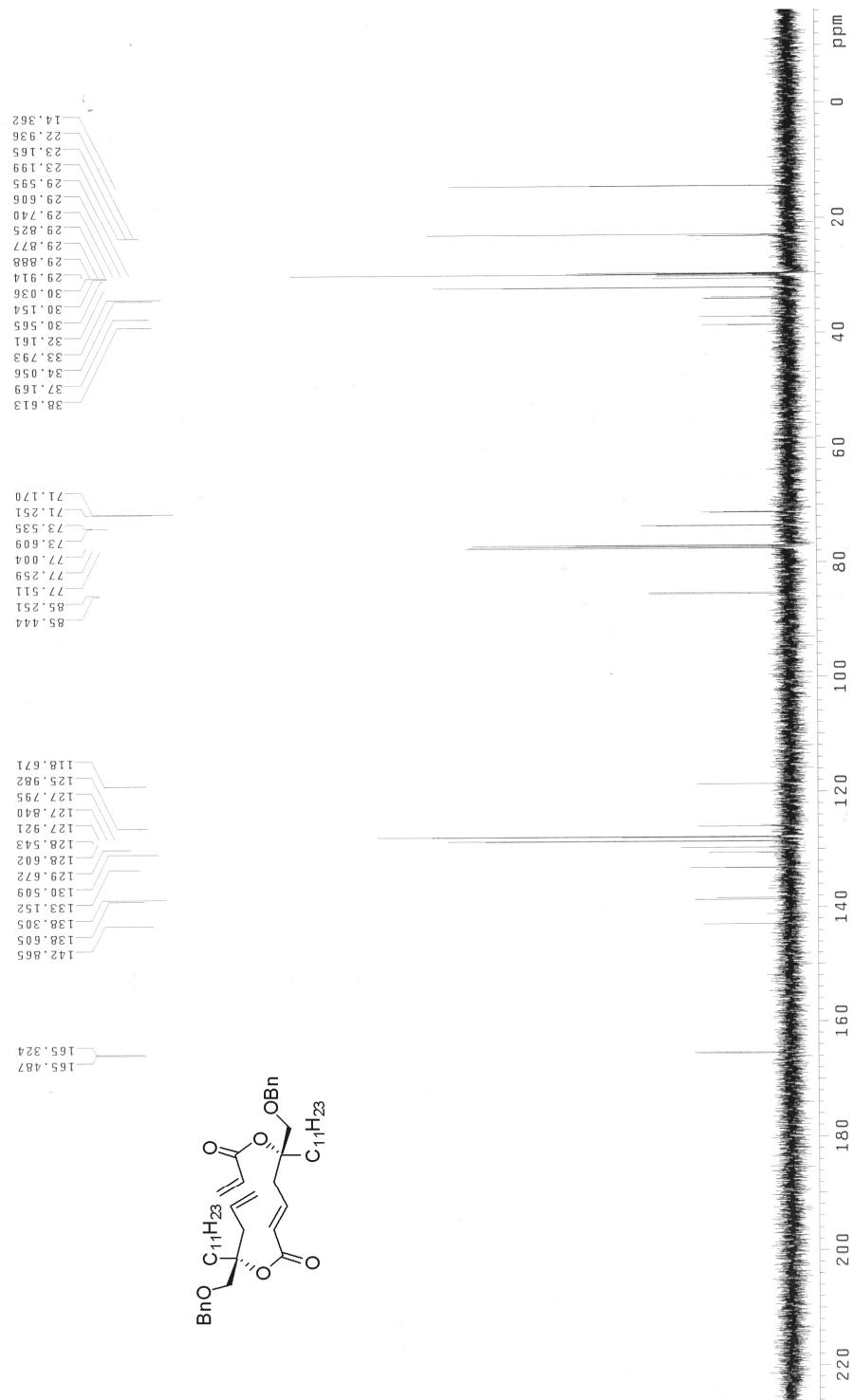
<sup>1</sup>H-NMR (500MHz, CDCl<sub>3</sub>) of diene product of **10**



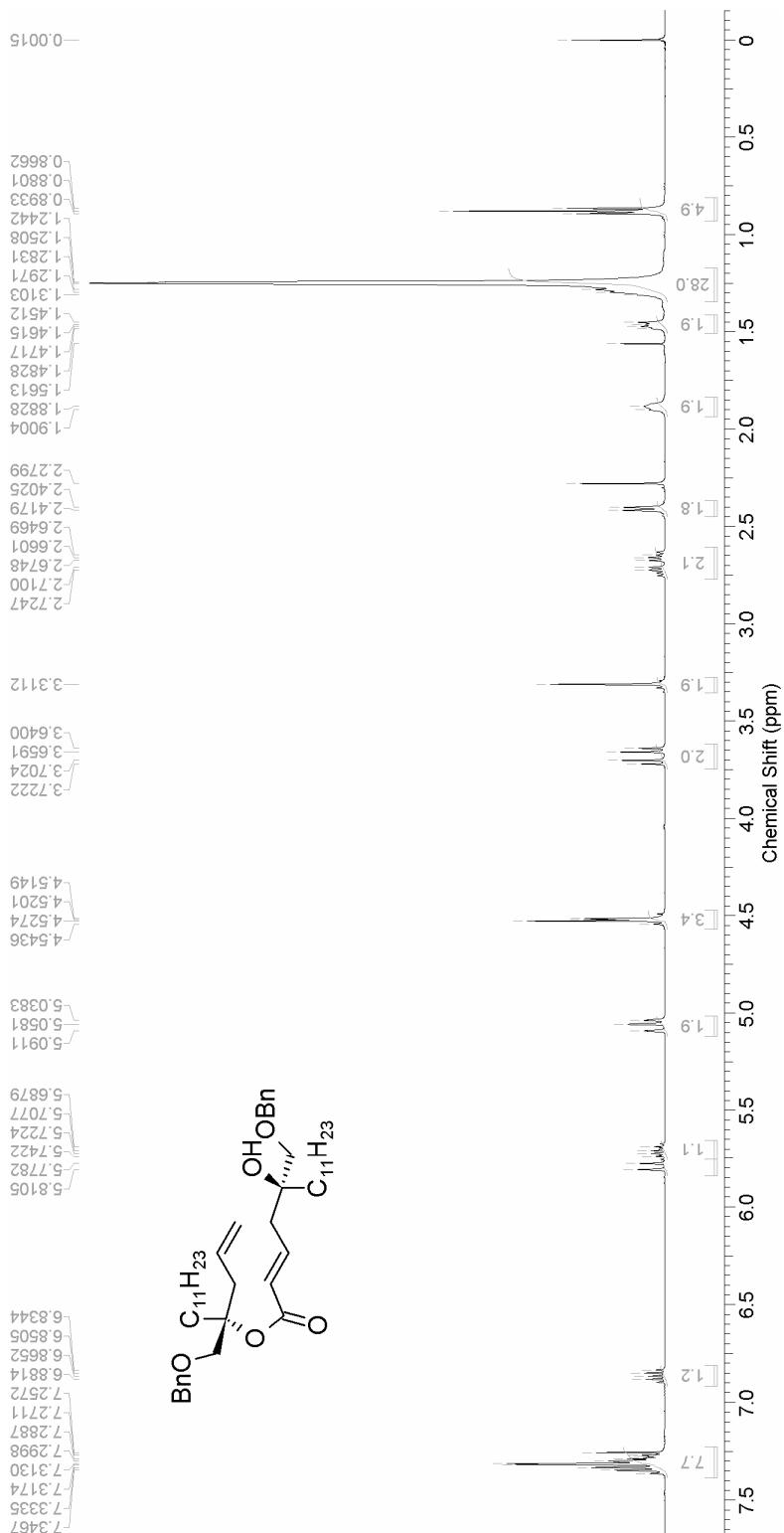
<sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>) of diene product of 10



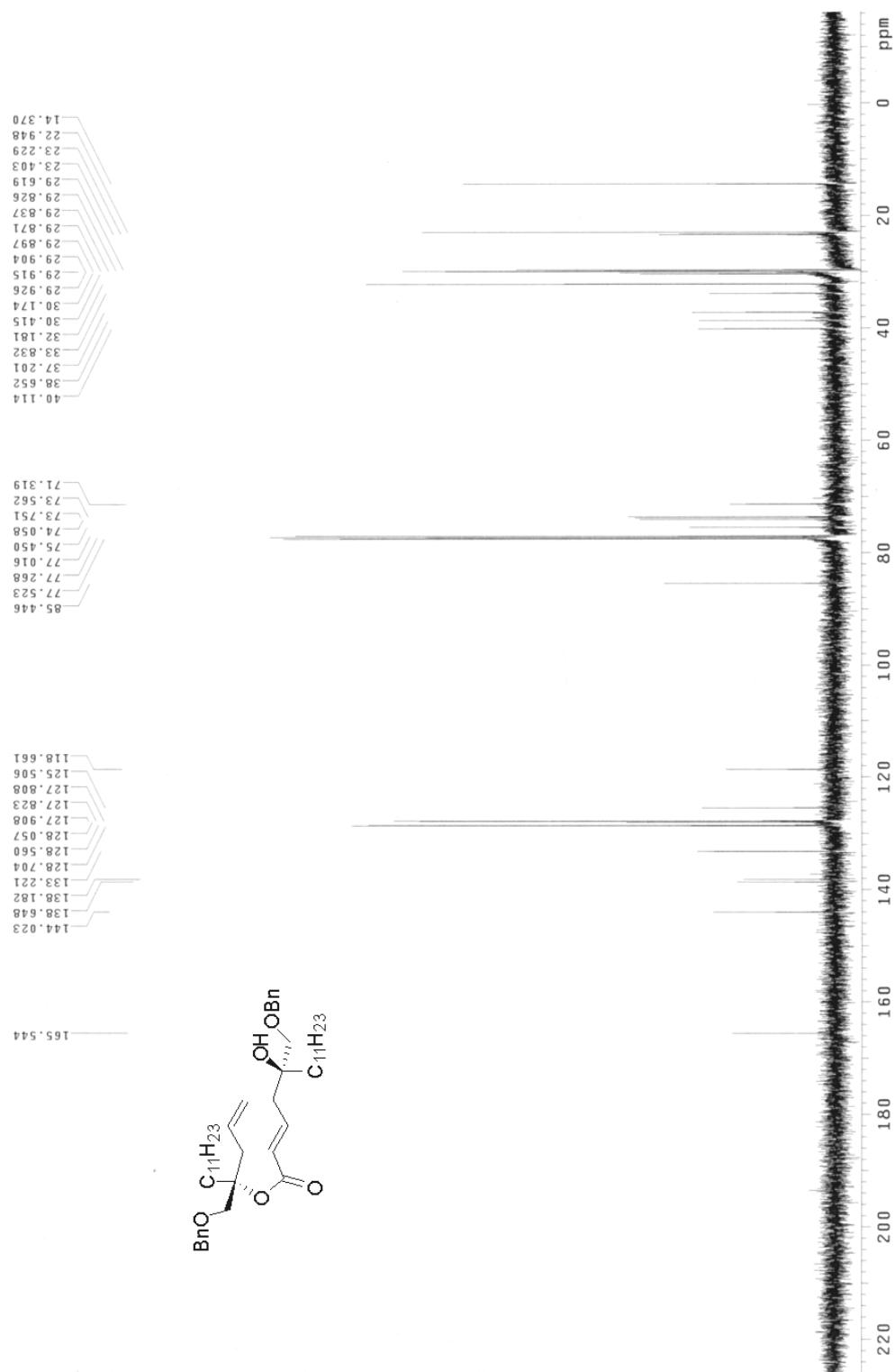
$^1\text{H}$ -NMR (500 MHz,  $\text{CDCl}_3$ ) of **11**



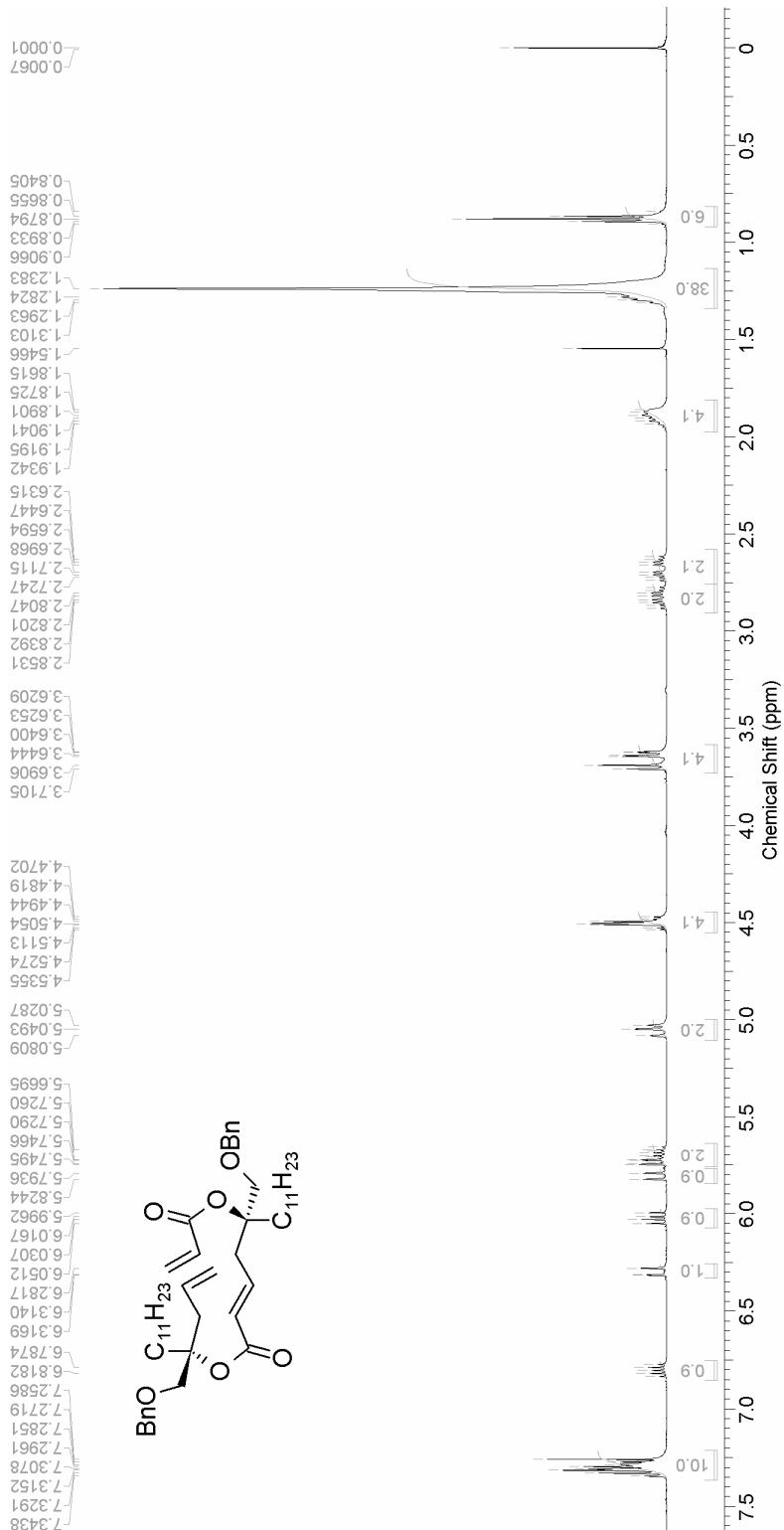
$^{13}\text{C}$ -NMR (125 MHz,  $\text{CDCl}_3$ ) of **11**

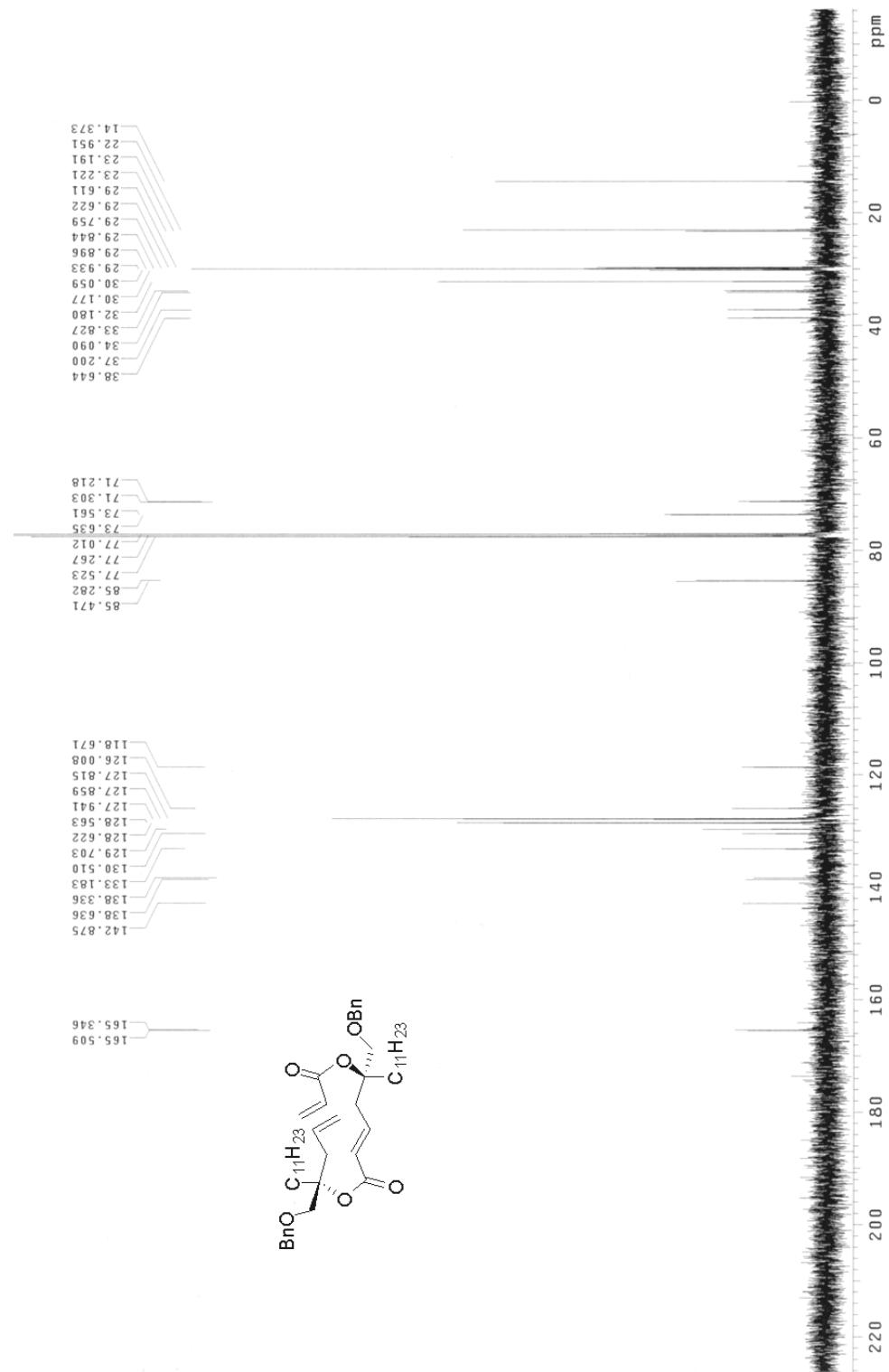


<sup>1</sup>H-NMR (500MHz, CDCl<sub>3</sub>) of diene product of **13**

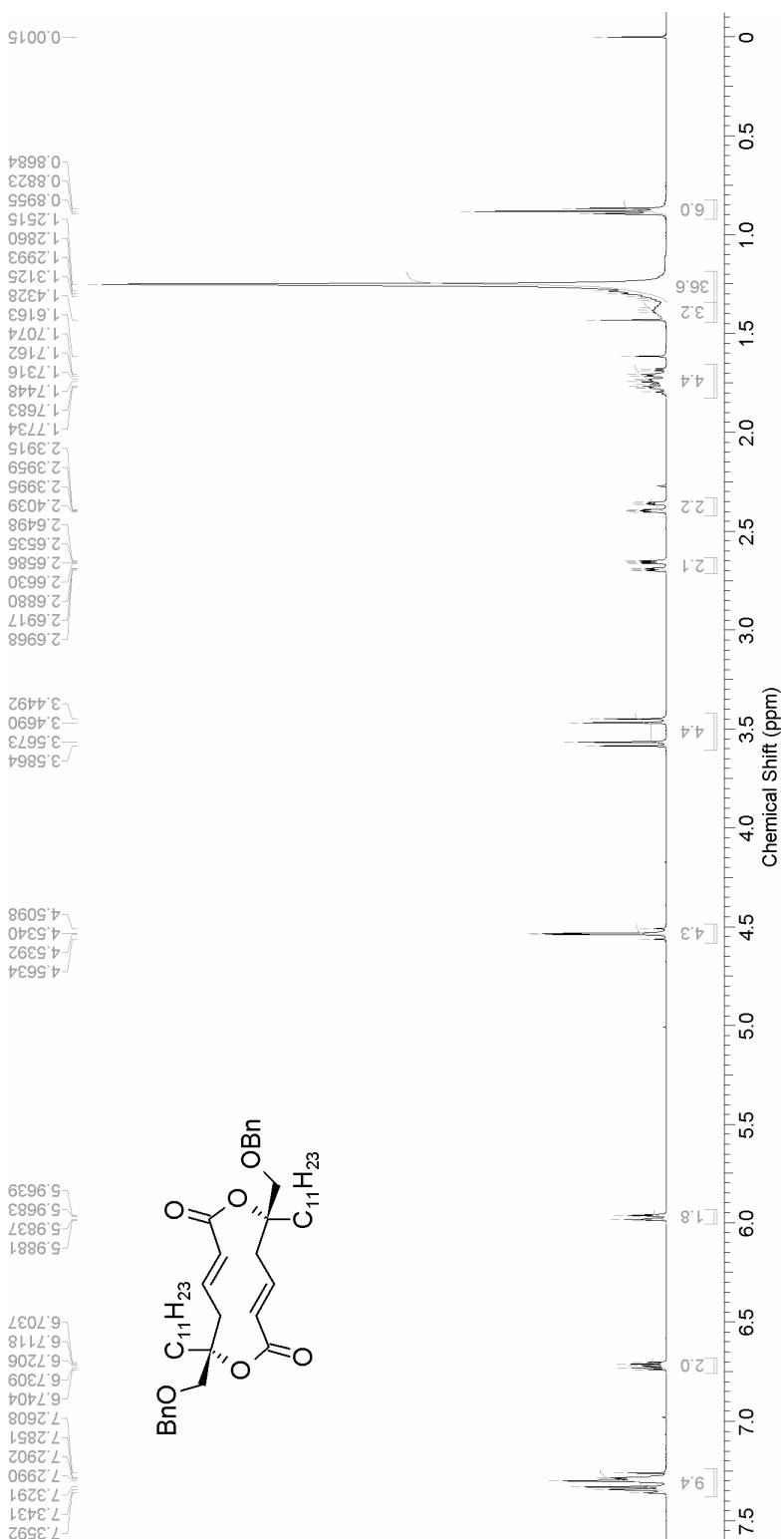


$^{13}\text{C}$ -NMR (125 MHz, CDCl<sub>3</sub>) of diene product of **13**

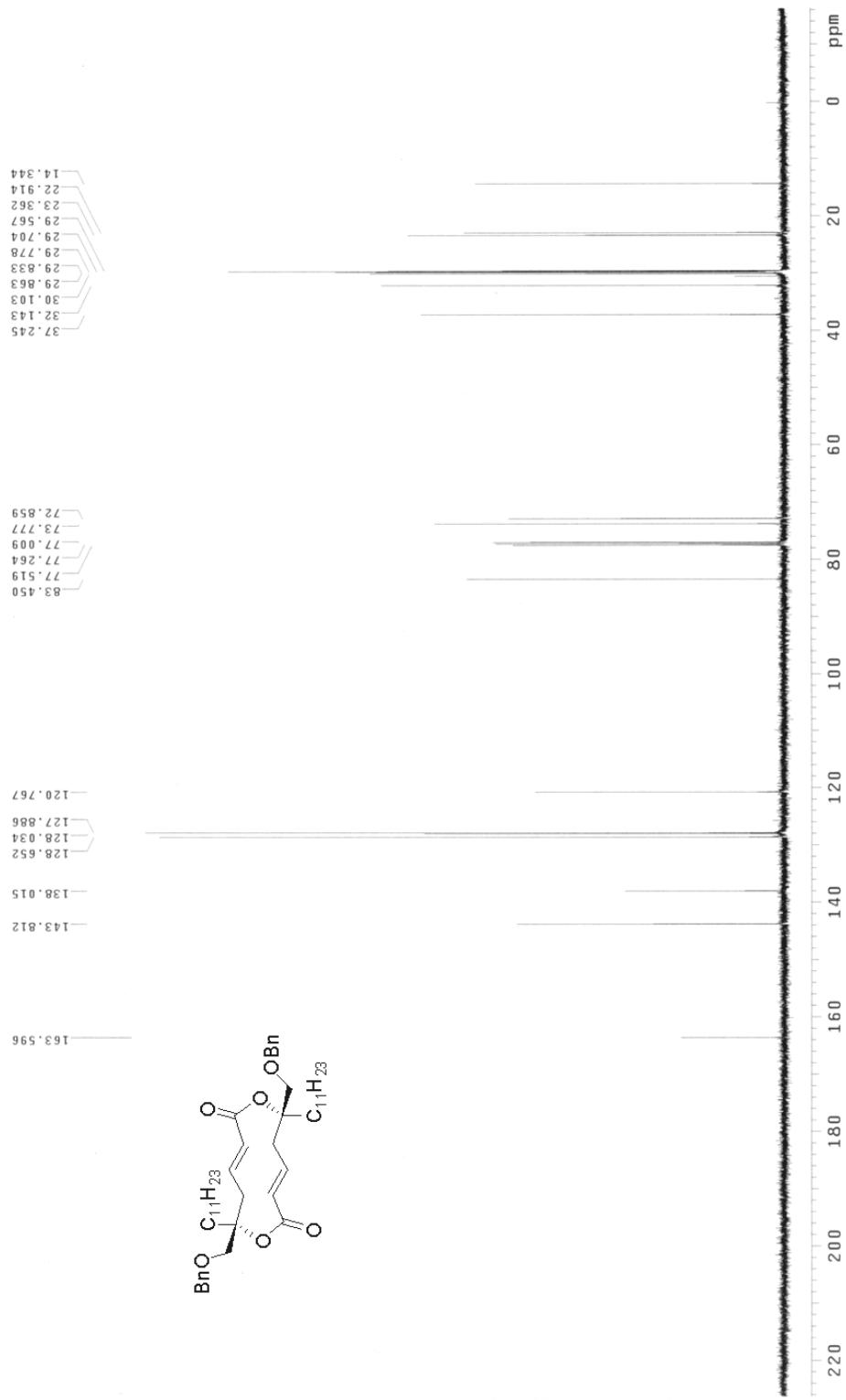




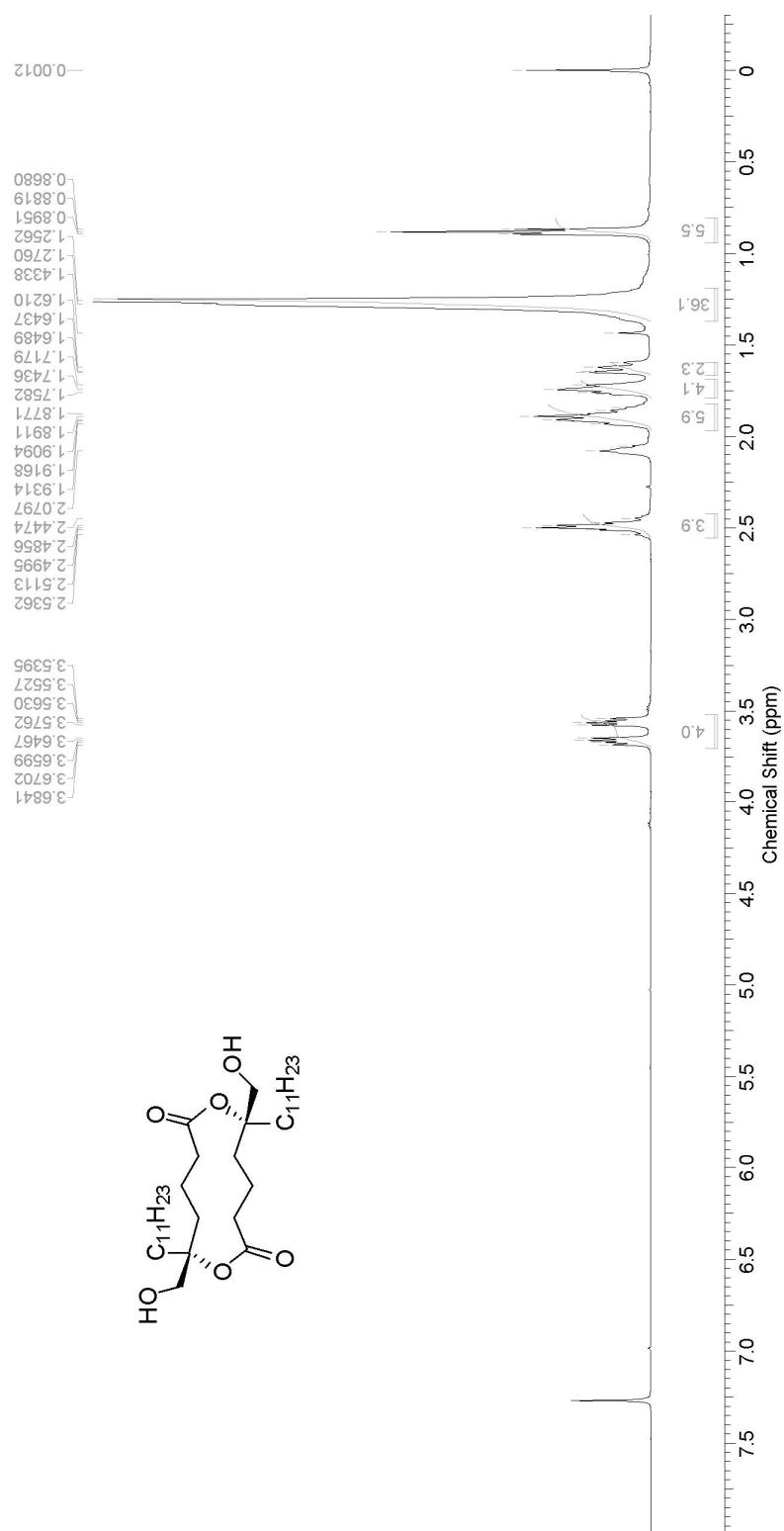
$^{13}\text{C}$ -NMR (125 MHz,  $\text{CDCl}_3$ ) of **14**

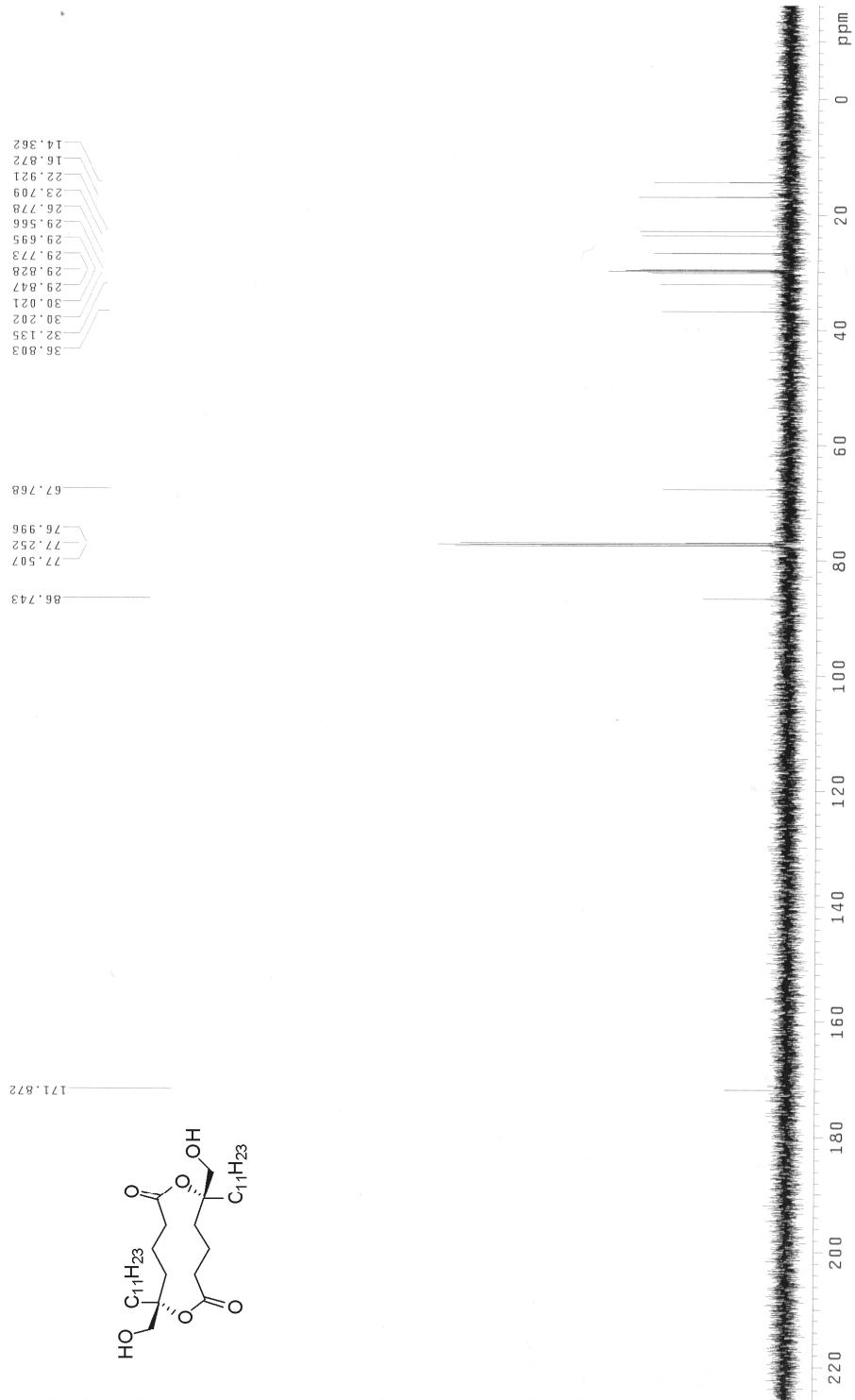


<sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>) of **12**

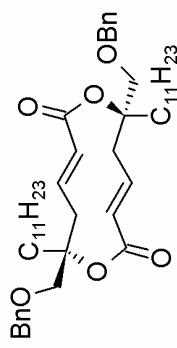
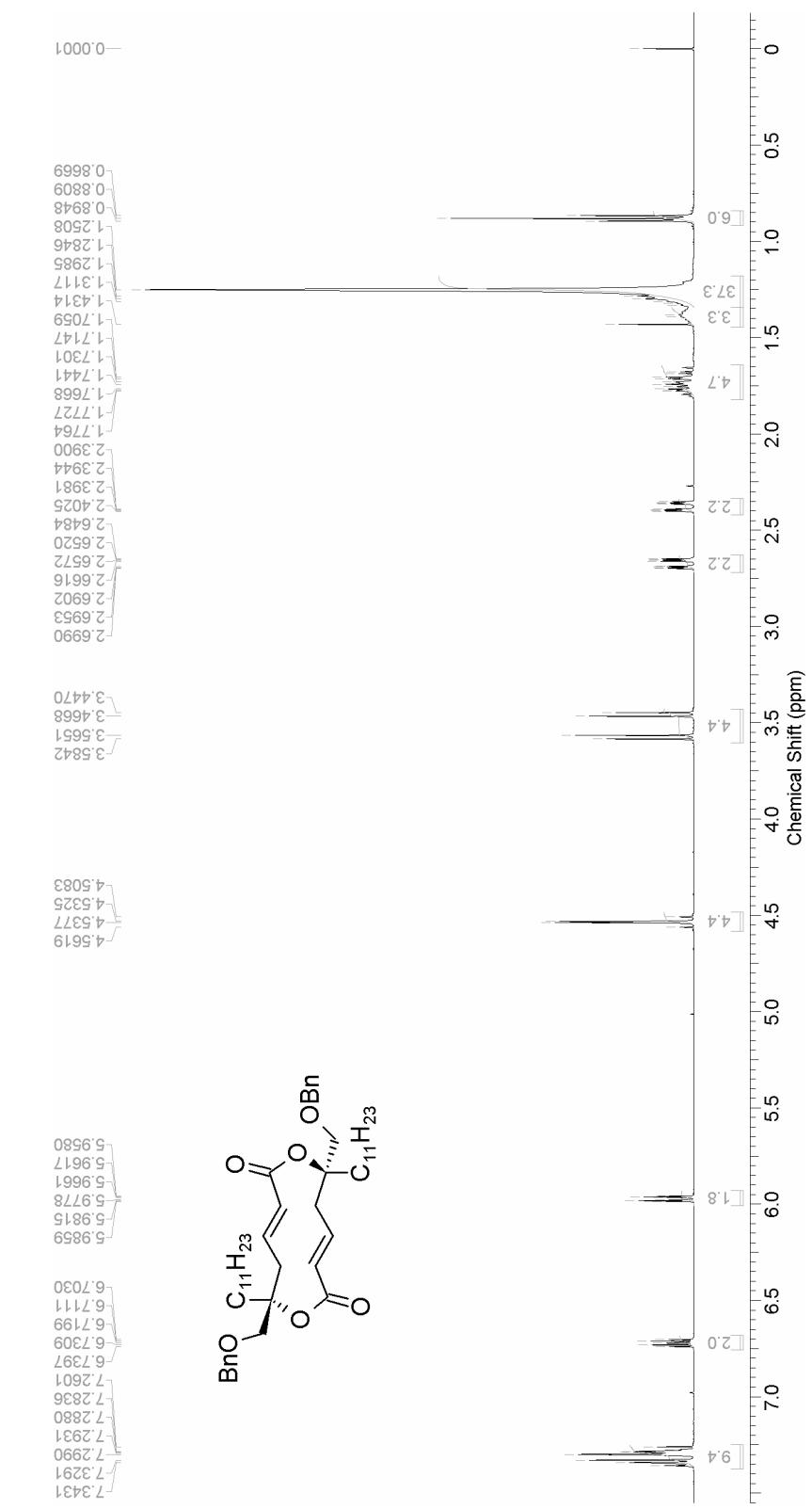


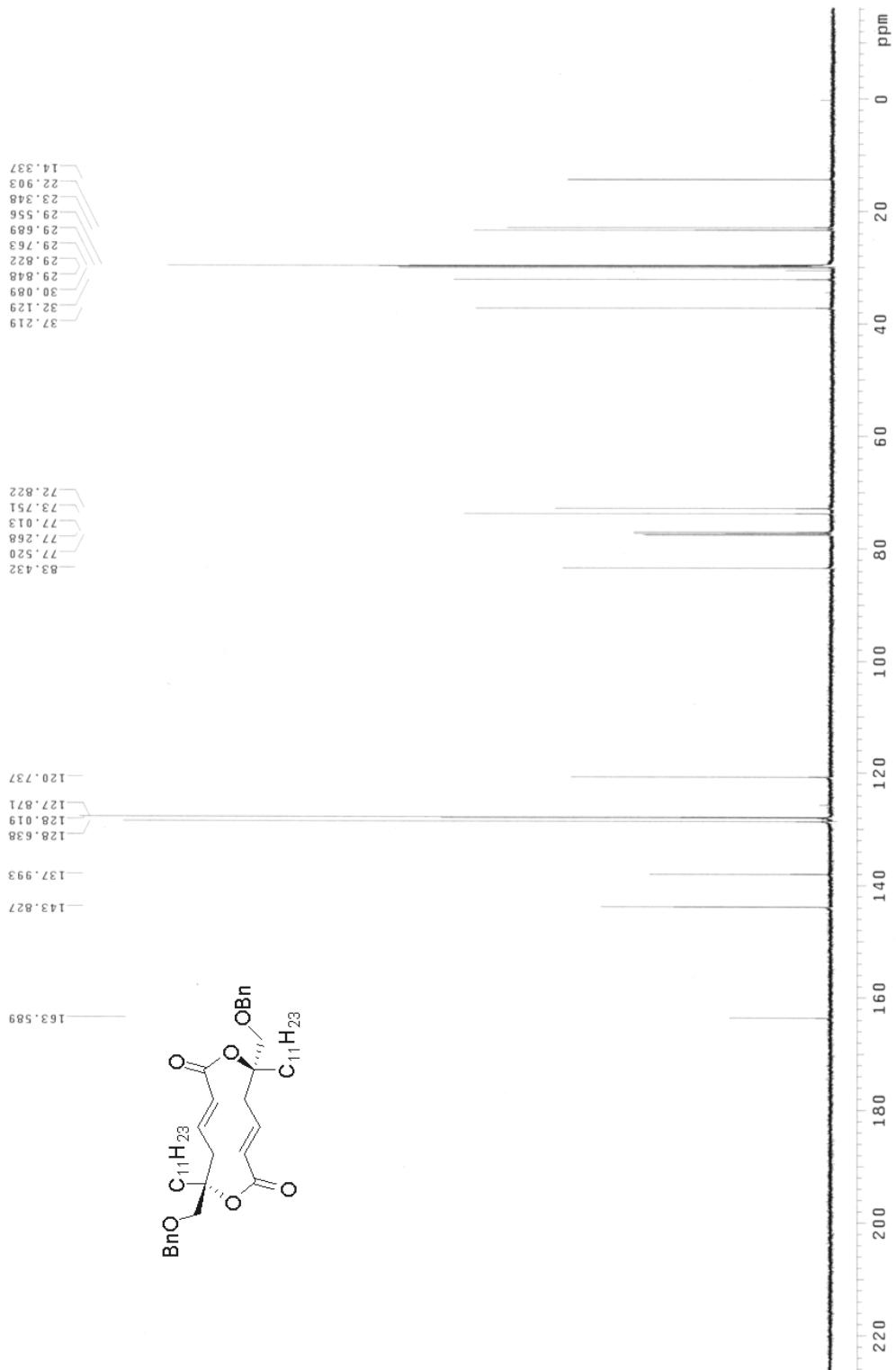
<sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>) of **12**

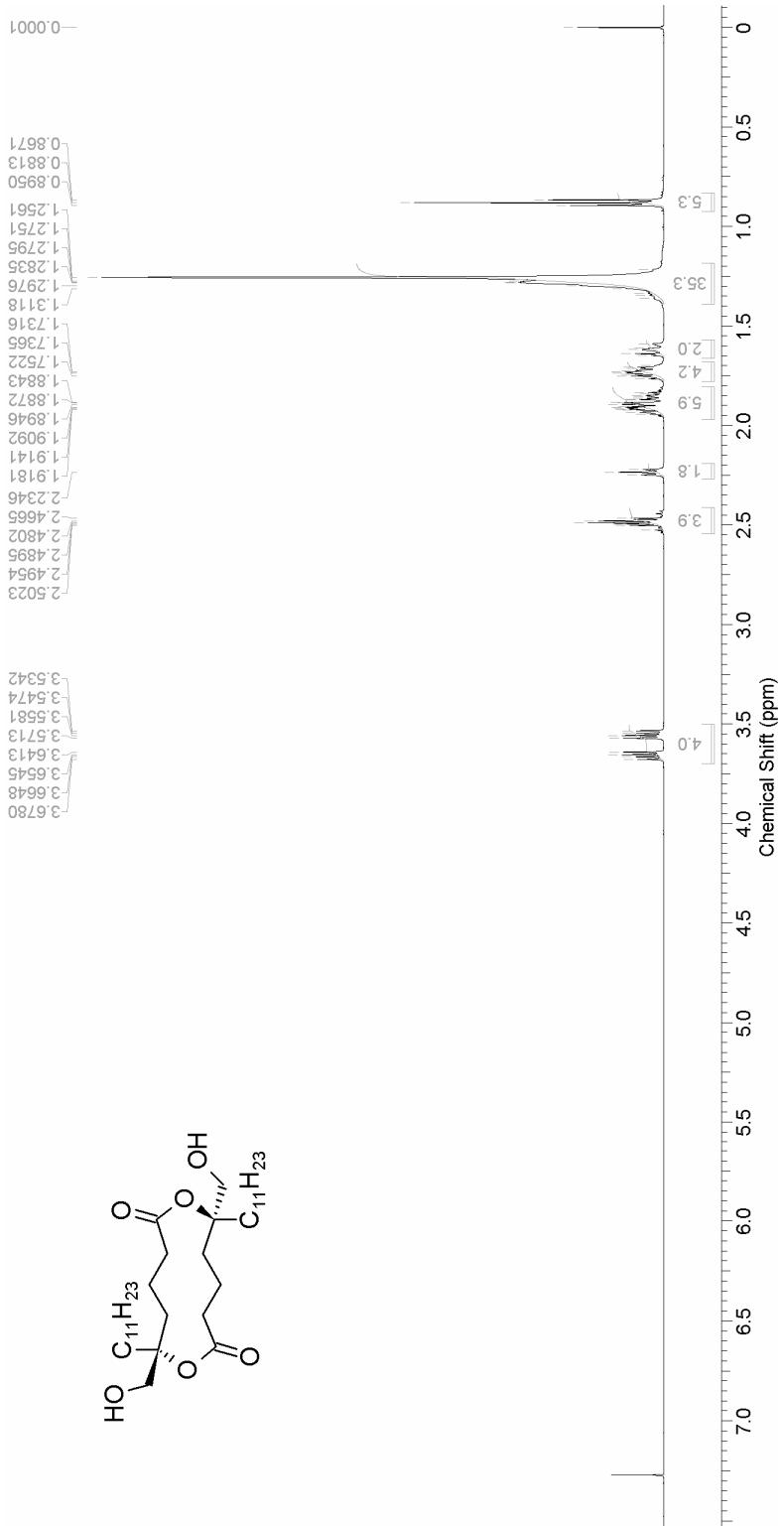




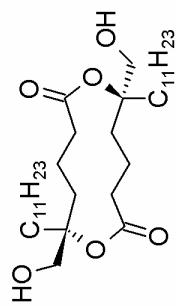
$^{13}\text{C}$ -NMR (125 MHz,  $\text{CDCl}_3$ ) of **5**

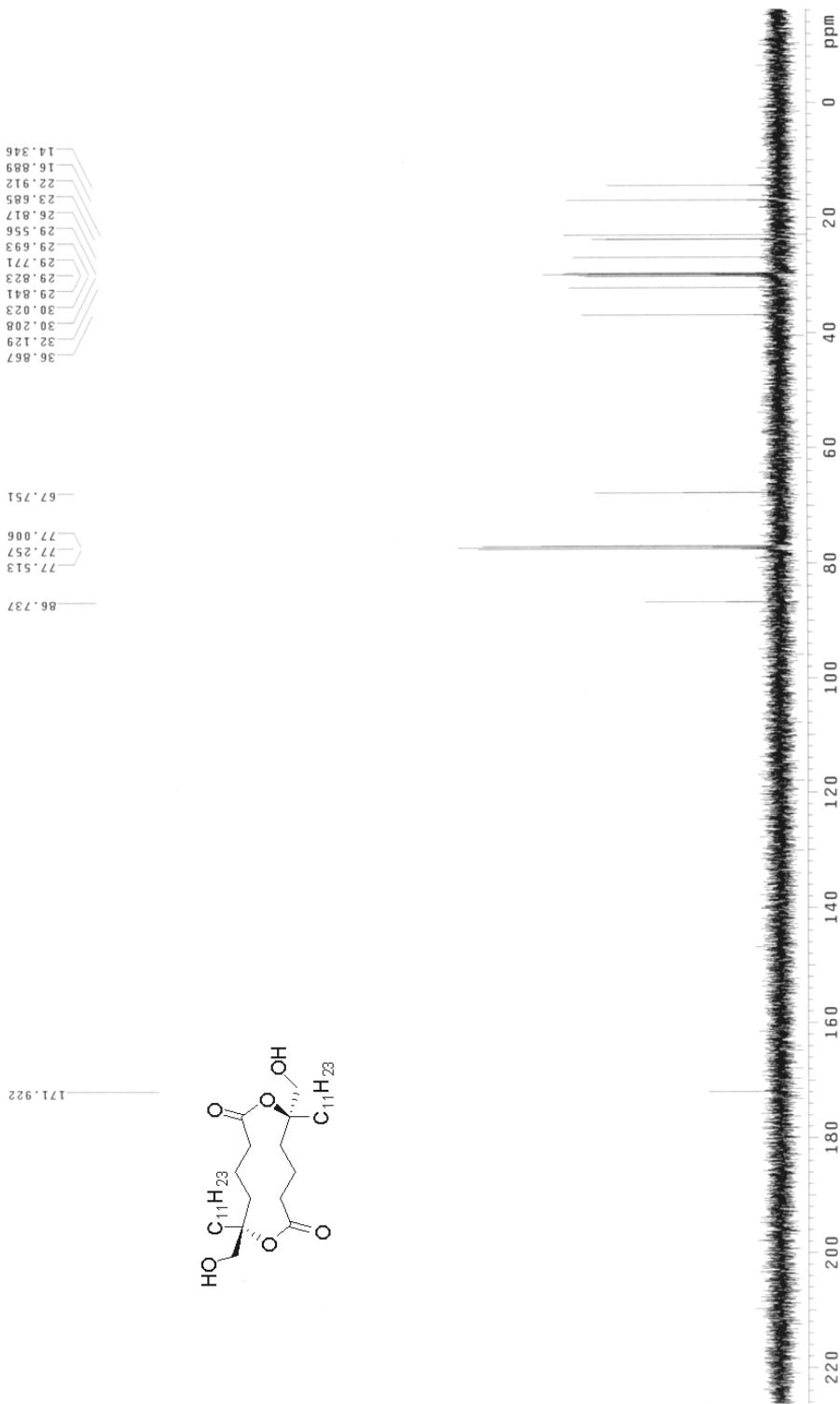






<sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>) of **4**



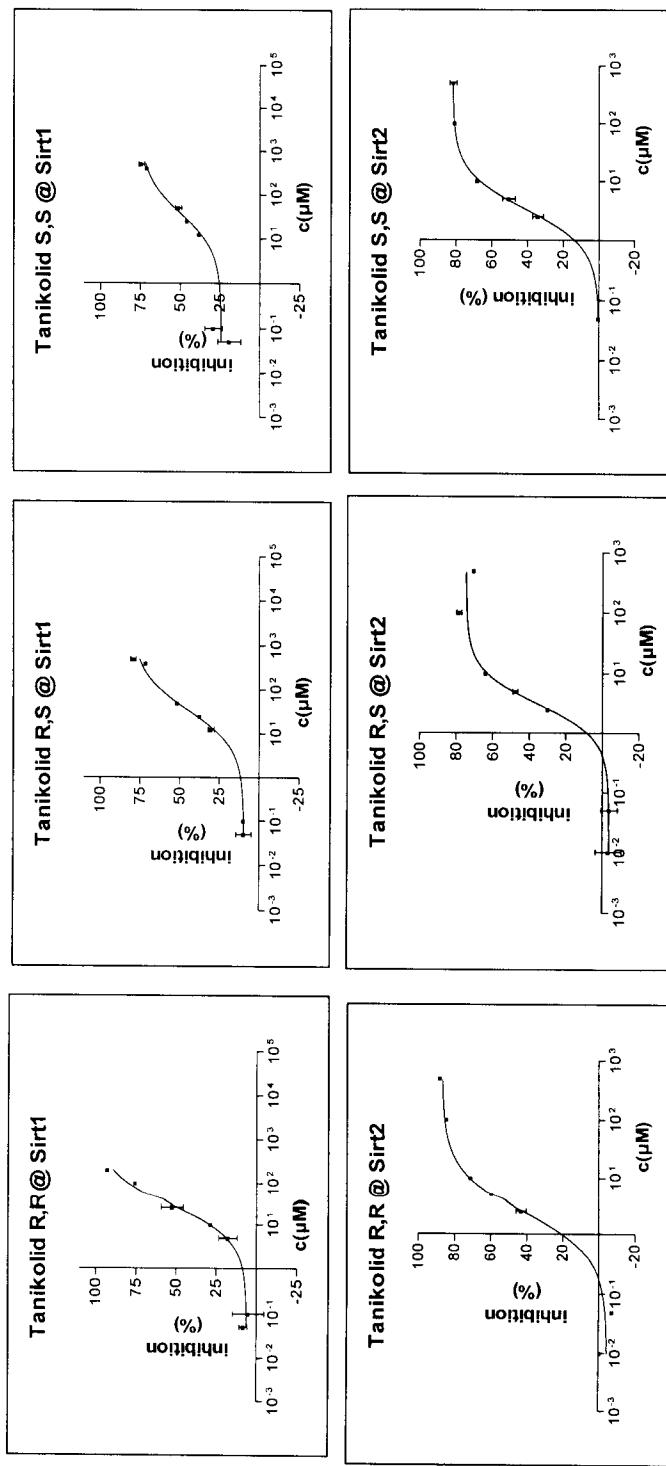


$^{13}\text{C}$ -NMR (125 MHz,  $\text{CDCl}_3$ ) of **4**

### Test results of Tanikolide Dimers Against Sirt1 and Sirt2

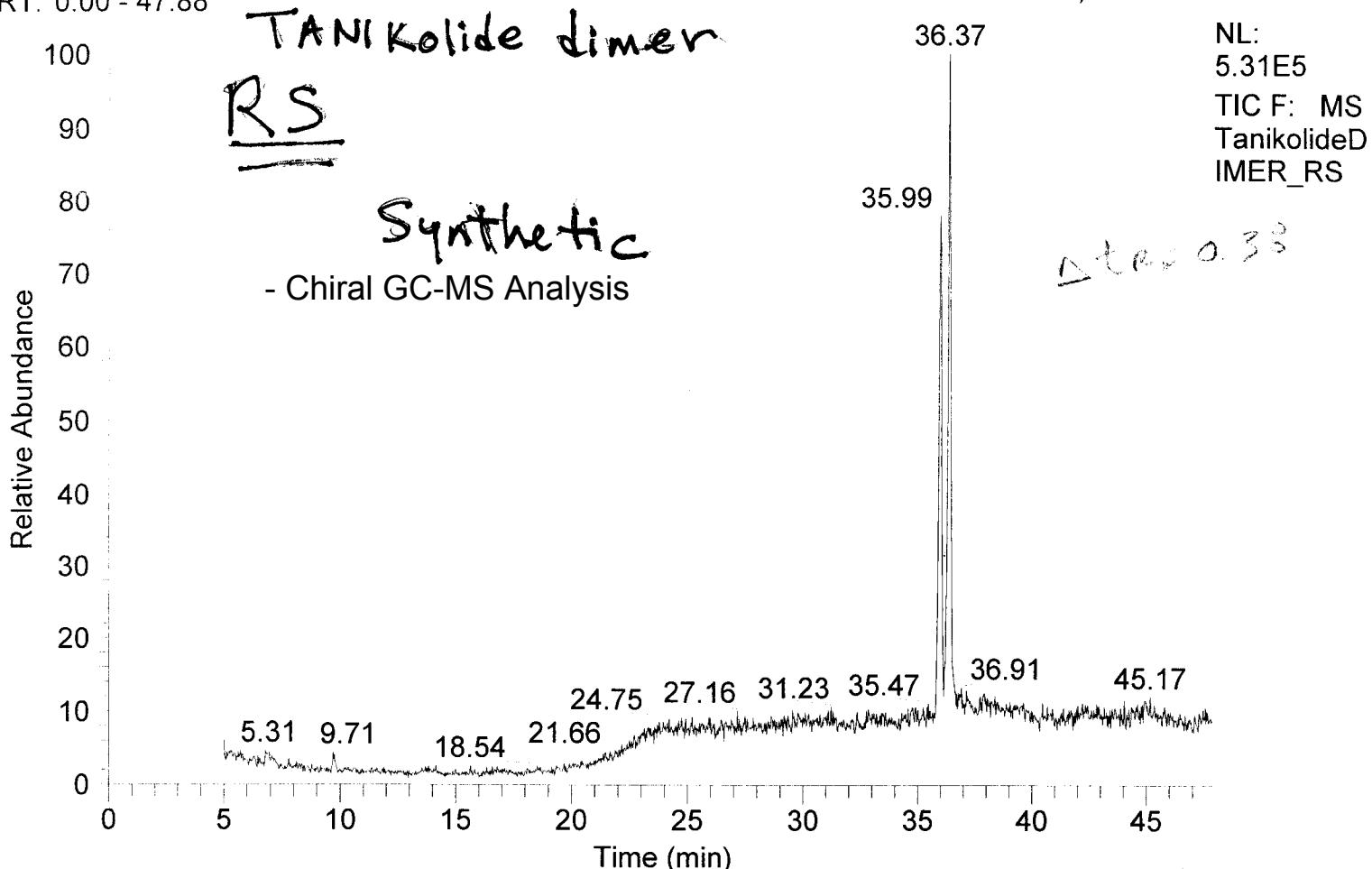
	$IC_{50} \pm SE [\mu M] @ Sirt1$	$IC_{50} \pm SE [\mu M] @ Sirt2$
R,R	28,8 ± 4,3	2,4 ± 0,2
S,S	36,4 ± 8,1	3,3 ± 0,2
R,S	34,5 ± 3,5	3,1 ± 0,3

Inhibition dose response curves for compounds **4**, **5** and *ent*-**5** to SIRT1 and SIRT2



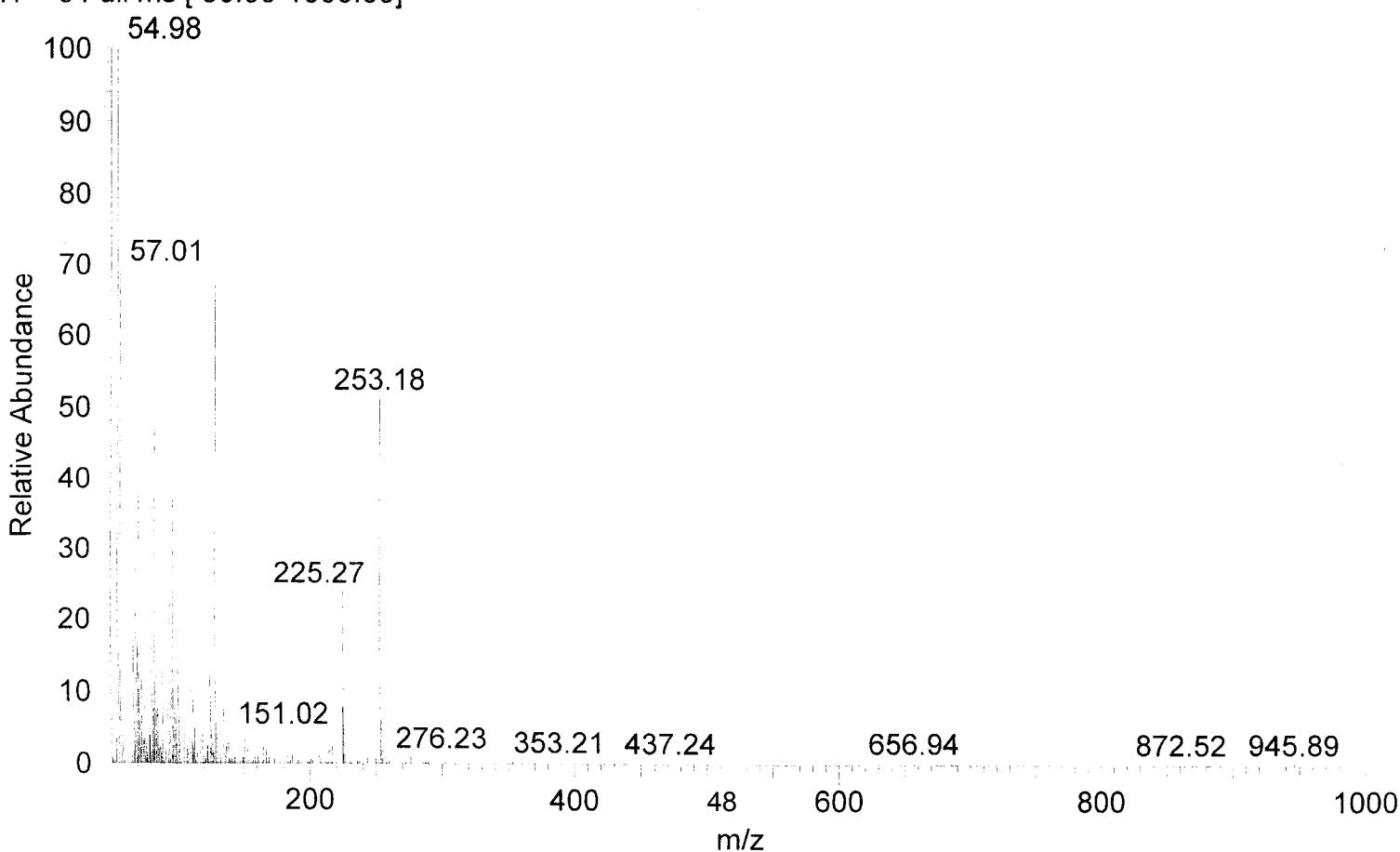
Copy

RT: 0.00 - 47.88

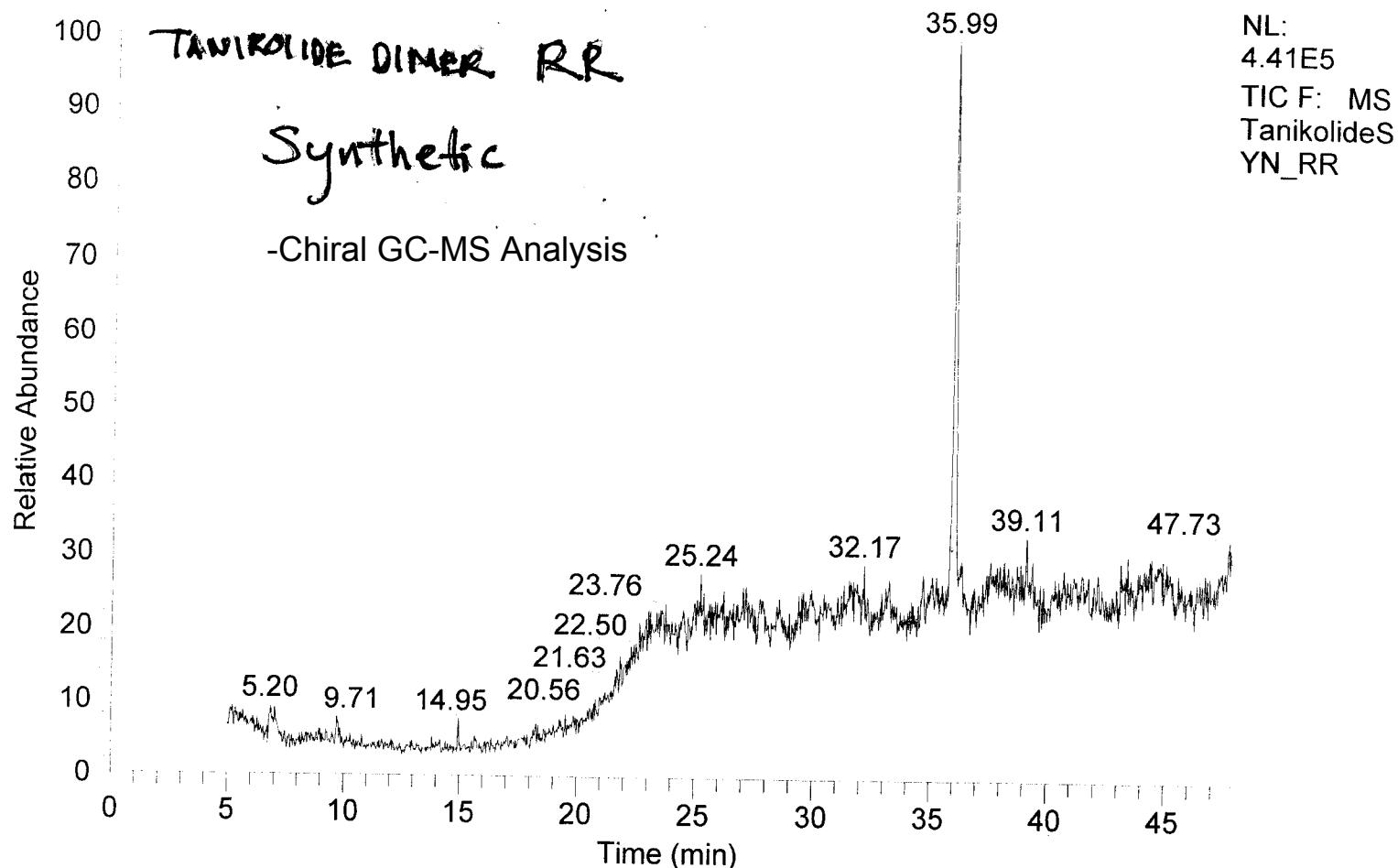


TanikolideDIMER\_RS #1212 RT: 35.99 AV: 1 NL: 4.13E4

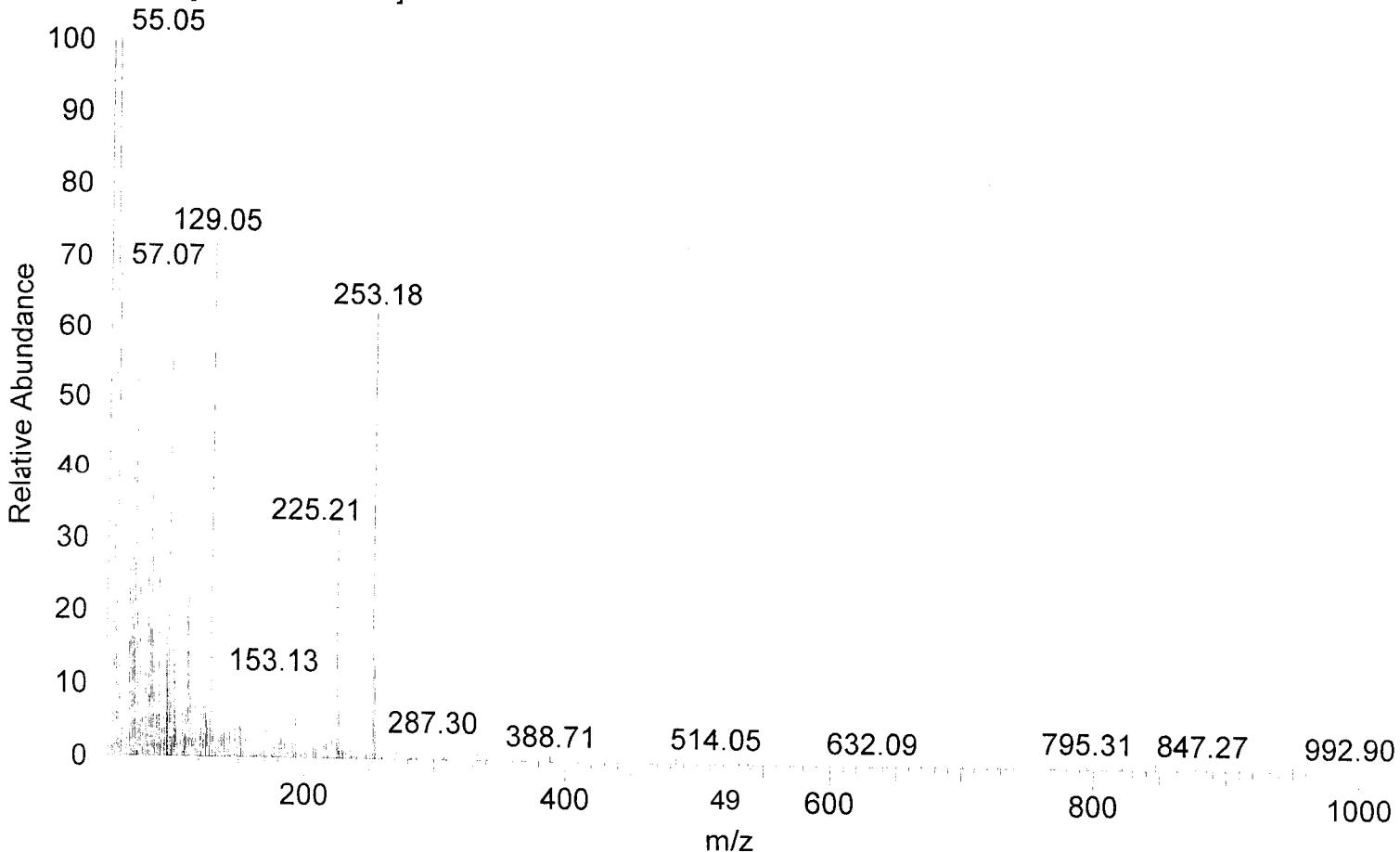
T: + c Full ms [ 50.00-1000.00]



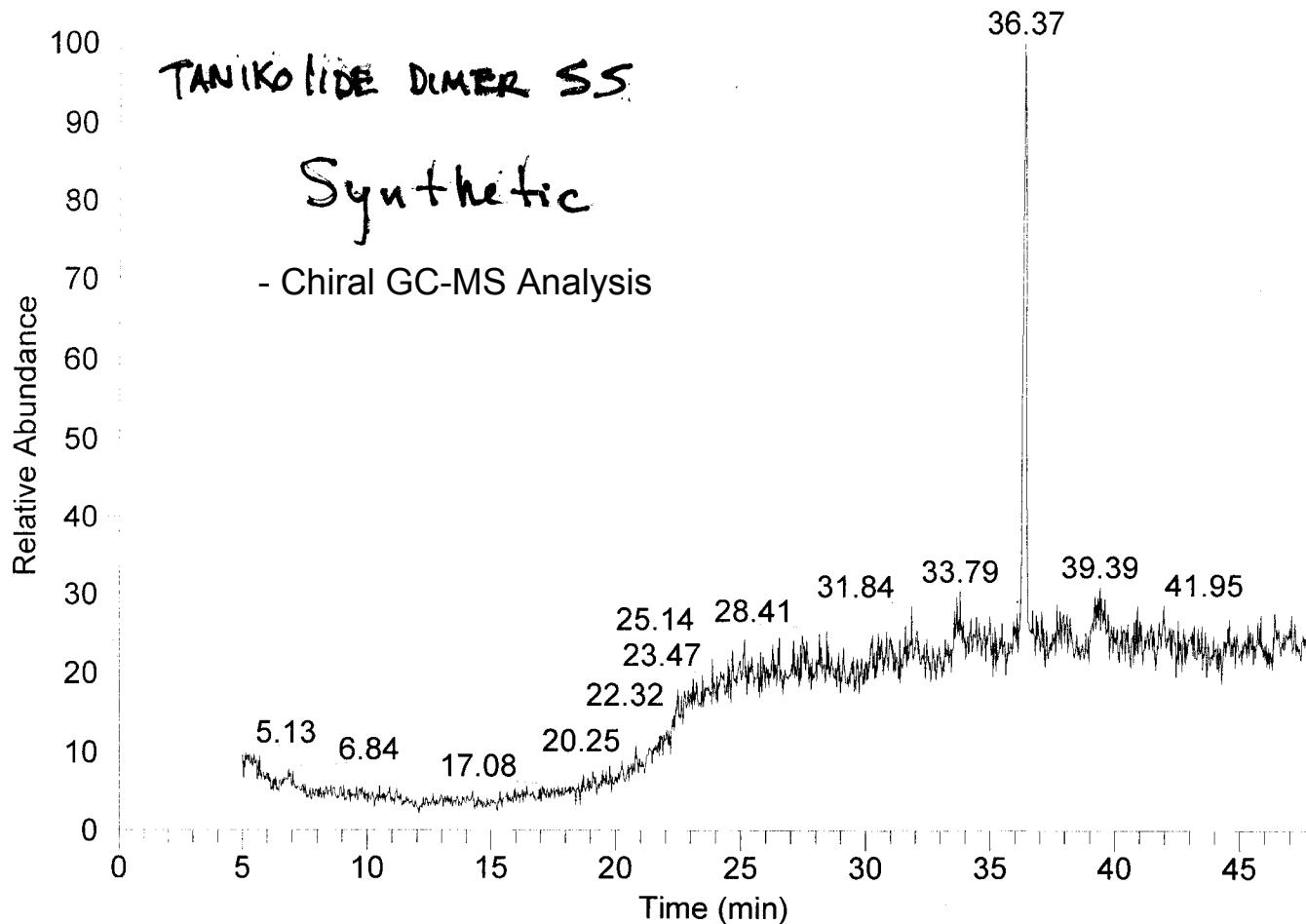
RT: 0.00 - 47.88



TanikolideSYN\_RR #1212 RT: 35.99 AV: 1 NL: 3.19E4  
T: + c Full ms [ 50.00-1000.00]



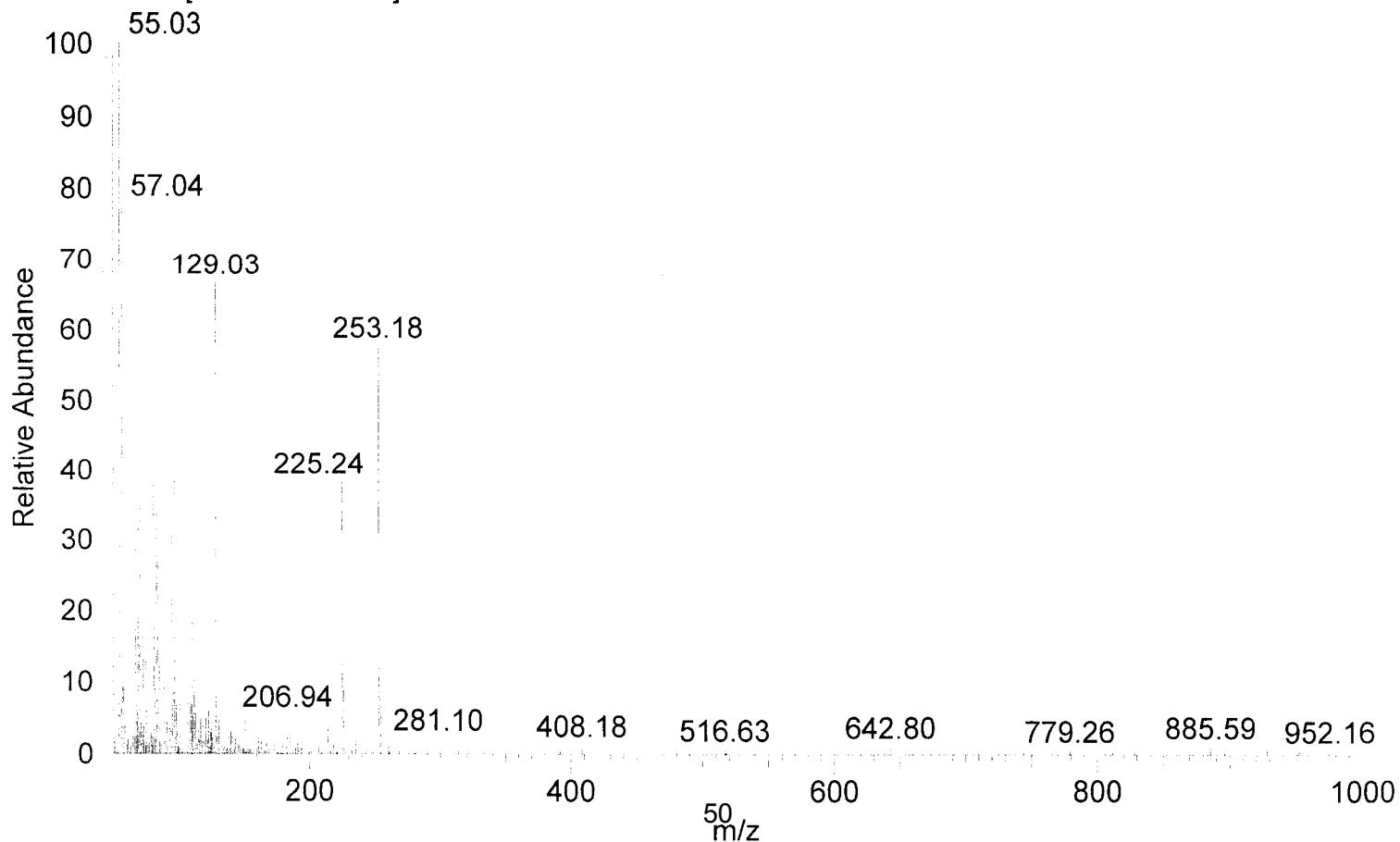
RT: 0.00 - 47.86



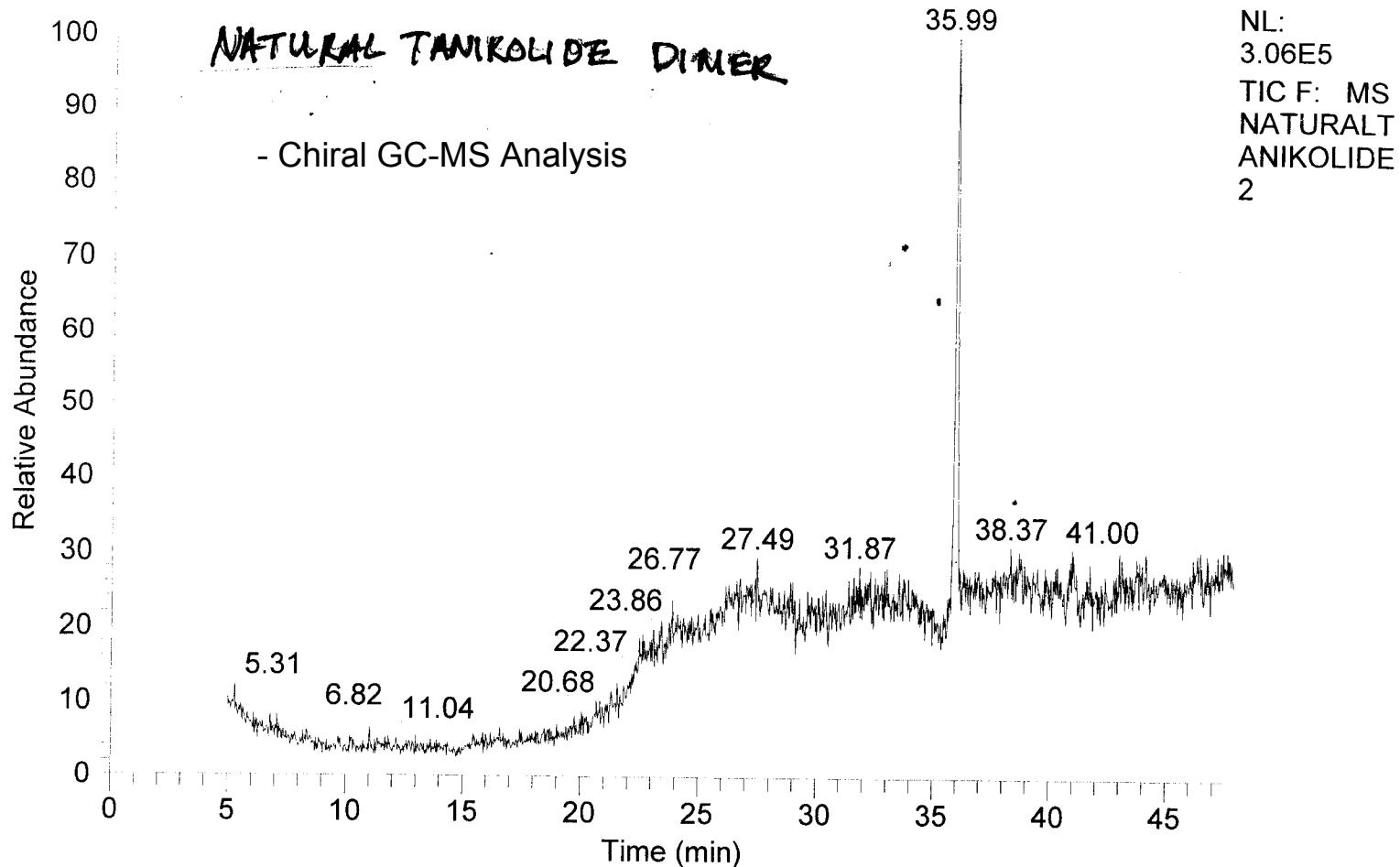
NL:  
3.57E5  
TIC F: MS  
TanikolideS  
YN\_SS

TanikolideSYN\_SS #1227 RT: 36.37 AV: 1 NL: 3.03E4

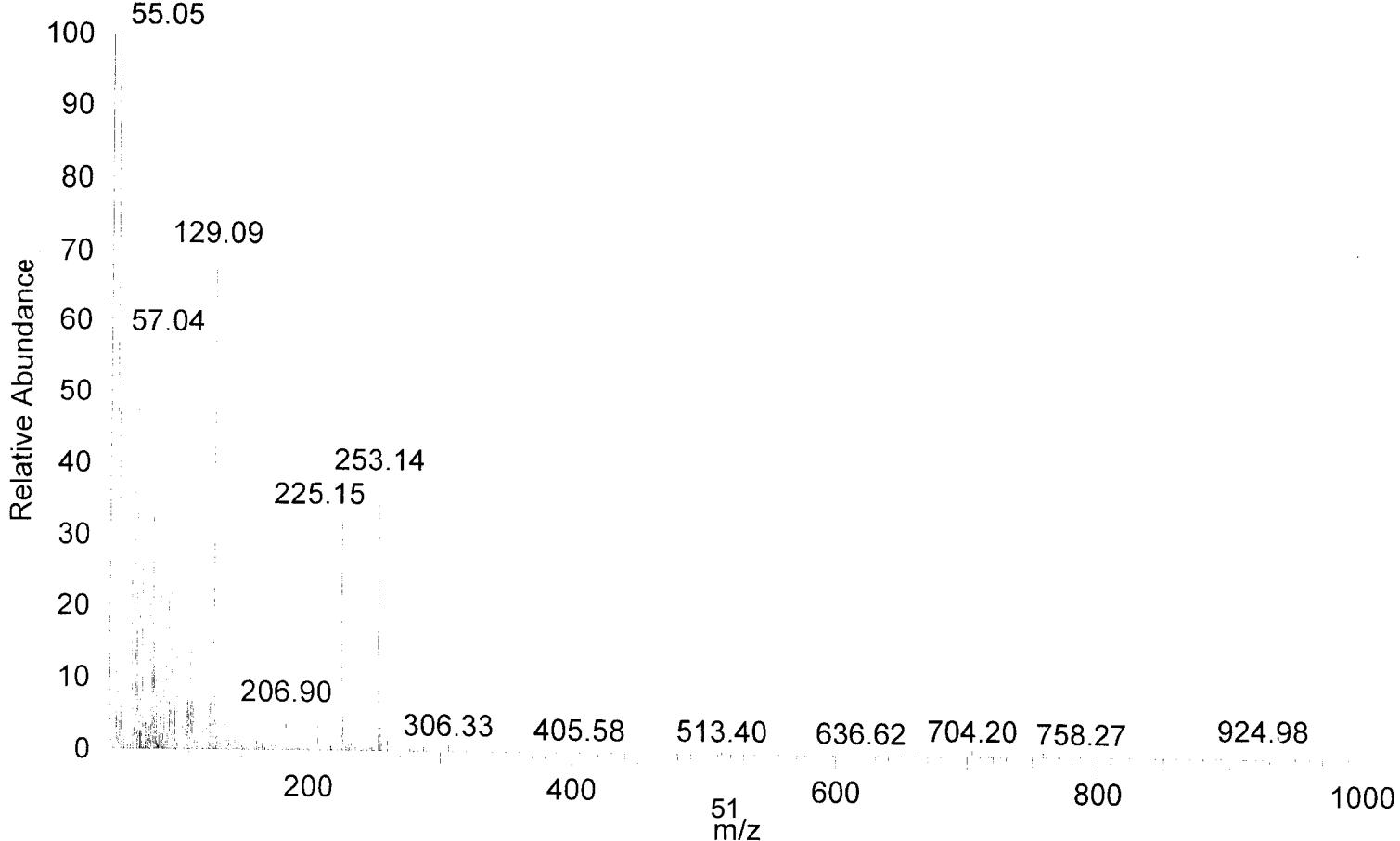
T: + c Full ms [50.00-1000.00]



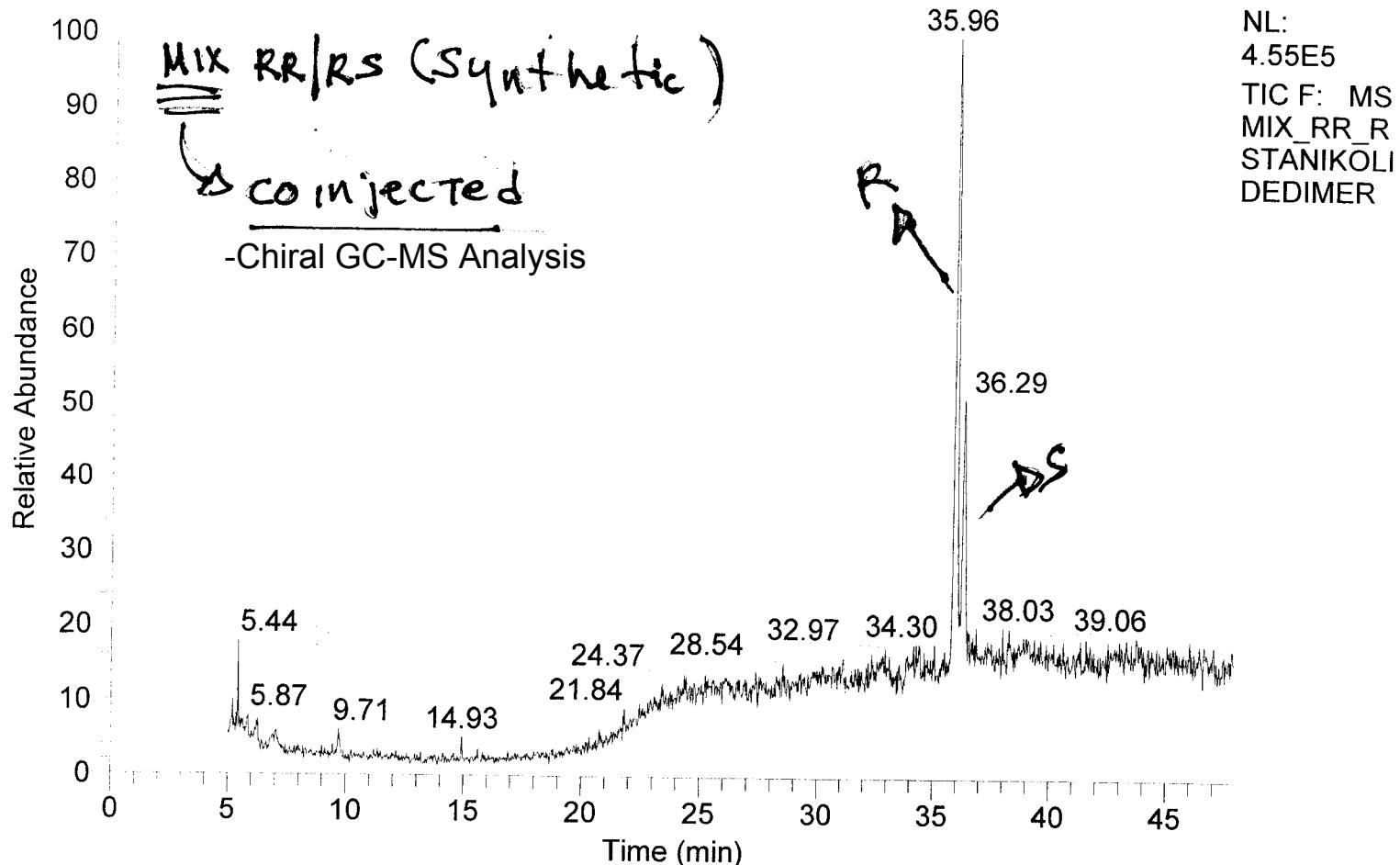
RT: 0.00 - 47.88



NATURAL TANIKOLIDE2 #1212 RT: 35.99 AV: 1 NL: 2.82E4  
T: + c Full ms [ 50.00-1000.00]



RT: 0.00 - 47.88



MIX\_RR\_RSTANIKOLIDEDIMER #1211 RT: 35.96 AV: 1 NL: 4.89E4  
T: + c Full ms [ 50.00-1000.00]

