

**Supplementary data**

**Table 1:** The list of reaction kinetics for all nodes in the dynamic integrated model linking protein aggregation and proteasomal degradation of substrate proteins and its inhibition, mitochondrial dysfunction and GSH synthesis/metabolism.

<b>1) Protein aggregation and proteasomal degradation</b>				
<b>(a) Protein aggregation and Lewy body formation</b>				
NODE	Equation	Parameters (units)	Mechanism and assumption	Reference(s)
Seed formation (control $\alpha$ -syn)	syn_control --> seeds	Kf = 1.2E-6 (1/sec)	Irreversible mass action. Here during aggregation of a-syn a condition is given that aggregation will occur when concentration of A-Syn $\leq$ 8uM	[3,7]
Seed formation (WT $\alpha$ -syn)	syn_WT --> seeds	Kf = 4.4E-2 (1/sec)	Irreversible mass action.	[1,2,4]
Seed formation (A53T $\alpha$ -syn)	syn_A53T --> Seeds	Kf = 8.79E-2 (1/sec)	Irreversible mass action. Assumed as the kinetics of Mutant A-syn aggregation is 2 fold higher than WT.	[1,2,4]
Seeds oligomerization	Seeds --> Oligomers	Kf = 3.9E-6 (1/sec)	Irreversible mass action.	[3,7]
Protofibril formation (oligomer polymerization)	Oligomers --> Protofibrils	Kf = 9.5E-7 (1/sec)	Irreversible mass action.	[3,7]
syn-agg Formation (Protofibril oligomerization)	Protofibrils --> synagg	Kf = 1E-4 (1/sec)	Irreversible mass action.	[3,7]

Lewy body formation (syn-agg polymerization)	syn-agg --> Lewy body	$K_f = 8.35E-6$ (1/sec)	Irreversible mass action.	[3,7]
<b>(b) Proteasomal degradation and its inhibition by aggregates or exogenous inhibitor (PI)</b>				
Protein Degradation	Proteins ( $\alpha$ -syn/ synphilin / Pael-R) --> Sink	$K_{catf\_proteasome} = 0.5$ ( $\mu\text{mol}/\text{sec}$ ) $K_m\_protein = 5\mu\text{M}$ $K_i\_toxic\ aggregates = 0.001\ \mu\text{M}$ $K_i\_PI = 1\mu\text{M}$	Michaelis menten kinetics. 1. Sink is shown for the degradation of protein 2. Toxic aggregates (which comprise intermediates of aggregation namely, oligomers, protofibrils, synagg and their dopamine adducts ) are shown to inhibit the proteasomal degradation. 3. Proteasoamal activity is modeled as michelis-menten kinetics.	[8,9]
<b>2) Mitochondrial Bioenergetics and Dysfunction</b>				
Electron transport chain-NADH oxidation	(a) $CI + NADH \rightarrow CIbnd + NAD$ (b) $3Hplus + UQ + CIbnd \rightarrow 3 Hplus + Uqbnd + CI$	(a) $K_m\_CI = 0.3\ \mu\text{M}$ , $K_m\_NADH = 300\ \mu\text{M}$ , $V_f = 7.5E-2$ ( $\mu\text{m}/\text{sec}$ ) (b) $V_f = 8.5\ E-2$ ( $\mu\text{m}/\text{sec}$ ), $K_m\_CIbnd = 0.15\ \mu\text{M}$ , $K_m\_UQ = 1.5\ E-1$ $K_i\_Toxic\_Aggregate = 3.33E-6\ \mu\text{M}$ $K_i\_PN = 7E-3\ \mu\text{M}$	Coupled reactions. Both reactions are modified Michelis-Menten equations.	[10-18]
ROS production coupled with CI oxidation	$CIbnd + O_2Mt \rightarrow CI + ROSmt$	$K_f = 5E-5$ ((Liter/ $\mu\text{M}$ )*(1/sec))	Irreversible mass action activated by ROS. This	[12,21-24]

		$K_{act\_ROS\_C1} = 0.1 \mu M$	reaction remains coupled with CI reduction. Mitochondria produces SO and primary ROS at two complexes I and III .CI and CIII produces SO on the matrix side of membrane.	
CIII reduction coupled with UQ oxidation.	$UQ_{bnd} + CIII \rightarrow UQ + CIII_{bnd}$	$V_f = 0.1 (\mu m/sec),$ $K_m_{UQ_{bnd}} = 0.15 (\mu M),$ $K_m_{C3} = 0.15 (\mu M)$	Modified Michaelis-Menten reaction.	[13-16,19,20]
CIII oxidation coupled with ROS production.	$C3_{bnd} + O2_{Mt} \rightarrow CIII + ROS_{mt}$	$K_f = 5E-5$ $((Liter/\mu M)*(1/sec))$ $K_{act\_ROS\_C3} = 0.1 \mu M$	Irreversible mass action activated by ROS. This reaction remains coupled with CIII reduction. Mitochondria produces SO and primary ROS at two complexes I and III .CI and CIII produces SO on the matrix side of membrane.	[12,21-24]
UQ oxidation coupled with cytochrome C (CYC <i>S</i> <sub>i</sub> ) reduction (During the process of electron flow, each complexes catalyzes the translocation of protons across inner membrane).	$CIII_{bnd} + 3$ $Mt_{matrix.Hplus_i} + CYC_{S_i}$ $\rightarrow CYC_{S_{bnd}} + 3$ $IMS.Hplus_o + CIII$	$V_f = 0.1 (\mu m/sec),$ $K_m_{CIII_{bnd}} = 0.15 (\mu M)$ $K_m_{CYC_{S_i}} = 0.15 (\mu M)$ $K_i_{PN} = 0.007 (\mu M)$	Modified Michaelis-Menten reaction. During the process of electron flow, each complex catalyzes the translocation of protons across IM.	[13-16,20]

CIV reduction coupled with cytochrome C Oxidation	CYCSbnd + CIV --> CYCSi + CIVbnd	Vf = 0.1 (μm/sec), Km_CYCSbnd = 0.15 (μM) Km_C4 = 0.15 (μM)	Modified Michaelis-Menten reaction. During the process of electron flow, each complex catalyzes the translocation of protons across IM.	[13-16,20}
CIV oxidation coupled with ROS production	O <sub>2</sub> Mt + CIVbnd --> CIV + ROSmt	Kf= 5E-5 ((Liter/μM)*(1/sec)) Kact_ROS_C3 = 0.1 μM	Irreversible mass action.	[12,21-24]
H <sub>2</sub> O Formation coupled with CIV oxidation.	CIVbnd + 5 Mtmatrix.Hplusi + Mtmatrix.O2Mt --> CIV + 3 IMS.Hpluso + Mtmatrix.H <sub>2</sub> Omt	Vf = 1.25 E-1 (μmol/sec), Km_CIVbnd = 0.15 μM, Ki_NO = 20 μM Ki_PN = 0.007 μm.	Modified Michaelis-Menten.	[13-16,20}
<b>(3) GSH Biosynthesis and metabolism (including GCL mRNA transcription)</b>				
Glutathione synthesis Step 1- mediated by γ-GCL	Glut + Cys + ATPcyto --> GluCys + ADPcyto + Pi	Kcatf_GCL = 7.344 1/sec Km_Glut = 3600 μM, Km_Cys = 100 μM, Km_ATPcyto = 400 μM, Ki_GSH = 500 μM.	Modified Michaelis-Menten kinetics with product inhibition by GSH.	[28-32]
Glutathione synthesis Step 2- mediated by GS (glutathione synthetase)	GluCys + Gly + ATPcyto -> Pi + ADPcyto + GSH	Kcatf_GS = 4.9 E-1, Km_ATPcyto = 4.5 μM, Km_Gly = 2000 μM.	This is modeled as modified Michaelis-Menten reaction.	[29,30,33,34]

GPx – Glutathione Peroxidase	$\text{H}_2\text{O}_2 + \text{GSH} \rightarrow \text{H}_2\text{O} + \text{GSSG}$	Kcatf_GPx = 21 (1/sec), Km_GSH= 3000 $\mu\text{M}$ , Km_H <sub>2</sub> O <sub>2</sub> = 10 $\mu\text{M}$	This reaction is modeled as two substrate Michaelis-Menten mechanism	Values from Brenda and optimized [EC no. 1.11.1.9]
GR – Glutathione Reductase	$\text{GSSG} + \text{NADPH} \rightarrow \text{GSH} + \text{NADP}$	Kcatf_GR = 100 1/sec, Km_NADPH = 85 $\mu\text{M}$ , Km_GSSG = 65 $\mu\text{M}$ .	This reaction is modeled as two substrate Michaelis-Menten mechanism	Values from Brenda and optimized [EC no. 1.8.1.7]
S-Nitroso-glutathione formation	(a) $\text{GSH} + \text{ONOO} \rightarrow \text{GSNO}$ (b) $\text{GSHc} + \text{N}_2\text{O}_3 \rightarrow \text{GSNOc} + \text{Hplus} + \text{NO}_2$	(a) Kf = 3 E-4 (1/sec) (b) Kf = 1E-5 (1/sec)	Irreversible mass action.	[35,36]
<b>GCL upregulation</b>				
Keap1_Nrf2 complex dissociation	$\text{Keap1\_Nrf2} \rightarrow \text{Keap1} + \text{Nrf2c}$	Vf = 0.01 1/sec, km_Keap1_Nrf2 = 50 $\mu\text{M}$ Ka = 2 E-6 uM	Modified Michaelis Menten which is activated by ROS/electrophile stress	[38- 44]
Nrf2 translocation	$\text{Nrf2c} \rightarrow \text{Nrf2n}$	Kf = 1.0E-5 (1/sec)	Irreversible mass action.	[38-40]
EpRE complex formation	$\text{Nrf2n} + \text{JUNn} + \text{EpRE} \rightarrow \text{EpRE\_Nrf2\_JunD}$	Kf = .01 $\mu\text{M}$ Kr = .008 $\mu\text{M}$	The binding of Nrf2 in nucleus to form EpRE complex is modeled with reversible mass action kinetics.	[38-40]
Transcription of $\gamma$ - GCL gene induced by EpRE Transcription factor .	$\text{DNA} \rightarrow \text{GCL mRNA}$	Km_EpRE_Nrf2_JunD = 19 $\mu\text{M}$ Vf = 0.0082 $\mu\text{mol/sec}$	This reaction is modeled as modified Michaelis-Menten mechanism.	[38-40]

Phosphorylation of Jun by JNK in the FosJun complex formation	FOSpn_JUNn --> FOSpnJUNpn	Kcatf_JNK = 25 (1/sec), Km_FosPn_JunN = 10 μM	This reaction is modeled as modified Michaelis-Menten mechanism.	[37-40]
Transcription of γ- GCL gene induced by AP1 Transcription factor.	DNA --> GCL mRNA	V max = 0.004 μmol/sec Km_FOSpn_JUNn = 19 μM	This reaction is modeled as Michaelis-Menten mechanism.	[41-43]
Translation of GCL enzyme	GCLmRNAc --> GCL	Kf = 3.5E-2 (1/sec)	The reaction is modeled with mass action kinetics.	[41-43]
Inhibition of proteasomal degradation of Nrf2 by toxic Aggregates.	Nrf2c --> Sink	Kcatf_proteasome = 0.01 (1/sec) Km_Nrf2c = μm Ki_Toxic aggregate = 1E-5 μM Ki_PI = 1 μM	This reaction is modeled as Modified Michaelis-Menten.	[8,40-45]

**List of abbreviations:**

CI – NADH Oxidoreductase (Complex I of Electron Transport Chain)  
CIbnd – Electron bound CI  
CIII – Cytochrome c Oxidase (Complex III of Electron Transport Chain)  
CIIIbnd- Electron bound CIII  
CIV- Cytochrome c Reductase (Complex IV of Electron Transport Chain)  
CIVbnd- Electron bound CIV  
UQ – Ubiquinone  
Cytc – Cytochrome c  
UQbnd - Electron bound Ubiquinone  
cytcband - Electron bound Cytochrome c  
ATP – Adenosine triphosphate  
ADP – Adenosine diphosphate  
Pi – Inorganic phosphate  
NADP – Nicotinamide adenosine diphosphate  
Arg – Arginine  
Cit- Citrulline  
NO -Nitric oxide  
O<sub>2</sub> – Oxygen  
ROS – Superoxide  
H<sub>2</sub>O<sub>2</sub> – Hydrogen peroxide  
PN – Peroxynitrite  
NO<sub>2</sub> – Nitrogen dioxide radical  
DOPAL – Metabolite of dopamine  
GSH – Glutathione  
GSSG – Glutathione disulphide  
GR – Glutathione reductase  
GPx – Glutathione peroxidase  
SOD – Super oxide dismutase  
MAO– Monoamine oxidase  
XO– Xanthine oxidase

NOS – Nitric oxide synthase  
Gly – Glycine  
GluCys -GlutamylCysteine  
Cys – Cysteine  
Glu – Glutamate  
GSNO – S- Nitrosoglutathione  
NO<sub>2</sub>R – Nitrogen Dioxide radicals  
N<sub>2</sub>O<sub>2</sub>- Anhydrous nitrous acid  
ONOOH - Peroxynitrous acid  
Hplusi – Proton concentration in mitochondrial matrix  
Hpluso - Proton concentration in mitochondrial membrane space.  
FOSpn\_JUNn – With Phosphorylated FOS in AP 1 complex.  
FOSpnJUNpn – Activated AP1  
DNA – Deoxyribonucleic acid.  
GCL mRNA – mRNA for GCL  
γ-GCL – Gamma glutamylcysteine ligase  
Nrf2c – Component of EpRE transcription factor in the cytosol.  
Keap1 -A cytoskeleton binding protein  
Keap1\_Nrf2 - Inactive Nrf2 bound with Keap1  
Nrf2n -Component of EpRE transcription factor in the nucleus.  
JUNn - Nuclear JUN protein  
EpRE - Electrophile Response Element or ARE  
EpRE\_Nrf2\_JunD -Electrophile Response Element or ARE bound with JunD and Nrf2 to form an active Transcription factor  
AP1- Activator Protein 1 - Activator protein 1 complex

**Parameters:**

Kf = Forward rate constant  
Kcat = Catalytic rate constant  
V max = The maximum rate of reaction  
Km = Michaelis Menten constant  
Ka = Activation constant  
Ki = Inhibition constant



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