Asymmetric Total Synthesis of (-)-Plicatic Acid via a Highly Enantioselective and Diastereoselective Nucleophilic Epoxidation of Acyclic Trisubstitued Olefins

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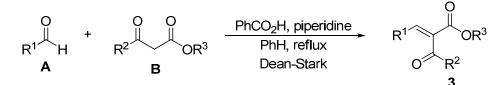
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General Information. ¹H NMR spectra were recorded on a Varian Inova 400 (400 MHz) or Varian Inova 500 (500 MHz) spectrometers. Chemical shifts are reported in ppm down field from TMS, using TMS (0.00 ppm) or residual CDCl₃ (7.26 ppm) as an internal standard. Data are reported as: (b = broad, s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet; coupling constant in Hz, integration). ¹³C NMR spectra were recorded on a Varian Inova 400 (100 MHz) or Varian Inova 500 (125 MHz) spectrometers, using proton decoupling unless otherwise noted. Chemical shifts are reported in ppm down field from TMS, using the central resonance of CDCl₃ (77.00 ppm) or CD₃OD (49.00 ppm) as the internal standard. Infrared spectra were recorded on a Perkin Elmer FT-IR spectrometer and are reported in frequency of absorption. Mass spectrometry was performed by Mass Spectrometry Laboratory, University of Illinois at Urbana-Champaign. Optical rotations were measured on a Jasco Digital Polarimeter with a sodium lamp. CD spectra were recorded on Jasco J-810 circular dichroism spectrometer.

Materials. Unless otherwise noted, reagents and solvents were commercially available and used without further purification. Anhydrous DMF was obtained by distillation from calcium hydride. Anhydrous THF was obtained by distillation from sodium metal (benzophenone indicator).

General Procedure for the Preparation of 3¹



The mixture of aldehyde **A** (11.0 mmol equiv.), 1,3-dicarbonyl compound **B** (10.0 mmol), benzoic acid (1.0 mmol), and piperidine (1.0 mmol) in benzene (50.0 mL) was heated at reflux in a round bottom flask connected to a Dean-Stark apparatus until the starting materials were consumed as indicated by TLC analysis. After being cooled down to room temperature, the reaction mixture was concentrated under reduced pressure. The residue was purified by silica gel chromatography to afford (*E*)-**3** in 40-60% yield. The stereochemistry of **3c-g** was assigned based on the ${}^{3}J_{C-H}$ coupling constants of C_{ketone}-H and C_{ester}-H.²

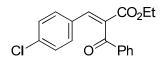
(E)-Ethyl 2-benzoyl-3-phenylacrylate (3b)³

¹ Allen, C. F. H.; Spangler, F. W. Org. Syn. 1955, Coll. Vol. 3, 377.

² Kingsbury, C. A.; Draney, D.; Sopchik, A.; Rissler, W.; Durham, D. J. Org. Chem. 1978, 41, 3863.

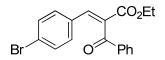
³ Li, Z.; Li, H.; Guo, X.; Cao, L.; Yu, R.; Li, H.; Pan, S. Org. Lett. 2008, 10, 803.

¹**H** NMR (400 MHz, CDCl₃) δ 7.93 (d, J = 6.8 Hz, 2H), 7.56-7.20 (m, 9H), 4.20 (q, J = 7.2 Hz, 2H), 1.15 (t, J = 7.2 Hz, 3H); ¹³**C** NMR (100 MHz, CDCl₃) δ 195.66, 164.99, 142.60, 136.09, 133.87, 132.78, 131.24, 130.35, 130.17, 129.12, 128.80, 128.76, 61.54, 14.00; **IR** (thin film) 3061, 2980, 1720, 1702, 1674, 1624, 1597, 1449, 1253, 1231, 1198, 1092 cm⁻¹; **LRMS (ESI)**: 303 (M + Na⁺); **HRMS (ESI)**: Exact mass calcd for C₁₈H₁₆O₃Na (M + Na), 303.0997. Found 303.0997.



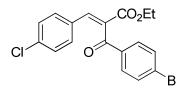
(*E*)-Ethyl 2-benzoyl-3-(4-chlorophenyl)acrylate (3c)

¹**H NMR** (400 MHz, CDCl₃) δ 7.94-7.90 (m, 2H), 7.59-7.20 (m, 7H), 4.22 (q, J = 7.2 Hz, 2H), 1.17 (t, J = 7.2 Hz, 3H); ¹³**C NMR** (100 MHz, CDCl₃) δ 195.38, 164.76, 141.04, 136.44, 135.90, 134.09, 131.84, 131.30, 129.09, 128.90, 61.68, 14.00; **IR** (thin film) 3061, 2981, 1720, 1673, 1624, 1592, 1492, 1252, 1093, 1013 cm⁻¹; **LRMS** (**ESI**): 315 (M + H⁺); **HRMS** (**ESI**): Exact mass calcd for C₁₈H₁₆ClO₃ (M + H), 315.0788. Found 315.0794. ³*J*_{C-H} = 9.9 Hz for C_{ketone}-H, ³*J*_{C-H} = 7.6 Hz for C_{ester}-H.



(E)-Ethyl 2-benzoyl-3-(4-bromophenyl)acrylate (3d)

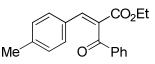
¹**H NMR** (400 MHz, CDCl₃) δ 7.93 (d, J = 7.2 Hz, 2H), 7.88 (s, 1H), 7.58-7.41 (m, 3H), 7.36 (d, J = 7.2 Hz, 2H), 7.21 (d, J = 7.6 Hz, 2H), 4.22 (q, J = 7.2 Hz, 2H), 1.16 (t, J = 7.2 Hz, 3H); ¹³**C NMR** (100 MHz, CDCl₃) δ 195.24, 164.65, 140.99, 135.78, 134.03, 131.95, 131.92, 131.62, 131.38, 128.99, 128.82, 124.81, 61.59, 13.91; **IR** (thin film) 2978, 2366, 2346, 1720, 1707, 1702, 1672, 1250, 1230, 1196, 1073, 1010, 818 cm⁻¹; **LRMS** (**ESI**): 359, 361 (M + H⁺); **HRMS** (**ESI**): Exact mass calcd for C₁₈H₁₆BrO₃ (M + H), 359.0283. Found 359.0280. ³ $J_{C-H} = 9.9$ Hz for C_{ketone}-H, ³ $J_{C-H} = 7.6$ Hz for C_{ester}-H.



(E)-Ethyl 2-(4-bromobenzoyl)-3-(4-chlorophenyl)acrylate (3e)

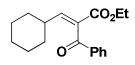
¹**H** NMR (400 MHz, CDCl₃) δ 7.91 (s, 1H), 7.80 (d, J = 8.0 Hz, 2H), 7.57 (d, J = 8.4 Hz, 2H), 7.27 (d, J = 8.4 Hz, 2H), 7.21 (d, J = 8.4 Hz, 2H), 4.23 (q, J = 7.2 Hz, 2H), 1.18 (t, J = 7.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 194.17, 164.34, 141.21, 136.48, 134.51, 132.15, 131.14, 131.11, 130.94, 130.97, 129.36,

129.02, 61.63, 13.90; **IR** (thin film) 2980, 2361, 2345, 1720, 1702, 1672, 1584, 1252, 1195, 1011 cm⁻¹; **LRMS (ESI)**: 393 (M + H⁺); **HRMS (ESI)**: Exact mass calcd for $C_{18}H_{15}ClBrO_3$ (M + H), 392.9893. Found 392.9898. ${}^{3}J_{C-H} = 9.9$ Hz for C_{ketone} -H, ${}^{3}J_{C-H} = 7.7$ Hz for C_{ester} -H.



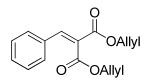
(E)-Ethyl 2-benzoyl-3-p-tolylacrylate (3f)

¹**H NMR** (400 MHz, CDCl₃) δ 7.94 (d, J = 8.0 Hz, 2H), 7.54 (t, J = 7.2 Hz, 1H), 7.41 (t, J = 7.6 Hz, 2H), 7.23 (d, J = 8.0 Hz, 2H), 7.02 (d, J = 8.0 Hz, 2H), 4.19 (q, J = 7.2 Hz, 2H), 2.25 (s, 3H), 1.15 (t, J = 7.2 Hz, 3H); ¹³**C NMR** (100 MHz, CDCl₃) δ 196.16, 165.36, 142.82, 141.19, 136.42, 134.06, 130.53, 130.35, 130.24, 129.78, 129.35, 129.04, 61.64, 21.60, 14.26; **IR** (thin film) 2961, 2930, 2870, 2362, 2345, 1719, 1702, 1676, 1253, 1230, 1200, 1183, 1085 cm⁻¹; **LRMS** (**ESI**): 295 (M + H⁺); **HRMS** (**ESI**): Exact mass calcd for C₁₉H₁₉O₃ (M + H), 295.1334. Found 295.1324. ³ $J_{C-H} = 9.8$ Hz for C_{ketone}-H, ³ $J_{C-H} = 7.6$ Hz for C_{ester}-H.



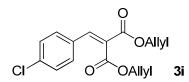
(E)-Ethyl 2-benzoyl-3-cyclohexylacrylate (3g)

¹**H NMR** (400 MHz, CDCl₃) δ 7.90 (d, J = 7.6 Hz, 2H), 7.61-7.44 (m, 3H), 6.90 (d, J = 10.8 Hz, 1H), 4.14 (q, J = 7.2 Hz, 2H), 2.12 (m, 1H), 1.67-1.01 (m, 13H); ¹³**C NMR** (100 MHz, CDCl₃) δ 194.49, 164.79, 152.80, 136.91, 133.54, 131.72, 128.98, 128.59, 61.03, 38.55, 31.60 (2), 25.47, 24.90 (2), 13.90; **IR** (thin film) 2929, 2853, 1724, 1675, 1639, 1597, 1449, 1252, 1233, 1221 cm⁻¹; **LRMS (ESI)**: 287 (M + H⁺, 100); **HRMS (ESI)**: Exact mass calcd for C₁₈H₂₃O₃ (M +H), 287.1647. Found 287.1635. ³*J*_{C-H} = 9.8 Hz for C_{ketone}-H, ³*J*_{C-H} = 7.6 Hz for C_{ester}-H.



Diallyl 2-benzylidenemalonate (3h)

¹**H NMR** (400 MHz, CDCl₃) δ 7.79 (s, 1H), 7.47-7.35 (m, 5H), 6.01-5.86 (m, 2H), 5.39-5.24 (m, 4H), 4.78-4.74 (m, 4H); ¹³**C NMR** (100 MHz, CDCl₃) δ 166.21, 163.28, 142.96, 132.69, 131.55, 131.15, 130.69, 129.53, 128.80, 125.60, 119.33, 118.43, 66.28, 66.04; **IR** (thin film) 2980, 2962, 2363, 1730, 1629, 1257, 1198 cm⁻¹; **LRMS** (**ESI**): 295 (M + Na⁺, 100); **HRMS** (**ESI**): Exact mass calcd for $C_{16}H_{16}O_4Na (M + Na)$, 295.0946. Found 295.0944.



Diallyl 2-(4-chlorobenzylidene)malonate (3i)

¹**H NMR** (400 MHz, CDCl₃) δ 7.72 (s, 1H), 7.41-7.34 (m, 4H), 6.00-5.86 (m, 2H), 5.39-5.26 (m, 4H), 4.78-4.75 (m, 4H); ¹³**C NMR** (100 MHz, CDCl₃) δ 165.91, 163.43, 141.41, 136.76, 131.42, 131.12, 131.00, 130.73, 129.10, 126.10, 119.55, 118.53, 66.36, 66.12; **IR** (thin film) 3086, 3021, 2947, 1734, 1648, 1630, 1591, 1492, 1385, 1254, 1197, 1093, 1060 cm⁻¹; **LRMS** (**ESI**): 329 (M + Na⁺); **HRMS** (**ESI**): Exact mass calcd for $C_{16}H_{15}ClO_4Na$ (M + Na), 329.0557. Found 329.0550.

General Procedure for Base Screening

The base (0.010 mmol) was added to a stirring mixture of (+)-(*S*, *S*)-TADOOH (0.12 mmol) and **3b** (0.10 mmol) in THF (0.50 mL) via a syringe at 0 °C under air. After 36 hours, the reaction mixture was allowed to pass rapidly through a plug (20 mm) of silica gel. The silica gel plug was washed with ethyl acetate (10 mL). The filtrate was concentrated under reduced pressure. The residue was subjected to ¹H NMR and HPLC analysis to determine the conversion. The ee of **4b** was determined by HPLC analysis [Chiralcel AD-H column, Hexanes/*i*-PrOH = 98/2, 1.0 mL/min, 254 nm, 15 °C, $t_R(1) \approx 22 \text{ min}$, $t_R(2) \approx 33 \text{ min}$]. The results were listed in Table 1.

Table	e 1.	Base	Scre	ening
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	CO ₂ Et O Ph 3b	+)-TADOOH → 0°C, 36 hr	O CO ₂ Et	
Entry 1	Base (10 mol %) LiOH (5.0 M, aq.)	conv. (%, HPLC) 99	ee (%) 92	Major peak <i>t</i> _R (2)
2	Li ₂ CO ₃ (5.0 M, aq.)	trace	/	/
3	NaOH (5.0 M, aq.)	12	44	$t_{\rm R}(1)$
4	Na ₂ CO ₃ (5.0 M, aq.)	trace	/	/
5	KOH (5.0 M, aq.) (indissolvable)	trace	/	/
6	K ₂ CO ₃ (5.0 M, aq.)	21	73	<i>t</i> _R (2)
7	CsOH (5.0 M, aq.)	94	24	<i>t</i> _R (1)
8	Cs ₂ CO ₃ (5.0 M, aq.)	92	30	<i>t</i> _R (1)
9	DBU	35	42	<i>t</i> _R (2)
10	DABCO	4	/	/
11	Et ₃ N	4	/	/

General Procedure for Solvent Screening

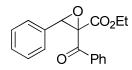
An aqueous solution of LiOH (0.010 mmol, 2.0 μ L, 5.0 M) was added via a syringe at 0 °C to a stirring solution of (+)-TADOOH (0.12 mmol) and **3b** (0.10 mmol) in a solvent (0.50 mL) specified in Table 2 under air. The resulting reaction mixture was stirred for 36 hr, after which the reaction mixture was allowed to pass rapidly through a plug (20 mm) of silica gel. The silica gel was washed with ethyl acetate (10 mL). The filtrate was concentrated under reduced pressure. The residue was subjected to ¹H NMR and HPLC analysis to determine the conversion. The ee of **4b** was determined by HPLC analysis [Chiralcel AD-H column, Hexanes/*i*-PrOH = 98/2, 1.0 mL/min, 254 nm, 15 °C, $t_R(1) \approx 22 \text{ min}$, $t_R(2) \approx 33 \text{ min}$]. The results were listed in Table 2.

Tab	le 2.	Solvent	Screening
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	O Ph 3b	- (+)-TADOOH 0°C, 36 hr	O CO ₂ Et O Ph 4b	
Entry	Solvent (0.20 M)	conv. (%, HPLC)	ee (%)	Major peak
1	THF	99	92	<i>t</i> _R (2)
2	Et ₂ O	77	80	<i>t</i> _R (2)
3	DCM	12	45	<i>t</i> _R (1)
4	Toluene (slurry)	35	39	<i>t</i> _R (2)
5	EtOH	4	/	<i>t</i> _R (1)
6	EtOAc	50	93	<i>t</i> _R (2)
7	DMF (slurry)	43	32	<i>t</i> _R (2)

General Procedure for the Asymmetric Epoxidation of 3

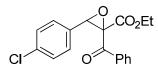
An aqueous solution of LiOH (0.010 mmol, 2.0 μ L, 5.0 M) was added via a syringe to a solution of (+)-TADOOH (0.12 mmol) and **3** (0.10 mmol) in THF (0.50 mL) at 0 °C (or indicated temperature), under air. The reaction mixture was stirred vigorously at this temperature for indicated time. After the reaction was complete as indicated by TLC analysis, the reaction mixture was allowed to pass rapidly through a plug (20 mm) of silica gel. The silica gel was washed with ethyl acetate (10 mL). The filtrate was concentrated under reduced pressure and the residue was purified by flash chromatography or preparative TLC to give pure epoxide **4**.



(-)-Ethyl 2-benzoyl-3-phenyloxirane-2-carboxylate (4b)

0 °C, 1.5 days, 98% yield, 92% ee.

¹**H** NMR (400 MHz, CDCl₃) δ 7.85 (d, J = 8.0 Hz, 2H), 7.52 (t, J = 7.2 Hz, 1H), 7.38 (t, J = 8.0 Hz, 2H), 7.19 (s, 5H), 4.73 (s, 1H), 4.24 (q, J = 6.8 Hz, 2H), 1.17 (t, J = 6.8 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 189.81, 166.40, 135.15, 133.81, 132.04, 128.87, 128.70, 128.45, 128.27, 126.21, 65.80, 62.97, 62.74, 13.73; **IR** (thin film) 2925, 2853, 2360, 2342, 1748, 1730, 1693, 1450, 1256, 1235, 696 cm⁻¹; **LRMS** (**ESI**): 319 (M + Na⁺); **HRMS (ESI**): Exact mass calcd for C₁₈H₁₆O₄Na (M + Na), 319.0946. Found 319.0933; $[\alpha]_D^{25} = -15.3^\circ$ (*c* 1.0, CHCl₃, 92% ee), 92% ee was determined by HPLC analysis [Chiralcel OD-H column, Hexanes/*i*-PrOH = 96/4, 1.0 mL/min, 220 nm, 20 °C, t_R (1) = 11.3 min (major), t_R (2) = 20.6 min (minor)].



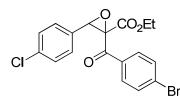
(+)-Ethyl 2-benzoyl-3-(4-chlorophenyl)oxirane-2-carboxylate (4c)

0 °C, 2 days, 99% yield, 94% ee.

¹**H NMR** (400 MHz, CDCl₃) δ 7.85 (d, J = 7.2 Hz, 2H), 7.56 (t, J = 7.2 Hz, 1H), 7.42 (t, J = 7.6 Hz, 2H), 7.19 (d, J = 8.8 Hz, 2H), 7.14 (d, J = 8.8 Hz, 2H), 4.70 (s, 1H), 4.26 (q, J = 6.8 Hz, 2H), 1.18 (t, J = 6.8 Hz, 3H); ¹³**C NMR** (100 MHz, CDCl₃) δ 189.62, 166.24, 135.05, 134.91, 134.11, 130.73, 128.79, 128.67, 128.65, 127.66, 65.80, 62.96, 62.36, 13.80; **IR** (thin film) 2360, 2334, 1748, 1733, 1691, 1256, 1234 cm⁻¹; **LRMS (ESI)**: 353 (M + Na⁺, 100); **HRMS (ESI)**: Exact mass calcd for C₁₈H₁₅ClO₄Na (M + Na), 353.0557. Found 353.0554; $[\alpha]_D^{25} = +0.8^\circ$ (*c* 1.0, CHCl₃, 94% ee), 94% ee was determined by HPLC analysis [Chiralcel OD-H column, Hexanes/*i*-PrOH = 90/10, 1.0 mL/min, 220 nm, 20 °C, *t*_R (1) = 9.5 min (major), *t*_R (2) = 15.7 min (minor)].

(+)-Ethyl 2-benzoyl-3-(4-bromophenyl)oxirane-2-carboxylate (4d) 0 °C, 2 days, 83% yield, 93% ee.

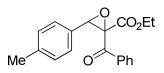
¹**H NMR** (400 MHz, CDCl₃) δ 7.85 (d, *J* = 8.0 Hz, 2H), 7.56 (t, *J* = 7.6 Hz, 1H), 7.42 (t, *J* = 7.6 Hz, 2H), 7.34 (d, *J* = 8.0 Hz, 2H), 7.07 (d, *J* = 7.6 Hz, 2H), 4.68 (s, 1H), 4.26 (q, *J* = 7.6 Hz, 2H), 1.18 (t, *J* = 7.2 Hz, 3H); ¹³**C NMR** (100 MHz, CDCl₃) δ 189.61, 166.21, 135.02, 134.13, 131.59, 131.23, 128.78, 128.65, 127.92, 123.16, 65.75, 62.97, 62.40, 13.80; **IR** (thin film) 2982, 1748, 1734, 1690, 1596, 1490, 1256, 1233 cm⁻¹; **LRMS** (**ESI**): 397, 399 (M + Na⁺); **HRMS** (**ESI**): Exact mass calcd for C₁₈H₁₅BrO₄Na (M + Na), 397.0051. Found 397.0056; $[\alpha]_D^{25} = +5.4^\circ$ (*c* 1.0, CHCl₃, 93% ee), 93% ee was determined by HPLC analysis [Chiralcel OD-H column, Hexanes/*i*-PrOH = 90/10, 0.8 mL/min, 220 nm, 20 °C, *t*_R (1) = 11.2 min (major), *t*_R (2) = 18.6 min (minor)].



(+)-Ethyl 2-(4-bromobenzoyl)-3-(4-chlorophenyl)oxirane-2-carboxylate (4e)

0 °C, 2 days, 99% yield, 93% ee.

¹**H** NMR (400 MHz, CDCl₃) δ 7.72 (d, J = 8.4 Hz, 2H), 7.57 (t, J = 8.4 Hz, 1H), 7.21 (t, J = 8.4 Hz, 2H), 7.11 (d, J = 8.4 Hz, 2H), 4.68 (s, 1H), 4.27 (q, J = 7.2 Hz, 2H), 1.21 (t, J = 7.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 188.79, 166.00, 135.09, 133.77, 132.09, 130.51, 130.23, 129.61, 128.79, 127.56, 65.62, 63.12, 62.37, 13.87; **IR** (thin film) 2984, 2933, 2360, 2341, 1749, 1734, 1691, 1584, 1494, 1398, 1257, 1236, 1012, 832 cm⁻¹; **LRMS (ESI)**: 431, 433 (M + Na⁺); **HRMS (ESI)**: Exact mass calcd for C₁₈H₁₄ClBrO₄Na (M + Na), 430.9662. Found 430.9670; $[\alpha]_D^{25} = +25.0^\circ$ (*c* 1.0, CHCl₃, 93% ee), 93% ee was determined by HPLC analysis [Chiralcel OD-H column, Hexanes/*i*-PrOH = 96/4, 0.8 mL/min, 220 nm, 20 °C, *t*_R (1) = 19.2 min (major), *t*_R (2) = 42.7 min (minor)].



(+)-Ethyl 2-benzoyl-3-p-tolyloxirane-2-carboxylate (4f)

0 °C, 2 days, 60% yield, 96% ee.

¹**H NMR** (400 MHz, CDCl₃) δ 7.86 (d, J = 7.6 Hz, 2H), 7.53 (t, J = 7.6 Hz, 1H), 7.40 (t, J = 7.6 Hz, 2H), 7.08 (d, J = 8.0 Hz, 2H), 7.00 (d, J = 8.0 Hz, 2H), 4.69 (s, 1H), 4.24 (q, J = 7.6 Hz, 2H), 2.22 (s, 3H), 1.17 (t, J = 7.6 Hz, 3H); ¹³**C NMR** (100 MHz, CDCl₃) δ 189.99, 166.51, 138.80, 135.26, 133.80, 129.04, 129.02, 128.76, 128.48, 126.18, 65.86, 63.16, 62.72, 21.10, 13.76; **IR** (thin film) 2983, 1746, 1730, 1601, 1450, 1256, 1233, 1177 cm⁻¹; **LRMS (ESI)**: 311 (M + H⁺); **HRMS (ESI)**: Exact mass calcd for C₁₉H₁₉O₄

(M + H), 311.1283. Found 311.1278; $[\alpha]_D^{25} = +0.7^\circ$ (*c* 1.0, CHCl₃, 96% ee), 96% ee was determined by HPLC analysis [Chiralcel OD-H column, Hexanes/*i*-PrOH = 90/10, 0.8 mL/min, 220 nm, 20 °C, t_R (1) = 10.0 min (major), t_R (2) = 12.9 min (minor)].

(+)-Ethyl 2-benzoyl-3-cyclohexyloxirane-2-carboxylate (4g)

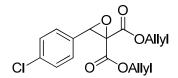
-40 °C, 3 days, 86% yield, 86% ee.

¹**H** NMR (400 MHz, CDCl₃) δ 8.04 (d, J = 8.0 Hz, 2H), 7.61 (t, J = 7.6 Hz, 1H), 7.49 (t, J = 8.0 Hz, 2H), 4.22 (q, J = 7.2 Hz, 2H), 3.34 (d, J = 8.8 Hz, 1H), 1.87 (d, J = 7.6 Hz, 2H), 1.69-0.97 (m, 12H); ¹³**C** NMR (100 MHz, CDCl₃) δ 190.94, 167.50, 135.06, 134.02, 129.23, 128.62, 67.06, 63.69, 62.55, 38.04, 30.16, 28.00, 25.93, 25.03, 24.97, 13.81; **IR** (thin film) 2976, 2931, 2854, 2360, 2338, 1749, 1730, 1691, 1450, 1285, 1256, 1233, 698 cm⁻¹; **LRMS (ESI)**: 303 (M + H⁺); **HRMS (ESI)**: Exact mass calcd for C₁₈H₂₃O₄ (M + H), 303.1596. Found 303.1585; $[\alpha]_D^{25} = +7.4^\circ$ (*c* 1.0, CHCl₃, 86% ee), 86% ee was determined by HPLC analysis [Chiralcel As-H + Chialpak AD + Chiralcel AD-H column, Hexanes/*i*-PrOH = 99/1, 0.8 mL/min, 254 nm, 20 °C, *t*_R (1) = 50.2 min (major), *t*_R (2) = 54.65 min (minor)].

(-)-Diallyl 3-phenyloxirane-2,2-dicarboxylate (4h)

0°C, 3 days, 90% yield, 92% ee.

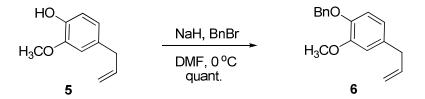
¹**H** NMR (400 MHz, CDCl₃) δ 7.33 (s, 5H), 5.98-5.88 (m, 1H), 5.60-5.50 (m, 1H), 5.38 (dd, J = 16.8, 1.2 Hz, 1H), 5.30 (d, J = 10.4 Hz, 1H), 5.08 (dd, J = 14.8, 2.8 Hz, 2H), 4.76-4.74 (m, 2H), 4.59 (s, 1H), 4.45 (d, J = 5.6 Hz, 2H); ¹³**C** NMR (100 MHz, CDCl₃) δ 165.04, 163.13, 131.83, 130.69, 129.11, 128.38, 126.18, 119.41, 119.05, 67.10, 66.24, 63.12, 62.21; **IR** (thin film) 3028, 2918, 1753, 1750, 1650, 1455, 1264, 1227, 1122, 1045, 936, 754 cm⁻¹; **LRMS (ESI)**: 311 (M + Na⁺); **HRMS (ESI)**: Exact mass calcd for C₁₆H₁₆O₅Na (M + Na), 311.0895. Found 311.0894; $[\alpha]_D^{25} = -57.2^\circ$ (*c* 1.0, CHCl₃, 92% ee), 92% ee was determined by HPLC analysis [Chiralcel OD-H column, Hexanes/*i*-PrOH = 96/4, 1 mL/min, 254 nm, 20 °C, *t*_R (1) = 11.2 min (minor), *t*_R (2) = 12.3 min (major)].



(-)-Diallyl 3-(4-chlorophenyl)oxirane-2,2-dicarboxylate (4i)

-20 °C, 3 days, 70% yield, 84% ee.

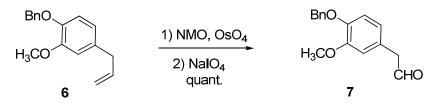
¹**H NMR** (400 MHz, CDCl₃) δ 7.32 (d, J = 8.8 Hz, 2H), 7.28 (d, J = 8.4 Hz, 2H), 5.98-5.88 (m, 1H), 5.65-5.55 (m, 1H), 5.39 (d, J = 17.2 Hz, 1H), 5.30 (d, J = 10.4 Hz, 1H), 5.13 (d, J = 11.6 Hz, 2H), 4.80-4.71 (m, 2H); 4.56 (s, 1H), 4.48 (d, J = 6.0 Hz, 2H); ¹³**C NMR** (100 MHz, CDCl₃) δ 164.78, 162.92, 135.16, 130.63, 130.57, 130.40, 128.67, 127.65, 119.53, 119.35, 67.19, 66.39, 63.03, 61.54; **IR** (thin film) 2362, 2338, 1751, 1270, 1225 cm⁻¹; **LRMS** (**ESI**): 345 (M + Na)⁺; **HRMS** (**ESI**): Exact mass calcd for C₁₆H₁₅ClO₅Na (M + Na), 345.0506. Found 345.0497; $[\alpha]_D^{25} = -62.7^\circ$ (*c* 1.0, CHCl₃, 84% ee.), 84% ee was determined by HPLC analysis [Chiralcel OD-H column, Hexanes/*i*-PrOH = 99/1, 0.8 mL/min, 254 nm, 20 °C, *t*_R (1) = 13.5 min (major), *t*_R (2) = 16.8 min (minor).



To a suspension of NaH (55% assay, 1.05 g, 24.0 mmol) in anhydrous DMF (20.0 mL) at 0 °C was added dropwise a solution of eugenol (**5**) (3.28 g, 20.0 mmol) in anhydrous DMF (10.0 mL) via syringe over 15 min. The resulting deep green mixture was stirred at 0 °C for 15 min, and then to this solution was added dropwise a solution of benzylbromide (2.85 mL, 24.0 mmol) in anhydrous DMF (10.0 mL) via syringe over 2 min. The resulting mixture was stirred at 0 °C for 1 hr. Then the reaction was quenched by a slow addition of saturated aqueous NH₄Cl (50 mL) and then ethyl acetate (50 mL). The organic layer was collected, and the aqueous layer was extracted with ethyl acetate (3×50 mL). The combined extracts were washed with water (4×50 mL), brine (40 mL), dried over Na₂SO₄ and concentrated under reduced pressure. The residue was subjected to silica gel chromatography (petroleum ether/EtOAc = 8/1) to give **6** (5.07 g, quant.) as a coloress oil.

¹**H NMR** (400MHz, CDCl₃) δ 7.43 (d, *J* = 7.2 Hz, 2H), 7.35 (t, *J* = 7.2 Hz, 2H), 7.28 (t, *J* = 7.2 Hz, 1H), 6.80 (d, *J* = 8.0 Hz, 1H), 6.72 (s, 1H), 6.65 (d, *J* = 8.0 Hz, 1H), 5.93 (m, 1H), 5.12 (s, 2H), 5.07 (d, *J* = 11.6 Hz, 1H), 5.04 (d, *J* = 2.8 Hz, 1H), 3.86 (s, 3H), 3.31 (d, *J* = 7.2 Hz, 2H);

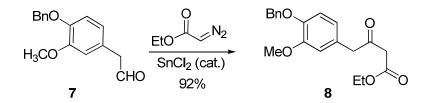
IR (thin film) 3064, 3034, 3003, 2976, 2935, 2907, 2874, 2833, 1638, 1606, 1591, 1514, 1506, 1464, 1454, 1418, 1335, 1260, 1230, 1156, 1140, 1034, 914, 850, 806, 739, 697, 654, 597 cm⁻¹; **LRMS** (**ESI**): 277.1 (M + Na)⁺; **HRMS** (**ESI**): Exact mass calcd for $C_{17}H_{18}O_2Na$ (M + Na), 277.1204. Found 277.1196.



To a solution of 6 (508 mg, 2.0 mmol) in t-BuOH/H₂O/THF (2/1/4) (7.0 mL) was added a solution of OsO₄ in *t*-BuOH (0.20 M, 0.50 mL, 0.010 mmol, 5 mol %) at 10 °C. To the resulting mixture, N-methylmorpholine N-oxide (NMO) (357 mg, 3.0 mmol) was added in one portion. The resulting reaction mixture was stirred at 10 °C for 15 min and then warmed to room temperature. The stirring was continued for another 1 hr before the reaction was quenched with addition of sat. NaHSO₃. The resulting mixture was stirred at room temperature overnight. The mixture was diluted with ethyl acetate (30 mL), and then washed with sat. NaHSO₃(15 mL). The aqueous layer was extracted with ethyl acetate (2×20 mL). The organic phase was collected and washed with brine (20 mL), dried over Na_2SO_4 and concentrated under reduced pressure. The residue was dissolved in CH_2Cl_2 (5.0 mL), and the resulting solution was added into a solution of NaIO₄ (630 mg, 3.0 mmol) in water (5.0 mL) at 0 °C. The resulting light yellow mixture was stirred at 0 °C for 40 min and then quenched with saturated aqueous NaHSO₃. The resulting mixture was stirred at room temperature for 30 min, and then extracted with ethyl acetate (3×20) mL). The organic phase was collected, washed with brine (20 mL), dried over Na₂SO₄ and concentrated under reduced pressure. The residue was subjected silica gel chromatography (petroleum ether/EtOAc = 4/1) to give aldehyde 7 (507 mg, quant.) as a coloress oil.

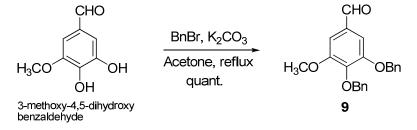
¹**H NMR** (400MHz, CDCl₃) δ 9.68 (t, J = 2.4 Hz, 1H), 7.41 (d, J = 7.2 Hz, 2H), 7.34 (t, J = 7.2 Hz, 2H), 7.30 (t, J = 7.2 Hz, 1H), 6.87 (d, J = 8.0 Hz, 1H), 6.72 (s, 1H), 6.70 (d, J = 8.0 Hz, 1H), 5.12 (s, 2H), 3.85 (s, 3H), 3.57 (d, J = 2.4 Hz, 2H); ¹³**C NMR** (100 MHz, CDCl₃) δ 199.418, 149.965, 147.506, 136.987, 128.479, 127.796, 127.181, 124.669, 121.740, 114.401, 1 13.118, 71.012, 55.947, 50.073; **IR** (thin film) 3063, 3034, 3007, 2937, 2876, 2833, 2728, 1723, 1682, 1591, 1514, 1464, 1454, 1383, 1332, 1231, 1261, 1141, 1034, 854, 810, 741, 698 cm⁻¹;

LRMS (ESI): 279.1 (M + Na)⁺; **HRMS (ESI)**: Exact mass calcd for $C_{16}H_{16}O_3Na$ (M + Na), 279.0997. Found 279.0990.



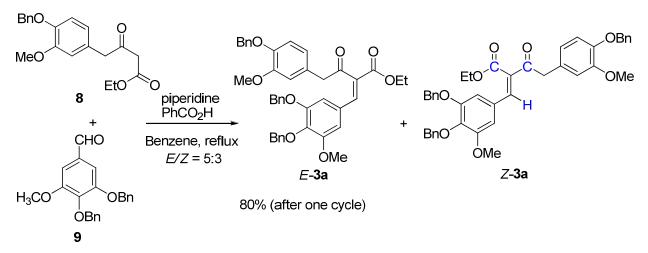
SnCl₂·2H₂O (45 mg, 2.0 mmol, 10 mol %) was suspended in anhydrous CH₂Cl₂ (3.0 mL) at -72 °C. To this suspension, ethyl diazoacetate (0.23 mL, 2.2 mmol) was added dropwise via syringe. The resulting yellow mixture was stirred at this temperature for 10 min and warmed up to room temperature for another 15 min. The mixture was recooled to -72 °C and a solution of **7** (507 mg, 2.0 mmol) in CH₂Cl₂ (5.0 mL) was introduced via syringe over 5 min. The resulting mixture was stirred while warming up to room temperature over 1 hr and stirred at room temperature overnight. The reaction was quenched with addition of aq. HCl (1.0 N, 2.0 mL) and extracted with ethyl acetate (3×25 mL). The combined extracted were washed with 5% aq. NaHCO₃ (3×15 mL), brine (20 mL), dried over Na₂SO₄ and concentrated at reduced pressure. The residue was subjected to silica gel chromatography (petroleum ether/EtOAc = $8/1 \rightarrow 4/1$) to give β-keto ester **8** (640 mg, 92% yield) as a white solid.

¹**H NMR** (400MHz, CDCl₃) δ 7.40 (d, J = 7.2 Hz, 2H), 7.33 (dd, J = 7.6 Hz, 7.2 Hz, 2H), 7.27 (d, J = 7.6 Hz, 1H), 6.81 (d, J = 8.6 Hz, 1H), 6.71 (s, 1H), 6.65 (d, J = 8.6 Hz), 5.11 (s, 2H), 4.14 (q, J = 7.4 Hz, 2H), 3.85 (s, 3H), 3.71 (s, 2H), 3.41 (s, 2H), 1.22 (t, J = 7.4 Hz, 3H); ¹³**C NMR** (100MHz, CDCl₃) δ 200.722, 167.029, 149.654, 147.324, 136.910, 128.394, 127.696, 127.088, 126.056, 121.593, 113.980, 112.834, 70.821, 61.220, 55.815, 49.470, 47.914, 13.940; **IR** (thin film) 3064, 3033, 2982, 2938, 1745, 1716, 1606, 1591, 1515, 1464, 1454, 1421, 1385, 1368, 1317, 1262, 1232, 1142, 1032, 855, 805, 741, 698 cm⁻¹; **LRMS** (**ESI**): 343.2 (M + H)⁺; **HRMS** (**ESI**): Exact mass calcd for C₂₀H₂₃O₅ (M + H), 343.1545. Found 343.1541.



To a brown mixture slurry of 3-methoxy-4,5-dihydroxy-benzaldehyde (1.00 g, 6.0 mmol) and K_2CO_3 (3.00 g, 21.7 mmol, 3.5 eq.) in acetone (20 mL) was added benzyl bromide (1.78 mL, 15.0 mmol, 2.5 eq.) at 0 °C. The corresponding yellow mixture was refluxed for 7 hr. The resulting green mixture was cooled down to 0 °C and quenched carefully with aq. HCl (1N, 35.0 mL). The mixture was diluted with ethyl acetate, washed with water (3×20 mL), saturated NaHCO₃ (3×10 mL), brine (25 mL) and dried over Na₂SO₄. After concentration at reduced pressure, the residue was subjected to silica gel chromatography (petroleum ether/EtOAc = 4/1) to give aldehyde **9** (2.1 g, quant.) as an orange oil.

¹**H NMR** (400MHz, CDCl₃) δ 9.81 (s, 1H), 7.43-7.27 (m, 10H), 7.15 (s, 1H), 7.12 (s, 1H), 5.14 (s, 4H), 3.89 (s, 3H); ¹³**C NMR** (100MHz, CDCl₃) δ 191.014, 154.125, 152.926, 142.929, 137.138, 136.295, 131.741, 128.538, 128.386, 128.159, 128.060, 127.999, 127.415, 108.750, 106.534, 75.019, 71.110, 56.181; **IR** (thin film) 3063, 3032, 2939, 2841, 1693, 1587, 1498, 1454, 1428, 1386, 1327, 1232, 1141, 1120, 979, 833, 737, 697 cm⁻¹; **LRMS** (**ESI**): 349.1 (M + H)⁺; **HRMS** (**ESI**): Exact mass calcd for $C_{22}H_{21}O_4$ (M + H), 349.1440. Found 349.1441.

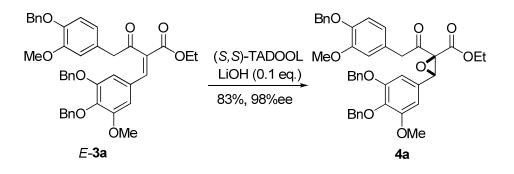


A mixture of β -keto ester **8** (172 mg, 0.50 mmol), aldehyde **9** (174 mg, 0.50 mmol), piperidine (distilled, 8.5 mg, 0.10 mmol) and benzoic acid (9.2 mg, 0.075 mmol) in benzene (20 mL) in a round bottom flask (50 mL) connected to a Dean-Stark apparatus was heated to reflux. After 2.5 hr, the reaction mixture was cooled to room temperature and concentrated at reduced pressure. The residue was subjected to silica gel chromatography (petroleum ether/diethyl ether = $3/1 \rightarrow 1/1$) to give *E*-**3a** (199 mg, 59% yield) as a light yellow oil and *Z*-**3a** (66 mg, 20% yield) as a white solid. Ketoester **8** and aldehyde **9** were recovered (ca. 23 mg for each) in 13% yield.

Isomerization of Z-isomer: A solution of Z-**3a** (66 mg, 0.10 mmol) and pyridine (1 drop) in benzene (5 mL) was refluxed for 5 hr. The solution was cooled to room temperature and diluted with ether. The mixture was concentrated at reduced pressure and the residue was subjected to silica gel chromatography (petroleum ether/EtOAc = 2/1) to give *E*-**3a** (38 mg, 58% yield) and *Z*-**3a** (25 mg, 38% yield), respectively.

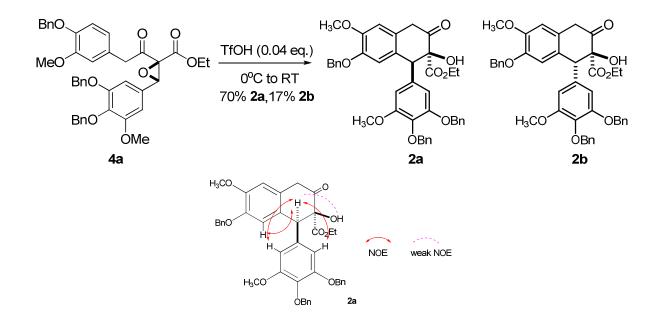
E-**3a**: ¹**H NMR** (400 MHz, CDCl₃) δ 7.57 (s, 1H), 7.42-7.24 (m, 15H), 6.70 (d, *J* = 8.0 Hz, 1H), 6.01 (br, 1H), 6.59 (br, 1H), 6.55 (br, 1H), 6.46 (dd, *J* = 8.4 Hz, 0.8 Hz, 1H), 5.06 (s, 2H), 5.04 (s, 2H), 4.92 (s, 2H), 4.23 (q, *J* = 7.2 Hz, 2H), 3.74 (s, 3H), 3.70 (s, 2H), 3.69 (s, 3H), 1.28 (t, 7.2 Hz, 3H); ¹³**C NMR** (100 MHz, CDCl₃) δ 203.447, 164.380, 153.632, 152.531, 149.389, 147.256, 141.138, 139.476, 137.366, 137.055, 136.432, 132.675, 128.500, 128.462, 128.402, 128.143, 127.984, 127.931, 127.749, 127.438, 127.172, 126.064, 121.897, 113.676, 113.418, 108.705, 107.301, 75.004, 70.890, 70.875, 61.501, 56.013, 55.785, 49.956, 14.145; **IR** (thin film) 3063, 3032, 2937, 2873, 1722, 1580, 1513, 1504, 1462, 1454, 1423, 1377, 1333, 1259, 1232, 1158, 1122, 1027, 912, 855, 736, 697 cm⁻¹; **LRMS (ESI)**: 673.3 (M + H)⁺; **HRMS (ESI)** Exact mass calcd for C₄₂H₄₁O₈ (M + H), 673.2801. Found 673.2812.

Z-**3a**: ¹**H NMR** (400 MHz, CDCl₃) δ 7.56 (s, 1H), 7.42-7.24 (m, 15H), 6.82 (d, *J* = 8.0 Hz, 1H), 6.75 (s, 1H), 6.74 (s, 1H), 6.70 (s, 1H), 6.87 (d, *J* = 8.0 Hz, 1H), 5.13 (s, 2H), 5.08 (s, 2H), 5.05 (s, 2H), 4.17 (q, *J* = 7.2 Hz, 2H), 3.95 (s, 2H), 3.86 (s, 3H), 3.80 (s, 3H), 1.19 (t, *J* = 7.2 Hz, 3H); ¹³C **NMR** (100 MHz, CDCl₃, C-H coupling) δ 194.02 (td, *J* = 6.8, 6.1 Hz, C=O), 168.07 (dt, *J* = 12.2 Hz, 3.0 Hz, COOEt), 153.65, 152.61, 149.58, 147.21, 141.65, 137.27, 137.08, 136.42, 132.71, 128.51, 128.46, 128.40, 128.29, 128.12, 128.00, 127.92, 127.73, 127.41, 127.34, 127.14, 126.47, 121.63, 113.90, 113.02, 109.05, 107.17, 77.21, 75.03, 71.10, 70.91, 61.71, 56.04, 55.89, 45.15, 13.85. ³*J*_{C-H} = 6.1 Hz for C_{ketone}-H, ³*J*_{C-H} = 12.2 Hz for C_{ester}-H.



To a solution of *E*-**3a** (17.0 mg, 0.0250 mmol) and (+)-TADOOL (13.3mg, 0.0275 mmol) in THF (0.15 mL) at 0 °C was added a solution of LiOH (5.0 M, 0.5 μ L, 0.50 mmol) via syringe. The resulting mixture was stirred at 0 °C for 6 hr, and then warmed up to RT over night. The mixture was subjected to silica gel chromatography (petroleum ether/EtOAc = 10/1 \rightarrow 2/1) to give the desired epoxide **4a** (14.5 mg, 83% yield) as a light yellow oil. The 98% ee of **4a** was determined by converting **4a** into **2a** and then an ee analysis of **2a**. TADDOL was recycled as a white solid in > 90 % yield.

 $[\alpha]^{23}{}_{D}$ = - 61° (*c* 0.96, CHCl₃); ¹**H** NMR (400 MHz, CDCl₃) δ 7.43-7.25 (m, 15H), 6.66 (d, *J* = 8.0Hz, 1H), 6.51 (s, 1H), 6.45 (s, 1H), 6.41 (s, 1H), 6.25 (d, *J* = 8.0 Hz, 1H), 5.05 (s, 2H), 5.04 (d, *J* = 11.6 Hz, 1H), 5.01 (s, 2H), 4.96 (d, *J* = 11.6 Hz, 1H), 4.45 (s, 1H), 4.26 (q, *J* = 7.2 Hz, 2H), 3.76 (s, 3H), 3.69 (d, *J* = 16.8 Hz, 1H), 3.41 (d, *J* = 16.8 Hz, 1H), 1.29 (t, *J* = 7.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 197.974, 165.943, 153.905, 152.744, 149.381, 147.233, 138.003, 137.548 137.047, 136.554, 128.470, 128.432, 128.356, 128.113, 127.916, 127.832, 127.711, 127.415, 127.180, 124.728, 121.768, 113.623, 113.183, 105.099, 103.596, 74.935, 70.958, 70.799, 66.662, 62.859, 62.799, 56.066, 55.846, 47.906, 13.947; **IR** (thin film) 3063, 3032, 2938, 1748, 1726, 1592, 1513, 1454, 1430, 1374, 1262, 1235, 1121, 1026, 850, 736, 697 cm⁻¹; **LRMS** (**ESI**): 689.3 (M + H)⁺; **HRMS (ESI**): Exact mass calcd for C₄₂H₄₁O₉ (M + H), 689.2751. Found 689.2744.



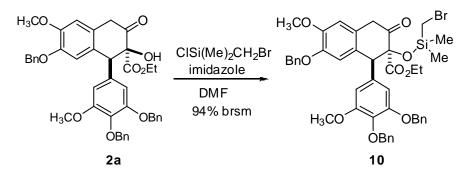
To a solution of epoxide **4a** (344 mg, 0.500 mmol) in CH₂Cl₂ (35 mL) at -10 °C was added dropwise a freshly prepared solution of TfOH in Et₂O (0.50 M, 40 μ L, 0.020 mmol) via syringe with vigorous stirring. After stirring at this temperature for 15 min, a TLC test indicated the starting material **4a** was completely consumed. The reaction was quenched with addition of Et₃N (3 μ L). The reaction mixture was concentrated under reduced pressure, and the residue was subjected to silica gel chromatography (petroleum ether/EtOAc = 5/1 \rightarrow 3/1) to give light blue **2a** (240 mg, 70% yield) and its distereoisomer **2b** (58 mg, 17% yield), respectively. **2a** was determined to be > 98% ee by HPLC analysis [Chiralpak AD, Hexanes/*i*-PrOH = 65:35, 1 mL/min, 220 nm, 20 °C, *t*_R (1) = 25 min, *t*_R (2) = 33 min].

2a: $[\alpha]_{D}^{20} = + 66 (c \ 0.43, CHCl_3)$; ¹H NMR (400 MHz, CDCl₃) δ 7.23-7.45 (m, 15H), 6.64 (s, 1H), 6.62 (s, 1H), 6.37 (d, $J = 2.0 \ Hz$, 1H), 6.08 (d, $J = 1.2 \ Hz$, 1H), 5.02 (s, 2H), 5.00 (s, 2H), 4.93 (AB, $J = 12 \ Hz$, 2H), 4.63 (s, 1H, H-7'), 4.13 (s, 1H, C8'-OH), 4.04 (q, $J = 6.8 \ Hz$, 2H), 3.91 (s, 3H), 3.88 (d, $J = 22 \ Hz$, H-7 α), 3.77 (s, 3H), 3.48 (d, $J = 22 \ Hz$, H-7 β), 1.02 (t, $J = 6.8 \ Hz$, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 204.61, 168.78, 153.11, 151.87, 149.40, 147.60, 137.78, 137.07, 136.93, 136.50, 131.97, 128.45, 128.35, 128.06, 127.90, 127.69, 127.65, 127.42, 127.03, 124.68, 114.58, 111.00, 108.46, 107.51, 82.13, 74.92, 70.82, 70.76, 62.02, 56.07, 55.96, 52.75, 40.99, 13.72; **IR** (thin film) 3436, 2936, 1724, 1589, 1514, 1504, 1454, 1372, 1328, 1298, 1246, 1115, 1014, 736, 697 cm⁻¹; **LRMS (ESI**): 689.3 (M + H)⁺; **HRMS (ESI**): Exact mass calcd for C₄₂H₄₁O₉ (M + H), 689.2751. Found 689.2753.

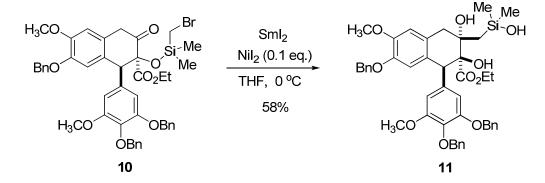
2b: ¹**H NMR** (400 MHz, CDCl₃): see spectrum; ¹³**C NMR** (100 MHz, CDCl₃) δ 203.76, 168.43, 153.44, 152.40, 148.92, 146.77, 137.89, 136.92, 136.75, 133.00, 128.46, 128.44, 128.39, 128.15, 127.86, 127.84, 127.79, 127.38, 127.23, 125.27, 114.13, 110.74, 109.19, 107.71, 83.23, 75.01, 70.92, 70.89, 62.24, 58.12, 56.06, 56.03, 42.24, 13.73.

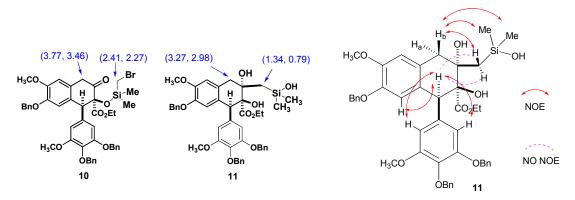
NOTE: Various Lewis acids were screened for their ability to effect the intramolecular epoxide ring opening Friedel-Crafts cyclization with **4a** to form the 2,7'-cyclolignane skeleton with concomitant installation of the C7' stereocenter. Although both $SnCl_4^{4a-c}$ and $BF_3 Et_2O^{4d}$ have been documented for promoting Friedel-Crafts reaction with epoxides, only the latter promoted the desired cyclization when employed in large excess (six equivalents). However, the yield of the BF₃-promoted Friedel-Crafts reaction was found to be inconsistent at a scale of greater than 100 mg.

⁴ (a) Nicolaou, K. C.; Wu, T. R.; Kang, Q.; Chen, D. Y.-K. *Angew. Chem. Int. Ed.* **2009**, *48*, 3440. (b) Kraus, G. A.; Kim, I. *Org. Lett.* **2003**, *5*, 1991. (c) Amupitan, J.; Sutherland, J. K. *J. Chem. Soc., Chem. Commun.* **1978**, 852. (d) Ono, M.; Suzuki, K.; Akia, H. *Tetrahedron Lett.* **1999**, *40*, 8223.



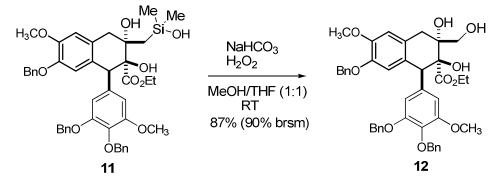
To a solution of hydroxyketone 2a (180 mg, 0.26 mmol) in anhydrous DMF (1.0 mL) at RT was added ClSi(Me)₂CH₂Br (113 µL, 0.83 mmol), followed by imidazole (54 mg, 0.78 mmol). The resulting mixture was stirred at this temperature for 1 hr and then subjected to silica gel chromatography directly eluted with petroleum ether/EtOAc = $10/1 \rightarrow 3/1$ to give silvl ether 10 (165 mg, 94% yield brsm) as light green foam, and the starting material (37 mg, 20% recovery). $[\alpha]_{D}^{20} = +42.4 \ (c \ 1.34, \ CHCl_3); \ ^{1}H \ NMR \ (400 \ MHz, \ CDCl_3) \ \delta \ 7.50 \sim 7.15 \ (m, \ 15H), \ 6.59 \ (s, \ 2H),$ 6.36 (s, 1H), 6.03 (s, 1H), 5.00(s, 2H), 4.99 (s, 2H), 4.93 (AB, J = 12.4 Hz, 2H), 4.53 (s, 1H), 4.02 (m, 2H), 3.89 (s, 3H), 3.77 (A of AB, J = 22.0 Hz, 1H), 3.77 (s, 3H), 3.46 (B of AB, J = 22.0 Hz, 1H), 2.41 (A of AB, J = 12.7 Hz, 1H), 2.27 (B of AB, J = 12.7 Hz, 1H), 1.02 (t, J = 7.1 Hz, 3H), 0.13 (s, 3H), 0.06 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 204.16, 168.90, 153.01, 151.97, 149.35, 147.49, 137.67, 137.06, 137.03, 136.53, 132.78, 128.45, 128.42, 128.40, 128.09, 127.86, 127.75, 127.68, 127.39, 127.22, 126.99, 124.76, 114.50, 110.76, 108.57, 107.94, 85.71, 75.06, 71.01, 70.75, 61.92, 56.11, 55.95, 54.87, 42.08, 18.63, 13.72, -1.01, -1.10; **IR** (thin film) 3088, 3064, 3032, 2936, 1731, 1590, 1514, 1454, 1248, 1114, 1028 cm⁻¹; LRMS (ESI): 841.2, 839.2 $(M + H)^+$; **HRMS (ESI)**: Exact mass calcd for C₄₅H₄₈O₉BrSi (M + H), 839.2251.Found 839.2239.





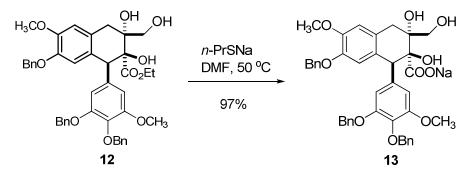
To a solution of bromide **10** (120 mg, 0.143 mmol) in anhydrous THF (10 mL) under N₂ at 0 °C was added SmI₂ (0.10M THF soln, containing 1 mol % NiI₂, 5.0 mL, 0.50 mmol) dropwise. The resulting mixture was stirred at 0 °C for 1 hr and then let warm to RT. The solvent was evaporated and the residue was taken up in EtOAc, rinsed with aq. NaHCO₃ (sat.) and brine successively. The organic layer was dried over anhydrous Na₂SO₄. After the solvent was removed by rotavap, the residue was subjected to purification on silica gel eluted with petroleum ether/EtOAc = $3/1 \rightarrow 2/1$ to give silanol **11** (65 mg, 58% yield) as dense colorless liquid. Splitting and broadening peaks are observed in both the ¹H NMR and ¹³C NMR spectra due to the existence of atropisomers arising from hindered rotation along C1'-C7' bond.

[α]_D²⁰ = -2.98 (*c* 1.17, CHCl₃); ¹**H** NMR (400 MHz, CDCl₃) δ 7.50~7.18 (m, 15H), 6.65 (s, 1H), 6.44 (br, 0.4H), 6.38 (br, 0.6H), 6.29 (br, 0.6H), 6.25 (br, 1.4H), 5.09 (br, 1H), 5.08 (s, 2H), 4.89~4.78 (m, 2H), 4.81 (s, 1H), 4.57 (s, 1H), 4.10 (m, 1.2H), 3.98 (m, 0.8H), 3.89 (s, 3H), 3.82 (s, 1.8H), 3.67 (s, 1.2H), 3.55 (br, 1H), 3.27 (A of AB, J = 16.9 Hz), 2.98 (B of AB, J = 16.9 Hz), 2.86 (br, 1H), 1.34 (A of AB, J = 14.6 Hz), 1.11 (t, J = 6.8 Hz), 0.79 (B of AB, J = 14.6 Hz), 0.30 (s, 3H), 0.22 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 174.71, 153.80 (br), 153.19 (br), 152.34 (br), 152.04 (br), 148.40, 146.19, 137.78, 137.15, 136.86 (br), 134.66, 128.46 (br), 128.39, 128.30, 128.11, 128.07 (br), 127.78, 127.61, 127.28, 127.14, 127.06 (br), 126.66, 114.94, 111.77, 110.36 (br), 108.52, 107.02 (br), 81.53, 75.54, 74.93, 71.16 (br), 70.90 (br), 70.33 (br), 62.35, 56.07 (br), 55.90, 50.39 (br), 39.00, 27.82 (br), 14.09, 2.29, 2.14; **IR** (thin film) 3505, 3063, 3031, 2936, 1712, 1589, 1514, 1454, 1373, 1330, 1252, 1213, 1119, 1028cm⁻¹; **LRMS (ESI)**: 801.3 (M + Na)⁺; **HRMS (ESI)**: Exact mass calcd for C₄₅H₅₀O₁₀NaSi (M + Na), 801.3071. Found 801.3062.



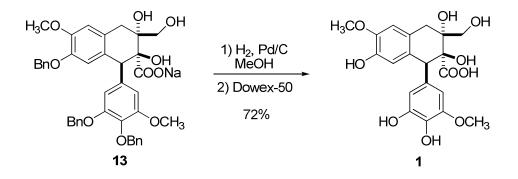
To a solution of silanol **11** (65 mg, 0.084 mmol) in a mixed solvent of MeOH/THF (1:1, 6.0 mL) at RT was added NaHCO₃ (45 mg, 0.54 mmol) in one portion and H₂O₂ (35%, 150 μ L, ca. 1.5 mmol) successively. The resulting mixture was stirred at this temperature over night. The solvent was evaporated at reduced pressure and the residue was subjected to silica gel chromatography eluted with petroleum ether/EtOAc = $3/1 \rightarrow 1/1$ to give triol **12** (50 mg, 87% yield, 90% yield brsm) as dense colorless liquid, along with recovered substrate (3mg). Splitting and broadening peaks are observed in both the ¹H NMR and ¹³C NMR spectra due to the existence of atropisomers arising from hindered rotation along C1'-C7' bond.

 $[\alpha]_{D}^{20} = +5.31 (c 0.833, CHCl_3);$ ¹H NMR (400 MHz, CDCl₃) δ 7.49~7.16 (m, 15H), 6.65 (s, 1H), 6.39 (br, 2H), 6.23 (s, 1H), 5.08 (s, 2H), 5.05 (m, 1H), 4.89~4.79(3H), 4.58 (s, 1H), 4.08 (br, 2H), 3.88 (s, 3H), 3.82 (br, 1.4H), 3.74 (br, 1.6H), 3.55~3.53(br, 2H), 3.25 (br, 1H), 3.14 (A of AB, *J* = 17.2 Hz, 1H), 2.68 (B of AB, *J* = 17.2 Hz, 1H), 1.89 (br, 1H), 1.17 (br, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 174.90, 153.64-153.04 (br), 152.29-151.94 (br), 148.33, 146.27, 137.83, 137.13, 136.92 (br), 134.20, 128.40, 128.31, 128.27, 128.10, 127.76, 127.67, 127.62, 127.28, 127.15, 125.52, 114.92, 111.80, 110.89 (br), 108.88 (br), 107.40 (br), 77.76, 74.91, 74.83, 71.14 (br), 70.86, 70.37 (br), 66.11, 62.50, 56.06 (br), 55.94, 49.94, 34.95, 29.66, 14.03; **IR** (thin film) 3506, 3064, 3031, 2936, 1712, 1589, 1514, 1453, 1373, 1330, 1252, 1213, 1119, 1028 cm⁻¹; **LRMS** (**ESI**): 721.3 (M + H)⁺; **HRMS (ESI**): Exact mass calcd for C₄₃H₄₅O₁₀ (M + H), 721.3013. Found 721.3010.



Triol **12** (50 mg, 0.069 mmol) was placed in a dry flask under N₂ and a freshly prepared solution of *n*-PrSNa in anhydrous DMF (0.15 M, 0.60 mL, 0.090 mmol) was introduced. The resulting mixture was stirred at 50 °C for 24 hr. DMF was then evaporated at reduced pressure and the residue was subjected to silica gel chromatography eluted with $CH_2Cl_2/CH_3OH = 5/1 \rightarrow 0/1$ to give sodium carboxylate **13** (48 mg, 97% yield) as white amorphous powder. Splitting and broadening peaks are observed in both the ¹H NMR and ¹³C NMR spectra due to the existence of atropisomers arising from hindered rotation along C1'-C7' bond.

[α]_D²⁰ = +62.5 (*c* 0.233, CHCl₃); ¹**H NMR** (400 MHz, CD₃OD) δ 7.42~7.13 (m, 15H), 6.73 (s, 1H), 6.69-6.62 (split br, 1H), 6.46-6.36 (split br, 1H), 6.23 (s, 1H), 5.10~4.70 (6H), 4.64 (s, 1H), 3.82 (s, 3H), 3.77-3.51 (split br, 3H), 3.68 (A of AB, J = 11.2 Hz, 1H), 3.60 (B of AB, J = 11.2 Hz, 1H), 3.13 (A of AB, J = 17.0 Hz, 1H), 2.75(A of AB, J = 17.0 Hz, 1H); ¹³C **NMR** (125 MHz, CD₃OD) δ 179.94 (br), 154.05 (br), 152.96 (br), 149.48, 147.24, 139.31, 138.74, 137.90, 137.25, 131.40, 129.67, 129.36, 129.13, 129.02, 128.87, 128.72, 28.47, 116.87, 113.66, 112.59 (br), 111.02 (br), 110.78 (br), 109.43 (br), 78.82 (br), 76.14, 75.87, 72.02 (br), 71.45 (br), 67.86, 56.57, 51.19, 35.61, 30.75 (br); **IR** (thin film) 3439, 3089, 3063, 3031, 2922, 2852, 1651, 1598, 1514, 1504, 1454, 1378, 1327, 1255, 1218, 1119, 1100, 1027cm⁻¹; **LRMS (ESI)**: 715.3 (M + H)⁺; **HRMS (ESI)**: Exact mass calcd for C₄₁H₄₀O₁₀Na (M + H), 715.2519. Found 715.2507.



To a solution of sodium carboxylate **13** (47 mg, 0.066 mmol) in MeOH (25 mL) was added Pd/C (15%, 20 mg). A balloon filled with H_2 was connected to the flask. The reaction mixture was flushed with H_2 by 3 cycles of vacuuming/refilling and then kept stirring at RT for 4 hr. The mixture was then filtered through a Dowex-50 column, which was further eluted with MeOH. The filtrate was collected and concentrated at reduced pressure to give a light brown solid (28 mg, 100% crude yield). The crude product was further purified on preparative TLC plate developing with CH_2Cl_2/CH_3OH (4:1) to give a light tan amorphous solid (20 mg, 72% yield). Splitting and broadening peaks are observed in both the ¹H NMR and ¹³C NMR spectra due to the existence of atropisomers arising from hindered rotation along C1'-C7' bond.

Synthetic Sample

 $[\alpha]_{D}^{20} = -40 \ (c \ 0.040, \ water), -130 \ (c \ 0.017, \ water); {}^{1}\mathbf{H} \ \mathbf{NMR} \ (400 \ MHz, \ CD_{3}OD, 21 \ ^{\circ}C) \ \delta \ 6.67$ (s, 1H), 6.51 (br, 1H), 6.36 (br, 1H), 6.22 (s, 1H), 4.59 (m, 1H), 3.81 (s, 3H), 3.747-3.66 (4H), 3.51 (br, 1H), 3.11 (A of AB, $J = 16.2 \ Hz$, 1H), 2.65 (B of AB, $J = 16.2 \ Hz$, 1H); ${}^{1}\mathbf{H} \ \mathbf{NMR} \ (400 \ MHz, \ CD_{3}OD, 70 \ ^{\circ}C) \ \delta \ 6.67 \ (s, 1H), 6.47 \ (br, 2H), 6.25 \ (s, 1H), 4.57 \ (s, 1H), 3.82 \ (s, 3H), 3.73$ (br, 4H), 3.53 (br, 1H), 3.15 (A of AB, $J = 17.2 \ Hz$, 1H), 2.64 (B of AB, $J = 17.2 \ Hz$, 1H); ${}^{13}C \ \mathbf{NMR} \ (125 \ MHz, \ CD_{3}OD, 21 \ ^{\circ}C) \ \delta \ 181.20 \ (br), 149.25 \ (br), 148.58 \ (br), 147.50, 146.00 \ (br), 145.32 \ (br), 145.25, 134.02 \ (br), 132.65 \ (br), 132.18 \ (br), 126.11 \ (br), 117.30, 114.43 \ (br), 113.03, 112.62, 109.38 \ (br), 107.37 \ (br), 78.82 \ (br), 76.09 \ (br), 67.41 \ (br), 56.66 \ (br), 56.48, 50.52 \ (br), 36.03; IR \ (thin film) 3392 \ (br, strong), 2954, 1611 \ (split), 1515, 1350 \ (br), 1236 \ (br, multiple), 1104, 870 \ (weak) \ cm^{-1}; \ LRMS \ (ESI): 445.1 \ (M + Na)^+; \ HRMS \ (ESI): Exact mass calcd for <math>C_{20}H_{22}O_{10}Na \ (M + Na), 445.1111.$ Found 445.1112.

Authentic Sample

 $[α]_D^{21} = -9.99$ (water) ⁵; $[α]_D^{20} = -18$ (*c* 0.040, water); ¹H NMR (400 MHz, CD₃OD, 21 °C) δ 6.68 (s, 1H), 6.50 (br, 1H), 6.33 (br, 1H), 6.24 (s, 1H), 4.54 (m, 1H), 3.81 (s, 3H), 3.81-3.67 (4H), 3.58 (br, 1H), 3.12 (A of AB, *J* = 16.6Hz, 1H), 2.79 (B of AB, *J* = 16.6Hz, 1H); ¹H NMR (400 MHz, CD₃OD, 70 °C) δ 6.67 (s, 1H), 6.45 (s, 2H), 6.27 (s, 1H), 4.55 (m, 1H), 3.82 (s, 3H), 3.77 (br, 3H), 3.68 (A of AB, *J* = 11.2 Hz, 1H), 3.61 (B of AB, *J* = 11.2 Hz, 1H), 3.17 (A of AB, *J* = 16.6 Hz, 1H); ¹³C NMR (125 MHz, CD₃OD, 21 °C) δ 180.52, 149.14, 148.67, 147.42, 145.80, 145.38, 145.06, 133.91, 132.72, 132.12, 126.78, 117.23, 114.17, 113.00, 112.68, 109.12, 107.50, 79.04, 75.72, 68.38, 56.63, 56.44, 50.67, 35.19; **IR** (thin film)

⁵ Gradner, J. A. F.; Barton, G. M.; MacLean, H. Can. J. Chem. **1959**, 37, 1703.

3391 (br, strong), 2957, 1610 (split), 1515, 1361 (br), 1236 (br), 1097, 868 (weak) cm⁻¹; LRMS (ESI): 445.1 (M + Na)⁺; HRMS (ESI): Exact mass calcd for $C_{20}H_{22}O_{10}Na$ (M + Na), 445.1111. Found 445.1111.

Mixture Sample* [prepared by mixing the Authentic Sample (~10 mg) and Synthetic Sample (~10 mg)]

¹**H NMR** (400 MHz, CD₃OD, 21 °C) δ 6.65 (s, 1H), 6.48 (br, 1H), 6.32 (br, 1H), 6.21 (s, 1H), 4.54 (m, 1H), 3.79 (s, 3H), 3.79-3.64 (4H), 3.57 (br, 1H), 3.11 (A of AB, J = 16.6 Hz, 1H), 2.72 (B of AB, J = 16.6 Hz, 1H); ¹**H NMR** (400 MHz, CD₃OD, 70 °C) δ 6.67 (s, 1H), 6.46 (s, 2H), 6.26 (s, 1H), 4.56 (m, 1H), 3.83 (s, 3H), 3.75 (br, 3H), 3.66 (br, 1H), 3.62 (br, 1H), 3.16 (A of AB, J = 16.6 Hz, 1H), 2.70 (B of AB, J = 16.6 Hz, 1H); ¹³C **NMR** (125 MHz, CD₃OD, 21 °C) δ 180.77, 149.16, 148.60, 147.44, 145.85, 145.29, 145.12, 133.96, 132.69, 132.13, 126.47, 117.26, 114.27, 113.00, 112.65, 109.24, 107.41, 78.95, 75.86, 67.96, 56.63, 56.45, 50.59, 35.53;

Comparison of ¹³C NMR

	AUTHENTIC	SYNTHETIC	MIXTURE*
1	180.52	181.20	180.77
2 ^a	149.14	149.25	149.16
3 ^a	148.67	148.58	148.60
4	147.42	147.50	147.44
5 ^b	145.80	146.00	145.85
6 ^b	145.38	145.32	145.29
7	145.06	145.25	145.12
8	133.91	134.02	133.96
9	132.72	132.65	132.69
10	132.12	132.18	132.13
11	126.78	126.11	126.47
12	117.23	117.30	117.26
13 [°]	114.17	114.43	114.27
14	113.00	113.03	113.00
15 [°]	112.68	112.62	112.65
16 ^d	109.12	109.38	109.24
17 ^d	107.50	107.37	107.41
18	79.04	78.82	78.95
19	75.72	76.09	75.86
20	68.38	67.41	67.96
21	56.63	56.66	56.63
22	56.44	56.48	56.45
23	50.67	50.52	50.59
24	35.19	36.03	35.53

* Mixture Sample was prepared by mixing the Authentic Sample (~10 mg) and Synthetic Sample (~10 mg).

^a probably splitting from one carbon due to existence of the rotamers.

^b probably splitting from one carbon due to existence of the rotamers.

^c probably splitting from one carbon due to existence of the rotamers.

^d probably splitting from one carbon due to existence of the rotamers.

Characterization of synthetic plicatic acid by spiking experiments and CD spectrometry

The ¹H and ¹³C NMR spectra of natural plicatic acid were not reported. Perhaps due to numerous possible patterns of intra or inter molecular hydrogen bonding interactions between various functionalities, the ¹H and ¹³ C NMR spectra were found to vary with either the synthetic or the authentic sample. The identity of the synthetic (-)-plicatic acid was established by spiking experiments. A sample of an approximately 1:1 mixture of synthetic and authentic plicatic acids was prepared. Spectroscopic and chromatographic studies of this mixture, including ¹H NMR, ¹³C NMR as well as HPLC chromatograms obtained with two different columns, showed that the mixture was homogeneous. The exact mass of both the synthetic and natural plicatic acid was determined to be consistent with the molecular formula of plicatic acid by HRMS analysis. The absolute configuration of the synthetic sample was determined to be the same as the natural plicatic acid through a comparison of their respective CD spectrum.

1. ¹H NMR and ¹³C NMR (see page S73-S99)

First, NMR (¹H NMR and ¹³C NMR) spectra of Synthetic Sample and Authentic Sample were recorded respectively; they were found to be slightly different. Then, NMR (¹H NMR and ¹³C NMR) spectra of Mixture Sample comprising Synthetic Sample and Authentic Sample in approximately 1:1 ratio were recorded; only one set of peaks was observed in the NMR spectra, which indicated that the mixture sample was homogeneous. It's worth mentioning that both ¹H NMR and ¹³C NMR spectra of these samples show splitting and broadening peaks probably due to atropisomerism (retarded rotation along C1'-C7' bond) and extensive hydrogen bonding interactions as well. Similar splitting and broadening patterns are present in the ¹H NMR and ¹³C NMR spectra of **11**, **12**, and **13**.

2. VT-¹H NMR (see page S100)

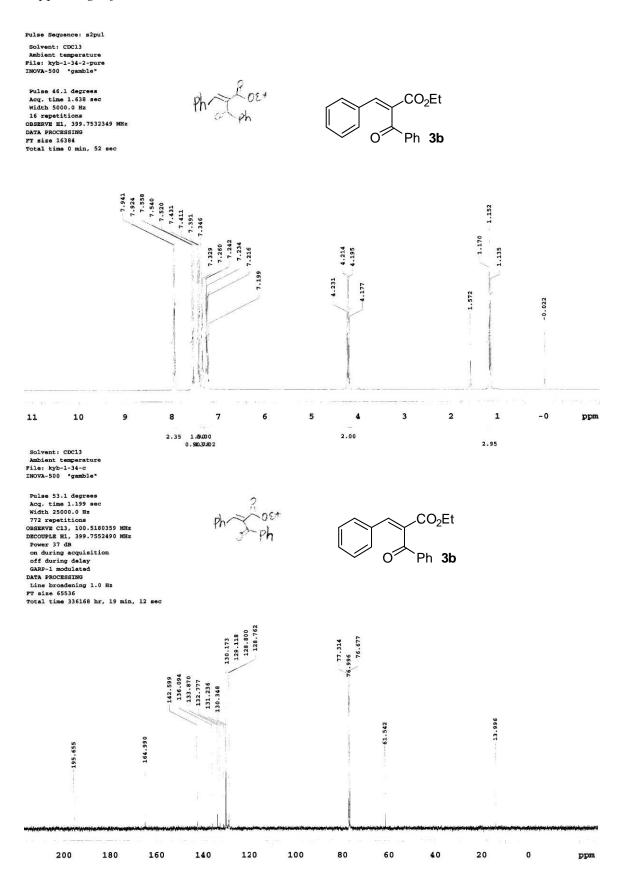
VT-¹H NMR experiments were performed on Authentic Sample, Synthetic Sample, Mixture Sample, respectively, at 21, 30, 40, 50, 60, and 70 °C in stages. Similar patterns and trends in the NMR spectra were observed for all three samples.

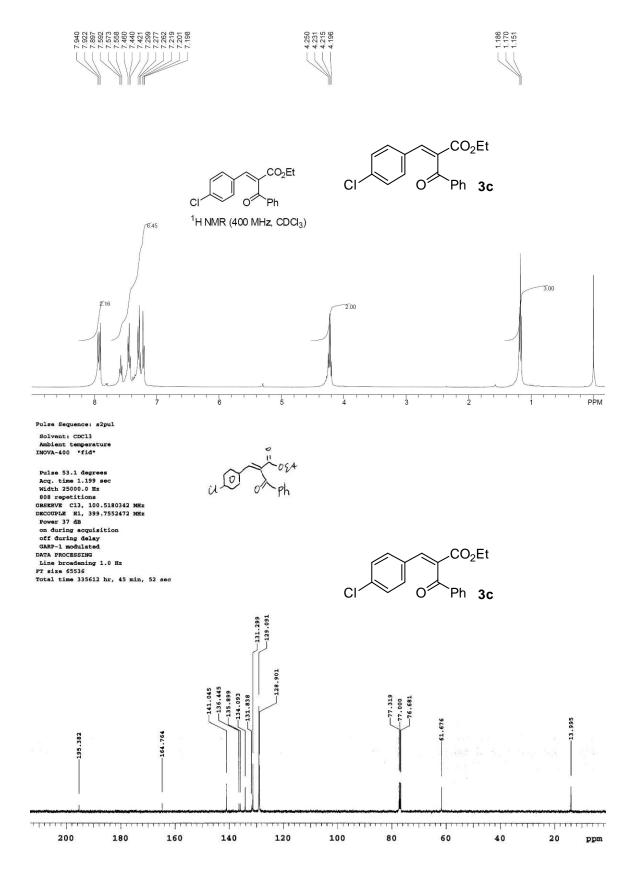
Take Mixture Sample for instance. Two separate peaks (1H for each by integration) at δ 6.48 and δ 6.32 at 21 °C gradually merge into one at δ 6.46 (2H by integration) as the temperature steadily rises to 70 °C; CH₃O (on pending ring C) gives split peaks at δ 3.79 and δ 3.64 (3H combined, by integration) at 21 °C which gradually unite at δ 3.75 (3H by integration) as the temperature steadily rises to 70 °C.

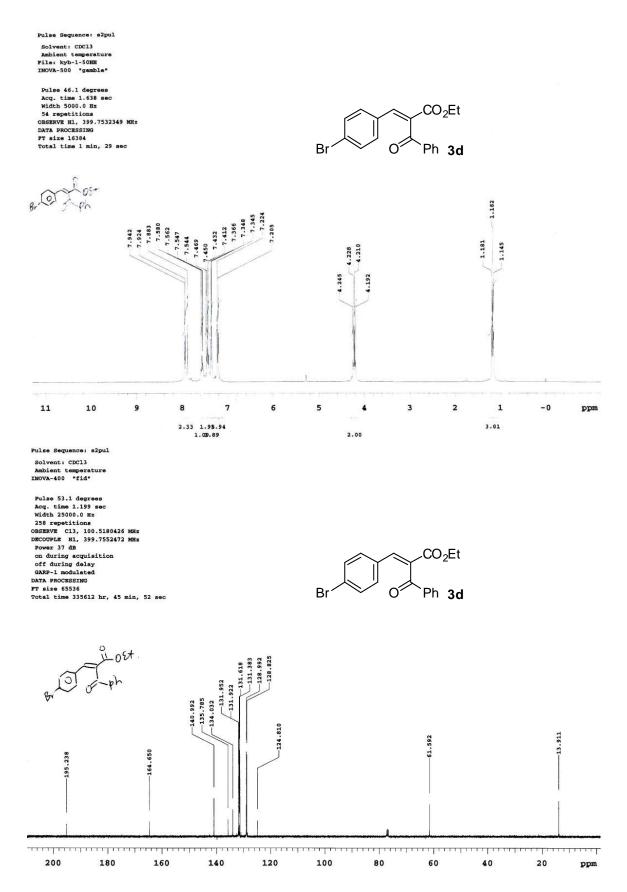
3. HPLC (see page S106)

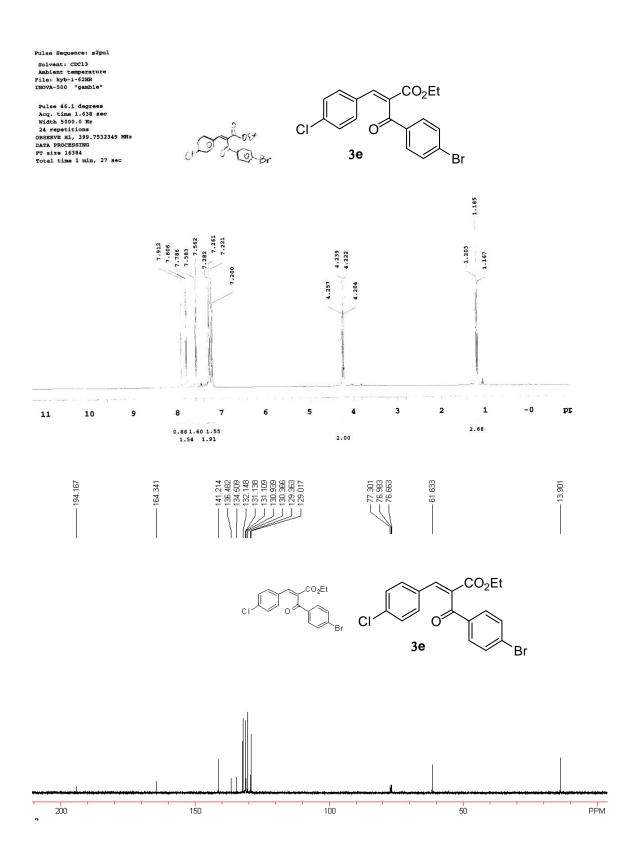
HPLC analysis was performed using two different columns, Zorbax Rx C8 and Symmetry® C18, for Synthetic Sample, Authentic Sample and Mixture Sample, respectively. On Zorbax Rx C8 column (100% water, 0.25 mL/min, 254 nm), the retention time for Synthetic, Authentic, Mixture Sample are 7.91, 7.91, 8.05 min. On Symmetry® C18 column (100% water, 0.20 mL/min, 254 nm), the retention time for Synthetic, Authentic, Mixture Sample are 9.19, 8.97, 9.09 min, respectively. Thus, only one peak was shown by HPLC analysis of Mixture Sample. *4. CD (see page S106)*

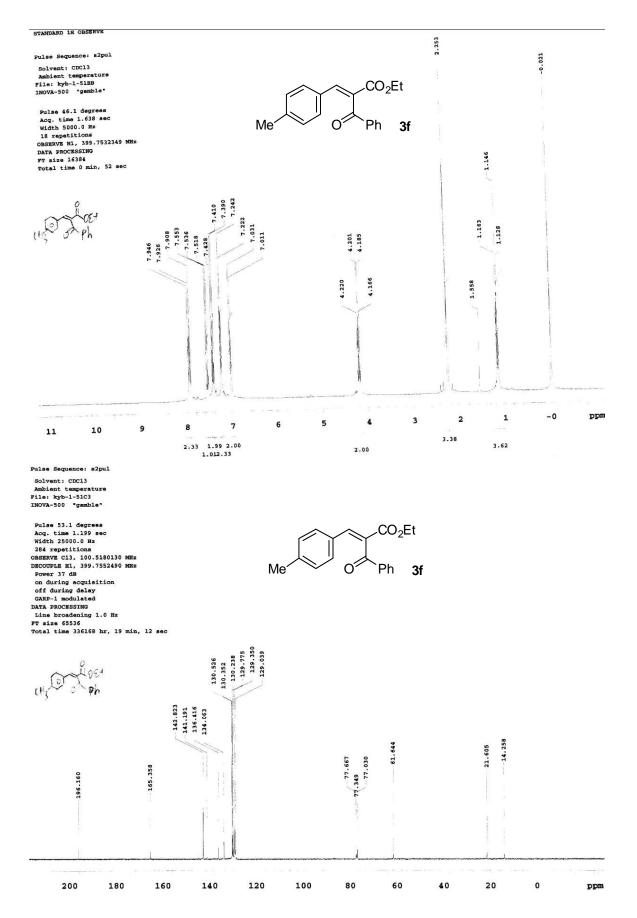
A comparison of CD spectrum of Synthetic Sample and Authentic Sample indicates they are of the sample absolute configuration.



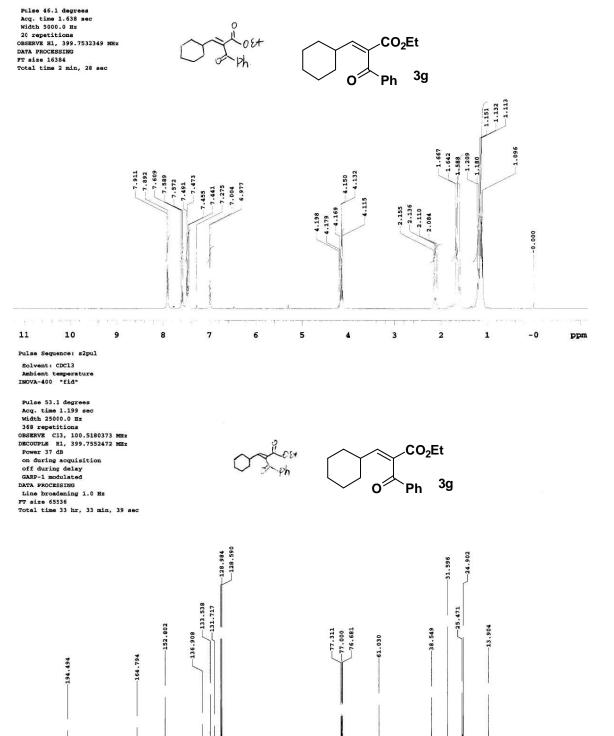






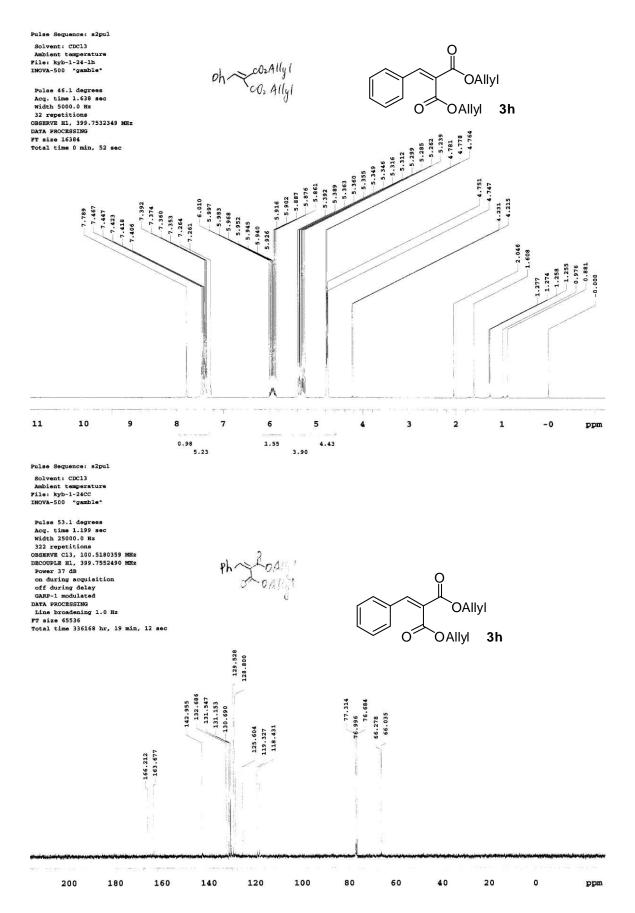


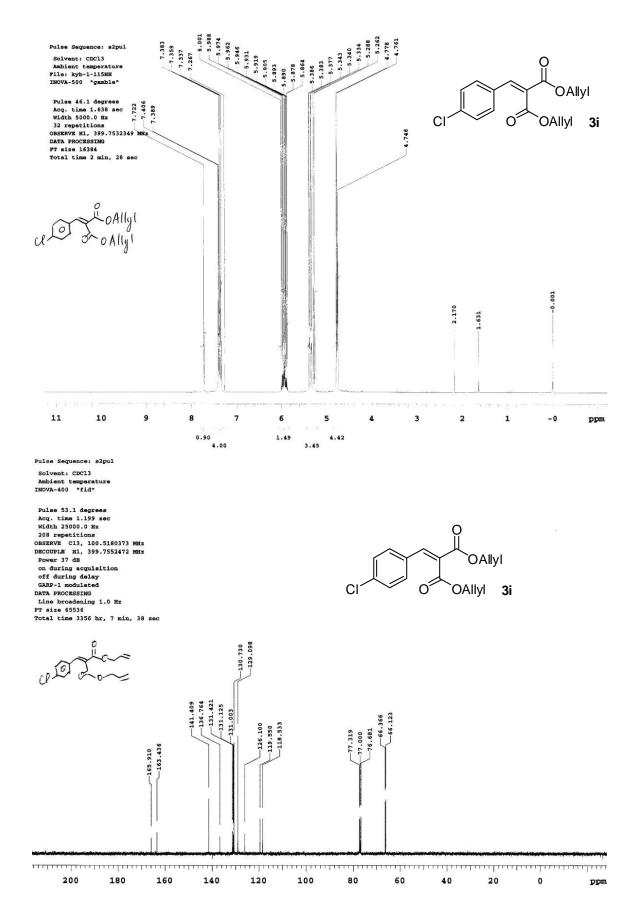
Pulse Sequence: s2pul Sclvent: CDCl3 Ambient temperature File: kyb-1-127B-better INCVA-500 *gamble*



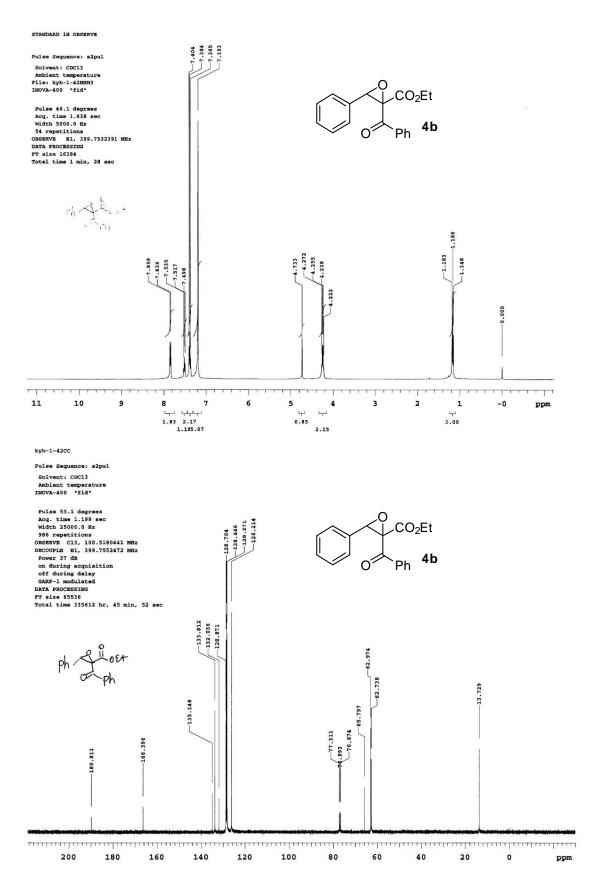
TTTT

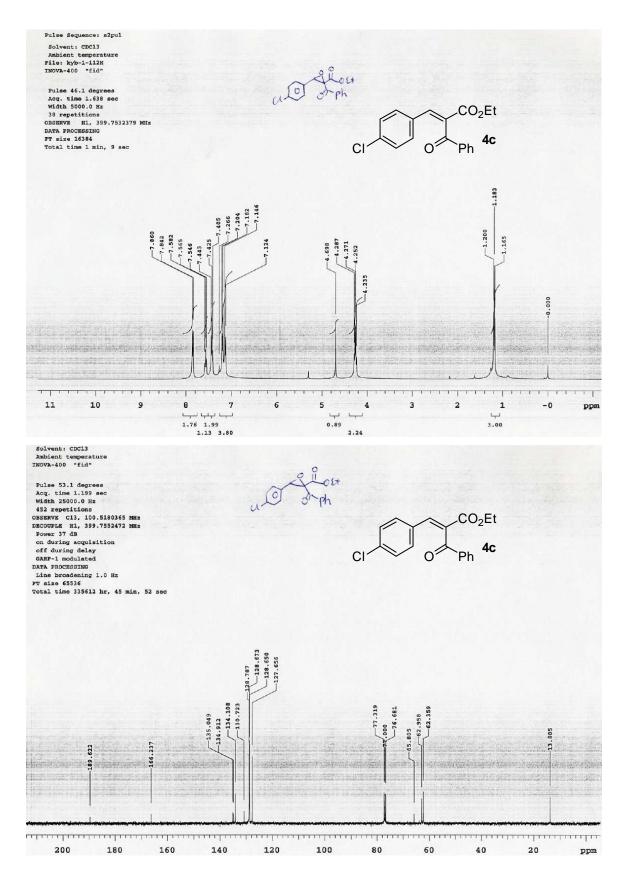
ppm

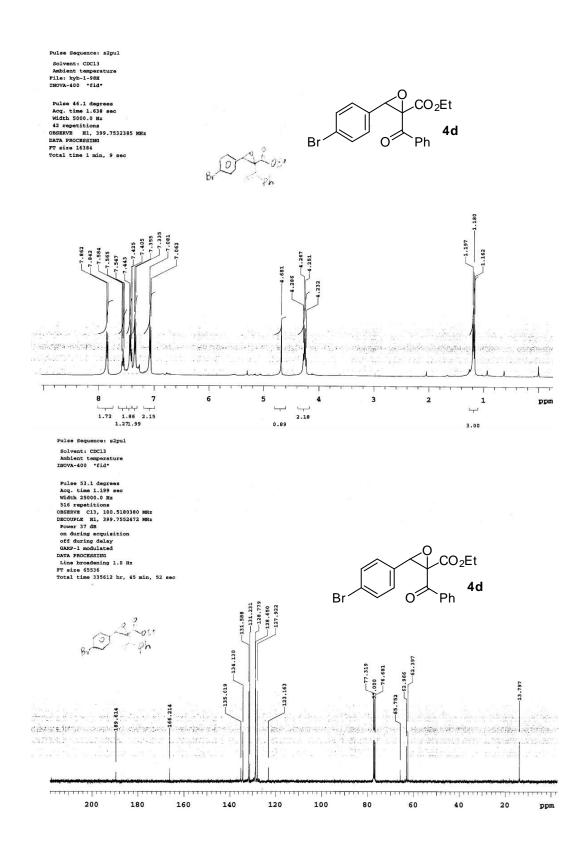


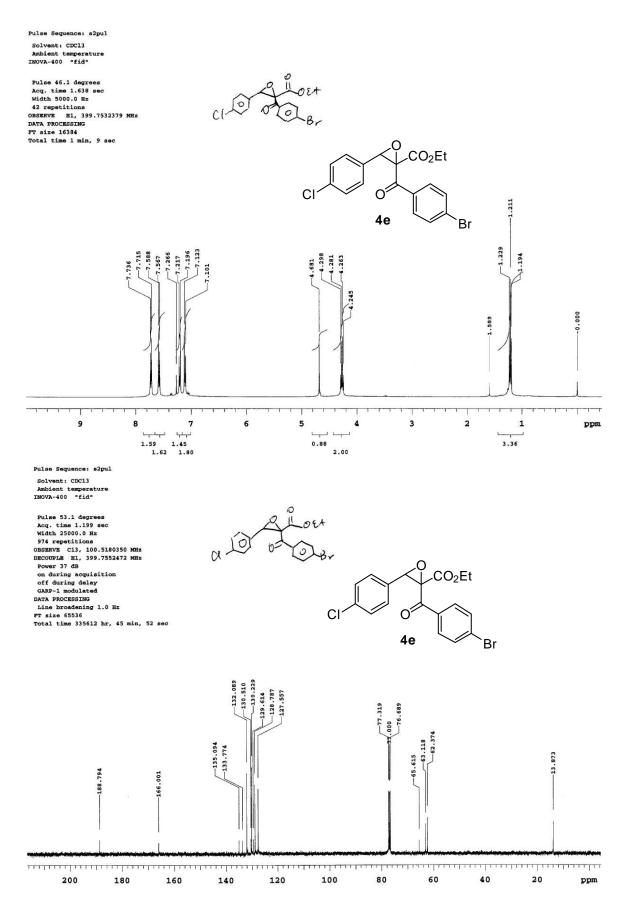


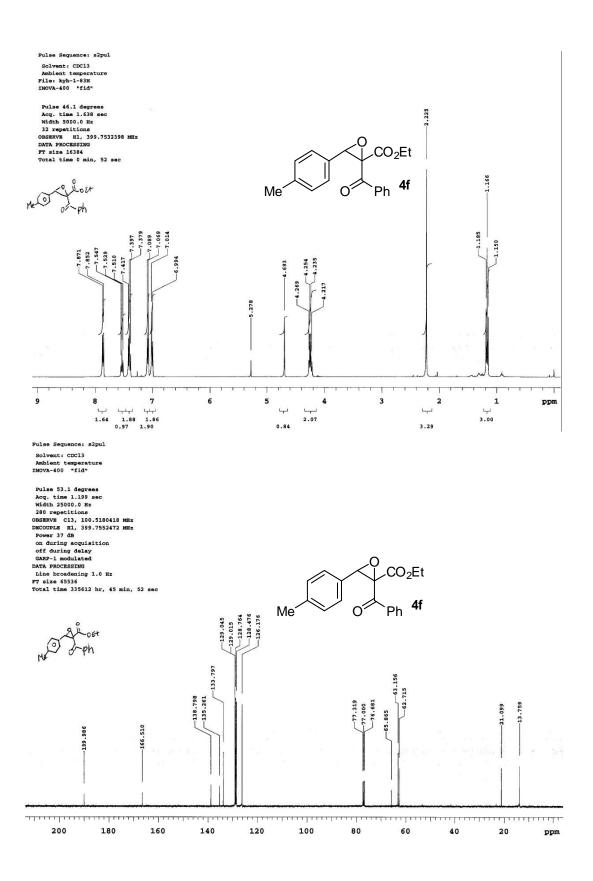
Supporting Information

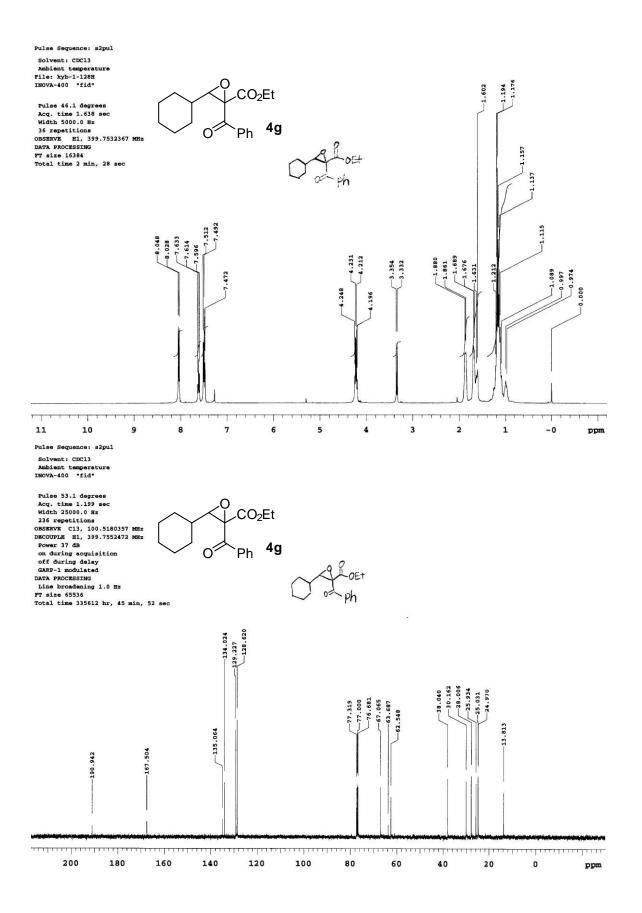


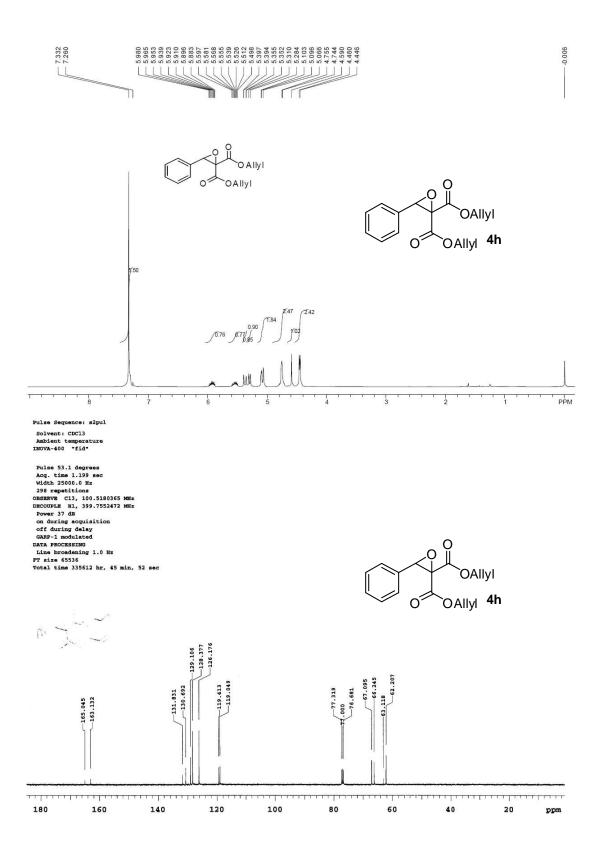




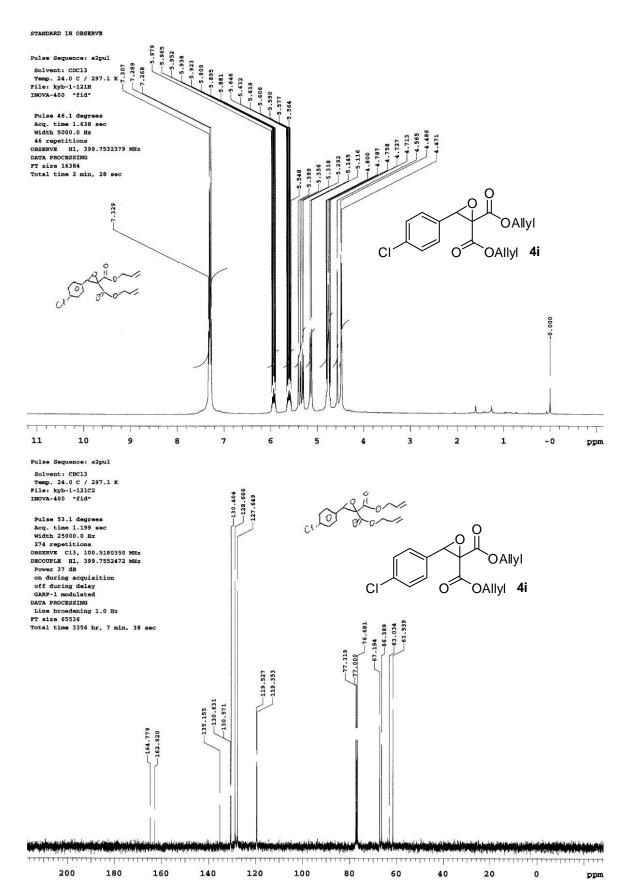


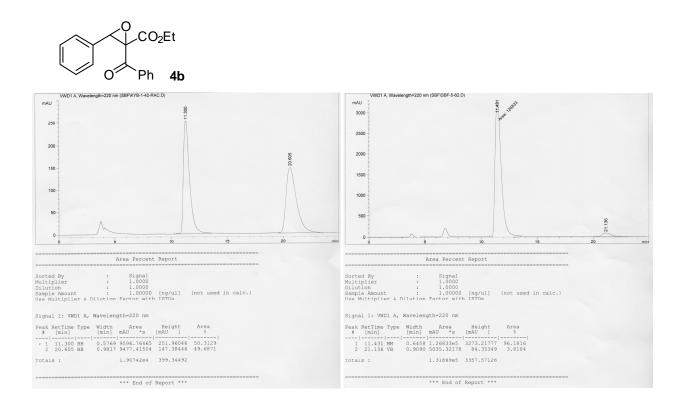




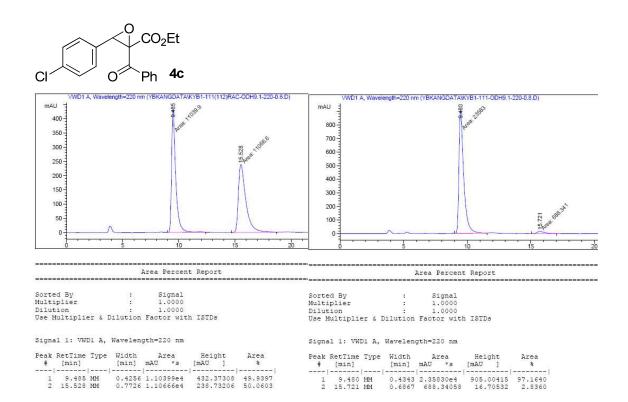


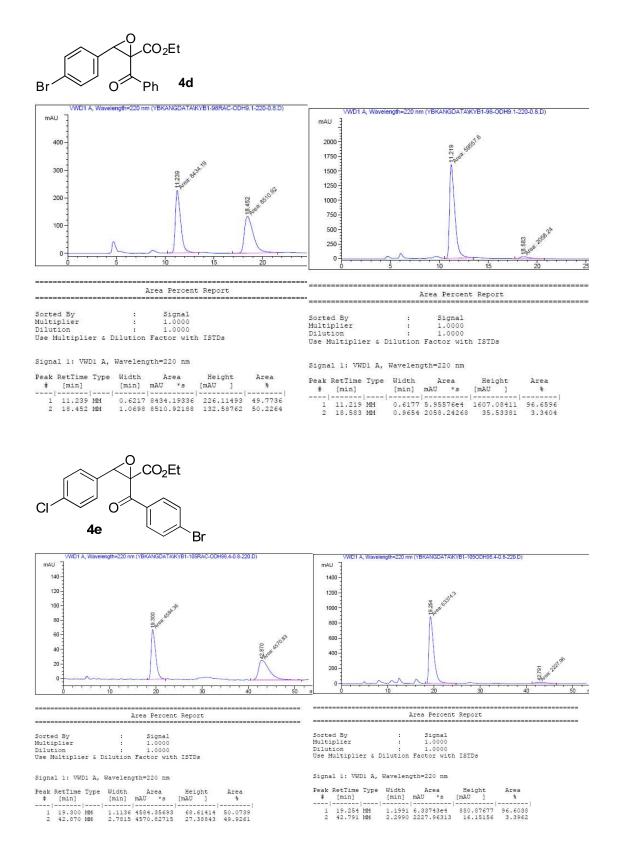
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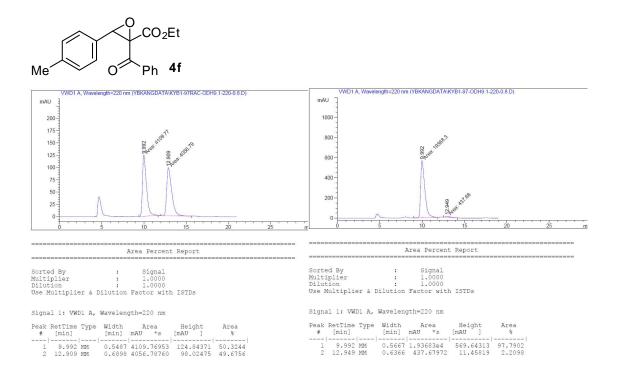


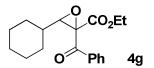


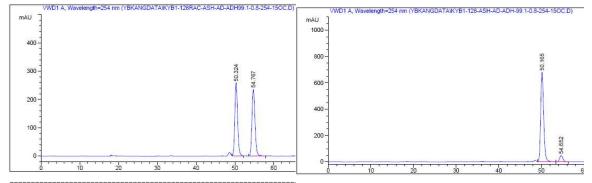
HPLC Spectra for 4b-i











Area Percent Report Area Percent Report

Sort	ed By		: Signal			
Multiplier			: 1.0000			
Dilution			3 3	1.0000		
Use	Multiplier	\$	Dilution	Factor	with	ISTDs

Signal 1: VWD1 A, Wavelength=254 nm

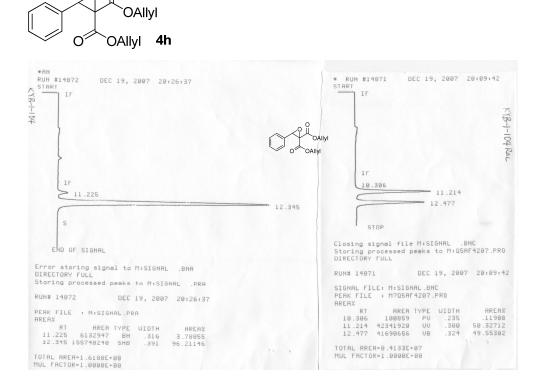
Peak	RetTime Type		Width Area		rea	Height		Area
#	[min]		[min]	mAU	* 3	[mAU	1	왐
1	50.324	VV	0.7735	1.30	278e4	258.	87851	49.9211
2	54.767	VV	0.8571	1.30	689e4	235.	49992	50.0789

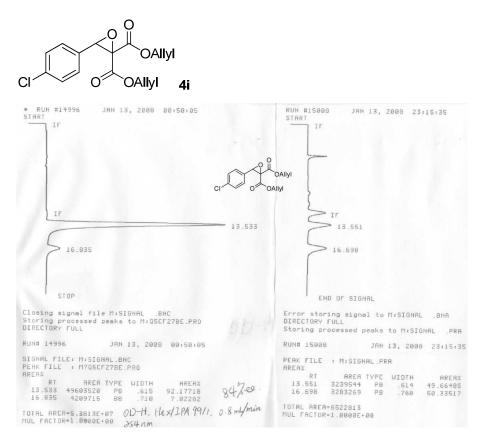
Sorted By		: Signal				
Multiplier	:	1.0000				
Dilution		:	1.0	000		
Use Multiplier	5	Dilution	Factor	with	ISTDs	

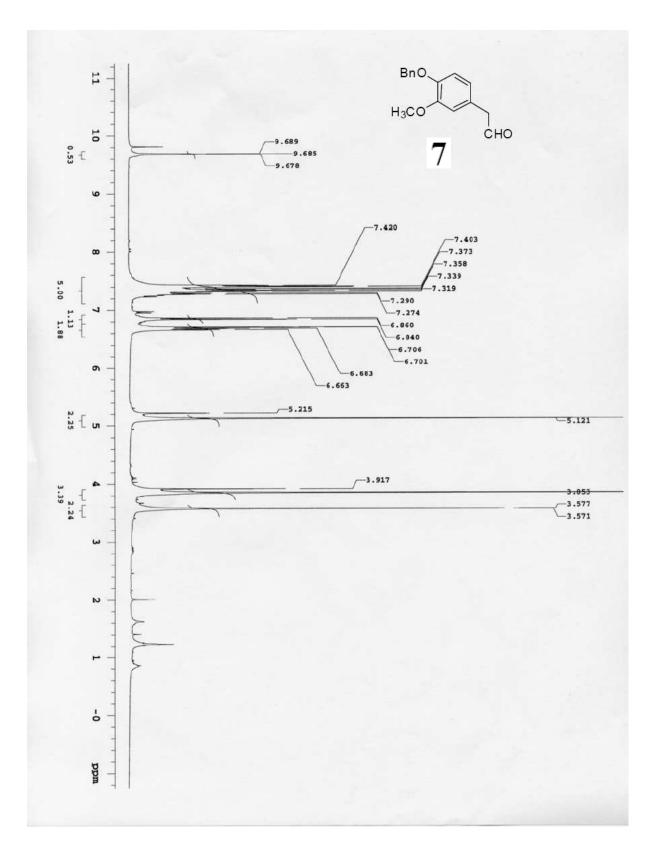
Signal 1: VWD1 A, Wavelength=254 nm

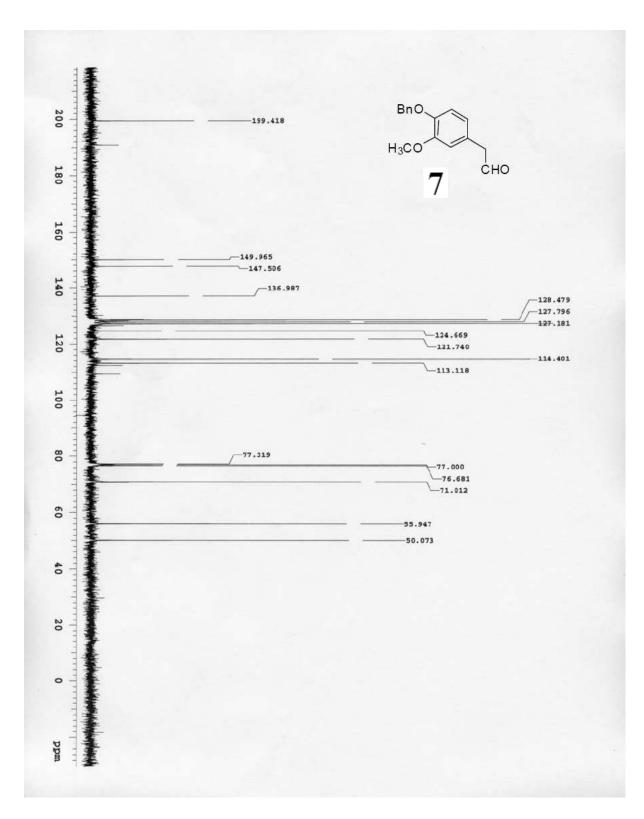
Peak	RetTime	Type	Width	Area		Height		Area	
ŧ	[min]		[min]	mAU	*s	[mAU	1	*	
1	50.165	VV	0.7617	3.354	28e4	680.3	19843	92.8466	
2	54.652	VV	0.8244	2584.	33081	48.3	13493	7.1534	

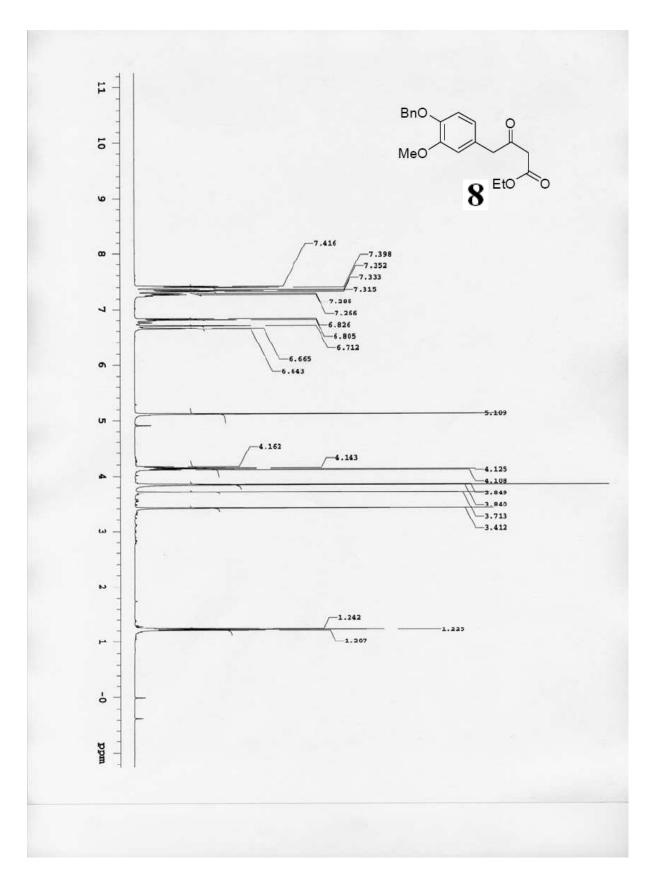
 $^{\circ}$

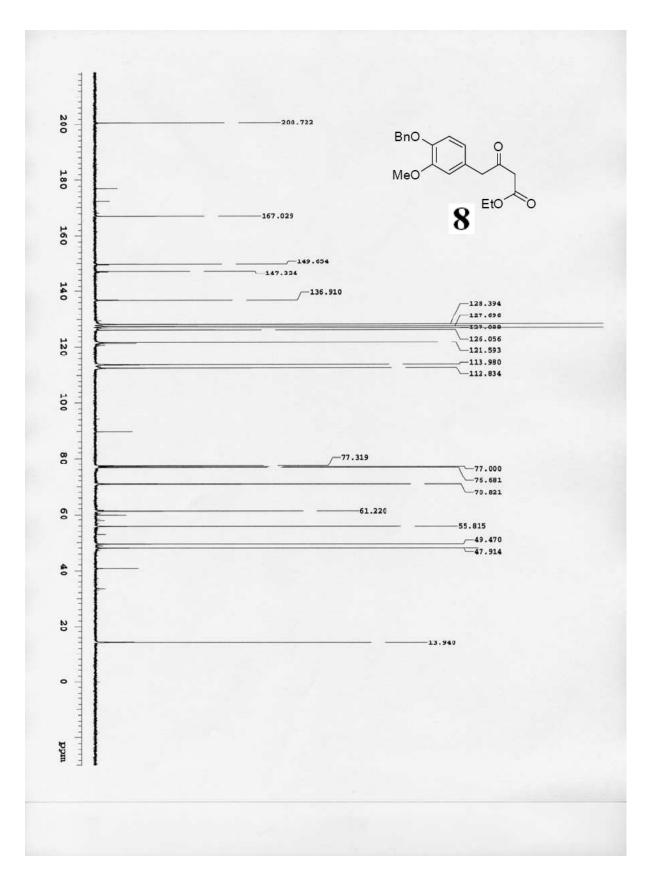


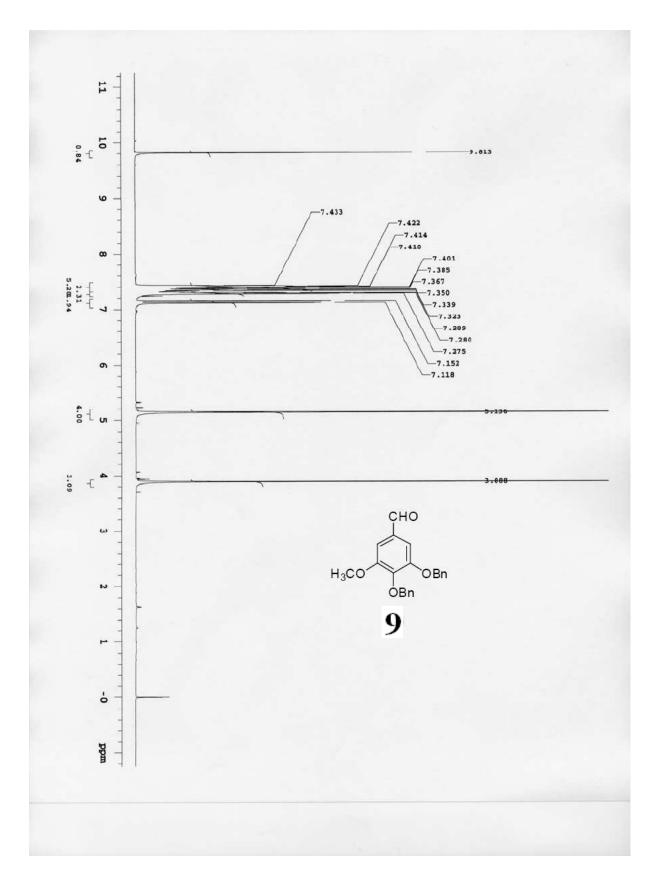


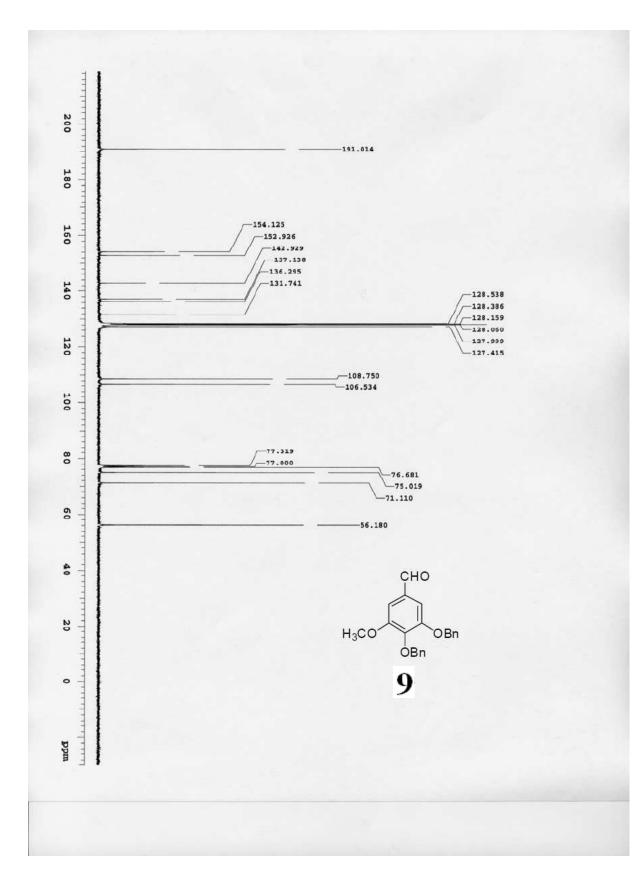


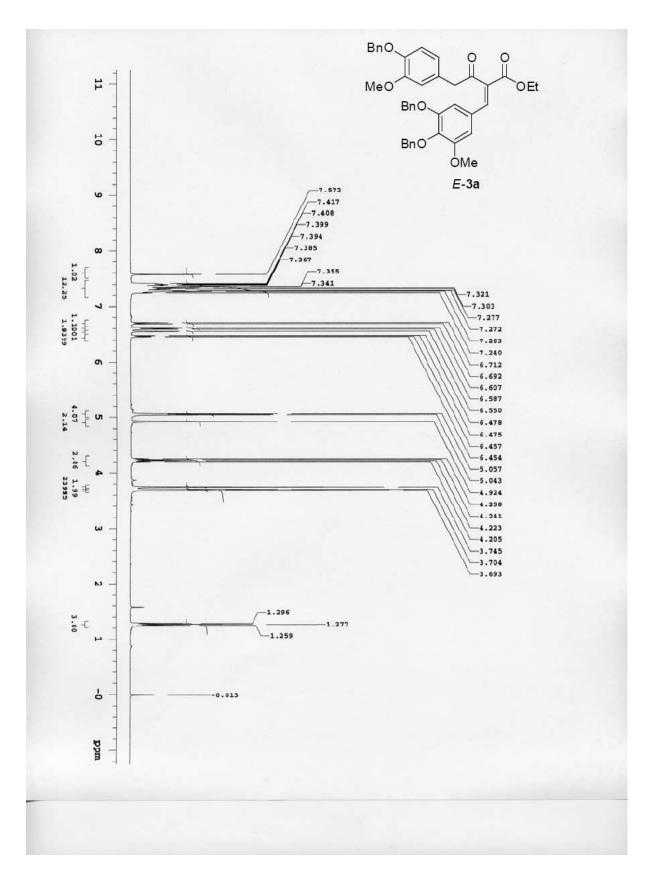


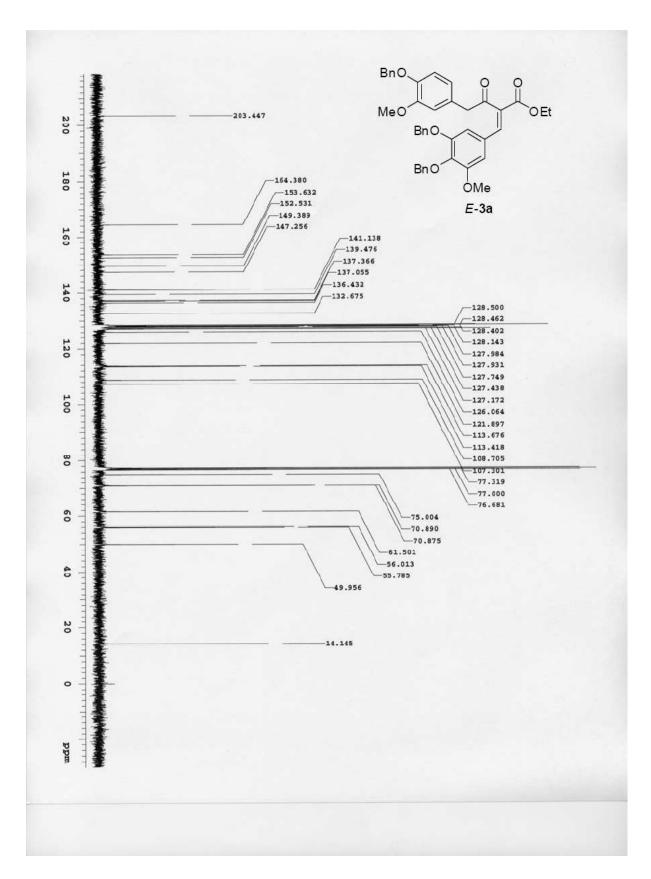




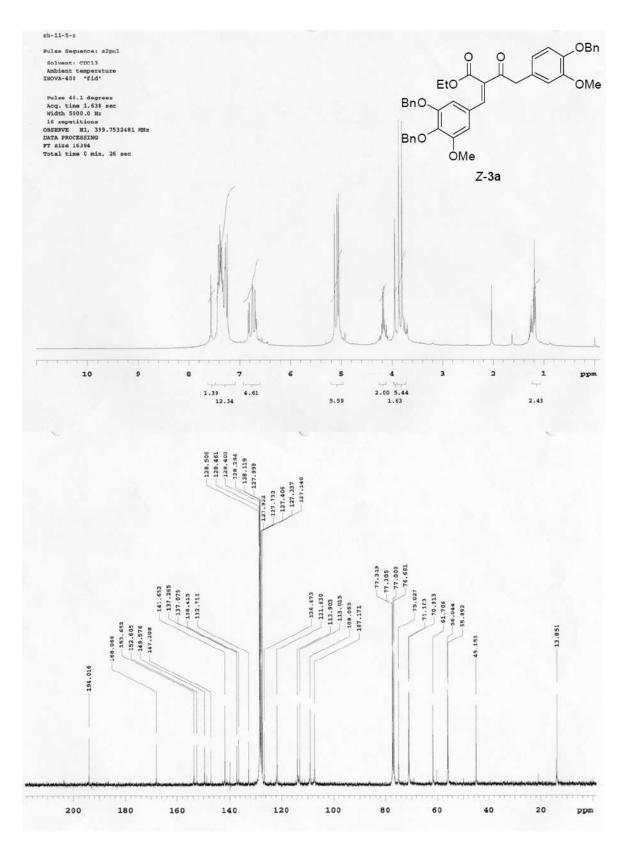


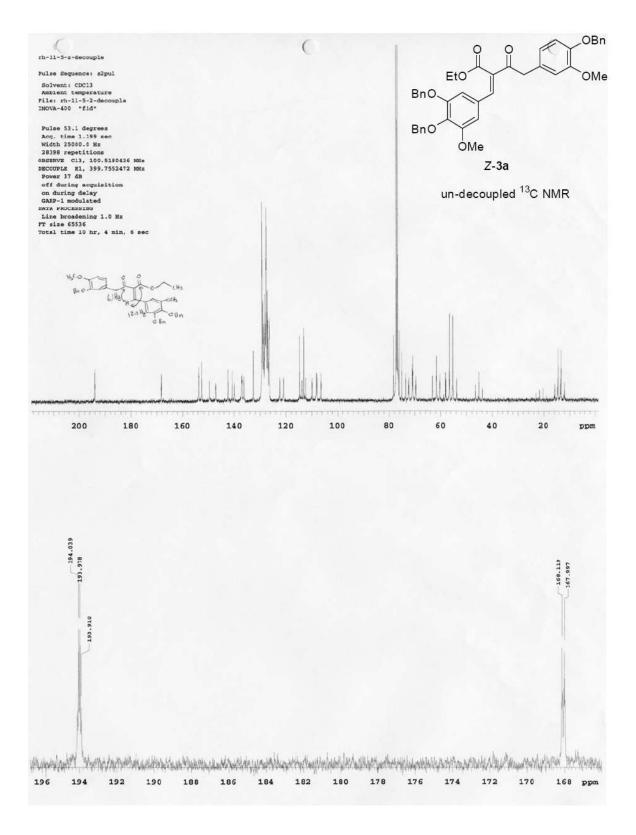




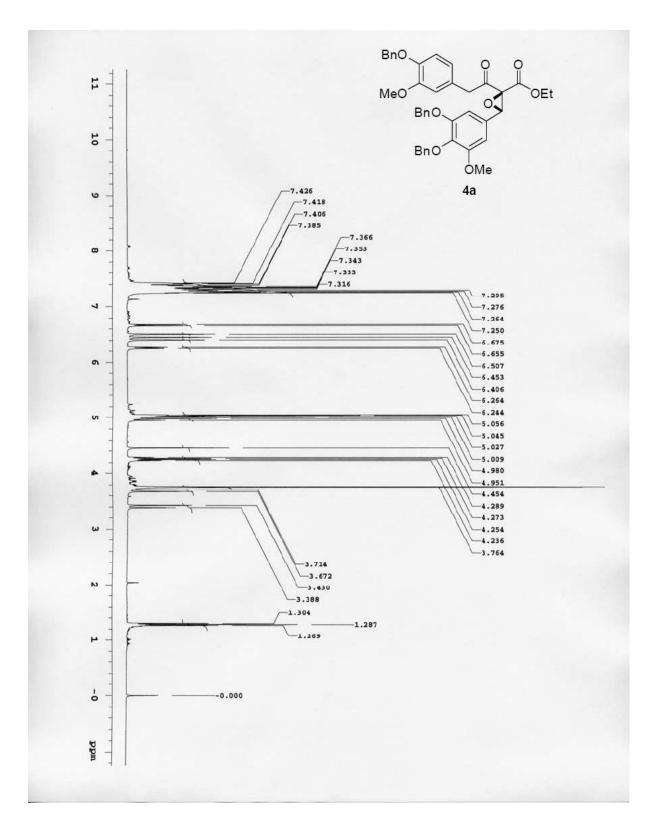


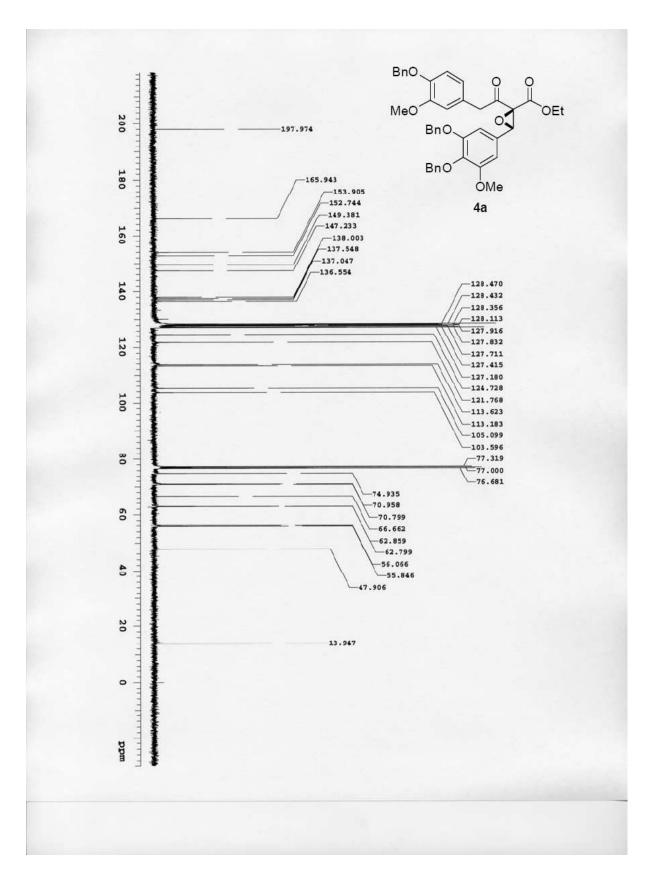
Supporting Information

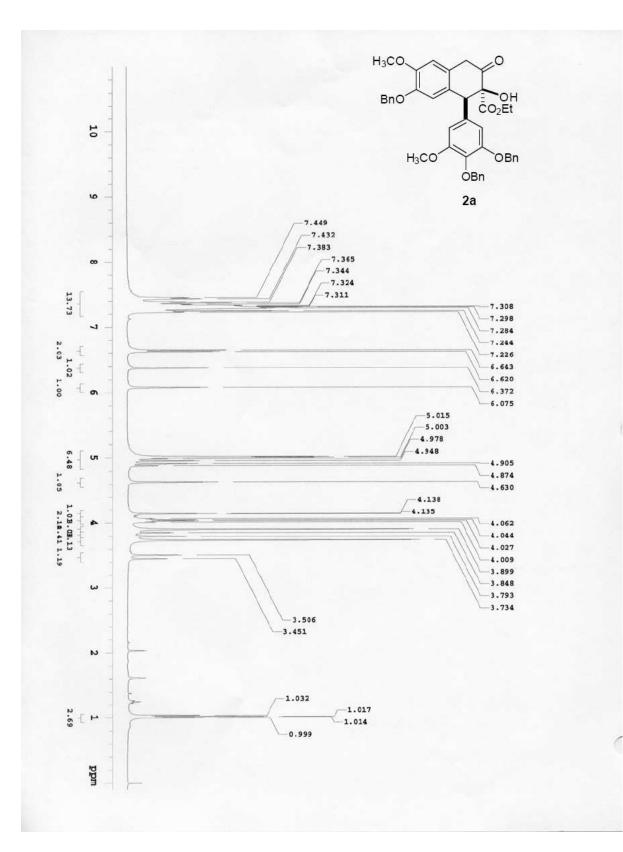


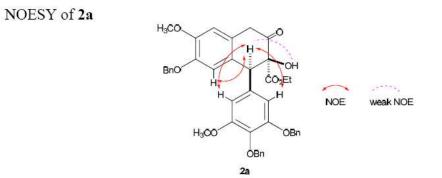


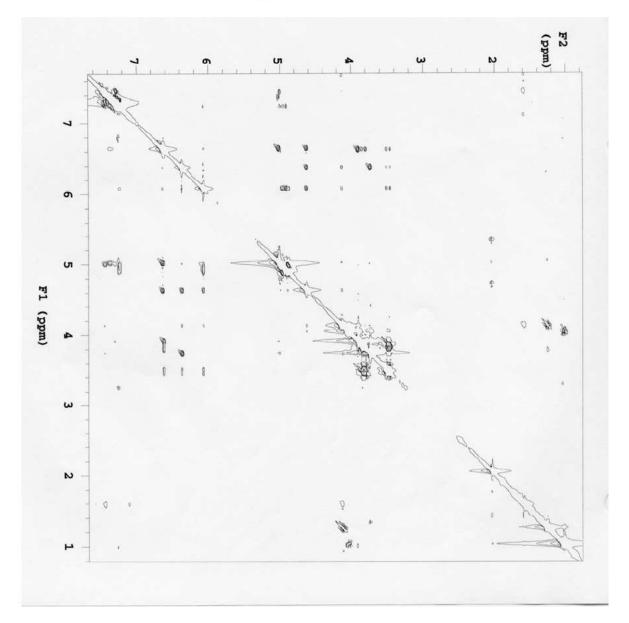
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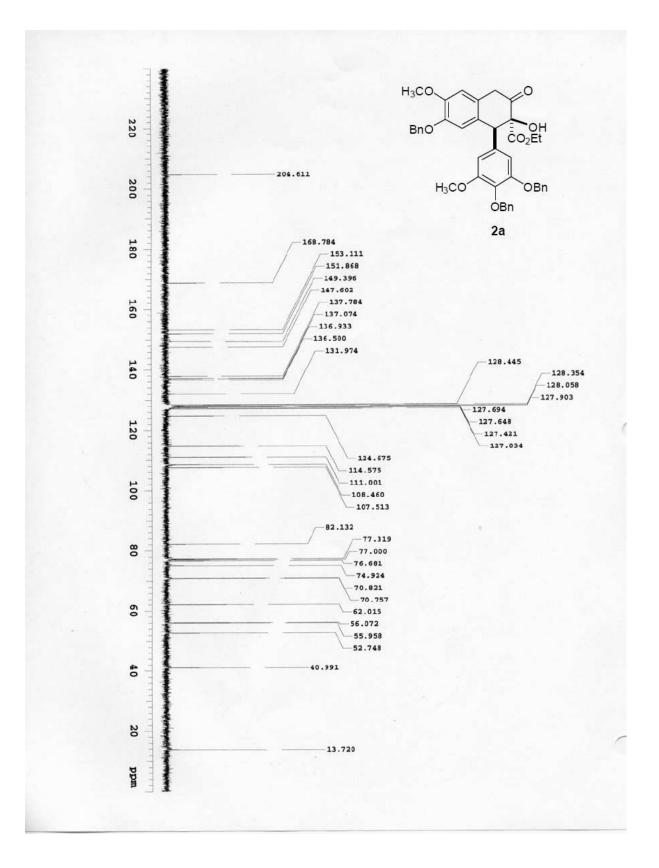


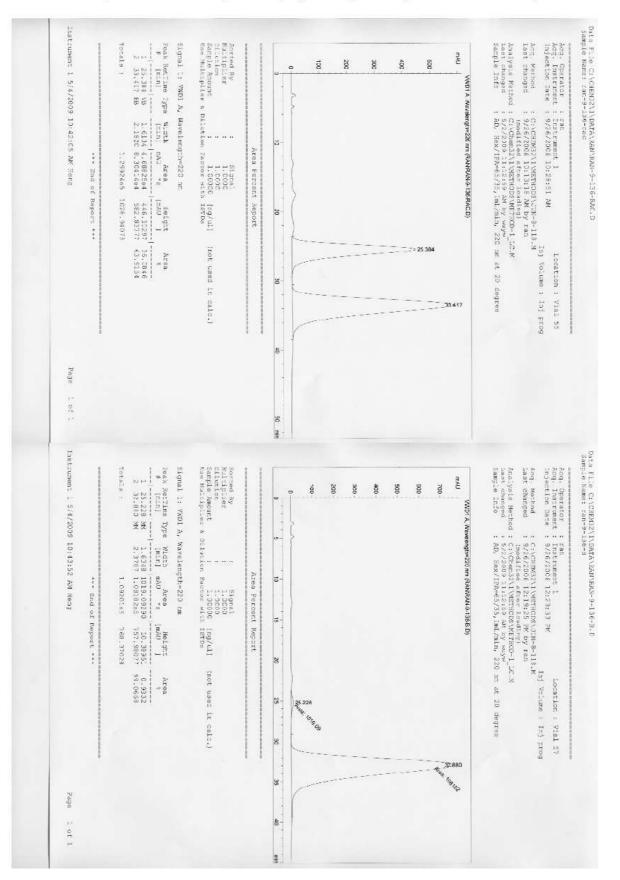




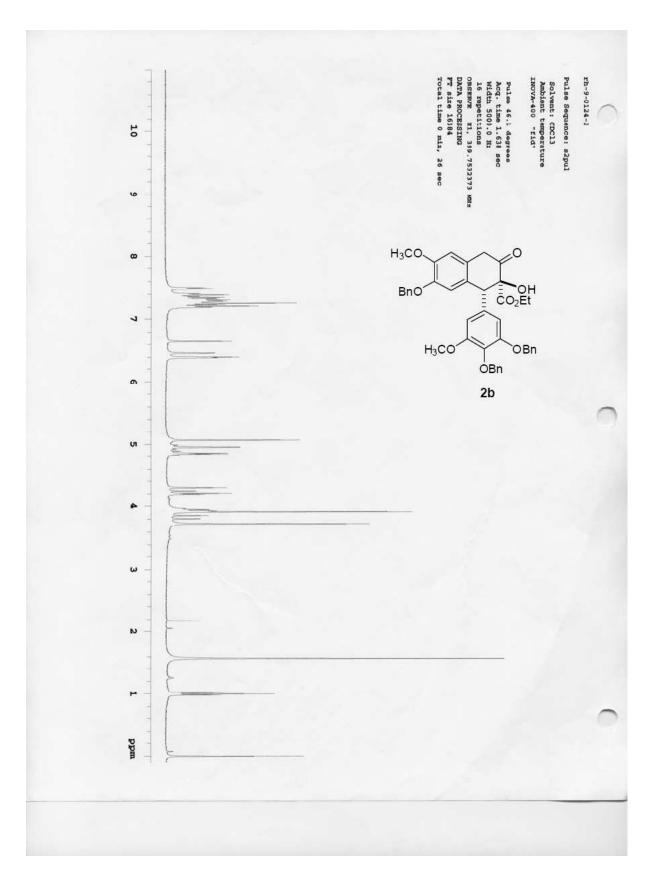


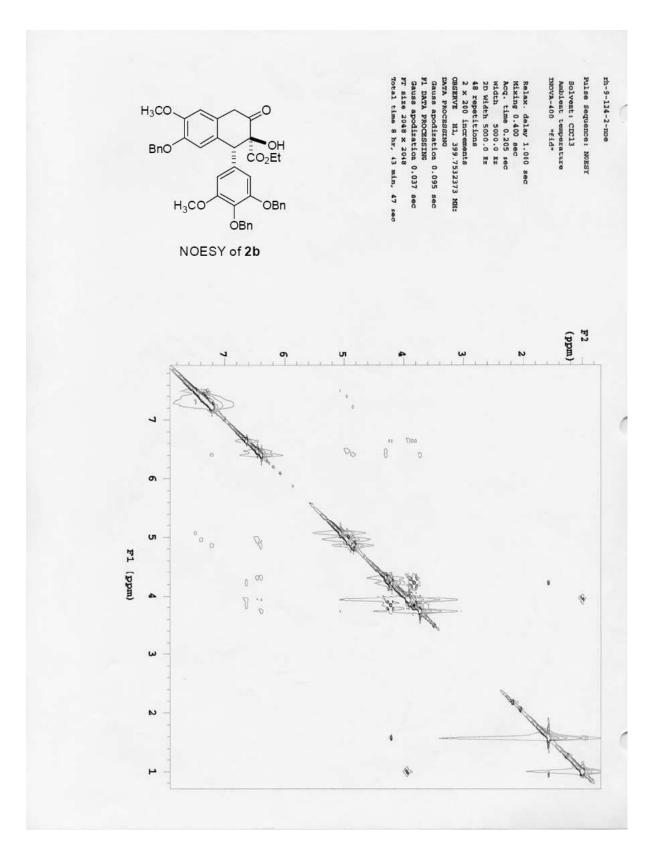


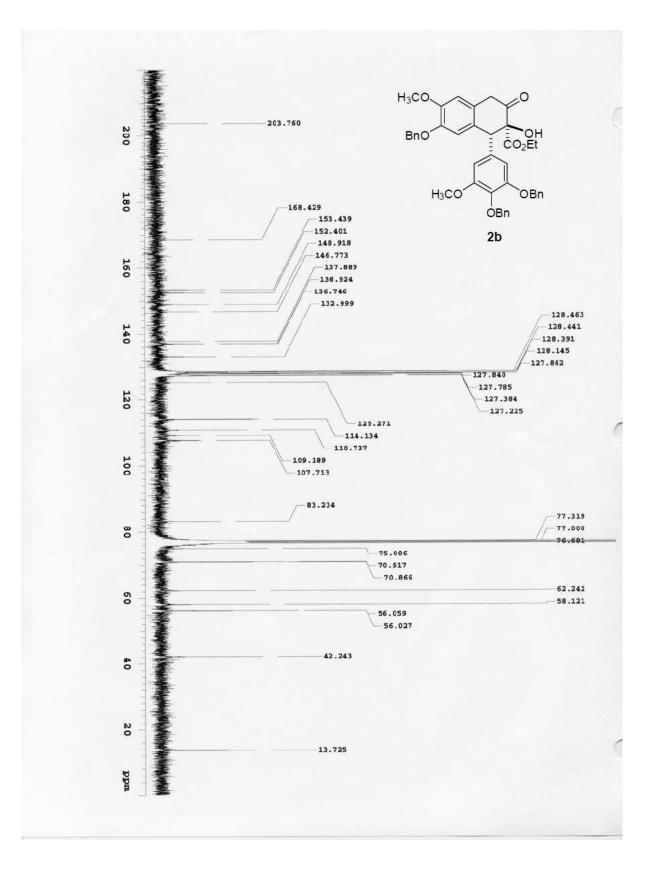


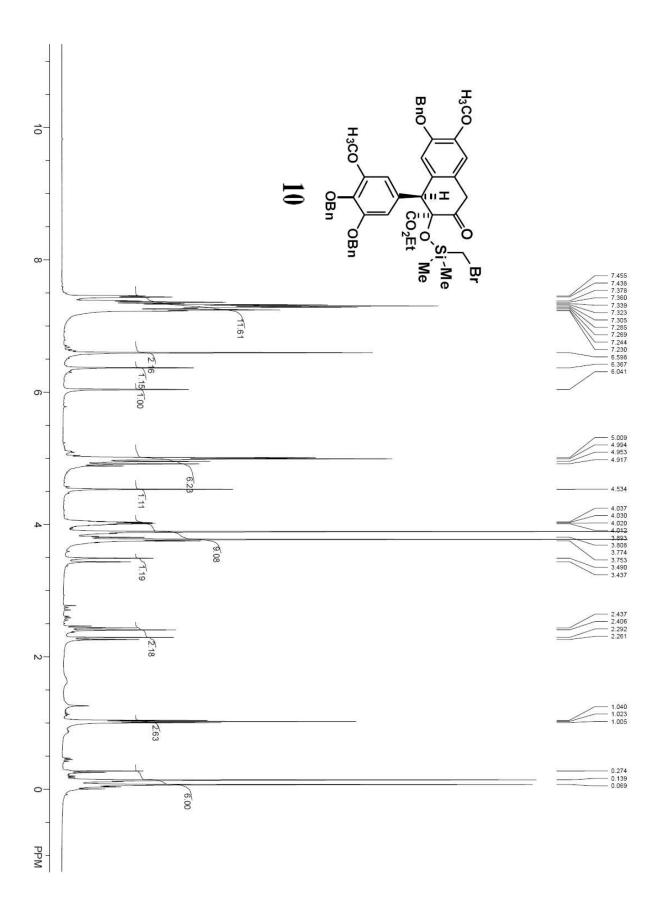


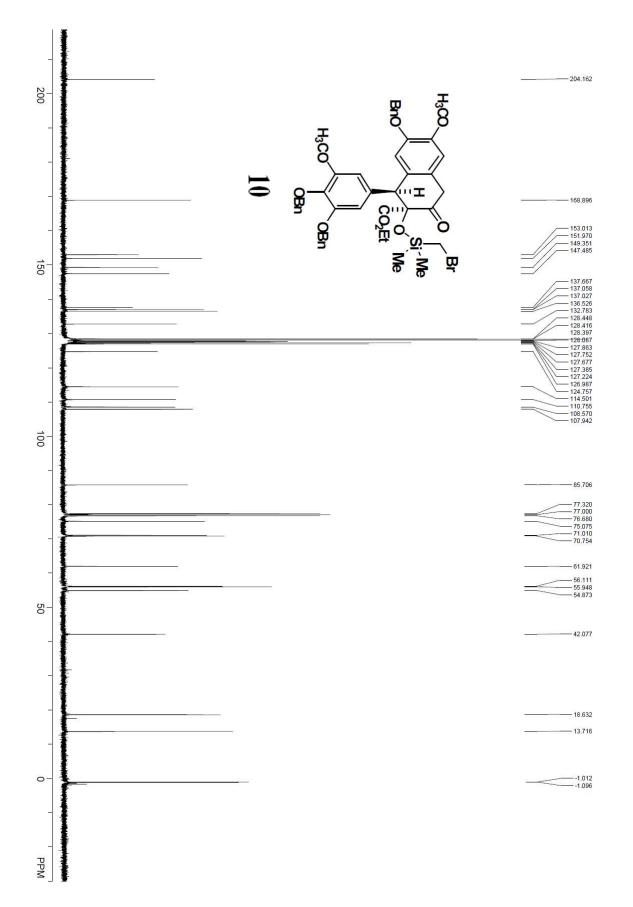
HPLC of 2a: (The racemic sample was prepared by mixing both enantiomers.)

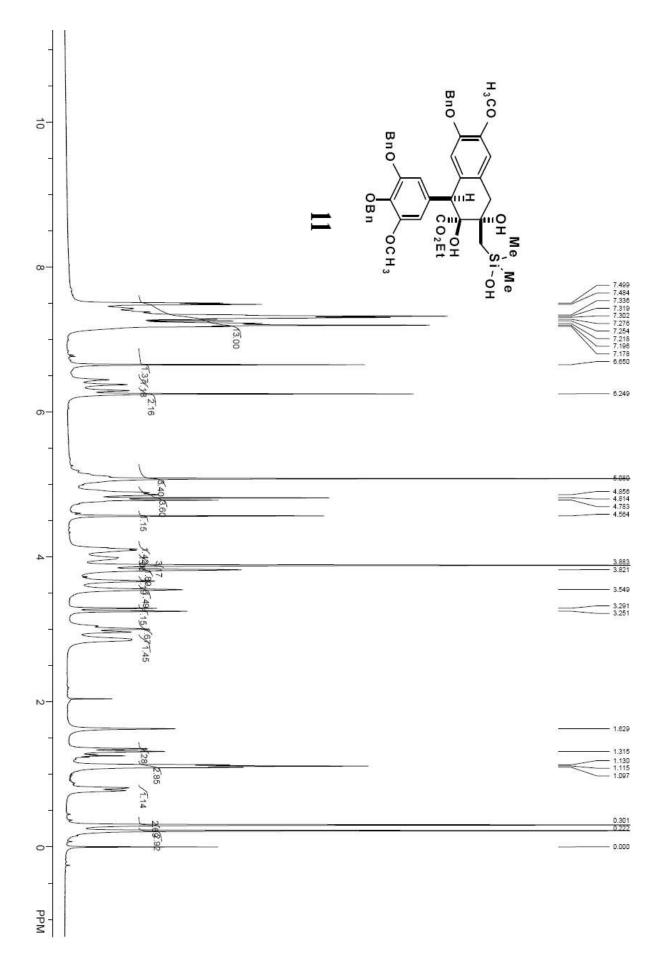


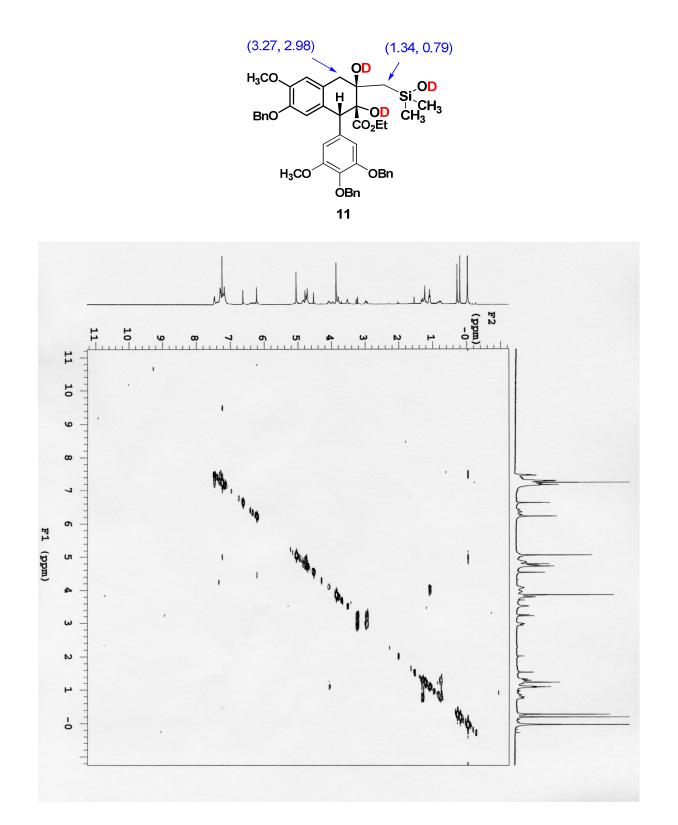




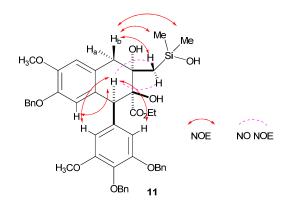


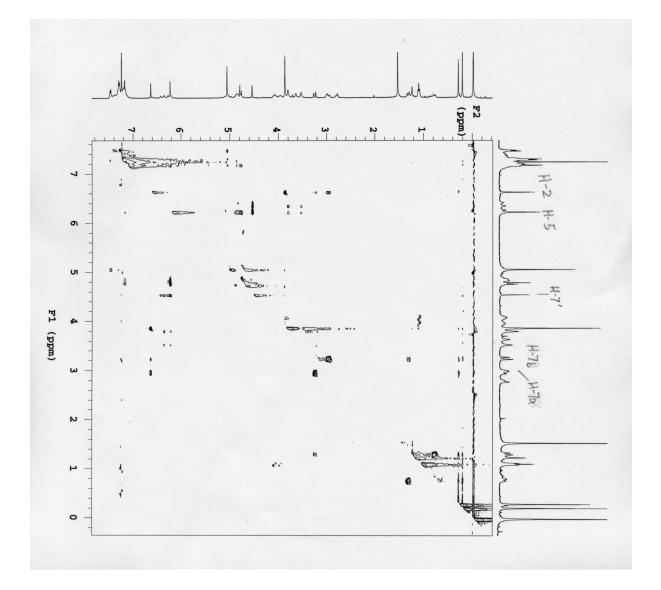


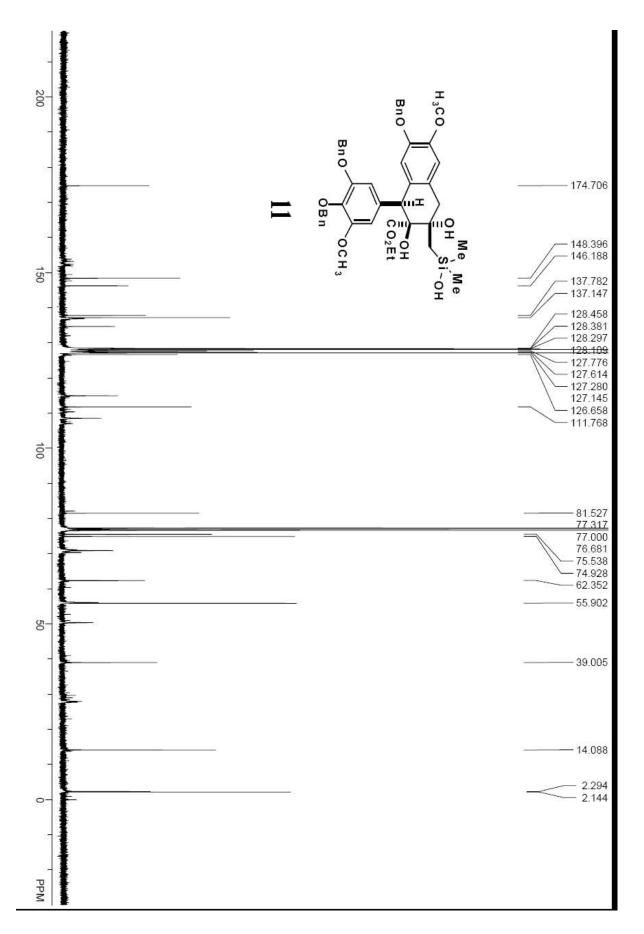


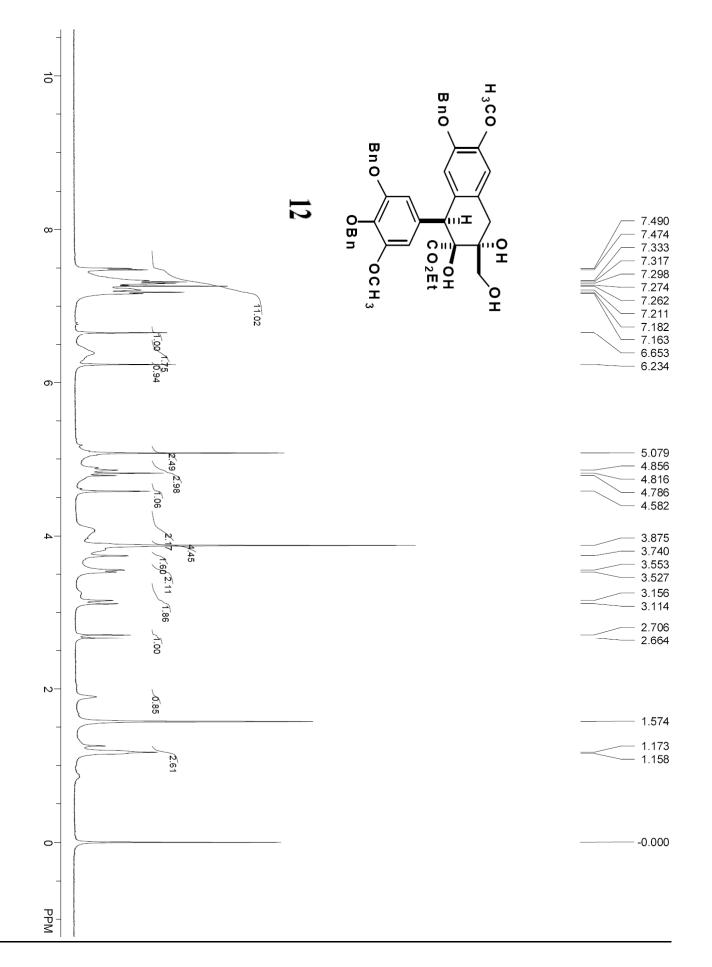


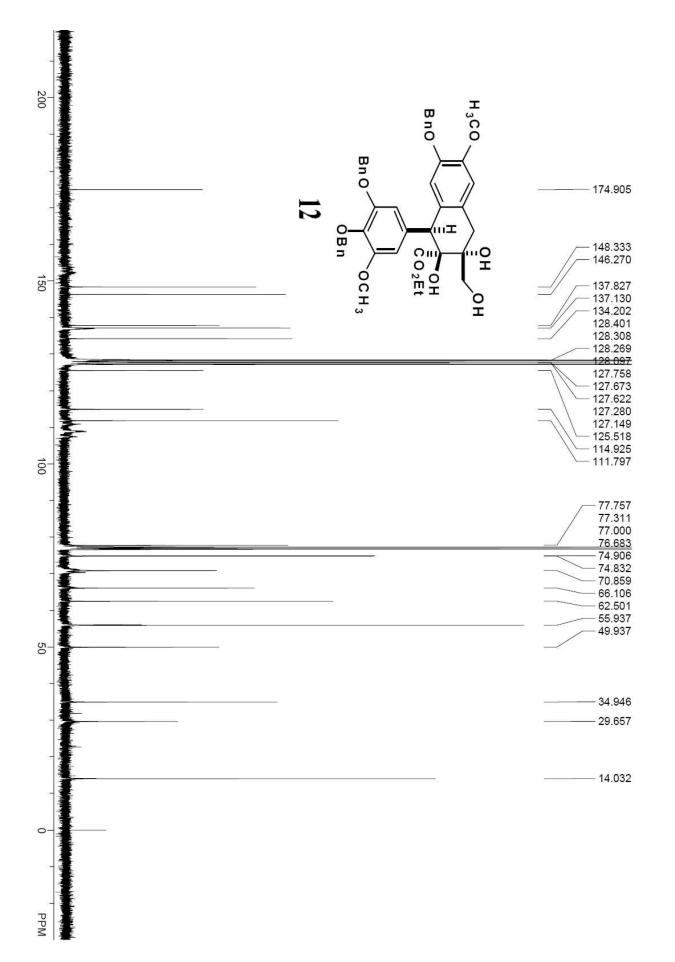
NOESY of 11

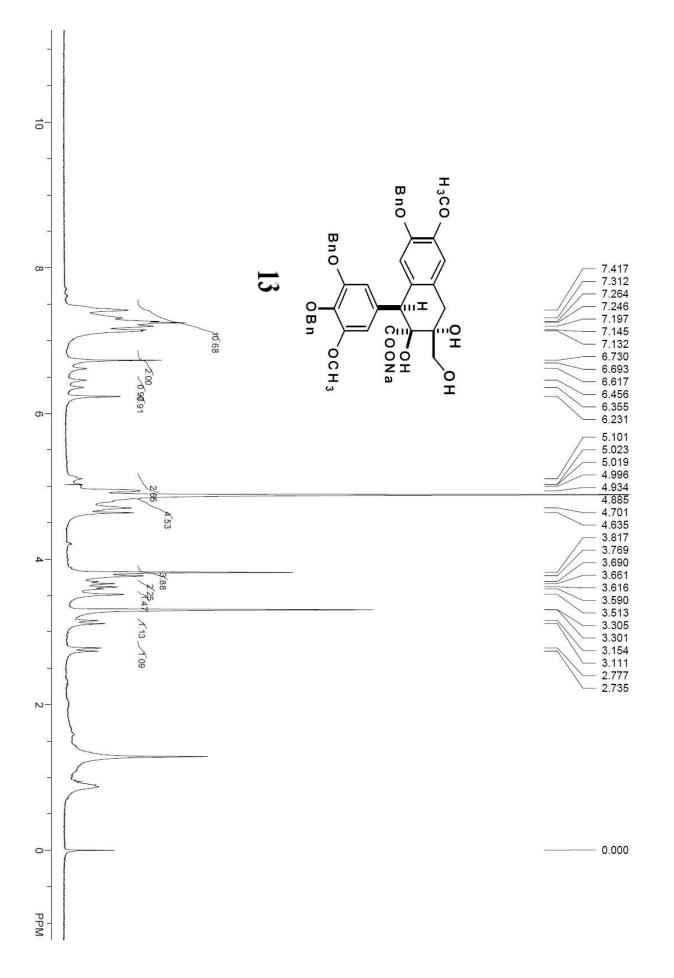


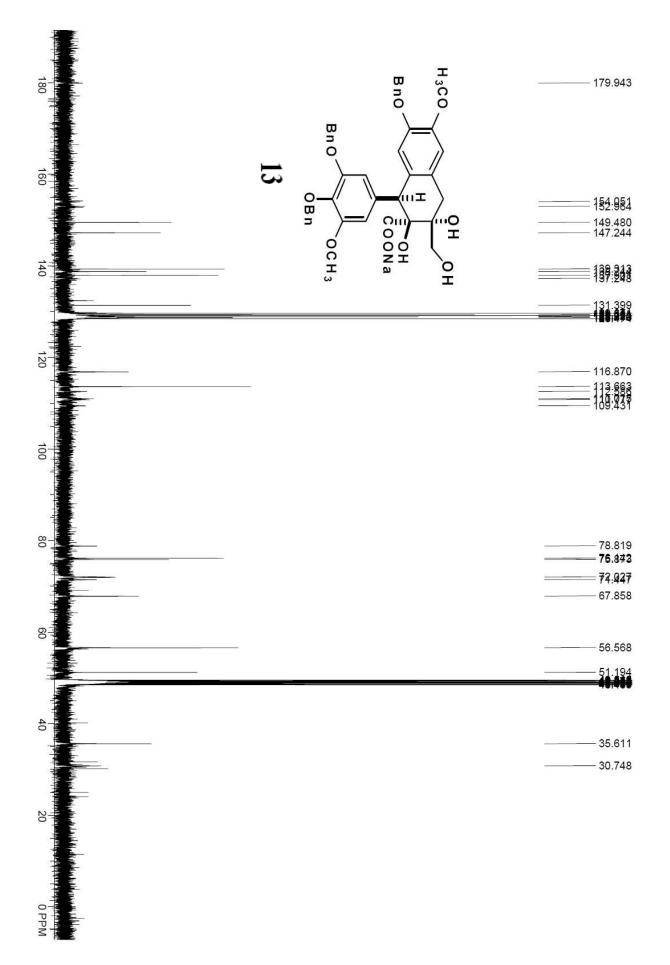




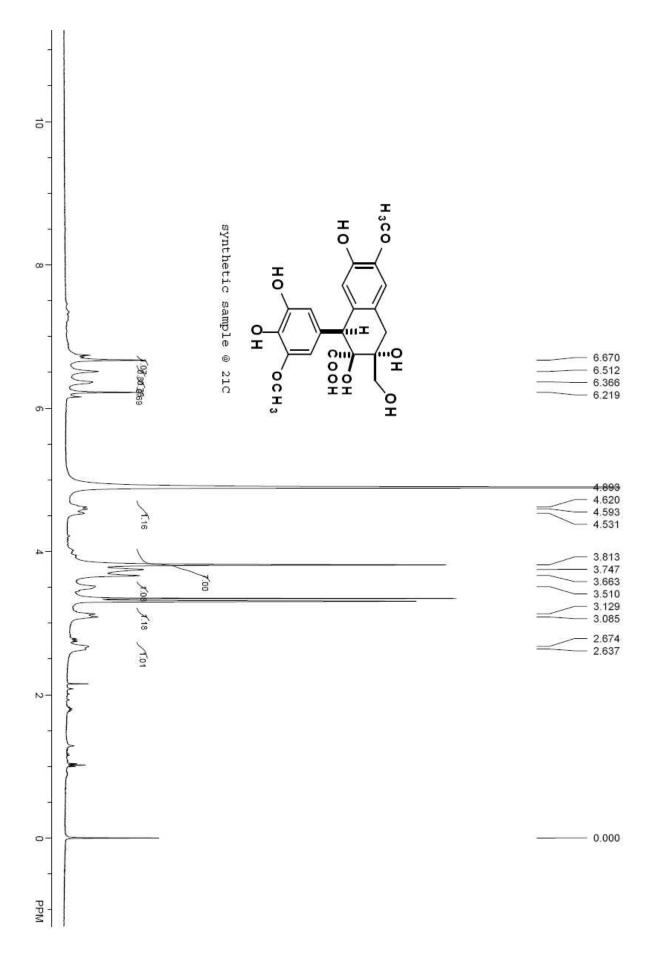




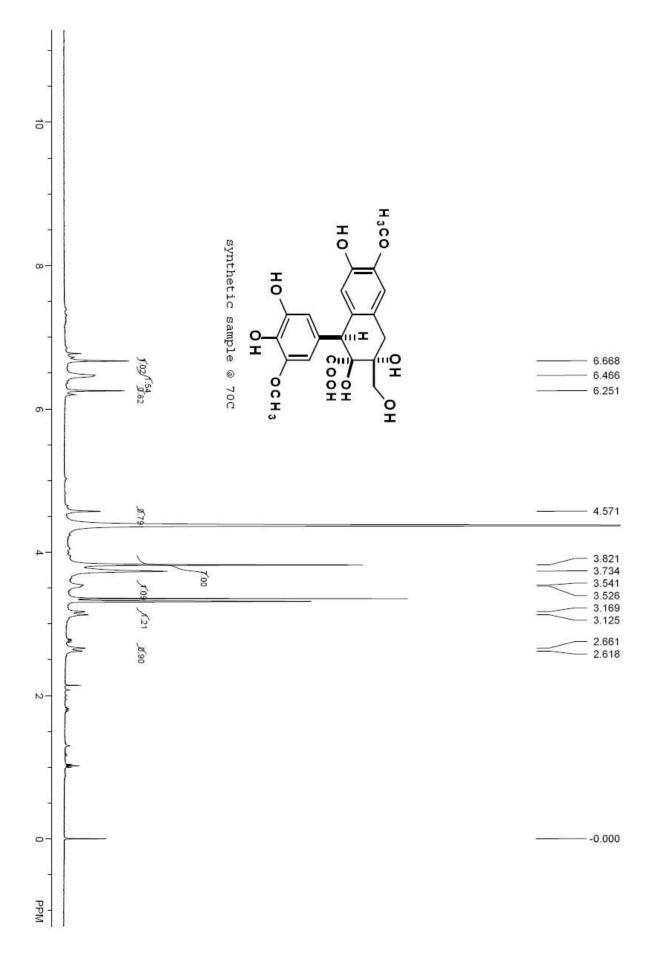


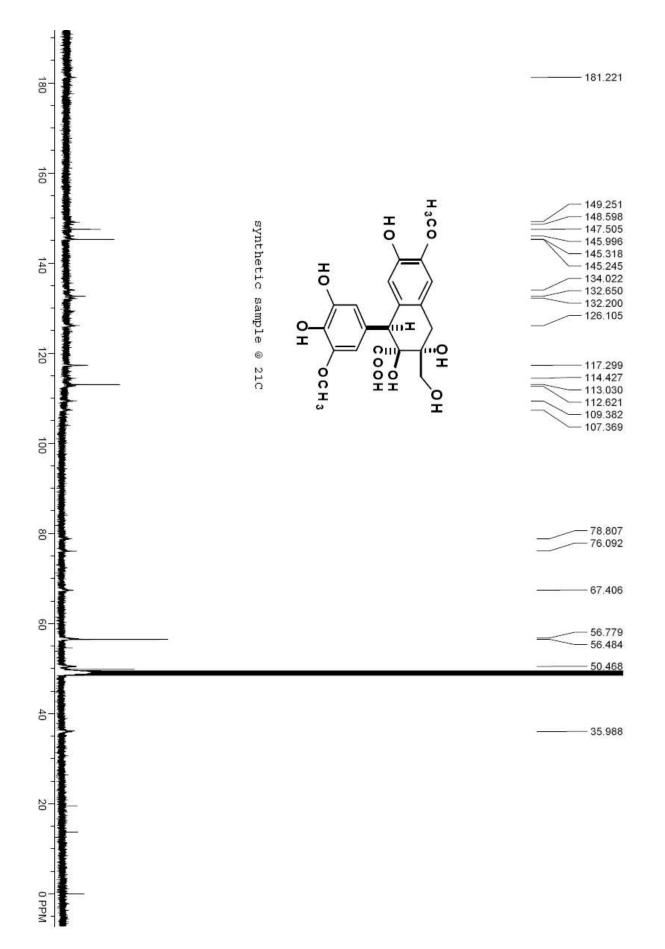


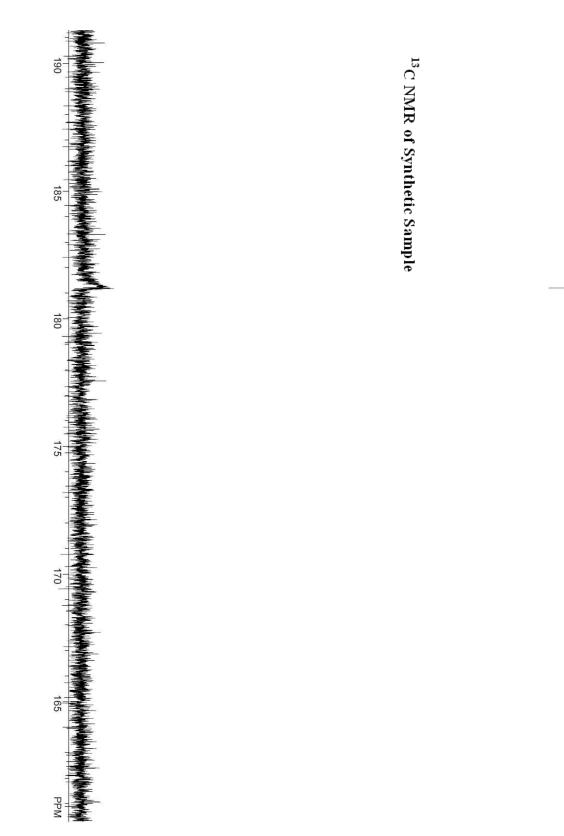
Supporting Information



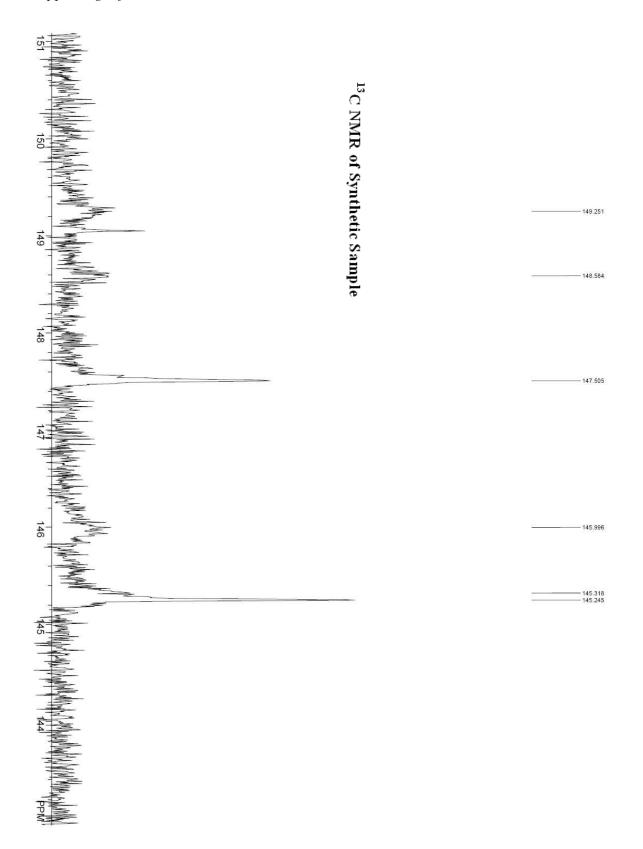
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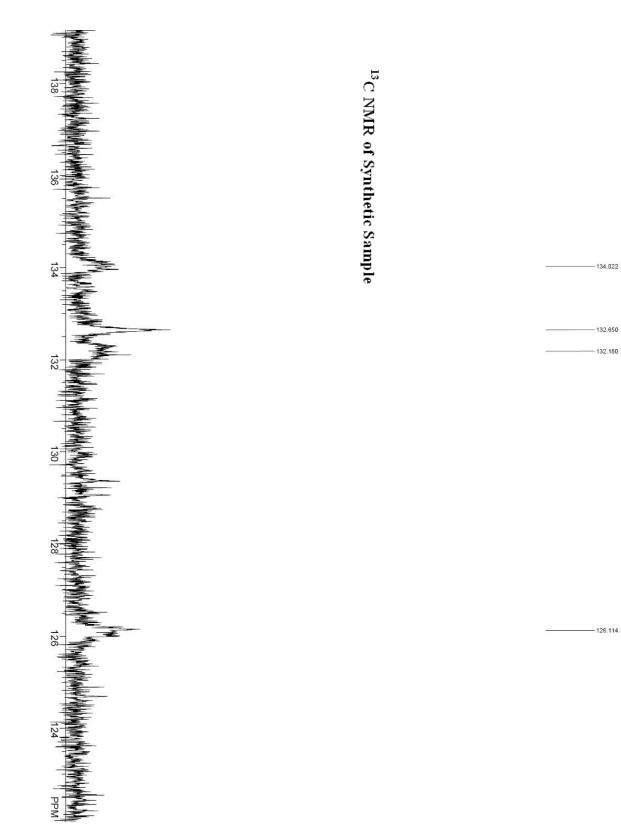


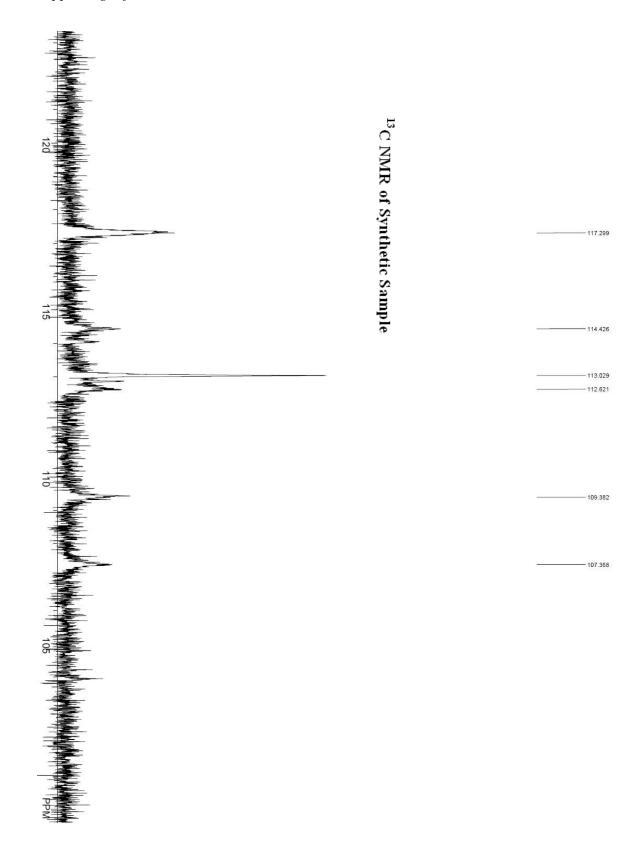


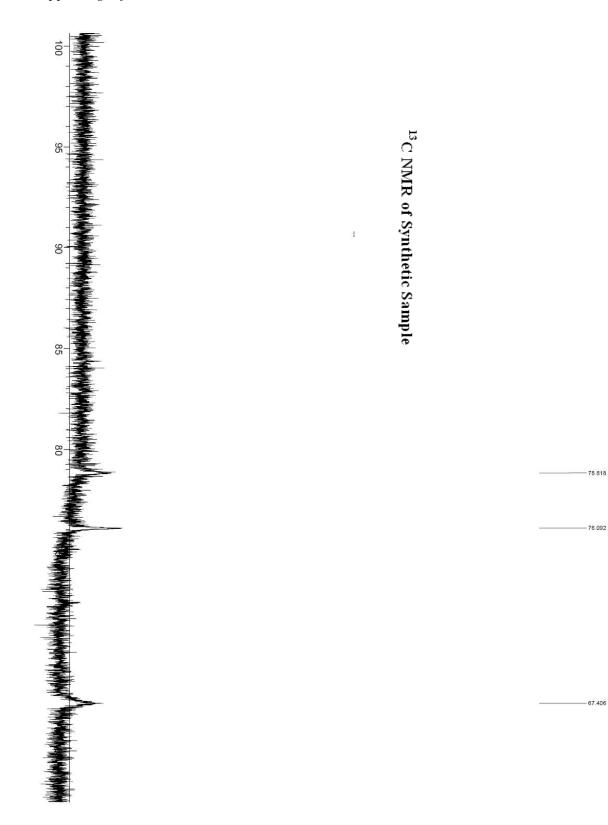


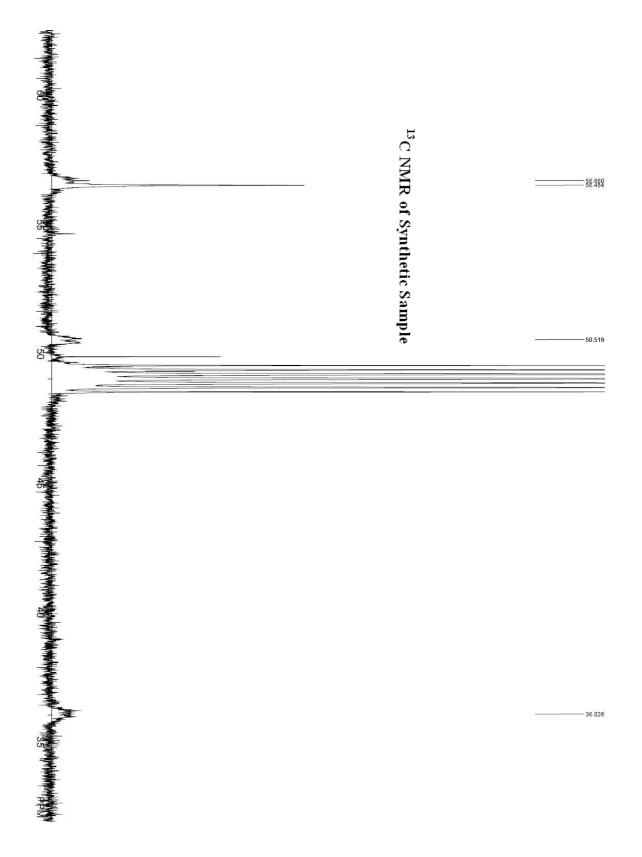
- 181.197

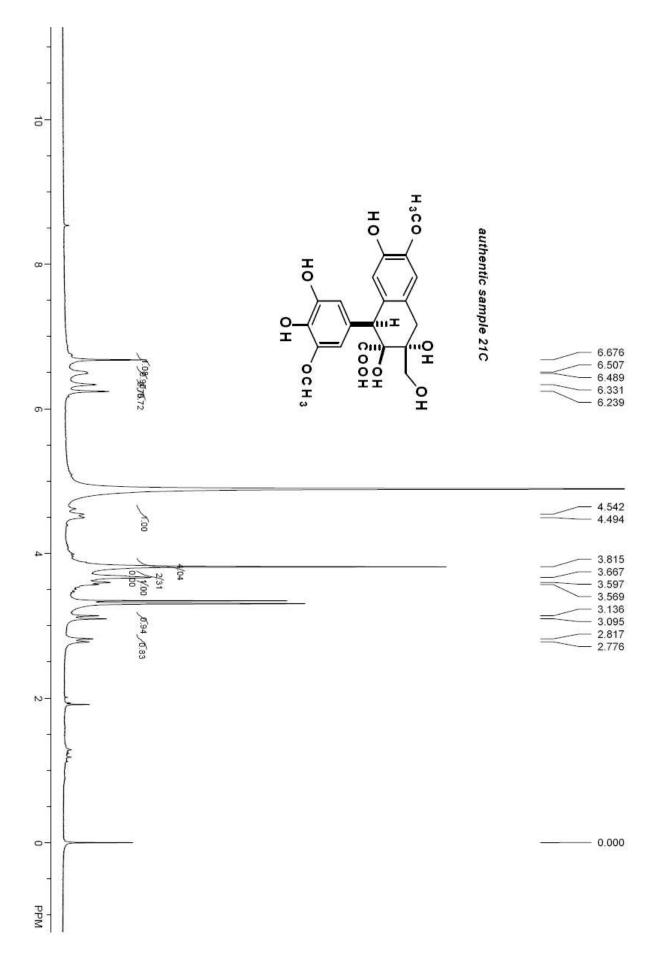




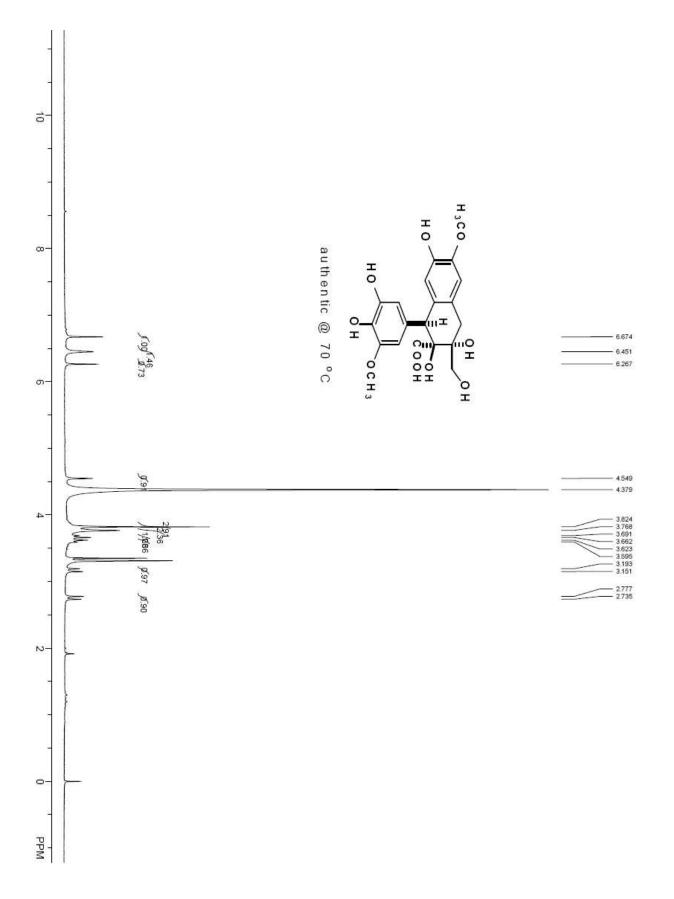


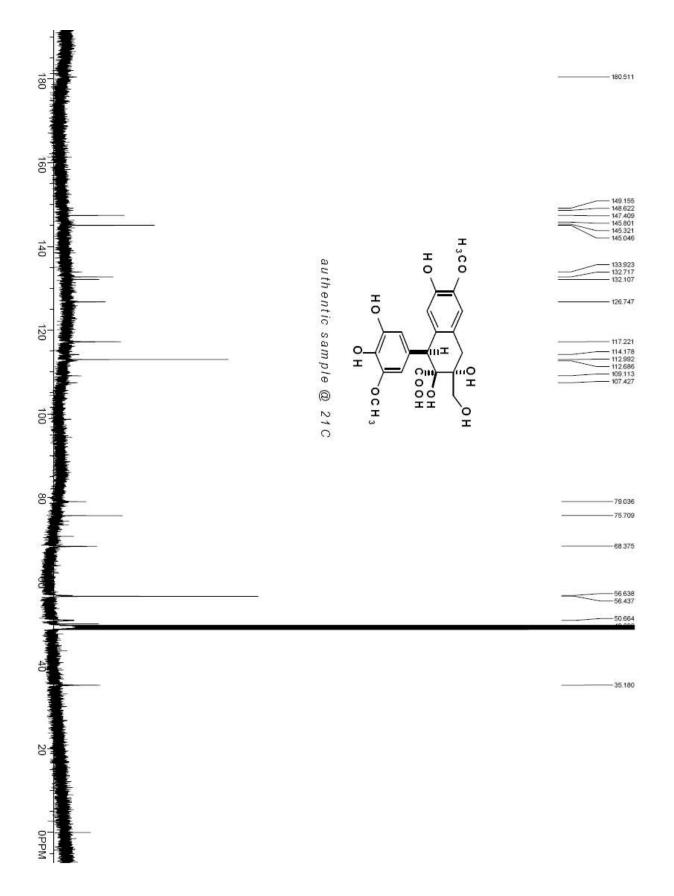


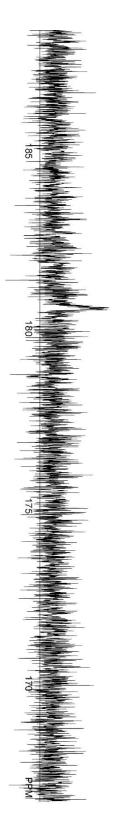




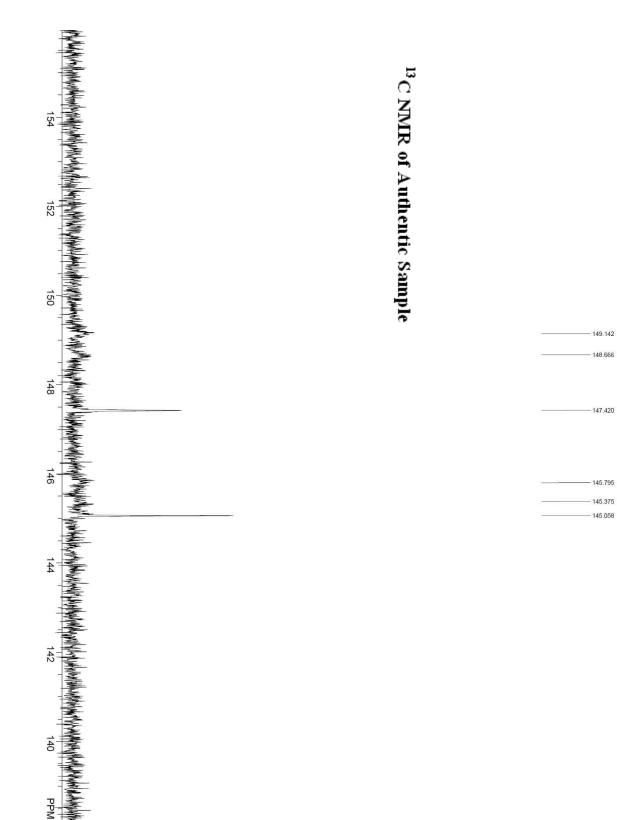
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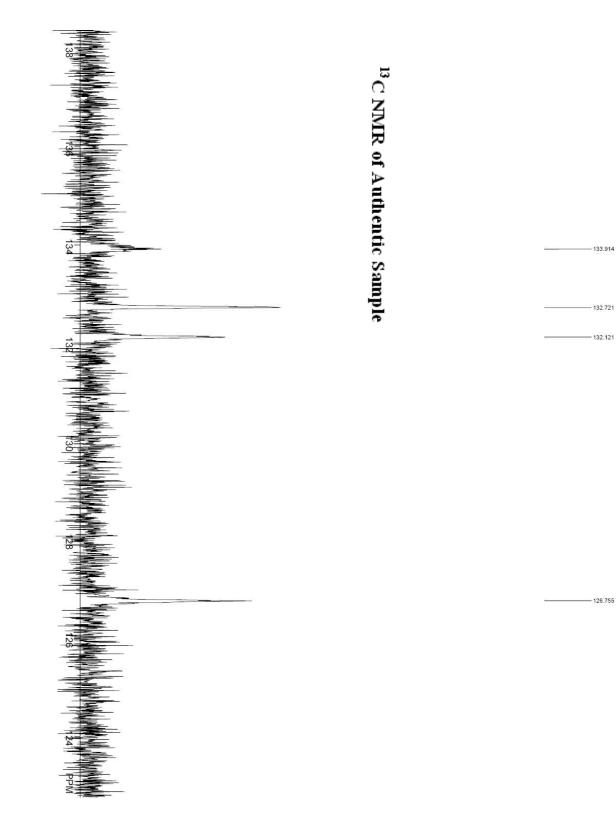
¹³C NMR of Authentic Sample

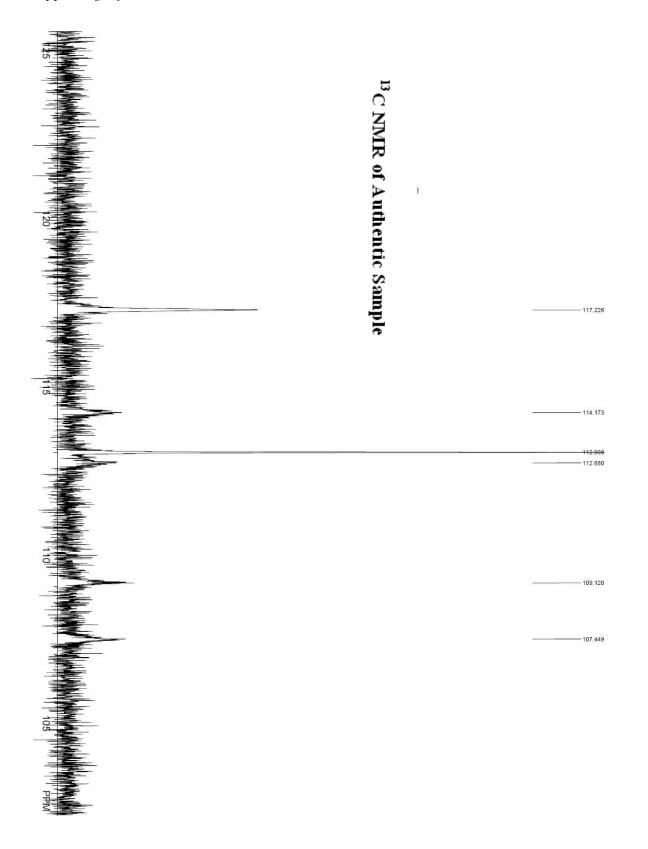


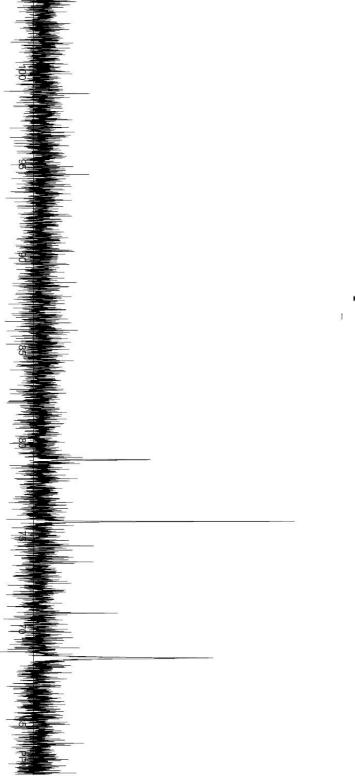
148.666

- 147.420

145.795 145.375 145.058



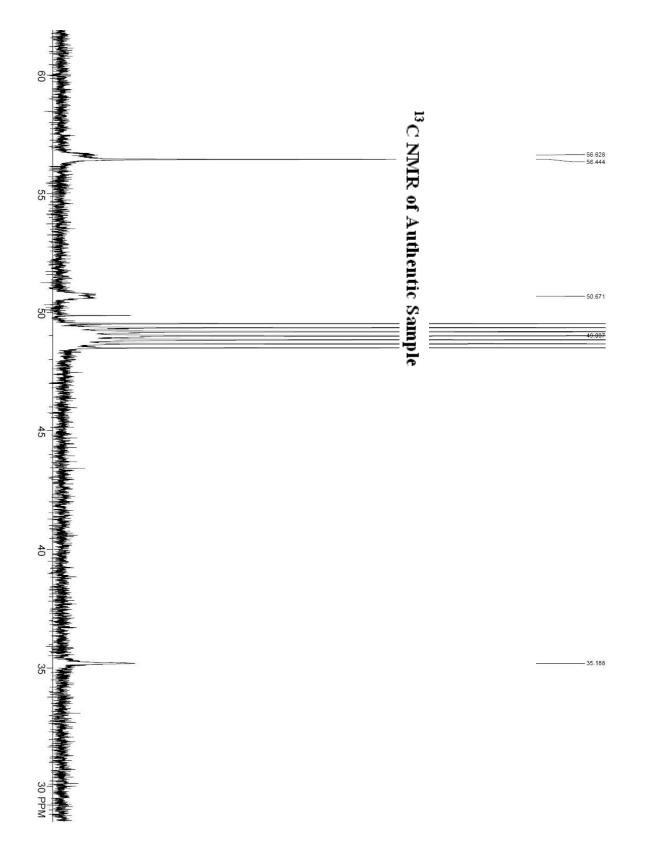


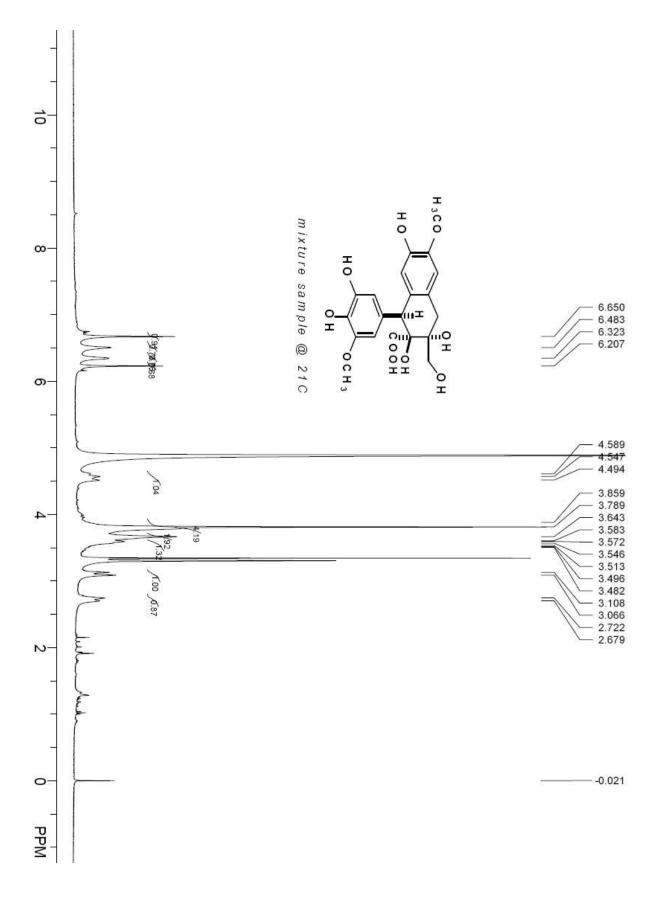


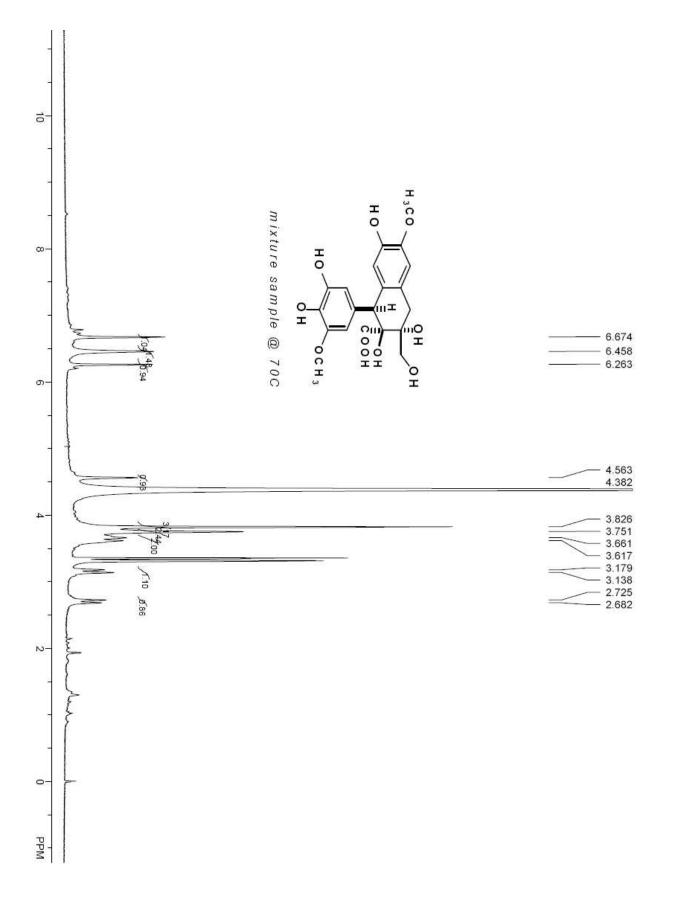
¹³C NMR of Authentic Sample

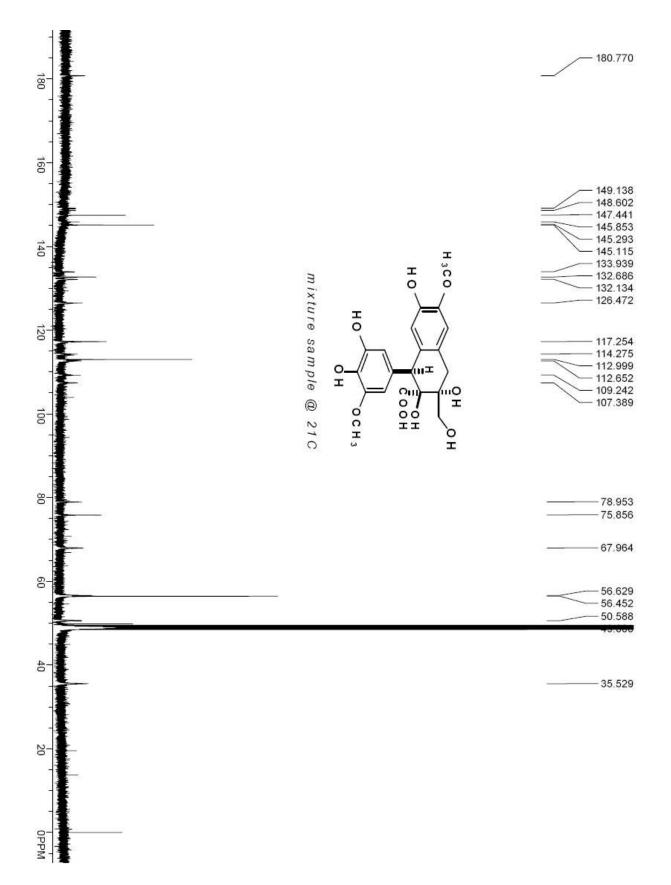
- 79.044

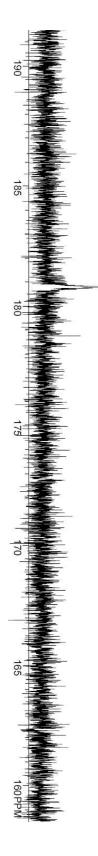
75.716





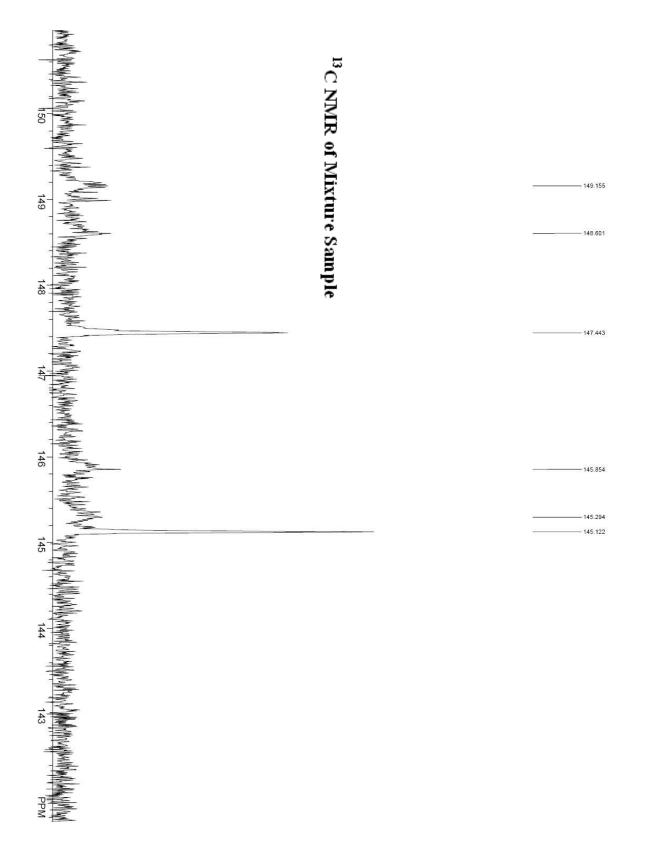


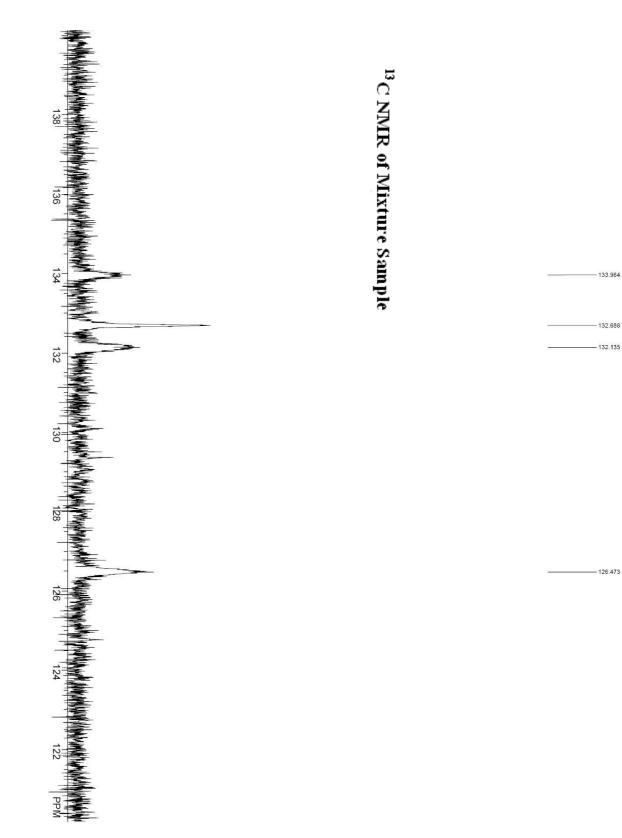


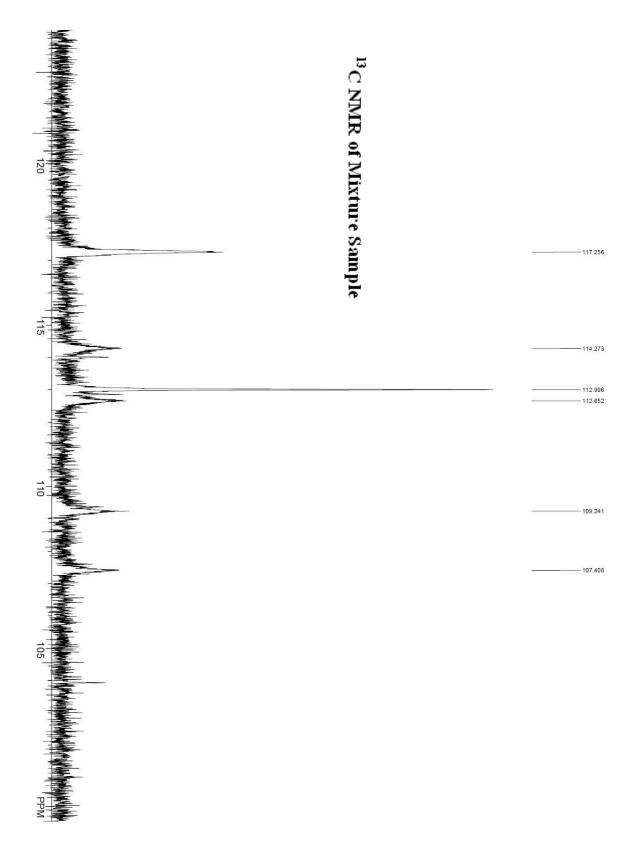


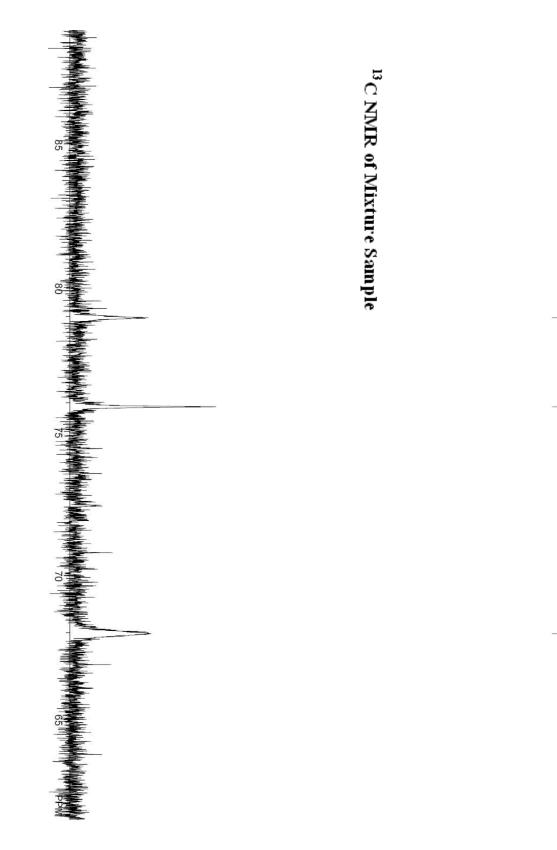


I



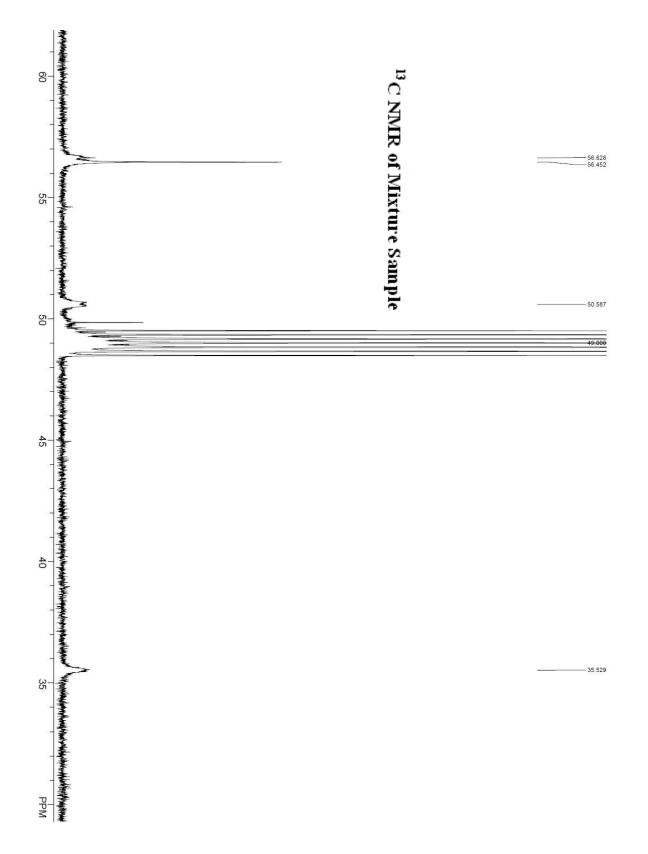


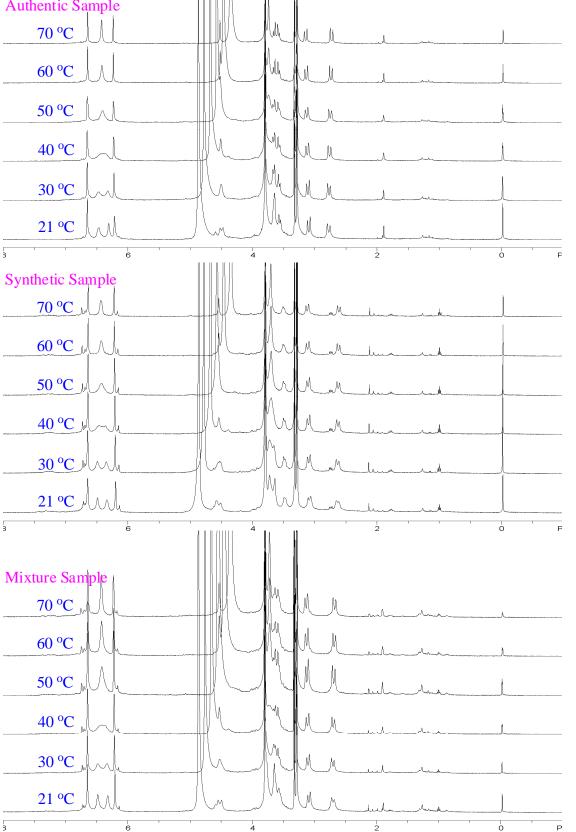




78.954

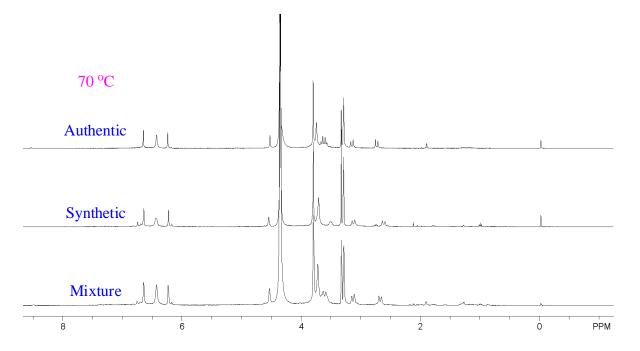
75.857

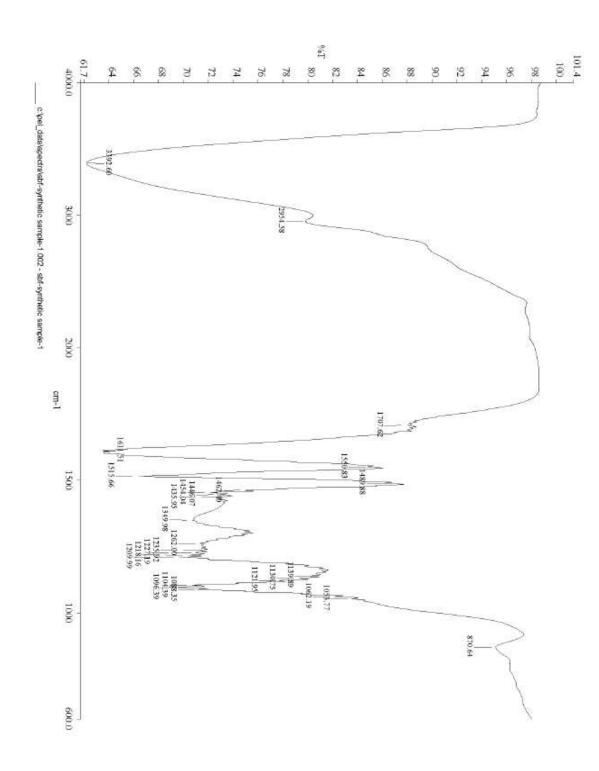


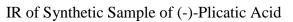


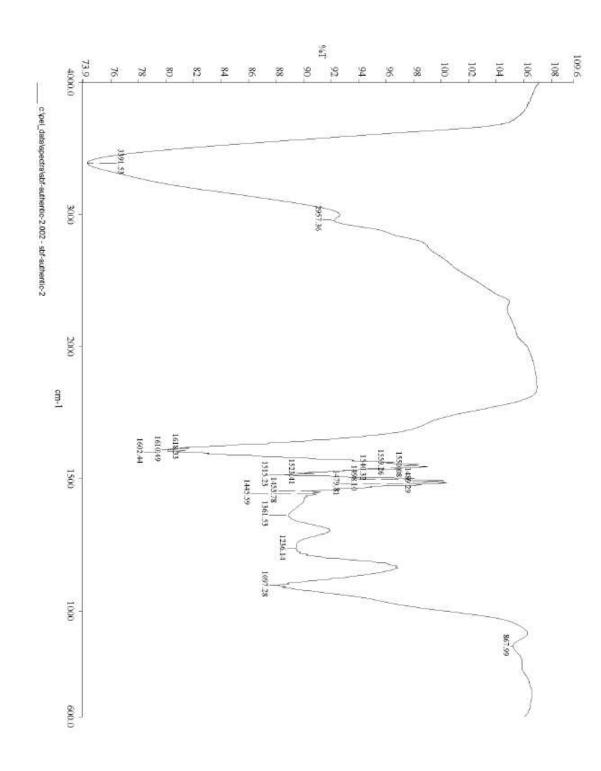
VT-¹H NMR of Authentic Sample, Synthetic Sample and Mixture Sample of (-)-Plicatic Acid Authentic Sample

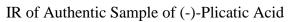
 ^1H NMR Spectra of Authentic Sample, Synthetic Sample and Mixture Sample of (-)-Plicatic Acid (1) at 70 $^{\rm o}\text{C}$



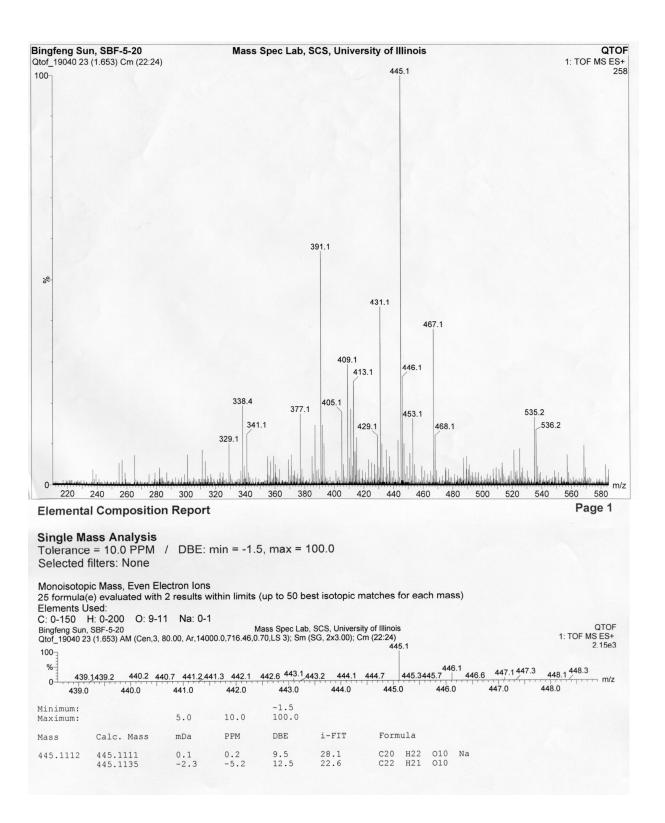




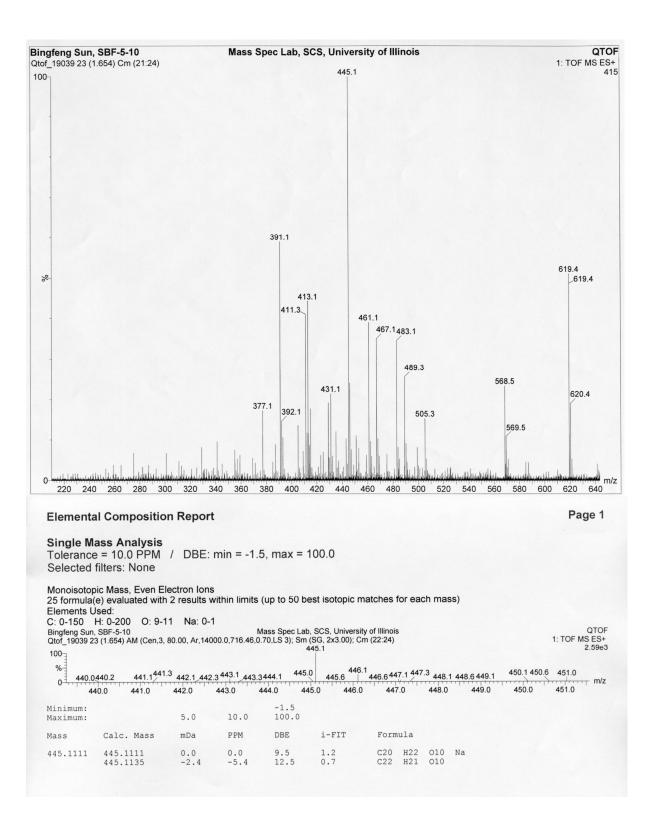


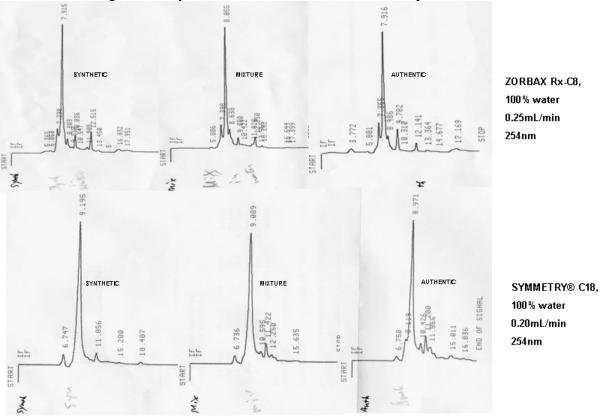


MS and HRMS of Synthetic Sample of (-)-Plicatic Acid



MS and HRMS of Authentic Sample of (-)-Plicatic Acid





HPLC Chromatograms of Synthetic, Mixture and Authentic Sample on Two Columns

CD Spectra of Synthetic and Authentic Sample of (-)-Plicatic Acid

