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**Supporting Material**

**Protein stabilization and the Hofmeister effect. The role of hydrophobic salvation**

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## SUPPLEMENTARY MATERIAL

*Table S1: Legend for the multiple K to Q replacements.*

<i>Name</i>	<i>Legend</i>
<i>Kx2Q</i>	K42Q + K54Q
<i>Kx3Q</i>	K28Q + K42Q + K54Q
<i>Kx4Q</i>	K28Q + K42Q + K54Q + K61Q
<i>Kx5Q</i>	K23Q + K28Q + K42Q + K54Q + K61Q
<i>Kx6Q</i>	K23Q + K28Q + K41Q + K42Q + K54Q + K61Q

**Table S2.** Summary of experimental restraints and statistics of the structure determination of the Kx5Q mutant. Average values over the 20 energy-refined conformers.

<i>NOE upper distance limits:</i>	1335
<i>Short-range, <math> i-j  \leq 1</math></i>	631
<i>Medium-range, <math>1 &lt;  i-j  &lt; 5</math></i>	226
<i>Long-range, <math> i-j  \geq 5</math></i>	498
<i><math>\phi/\psi</math> dihedral angle restraints from TALOS</i>	76
<i>Maximal violation (<math>\text{\AA}</math>)</i>	0.14
<i>Violations <math>&gt; 0.2 \text{\AA}</math></i>	0
<i>CYANA target function (<math>\text{\AA}^2</math>)</i>	$0.54 \pm 0.11$
<i>AMBER energy (kcal/mol)</i>	$-3051.70 \pm 55.66$
<i>Ramachandran plot statistics (%)</i>	
<i>residues in:</i>	
<i>most favoured regions</i>	88.4
<i>additionally allowed regions</i>	11.1
<i>generously allowed regions</i>	0.6
<i>disallowed regions</i>	0
<i>RMSD to mean coordinates (<math>\text{\AA}</math>)</i>	
<i>backbone/heavy atoms (from 4 to 64)</i>	0.34 / 0.70
<i>RMSD to wild type ProtL, PDB ID 1HZ6 (<math>\text{\AA}</math>)</i>	
<i>backbone (from 4 to 64)</i>	0.77
<i>backbone (secondary structure)</i>	0.58

**Table S3:** Experimental data for the single point mutants considered in the present study ( in the absence of cosolute).

<b>Mutant</b>	<b>Average <math>T_m</math> / °C</b>	<b>Number of independent measurements (CD / Fluorescence)</b>
<i>K7Q</i>	64.6 ± 2.1	3 / 3
<i>K23Q</i>	66.1 ± 0.5	3 / 3
<i>K28Q</i>	71.3 ± 0.9	3 / 3
<i>K41Q</i>	69.4 ± 0.9	3 / 3
<i>K42Q</i>	63.8 ± 1.4	3 / 3
<i>K54Q</i>	67.1 ± 0.5	3 / 3
<i>K61Q</i>	65.5 ± 1.4	3 / 3
<i>E2D</i>	70.0 ± 0.8	4 / 4
<i>E3D</i>	68.0 ± 0.4	2 / 1
<i>E21D</i>	63.9 ± 1.4	2 / 3
<i>E32D</i>	70.9 ± 0.6	4 / 4
<i>E46D</i>	70.3 ± 0.7	4 / 4
<i>K23A</i>	63.1 ± 0.6	3 / 2
<i>Kx2Q<sup>(a)</sup></i>	66.1 ± 0.5	3 / 3
<i>Kx3Q<sup>(a)</sup></i>	69.4 ± 0.8	3 / 3
<i>Kx4Q<sup>(a)</sup></i>	62.3 ± 0.6	3 / 3
<i>Kx5Q<sup>(a)</sup></i>	61.9 ± 0.9	3 / 3
<i>Kx6Q<sup>(a)</sup></i>	60.2 ± 0.4	3 / 3

(a) Legend: Kx2Q = K42Q, K54Q; Kx3Q = K28Q, K42Q, K54Q; Kx4Q = K28Q, K42Q, K54Q, K61Q; Kx5Q = K23Q, K28Q, K42Q, K54Q, K61Q; Kx6Q = K23Q, K28Q, K41Q, K42Q, K54Q, K61Q.

**Table S4: Solvent accessibilities and residues non-polar areas employed in equation 2.**

Mutant	$\chi_{np,Mut}^F$ <sup>(a)</sup>	$\chi_{np,WT}^F$ <sup>(a)</sup>							
K7Q	0.867 ± 0.06	<b>0.738 ± 0.17</b>							
K23Q	<b>0.751 ± 0.03</b>	<b>0.660 ± 0.18</b>							
K28Q	<b>0.665 ± 0.10</b>	<b>0.363 ± 0.06</b>							
K41Q	0.852 ± 0.08	<b>0.492 ± 0.20</b>							
K42Q	<b>0.249 ± 0.07</b>	<b>0.375 ± 0.09</b>							
K54Q	<b>0.313 ± 0.05</b>	<b>0.328 ± 0.07</b>							
K61Q	<b>0.709 ± 0.14</b>	<b>0.623 ± 0.14</b>							
E2D	0.448 ± 0.06	<b>0.307 ± 0.15</b>							
E3D	0.384 ± 0.04	<b>0.416 ± 0.23</b>							
E21D	0.682 ± 0.07	<b>0.698 ± 0.22</b>							
E32D	0.757 ± 0.10	<b>0.522 ± 0.27</b>							
E46D	0.539 ± 0.04	<b>0.504 ± 0.19</b>							
K23A	0.696 ± 0.05	<b>0.660 ± 0.18</b>							
$\chi_{np,K}^U$	$\chi_{np,Q}^U$	$\chi_{np,E}^U$	$\chi_{np,D}^U$	$\chi_{np,A}^U$	$A_{np}^K / \text{\AA}^2$	$A_{np}^Q / \text{\AA}^2$	$A_{np}^E / \text{\AA}^2$	$A_{np}^D / \text{\AA}^2$	$A_{np}^A / \text{\AA}^2$
0.248 <sup>(b)</sup>	0.295 <sup>(b)</sup>	0.277 <sup>(b)</sup>	0.222 <sup>(b)</sup>	0.360 <sup>(b)</sup>	122 <sup>(c)</sup>	66 <sup>(c)</sup>	69 <sup>(c)</sup>	45 <sup>(c)</sup>	86 <sup>(c)</sup>

(a) Bold values are calculated from high resolution NMR structures (2PTL{Wikstrom, 1994 #142} or 2JZP). The 12 conformations of minimal energy have been used to estimate the error. Plain values have been obtained from homology models (using the servers' phyre {Bennett-Lovsey, 2007 #139} and swiss model workspace {Arnold, 2006 #125}). The error bars reflect the discrepancies between the models.

(b) Calculated from the expression:  $\chi_{np}^U = A_{np}^U / A_{np}$ . Values for  $A_{np}$  are taken from ref. {Bernado, 2006 #140}.

(c) Obtained from {Karplus, 1997 #134}.

**Table S5:** Experimental  $T_m$  values for sodium sulfate.

<b>Sodium sulfate</b>					
<i>Mutant</i>	<i>Conc. / mM</i>	<i>T<sub>m</sub> / °C</i>	<i>Mutant</i>	<i>Conc. / mM</i>	<i>T<sub>m</sub> / °C</i>
<i>K7Q</i>	250	72.6 ± 0.6	<i>K61Q</i>	500	78.5 ± 0.6
	500	78.2 ± 0.5		750	84.0 ± 0.7
	750	83.7 ± 0.4		1000	87.7 ± 0.4
	1000	88.2 ± 0.9		<i>Kx2Q<sup>(a)</sup></i>	100
<i>K23Q</i>	250	71.6 ± 1.4	250		71.9 ± 1.0
	500	77.0 ± 1.1	330		72.3 ± 0.3
	750	82.3 ± 0.1	450		76.3 ± 0.3
<i>K28Q</i>	200	74.0 ± 0.8	<i>Kx3Q<sup>(a)</sup></i>	600	79.7 ± 0.4
	400	77.1 ± 0.7		50	70.3 ± 0.1
	800	82.5 ± 0.3		100	71.5 ± 0.1
<i>K41Q</i>	200	71.8 ± 1.3		200	74.2 ± 0.6
	400	77.6 ± 1.0		400	78.1 ± 1.4
	600	81.1 ± 1.0		750	82.9 ± 1.8
<i>K42Q</i>	200	65.9 ± 1.5	<i>Kx4Q<sup>(a)</sup></i>	250	70.1 ± 0.2
	250	67.4 ± 0.4		400	73.1 ± 0.2
	400	70.1 ± 1.4		750	79.7 ± 1.3
	500	73.2 ± 1.1		850	82.0 ± 0.3
	600	75.4 ± 1.3		<i>Kx5Q<sup>(a)</sup></i>	250
750	76.5 ± 0.5	500	76.7 ± 0.2		
<i>K54Q</i>	200	73.2 ± 0.7	750		82.2 ± 1.6
	400	76.3 ± 0.6	<i>Kx6Q<sup>(a)</sup></i>	250	67.2 ± 0.2
	600	79.4 ± 0.9		450	72.8 ± 0.3
800	84.5 ± 0.7	660		79.9 ± 0.3	
<i>K61Q</i>	250	72.0 ± 0.5			

(a) Legend: Kx2Q = K42Q, K54Q; Kx3Q = K28Q, K42Q, K54Q; Kx4Q = K28Q, K42Q, K54Q, K61Q; Kx5Q = K23Q, K28Q, K42Q, K54Q, K61Q; Kx6Q = K23Q, K28Q, K41Q, K42Q, K54Q, K61Q.

**Table S6:** Experimental  $T_m$  values for sodium phosphate.

<b>Sodium phosphate</b>					
<i>Mutant</i>	<i>Conc. / mM</i>	<i>T<sub>m</sub> / °C</i>	<i>Mutant</i>	<i>Conc. / mM</i>	<i>T<sub>m</sub> / °C</i>
<i>K7Q</i>	200	67.4 ± 0.3	<i>K61Q</i>	400	71.2 ± 0.9
	400	70.9 ± 0.5		600	75.2 ± 0.8
	600	74.3 ± 0.3		800	77.0 ± 1.0
	800	78.4 ± 0.6		<i>Kx2Q<sup>(a)</sup></i>	200
<i>K23Q</i>	200	69.1 ± 0.3	400		71.2 ± 1.6
	400	71.7 ± 0.2	600		75.9 ± 0.4
	600	75.7 ± 0.3	800		77.0 ± 1.0
	800	77.2 ± 0.4	<i>Kx3Q<sup>(a)</sup></i>	200	73.3 ± 0.8
<i>K28Q</i>	200	72.9 ± 0.8		400	75.7 ± 0.9
	400	75.0 ± 0.6		600	79.3 ± 0.9
	800	78.6 ± 0.4		800	81.7 ± 0.7
<i>K41Q</i>	200	70.6 ± 0.8	<i>Kx4Q<sup>(a)</sup></i>	200	66.7 ± 0.2
	400	75.8 ± 0.5		400	68.8 ± 0.1
	600	76.6 ± 1.5		600	72.0 ± 0.1
	800	75.4 ± 3.9		800	74.9 ± 0.5
<i>K42Q</i>	200	64.7 ± 0.9	<i>Kx5Q<sup>(a)</sup></i>	900	77.2 ± 0.4
	400	69.7 ± 1.6		200	66.9 ± 0.3
	600	72.2 ± 1.1		400	70.7 ± 0.5
	800	75.2 ± 0.7		600	76.1 ± 0.8
<i>K54Q</i>	200	70.8 ± 0.7	<i>Kx6Q<sup>(a)</sup></i>	800	77.9 ± 0.1
	400	72.7 ± 0.6		200	64.5 ± 0.4
	600	75.5 ± 0.7		400	68.9 ± 1.3
	800	77.4 ± 0.9		600	74.5 ± 0.5
<i>K61Q</i>	200	69.1 ± 0.7	800	80.5 ± 0.5	

(a) Legend: Kx2Q = K42Q, K54Q; Kx3Q = K28Q, K42Q, K54Q; Kx4Q = K28Q, K42Q, K54Q, K61Q; Kx5Q = K23Q, K28Q, K42Q, K54Q, K61Q; Kx6Q = K23Q, K28Q, K41Q, K42Q, K54Q, K61Q.

**Table S7: Experimental  $T_m$  values for sodium fluoride.**

<b>Sodium fluoride</b>					
<i>Mutant</i>	<i>Conc. / mM</i>	<i>T<sub>m</sub> / °C</i>	<i>Mutant</i>	<i>Conc. / mM</i>	<i>T<sub>m</sub> / °C</i>
<i>K7Q</i>	200	68.0 ± 0.8	<i>K61Q</i>	200	68.5 ± 1.1
	400	71.1 ± 0.6		400	70.5 ± 1.0
	600	73.4 ± 1.5		600	74.3 ± 0.8
	800	75.6 ± 1.4		800	76.0 ± 0.8
<i>K23Q</i>	200	68.7 ± 1.1	<i>Kx2Q<sup>(a)</sup></i>	200	69.3 ± 0.9
	400	72.1 ± 0.3		300	70.1 ± 0.4
	600	74.1 ± 0.3		400	71.3 ± 0.1
	800	76.1 ± 0.2		600	74.8 ± 0.4
<i>K28Q</i>	200	73.8 ± 1.7	<i>Kx3Q<sup>(a)</sup></i>	200	72.1 ± 0.1
	400	76.1 ± 1.3		400	73.8 ± 0.6
	600	78.5 ± 1.6		600	77.1 ± 1.7
<i>K41Q</i>	200	72.2 ± 0.4	<i>Kx4Q<sup>(a)</sup></i>	760	79.5 ± 3.3
	400	74.7 ± 1.3		200	66.0 ± 0.7
	600	79.2 ± 1.2		400	69.1 ± 0.5
	760	79.7 ± 0.9		600	71.6 ± 1.2
<i>K42Q</i>	200	64.2 ± 2.7	<i>Kx5Q<sup>(a)</sup></i>	760	74.9 ± 0.2
	400	67.6 ± 0.5		200	66.9 ± 0.6
	600	69.2 ± 1.3		400	71.0 ± 0.4
	760	70.5 ± 0.6		640	74.3 ± 0.6
<i>K54Q</i>	200	71.1 ± 1.0	<i>Kx6Q<sup>(a)</sup></i>	100	62.6 ± 2.1
	400	72.1 ± 1.0		250	65.0 ± 0.2
	600	75.3 ± 0.7		300	65.6 ± 0.7
	800	77.8 ± 1.3		500	70.4 ± 0.5

(a) Legend: Kx2Q = K42Q, K54Q; Kx3Q = K28Q, K42Q, K54Q; Kx4Q = K28Q, K42Q, K54Q, K61Q; Kx5Q = K23Q, K28Q, K42Q, K54Q, K61Q; Kx6Q = K23Q, K28Q, K41Q, K42Q, K54Q, K61Q.



**Table S8: Experimental  $T_m$  values for sodium nitrate.**

<b>Sodium nitrate</b>					
<i>Mutant</i>	<i>Conc. / mM</i>	<i>T<sub>m</sub> / °C</i>	<i>Mutant</i>	<i>Conc. / mM</i>	<i>T<sub>m</sub> / °C</i>
<i>K7Q</i>	200	63.6 ± 0.1	<i>K54Q</i>	750	63.5 ± 1.3
	400	62.3 ± 0.4		1000	62.4 ± 1.3
	600	61.8 ± 0.3	<i>K61Q</i>	250	65.2 ± 1.1
	800	60.0 ± 0.3		500	63.5 ± 1.0
<i>K23Q</i>	200	64.2 ± 0.5	<i>Kx2Q<sup>(a)</sup></i>	1000	60.3 ± 0.8
	400	63.5 ± 0.7		500	63.3 ± 0.2
	600	62.3 ± 0.5	750	61.8 ± 0.5	
	800	60.6 ± 0.3	1000	61.4 ± 0.3	
<i>K28Q</i>	200	69.5 ± 0.8	<i>Kx3Q<sup>(a)</sup></i>	250	67.8 ± 0.01
	400	68.7 ± 1.2		500	65.6 ± 0.4
	600	66.8 ± 0.3	750	63.2 ± 0.3	
	800	65.6 ± 0.3	<i>Kx4Q<sup>(a)</sup></i>	100	61.5 ± 0.1
<i>K41Q</i>	250	67.3 ± 2.7		250	61.1 ± 0.4
	500	65.5 ± 0.5		350	60.9 ± 0.01
	1000	63.1 ± 0.6	500	59.4 ± 0.9	
<i>K42Q</i>	500	61.6 ± 1.3	<i>Kx5Q<sup>(a)</sup></i>	250	60.2 ± 0.4
	1000	59.1 ± 0.9		500	58.1 ± 0.5
<i>K54Q</i>	250	66.2 ± 1.0		750	57.3 ± 0.2
	500	65.8 ± 0.7	1000	55.3 ± 0.7	

(a) Legend: Kx2Q = K42Q, K54Q; Kx3Q = K28Q, K42Q, K54Q; Kx4Q = K28Q, K42Q, K54Q, K61Q; Kx5Q = K23Q, K28Q, K42Q, K54Q, K61Q.

**Table S9: Experimental  $T_m$  values for sodium perchlorate.**

<b>Sodium perchlorate</b>					
<i>Mutant</i>	<i>Conc. / mM</i>	<i>T<sub>m</sub> / °C</i>	<i>Mutant</i>	<i>Conc. / mM</i>	<i>T<sub>m</sub> / °C</i>
<i>K7Q</i>	200	63.8 ± 0.8	<i>K61Q</i>	800	56.8 ± 1.0
	400	61.8 ± 1.0		200	63.4 ± 0.8
	600	58.8 ± 0.3		400	61.6 ± 0.8
	800	58.0 ± 0.9		600	58.7 ± 0.3
<i>K23Q</i>	200	63.4 ± 0.5	<i>Kx2Q<sup>(a)</sup></i>	800	57.7 ± 0.5
	400	61.7 ± 0.4		250	61.9 ± 0.04
	600	60.2 ± 0.5		500	57.8 ± 0.2
	800	58.2 ± 0.4		1000	52.8 ± 1.5
<i>K28Q</i>	200	69.2 ± 0.3	<i>Kx3Q<sup>(a)</sup></i>	100	67.7 ± 0.2
	400	66.7 ± 0.5		250	65.9 ± 0.2
	600	64.6 ± 0.4		750	60.8 ± 0.2
	800	61.8 ± 0.8		200	58.9 ± 0.9
<i>K41Q</i>	200	65.1 ± 0.6	<i>Kx4Q<sup>(a)</sup></i>	600	56.5 ± 0.3
	400	59.6 ± 1.6		700	53.0 ± 0.9
	800	58.6 ± 1.5		250	59.0 ± 0.3
<i>K42Q</i>	200	59.9 ± 0.7	<i>Kx5Q<sup>(a)</sup></i>	500	57.4 ± 0.7
	400	55.2 ± 0.6		750	54.5 ± 0.4
	600	55.1 ± 1.0		1000	53.3 ± 0.9
	800	53.9 ± 0.8		500	56.0 ± 0.7
<i>K54Q</i>	200	62.6 ± 0.8	<i>Kx6Q<sup>(a)</sup></i>	750	52.0 ± 0.5
	400	61.0 ± 0.2		1000	49.5 ± 0.6
	600	57.6 ± 0.7			

(a) Legend: Kx2Q = K42Q, K54Q; Kx3Q = K28Q, K42Q, K54Q; Kx4Q = K28Q, K42Q, K54Q, K61Q; Kx5Q = K23Q, K28Q, K42Q, K54Q, K61Q; Kx6Q = K23Q, K28Q, K41Q, K42Q, K54Q, K61Q.

**Table S10: Experimental  $T_m$  values for sodium thiocyanate.**

<b>Sodium thiocyanate</b>						
<b>Mutant</b>	<b>Conc. / mM</b>	<b><math>T_m</math> / °C</b>	<b>Mutant</b>	<b>Conc. / mM</b>	<b><math>T_m</math> / °C</b>	
<i>K7Q</i>	200	63.0 ± 0.3	<i>K61Q</i>	600	53.1 ± 0.5	
	400	57.7 ± 1.3		800	49.5 ± 0.8	
	600	53.8 ± 0.8	<i>Kx2Q<sup>(a)</sup></i>	250	58.9 ± 0.3	
	800	49.9 ± 0.5		500	54.4 ± 0.8	
<i>K23Q</i>	200	61.3 ± 0.7	750	50.7 ± 0.3		
	400	57.9 ± 0.3	1000	47.1 ± 0.4		
	600	52.2 ± 0.8	<i>Kx3Q<sup>(a)</sup></i>	100	67.6 ± 0.2	
<i>K28Q</i>	200	65.9 ± 0.5		250	64.9 ± 0.9	
	400	60.6 ± 1.3		300	64.7 ± 0.5	
	800	54.2 ± 0.6		375	63.4 ± 0.3	
<i>K41Q</i>	200	64.4 ± 1.2	<i>Kx4Q<sup>(a)</sup></i>	500	60.7 ± 0.5	
	400	62.1 ± 0.8		100	60.1 ± 0.5	
	600	58.7 ± 1.3		300	56.3 ± 0.3	
	800	56.2 ± 0.6		400	54.0 ± 0.9	
<i>K42Q</i>	200	59.5 ± 0.8	<i>Kx5Q<sup>(a)</sup></i>	100	60.9 ± 0.4	
	400	56.4 ± 0.6		200	58.0 ± 1.3	
	600	53.3 ± 0.2		450	55.4 ± 0.8	
	800	51.0 ± 0.7		600	52.2 ± 1.3	
<i>K54Q</i>	250	59.4 ± 0.8		<i>Kx6Q<sup>(a)</sup></i>	750	48.0 ± 0.6
	500	55.0 ± 0.1			200	55.7 ± 0.5
	750	50.7 ± 0.5	400		53.0 ± 0.8	
	1000	46.5 ± 0.7	500		50.9 ± 0.3	
<i>K61Q</i>	200	62.4 ± 0.8	650	47.0 ± 0.8		
	400	57.3 ± 0.5	750	43.8 ± 0.4		

(a) Legend: Kx2Q = K42Q, K54Q; Kx3Q = K28Q, K42Q, K54Q; Kx4Q = K28Q, K42Q, K54Q, K61Q; Kx5Q = K23Q, K28Q, K42Q, K54Q, K61Q; Kx6Q = K23Q, K28Q, K41Q, K42Q, K54Q, K61Q.

**Table S11:** Comparison of the  $T_m$  variation produced by the cosolute ( $\partial T_m / \partial C_3$ ) and the salting in coefficient for K7Q.

Anion	$\partial T_m / \partial C_3$ $^{\circ}\text{C}\cdot\text{M}^{-1}$	Salting-in constant(a) $\text{M}^{-1}$
<i>Sulfate</i>	2.33	0.013
<i>Phosphate</i>	1.76	n.a.
<i>Fluoride</i>	1.43	0.027 (KF)
<i>Nitrate</i>	-5.5	n.a.
<i>Perchlorate</i>	-9.1	0.097
<i>Thiocyanate</i>	-19.3	0.077

(a) Data taken from ref. 4.