

Supporting Information

Human Manganese Superoxide Dismutase Tyrosine 34

Contribution to Structure and Catalysis

J. Jefferson P. Perry,^{†,‡} Amy S. Hearn[§], Diane E. Cabelli,[‡] Harry S. Nick,[§]

John A. Tainer,^{†,‡} and David N. Silverman^{*}

[†]Department of Molecular Biology, The Scripps Research Institute, La Jolla, CA 92037

[‡]School of Biotechnology, Amrita University, Kollam, Kerala 690525, India

[‡]Life Sciences Division, Department of Molecular Biology, Lawrence Berkeley National Laboratory, Berkeley, CA, 94720

[‡]Department of Chemistry, Brookhaven National Laboratory, Upton, New York 11973

[§]Department of Neuroscience, University of Florida, Gainesville, FL 32510

^{*}Department of Pharmacology, University of Florida, Gainesville, FL 32610

Contents: Three pages total containing a figure of the rate of change of absorbance upon introduction of superoxide to Y34N and Y34F human MnSOD, and a figure of extinction coefficients at different wavelengths for Y34A human MnSOD.

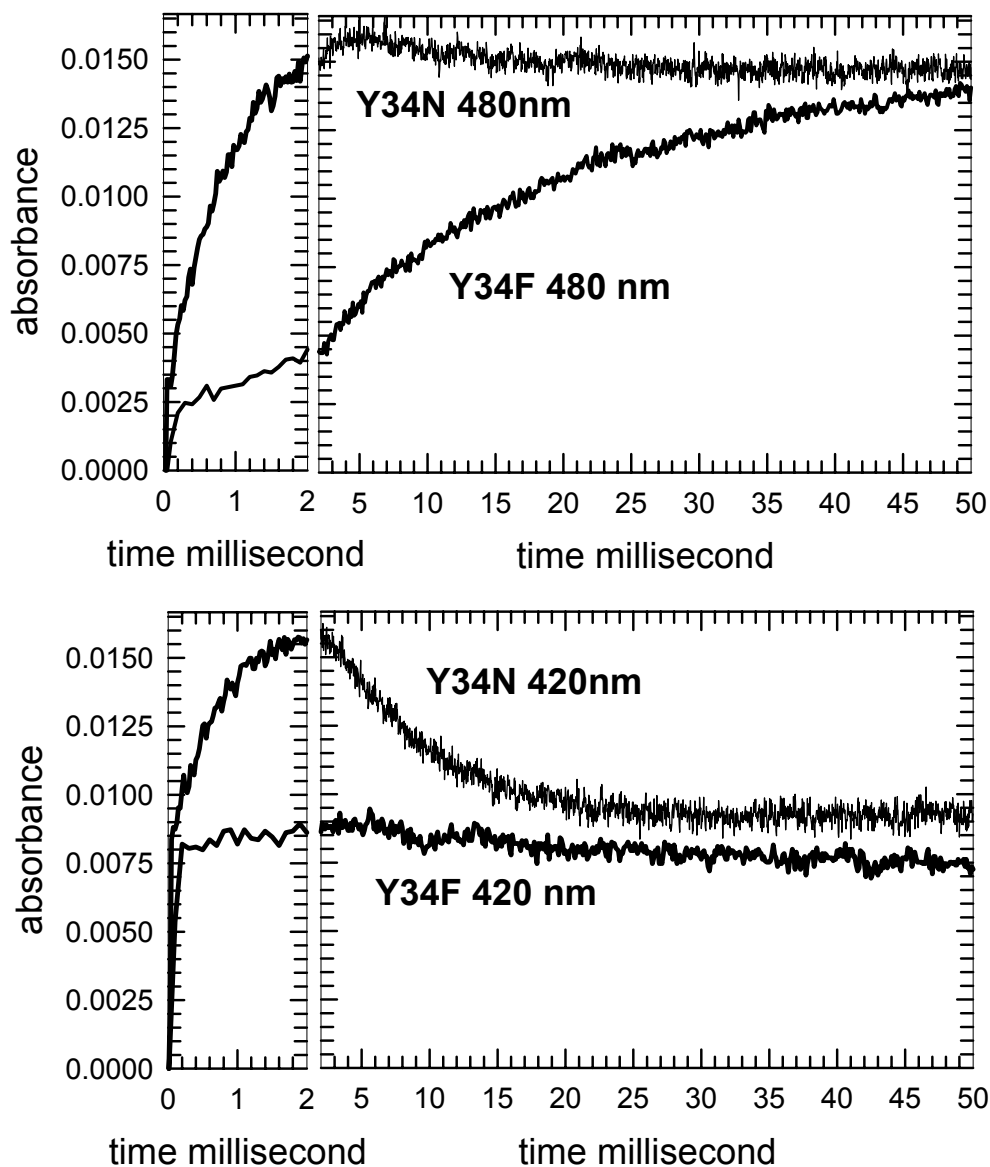


Figure S1. The increase in absorbance (*top*) at 480 nm of 182 μM Y34N MnSOD and 170 μM Y34F MnSOD and (*bottom*) at 420 nm after pulse radiolysis generated superoxide. Solutions contained 30 mM formate, 50 μM EDTA, and 2 mM Mops at pH 7.4 for Y34F and pH 7.1 for Y34N. The enzymes had been reduced prior to the experiment with H_2O_2 .

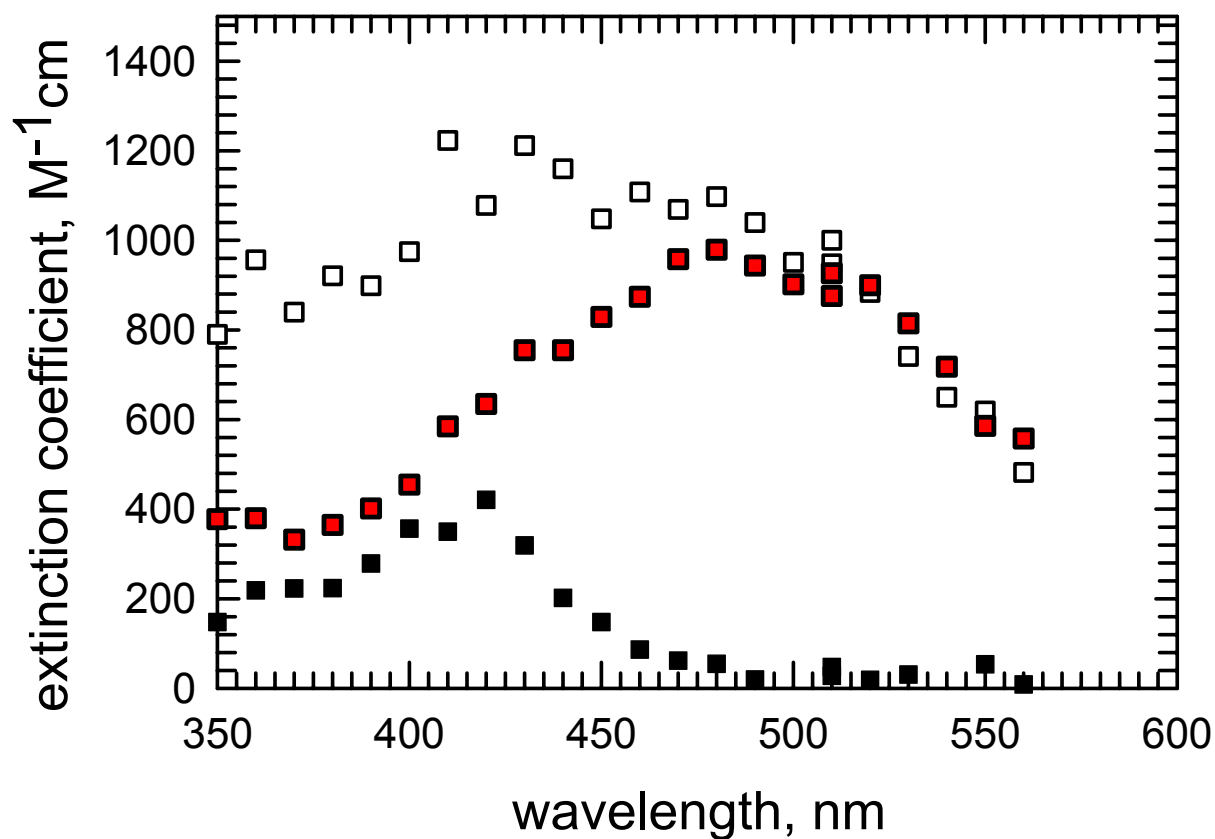


Figure S2. Molar extinction coefficient ($M^{-1}cm^{-1}$) observed after pulse radiolysis of a solution of Y34A MnSOD ($120 \mu M$) pre-reduced with H_2O_2 . (Red squares) spectrum of $Mn^{3+}(OH)SOD$ obtained by extrapolating to long times as in Figure 2A; (black squares) spectrum of the product-inhibited complex obtained by extrapolating the initial increase ($t \neq 0.2$ ms) as in Figure 2A; (open squares) spectrum of a newly-detected intermediate obtained by extrapolating the second kinetic process at times near 2 ms as in Figure 2B. Conditions were as described in Figure 2.