Total Synthesis of (–)-Himandrine

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Supporting Information

General procedure. All reactions were performed in oven-dried or flame-dried round-bottomed flasks, modified Schlenk (Kjeldahl shape) flasks, or glass pressure vessels. The flasks were fitted with rubber septa and reactions were conducted under a positive pressure of argon. Stainless steel syringes or cannulae were used to transfer air- and moisture-sensitive liquids. Flash column chromatography was performed as described by Still et al. using silica gel (60 Å pore size, 40-63 μ m,

4-6% H_2O content, Zeochem).¹ Where necessary (so noted), silica gel was neutralized by treatment of the silica gel prior to chromatography with the eluent containing 1% triethylamine or 1% ammonium hydroxide. Analytical thin-layer chromatography was performed using glass plates precoated with 0.25 mm 230–400 mesh silica gel impregnated with a fluorescent indicator (254 nm). Where necessary (so noted), silica gel plates were neutralized by treatment with a solution of 1% triethylamine or 1% ammonium hydroxide in dichloromethane followed by heating on a hot plate $(\sim 250 \degree C)$. Thin layer chromatography plates were visualized by exposure to ultraviolet light and/or by exposure to an ethanolic phosphomolybdic acid (PMA), an acidic solution of *p*-anisaldehyde (anis), an aqueous solution of ceric ammonium molybdate (CAM), an aqueous solution of potassium permanganate (KMnO₄) or an ethanolic solution of ninhydrin followed by heating (≤ 1 min) on a hot plate (\sim 250 °C). Organic solutions were concentrated on Büchi R-200 rotary evaporators at \sim 20 Torr at 25–35 °C unless otherwise indicated.

Materials. Commercial reagents and solvents were used as received with the following exceptions: dichloromethane, diethyl ether, tetrahydrofuran, acetonitrile, and toluene were purchased from J.T. Baker (CycletainerTM) and were purified by the method of Grubbs et al. under positive argon pressure.² Triethylamine, diisopropylethylamine, and benzene were distilled over calcium hydride immediately before use. Acrolein was distilled over calcium sulfate immediately before use. Methyl vinyl ketone was distilled over potassium carbonate and calcium chloride immediately prior to use. Martin sulfurane was purchased from Aldrich and stored in a glove box under nitrogen atmosphere. *N*-Chlorosuccinimide (NCS) was recrystallized from benzene prior to use. Phosphorus oxychloride was distilled under reduced pressure before use. The molarity of *n*-butyllithium solutions was determined by titration using diphenylacetic acid as an indicator (average of three determinations).³ Ammonia saturated dichloromethane was obtained by agitation of dichloromethane in the presence of ammonium hydroxide followed by drying over anhydrous sodium sulfate. Where necessary (so noted) solutions were deoxygenated by alternate freeze (liquid nitrogen)/evacuation/argon-flush/thaw cycles (FPT, three iterations) or degassed by purging with argon for several minutes.

Instrumentation. Proton nuclear magnetic resonance (¹H NMR) spectra were recorded with a Varian 300 Mercury or a Varian inverse probe 500 INOVA spectrometer or a Bruker 400 spectrometer or a Bruker inverse probe 600 Avance spectrometer. Chemical shifts are recorded in parts per million on the δ scale and are referenced from the residual protium in the NMR solvent (CHCl₃: δ 7.27, C₆D₅H: δ 7.16). Data is reported as follows: chemical shift [multiplicity (s = singlet, d $=$ doublet, t = triplet, q = quartet, m = multiplet, app = apparent, br = broad), coupling constant(s) in Hertz, integration, assignment]. Carbon-13 nuclear magnetic resonance $(^{13}C$ NMR) spectra were recorded with a Bruker 600 Avance spectrometer, a Varian 500 INOVA spectrometer or a Bruker 400 spectrometer with a Magnex Scientific superconducting magnet and are recorded in parts per million on the δ scale and are referenced from the carbon resonances of the solvent (CDCl₃: δ 77.2, benzene-*d*6: δ 128.4). Infrared data were obtained with a Perkin-Elmer 2000 FT-IR and are reported as follows: [frequency of absorption (cm^{-1}) , intensity of absorption (s = strong, m = medium, w = weak, br = broad), assignment]. Optical rotations were measured on a Jasco-1010 polarimeter. We are grateful to Dr. Li Li for obtaining the mass spectroscopic data at the Department of Chemistry's Instrumentation Facility, Massachusetts Institute of Technology. High-resolution mass spectra (HRMS) were recorded on a Bruker APEX 4.7 Tesler FTMS spectrometer using electronspray ion source (ESI) or electronspray (ES). The structure of (–)-himandrine was obtained with the assistance

 ¹ Still, W. C.; Kahn, M.; Mitra, A. *J. Org. Chem.* **¹⁹⁷⁸**, *⁴³*, 2923.

² Pangborn, A. B.; Giardello, M. A.; Grubbs, R. H.; Rosen, R. K.; Timmers, F. J. *Organometallics* **1996**, *15*, 1518.

³ Kofron, W. G.; Baclawski, L. M. *J. Org. Chem*. **1976**, *41*, 1879.

of Dr. Peter Muller at the X-ray diffraction facility of Department of Chemistry, Massachusetts Institute of Technology, and Justin Kim of the Movassaghi group.

Additional Notes. Positional numbering system: For ease of direct comparison, particularly from *trans*-decalin $(-)$ -14 to himandrine $(-)$ -1, the numbering scheme used by Taylor and coworkers in the isolation paper 4 is used in this supporting document. In key instances the products are accompanied by the numbering system as shown below for this document.

 ⁴ Ritchie, E.; Taylor, W. C. *In the Alkaloids*; Manske, R. H. F., Ed.; Academic Press; New York, 1967; Vol. 9, Chapter 14.

(–)-(*S***)-2-(***N***-Phenyl-aminooxy)-hept-6-en-1-ol (S2):**

Nitrosobenzene (9.7 g, 0.090 mmol, 1 equiv) was added as a solid to a suspension of Dproline (0.46 g, 4.0 mmol, 4.0 mol%) in chloroform (50 mL) at 0° C, and the resulting mixture was sealed under an argon atmosphere. After 15 min, hept-6-enal⁵ (11.2 g, 10.0 mmol, 1.10 equiv) was added drop-wise via additional funnel to the bright green solution. After 3 h, the resulting brown reaction mixture was added dropwise via additional funnel to a suspension of sodium borohydride (14.2 g, 0.375 mol, 4.17 equiv) in methanol (50 mL) at 0 °C. After 30 min, saturated aqueous sodium bicarbonate solution (100 mL) was added, and the layers were separated. The aqueous layer was extracted with dichloromethane $(3 \times 150 \text{ mL})$. The combined organic layers were washed with brine (200 mL), were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The resulting oil was purified by flash column chromatography (silica gel: diam. 7 cm, ht. 7 cm; eluent: 33% EtOAc in hexanes) to afford alcohol (–)-**S2** (18.1 g, 91%) as a yellow oil $([\alpha]^{22}$ _D = –26.4 (*c* 1.0, CH₂Cl₂)). This compound was determined to be of >98.5% ee by chiral HPLC analysis (Chirapak AD-H, 95% hexanes / 5% *iso-propanol*, 3 mL/min, 215 nm t_R (major) = 18.03 min; t_R (minor) = 22.15 min).

The corresponding enantiomer, (+)-(*R*)-2-(*N*-Phenyl-aminooxy)-hept-6-en-1-ol (3.22 g, 80%, $[\alpha]^2$ ²_D = +26 (*c* 0.90, CH₂Cl₂)), was prepared according to the same procedure using L-proline as the catalyst. This compound was determined to be of >98.5% ee by chiral HPLC analysis (Chirapak AD-H, 95% hexanes / 5% *iso-propanol*, 3 mL/min, 215 nm t_R (minor) = 18.12 min; t_R (major) = 22.02 min). Structural assignment utilized additional information from gCOSY and HSQC.

 5 6-Heptenal was prepared from 7-octene-1,2-diol (commercially available), sodium metaperiodate, diethyl ether, water, 1h, 93%. Spectroscopic data matched those in the literature; see: Taylor, R. E.; Galvin, G. M.; Hilfiker, K. A.; Chen, Y. *J. Org. Chem.* **1998**, *63*, 9580.

(–)-(*S***)-Hept-6-ene-1,2-diol (6):**

Zinc powder (8.89 g, 136 mmol, 2.00 equiv) was added as a solid to a solution of alcohol $(-)$ -**S2** (15.1 g, 68.0 mmol, 1 equiv) in a mixture of ethanol and acetic acid (3:1, 340 mL) at 23 °C. After 2 h, the resulting mixture was filtered through a plug of celite (diam. 8.5 cm, ht. 2 cm), and the residue was washed with ethanol $(3 \times 150 \text{ mL})$. The filtrate was concentrated under reduced pressure at 30 °C. The residue was dissolved in ethyl acetate (400 mL), was washed with saturated aqueous sodium bicarbonate solution (100 mL), was dried over anhydrous sodium sulfate, was filtered, and was concentrated under reduced pressure. The resulting light yellow oil was purified via flash column chromatography (silica gel: diam. 5 cm, ht. 25 cm; eluent: 5% EtOAc in hexanes to 75% EtOAc in hexanes) to afford diol (-)-6 (7.1 g, 80%) as a yellow oil $([\alpha]^{2}$ ^D = -21 (*c* 0.44, EtOH)). The spectroscopic data was consistent with the literature.⁶ Structural assignment utilized additional information from gCOSY and HSQC.

 ⁶ Takahata, H.; Takahashi, S.; Kouno, S.; Momose, T. *J. Org. Chem.* **¹⁹⁹⁸**, *⁶³*, 2224.

(+)-(*S***)-1-(***tert***-Butyl-dimethyl-silanyloxy)-hept-6-en-2-ol (7):**

tert-Butylchlorodimethylsilane (6.4 g, 42 mmol, 1 equiv) was added as a solid to a solution of diol (–)-**6** (6.10 g, 47.0 mmol. 1.05 equiv), 4-dimethylaminopyridine (229 mg, 1.90 mmol, 4.00 mol%), and imidazole (4.1 g, 60 mmol, 1.5 equiv) in *N,N*-dimethylformamide (230 mL) at 0 ºC, and the reaction mixture was sealed under an argon atmosphere. After 3 h, the reaction mixture was diluted with diethyl ether (300 mL) and brine (150 mL), and the layers were separated. The aqueous layer was extracted with diethyl ether $(3 \times 150 \text{ mL})$. The combined organic layers were washed with brine (250 mL), were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The resulting oil was purified via flash column chromatography (silica gel: diam. 5 cm, ht. 17 cm; eluent: 10% EtOAc in hexanes) to provide silyl ether (+)-**7** (9.6 g, 94%) as a pale yellow oil $([\alpha]^{22}{}_{D} = +3.5$ (*c* 1.4, CH₂Cl₂)).

(–)-(*S***)-***tert***-Butyl-(2-methoxy-hept-6-enyloxy)-dimethyl-silane (S3):**

Oven-dried 4Å molecular sieves (19.6 g, 2:1, wt/wt), Proton Sponge® (25.5 g, 118 mmol, 3.00 equiv), and trimethoxyl oxonium tetrafluoroborate (14.5 g, 98.0 mmol, 2.51 equiv) were added sequentially to a solution of alcohol (+)-**7** (9.6 g, 39 mmol, 1 equiv) in dichloromethane (392 mL) at 23 °C, and the reaction mixture was sealed under an argon atmosphere. After 3 h, the reaction mixture was filtered through a plug of celite (diam. 8.5 cm, ht. 3 cm), and the residue was washed with dichloromethane $(3 \times 100 \text{ mL})$. The filtrate was concentrated under reduced pressure. The residue was dissolved in a mixture of hexanes and ethyl acetate (1 : 1, 400 mL), and the residual insoluble light brown solid was removed by filtration, and was washed with a mixture of hexanes and ethyl acetate (1 : 1, 2 \times 100 mL). The filtrate was washed with saturated aqueous copper sulfate solution (150 mL) and brine (150 mL), was dried over anhydrous sodium sulfate, was filtered, and was concentrated under reduced pressure. The resulting oil was purified via flash column chromatography (silica gel: diam. 5 cm, ht. 17 cm; eluent: 3% EtOAc in hexanes) to afford methyl ether (-)-**S3** (9.3 g, 93%) as a colorless oil ($[\alpha]^{22}$ _D = -11 (*c* 1.4, CH₂Cl₂)). Structural assignment utilized additional information from gCOSY and HSQC.

(+)-(*S***)-2-Methoxy-hept-6-en-1-ol (8):**

Thionyl chloride (0.495 mL, 13.6 mmol, 0.400 equiv) was added dropwise to methanol (340 mL) at 23 °C. After 5 min, the resulting methanolic hydrochloric acid solution (0.04 M) was added to a solution of silyl ether (–)-**S3** (8.6 g, 34 mmol, 1 equiv) in methanol (340 mL) at 23 °C. After 15 min, the reaction solvent was removed under reduced pressure, and the residue was purified by flash column chromatography (silica gel: diam. 3 cm, ht. 10 cm; eluent: 33% EtOAc in hexanes) to afford alcohol (+)-**8** (4.9 g, 98%) as a pale yellow oil $([\alpha]^{22}{}_{D} = +22$ (*c* 0.70, CH₂Cl₂)).

(–)-1,1-Dibromo-3-methoxy-octa-1,7-diene (9):

Dimethyl sulfoxide (24.2 mL, 340 mmol, 10.0 equiv), diisopropylethylamine (30.5 mL, 170 mmol, 5.00 equiv) and sulfur trioxide pyridine complex (16.2 g, 102 mmol, 3.00 equiv) were added sequentially to a solution of alcohol (+)-**8** (4.9 g, 34 mmol, 1 equiv) in dichloromethane (170 mL) at 23 °C, and the reaction mixture was sealed under an argon atmosphere. After 15 min, the reaction mixture was diluted with diethyl ether (250 mL) and water (100 mL), and the layers were separated. The aqueous layer was extracted with diethyl ether $(3 \times 150 \text{ mL})$. The combined organic layers were washed sequentially with aqueous hydrochloric acid solution (1M, 100 mL), saturated aqueous sodium bicarbonate solution (100 mL), and brine (100 mL), were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The resulting oil was purified via flash column chromatography (silica gel: diam. 3 cm, ht. 10 cm; eluent: 33% diethyl ether in hexanes) to afford aldehyde **S4** as a colorless oil. 7

Triphenylphosphine (21.4 g, 81.6 mmol, 2.40 equiv) was added as a solid to a solution of carbon tetrabromide (13.5 g, 40.8 mmol, 1.20 equiv) in dichloromethane at 0 $^{\circ}$ C, and the reaction mixture was sealed under an argon atmosphere. After 15 min, the solution of aldehyde **S4** in dichloromethane (10 mL) was added dropwise via cannula to the resulting orange reaction mixture. After 15 min, excess dibromophosphorane was quenched by sequential addition of triethylamine (11.5 mL, 81.6 mmol, 2.40 equiv) and methanol (3.5 mL, 81.6 mmol, 2.40 equiv). The reaction mixture was added dropwise to a mixture of hexanes and ethyl acetate (5:1, 400 mL). The resulting light brown solid was removed by filtration, and was washed with a mixture of hexanes and ethyl acetate (5:1, 100 mL). The filtrate was concentrated, and the residue was purified by flash column chromatography (silica gel: diam. 5 cm, ht. 17 cm; eluent: 10% diethyl ether in hexanes) to provide dibromide (-)-9 (6.1 g, 65% 2-steps) as a colorless oil $([\alpha]^{22}$ _D = -19 (*c* 2.8, CH₂Cl₂)). Structural assignment utilized additional information from gCOSY.

¹ H NMR (500 MHz, CDCl ₃ , 20 °C):	6.30 (d, $J = 8.5$ Hz, 1H, CBr ₂ =CH), 5.83-5.72 (m, 1H, CH ₂ =CH), 5.03-4.91 (m, 2H, CH ₂ =CH), 3.89-3.84 (m, 1H, CHCH=CBr ₂), 3.28 (s, 3H, OCH ₃), 2.08-2.03 (m, 2H, CH ₂ =CHCH ₂), 1.65-1.57 (m, 1H, CHH'CH ₂), 1.53- 1.37 (m, 3H, CHH'CH ₂).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C):	140.1 (CH ₂ =CH), 138.7, 115.1 (CH ₂ =CH), 91.4, 81.5 $(OCH3), 57.2, 33.9, 33.8, 24.3.$
FTIR (thin film) cm^{-1} :	2931 (s), 2822 (w), 1641 (m), 1617 (m), 1458 (m), 1105 (s) , 912 (s), 782 (s).
Elemental Analysis:	calc'd for $C_9H_{14}Br_2O$: C, 36.27; H, 4.74, found: C, 35.98; H, 4.70.
TLC (10% Et_2O in hexanes), Rf.	0.78 (UV, CAM).

⁷ Reduction of a sample of aldehyde **S4** (NaBH₄) returned the alcohol $(+)$ -**8** with the same optical activity as compared to the starting alcohol (+)-**8**.

(4-Bromo-6-methoxy-1-methyl-undeca-2,4,10-trienyloxy)-*tert***-butyl-dimethyl-silane (11):**

Tetrakis(triphenylphosphine)palladium (1.54 g, 1.30 mmol, 8.00 mol%) and thallium carbonate (15.7 g, 33.0 mmol, 2.00 equiv) were added sequentially to a degassed solution of dibromide (–)-**9** (4.99 g, 17.0 mmol, 1 equiv) and boronic acid **10**⁸ (4.04 g, 17.6 mmol, 1.10 equiv) in a mixture of tetrahydrofuran and water (2:1, 68 mL) at 23 °C in the dark, and the reaction mixture was sealed under an argon atmosphere. After 10 h, the pale yellow heterogeneous reaction mixture was diluted with ethyl acetate, was filtered through a plug of silica gel (diam. 5 cm, ht. 3 cm), and the residue was washed with ethyl acetate $(3 \times 100 \text{ mL})$. The filtrate was washed with saturated aqueous sodium bicarbonate solution (100 mL) and brine (100 mL), was dried over anhydrous sodium sulfate, was filtered and was concentrated under reduced pressure. The resulting oil was purified by flash column chromatography (silica gel: diam. 5 cm, ht. 10 cm; eluent: 33% EtOAc in hexanes) to afford vinyl bromide **11** (6.7 g, 97%) as a 1:1 mixture of two diastereomers. Structural assignment utilized additional information from gCOSY.

¹H NMR (500 MHz, CDCl₃, 20 °C, one diastereomer noted by *): 6.23-6.19 (m, 1H, CH=C**H**–CBr;

¹³C NMR (100 MHz, CDCl₃, 20 °C): 140.5, 140.4, 138.8, 138.8, 134.2, 134.1, 127.0, 126.9, 126.9, 126.8, 114.9, 80.5, 77.4, 68.3, 56.9, 34.4, 33.9, 26.1, 24.6, 18.5, –4.5.

FTIR (thin film) cm^{-1} : : 2929 (s), 1470 (m), 1368 (w), 1253 (s), 1147 (s), 1093 (s), 835 (m), 776 (m).

HRMS (ESI) calc'd for C₁₉H₃₅BrNaO₂Si [M+Na]⁺: 425.1482, found: 425.1491.

TLC (10% EtOAc in hexanes), *Rf*: 0.65(UV, CAM).

 ⁸ The boronic acid was prepared as described previously; see Movassaghi, M.; Hunt, D. K.; Tjandra, M. *J. Am. Chem. Soc.* **²⁰⁰⁶**, *¹²⁸*, 8126.

1-{1-[3-(*tert***-Butyl-dimethyl-silanyloxy)-but-1-enyl]-3-methoxy-octa-1,7-dienyl}-azetidin-2-one**

(S5): 2-Azetidinone (1.16 g, 16.3 mmol, 2.50 equiv), copper iodide (1.58 g, 8.30 mmol, 50.0 mol%), potassium carbonate (5.74 g, 41.5 mmol, 2.50 equiv) and *N,N*-dimethyl ethylene diamine (4.50 mL, 41.5 mmol, 2.50 equiv) were added sequentially to a solution of vinyl bromide **11** (6.72 g, 16.6 mmol, 1 equiv) in anhydrous toluene (16 mL) at 23 °C in a 50-mL schlenk flask. The reaction vessel was sealed under an argon atmosphere, and it was heated to 120 °C. After 16 h, the reaction mixture was cooled to 23 °C and filtered through a plug of silica gel (diam. 3 cm, ht. 3 cm), and the residue was washed with ethyl acetate $(3 \times 200 \text{ mL})$. The filtrate was concentrated under reduced pressure, and the residue was purified by flash column chromatography (silica gel: diam. 3 cm, ht. 15 cm; eluent: 33% EtOAc in hexanes) to afford triene **S5** (5.4 g, 85%) as a 1:1 mixture of two diastereomers. Structural assignment utilized additional information from gCOSY.

(–)-(*S***)-1-[3-Methoxy-1-(3-oxo-but-1-enyl)-octa-1,7-dienyl]-azetidin-2-one (S7):**

Tetrabutylammonium fluoride solution in tetrahydrofuran (1.0 M, 21 mL, 21 mmol, 1.5 equiv) was added via syringe to a solution of triene **S5** (6.9g, 17.6 mmol) in tetrahydrofuran (176 mL) at 0 °C under an argon atmosphere, and the reaction mixture was allowed to warm to 23 °C. After 2 h, the reaction mixture was diluted with diethyl ether (400 mL) and brine (150 mL), and the layers were separated. The aqueous layer was extracted with diethyl ether $(3 \times 150 \text{ mL})$. The combined organic layers were washed with brine (100 mL), were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The residue was purified by flash column chromatography (silica gel: diam. 3 cm, ht. 8 cm; eluent: 75% EtOAc in hexanes) to afford alcohol **S6** (4.6 g, 95%) as a pale yellow oil, which was used directly in the following oxidation step.

Dimethyl sulfoxide (12.5 mL, 175 mmol, 10.0 equiv), diisopropylethylamine (15.7 mL, 87.5 mmol, 5.00 equiv), and sulfur trioxide pyridine complex (8.40 g, 52.5 mmol, 3.00 equiv) were added sequentially to a solution of alcohol **S6** (4.62 g, 16.6 mmol, 1 equiv) in dichloromethane (176 mL) at 23 °C, and the reaction mixture was sealed under an argon atmosphere. After 15 min, the reaction mixture was diluted with diethyl ether (250 mL) and brine (100 mL), and the layers were separated. The aqueous layer was extracted with diethyl ether $(3 \times 100 \text{ mL})$. The combined organic layers were washed with brine (200 mL), were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The resulting oil was purified by flash column chromatagraphy (silica gel: diam. 3 cm, ht. 10 cm; eluent: 75% EtOAc in hexanes) to afford ketone (–)-**S7** (3.9 g, 83%) as a pale yellow oil $([\alpha]^{22}{}_{D} = -29$ (*c* 0.27, CH₂Cl₂)). Structural assignment utilized additional information from gCOSY and HSQC.

(–)-(*S***)1-{1-[3-(***tert***-Butyl-dimethyl-silanyloxy)-buta-1,3-dienyl]-3-methoxy-octa-1,7-**

dienyl}-azetidin-2-one (12): Triethylamine (2.1 mL, 15 mmol, 1.5 equiv) and *tert*butyldimethylsilyl trifluoromethanesulfonate (2.9 mL, 12 mmol, 1.2 equiv) were added sequentially to a solution of ketone (–)-**S7** (2.77g, 10.0 mmol, 1 equiv) in dichloromethane (100 mL) at −78 °C under an argon atmosphere. After 2 h, saturated aqueous sodium bicarbonate solution (40 mL) was added, and the reaction mixture was allowed to warm to 23 °C. The layers were separated, and the aqueous layer was extracted with dichloromethane $(3 \times 100 \text{ mL})$. The combined organic layers were washed with brine, were dried over anhydrous sodium sulfate were filtered, and were concentrated under reduced pressure. The resulting oil was then purified by flash column chromatography (silica gel, treated with 1% NEt₃ in [49% EtOAc in hexanes], diam. 3 cm, ht. 10 cm; eluent: 1% of NEt₃ in [49% EtOAc in hexanes]) to afford silyl enol ether $(-)$ -12 (3.4 g, 86%) as a pale yellow oil $([\alpha]^{22}$ _D = -28 (c 0.27, CH₂Cl₂)). Structural assignment utilized additional information from gCOSY and HSQC.

¹H NMR (600 MHz, C_6D_6 , 20 °C):

6.67 (d, $J = 15$ Hz, 1H, CH=CHCN), 6.19 (d, $J = 15.6$ Hz, 1H, C**H**=CHCN), 5.81-5.72 (m, 1H, CH2=C**H**), 5.48 (d, *J* = 9.0 Hz, 1H, NC=C**H**), 5.04-4.95 (m, 2H, C**H**2=CH), 4.42 (s, 1H, C**H**H'=COTBS), 4.38 (s, 1H, CH**H'**=COTBS), 4.09-4.04 (m, 1H, C**H**OCH3), 3.21 (s, 3H, OC**H**3), 2.87 (app-q, *J* = 4.8 Hz, 1H, C**H**2C(=O)N), 2.80 (app-q, $J = 4.8$ Hz, 1H, CH₂C(=O)N), 2.45 (t, $J =$ 4.2 Hz, 2H, C**H**2NC=O), 2.00 (app-t, *J* = 6.6 Hz, 2H, CH**H**'CH=CH2), 1.74-1.49 (m, 4H, C**H**2C**H**2), 0.98 (s, 9H, SiC(C**H**3)3), 0.14 (s, 6H, Si(C**H**3)2).

¹³C NMR (100 MHz, C_6D_6 , 20 °C): 165.3, 155.4, 139.4, 135.9, 131.5, 129.2, 126.6, 115.1, 98.0, 77.6, 56.8, 41.8, 37.1, 35.5, 34.4, 26.3, 25.6, 18.8, $-4.2, -4.2$.

FTIR (thin film) cm^{-1} : : 2930 (m), 1760 (s), 1622 (w), 1583 (w), 1396 (m), 1318 (m), 1254 (m), 1102 (m), 1028 (m), 840 (m), 782 (m). HRMS (ESI) calc'd for $C_{22}H_{37}NNaO_3Si$ [M+Na]⁺: 414.2435, found: 414.2436.

TLC (1% NEt3 in [32% EtOAc in hexanes]), *Rf*: 0.55 (UV, CAM).

(–)-(*S***,2***E***,8***Z***,10***E***)-12-(***tert***-Butyl-dimethyl-silanyloxy)-7-methoxy-9-(2-oxo-azetidin-1-yl)-**

trideca-2,8,10,12-tetraenal (13): Acrolein (0.80 mL, 12 mmol, 5.0 equiv) and the Grubbs-Hoveyda catalyst (150 mg, 0.240 mmol, 10.0 mol%) were added sequentially to a solution of silyl enol ether $(-\frac{1}{2})$)-**12** (0.95 g, 2.4 mmol, 1 equiv) in dichloromethane (8 mL) at 23 °C, and the reaction vessel was sealed under an argon atmosphere. After 1 h, the reaction mixture was directly loaded onto and purified by flash column chromatography (silica gel, treated with 1% NEt₃ in [32% EtOAc in hexanes], diam. 5 cm, ht. 15 cm; eluent: 1% NEt₃ in [32% EtOAc in hexanes]) to afford tetraenal (-)-**13** (855 mg, 85%) as a pale yellow oil $([\alpha]^{22}{}_{D} = -70$ (*c* 0.20, benzene)). The starting material (-)-12 (140 mg, 15%) was also recovered.

HRMS (ESI) calc'd for $C_{23}H_{37}NNaO_4Si$ [M+Na]⁺: 442.2384, found: 442.2381.

TLC (1% NEt3 in [32% EtOAc in hexanes]), *Rf*: 0.33 (UV, CAM).

*trans***-Decalin aldehyde (–)-14:**

2,6-Di-*tert*-butyl-4-methylphenol (10 mg, 45 µmol, 0.56 mol%) and *N,N*-diethyl aniline (0.13 mL, 0.81 mmol, 10 mol%) were added sequentially to a solution of tetraenal (–)-**13** (3.4 g, 8.1 mmol, 1 equiv) in acetonitrile (800 mL). The resulting solution was degassed thoroughly by passage of a stream of argon. The resulting pale yellow solution was partitioned into two 500-mL pressure vessels. The vessels were sealed under an argon atmosphere and heated to 95 °C. After 7 h, the reaction vessels were allowed to cool to 23 °C, and the combined mixture was concentrated under reduced pressure. The residue was purified by flash column chromatography (silica gel, treated with 1% NEt₃ in [32% EtOAc in hexanes], diam. 5 cm, ht. 10 cm; 1% NEt₃ in [32% EtOAc in hexanes]) to afford the desired *trans*-decalin aldehyde (-)-14 (2.1 g, 63%) as a pale yellow oil $([\alpha]^{22}$ _D = -39 (*c* 1.5, CH₂Cl₂)). The minor diastereomer (+)-S8 (420 mg, 13%) was also isolated ($[\alpha]^{22}$ _D = +66 (*c* 0.45, $CH₂Cl₂$)). Structural assignment utilized additional information from gCOSY, HSQC, and HMBC.

Data for the major and desired diastereomer (–)-**14**:

TLC (1% NEt3 in [32% EtOAc in hexanes]), *Rf*: 0.25 (UV, CAM).

TLC (1% NEt3 in [32% EtOAc in hexanes]), *Rf*: 0.40 (UV, CAM).

Tricyclic Enone (–)-15:

A freshly prepared solution of titanium tetrachloride in dichloromethane (1.0 M, 0.36 mL, 0.36 mmol, 2.0 equiv) was added in one portion via syringe to a suspension of *trans*-decalin aldehyde (–)-**14** (75 mg, 0.18 mmol, 1 equiv) and oven-dried 4Å-molecular sieves (100 mg) in dichloromethane (8.9 mL) at −78 °C under an argon atmosphere. After 2 min, saturated aqueous sodium chloride solution (10 mL) was added in one portion via syringe. The resulting mixture was allowed to warm to 23 \degree C, and the layers were separated. The aqueous layer was extracted with dichloromethane $(3 \times 50 \text{ mL})$. The combined organic layers were washed with brine (25 mL) , were dried over anhydrous sodium sulfate, were filtered and were concentrated under reduced pressure to afford the desired crude intramolecular aldol addition product as an oil. The residue was dried by concentration from anhydrous benzene $(2 \times 5 \text{ mL})$ and was directly used in the following dehydration step.

A solution of the Martin sulfurane reagent (133 mg, 0.198 mmol, 1.10 equiv) in anhydrous benzene (3.6 mL) was added via cannula to the crude solution of aldol product in anhydrous benzene (3.6 mL) at 23 °C. After 30 min, the reaction mixture was directly loaded onto and purified via flash column chromatography (silica gel: diam. 1.5 cm, ht. 4 cm; eluent: 75% EtOAc in hexanes) to afford enone (-)-15 (30 mg, 57%) as an oil $([\alpha]^{22}{}_{D} = -18$ (*c* 0.65, CH₂Cl₂)). Structural assignment utilized additional information from gCOSY, and HSQC.

Hydroxycarbamate (–)-20:

A solution of *n*-butyl lithium in hexanes (2.5 M, 0.80 mL, 2.0 mmol, 4.0 equiv) was added dropwise via syringe to a degassed suspension of the iminium chloride $(-)$ - 5^9 (145 mg, 0.980 mmol, 2.00 equiv) in tetrahydrofuran (1.4 mL) at -78 °C under an argon atmosphere. After 15 min, the reaction mixture was allowed to warm to 0 °C. Complete dissolution of the iminium chloride was detected after 15 min at which time the reaction mixture was cooled to -78 °C. The brown solution of the lithioenamine was transferred via cannula under positive argon pressure to a degassed suspension of copper bromide dimethyl sulfide (101 mg, 0.490 mmol, 1 equiv) in tetrahydrofuran (0.7 mL) at $-$ 78 °C. The reaction mixture was allowed to gradually warm to –40 °C over 1 h. The resulting brown reaction mixture was cooled to -78 °C, and a degassed solution of enone $(-)$ -15 (150 mg, 0.520) mmol, 1.05 equiv) in tetrahydrofuran (0.5 mL) was added via cannula. The resulting reaction mixture was allowed to warm to -10 °C over 1.5 h. A solution of degassed thiophenol (0.11 mL, 1.0 mmol, 2.2 equiv) in absolute ethanol (200 proof, 1 mL) was added to the reaction mixture. The resulting mixture was diluted with a degassed aqueous ammonium hydroxide in a saturated aqueous ammonium chloride solution (1:5, 2.4 mL), and the reaction was allowed to warm to 23 °C. After 1.5 h of vigorous stirring, the reaction mixture was diluted with degassed dichloromethane (8 mL), and the layers were separated under an argon atmosphere. The organic layer and the aqueous layer were partitioned, and the aqueous layer was extracted with degassed dichloromethane $(3 \times 8 \text{ mL})$ under an argon atmosphere. The combined organic layers were concentrated under reduced pressure, and the residue was dried by concentration from degassed anhydrous benzene $(2 \times 5 \text{ mL})$ and was directly used in the following reduction step.

Sodium borohydride (59 mg, 1.6 mmol 3.2 equiv) was added as a solid to a degassed solution of the crude pentacylic imine in ethanol (8 mL) at 0 °C under an argon atmosphere. After 30 min, aqueous sodium carbonate solution (1.0 M, 10 mL) was added, and the layers were separated. The aqueous layer was extracted with dichloromethane $(3 \times 50 \text{ mL})$. The combined organic layers were washed with aqueous sodium carbonate solution (1.0 M, 15 mL), and were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The residue was filtered through a plug of silica gel (silica gel, treated with 1% NH₃ in [3% MeOH in CH₂Cl₂], diam. 1.5 cm, ht. 3 cm; eluent: 1% NH₃ in [3\% methanol in dichloromethane]) to afford crude pentacyclic compound (–)-**19** and was directly used in the following step.

Benzyl chloroformate (0.32 mL, 2.2 mmol, 4.5 equiv) was added via syringe to a heterogeneous mixture of crude pentacyclic amine $(-)$ -19, a solution of potassium carbonate (1.3 g, 9.4 mmol, 19 equiv) in water (9.2 mL), and diisopropylethyl amine (1.3 mL, 7.4 mmol, 15 equiv) in tetrahydrofuran (9.2 mL) at 0 °C. The reaction vessel was sealed under an argon atmosphere, and the reaction mixture was allowed to warm to 23 °C. Additional portions of benzyl chloroformate (2 \times 0.32 mL) were added at 0 °C at 30 min intervals. Morpholine (0.58 mL, 6.7 mmol, 14 equiv) was added to quench excess benzyl chloroformate. The reaction mixture was diluted with dichloromethane (50 mL), and the layers were separated. The aqueous layer was extracted with

 ⁹ Movassaghi, M.; Hunt, D. K.; Tjandra, M. *J. Am. Chem. Soc.* **²⁰⁰⁶**, *¹²⁸*, ⁸¹²⁶

dichloromethane $(2 \times 35 \text{ mL})$. The combined organic layers were washed with aqueous sodium carbonate solution (1.0 M, 20 mL), were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The residue was purified by flash column chromatography (silica gel, treated with 1% NEt₃ in [5% acetone in hexanes], diam. 1.5 cm, ht. 4 cm; eluent: 1% NEt₃ in [5% acetone in hexanes] to 1% NEt₃ in [35% acetone in hexanes]) to afford hydroxy carbamate $(-\frac{1}{2})$ $(-132 \text{ mg}, 50\%)$ as a white solid $([\alpha]^{22}$ _D = -22 (*c* 1.0, CH₂Cl₂)). Structural assignment utilized additional information from gCOSY, HSQC, HMBC, and NOESY.

TLC (40% acetone in hexanes), *Rf*: 0.50 (UV, KMnO₄, CAM).

Ketoalcohol (–)-21:

p-Toluenesulfonic acid monohydrate (8.3 mg, 0.043 mmol, 30 mol%) was added as a solid to a solution of hydroxy carbamate (–)-**20** (77 mg, 0.14 mmol, 1 equiv) in benzene (15 mL) at 23 °C, and the reaction mixture was sealed under an argon atmosphere. Because of the sensitivity of the product to acid, aqueous work-up was avoided. After 1.5 h, the reaction mixture was directly loaded onto and purified via flash column chromatography (silica gel, treated with 1% NEt₃ in [35% acetone in hexanes]: diam. 1.5 cm, ht. 4 cm; eluent: 1% NEt₃ in [35% acetone in hexanes]) to afford ketone (– -21 (54.8 mg, 81%) as a white solid ($\left[\alpha\right]^{22}$ _D = -62 (*c* 1.1, CH₂Cl₂)). Structural assignment utilized additional information from gCOSY, HSQC, HMBC, and NOESY.

Vinyl ether (+)-23:

Freshly distilled phosphorus oxychloride (13 µL, 0.14 mmol, 2.0 equiv) was added dropwise via syringe to *N,N*-dimethylformamide (450 μ L, 5.61 mmol, 81.0 equiv) at 0 °C under an argon atmosphere. After 30 min, a solution of ketone (–)-**21** (33.3 mg, 69.0 µmol, 1 equiv) in dichloromethane (1.4 mL) was added dropwise via cannula to the reaction mixture at 0 °C, and the resulting yellow solution was allowed to warm to 23 °C. After 30 min, saturated aqueous sodium bicarbonate solution (4 mL) was added to quench excess acid, and the layers were separated. The aqueous layer was extracted with dichloromethane $(3 \times 10 \text{ mL})$. The combined organic layers were washed with brine (10 mL), were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. Purification of the resulting yellow oil via flash column chromatography (silica gel: diam. 1.5 cm, ht. 3 cm; eluent: 40% EtOAc in hexanes then 50% acetone in hexanes afforded the vinyl ether (+)-23 (24 mg, 71%) as a white film ($\left[\alpha\right]_{\text{D}}^{2} = +19$ (*c* 0.50, CH₂Cl₂)). Structural assignment utilized additional information from gCOSY, HSQC, HMBC, and NOESY.

¹H NMR (600 MHz C_6D_6 20 °C)⁻¹

42.0 (**C**19), 41.1 (**C**10), 40.1 (**C**5), 35.4 (**C**7), 34.5 (**C**8), 32.4 (**C**11), 32.0 (**C**21), 31.3 (**C**13), 29.1 (**C**4), 23.1 (**C**12), 20.8 (C_1H), 18.0 (C_3).

: 2932 (s), 1693 (s, C=O), 1596 (s), 1312 (w), 1120 (m).

FTIR (thin film) cm^{-1} :

HRMS (ESI) calc'd for $C_{30}H_{38}NO_5 [M+H]^+$: 492.2744,

found: 492.2745.

TLC (50% EtOAc in hexanes), *Rf*: 0.49 (UV, CAM).

Ketoester (–)-26:

A solution of vinyl ether (+)-**23** (7.5 mg, 15 µmol, 1 equiv) in mixture of acetonitrile and water (5:1, 150 μ L) was treated sequentially with silica gel (1.7 mg) and 2,3-dichloro-5,6-dicyano-*p*benzoquinone (DDQ, 3.5 mg, 16 µmol, 1.1 equiv) at 23 °C, and the reaction vessel was sealed under an argon atmosphere. After 6 h, the reaction mixture was filtered through a plug of cotton to remove the silica gel and the filtrate was partitioned between water (1 mL) and dichloromethane (8 mL). The aqueous layer was extracted with dichloromethane $(3 \times 5 \text{ mL})$. The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure to afford the crude ketoaldehyde **24**. The ketoaldehyde was directly used in the following oxidation step.

To a solution of the crude ketoaldehyde **24** in *tert*-butanol (380 µL) at 23 °C was added 2 methyl-2-butene (16 µL, 0.15 mmol 10 equiv) and a solution of sodium phosphate monohydrate monobasic (21 mg, 0.15 mmol, 10 equiv) in water (150 µL) followed by a solution of sodium chlorite (14 mg, 0.15 mmol, 10 equiv) in water (150 μ L) via syringe. After 1 h, saturated aqueous sodium thiosulfate solution (1 mL) was added to quench excess oxidant, and the layers were separated. The aqueous layer was extracted with dichloromethane $(3 \times 5 \text{ mL})$. The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure to afford the crude ketoacid that was directly used in the following methylation step.

Freshly prepared diazomethane solution in ether (1.50 mL, 1.60 mmol, 100 equiv) was added to a solution of the crude sample and acetic acid $(31 \mu L, 61 \mu mol, 4.0 \text{ equiv})$ in THF $(100 \mu L)$ at 0 °C. After 30 min, a tetrahydrofuran solution of acetic acid (2 M, 0.5 mL) was added to quench excess diazomethane and the volatiles were removed under reduced pressure. Purification of the resulting yellow oil via flash column chromatography (silica gel: diam. 1.5 cm, ht. 2 cm; eluent: 50% EtOAc in hexanes to 75% EtOAc in hexanes then 50% acetone in hexanes afforded the ketoester (–)-**26** (5 mg, 61%) as a clear film $([\alpha]^{22}$ _D = -64 (*c* 0.42, CH₂Cl₂)). Structural assignment utilized additional information from gCOSY, HSQC, and HMBC.

¹H NMR (500 MHz, C_6D_6 , 20 °C):

7.29 (app-d, $J = 7.2$ Hz, 2H, Ar**H**), 7.16-7.14 (m, 2H, Ar**H**), 7.10-7.05 (m, 1H, Ar**H**), 5.22 (d, *J* = 12.5 Hz, 1H, PhC**H**H'OC=ON), 5.18 (d, *J* = 12.5 Hz, 1H, PhCH**H**'OC=ON), 4.66 (br-s, 1H, C6**H**), 4.38 (br-s, 1H, C2**H**), 3.57 (s, 3H, OC**H**3), 3.55 (s, 3H, COOC**H**3), 3.15 (dt, $J = 5.5$, 10.5 Hz, 1H, C₁₄H), 2.58-2.40 (m, 2H, C_7 **H**H', O**H**), 2.10-1.94 (m, 3H, C₅**H**, C₁₃**H**H', C₁₅**H**), 1.94-1.86 (m, 1H, C4**H**H'), 1.78 (app-d, *J* = 12.0 Hz, 1H, C_9H), 1.72 (br-s, 1H, $C_{21}HH'$), 1.57 (br-s, 1H, C_8H), 1.39-1.20 (m, 4H, C3**H**H', C12**H**H', C11**H**H', C3H**H**'), 1.13-0.87 (m, 5H, $C_{13}HH$, $C_{4}HH$, $C_{7}HH$, $C_{21}HH$ ', C_{10} **H**), 1.02 (d, $J = 7.0$ Hz, 3H, C_1 **H**₃), 0.75-0.64 (m, 1H, $C_{12}HH'$, 0.48-0.38 (m, 1H, $C_{11}HH'$).

Amino Ketoester (–)-S9:

Iodotrimethylsilane $(26 \mu L, 0.18 \text{ mmol}, 14 \text{ equiv})$ was added via syringe to a solution of keto ester (–)-**26** (7.1 mg, 13 mmol, 1 equiv) and 2,6-di-*tert*-butyl-4-methyl-pyridine (670 mg, 3.25 mmol, 250 equiv) in dichloromethane (500 μ L) at 0 °C under an argon atmosphere. Additional portions of iodotrimethylsilane $(8 \times 26 \mu L)$ were added at 1 h intervals until complete consumption of $(-)$ -26 was observed by TLC analysis (~ 8 h). Isopropanol (300 μ L) and aqueous sodium carbonate solution (1 M, 6 mL) were added, and the biphasic reaction mixture was stirred vigorously at 23°C. After 2 h, the organic layer and the aqueous layer were separated. The aqueous layer was extracted with dichloromethane $(4 \times 5 \text{ mL})$. The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. Purification of the resulting yellow oil via flash column chromatography (silica gel, treated with 1% NEt₃ in [35% EtOAc in hexanes], diam. 1.5 cm, ht. 4 cm; eluent: 1% NEt₃ in [5% EtOAc in hexanes] to 1% NEt₃ in [35% EtOAc in hexanes] then 5% methanol in CH_2Cl_2) afforded the pentacyclic amino ketoester $(-)$ -**S9** (4.1 mg, 66%) as a clear film $([\alpha]^{22}$ _D = -4.7 (*c* 0.15, CH₂Cl₂)). Structural assignment utilized additional information from gCOSY, HSQC, and HMBC.

HRMS (ESI) calc'd for $C_{26}H_{42}NO_5Si$ [M+H]⁺: 476.2827, found: 476.2820.

TLC (5% Methanol in CH_2Cl_2), *Rf*: 0.33 (UV, CAM).

Pentacyclic amino alcohol (–)-3:

Triethylamine trihydrogen fluoride (0.114 mL, 0.700 mmol, 82.0 equiv) was added via syringe to a solution of pentacyclic amine (–)-**S9** (4.0 mg, 8.4 µmol, 1 equiv) in tetrahydrofuran (100 µL) at 23°C under an argon atmosphere. After 4 h, aqueous sodium carbonate solution (1 M, 5 mL) was added to quench excess acid. The reaction mixture was diluted with dichloromethane (5 mL), and the layers were separated. The aqueous layer was extracted with dichloromethane $(4 \times 5 \text{ mL})$. The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. Purification of the resulting residue via flash column chromatography (silica gel: diam. 1.5 cm, ht. 3 cm; eluent: 2% to 5% to 10% methanol in CH_2Cl_2) afforded the pentacyclic amino alcohol (-)-3 (3 mg, 90%) as a clear film ($[\alpha]_{\text{D}}^{22} = -24$ (*c* 0.080, CH₂Cl₂). Structural assignment utilized additional information from gCOSY, HSQC, and HMBC.

FTIR (thin film) cm^{-1} : : 2921 (s), 2850 (m), 1734 (m, COOMe), 1671 (m, C=O), 1460 (m), 1261 (w), 1111 (w).

HRMS (ESI) calc'd for $C_{23}H_{34}NO_5 [M+H]^+$: 404.2431, found: 404.2432.

TLC (8% MeOH in CH_2Cl_2), *Rf*: 0.38 (UV, CAM).

(+)-Hexacyclic ketoester (28):

Freshly recrystallized *N*-chlorosuccinimide (1.6 mg, 12 µmol, 2.0 equiv) was added as a solid to a solution of amino alcohol $(-)$ -3 (2.5 mg, 6.2 µmol, 1 equiv) in acetonitrile (0.3 mL) at 23 °C, and the reaction mixture sealed under an argon atmosphere. After 45 min, the reaction solvent was removed under reduced pressure. The residue was immediately purified by flash column chromatography (silica gel: diam. 1.25 cm, ht. 2 cm; eluent: 3% methanol in dichloromethane) to afford hexacyclic ketoester (+)-28 (2.2 mg, 89%) as a white solid ($[\alpha]^{22}$ _D = +20 (*c* 0.11, CH₂Cl₂)). Structural assignment utilized additional information from gCOSY, HSQC, HMBC, ROESY, and NOESY.

(+)-16-Debenzoyl-himandrine (29):

Sodium borohyride (4.0 mg, 0.11 mmol, 10 equiv) was added as a solid to a solution of hexacyclic ketoester (+)-**28** (4.0 mg, 9.9 mmol, 1 equiv) in ethanol (150 µL) at 0 °C, and the reaction mixture was sealed under an argon atmosphere. After 30 min, aqueous sodium carbonate solution (1 M, 0.5 mL) was added, and the layers were separated. The aqueous layer was extracted with dichloromethane $(4 \times 5 \text{ mL})$. The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. Purification of the resulting residue via flash column chromatography (silica gel: diam. 1.25 cm, ht. 1.5 cm; eluent: 5% methanol in CH_2Cl_2) afforded the hexacyclic diol (+)-29 (3.6 mg, 90%) as a single diastereomer ($\left[\alpha \right]^{22}$ _D = +28 (*c* 0.035, CHCl3). Structural assignment utilized additional information from gCOSY, HSQC, and HMBC.

1459 (m), 1281 (m), 1262 (m), 1080 (m).

found: 404.2428.

HRMS (ESI) calc'd for $C_{23}H_{34}NO_5 [M+H]^+$: 404.2431,

TLC (10% Methanol in CH₂Cl₂), *Rf*: 0.19 (UV, CAM).

Comparison of our assignments for (+)-16-debenzoyl-himandrine (29) with literature data:

 ¹⁰ Chemical degradation of himandrine (**1**) gave (+)-16-debenzoyl-himandrine (**29**); see Mander, L. N.; Prager, R. H.; Rasmussen, M.; Ritchie, E.; Taylor, W. C. *Aust. J. Chem.,* **1967**, *20*, 1473.

 11 Our assignment of the C2 methine is supported by our gCOSY, HSQC and HMBC data. The original paper (ref. 10) listed both C2 and C6 methines at 3.30 ppm (br). Our 2D data reveals the C2 methine is actually obscured by the methyl ether signal (3.40 ppm), while C6 methine alone corresponds to the signal at 3.31 ppm (br).

(–)-Himandrine (1):

Freshly distilled benzoyl chloride (0.1 mL) was added to a solution of alcohol (+)-**29** (2.0 mg, 4.9 µmol, 1 equiv) in pyridine (0.12 mL) at 23 °C under an argon atmosphere. After 7 d, the reaction mixture was diluted with dichloromethane (5 mL) and aqueous sodium carbonate solution (1.0 M, 2 mL). After 30 min, the layers were separated, and the aqueous layer was extracted with dichloromethane $(3 \times 10 \text{ mL})$. The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The crude product was purified by flash column chromatography (silica gel: diam. 0.5 cm, ht. 5 cm; eluent: 3% methanol in CH_2Cl_2) to afford (-)-himandrine (1, 2.0 mg, 87%) ($[\alpha]^{22}$ _D = -21 (*c* 0.12, CHCl₃)).¹² Structural assignment utilized additional information from gCOSY, HSQC, HMBC, and ROESY. Crystals suitable for Xray diffraction were obtained from dichloromethane–hexanes (5:1). For a thermal ellipsoid representation of (–)-himandrine (**1**) see page S31.

¹H NMR (600 MHz, CDCl₃, 20 °C):

7.94 (app-d, $J = 7.8$ Hz, 2H, Ar**H**), 7.49 (app-t, $J = 7.2$) Hz, 1H, Ar**H**), 7.39 (app-t, *J* = 7.2 Hz, 2H, Ar**H**), 6.18 $(d, J = 7.8 \text{ Hz}, 1H, C_{16}H), 4.54 \text{ (br-s, 1H, C₂₀OH), 3.58)$ (s, 3H, COOC**H**3), 3.53-3.44 (m, 1H, C2**H**), 3.35 (br-s, 1H, C6**H**), 3.12 (s, 3H, OC**H**3), 3.05 (dt, *J* = 4.2, 10.8 Hz, 1H, C14**H**), 2.44-2.37 (m, 1H, C15**H**), 2.34-2.27 (m, 1H, C4**H**H'), 2.23-2.08 (m, 3H, C8**H**, C13**H**H', C3**H**H'), 2.03 (br-s, 1H, C5**H**), 1.91-1.75 (m, 4H, C7**H**H', C12**H**H', C_4 H**H'**, C_{21} **H**H'), 1.68-1.62 (m, 2H, C_{11} **H**H', C_{21} H**H'**), 1.57 (app-d, *J* = 11.4 Hz, 1H, C7H**H**'), 1.45 (d, *J* = 6.6 Hz, 3H, C1**H**), 1.45-1.42 (m, 2H, C10**H**, C11H**H**'), 1.41- 1.33 (m, 1H, C3H**H'**), 1.33-1.27 (m, 1H, C12H**H**'), 1.13- 1.04 (m, 1H, C13H**H**').

TLC (15% MeOH in CH_2Cl_2), *Rf*: 0.60 (UV, CAM)

¹² The magnitude of the optical rotation of (-)-himandrine (**1**) is sensitive to concentration: $[\alpha]_{D}^{22} = -12$ (*c* 0.060, CHCl₃).

Comparison of our assignments for (–)-himandrine (1) with literature:

¹³ The original structure of himandrine was based on X-ray crystallographic analysis of the corresponding hydrobromide salt of 1; see (a) Guise, G. B.; Mander, L. N.; Prager, R. H.; Rasmussen, M.; Ritchie, E.; Taylor, W. C. *Aust. J. Chem.*, **1967**, *20*, 1029, and (b) Willis, A. C.; O'Connor, P. D.; Taylor, W. C.; Mander, L. N. *Aust. J. Chem*., **2006**, *59*, 629.

¹⁴ We confirmed the structure of our synthetic (-)-himandrine (1) by both X-ray crystallographic analysis and extensive 2D NMR data, ¹⁵ Our assignment of the C2 methine is supported by our gCOSY, HSQC, HMBC, and ROESY data. The isolation paper (ref. 13) listed both C2 and C6 methines at 3.38 ppm (br). Our 2D data reveals the C2 methine is actually at 3.53-3.44 ppm (m), while C6 methine alone corresponds to the signal at 3.35 ppm (br-s).

View 1:

Table S1. Crystal data and structure refinement for (–)-himandrine (**1**).

C(26)	14848(2)	$-4000(2)$	$-5428(2)$	24(1)	
C(27)	16546(2)	$-5528(2)$	$-5694(2)$	26(1)	
C(28)	16881(2)	$-6760(2)$	$-4668(2)$	25(1)	
C(29)	15491(2)	$-6478(2)$	$-3377(2)$	25(1)	
C(30)	13799(2)	$-4961(2)$	$-3103(2)$	22(1)	

Table S3. Bond lengths [Å] and angles [°] for (–)-himandrine (**1**).

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}	
O(1)	27(1)	25(1)	22(1)	$-4(1)$	$-8(1)$	$-12(1)$	
O(2)	26(1)	19(1)	30(1)	$-4(1)$	$-7(1)$	$-10(1)$	
O(3)	18(1)	17(1)	21(1)	$-2(1)$	$-6(1)$	$-7(1)$	
O(4)	24(1)	27(1)	27(1)	$-3(1)$	$-9(1)$	$-13(1)$	
O(5)	28(1)	24(1)	22(1)	$-1(1)$	$-5(1)$	$-16(1)$	
O(6)	28(1)	21(1)	22(1)	1(1)	$-6(1)$	$-9(1)$	
N(1)	21(1)	20(1)	20(1)	$-1(1)$	$-6(1)$	$-9(1)$	
C(1)	23(1)	20(1)	26(1)	1(1)	$-11(1)$	$-7(1)$	
C(2)	21(1)	20(1)	20(1)	1(1)	$-7(1)$	$-8(1)$	
C(3)	22(1)	22(1)	23(1)	1(1)	$-9(1)$	$-10(1)$	
C(4)	28(1)	23(1)	26(1)	$-1(1)$	$-11(1)$	$-11(1)$	
C(5)	25(1)	21(1)	22(1)	$-6(1)$	$-6(1)$	$-8(1)$	
C(6)	23(1)	22(1)	18(1)	$-2(1)$	$-3(1)$	$-9(1)$	
C(7)	22(1)	23(1)	26(1)	$-4(1)$	$-4(1)$	$-8(1)$	
C(8)	20(1)	20(1)	26(1)	$-1(1)$	$-7(1)$	$-6(1)$	
C(9)	18(1)	17(1)	22(1)	0(1)	$-6(1)$	$-7(1)$	
C(10)	20(1)	18(1)	23(1)	0(1)	$-8(1)$	$-8(1)$	
C(11)	20(1)	23(1)	30(1)	0(1)	$-8(1)$	$-10(1)$	
C(12)	25(1)	22(1)	37(1)	$-1(1)$	$-13(1)$	$-12(1)$	
C(13)	27(1)	22(1)	32(1)	$-3(1)$	$-12(1)$	$-12(1)$	
C(14)	25(1)	17(1)	24(1)	$-1(1)$	$-12(1)$	$-8(1)$	
C(15)	20(1)	16(1)	22(1)	0(1)	$-10(1)$	$-8(1)$	
C(16)	19(1)	16(1)	20(1)	0(1)	$-7(1)$	$-6(1)$	
C(17)	19(1)	16(1)	22(1)	0(1)	$-9(1)$	$-6(1)$	
C(18)	19(1)	13(1)	25(1)	$-3(1)$	$-7(1)$	$-4(1)$	
C(19)	22(1)	14(1)	22(1)	1(1)	$-12(1)$	$-6(1)$	
C(20)	21(1)	16(1)	24(1)	$-4(1)$	$-5(1)$	$-8(1)$	
C(21)	22(1)	18(1)	26(1)	$-3(1)$	$-7(1)$	$-5(1)$	
C(22)	26(1)	28(1)	30(1)	1(1)	$-3(1)$	$-16(1)$	
C(23)	43(1)	38(1)	24(1)	$-5(1)$	$-13(1)$	$-17(1)$	
C(24)	23(1)	20(1)	20(1)	$-1(1)$	$-7(1)$	$-13(1)$	
C(25)	22(1)	20(1)	24(1)	$-3(1)$	$-8(1)$	$-12(1)$	

Table S4. Anisotropic displacement parameters $(\hat{A}^2 \times 10^3)$ for (-)-himandrine (1). The anisotropic displacement factor exponent takes the form: $-2p^2$ [h² a^{*2}U¹¹ + ... + 2 h k a^{*} b^{*} U¹²]
C(26)	28(1)	22(1)	24(1)	0(1)	$-7(1)$	$-14(1)$
C(27)	25(1)	26(1)	25(1)	$-6(1)$	$-2(1)$	$-13(1)$
C(28)	24(1)	19(1)	33(1)	$-8(1)$	$-9(1)$	$-7(1)$
C(29)	29(1)	21(1)	28(1)	1(1)	$-14(1)$	$-12(1)$
C(30)	23(1)	23(1)	23(1)	$-4(1)$	$-6(1)$	$-12(1)$

Table S5. Hydrogen coordinates (x 10⁴) and isotropic displacement parameters (\AA ²x 10³) for (–)-himandrine (**1**).

Table S6. Hydrogen bonds for (–)-himandrine (**1**) [Å and °].

Symmetry transformations used to generate equivalent atoms:

2 22.017 MM 0.9094 1.19103e4 218.27008 99.4568 Totals :

1.19753e4 220.34498

Results obtained with enhanced integrator! Summed Peaks Report

Signal 1: MWD1 E, Sig=215,16 Ref=360,100

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